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1 The magmatic evolution and the regional context of the 1835
2 AD Osorno volcano products (41°06'S, Southern Chile)

3

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23 ABSTRACT

24 Osorno volcano (41°06'S, 72°20'W) is a composite stratovolcano of the Central Southern
25 Volcanic Zone of the Chilean Andes. It is the southernmost member of a NE-SW trending
26 alignment of volcanic edifices including La Picada and Puntigudo volcanoes and the Cordón
27 Cenizos chain. According to contemporary descriptions recorded by Charles Darwin in 1835,
28 two eruptive events occurred: the first during January-February, and the second during
29 November-December 1835 and January 1836. The volcano erupted basaltic andesite lavas
30 and tephra fall deposits (52.4 to 52.9 SiO₂ wt. %) which contain phenocrysts of olivine,
31 plagioclase, clinopyroxene, and spinel. The compositions of these phenocryst phases,
32 together with those of olivine-hosted melt inclusions, allowed us to constrain intensive
33 parameters for the pre-eruptive magmas. These varied from 1060 to 1140 °C, with an oxygen
34 fugacity buffer of $\sim\Delta\text{QFM} + 1.1$, dissolved water concentrations of up to 5.6 wt. % (average
35 of ~ 4.2 wt. %) and **maximum** pressures equivalent to ~ 7 km depth. Textural relations, such
36 as crystal accumulations and clots, zoning in crystals and other indications of disequilibrium,
37 lead us to infer the involvement of a crystal mush, rich in individual crystals and clots of
38 crystals, which underwent a degree of disaggregation and entrainment into the transiting
39 magma prior to eruption. Comparison of trace element abundances, including rare earth
40 elements, fluid-mobile elements, and relatively fluid-immobile elements, combined with
41 $^{87}\text{Sr}/^{86}\text{Sr}$ and $^{143}\text{Nd}/^{144}\text{Nd}$ isotope ratios, allow us to consider variations in slab-derived fluid
42 input and the minor role of crustal contamination on the Osorno eruptive products and those
43 from neighbouring volcanic systems. Our results suggest both a greater contribution from
44 slab-derived fluid and a higher degree of partial melting in the systems supplying

45 stratovolcanoes (Osorno, Calbuco, and La Picada) relative to those supplying small eruptive
46 centres built over the major regional Liquiñe-Ofqui Fault Zone.

47 **INTRODUCTION**

48 In subduction zones, geochemical features of relatively evolved volcanic products across-arc
49 may be preserved (Watt et al., 2013). This allows to contrast whole-rock chemical
50 compositions of products from different volcanoes to understand mantle- and slab-related
51 contributions. Specifically in Southern Andean volcanoes, several studies (e.g., Hickey-
52 Vargas et al., 1989; 2002; López-Escobar et al., 1992; 1995a; Watt et al., 2013; Jacques et
53 al., 2014; Morgado et al., 2015; Rawson et al., 2016; McGee et al., 2017) have reported
54 chemical variations of erupted material over small spatial scales and have attributed crustal
55 and subcrustal processes as causes for the observed variations in magma composition.
56 However, studies combining major and trace elements, isotopic ratios, and mixing
57 modellings of products from different volcanoes at the same latitude, are uncommon.

58 Besides whole-rock chemistry, recent articles (e.g., Ruth et al., 2018; Weller and Stern, 2018;
59 Morgado et al., 2019a; Couperthwaite et al., 2020; Tassara et al., 2020) have also considered
60 melt inclusion composition, mineral chemistry, and diffusion chronometry in volcanic rocks,
61 which have placed observed chemical variabilities in a geological context, permitting events
62 occurring during the magmatic evolution to be identified. Particularly, melt inclusion
63 compositions, which have been used to constrain pressure of the magmatic storage and water
64 dissolved in melt, could undergo significant reequilibration after recrystallisation in the rims
65 and volatile loss, leading to misinterpretations of magmatic processes.

66 In Southern Chile, from 1575 to 1835, Osorno volcano had an eruptive frequency of one
67 event per ~25 years, but its last eruption occurred in 1835 (Darwin, 1839a; Moraleda y
68 Montero, 1888; Martin, 1901; O'Compley, 1936; Petit-Breuilh, 1999). Despite this relatively
69 long quiescence, Osorno has shown an elevation of seismicity base level in recent years,
70 interpreted as a sign of volcanic unrest (SERNAGEOMIN–OVDAS 2017a; 2018a, b, c;
71 2019; 2022). Here, we report our findings on the last eruption cycle (in 1835 AD) of Osorno
72 volcano (Central Southern Volcanic Zone of the Southern Andes of Chile; Fig. 1); eruptions
73 that lack detailed studies of the magmatic conditions that preceded volcanic activity. We
74 describe crustal magmatic processes and assess the pre-eruptive physical magmatic
75 conditions via mineral chemistry and coupling melt inclusion compositional profiles and
76 diffusion chronometry. Finally, to place the Osorno volcano into a wider context with
77 neighbouring volcanic activity, we compare and contrast the nature of the chemical
78 enrichment/depletion of the erupted products across-arc from volcanoes at roughly the same
79 latitude (41°06'S) from W to E, i.e., from Calbuco, Osorno, La Picada and La Viguería
80 volcanoes. Comparison of volcanoes at the same latitude decreases the uncertainties related
81 to the mantle source and minimises the potential effects of latitudinal variability in the supply
82 of material associated with trench sediments (Watt et al., 2013) or other geodynamic
83 anomalies (e.g., subduction of oceanic fracture zones; Sellés et al., 2004).

