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## Supplementary material

for

### Atomic-scale modelling of organic matter in soil: Adsorption of organic molecules and biopolymers on the hydroxylated $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface

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**Figure S1.** Adsorption configurations of methanol on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface. Red spheres – O atoms, light grey – Al, dark grey – C, white – H atoms.

**Table S1.** Adsorption energies and hydrogen bond lengths for methanol adsorbed on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface. Here and in the following tables, the label “m” refers to the atoms of the adsorbed molecule; “s” refers to surface hydroxyls (OH attached to surface Al atoms), and “ss” refers to surface hydroxyls formed by H atoms attached to surface O atoms.

**Figure S2.** Adsorption configurations of methylamine on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface. Red spheres – O atoms, light grey – Al, dark grey – C, blue – N, white – H atoms.

**Table S2.** Adsorption energies and hydrogen bond lengths for methylamine adsorbed on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface.

**Figure S3.** Adsorption configurations of acetamide on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface.

**Table S3.** Adsorption energies and hydrogen bond lengths for acetamide adsorbed on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface.

**Figure S4.** Adsorption configurations of methyl acetate on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface.

**Table S4.** Adsorption energies and hydrogen bond lengths for methyl acetate adsorbed on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface.

**Figure S5.** Adsorption configurations of acetic acid hydrogen-bonded on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface.

**Table S5.** Adsorption energies and hydrogen bond lengths for acetic acid adsorbed on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface by hydrogen bonding.

**Figure S6.** Adsorption configurations of acetic acid covalently bonded on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface.

**Table S6.** Adsorption energies and bond lengths for acetic acid covalently adsorbed on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface. Labels “C1, C2, C3” refer to covalently bonded acetic acid co-adsorbed with a surface hydroxyl (OH) or with a water molecule, while “P1” refers to covalently bonded acetic acid where the co-adsorbed water was desorbed but remained hydrogen bonded to the surface.

**Table S7.** Adsorption energies and hydrogen bond lengths of the most stable adsorbed configurations of acetic acid, methylamine, acetamide, methanol and methyl acetate calculated using PBE+D3, compared to the same configurations re-optimised using PW91+D3 and vdW-DF2, and B3LYP+D3 energies for vdW-DF2-optimised geometries.

**Figure S7.** Summary of adsorption energies of the most stable adsorbed configurations of acetic acid, methylamine, acetamide, methanol and methyl acetate calculated using PBE+D3, compared to these configurations’ energies calculated using PW91+D3, vdW-DF2 and B3LYP+D3.

**Figure S8.** Adsorption configurations of cellulose on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface.

**Table S8.** Adsorption energies and hydrogen bond lengths for cellulose adsorbed on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface. In this and the following tables, labels “(m)” refer to intramolecular hydrogen bonds, labels “(s)” refer to hydrogen bonds between atoms of the surface, and all other hydrogen bonds are surface-adsorbate hydrogen bonds.

**Figure S9.** Adsorption configurations of pectin on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface.

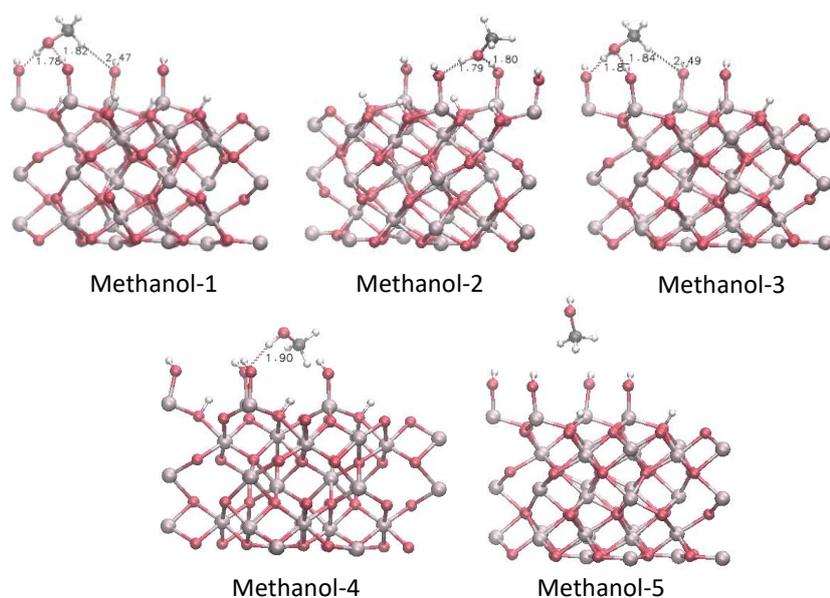
**Table S9.** Adsorption energies and hydrogen bond lengths for pectin adsorbed on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface.

**Figure S10.** Adsorption configurations of chitin on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface.

**Table S10.** Adsorption energies and hydrogen bond lengths for chitin adsorbed on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface.

**Figure S11.** Adsorption configurations of chitosan on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface.

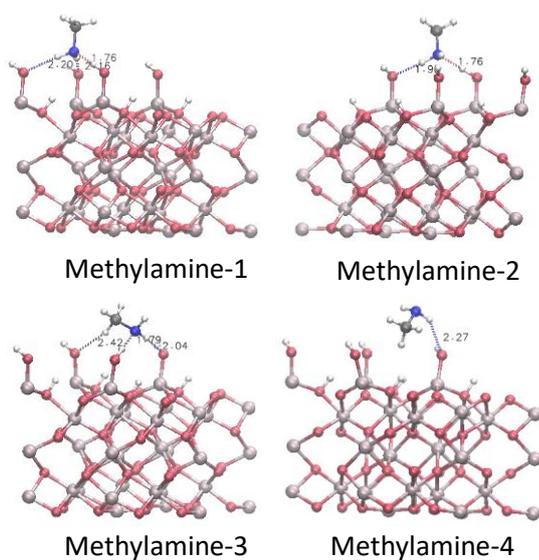
**Table S11.** Adsorption energies and hydrogen bond lengths for chitosan adsorbed on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface.



