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Searching for an oxygenation event in the fossiliferous Ediacaran of northwestern

Canada

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Late Neoproterozoic (Ediacaran) strata from northwestern Canada provide

a thick and rich sedimentological record, preserving intercalated carbonates and

shale extending from the ~ 635 million year old Marinoan glacial deposits up

through the ~ 541 million year old Precambrian – Cambrian boundary. This region

also holds one of the classic localities for the study of early animal life, with the

ensuing suggestion that this temporal interval captures a gross change in the O$_2$

content of Earth's atmosphere.   To test this hypothesis and bring records of

northwestern Canada into line with other Ediacaran, fossil-bearing basins, we provide a detailed geochemical reconstruction from the Wernecke Mountains of the Yukon. Where possible, we also extend these records to the Ogilvie Mountains to the west and previously published data from the Mackenzie Mountains to the east.

Our work in the Wernecke Mountains is set against a composite $\delta^{13}C$ record for carbonate that preserves three distinct Ediacaran isotope excursions, the lowermost of which (preserved in the Gametrail Formation) is a putative Shuram excursion equivalent. What emerges from a multi-proxy (Fe speciation, sulfur isotopes, major and trace elements analyses) reconstruction is a picture of a persistently anoxic and ferruginous Ediacaran ocean. Notably absent is geochemical evidence for a prominent oxygenation event, an expectation given the appearance of animals and large swings in $\delta^{13}C$. The new insight gained through these data challenge the idea of an Ediacaran jump in atmospheric oxygen, which in turn muddles the link between animal evolution and local geochemical environments.

1.0 INTRODUCTION:

Interest in understanding Earth surface change in the late Proterozoic is rooted in trying to pinpoint the mechanisms and feedbacks associated with the origin of animals (Canfield et al., 2007; Cloud and Drake, 1968; Holland, 1984; McFadden et al., 2008; Nursall, 1959). Decades of work has combed Neoproterozoic successions and provides a robust paleontological roadmap for both the distribution of classic Ediacaran-type fossils

44   as well as the underlying, earlier Ediacaran acritarch record (Fedonkin and Waggoner,

45   1997; Grey, 2005; Hofmann et al., 1990; Narbonne and Aitken, 1990b). Despite

46   numerous models for the mechanisms driving the transition in biological systems

47   (Butterfield, 2009; Canfield et al., 2007; Johnston et al., 2012b), the clues about the

48   critical events that led to the emergence of animals remains at least partially locked

49   within marine sediments. As a result, a suite of studies employing redox sensitive

50   proxies have developed a framework whereby it is proposed that roughly coincident with

51   the first animals, the deep ocean became oxygenated (Canfield et al., 2008; Canfield et

52   al., 2007; Fike et al., 2006; Scott et al., 2008); this was Earth's canonical second great

53   oxidation (DesMarais et al., 1992; Holland, 1984; Och and Shields-Zhou, 2012).

54   Assaying large-scale changes in atmospheric oxygen and geochemical cycles is

55   classically a problem resolved through carbon isotope reconstructions (Broecker, 1970;

56   DesMarais et al., 1992; Fike et al., 2006; Johnston et al., 2012a; Knoll et al., 1986).

57   Here, the isotopic composition of carbonate and bulk organic carbon ($\delta^{13}C_{carb}$ and $\delta^{13}C_{org}$,

58   respectively), when placed in a steady-state framework for the operation of the carbon

59   cycle (Hayes et al., 1999), provide insight on the net production of oxygen. This O_2 then

60   propagates throughout surface environments, often leaving an imprint directly on pO_2,

61   but equally as important it controls sulfate delivery to the oceans and ferric oxide

62   production on the continents (Hayes and Waldbauer, 2006; Holland, 1984; Holland,

63   2006; Holland et al., 1986). The net gain in these three oxidized reservoirs is the true

64   measure of surface oxidation, and requires an expansion of the geochemical toolbox.

65   Fortunately, the sulfur cycle can be indirectly tracked through isotopic reconstructions of

66   sedimentary pyrite records and where possible, sulfates from evaporites, fluid inclusions,

67   or that trapped within the carbonate lattice (Canfield, 1998; Canfield and Teske, 1996;

68   Fike and Grotzinger, 2008; Fike et al., 2006; Kampschulte and Strauss, 2004).   The

69   influence of oxygen on global Fe budgets is more difficult to measure, as much of the

70   ferric iron resides on the continents and in crustal materials.   However, a refined

71   extraction scheme provides a snapshot of iron cycling along a continental margin and

72   serves as a powerful proxy for this budget.   If the commonly posited increase in pO_2

73   accompanies or slightly predates the Ediacaran appearance of animals, it should be

74   detectable in the C-S-Fe geochemistry of strata in northwestern Canada.

75   The last decade has seen the pervasive application of a revised sedimentary iron

76   extraction scheme capable of diagnosing the chemical structure of the marine water

77   column (Canfield et al., 2008; Canfield et al., 2007; Johnston et al., 2010; Johnston et al.,

78   2012b; Li et al., 2010; Planavsky et al., 2011; Poulton et al., 2004; Poulton et al., 2010).

79   Through quantifying the partitioning of iron into different Fe-bearing minerals, it is

80   possible to identify anoxia and further distinguish between ferruginous (ferrous iron

81   bearing) and euxinic (sulfide bearing) conditions (Poulton and Canfield, 2005).   Applied

82   to shale facies from a number of Neoproterozoic basins, Fe geochemistry is revealing a

83   complicated picture of marine redox between the end of the Marinoan glaciation roughly

84   635 Million years ago (Ma) and the Precambrian − Cambrian boundary at 541 Ma.

85   Reconstructions of fossiliferous basins in modern day Russia and Newfoundland provide

86   internally consistent pictures of a water column that is sometimes oxygenated, but when

87   anoxic, is always ferruginous (Canfield et al., 2007; Johnston et al., 2012b).   As it relates

88   to the origin of metazoans, up-section changes in the redox state of the EEP basin

89   (targeting transitions that antedate or that are coincident with the first appearance of

90   animals) are not profound, leading to the subtler hypothesis that the geochemical/redox

91   stabilization of these environments may have also been important to the development of

92   more complex life and ecologies (Johnston et al., 2012b). In contrast, condensed fossil-

93   bearing stratigraphic sections from South China capture much more reducing marine

94   conditions, with an almost episodic fluctuation between ferruginous and euxinic

95   conditions (Li et al., 2010; Shen et al., 2008a) and even some indication of oxic

96   conditions (Sahoo et al., 2012). Inferentially, marine chemical evolution through the

97   Ediacaran appears to be a basin-by-basin affair (Johnston et al., 2012b; Kah and Bartley,

98   2011). Interestingly, and appreciating that taphonomy is also variable, the distribution of

99   fossil-bearing assemblages is similarly heterogeneous.

100   Marine redox/geochemical variability will always reflect the interplay between

101   equilibrium with the overburden of atmospheric oxygen and the move toward

102   disequilibria in water column dissolved oxygen (DO) as a result of differential nutrient

103   inputs and productivity-remineralization regimes. How this context then feeds back on

104   the link to animals and their unique physiological requirements is less clear, but must be

105   related. Finding continuity between the geochemical and biological records through

106   Ediacaran successions, especially those containing early metazoan records, requires high-

107   resolution sampling, detailed and explicit correlations to fossil bearing horizons, and an

108   expansion of geochemical reconstructions to include an understanding of sediment

109   protolith.

110   One of the cornerstone Ediacaran paleontological records comes from mixed

111   siliciclastic and carbonate sequences in northwestern Canada (Hofmann et al., 1990;

112   Kaufman et al., 1997; Macdonald et al., this issue; Narbonne and Aitken, 1990b;

113   Narbonne et al., 1994; Pyle et al., 2004). Work on C-S-Fe in early Ediacaran stratigraphy

114   from the Mackenzie Mountains has already hinted at evolving water column chemistry

115   (Canfield et al., 2008; Shen et al., 2008b), but how that geochemical setting varies

116   spatially (across the basin) or in time (up toward the Precambrian – Cambrian boundary)

117   is unclear. In what follows we describe a detailed geochemical study, in conjunction with

118   a revised stratigraphic context (Macdonald et al., this issue), where we track major and

119   trace element geochemistry and key isotopic metrics. Taken together, and when

120   assimilated with data from other Ediacaran paleo-basins, a more lucid picture of Earth

121   surface change and biological innovation is realized.

122

123   **2.0 GEOLOGICAL SETTING**:

124   2.1 THE WERNECKE MTNS.

125   In northwestern Canada, Ediacaran strata begin with micropeloidal dolomite of

126   the Ravensthroat Formation (James et al., 2001), which caps a lowstand systems tract at

127   the top of the Cryogenian Keele Formation (Day et al., 2004) and glacigenic diamictite

128   wedges of the Marinoan Icebrook Formation (Aitken, 1991b). The Ravensthroat is up to

129   30 m thick and is locally overlain by ≤10 m of limestone with pseudomorphosed

130   aragonite fans of the Hayhook Formation (James et al., 2001). In the central Mackenzie

131   Mountains, sea-floor barite discontinuously mantles the contact between the Ravensthroat

132   and Hayhook formations, which together comprise the 'cap carbonate' (Hoffman and

133   Halverson, 2011). Above the cap carbonate, the Ediacaran stratigraphy is a mix of

134   siliciclastic and carbonate rocks (defined as the Sheepbed, Gametrail, Blueflower, and

135   Risky formations, along with the informal, newly described Sheepbed carbonate and June

136   beds (Macdonald et al., this issue)) preserving large lateral variability in exposure and

137   thickness (Fig. 1).

138   In the Wernecke Mountains, the lower Sheepbed Formation consists of shale with

139   thin limestone interbeds and measures ~350 m thick. Similar to exposures in the

140   Mackenzie Mountains, the lower ~100 m of shale are the darkest and most fissile and

141   likely contain the maximum flooding surface (MFS) of the glacio-eustatic marine

142   transgression. A distinct coarsening occurs roughly 180 m into the Sheepbed in our

143   Wernecke section (same as Goz A from (Pyle et al., 2004)), where siltstone interbeds and

144   platform-derived debris flows begin to appear. Upsection from this surface, limestone is

145   increasingly more abundant and finally gives way to shallow water dolostone of the

146   informal Sheepbed carbonate (Aitken, 1991a; Macdonald et al., this issue).   This

147   carbonate unit was previously correlated with the Gametrail Formation (Pyle et al.,

148   2004), but a recent examination of this correlation has identified an additional sequence

149   boundary between the Sheepbed carbonate and the Gametrail Formation at its type

150   locality (Macdonald et al., this issue). The lower ~150 m of the Sheepbed carbonate

151   consists of a monotonous, massive dolomite with rare cross-bedding and giant ooids.

