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1           **The role of the hole-extraction layer in determining the operational stability of a**  
2           **polycarbazole:fullerene bulk-heterojunction photovoltaic device**

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8  
9           **Abstract**

10          We have made a comparative study of the relative operational stability of bulk-heterojunction  
11          organic photovoltaic (OPV) devices utilising different hole transport layers (HTLs). OPV devices  
12          were fabricated based on a blend of the polymer PCDTBT with the fullerene PC<sub>70</sub>BM, and  
13          incorporated the different HTL materials PEDOT:PSS, MoO<sub>x</sub> and V<sub>2</sub>O<sub>5</sub>. Following 620 hours of  
14          irradiation by light from a solar simulator, we find that devices using the PEDOT:PSS HTL retained  
15          the highest efficiency, having a projected T<sub>80</sub> lifetime of 14,500 hours.

16

1 The commercialisation of polymer:fullerene bulk-heterojunction photovoltaics will require  
2 the demonstration of high device efficiency together with extended device lifetime. Indeed, it has  
3 been estimated <sup>1</sup>, that it will be necessary for organic photovoltaic devices (OPVs) to demonstrate  
4 power conversion efficiencies (PCE) and operational lifetimes of 10% and 10 years respectively  
5 before commercialisation is possible. Following recent advances in device efficiency, attention has  
6 increasingly focused on OPV lifetime and the mechanisms that underpin device degradation <sup>2</sup>.  
7 Much work on degradation processes have studied OPV devices based on the material poly(3-  
8 hexylthiophene) (P3HT) <sup>3-6</sup>, with lifetimes of up to 3 years reported <sup>6</sup>. More recently, outdoor inter-  
9 laboratory studies on P3HT based devices reveal extrapolated lifetimes in excess of 5 years <sup>7</sup>, with  
10 other laboratory studies on poly[9-(heptadecan-9-yl)-9H-carbazole-2,7-diyl-alt-(4',7'-di-2-thienyl-  
11 2',1',3'-benzothiadiazole)-5,5-diyl] [PCDTBT] based OPVs demonstrating extrapolated device  
12 lifetimes of up to 10 years <sup>6,8</sup>.

13 To improve OPV stability and lifetime, several studies now report the effect of changing  
14 other interlayers within P3HT based devices <sup>9-12</sup>. Notably, PCDTBT <sup>6,11,12,18</sup> and other  
15 polycarbazole polymers can be used to create devices having extended operational lifetimes <sup>6</sup>, with  
16 the comparative stability afforded by PCDTBT presenting an opportunity to determine the relative  
17 role of the hole transport layer (HTL) in contributing to device stability. We have therefore  
18 fabricated PCDTBT:PC<sub>70</sub>BM OPV devices that utilise a range of HTLs including poly(3,4-  
19 ethylenedioxythiophene):poly(styrene sulfonic acid) (PEDOT:PSS), molybdenum oxide (MoO<sub>x</sub>)  
20 and vanadium oxide (V<sub>2</sub>O<sub>5</sub>) and have measured device PCE under simulated solar radiation over a  
21 period of 620 hours. We find that in contrast to previous work <sup>9,10,12</sup>, devices utilising a  
22 PEDOT:PSS HTL have higher stability than those incorporating MoO<sub>x</sub> or V<sub>2</sub>O<sub>5</sub> HTLs, having an  
23 extrapolated T<sub>80</sub> lifetime of 14,500 hours.

24 The devices explored are based on the structure ITO/HTL/PCDTBT:PC<sub>70</sub>BM/Ca/Al in  
25 which the hole transport layer was composed of MoO<sub>x</sub>, PEDOT:PSS or V<sub>2</sub>O<sub>5</sub>. PEDOT:PSS is a  
26 material that has been widely explored as a hole-transporting layer due to its large work-function,  
27 high conductivity and good film transparency <sup>19</sup>. Thin films of PEDOT:PSS are however extremely  
28 hygroscopic <sup>20</sup>, and post deposition thermal treatments are needed to remove water remaining in the  
29 films.

