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1 An improved numerical model for early detection of bed
2 agglomeration in fluidized bed combustion

3 Nik Nor Aznizam Nik Norizam^{a,*}, Xin Yang^{b,c}, Nik Azri^d, Derek Ingham^a, Janos
4 Szuhánszki^a, Lin Ma^a, Mohamed Pourkashanian^a

5 ^a Energy 2050, Translational Energy Research Centre (TERC), School of Mechanical,
6 Aerospace and Civil Engineering, University of Sheffield, Sheffield S3 7RD, UK.

7 ^b School of Mechatronical Engineering, Beijing Institute of Technology, Beijing
8 100081, People's Republic of China.

9 ^c Advanced Research Institute of Multidisciplinary Science, Beijing Institute of
10 Technology, Beijing, 100081, People's Republic of China.

11 ^d Project Delivery & Technology Division, PETRONAS Research Sdn Bhd, Selangor,
12 43000, Malaysia

13 **Abstract**

14 An improved predictive numerical index has been developed to predict the tendency of
15 bed agglomeration in fluidized bed boilers. The index was developed based on the melt fraction
16 resulting from the thermodynamic equilibrium model of fuel ash compositions together with
17 SiO₂ as the bed material at temperatures ranging from 700 to 900 °C. The partial least square
18 regression (PLSR) coupled with the cross-validation technique is utilized to establish the
19 correlation for the bed agglomeration index, I_a. The improved index, I_a has been validated by
20 experimental observations found in various literature sources. The results obtained using the
21 improved index, I_a demonstrated a significantly higher success rate in predicting the bed

22 _____
23 * Corresponding author.

24 E-mail address: nnaniknorizam1@sheffield.ac.uk.

25 agglomeration tendency of biomass fuel ash compared to the other four conventional bed
26 agglomeration indices. In addition, K_2O is the main element that accelerates the formation of
27 bed agglomeration in the biomass firing while CaO was found to reduce the tendency of bed
28 agglomeration in the fluidized bed combustion system.

29 **Keywords:** Biomass; Bed agglomeration index; Thermodynamic equilibrium model; Silica
30 sand; Fluidized bed

31 **1 Introduction**

32 Over the past few years, research has been focused on a comprehensive analysis of the
33 agglomeration mechanisms occurring during biomass's fluidized bed combustion (FBC). This
34 is included by looking at the aspects of potential early detection methods and operational
35 countermeasures for agglomeration mechanisms in the FBC system [1]. First of all, it is
36 important to understand the fundamental root cause of the agglomeration problems in the FBC
37 systems. Agglomeration is fundamentally driven by the formation of eutectic alkali silicates
38 due to the interaction between SiO_2 in the bed material or ash and alkali species such as K_2O
39 and Na_2O [2-4]. The alkali silicate formation will determine the eutectic melting temperature
40 as the primary instigator of bed agglomeration. The low melting temperature of eutectic
41 compounds ($<800\text{ }^\circ\text{C}$) [5] will cause bed agglomeration in the FBC due to the formation of
42 melt at temperatures $800\text{-}900\text{ }^\circ\text{C}$ [6]. In addition, it has been reported that the emergence of
43 alkali phosphates also results in bed agglomeration [7, 8]. Bed agglomeration can cause an
44 unfavourable flop of the fluidized bed, which is described as defluidization. As a result, several
45 cases have been reported in the literature that the defluidization of the bed will lead to the
46 shutdown of the fluidized bed combustors.

47 Visser et al. [9] identified two mechanisms responsible for the agglomeration of bed
48 particles in the FBC system which are coating-induced and melt-induced agglomeration.
49 Melting phenomena are crucial in both mechanisms. In the coating-induced mechanism, ash

50 deposited on bed particles will form a liquid phase as a result of chemical reactions which
51 causes multiple particles to stick to each other. Melt-induced agglomeration means the straight
52 adherence of the bed elements through relatively liquified fuel-derived cinders. Melt-induced
53 agglomeration relies on larger molten ash particles colliding with the bed particles, with the
54 molten ash acting as a viscous glue [10-12]. However, coating-induced agglomeration has been
55 identified as the predominant mechanism in the FBC system [2, 3]. Various researchers
56 assessed bed particle composition and subsequently used thermodynamic calculations to
57 identify potential melt phases. This suggests that all chemical reactions within the ash have
58 achieved thermodynamic equilibrium corresponding to coating-induced agglomeration.

59 According to F. Scala et al. [1], the method of combination between thermodynamic
60 equilibrium analysis and compression strength test was unable to predict the sintering tendency
61 of biomass ash, this is due to the method not taking into consideration the interaction between
62 ash and quartz as bed material. This also happened the same in the past when the researchers
63 tested the ash fusion accuracy to predict the bed agglomeration temperatures. Therefore, it was
64 concluded that the ex-situ method failed to predict the bed agglomeration behaviours [1]. Over
65 the years, researchers have proposed various empirical indices for bed agglomeration in the
66 FBC system. The alkali index is generally effective in predicting agglomeration, however, its
67 accuracy diminishes when considering factors such as alternative bed materials or additives, as
68 noted by various researchers [13-15]. Fernández Llorente et al. [16] determined that the ratio
69 of alkaline earth oxides to alkaline oxides is a weak indicator of ash sintering likelihood and
70 severity. Moreover, the bed agglomeration index (BAI) was introduced to measure the
71 operational issues in FBC technology. Several publications report that silica-dominated in-bed
72 agglomerations form in fluidized bed combustors at 760-900 °C [14, 17]. The BAI index,
73 however, does not take this element into account when making predictions about deposit
74 generation. P. Billen et al. [18] proposed a method to predict the agglomeration in FBC by

75 predicting melt formation and agglomeration using phase diagrams as a result of
76 thermodynamic calculations. Recently, Nik Norizam et al. [19] successfully produced a
77 numerical model to predict slagging propensity in the fixed bed reactor by using the melt
78 formation theory. Melt formation is the formation of a liquid phase when the temperature
79 exceeds the melting point of compounds in the ash. This can cause the ash and bed particles to
80 stick together and agglomerate leading to melt-induced agglomeration. On the other hand,
81 coating-induced agglomeration occurs when the ash reacts with the bed material and forms a
82 coating on the bed particles. The melt phase in this case only forms after the ash interacts with
83 the bed material. This coating facilitates the agglomeration of the ash and bed material.

