

1 **Cadmium isotope constraints on primary productivity and environmental perturbations**
2 **across the Late Ordovician mass extinction**

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4 He Zhao¹, Simon W. Poulton^{2,3}, Xiangdong Wang^{2,*}, Jun Sun¹, Yuangeng Huang⁴ and Fang Hao⁵

5
6 ¹College of Marine Science and Technology, China University of Geosciences, Wuhan 430074,
7 China

8 ²State Key Laboratory of Geological Processes and Mineral Resources, China University of
9 Geosciences, Wuhan 430074, China

10 ³School of Earth and Environment, University of Leeds, Leeds LS2 9JT, UK

11 ⁴State Key Laboratory of Biogeology and Environmental Geology, China University of
12 Geosciences, Wuhan, Hubei, 430074, China

13 ⁵National Key Laboratory of Deep Oil and Gas, China University of Petroleum (East China),
14 China

15
16 E-mail: wangxiangdong@cug.edu.cn (XDW)

17
18 **ABSTRACT**

19 The Late Ordovician mass extinction (LOME) constitutes the second largest of the “Big Five”
20 extinctions of the Phanerozoic. The LOME comprised two extinction pulses associated with sea
21 level changes linked to the Hirnantian glaciation. Although climate change has been implicated as
22 a potential driver of the mass extinction, uncertainty remains as to its precise impact relative to the
23 concurrent development of ocean anoxia. Here, we investigate the behaviour of the oceanic
24 cadmium (Cd) cycle, as a key element involved in a number of biological processes, across the
25 LOME and into the Early Silurian. Our focus is on the Wangjiawan section, South China, which is
26 the Global Stratotype Section and Point (GSSP) marking the base of the Hirnantian Stage. We
27 combine authigenic Cd isotope analyses ($\delta^{114}\text{Cd}_{\text{auth}}$) with total organic carbon (TOC) concentrations
28 and isotopes, and major and trace element systematics, to determine the evolution of marine
29 productivity across the LOME, and to provide insight into the mass extinction and relationships

30 between climatic and environmental change. Our $\delta^{114}\text{Cd}_{\text{auth}}$ data display a gradually decreasing trend
31 from the late Katian to the Katian–Hirnantian boundary, suggesting enhanced biological
32 assimilation of isotopically light Cd followed by export to the sediments. This interpretation is
33 supported by an increase in organic carbon isotope ($\delta^{13}\text{C}_{\text{org}}$) compositions, as well as a progressive
34 decrease in P/Al ratios and increasing $\text{C}_{\text{org}}/\text{P}$ ratios in the early part of the late Katian. A slight
35 increase in Cd isotope values in the early Hirnantian was likely caused by drawdown of light Cd (as
36 CdS) in euxinic shallower seawater settings. During glacial melting in the late Hirnantian, organic
37 carbon burial declined, consistent with lower Cd/Al, Zn/Al and Ni/Al ratios. However, $\delta^{114}\text{Cd}_{\text{auth}}$
38 values remain low across this interval, possibly due to an increase in the supply of isotopically light
39 Cd from enhanced weathering and rising sea level, as supported by elevated Al contents and
40 Chemical Index of Alteration (CIA) values. Elevated $\delta^{114}\text{Cd}_{\text{auth}}$ values subsequently occurred in the
41 Early Silurian (Rhuddanian), alongside a decline in CIA and Al values, suggesting that the Cd sink
42 was gradually balanced by a decline in the weathering input of Cd and lower rates of primary
43 productivity. Our data provide new insight into the Cd cycle through the Late Ordovician to Early
44 Silurian, and suggest that elevated marine productivity drove enhanced burial of organic matter,
45 which likely contributed to CO_2 drawdown and the initiation of the Hirnantian glaciation.

46

47 INTRODUCTION

48 The first of the “Big Five” mass extinctions occurred during the Hirnantian – the terminal stage
49 of the Ordovician – which represents an interval of major climatic and environmental change (e.g.,
50 Yan et al., 2010; Melchin et al., 2013; Harper et al., 2014; Zhou et al., 2015; Luo et al., 2016;
51 Smolarek et al., 2017). The late Ordovician mass extinction (LOME) occurred in two pulses
52 associated with sea level change linked to the Hirnantian (Gondwanan) glaciation, and resulted in
53 the loss of ~53% of marine species (Sheehan, 2001; Yan et al., 2010; Fan et al., 2020). The first
54 pulse occurred at the beginning of the Hirnantian stage during climatic cooling, which occurred
55 coincident with the onset of positive excursions in both inorganic carbon ($\delta^{13}\text{C}_{\text{carb}}$) and organic
56 carbon ($\delta^{13}\text{C}_{\text{org}}$) isotope values (Fan et al., 2009; Yan et al., 2009; Jones et al., 2011; Finnegan et al.,
57 2011), as well as widespread sulfidic (euxinic) anoxia (Hammarlund et al., 2012; Ahm et al., 2017;
58 Zuo et al., 2018; Bond and Grasby, 2020; Kozik et al., 2022a; Liu et al., 2022). The second extinction
59 pulse began with the onset of glacial melting, which occurred coincident with a rapid rise in eustatic

60 sea-level and the re-development of widespread euxinia (Yan et al., 2010; Finnegan et al., 2011;
61 Jones et al., 2011; Hammarlund et al., 2012; Melchin et al., 2013; Zhou et al., 2015; Luo et al., 2016;
62 Smolarek et al., 2017; Zuo et al., 2018; Bond and Grasby, 2020; Sánchez-Roda et al., 2024).

63 Although climate change has been implicated as a driver of the late Ordovician mass extinction
64 (LOME), the role of ocean anoxia has increasingly been stressed (Hammarlund et al., 2012; Ahm et
65 al., 2017; Zuo et al., 2018; Bond and Grasby, 2020; Liu et al., 2022; Kozik et al., 2022a). As such,
66 the causal relationships between climate change, marine redox changes and the biotic response are
67 heavily debated. For instance, one model suggests the first pulse of the LOME was driven by cooling
68 due to drawdown of atmospheric CO₂ caused by global silicate weathering and/or organic carbon
69 burial (Kump et al., 1999; Melchin and Holmden, 2006; Yan et al., 2010; Finnegan et al., 2012;
70 Lenton et al., 2012; Lv et al., 2022). By contrast, other studies suggest that the initial decline in
71 biodiversity was caused by warming and anoxia, rather than cooling itself (Hammarlund et al., 2012;
72 Zuo et al., 2018; Bond and Grasby, 2020; Kozik et al., 2022a; Liu et al., 2022; Sánchez-Roda et al.,
73 2024), with global cooling playing a secondary role in the second extinction pulse (Zuo et al., 2018).
74 Nevertheless, enhanced nutrient supply from intense volcanism, weathering and/or upwelling is
75 thought to have stimulated marine primary productivity and contributed to the Hirnantian positive
76 carbon isotopic excursion and the cooling event itself (Longman et al., 2021; Qiu et al., 2022). In
77 this case, changes in marine primary productivity could have played a significant role in both climate
78 change and marine redox conditions. However, the evolution of marine productivity across the
79 Ordovician–Silurian boundary (LOME) is poorly understood (Brenchley et al., 2003; Saltzman and
80 Young, 2005; Zhou et al., 2015; Bond and Grasby, 2020; Chen et al., 2021; Wang et al., 2022),
81 which has hampered understanding of the mass extinction, as well as the mechanisms that drove the
82 carbon cycle dynamics that are a hallmark of the Hirnantian.