**Figure S1.** Adsorption configurations of methanol on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface. Red spheres – O atoms, light grey – Al, dark grey – C, white – H atoms.

**Table S1.** Adsorption energies and hydrogen bond lengths for methanol adsorbed on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface. Here and in the following tables, the label “m” refers to the atoms of the adsorbed molecule; “s” refers to surface hydroxyls (OH attached to surface Al atoms), and “ss” refers to surface hydroxyls formed by H atoms attached to surface O atoms.

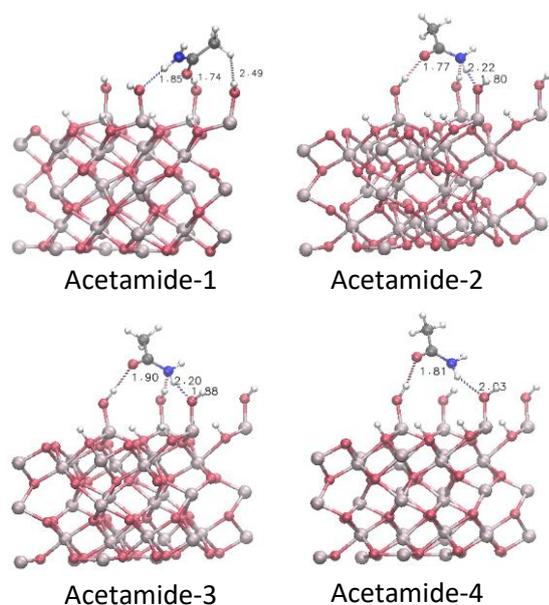
Adsorbate	Adsorption energy, eV	Hydrogen bonds	Hydrogen bond lengths, Å
Methanol-1	-0.71	Om...HOs, OHm...OHs, CHm...OHs	1.82, 1.78, 2.47
Methanol-2	-0.70	Om...HOs, OHm...OHs	1.80, 1.79
Methanol-3	-0.70	Om...HOs, OHm...OHs, CHm...OHs	1.84, 1.88, 2.49
Methanol-4	-0.42	OHm...OHs	1.90
Methanol-5	-0.09	N/A	N/A



**Figure S2.** Adsorption configurations of methylamine on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface. Red spheres – O atoms, light grey – Al, dark grey – C, blue – N, white – H atoms.

**Table S2.** Adsorption energies and hydrogen bond lengths for methylamine adsorbed on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface.

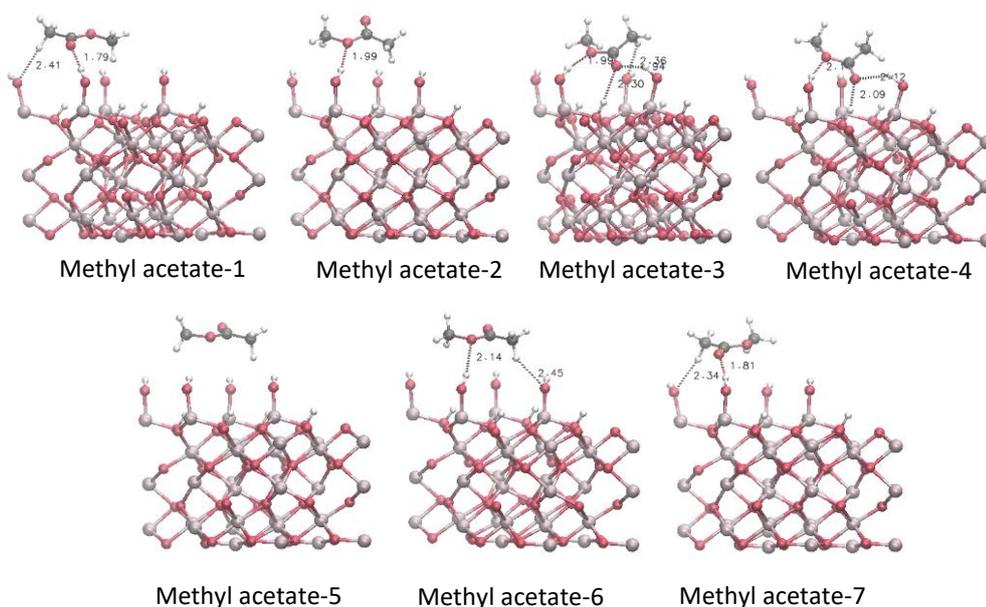
Adsorbate	Adsorption energy, eV	Hydrogen bonds	Hydrogen bond lengths, Å
Methylamine-1	-0.79	Nm...HOs, NHm...OHs, NHm...OHs	1.76, 2.16, 2.20
Methylamine-2	-0.59	Nm...HOs, NHm...OHs,	1.76, 1.95
Methylamine-3	-0.58	Nm...HOs, NHm...OHs, CHm...OHs	1.79, 2.04, 2.42
Methylamine-4	-0.16	NHm...OHs	2.27



**Figure S3.** Adsorption configurations of acetamide on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface.

**Table S3.** Adsorption energies and hydrogen bond lengths for acetamide adsorbed on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface.