84 To the best of the authors' knowledge, a number of research have been done in the past
85 to develop bed agglomeration prediction indices in the FBC system, however, the applicability
86 of the existing indices across different types of biomass to estimate the bed agglomeration
87 tendency remains limited. This is because previous researchers have not focused sufficiently
88 on the chemical reactions of fuel ash compositions in response to temperature changes within
89 the system. Also, this can be supported by Morris et al. [2], who recommended determining
90 bed agglomeration behaviour influenced by the fuel ash composition. In this paper, an
91 improved semi-empirical index based on the thermodynamic equilibrium model (TEM) is
92 developed for early detection of bed agglomeration combustion in FBC technology. The
93 research aims to predict bed agglomeration in FBC biomass technology by analyzing chemical
94 ash compositions, using TEM to assess the melting degree of biomass fuels with SiO₂ (quartz)
95 as the bed material. Then the partial least square regression (PLSR) coupled with cross-
96 validation has been employed to create a numerical model, I_a to estimate the bed agglomeration
97 index based on the experimental ash composition and the degree of melt from TEM. The results
98 obtained with the improved numerical model, I_a demonstrate a significantly greater success
99 rate in forecasting the tendency of bed agglomeration as compared to experimental

100 observations from the existing literature. This predictive tool would allow a better selection of
101 fuels and a priori incorporation of countermeasures in the FBC system for the industrial
102 operators.

103 **2 Material and methods**

104 In this section, the discussion focuses on the datasets employed to develop an improved
105 index to predict the bed agglomeration tendency of biomass. A total of 35 datasets containing
106 biomass ash compositions were gathered from relevant literature sources [7, 20-27], as
107 illustrated in Table 1, which consists of ash compositions, ash content (%) and experimental
108 observations of bed agglomeration tendency in the FBC boiler. 9 major ash compositions
109 (Na_2O , MgO , Al_2O_3 , SiO_2 , P_2O_5 , K_2O , CaO , SO_3 , Fe_2O_3) and quartz (SiO_2) as a bed material
110 were included in the model, however, TiO_2 was excluded due to its low content in the ash. The
111 initial 20 datasets (1-20) represent experimental ash composition data and have been designated
112 as training datasets, while the subsequent 15 datasets (21-35) will serve as testing datasets. The
113 training and testing datasets will be analysed through thermodynamic equilibrium modelling
114 in Section 2.1 to predict the melting fraction based on the ash compositions. The obtained
115 results will be utilized in Section 2.2, where PLS regression analysis will be employed to
116 formulate an expression, denoted as I_a , for predicting the bed agglomeration behaviour without
117 the necessity for conducting combustion tests in the future. Subsequently, the effectiveness of
118 I_a will be validated against experimental observations of bed agglomeration in the FBC system.

119 *2.1 Thermodynamic equilibrium model*

120 The application of thermodynamic equilibrium modelling (TEM) has become extensive
121 in the industry as a tool to predict the ash transformation behaviour and the chemical and
122 physical characteristics of ash in various ash-related processes, such as the issues of bed
123 agglomeration in fluidized beds, heat transfer surfaces corrosion, and smelt bed behaviour
124 patterns in boilers. The modelling was carried out using FactSage and the experimental data

125 gathered during the measurements were utilised as model inputs to simulate the formation of
126 the bed agglomeration in the FBC technology. It offers a reasonably accurate prediction of the
127 ash conversion process without the need for intricate experiments for each biomass. However,
128 it is important to note that the process in a real fluidized bed furnace may not entirely reach
129 equilibrium although particles in a fluidized bed have much longer residence time than those
130 in a pulverised combustor [28-31]. The FactSage thermochemical software is extensively
131 employed for the analysis of solid fuel firing. It facilitates the computation of multiphase
132 multicomponent equilibrium conditions, providing accurate results when configured with the
133 appropriate fuel composition, atmosphere, and temperature settings [19]. It builds upon the
134 Gibbs energy reduction and incorporates extensive databases for oxide, silicate, and salt
135 composition structures [28-31]. The databases contained both unmixed compounds and
136 solution stages. The pure compounds consist of stoichiometric element compositions, while
137 the solution databases feature optimized frameworks for the solution stages. The methods
138 employed to establish the thermodynamic database have been extensively examined in prior
139 studies [28-31]. The FactSage 8.1 software was utilized in this study to predict the melting
140 fraction, M_f of biomass fuels as listed in Table 1. The model extended the configuration set by
141 Nik Norizam et al. [19] by integrating the bed material into the present model (iii) to simulate
142 the bed agglomeration model in the FBC boiler, and the setting of the TEM is as follows:

- 143 i. The model is configured under the assumption of equilibrium conditions, utilizing the
144 FACTPS, FTsalt, and FToxid databases.
- 145 ii. Table 1 includes 9 ash compositions (MgO, Al₂O₃, SiO₂, K₂O, CaO, Fe₂O₃, Na₂O, P₂O₅,
146 SO₃) derived from experimental data. These compositions were normalised and labelled
147 as input-stream 1 in the model.

