



UNIVERSITY OF LEEDS

This is a repository copy of *Saccharomyces cerevisiae-incorporated and sucrose-rich sodium alginate film: An effective antioxidant packaging film for longan preservation*.

White Rose Research Online URL for this paper:

<https://eprints.whiterose.ac.uk/196822/>

Version: Accepted Version

Article:

Yang, Z, Zhai, X, Li, M et al. (10 more authors) (2022) *Saccharomyces cerevisiae-incorporated and sucrose-rich sodium alginate film: An effective antioxidant packaging film for longan preservation*. *International Journal of Biological Macromolecules*, 223 (Part A). pp. 673-683. ISSN 0141-8130

<https://doi.org/10.1016/j.ijbiomac.2022.11.039>

© 2022 Published by Elsevier B.V. This manuscript version is made available under the CC-BY-NC-ND 4.0 license <http://creativecommons.org/licenses/by-nc-nd/4.0/>.

Reuse

This article is distributed under the terms of the Creative Commons Attribution-NonCommercial-NoDerivs (CC BY-NC-ND) licence. This licence only allows you to download this work and share it with others as long as you credit the authors, but you can't change the article in any way or use it commercially. More information and the full terms of the licence here: <https://creativecommons.org/licenses/>

Takedown

If you consider content in White Rose Research Online to be in breach of UK law, please notify us by emailing eprints@whiterose.ac.uk including the URL of the record and the reason for the withdrawal request.



eprints@whiterose.ac.uk
<https://eprints.whiterose.ac.uk/>

Highlights

S. cerevisiae with satisfactory GSH production was selected as the antioxidant agent.

Active films made from sodium alginate, sucrose and *S. cerevisiae* were developed.

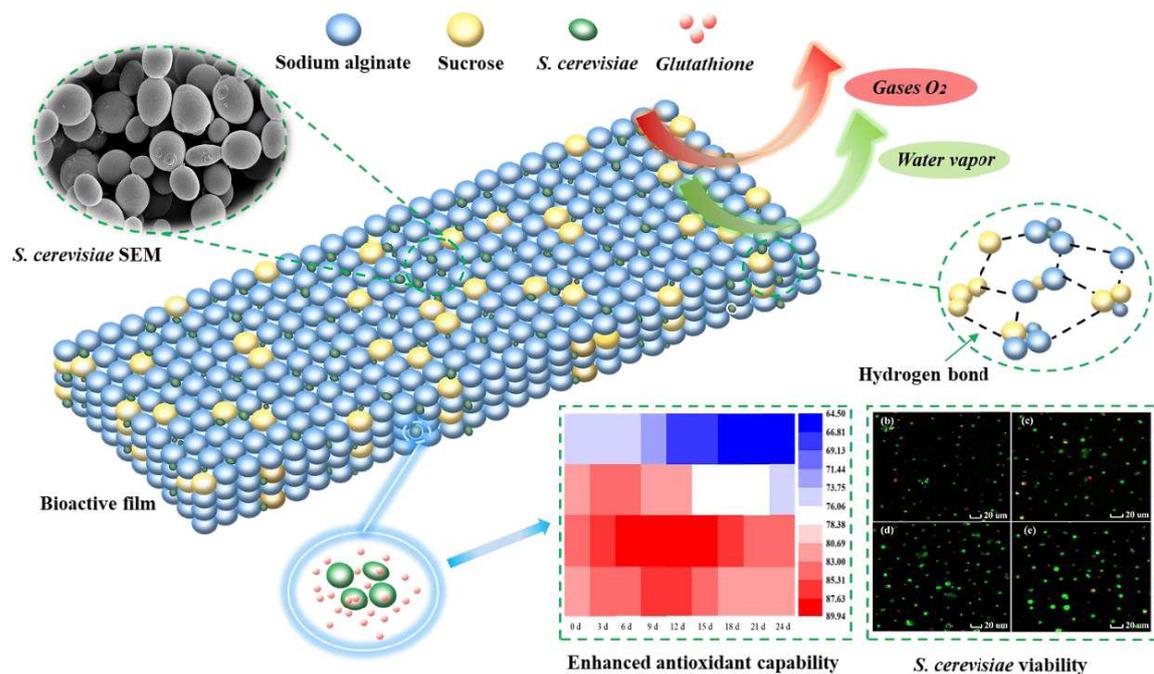
Sucrose significantly improved the mechanical property of the films.

Sucrose maintained the antioxidant property of the films during the storage period.

Active films improve the quality and pericarp browning of longan.

Abstract

A sodium alginate (SA) film incorporated with *Saccharomyces cerevisiae* (SE) and sucrose (SU) was fabricated to control the quality and pericarp browning of longan. The SE with satisfactory glutathione production was selected as the antioxidant agent. The scanning electron microscopy (SEM) results revealed that the SU-rich SA film could be used as an effective carrier to protect the cell integrity of SE. The FTIR and mechanical property results indicated that the SA-SE film with the incorporation of SU has good flexibility due to the existence of hydrogen bonds. Notably, the cell viability of the SE was significantly improved with the addition of SU, which positively affects the antioxidant property of the film during the storage period. Finally, the SA-SU-SE films obviously improved the quality and pericarp browning of longan. The SA-based film incorporated with SU and SE may be established as a novel antioxidant fruit packaging material.



1 ***Saccharomyces cerevisiae*-encapsulated and sucrose-rich sodium alginate**
2 **film: an effective antioxidant packaging film for longan preservation**

3
4 4 Zhikun Yang ^a, Xiaodong Zhai ^a, Mingrui Li ^a, Zhihua Li ^a, Jiyong Shi ^a, Xiaowei Huang ^{a,*}, Xiaobo Zou ^{a,*}, Yunyun Gong ^b,

5
6 5 Melvin Holmes ^b, Megan Povey ^b, Jianbo Xiao ^c

7
8 6 ^aAgricultural Product Processing and Storage Lab, School of Food and Biological Engineering, Jiangsu

9
10 7 University, Zhenjiang, Jiangsu 212013, China

11
12 8 ^bSchool of Food Science and Nutrition, University of Leeds, Leeds LS2 9JT, United Kingdom

13
14 9 ^cInstitute of Chinese Medical Sciences, State Key Laboratory of Quality Research in Chinese Medicine,

15
16 10 University of Macau, Taipa, Macau, China

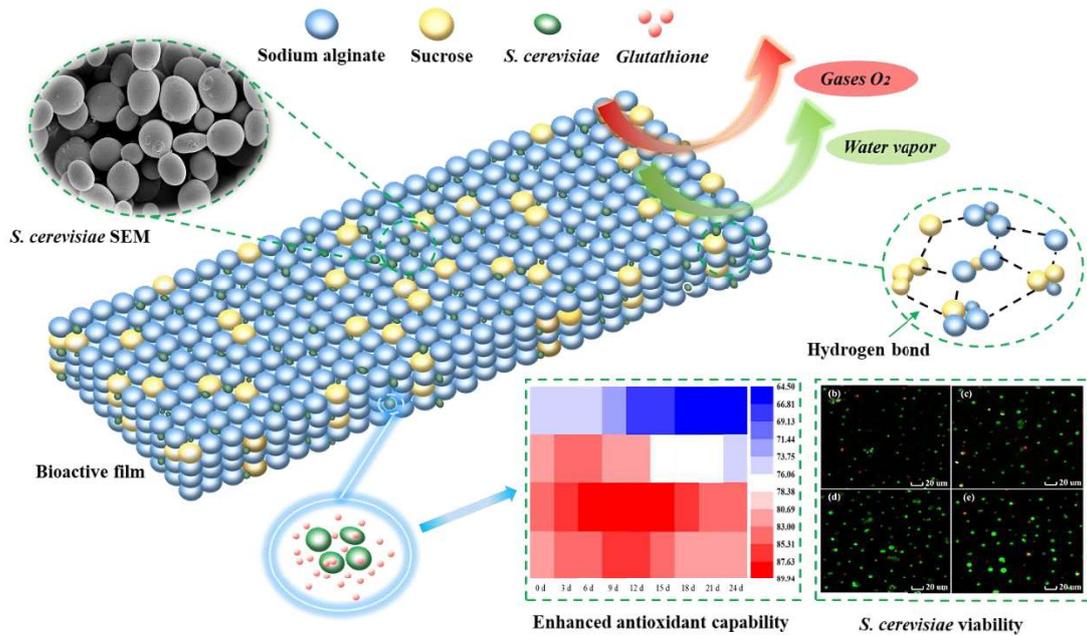
17
18 11 *Corresponding author. Email: zou_xiaobo@ujs.edu.cn (Xiaobo Zou)

19
20 12 *Corresponding author. Email: huangxiaowei@ujs.edu.cn (Xiaowei Huang)

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
61
62
63
64
65

13 **Abstract**

14 A sodium alginate (SA) film incorporated with *Saccharomyces cerevisiae* (SE) and sucrose (SU) was
15 fabricated to control the quality and pericarp browning of longan. The SE with satisfactory glutathione
16 production was selected as the antioxidant agent. The scanning electron microscopy (SEM) results
17 revealed that the SU-rich SA film could be used as an effective carrier to protect the cell integrity of SE.
18 The FTIR and mechanical property results indicated that the SA-SE film with the incorporation of SU
19 has good flexibility due to the existence of hydrogen bonds. Notably, the cell viability of the SE was
20 significantly improved with the addition of SU, which positively affects the antioxidant property of the
21 film during the storage period. Finally, the SA-SU-SE films obviously improved the quality and pericarp
22 browning of longan. The SA-based film incorporated with SU and SE may be established as a novel
23 antioxidant fruit packaging material.