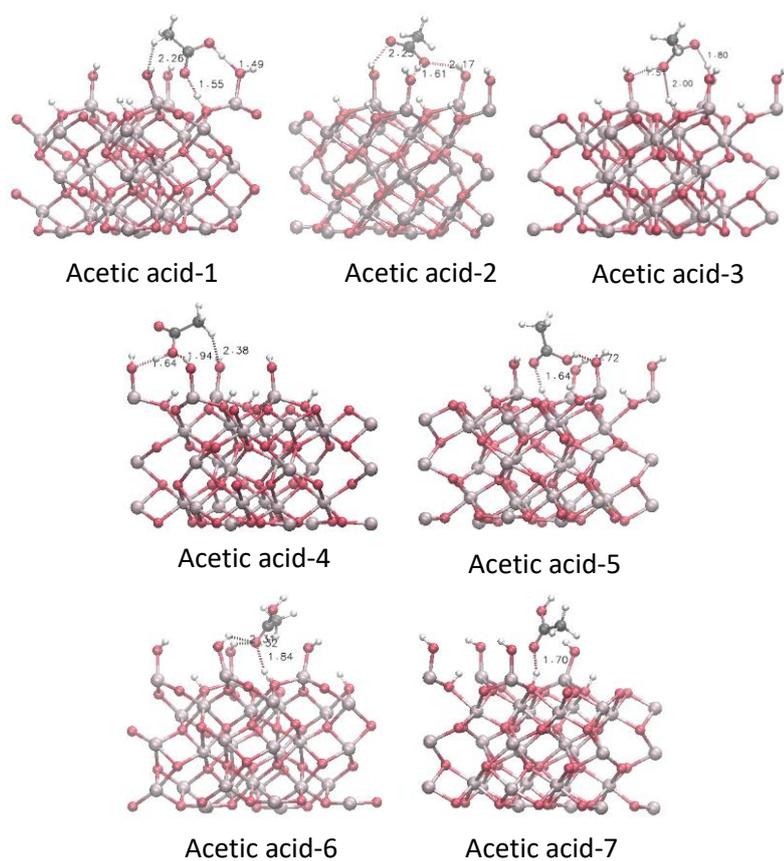
Adsorbate	Adsorption energy, eV	Hydrogen bonds	Hydrogen bond lengths, Å
Acetamide-1	-0.74	Om...HOs, NHm...OHs, CHm...OHs	1.74, 1.85, 2.49
Acetamide-2	-0.70	Om...HOs, NHm...OHs, Nm...HOs	1.77, 1.80, 2.22
Acetamide-3	-0.67	Om...HOs, NHm...OHs, Nm...HOs	1.90, 1.88, 2.20
Acetamide-4	-0.55	Om...HOs, NHm...OHs	1.81, 2.03



**Figure S4.** Adsorption configurations of methyl acetate on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface.

**Table S4.** Adsorption energies and hydrogen bond lengths for methyl acetate adsorbed on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface.

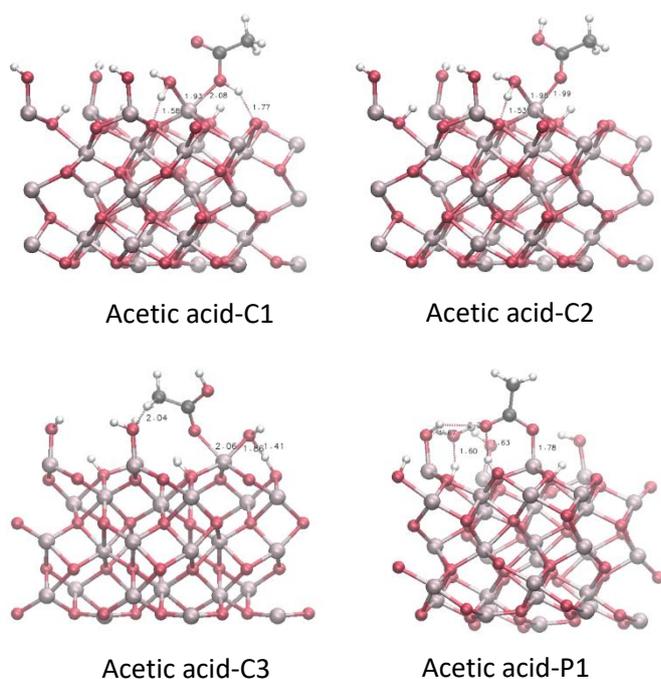
Adsorbate	Adsorption energy, eV	Hydrogen bonds	Hydrogen bond lengths, Å
Methyl acetate-1	-0.49	O(carbonyl)m...HOs, CHm...HOs	1.79, 2.41
Methyl acetate-2	-0.47	O(ether)m...HOs	1.99
Methyl acetate-3	-0.43	O(ether)m...HOs, O(carbonyl)m...HOs, O(carbonyl)m...HOss, CHm...OHs	1.99, 1.94, 2.30, 2.39
Methyl acetate-4	-0.37	O(ether)m...HOs, O(carbonyl)m...HOs, O(carbonyl)m...HOss	2.11, 2.12, 2.09
Methyl acetate-5	-0.25	N/A	N/A
Methyl acetate-6	-0.19	O(ether)m...HOs, CHm...HOs	2.14, 2.45
Methyl acetate-7	-0.11	O(carbonyl)m...HOs, CHm...HOs	1.81, 2.34



**Figure S5.** Adsorption configurations of acetic acid hydrogen-bonded on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface.

**Table S5.** Adsorption energies and hydrogen bond lengths for acetic acid adsorbed on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface by hydrogen bonding.

