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129 **Overcoming recycling barriers to transform global phosphorus management**

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Abstract

The global phosphorus challenge arises from the uneven distribution of phosphorus resources, environmental impacts from phosphorus losses and unsustainable linear management. Despite progress in advanced phosphorus recycling, less than 1% of secondary phosphorus resources produced globally are recycled. In this Review, we comprehensively explore global barriers to phosphorus recycling. Manure (15–20 Mt P yr⁻¹), mining and fertilizer industry waste (6–12 Mt P yr⁻¹), wastewater (~3.7 Mt P yr⁻¹) and food waste (~1.2 Mt P yr⁻¹) are the major secondary phosphorus resources worldwide. In addition, accumulated legacy phosphorus in soil and sediment comprises a combined stock of more than 3,200 Mt P. Phosphorus mismanagement and losses cost stakeholders US\$265 billion yearly, yet substantial barriers to phosphorus recycling remain. Key challenges to be overcome include low competitiveness of recycled phosphorus products, complex waste handling, limited legacy phosphorus recovery and fragmented collaboration among stakeholders. A shift is needed toward an integrated, systems-based approach that simultaneously addresses technical, economic and societal challenges. Transdisciplinary strategies and research will advance phosphorus recycling and development of a sustainable, circular phosphorus economy. Incorporating the perspectives of diverse stakeholders will help drive increasingly sustainable phosphorus management.

Key points

- Mineral phosphorus dependency, uneven global distribution, eutrophication and linear nutrient management are fundamental and deeply interconnected challenges in managing phosphorus.
- Efficient phosphorus use and recycling are essential to closing the phosphorus cycle, but numerous barriers stand in the way of achieving this goal.
- Conflicting objectives among stakeholders is a key barrier to developing and implementing effective strategies for sustainable phosphorus use.
- Successful strategies for circular management of phosphorus require improved communication, interdisciplinary research, and transdisciplinary processes that incorporate the needs of all stakeholders.
- Inclusive policies are vital to align incentives, foster collaboration and promote sustainable phosphorus-use practices.

[H1] Introduction

Phosphorus (P) is crucial for supporting food and industrial production worldwide. Global P demand currently totals 26.5 Mt P yr⁻¹, driven by food production (~80% for fertilizers and 6% for food additives) and industrial applications (14%)^{1,2}. Most consumed P originates from phosphate rock^{3,4}, a finite resource with known high-quality reserves that are expected to be depleted within the next few centuries⁵. In addition, more than 85% of global P deposits are concentrated in just five countries, limiting equitable access to this critical resource, particularly during periods of geopolitical uncertainty^{6,7}. Furthermore, 80–95% of all mined P is lost owing to inefficient P management^{8,9}. Much of this loss occurs on farmers' fields, where unconsumed P from past inputs accumulates as legacy P in soil or freshwater sediment. Excess P in aquatic environments causes severe environmental damage. Globally, P mismanagement costs stakeholders approximately US\$265 billion annually (ref.⁶).

In response to these sustainability, equity and environmental challenges, efforts to establish a circular phosphorus economy are gaining momentum¹⁰. For example, China is reducing P waste streams by using industrial sludge as fertilizer¹¹. Brazil's National Fertilizer Program is diminishing P inputs in agriculture and reliance on imports by establishing governance and monitoring tools, promoting research and innovation and exploring domestic sedimentary phosphate basins^{12,13}. The European Union's Circular Economy Action Plan¹⁴, adopted in 2015, provides an initial framework for nutrient recycling within its territory. Globally, the potential for P recycling is huge, with the amount of P trapped annually in recyclable resources comprising 143% of the current (2024) yearly P demand¹.

Cascading innovations are focused on the recovery of P from diverse waste streams. The economic, societal and environmental benefits of P recovery¹⁵, the potential of secondary sources of P to partially substitute mineral P in the production of fertilizers^{16,17} and a role for recycling in closing nutrient loops¹⁸ highlight the importance and advantages of P recycling. Concurrently, technological breakthroughs have facilitated the transition of large-scale operations to the use of secondary P in manufacturing industrial products^{16,17}. Despite notable progress in infrastructure^{19,20}, policy development^{21,22} and recycling technologies^{17,18} for circular nutrient management, challenges in P management persist. Contextual differences, disciplinary fragmentation and limited stakeholder coordination hinder the development of well-defined pathways to achieve greater phosphorus circularity worldwide^{10,23}, reflecting its nature as a wicked problem.

In this Review, we explore the reasons why P recycling from secondary sources remains limited. We discuss a range of technical, economic and societal barriers to global P recycling^{24–26} and argue that fostering transdisciplinary collaborations is essential to improving P sustainability worldwide, precisely because such collaborations are best positioned to align the diverse interests of stakeholders across sectors and disciplines.

[H1] The need to improve phosphorus management

The ideal P value chain is circular, but in practice it is predominantly linear. The cycle begins when P mined from phosphate rock is used to produce fertilizers and other goods. After manufacturing, P is consumed by humans, crops or animals, but substantial amounts of P are lost to secondary P waste streams at every stage of the P life cycle. Phosphorus losses to the environment are also pervasive, leading to P accumulation in soil and sediment, where it can contribute to environmental degradation^{8,27–29}.

Furthermore, as access to phosphate rock reserves diminishes, both economically and geographically, the risks to global food production also rise^{30,31}. These challenges highlight the need to develop strategies for recovering and recycling P. The EU has some of the most advanced technology, data availability and legislation relating to P recycling, yet, even in the EU, the P cycle remains essentially linear (Fig. 1).

[H2] Geopolitical dependence on primary phosphorus imports

Most regions worldwide rely heavily on P imports from a limited number of countries (Morocco alone contains ~67% of the world's known reserves^{2,32,33}), creating pronounced geopolitical vulnerabilities in food security^{34,35}. Political, economic and environmental disruptions can lead to global price shocks³⁵. For example, in 2008, phosphate fertilizer prices spiked by 800%, mostly due to rising energy costs and geopolitical trade policies³⁶, generating supply constraints and threatening agricultural production and food security in several parts of the world^{37,38}. More recent supply chain disruptions include the COVID-19 pandemic³⁹, the Ukraine-Russia conflict⁴⁰, reductions in P fertilizer exports by China and Russia^{34,41}, and trade wars between key stakeholders (such as China and the USA⁴²). The negative impacts of these events disproportionately affected countries in Africa⁴³. Asymmetric risks associated with primary P dependence underscore the need for diversified P sources and more resilient supply chains.

[H2] Agricultural needs and phosphorus wastage

Approximately 90% of all mined P resources are used in fertilizers^{44,45}, and global P use since the Green Revolution has increased more than six-fold^{1,46}. However, quantifying global P flows and stocks is challenging owing to major regional differences in P resource availability, prices and efficiency of use, all of which are shaped by economic, political and environmental factors^{10,46}. The most comprehensive accountings of European and global P mass describe conditions in 2005⁴⁷ and 2009³⁸, respectively. Both assessments highlight increasing agricultural demand and pervasive losses throughout the P life cycle.

Where intensive agricultural practices coincide with widespread access to mined P resources, high P input and legacy P often result⁴⁸. China and Brazil use substantial amounts of mineral fertilizer (more than 32 kg P ha⁻¹ of cultivated land yr⁻¹)⁴⁸ as P input compared with 5–10 kg P ha⁻¹ yr⁻¹ of manure, and P removal through crop harvests is also high (24 and 26 kg P ha⁻¹ yr⁻¹, respectively). Conversely, the low agricultural productivity in most African countries is partially attributed to low P input⁴⁹ (Fig. 2).

Most European countries have moderate or low mineral fertilizer inputs ($0\text{--}16\text{ kg P ha}^{-1}\text{ yr}^{-1}$) and moderate or high manure inputs ($8\text{--}24\text{ kg P ha}^{-1}\text{ yr}^{-1}$), except for the Netherlands and Belgium, where manure application is remarkably high (65 and $35\text{ kg P ha}^{-1}\text{ yr}^{-1}$, respectively). In the USA, mineral fertilizer application is moderate ($8\text{--}16\text{ kg P ha}^{-1}\text{ yr}^{-1}$), manure input is low ($0\text{--}8\text{ kg P ha}^{-1}\text{ yr}^{-1}$) and crop removal is moderate ($8\text{--}16\text{ kg P ha}^{-1}\text{ yr}^{-1}$)⁴⁸. Importantly, P losses are extensive almost everywhere. For example, the terrestrial P surplus (that is, P that was applied to fields but not exported by crop harvest) in the USA in 2012 is estimated at 1.85 Mt P, with many agricultural areas exhibiting high surpluses, particularly in the upper Midwest⁵⁰.

[H2] Environmental impact of phosphorus mismanagement

Phosphorus mismanagement results in massive environmental degradation. Some mismanaged P ends up in water bodies and causes eutrophication^{51,52}, resulting in harmful algal blooms, dead zones and biodiversity loss. Annually, approximately 1.5 Mt of anthropogenic P are lost to freshwater systems⁵³. The EU released a Water Framework Directive more than 25 years ago⁵⁴, yet 60% of lakes in the region do not meet the directive's 'good' standard. Pervasive eutrophication and ineffective water restoration highlight the urgent need to rethink P use practices and develop sustainable P management solutions.