24

25 **Keywords:** Bioactive film; *Saccharomyces cerevisiae*; Glutathione; Antioxidant; Longan fruit;

26

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
61
62
63
64
65

27 Introduction

28 Longan (*Dimocarpus longan Lour.*) is popular with consumers due to its unique flavor and high nutrient
29 value. However, the shelf life and market value of the longan fruit could be significantly reduced due to
30 post-harvest physiological changes, especially pericarp browning caused by enzyme[1]. Hence, it is
31 essential to control the postharvest quality of longan fruit.

32 When it comes to the preservation and browning control of fruits, the active packaging materials are
33 always attracted more attention than traditional materials due to they can interact positively with food
34 and the environment so that improve the quality of the foods [2]. The active packaging generally consists
35 of the film-forming polymer matrix and the natural active substance. Several natural polymers have been
36 widely used as the film-forming matrix to develop active packaging materials [3]. Sodium alginate (SA)
37 is an anionic polysaccharide that contains two structural units of 1–4 linked α -l-guluronic acid and β -d-
38 mannuronic acid. it has been widely used as an effective active substance carrier for developing
39 packaging materials because it has low price and good film-forming properties [4].

40 Various natural substances have been added to the film-forming polymer matrix to develop active
41 packaging materials [5]. Recently, numerous natural bioactive agents made from microorganisms or their
42 derivatives have received considerable attention in food application due to their effective biological
43 activities and health benefits, such as antimicrobial activity, antioxidant activity, and anticancer [6]. The
44 European Union Novel Food regulation (9258/1997 EEC) states that the microorganism used in food
45 industry (including food products or food packaging) should be (Qualified Presumption of Safety) QPS
46 [6]. Several studies have tried to use the some QPS microorganisms as the potential active substance to
47 develop active packaging films or coatings [7, 8].

48 *Saccharomyces cerevisiae* as a QPS microbial with good biological activity has been widely applied in
49 foods manufacture and medicine production industry [8-12]. Remarkably, *S. cerevisiae* produces
50 glutathione (GSH) is a low-molecular-mass thiol, which has demonstrated antioxidant capability, anti-
51 aging, and detoxification [13]. Moreover, several studies have reported that the GSH could effectively
52 inhibit the enzymatic browning and enzyme activity in fruits [14]. Hence, *S. cerevisiae* cells with
53 satisfactory GSH production could be used as a novel antioxidant agent to develop active packaging
54 materials.

55 Actually, one of the main concerns of packaging materials containing microorganisms is the cell viability
56 of the incorporating microorganism during the storage period, because the viability of the microorganism

1 57 affects their cell numbers or biomass production, which affects the bioactive property of the film [2].
2 58 Hence, the improvement of microorganism viability is critical for strengthening the bioactive property
3
4 59 of the film. On the other hand, SA film as a common food preservation film has poor mechanical property
5
6 60 and low oxygen permeability, which restrict its applications. Sucrose (α -D-glucopyranosyl-(1 \leftrightarrow 2)- β -D-
7
8 61 fructofuranoside) is the world's most abundant free low molecular weight carbohydrate. Several studies
9
10 62 have reported that sucrose could retain the cell viability of *S. cerevisiae* under harsh storage conditions
11
12 63 and increase the biomass production of *S. cerevisiae* [15-17]. Moreover, studies have demonstrated that
13
14 64 sucrose could be used as a plasticizer to increase polymer chain mobility so that improve the mechanical
15
16 65 property of the natural polymer film [18]. Therefore, sucrose could be added into the SA film containing
17
18 66 *S. cerevisiae* to improve the bioactive and mechanical properties of the film.
19
20 67 However, to our best knowledge, no works regarding the use of sucrose to improve the bioactive and
21
22 68 mechanical properties of SA-*S. cerevisiae* films for controlling the quality and pericarp browning of
23
24 69 longan fruits. Hence, the present work aimed to fabricate the SA active films enriched with *S. cerevisiae*
25
26 70 and sucrose. The effect of sucrose on the viability of *S. cerevisiae* in SA films was explored. The
27
28 71 mechanical and biological properties of the films were also evaluated. Finally, the bioactive film was
29
30 72 applied to control the quality and pericarp browning of longan fruits at ambient temperature (25 ± 1 °C).

31 73 **2. Materials and methods**

32 74 **2.1. Materials**

33 75 *Saccharomyces cerevisiae* (CICC 1048) was obtained from China Center of industrial Culture collection.
34
35 76 All chemical agents including sodium alginate, sucrose, methanol, GSH standards, fluorescein diacetate
36
37 77 (FDA) and propidium iodide (PI) were obtained from Nanjing Jiancheng Technology Co., Ltd. (Jiangsu,
38
39 78 China).

40 79 **2.2. Preparation of *S. cerevisiae* culture**

41 80 The yeast was cultured in nutrient yeast extract peptone sucrose broth. The prepared yeast cultures were
42
43 81 packed in an Erlenmeyer flask and incubated at 25 °C for 24 h. The *S. cerevisiae* culture was centrifuged
44
45 82 at 5000 rpm for 10 min to obtain the precipitate. The obtained precipitate was washed two times with
46
47 83 sterile normal saline and then centrifuged to prepare the *S. cerevisiae* pellet.

48 84 **2.3. Determination of growth curve and GSH production**

49 85 The growth curve was measured by noting the absorbance of *S. cerevisiae* cultures at 600 nm every
50
51 86 2 hours. The nutrient yeast extract peptone sucrose broth was served as the control. The intracellular
52
53
54
55
56
57
58
59
60
61
62
63
64
65

1 87 GSH production curve was measured using the 5,5'-Dithiobis-(2-nitrobenzoic acid) (DTNB) method.
2 88 The content of GSH in the film was measured using the DTNB method. The *S. cerevisiae* pellet (0.05 g)
3 89 was mixed with 4 mL of (40% v/v) menthol and cultured at 25 °C for 2 hours. Then, the supernatant was
4 90 obtained by centrifuging the mixture at 6000 rpm, 4 ± 1 °C for 15 min. Tris-HCl buffer and DTNB
5 91 solution were then added to 1 mL of the supernatant and reacted at 25°C for 15 min. The absorbance value
6 92 at 412 nm was measured using 2 mL of deionization water as a blank control. The absorbance value was
7 93 substituted into Eq. (1) to calculate the GSH content. The GSH content was determined every 2 hours
8 94 within 36 hours in this work.

$$17 \quad y_1 = (x + 0.0228)/1.17 \quad (1)$$

18 96 Where x is the absorbance value at 412 nm. y_1 : the content of GSH in the pellet (mg/100g *S. cerevisiae*
19 97 pellet).

23 98 **2.4. The fabrication of the films**

24 99 SA powder (2.0% w/v) was dissolved in distilled water at 60 °C under continuous stirring to obtain an
25 100 SA solution. The SA-*S. cerevisiae* solutions (SA-SE) were prepared by adding *S. cerevisiae* pellet into
26 101 the SA solution at 30 °C. The concentration of *S. cerevisiae* in SA-SE solutions was 1×10^8 cells
27 102 mL⁻¹. The different concentrations of sucrose (1.0% w/v, 3.0% w/v, and 6.0% w/v) were respectively
28 103 added into the SA-SE solutions at 30 °C to prepare the SA-*S. cerevisiae*-1.0% sucrose (SA-SE-1.0%SU),
29 104 SA-*S. cerevisiae*-3.0% sucrose (SA-SE-3.0%SU) and SA-*S. cerevisiae*-6.0% sucrose (SA-SE-6.0%SU)
30 105 solutions. The films were obtained by casting all prepared solutions into the sterile plates and then drying
31 106 at 30 °C for 12 h in a sterile oven.

