



## Research



**Cite this article:** Yenen TB *et al.* 2025 Mineral phase changes during intervertebral disc degeneration. *Proc. R. Soc. A* **481**: 20240488. <https://doi.org/10.1098/rspa.2024.0488>

Received: 11 July 2024

Accepted: 2 April 2025

**Subject Areas:**

chemical biology

**Keywords:**

minerals, phase, intervertebral disc, degeneration

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Electronic supplementary material is available online at <https://doi.org/10.6084/m9.figshare.c.7765060>.

# Mineral phase changes during intervertebral disc degeneration

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Intervertebral disc disease is a common cause of pain and neurological deficits known to be associated with degeneration and calcification of discs in humans and animals. Here, we analysed samples of herniated disc material and compared it to material taken from non-herniated discs following surgical treatment in dogs. Our clinical approach to these cases allowed collection of approximately 100 samples providing a unique opportunity for a substantial case-controlled

study, an opportunity not available to the human neurosurgeon. We analysed all samples using Fourier transform infrared (FTIR) spectroscopy, as well as subsets of 10 samples by X-ray diffraction (XRD) and two samples by transmission electron microscopy (TEM). We show the majority of herniated samples have FTIR spectra consistent with the presence of crystalline hydroxyapatite (HAp), whereas most non-herniated discs showed spectra consistent with amorphous phosphate. XRD and TEM confirmed these findings and the two samples examined with TEM identified the amorphous material as amorphous calcium phosphate (ACP) nanoparticle clusters of approximately 20 nm diameter and the crystalline HAp as polycrystalline needles up to 100 nm in length. Our data suggest that the intervertebral disc degenerative process involves conversion of ACP into crystalline HAp, which might predispose a disc to herniate.

## 1. Introduction

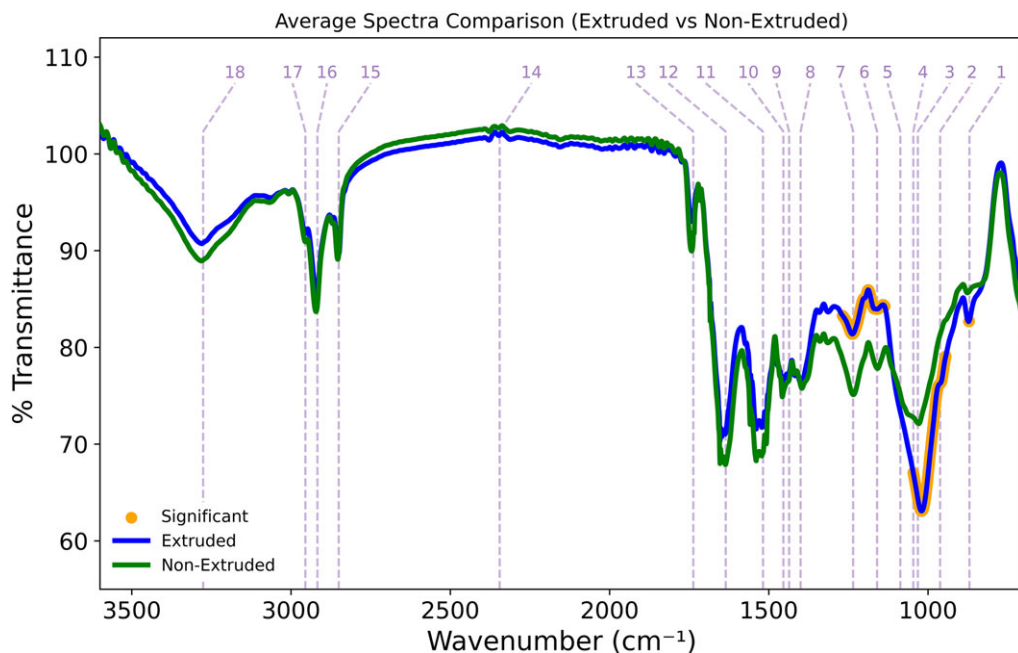
Intervertebral disc disease is characterized by the degeneration of disc(s) that separate the bones of the spine leading to localized pain that frequently extends to the legs and arms. It affects approximately 5% of the population in high-income countries and costs the UK National Health Service an estimated £5 billion p.a. (source Manchester Metropolitan University). Moreover, the disease is not restricted to humans. Chondrodystrophic dogs, characterized by a short-legged conformation, such as the dachshund and French bulldog, are especially prone to frequent, severe disease.

The intervertebral disc of mammals consists of the inner gelatinous nucleus pulposus, the fibrous annulus fibrosus, which creates a border around the nucleus, and the cartilaginous endplates, which bind the disc to the adjacent vertebral bodies. The healthy nucleus pulposus contains mainly notochordal cells surrounded by a matrix rich in proteoglycan and collagen (type II) as well as hyaluronic acid, creating a high osmotic pressure leading to high water content [1–6]. Calcification of intervertebral discs, in both the annulus fibrosus and the nucleus pulposus, is seen in disc degeneration of dogs and humans [7–11]. In dogs, calcification is associated with disc extrusion [12,13], which can lead to paralysis and incontinence or even death, while in humans, it is associated with discogenic pain [14–16]. However, while calcification is well recognized in the pathogenesis of intervertebral disc disorders, the relationship of the mineral composition to clinical disease is not known.

At the cellular level, amorphous calcium phosphate (ACP) particles are benign but ectopic hydroxyapatite (HAp) crystals are pro-inflammatory so the mineral phase could well dictate outcomes in calcified discs [17]. Indeed, biological HAp has been found in both degenerated [18] human intervertebral discs and in joints of patients with osteoarthritis [19]. In addition, large crystal-sized HAp accumulations (80–130 nm) were found in ovine intervertebral discs and were considered to be related to ageing [20]. By contrast, little is known about the presence of ACP in intervertebral discs despite many studies supporting the possibility that ACP could be the precursor form of HAp in other biological situations such as bone formation [21–24].

During surgery for extruded intervertebral discs in dogs, in addition to removing herniated nucleus pulposus material from the vertebral canal, it is good practice to clear out non-extruded neighbouring discs, a process known as disc fenestration which has been shown to be prophylactic in preventing future extrusion [25,26]. This provides a unique opportunity to compare the mineral phase of the calcified deposits in a clinically degenerative situation with very closely matched non-extruded discs, in a way that is not ethically possible in humans.