By recovering P from nutrient-rich waste streams, such as wastewater and animal manure, using methods such as chemical precipitation, advanced composting and biochar production, P recycling from agricultural systems can be increased^{1,17,31}. Furthermore, recovering recalcitrant P stocks in soil and sediment can be crucial to ensuring global P circularity.

These geopolitical, wastage and environmental aspects reveal a global P landscape marked by stark regional contrasts, ranging from over-application and environmental losses to chronic underuse owing to limited access to P resources. Addressing these imbalances is essential to improving agricultural productivity and the sustainability of P use worldwide.

[H1] Recovery of secondary phosphorus

Improving P circularity is indispensable to overcoming the geopolitical, agricultural and environmental challenges associated with P mismanagement^{8,10}. Achieving this goal requires optimizing the use of primary P resources (mined phosphate rock), utilizing P stocks accumulated in soil^{27,28} and sediment⁵⁵, and enhancing P recycling from P-rich secondary sources, including waste from the mining and fertilizer industries^{1,56}, livestock manure¹⁶, wastewater and sewage sludge^{57,58} and food waste⁵⁹ (Table 1).

[H2] Mining and fertilizer industry waste

The mining industry produces substantial amounts of secondary P contained in byproducts and waste from rock mining ($1.1\text{--}3.0\text{ Mt yr}^{-1}$), and phosphoric tailings generated during rock beneficiation ($2.3\text{--}4.6\text{ Mt yr}^{-1}$), which together represent 12–16% of total P mined globally^{2,9}. Although P recovery from mining waste using pyrolysis, leaching and precipitation have been proposed, these approaches are complex and expensive^{60,61}, and large-scale recovery is currently not viable⁶². Typically, mining waste is landfilled,

and usually covered with vegetation to reduce environmental risks such as erosion, dust production and P runoff.

The fertilizer industry also generates P-rich waste such as P slag ($\sim 1.8 \text{ Mt P yr}^{-1}$) and ferrophosphorus ($\sim 0.3 \text{ Mt P yr}^{-1}$)¹. In addition, $\sim 300 \text{ Mt}$ of phosphogypsum (containing 6–9.8 Mt P) is produced annually by acidifying phosphate rock^{1,63}. Although phosphogypsum has agronomic applications, 58% is landfilled⁶⁴ and 28% is discharged into the sea⁶⁵, comprising a global stockpile of 60–160 Mt of P^{56,65}, of which only 14% is further treated⁵⁶.

[H2] Livestock manure

Livestock manure is the largest secondary P resource ($15\text{--}20 \text{ Mt P yr}^{-1}$)^{1,47}, accounting for more than 50% of the annual secondary P generated worldwide. In the EU, $\sim 2.2 \text{ Mt P yr}^{-1}$ comes from manure, of which $\sim 90\%$ is directly applied to agricultural land^{16,66,67}. However, a substantial portion of manure is difficult to recover because grazing animals deposit it directly onto grasslands⁶⁷. Intentional application is often concentrated near livestock production areas, serving more as a means of waste disposal than as a targeted agronomic strategy¹⁶. This practice can lead to the accumulation of soil legacy P. Moreover, manure typically has a low N:P ratio, which can contribute to P overapplication as farmers prioritize addressing crop nitrogen requirements. In China, $\sim 2.14 \text{ Mt P yr}^{-1}$ from manure is applied on agricultural land⁶⁸, representing 26% of the country's total P demand, but this is only $\sim 50\%$ of the P that could be harnessed from manure in China⁶⁹.

Owing to the low P concentration in manure ($<1\%$ of P per fresh weight), concentrating P from this source can improve P recovery and subsequent recycling, thereby facilitating efficient handling, transportation and application^{16,67}. The water content and volume of manure can be reduced using non-thermal methods, such as flocculation, settling, screw pressing, belt filtration centrifugation, and dissolved air flotation after anaerobic digestion^{70,71}, but technical, logistic and financial challenges remain.

Biogas production from manure is common in developed countries, yielding a digestate containing $\sim 2\% \text{ P}$ ⁷². To further concentrate P in fresh manure and its digestate, solid–liquid separation techniques have proved effective⁷³. About 70–75% of P can be recovered in the solid fraction without flocculation, whereas flocculation increases the recovery rate to 80–90%.

[H2] Food and biorefining wastes

As much as 1.2 Mt P could be recycled annually from farming, food manufacturing, biorefining, and consumer wastes and byproducts¹. However, resource heterogeneity complicates recycling from food waste²⁵. Recovery techniques such as composting, anaerobic digestion and fermentation are the most common strategies for nutrient recovery from food waste. Other methods, such as incineration, are promising, but come with their own drawbacks such as increased pollution^{59,74}.

Industrial food manufacturing is a major source of food waste, accounting for almost 50% of the food waste in the entire supply chain in the UK⁷⁵. This waste often takes the form of useful materials such

as whey and starch residues⁷⁵. Meat and bone meal (a slaughterhouse byproduct) has high P content (3–5%) and are already used as a P fertilizer in some countries, but concerns about pathogen risk and public perception still limit its use in other countries⁷⁶.

Biorefinery residuals, such as waste from bioethanol production or breweries, are also rich in P. A prominent example is distillers' dried grains with solubles, a byproduct of corn ethanol production, of which ~38 Mt yr⁻¹ is produced in the USA alone. With a typical P concentration of ~1% of dry weight^{74,76}, this amounts to ~0.38 Mt P yr⁻¹, making it one of the largest flows of concentrated organic secondary P in North America⁷⁷. Although commonly used in livestock feed, which aids P recycling to some extent, the spatial disconnect between production and consumption sites requires complex transport logistics and high costs, often resulting in ineffective P recovery⁷⁷.

Post-consumer food waste, primarily from households and the food service industry, is the largest global food waste stream, amounting to ~570 Mt annually⁷⁸. This waste is typically heterogeneous and often contaminated with plastics or packaging residues⁷⁹, posing major logistical and regulatory challenges for safe reuse⁸⁰. In Barcelona, compost derived from household food waste is applied to urban agriculture plots, but contamination and legal hurdles intended to lower the risk of contamination limit broader nutrient recycling efforts⁸⁰. Another example is seen in Thailand, where food waste composting and direct use as animal feed recovered up to 71% of the P content in food waste from retail and wholesale markets⁸¹. Despite its potential, P recovery from food waste remains underdeveloped, partially owing to severe logistical, environmental and behavioural challenges, low economic incentives and weak regulatory enforcement^{74,76}.

[H2] Wastewater

Wastewater refers to water discarded from households, businesses and industries, as well as stormwater runoff. Globally, ~360 billion m³ of wastewater are produced annually, containing 3.7 Mt P^{82,83}. However, in current market conditions, the economic feasibility of P recycling from wastewater remains limited⁸³, so most wastewater treatment plants prioritize meeting discharge requirements for treated wastewater rather than actively recovering P²³. P is typically removed from wastewater through chemical, biological⁸⁴ or combined treatment methods⁸⁵, with ~90% ending up in sewage sludge⁸⁶.

Sewage sludge is sometimes applied directly to agricultural fields⁸⁷. However, this practice is being increasingly restricted owing to concerns about contaminants, including heavy metals, pathogens, microplastics and toxic organic compounds. Although only a small fraction of P from wastewater is recovered using advanced recovery methods⁸⁸, the potential of numerous technologies has been demonstrated at full or pilot scale^{88,89}, particularly in North America, Europe and Asia. For effective recovery, P must first be dissolved and then concentrated and recovered through precipitation, crystallization or adsorption⁸⁹. Typically, sludge liquor, sludge or sludge ash are used to produce P-recovered products such as struvite, calcium phosphate or phosphoric acid⁹⁰.

Recovering P directly from sludge tends to yield lower-purity struvite (with lower P and higher impurity concentrations), whereas recovery from the water separated from sludge generally produces a higher-quality product and higher yield, but combined, these methods only capture 10–40% of influent P¹⁷. High recovery yields, ranging from 60–70%, can be achieved through vivianite precipitation⁹⁰ or sludge acidification, followed by solid–liquid separation and precipitation⁹¹. However, these processes are energy-intensive and thus contribute to global warming, and they require substantial amounts of acid⁹². Although the P recovery efficiency is higher than for struvite-based technologies, these methods have yet to progress beyond pilot-scale demonstrations⁹⁰.

High yields of micropollutant-free P, typically 80–90%, can also be recovered from sludge ash⁸⁹. Current technologies for P extraction from sludge ash can be categorized as wet-chemical, thermochemical and electrodialysis methods⁹³. These methods vary in their effectiveness in heavy metal removal, emissions production and energy demands⁹². P extraction from sludge ash can recover high volumes of P, making centralized facilities feasible. Other methods, such as pyrolysis or hydrothermal carbonization, produce char that can be used directly on agricultural fields, if inorganic and organic pollutants are below local legal thresholds⁹³. Despite the high potential for P recovery from wastewater and its byproducts, only 61 facilities worldwide currently pursue advanced P recovery that goes beyond conventional sludge recycling or land application of biosolids, collectively recovering $\sim 4.2 \times 10^{-3}$ Mt P yr⁻¹ from secondary sources⁸⁸ ([H1] 3). High capital expenditure requirements hinder investment in these promising technologies^{17,89}.