30 MoO<sub>x</sub>, along with other transition metals, has been used extensively with polymers having  
31 low lying HOMO levels to improve charge extraction due to its large work function (in the region  
32 of 5.5 eV <sup>21</sup>). MoO<sub>x</sub> can be deposited by a variety of processes, including evaporation <sup>22</sup>, sputtering  
33 <sup>23</sup> and solution-based deposition <sup>24-26</sup>. When combined with polymers having a deep HOMO level,

1 such as PCDTBT (-5.5eV<sup>27</sup>), device efficiencies of over 7% have been achieved<sup>12</sup>. In this study we  
2 have used thermally-evaporated MoO<sub>x</sub>; a form of this metal oxide most often utilised in OPVs.

3 V<sub>2</sub>O<sub>5</sub> is a transition metal oxide that has been extensively used as a HTL in OPVs. It can be  
4 deposited by a variety of processes and has a work function of 5.6 eV<sup>28</sup> making it well matched for  
5 hole extraction from PCDTBT. To deposit V<sub>2</sub>O<sub>5</sub>, we have used a facile synthetic process in air to  
6 form uniform thin layers of V<sub>2</sub>O<sub>5</sub> from an oxytriisopropoxide precursor solution<sup>29-31</sup>, forming an  
7 amorphous, smooth V<sub>2</sub>O<sub>5</sub> layer.

8 OPV devices were fabricated on ITO substrates provided by Ossila Ltd. Devices utilising a  
9 PEDOT:PSS HTL (Heraeus Clevios™ P VP AI 4083) layer were spin-cast at to form a ~ 30 nm  
10 thick film which were annealed for 10 minutes at 150°C, following which they were transferred to a  
11 nitrogen filled glovebox. The MoO<sub>x</sub> HTLs were deposited by thermal evaporation, using  
12 molybdenum (VI) oxide pellets placed in an aluminium oxide crucible. Vanadium  
13 oxytriisopropoxide was dissolved in 2-propanol at a concentration of 5 μL/mL and spin cast in air  
14 to form a ~ 5 nm thick thin film. The film was left in air for 45 minutes to fully hydrolyse, after  
15 which it was transferred to a glovebox for further processing.

16 The active layer of all devices was deposited from a solution of PCDTBT (synthesised using  
17 previously described techniques<sup>32</sup>) blended with PC<sub>70</sub>BM at a total concentration of 25 mg mL<sup>-1</sup> at  
18 1:4 blend ratio in chlorobenzene. This active layer solution was spin-cast onto each HTL layer in  
19 the glovebox to form a 70 nm thick film. A cathode consisting of a calcium / aluminium film (5 /  
20 100 nm) was then thermally evaporated onto the active layer creating 6 pixels on each substrate,  
21 each having an active area of 0.04 cm<sup>2</sup>. Finally, the devices were encapsulated using a glass cover  
22 slip and a UV-curable epoxy in the glovebox.

23 The completed devices were measured using a Newport 92251A-1000 AM 1.5 solar  
24 simulator calibrated against an NREL certified silicon reference cell. An aperture mask having an  
25 area of 0.0261 cm<sup>2</sup> was used to define the area irradiated on each pixel. The values of PCE, J<sub>sc</sub>, FF  
26 and V<sub>oc</sub> we quote represent an average of parameters recorded from 12 pixels defined on two  
27 separate substrates in which the worst 25% of pixels have been omitted. The errors quoted are  
28 defined by the standard deviation about the mean. External quantum efficiency (EQE)  
29 measurements were made using a halogen lamp as a light source in conjunction with a  
30 monochromator. A Newport 818-UV calibrated silicon photodiode was used to measure a reference  
31 spectrum of the light source.

32 Lifetime testing was carried out in an ATLAS Suntest CPS+ solar simulator with a 1500 W  
33 xenon lamp in which a quartz filter was used to reduce the IR portion of the emitted light and more  
34 closely replicate the solar spectrum. Note that devices placed in the ATLAS solar simulator were

1 not covered by an aperture mask during irradiation or during JV testing. Internal reflectors in the  
2 ATLAS test chamber were used to produce a uniform light irradiation of  $1000 \text{ Wm}^{-2}$  over all the  
3 devices, with differences in the light flux impinging on the different OPV devices being less than  
4 3%. The temperature inside the ATLAS testing chamber was held at  $(40 \pm 2)^\circ\text{C}$ . A series of  
5 temperature sensors placed over the device test area indicated that variations in the temperature  
6 between different devices was  $< 1^\circ\text{C}$ . The typical humidity in the lifetime testing system was  $(25 \pm$   
7  $5)\%$ . A photograph of the OPV test-board, temperature sensors and reference photodiodes is shown  
8 in **Figure 1 (a)**.