83 As a biologically active trace element, cadmium (Cd) is involved in a number of biological
84 processes in marine micro-organisms (e.g., assimilation by some phytoplankton species), and it has
85 been demonstrated that biological uptake of Cd is associated with a significant isotopic fractionation
86 (Ripperger et al., 2007; Abouchami et al., 2011, 2014; Conway and John, 2015). Biological
87 assimilation in surface waters results in uptake of the lighter Cd isotopes, and these lighter isotopes
88 are regenerated in the deep sea or preserved in the sediments along with organic matter, which
89 overall leads to a monotonic transition to lighter values in the deep ocean (Abouchami et al., 2014;

90 Conway and John, 2015).

91 Modern seawater dissolved $\delta^{114}\text{Cd}$ measurements show that the isotopic signature of the deep
92 ocean has low values of +0.2‰ to +0.5‰, whereas the surface ocean may exceed +2.0‰ (Ripperger
93 et al., 2007; Xue et al., 2013; Conway et al., 2013; Abouchami et al., 2011, 2014; Conway and John,
94 2015). Cadmium isotope compositions in carbonates from various settings in the late Permian also
95 display lower values in deeper water sections relative to shallow waters, demonstrating that this
96 isotopic gradient may be preserved in ancient marine sediments (Zhang et al., 2023). In contrast to
97 the Cd isotopic signature of seawater, potential inputs (upper continental crust) exhibit lighter
98 isotopic compositions ($\sim -0.04\%$; Schmitt et al., 2009), and in addition to the burial of isotopically-
99 light Cd in association with organic matter, CdS precipitation under anoxic conditions has been
100 proposed as an isotopically-light sink from ambient seawater (Janssen et al., 2014; Guinoiseau et
101 al., 2018). In addition to biological uptake and CdS precipitation, experiments have shown that Cd
102 can readily substitute for Ca in CaCO_3 , resulting in an isotopic fractionation factor ($\alpha_{\text{CaCO}_3\text{-Cd(aq)}}$) of
103 ~ 0.99955 during calcite precipitation in seawater, although such an effect is unlikely to play a
104 significant role when considered relative to Cd utilization by phytoplankton (Horner et al., 2011).
105 These investigations of Cd isotope compositions in modern oceans provide critical insight into
106 marine Cd cycling and the mechanisms controlling Cd uptake in the oceans, thus enabling the
107 application of Cd isotopes as an effective paleo-proxy for past primary production (e.g., Georgiev
108 et al., 2015; Zhang et al., 2018, 2023; Frederiksen et al., 2022; Dickson et al., 2022; Liu et al., 2024).

109 In order to investigate oceanic Cd cycling and the evolution of marine productivity across the
110 LOME, we present new Cd isotope data alongside organic carbon isotopes and elemental
111 compositions, for the Wangjiawan section in South China. This section is the Global Stratotype
112 Section and Point (GSSP) for the basal Hirnantian Stage (Chen et al., 2006), and is well-studied in
113 terms of its bio- and chemo-stratigraphy. These data enable an assessment of possible controls on
114 measured $\delta^{114}\text{Cd}$ values, and in addition to providing a perspective on marine productivity, our
115 approach provides insight into the relationships between climatic and environmental change in the
116 Late Ordovician, as well as their role in the first mass extinction of the Phanerozoic.

117

118 **GEOLOGICAL SETTING**

119 During the Late Ordovician to Early Silurian, the South China craton was a microcontinent on

120 the northwestern margin of Gondwana, situated approximately at the paleoequator (Fig. 1A). The
121 Wangjiawan section lies to the north of Yichang City, Hubei Province, and was located on the Upper
122 Yangtze Platform in an unrestricted, deep outer-shelf setting (Chen et al., 2006; Fig. 1B). The section
123 comprises the Wufeng, Kuanyinchiao and Lungmachi formations in ascending order. The Wufeng
124 Formation is composed mainly of black shale with abundant graptolites (Chen et al., 2006), whereas
125 the Kuanyinchiao Formation, which is characterized by the Hirnantian Fauna, is dominated by
126 carbonaceous mudstone containing many brachiopod shells, and sporadic argillaceous limestone
127 turbidites that record lowstand conditions during the Hirnantian glaciation (Rong et al., 2002; Chen
128 et al., 2006; Yan et al., 2010; Zhou et al., 2015; Luo et al., 2016). The overlying Lungmachi
129 Formation represents an interval of rising and subsequently stable sea level, following Hirnantian
130 glacial melting, and is similar lithologically to the Wufeng Formation, except for higher contents of
131 siliceous minerals (Zhou et al., 2015; Luo et al., 2016). The complete biozones established for this
132 section provide the international reference for the identification and correlation of strata across the
133 LOME (Chen et al., 2006). The base and termination of the Hirnantian glaciation occur in the
134 uppermost Wufeng and Kuanyinchiao formations, and are constrained by the basal boundary of
135 *Metabolograptus (M.) extraordinarius* and the upper boundary of *Normalograptu (N.) ojsuensis*
136 graptolite biozones, respectively (Fig. 2).

137

138 **METHODS AND ANALYTICAL TECHNIQUES**

139 **Major and Trace Element Analyses**

140 Whole-rock major element analyses were performed at the State Key Laboratory of Geological
141 Processes and Mineral Resources (SKLGPMR), China University of Geosciences (CUG-Wuhan).
142 Chemical dissolution and separation methods for each element followed the national standard
143 procedure (GB/T 14506-2010) that includes alkali fusion, extraction, titration, weight loss and
144 colorimetry. In brief, weight loss and titration were used for determining Al contents, and a Hitachi
145 atomic absorption spectrophotometer (180-70) was utilized for Ca, K and Na analysis, with an
146 ultraviolet-visible spectrophotometer (UV-754) used for P analysis. Uncertainties of less than 5%
147 were achieved for all reported elements, as monitored by national standard samples, GBW07114
148 and GBW07122. Trace elements were analyzed at SKLGPMR via ICP-MS (Agilent 7500a). The
149 dissolution procedure follows that described by Liu et al. (2008). The sample powder was weighed

150 into Teflon bombs and 1 mL of HNO₃ and 1 mL of HF were added. The sealed bombs were heated
151 at 190°C for 48 h, and then the solution was evaporated on a hot plate, followed by the addition of
152 1 ml of HNO₃ and evaporation to dryness. The resultant salt was re-dissolved by adding ~3 ml of
153 30% HNO₃, and the bomb was resealed and heated at 190°C for 12–24 h. The final solution was
154 diluted to 100 mL with 2% HNO₃ for ICP-MS analysis. Blanks and national standards AGV-2,
155 BHVO-2, BCR-2 and RGM-2 were used for calibrating elemental concentrations, with an analytical
156 uncertainty of <2% for all elements of interest.