Table 1 Chemical ash compositions for various types of biomass [7, 20-27].

Num.	Biomass	Na ₂ O	MgO	Al ₂ O ₃	SiO ₂	P ₂ O ₅	K ₂ O	CaO	SO ₃	Fe ₂ O ₃	M _r	Ash content (%)	Agglomeration tendencies (experimental)
1	Logging residues	0.96	5.16	3.08	31.63	5.37	10.44	36.38	5.22	1.75	0.27	2.4	Low
2	Bark I	1.50	3.69	5.08	37.16	2.95	7.95	36.12	3.47	2.09	0.28	3.7	Low
3	Wheat straw	0.85	3.47	0.21	35.86	6.24	31.55	11.73	9.94	0.15	1.00	5.7	High
4	Sunflower husks	0.10	8.57	14.62	17.94	9.48	21.27	14.72	6.85	6.45	0.60	1.9	High
5	DDGS-logging residues	0.43	6.68	0.99	12.06	22.15	17.14	12.77	27.08	0.69	0.50	3.2	Low
6	DDGS-wheat straw	0.48	5.64	0.12	17.25	19.71	24.82	6.27	25.50	0.20	0.79	5.05	High
7	Logging residues-PA	0.95	4.99	3.03	31.12	7.82	10.27	35.09	5.01	1.72	0.28	2.4	Low
8	Pepper waste	1.05	4.55	8.40	15.40	11.20	35.36	10.04	10.62	3.38	0.95	7.4	High
9	Rice straw	0.70	9.24	3.34	33.66	0	10.33	36.62	0	6.11	0.19	n.m.	n/a
10	Corn cob	3.27	10.34	5.21	46.27	0	21.94	5.24	0	7.71	0.70	n.m.	n/a
11	DDGS	0.20	6.99	0.03	3.27	28.65	19.35	2.31	38.98	0.22	0.70	4.4	High
12	DDGS-willow	0.37	6.67	0.38	5.06	25.35	19.69	10.73	31.39	0.36	0.55	3.3	High
13	Coconut shell	4.62	1.54	8.48	66.76	1.54	8.48	2.41	0.01	6.17	0.96	3.1	High
14	Rapeseed mean-bark	0.86	7.41	2.15	17.30	19.51	13.59	22.03	15.94	1.22	0.40	4.8	Low
15	Coffee (mbuni) husks	0.52	5.23	5.10	17.65	4.84	49.80	13.99	n.m.	2.88	1.3	4.1	High
16	Bark II	1.74	4.28	5.42	31.72	3.69	8.54	43.30	0	1.31	0.18	4.9	Low
17	Forest residues	1.23	6.18	2.48	18.04	6.44	16.08	42.28	5.26	2.01	0.16	n.m.	Low
18	Cotton stalk	4.52	11.08	4.06	7.80	9.03	40.23	21.34	0	1.93	0.67	1.75	High
19	Thistle	11.73	5.03	3.50	18.27	2.13	13.86	44.16	0	1.32	0.77	8.9	High
20	Almond shell	0.86	4.58	0.86	6.17	4.23	54.63	28.19	0	0.48	0.65	0.94	High

21	Rapeseed meal	0.20	9.95	0.24	2.16	32.29	17.83	11.31	25.48	0.55	0.40	7.4	Low
22	RM	0.19	9.91	0.23	2.38	32.16	17.74	11.47	25.33	0.60	0.43	6	Low
23	Logging residues-PA 2	0.89	4.71	2.86	29.36	13.01	9.69	33.11	4.73	1.50	0.39	2.4	Low
24	Rapeseed cake	7.32	8.41	0.20	1.29	35.69	18.32	13.35	15.07	0.35	0.70	7.5	High
25	Wheat straw 2	0.49	2.43	1.05	43.12	5.20	32.07	10.95	3.65	1.05	1.11	7.3	High
26	Brassica	1.25	3.82	2.25	14.05	6.59	27.75	43.37	0	0.92	0.34	7.7	Low
27	Cotton husks	1.32	7.59	1.32	10.93	4.05	50.20	20.95	1.72	1.92	0.90	3.2	High
28	Coffee husks	0.84	3.06	10.45	5.29	5.71	61.00	12.81	0.56	0.28	1.19	n.m.	High
29	Soy husks	6.27	8.40	8.76	2.01	5.80	36.09	25.33	4.38	2.96	0.70	5.1	High
30	Coffee (parchment) husks	0.64	4.76	5.78	21.34	4.37	47.43	12.60	0	3.08	1.1	0.9	High
31	Wood	0.84	1.26	5.74	17.93	2.94	0.70	63.31	0	7.28	0.1	0.5	Low
32	Peat	0.50	1.50	10.10	30.67	6.73	0.75	39.53	0	10.22	0.1	4.3	Low
33	RC-Bark10	7.24	8.26	0.22	2.52	35.13	18.31	14.85	13.09	0.37	0.68	7.2	High
34	RC-Bark20	6.89	8.01	0.46	3.84	32.91	17.47	16.04	13.79	0.59	0.67	7.0	High
35	RC-Bark30	6.75	7.80	0.76	5.52	31.98	17.36	17.83	11.36	0.65	0.66	6.7	High

149
150
151

a) All chemical compositions of ash were measured on dry basis in wt%.
b) M_f : Melt fraction obtained from equilibrium model.
c) n.m.: Not measured

- 152 iii. Quartz (SiO_2) was introduced into the system as input-stream 2 with a 1:1 ratio to the
153 fuel. The temperature range of the equilibrium modelling was set between 700 and 900
154 $^\circ\text{C}$ in oxidizing conditions (excess of 10% O_2). This temperature range has been chosen
155 based on the typical operating conditions of biomass FB boilers [2].
- 156 iv. The simulations utilized the "FToxid-SLAGA" and "FTsalt-SALTF" model,
157 incorporating two-phase immiscibility as the solution database [32].
- 158 v. Record the weight range of the solid-liq phase formed for each fuel as a result of the
159 equilibrium modelling.
- 160 vi. The solid-liq phase formed during the equilibrium calculation is fractioned by 1g of the
161 ash to calculate the melt fraction, M_f .