84 **GEOLOGICAL SETTING**

85 **Southern Volcanic Zone**

86 The Southern Volcanic Zone (SVZ; Fig. 1) of the Andes is a volcanic chain produced by the
87 subduction of the oceanic Nazca plate below the South American continental plate (e.g.,

88 Jarrard 1986; Dewey and Lamb 1992; DeMets et al., 1994; Angermann et al., 1999). The
89 SVZ comprises over 70 Pleistocene and Holocene stratovolcanoes and several large volcanic
90 fields (Stern et al., 2007). The whole volcanic segment is ~1400 km long (33° to 46°S) and
91 is accompanied by the NS regional Liquiñe-Ofqui Fault Zone (LOFZ), which is an intra-arc
92 fault that runs along-strike, just like the much of the arc volcanism between ~38° and ~46°S
93 (Cembrano and Lara, 2009; Sánchez et al., 2013; Pérez-Flores et al., 2016). The SVZ has
94 been subdivided into four segments on the basis of tectonic setting, geochemical and
95 petrologic considerations (Tormey et al., 1991; López-Escobar et al., 1995a; Stern, 2004;
96 Stern et al., 2007). These subdivisions are shown in Fig. 1 and comprise the Northern (NSVZ,
97 33.3 to 34.4°S), Transitional (TSVZ, 34.4 to 37°S), Central (CSVZ, 37 to 42°S), and Southern
98 (SSVZ, 42 to 46°S) zones.

99 The present study focuses on volcanic centres from the CSVZ, the most active region of the
100 SVZ (and the whole Andes; Stern et al., 2007), in particular on Osorno volcano considering
101 a geochemical comparison with products from other volcanoes at the latitude of ~41°S: La
102 Picada, and Calbuco stratovolcanoes and La Viguería cone. Osorno volcano is the
103 westernmost member of a NW alignment together with La Picada, Puntigudo, and Cordon
104 Cenizos (Moreno et al., 2010; Vander Auwera et al., 2019; Fig. 2). The nearby Calbuco
105 volcano is not located directly upon any major regional structure (Sellés and Moreno, 2011).
106 Conversely, La Viguería cone (41°22'S) belongs to Cayutué-La Viguería small eruptive
107 centres built over the LOFZ master fault. According to the conceptual model of Cembrano
108 and Lara (2009), NW alignments are related to tension cracks at depth, which may reach the
109 surface and then provide pathways for magma ascent and, therefore, allow the building of
110 either a stratovolcano or an elongated cluster of minor eruptive centres, depending on the

111 balance between strain rate and magma input (cf. Takada, 1994). Magmas erupted at small
112 eruptive centres can be shown to have ascended rapidly (e.g. Morgado et al., 2017), and with
113 no pronounced interaction with the crust (e.g., McGee and Smith, 2016; Hickey-Vargas et
114 al., 2016; Smith and Németh, 2017). This style of volcanism is usually related to extensional
115 domains (e.g., Nakamura, 1977; Takada, 1994; Piochi et al., 2005).

116 **Osorno volcano: general features**

117 Osorno is an active stratovolcano of the Central Southern Volcanic Zone of the Chilean
118 Andes. Its maximum height is 2661 meters above sea level (m.a.s.l.) and its base level is
119 between 50 and 250 m.a.s.l. with an estimated volume of 130 to 160 km³, covering an area
120 of 250 km² (Moreno et al., 2010; Völker et al., 2011). It is located at 41°06'S, 72°20'W, and
121 lies ~13 km west of the main trace of the LOFZ (Fig. 2). Osorno volcano is the southernmost
122 member of an NE-SW alignment that includes La Picada volcano and the Puntigudo–
123 Cordón Cenizos volcanic chain (Fig. 2).

124 Osorno volcano has been active since the Middle Pleistocene (ca. 200 ka; Moreno et al.,
125 2010). Its postglacial activity has been reported as summit crater eruptions, fissure-fed lava
126 flows, parasitic cones and dacitic domes on the flanks (Moreno et al., 2010). The basement
127 beneath Osorno volcano comprises Cretaceous granitoids (ca. 70 Ma; Moreno et al., 2010)
128 and Miocene plutonic rocks from the Northern Patagonian Batholith (López-Escobar et al.,
129 1992; Adriasola et al., 2006). Vander Auwera et al (2019) found Miocene-age (ca. 9.6 Ma)
130 gabbroic xenolith blocks within the volcanic products of the adjacent La Picada volcano and
131 Hickey-Vargas et al. (1995) described gabbroic and granulitic xenoliths recovered from

132 volcanic material of the 1961 eruption of the neighbouring Calbuco volcano. All the xenoliths
133 from previous studies have been interpreted as being derived from the local basement.