Here, we found that the major mineral phase present in extruded canine discs is crystalline HAp, whereas a significant proportion of non-extruded discs in the same animals contain ACP, providing evidence that a switch from amorphous to crystalline calcification appears to precede and might possibly trigger disc extrusion.



**Figure 1.** Group averaged FTIR spectra for extruded (blue,  $n = 58$ ) and non-extruded (green,  $n = 46$ ) intervertebral disc samples. Inter-group differences in transmittance were evaluated by two-sample  $t$ -tests; wavenumbers with Benjamini–Hochberg corrected ( $\alpha 0.1\%$ )  $p$ -values below 0.05 are highlighted (orange) and show consistent differences between extruded and non-extruded samples in (i) the carbonate  $\nu_2$  peak ( $870\text{ cm}^{-1}$ ), (ii) the phosphate peak region between  $800$  and  $1100\text{ cm}^{-1}$  and (iii) the amide region between  $1200$  and  $1300\text{ cm}^{-1}$ . Peak assignments are detailed in Table 1.

## 2. Results

In total, 57 clinical cases were included from 56 individual dogs (one dog suffered two separate episodes), all of which were chondrodystrophic or chondrodystrophic-mix dogs. Clinical details are contained in electronic supplementary material, table S1. The clinical cases generated 104 individual samples of intervertebral disc material, of which 58 were samples of extruded (herniated) material removed at surgical decompression, and 46 were non-extruded (control) samples of nucleus pulposus taken by fenestration of non-herniated discs.

### (a) Fourier transform infrared analysis of intervertebral discs

Fourier transform infrared (FTIR) spectra of 104 disc samples, 58 extruded and 46 non-extruded, were produced. The raw data were afflicted with high-frequency noise and all spectra were pre-processed via wavelet denoising as described in the Materials and methods and as shown in electronic supplementary material, figures S2 and S3. No other pre-processing techniques, such as baseline normalization, were employed. Group average spectra were generated for the extruded and non-extruded sample groups (figure 1) and their transmittance profiles were found to overlay closely across much of the IR spectrum—aside from several key loci where pronounced inter-group differences were noted. Statistically significant inter-group spectral differences were found at (i) the carbonate ( $\text{CO}_3^{2-} \nu_2$ ) absorption band at  $870\text{ cm}^{-1}$ , (ii) the phosphate absorption region between  $950$  and  $1100\text{ cm}^{-1}$  ( $\text{PO}_4^{3-} \nu_{3a-c}$  at  $1030$ – $1090\text{ cm}^{-1}$  and  $\text{PO}_4^{3-} \nu_1$  at  $962\text{ cm}^{-1}$ ), (iii) the glycosidic bond absorbance band at  $1160\text{ cm}^{-1}$  and (iv) the amide absorption region between  $1200$  and  $1300\text{ cm}^{-1}$ . Transmittance differences were also visible at the amide II ( $1518\text{ cm}^{-1}$ ) and amide I ( $1635\text{ cm}^{-1}$ ) regions, but these did not meet statistical significance after applying Benjamini–Hochberg correction.

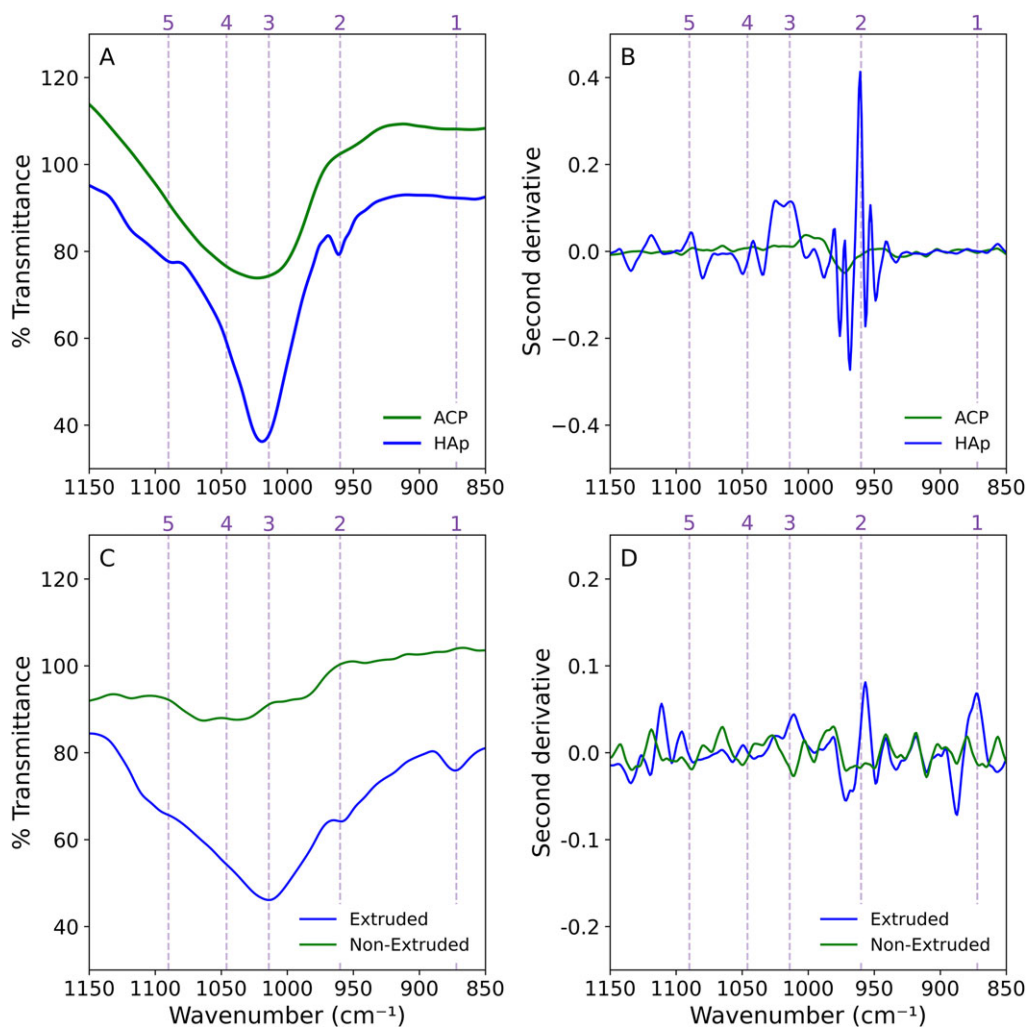
**Table 1.** FTIR absorption peak assignments.