9 We have characterised the spectral distribution of the radiation inside the ATLAS solar  
10 simulator as shown in **Figure 1 (b)**, and find a reasonably close correspondence between the  
11 ATLAS and the AM1.5 solar spectrum over the wavelength range 350 - 600 nm, corresponding to  
12 the absorption maxima of the polymer:fullerene active layer. However, small differences in the  
13 spectra resulted in a difference in OPV  $J_{\text{sc}}$  measured using the different systems of  $\leq 10\%$ . To  
14 eliminate errors resulting from the spectral mismatch, all tabulated device metrics quoted (PCE, FF,  
15  $V_{\text{oc}}$ ,  $J_{\text{sc}}$  and  $T_{80}$ ) were determined from separate JV scans measured using an aperture mask and the  
16 calibrated Newport solar simulator.

17 In our measurements, we made a side-by-side comparison between six devices based on  
18 PEDOT:PSS,  $\text{MoO}_x$  and  $\text{V}_2\text{O}_5$  HTLs (two substrates per HTL comprising 12 pixels). During  
19 lifetime testing, a JV sweep (-1.0 V to 1.0 V) was recorded on each pixel every 10 minutes, with the  
20 devices held at open circuit between measurements. The devices were tested for a total of 620 hours  
21 under constant illumination, during which time they were removed every  $\sim 3$  days to measure EQE  
22 and record JV scans under the Newport solar simulator.

23 In **Figure 2**, we plot the average values PCE, FF,  $V_{\text{oc}}$  and  $J_{\text{sc}}$ , determined from the different  
24 devices as a function of irradiation time. Here, data recorded using the ATLAS xenon lamp is  
25 plotted using blue, red and green lines for  $\text{MoO}_x$ , PEDOT:PSS and  $\text{V}_2\text{O}_5$  HTL-based devices  
26 respectively. In each figure, we also plot data recorded using the Newport solar simulator with solid  
27 symbols using the same colour scheme. In all cases, PCE is normalised to its initial value. Notably,  
28 there is a discrepancy between data recorded between ATLAS and Newport systems that derives  
29 both from the spectral mismatch between the systems and the fact that an aperture mask was not  
30 used to define that active area during measurements in the ATLAS system. For this reason, data  
31 recorded using the ATLAS system is only included to highlight general trends in device metrics as  
32 they undergo aging.

33 It can be seen that all devices undergo a burn in phase that lasts for  $\sim 250$  hours in which the  
34 FF and  $J_{\text{sc}}$  decrease at an exponential rate, after which the decay follows a linear trend. We make a

1 linear-fit to the PCE data post burn-in plotted using a dashed line that allows us to determine the  $T_{80}$   
2 lifetime (defined as the point at which the efficiency of the device falls below 80% of its efficiency  
3 at the end of the burn-in period).

4 In **Table 1** we display the average values of PCE, peak EQE, FF,  $V_{oc}$  and  $J_{sc}$  (and their  
5 uncertainties) for all devices before and after lifetime testing (determined using the Newport solar  
6 simulator), with corresponding JV and EQE curves shown in **Figure 3**. We also report device PCE  
7 after the burn-in period and the relative loss in the PCE after 250 and 620 hours and  $T_{80}$  in **Table 2**.  
8 The initial burn-in (see **Figure 2**) is believed to originate from photochemical reactions in the  
9 PCDTBT:PC<sub>70</sub>BM blend that adversely affect its charge transport properties<sup>6,11,18,33</sup>. The EQE  
10 spectra recorded before and after irradiation (see **Figure 3 (b) and (d)** respectively), indicate a  
11 general reduction in quantum efficiency commensurate with the observed reduction in  $J_{sc}$  without  
12 any change in spectral shape; a result consistent with some photooxidation. Here, photochemical  
13 reactions in PCDTBT result in an increase in the density of sub-bandgap states that lead to a  
14 reduction in hole mobility<sup>11</sup>. The formation of trap states saturates over time, which results in an  
15 end to the burn-in phase and subsequent linear decay<sup>11</sup>.