Adsorbate	Adsorption energy, eV	Hydrogen bonds	Hydrogen bond lengths, Å
Acetic acid-1	-1.11	OHm...OHs, O(carbonyl)m...HOss, CHm...OHs	1.49, 1.55, 2.26
Acetic acid-2	-0.98	OHm...OHs, O(carbonyl)m...HOs, O(hydroxyl)m...HOs	1.61, 2.25, 2.17
Acetic acid-3	-0.95	OHm...OHs, O(carbonyl)m...HOs, O(hydroxyl)m...HOss	1.51, 1.80, 2.00
Acetic acid-4	-0.94	OHm...OHs, O(hydroxyl)m...HOs, CHm...OHs	1.64, 1.94, 2.38
Acetic acid-5	-0.93	OHm...OHs, O(carbonyl)m...HOss	1.72, 1.64
Acetic acid-6	-0.55	O(carbonyl)m...HOss, O(carbonyl)m...HOs, O(carbonyl)m...HOs	1.84, 2.31, 2.32
Acetic acid-7	-0.45	O(carbonyl)m...HOss	1.77



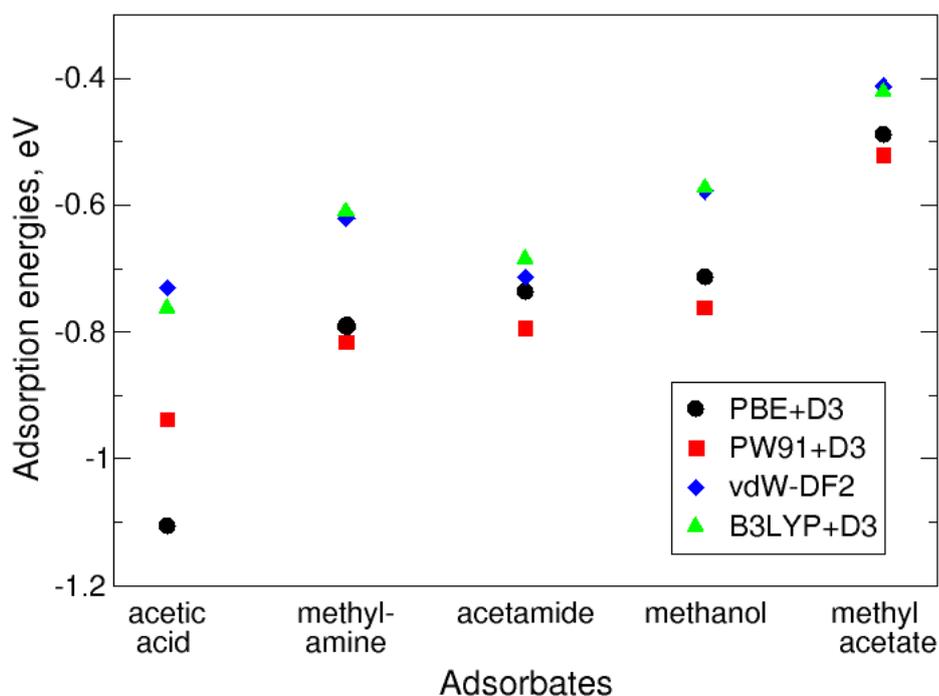
**Figure S6.** Adsorption configurations of acetic acid covalently bonded on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface.

**Table S6.** Adsorption energies and bond lengths for acetic acid covalently adsorbed on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface. Labels “C1, C2, C3” refer to covalently bonded acetic acid co-adsorbed with a surface hydroxyl (OH) or with a water molecule, while “P1” refers to covalently bonded acetic acid where the co-adsorbed water was desorbed but remained hydrogen bonded to the surface.