index	wavenumber (cm <sup>-1</sup> )	assignment	reference
1	870	carbonate, $\nu_2$	[27,28]
2	962	phosphate, $\nu_1$	[29]
3	1032	phosphate, $\nu_{3c}$	[29]
4	1046	phosphate, $\nu_{3b}$	[29]
5	1087	phosphate, $\nu_{3a}$	[28,29]
6	1160	glycoside, C-O-C	[30]
7	1235	amide III	[30]
8	1400	carbonate, $\nu_3$	[27,28]
9	1436	carbonate, $\nu_3/\nu_4$	[28]
10	1454	carbonate, $\nu_3/\nu_4$	[28]
11	1518	amide II	[31]
12	1635	amide I	[30]
13	1737	ester fatty acid, C=O	[30]
14	2345	carbon dioxide, $\nu_3$	[32]
15	2850	aliphatic CH/CH <sub>2</sub> , $\nu_1$	[30]
16	2917	aliphatic CH/CH <sub>2</sub> , $\nu_3$	[30]
17	2955	aliphatic CH <sub>3</sub> , $\nu_3$	[33]
18	3276	hydroxyl group, $\nu_3$	[34]

Notably, the directionality of the inter-group absorption differences was not uniform. The extruded group average spectrum exhibited greater absorbance at the carbonate and phosphate loci and simultaneously reduced absorbance at the glycoside and amide loci. This finding was indicative of compositional sample differences between the groups and prompted close inspection of individual spectra at the significant wavenumber regions.

Case-by-case visual assessment revealed that while the vast majority of both extruded and non-extruded spectra contained a prominent  $\text{PO}_4^{-3} \nu_3$  peak suggesting the presence of calcium phosphate in both groups, the intensity and shape differed greatly between groups: 50 of 58 of the extruded spectra evinced a sharp  $\text{PO}_4^{-3} \nu_3$  peak, while only 10 of 46 non-extruded samples exhibited this peak shape. Furthermore, 47 of the 58 extruded spectra contained a distinct  $\text{PO}_4^{-3} \nu_1$  peak at approximately  $960 \text{ cm}^{-1}$ , while in the non-extruded group this vibration was only resolvable in seven of 46 samples. Since the  $\text{PO}_4^{-3} \nu_1$  peak at approximately  $960 \text{ cm}^{-1}$  is attributed specifically to phosphate vibrations in a crystalline (hydroxy)apatite environment [35], and  $\text{PO}_4^{-3} \nu_3$  signal was observable in the majority of samples, the disparity in  $\text{PO}_4^{-3} \nu_1$  intensity and resolvable features was suggestive of inter-group differences in mineral crystallinity.

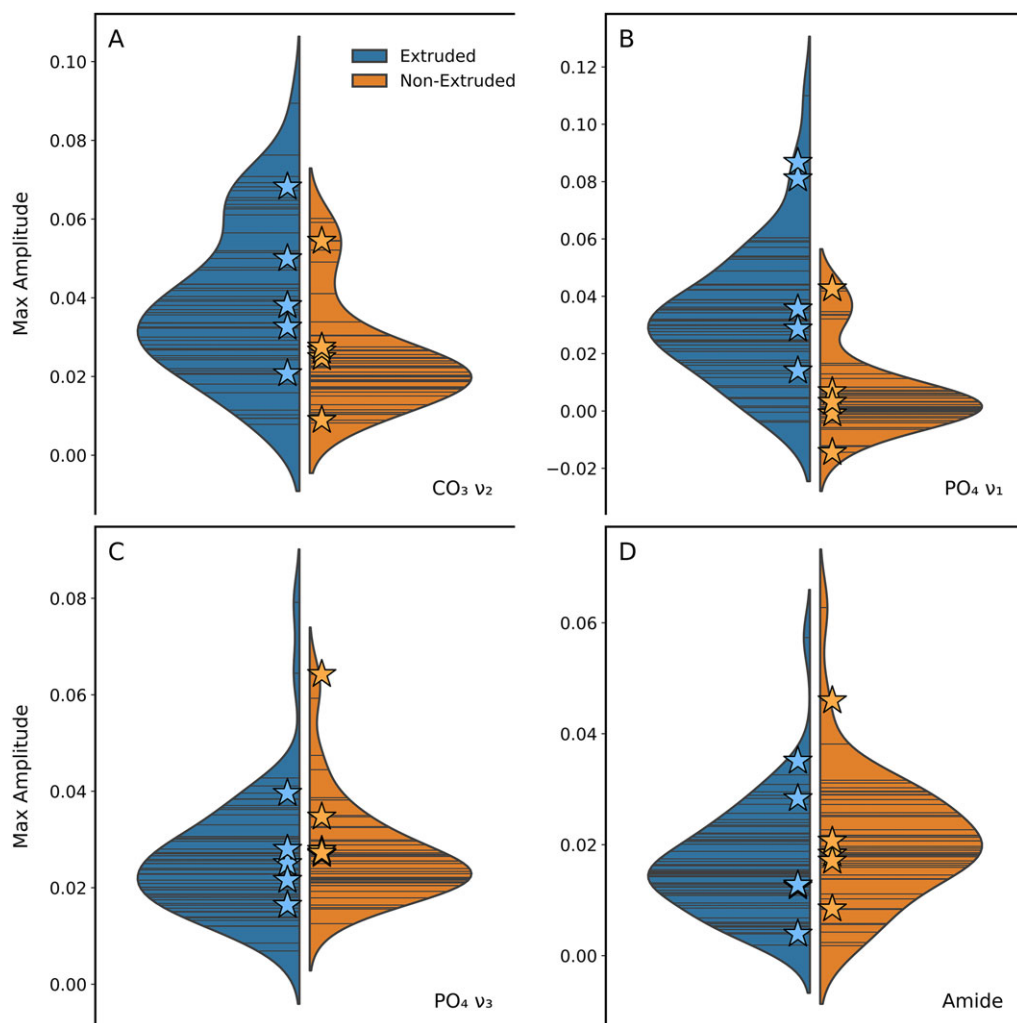
To verify phosphate mineral spectral features, FTIR spectroscopy was conducted with reference calcium phosphate minerals, ACP and crystalline HAp. In accordance with Gadaleta *et al.* [35], this analysis revealed clear differences in the phosphate band signal structure (at  $800\text{--}1100 \text{ cm}^{-1}$ )—the crystalline material evincing a sharper and more intense  $\text{PO}_4^{-3} \nu_3$  absorption peak (figure 2A) along with a distinct or well-resolved  $\text{PO}_4^{-3} \nu_1$  absorbance band and correspondingly pronounced maxima in the second derivative (figure 2B). Conversely, ACP exhibited a broad phosphate absorption band in the primary spectrum (figure 2A) and negligible second derivative signal at the expected  $\text{PO}_4^{-3} \nu_3$  and  $\text{PO}_4^{-3} \nu_1$  wavenumbers (figure 2B). Similar, albeit less extreme, differences in spectral profiles were also noted in the IR spectra of most extruded and non-extruded clinical samples, suggesting that extruded samples are more likely