41 107 **2.5. Viability of *S. cerevisiae* in the films during the storage period**

42 108 The bioactive film incorporated with *S. cerevisiae* was stored at 25 ± 1 °C and 75% relative humidity
43 109 (RH) for 24 d, and the viability of the yeast was evaluated every three days by determining the colony-
44 110 forming units using the plate-count method. Briefly, film samples (1 g) were aseptically transferred into
45 111 9 mL of the sterile saline solution and agitated to ensure that all yeasts were released into the solution.
46 112 Finally, a microorganism solution of an appropriate concentration was spread on a yeast extract peptone
47 113 dextrose (YEPD) medium.
48 114 Moreover, to check the cellular state of the yeast, the fluorescence of the films after storage for 24 days
49 115 was also checked with a confocal laser microscope (Leica TCS SP5) at 480 nm excitation wavelength.
50 116 The FDA was dissolved in acetone at a concentration of 10 µg/mL. PI was dissolved in distilled water at

1 117 a concentration of 5 µg/mL. Films were first stained with FDA and then with PI, both for 20 min at room
2 118 temperature in the dark. The viable yeast cells were presented with green fluorescence; dead yeast cells
3
4 119 were presented with red fluorescence.
5

6 120 **2.6. Microstructure observation of the film and yeast cell**

7
8 121 A scanning electron microscope (JEOL, JSM-6360) was used to observe the film morphology. The tested
9 122 films were fixed on bronze stub using double-side adhesive and then sputtered with gold in a vacuum
10 123 evaporator.
11

12 124 Cell morphology was observed by a cold field emission SEM (SU8200, HITACHI, Japan). The *S.*
13 125 *cerevisiae* cultures were centrifuged at 4 °C and 2000 rpm for 10 min and then fixed with 5%
14 126 glutaraldehyde. After fixation, the samples were rinsed with phosphate buffer (0.1 M). Subsequently, the
15 127 samples were fixed with 1% ruthenium tetroxide and rinsed thrice in the same buffer. After the
16 128 pretreatment, the samples were dehydrated with ethanol and then dried at the CO₂ critical point to obtain
17 129 the dried samples. The dried samples were attached with an aluminum foil for ion sputtering.
18

19 130 **2.7. Fourier transform infrared (FTIR) spectroscopy**

20 131 FT-IR spectrum of the film was measured to evaluate the chemical structure of the bioactive film. The
21 132 spectrum of the dried film sample was analyzed in the range of 525–4000 cm⁻¹ with a Nicoletis50
22 133 infrared spectrometer (Perkine Elmer 16 PC spectrometer, Boston, USA).
23

24 134 **2.8. Antioxidant property of the films**

25 135 The antioxidant property of the film was evaluated by analysis the scavenging ability of DPPH radicals.
26 136 Briefly, 9 mL of the film extract solution was mixed with 2 mL of DPPH solution (0.1 mM solution in
27 137 ethanol) and incubated under the dark for 45 minutes. The absorbance of the solution at 517 nm was
28 138 noted with a UV spectrophotometer. The value of DPPH scavenging activity was evaluated based on Eq.
29 139 (2):
30

$$31 140 \text{ DPPH scavenging activity (\%)} = (A_0 - A_1)/A_0 \times 100\% \quad (2)$$

32 141 Where A_1 and A_0 signify the absorbance of the DPPH solution mixed with or without film extract solution.
33

34 142 **2.9. The barrier and mechanical properties**

35 143 The water vapor permeability (WVP) of all samples was measured referred to the method reported by
36 144 Yang Zhikun et al. [19]. Firstly, a centrifuge tube (15 mL) was filled with 10 mL of distilled water and
37 145 covered with the fabricated film. After that, the centrifuge tube was placed in the dryer. The weight of
38 146 the centrifuge tube was weighted every two hours. The following formula obtained the WVP:
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
61
62
63
64
65

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
61
62
63
64
65

147 $WVP = (\Delta m \times \Delta P) / (A \times \Delta t \times x)$ (3)

148 Where Δm represents the weight gain of the tube per unit time Δt (s); A is the area of exposed film; x is
149 the thickness of the film, and Δp is the partial water vapor pressure difference on both sides of the film.

150 The oxygen permeability (OP) of the film sample was determined with an automated oxygen
151 permeability testing machine according to Zhai Xiaodong et al. [20].

152 Tensile strength (TS) and elongation percentage at break (EB) of the film samples (20 × 60 mm
153 rectangular shapes) were conducted using a Texture analyzer (TA-XT2i, Stable Micro Systems Ltd,
154 Surrey, UK) according to the ASTM method. The initial grip separation distance was 40 mm, and the
155 cross-head speed was 1 mm/s. The TS and EB were obtained by the following formula:

156 $TS = F/S$ (4)

157 $EB(\%) = \Delta l/l_0$ (5)

158 **2.10. Visual appearance and transmittance**

159 The visual appearance of the film was obtained with a scanner. The transmittance of films was measured
160 at 200–800 nm wavelengths using a spectrophotometer.

161 **2.11. Potential application on longan fruits**

162 Fresh longan fruits (cultivar “Shixia”) with healthy outer and uniform size were provided by local
163 producers. The effect of browning control of longan fruits by the fabricated film was measured in this
164 work. The longan fruits were washed with distilled water and then air-dried. The air-dried fruits were
165 immersed in different film-forming solutions and air-dried again. The fruits only washed with distilled
166 water served as the control. After that, all treated fruits were placed in a polyethylene tray and covered
167 with polyethylene film, and then stored at 25 ± 1 °C and 75% RH for eight days.

168 **2.11.1. Browning index**

169 The pericarp browning of longan fruits was assessed by the analysis of the extent of total browning area
170 on each fruit pericarp and evaluated by using the following scales: where 1 represents no browning; 2
171 represents slight browning; 3 represents $\leq 25\%$ browning; 4 represents 25%~50% browning; 5 represents
172 $\geq 50\%$ browning. The browning index was calculated using the following Eq: \sum (browning scale ×
173 percentage of corresponding fruit within each class).

174 **2.11.2. Polyphenol oxidase (PPO) and peroxidase (POD) enzyme activity**

175 The previously described method with minor modifications was used to determine the PPO and POD

176 enzyme activities [21]. Fruit pericarp was blended with a high-speed blender and mixed with phosphate
177 buffer (0.2 M, pH 6.5). The supernatant was obtained by centrifuging the homogenate at $10,000 \times g$, $4 \pm$
178 $1 \text{ }^\circ\text{C}$ for 15 min. For PPO activity, the obtained supernatant (50 μL) was pipetted into a solution
179 containing 1 mL of catechol (0.1 M) and 1.95 mL of phosphate buffer (0.2 M) and then reacted in the
180 dark. The absorbance of the reaction solution at 410 nm was noted. For POD activity, the supernatant
181 (50 μL), 0.15 mL of guaiacol (0.01 M), 0.15 mL of H_2O_2 , and 2.66 mL of phosphate buffer (0.2 M) were
182 mixed to obtain the assay solution. After that, the assay solution was reacted in the dark for 25 min, and
183 then the absorbance at 470 nm was noted. The unit (U) of PPO and POD activity defined as a change of
184 0.01 in absorbance per minute. The enzyme activity expressed as U g^{-1} of fresh weight (FW).

185 2.11.3. Weight loss

186 The weight loss (%) was determined based on the ratio of the decreased weight of longan fruits (W_d) over
187 the original weight of longan fruits (W_o).

$$188 \text{Weight loss (\%)} = W_d/W_o \times 100 \quad (6)$$

189 2.12. Statistical analysis

190 Significant differences were analyzed with SPSS 19.0 software using one-way analysis of variance
191 (ANOVA). The statistical level of significance was $P < 0.05$. All experiments were performed five times.

192 3. Results and discussion

193 3.1. Characterization and GSH content determination of *S. cerevisiae*

194 The micrographs (400 \times) of *S. cerevisiae* cells were oval or spherical (Fig. 1a). The colony morphology
195 of *S. cerevisiae* was white, smooth, and sticky appearance (Fig. 1b). The growth curve of *S. cerevisiae*
196 was also presented in Fig. 1c. The results showed that the lag phase of *S. cerevisiae* was within 4 h, the
197 logarithmic phase was 6-18 h, and the stationary period was after 18 h. The GSH is a biologically active
198 tripeptide with excellent antioxidant capability, which is the secondary metabolites of *S. cerevisiae* [13].
199 The intracellular GSH production curve of *S. cerevisiae* was presented in Fig. 1d. In the logarithmic
200 growth phase (4-18 h), the intracellular GSH content rapidly accumulated, and the GSH content in the
201 18-h *S. cerevisiae* pellet was $497 \pm 16.8 \text{ mg/100g}$. The intracellular GSH content reached the highest
202 value ($521 \pm 26.0 \text{ mg/100g}$) after 24 hours. The intracellular GSH content in *S. cerevisiae* pellet did
203 not obviously change within 18-36 hours. This result indicated that the feasibility of *S. cerevisiae* to
204 produce GSH with sucrose as the main carbon source and the intracellular GSH content is related to the
205 growth of yeast. Since the yeast strains in the logarithmic growth phase were most active and with higher

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
61
62
63
64
65

206 intracellular GSH content, the 18-h seed solution was selected to prepare the yeast pellet as the
207 antioxidant agent.

208 **3.2. Scanning electron microscopy**

209 Fig. 2a presented that the cross-section of SA film was completely uniform and smooth. The cell
210 morphology figure of *S. cerevisiae* was observed in Fig. 2b, all the *S. cerevisiae* cells are oval, and their
211 average particle size was around 5~15 μm . The surface of the *S. cerevisiae* cell was smooth and flat.
212 Some intact *S. cerevisiae* cells could be obviously observed in the cross-section of the SA-SE film (Fig.
213 2c). This result indicated that the *S. cerevisiae* cell could be well alive and dispersed in the SA polymer.
214 As can be seen from Fig. 2d, the cross-section of 1.0%SU loaded SA-SE film was denser. This
215 phenomenon might be caused by intermolecular hydrogen bonds between sucrose and the natural
216 polysaccharide matrix [22]. Some irregular semi-crystalline structures was observed in cross-section of
217 the film when the incorporating SU concentration increased to 3.0% w/v (Fig. 2e). Similar structure was
218 observed in the cross-section of the SA-SE-6.0%SU film (Fig. 2f). Jiyuan Xu et al. [23] also found that
219 the cross-section of chitosan became rough with the addition of xylooligosaccharides. Moreover, the
220 average size of *S. cerevisiae* cells in all films was reduced. This is because the size of the microorganism
221 will be changed under different stresses. Similar founding was also reported by Soto-Reyes, N et al [24].
222 It is interesting to note that the *S. cerevisiae* cell density in the film was obviously increased when the
223 incorporating SU content was 3.0% w/v and 6.0% w/v. This could be due to the fact that sucrose as an
224 effective protectant with small molecular structures could easily replace the water molecules removed
225 during the drying process so that protect the yeast cell integrity [25]. This result revealed that the SA-
226 based bioactive films incorporated with SU and *S. cerevisiae* were fabricated successfully in this work,
227 which was expected to be a novel ideal carrier.