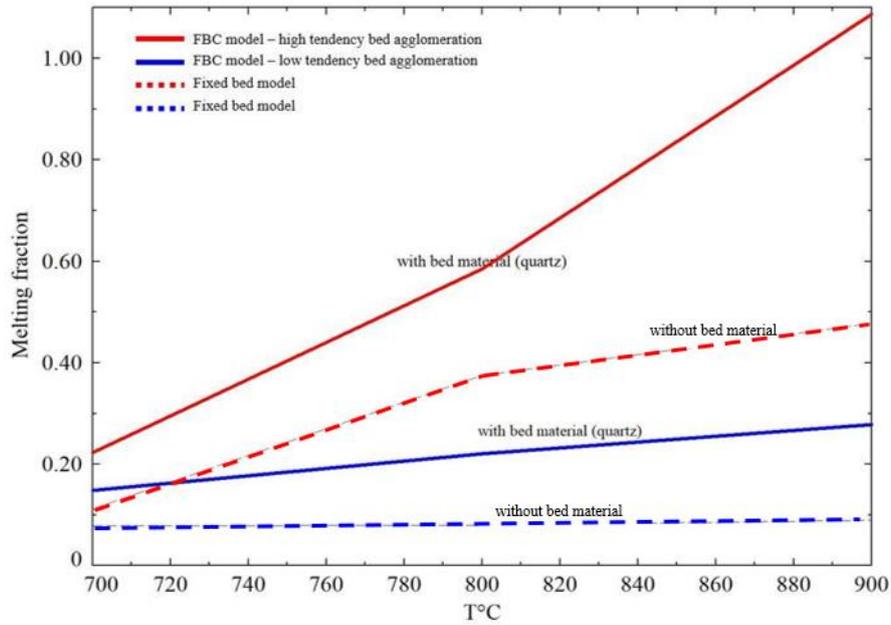
162 *2.2 Partial Least Square Regression Coupled with Cross Validation*

163 The mathematical approach of Partial Least Squares Regression (PLSR) incorporates
164 and generalizes the multivariate regression along with the analysis of the principal components
165 [33]. It excels in analysing extensive datasets and multivariate systems with a high correlation,
166 outperforming the multiple regression method [34]. This method is commonly employed to
167 determine the stopping criterion and the total number of latent variables during cross-
168 validation, considering only one observation at a time [34-36]. The PLSR is especially well-
169 suited for problems with limited observational data, as encountered in the analysis of bed
170 agglomeration behaviour of biomass. For additional details on PLSR and the cross-validation
171 technique, please refer to the following references [33-35, 37-39]. The PLSR coupled with
172 cross-validation techniques, was utilized for the analysis of the training datasets, as illustrated
173 in Table 1. This process has led to the formulation of an equation, expressed as a function of
174 the biomass ash compositions, to predict the improved bed agglomeration index (BAI), denoted
175 as I_a .

176 **3 Results & Discussion**

177 *3.1 Comparison melt fraction with and without presence of SiO₂ (quartz)* 178 *predicted by TEM.*

179
180 Isaak et al. [40] introduced the concept of melt fraction, suggesting that higher
181 temperatures lead to an increased presence of liquid in deposits. Experimental findings by Isaak
182 et al. [40] revealed a correlation between deposit stickiness and temperature, indicating that
183 elevated temperatures result in stickier deposits due to an increased amount of liquid phase.
184 Zhou et al. [41] indicated that the melting curve of ash melting fraction increases with an
185 increase in temperatures. Moreover, the idea of melting fraction gained popularity in predicting
186 the deposition of biomass ash [41-43]. Recently, Nik Norizam et al. [19] successfully
187 developed a model to predict the slagging propensity in fixed bed boilers of woody biomass at
188 high-temperature regions by employing the melting fraction concept obtained from the
189 Factsage 8.1 equilibrium model. Based on Figure 1, the melt fraction was compared between
190 the current FBC (with bed material) and the fixed bed model (without bed material). The
191 motivation is to compare the melt fraction model with and without the presence of the bed
192 material (quartz). The presence of bed material (rigid lines) has higher melt fractions compared
193 to without the presence of quartz (dash lines). This proves that the presence of SiO₂ (quartz) as
194 bed material actively reacts with the fuel ash compositions and produces a significant amount
195 of agglomerates in FBC technology. Figure 1 demonstrates that Wheat straw (red line) exhibits
196 a higher melting fraction than Bark I (blue line). Experimentally, it was observed that Wheat
197 straw tends to have a high propensity for bed agglomeration, whereas Bark I exhibits a low
198 tendency for bed agglomeration. This is because Wheat straw has a high ratio of K₂O/CaO
199 compared to Bark I. The analysis of the chemical compositions of ash will be further explained
200 in Section 3.2. In addition, Figure 1 clearly illustrates the melting degree increases with an
201 increase in the temperature. Thus, the melting degree illustrated in TEM simulation is in



202
 203 **Figure 1 Comparison of the melt fraction between the FBC model with bed material and**
 204 **fixed-bed model without bed material.**

205 agreement with the common melting curve demonstrated by Zhou et al. [41] and the melting
 206 fraction of woody biomass in fixed bed boiler by Nik Norizam et al. [19]. This paper
 207 concentrates on determining the melt fraction exclusively in the low-temperature range of 700-
 208 900 °C based on typically FBC operating conditions [2].