152   This is capped by a major karst surface, and is succeeded by ~40 m of coarse-grained

153   sandstone and calc-arenite, which are assigned to the informal June beds. The overlying

154   Gametrail Formation (Peritidal member of (Pyle et al., 2004)) consists of ~50 m of

155   peritidal dolomite stacked in ~10 m thick parasequences capped by exposure surfaces.

156   These surfaces culminate in a major unconformity that defines the base of the Blueflower

157   Formation (Yuletide member of (Pyle et al., 2004)) and is incised by multiple beds of

158 cobble-clast quartzite conglomerate. The conglomerates are succeeded by a major

159 flooding surface and shale with interbedded normally graded beds of sand and silt

160 (interpreted as turbidites) that host simple bedding plane traces and Ediacaran disk fossils

161 (Narbonne and Aitken, 1990a). The Blueflower Formation shallows upwards into

162 hummocky cross-stratified sandstone and pink sandy dolomite of the Risky Formation.

163 The top of the Risky Formation is marked by a karstic unconformity. Above this

164 unconformity, small shelly fossils and a diverse microfossil assemblage have been

165 described in phosphatic carbonates in the Ingta Formation near the southern Wernecke

166 Mountain sections at Goz C and D (Nowlan et al., 1985; Pyle et al., 2006). This

167 stratigraphic framework is presented in Figure 1 and discussed in detail in a companion

168 study (Macdonald et al., this issue).

169

170 2.2 CORRELATION TO THE MACKENZIE AND OGILVIE MTNS

171 Correlative Ediacaran strata have been measured both to the east (in the central

172 and southeast Mackenzie Mountains) and to the west (Ogilvie Mountains). The details of

173 these correlations are published elsewhere (Macdonald et al., this issue) with key features

174 highlighted below. Near Shale Lake in the central Mackenzie Mountains, the Sheepbed

175 Formation (the target formation of this study) is between 450 and 700 m thick and

176 dominated by black shale with minor fine-grained sandstones to siltstone interbeds

177 interpreted as turbidites (Fig. 1). Recent workers reconstructed the C-S-Fe systematics in

178 this section and suggested a shift in Fe-speciation ratios – from anoxic to oxic – at ~180

179 m (Shen et al., 2008b). The general position of this geochemical transition roughly

180 corresponds to a coarsening in which siltstone interbeds become more common. In the

181   southeast Mackenzie Mountains, at Sekwi Brook the lower Sheepbed Formation is

182   incised by a cobble-clast debris flow that contains abundant giant ooids and has a sharp,

183   erosive base. Above this contact, the background sediment is dominated by carbonate-

184   rich shale and the first appearance datum of the Ediacaran Aspidella occurs within 15 m

185   on the sole of a normally graded fine-grained sandstone bed. The overlying middle

186   member as described by Dalrymple and Narbonne (1996) consists of carbonate-rich shale

187   and siltstone with common Aspidella (Narbonne and Aitken, 1990a). Ediacaran biota are

188   preserved primarily on the bottoms of fine-grained sandstone beds interpreted to

189   represent the Bouma C turbidite sub-division. Departing from previous correlations that

190   assigned these strata to the Sheepbed Formation (Narbonne and Aitken, 1990a; Shen et

191   al., 2008b), Macdonald and colleagues (this issue) assigned the Ediacaran-bearing,

192   carbonate-rich strata above the surface at ~200 m to the June beds, which elsewhere rest

193   above the Sheepbed carbonate, separated by a major sequence boundary. This

194   assignment invalidates the correlation of the shift in Fe speciation of Shen et al. (Shen et

195   al., 2008b) with the appearance of Ediacaran biota. Finally, in the Coal Creek inlier of

196   the Ogilvie Mountains, the Sheepbed Formation is ~250 m thick and is succeeded by ~25

197   m of massive white to buff-colored dolostone with pervasive cements and an additional

198   ~100 m of thinly bedded, pink dolo-ribbonite with hummocky cross-stratification,

199   grainstone, and stromatolites (unit PH 4). The dolomites are succeeded by ~10 m of dark-

200   colored, nodular, organic-rich limestone, which resembles the lower Blueflower

201   Formation at Sekwi Brook. This unit is unconfomably overlain by Cambrian, fossil-

202   bearing siliciclastic strata (unit PH5).

203

204   **3.0 METHODS:**

205   Carbon, sulfur, and iron data were generated through standard techniques.

206   Carbonate carbon content (% carbonate) was quantified by loss through an acid

207   dissolution, which also isolated a residue of siliciclastic material and organic matter.

208   These mass fractions were further distinguished through quantifying yields on a Carlo

209   Erba Elemental Analyzer linked to a Thermo Finnigan Delta –V configured in continuous

210   flow mode, which also yielded $\delta^{13}C_{org}$. The isotopic composition of carbonate ($\delta^{13}C_{carb}$)

211   is from a companion study (Macdonald et al., this issue). Uncertainties for $\delta^{13}C$ are

212   0.2‰, and better than 0.05 wt % for TOC.

213   The sedimentary iron cycle is evaluated through an operationally defined

214   extraction protocol optimized for siliciclastic rocks and marine sediments (Poulton and

215   Canfield, 2005). This procedure isolates the biogeochemically reactive iron minerals

216   (binned as oxides [Fe_{ox}], mixed valence Fe minerals such as magnetite [Fe_{mag}], and

217   ferrous iron carbonates [Fe_{carb}]). Extracted separately, but still related are iron sulfide

218   minerals (Fe_{py}) (Canfield et al., 1986). The sum of these extractions represents the highly

219   reactive pool (FeHr). Total Fe content and other major and trace element data were

220   analyzed commercially (SGS, Alberta Canada) by ICP-AES on bulk samples dissolved in

221   HF-HNO_3-$HClO_4$ solution.

222   Carbonates from the Sheepbed carbonate and overlying Gametrail Formation

223   were also processed for carbonate-associated sulfate according to established protocols

224   (Burdett et al., 1989; Gill et al., 2007). These sulfate isolates and pyrites extracted from

225   both the residues associated with the CAS extraction and from siliciclastic pyrite iron

226   (Fe_{py}) where analyzed through combustion via a Costech Elemental Analyzer linked to a

227   Delta V in continuous flow mode (measured as $SO-SO_2$) with a precision of < 0.2 ‰ in

228      $\delta^{34}S$.

229   Samples from northwestern Canada were also analyzed (and data manipulated) to

230   determine the chemical index of alteration, or CIA. This metric tracks the preferential

231   loss of particular cations through weathering reactions: $CIA=[Al_2O_3/(Al_2O_3 + CaO^* +$

232      $Na_2O + K_2O)]$ (Fedo et al., 1995; McLennan, 1993; McLennan and Taylor, 1991; Tosca

233   et al., 2010). For these calculations, oxide fractions are determined from ICP data and

234   corrections are taken from established methods. Namely, for CaO^* in siliciclastics with

235   variable mass fractions of carbonate, the silicate $[CaO]$ fraction is fixed to $[Na_2O]$

236   (Kronberg et al., 1986).

237

238   **4.0 RESULTS:**

239   Carbon isotopes values in carbonate and organic matter vary up-section, with

240   clear distinction between individual formations, despite some lithofacies dependence

241   (Fig. 2). Carbonate $\delta^{13}C$ values are typically depleted through the Hayhook cap

242   carbonate and into the lowermost Sheepbed Formation, but appear to recover steadily

243   back to ~5‰ by the Sheepbed carbonate unit, consistent with previous studies of this and

244   other basal Ediacaran successions (Hoffman et al., 2007; James et al., 2001). A negative

245      $\delta^{13}C_{carb}$ excursion is recorded in the Gametrail Formation, with two additional anomalies

246   preserved within the overlying Blueflower and Risky formations. The Gametrail

247   Formation excursion preserves a dramatic and coherent negative $\delta^{13}C_{carb}$ anomaly of >

248   10‰ (from +4 to -7‰), and recovers to near 0‰ before the Gametrail – Blueflower

249   sequence boundary. Sparse $\delta^{13}C_{carb}$ data from the Blueflower Formation preserves a

250   negative anomaly down to -8‰, and after a recovery to -2‰, another depletion event

251   down to < -5‰ in the Risky Formation. Carbon isotope data from PH4 in the Ogilvie

252   Mountains are also highly variable but systematic (Fig. 2), with $\delta^{13}C_{carb}$ beginning near

253   0‰ before plummeting to -8‰ and recovering up-section back to +4‰. Based on new

254   correlations (Macdonald et al., this issue), the Gametrail Formation excursion post-dates

255   the Sheepbed transgressive-regressive sequence and is tentatively regarded as time

256   equivalent to the global Shuram – Wonoka excursion (Grotzinger et al., 2011).

257   The isotopic composition of organic matter from the Wernecke and Ogilvie

258   Mountains is also presented (Fig. 2). Organic matter from the Sheepbed Formation in the

259   Wernecke Mountains is consistently offset from carbonate by roughly -30‰.   The

260   Blueflower and Gametrail formations preserve $\delta^{13}C_{org}$ values that do not vary

261   systematically with $\delta^{13}C_{carb}$, and range widely from -20‰ to -35‰. Similarly, total

262   organic carbon contents are quite variable. In the Sheepbed Formation, TOC is low at the

263   base (0.1 wt %) with a pronounced enrichment of > 0.4 wt % between 100-225 m. In the

264   uppermost Sheepbed Formation, values are less than 0.2 wt%.   In the overlying

265   Blueflower Formation, similar TOC concentrations are preserved, with 2 samples

266   preserving anomalous enrichments of > 0.4 wt%. In the Blueflower Formation, the

267     $\delta^{13}C_{org}$ is highly variable with most samples falling between -30‰ and -15‰. As in the

268   Wernecke Mountains, the TOC values are significant but highly variable in the Ogilvie

269   Mountains, with a mean of ~ 0.1 wt%.

270   Iron speciation data (Fig. 3) for siliciclastic samples (Sheepbed and Blueflower

271   formations) suggest that most of the reactive Fe is locked within iron carbonates (0.7±0.4

272   wt%) or oxide phases (1.0±0.4 wt%). Magnetite and pyrite iron represent only minor

273   contributions (0.11±0.07 wt% and 0.10±0.08 wt%, respectively). The δ^{34}S data from

274   pyrite span a large range of values in the Sheepbed Formation (Fig. 4), but are

275   stratigraphically coherent: more depleted isotopic compositions characterize the lower

276   portion of the unit (< -20‰), systematically increasing toward more enriched values (>

277   20‰) upsection and leveling off at ~ 250 m. Above this horizon, sulfides are variable

278   (generally between 10-30‰) but are always positive. Sulfates extracted from the

279   Sheepbed carbonate contain consistent CAS values (in δ^{34}S) of 30‰.

280   Major and trace element chemistry through the Wernecke Mountains section

281   broadly tracks lithology. Redox sensitive elements (e.g., Mn, V, Zn, Cr) and those that

282   participate more actively in biogeochemistry (P) show some systematic variability.