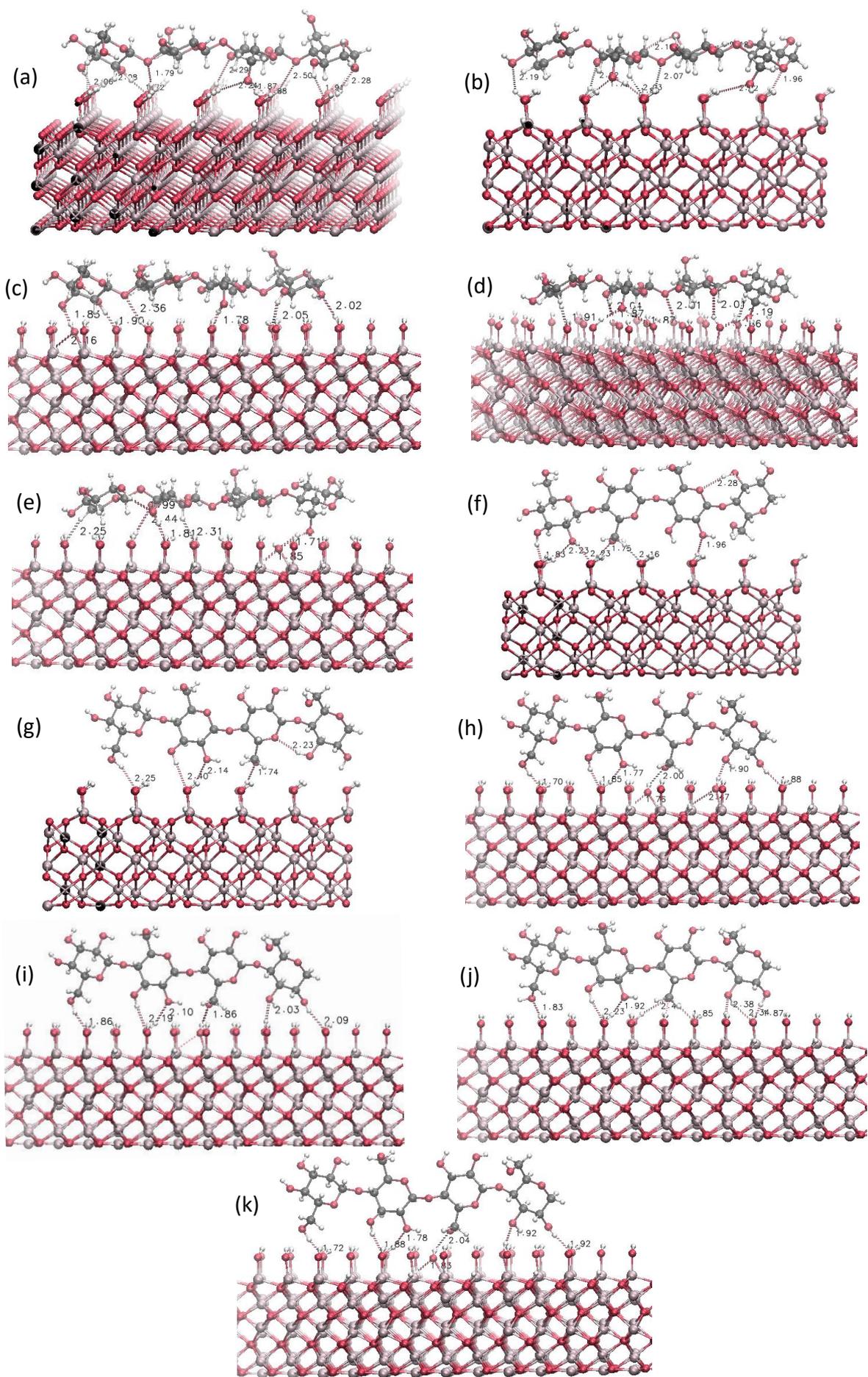
Adsorbate	Adsorption energy, eV	Surface-adsorbate covalent bonds	Covalent bond lengths, Å	Hydrogen bonds	Hydrogen bond lengths, Å
Acetic acid-C1	-0.64	Al-O(acid OH), Al-O(water)	2.08, 1.93	OHm...Oss, OH(water)...Oss	1.77, 1.58
Acetic acid-C2	-0.59	Al-O(acid O=), Al-O(water)	1.99, 1.98	OH(water)...Oss	1.53
Acetic acid-C3	-0.56	Al-O(acid O=), Al-O(OH)	2.06, 1.86	CHm...OHs, CHm...OHs, OHs...HOss	2.04, 2.19, 1.41
Acetic acid-P1	-1.08	Al-O(acid O=)	1.87	OHm...Oss, OH(water)...OHs, O(hydroxyl)m... HO(water), O(water)...HOs	1.63, 1.67, 2.29, 2.39

**Table S7.** Adsorption energies and hydrogen bond lengths of the most stable adsorbed configurations of acetic acid, methylamine, acetamide, methanol and methyl acetate calculated using PBE+D3, compared to the same configurations re-optimised using PW91+D3 and vdW-DF2, and B3LYP+D3 energies for vdW-DF2-optimised geometries.

Adsorbate	Adsorption energies, eV				Hydrogen bond lengths, Å			
	PBE +D3	PW91 +D3	vdW-DF2	B3LYP +D3	Hydrogen bonds	PBE +D3	PW91 +D3	vdW-DF2
Acetic acid	-1.11	-0.94	-0.73	-0.76	OHm...OHs,	1.49,	1.49,	1.66,
					O(carbonyl)m...HOss	1.55	1.56	1.91
Methylamine	-0.79	-0.82	-0.62	-0.61	Nm...HOs,	1.76,	1.76,	1.84,
					NHm...OHs,	2.16,	2.16,	2.19,
					NHm...OHs	2.20	2.22	2.25
Acetamide	-0.74	-0.79	-0.71	-0.68	Om...HOs,	1.74,	1.73,	1.84,
					NHm...OHs	1.85	1.86	1.91
Methanol	-0.71	-0.76	-0.58	-0.57	Om...HOs,	1.82,	1.83,	1.84,
					OHm...OHs	1.78,	1.78	1.90
Methyl acetate	-0.49	-0.52	-0.41	-0.42	O(carbonyl)m...HOs,	1.79,	1.79,	1.86,
					CHm...HOs	2.41	2.43	2.51