134 The products of Osorno volcano vary from basalts to dacites (~50-70% SiO₂; e.g., Tagiri et
135 al., 1993; Moreno et al., 2010; Bechon et al., 2022) and have been classified into four units
136 (1, 2, 3, and 4) depending on the age of the deposits. In the unit Osorno 4, several historical
137 eruptions during the years 1575, 1640, 1644(?), 1719, 1737, 1765, 1778 to 1779, 1790, and
138 1834 to 1835 have been recognised and briefly described, whereas activity since 1834 has
139 been confined to fumarolic episodes in 1852, 1882 to 1883, 1911 to 1915, 1963, 1969, 1985,
140 and 1993 (Petit-Breuilh, 1999).

141 **The 1835 AD Osorno eruption: History**

142 Charles Darwin (1839b; 1840) described two eruption stages of Osorno volcano during 1835.
143 Although he recognised release of “smoke” in November 1834 (Darwin, 1839a), the first
144 stage of the 1835 Osorno eruption started on January 19th with moderate to vigorous
145 strombolian activity (VEI 2; according to the Pyle (2000) catalogue; Lara et al., 2012). The
146 volcano was still erupting in this manner on February 18th, two days before the large-
147 magnitude earthquake (M_w 8.0-8.3; Watt et al., 2009) of Concepción, Southern Chile
148 (Darwin, 1840). During the shock period, Darwin (1839b) described the emission of a dark
149 blue “smoke” column from the summit, following the eruption cessation, ending the first
150 period of eruption. Basaltic lavas are associated with this first period (Moreno et al., 2010),
151 which were emitted from an array of eruptive fissures and small pyroclastic cones (see Fig.
152 3). According to Darwin (1840), the second stage of eruptive activity, which started on
153 November 11th, resumed from the vents of the first stage (Lara et al., 2012). During December

154 5th, a fissure eruption occurred, associated with lava flows (Gillis, 1855) and lahars, which
155 burnt trees (Petit-Breuilh, 1999). The volcano was still erupting on the January 20th 1836
156 (Gillis, 1855).

157 **SAMPLING AND ANALYTICAL METHODS**

158 **Sample collection**

159 We collected samples from three volcanic systems: Calbuco, Osorno, and La Viguera.

160 *Calbuco volcano products.*

161 Calbuco samples correspond to four lapilli-sized pyroclasts from fall deposits and five bombs
162 from pyroclastic density current deposits from the 2015 eruption. The locations from which
163 they were taken, detailed quantification of the main petrographic features, and whole-rock
164 chemistry (major, minor, and trace elements) of Calbuco samples are available in Morgado
165 et al. (2019a).

166 *1835 AD Osorno volcano products.*

167 The collected samples from the 1835 AD Osorno eruption correspond to ten from lava flows
168 erupted from the main crater (reaching lengths from 10 to 14 km long) from five locations
169 (Table 1) and lapilli fall deposits erupted from parasitic cones in one location (Fig. 3; Table
170 1). Examination and quantification of the main petrographic features (e.g. crystallinity,
171 vesicularity, and modal percentages of phenocryst phases) from the collected samples was
172 done by optical microscope and image processing of pseudo-coloured images via the
173 freeware JmicroVision, respectively (Roduit, 2005).

174 *La Viguera cone.*

175 The collected samples from La Viguería cone correspond to one from a lava flow (~1.2 km
176 length) and one from fall deposits (locations in Table 1). We performed the quantification of
177 the main petrographic features using the same methods described for Osorno samples.

178 **Whole-rock major elements**

179 Whole-rock major elements compositions of lapilli deposits of one location (Os-164), three
180 lava samples from three locations from the 1835 AD Osorno eruption (Os-144, Os-147, and
181 Os-154), and two samples (one lava and one bomb) from La Viguería monogenetic cone
182 (Fig. 2) were determined on fused glass discs by XRF (major elements) at ACME labs (now
183 Bureau Veritas Minerals, Canada) using OREAS-184 and SY-4 standards for all the major
184 elements except for FeO, for which CCU-1D, and OREAS-700 standards were used for FeO.
185 Accuracy for major elements in all standards is better than 3% (relative) except for TiO₂ in
186 SY-4 which is 4.5% (relative) and P₂O₅ in OREAS-184 and SY-4, which were 14% and 17%
187 (relative), respectively. The detection limits are reported in Table 2.

188 **Whole-rock trace element abundances**

189 The major and trace element chemistry of the 1835 Osorno volcano products and La Viguería
190 cone was obtained from bulk-rock powders. All trace element measurements were performed
191 on an ICP-MS Agilent 7500a at The Open University (Milton Keynes, UK). 0.1 g of finely
192 powdered sample was weighed out and 1 mL of Romil UpA HNO₃ and 4 mL of Romil UpA
193 HF were added. The samples were heated in sealed Teflon vials to 130 °C and agitated in an
194 ultrasonic bath for 20 minutes every 12 hours until they had completely dissolved. The
195 samples were then dried down to incipient dryness and the resulting gel brought up in 2 mL
196 of Romil UpA HNO₃ and agitated in an ultrasonic bath for 20 minutes then returned to the

197 hot plate overnight. After drying down for a second time the samples were redissolved in
198 concentrated UpA Romil HCl and returned to the hot plate overnight, before drying down a
199 third time. The residue was converted to a nitrate matrix using 1 mL of concentrated UpA
200 HNO₃, dried down and re-dissolved in 100 mL of 2 % UpA HNO₃ to give a final solution
201 with a thousandfold dilution, ready for mass spectrometry.