157

158 **Organic Carbon Isotopes**

159 Before analysis, sample powders were digested in 2 N HCl at 50°C for 12 h to dissolve
160 carbonate phases. Then ~0.5–0.8 mg of powder was analyzed using a MAT 253 IR-MS interfaced
161 with a Flash EA 2000 auto-sampler. $\delta^{13}\text{C}_{\text{org}}$ results are reported relative to the Vienna Pee Dee
162 belemnite (V-PDB) standard, with a precision better than $\pm 0.1\text{‰}$, based on repeat analyses ($n = 2$)
163 of national standard GBW04407 ($\delta^{13}\text{C} = -22.4\text{‰}$).

164

165 **Chemical Separation and Cd Isotope Measurements**

166 Sample digestions, chemical separation and Cd isotope measurements were conducted at the
167 State Key Laboratory of Ore Deposit Geochemistry, Institute of Geochemistry, Chinese Academy
168 of Sciences (Guiyang). The detailed procedure follows that described by Wen et al. (2015) and
169 Zhang et al. (2018, 2023). In brief, powdered sample containing approximately 0.15 μg of Cd was
170 placed in a centrifuge tube, then enough aqua regia (a 3:1 volume mixture of HCl and HNO₃) was
171 added to dissolve Cd. After 48 h, the mixture was centrifuged to separate the supernatant from the
172 residue. A $^{110}\text{Cd}/^{111}\text{Cd}$ double-spike solution, with $^{110}\text{Cd}/^{111}\text{Cd} \approx 0.5$ and a Cd concentration of 200
173 $\mu\text{g}/\text{mL}$, was then added to the supernatant to achieve a Cd spike–sample ratio of approximately 2,
174 followed by shaking to ensure isotopic equilibrium (Zhang et al., 2018). Finally, the supernatant was
175 dried in a 30 mL beaker, then 5 mL of 2 M HCl was added and the solution was evaporated to
176 dryness. After this, 5 mL of 2 M HCl was added to prepare the sample for chemical anion-exchange
177 separation. A blank sample was prepared by this method to monitor the sample digestion and
178 chemical separation procedure.

179 Strong anion resin AG-MP-1M was used in ion-exchange chromatography to purify Cd from

180 the matrix, to remove elements that could potentially interfere with the determination of Cd isotopes,
181 including Sn, In, Zn and Pb (the detailed procedure is reported in Wen et al., 2015). This step was
182 repeated twice in order to thoroughly purify the Cd. Cadmium isotope analysis was conducted using
183 a Neptune Plus multi-collector inductively coupled plasma mass-spectrometer (MC-ICP-MS). A
184 0.15 mol/L HNO₃ solution was used as a carrier for isotopic analysis, with Cd concentrations of
185 approximately 1 ppm. High-sensitivity (X) Ni cones were used for increasing transmission to ensure
186 that signal intensity was at a level of ~65 V/1 ppm. After each analysis, the nebulizer and spray
187 chamber were rinsed using 0.6 mol/L HNO₃, in order to prevent cross contamination between
188 samples. Reference standards, Spex and JMC, were analyzed during the course of the study and
189 yielded $\delta^{114}\text{Cd}$ values of $-0.08 \pm 0.04\text{‰}$ (n=2) and $-1.74 \pm 0.09\text{‰}$ (n=2), respectively, which agree
190 with values reported by Zhang et al. (2018) ($-0.10 \pm 0.07\text{‰}$ and $-1.71 \pm 0.06\text{‰}$), indicating that Cd
191 was completely purified by the column chromatography and no Cd isotope fractionation occurred
192 during the measurements. Double-spike data reduction was calculated according to a MATLAB-
193 based script described by Ripperger and Rehkämper (2007). The reported $\delta^{114}\text{Cd}/^{110}\text{Cd}$ values are
194 the mean of replicate measurements of the same samples. Cd isotopic compositions were calculated
195 relative to the NIST SRM 3108 Cd solution using the following delta notation:

$$196 \quad \delta^{x/110}\text{Cd} (\text{‰}) = [({}^x\text{Cd}/^{110}\text{Cd})_{\text{sample}}/({}^x\text{Cd}/^{110}\text{Cd})_{\text{standard}} - 1] \times 1000 \quad (1)$$

197 where x = 114, 112 or 111.

198

199 **RESULTS**

200 **Organic Carbon Isotopes**

201 Values of $\delta^{13}\text{C}_{\text{org}}$ vary from -30.3 to -27.8‰ through the section, with a gradually elevated trend
202 through the late Katian (Fig. 2; Table S1). This is followed by a positive shift that marks the global
203 Hirnantian $\delta^{13}\text{C}_{\text{org}}$ excursion (HICE) (Underwood et al., 1997; LaPorte et al., 2009). High values
204 then persist until the end of the Hirnantian glaciation, when $\delta^{13}\text{C}_{\text{org}}$ returns to pre-extinction levels
205 through the latest Ordovician and Early Silurian (Fig. 2).

206

207 **U/Th Ratios, TOC Contents and C_{org}/P Ratios**

208 Uranium (U) is a redox-sensitive element and occurs in solution as U⁶⁺ under oxic conditions.
209 In anoxic environments, however, U⁶⁺ is generally reduced to U⁴⁺, and may be enriched in sediments

210 (Tribovillard et al., 2006). By contrast, the behaviour of thorium (Th) is not affected by redox
211 conditions and usually occurs in the form of insoluble Th⁴⁺ (Wignall and Myers, 1988). Thus, U/Th
212 ratios can be used to evaluate redox-driven changes in U drawdown to the sediments (Wignall and
213 Myers, 1988; Sun et al., 2015). U/Th ratios vary from 0.94 to 4.92 through the Wangjiawan section,
214 with a declining trend from the late Katian to minimum values in the early Hirnantian, after which
215 ratios rise in a fluctuating manner to relatively high values (~3.0) in the late Hirnantian, with
216 maximum values in the Rhuddanian (Fig. 2; Table S1).

217 The same samples from the Wangjiawan section have previously been measured for TOC
218 contents (Gong et al., 2017). Relatively high values (generally >1 wt. %) occur throughout the
219 section, with a similar trend to U/Th ratios, notably with minimum values in the early Hirnantian
220 (Fig. 2; Table S1). However, in contrast to U/Th ratios, TOC contents yield maximum values in the
221 Late Hirnantian. A plot of TOC versus U/Th displays a positive correlation (Fig. 3a), indicating that
222 TOC preservation and/or production rates were likely controlled by redox conditions.

223 The molar C_{org}/P ratio is another paleoredox proxy that may be applied to shale and mudstone
224 formations (Algeo and Ingall, 2007). Values of C_{org}/P vary from 16 to 826 through the Wangjiawan
225 section, with a generally similar profile to TOC, displaying a declining trend from relatively high
226 values (mostly more than 100) in the late Katian, to lower values (less than 50) at the Katian-
227 Hirnantian boundary (Fig. 2). The ratios rise again to high values (~100 to 800) in the late Hirnantian
228 and through most of the early Rhuddanian.