228 **3.3. FTIR studies**

229 The FTIR spectrum of SA, SA-SE, SA-SE-1.0%SU, SA-SE-3.0%SU and SA-SE-6.0%SU films were
230 presented in Fig. 3. For the neat SA film, the characteristic band at 1026, 1407, 1593, and 2929 cm^{-1}
231 were assigned to the elongation of C-O groups, the asymmetric stretching vibration of COO- groups, and
232 the CH stretching, respectively. The broad peak in the range of 3000–3667 cm^{-1} corresponding to the
233 hydroxyl groups (-OH). The weak bands at 948 indicated the presence of uronic acid functional group.
234 As observed in the spectrum of SA-SE, the incorporation of *S. cerevisiae* did not affect the FTIR spectra
235 of the pure SA films. This result could be due to the fact that there are no interactions between the yeast

1 236 and the carrier material. Orozco-Parra, Mejía, and Villa [26] also reported that the addition of
2 237 microorganisms did not affect the FTIR spectrum and crystalline structures of polysaccharide film. After
3
4 238 adding 1.0% SU to SA-SE film, the FTIR spectrum of the SA-SE-1.0%SU film shows a combination of
5
6 239 characteristics similar to that of the pure SA film, and the characteristic peaks corresponding to SA at
7
8 240 1593 cm^{-1} and 1407 cm^{-1} were slightly blue-shifted due to hydrogen bonding between SA and SU
9
10 241 molecules. The decrease in the intensity of peak at 1593 cm^{-1} and 1407 cm^{-1} of SA-SE-1.0%SU film can
11
12 242 be indicative of the greater degree of disorder in SA (Fig. 2d, 2e, and 2f) and could be related to the
13
14 243 plasticizing effect of the SU on the SA matrix. Pępczyńska et al. [27] also reported that the addition of
15
16 244 SU could decrease the intensity of the Amide I band of salmon gelatin film.
17
18 245 The peak in the SA-SE-3.0%SU film shift from 947 to 924 cm^{-1} was attributed to the Csingle bondO
19
20 246 bonding stretch associated with plasticization. Noticeably, The O-C stretching band of SA-SE-3.0%SU
21
22 247 film at 989-1027 cm^{-1} was double-peaked instead of the single peak in pure SA-SE film. The double
23
24 248 peaks occurred because the sucrose formed stable hydrogen bonds at CH bends and with both 'O' of C-
25
26 249 O-H and 'O' of anhydroglucose ring in SA molecules. Moreover, the characteristic peak corresponding
27
28 250 to the hydroxyl groups (-OH) in the SA film was broadened and strengthened, indicating more hydroxyl
29
30 251 groups (-OH) were formed in SA-SE-3.0%SU film due to the hydrogen bonding. This result indicated
31
32 252 that the 3% SU concentration could form stronger bonds with SA than 1.0% SU. Similar founding was
33
34 253 also reported by Pushpadass, Marx, and Hanna [28], who added SU into the starch-based film. No
35
36 254 significant difference was observed in the FTIR spectrum of SA-SE-3.0%SU and SA-SE-6.0%SU films.
37
38 255 Thus, it could be assumed that the addition of a certain content (3.0%~6.0%) of SU into SA film involved
39
40 256 the formation of hydrogen bonds between the SA and SU, replacing some of the original strong bonds in
41
42 257 the hydroxyl groups of pure SA.

43 258 **3.4. *S. cerevisiae* cell number and antioxidant capability in stored films**

44
45
46 259 The viability of the yeast is an essential parameter which had a significant influence on the property of
47
48 260 the yeast films. The cell number of the *S. cerevisiae* in the film during storage at 25 ± 1 °C and 75% RH
49
50 261 was presented in Fig. 4a. The initial cell number of *S. cerevisiae* in SA-SE film was 7.22 ± 0.01 Log
51
52 262 CFU/g and it presented a decreasing trend through the storage period. The viable *S. cerevisiae* number
53
54 263 in the SA-SE film was lower than that of SA-SE-SU films. This is because the presence of SU could
55
56 264 effectively maintain the *S. cerevisiae* cell integrity during the drying process (Fig. 2). It is interesting to
57
58 265 observe that the viability of *S. cerevisiae* in all SA-SE-SU films exhibited a trend of rising first and then
59
60
61
62
63
64
65

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
61
62
63
64
65

266 falling. This is because the yeast could rapidly proliferate by using sufficient nutrients in the film at the
267 early stage of storage. However, with the increase of storage time, the yeasts in the film will produce
268 large amounts of metabolites which are detrimental to cells and eventually cause cell death. The viable
269 cell numbers in SA-SE-1.0%SU film reached the peak value (9.32 ± 0.4 Log CFU/g) at 6 d, and then
270 gradually decreased. Remarkably, the SA-SE-3.0%SU and SA-SE-6.0%SU films could delay the arrival
271 of the peak value. The peak value of viable *S. cerevisiae* cells in SA-SE-3.0%SU, and SA-SE-6.0%SU
272 films was 10.20 ± 0.1 Log CFU/g and 9.41 ± 0.2 Log CFU/g, respectively. After storage for 24 days, the
273 viability of *S. cerevisiae* in SA-SE-SU films was obviously higher over that of SA-SE film. This result
274 indicated that the addition of SU could improve the viability of the yeast in the SA film. The viable cell
275 numbers in the SA-SE-3.0%SU and SA-SE-6.0%SU films were higher over that of SA-SE-1.0%SU film,
276 and no obvious difference was observed in SA-SE-3.0%SU and SA-SE-6.0%SU films. These results
277 indicated that the appreciated concentration of SU could promote the growth of *S. cerevisiae* in the stored
278 film. However, the excess SU content ($\sim 6\%$ w/v) could not furtherly increase the cell viability of *S.*
279 *cerevisiae*. This is because high concentrations of SU may alter the osmotic pressure in the polymer
280 matrix, thereby inhibiting the growth of yeast [29].

281 The antioxidant capability of the active film is essential for controlling the oxidant damage of foods. The
282 DPPH scavenging activity of the films during 24 days of storage period was presented in Fig. 4b. The
283 DPPH scavenging activity of pure SA film was $8.6 \pm 0.62\%$, and it hardly changed during the test period
284 (Fig. S2). This could be due to the presence of hydrogen bond donating functional groups in SA that
285 could also scavenge some free radicals [30]. Notably, the initial antioxidant property (0 d) of the SA film
286 was obviously increased to $76.0 \pm 1.1\%$ by the addition of *S. cerevisiae*. This is due to the fact that *S.*
287 *cerevisiae* itself could be used as an effective antioxidant [31]. Moreover, *S. cerevisiae* will produce some
288 metabolites with antioxidant ability, especially glutathione (Fig. 1d). A similar result was also reported
289 by Siying Li et al. [32], who observed that the exopolysaccharide produced by lactic acid bacteria could
290 improve the antioxidant property of the film. The initial DPPH scavenging activity (0 d) of the SA-SE
291 film was further strengthened with the addition of SU. No significant difference was observed in the
292 initial antioxidant property of SU-rich SA films. This is because the cell number and integrity of *S.*
293 *cerevisiae* were well-maintained during the film-forming process by the addition of SU (Fig. 1 and 3a)
294 so that maintain the biological capability of *S. cerevisiae* cell. The antioxidant property of SA-SE film
295 exhibited a decreasing trend during the storage period. It is interesting to observe that the antioxidant

1 296 property of sucrose-rich SA-SE film presented a trend of rising first and then falling during the storage
2 297 period. After storage for 24 days, the DPPH scavenging activity of SA-SE-1.0%SU films decreased to
3
4 298 $75.5 \pm 1.0\%$, while the value of SA-SE-3.0%SU and SA-SE-6.0%SU were $\sim 85.4\%$ and $\sim 81.6\%$,
5
6 299 respectively. The changing trend in antioxidant property of the films during the storage period was
7
8 300 consistent with that of the number of viable yeast cells in the films (Fig. 4a). Our previous research
9
10 301 demonstrated that the antioxidant property of the film during the storage was related to the content of the
11
12 302 active substance [4]. Thus, it could be concluded that the presence of SU maintains the number of *S.*
13
14 303 *cerevisiae* cells with satisfactory GSH production in the stored film, which offer a great potential for
15
16 304 maintaining the antioxidant property of the films during the storage period. While the mechanism
17
18 305 between *S. cerevisiae* cell numbers and the antioxidant properties of the films is unclear, the active film
19
20 306 with enhanced antioxidant properties is beneficial for reducing the food oxidative damage.