16 It can be seen in **Figure 2** that that devices utilising a PEDOT:PSS HTL underwent a rapid,  
17 initial burn in phase that was dominated by a reduction in FF and  $J_{sc}$  which then stabilised after 60  
18 hours. Significantly, the  $V_{oc}$  of such devices also underwent an initial and rapid reduction (over a  
19 period 5 hours) but then recovered and stabilised over a period of 200 hours to a value that was  
20 some 15 mV higher than the initial  $V_{oc}$ . This, we speculate, results from charge transfer from the  
21 PEDOT:PSS to the active layer<sup>34,35</sup>; a process that alters the work function of the PEDOT:PSS  
22 surface and thus facilitates charge transfer. Note that the PEDOT:PSS used here is known to be  
23 acidic (pH ~ 2<sup>36,37</sup>), and has been shown to react with the ITO anode<sup>38-40</sup>, resulting in degradation  
24 of device efficiency. It is possible that the use of different grades of PEDOT:PSS having a more  
25 neutral pH may well reduce the drop in efficiency during the initial burn-in period.

26 We find that the PCE, FF,  $J_{sc}$ , and  $V_{oc}$  of devices based on MoO<sub>x</sub> or V<sub>2</sub>O<sub>5</sub> HTLs degraded at  
27 a significantly faster rate than that of the PEDEOT:PSS based devices, both during the burn-in  
28 period and in the subsequent linear decay. In MoO<sub>x</sub> HTL devices, the overall reduction in PCE is  
29 dominated by a reduction in FF and  $J_{sc}$ , with the  $V_{oc}$  only undergoing a relatively small reduction  
30 (by 4%) over 620 hours. At present, the precise mechanisms by which the MoO<sub>x</sub> HTLs undergo  
31 degradation is unclear, however we speculate that such nanocrystalline metal-oxide films undergo  
32 thermally-assisted break-down followed by diffusion into the active semiconductor layer; a process  
33 that could result in both charge trapping and reduced hole-extraction efficiency.

1 In  $V_2O_5$  HTL devices, the  $V_{oc}$  reduced by 40% over the 620 hours of testing. This reduction  
2 was also accompanied by significant reductions in PCE, FF and  $J_{sc}$ , as can be seen in **Table 1**,  
3 resulting in the largest loss in device PCE. We note that the vanadium oxytriisopropoxide precursor  
4 is a Lewis acid, and has been shown to damage conjugated polymers, resulting in the formation of  
5 main-chain defects and trap states and a reduction in device PCE<sup>41</sup>. We speculate that residues of  
6 the precursor material in the HTL film, along with other organic by-products of the hydrolysis  
7 process, may be the cause of such device degradation. This suggests that the use of  $V_2O_5$  HTLs  
8 prepared by this synthetic process would be problematic for practical applications. The use of other  
9 techniques to deposit  $V_2O_5$  layers<sup>41-44</sup> may address this issue.

10 In **Table 2**, we display the values for  $T_{80}$  that we have extracted from our measurements  
11 from the Suntest and Newport solar simulators. The discrepancy between the extrapolated lifetimes  
12 results from differences in the PCE as a result of the spectral mismatch and the use of an aperture  
13 mask. Using the Newport solar simulator we determine an average lifetime for devices using a  
14 PEDOT:PSS HTL of 14,500 hours. If we neglect other possible catastrophic device degradation  
15 processes (e.g. failure of the device encapsulation), our extrapolated lifetime is equivalent to 7.2  
16 years of practical operation, assuming an operational device would receive 5.5 hours of sun per day  
17 at one sun intensity<sup>6</sup>. For  $MoO_x$ , however, the device  $T_{80}$  lifetime was found to be 1000 hours  
18 (equivalent to 6 months operation); a value reduced to 350 hours (63 days) for the  $V_2O_5$  HTL  
19 devices. We emphasise that lifetime figures we determine here are for devices aged indoors under  
20 the ISOS-L-1 specification<sup>45</sup>, as outdoor testing results in thermal cycling and light intensity  
21 variations.

22 Our experiments demonstrate therefore that the rate of degradation of the device metrics is a  
23 significant function of the nature of the HTL material; a result that confirms that degradation in the  
24 active layer does not completely account for the observed reduction in device performance. Rather,  
25 the changes in FF,  $J_{sc}$ , and  $V_{oc}$  that we observe also result from the formation of additional trap  
26 states at the interface between the active layer and the hole transport layers, due to generation of  
27 structural and electronic defects<sup>18</sup>.