229

230 **Chemical Index of Alteration (CIA) and Aluminum**

231 The Chemical Index of Alteration (CIA) may be used to evaluate the degree of chemical
232 weathering in siliciclastic sediments, and is widely used to evaluate climatic controls on chemical
233 weathering in the sediment source region (Nesbitt and Young, 1982). High CIA values (85–100)
234 generally reflect intense chemical weathering under warm and humid conditions. Conversely, lower
235 values (less than 65) commonly indicate low chemical weathering rates under cool and dry
236 conditions (Nesbitt and Young, 1982; Yan et al., 2010). The CIA was calculated as (Nesbitt and
237 Young, 1982):

$$238 \text{ CIA} = \text{Al}_2\text{O}_3 / (\text{Al}_2\text{O}_3 + \text{CaO}^* + \text{K}_2\text{O} + \text{Na}_2\text{O}) \quad (2)$$

239 where all values represent molar concentrations and CaO* represents CaO in silicate minerals only.

240 To correct measured CaO contents for the presence of carbonates and apatite, CaO* was calculated
241 by determining the CaO/Na₂O molar ratio of the bulk sediment followed by comparison with an
242 assumed CaO/Na₂O molar ratio of 1.0 for the detrital silicate fraction (McLennan, 1993). If the
243 measured CaO/Na₂O molar ratio was <1.0, the measured CaO concentration was adopted as CaO*.
244 If the measured CaO/Na₂O molar ratio was >1.0, the concentration of Na₂O replaced CaO*:

$$245 \quad \text{CIA} = \text{Al}_2\text{O}_3 / (\text{Al}_2\text{O}_3 + \text{K}_2\text{O} + 2 \times \text{Na}_2\text{O}) \quad (3)$$

246 Calculated CIA values vary from 0.69 to 0.80 through the Wangjiawan section (Fig. 2; Table
247 S1). During the Katian stage, CIA values gradually increase to relatively high values (~0.79) in the
248 late Katian, followed by a sharp decline to minimum values (~0.69) at the base of the Hirnantian
249 stage. The CIA values then rapidly return to higher levels (maximum of ~0.80) through the rest of
250 the Hirnantian, followed by a progressive decline to relatively low values through the Rhuddanian
251 stage (Fig. 2).

252 Aluminium (Al) is commonly considered a proxy for the terrigenous flux due to its
253 incorporation into aluminosilicate minerals such as clays and its stability during transfer and
254 diagenesis (Shi et al., 2016). The Al₂O₃ contents range from 2.97 to 25.37 wt. % in the Wangjiawan
255 section, with a similar trend to the CIA values, showing minimum values at the bottom of the
256 Hirnantian stage and maximum values in the late Hirnantian (Fig. 2; Table S1).

257

258 **Nutrient Elements**

259 Phosphorus (P) is an essential macronutrient that is commonly considered to limit primary
260 productivity on geological timescales (Tyrell, 1999), while cadmium (Cd), zinc (Zn) and nickel (Ni)
261 are micronutrients that can be used as proxies for relative changes in oceanic primary productivity
262 (Tribovillard et al., 2006; Grasby et al., 2016). P/Al, Cd/Al, Zn/Al and Ni/Al ratios vary widely
263 through the section, from 2×10^{-3} to 104×10^{-3} (wt. %/wt. %), from 4.6 to 855 (ppb/wt. %), from 2.7
264 to 409 (ppm/wt. %), and from 3.7 to 152 (ppm/wt. %), respectively (Fig. 4; Table S1). All four ratios
265 have similar profiles, with relatively low values in the late Katian (although P/Al ratios show a
266 gradual decrease), followed by a rapid rise at the end of the Katian to maximum values in the early
267 Hirnantian. Ratios then decrease in the mid-Hirnantian, remaining relatively constant through the
268 Rhuddanian (Fig. 4).

269

270 **Cadmium Isotopes**

271 Bulk Cd isotope ($\delta^{114}\text{Cd}/^{110}\text{Cd}_{\text{bulk}}$) values range from $-0.22 \pm 0.11\text{‰}$ to $1.0 \pm 0.1\text{‰}$ through the
272 section, with an overall progressive decrease from the late Katian to the Hirnantian, followed by a
273 slight increase in the early Hirnantian and then a sharp decrease at the end of the Hirnantian
274 glaciation (Fig. 4; Table S1). Low values ($\sim -0.2\text{‰}$) are then maintained until the early Rhuddanian,
275 followed by a progressive increase to higher values ($\sim 0.6\text{‰}$).

276

277 **DISCUSSION**

278 **Carbon Cycling and Redox Conditions Across the Late Ordovician to Early Silurian**

279 The Hirnantian glaciation is characterized by a global positive excursion in both inorganic and
280 organic carbon isotopes (e.g., Kump et al., 1999; Chen et al., 2006; Yan et al., 2009; Zou et al., 2018;
281 Lv et al., 2022). This excursion begins near the Katian-Hirnantian boundary and peaks during the
282 Hirnantian *M. extraordinarius*–*M. persculptus* graptolite zone (Chen et al., 2006; Liu et al., 2022),
283 consistent with the $^{13}\text{C}_{\text{org}}$ profile at the Wangjiawan section (Fig. 2). A global increase in organic
284 matter burial has been invoked as a mechanism to explain the positive excursion, which is supported
285 by a positive shift in Zn isotopes (as a tracer for changing organic carbon fluxes in ancient oceans)
286 during the Hirnantian in South China (Lv et al., 2022). Both marine anoxia and high marine primary
287 productivity aid in the burial of organic matter. Hence, reconstructing marine redox conditions
288 across the Late Ordovician to Early Silurian is pivotal to understand the cause of elevated organic
289 matter burial during the Hirnantian.

290 Ratios of U/Th can be used to assess bottom water redox conditions, with U/Th ratios <0.75 ,
291 0.75 – 1.25 and >1.25 being indicative of oxic, dysoxic (2.0 – 0.2 mL/L O_2) and suboxic–anoxic (0.2 –
292 0.0 mL/L O_2) environments, respectively (Tyson and Pearson, 1991; Jones and Manning, 1994).
293 Ratios of $\text{C}_{\text{org}}/\text{P}$ can provide further insight into benthic redox conditions as oxygen depletion
294 simultaneously enhances the preservation of organic C and may cause a diffusive loss of
295 remineralized organic P, with values of <50 , ~ 50 – 100 and >100 being indicative of oxic, suboxic
296 and anoxic environments, respectively (Algeo and Ingall, 2007). Values of U/Th and $\text{C}_{\text{org}}/\text{P}$ range
297 from 1.96 to 2.46 and from 16 to 459 (mostly more than 100), respectively, throughout most of the
298 late Katian (through the *Dicellograptus complexus* and all but the uppermost *Paraorthograptus*
299 *pacificus* graptolite zones; Fig. 2) at the Wangjiawan section, suggesting a suboxic/anoxic

300 environment. This suggestion is consistent with $\delta^{98}\text{Mo}$ records at the Wangjiawan and Nanbazi
301 sections, South China (Zhou et al., 2015), and a $\delta^{238}\text{U}_{\text{carb}}$ record for the Wuhe section in South China
302 (Liu et al., 2022). In addition, a detailed study of iron speciation and Mo data from inner-shelf to
303 outer-shelf settings in South China also shows persistent anoxia during the late Katian (Zou et al.,
304 2018), indicating widespread anoxia during this time in South China.