**Figure 2.** FTIR denoised spectral (A) and second derivative signal (B) comparison of the phosphate absorption region for non-crystalline and crystalline calcium phosphate mineral reference compounds, and for exemplar extruded and non-extruded clinical samples (C,D). The sharp  $\text{PO}_4^{3-} \nu_1$  signal (label 2; approx.  $960 \text{ cm}^{-1}$ ) in spectrum (A) and in the second derivative signal (B) is diagnostic for crystalline HAp and absent for ACP. This crystallinity marker is present in the extruded sample and not evident in the non-extruded sample (C,D). Peak labels are: (1)  $\text{CO}_3^{2-} \nu_2$ , (2)  $\text{PO}_4^{3-} \nu_1$ , (3)  $\text{PO}_4^{3-} \nu_{3c}$ , (4)  $\text{PO}_4^{3-} \nu_{3b}$  and (5)  $\text{PO}_4^{3-} \nu_{3a}$ .

than their non-extruded counterparts to be abundant in the crystalline calcium phosphate mineral HAp (figure 2C,D at vertical line traces 2 and 3; electronic supplementary material, figure S4). Extruded samples also exhibited a distinct  $\text{CO}_3^{2-} \nu_2$  signal at  $870 \text{ cm}^{-1}$ , consistent with the presence of carbonated, biological HAp [36]. This feature was expectedly absent from the synthetic reference material.

To assess the second derivative signal at group level, peak amplitudes (in the second derivative spectra) were quantified at key wavenumbers for the  $\text{CO}_3^{2-} \nu_2$ ,  $\text{PO}_4^{3-} \nu_1$ ,  $\text{PO}_4^{3-} \nu_3$ , and amide peaks, and group probability density functions estimated via kernel density estimation (KDE) (figure 3). Consistent with the denoised spectral analysis reported above (figures 1 and 2), the second derivative spectra of extruded samples (blue) exhibited greater peak amplitudes at the  $\text{CO}_3^{2-} \nu_2$  (figure 3A) and  $\text{PO}_4^{3-} \nu_1$  (figure 2B) absorption bands than their non-extruded (orange) counterparts. This difference was particularly striking at the  $\text{PO}_4^{3-} \nu_1$  band (Welch's  $t$ -test  $p$ -value

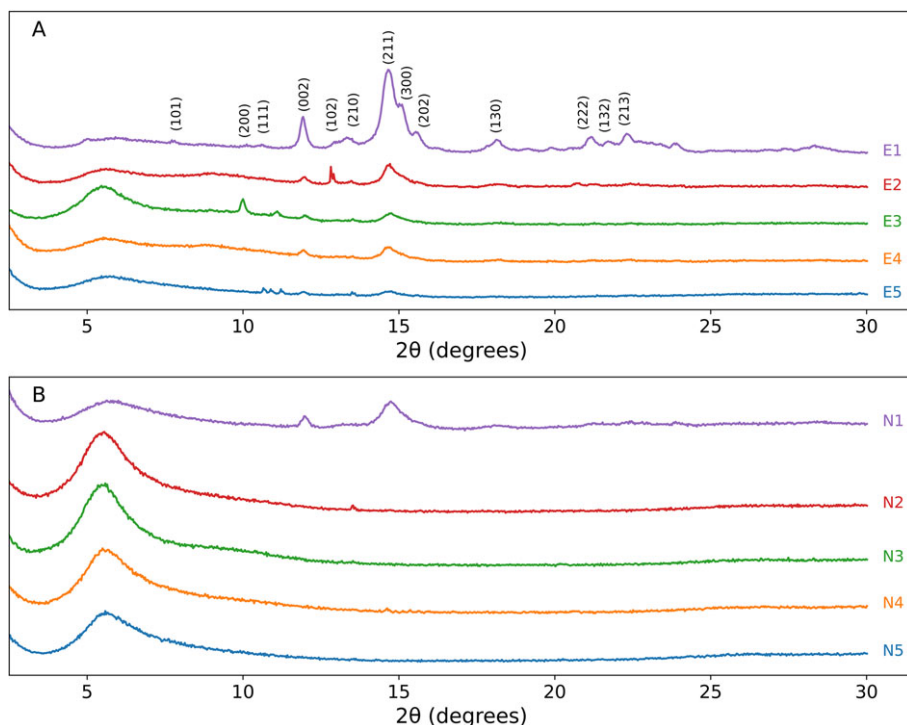


**Figure 3.** Violin plots displaying the distribution of FTIR spectra second derivative maximum amplitude values for extruded (blue) and non-extruded (orange) samples across four vibrational regions (A)  $\text{CO}_3^{2-} \nu_2$ , (B)  $\text{PO}_4^{3-} \nu_1$ , (C)  $\text{PO}_4^{3-} \nu_3$  and (D) amide. The width of each violin indicates the KDE, representing the probability density of the data at different amplitude values. Internal stick lines represent the individual observations within each category, while the overlaid star markers highlight the specific samples down-selected for XRD analysis. The density is normalized by count to reflect the number of observations in each region. A statistically significant difference between extruded and non-extruded samples was found in the  $\text{PO}_4^{3-} \nu_1$  region (subplot b;  $p = 0.0483$ , Welch's  $t$ -test).