21 307 **3.5. *S. cerevisiae* cellular state in the films**

22
23 308 To check the cellular state of the *S. cerevisiae* in the film, the fluorescence of the stored films after storage
24
25 309 for 24 days was presented in Fig. S1. The green fluorescence was the predominant color in all figures,
26
27 310 indicating most *S. cerevisiae* could be alive in the developed film during the room temperature storage.
28
29 311 This could be in accord with the fact that *S. cerevisiae* is a model organism with higher resistance to
30
31 312 harsh environmental conditions. It was easy to observe that the red point numbers in Fig. S1a were
32
33 313 obviously higher than that in other figures, indicating that the viable *S. cerevisiae* cell concentration in
34
35 314 SA film was obviously lower than that in SA-SE-SU films. It is interesting to observe the cell density in
36
37 315 the SA-SE-3.0%SU was higher than that of other SA-SE-SU films. This result was consistent with plate
38
39 316 counts result (Fig. 4a). Oluwatosin, Tai, and Fagan-Endres [25] also found that certain content of sucrose
40
41 317 could effectively improve the final cell density of *Lactobacillus Plantarum* during the storage. These
42
43 318 results indicated that the addition of SU to SA film could improve the viability of the *S. cerevisiae* cell
44
45 319 during the storage period.

46 320 **3.6. Mechanical property**

47 321 The tensile strength of the SA film exhibited a little change after the addition of *S. cerevisiae* (Fig. 5a).
48
49 322 Similar result was also reported by Soukoulis et al. [33], who added the *Lactobacillus rhamnosus* into
50
51 323 the starch-protein film. The TS of the SA-SE film significantly decreased with the incorporation of SU
52
53 324 ($P < 0.05$), and it furtherly decreased with the gaining concentration of SU. When the incorporating SU
54
55 325 concentration exceeds 3.0% w/v, the TS of the film hardly decreased anymore. By contrast, the EB of
56
57
58
59
60
61
62
63
64
65

1 326 the film obviously increased with the incorporation of SU ($P < 0.05$), and it reached the highest value
2 327 ($37.6 \pm 2.4\%$) when the SU concentration was 3.0% w/v (Fig. 5a). No significant difference was observed
3
4 328 between SA-SE-3.0%SU film and SA-SE-6.0%SU films ($P > 0.05$). This result accord with the fact that
5
6 329 sucrose as a plasticizer could increase the polymer chain mobility so that it strength the extensibility of
7
8 330 the film [18]. Similar result was also studied by Fadini et al. [34]. This result showed that certain content
9
10 331 of SU could be used to improve the flexibility of SA-SE film, and the SA-SE-3%SU exhibited the optimal
11
12 332 extensibility.

13 333 3.7. Barrier property

14
15
16 334 Fig. 5b showed the water barrier property of the tested films. The addition of *S. cerevisiae* presented little
17
18 335 influence on the WVP of SA film. Li Siying et al. [24] also observed that the WVP of the cassava
19
20 336 starch/carboxymethylcellulose edible films altered little with the incorporation of lactic acid bacteria.
21
22 337 Notably, it significantly decreased with the accession of SU. However, no significant difference was
23
24 338 observed in the WVP value of the SA-SE-SU films ($P > 0.05$). Regarding the OP of the films (Fig. 5b),
25
26 339 it altered little with the incorporation of *S. cerevisiae*. The OP of the SA-SE film obviously decreased to
27
28 340 $19.1 \pm 0.51 \text{ cm}^3 \cdot \text{um} \cdot \text{m}^{-2} \cdot \text{d}^{-1} \cdot \text{Kpa}^{-1}$ when the incorporating SU was 1.0% w/v ($P < 0.05$), and it reached
29
30 341 the lowest value (14.5 ± 1.1) when the SU content was 3.0% w/v. However, the OP of the 6.0% w/v SU
31
32 342 loaded film increased. This result could be due to the addition of high concentration sucrose may decrease
33
34 343 interaction between the polymer chains so that it facilitates the migration of oxygen [35]. This result
35
36 344 indicated that the addition of the appropriate amount (1.0~3.0% w/v) of sucrose positively affected the
37
38 345 barrier property of SA-SE film.

39 346 3.8. Optical property

40
41
42 347 The visual appearance of all the developed films was presented in Fig. 5c. The pure SA film was light
43
44 348 and transparent, and it became light-yellow with the incorporation of *S. cerevisiae*. The appearance of
45
46 349 the SA-SE film was hardly altered by the addition of SU, and no significant difference was found in the
47
48 350 appearance of all SU-rich SA-SE films. Similar result was also observed in the transmittance of the films
49
50 351 (Fig. 5d). The transmission of all the film was $> 75.0\%$ at 600 nm, and the pure SA film presented the
51
52 352 best transparency. The transparency of the SA at 600 nm slightly decreased to $\sim 81.1\%$ by the
53
54 353 incorporation of *S. cerevisiae*, and it further slightly decreased to $\sim 78.2\%$ with the incorporation of SU.
55
56 354 All the films presented the similar UV-barrier property. This result indicated that all the fabricated films
57
58 355 in this work presented the satisfactory appearance and transmittance.
59
60
61
62
63
64
65

356 3.9. Application for longan fruits

357 3.9.1. Browning index (BI) and enzyme activity

358 The pericarp browning of longan is a decisive parameter for its edible quality and market value. PPO and
359 POD are the key enzymes that cause the pericarp browning of longan [36]. Therefore, the effects of
360 different treatments on the PPO and POD activities of longan fruits were checked in this work. As can
361 be seen in Fig. 6a, the PPO activity of all fruits increased firstly and decreased thereafter. The PPO
362 activity of the untreated and SA-treated fruits was rapidly increased within the first 4 d. The PPO activity
363 of SA-SE and SA-SE-SU treated fruits achieved a maximum value at 2 d, and it slowly decreased
364 thereafter. At the end of storage, the PPO activity of untreated fruits was 582 ± 25 U g⁻¹FW while the
365 value of SA-SE-3.0%SU was only 400 ± 35 U g⁻¹FW, and no significant difference was found in SA-
366 SE-SU films ($P > 0.05$). Similar findings were also presented in the POD activity of longan fruits (Fig.
367 6b). The untreated and SA-treated fruits exhibited a sharp increase in POD activity in the initial four days.
368 The peak value of POD activity in untreated and SA-treated fruits were 1075 ± 24 U g⁻¹FW and $976 \pm$
369 20 U g⁻¹FW, respectively. The POD in SA-SE and SA-SE-SU treated fruits were increased at the
370 beginning of storage, reached the maximum value at two days of storage, and then decreased gradually
371 from the second day to the eighth day of storage. At the end of storage, the POD activity in untreated and
372 SA treated fruits were 782 ± 45 U g⁻¹FW and 715 ± 40 U g⁻¹FW while the value of SA-SE-SU films was
373 between 456 and 475. The POD activity between SA-SE treated samples and SA-SE-SU treated samples
374 was not significant ($P > 0.05$). At the beginning of storage, the difference of BI values between untreated
375 fruits and treated fruits were not significant ($P > 0.05$) (Fig. 6c). The significantly lower BI value was
376 observed in the 4-d longan fruits treated with SA-SE and SA-SE-SU films over that of untreated 4-d
377 longan fruits ($P < 0.05$). The SA treatment can also reduce the browning of peel to some extent. This
378 could be in accord with the fact that the SA could act as a natural barrier to control the exchange of
379 oxygen so that reduce the browning on the fruit peel. After 12 d of storage, the fruits treated with *S.*
380 *cerevisiae*-encapsulated and sucrose-rich sodium alginate film-forming solutions presented the
381 significantly lower BI value than that of control fruits. By analyzing the results of PPO, POD, and BI,
382 the potential mechanism of inhibiting the pericarp browning of longan fruits by SA-SE-SU treatment
383 might be due to that SA-SE-SU film as a barrier could effectively reduce the oxygen exchange (Fig. 6b)
384 so that decrease the oxidative reaction rates of the longan fruits. On the other hand, the presence of GSH
385 produced by *S. cerevisiae* with proven enzyme inhibition ability in the active film could slowly release

1 386 into the surface of the fruits so that inhibit the PPO and POD activities (Fig. 6a and 6b). Similar founding
2 387 was also reported by Shi Shengyou et al. [37], who observed that chitosan/nano-silica coating treatment
3
4 388 could effectively reduce the browning index of longan fruits by inhibiting the enzyme activities.
5
6 389 Generally, an average BI values of more than three were deemed unsuitable for marketing [38]. Hence,
7
8 390 the obtained results revealed that the addition of *S. cerevisiae* and sucrose to SA could improve the market
9
10 391 value of longan fruits by reducing the browning index of the pericarp.

11
12 392 These results indicated that the SA-SE and SA-SE-SU treatments exhibited the satisfactory effect on
13
14 393 inhibiting the PPO, POD activities, and pericarp browning of longan fruits, and the SA-SE-3.0%SU
15
16 394 treatment presented the optimal effect.

17 395 **3.9.2. Weight loss**

18
19 396 The weight change is an important indicator to assess the quality of longan. Weight loss of the longan
20
21 397 fruit was presented in Fig. 6d. The weight loss of longan samples gradually increased with the increasing
22
23 398 storage time. Compared with untreated longan fruits, the weight loss of all treated fruits obviously
24
25 399 decreased. The SA-SE-SU treated fruits achieved the lowest weight loss value at the end of storage, and
26
27 400 no significant difference was found in SA-SE-SU treatments ($P > 0.05$). This decreased weight loss in
28
29 401 SA-SE-SU treated fruits could be due to that the enhanced water vapor barrier property of the SA-SE-
30
31 402 SU film could offer great potential in reducing the loss of moisture from fruits [39].

