# Reactive Fluid Ferroelectrics: A Gateway to the Next Generation of Ferroelectric Liquid Crystalline Polymer Networks

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#### Abstract

Herein we report the first examples of reactive mesogenic materials (RMs) which exhibit fluid ferroelectric order based on the recently discovered ferroelectric nematic (N<sub>F</sub>) phase. We term these materials N<sub>F</sub> RMs and they provide the first steps towards the next generation of ferroelectric liquid crystalline polymer networks. We report the chemical synthesis and characterisation of the liquid crystalline properties of these materials, demonstrating that they have the lowest longitudinal molecular dipole moments ( $\mu$ ) of any reported N<sub>F</sub> material of 7.39 D. We go on to demonstrate a potential use case of this new class of reactive material through the polymer stabilisation of a matrix which exhibits the N<sub>F</sub> phase, increasing the phase range of the ferroelectric phase from 75 °C to 120 °C. The N<sub>F</sub> RMs reported herein are an exciting step forward in ferroelectric liquid crystal research, demonstrating that reactive N<sub>F</sub> materials are achievable, allowing for the future development of liquid crystalline ferroelectric networks, elastomers and polymers.

Reactive mesogens (RMs) are functional materials displaying liquid crystalline (LC) mesophases that possess reactive groups, allowing for the formation of polymers, elastomers and network structures with liquid crystalline order. The first RM was synthesised in 1989 [1] as a means of protecting optical fibres, but it was not long until applications in the fabrication of liquid crystal displays (LCDs) emerged, leading to far reaching implications in the multibillion-dollar LCD industry [2–5]. Over the past 35-years, the application of reactive mesogenic materials has continued to grow giving rise to new fields of study, such as: liquid crystal elastomers (LCEs) [6,7], liquid crystal templating [8], polymer dispersed LC networks [9], and the stabilisation of LCs for device applications – commonly referred to as polymer stabilised liquid crystals (PSLCs) [10].

The recent discovery of the ferroelectric nematic ( $N_F$ ) phase, which combines nematic orientational order and fluidity with almost perfect parallel polar order of its constituent molecules [11,12], has garnered significant scientific interest due to its potential applications such as fluid piezoelectrics [13], tuneable lasers [14] and reflectors [15], non-linear electrooptical devices [16,17], and the generation of entangled photon pairs [18]. The potential scope of the  $N_F$ phase would be further increased through its incorporation into network structures. Whilst some work has already been done on macromolecular N<sub>F</sub> materials [19–21], a polymer network with inherent N<sub>F</sub> order has yet to be reported but would be desirable [22]. An obvious direction would be polymer stabilisation of the N<sub>F</sub> phase. So far, it has been reported that ferroelectric nematogens have poor miscibility with the non-mesogenic network structures typically applied, resulting in phase separation [23]. Some successes have been achieved using blue-phase networks [23], and optically isotropic polymer stabilised materials [24,25]. An N<sub>F</sub> exhibiting RM would be desirable as it would not only improve miscibility with N<sub>F</sub> phases for polymer stabilisation but would also give a pathway to new polymer networks that are polar in their own right. To date, there is no such RM reported within the literature and so we elected to design and synthesise a ferroelectric nematogen with reactive functionality.

As research into the N<sub>F</sub> phase is still largely in its infancy, there are only a few structure spaces currently known to promote the formation of the phase [26–28]. 4-(difluoro(3,4,5-trifluorophenoxy)methyl)-1,1'-biphenyl) is a common molecular feature present in several materials exhibiting the N<sub>F</sub> [29,30] (and other polar mesophase [31,32]) and so we chose this as the basic unit for our RMs. Coupling this to a basic acrylate unit afforded the four reactive materials **1-4 (Table 1)**. Of the four RMs prepared, two of the structures, **2** and **3**, exhibit the N<sub>F</sub> phase. For the sake of brevity, we will focus our discussion on **2** while data for the remaining materials are provided in the ESI to this article.