28 We have shown therefore that OPV devices utilising a PEDOT:PSS HTL have a higher  
29 stability than comparable devices using a  $MoO_x$  or  $V_2O_5$  HTLs, with extrapolated  $T_{80}$  lifetimes for  
30 devices utilising a PEDOT:PSS HTL being 14,500 hours. Such lifetimes are in good accord with  
31 the results of previous studies<sup>6</sup>. Other work<sup>9,10,12</sup> however has suggested that OPVs having a  
32 PEDOT:PSS HTLs have relatively poor operational stability. We believe this apparent contradiction  
33 results from the differing sensitivity of the donor-polymer to trapped moisture. We note that  
34 previous studies explored OPVs based on the polymer P3HT; a material known to be sensitive to

1 the presence of moisture that is likely introduced into the device by the highly hygroscopic  
2 PEDOT:PSS<sup>46</sup>. In contrast, however, PCDTBT is more stable to the presence of water and oxygen  
3 even under elevated temperatures<sup>11</sup>, and thus the effect of trapped moisture within the PEDOT:PSS  
4 appears less problematic.

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		PCE (%)	Peak EQE (%)	FF	V <sub>oc</sub> (V)	J <sub>sc</sub> (mAcm <sup>-2</sup> )
<b>Molybdenum Oxide (MoO<sub>x</sub>)</b>	Initial	5.19±0.07	54.7±0.1	64.6±0.7	0.91±0.01	-8.9±0.1
	Final	2.7±0.1	47.4±0.2	43.9±0.9	0.87±0.01	-7.0±0.2
<b>PEDOT:PSS</b>	Initial	5.6±0.2	63.7±0.2	65.9±0.9	0.85±0.01	-9.9±0.2
	Final	3.4±0.2	56.6±0.3	46±1	0.89±0.001	-8.3±0.4
<b>Vanadium</b>	Initial	5.2±0.2	58.4±0.1	65±1	0.85±0.01	-9.4±0.1

Oxide ( $V_2O_5$ )	Final	1.18±0.04	43.7±0.1	36.8±0.8	0.51±0.2	-6.3±0.1
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1

2 Table 1. Average values for PCE, FF,  $V_{oc}$  and  $J_{sc}$  were calculated from 12 pixels across two  
3 substrates, as measured using the Newport solar simulator, where the worst 25% of pixels were  
4 discarded due to film defects. The error quoted on all measurements is based on the standard  
5 deviation around the mean.

6

	PCE after burn- in (250hrs)	PCE Loss over burn-in (250hrs)	PCE Loss over 620 hours	$T_{80}$ (hours) Newport	$T_{80}$ (hours) Suntest
<b>Molybdenum Oxide (<math>MoO_x</math>)</b>	2.9±0.1%	44±3%	48±3%	1000	650
<b>PEDOT:PSS Annealed</b>	3.4±0.4%	39±6%	39±5%	14500	20000
<b>Vanadium Oxide (<math>V_2O_5</math>)</b>	1.5±0.1%	72±4%	77±4%	350	236