[H1] Legacy phosphorus

In agriculture, P input through the application of mineral and organic P fertilizers often exceeds the amount absorbed by crops, resulting in a gradual accumulation of legacy P in the soil^{27,28}. Most legacy P is strongly bound to the soil matrix, primarily to inorganic soil constituents, or is lost to aquatic environments through erosion, runoff or leaching^{48,55}, eventually accumulating legacy P in sediment^{28,55,94}, where it often contributes to environmental issues, such as eutrophication. Although legacy P is often not immediately available to plants, physical, chemical and biological pathways exist that can mobilize this P, allowing a portion of legacy P to be taken up by crops in the future⁹⁵.

[H2] Legacy phosphorus in soil

Globally, excessive fertilizer use between 1965 and 2007 resulted in the accumulation of over 815 Mt of legacy P in the soil⁹⁶. The high reliance on P fertilizers is particularly evident in Western Europe, Brazil, North America and Asia, where anthropogenic P accounted for nearly 60% of the available soil P in 2017⁹⁷. By contrast, Africa relies on these P inputs for only approximately 30% of the total available P in the soil⁹⁸. Although P fertilization is projected to rise from 20 Mt P in 2023 to 22 Mt P in 2025², only ~30% of the P supplied to crops is absorbed in the year of application, and ~45% becomes legacy P²⁹ (the remaining 25% is lost through erosion, runoff, and leaching). For example, soil legacy P was ~9 Mt in 2023⁴⁸. In the

EU, excessive fertilizer use has led to an accumulation of ~222 Mt of legacy P in topsoil⁶⁶, and legacy P accumulated in agricultural soil could reach ~107 Mt by 2050 in Brazil²⁹.

Efforts have been made to utilise or ‘mine’ soil legacy P. One promising approach is cover cropping with species that are capable of accessing less-available forms of P⁹⁹ through adaptations such as altered root structures and architectures¹⁰⁰, high rhizosphere acidification capacity¹⁰¹ and increased root exudation of compounds such as phosphatases and carboxylates that enhance P mobilization and availability¹⁰². The biomass of cover crops is then incorporated into the soil, where it releases bioavailable P that can be used by subsequent crops⁹⁹. Bioengineering and breeding crop varieties with higher P mobilization potential are also being investigated^{103,104}. However, these approaches are complex and heavily influenced by specific environmental conditions and soil P status¹⁰⁵, which can limit their reliability. Furthermore, newly dissolved P resulting from the effects of freeze–thaw cycles on cover crop biomass can reach nearby waterways¹⁰⁶.

Biostimulants are another promising strategy for mobilizing legacy P¹⁰⁷. These substances alter the soil microbiota or promote P scavenging by roots to enhance crop P uptake¹⁰⁸. Biostimulants work by mobilizing relatively stable forms of P into plant-available forms, which can considerably increase the efficiency of P use in agricultural systems, particularly in soil with high legacy P levels.

[H2] Legacy phosphorus in sediment

Historical P loading from wastewater, runoff and erosion from agricultural soil has led to the accumulation of ~2,686 Mt of legacy P in aquatic sediment, with an estimated current accumulation rate of 1.5 Mt P yr⁻¹⁵⁵. Waterbody management practices focus on maintaining or restoring water quality by immobilizing P in the sediment, which is typically achieved by adding P-binding agents or aerating lake water to promote the binding of P to iron in sediment¹⁰⁹. Although these practices reduce P reactivity and mitigate potential environmental damage, they do not facilitate P recycling. By contrast, reusing lake sediment⁹⁴ or filtering and recycling P-rich lake water would help to achieve P circularity.

Despite the substantial P stock in sediment, exploiting this resource by recycling P faces notable challenges that hinder its feasibility. The P recovery process involves complex and costly steps, including dredging, flocculation and dewatering of sediment¹¹⁰. Then, the treated sediment must be transported and applied to agricultural soil, which adds new layers of complexity, expense and risk if the sediment contains contaminants. Although using P in sediment holds promise and could contribute to P supply, more research and technological development are needed to overcome these hurdles.

[H1] Barriers to phosphorus recycling

In Europe, China, Japan and North America, financial capacity is adequate for developing and deploying efficient P recycling technologies^{10,11,25,31}. However, numerous interconnected barriers pose considerable challenges to implementing efficient P recycling (Fig. 3). These obstacles also increase reliance on imported mineral phosphates, exacerbating geopolitical and environmental vulnerabilities. Acknowledging the interconnectedness of these barriers underscores the need for transdisciplinary

approaches, where diverse sectors (including academia, industry, consumers, farmers and others) collaborate to develop comprehensive and sustainable solutions for P management and recycling.

[H2] Technological barriers

Developing P-recycling capabilities is the most apparent technological challenge. The heterogeneity of secondary P resources complicates technology transfer between waste streams. For example, P-recovery technologies designed for wastewater are not easily adaptable to recovering P from manure or lake sediment, owing to different requirements regarding removal of pollutants, pathogens and contaminants¹¹¹. Furthermore, some recycling methods can separate contaminants from waste but often result in poor P recovery or reduced P availability. The best available techniques for recovering P, such as incineration, pyrolysis, thermochemical processing and electrodialysis, are typically chemically intensive, operationally complex and costly^{112,113}, and all yield products that have downsides, notably low P bioavailability without further chemical processing¹⁷, and the continued presence of some heavy metals and other pollutants.

Nonetheless, various P recovery technologies are available and are increasingly being implemented at full scale⁸⁸. The challenge lies in balancing treatment costs with performance. Ensuring high P recovery rates from secondary resources while producing clean, safe and highly efficient fertilizers remain difficult to optimize simultaneously.

A less obvious technological barrier relates to the production of knowledge. Standardized analytical methods for assessing P availability do not effectively characterize highly complex, heterogeneous products^{114,115}. Worse yet, no standardized method for P speciation (the chemical forms in which P exists) in recovered products currently exists, although X-ray diffraction is often used. Although effective for identifying mineral P forms, this method is unsuitable for detecting organic P, is inherently qualitative, and fails to provide accurate results when mineral phases are poorly crystalline or amorphous.

These data are crucial for generating, evaluating and comparing the efficiency of recovered P sources, relative to conventional fertilizers^{115,116}. Furthermore, the heterogeneity and complex chemical composition of secondary P resources hinder their evaluation as fertilizer alternatives. For example, in the Netherlands, increasing amounts of macerated food waste are being diverted to sewage systems, contributing to mixed flows being treated in wastewater treatment plants¹¹⁷. This practice blurs the line between household and municipal waste streams, complicating the traceability and purity of recovered P.

[H2] Knowledge barriers

These technical barriers result in knowledge barriers: gaps in understanding of the dynamics of recycled materials in the environment and their efficiency as fertilizers. For example, lack of knowledge of P availability and speciation hinders developers of models for improved P management and regulators, who need these data to proceed sensibly. Recycled products often have lower water solubility and slower P dissolution dynamics than conventional mineral P fertilizers, but can release similar amounts of P over longer periods¹¹⁸ and achieve similar agronomic efficiency^{119,120}. However, the long-term effects of

continuous use of recycled fertilizers are still mostly unknown. The characteristics, production process, application method and environmental risks of the various recycled P sources are not sufficiently understood to anticipate how they will affect fertilizer performance and the environment over time^{16,121}. Therefore, understanding the dissolution kinetics and overall behaviour of recycled fertilizers in different environments is essential for modelling their transport in fields and watersheds.

In addition, the complex interplay between soil particles (such as Fe and Al oxyhydroxides and clay minerals) and the P in fertilizers is not sufficiently understood. Phosphorus phytoavailability and legacy P accumulation rates vary considerably. Although slower fertilizer solubilization can be advantageous in acidic soil, improving crop uptake and reducing leaching and runoff, P release from biochar and struvite, for example, is considerably delayed or reduced in alkaline soil, and therefore is potentially too slow for optimal plant growth^{122,123}. Furthermore, soils with high P sorption capacity, such as those rich in iron and aluminium oxides, can tightly bind P, increasing soil legacy P, whereas sandy or organic-rich soils often retain less fertilizer-derived P, promoting higher bioavailability.

Current understanding of mineral-bound P species is uneven. Whereas iron phosphates have received increasing attention over the past decade, particularly those precipitated from sewage sludge, research on aluminium-P forms, which is also highly utilized for P precipitation in wastewater treatment plants, remains underexplored. This gap limits the development of P-recovery processes from aluminium-rich waste streams and further complicates efforts to assess the potential and limitations of different recycled P sources in a comparative, system-wide manner.

These knowledge gaps complicate the accurate measurement of P circularity. For this reason, the true economic, environmental and climate-related impacts of failing to recycle P are still not fully understood. These uncertainties impede wider adoption of recycled P alternatives, even by early adopters⁸⁸. Without sufficient production, potential users cannot fully test P sources in real production scenarios, which raises concerns about their effectiveness in agricultural systems. This uncertainty lowers farmer demand, which lowers profit expectations for producers, thereby further hindering effective P recycling^{23,124}.