209 *3.2 Analysis and application of an improved BAI, I_a*

210
 211 Predicting the bed agglomeration tendency in FBC can assist power plant operators in
 212 anticipating the fuel quality before biomass firing. The objective of this section is to create an
 213 improved numerical model for a bed agglomeration index, I_a . This numerical model is intended
 214 for predicting the bed agglomeration behaviour of biomass based on the fuel ash composition,
 215 eliminating the need for a thermodynamic equilibrium calculation process. This is particularly
 216 relevant as the FactSage software is not extensively utilized among power plant operators. The
 217 development of bed agglomeration index, I_a is based on the melting fraction, M_f and chemical
 218 ash compositions (training datasets 1-20) as shown in Table 1 by applying the method of PLS
 219 regression (PLSR) analysis, coupled with cross-validation. The root means square error

220 (RMSE), R^2 , and slope for the training data were 0.031, 0.988 and 0.988, respectively which
221 suggested a good fit of the model [44]. The expression of I_a is acquired as follows:

$$222 \quad I_a = |[0.34 - 1.2055(MgO + CaO) + 1.6228(K_2O + Na_2O) + 7.113 \times 10^{-1} SiO_2 + \\ 223 \quad 1.627 \times 10^{-1}(SO_3 + P_2O_5)]| \quad (3.1)$$

224 **Condition applying eq 3.1:*

225 *The mass fractions of the ash compositions must be applied for the oxide parameters in the Equation 3.1.*

226

227 The negative coefficient in Equation (3.1) for MgO+CaO indicates that an increase in

228 the values of MgO+CaO will result in a decrease in the predicted value of BAI, I_a . Conversely,

229 the parameters (SiO_2 , K_2O+Na_2O , $SO_3+P_2O_5$) associated with the bed agglomeration

230 tendencies of biomass fuel exhibit positive coefficients. This implies an increased value of

231 these three parameters will result in a higher predicted number of BAI, I_a . Elements exhibiting

232 positive coefficients are highly prone to the formation of bed agglomeration, with potassium

233 being the most prevalent component in the agglomerates [2, 7, 20, 21, 24, 45]. Among the

234 positive coefficients, the highest value is +1.6228, and it corresponds to the regression

235 coefficient of potassium combined with sodium. This suggests that the presence of the

236 potassium element significantly influences the bed agglomeration behaviour. Morris et al. [2]

237 explained that agglomeration caused by melt-induced mechanism occurs when there is a

238 sufficient content of silica and alkali metal in the fuel ash. Lin et al. [20] found that high

239 potassium content in straw leads to the development of agglomerates and defluidization.

240 According to Grimm et al. [7], the wheat straw (potassium rich fuel) showed high tendency of

241 bed agglomeration upon firing in a 5 kW bubbling fluidized bed reactor. Figure 2 illustrates

242 the solid-liquid phase for wheat straw obtained by the TEM simulation, it is clearly shown that

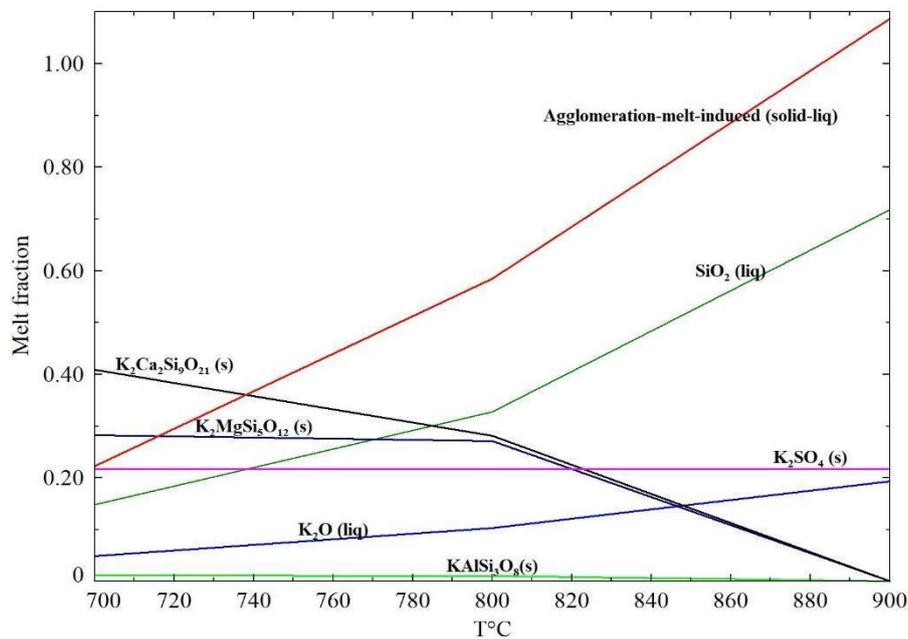
243 most of the compounds containing potassium which consist of $K_2Ca_2Si_9O_{21}$, $K_2MgSi_5O_{12}$,