283   Major element chemistry is also variable, in part recorded by the chemical index of

284   alteration: CIA (Fig. 5).

285

286   **5.0 DISCUSSION**:

287   5.1 THE EDIACARAN CARBON CYCLE

288   Understanding the nature of the Ediacaran ocean-atmosphere system begins with

289   a critical evaluation of the behavior of the carbon cycle. In many ways, the pattern of

290     $\delta^{13}C_{carb}$ through the basal Hayhook-Icebrook cap carbonate sequence and overlying

291 Sheepbed Formation is typical of post-glacial successions globally (Halverson et al.,

292 2005). Here, depleted and stable $\delta^{13}C_{carb}$ (~ -5‰) in the Hayhook Formation continue

293 into the basal Sheepbed Formation (Fig. 2), which records a modest isotopic enrichment

294 up section (~ 0.1‰ per 10 m). This gradual change is punctuated by a larger and

295 terminal increase to roughly 6‰ (~ 1‰ per 10 m) through the transition into the

296 Sheepbed carbonate. The apparent disparity in the rate of isotopic change may be an

297 artifact of fluctuations in sedimentation rates rather than a true regime shift. The

298 progressive enrichment through the highstand and ensuing shallowing-upward sequence

299 is neatly tracked by $\delta^{13}C_{org}$ values from shale and (where available) carbonate, which are

300 systematically offset by roughly 30‰ (Fig. 2). Interestingly, total organic carbon

301 contents vary with a distinct maximum at ~ 120 m (Fig. 4), with no effect on $\delta^{13}C_{org}$

302 (Johnston et al., 2012a). Estimates for the timescale of deposition for the entire post-

303 glacial sequence (Hayhook – Sheepbed) range from a maximum duration of 635 Ma –

304 580 Ma, to a shorter duration of a few million to tens of million years (Macdonald et al.,

305 this issue).

306 The sheer abundance of carbon isotopic variability within the middle and upper

307 stratigraphy of the Wernecke Mountains, although stratigraphically coherent, requires

308 attention. The most prominent feature of roughly time correlative strata to the Gametrail

309 Formation is the ca. 580 Ma Shuram-Wonoka excursion (Grotzinger et al., 2011). Like

310 in the Wernecke and Ogilvie Mountains, a ~ 10‰ negative $\delta^{13}C_{carb}$ excursion reaching a

311 minimum value of -6‰ is broadly consistent with the record in Oman (Fike et al., 2006),

312 Australia (Calver, 2000), Namibia (Workman et al., 2002), western USA (Corsetti and

313 Kaufman, 2003), and South China (McFadden et al., 2008). Notably, the $\delta^{13}C$ signal in

314   the Ogilvie Mountains is ~2‰ offset from the Wernecke Mountains toward more

315   depleted values suggesting a reverse basinal gradient of some variety or that the two are

316   not correlative (Macdonald et al., this issue). In viewing the Gametrail Formation

317   excursion as representative of this basin, this fits with a correlation to the Shuram –

318   Wonoka excursion; an assignment that is more thoroughly described in (Macdonald et al.,

319   this issue), but whose consequences for global surface environments are explored herein.

320   The uppermost excursion in the Risky Formation of the Wernecke Mountains

321   shares some similarities with the Precambrian – Cambrian boundary excursion (Knoll

322   and Walter, 1992). However, the isotope excursion in the Blueflower Formation is more

323   difficult to unique correlate or extend to a global framework, as it is developed in

324   carbonate cements within siliciclastic strata. Carbonate preserved within these settings

325   likely reflect an imprint of both global DIC adopted from overlying seawater and from

326   local (sedimentary) remineralization reactions (Schrag et al., 2013). Why the Blueflower

327   is influenced by authigenic carbonate whereas other siliciclastic strata are less affected

328   (see the Sheepbed Formation, for instance) is most probably the result of the local

329   controls on remineralization and the openness of local pore waters to overlying seawater

330   (Schrag et al., 2013).

331   One means of testing the fidelity of these interpretations is through an

332   examination of $\delta^{13}C_{org}$ within both the carbonate and siliciclastic units. The early

333   Ediacaran Sheepbed Formation preserves a snapshot of typical carbon isotopic behavior,

334   where $\delta^{13}C_{org}$ tracks carbonate carbon closely (Hayes et al., 1999; Johnston et al., 2012a;

335   Knoll et al., 1986). Conversely, the upper portion of the stratigraphy (Gametrail through

336   Blueflower Formation) does not conform to this simple behavior, as the isotopic

337   composition of the carbonate and organic carbon appear to varying independently. As we

338   interpret the Blueflower Formation excursion as the product of secondary influences, no

339   concrete prediction exists for coincident $\delta^{13}C_{org}$. This is not the case for the putative

340   Shuram anomaly in the Gametrail Formation or the boundary excursion in the Risky

341   Formation. For the Shuram-like excursion in the Ogilvie Mountains, the isotopic

342   composition of the organic matter is disjoined from the carbonate isotopes, much like that

343   observed in Oman and South China (Fike et al., 2006; McFadden et al., 2008). Unlike

344   that preserved elsewhere, however, the $\delta^{13}C_{org}$ from the Ogilvie Mountains is highly

345   variable rather than being remarkably invariant (Fike et al., 2006; McFadden et al., 2008).

346   Setting aside for the moment the interpretation of $\delta^{13}C_{carb}$, it is difficult to uniquely

347   diagnose the source(s) of huge variability in organic matter through the Gametrail,

348   Blueflower and Risky formations. One can envision fluctuating contributions from

349   contemporaneous primary production, selective remineralization, detrital fluxes (assayed

350   through determining protolith) and perhaps the later delivery of organic rich fluids, with

351   only primary organic matter tracking the isotopic composition of DIC (and hence

352   preserving the same excursion as $\delta^{13}C_{carb}$) (Johnston et al., 2012a; Schrag et al., 2013).

353   The primary variability in $\delta^{13}C_{carb}$ must be interpreted against the backdrop of

354   uncertainty about the sources and mechanisms influencing the $\delta^{13}C_{org}$. The traditional

355   means of interpreting carbonate carbon isotopic excursions invokes changes in the

356   fractional burial of organic matter, f_{org}. That is, given isotopic mass balance (Hayes et al.,

357   1999; Johnston et al., 2012a), the isotopic difference between carbon outputs (commonly

358   simplified as carbonate and organic carbon) must sum back to the composition of mantle

359   inputs. Using the stoichiometry of primary production, this estimated flux is roughly

360 equivalent to the production of O_2 (Bristow and Kennedy, 2008; Knoll et al., 1986).

361 Confidence in f_{org} is lessened, however, when $\delta^{13}C_{carb}$ plunges below the oft-assumed

362 mantle input value of -6‰ (Grotzinger et al., 2011) or remains at significantly elevated

363 values for 10^7-10^8 years (Halverson et al., 2005). This framework underlies hypotheses

364 about large-scale 'events' as driving the profound isotopic change preserved within

365 Ediacaran records, most notably the Shuram excursion (Fike et al., 2006; Rothman et al.,

366 2003). Both a change in f_{org} or a massive remineralization 'event' would carry direct and

367 predictable consequences for oxidant budgets (Bristow and Kennedy, 2008).

368 Redox reconstructions serve as a proxy for surface oxidizing capacity (i.e. the

369 availability of O_2, SO_4^{2-}, NO_3^-, Fe^{3+}) and represent a test of these models. Put differently,

370 if Shuram-like excursions from the Wernecke and Ogilvie mountains were in fact

371 reflecting a real change in the oxidation state of the world's oceans (and atmosphere) –

372 through a massive remineralization event (consuming oxidants) or through a crash in f_{org}

373 (turning down oxidant production) – redox sensitive metrics should record a change.

374 Conversely, if for instance the Gametrail Formation $\delta^{13}C_{carb}$ excursion is reflecting the

375 ingrowth of significant authigenic carbonate precipitation, which alters the overall

376 isotopic mass balance underlying f_{org} (Schrag et al., 2013), then a massive change in

377 oxidants is not required. We perform this test below.

378

379   5.2 THE BEHAVIOR OF THE EARLY EDIACARAN FE CYCLE

380 The redox character of Ediacaran ocean chemistry is queried by investigating

381 biogeochemically cycled elements, particularly iron and sulfur. Recall that Earth's total

382   oxidant budget in the modern is dominated by ferric iron and sulfate, with pO_2 coming in

383   third in sheer oxidizing capacity (Hayes and Waldbauer, 2006). To assay Fe budgets in

384   the past, speciation techniques track the accumulation of mobile, highly reactive iron

385   (FeHr) enrichments in anoxic environments (FeHr/FeT > 0.38 is deemed anoxic and <

386   0.22 as oxic (Poulton and Raiswell, 2002; Raiswell and Canfield, 1998)). Related to this,

387   a predominance of FeHr residing in pyrite (> ~70-80%) suggests euxinic conditions

388   (Anderson and Raiswell, 2004; Poulton and Canfield, 2011), whereas lower sulfide

389   contents (FeHr in carbonates and oxides) are indicative of a ferruginous water column

390   (Poulton et al., 2004). For sulfur, isotopic reconstructions serve as a complementary tool

391   in assaying the behavior of the biosphere and the generation of biogenic iron sulfides.

392   Both approaches carry implications for the location and dominance of particular

393   heterotrophic metabolisms, namely dissimilatory sulfate and iron reduction, which in turn

394   inform environmental conditions.

395   We begin by interpreting siliciclastic sediments from the Ediacaran of the

396   Wernecke Mountains. Near the base of the Sheepbed Formation, FeHr/FeT > 0.38 and

397   Fe_{py}/FeHr is much less than 0.8 (Fig. 3). The conventional interpretation would then

398   point to a generally anoxic and ferruginous water column. As deposition of the lowermost

399   ~100m of the Sheepbed Formation may have been relatively rapid, considering that the

400   necessary accommodation space was developed during glacial times, the anoxia signal

401   may reflect a transient state associated with the ventilation of the post-glacial ocean.

402   Throughout the remainder of the Sheepbed shale, FeHr/FeT straddles the geochemical

403   threshold defining anoxia (mean FeHr/FeT ≈ 0.4). Although the mean values throughout

404   fall near the calibrated anoxia threshold, the variability within the any given stratigraphic

405   interval does allow for the possibility of intervening (and transient) oxic conditions.

406   Values near the oxic threshold (left most red dashed line in Figure 3) become more

407   common toward the top of the Sheepbed, perhaps suggesting an influence from a well

408   mixed and oxygenated surface-ocean. The Sheepbed can then be interpreted as recording

409   a background anoxic, ferruginous deep-water condition that may be punctuated by

410   occasional communication with the surface ocean. This nicely fits with the sequence

411   stratigraphic architecture.