0.048), where 50 of 58 extruded samples had a second derivative peak amplitude (median 0.033 units) greater than the upper quartile value (0.018 units) of the non-extruded samples (median 0.007 units). The  $\text{PO}_4^{3-} \nu_1$  second derivative peak amplitudes of the non-extruded samples appeared to exhibit a bimodal distribution, with greatest density at ca. zero units and a smaller subsidiary probability peak at 0.04 units. In contrast to the  $\text{CO}_3^{2-} \nu_2$  and  $\text{PO}_4^{3-} \nu_1$  derivative data, and despite large differences in group average spectra absolute signal intensity (figure 1), second derivative peak amplitudes at the  $\nu_3 \text{PO}_4^{3-}$  (figure 3C) and amide (figure 3D) absorbance regions were similar for both groups—indicating the presence of similar compositional components in both groups.

The second derivative analyses accorded with findings from visual spectral classification, indicating a substantial inter-group difference in the frequency of samples containing detectable





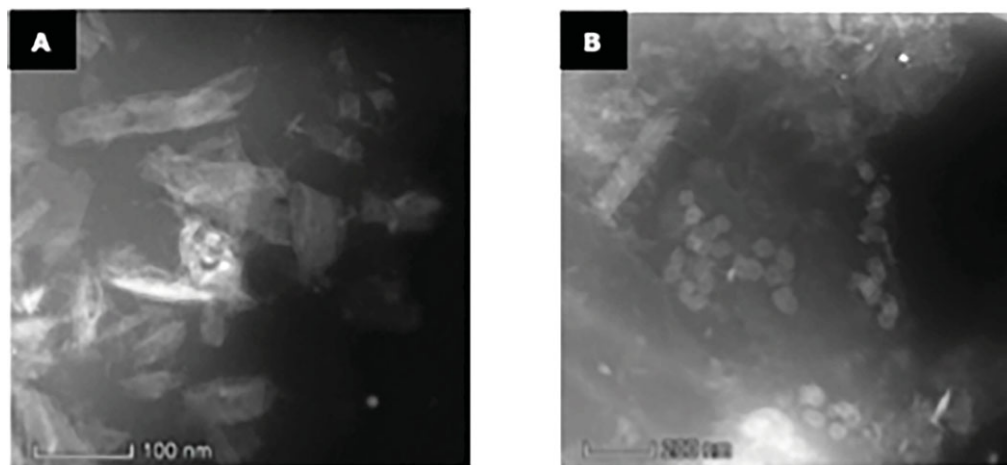
**Figure 4.** Powder X-ray diffraction patterns of representative extruded (A) and non-extruded (B) samples, indexed against hydroxyapatite reference structure ICDD 74-0565. All extruded samples E1–E5 and one non-extruded sample (N1) have crystalline hydroxyapatite peaks, while the remaining non-extruded samples (N2–N5) are all X-ray amorphous.

crystalline HAp. Notably, intra-group spectral differences were also sustained, with a sub-population of non-extruded controls appearing to exhibit crystalline spectral features too.

To corroborate the spectroscopic analyses and confirm the presence/absence of crystalline material, a series of samples from both groups (extruded and non-extruded) were selected for X-ray diffraction (XRD). These samples (figure 3, star markers) were selected based on the second derivative distributions of both groups, in order to represent the range of spectral profiles displayed and especially to ensure that distinct sample sub-populations (e.g. non-extruded  $\text{PO}_4^{3-} \nu_1$  low-signal versus high-signal within each group) were represented.

## (b) Powder X-ray diffraction results

Five extruded and five non-extruded discs were selected for XRD analysis because they were considered to be representative of the range of second derivative maxima signal at the  $\text{PO}_4^{3-} \nu_3$ ,  $\text{PO}_4^{3-} \nu_1$  and  $\text{CO}_3^{2-} \nu_2$  functional group regions of the FTIR spectra of the 58 extruded and 46 non-extruded tested above (figure 3). The XRD profiles of all five of the extruded (figure 4A, traces E1–E5) and one of the five non-extruded disc materials (figure 4B, trace N1) were indicative of crystalline HAp by the presence of significant  $2\theta$  peaks at around  $15^\circ$ , indexed according to the Inorganic Crystal Structure Database file 74-0565 [37]. Phase identification was confirmed via structural Rietveld refinements. The details of the powder X-ray diffraction (PXRD) data fitting strategy are reported in the electronic supplementary material together with the Rietveld plots for all the samples containing a crystalline fraction (electronic supplementary material, figures S5–S10). The other four non-extruded samples showed profiles without any significant Bragg peaks indicating that these contained amorphous material only (figure 4B, traces N2–N5; electronic supplementary material, figure S11). Structural and microstructural parameters as obtained from



**Figure 5.** HAADF-STEM images of the extruded disc (A) material showing agglomerate of needles indicative of hydroxyapatite nanocrystals (A) and non-extruded (B) disc material showing small amorphous spheres and only some needle-like particles. Scale marker bars are 100 nm and 200 nm in A and B, respectively.

the refinements are reported in electronic supplementary material, table S2. Crystal sizes of the crystalline HAp mineral fraction in the extruded specimens ranged from 10 to 15 nm, as estimated by sample contribution to diffraction peak broadening by implementing the Scherrer equation in the whole-pattern analysis (electronic supplementary material, table S2 and figures S5–S10).

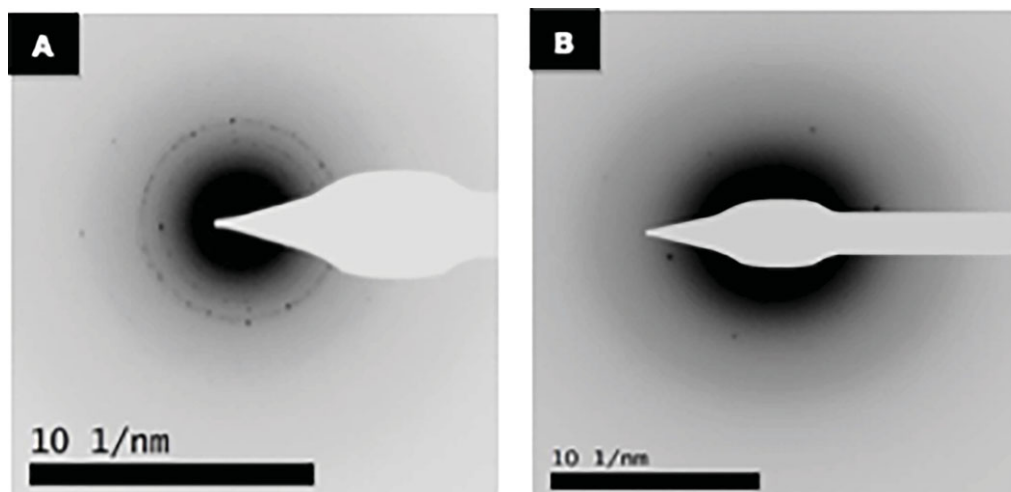
Importantly, the non-extruded sample that contains crystalline HAp (figure 4B, trace N1; electronic supplementary material, figure S10) was the sample selected to represent the non-extruded  $\text{PO}_4^{3-}$   $\nu_1$  positive sub-population (figure 3B, highest orange star), supporting the sensitivity of that spectral feature for HAp detection across clinical groups. The crystallite size for this sample is estimated to be less than 8 nm (electronic supplementary material, table S2), slightly smaller than the crystallite size range of the extruded samples.