202 All element concentrations were corrected for blank effects (which resulted in corrections of
203 less than 1 % for all elements) and subsequently corrected offline for instrumental drift.
204 Where more than one analysis of a sample was undertaken during the same run the mean
205 value was taken prior to offline corrections were performed. Detection limits were estimated
206 to be five times the blank concentration for each element analysed and all data points with
207 concentrations lower than the detection limits were discarded. The precision was better than
208 8% (relative, 2 σ) and accuracy in the trace elements of standards JB-2 (n = 5), BHVO-2 (n =
209 2) and AGV-1 (n = 2) is better than 10% (relative) except in Ni (21% relative, in AGV-1)
210 and Pb (13% relative, in BHVO-2). The uncertainties (2 σ) of measurements are reported in
211 Table 2.

212 **Strontium and neodymium isotope measurements**

213 Strontium and neodymium isotopes were measured in volcanic products from Osorno
214 volcano, La Viguería cone (see section 4.1), and six samples from 5 locations on the nearby
215 Calbuco stratovolcano (Cal-149Ta, Cal-157, Cal-158, Cal-159, Cal-160; Morgado et al.,
216 2019a). All ⁸⁷Sr/⁸⁶Sr and ¹⁴³Nd/¹⁴⁴Nd analyses in this study were conducted on a Thermo
217 Scientific Triton Thermal Ionization Mass Spectrometer (TIMS) fitted with 10¹¹ Ohm
218 amplifiers at the University of Leeds. Analyses were performed on ca. 100 mg of bulk rock
219 powders digested using a combination of Romil-UpA HF, HNO₃ and HCl, using the same

220 method as described for the trace element determinations. Once in solution, samples were
221 centrifuged prior to loading onto Eichrom Sr SpecTM columns with a bed volume of
222 approximately 100 μ L. The REE fraction was collected before elution of the Sr fractions into
223 different vials. Prior to loading onto W filaments with a TaCl₅ activator, the Sr fraction was
224 passed through the Sr spec column a second time and dried with two drops of dilute HClO₄
225 to remove any organic contaminants derived from the resin. The REE fraction was passed
226 first through a TRU SpecTM column (ca. 100 μ L bed volume) and finally a LN SpecTM column
227 (bed volume ca. 2 mL) before loading onto Re filaments. The USGS reference material
228 BHVO-2 was processed alongside the samples (Table 3), together with two total procedural
229 blanks spiked with highly enriched ⁸⁴Sr, and ¹⁵⁰Nd solutions. Mean total procedural blanks
230 for Sr and Nd were 326 pg and 153 pg, respectively. Accuracy of the radiogenic isotope
231 analyses was assessed by analysing standard solutions of known composition alongside the
232 unknowns – NBS987 for Sr, La Jolla for Nd, and the reference material BHVO-2 for Sr and
233 Nd, which yielded $^{87}\text{Sr}/^{86}\text{Sr} = 0.710241 \pm 0.000006$ for NBS987, $^{143}\text{Nd}/^{144}\text{Nd} = 0.511842 \pm$
234 0.000002 for La Jolla, $^{87}\text{Sr}/^{86}\text{Sr} = 0.703461 \pm 0.000007$ and $^{143}\text{Nd}/^{144}\text{Nd} = 0.512977 \pm$
235 0.000004 for BHVO-2, in agreement with published results (e.g., GeoReM and GEOROC
236 databases; Weis et al., 2005).

237 A static multicollection routine with standard cup configurations was used for all of the
238 isotopic analyses. All measured $^{87}\text{Sr}/^{86}\text{Sr}$ ratios were corrected for mass bias relative to
239 $^{88}\text{Sr}/^{86}\text{Sr}$ of 8.375209, using an exponential law. Data acquisition comprised the integration
240 of two hundred 8 second measurements of masses 84, 85, 86, 87 and 88 for Sr isotope
241 analyses. Mass 85 was measured to correct for any Rb interference on mass 87 assuming a
242 natural $^{87}\text{Rb}/^{85}\text{Rb}$ of 0.386. For Nd isotope measurements, masses 143, 144, 146, 147, 148

243 and 150 were acquired in the same way as Sr isotopes. Mass 147 was measured to correct for
244 Sm interferences on mass 144, 148 and 150. Natural $^{144}\text{Sm}/^{147}\text{Sm}$, $^{144}\text{Sm}/^{148}\text{Sm}$, and
245 $^{144}\text{Sm}/^{150}\text{Sm}$ of 0.2048, 0.7498 and 0.4923, respectively were assumed for this correction.