412   After passing through the Sheepbed carbonate and the carbonate dominated

413   Gametrail Formation, siliciclastic sedimentation returns in the Blueflower Formation.

414   The Blueflower Formation consists of multiple small-scale sequences superimposed on a

415   larger shallowing upward sequence (Macdonald et al., this issue), and is relatively thin in

416   the Wernecke Mountains, but thickens to the southwest into the Selwyn Mountains (Fig.

417   1). Above the sandstone and conglomerate that constitute the base of the succession, the

418   unit is shale and siltstone dominated and transitions upward into coarser grained siltstone

419   and sandstone. Importantly, it is within the Blueflower Formation that the first Ediacaran

420   biota appear in the Wernecke Mountains (Pyle et al., 2004). One might expect then,

421   given the oxygen requirements of animals (Raff and Raff, 1970; Runnegar, 1991), that

422   redox measures from the Blueflower Formation would all point toward abundant O_2 at

423   this time. In contrast to this prediction, the mean FeHr/FeT far exceeds 0.4 with no

424   samples falling below the oxic threshold (Fig. 3). Despite the similar suggestion of

425   predominant anoxia and low sulfide contents in the Blueflower Formation, the

426   distribution of Fe is quite different from the Sheepbed Formation. The Sheepbed

427   Formation preserves largely invariant FeT contents, with iron carbonate representing the

428   primary FeHr phase. The Blueflower Formation, on the other hand, carries a wide range

429   of FeT with oxides being the dominant phase. Facies within the Blueflower Formation,

430   which include hummocky cross stratification, suggest a near shore environment, perhaps

431   indicating that riverine-derived Fe-oxides may have been scavenged in a fashion similar

432   to today (Poulton and Raiswell, 2002). Alternatively, the source of sediments (protolith)

433   or the weathering regime as a whole (sheer mass flux) may be different between the two

434   units. If true, this would also impact local sediment biogeochemistry and may help

435   explain the distribution of authigenic carbonate.

436   Sequence stratigraphy and an expanded geochemical toolbox help inform the

437   depositional environment of the post-Marinoan (earliest Ediacaran) Sheepbed Formation.

438   The deposition of the lower ~ 100 m of the Sheepbed Formation followed the alkalinity

439   pulse that generated the cap carbonate (Higgins and Schrag, 2003) and may have been

440   relatively rapid. This is succeeded by a MFS approximated within 100 to 180 m from the

441   base and a shallowing upward sequence that grades into the Sheepbed carbonate. Similar

442   to Fe speciation, Fe/Al ratios track net iron enrichment over a crustal background

443   composition (Lyons and Severmann, 2006). Iron - Aluminum ratios are lowest in the

444   basal Sheepbed Formation (< 0.5), but quickly increase and stabilize at ~ 0.5 for the

445   remainder of the siliciclastic deposition: values generally indicative of typical oxic

446   marine deposition. The slightly more depleted Fe/Al values captured during initial

447   transgression may reflect that the terrestrial input of unreactive Fe was lower than

448   average shale in this locality (driving Fe/Al down and FeHr/FeT up). If true, the then

449   maximum of FeHr/FeT in the basal Sheepbed Formation would be an artifact of

450   weathering through this interval and not reflect a marine condition different from the

451   overlying middle and upper Sheepbed Formation, where both FeHr/FeT and Fe/Al vary

452   around a stable mean value.

453   Adding complexity to this story is a notable maximum in manganese (Mn)

454   coincident with the MFS (Fig 5b). Punctuated Mn enrichments are often interpreted as

455   reflecting an increase in net oxidation state, such as is observed in the early

456   Paleoproterozoic in association with the Great Oxidation Event (Holland, 2006; Holland

457   et al., 1986). However, we note that this signature occurs within the middle of the post-

458   glacial transgressive-regressive sequence, meaning that relative water depth may play a

459   role. Rather than indicating wholesale oxygenation, which would be in contrast to Fe

460   data, we interpret the Mn maxima as reflecting an oxic surface ocean Mn enrichment that

461   encroaches on the seafloor over this interval. This would require the Fe^{2+} content of the

462   deeper, anoxic ocean be modest (but still ferruginous) as to not reduce the Mn as it settles

463   and is deposited. This is consistent with the stable signal from other redox sensitive

464   metals (V, Ni, Cr, Co, Cu and Zn: Fig. 6), none of which independently suggest a mid-

465   Sheepbed oxidation event. Chalcophile elements like Mo are notably absent from these

466   samples, likely as a result of generally low sulfide contents rather than being related to

467   the size of the marine redox sensitive metal reservoir. Together, the Sheepbed Formation

468   records a relatively stable geochemical regime where deeper basinal waters were

469   generally anoxia (and ferruginous) and controlled in large part by changes in relative sea-

470   level and local biogeochemistry.

471   The increases in TOC and total P contents over the initial trangression and MFS

472   tracks increases in pyrite abundance, consistent with the expectation that with a greater

473   flux of organic carbon through the water column and to the sediments, the more

474   heterotrophic sulfate reduction will follow. It is notably that C:P ratios are quite high,

475   similar to that seen in the Ediacaran from Russia (Johnston et al., 2012b) and Cambrian

476   of Australia (Creveling et al., 2013), and here likely are more a function of lower relative

477   TOC contents.   That noted, the inner-workings and controls on the Neoproterozoic P

478   cycle and deposition remains murky. The sulfur isotopic composition of pyrite provides

479   additional insight into Ediacaran biogeochemistry. The $\delta^{34}S_{sulfide}$ is very negative in the

480   basal Sheepbed Formation and increases systematically over the lower ~ 100 m (Fig 5a).

481   In the upper 300 m of shale, $\delta^{34}S_{sulfide}$ values are always positive and often approximate

482   that of seawater sulfate.   Carbonate associated sulfate extracted from the Sheepbed

483   carbonate carries a $\delta^{34}S$ near 30‰. We interpret the general change in pyrite $\delta^{34}S$ as

484   reflecting the migration of the zone of sulfate reduction from very near the sediment

485   water interface (sulfate replete), to some depth in the sediments. Once in the sediments, it

486   is more realistic for sulfate to become limiting, as it would be diffusion controlled in the

487   absence of bioturbation (Canfield and Farquhar, 2009). These fractionation patterns may

488   also reflect changes in local sulfate reduction rates (Chambers et al., 1975; Leavitt et al.,

489   2013). Given that TOC fluctuates wildly and does not track the $\delta^{34}S$ of pyrite, the role

490   for variable sulfate reduction rates is, however, less likely. Finally, the progression of

491     $\delta^{34}S$ through the lower Sheepbed Formation could be interpreted as preserving a crash in

492   the marine SO_4 pool, rather than the migration of sulfate reduction deeper into sediments.

493   This interpretation is also not preferred, given the longer residence times of sulfate at

494   even 5 or 10% that of the modern ocean and the expectation that the $\delta^{34}S$ of sulfate would

495   drastically increase as the concentration dropped. Regardless, it is important to note that

496   despite increases in pyrite abundances, $Fe_{py}/FeHr$ remains relatively low through the

497   entire Sheepbed Formation (Fig. 3b, 3c), leaving iron reduction to quantitatively outpace

498   sulfate reduction. Surely, biogenic sulfide generated deeper in sediments will be more

499   likely to be pyritized, perhaps explaining some of the pyrite enrichment and further

500   linking $\delta^{34}S$ to the iron cycle.

501   **5.2.1 Extensions to the Mackenzie Mountains**

502   The interpretation of the Sheepbed Formation shale in the Wernecke Mountains

503   can be compared to similar datasets from the Mackenzie Mountains at Shale Lake (see

504   Figs. 1, 7a). In one such study (Shen et al., 2008b), it was argued that a distinct shift in

505   the FeHr/FeT at ~ 170 m captured the oxygenation of the basin's water column. Because

506   this FeHr/FeT shift was thought to correlate to the appearance of Ediacaran fossils,

507   primarily Aspidella from Sekwi Brook (Narbonne and Aitken, 1990b)(also within the

508   Selwyn Mountains), this apparent Sheepbed oxygenation was interpreted to be a major

509   driver for the biological innovation that resulted in the appearance of these organisms.

510   Evidence for Ediacaran oxygenation in northwestern Canada is also important, because

511   similar findings from fossil-bearing units in Newfoundland (Canfield et al., 2007), and

512   the Eastern European platform of Russia (Johnston et al., 2012b) provide complementary

513   tests on the linkage between animals and changes in ocean chemistry.

514   There are two critical considerations in deriving a basinal-scale picture of

515   Ediacaran redox from the Mackenzie and Wernecke Mountains. Foremost, the re-

516   correlation of both Shale Lake and Goz Creek to Sekwi Brook (the type fossil locale in

517   the Mackenzie Mtns.) now leaves all siliciclastic data from the Sheepbed Formation as

518   underlying the first appearance of animals. This reassignment removes the tight

519   stratigraphic correlation called upon by Shen and colleagues between a shift in FeHr/FeT

520   and Aspidella. There is also a methodological difference between Shen et alias and our

521   study that lessens direct comparison. Shen et al. employed an older iron extraction

522   method that does not isolate or quantify Fe$_{carb}$. Iron carbonate phases are perhaps the

523   most quantitatively significant fraction in the Wernecke Mountains sections (Fig. 7b,c)

524   and dominate a vast majority of the Neoproterozoic (Canfield et al., 2008). Fortunately, a

525   small Sheepbed Formation sample set from the same locality as Shen et al. in the

526   Mackenzie Mountains was published by Canfield and colleagues (Canfield et al., 2008)

527   and matches quite nicely with our data from Goz Creek (Fig. 7a).

528

529   5.3 FINGERPRINTING SEDIMENT PROTOLITH

530   A majority of the major element data from the Sheepbed Formation follow the

531   same first order pattern: a constant non-zero rate of change over the lower ~ 100 m of the

532   Sheepbed Formation, an inflection near the MFS, and a return to values similar to the

533   basal Sheepbed Formation at the top of the unit. For instance, Al contents start high at

534   the base of the section and decline modestly over the first 100 m of shale (also leading to

535   lower Fe/Al: Fig. 3d). On the same length-scale, the chemical index of alteration (CIA)

536   for the siliciclastic flux decreases from ~ 75 to 70 over the lower 100 meters, only to

537   climb again over the upper 200 m of the Sheepbed Formation back to ~ 75 (Fig. 5a). This

538   trend obviously holds for [Ca^{2+}], [Mg^{2+}], [Na$^+$], and [K$^+$], as they contribute to the

539   reported CIA, but is also present in TOC (wt%), total P, pyrite (all in Fig. 4) and

540   exceptionally so in total manganese (Fig. 5b). Based on our reading of the Mn data, we

541   interpret all these patterns as reflecting a first order control of water depth on the

542   chemistry of the Sheepbed shale in the Wernecke Mountains.