[H2] Logistics-related barriers

Logistical barriers refer to obstacles that impede the collection, handling, processing and redistribution of waste streams and recycled P. A notable logistical challenge is the distance between the sources of recycled P and arable land where it is needed. For example, densely populated areas generate substantial amounts of P-rich wastewater, but these regions are frequently far from agricultural lands that require P inputs⁶⁶. For example, in the Netherlands, sewage sludge application on farmland has been banned since the 1990s owing to concerns about contaminants. Following an incineration capacity shortfall in 2020, the Netherlands exported 27.5 kt of sewage sludge to the UK, one of the few nearby countries that still permit land application of treated sludge¹²⁵. Besides being logistically difficult, this trade was economically impractical, resulting in a financial loss.

Regional generation and distribution imbalances are an issue with the use of manure^{16,67,126}. The high volume and moisture content (85–95%) of manure create logistical challenges for its handling, storage and long-distance transport^{127–129}. The water content of manure can be reduced, thereby increasing P concentration, but logistical shortcomings continue to limit the accessibility of manure for P recycling^{130,131}. Existing techniques such as thermal drying can reduce moisture content to 10–15% but are expensive and energy-intensive¹³². The use of more accessible, non-thermal techniques results in a product with 65–75% water content, which is still too high for efficient, cost-effective application and transportation. In Sweden, if fertilizer prices remain stable and recycling processes are unchanged, transportation costs would need to be reduced by 73% for manure recycling to be cost-effective¹³³.

Last, recycled P fertilizers are cumbersome, complicating their application^{129,134}. Recycled P fertilizers are often bulky or dusty, which makes them difficult to handle and requires specialized and/or multiple machines to manage large volumes. For example, although food-waste-based compost and digestate can improve soil structure and organic carbon content, their P content is low (~0.4% P on a dry matter basis^{25,76}). As more concentrated P fertilizer options exist, farmers resist extra investments that would facilitate the use of more bulky, recycled options. In addition to P, recycled fertilizers can contain variable nitrogen, potassium and micronutrient content, and this variability, coupled with differing nutrient-release dynamics and logistical costs, tends to dissuade farmers from using recycled products¹³⁵.

[H2] Economic and financial barriers

The economic feasibility of safely recovering and using P from secondary sources is another barrier to P recycling¹³⁶. High initial investments, elevated production costs and uncertain returns deter the implementation of P-recovery technologies²³. The financial magnitude of full-scale P recycling from sewage sludge ash into technical-grade phosphoric acid (75% H₃PO₄, containing 23% P) has been demonstrated in Switzerland. As of 2023, the projected capital expenditure for a facility producing 40 kt of technical-grade phosphoric acid annually is ~US\$190 million, with operational costs of ~US\$29.5 million yr⁻¹ to produce 12 kt of phosphoric acid annually (2,760 t P yr⁻¹). Thus, operational costs are ~US\$2,460 t⁻¹ of phosphoric acid and annualized capital expenditure is ~US\$9.3 million (~US\$775 t⁻¹ of phosphoric acid)¹³⁷. At the time, the market price of phosphoric acid was below US\$1,100 t⁻¹, about three times less than the production cost of recycled phosphoric acid^{137,138}.

The higher cost of recycled P fertilizers is related to the physicochemical characteristics of the secondary materials, the smaller operation size (economy of scale) and the increased complexity of most recovery processes¹³⁷. For example, advanced P recovery from manure yields a final product that is too expensive for most farmers, compared with raw manure¹⁶. Similarly, fertilizers produced with struvite precipitation technologies can be 2–14-fold more costly than those derived from phosphate rock^{17,139,140}. Most attempts at struvite precipitation (related to clogged pipes from uncontrolled struvite crystallization) have been motivated by an interest in reducing maintenance costs, not in producing fertilizers^{141,142}. The

higher cost of recycled fertilizers compared with those from conventional sources and their uncertain benefits in terms of crop yield can considerably discourage farmers from using recycled materials¹⁴⁰.

Phosphorus recycling is also markedly influenced by global economic disparities. For farmers, who often even struggle with the cost of conventional fertilizers, investing in recycled P sources is risky, with an uncertain and potentially delayed return on investment^{143,144}. This scenario leads to reluctance in committing resources to adopt recycled alternatives, hindering P recycling implementation in developing regions^{18,145}.

[H2] Societal barriers

Societal barriers encompass poor food planning, misinterpretation of ‘best-before’ dates for perishable foods, and over-purchasing, all of which are core contributors to household waste^{78,79}. Furthermore, perceptions of recycled fertilizers, including their safety, can lead to acceptance-hindering stigmas, including concerns about contaminants and pathogens, as well as ‘yuck factors’ such as odour^{146,147}. These perceptions can discourage farmers from embracing recycled fertilizers and deter consumers from buying produce grown with them^{136,148}. In addition, farmers often resist shifting from established practices that they know are effective¹³⁵. They know that recycled fertilizers release nutrients more slowly but are uncertain about the implications of this difference. Their hesitance is exacerbated by a lack of expert advice and guidance that is tailored to their specific situations, leaving many farmers unsure about the most effective use of these products¹⁴⁹.

[H2] Regulatory barriers

Regulatory and legal barriers to P recycling are related to policies and legislation that fail to effectively support P recycling at various scales. Issues range from poorly focused regulations and guidelines for P use, management and recycling, the absence of quality assurance procedures, inadequate governmental incentives and lack of collaborative goal-setting¹⁵⁰, to the unexpected and adverse impacts on domestic P recycling of regulation designed to address ‘unfair’ agricultural trade policies at the international level¹⁵¹. All of these issues hinder the development of strategies and clear frameworks for P recycling globally¹⁵².

Market access to recycled fertilizers is often impeded by existing subsidies for mineral P fertilizers, but incentives to adopt recycled products are also insufficient, perpetuating their low competitiveness¹⁵³. These policies discourage the use of recycled products and limit their global trade potential, further stalling efforts to create a circular economy for P recycling. One notable regulatory barrier is the lack of authorization for using recycled fertilizers in agriculture. For example, countries such as China, Japan and the USA have made technological advances in P recovery from wastewater, but lack policies that actively support or require P recovery from the wastewater sector⁵⁸. A major reason is quality concerns regarding products that are feared to contain heavy metals, pathogens, microplastics and toxic organic compounds, such as per- and polyfluoroalkyl substances (PFAS). These fears lead to regulatory and market barriers⁸⁰. Only a few countries worldwide (that is, Germany¹⁵⁴, Austria¹⁵⁵ and Switzerland¹⁵⁶) have implemented

regulatory mandates for P recovery from sewage sludge¹⁵⁷. This regulatory lag limits the widespread adoption of P-recovery technologies in many regions.

Similarly, up until the 2020s, legal approval for struvite to be used as a fertilizer was not forthcoming, but this use is gradually being accepted globally. In the EU, legislation is following this trend: an amendment of the Fertilising Products Regulation (EC2019/1009)¹⁵⁸ now enables the use of struvite as a fertilizer. However, other major P-rich secondary sources still lack approval. An example is Category 1 animal byproducts (animal parts suspected or confirmed to be infected by biological hazards), which can have high P concentrations, but are not currently authorized under the EU's Fertilising Products Regulation owing to health concerns. For example, there is a potential risk of contamination with prions that cause bovine spongiform encephalopathy, and despite certain thermal treatments and downstream acidulation showing certain efficiency in eliminating such risks¹⁵⁹, other studies are less conclusive¹⁶⁰ and the techniques lack validation at scale. More complex certification processes for these products place an additional burden on producers, who must ensure compliance to gain market access, and ultimately discourage P recycling¹⁶¹.

There is a lack of harmony between the regulations and guidelines for different P-related sectors. Numerous sector-specific regulations exist for P management but there is no overarching governance framework¹⁶². For example, in the EU, the Fertilising Products Regulation (EC2019/1009) now includes recovered P fertilizers, but frameworks that regulate secondary P resources, such as the Urban Wastewater Treatment Directive 91/271/EEC¹⁶³, offer no guidelines about recovered P. Using manure or sewage sludge, farmers often apply the maximum nitrogen dose allowed by the EU Nitrates Directive (91/676/EC)^{164,165}, potentially resulting in P overapplication⁶⁷. Although many EU Member States provide P application guidelines, relatively few have legislation focused on the use of P fertilizers¹⁶⁶. This sectoral approach perpetuates technocratic objectives focused on narrow goals²², relying heavily on command-and-control instruments that are weakly enforced. These hindrances result in a fragmented policy landscape that impedes the systemic adoption of P recycling practices²¹.