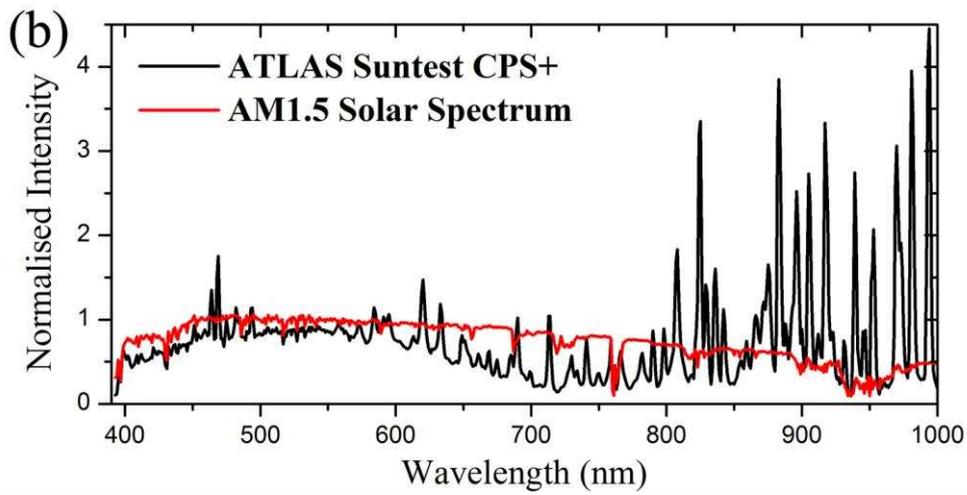
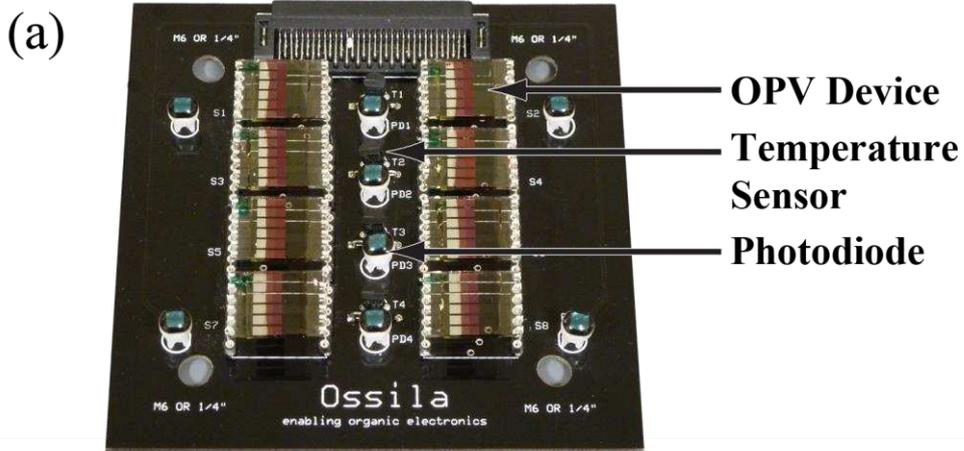
7

8 Table 2. PCE loss on burn in and over the full 620 hrs of testing with calculated  $T_{80}$  lifetimes  
9 determined using the Newport solar simulator data and the ATLAS Suntest CPS+ data. All PCE  
10 values and losses were calculated using data from the Newport solar simulator.

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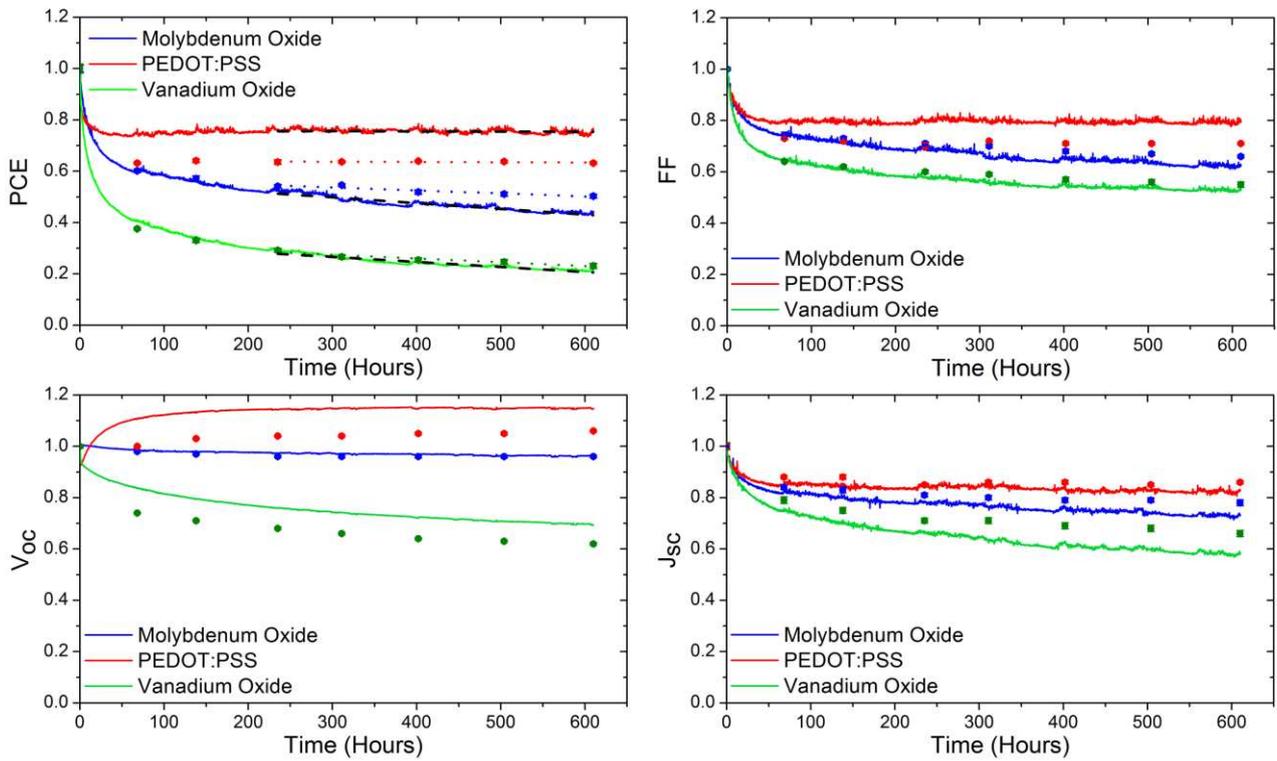
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2 Figure 1. (a) Lifetime test board. (b) Spectrum of the ATLAS Suntest CPS+ and AM1.5 solar  
3 spectrum.