### (c) Electron microscopy analysis of two representative samples

TEM analysis of one extruded (E2) and one non-extruded (N3) intervertebral disc sample was supportive of both the FTIR and XRD findings. STEM images of the extruded disc material revealed a continuous, gel-like or amorphous phase containing many nanoscale, needle-like agglomerates or deposits within the matrix (figure 5A; the crystalline needles scatter more strongly than the amorphous matrix and so have a white contrast in the HAADF images (see also electronic supplementary material, figure S12a,b). STEM-EDX elemental analysis indicated that the needle-like agglomerates were Ca, P and O rich (electronic supplementary material, figure S13). Furthermore, electron diffraction of the extruded sample showed four polycrystalline rings of spacing 2.82 (strong ring), 3.2, 3.5 and approximately 4.4 Å consistent with those expected for HAp [36] (figure 6A).

STEM images of the non-extruded disc material revealed gel-like deposits similar to the extruded sample. The deposits within the matrix consisted of nanoscale, spherical agglomerates identified as amorphous material by (i) the weaker scattering in the image (figure 5B; the deposits are smaller, rounder and less white than the needles in the extruded material in figure 5A) and (ii) no visible atomic lattice in high-resolution imaging (electronic supplementary material, figure S12c,d). In addition, electron diffraction analysis confirmed that the deposits in the gel are predominately amorphous by a lack of sharp rings and spots (figure 6B). The spherical agglomerates in the non-extruded sample that were amorphous by electron diffraction were found to be Ca, P and O rich by energy dispersive X-ray spectroscopy (EDX), consistent with an ACP phase (electronic supplementary material, figure S14).





**Figure 6.** Electron diffraction pattern of extruded (A) and non-extruded (B) intervertebral disc material note. (A) Electron diffraction pattern of extruded IVD material shows four rings at 2.82 (strong ring), 3.2, 3.5 and approximately 4.4 Å for which the ring spacings are indicative of hydroxyapatite (see text). (B) Electron diffraction pattern of non-extruded IVD material showing diffuse rings with very few diffraction spots that indicate the tested material is predominately amorphous. The scale marker bars are  $10 \text{ nm}^{-1}$  in both patterns shown.

### 3. Discussion

This study produces new information regarding the mineral form of calcified degenerate intervertebral discs in 56 chondrodystrophic dogs, revealing that the discs contain either crystalline HAp or ACP nanoparticles. ACP nanoparticles were found only in discs that had not undergone extrusion (or herniation), whereas crystalline HAp was found in both herniated and non-herniated discs. These findings suggest that canine intervertebral disc degeneration, at least in this group of chondrodystrophic dogs, involves a transformation (change of phase) from an initially ACP phase into crystalline HAp nanoparticles. It is possible that this phase change may then predispose the discs to extrusion.

To minimize the possibility of mineral phase changes post-sample collection, freeze-drying was used for sample preparation prior to analysis. Freeze-drying has been shown to conserve the phase and morphology of minerals, although it can cause particle size reduction [27]. It is also possible that the mineral phase of samples of nucleus pulposus changed following extrusion, i.e. once inside the vertebral canal and outside the disc, a possibility that could be tested in the future by comparing nuclear material remaining within an extruded disc with the herniated, extruded material. This, however, seems unlikely given that some non-extruded samples also contained crystalline material.

In this study, all 102 samples were analysed by FTIR spectroscopy to demonstrate statistical differences in the second derivative maxima signals of the phosphate band at  $800\text{--}1100 \text{ cm}^{-1}$  that suggest the extruded disc material contained crystalline HAp (figures 2 and 3). A sample set of five extruded and five non-extruded discs shown to be representative of this FTIR spectroscopy analysis was taken forward for XRD measurement (figure 4). Results confirmed the findings that ACP nanoparticles were found only in discs which had not undergone extrusion, whereas crystalline HAp was found in all five extruded discs and one non-extruded disc. A single sample from each XRD group, which from visual evaluation of FTIR spectra appeared representative of the bulk of samples from each group (figure 3), were further submitted for transmission electron microscope analysis to reveal the nanoscale phase and internal structure of the investigated materials (figure 5) and to confirm that when crystalline the HAp phase forms needles (figures 5 and 6).

A small number of FTIR spectra from the extruded samples did not reveal any of the prominent phosphate or carbonate peaks consistent with the presence of biological HAp [28–31]. Although it is possible that these extruded samples were not calcified or did not include crystalline HAp, it is more likely that a sampling error in surgery might be the reason for these findings, given the clinical nature of this study.

Crystal sizes in the samples analysed by XRD were in the range of 5–15 nm in the five extruded and the single non-extruded sample, in line with previously reported crystal sizes for biological HAp [32]. The size of the crystalline needles observed by TEM was bigger (50–100 nm) suggesting these are polycrystalline agglomerates (that could have grown from the smaller ACP particles seen in the non-herniated disc material). Previous literature suggests that the size of the crystalline particles of HAp in bone increases with maturity [31,33,34]. It is possible that crystal, and agglomerate sizes of the HAp within the intervertebral discs increases with the stage of degeneration, and may also be associated with extrusion, but larger sample size analysis is required to make a more definitive statement regarding this.

A few of the non-extruded samples showed FTIR spectra consistent with crystalline HAp (and one had an XRD pattern confirming this; N1 in figure 4). This is not unexpected if the degenerative process involves a transformation of ACP to crystalline HAp prior to extrusion, since it may be that a critical mass and/or size of HAp is needed before herniation/extrusion occurs. Clearly, it is likely that multiple other factors such as mechanical stress, dehydration and loss of proteoglycans and genetic predisposition are also involved in intervertebral disc disease and herniation [38]. However, if discs containing crystalline HAp could be identified before they have herniated, e.g. by calibration to MRI, it may be possible to target them with a prophylactic therapy such as surgical fenestration, chemonucleolysis or laser ablation [25,26,38,39].