The fragmentation in legislation is further exacerbated by stakeholders operating in disciplinary silos with a narrow focus on their specific concerns rather than collaborating to establish a holistic framework for P management¹⁶⁷. For example, stakeholders concerned with contaminants in waste-derived fertilizers, such as environmental advocacy groups, or the general public concerned with the 'yuck factor' of these fertilisers, tend to advocate for strict regulations regarding P recycling without fully considering the economic challenges faced by farmers and wastewater companies¹³⁶. These concerns highlight the broader challenge of determining when such materials can be considered safe to use, which complicates regulation and policymaking, a key issue in the ongoing 'End-of-Waste' debate. This dilemma points to the need for shared knowledge, open dialogue and a community-driven willingness to compromise in pursuit of more balanced and workable policies.

[H1] Strategies for moving forward

[H2] Improving phosphorus flows

Addressing the global P challenge requires a multifaceted, holistic approach that harnesses the expertise of individual disciplines²². The first pathway to better P management is to reduce the volume of secondary P. In agriculture, this approach includes developing precision farming techniques that increase application efficiency and reduce runoff to water bodies¹⁶⁸. In addition, better manure recycling, and the integration of livestock and crop production, are needed to use manure more efficiently as a P fertilizer. In urban contexts, this approach includes decreasing food waste and broadening the collection and treatment of wastewater to recover P and reintroduce it into the cycle. P recovery through waste treatment should be expanded globally, with investments in infrastructure that support P recycling from agricultural, industrial and urban waste¹⁶⁹.

Increasing the recovery from secondary P resources must be complemented by reducing demand for P resources. Transformative changes in human diets and consumption patterns are essential to mitigate the impacts of P-intensive food production. Over the past 50 years, per capita P footprints have surged owing to dietary shifts, primarily rising consumption of meat, which now comprises 72% of the global P footprint. This shift has driven a 38% increase in global P demand between 1961 and 2007, with substantial variations across countries¹⁷⁰. Reducing meat consumption, especially in countries with high P footprints, could substantially decrease P demand¹⁷⁰. This change would both help conserve finite P resources and reduce the risk of eutrophication, thus aligning with broader environmental and health sustainability goals. In addition to dietary changes, improving P use efficiency and enhancing recycling at each stage of food production are crucial complementary strategies¹⁷⁰.

In addition to improved production, the challenges posed by the damage done by P that is lost to the environment remain. Technologies are needed to reduce P runoff into waterbodies. Lower P losses can be achieved by effective exploitation of legacy P in soil through intensive farming, cover cropping, biostimulant use and efficient P management (for example, fertilizer choice, application timing and doses). Waterbody restoration techniques should focus on recycling legacy P in sediment for use in agriculture. For example, restoring lakes through sediment dredging could provide P while substantially reducing methane emissions¹⁷¹. Therefore, tapping into P stocks in soil and sediment could help address both the P and climate challenges. Advanced P recycling can also play a major part in reducing both legacy P accumulation and losses to water bodies. Widespread P recycling would increase the market availability of less water-soluble fertilizers (such as struvite, (bio)chars and sludge ashes), decreasing reliance on water-soluble fertilizers that are prone to runoff¹⁷². In some cropping systems, this shift could better align P release with plant demand, reducing P fixation in soil and losses to water bodies¹⁷³.

Improving P use efficiency in animal feed can substantially reduce P losses across the agricultural system. One well-established approach is the addition of enzymes such as phytase, which increase P availability to animals, thereby lowering P additive to feed and therefore total P intake by livestock and reducing P concentrations in manure¹⁷⁴. Another strategy is to remove excess P from feed ingredients. For example, in the USA, P is removed from dry distillers' grains used as feed to reduce its P content before

consumption⁷⁷. However, the need for more efficient use of manure remains. Once manure is produced, recovering P from digestate can further improve overall nutrient recycling¹⁷⁵. Advanced P recovery from manure often involves solubilizing and precipitating P compounds such as calcium phosphate and struvite^{176,177}. Emerging methods such as vivianite separation, vacuum evaporation, membrane filtration and ion exchange are promising¹⁷⁵ but not yet fully developed or widely implemented.

[H2] Policy and pricing instruments

Prescriptive regulations to reduce harmful practices, mandate P recovery¹⁷⁸ and enforce the use of recovered products are widely advocated^{11,16,18} but not often implemented. This hesitance has spurred calls for broader approaches, including price-based policy instruments such as auction or tender systems that allocate public funds to support environmental services, and quantity-based mechanisms, including offset programmes¹⁷⁹. The major challenge for any price-based incentive scheme designed to change P use at the country level is preparing for and responding to unexpected impacts from World Trade Organization (WTO) rulings regarding food production for export.

Under the rules of the WTO Agreement on Agriculture, pricing policies interpreted as ‘market-distorting’ are prohibited. Government policies that manipulate the price of P fertilizers might fit this description. Fortunately, exemptions are possible for price-based incentive schemes that do not exceed specific limits, or have “no or at most minimal trade-distorting effects on production” and do “not have the effect of providing price support to farmers”¹⁸⁰. And, to the extent that any price incentive scheme for recycling P could be part of a “clearly defined” governmental environmental programme, the amount paid to farmers must be “limited to the extra costs or loss of income involved in complying with the government programme”¹⁸¹. However, the risk of an adverse WTO ruling remains.

Cap-and-trade systems for P contained in manure (as for CO₂ emissions) have also been suggested for improving P management²¹ and have even been implemented in the Netherlands¹⁸². However, this approach can be problematic because watersheds and waterbodies are affected by local pollution. Therefore, unlike CO₂ emissions, choosing where to cap and trade becomes crucial, especially in light of the need for improved manure management and transportation logistics. Regulations should support safe, new technologies and recovered P products in agriculture and industry, despite inter-sectoral legal difficulties. For example, the EU’s Carbon Border Adjustment Mechanism (CBAM) attempts to address regulatory issues regarding CO₂ emissions on imported goods, but questions remain about the feasibility of CBAM’s implementation and compatibility with WTO trade rules. Alternatively, subsidy reforms, such as reducing mineral fertilizer subsidies (as in China¹⁸³) or transforming agricultural subsidies (as in the EU and USA) can encourage the use of recovered fertilizers and sustainable practices, shifting subsidies toward improved P management, provided that challenges around subsidy design to ensure compatibility with WTO trade rules can be overcome^{178,184}.

Large, centralized facilities could take advantage of economies of scale to make P recycling economically viable in areas that are highly urbanized or have dense livestock production¹⁸⁵. Such facilities optimize labour, energy and raw material use, reducing operational costs and maximizing output efficiency. However, centralization comes with its own set of financial and operational risks, such as supply chain interruptions and regulatory compliance issues, as well as limited ability to adapt to local market conditions¹⁸⁵. By contrast, decentralized, small-scale solutions such as on-farm urine sterilization or manure processing can be more suitable in rural or low-income settings where nutrient demand and supply are more locally balanced, and transport distances are shorter. This contrasting suitability is due to the high capital investment required for constructing and maintaining large-scale facilities, which can be a notable financial burden that affects the decision-making of implementing P recycling.

[H2] Transdisciplinarity in phosphorus recycling

Interdisciplinary frameworks and decision support tools are vital for developing comprehensive P sustainability strategies. The EU-centred 5R phosphorus stewardship framework offers a broad model that includes realigning P inputs, reducing losses, recycling secondary resources, recovering P from waste, and redefining food systems through demand shifts¹⁸⁶. Technologies and practices must achieve both technical and societal acceptance, addressing both community impacts and agronomic needs¹⁸⁷. Similarly, the ‘net-zero phosphorus cities’ framework emphasizes the role of urban areas in creating circular P economies by capturing wastewater P¹⁵³. These frameworks could be improved with better empirical data^{188–190} and holistic integration of disciplinary perspectives.

We call for a transdisciplinary approach that incorporates non-academic stakeholders to ensure full co-production of strategies (Box 1). Collaboration must transcend individual interests and focus on shared goals, such as reducing runoff, improving nutrient-use efficiency and promoting recycling¹⁹¹. Transdisciplinarity enables a holistic understanding of the P cycle by integrating insights from agriculture, waste, environment, economics and policy sectors. Although transdisciplinarity faces challenges such as securing long-term funding, time-consuming coordination and communication among diverse stakeholders^{57,148}, it fosters innovative solutions for P recycling and pollution reduction, ensuring that strategies are practical and sustainable. Transdisciplinary efforts must inform policies that address P management complexities from local to global levels^{21,22}.

Phosphorus-focused organizations, such as the Global Phosphorus Research Initiative, the European Sustainable Phosphorus Platform, the US Sustainable Phosphorus Alliance, the Australian Sustainable Phosphorus Futures and the Phosphorus Industry Development Organisation of Japan), have a key role in coordinating efforts by connecting stakeholders and facilitating P-recovery strategies¹⁹². These agencies should serve as ‘mediators’ in stakeholder relationships. In regions where they do not exist, efforts should focus on engaging stakeholders beyond sectoral boundaries, thus paving the way for technological scaling, supportive policies and regulatory frameworks¹⁴⁸ while ensuring a transition to circular P economies⁵⁸.