246 **Mineral chemistry**

247 The phenocryst and microlite phases were identified using Scanning Electron Microscopy
248 (FEI Quanta 650) at Leeds Electron Microscopy and Spectroscopy Centre at the University
249 of Leeds (LEMAS, Leeds, UK). The compositions of olivine, pyroxene, plagioclase, spinel,
250 ilmenite, titanomagnetite, and glass were measured using an electron probe microanalyser
251 (JEOL JXA8230) at LEMAS (Leeds, UK). Details of the analytical conditions of electron
252 microprobe measurements are available in [Table 4](#). For ilmenite, pyroxene, and spinel the
253 values of Fe^{3+} were obtained following the formulations of Droop (1987) and for
254 titanomagnetites, using the formulations of Stormer (1983).

255 **Crystallographic orientation**

256 Crystallographic orientations of olivine crystals were determined using electron back-scatter
257 diffraction in olivine crystals (EBSD; Prior et al. 1999) on a FEI Quanta 650 FEGSEM
258 equipped with a Nordlys EBSD camera at the School of Earth and Environment, University
259 of Leeds (Leeds, UK).

260 **RESULTS**

261 **Petrography**

262 *Osorno 1835 AD products*

263 The lava flow samples have a vesicularity of 4 to 17 vol. % and the lapilli samples have a
264 higher vesicularity of 31 to 61 vol. % (Table 5). The main phenocryst phases are plagioclase,

265 olivine, and clinopyroxene, as described for other Osorno materials in previous studies
266 (Tagiri et al., 1993; Bechon et al., 2022). The nature of the groundmass depends on the
267 sample; those collected from lava flows (Table 5) possess a glass-free microcrystalline
268 groundmass, and those corresponding to lapilli samples contain microlites and interstitial
269 glass. The microlite phases comprise plagioclase, olivine, clinopyroxene and
270 titanomagnetite. We also recognised several volcanic cognate xenoliths (up to ~1 cm long)
271 in the lapilli deposits at location Os-164 (Fig. 3). Their mineralogy comprises plagioclase,
272 orthopyroxene, olivine, ilmenite, and titanomagnetite as phenocrysts, with plagioclase,
273 clinopyroxene and titanomagnetite as microlites (details in Supplementary Data 1) in a glass-
274 free groundmass.

275 *La Viguera cone*

276 The lava flow sample has a vesicularity of ~12 vol. % and the lapilli sample has a higher
277 vesicularity of ~43 vol. % (Table 5). The main phenocryst phases are olivine and
278 clinopyroxene and the microlite phases identified using Scanning Electron Microscopy
279 comprise plagioclase, olivine, and clinopyroxene (Table 5).

280 **Mineral chemistry and textural features of Osorno 1835 products**

281 The main features of the mineral phases from the samples of the 1835 AD Osorno eruption
282 are available in the Table 6.

283 *Plagioclase.*

284 Plagioclase crystals are found as isolated crystals, as inclusions in olivine and clinopyroxene,
285 and as crystal clots, together with olivine and clinopyroxene. They are characterized by
286 compositional zoning that can be categorized into three distinct zones: Zone 1 corresponds

287 to cores of phenocrysts with composition of An₈₀₋₈₉. Zone 2 corresponds to plagioclase
288 surrounding Zone 1 or core composition where Zone 1 is absent, with compositions of An₇₀₋
289 ₇₉. Zone 3 (An₆₀₋₆₆) corresponds to plagioclase rims, mainly observed in plagioclase
290 phenocrysts from lava samples (Fig. 4). Plagioclase exhibit a composition equivalent to Zone
291 3 of plagioclase phenocrysts and display no zoning patterns. Plagioclase microlites have
292 compositions of An₅₂₋₅₉.

293 Several plagioclase phenocrysts exhibit resorption textures in Zone 1 and Zone 2.
294 Additionally, the composition observed in plagioclase phenocryst rims and that filling the
295 spaces created because of resorption are the same and correspond to Zone 3 (Fig. 4).

296 *Olivine*

297 Olivine phenocrysts are present as both isolated crystals (e.g., Figure 5) and clots of crystals
298 (together with plagioclase, and clinopyroxene). There are two groups of olivine phenocrysts
299 regarding core compositions: Group 1 corresponds to crystals in the compositional range of
300 Fo₇₆₋₇₉. Group 2 are crystals in the compositional range of Fo₆₉₋₇₃. All the olivine phenocrysts
301 have normal zoning consisting of thin rims (< 15 μm) of composition Fo₆₇₋₇₅. EBSD maps
302 (cf. Prior et al., 1999) show no measurable deformation in olivine phenocrysts (Fig. 6).
303 Olivine microphenocrysts exhibit the same compositions as Group 1 and Group 2 of olivine
304 phenocrysts (including ~3 μm thick rims with normal zonation). Olivine microlites have a
305 composition of Fo₅₂₋₆₀.

306 *Pyroxene*

307 Clinopyroxene phenocrysts occur as isolated crystals, inclusions in olivine, and as members
308 of crystal clots. The clinopyroxene compositional range corresponds to augite: $\text{En}_{43-52}\text{Wo}_{33-}$
309 $_{44}\text{Fs}_{8-16}$. The pyroxenes coupled with olivine grains from Group 1 in crystal clots have a
310 composition in the range of $\text{En}_{48-49}\text{Wo}_{40-42}\text{Fs}_{8-9}$ and those coupled with olivine grains from
311 Group 2 have a composition in the range of $\text{En}_{43-52}\text{Wo}_{33-44}\text{Fs}_{11-16}$.

