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Supporting Information

Fenton-Like Oxidation of 4−Chlorophenol: Homogeneous or Heterogeneous?

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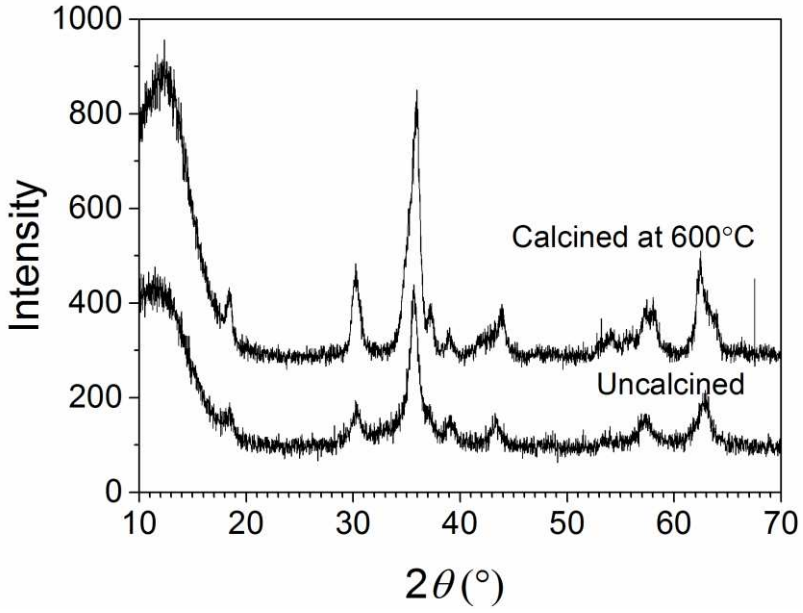

Figure S1. Powder XRD patterns of the calcined and uncalcined $CuFe_2O_4$ powder measured using a Siemens D500 Kristalloflex Diffractometer. The step size was 0.02°.

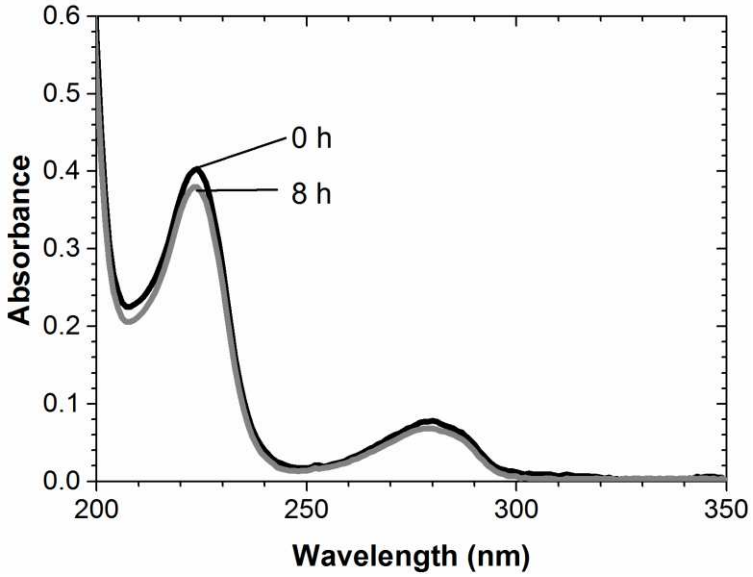

Figure S2. Blank test using 0.48 mM 4-CP only showing negligible 4-CP loss due to volatilisation at 60°C. More datapoints in between but only show the initial and final one because the intermediate ones are similar for clarity,