Transdisciplinary research must promote broad and inclusive participation, ensuring that diverse perspectives are represented in decision-making. This approach supports fair processes and fosters more widely acceptable solutions. The potential of the transdisciplinary approach is exemplified by the UK Phosphorus Transformation Strategy¹⁹³ that involves extensive stakeholder engagement and has already produced a credible roadmap for multi-sectoral action across the P value chain, which can be broadened to other regions (Fig. 4). Future efforts should foster knowledge exchange to identify stakeholder interests and challenges, create local pathways, assign responsibilities and develop realistic timelines for P recycling targets¹⁴⁸.

[H1] Summary and future perspectives

Recycling is an important step towards P circularity, but global adoption remains limited owing to economic, technical, societal, regulatory and political barriers. The technical complexity of P recycling technologies⁸⁹, small scale of operations and high production costs make recycled fertilizers considerably more expensive than conventional, rock-derived fertilizers, as shown in Switzerland¹³⁷. Recycled P products are also more chemically and physically complex, complicating attempts to understand their agricultural efficiency, impacts on human health and long-term environmental safety¹²⁹. Limited production, logistical hurdles in transport and utilization, and uncertain financial returns further discourage adoption, as demonstrated in Sweden¹³³. Meanwhile, narrow, disciplinary stakeholder focus and diverse socioeconomic and environmental realities across regions hinder collaboration and stifle global progress in P recycling¹⁴⁸.

Despite growing awareness of the importance of P recycling, challenges in accurately determining global P flows and uncertainties around the agronomic efficiency, safety, and environmental impacts of recycled P fertilizers continue to hamper progress. In addition, diverse agricultural systems and regulatory frameworks further complicate the development of coherent global strategies. This fragmentation extends across the whole P system: researchers, farmers, policymakers, environmental agencies and consumers each prioritise different goals, creating misalignments that undermine coordinated action. Bridging these gaps requires improved communication, transdisciplinary collaboration, and context-specific solutions that balance technical, societal and environmental needs.

Transforming P management requires a holistic approach in which stakeholders prioritize the shared objective of tackling the wicked problem of P circularity. Key goals include minimising waste generation, reducing P runoff, enhancing nutrient-use efficiency and promoting P recycling efforts. Breaking down barriers across sectors is essential: improved communication, transdisciplinary research and diverse perspectives can lay the foundation for an integrated approach to P circularity. Inclusive policies that address the needs of all stakeholders can drive collective action. Strategic research developments and actionable priorities will support progress over the next decade (Table 2).

Supporting these efforts requires a concerted push, through targeted incentives for sustainable P use, coherent regulation at local, regional and international scales, investments in innovation and research and the creation of inclusive platforms that facilitate transparent dialogue across sectors. Ultimately, the transition to circular P systems will depend on the ability of stakeholders to break down disciplinary, institutional and geographic silos, fostering collaboration that bridges science, policy and practice. Only then can resilient, sustainable P management strategies be built that are capable of reshaping the current linear supply chains into circular systems for the future.

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The authors declare no competing interests.

617 Disclaimer

618 The results expressed in this article are those of the authors only and do not necessarily reflect those of the
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633 Display items

634 **Table 1. Major secondary phosphorus resources and recycling techniques and challenges**

Secondary P waste stream or legacy P stock	Global amount generated	Current fate	Potential advanced recycling techniques for secondary P or drawdown strategies for legacy P	Challenges
Secondary phosphorus resources				
Mining and fertilizer industry	6–12 Mt P yr ⁻¹ (1,2,8,64)	Disposal in landfills or coastal waters, or stacked on land	Pyrolysis, leaching and precipitation	High complexity and cost
Livestock manure	15–20 Mt P yr ⁻¹ (1)	Direct application (~90%)	Flocculation, settling, screw pressing, belt filtration, centrifugation and anaerobic digestion (and subsequent dissolved air flotation)	Large volumes produced only in a few regions Geographic disconnect between generation and use locations Dewatering is energy-intensive and transportation is logistically complex
Food waste	~1.2 Mt P yr ⁻¹ (1,59)	Landfill (~60%) Animal feed (5–10%)	Composting, anaerobic digestion and incineration	Low phosphorus concentration High heterogeneity of materials
Wastewater	~3.7 Mt P yr ⁻¹ (82,83)	Discharge into surface waters, irrigation and agricultural reuse ^b	Biological removal and chemical precipitation	Requires infrastructure Processes might substantially differ depending on waste characteristics and regions
Sewage sludge ^a	~3 Mt P yr ⁻¹ (86)	Landfill, direct application ^b	Incineration, pyrolysis and composting	Yuck factor Low phosphorus availability in sludge-derived products Chemically intensive
Sewage sludge ash ^a	0.13–0.22 Mt P yr ⁻¹ (194)	Landfill, direct application ^b	Chemical and thermochemical treatments	Energy- and chemically intensive High costs

Legacy phosphorus stocks				
In soil	~815 Mt P ⁽⁹⁶⁾	Accumulation, erosion and runoff	Phosphorus-mining crops, biostimulants, bioengineering, plant breeding and biofertilizers	Low bioavailability Geographical dispersion
In sediment	>2600 Mt P ⁽⁵⁵⁾	Accumulation	Direct application	Dredging lakes is costly and complex Contamination is possible

^aThe phosphorus reported in sewage sludge and sludge ash originates from the total phosphorus in wastewater. These values are not additive; instead, the phosphorus content in sludge and ash is contained within the original wastewater phosphorus flow. ^bSubject to local legislation. P, phosphorus.

Table 2. Strategic priorities for advancing phosphorus recycling and sustainability

Goal	Key developments needed	Research priorities	Methods and techniques
Minimize waste	Improve food waste management practices Redouble efforts to keep sewage clean to enable safe reuse of sludge Reduce industrial phosphorus losses during food processing	Investigate success factors in household food waste separation and reduction Study societal, economic and infrastructural barriers to source-separated sanitation Develop comprehensive national and regional phosphorus balances to quantify and locate key phosphorus losses	Behavioural analysis: surveys, interviews, smart bin sensor data analysis Spatial mapping: GIS mapping of waste and sanitation access patterns Economic evaluation: cost-effectiveness analysis, pilot incentive schemes Infrastructure and policy integration: collaboration with utilities, regulatory framework co-design Material flow analysis: mass balance studies, integration of corporate and national phosphorus data Modeling and prediction: nutrient dynamic models (APSIM or DNDC), machine learning for waste prediction
Reduce phosphorus runoff	Adopt more efficient fertilizer use practices Develop slow-release and enhanced-efficiency fertilizers Exploit legacy phosphorus in soil and sediment	Advance precision fertilization and controlled-release technologies Improve understanding of phosphorus mobility under different soil and climate conditions Quantify the effectiveness of cover crops in legacy phosphorus mobilization Assess the agronomic potential of phosphorus recovery from lake sediment	Fertilizer efficiency improvement: precision fertilization trials, slow-release fertilizer testing, meta-analysis under various conditions Sensor and technology integration: sensor-based fertilization systems, variable-rate application with drones or satellites Phosphorus movement tracking: lysimeter and field leaching studies Nutrient dynamics modelling: simulation using APSIM or DNDC models Legacy phosphorus mobilization: cover crop experiments, microbial and enzymatic pathway studies Sediment phosphorus recovery: collection and laboratory testing of sediment cores from eutrophic lakes
Enhance nutrient use efficiency	Introduce nutrient budgeting tools at the farm scale Increase farmer awareness of efficient phosphorus use Design more efficient and sustainable phosphorus fertilizers Improve phosphorus management on-farm	Evaluate policy instruments that promote balanced phosphorus inputs, such as nutrient caps, certification schemes Identify best practices and barriers in farmer education and extension programs Explore digital tools and decision-support systems for optimising farm-level nutrient use	Farm-scale nutrient planning: nutrient budgeting tools, whole-farm simulation (NuGIS, FarmDESIGN) Policy and governance evaluation: policy assessment (nutrient caps, certification), agent-based modelling of policy impacts Farmer education and outreach: surveys and/or interviews on extension strategies, identification of best practices and peer learning champions Digital tool development: decision-support systems, mobile apps, AI-powered nutrient management platforms

			Precision agriculture integration: incorporation of weather and soil data into recommendations, collaboration with ag-tech startups, integrate weather and soil data layers for precision recommendations Digital tool development: decision-support systems, mobile apps, AI-powered nutrient management platforms, co-designed farmer-friendly interfaces
Promote phosphorus recycling	Reduce land application of sewage sludge due to micropollutant concerns Adopt targeted phosphorus recovery (e.g. struvite and ash-derived products)	Investigate financial models and policy tools to reduce investment risk in phosphorus recovery technologies Support the harmonization and implementation of product standards, such as the EU Fertilising Products Regulation Research farmer perceptions and agronomic performance of recovered phosphorus products Develop certification schemes to increase market acceptance	Technology development and evaluation: techno-economic analysis of phosphorus recovery technologies, digital twin models for wastewater tracking Policy and financial instruments: financial model investigation, targeted subsidies, tax incentives, certification schemes, harmonization of product standards Farmer acceptance and agronomic testing: farmer field trials with recovered phosphorus products, surveys on farmer perceptions Governance and multi-actor collaboration: stakeholder platforms, comparative governance analysis (ESPP, UK and Germany), systems thinking tools (e.g. causal loop diagrams) Monitoring and regulatory frameworks: municipal monitoring programmes, micropollutant threshold co-definition with regulatory agencies Knowledge sharing and foresight: global case study repository, foresight workshops on scaling collaboration platforms
Foster transdisciplinary collaboration	Model new platforms on existing ones (e.g. ESPP, Dutch and German Nutrient Platforms and UK Phosphorus Transformation Strategy)	Evaluate the effectiveness and governance models of multi-stakeholder platforms Identify key success factors and lessons learned from regional initiatives Assess mechanisms for scaling and replicating collaborative governance structures	Governance analysis: evaluation of multi-stakeholder platform models, analysis of governance structures, funding and communication strategies Scaling and replication studies: assessment of scaling mechanisms, testing replicability via policy labs, publishing blueprints with adaptation guidelines Stakeholder engagement: interviews with platform coordinators and stakeholders, synthesis of lessons learned for guidelines Participatory methods: use of living laboratories, participatory foresight, and scenario-building for adaptive governance design International collaboration: organize workshops for model exchange and best practice sharing across regions

617 APSIM, Agricultural Production Systems sIMulator; DNDC, DeNitrification-DeComposition; ESPP,
618 European Sustainable Phosphorus Platform.