This study sheds some light on the process of calcification that is an important part of intervertebral disc degeneration in dogs and people and has been associated with disease in both species. Our findings lead us to speculate that calcification is a process in which initially an ACP phase is formed within the nucleus pulposus, and that this then transforms into a crystalline HAp. This is not unexpected given what is known about the maturation of bone and teeth and the role of HAp in these biological materials. We can only speculate, at this stage, on why the calcium phosphate undergoes such a phase change; possibly, it is associated with the loss of a stabilizing substance that maintains the amorphous phase only until a certain level of degeneration of the disc.

A recent review of the role of calcification in human intervertebral disc disease also points to the role of calcium pyrophosphate [11], but we did not identify this specific material in our samples. There may indeed be some differences in the pathological process between dogs and humans, although it is clear that calcium has a significant role in intervertebral disc disease in both species, with canine intervertebral disc disease having been suggested to be a good natural model of the disease in humans [40]. If our hypothesis regarding the maturation of the calcification in intervertebral discs is correct, it may further be possible to intervene in this process to slow down or even prevent or reverse the process and perhaps reduce the rate of intervertebral disc herniation as a result.

Further investigation into the topographical and mechanical properties of intervertebral disc samples is needed to better describe and verify these findings, as well as attempt to define the biomechanical features of intervertebral disc material.

## 4. Conclusion

Our results shed new light on the process of calcification in intervertebral disc degeneration and disease. It appears that an ACP phase within discs is transformed into crystalline HAp, and this might either be due to an unidentified trigger or to the loss of a stabilizing agent. This change in the mineral phase within the intervertebral disc appears to be associated with herniation of the disc, perhaps due to alterations in biomechanical properties. The identification of this transformation as an integral part of the degenerative process requires further verification

potentially by analysis of a larger number of samples with techniques such as XRD and TEM (as illustrated here), but if confirmed opens up the possibility for further work, which might lead to new therapies and/or prophylactic measures to treat and/or reduce the incidence of this frequently disabling disease in both humans and dogs.

## 5. Material and methods

### (a) Inclusion criteria

The study population consisted of chondrodystrophic dogs with intervertebral disc extrusion that underwent surgical treatment at two different institutions, the Queen's Veterinary School Hospital, University of Cambridge and Dick White Referrals, Newmarket, between October 2020 and June 2021. Dogs were included if they had a diagnosis of intervertebral disc extrusion made by either magnetic resonance imaging (MRI) or computed tomography (CT) and confirmed surgically. Details recorded for each case included breed, age, sex and location of extrusion. The study was approved by the Ethics and Welfare Committee of the Department of Veterinary Medicine, University of Cambridge, and all methods were performed in accordance with the relevant guidelines and regulations. The study is reported in accordance with relevant ARRIVE guidelines. All data associated with this paper are openly available from the University of Cambridge Data Repository.

### (b) Study and control samples

Study samples consisted of extruded disc material collected during decompressive surgery. Material was collected directly from the surgically exposed vertebral canal by physical extraction with a variety of instruments following laminectomy performed with a high-speed burr [41].

Control samples consisted of material collected during prophylactic fenestration of adjacent non-extruded discs [42,43]. These discs were confirmed to be non-extruded following examination of their imaging (MRI or CT). Both extruded and non-extruded disc materials were placed into sterile plain plastic sample pots and held at +4°C for a maximum of 48 h. They were then transferred and stored in a -20°C freezer until the freeze-drying process.

If the sample was large enough, half of the collected material was separated for histopathology. These samples were fixed in 10% formalin for 24–48 h and then transferred into pots with 70% ethanol until histopathologic investigation. For slide preparation, samples were embedded in paraffin wax, cut to thicknesses of 4–5 µm, and stained with haematoxylin and eosin. None of the samples were decalcified during preparation. Slides were then observed with an optical microscope at a magnification of 40× and were evaluated by a European Boarded Anatomic Pathologist. Twenty-seven samples, of which 21 were extruded and six were non-extruded, were evaluated in this way, and all were identified as nucleus pulposus because they contained notochordal cells and/or chondrocyte-like cells.

### (c) Sample preparation

All samples were freeze-dried prior to mineral analysis for a total of 17 h. After placing into the freeze-dryer (VirTis Advantage Plus), they were cooled from +5°C to -35°C at a rate of -0.19°C min<sup>-1</sup> and at a pressure of 400 mTorr. This was followed by the drying phase, where the temperature was gradually increased to +20°C while the pressure was held at 80 mTorr. Samples were then brought to room temperature and stored until mineral analysis.

### (d) Fourier transform infrared analysis

FTIR analysis was performed with a Shimadzu IRPrestige21 Fourier Transform Infrared Spectrophotometer equipped with a Golden Gate ATR accessory (Specac Ltd.). FTIR scans of

samples covered wavenumbers  $400\text{--}4000\text{ cm}^{-1}$ . The spectrometer was set to a scan speed of  $2.8\text{ mm s}^{-1}$  and had a resolution of  $2\text{ cm}^{-1}$ . The ATR sample holder was cleaned twice with 100% ethanol between each sample analysis. Samples were analysed in the form they left the freeze-drying process, with no grinding or further handling/sample prep. Due to the small size of the ATR sample holder and the lack of sample homogeneity, most samples (i.e. where possible) were subsampled up to three times and each scan recorded. For samples with multiple scans, when the three spectra differed (21 of 104 samples), the best-conditioned spectrum was included in the data analysis, with selection considering the following factors: baseline stability, degree of high-frequency noise and resolution of the phosphate absorbance region ( $900\text{--}1200\text{ cm}^{-1}$ ).

### (e) Interpretation of Fourier transform infrared spectra of intervertebral discs

All FTIR spectra were treated via wavelet denoising using the PyWavelets library in Python, employing the *coif5* wavelet, with level 3 decomposition, and hard thresholding at threefold of the Universal Threshold (see electronic supplementary material, figure S3 and figure S15 for Python code). Group average spectra were calculated for extruded and non-extruded samples by averaging the denoised spectra across all samples within each group. Two-sample *t*-tests were performed at each wavenumber to identify statistically significant differences between the groups. The assumption of unequal variances was applied. To account for multiple comparisons, the *p*-values were corrected using the Benjamini–Hochberg false-discovery rate method with an  $\alpha$  threshold of 0.0001.