312 *Spinel*

313 Spinel crystals are found as octahedral crystals typically preserved as inclusions in olivine
314 phenocrysts and rarely (< 5% of the Cr-spinel crystals) as isolated crystals. The Mg# and Cr#
315 (Cr# = molar Cr/(Fe³⁺+Al+Cr)) values yield ranges from 0.28 to 0.44 and from 0.27 to 0.36,
316 respectively. Spinel as inclusions in olivine phenocrysts from Group 1 have a Mg# ranging
317 from 0.36 to 0.44 and Cr# from 0.32 to 0.36, whereas those preserved as inclusions in olivine
318 phenocrysts from Group 2 have a Mg# from 0.28 to 0.29 and Cr# from 0.27 to 0.28. Measured
319 SiO₂ concentrations were low (< 0.2 wt. %) suggesting that analyses were not affected by
320 significant secondary fluorescence (Coogan et al., 2014) and that all the measured spinel
321 contents are reliable.

322 *Glass*

323 Glass occurs in the groundmass, interstitial between crystal clots, and as melt inclusions in
324 olivine and plagioclase phenocrysts (Fig. 4; 5). Olivine-hosted melt inclusions are generally
325 spherical with radii from 15 to 150 μm, whereas plagioclase-hosted melt inclusions have
326 elliptical outlines with long-axis lengths from 40 to 160 μm and short-axis widths from 20 to
327 50 μm. Different compositional groups are recognised in glasses; glass from groundmass,
328 which has SiO₂ contents of 53.9 to 57.1 wt. %, versus interstitial glass between crystals from

329 clots and olivine-hosted melt inclusions, which both exhibit SiO₂ contents of 51.9 to 54.7 wt.
330 % (see Table 7, details in Supplementary Data 2).

331 **Major and trace element geochemistry**

332 The collected 1835 Osorno eruption samples are comparatively uniform with respect to their
333 major element concentrations. All are of basaltic andesite composition (52.4 to 52.9 SiO₂ wt.
334 %; Fig. 7) with high Al₂O₃ contents (19.6 to 20.7 wt. %), MgO contents from 4.09 to 4.55
335 wt. %, Mg# (Mg# = molar Mg/(Mg+Fe²⁺)) between 0.56 and 0.59, K₂O from 0.49 to 0.51 wt.
336 % and CaO contents from 10.2 to 10.9 wt. %. The 1835 Osorno products plot in the tholeiitic
337 trend (Fig. 8; Calbuco products in the tholeiitic trend and La Viguera products plot in the
338 calc-alkaline trend). Comparison with other nearby volcanic products erupted during the
339 Holocene can be seen in Fig. 7. Trace element concentrations are also similar between
340 samples, with invariant multi-element diagram and Rare Earth Element (REE) patterns and
341 a narrow range of La_N/Yb_N (2.1 to 2.2; Fig. 9, where the subscript N refers to values
342 normalised to Primitive Mantle concentrations; Sun and McDonough, 1989). The 1835
343 Osorno products show similar values and REE patterns to volcanic materials from the nearby
344 Calbuco and La Picada volcanoes, and the La Viguera cone (Fig. 9). In addition, the
345 incompatible trace element diagram of the 1835 Osorno products shows similar trends in all
346 samples with positive Pb and Sr anomalies and negative Nb, P, and Ti anomalies, typical of
347 magmas from arc settings (Sun and McDonough, 1989; Fig. 9). The trends are consistent
348 with those reported by Bechon et al. (2022).

349 **Sr-Nd isotopic compositions**

350 Osorno (samples Os-144a, Os-147a, Os-154a, and Os-164a; see locations in Fig. 3) and
351 Calbuco (samples Cal-149Ta, Cal-149Tb, Cal-157a, Cal-158a, Cal-159a, and Cal-160; see
352 location in Morgado et al., 2019a) volcanic products each show a narrow range of $^{87}\text{Sr}/^{86}\text{Sr}$
353 values (0.704309 to 0.704333, Osorno; 0.704384 to 0.704438, Calbuco), which though very
354 close, do not overlap (Fig. 10). $^{143}\text{Nd}/^{144}\text{Nd}$ in Osorno and Calbuco volcanic products show
355 ranges of 0.512741 to 0.512784 (Osorno) and 0.512680 to 0.512835 (Calbuco). Two samples
356 from La Viguera (LV-152a and LV-153a), a volcanic cone overlying the LOFZ, exhibit
357 lower $^{87}\text{Sr}/^{86}\text{Sr}$ (0.703705 to 0.703753), but $^{143}\text{Nd}/^{144}\text{Nd}$ (0.512763 to 0.512829) in the range
358 of those obtained in Osorno and Calbuco volcanic samples (Fig. 10). Non-volcanic crustal
359 rocks of the region are potential contaminants of Osorno and Calbuco products (granulites,
360 gabbros, and granitoids, Hickey-Vargas et al., 1995; Aragón et al., 2011) and most of their
361 isotopic values differ from the ranges of $^{87}\text{Sr}/^{86}\text{Sr}$ and $^{143}\text{Nd}/^{144}\text{Nd}$ of Osorno and Calbuco
362 products (see Table 3).