244 K_2SO_4 , and $KAlSi_3O_8$. This is in agreement with the study by Lin et al. [20] who suggested

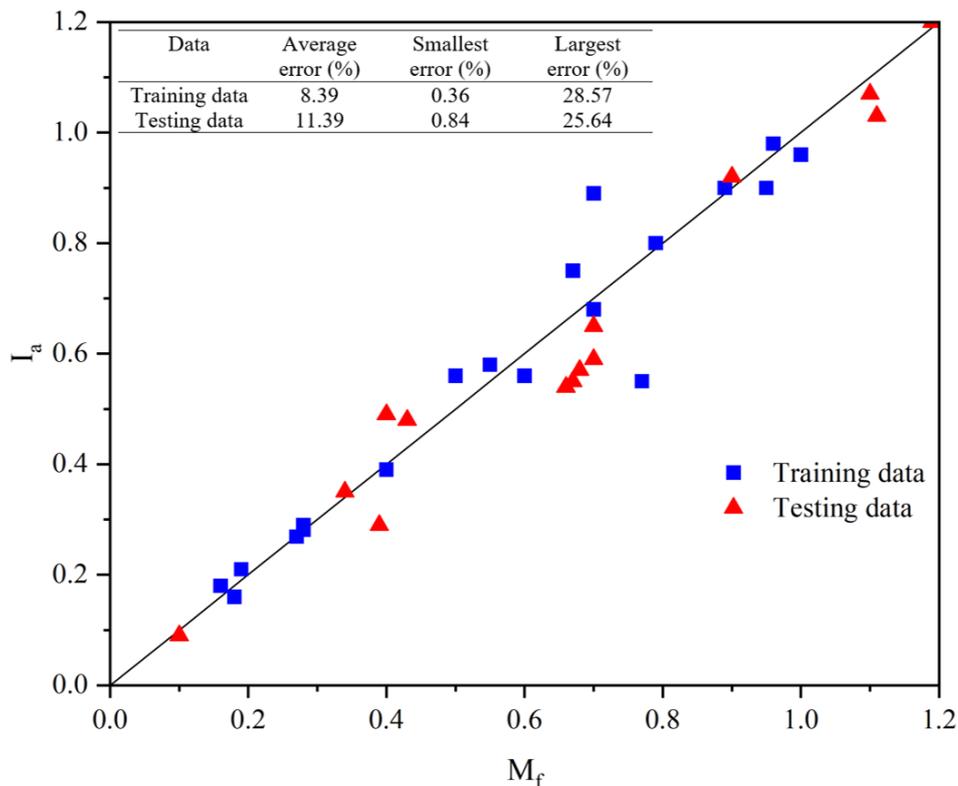
245 that compounds containing potassium have a tendency to persist in the bed and create

246 potassium-rich ash with a low melting point during the combustion process. Moreover,

247 experimental observation from Table 1 shows that about 95% of high bed agglomeration
 248 tendency was observed when the potassium value is higher than 10 wt%.



249 **Figure 2 Solid-liquid phase for Wheat straw (3) obtained by the FactSage equilibrium**
 250 **modelling.**



251 **Figure 3 Comparison between melt fraction by TEM, M_f and predictive BAI index, I_a .**

252 Figure 3 demonstrates a comparison between the melt fraction values, M_f and the
 253 predicted agglomeration index, I_a and the estimated error. The melt fraction for the 20 biomass

254 fuels (training datasets) were determined using the TEM. The M_f value will serve as a
 255 quantitative measure of bed agglomeration behaviour and will be compared with the qualitative
 256 measurement (low and high) based on experimental observations. As explained in Section 3.1,
 257 the value of the melt fraction, M_f will increase corresponding to experimental observations of
 258 bed agglomeration from low to high tendency. For a comprehensive explanation of the
 259 classifications of the experimental observations, please refer to Table 2 [7, 20-24]. It can be
 260 observed that the predicted index, I_a closely matches the melt fraction, M_f for both the training
 261 and testing datasets. It was found that the average error for prediction bed agglomeration index,
 262 I_a is 11.39%, which indicates that both of the indices M_f and I_a are in agreement with each other
 263 and exhibit the same trend as the calculated agglomeration index.

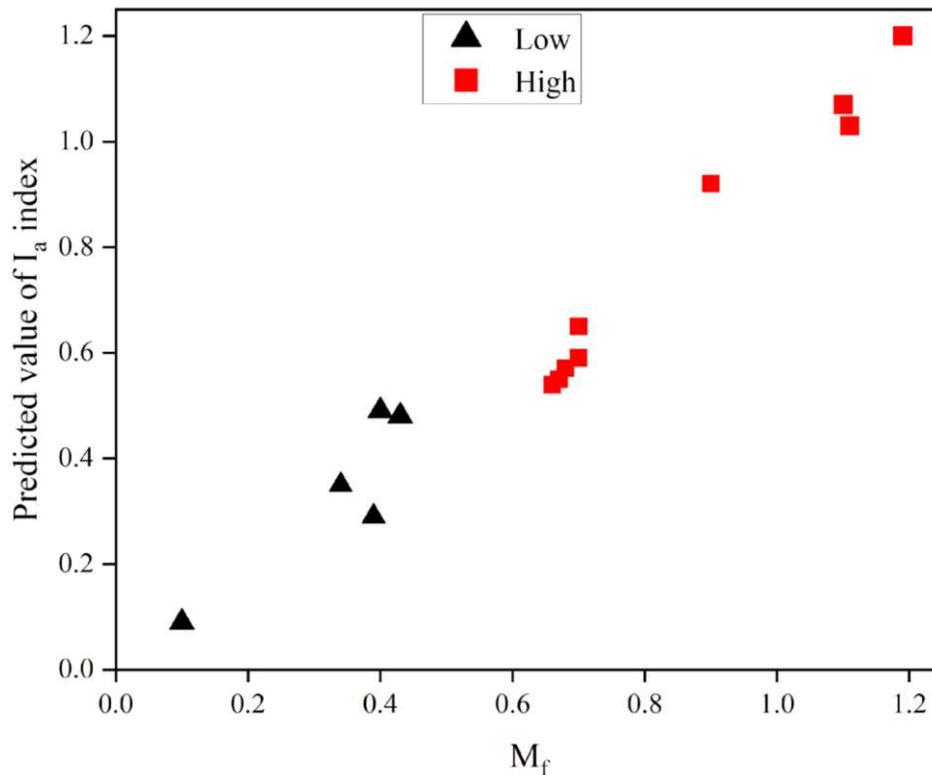
264 **Table 2 A comprehensive explanation of the classification for the bed agglomeration**
 265 **tendency by experimental observations.**