32 403 **3.9.3. Appearance of longan fruits**

33
34 404 The appearance of the stored-longan fruits was presented in Fig. 6e. At the beginning of storage, all
35
36 405 longan fruits are full and without damage, and the pericarp of the fruits are bright color and healthy. No
37
38 406 obvious difference was found in the pericarp of all longan fruits. By contrast, the pericarp of all fruits
39
40 407 exhibited browning to some extent after storage for eight days. These changes could be due to the
41
42 408 synthesis of brown pigments in the longan pericarp during enzymatic browning. The color of all untreated
43
44 409 fruits was changed to dark brown, indicating severe browning and poor quality. The browning degree of
45
46 410 SA-treated fruits was lower than that of untreated fruits, which is consistent with the result of BI (Fig.
47
48 411 6c). Remarkably, the SA-SE and SA-SE-SU treatments effectively reduce the browning degree in longan
49
50 412 pericarp, and SA-SE-SU films showed the optimal effect. This valid inhibition effect could be in accord
51
52 413 with the fact that the *S. cerevisiae*-loaded films with excellent antioxidant properties (Fig. 4b) effectively
53
54 414 inhibit the enzyme activity in longan pericarp (Fig. 6a and 6b) so that decrease the browning degree and
55
56 415 maintain good appearance quality. Li Siying et al. [32] also observed that the probiotic load films could
57
58
59
60
61
62
63
64
65

1 416 reduce the browning of fruits. This result indicated that SA-SE-SU films might be applied for controlling
2 417 the pericarp browning of longan fruits.
3

4 418 **4. Conclusion**

5
6 419 *S. cerevisiae*-encapsulated and sucrose-rich sodium alginate film were developed successfully in this
7
8 420 work. The SEM, FTIR, and mechanical properties revealed that SA-SE film with the incorporation of
9
10 421 SU has good flexibility due to the existence of hydrogen bonds. The addition of SU to the SA-SE film
11
12 422 improved the WVP and OP of the film. Remarkably, the viable cell number of *S. cerevisiae* in the SA-
13
14 423 SE active film was obviously improved with the addition of SU. This improvement of *S. cerevisiae*
15
16 424 viability positively affected maintaining the excellent antioxidant property of the stored film. A threshold
17
18 425 of 3.0% w/v was established as the effect of SU concentration was not significantly above that
19
20 426 concentration. Finally, the SA-SE-SU packaging material effectively improved the quality and pericarp
21
22 427 browning of longan fruits by inhibiting the PPO and POD enzyme activities. The SA-SE-SU film could
23
24 428 be used as a novel antioxidant packaging material for food preservation.
25

26 429 **Acknowledgments**

27
28
29 430 This work was supported by National Natural Science Foundation of China (32102080, 31801631,
30
31 431 31671844, 1601360061); Jiangsu Natural Science Foundation for Excellent Young Scholars
32
33 432 (BK20200103); China Postdoctoral Science Foundation (2020M683372); Natural Science Foundation
34
35 433 of Jiangsu Province (BK20160506, BK20180865); Postgraduate Research & Practice Innovation
36
37 434 Program of Jiangsu Province (KYCX21_3395); Jiangsu Association for Science and Technology youth
38
39 435 Talent Promotion Project; The Earmarked Fund for China Agriculture Research System, Young talent
40
41 436 Development Program of Jiangsu University.
42

43 437 **Supplementary data**

44
45 438 Some supplementary materials to this article were added.
46
47

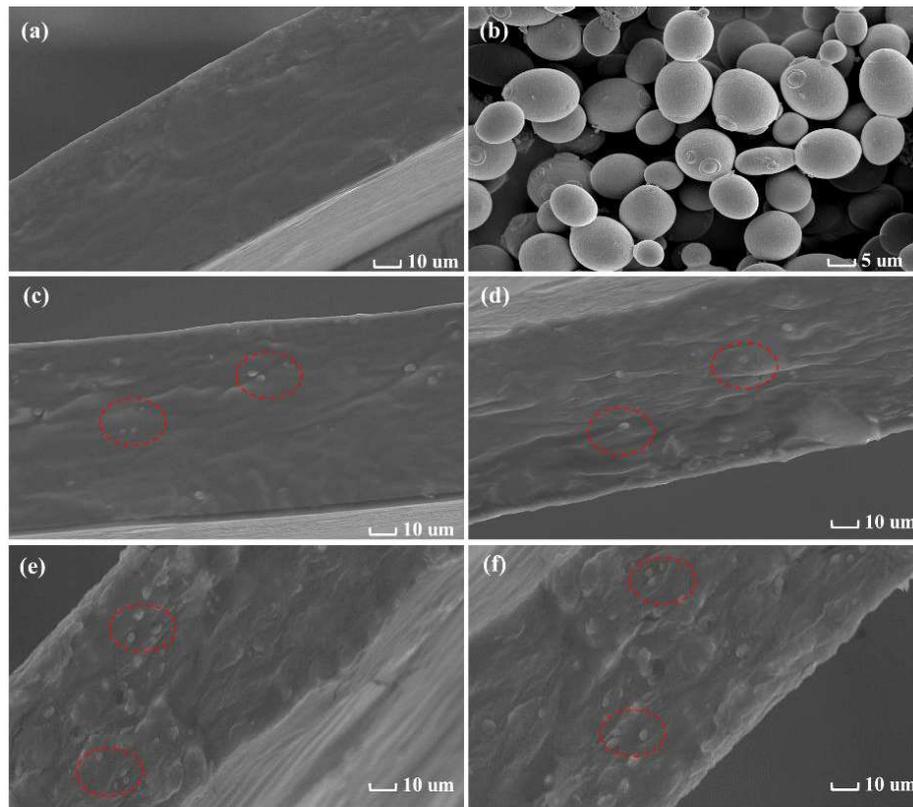
48 439 **References**

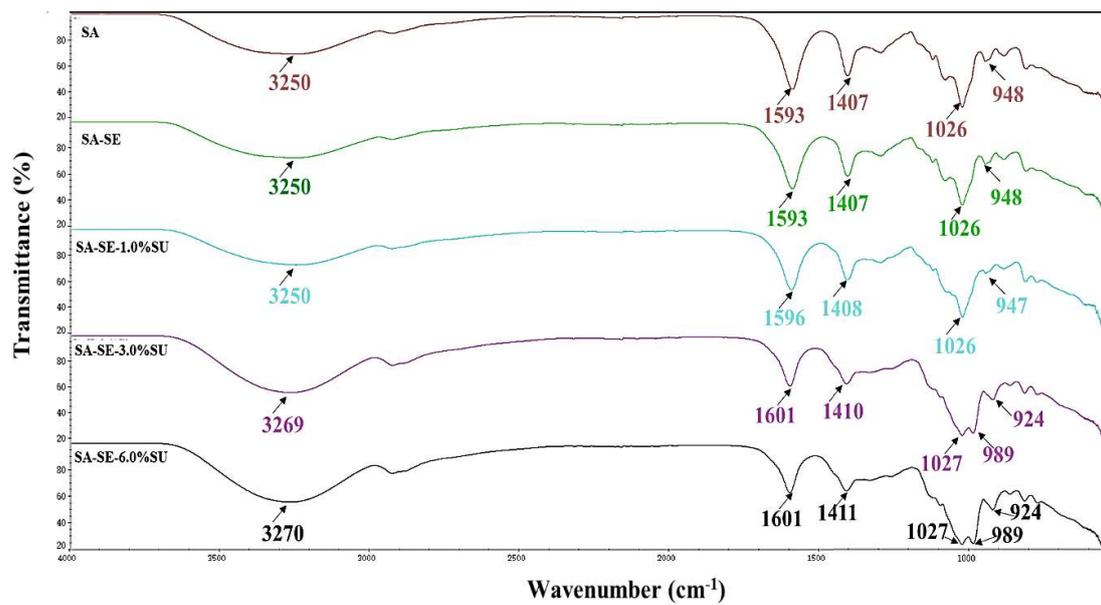
- 49
50 440 [1] Y. Lin, Y. Lin, H. Lin, S. Zhang, Y. Chen, J. Shi, LWT - Food Science and Technology, 60 (2015)
51
52 441 1122-1128.
53
54
55 442 [2] P.J.P. Espitia, R.A. Batista, H.M.C. Azeredo, C.G. Otoni, Food Research International, 90 (2016) 42-
56
57
58 443 52.
59
60
61
62
63
64
65