305 Both U/Th and C_{org}/P show a declining trend in the latest Katian, with relatively low values
306 (0.94–1.83 for U/Th; mostly less than 100 for C_{org}/P) persisting through the Hirnantian glaciation,
307 indicative of fluctuations between oxic to suboxic bottom waters (Fig. 2). This result is in agreement
308 with a positive shift in $\delta^{98}\text{Mo}$ values during the latest Katian to early Hirnantian in South China
309 (Zhou et al., 2015), as well as iron speciation data reported for the Wangjiawan section, that reveal
310 a shift from anoxic bottom waters in the Katian to a more oxygenated deep outer-shelf in the early
311 Hirnantian (Yan et al., 2012). By contrast, Fe speciation systematics alongside elevated Mo
312 concentrations on the Shuanghe inner shelf section in South China indicate persistent euxinia during
313 the early Hirnantian (*Metabolograptus extraordinarius* graptolite zone), while less reducing
314 (ferruginous) conditions were prevalent in mid-shelf and outer-shelf sections (Zou et al., 2018). This
315 suggests a redox gradient, with shallow water euxinia during the latest Katian to early Hirnantian,
316 giving way to better ventilated deeper waters.

317 In the upper part of the Hirnantian glacial interval, where an increase in CIA values documents
318 the deglaciation phase, initially elevated U/Th and C_{org}/P ratios suggest a return to water column
319 anoxia (Fig. 2). While U/Th ratios remain slightly elevated in the upper part of the Hirnantian glacial
320 interval, indicating persistent water column anoxia, C_{org}/P ratios fall to low values. Low C_{org}/P ratios
321 at this time are also observed in the Yangtze shelf sea, where a detailed study of the phase
322 partitioning of P demonstrated that these low ratios occur as a result of limited chemical weathering
323 (hence retaining high concentrations of detrital P phases) of source sediments produced during the
324 glacial phase (Qiu et al., 2022). A subsequent return to elevated C_{org}/P ratios, alongside a transition
325 to particularly high U/Th ratios, support continued water column anoxia as chemical weathering
326 intensity increased across the base of the *Metabolograptus persculptus* graptolite zone
327 (corresponding to the LOME-2 horizon; Fig. 2). Anoxia was also widespread in South China across
328 this interval, and spanned from shallow to deep water settings, as recorded by multiple independent
329 proxies (Yan et al., 2012; Zhou et al., 2015; Zou et al., 2018; Liu et al., 2022). Anoxic oceanic

330 conditions persisted through the Rhuddanian stage of the Early Silurian, as shown by continually
331 high U/Th and C_{org}/P values in the Wangjiawan section (Fig. 2), and multiple proxies from other
332 global localities (e.g., Hammarlund et al., 2012; Melchin et al., 2013; Zhou et al., 2018; Stockey et
333 al., 2020; Qiu et al., 2022; Sánchez-Roda et al., 2024).

334

335 **Evaluation of Authigenic Cd Concentrations and Isotopes**

336 Evaluation of the Cd flux from seawater to the sediments requires correction for the lithogenic
337 (terrigenous) component of the sediment. The samples generally have high Cd concentrations (up
338 to 2700 ppb), which are significantly elevated above the average value for upper continental crust
339 (~90 ppb), although there are three samples (WS-10, WS-4 and WOO3L-13) that have low Cd
340 concentrations (less than 90 ppb) (Fig. 4; Table S1). Furthermore, a plot of Cd versus Al reveals no
341 correlation (Fig. 3b), indicating that lithogenic Cd played an insignificant role in controlling Cd
342 concentrations, and the sedimentary Cd budget is instead largely controlled by authigenic Cd
343 enrichments. Nevertheless, we correct Cd concentrations for the lithogenic component by assuming
344 an homogenous Cd/Al ratio ((Cd/Al)_{lith}) and Cd isotopic composition for the lithogenic component.
345 Thus, authigenic Cd contents (Cd_{auth}) are expressed as:

$$346 \quad \text{Cd}_{\text{auth}} = \text{Cd}_{\text{sample}} - \text{Al}_{\text{sample}} \times (\text{Cd}/\text{Al})_{\text{lith}} \quad (4)$$

347 where Cd_{sample} and Al_{sample} are the sample concentrations of each element, and (Cd/Al)_{lith} is the global
348 average for upper continental crust (11.3 ppb/wt. %; Rudnick and Gao, 2003). The fraction of
349 authigenic Cd (f_{auth}) and lithogenic Cd (f_{lith}) can then be calculated using the formulae:

$$350 \quad f_{\text{auth}} = \text{Cd}_{\text{auth}}/\text{Cd}_{\text{sample}} \quad (5)$$

$$351 \quad f_{\text{auth}} + f_{\text{lith}} = 1 \quad (6)$$

352 Finally, the isotopic composition of authigenic Cd ($\delta^{114}\text{Cd}_{\text{auth}}$) is calculated from the equation:

$$353 \quad \delta^{114}\text{Cd}_{\text{bulk}} = \delta^{114}\text{Cd}_{\text{auth}} \times f_{\text{auth}} + \delta^{114}\text{Cd}_{\text{lith}} \times f_{\text{lith}} \quad (7)$$

354 where $\delta^{114}\text{Cd}_{\text{lith}}$ (-0.04‰; Schmitt et al., 2009) and $\delta^{114}\text{Cd}_{\text{bulk}}$ are the isotopic compositions of the
355 lithogenic component and the measured sample values, respectively.

356 The results from these calculations show that authigenic contributions to the sedimentary Cd
357 budget dominate in the Wufeng and Kuanyinchiao Formations, with only one sample having an f_{auth}
358 value of less than 0.8 (0.6) (Table S1). By contrast, the Lungmachi Formation generally has lower
359 proportions of authigenic Cd, with most f_{auth} values being less than 0.5 (Table S1). This indicates an

360 increased lithogenic Cd contribution to the sediments during the transition period from the
361 Kuanyinchiao Formation to the Lungmachi Formation, which corresponds to the melting phase of
362 the Hirnantian glaciation (as documented by the return to higher CIA and Al values), as well as the
363 second mass extinction pulse (Fig. 2; Table S1). Authigenic $\delta^{114}\text{Cd}_{\text{auth}}$ values show a similar trend
364 to the $\delta^{114}\text{Cd}_{\text{bulk}}$ values, but slightly offset by the lithogenic correction (Fig. 4).

365

366 **The Cd Cycle Across the LOME**

367 Studies of the modern oceanic Cd cycle show that biological uptake of isotopically-light Cd in
368 surface waters plays an important role in the distribution of Cd and its isotopes (Ripperger et al.,
369 2007; Abouchami et al., 2011; 2014; Conway and John, 2015), and this results in lower Cd
370 concentrations and heavier Cd isotope signatures in the mixed surface layer, compared to deeper
371 waters. In addition, CdS precipitation under anoxic conditions can also have a significant local
372 impact on Cd cycling, resulting in removal of isotopically-light Cd to the sediments (Janssen et al.,
373 2014; Guinoiseau et al., 2018). Both of these output mechanisms have been invoked to explain
374 higher $\delta^{114}\text{Cd}/^{110}\text{Cd}$ seawater values ($\sim 0.38\text{‰}$; Schmitt et al., 2009) relative to the terrestrial influx
375 (-0.04‰ ; Schmitt et al., 2009). Here, we utilize this understanding of Cd isotope systematics to
376 interpret our $\delta^{114}\text{Cd}_{\text{auth}}$ data in terms of the major local and global perturbations that occurred during
377 the Late Ordovician and into the Early Silurian.