Authors	Technique used to determine the bed agglomeration in FBC
Grimm et al. [7]	The study evaluates the bed agglomeration tendency based on the formation and characteristics of coating layers on bed particles, as well as the temperature at which agglomeration leads to defluidization.
Lin et al. [20]	The paper uses defluidization time as the primary metric to assess the agglomeration tendency in FBC, with temperature being a critical factor influencing this behavior. The shorter the time before defluidization occurs, the higher the agglomeration tendency.
Yu et al. [21]	The study is to determine the tendency of agglomeration in FBC of biomass by measuring the slip resistance between particles. An increase in slip resistance between particles will correspond to the defluidization and high tendency to bed agglomeration.
Llorente et al. [22]	The sintering degree of agglomerates of bed material was determined visually and the assessment of ash disintegration in the agglomerates and deposits.
Liu et al. [23]	The bed agglomeration tendency was measured by monitoring the chemical composition changes, observing the physical changes in the bed material particles, and correlating these with the operational stability of the CFB during the experiments.

Piotrowska et al. [24]	The bed agglomeration tendency was concluded by analysing the temperature and pressure to determine the initial and total defluidization temperature. The initial defluidization temperature indicated the growth of agglomerates. The bed agglomeration tendency will reduced with an increased in defluidization temperature.
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266 Figure 4 shows the predicted value of the agglomeration index, I_a compared to the 15
267 experimental observations datasets (Table 1, numbers 21-35) from the literature [7, 20-27]. The
268 results indicate that the value of I_a index below 0.5 corresponds to a low bed agglomeration
269 tendency based on experimental observations. For example, Brassica (26) and Wood (31) were
270 observed as low tendency of bed agglomeration in the experimental observation [22, 27]. This
271 is due to the low ratio of K_2O/CaO . In contrast, biomass fuels with I_a value above 0.5 tend to
272 show high bed agglomeration during combustion. Most of the biomass fuels that were observed
273 to exhibit high bed agglomeration have a high K_2O/CaO ratio. For example, Cotton husks (27)
274 and coffee husks (28) have high amounts of potassium which are 50.20 wt% and 61.00 wt%,
275 respectively while the amount of calcium for both fuels are lower than the potassium content.
276 The findings suggest that potassium-rich ash compositions typically led to higher bed
277 agglomeration risks while biomass fuels with a high amount of calcium content shows lower
278 bed agglomeration potential in fluidised quartz bed combustors.

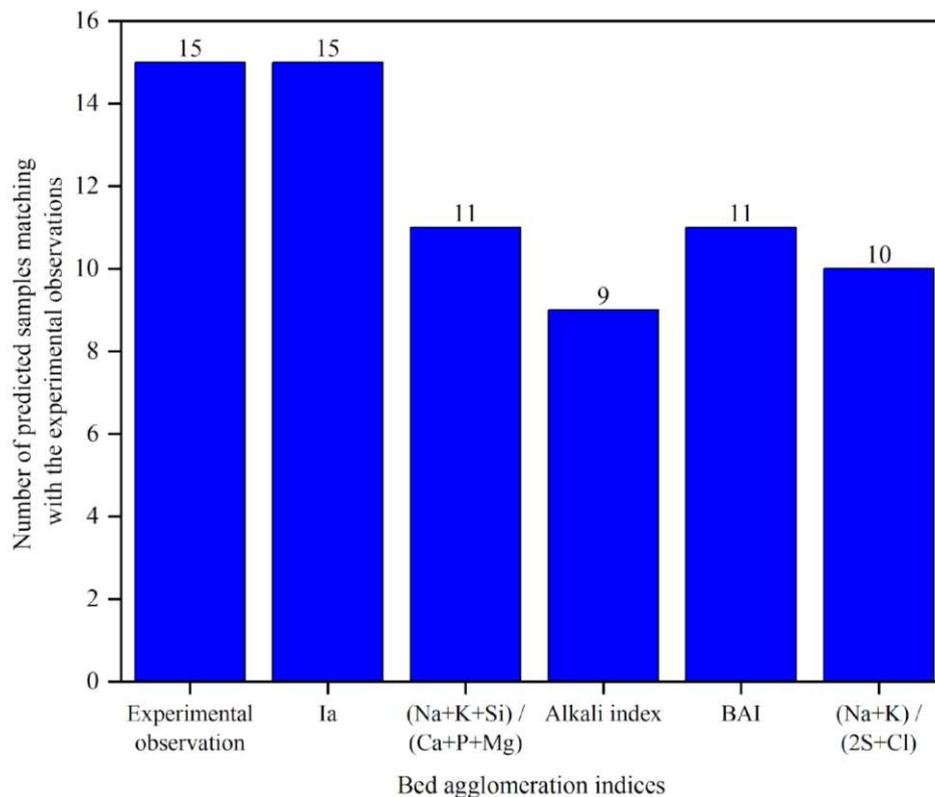
279 However, the K_2O/CaO ratio alone is not sufficient to accurately predict bed
280 agglomeration propensities. For example, Thistle (19) observed high agglomeration tendencies
281 during the experiments despite having a low K_2O/CaO ratio. This discrepancy is because the
282 K_2O/CaO ratio does not account for the elements that are prone to agglomeration such as
283 sodium. Furthermore, Thistle (19) also showed a high melting fraction (0.77) in TEM analysis
284 which indicated high tendency of bed agglomeration in fluidized bed reactor. On the other
285 hand, Rapeseed meal (21) and RM (22) had high K_2O/CaO ratios, yet they showed low
286 agglomeration propensities in the experiments. When their ash compositions were applied to