Table 1.Transition temperatures and their associated enthalpy changes [in brackets] for 1-4. Their longitudinal molecular dipole<br/>moments ( $\mu$  / D) (at the DFT:B3LYP-GD3BJ/cc-pVTZ level [35–38]) are also given in the left-hand column. Crystallisation<br/>temperatures (Crystal) recorded on cooling.



|             | Melt            | Crystal           | SmA <sub>F</sub> - N <sub>F</sub> | N <sub>F</sub> -I | N-I               |
|-------------|-----------------|-------------------|-----------------------------------|-------------------|-------------------|
| Compound    | т/℃             | т/℃               | т/℃                               | т/℃               | т/℃               |
| compound    | [ΔH / KJ        | [ <b>ДН / К</b> Ј | [ <b>ДН / К</b> Ј                 | [ <b>ДН / К</b> Ј | [ <b>ДН / К</b> Ј |
|             | mol⁻¹]          | mol⁻¹]            | mol⁻¹]                            | mol⁻¹]            | mol⁻¹]            |
|             |                 |                   |                                   |                   |                   |
|             | 73.9            | 53.3              |                                   |                   | 77.1              |
| 1           | [25.8]          | [21.7]            | -                                 | -                 | [0.5]             |
| μ = 6.87 D  |                 |                   |                                   |                   |                   |
| F           |                 |                   |                                   |                   |                   |
| 0-√ }       | 81.6            | 23.4              | 27.7                              | 84.3              | -                 |
| · · · · · · | [15.2]          | [14.8]            | [0.5]                             | [2.9]             |                   |
| 2<br>       |                 |                   |                                   |                   |                   |
| μ = 7.39 D  |                 |                   |                                   |                   |                   |
|             | 112.4<br>[21.6] | 99.7<br>[25.4]    | -                                 | 116.2<br>[5.7]    | -                 |
| μ = 7.98 D  |                 |                   |                                   |                   |                   |
| F<br>       | 103.0<br>[32.4] | 90.6<br>[32.2]    | -                                 | -                 | -                 |

Briefly, before beginning the discussion of **2**, it is of note that all four homologues have relatively low longitudinal molecular dipole moments ( $\mu$ ) (**Table 1**) compared to the vast majority of reported ferroelectric nematogens [26,27]. Although the importance of large values of  $\mu$  with regards to the formation of the polar phases is currently debated[39–43] all existing reported N<sub>F</sub> materials possess dipole moments greater than 8.5 D with the exceptions generally belonging to polar smectic materials[41–43]. For example, the archetypal materials RM734 and DIO have dipole moments of ~11 D and ~9 D, respectively [26]. Here **2** and **3** possess dipole moments of 7.39 D and 7.98 D, respectively, making them the lowest of any reported material exhibiting the N<sub>F</sub> phase to date. Empirically, this reinforces the emerging trend that the magnitude of  $\mu$  may not be a limiting factor for thermal stability of the N<sub>F</sub> phase for all materials [41].

The phase transition properties of the **2** were determined from polarized optical microscopy (POM), differential scanning calorimetry (DSC) and current reversal measurements (experimental details included in the ESI). When viewed between untreated glass, cooling **2** from the isotropic (I) phase sees the N<sub>F</sub> phase form from small droplets (**Fig. 1a [left]**), characteristic of the N<sub>F</sub> phase [42]. Further cooling sees the coalescence of the droplets into a banded texture, synonymous with the assignment of the N<sub>F</sub> phase (**Fig. 1a [centre]**). Rapid cooling to 25 °C showed a further transition to a phase with a blocky mosaic texture associated with the SmA<sub>F</sub> phase (**Fig. 1a [right]**) [43,44]. Attempts to confirm this assignment via X-ray scattering were unsuccessful due to crystallisation of the sample during measurements therefore the assignment of the SmA<sub>F</sub> phase here is tentative. Transition temperatures for these phases were obtained by DSC, with transitions appearing as first-order peaks

in the relevant DSC thermograms (**Table 1, Fig. 1b, and Fig. S5**). Ferroelectric order was confirmed via current response measurements (**Fig. 1c and Fig. S7**), where a single polarisation reversal peak was observed.