543   Major and trace element chemistry provides significant insight into both sediment

544   provenance and the addition of exogenous components throughout the entire Wernecke

545   Mountain succession. Because some trace elements are known to carry signatures of

546   carbonate mineral formation and subsequent diagenesis (McLennan, 1993; McLennan

547   and Taylor, 1991), we target elements that are more reflective of siliciclastic deposition

548   and whose utility as tracers (of provenance) will not been compromised by carbonate

549   diagenesis. The elements/oxides most reflective of siliciclastic deposition and resistant to

550   post-depositional alteration are Al_2O_3, K_2O, La, Sc, Zr and Ba (McLennan et al., 2003;

551   Taylor and McLennan, 1985). These elements have often been used in studies of trace

552   element geochemistry of carbonate rocks to detect small amounts of siliciclastic

553   contamination (e.g., (Webb and Kamber, 2000)). For instance, when plotted against

554   Al_2O_3, the elements La, Sc, Zr, Ba and K_2O all show strong positive correlations (R^2

555   ranging from 0.78 to 0.93) indicating detrital siliciclastic sources as the primary control

556   on their behavior. In contrast, elements that are known or suspected to be controlled by

557   carbonate diagenesis such as Sr, exhibits no discernible linear trend when compared to

558   Al.

559   With the effects of lithology carefully screened for, element profiles through the

560   Wernecke Mountains section can be used to identify the influence of sedimentary

561   components with differing provenance. The La/Sc ratio (Fig. 8a), for example, is a useful

562   provenance indicator because under most igneous differentiation processes, La behaves

563   as an incompatible element and typically becomes enriched relative to the more

564   compatible element Sc (Taylor and McLennan, 1985). Beginning with the Sheepbed

565   Formation, La/Sc is poised at a value typically found in shales that sample the average

566   post-Archean upper crust (Fig. 8, 9). Well above the MFS in the Sheepbed Formation, but

567   before fully transitioning into the Sheepbed carbonate, La/Sc sharply increases and

568   remains at elevated and highly variable levels. High La/Sc ratios continue through the

569   June Beds, Gametrail Formation and into the Blueflower siliciclastics. Other measures of

570   incompatible element enrichment, including Zr/Sc (Fig. 8b), K_2O/Al_2O_3 (Fig. 8c) and

571   Ba/Al_2O_3 (not shown) follow similar trends through the succession and reflect the

572   addition of a compositionally-distinct siliciclastic component to both carbonate- and

573   shale-dominated lithologies (Fig. 9). Importantly, this siliciclastic flux is different from

574   that comprising the lower (majority) of Sheepbed Formation deposition, which is

575   compositionally similar to typical post-Archean shale.

576   Comparisons of the various indicators of igneous differentiation show that

577   enrichments in La, Zr, Ba, and K_2O can be attributed to the same component. Plots of

578   La/Sc, Zr/Sc, and Ba/Al_2O_3 versus K_2O/Al_2O_3 are all consistent with two-component

579   mixing between an end-member enriched in incompatible elements and a component that

580   is compositionally similar to the post-Archean average upper crust: PAAS (Fig. 9). These

581   plots also highlight the differences between the lower part of the Sheepbed Formation (<

582   377 m), which approximates PAAS, and the upper part of the succession, which received

583   variable proportions of sediment from compositionally distinct sources.

584   The effects of chemical weathering, diagenesis and sediment provenance on major

585   element chemistry can be evaluated in the Al_2O_3-(CaO^*-Na_2O)-K_2O (or A-CN-K)

586   geochemical system. Figure 10a-c compares the lower (a: < 127 m) and upper (b: 127-

587 368 m) portions of the Sheepbed Formation with siliciclastic samples from the upper

588 stratigraphy (Fig 10c). The Sheepbed siliciclastics show a linear trend cannot be

589 attributed to chemical weathering alone, which typically produces trends parallel to the

590 A-CN join. This indicates that Sheepbed samples have suffered a minor amount of K_2O

591 addition, probably during late diagenesis, and this has pulled samples toward the K_2O

592 apex (Fedo et al., 1995). This sort of K addition is common (Johnston et al., 2012b;

593 Tosca et al., 2010). The > 377 m siliciclastics (namely the Blueflower Formation),

594 however, are chemically distinct from Sheepbed siliciclastics and nearly all are

595 compositionally equivalent to 100% illite. Including Fe and Mg in the analysis of major

596 element systematics (as FeO and MgO, respectively: Fig 10d-f)) points to a similar

597 distinction between Sheepbed and Blueflower siliciclastic sediments. Sheepbed

598 Formation samples are consistent with derivation from a largely upper crustal source,

599 whereas the Blueflower Formation samples are nearly pure illite with variable Fe

600 enrichments (reflected by mixing between the FM apex and Fe-free illite).

601   Major and trace element systematics show that a highly differentiated,

602 compositionally distinct sedimentary component is being added to this basin later in the

603 Ediacaran. Although the highly differentiated character of this component is evident

604 from La/Sc ratios, enrichment in Zr/Sc ratios (Fig. 8) can also result from the addition of

605 highly recycled sediments that tend to concentrate heavy minerals during sediment

606 transport (e.g., (McLennan et al., 2003)). Thus, Zr enrichment might result from the

607 enrichment of zircon in extensively recycled sediments, but enrichments in other heavy

608 minerals, such as ilmenite, titanite or rutile would be expected if this were the case. No Ti

609 enrichment above average crustal levels is observed through the section. The major

610   element characteristics of this component are also inconsistent with the addition of highly

611   recycled sediment and instead reflect the addition of illite to siliciclastics sampling the

612   average upper crust (Figs. 9, 10).

613   Taken together, incompatible element enrichment and major element chemistry

614   indicate that the sedimentary component being added to the upper portions of the

615   Ediacaran stratigraphy is compositionally similar to a acidic (i.e. felsic) volcanic rock. In

616   a geochemical and isotopic study of core samples taken from the Paleoproterozoic

617   Animikie Basin, Hemming et al. (1995) observed many of the same major and trace

618   element characteristics in sampled ash horizons and stratigraphically proximal shale;

619   incompatible element enrichment and major element compositions reflecting illite

620   addition were clear geochemical indicators pointing to the influence of ash deposition on

621   sedimentary geochemistry. However, no volcanic ashes have been identified in the field

622   or through preliminary geochronology efforts in northwestern Canada, nor is there a

623   detrital zircon signature of late Neoproterozoic volcanism.

624   The geochemical effects of this distinct siliciclastic input can be observed across a

625   number of exposure surfaces and major unconformities. As such, this sediment source to

626   the basin is temporally long-lived, likely spanning as much as tens of millions of years

627   depending on the amount of time lost to non-deposition. The effects of ash deposition

628   also provide a means to reconcile the geochemical variability reflected in the upper

629   portions of the stratigraphy in comparison to the Sheepbed Formation. For example,

630   variable uptake and release of Fe by illite can heavily influence highly reactive and total

631   Fe budgets, effectively providing a reactive Fe silicate pool. Typically, the deposition of

632   volcanic ash into seawater results in the rapid alteration of volcanic glass to smectite with

633 minor amounts of zeolite (Weaver, 1989). The exchange of major and trace elements

634 during this process is largely a function of initial composition of volcanic glass. In

635 particular, Fe levels in smectite produced from volcanic ash are likely to be controlled by

636 redox conditions in co-existing pore water as well as igneous controls on composition.

637 Given the multiple lines of evidence in our samples pointing to a highly differentiated

638 composition for volcanic components, the initial volcanic glass and the smectite produced

639 from it are expected to be relatively Fe-poor (e.g., (Christidis and Dunham, 1997;

640 Weaver, 1989)), and thus should reflect a marine Fe sink. Consistent with this

641 interpretation, Figure 10f shows considerable Fe enrichment relative to typical illite

642 compositions, pointing to Fe addition from pore fluids as an important factor in

643 producing relatively Fe-rich illitic compositions. Again, these changes are likely reflected

644 in the composition of the highly reactive Fe pool, in this case biasing FeHr toward lower

645 values, and would also be evident in Fe/Al. Indeed, Fe/Al ratios are elevated in some

646 samples, but overall quite variable, perhaps reflecting the discontinuous nature of

647 volcanic ash addition to Blueflower siliciclastics, or the variable uptake of Fe by illite.

648 A change in the sedimentary sources from the Sheepbed through to the

649 Blueflower Formation is in some fashion a predictable result of late Ediacaran rifting on

650 the western margins of North America. As discussed elsewhere in this volume

651 (Macdonald et al., this issue), overlapping unconformaties and the appearance of

652 conglomerate units in the June beds and Blueflower Formation, along with the rapid

653 facies change to the southwest, suggest late Ediacaran extension in northwestern Canada

654 (MacNaughton et al., 2000) and are consistent with evidence for rift-related volcanism in

655 the southeast Canadian Cordillera (Colpron et al., 2002) and thermal subsidence models

656 (Bond and Kominz, 1984).

657

658 **6. CONCULSIONS AND IMPLCIATIONS**

659   The common approach to determining paleo-redox follows from a strict, up-

660 section reading of iron speciation records, which we complement with adjoining chemical

661 and stratigraphic metrics. Major element chemistry and CIA interpretations from the

662 majority of the Sheepbed Formation point to sedimentation from a common source − a

663 protolith that appears similar to typical post-Archean shale. Iron chemistry of the

664 Sheepbed Formation is diagnostic of an anoxic water column throughout Sheepbed

665 deposition. However, multiple sedimentological factors, such as variable water depth or

666 rapid infilling of accommodation space can distort and/or dilute Fe-based proxy

667 interpretations. Despite this, the continuity between the redox signals from the lower

668 and upper Sheepbed Formation is most conservatively interpreted as reflecting ocean

669 chemistry. The preservation of ferruginous conditions with low sulfide contents lends to

670 depletions in many biogeochemically important trace elements (V, Zn, and Mo; Fig. 6).

671 Nevertheless, the MFS nicely captures a pulse of biological activity recorded by a local

672 maximum of TOC and P contents. Manganese concentrations are elevated in the MFS,

673 which is interpreted as a shoaling of the chemocline with ensuing contributions from an

674 oxygenated surface ocean. Sulfur isotope records fit with the coordinated sequence

675 stratigraphic-geochemical story, suggesting that by the time the high-stand tract was

676   established, sulfate reduction was restricted to marine sediments and appears to be

677   diffusion (sulfate) limited.