363 **DISCUSSION**

364 **Calculated groundmass composition**

365 Oxygen fugacity modelling was performed using the compositions of both glass inclusions
366 in olivine phenocrysts (Group 1 and Group 2) and the calculated anhydrous matrix
367 composition (Os-144* composition, calculated from the composition of the sample Os-144,
368 representing the more evolved residual melt, present immediately before eruption; see Table
369 8). Being low in the sequence, the sample Os-144 has experienced limited interaction with
370 the atmosphere and most likely preserves a record of the oxygen fugacity most similar to pre-

371 eruptive conditions (Mollo et al., 2013). In addition, this sample has a representative whole-
372 rock chemistry, typical mineral assemblages and displays textural features observed in all the
373 volcanic products analysed (see Tables 2 and 5), making Os-144 a representative effusive
374 product of the 1835 AD eruption. The fall deposits corresponding to the second eruptive
375 phase (Os-164) are excluded from this type of modelling because they are more prone to
376 rapid oxidation during the eruption than those from lava flows (Mollo et al., 2013). We
377 determined the groundmass composition (glass and microlites) by considering Os-144 (fresh
378 and with no evident signs of oxidation) as representative of the mean whole-rock composition
379 (Table 2). To obtain groundmass composition, we subtracted the composition of
380 representative phenocryst mineral phases, considering mass balance and modal abundances
381 (Table 5). From now on that composition is named as Os-144* (Table 8, details of calculation
382 are available in Supplementary Data 2). The calculated composition of Os-144* is similar to
383 those reported for several aphyric products of Osorno volcano from Unit 3 reported by Tagiri
384 et al. (1993) and Group 2 of measured glasses measured as part of this study.

385 **Thermodynamic equilibrium**

386 Chemical equilibrium between olivine and clinopyroxene was tested using equations from
387 Grove et al. (1997). According to these equations, we calculated a hypothetical melt
388 composition in equilibrium with each measured crystal composition. We did this by checking
389 their partition coefficient K_D^{Fe-Mg} , where $K_D^{Fe-Mg} = [X_{Fe}^{Mineral\ phase} \times X_{Mg}^{Melt}] /$
390 $[X_{Mg}^{Mineral\ phase} \times X_{Fe}^{Melt}]$ (Roeder and Emslie, 1970). If the phases are in equilibrium with
391 the same melt, then we considered the whole system, consisting of those solid phases to be
392 in equilibrium too (details in Supplementary Data 3).

393 **Olivine-melt equilibrium and diffusive reequilibrium**

394 The equilibrium conditions between melt inclusions and the olivine host (Fig. 11) are tested
395 using the Toplis (2005) equation for K_D^{Fe-Mg} , considering Fe^{2+} and Fe^{3+} values in melt
396 following the methods of Maurel and Maurel (1982). This equilibrium filter is used prior to
397 the application of other methods that will be used to estimate intensive parameters. Most of
398 the olivine compositions (n=6) from Group 1 (Fo₇₆₋₇₉) and the two olivine crystals from
399 Group 2 (Fo₆₉₋₇₃) are in equilibrium with the hosted melt inclusions (Fig. 11). We only
400 consider glassy melt inclusions in olivine to ensure their compositions lie on the liquid line
401 of descent (Kent, 2008). Although inclusions with visible daughter and/or co-trapped
402 minerals exist, they are discarded from our attempt to obtain representative host melt contents
403 (Sinton et al., 1993; Danyushevsky et al., 2002a; Kent, 2008). We also discarded melt
404 inclusions linked to the exterior of the host mineral. The host mineral (in this case, olivine)
405 crystallises in melt inclusions at the boundary between the melt inclusion and the host
406 mineral, which has a lower nucleation energy than homogeneous nucleation (cf. Roedder,
407 1979). According to Kent (2008), this growth on the inclusion wall may be apparent in
408 compositional profiles and also BSE images. This growth causes fractionation of the melt
409 within the inclusion, which can subsequently be modified by diffusion during magma
410 residence at high temperatures (Danyushevsky et al., 2000; 2002a). Larger melt inclusion
411 diameters are associated with longer re-equilibration timescales (Danyushevsky et al. 2000;
412 2002b). To study pre-eruptive conditions via melt inclusions, we examine the core
413 compositions of the largest melt inclusions (diameter > 50 μ m), which we interpret would
414 represent pre-eruptive composition before re-equilibration (cf. Danyushevsky et al., 2002b).
415 The only evidence of post-entrapment crystallisation is a thin rim (< 1 μ m) of olivine growth.