378 We first evaluate the potential effects of biological organic matter and CdS precipitation on
379 $\delta^{114}\text{Cd}/^{110}\text{Cd}$ compositions. TOC shows a weak negative correlation with Cd concentrations ($R^2 =$
380 0.16 , $p=0.04$; Fig. 5a) and no correlation with $\delta^{114}\text{Cd}/^{110}\text{Cd}$ ($R^2 = 0.01$, $p>0.1$; Fig. 5b), indicating
381 that Cd derived from organic matter likely had little effect on the Cd isotopic composition of the
382 studied samples. As mentioned above, TOC preservation and/or production rates at the Wangjiawan
383 section were mainly controlled by bottom water redox conditions, as indicated by a strong positive
384 correlation between TOC and U/Th (Fig. 3a). The lack of correlation between TOC contents and
385 $\delta^{114}\text{Cd}/^{110}\text{Cd}$ values thus also implies that CdS precipitation likely had a minor effect on the Cd
386 isotopic composition of the studied samples. Thus, our $\delta^{114}\text{Cd}/^{110}\text{Cd}$ values appear to provide a
387 robust approximation of seawater Cd isotopic compositions.

388 The gradual decrease in $\delta^{114}\text{Cd}_{\text{auth}}$ compositions at the Wangjiawan section, from 1.1‰ in the
389 late Katian to 0.3‰ at the Katian-Hirnantian boundary (Fig. 4), is consistent with sinking of

390 isotopically-light organic-bound Cd to deeper water depths. As mentioned above, the late Katian
391 ocean was characterized by widespread anoxia, which can draw down isotopically light Cd, leaving
392 the water column enriched in heavier Cd isotopes, rather than resulting in the observed decline in
393 $\delta^{114}\text{Cd}_{\text{auth}}$ values. Therefore, the Cd isotope cycle during the late Katian was likely dominated by
394 preferential assimilation of isotopically-light Cd by phytoplankton in surface waters, which is
395 consistent with Cd isotope cycling in the modern ocean (Ripperger et al., 2007; Abouchami et al.,
396 2011, 2014; Conway and John, 2015). The negative $\delta^{114}\text{Cd}_{\text{auth}}$ shift during the late Katian thus likely
397 indicates a progressive increase in marine primary productivity. This is consistent with the
398 progressive decrease in P/Al ratios (Fig. 4) and increasing C_{org}/P ratios (Fig. 2) observed throughout
399 the early part of the late Katian. This suggests progressively enhanced redox-promoted recycling of
400 P from the sediments back to the water column, which would be expected to promote a positive
401 productivity feedback (e.g., Ingall et al., 1993; Ingall and Jahnke, 1994; Van Cappellen and Ingall,
402 1994)

403 The Wangjiawan section shows a slight increase in Cd isotopes in the early Hirnantian (from
404 0.3‰ to 0.6‰; Fig. 4), although the values are lower than those of the late Katian. Previous studies
405 have shown the development of more widespread euxinia in shallow water settings at this time (e.g.,
406 Zou et al., 2018). In this case, drawdown of light Cd (as CdS) in shallower water settings likely
407 contributed to the observed enrichment in heavier Cd isotopes in the oxygenated, deep outer-shelf
408 Wangjiawan setting. The global positive shift in $\delta^{13}\text{C}_{\text{org}}$ at this time (Kump et al., 1999; Chen et al.,
409 2006; Yan et al., 2009; Zou et al., 2018; Lv et al., 2022; Fig. 2), indicates increased burial of organic
410 ^{12}C due to high primary marine productivity, which is consistent with elevated Cd/Al, Zn/Al and
411 Ni/Al ratios in the early Hirnantian (Fig. 4). These high ratios occur despite only a modest increase
412 in preserved TOC in the early Hirnantian (Fig. 2), which is likely due to relatively decreased
413 preservation of TOC as the regional water column (and shallow porewaters) became oxygenated
414 (e.g., Canfield, 1994). High rates of primary productivity were likely enhanced by preferential
415 recycling of P back to the water column from sediments deposited under euxinic conditions
416 elsewhere (Qiu et al., 2022), although as expected, low C_{org}/P ratios and high P/Al ratios occur at
417 Wangjiawan (less than the Redfield ratio of 106:1; Fig. 2), due to the local oxic depositional setting
418 (e.g., Ingall et al., 1993). Overall, our data suggest that global marine primary productivity increased
419 from the latest Katian to early Hirnantian, thereby resulting in the progressive decrease in $\delta^{114}\text{Cd}_{\text{auth}}$

420 values observed through the late Katian, with the slight increase in $\delta^{114}\text{Cd}_{\text{auth}}$ values during the early
421 Hirnantian being due to globally-enhanced CdS precipitation. This increased CdS precipitation
422 masked the expression of isotopically-light $\delta^{114}\text{Cd}_{\text{auth}}$ values originating from particularly enhanced
423 primary productivity during the early Hirnantian.

424 In the mid to late Hirnantian (during and following glacial retreat), global rates of primary
425 productivity progressively declined, as indicated by a decrease in both organic ^{12}C burial (Fig. 2),
426 and Cd/Al, Zn/Al and Ni/Al ratios (Fig. 4). Thus, the increase in TOC contents and C_{org}/P ratios
427 across the LOME-2 interval (Fig. 2) likely reflects a local response to the re-establishment of water
428 column anoxia on the deep outer-shelf. This water column anoxia persisted throughout most of the
429 overlying succession, with the exception of a temporally-restricted return to better oxygenated
430 conditions during the earliest Silurian (indicated by low U/Th, decreased TOC contents, and low
431 C_{org}/P ratios in the *Akidograptus ascensus* graptolite zone; Fig. 2). However, given that the ocean
432 was characterized by particularly widespread (commonly euxinic) anoxia at this time (e.g.,
433 Hammarlund et al., 2012; Melchin et al., 2013; Zhou et al., 2018; Stockey et al., 2020; Qiu et al.,
434 2022; Sánchez-Roda et al., 2024), it is likely that global rates of primary productivity remained
435 relatively high. Nevertheless, the particularly low $\delta^{114}\text{Cd}_{\text{auth}}$ values (as low as -0.6‰) require further
436 consideration.