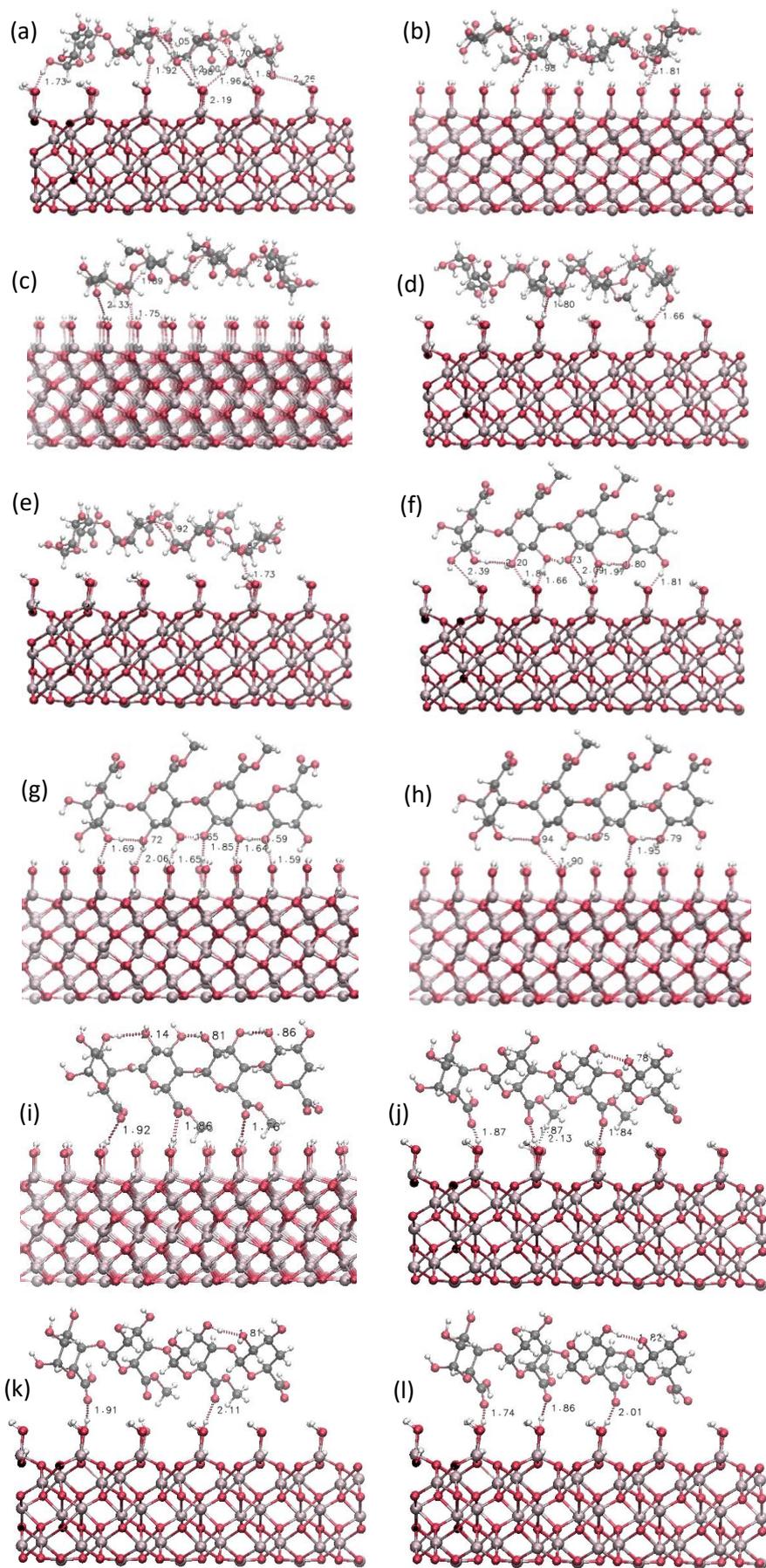
**Figure S7.** Summary of adsorption energies of the most stable adsorbed configurations of acetic acid, methylamine, acetamide, methanol and methyl acetate calculated using PBE+D3, compared to these configurations' energies calculated using PW91+D3, vdW-DF2 and B3LYP+D3.



**Figure S8.** Adsorption configurations of cellulose on the hydroxylated  $\alpha\text{-Al}_2\text{O}_3$  (0001) surface.

**Table S8.** Adsorption energies and hydrogen bond lengths for cellulose adsorbed on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface. In this and the following tables, labels “(m)” refer to intramolecular hydrogen bonds, labels “(s)” refer to hydrogen bonds between atoms of the surface, and all other hydrogen bonds are surface-adsorbate hydrogen bonds.