Figure 1.(a) POM micrographs taken for 2 depicting [left] the I-NF phase transition where the NF phase forms from<br/>droplets, [centre] the banded texture of the NF phase, and [left] the blocky mosaic structures associated<br/>with the SmAF phase. Images were taken within thin cells with no alignment layer at 84 °C, 50 °C, 25 °C,<br/>respectively; (b) DSC thermogram of 2 showing heating and cooling traces; and (c) current response trace<br/>measured for 2 measured at 100 Hz in the NF phase at 80 °C.

Presently, polymer stabilisation of the N<sub>F</sub> phase has not yet been reported. We elected to demonstrate the potential viability of our N<sub>F</sub> RMs by polymer stabilising a known, unreactive N<sub>F</sub> material. We envisaged that if the network contains repeat units that exhibit the N<sub>F</sub> phase, there will be two advantages. Firstly, the issue of poor affinity will be mitigated by the ideal mixing behaviour that occurs between N<sub>F</sub> molecules in mixtures [39,45]. Secondly, it is likely that a key component for the formation of a polar network is to polymerise in a polar phase. The addition of a non-N<sub>F</sub> RM would cause a significant reduction in the N<sub>F</sub> transition temperature preventing polymerisation in the N<sub>F</sub> phase. By using an N<sub>F</sub> RM, this is avoided completely.

For our test, we elected to stabilise the phase behaviour of the simple mixture **F7 (Fig. 2a)** [31], a simple binary mixture of two polar materials which exhibits the N<sub>F</sub> phase from room temperature up to 95 °C before transitioning to the splay nematic (N<sub>s</sub>) phase (an antiferroelectric nematic phase sometimes referred to as N<sub>X</sub>, N<sub>AF</sub> or SmZ<sub>A</sub> [46,47]). A series of mixtures containing varying quantities of **2 (Fig. 2b)**, the photoinitiator **MBF (Fig. 2c)**, and the commercially available crosslinker **RM82 (Fig. 2d)** in **F7** were produced, ensuring the concentrations of **F7** and **MBF** were kept constant (**Table 2**). Prior to polymerisation, each mixture was examined using the same physical characterisation techniques as materials **1-4**, confirming that the N<sub>F</sub> phase was exhibited by each mixture (**Fig. 3 a-c**, **Fig. S8** and **9**). For **PS3** and **PS4** the transition temperatures associated with the N<sub>F</sub> phase fall dramatically compared to **F7** due to the increased quantity of apolar **RM82** destabilising polar order.



Figure 2. The chemical structures and their transition temperatures, where appropriate, used in the formulation of mixtures **PS1-4**.

Table 2.The compositions of the polymer stabilised mixtures PS1-4, and the phase transition temperatures for<br/>the K-N<sub>F</sub> and N<sub>F</sub>-N<sub>s</sub> phase transitions pre- and post-polymerisation.

|                 |             |                |               |              | Pre-<br>polymerisation | Pre-<br>polymerisation | Post-<br>Polymerisation | Post-<br>Polymerisation |
|-----------------|-------------|----------------|---------------|--------------|------------------------|------------------------|-------------------------|-------------------------|
| Mixture<br>Name | 2<br>(wt %) | RM82<br>(wt %) | MBF<br>(wt %) | F7<br>(w t%) | T <sub>K-NF</sub> (°C) | T <sub>№</sub> (°C)    | Т <sub>к-N</sub> (°С)   | T <sub>№</sub> (°C)     |
| PS1             | 8.65        | 1.25           | 0.1           | 90           | -                      | 89.9                   | -                       | 89.8                    |
| PS2             | 7.4         | 2.5            | 0.1           | 90           | -                      | 86.6                   | -                       | 90.6                    |
| PS3             | 4.9         | 5.0            | 0.1           | 90           | 24.1                   | 78.5                   | -                       | 86.3                    |
| PS4             | 0           | 9.9            | 0.1           | 90           | 22.5                   | 54.5                   | 20.7                    | 86.4                    |
|                 |             |                |               |              |                        |                        |                         |                         |