678   Whereas the Sheepbed Formation preserves a clean snapshot of the immediately

679   post-Marinoan ocean, the later Ediacaran Blueflower and Risky formations capture a

680   more muddled picture of the latest Proterozoic. Understanding these units is paramount,

681   however, given the occurrence of Ediacaran fossils through this interval. That is, linking

682   geochemistry to the fossil record in northwestern Canada provides a unique opportunity

683   to assay the controls on early metazoan evolution. The high degree of intrinsic variability

684   in almost every major geochemical metric – $\delta^{13}C_{carb}$, $\delta^{13}C_{org}$, FeHr/FeT, CIA – makes

685   extracting an assuredly primary signal from the Blueflower and Risky formations

686   difficult. However, these data show no clear sign of oxygenation. In fact, the signature

687   of reducing, anoxic conditions is fully consistent with the inferred state of the oceans in

688   the Sheepbed Formation. In more detail, we note that later Ediacaran metazoan

689   preservation in the Wernecke Mountains is largely in detrital siltstone and sandstone (not

690   clays) and thus our data closely stratigraphically bracket the actual preservation of

691   fossiliferous material. Those bounding clays carry a composition similar to acidic

692   volcanic rock. It remains unclear how this change in sediment protolith could influence

693   local taphonomic or ecological conditions, but is testable in other Ediacaran, fossil-

694   bearing strata. The most parsimonious interpretation of the late Ediacaran points to

695   anoxia at depth up through the Precambrian – Cambrian boundary, with an uncertain tie

696   between animals and basinal redox.

697   The late Neoproterozoic response of redox sensitive proxies discussed above also

698   provides a test of whether the changes preserved in $\delta^{13}C_{carb}$ of the Gametrail Formation

699   are a local phenomenon (authigenesis) or global (f_{org}) in nature. If a global oxidation

700   event (f_{org} is robust), then the chemistry of the upper Sheepbed Formation should look

701   markedly different to that of the Blueflower Formation (the bounding units). This is in

702   stark opposition to most, if not all geochemical measures, which point to the infidelity of

703   extracting and interpreting f_{org}. Although siliciclastic data is not available through the

704   putative Shuram excursion in the Gametrail Formation, the lack of a profound change in

705   oxidation state between the Sheepbed and Blueflower formations by necessity carry the

706   same implications for the Gametrail $\delta^{13}C_{carb}$ excursion – that is, that it does not record a

707   quantitatively significant consumption of available oxidants.

708   The joined redox history of Ediacaran basins globally does show indications of

709   localized oxygenated conditions (e.g. Newfoundland, the Eastern European Platform, and

710   perhaps S. China), but some regions maintain evidence for anoxia throughout

711   (northwestern Canada and portions of S. China). Some level of local control is quite

712   possible, given variable nutrient inputs and resulting organic carbon flux to sediments

713   (Johnston et al., 2010), but in each case, these basinal signals start with the dissolved

714   oxygen load adopted from a well-mixed atmospheric reservoir of O_2. Important to

715   consider in these reconstructions is the remarkably low level of sulfide in siliciclastic

716   units, indicating the possibility of a depleted seawater sulfate pool, but more likely

717   suggesting minimal remineralization by sulfate reduction in sediments as a function of

718   limited organic carbon delivery (Johnston et al., 2010). More surprising than the

719   heterogeneous basinal chemistry is the lack of a strong oxygenation signal. Classic

720   storylines call upon a major oxygenation in the Ediacaran; one that appears to be missing

721   from many Ediacaran records.

722   If the classic call for a step-function change in atmospheric oxygen is not the

723   singular motivating force for the appearance of the first metazoans (Nursall, 1959), then

724   one of a number of issues surface. First, the patchy record of Ediacaran biota in basins

725   that all carry differing geochemical signals suggests that there may exist a more nuanced

726   physiological or ecological feature that is limiting or allowing for the move to

727   macroscopy (Johnston et al., 2012b). As much of the redox data falls near the

728   geochemical distinction between oxic and anoxic, perhaps an increase in pO_2 is in fact

729   housed in the Ediacaran, but that the rate or absolute magnitude of change is beneath that

730   which we can diagnose. Resurfacing is the reality that our geochemical thresholds for

731   'telling pO_2' lack the clarity to distinguish between 0.5 and 5% PAL, and it may be this

732   exact range that is critical for animals (Sperling et al., 2013). Taken together, the driver

733   for animal evolution remains elusive and although correlations can be drawn to

734   geochemical records, the interplay between biology and chemistry is either absent

735   (Butterfield, 2009) or more subtle than the sharpness of our geochemical tools.

736   These results and the preponderance of accumulating data argues against a great

737   Ediacaran increase in atmospheric oxygen. Perhaps rather than targeting the literal

738   reading of redox sensitive elements and counting electrons buried, as is done here and

739   elsewhere, the exercise should be extended to more formally include 'time,' bed-by-bed

740   analysis of fossiliferous sections, and consider the sheer volume of accumulating

741   sediments, not only the chemical content of that sediment package. Together, this may

742   hold the key to diagnosing both small changes in oxidation state and reveal the trigger for

743   metazoan life.

744

Figure 1: A geological map and stratigraphic correlation in northwestern Canada and Alaska (from (Macdonald et al., this issue)). Stratigraphic abbreviations are K: Keele, I: Icebrook, In: Ingta, H: Hayhook, and Ri: Risky, Gt; Gametrail, J/June; June beds, SC: Sheepbed carbonate and BB; Backbone. Also included is a classic lithological scale. The localities on the map correspond to the four sections presented as stratigraphic columns below (A-D). The primary target of this work is the Wernecke Mountains (column B), with the prominent fossil locality to the east at Sekwi Brook.

Figure 2: Carbon isotope chemostratigraphy from the Wernecke and Ogilvie Mountains. At left are data (carbonate carbon in red - scale at bottom; organic carbon in blue (Sheepbed Formation) and yellow (upper stratigraphy) – scale at top) from the Goz Creek section with data from the Ogilvie Mountains (same key) at right. Correlations and geologic/sedimentological key are discussed in Fig 1, with carbonate data from (Macdonald et al., this issue). Notably, Macdonald and colleagues describe the Blueflower Formation carbonate as nodular/banded cements and as a result, interpreted

768   these lenses as authigenically influenced. Inset (bottom right) are cross-plots of carbonate

769   versus organic carbon (see text for discussion).

770   **Figure 3:** Iron speciation and Al data through the Wernecke Mountains section at Goz

771   Creek. Both metrics of oxidation state (FeHr/FeT: red, bottom scale) and sulfide contents

772   (Fe_{py}/FeHr: blue, top scale) are plotted to the left. Data are in backdrop and binned

773   averages and standard deviations (per 50 meters) are the open, bold circles. Two

774   calibration lines are included for Fe-speciation: 0.22 and 0.38 (see text). Iron to

775   Aluminum follows the same scheme, with values from the carbonate (in black) following

776   the top X-axis scale. The remaining Fe/Al data (in white) is tracking the bottom X-axis

777   scale. The inset cross-plots represent a suite of geochemical relationships that describe

778   redox conditions in the Wernecke Mountains.

779   **Figure 4:** Sulfur, phosphorus and organic carbon data from the Wernecke Mountains. At

780   left are chemostratigraphic profiles with the data in the backdrop and the 50 m averages

781   as open circles in front (same as in Fig. 3). Open purple symbols correspond to sulfide

782   sulfur whereas closed purple circles are sulfates. Also included at right are three cross

783   plots relating the isotopic composition of sedimentary pyrite to TOC and iron speciation

784   metrics and pyrite contents.

785   **Figure 5:** Two chemical metrics critical to the interpretation of the Wernecke Mountain

786   siliciclastic fluxes. CIA aids in the interpretation and impact of chemical weathering on

787   source sediments, but can also be affected by diagenetic influences on Ca^{2+}, Na^+ or K^+.

788   Second is a notable enrichment at Mn at the MFS corresponds to an inflection in the CIA

789   profile. Both CIA and Mn show a large range of values in the Blueflower, but a coherent

790   signal in the Sheepbed Formation. See text for discussion.

791   **Figure 6:** Trace element data from the Wernecke Mountains. Data are broken out

792   stratigraphically. Notably, the Sheepbed and Blueflower formations are both siliciclastic

793   rich.   Notice the horizontal scale break in frames A and B.   Notably absent are

794   enrichments in Mo, which are not plotted here and never exceed 3 ppm.

795   **Figure 7:** Revisiting the Fe speciation data from the Wernecke (this study (a-c)) and

796   Mackenzie Mountains from previous authors (Canfield et al., 2008; Shen et al., 2008b).

797   The leftmost panel presents all the FeHr/FeT data from this study and the published

798   literature. The Shen et al. data employ an older extraction method and are from the same

799   locality as the Canfield et al. Mackenzie Mountain data. The center panel (b) presents the

800   fractions of highly reactive iron residing in carbonate and pyrite (this study), whereas the

801   right hand frame (c) relates iron carbonate to total Fe.

802   **Figure 8:**   Various ratios of incompatible-compatible element fractionation to diagnose

803   siliciclastic protolith. A full description of their interpretation is included in the text.

804   **Figure 9:** Cross-plots of the metrics presented stratigraphically in Figure 8, with the

805   extension to include CIA (Fig. 5). The key is included in Frame a, with PAAS being

806   post-Archean average shale (Taylor and McLennan, 1985). From each frame is it clear

807   that the Sheepbed Formation appears to consistently resemble PAAS in composition,

808   whereas above ~ 368 m (which includes the Upper Carbonate unit of the Sheepbed) the

809   siliciclastic flux reflects mixing between a PAAS-like component and a differentiated

810   component such as a volcanogenic ash.

811 **Figure 10:** Classic ternary diagrams used to determine the addition and loss of particular

812 elements over the life history of the sediment. Data are split by stratigraphic depth (see

813 frames). Frames a-c (with key in frame b) most notably illustrates K addition, pulling the

814 data in Frame a-b off the CN-Al contour. In frames d-f, contributions from Fe and Mg

815 are evaluated, leading to the conclusion that the siliciclastics in the upper 400 m of

816 stratigraphy appears similar to pure illite with variable Fe enrichment.