Thereafter, second derivative analysis was performed at absorbance regions which had been identified as significantly differing between the average spectra of the two clinical groups, namely:  $\text{CO}_3^{2-} \nu_3$  ( $865\text{--}890\text{ cm}^{-1}$ ),  $\text{PO}_4^{3-} \nu_1$  ( $955\text{--}965\text{ cm}^{-1}$ ),  $\text{PO}_4^{3-} \nu_3$  ( $1015\text{--}1095\text{ cm}^{-1}$ ) and amide ( $\text{cm}^{-1}$ ) absorbance bands. KDE was applied to visualize the distribution of maximum amplitude values across sample groups using a Gaussian kernel for smoothing. The bandwidth for the KDE was automatically selected using Scott's rule, adjusting the level of smoothing based on the data distribution. Violin plots were generated using seaborn, with width scaled according to the number of observations. Welch's *t*-test was used to test the statistical significance of inter-group differences in second derivative peak amplitude at critical loci. Before each statistical test/analysis, a Shapiro–Wilk test was performed to confirm that the data were normally distributed.

### (f) Calcium phosphate mineral reference compounds

Hydroxyapatite reference material, calcium chloride, magnesium chloride, disodium phosphate and tris buffer were all obtained from Sigma-Aldrich. Hydroxyapatite reference material was used as received without modification, while ACP was synthesized. Since ACP is metastable to HAP and undergoes mineral transformation rapidly at room temperature the following procedure was employed to prepare it: to 45 ml of tris ( $300\text{ mMol l}^{-1}$ ) buffered (pH 9.09) disodium phosphate solution ( $110\text{ mMol l}^{-1}$ ), 5 ml of tris ( $300\text{ mMol l}^{-1}$ ) buffered (pH 8.09) calcium solution, containing calcium chloride ( $800\text{ mMol l}^{-1}$ ) and magnesium chloride ( $200\text{ mMol l}^{-1}$ ) was added under vigorous stirring. The resulting precipitates were immediately collected by centrifugation, washed three times in ice-cold pH 9 water, three times in ice-cold acetone, dried under vacuum and then stored at  $-80^\circ\text{C}$ . Both the reference materials were analysed with XRD to verify that they were comprised from the appropriate mineral phase.

### (g) X-ray diffraction

PXRD analysis was performed on five extruded (study) and five non-extruded (control) samples that were previously analysed by FTIR spectroscopy. Samples were selected to represent the group ranges of FTIR spectral profile, comprising datum at the modes, upper and lower ends of the second derivative distributions. In particular, since the non-extruded samples exhibited a

bimodal distribution in second derivative amplitude at the critical  $\text{PO}_4^{3-} \nu_1$  absorbance band, we ensured that both sub-populations were represented in the XRD analyses—one sample (N1) was taken from the high-signal non-extruded subgroup and the other four from across the low-signal subgroup. The freeze-dried disc materials were gently crushed with an agate pestle & mortar and placed on a low background sample holder made of silicone. PXRD data were collected in Bragg–Brentano geometry with a Bruker D8 Advance diffractometer equipped with a Mo  $K\alpha$  radiation source, and a Lynxeye XE-T linear position sensitive detector with the following parameters:  $2\theta$  range: 5–40°; step size 0.015°; time per step 1 s; 50 kV  $\times$  40 mA.

## (h) Transmission electron microscopy analysis

Transmission electron microscopy (TEM) was performed on two representative samples (one extruded and one non-extruded), after they had been freeze-dried and analysed with FTIR spectroscopy and XRD. A Thermo Fisher Scientific Helios G4 CX scanning electron microscope was used for scanning electron microscopy energy dispersive X-ray (SEM-EDX) analysis, and a Thermo Fisher Scientific Titan [3] Themis G2 SuperX 300 kV scanning TEM with a Gatan OneView camera was used for TEM, scanning transmission electron microscopy (STEM), energy dispersive X-ray spectroscopy (STEM-EDX) and electron diffraction analysis.

SEM-EDX was initially performed on the extruded sample to determine the elements present in the material before any further sample preparation. A small amount of each of the two freeze-dried disc materials was then suspended in methanol after grinding in a pestle and mortar before being directly drop-cast onto a holey carbon TEM support film (EM Resolutions Ltd.) and air-dried. TEM images of the disc materials were generated with high-angle annular dark-field imaging by scanning transmission electron microscopy (HAADF-STEM). In addition, electron diffraction patterns were obtained for both samples using a selected area aperture of approximately 200 nm diameter at the image plane. Furthermore, STEM-EDX analysis was performed for elemental mapping (data acquired and processed in the Thermo software Velox).

## (i) Raw data

Raw data are available upon request.

**Ethics.** Ethical approval for this study was given by the Welfare and Ethics Committee of the Dept of Veterinary Medicine, University of Cambridge CR442. The study was approved by the Ethics and Welfare Committee of the Department of Veterinary Medicine, University of Cambridge, and all methods were performed in accordance with the relevant guidelines and regulations. The study is reported in accordance with relevant ARRIVE guidelines. All data associated with this paper are openly available from the University of Cambridge Data Repository.

**Data accessibility.** All raw data are available upon request and are deposited in full in data repository and as electronic supplementary material [44].

**Declaration of AI use.** We have not used AI-assisted technologies in creating this article.

**Authors' contributions.** T.B.Y.: writing—original draft; W.D.T.: data curation, formal analysis, writing—review and editing; R.J.: investigation, methodology, writing—review and editing; S.K.: data curation, formal analysis, writing—review and editing; V.R.S.: data curation, investigation; G.L.: formal analysis, methodology, writing—review and editing; K.H.: methodology; J.S.: methodology, writing—review and editing; D.C.: investigation, methodology, writing—review and editing; S.E.: investigation, writing—review and editing; G.B.C.: methodology, writing—review and editing; J.J.P.: conceptualization, methodology, writing—review and editing; A.B.: formal analysis, investigation, methodology, writing—review and editing; P.F.: conceptualization, funding acquisition, writing—review and editing.

All authors gave final approval for publication and agreed to be held accountable for the work performed therein.

**Conflict of interest declaration.** We declare we have no competing interests.

**Funding.** Work was part-funded by a grant from the Alice Noakes Trust, UK.



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