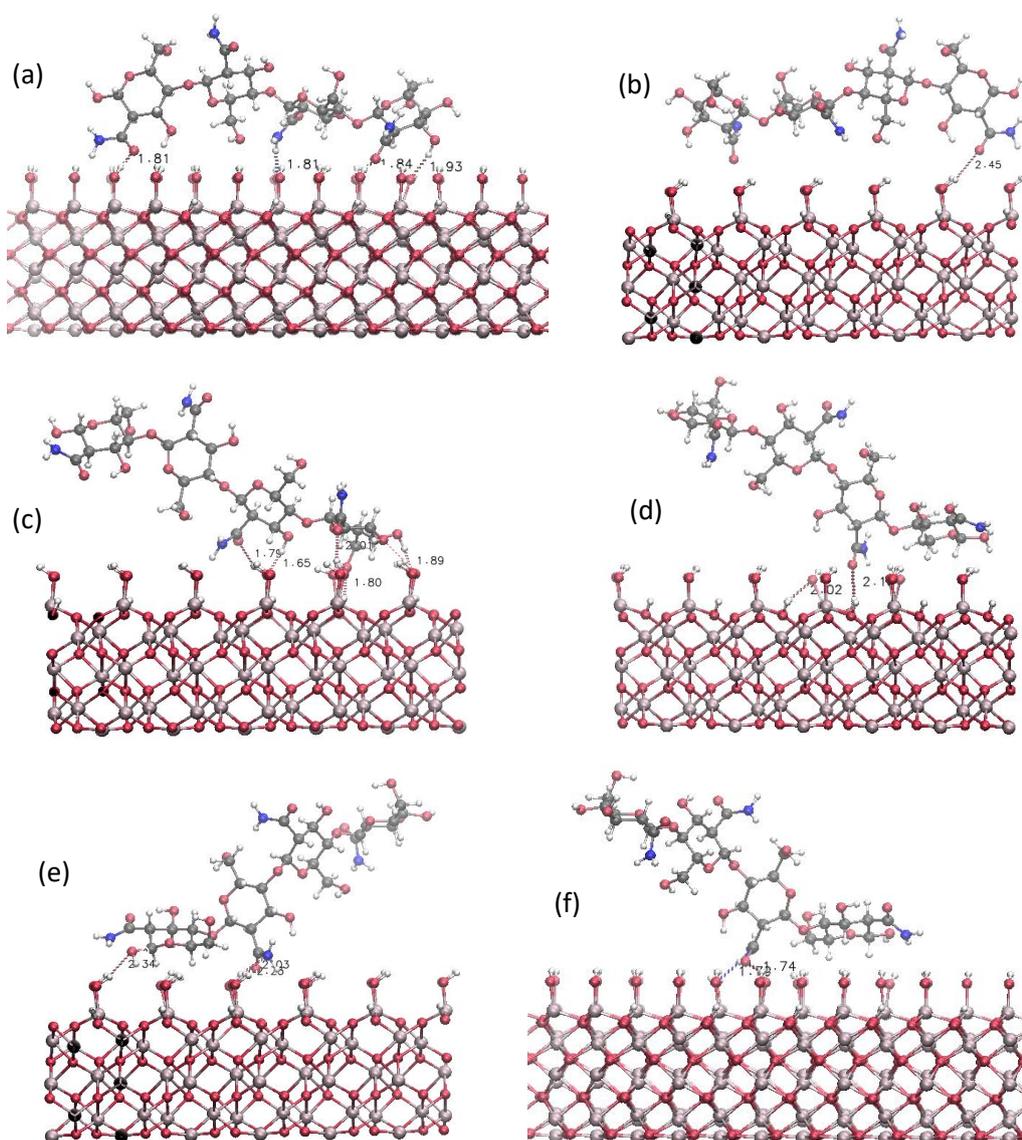
Number	Configuration	Adsorption energy, eV	Number of hydrogen bonds	Hydrogen bond lengths, Å
a	flat, along y	-2.92	11	2.06, 2.08, 1.72, 1.79, 2.29, 2.24, 1.87, 1.88, 2.50, 1.94, 2.28
b	flat, along y	-2.81	10	2.19, 2.02, 1.76, 1.85, 2.03, 2.07, 2.19(m), 2.18(m), 2.42, 1.96
c	flat, along x	-2.28	7	2.16(s), 1.83, 1.90, 2.36, 1.78, 2.05, 2.02
d	flat, along x	-1.90	8	1.91, 2.04, 1.87, 1.87, 2.01, 2.01, 1.86, 2.19
e	flat, along x	-1.31	7	2.25, 2.44, 1.99(m), 1.81, 2.31, 1.85(s), 1.71
f	edge (OH down), along y	-1.87	7	1.83, 2.23, 2.23, 1.75, 2.16, 1.96, 2.28(m)
g	edge (OH down), along y	-1.26	5	2.25, 2.40, 2.14, 1.74, 2.23(m)
h	edge (OH down), along x	-1.40	7	1.70, 1.85, 1.77, 1.90, 1.88, 1.76(s), 2.47(s)
i	edge (OH down), along x	-1.03	7	1.86, 2.19, 2.10, 1.86, 2.03, 2.09, 2.39(s)
j	edge (OH down), along x	-1.42	8	1.83, 2.23, 1.92, 2.43, 1.85, 2.38, 2.34, 1.87
k	edge (OH down), along x	-1.30	7	1.72, 1.88, 1.78, 1.83(s), 2.04, 1.92, 1.92



**Figure S9.** Adsorption configurations of pectin on the hydroxylated  $\alpha\text{-Al}_2\text{O}_3$  (0001) surface.

**Table S9.** Adsorption energies and hydrogen bond lengths for pectin adsorbed on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface.