Figure 3.(a) DSC thermograms of PS1-4, (b) POM micrograph of PS2 at 40 °C, and (c) current response trace for<br/>PS2 measured at 100 Hz in the NF phase at 30 °C pre-polymerisation. (d) DSC thermograms of PS1-4, (e)<br/>POM micrograph of PS3, and (f) current response trace for PS3 measured at 100 Hz in the NF phase at 30<br/>°C post-polymerisation.

Post-polymerisation using UV light (365 nm (2.5 W cm<sup>-2</sup>, 20 min), the N<sub>F</sub> phase is retained for all four mixtures and an increase in the temperature stability of the N<sub>F</sub> phase is observed for **PS2-4** with the transition temperature of the N<sub>F</sub> phase in **PS1** remaining comparable to the uncured sample (**Table 2, Fig. 3d**). This is due to the low concentration of **RM82** present in this mixture indicating a minimum degree of crosslinking is required for stabilisation. The characteristic banded texture of the N<sub>F</sub> is retained post-polymerisation by all mixtures (**Fig. 3e**) indicating that this polymer stabilisation mixture does not prevent the separation of the N<sub>F</sub> texture into domains of alternative polarisation. A stronger electric field is required to facilitate the switching of the polymerized materials (**Fig. 3f**) likely due to an increased elastic deformation cost which could be due to the creation of stronger polar anchoring [48] by the now polymerised network or some other restriction imposed by the resulting network structures. There is also likely to be an element of increased rotational viscosity.

The largest degree of stabilisation in the N<sub>F</sub> phase is seen for **PS4**, a mixture containing no **2** and, to a first approximation, it appears that an N<sub>F</sub> RM is not required to successfully stabilise the N<sub>F</sub>. However, both a significant broadening of the first order peak associated with the N<sub>F</sub> transition (**Fig. 3**) and the cold crystallisation of the sample on re-heating (**Fig. S8**) are evidence of significant phase separation [51]. Although the degree of stabilisation in **PS2** and **PS3** is lower than that observed in **PS4**, there is clearly significantly less de-mixing (**Fig. S8**). Whilst incorporating a N<sub>F</sub> RM into the mixture reduces the effect of the stabilisation, its presence is clearly beneficial, enhancing the miscibility between the network and the matrix.

To summarise, we have reported the synthesis of the first reactive mesogen materials to natively display the N<sub>F</sub> phase which we term N<sub>F</sub> RMs. We then show a potential use case for this new class of functional materials through the polymer stabilisation of the matrix **F7** [31]. When mixed in conjunction with a small amount of crosslinker (**RM82**) and photoinitiator (**MBF**), we demonstrate the successful stabilisation of the N<sub>F</sub> matrix increasing the effective N<sub>F</sub> phase range from 75 °C to 120 °C. By using inherently ferroelectric RMs we act to mitigate the phase separation issues that have been problematic in the previously attempted network structures [23]. We expect that N<sub>F</sub> RMs may be pivotal in the production of new ferroelectric liquid crystalline networks, elastomers and polymers, thereby increasing the possible use cases of the N<sub>F</sub> phase.

### Data availability

The data associated with this paper are openly available from the University of Leeds Data Repository at https://doi.org/10.5518/1635.

#### **Author Contribution Statement**

All authors discussed all results, and the manuscript was subsequently written, reviewed, and edited with contributions from all authors.

#### **Competing interests**

Authors declare that they have no competing interests.

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