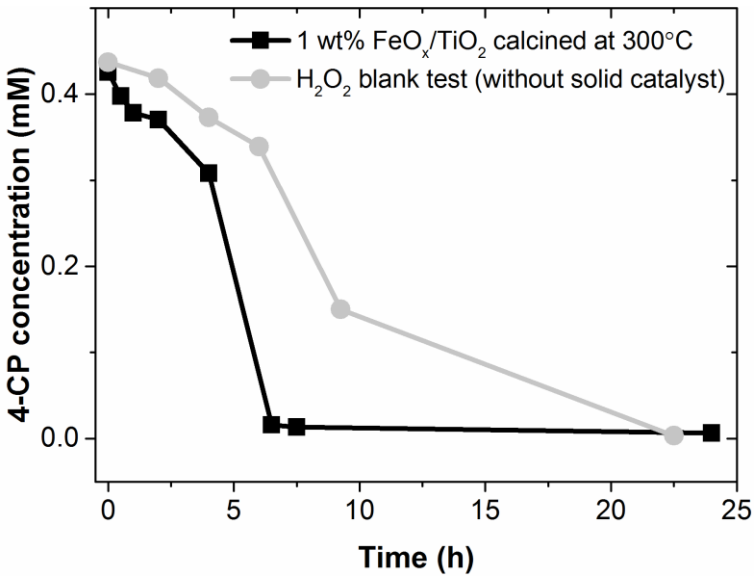

Figure S3. Blank test using 16 mM H_2O_2 without the presence of FeO_x/TiO_2 compared to reaction in the presence of FeO_x/TiO_2 catalyst (16 mM H_2O_2, 1g L^{-1} FeO_x/TiO_2 with1wt% Fe calcined at 300°C).