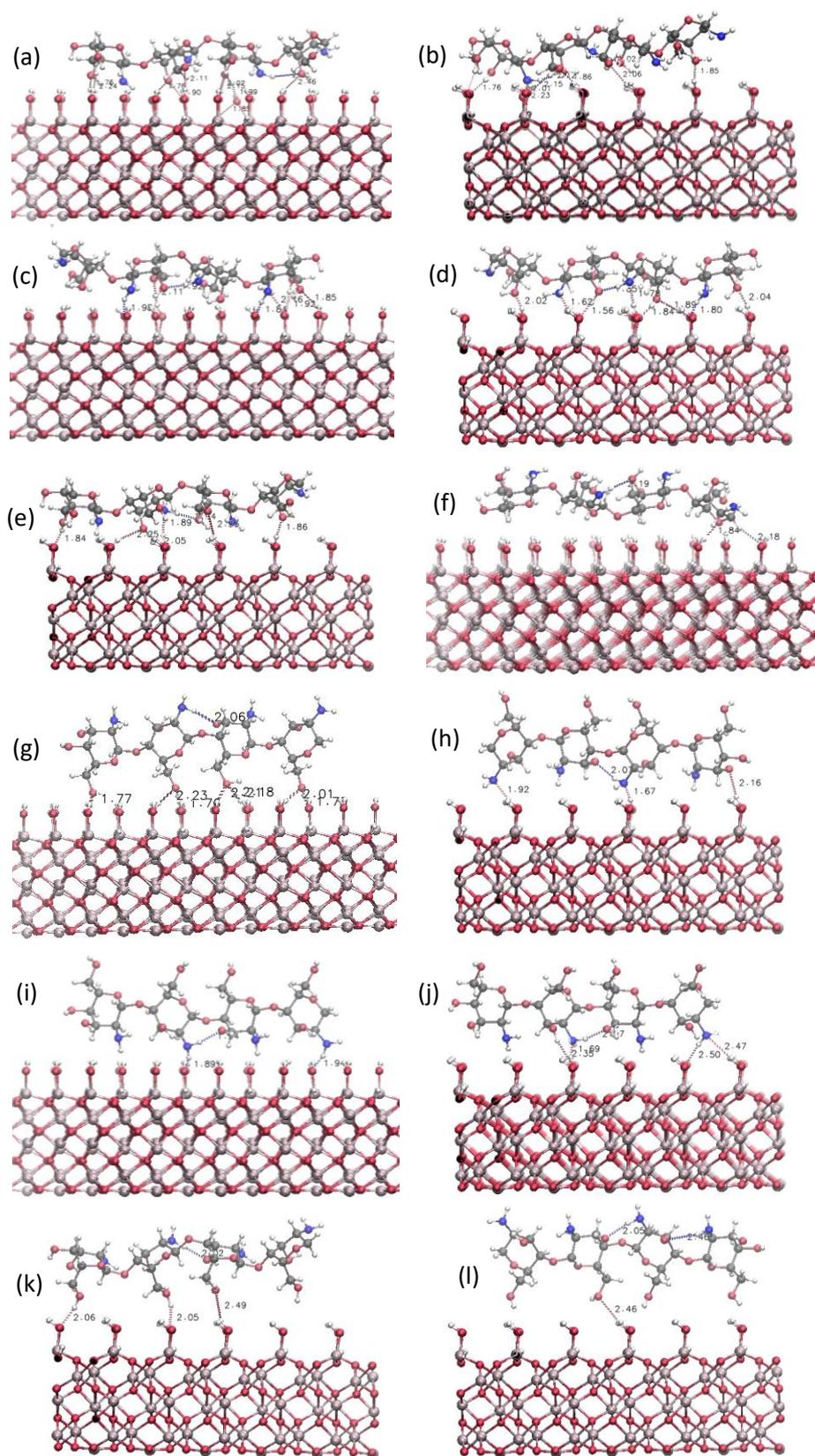
Number	Configuration	Adsorption energy, eV	Number of hydrogen bonds	Hydrogen bond lengths, Å
a	flat (methyl up), along y	-2.43	10	1.76, 1.92, 2.05(m), 1.98, 2.00, 1.70(m), 1.96, 2.19(s), 1.81, 2.26
b	flat (methyl down), along x	-2.05	5	1.98, 1.81, 1.91(m), 1.75(m), 1.94(m)
c	flat (methyl up), along x	-1.71	5	1.75, 2.33, 1.89(m), 1.83(m), 2.11(m)
d	flat (methyl down), along y	-1.46	3	1.66, 1.80, 1.80(m)
e	flat (methyl up), along y	-1.25	3	1.73, 1.82(m), 1.92(m)
f	edge, OH down, along y	-2.50	9	2.39, 2.20(m), 1.84, 1.66, 1.73(m), 2.09, 1.97, 1.80(m), 1.81
g	edge, OH down, along x above gap	-2.42	9	1.69, 2.06, 1.65, 1.86, 1.64, 1.59, 1.72(m), 1.65(m), 1.59(m)
h	edge, OH down, along x above OH	-1.90	5	1.90, 1.95, 1.94(m), 1.75(m), 1.79(m)
i	edge, ester down, along x above gap	-1.01	6	1.92, 1.86, 1.76, 2.14, 1.81, 1.86(m)
j	~edge, ester down, along y	-0.52	5	1.87, 1.87, 2.13, 1.84, 1.78(m)
k	~edge, ester down, along y	-0.50	3	1.91, 2.11, 1.81(m)
l	~edge, ester down, along y	-0.34	4	1.74, 1.86, 2.01, 1.82(m)



**Figure S10.** Adsorption configurations of chitin on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface.

**Table S10.** Adsorption energies and hydrogen bond lengths for chitin adsorbed on the hydroxylated  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> (0001) surface.

Number	Configuration	Adsorption energy, eV	Number of hydrogen bonds	Hydrogen bond lengths, Å
a	parallel, along x	-2.03	4	1.81, 1.81, 1.84, 1.93
b	parallel, along y	-1.05	1	2.45
c	half-adsorbed, along y	-1.71	6	1.89, 1.83, 1.80, 2.01, 1.65, 1.79
d	half-adsorbed, along x	-1.47	2	2.18, 2.02(s)
e	half-adsorbed, along y	-1.29	3	2.34, 2.23, 2.03
f	half-adsorbed, along x	-0.86	2	1.79, 1.74



**Figure S11.** Adsorption configurations of chitosan on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface.

**Table S11.** Adsorption energies and hydrogen bond lengths for chitosan adsorbed on the hydroxylated  $\alpha$ - $\text{Al}_2\text{O}_3$  (0001) surface

Number	Configuration	Adsorption energy, eV	Number of hydrogen bonds	Hydrogen bond lengths, Å
a	flat, amine and OH down, along x	-2.66	11	1.76, 2.24, 1.76, 1.90, 2.11, 2.02, 2.15, 1.99, 1.85(s), 2.35(m), 2.46
b	flat, amine and OH down, ~along y	-2.49	9	1.76, 1.21, 2.23, 2.12, 2.15, 1.86, 2.06, 1.85, 2.03(m)
c	flat, amine and OH down, along x	-2.39	7	1.98, 2.11, 1.92(m), 1.84, 2.16, 1.92, 1.85
d	flat, amine and OH down, along y	-2.29	9	2.02, 1.62, 1.56, 1.85(m), 1.78, 1.84, 1.89, 1.80, 2.04
e	flat, amine and OH down, along y	-2.09	7	1.84, 2.25, 2.05, 1.89, 1.94(m), 2.33, 1.86
f	flat, amine and OH up, along x	-1.30	3	1.84, 2.18, 2.19(m)
g	edge, OH down, along y	-2.39	8	1.77, 2.23, 2.76, 2.21, 2.18, 2.06(m), 2.01, 1.78
h	edge, amine down, along y	-1.69	4	1.92, 2.07(m), 1.67, 2.16
i	edge, amine down, along x	-1.65	3	1.89, 1.94, 1.94(m)
j	edge, amine down, along y	-1.61	5	2.35, 1.65, 2.07(m), 2.50, 2.47
k	~edge, OH down, along y	-1.12	4	2.49, 2.05, 2.06, 2.02
l	edge, OH down, along y	-0.73	4	1.74, 1.86, 2.01, 1.82(m)