437 Elevated precipitation of CdS under widespread anoxic/euxinic conditions would be expected
438 to sequester isotopically-light Cd. As such, similar to the increase observed in the early Hirnantian
439 (Fig. 4), $\delta^{114}\text{Cd}_{\text{auth}}$ values would be expected to display heavier values, rather than the observed low
440 values that characterize the lower two thirds of the Lungmachi Formation (Fig. 4). This suggests
441 that redox conditions had a limited effect on Cd isotope compositions across this interval. The Cd
442 isotope composition of the oceans can be expressed as (John et al., 2017):

$$443 \quad \delta^{114}\text{Cd}_{\text{ocean}} = \delta^{114}\text{Cd}_{\text{input}} - \Delta^{114}\text{Cd}_{\text{output}} \quad (8)$$

444 where $\delta^{114}\text{Cd}_{\text{ocean}}$ and $\delta^{114}\text{Cd}_{\text{input}}$ are the mean values for seawater and the input flux, respectively,
445 while $\Delta^{114}\text{Cd}_{\text{output}}$ represents the isotopic offset between $\delta^{114}\text{Cd}_{\text{ocean}}$ and the combined sinks of $\delta^{114}\text{Cd}$.
446 Thus, if an increased weathering influx of Cd is not balanced by a similar increase in the sinks,
447 $\delta^{114}\text{Cd}_{\text{ocean}}$ and $\delta^{114}\text{Cd}_{\text{auth}}$ values would both change. Thus, although speculative, one possibility is
448 that isotopically-light Cd from enhanced chemical weathering (as indicated by high CIA and Al
449 values; Fig. 2), which may have been exacerbated by rising sea level (Fig. 5), was imported to the

450 ocean. However, without a similar increase in the magnitude of the sinks, this would have lowered
451 seawater $\delta^{114}\text{Cd}$ in the late Hirnantian to early Silurian ocean, ultimately resulting in the particularly
452 low sediment $\delta^{114}\text{Cd}_{\text{auth}}$ compositions observed in the Wangjiawan section.

453 A rise in $\delta^{114}\text{Cd}_{\text{auth}}$ values then occurs in the upper part of the Rhuddanian, where values
454 approach pre-extinction levels towards the top of the sampled interval (Fig. 4). Correspondingly,
455 CIA values and Al contents show a declining trend across this interval, suggesting that the Cd sinks
456 were gradually balanced by a declining weathering input, in addition to biological assimilation of
457 isotopically-light Cd.

458

459 **Implications for Environmental Perturbations Across the LOME**

460 The integration of our Cd isotope and elemental data with previously published Late
461 Ordovician paleotemperature records, and local and global paleo-redox proxy ($\delta^{98}\text{Mo}$ and $\delta^{238}\text{U}$)
462 data (Fig. 5), enable a holistic assessment of marine environmental conditions across the LOME.
463 The negative trend in $\delta^{114}\text{Cd}_{\text{auth}}$ values through the late Katian occurs coincident with an overall
464 decrease in sea surface temperatures (Fig. 6; Finnegan et al., 2011; Kozik et al., 2022a). Cooling
465 during the Late Ordovician has commonly been considered to be due to drawdown of atmospheric
466 CO_2 through silicate weathering and/or organic carbon burial (Kump et al., 1999; Melchin and
467 Holmden, 2006; Lenton et al., 2012; Swanson-Hysell and Macdonald, 2017; Longman et al., 2021;
468 Qiu et al., 2022), which may have been enhanced through orogenesis (Swanson-Hysell and
469 Macdonald, 2017), the expansion of early land plants (Lenton et al., 2012), and/or enhanced
470 volcanism (Gong et al., 2017; Hu et al., 2020; Longman et al., 2021). These processes are thought
471 to have increased the delivery of P to the oceans, thereby stimulating increased primary productivity
472 and organic carbon burial, and hence CO_2 drawdown (Longman et al., 2021; Qiu et al., 2022).

473 The declining trend in $\delta^{114}\text{Cd}_{\text{auth}}$ and increasing trend in $\delta^{13}\text{C}_{\text{org}}$ through the late Katian (Fig. 6)
474 therefore likely reflect elevated marine primary productivity and organic carbon burial driven by
475 increased nutrient supply (Fig. 7), which is consistent with the observation of eukaryotic blooms
476 during this period (Shen et al., 2018). Indeed, the increasing CIA trend at the Wangjiawan section
477 supports increased chemical weathering on land (Fig. 6), and hence an enhanced terrestrial nutrient
478 input into the ocean system (Fig. 7). Since this increase in chemical weathering occurs during an
479 interval of cooling (rather than the general link to warming), this highlights the significant roles that

480 the expansion of early land plants (Lenton et al., 2012) and/or enhanced volcanism (Gong et al.,
481 2017; Hu et al., 2020; Longman et al., 2021) played in enhancing the weathering influx of P. The
482 potential influence of land plants and volcanism could have led to an increase in the extent of global
483 anoxia, as indicated by a negative shift in $\delta^{238}\text{U}$ values at the onset of the Hirnantian glaciation (Fig.
484 6; Bond and Grasby, 2020; Kozik et al., 2022a), although better oxygenated bottom waters are
485 indicated by decreased U/Th ratios and elevated $\delta^{98}\text{Mo}$ values for the deeper outer-shelf setting of
486 the Wangjiawan section (Fig. 6).

487 Following the declining trend in $\delta^{114}\text{Cd}_{\text{auth}}$ in the late Katian, a slight increase in $\delta^{114}\text{Cd}_{\text{auth}}$
488 values occurs in the early Hirnantian at the Wangjiawang section, likely due to drawdown of light
489 Cd (as CdS) due to widespread euxinia in shallow water settings at this time (Fig. 7; e.g., Zou et al.,
490 2018). The increases in Cd/Al, Zn/Al and Ni/Al ratios, along with the positive excursion in organic
491 carbon isotopes (Fig. 4), are indicative of high primary productivity and enhanced organic matter
492 burial (Fig. 7), coincident with the development of glacial deposits from Gondwana and records of
493 sea level regression (Caputo and Crowell, 1985; Holmden et al., 2013). This glacial event is thought
494 to have played an important role in organism losses because of climate cooling and habitat loss (Yan
495 et al., 2010; Finnegan et al., 2012), although widespread euxinia appears increasingly likely to have
496 played a major role in the first pulse of the LOME (LOME-1; e.g., Zou et al., 2018; Bond and Grasby,
497 2020). Climate cooling could also have helped promote a redox gradient from euxinic shallow
498 waters to better ventilated deeper waters (Zou et al., 2018), as indicated by iron speciation, U/Th
499 and $\delta^{98}\text{Mo}$ data presented in this study and previous work (Yan et al., 2012; Zhou et al., 2015; Zou
500 et al., 2018; Fig. 6).