287
 288 **Figure 4 The predicted values in the proposed index, I_a versus experimental observations**
 289 **of bed agglomeration tendency.**

290 the I_a index, the index accurately predicted their low agglomeration tendencies which aligned
 291 with experimental observations. This accuracy is because the I_a index considers multiple ash
 292 components beyond potassium and calcium, providing a more comprehensive prediction of
 293 agglomeration behaviour. In conclusion, the I_a index provides a more reliable assessment for
 294 predicting bed agglomeration in fluidized bed systems.

295 Figure 5 shows the performance of the improved model equation for I_a compared with
 296 four different existing index equations [10, 13-15, 46]. These four distinct expressions are
 297 created for predicting bed agglomeration behaviour in FBC technology [2]. Please refer to the
 298 appendix (Table 3) for a complete list of all the equations and their corresponding thresholds
 299 for each index. The Figure 5 illustrates the number of predicted samples, out of the 15 testing
 300 fuels (21-35), that correspond to the experimental observations for each index. It is observed
 301 that the ranking for the accuracy in the prediction performance, arranged from high to low, is
 302 as follows: $I_a > I_2 = (Na + K + Si) / (Ca + P + Mg) = BAI > I_1 = (Na + K) / (2S + Cl) > Alkali$



303 **Figure 5 Comparison of the number of predicted samples matching with the experimental**
 304 **observations between I_a and 4 existing indices.**
 305

306 index. The new improved index, I_a, successfully predicted a total of 15 out of 15 samples in
 307 accordance with the experimental observations. On the other hand, the four existing indices
 308 (I₂, Alkali index, BAI, I₁) could only accurately predict a maximum of 73% of the total
 309 samples.

310 **4 Conclusions**

311 This study has successfully developed an improved semi-empirical index to predict the
 312 tendency of bed agglomeration in a FBC system. The index, I_a takes into account the chemical
 313 ash compositions when biomass is fired in fluidized bed combustors. The predictive tool
 314 created is capable of assisting industrial users in determining bed agglomeration behaviours
 315 when firing various types of biomass in FBC boilers. The newly improved bed agglomeration
 316 index, I_a was created by analysing the predicted melting fraction through TEM, considering the
 317 quartz as bed material used in the FBC system and chemical compositions of fuel ash, by
 318 employing the numerical PLS regression coupled with a cross-validation. The presented

319 method has been verified using experimental observations from existing literature on biomass
320 in the FBC technology. The results indicate that the BAI propensity index, I_a can be categorized
321 into two primary groups: low bed agglomeration tendency when $I_a \leq 0.50$ and high bed
322 agglomeration propensity when $I_a > 0.50$. In addition, the I_a demonstrates a significantly higher
323 success rate compared to four existing indices in evaluating the tendency of bed agglomeration
324 in fluidized bed type of boiler.

325 It can be confirmed that the primary element contributing to bed agglomeration
326 formation is K_2O , while a high value of CaO will reduce the tendency of bed agglomeration in
327 FBC. The index should be utilized during the low-temperature (700-900 °C) operation of
328 fluidized bed boilers. In addition, the I_a index is developed based on fluidized quartz bed
329 combustion, therefore, caution should be exercised when extending the predicted indices, I_a in
330 this paper to other types of bed material used in FBC technology. The authors strongly
331 recommend obtaining more extensive datasets of experimental observations by conducting
332 tests at an industrial or full scale to improve the accuracy of the predictive indices in forecasting
333 the tendency of bed agglomeration of FBC.

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465

466 **Appendix**

467 **Table 3 list of all the equations and their corresponding thresholds for each index [10, 13-15, 46].**

Index	Equation	Threshold for each index
I_a	$I_a = [0.34 - 1.2055(MgO + CaO) + 1.6228(K_2O + Na_2O) + 7.113 \times 10^{-1} SiO_2 + 1.627 \times 10^{-1}(SO_3 + P_2O_5)] $	Bed agglomeration tendency, I_a : low $\leq 0.50 <$ high
Alkali index	$AI = (K_2O + Na_2O) \text{ kg/GJ}$	0.17 < AI < 0.34 agglomeration possible AI > 0.34 agglomeration near certain
Bed agglomeration index	$BAI = \frac{Fe_2O_3}{K_2O + Na_2O}$	Agglomeration when BAI < 0.15
Agglomeration index, I1	$I1 = \frac{Na + K}{2S + Cl}$	High agglomeration potential when I1 > 1
Agglomeration index, I2	$I2 = \frac{Na + K + Si}{Ca + P + Mg}$	High agglomeration potential when I2 > 1