619 **Figure legends**



Fig. 1. Phosphorus cycle flows and challenges in the European Union. Phosphorus flows in the EU in 2005, based on data from⁴⁷. The predominantly linear cycle begins with phosphate rock mining (mostly outside the EU), followed by phosphorus use in fertilizers, human and animal consumption, and eventual inefficient disposal as waste, leading to widespread losses to the environment as legacy phosphorus accumulates in soil and sediment. Partial recycling of phosphorus occurs through manure and sludge application in agriculture, but addressing environmental and health impacts remains technically and logistically challenging. No major advanced recycling efforts had been established when these data were collected. Collaboration-hindering stakeholder fragmentation is a major barrier to large-scale implementation of advanced phosphorus recycling initiatives.

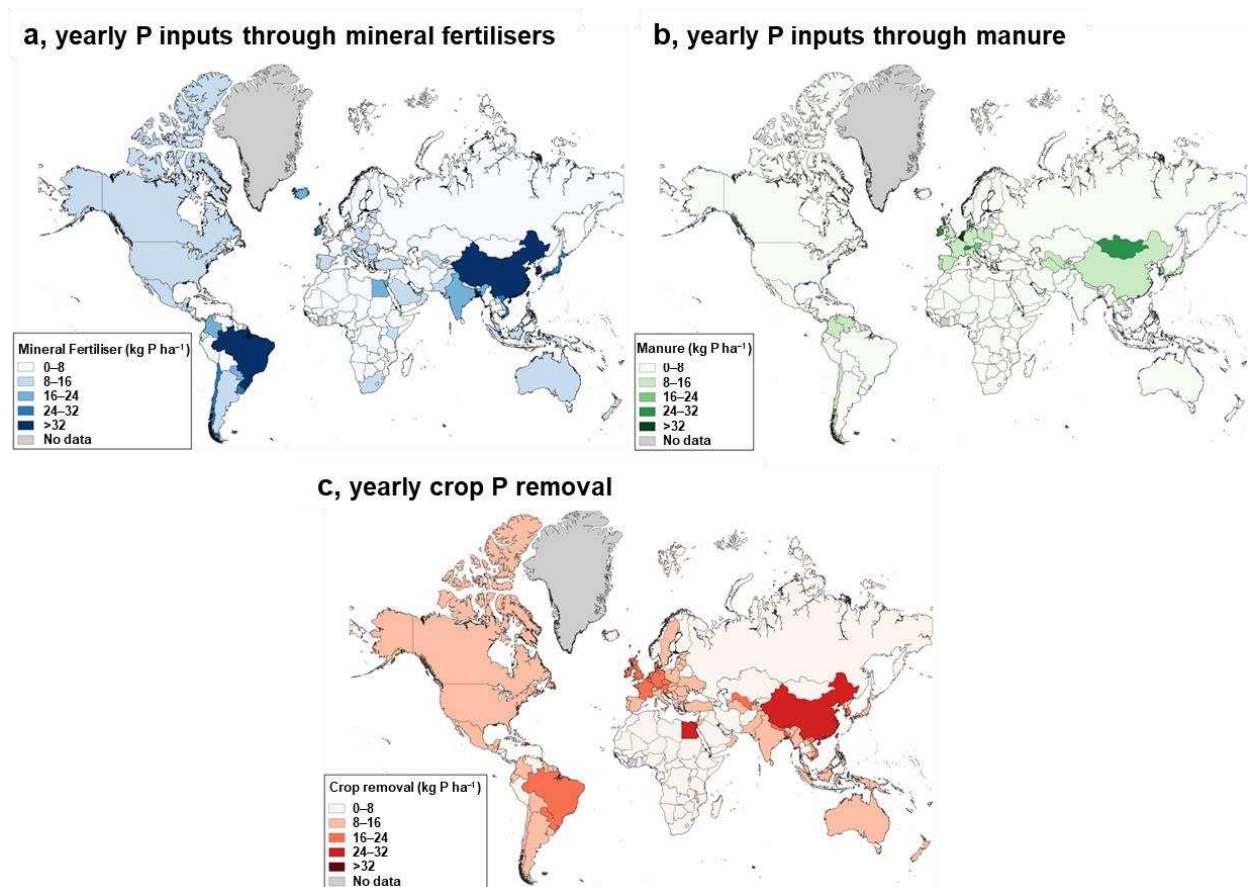
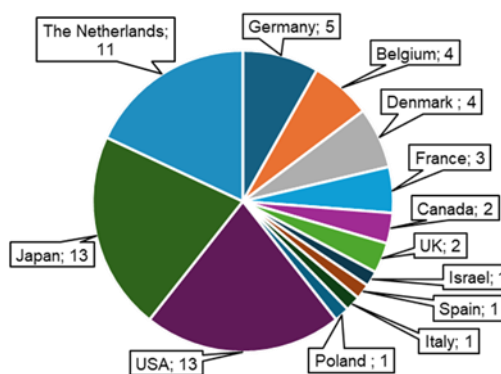


Fig. 2. Global phosphorus inputs and removals in agriculture. The major inputs of phosphorus in agriculture are through mineral fertilizer and manure application, whereas crop offtake is the main process of phosphorus removal. **a**, Global rates of mineral phosphorus fertilizer application. **b**, Global rates of manure application. **c**, Crop phosphorus removal. The maps highlight the uneven patterns of phosphorus consumption around the world, with disproportionately high phosphorus inputs in countries such as Brazil and China, higher manure application in Europe, China and Mongolia, and higher phosphorus uptakes in Europe despite low mineral phosphorus inputs, which is attributed mainly to legacy phosphorus drawdown. Based on data from ref.⁴⁸.

a, Advanced P recovery plants per country



b, Yearly P recovery by advanced techniques

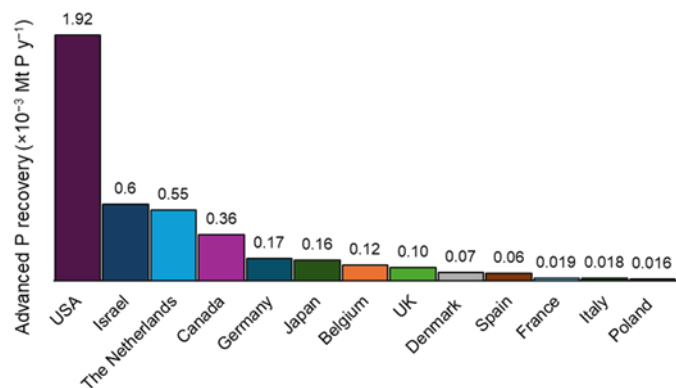


Fig. 3. Global distribution and recovery amounts of advanced phosphorus recovery plants. a, Number and distribution of operational advanced phosphorus recovery facilities in each country in 2023. **b,** Annual total phosphorus output by the 61 advanced P-recycling plants that reported their yields in 2016⁸⁸.