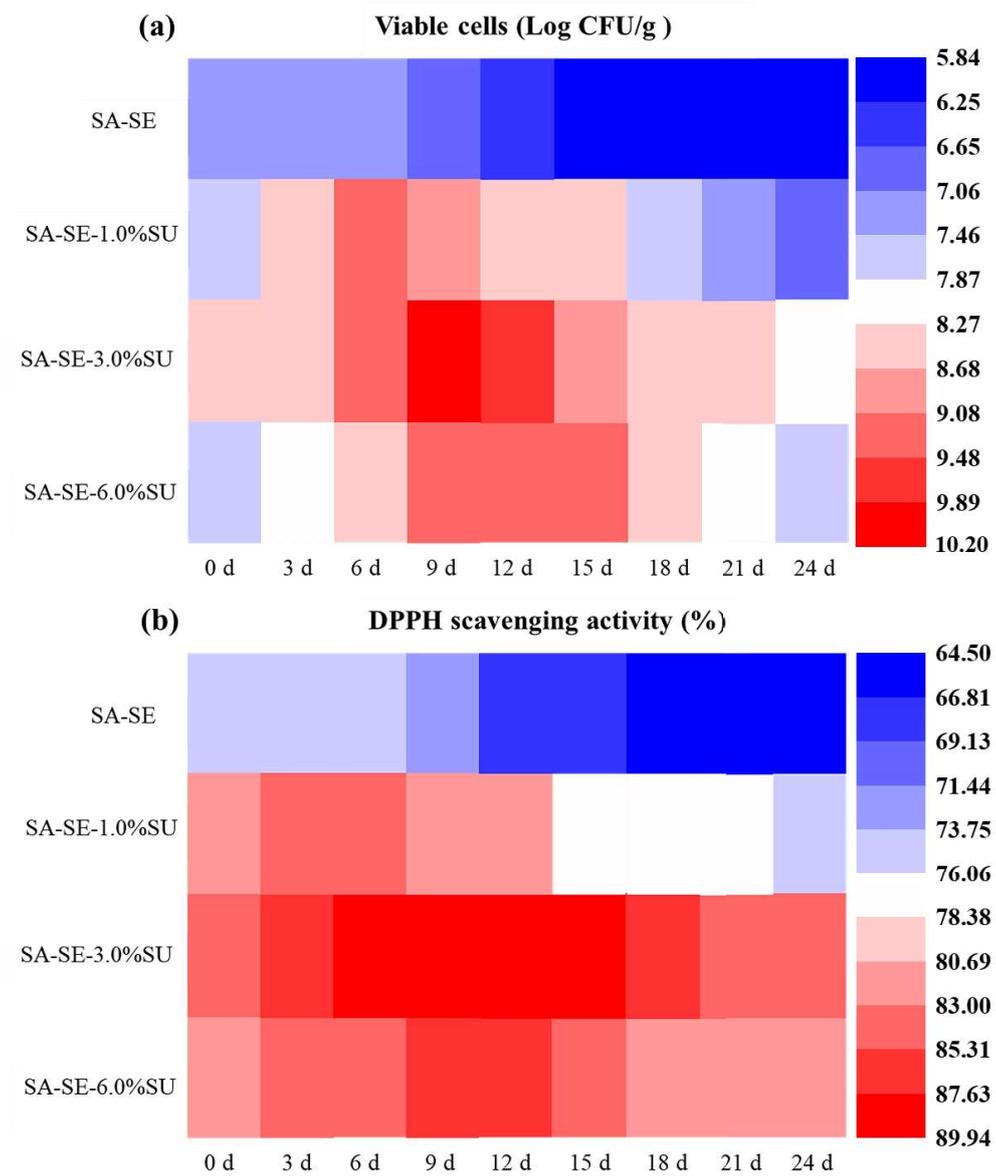
- 1 444 [3] M. Asgher, S.A. Qamar, M. Bilal, H.M.N. Iqbal, Food Research International, 137 (2020) 109625.
2
3 445 [4] Z. Yang, M. Li, X. Zhai, L. Zhao, H.E. Tahir, J. Shi, X. Zou, X. Huang, Z. Li, J. Xiao, International
4
5
6 446 Journal of Biological Macromolecules, 213 (2022) 145-154.
7
8 447 [5] C. Vilela, M. Kurek, Z. Hayouka, B. Röcker, S. Yildirim, M.D.C. Antunes, J. Nilsen-Nygaard, M.K.
9
10 448 Pettersen, C.S.R. Freire, Trends in Food Science & Technology, 80 (2018) 212-222.
11
12 449 [6] N. Soto-Reyes, M. Dávila-Rodríguez, A.C. Lorenzo-Leal, F. Reyes-Jurado, E. Mani-López, R.
13
14 450 Hernández-Figueroa, J.I. Morales-Camacho, A. López-Malo, Chapter 8 - Prospects for food applications
15
16 451 of products from microorganisms, in: B. Prakash (Ed.) Research and Technological Advances in Food
17
18 452 Science, Academic Press, 2022, pp. 195-229.
19
20 453 [7] P.J.P. Espitia, R.A. Batista, H.M.C. Azeredo, C.G. Otoni, Food Research International, 90 (2016) 42-
21
22 454 52.
23
24 455 [8] L. Cassani, A. Gomez-Zavaglia, J. Simal-Gandara, Food Research International, 129 (2020) 108852.
25
26 456 [9] B.D. Singu, P.R. Bhushette, U.S. Annapure, Food Bioscience, 36 (2020) 100668.
27
28 457 [10] M.Z.A. Chan, S.-Q. Liu, Current Opinion in Food Science, 43 (2022) 216-224.
29
30 458 [11] V.M. De, J. Schrezenmeir, Adv Biochem Eng Biotechnol, 22 (2002) 357-360.
31
32 459 [12] H. Borase, M.K. Dwivedi, R. Krishnamurthy, S. Patil, Chapter 30 - Probiotics: health safety
33
34 460 considerations, in: M.K. Dwivedi, N. Amaran, A. Sankaranarayanan, E.H. Kemp (Eds.) Probiotics in
35
36 461 the Prevention and Management of Human Diseases, Academic Press, 2022, pp. 449-463.
37
38 462 [13] J. Wang, K. Wang, Y. Wang, S. Lin, P. Zhao, G. Jones, Food Chemistry, 161 (2014) 361-366.
39
40 463 [14] S. Supapvanich, L. Samransuk, T. Somanusorn, N. Mesa, Effect of glutathione on browning of fresh-
41
42 464 cut wax apple fruit during refrigerated storage, 2014.
43
44 465 [15] W.L. Marques, V. Raghavendran, B.U. Stambuk, A.K. Gombert, FEMS Yeast Research, 16 (2016)
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
61
62
63
64
65

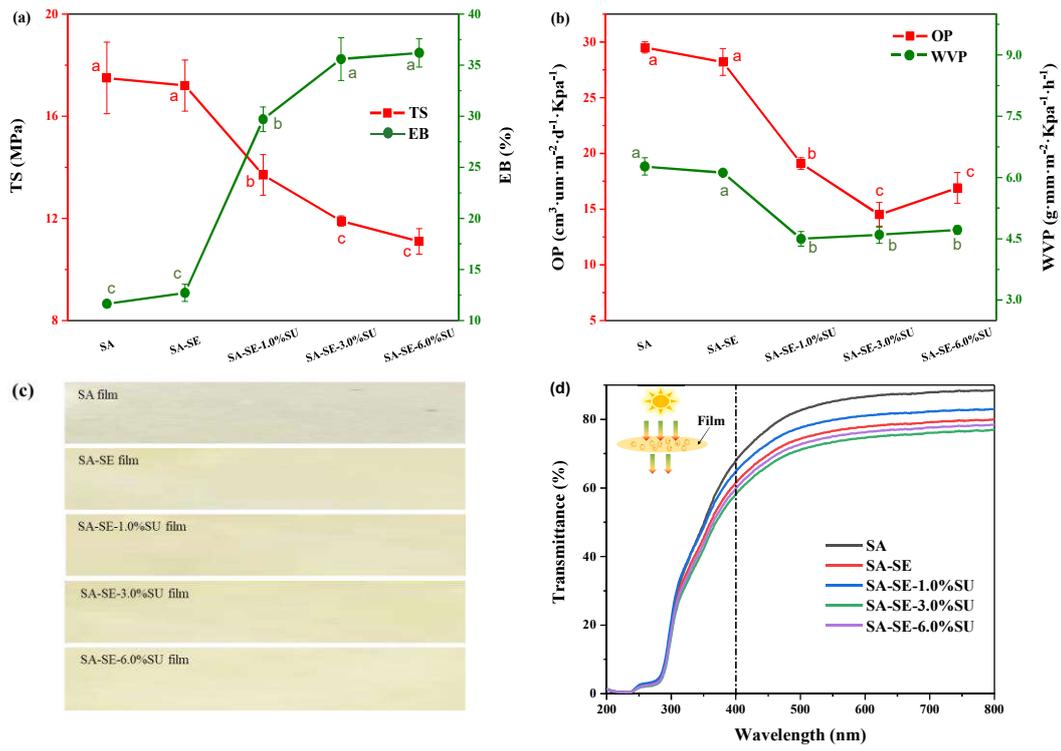
- 1 466 fov107.
2
3 467 [16] P. Suryabhan, K. Lohith, K.A. Anu-Appaiah, LWT, 107 (2019) 243-248.
4
5
6 468 [17] F. Badotti, M. Dário, S.L. Alves, M. Cordioli, L.C. Miletto, P. Araujo, B.U. Stambuk, Microbial Cell
7
8 469 Factories, 7 (2008) 4.
9
10
11 470 [18] P. Veiga-Santos, L.M. Oliveira, M.P. Cereda, A.R.P. Scamparini, Food Chemistry, 103 (2007) 255-
12
13 471 262.
14
15
16 472 [19] Z. Yang, X. Zhai, X. Zou, J. Shi, X. Huang, Z. Li, Y. Gong, M. Holmes, M. Povey, J. Xiao, Food
17
18 473 Chemistry, 336 (2021) 127634.
19
20
21 474 [20] X. Zhai, Z. Li, J. Zhang, J. Shi, M. Povey, Journal of Agricultural and Food Chemistry, 66 (2018).
22
23
24 475 [21] Z. Yang, X. Zou, Z. Li, X. Huang, H.E. Tahir, Food and Bioprocess Technology, 12 (2019).
25
26
27 476 [22] Y. Yang, H. Zhou, Y. Xiao, L. Feng, L. Yang, W. Mu, X. Peng, L. Bao, J. Wang, Carbohydrate
28
29 477 Polymers, 255 (2021) 117363.
30
31
32 478 [23] A. Jx, B. Rx, A. Ty, C. Rsa, Food Chemistry, 298.
33
34
35 479 [24] S. Li, Y. Ma, T. Ji, D.E. Sameen, Y. Liu, Carbohydrate Polymers, 248 (2020) 116805.
36
37
38 480 [25] S.O. Oluwatosin, S.L. Tai, M.A. Fagan-Endres, Biotechnology Reports, 33 (2022) e00696.
39
40
41 481 [26] J. Orozco-Parra, C.M. Mejía, C.C. Villa, Food Hydrocolloids, 104 (2020) 105754.
42
43
44 482 [27] M. Pępczyńska, P. Díaz-Calderón, F. Quero, S. Matiacevich, C. Char, J. Enrione, Food
45
46 483 Hydrocolloids, 97 (2019) 105207.
47
48
49 484 [28] H.A. Pushpadass, D.B. Marx, M.A. Hanna, Starch - Strke, 60 (2010) 527-538.
50
51
52 485 [29] T. Barnett, T. Atwood, B. Blasdel, C. Boker, T. Anderson, (2016).
53
54
55 486 [30] Y.B. Bhagath, M. Kola, A.M. Beulah, A. Rammohan, G.V. Zyryanov, Journal of Food Engineering,
56
57 487 301 (2021) 110566.
58
59
60
61
62
63
64
65