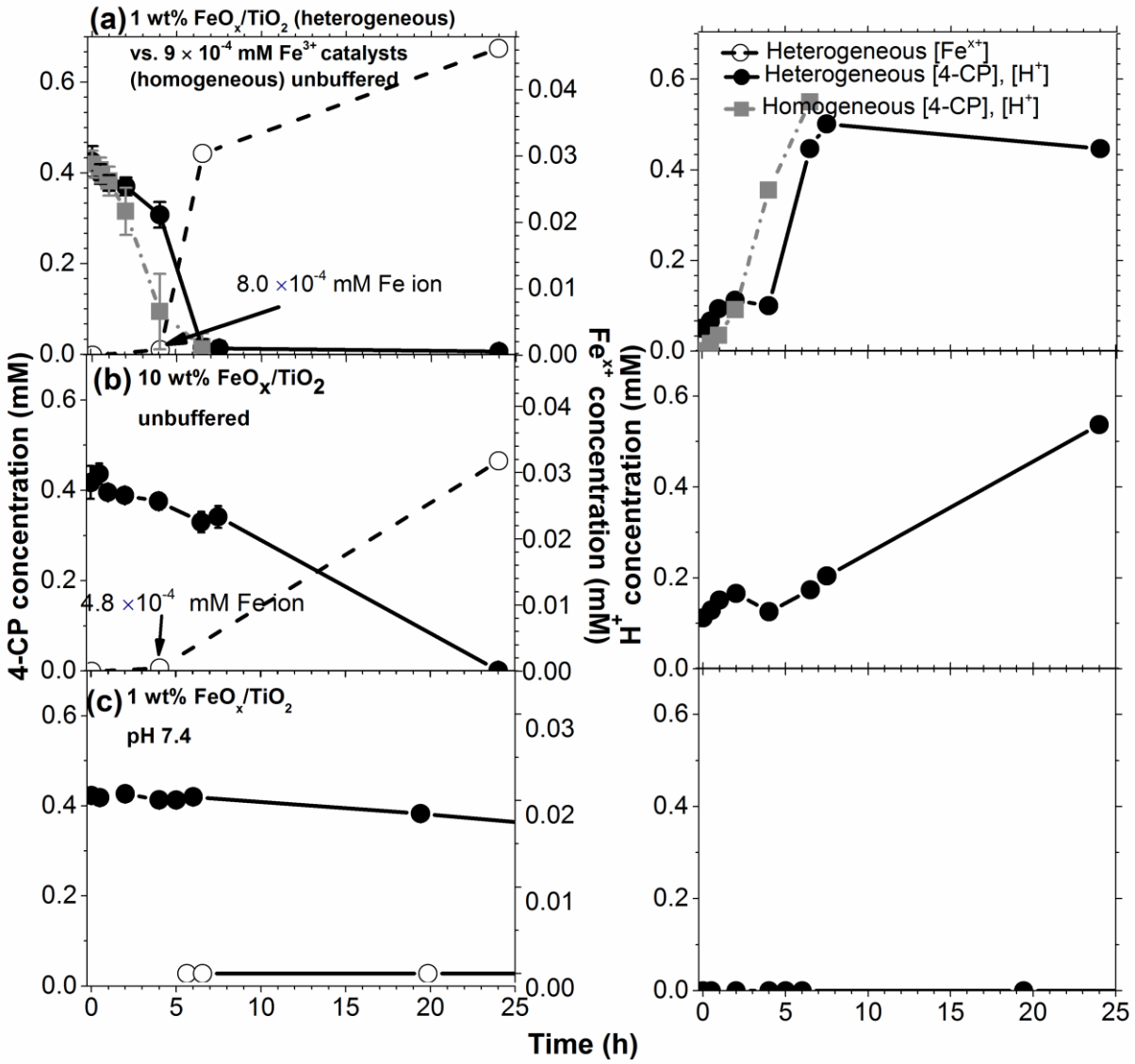

Figure S4. 4-CP degradation using FeO_x/TiO_2 catalysts. (a) The solid phase catalyst FeO_x/TiO_2 (1 wt% Fe) and the solution phase catalyst 9.0×10^{-4} mM $[Fe_2(SO_4)_3]$. (b) FeO_x/TiO_2 (10 wt% Fe) catalyst. (c) FeO_x/TiO_2 (1 wt% Fe) catalyst in a solution buffered at pH 7.4. All initial reactant solutions contained 16 mM H_2O_2 and 0.48 mM 4-CP.

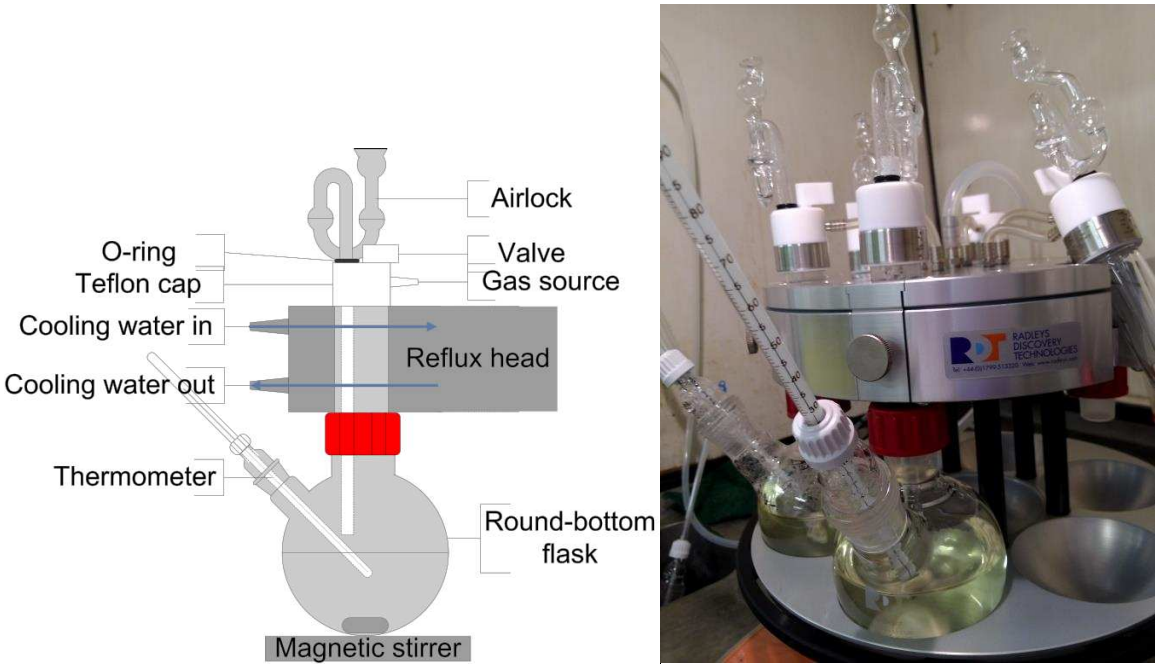

Figure S5. The glass reactor system.