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2 Figure 2 shows PCE,  $V_{oc}$ ,  $J_{sc}$  and FF for devices utilising the different HTL materials as a function  
 3 of irradiation time under the ATLAS solar simulator. All data is normalised to its initial value  
 4 determined at  $t = 0$ . In each part, we plot data measured every 10 minutes using the ATLAS solar  
 5 simulator (solid lines) and every 3 days using the calibrated Newport solar simulator (circular data  
 6 points). The decay in PCE (determined using both types of solar simulator) is fitted to a straight line  
 7 (dashed line or dotted line) for times beyond the burn-in period ( $t > 250$  hours). This linear fit to the  
 8 PCE is used to determine the  $T_{80}$  decay lifetime.

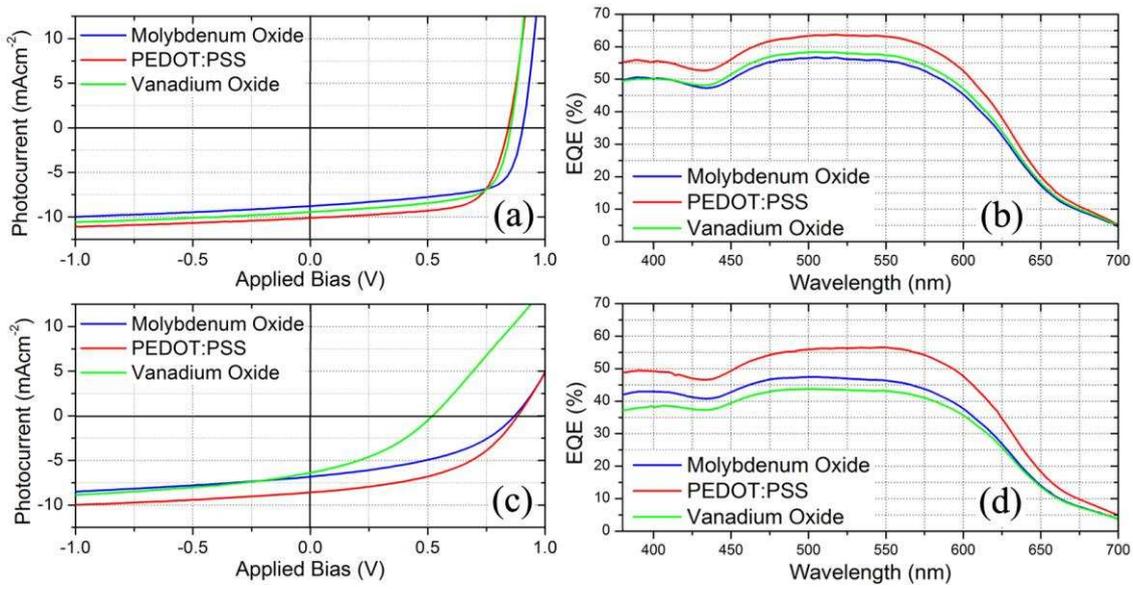
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2 Figure 3. Parts (a) and (b) show initial JV and EQE characteristics for devices respectively, with  
 3 parts (c) and (d) showing device JV and EQE characteristics after 620 hours of irradiation under the  
 4 ATLAS solar simulator.