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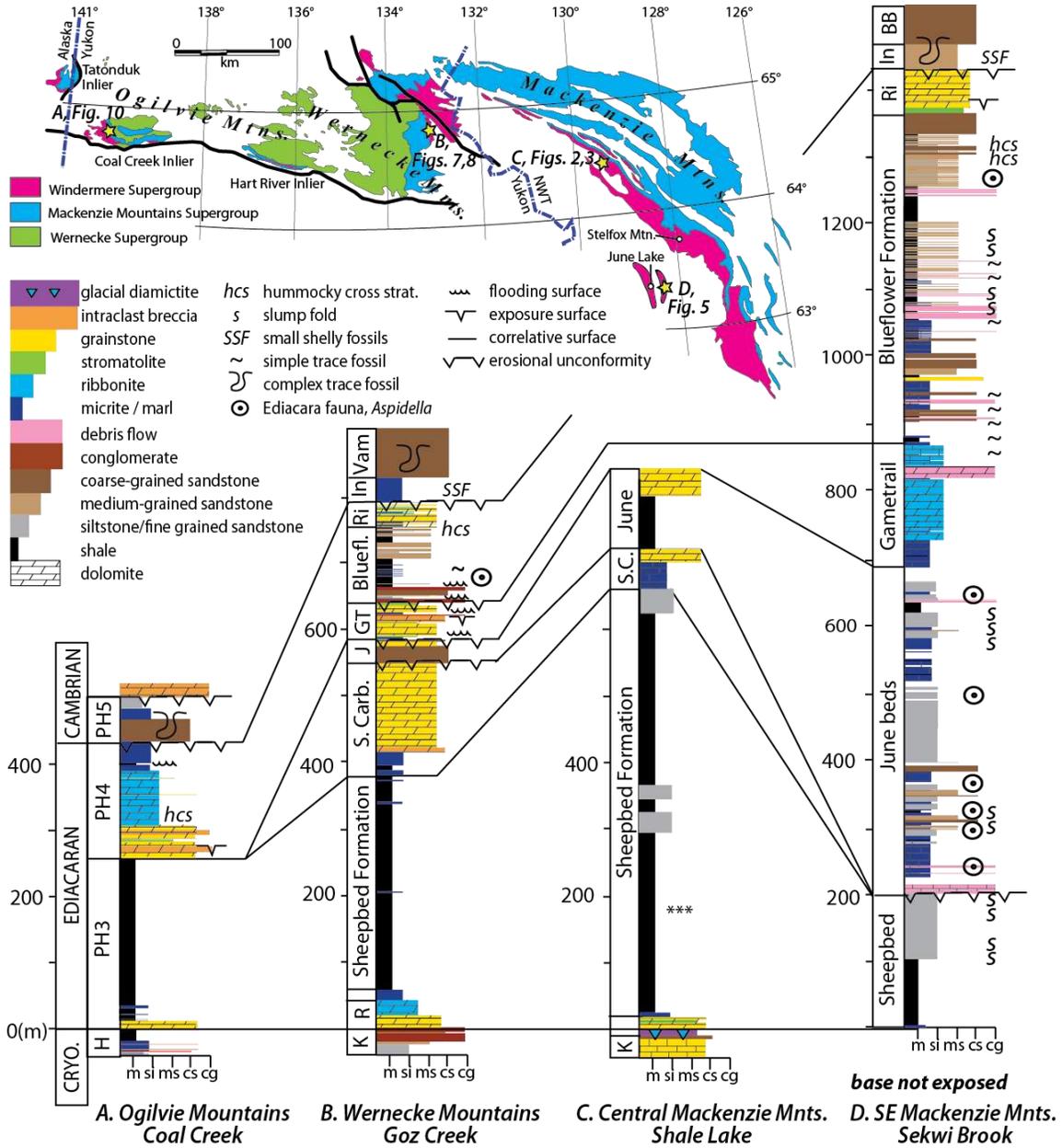1051   Acta, 64(9): 1557-1565.
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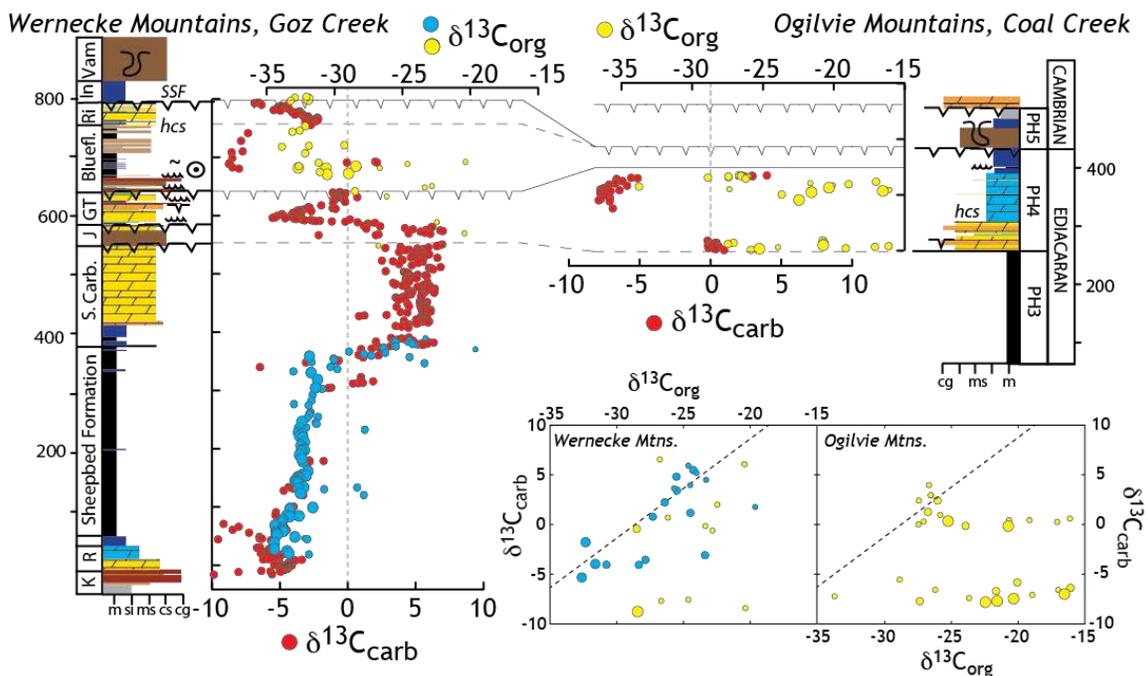1055   66(15A): A847-A847.

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1059 **Figure 1:** A geological map and stratigraphic correlation in northwest Canada and Alaska

1060 (from (Macdonald et al., in review)). Stratigraphic abbreviations are K: Keele, I:

1061 Icebrook, Ingta and R: Risky. The localities on the map correspond to the four sections

1062 presented as stratigraphic columns below. The primary target of this work is in the

1063 Wernecke Mountains (column B), with the prominent fossil locality to the east at Sekwi

1064 Brook.

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1068 **Figure 2:** Carbon isotope chemostratigraphy from the Wernecke and Ogilvie Mountains.

1069 At left are data (carbonate carbon in red - scale at bottom; organic carbon in blue

1070 (Sheepbed Fm.) and yellow (upper stratigraphy) – scale at top) from the Goz Creek

1071 section with data from the Ogilvie Mountains (same key) at right. Correlations and

1072 geologic/sedimentological key are discussed in Fig 1, with carbonate data from

1073 (Macdonald et al., in review). Inset (bottom right) are cross-plots of carbonate versus

1074 organic carbon (see text for discussion).

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1078 **Figure 3:** Iron speciation and Al data through the Wernecke Mountains section at Goz

1079 Creek. Both metrics of oxidation state (FeHr/FeT: red, bottom scale) and sulfide contents

1080 (FePy/FeHr: blue, top scale) are plotted to the left. Data are in backdrop and binned

1081 averages and standard deviations (per 50 meters) are the open, bold circles. Two

1082 calibration lines are included for Fe-speciation: 0.22 and 0.38 (see text). Iron to

1083 Aluminum follows the same scheme, with values from the carbonate (in black) following

1084 the top X-axis scale. The remaining Fe/Al data (in white) is tracking the bottom X-axis

scale. The inset cross-plots represent highly reactive to total iron and pyrite iron (and

1086 both to total organic carbon), and the ratios of aluminum to iron and to titanium.
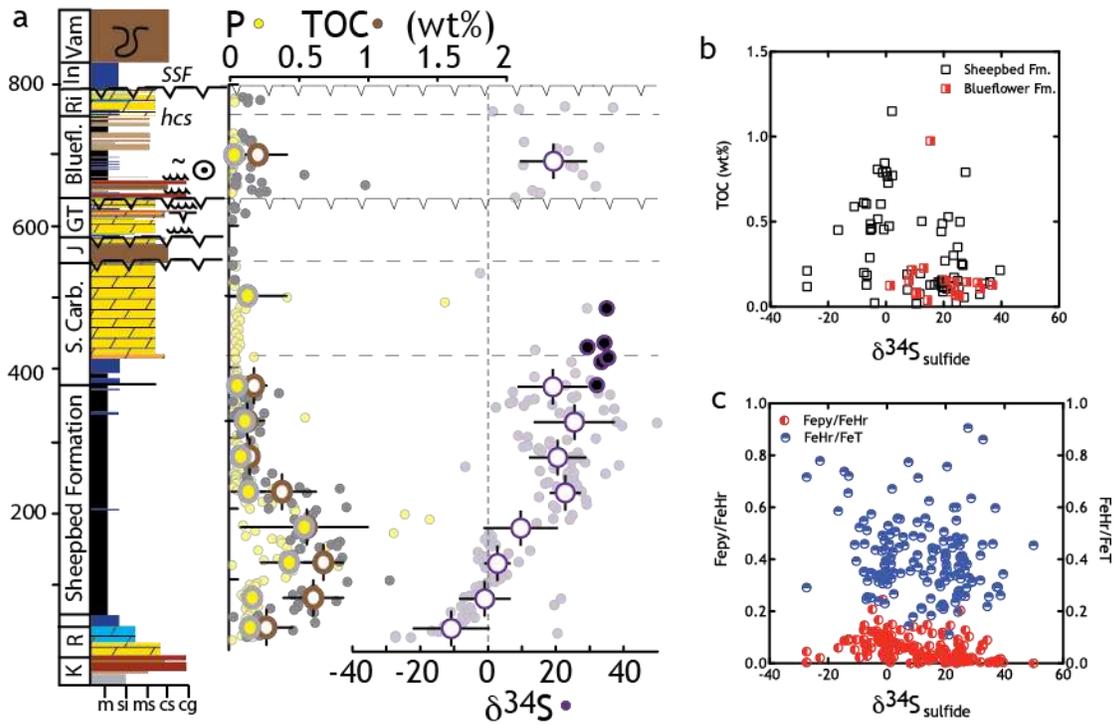

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Figure 4: Sulfur, phosphorus and organic carbon data from the Wernecke Mountains. At left are chemostratigraphic profiles with the data in the backdrop and the 50 m averages as open circles in front (same as in Fig. 2). Open purple symbols correspond to sulfide sulfur whereas closed purple circles are sulfates. Also included at right are two cross plots relating the isotopic composition of sedimentary pyrite to TOC and iron speciation metrics.

1102 **Figure 5:** Two chemical metrics critical to the interpretation of the Wernecke Mountain

1103 siliciclastic fluxes. CIA aids in the interpretation and impact of chemical weathering.