501 The $\delta^{114}\text{Cd}_{\text{auth}}$ values show the onset of a second negative shift in the middle Hirnantian where
502 elevated CIA values occur, likely indicating a global sea level rise associated with climatic warming
503 (Fig. 6). This climatic warming and sea level rise could have led to widespread oceanic anoxia in
504 the Early Silurian (Zhou et al., 2015; Stockey et al., 2020; Kozik et al., 2022a), as reflected by
505 elevated U/Th ratios and negative shifts in both $\delta^{98}\text{Mo}$ and $\delta^{238}\text{U}$ across the LOME-2 horizon (Fig.
506 6). Indeed, the expansion of anoxic/euxinic conditions associated with deglaciation has been
507 implicated as the most likely kill mechanism for the LOME-2 interval (Bond and Grasby, 2020; Liu
508 et al., 2022; Kozik et al., 2022a). Meanwhile, terrestrial chemical weathering was enhanced in
509 response to climate warming during the mid-Hirnantian to early Rhuddanian (Figs. 6, 7), which may

510 explain the low $\delta^{114}\text{Cd}_{\text{auth}}$ values due to input of isotopically-light continental Cd (Fig. 7). The Cd/Al,
511 Zn/Al and Ni/Al profiles, along with $\delta^{13}\text{C}_{\text{org}}$, gradually returned to lower values relative to those of
512 the early Hirnantian (Fig. 4), suggesting decreased productivity, which is similar to the record from
513 Dob's Linn, UK (Bond and Grasby, 2020).

514 Elevated $\delta^{114}\text{Cd}_{\text{auth}}$ values occurred in the Rhuddanian *Cystograptus vesiculosus* graptolite zone
515 (Fig. 6), corresponding to subdued terrestrial chemical weathering as indicated by decreased CIA
516 and Al values (Figs. 2, 6). Hence, a decreased isotopically-light Cd input to the oceans could have
517 contributed to the elevated $\delta^{114}\text{Cd}_{\text{auth}}$ values (Fig. 7). High U/Th ratios and invariable $\delta^{98}\text{Mo}$ values
518 imply that the oceans were still starved of oxygen at this time (Fig. 6), which may explain the
519 protracted and turbulent recovery of marine groups after the LOME (Rasmussen et al., 2019;
520 Stockey et al., 2020).

521

522 CONCLUSIONS

523 Cadmium and organic carbon isotopes, in combination with TOC and elemental
524 paleoenvironmental proxies, from the deep outer-shelf setting of the Wangjiawan section, South
525 China, provide a unique dataset for evaluating controls on marine primary productivity and
526 environmental perturbations across the LOME. A declining trend in $\delta^{114}\text{Cd}_{\text{auth}}$, coupled with a slight
527 positive shift in organic carbon isotopes, from the late Katian to the Katian-Hirnantian boundary,
528 suggest that elevated marine productivity resulted in enhanced sedimentation of isotopically-light
529 organic-bound Cd. Climate cooling in the early Hirnantian then appears to have promoted a redox
530 gradient from euxinic shallow waters to better ventilated deeper waters, as supported by a positive
531 shift in $\delta^{114}\text{Cd}_{\text{auth}}$ that was likely caused by drawdown of isotopically-light Cd (as CdS) in shallower
532 water settings.

533 The early Hirnantian was associated with high marine productivity, as indicated by elevated
534 Cd/Al, Zn/Al and Ni/Al ratios, which accounts for the global positive shift in $\delta^{13}\text{C}_{\text{org}}$ at this time.
535 Following this, deeper water anoxia was re-established (indicated by elevated U/Th ratios) during
536 the Hirnantian deglacial phase (indicated by increased CIA values), and enhanced terrestrial
537 chemical weathering likely contributed to low $\delta^{114}\text{Cd}_{\text{auth}}$ values. A subsequent increase in $\delta^{114}\text{Cd}_{\text{auth}}$
538 then occurred in the Rhuddanian, which can be explained by a decreased input of isotopically-light
539 Cd input to the oceans, as supported by low CIA and Al values indicative of low rates of chemical

540 weathering. Overall, our data provide new insight into the Late Ordovician to Early Silurian Cd
541 cycle, and support the hypothesis that an increase in marine productivity drove enhanced burial of
542 organic matter, potentially contributing to CO₂ drawdown and the initiation of the Hirnantian
543 glaciation.

544

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553

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837 **Figure Captions**

838 **Fig. 1 Palaeogeographic setting.** (A) Late Ordovician global palaeogeography (modified from
839 Zhou et al., 2015) showing the location of South China; (B) Late Ordovician map of the Yangtze

840 block area (modified from Gong et al., 2017) showing the locating of the Wangjiawan (WJW)
841 section.

842

843 **Fig. 2 Lithology and chemostratigraphic profiles for $^{13}\text{C}_{\text{org}}$, TOC, U/Th, molar $\text{C}_{\text{org}}/\text{P}$, CIA and**
844 **Al_2O_3 at the Wangjiawan section.** Se = Series, St = Stage, F = Formation, Lith = Lithology, GZ =
845 graptolite Zones, *D.c* = *Dicellograptus complexus*, *P. pacificus* = *Paraorthograptus pacificus*, *M. e*
846 = *Metabolograptus extraordinarius*, *N. o* = *Normalograptus ojsuensis*, *M. p* = *Metabolograptus*
847 *persculptus*, *A. a* = *Akidograptus ascensus*, *P. a* = *Parakidograptus acuminatus*, *C. v* = *Cystograptus*
848 *vesiculosus*, KYC = Kuanyinchiao. Biostratigraphy based on Chen et al. (2000; 2006).

849

850 **Fig. 3 Scatter plots of TOC versus U/Th (a) and Cd versus Al (b).**

851

852 **Fig. 4 Lithology and chemostratigraphic profiles for $\delta^{114}\text{Cd}/^{110}\text{Cd}_{\text{bulk}}$, $\delta^{114}\text{Cd}/^{110}\text{Cd}_{\text{auth}}$, Cd**
853 **concentrations, Cd/Al, Zn/Al, Ni/Al and P/Al at the Wangjiawan section.** Se = Series, St = Stage,
854 F = Formation, Lith = Lithology, GZ = graptolite Zones, *D.c* = *Dicellograptus complexus*, *P.*
855 *pacificus* = *Paraorthograptus pacificus*, *M. e* = *Metabolograptus extraordinarius*, *N. o* =
856 *Normalograptus ojsuensis*, *M. p* = *Metabolograptus persculptus*, *A. a* = *Akidograptus ascensus*, *P.*
857 *a* = *Parakidograptus acuminatus*, *C. v* = *Cystograptus vesiculosus*, KYC = Kuanyinchiao, UCC =
858 upper continental crust. Biostratigraphy based on Chen et al. (2000; 2006).

859

860 **Fig. 5 Scatter plots of Cd versus TOC (a) and $\delta^{114}\text{Cd}/^{110}\text{Cd}$ versus TOC (b).**

861

862 **Fig. 6 Summary profiles of $\delta^{114}\text{Cd}/^{110}\text{Cd}_{\text{auth}}$, $\delta^{13}\text{C}_{\text{org}}$, U/Th, $\delta^{98}\text{Mo}$, $\delta^{238}\text{U}$ and average sea**
863 **surface temperature (SST) across the LOME.** $\delta^{114}\text{Cd}/^{110}\text{Cd}_{\text{auth}}$, $\delta^{13}\text{C}_{\text{org}}$ and U/Th data are from
864 this study, $\delta^{98}\text{Mo}$ values for the Wangjiawan section are from Zhou et al. (2015), $\delta^{238}\text{U}$ data are
865 from Kozik et al. (2022b), and average sea surface temperatures are from Finnegan et al. (2011).

866

867 **Fig. 7 Summary of the Cd cycle and environment change across the LOME.**