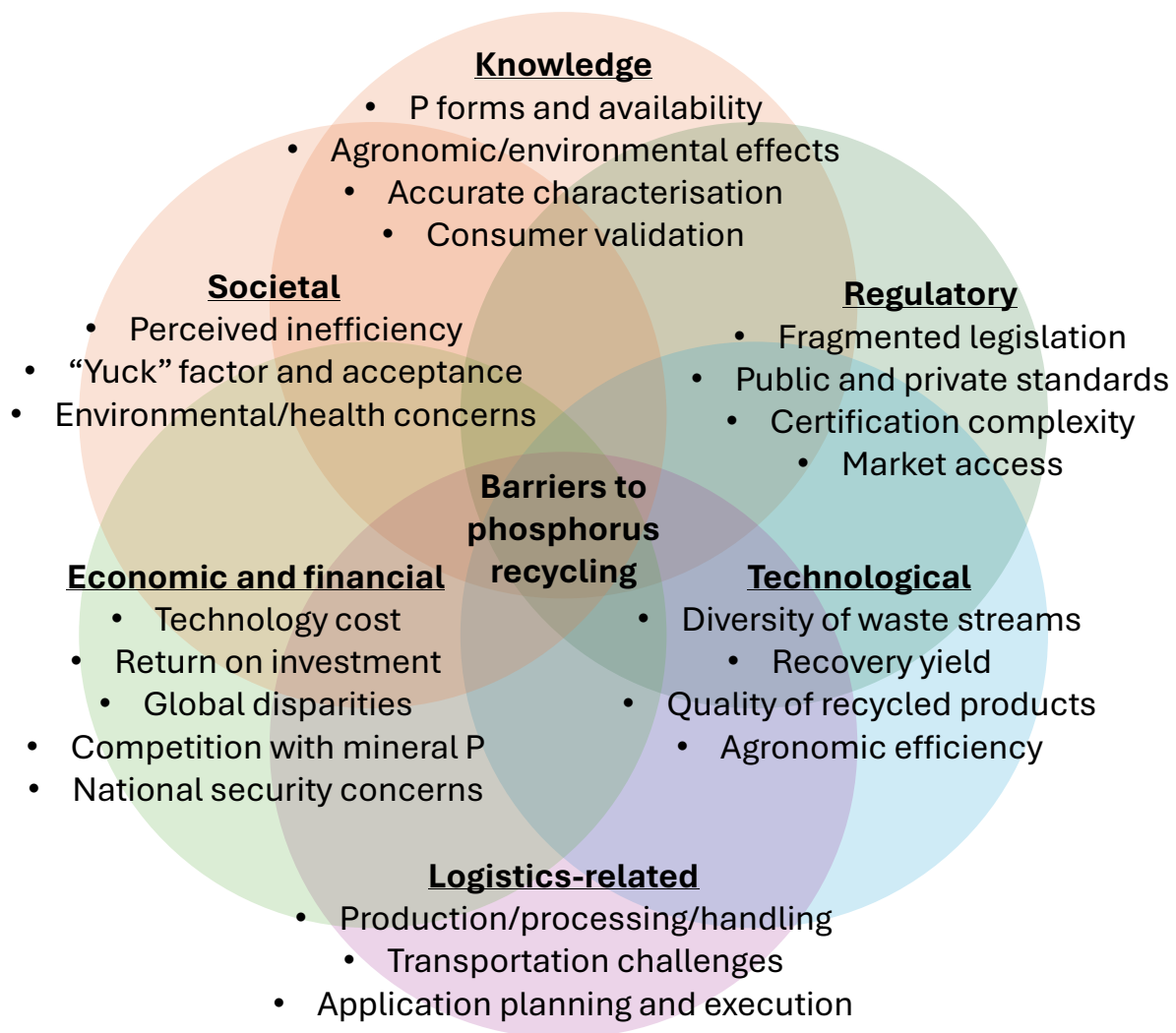


Fig. 4. Overview of key interconnected barriers to phosphorus recycling. Interconnected technological, economic and financial, regulatory, and societal barriers hinder phosphorus recycling. These barriers reinforce one another, sustain reliance on mineral phosphorus imports, contribute to inefficient waste management, and drive environmental phosphorus losses. Addressing them requires coordinated, cross-sector action to establish a more circular and sustainable phosphorus system.

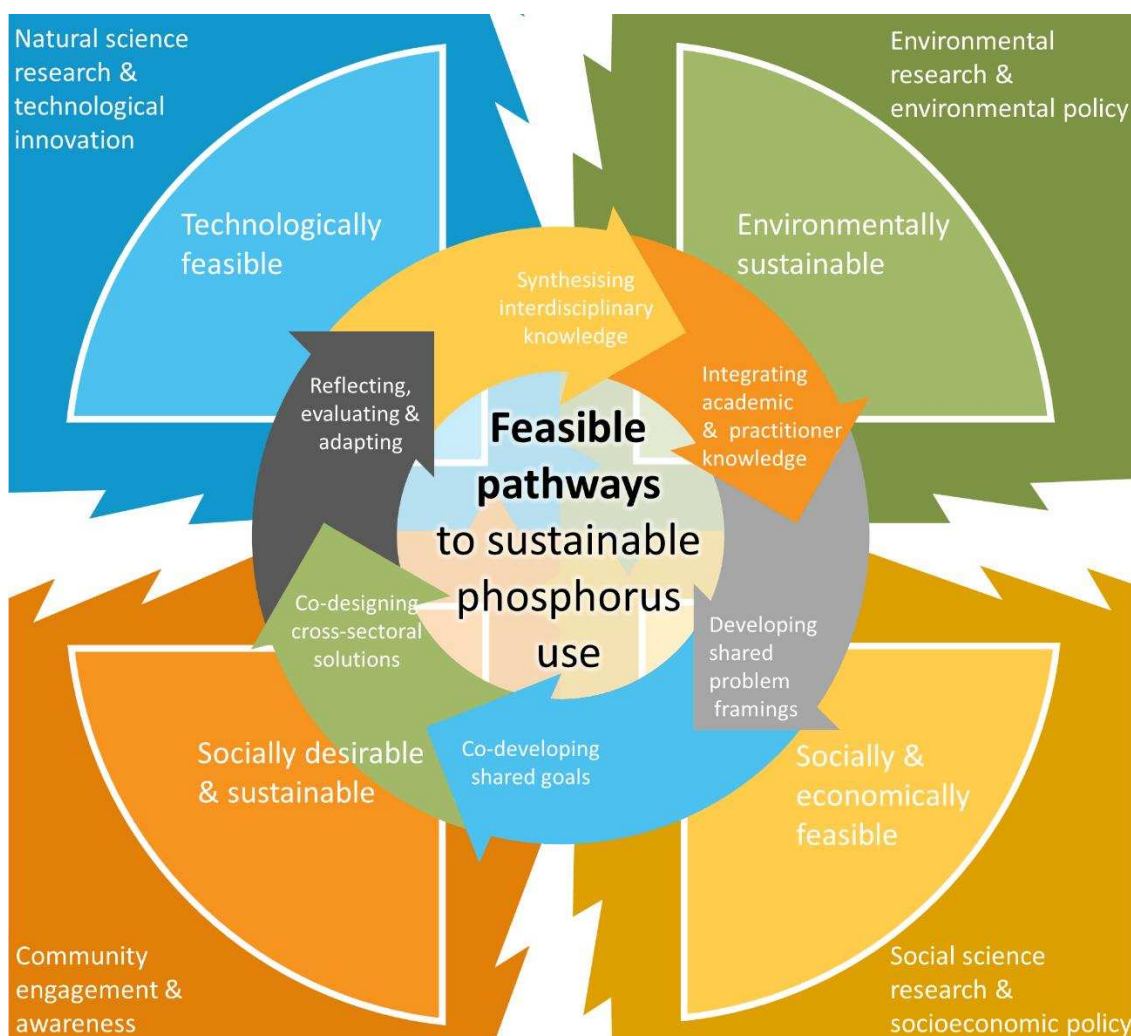


Fig. 4. Transdisciplinarity as a tool to overcome fragmentation through integration and stakeholder engagement. The role of a transdisciplinary approach to phosphorus management¹⁹³. Fragmentation across research fields and sectors hinders the development of sustainable phosphorus pathways that are technologically, environmentally, and socially feasible. Transdisciplinary approaches grounded in stakeholder engagement can overcome this by integrating disciplines, sectors, and actors. Co-developing solutions with stakeholders increases buy-in and the chances of successful implementation.

Box 1. Strategies to overcome transdisciplinarity barriers in phosphorus stewardship

To strengthen transdisciplinary research, several strategies can be implemented. First, enhancing communication and coordination is crucial. This aim can be achieved by organizing regular workshops and meetings to connect stakeholders from different disciplines¹⁹⁵, alongside developing shared digital platforms to facilitate collaboration and information exchange¹⁹⁶. Securing funding and resources is equally important. Advocating for dedicated funding programmes tailored to transdisciplinary research¹⁹⁵ and encouraging public–private partnerships to pool resources¹⁹⁶ can help address this need. Integrating

617 knowledge can be supported by establishing interdisciplinary training programmes, equipping researchers
618 with the skills to synthesize diverse expertise¹⁹⁵, and developing collaborative frameworks¹⁹⁶. Institutional
619 support also has a key role. Policy reforms that prioritize transdisciplinary efforts¹⁹⁵ and institutional
620 incentives that reward participation¹⁹⁶ can create a more supportive environment. Finally, stakeholder
621 engagement must be prioritized. Inclusive decision-making processes ensure that all relevant actors,
622 including local communities, are actively involved¹⁹⁵, and capacity-building initiatives empower
623 stakeholders to contribute meaningfully to knowledge production¹⁹⁶. Together, these actions can create a
624 robust foundation for transdisciplinary collaboration, ultimately leading to more innovative and impactful
625 outcomes.

626 **eToC blurb**

627 Achieving phosphorus circularity is a key challenge to realizing sustainable phosphorus use, and recycling
628 is a major route to accomplish this goal. This Review explores global barriers to phosphorus recycling and
629 discusses approaches to overcome the technical, economic and societal challenges in attaining sustainable
630 phosphorus management.