1104 Second is a notable enrichment at Mn at the MFS corresponds to an inflection in the CIA

1105 profile. Both CIA and Mn show a large range of values in the Blueflower, but a coherent

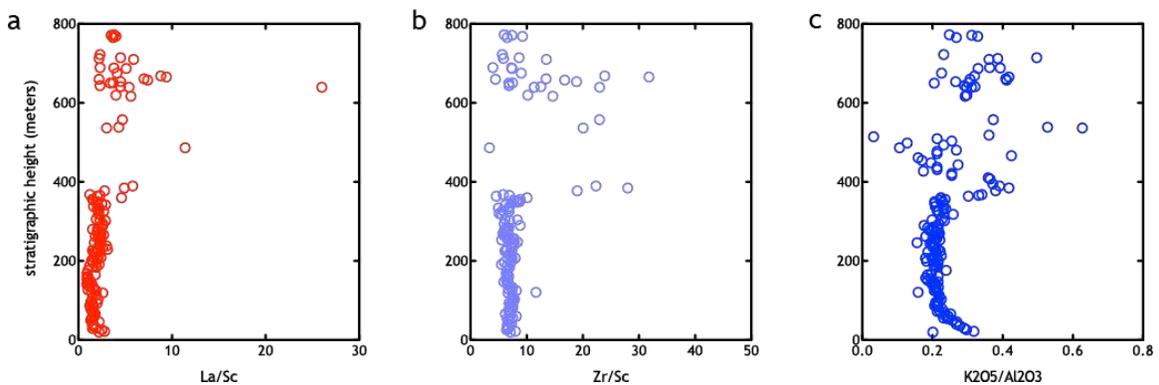1106 signal in the Sheepbed Formation. See text for discussion.

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1111 **Figure 6:** Classic major element plots to diagnose siliciclastic protolith. A full

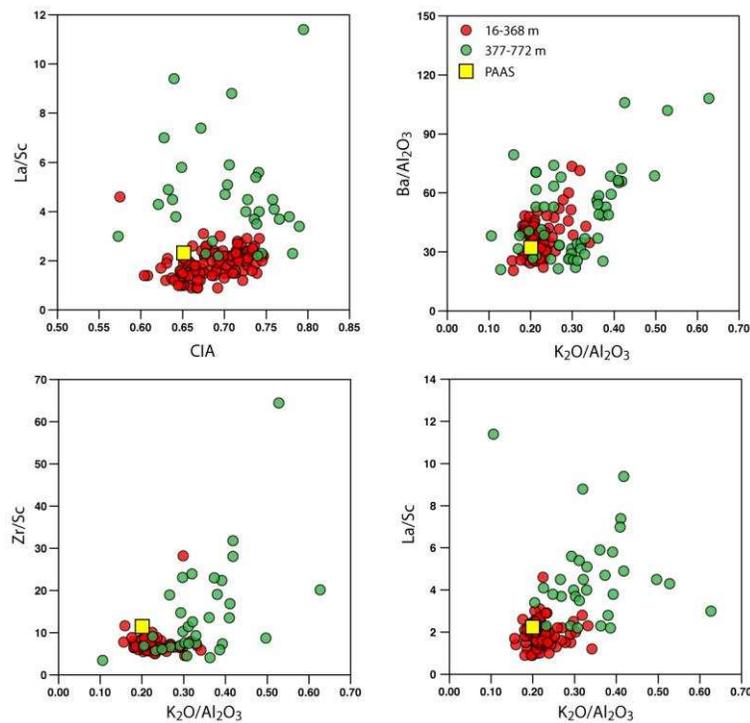1112 description of their interpretation is included in the text.

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Figure 7: Cross-plots of the metrics presented stratigraphically in Figure 6, with the extension to include CIA. The key is included in Frame b, with PAAS being post-Archean average shale. From each frame is it clear that the lower Sheepbed Formation appears to be quite typical in composition, whereas above ~ 368 m (which includes the Upper Carbonate unit of the Sheepbed) the siliciclastic flux more closely resembles a volcanogenic ash.
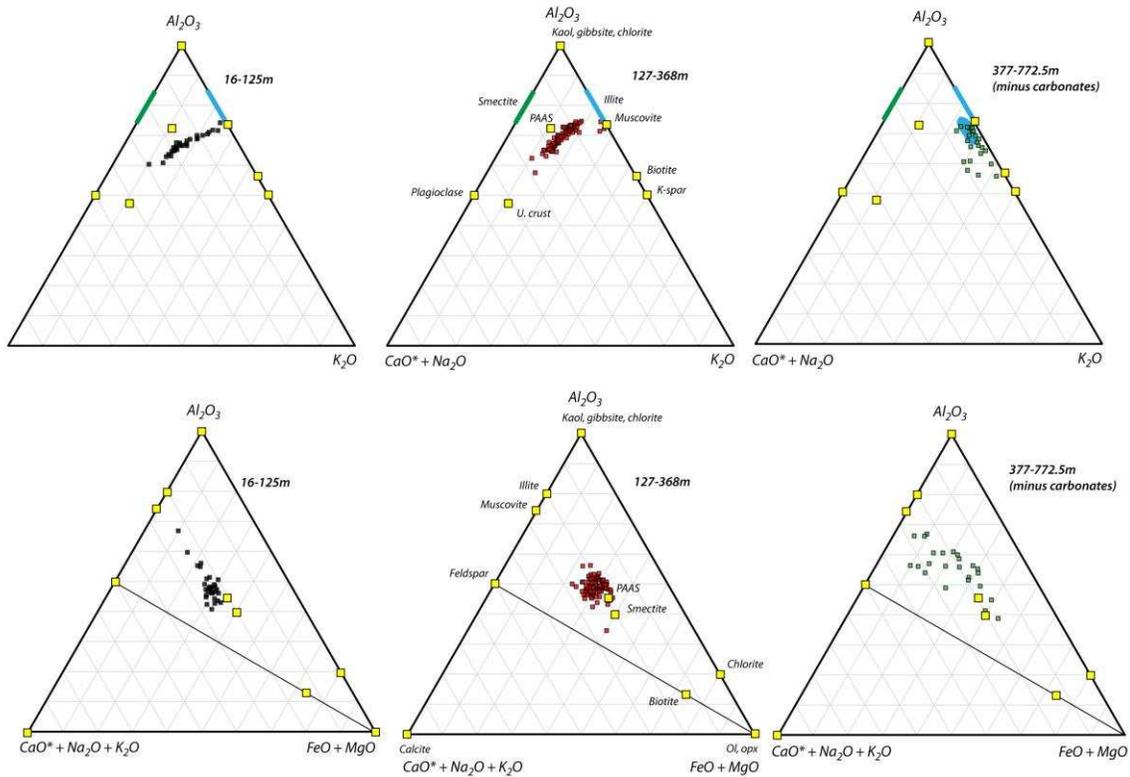

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1131   **Figure 8:** Classic ternary diagrams used to determine the addition and loss of particular

1132   elements over the life history of the sediment. Frames a-c (with key in frame b) most

1133   notably illustrates K addition, pulling the data in Frame a-b off the CN-Al contour. In

1134   frames d-f, contributions from Fe and Mg are evaluated, leading to the conclusion that the

1135   siliciclastics in the upper 400 m of stratigraphy appears similar to pure illite with variable

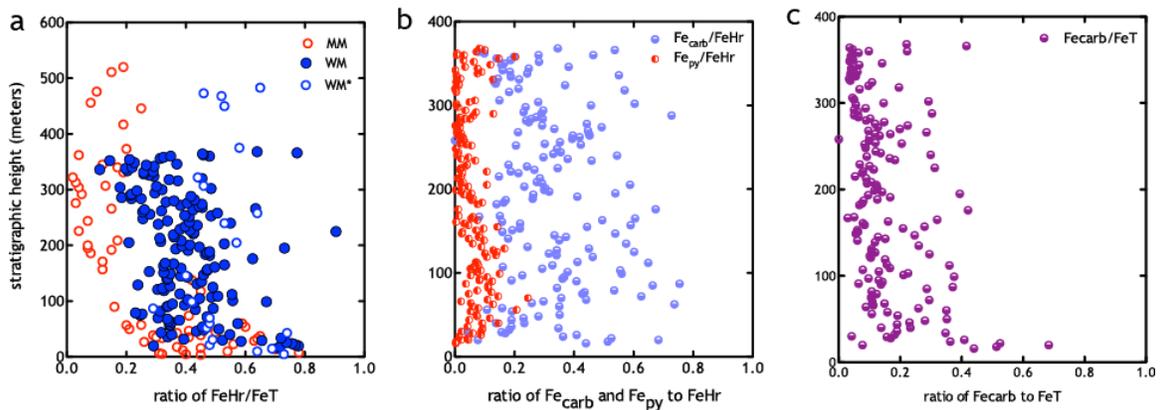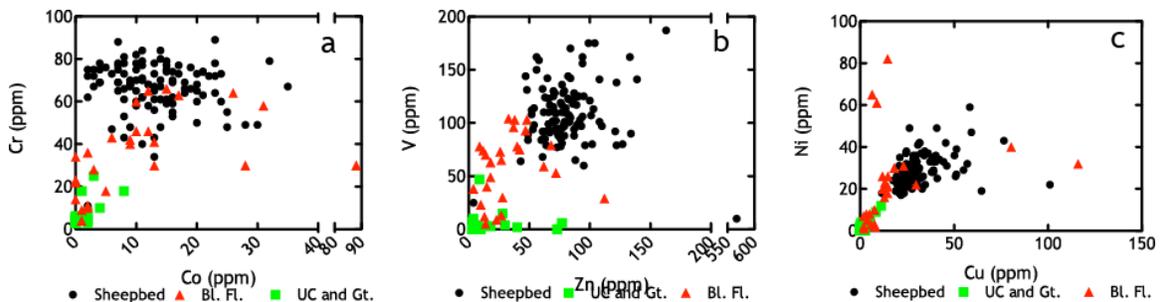1136   Fe.

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1144 **Figure 9:** Revisiting the Fe speciation data from the Wernecke (WM) and Mackenzie

1145 Mountains (MM)(Canfield et al., 2008; Shen et al., 2008). The leftmost panel presents

1146 all the FeHr/FeT data from this study and the published literature. Only the MM data are

1147 from older methods, where both WM sample sets used methods outlined in (Poulton and

1148 Canfield, 2005). The center panel presents the fractions of highly reactive iron residing in

1149 carbonate and pyrite, whereas the right hand frame relates iron carbonate to total Fe. See

1150 text for more discussion.

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1152 **Figure 10:** Trace element data from the Wernecke Mountains. Data are broken out

1153 stratigraphically and low values for the Upper Carbonate unit and Gametrail Formation

1154 (UC and Gt.) reflect a lithology change. The Sheepbed and Blueflower (Bl. Fl.)

1155 formations are both siliciclastic rich. Notice the horizontal scale break in frames A and

1156 B. Notably absent are enrichments in Mo.

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1162   Canfield, D.E. et al., 2008. Ferruginous conditions dominated later neoproterozoic
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