1 488 [31] Y. Wang, Y. Wu, Y. Wang, H. Xu, X. Mei, D. Yu, Y. Wang, W. Li, *Nutrients*, 9 (2017) 521.
2
3 489 [32] S. Li, Y. Ma, T. Ji, D.E. Sameen, S. Ahmed, W. Qin, J. Dai, S. Li, Y. Liu, *Carbohydrate Polymers*,
4
5
6 490 248 (2020) 116805.
7
8 491 [33] C. Soukoulis, P. Singh, W. Macnaughtan, C. Parmenter, I.D. Fisk, *Food Hydrocolloids*, 52 (2016)
9
10 492 876-887.
11
12 493 [34] A.L. Fadini, F.S. Rocha, I.D. Alvim, M.S. Sadahira, M.B. Queiroz, R.M.V. Alves, L.B. Silva, *Food*
13
14 494 *Hydrocolloids*, 30 (2013) 625-631.
15
16 495 [35] J. Xu, R. Xia, T. Yuan, R. Sun, *Food Chemistry*, 298 (2019) 125041.
17
18 496 [36] Y.M. Jiang, *Food Chemistry*, 66 (1999) 75-79.
19
20 497 [37] S. Shi, W. Wang, L. Liu, S. Wu, Y. Wei, W. Li, *Journal of Food Engineering*, 118 (2013) 125-131.
21
22 498 [38] S. Intarasit, B. Faiyue, J. Uthaibutra, K. Saengnil, *Scientia Horticulturae*, 293 (2022) 110678.
23
24 499 [39] Z. Yang, X. Zhai, C. Zhang, J. Shi, X. Huang, Z. Li, X. Zou, Y. Gong, M. Holmes, M. Povey, J.
25
26 500 Xiao, *Food Hydrocolloids*, 123 (2022) 107187.
27
28 501
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
61
62
63
64
65









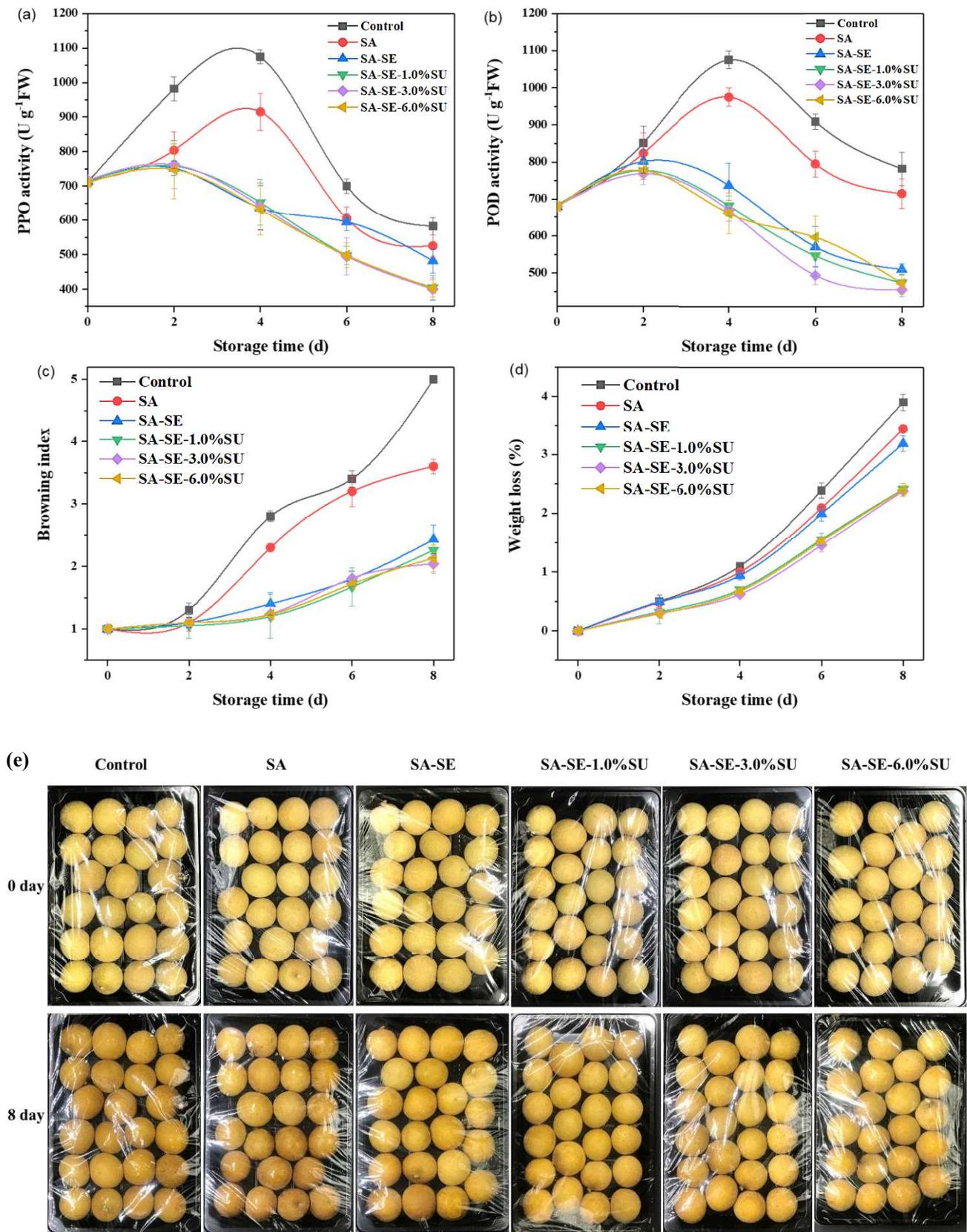


Figure. 1. The micrographs (400×) (a), colony morphology (b), growth curve (c) and intracellular GSH production curve (d) of *S. cerevisiae*.

Figure. 2. The scanning electron microscopy (SEM) microscopies of the cross section of SA (a), SA-SE (c), SA-SE-1.0%SU (d), SA-SE-3.0%SU (e) and SA-SE-6.0%SU films (f) and *S. cerevisiae* cell (b).

Figure. 3. FTIR spectrum of SA, SA-SE, SA-SE-1.0%SU, SA-SE-3.0%SU and SA-SE-6.0%SU films.

Figure. 4. The viable cells number of *S. cerevisiae* in the films (a) and the antioxidant capability of the films during the storage period (b).

Figure. 5. The mechanical property (TS and EB) (a), barrier property (WVP and OP) (b), visual appearance (c) and transmittance (d) of the films.

Figure. 6. The PPO (a), POD (b) activities, browning index (c), weight loss (d) and appearance quality (e) of stored-longan fruits treated with control, SA, SA-SE, SA-SE-1.0%SU, SA-SE-3.0%SU, and SA-SE-6.0%SU.

Conflict of Interest

Xiaobo Zou declares that he has no conflict of interest. Zhikun Yang declares that he has no conflict of interest. Jiyong Shi declares that he has no conflict of interest. Xiaodong Zhai declares that he has no conflict of interest. Xiaowei Huang declares that she has no conflict of interest. Zhihua Li declares that he has no conflict of interest. Yunyun Gong declares that she has no conflict of interest. Melvin Holmes declares that he has no conflict of interest. Megan Povey declares that she has no conflict of interest. Jianbo Xiao declares that he has no conflict of interest. Mingrui Li declares that he has no conflict of interest.



Click here to access/download
Supplementary Material
Supplementary material.docx