416 Compositional traverses were measured in olivine-hosted melt inclusions and show MgO
417 variations because the growth of the olivine thin rims were balanced by MgO consumption
418 from the melt inclusion. We can be confident that the compositional profiles of olivine-hosted
419 melt inclusions represent diffusion and are not affected by convolution effects because the
420 host olivine phenocrysts have higher MgO contents than the melt inclusions. Hence, they can
421 be used to calculate timescales of re-equilibration. If secondary fluorescence is affecting the
422 measurements, then the compositional profiles within melt inclusions would show an
423 apparent increase in MgO content towards the rims. In contrast, we recognise a decrease in
424 MgO content (Fig. 12). In small melt inclusions (diameter < 50 μm) the compositional
425 profiles do not show the MgO content depletion towards the rim that we observed in > 50
426 μm diameter melt inclusions. Compositional changes in melt inclusions from core to rim may
427 record the evolution of the melt close to the melt/mineral interface, influenced partially by a
428 high growth rate (Watson and Müller, 2009) and short timescales, hindering compositional
429 re-equilibrium. The calculated timescales of re-equilibration of six compositional traverses
430 for an unrealistically low temperature of 850 °C are only a few minutes (Fig. 12; Table 9)
431 and for more credible magmatic temperature of 1060 °C, are few seconds (Table 9). As such,
432 the compositions of the rims of olivine-hosted melt inclusions were modified only during
433 syn-eruptive process and the plateaus of uniform melt inclusion composition represent pre-
434 eruptive composition. Then the core compositions of olivine-hosted melt inclusions would
435 not be significantly affected by the olivine growth in the olivine-melt interface. According to
436 the experiments of Newcombe et al. (2014), concentration profiles could affect the centres
437 of large melt inclusions in up to 100 minutes.

438 Using finite element methods, we modelled the variation of Mg concentration (C_{MgO} , in wt.
439 %) resulting from Mg diffusion in the olivine-hosted melt inclusions, of the radial component
440 of the time-dependent diffusion equation in spherical polar coordinates (Crank, 1975),
441 following the equation 1:

$$442 \quad \left(\frac{\partial C_{MgO}}{\partial t} \right) = D_{MgO} \left(\frac{\partial^2 C_{MgO}}{\partial r^2} + \frac{2}{r} \cdot \frac{\partial C_{MgO}}{\partial r} \right) \quad \text{equation 1}$$

443 where r is radius (in m), D_{MgO} is diffusivity (in m^2/s), and t is time (in s). We calculated the
444 diffusivity of MgO in melt inclusions (D_{MgO}) following the equation of Chen and Zhang
445 (2008):

$$446 \quad \ln(D_{MgO}) = -7.895 - \frac{26\,257}{T} \quad \text{equation 2}$$

447 where T is in K. The modelled timescales, obtained for different input temperatures, are
448 available in Table 9. Details of all olivine-hosted melt inclusions can be found in
449 Supplementary Data 2 and example videos of modelled diffusion of MgO within melt
450 inclusions (and the corresponding MATLAB® codes) of the previously mentioned Osorno
451 samples are available in Supplementary Data 4. The uncertainties related to the Chen and
452 Zhang (2008) equation and glass chemistry measurement do not change significantly the
453 calculations performed. Therefore, the compositions of the core melt inclusions fit for
454 calculations of intensive variables.

455 **Physical intensive variables**

456 The propagations of uncertainties for the calculation of all the intensive conditions
457 (temperature, pressure, oxygen fugacity, and dissolved magmatic water content) were carried

458 out assuming a Gaussian distribution of uncertainties and considering both internal
459 uncertainties (associated with the experimental method) and external uncertainties
460 (associated with precision and accuracy of electron microprobe analyses) uncertainties. A
461 summary of the intensive variables is in Table 10.

462 *Olivine-augite thermometry*

463 We used the olivine-augite Fe-Mg exchange geothermometer of Loucks (1996) in the cores
464 of olivine-augite equilibrated pairs, which yielded temperatures from 1100 to 1140 °C (± 15
465 °C), with an average of 1114 °C ($\sigma = 16$ °C) for olivine-augite pairs involving Group 1 olivine
466 grains (Fo₇₆₋₇₉) and similar temperatures from 1091 to 1166 °C (± 15 °C), with an average of
467 1125 °C ($\sigma = 24$ °C) for olivine-augite pairs involving Group 2 olivine grains (Fo₆₉₋₇₃). We
468 determined the thermodynamic equilibrium state between olivine and augite (n=30) via the
469 Loucks (1996) parametrisation (see section Supplementary Data 3 for details).

470 *Ca-in-olivine thermometry*

471 We used the Ca-in-olivine thermometer of Shejwalkar and Coogan (2013) in olivine cores
472 (Fo₇₀₋₇₆), which yielded temperatures from 1094 to 1097 °C (± 22 °C), with an average of
473 1096 °C ($\sigma = 2$ °C) for Group 1 olivine phenocrysts (Fo₇₆₋₇₉) and temperatures from 1049 to
474 1173 °C (± 22 °C), with an average of 1129 °C ($\sigma = 40$ °C) for Group 2 olivine phenocrysts
475 (Fo₆₉₋₇₃). We assume contemporary growth in clots of olivine together with clinopyroxene
476 and plagioclase based on their textural relations.

477 *Olivine-spinel thermometry*

478 We used the olivine-spinel thermometer of Coogan et al. (2014) for spinel inclusions in
479 olivine phenocryst cores (see Wan et al., 2008), which yielded temperatures from 1051 to
480 1070 °C (± 42 °C) with an average of 1061 °C ($\sigma = 6$ °C) for Group 1 olivine phenocrysts
481 (Fo₇₆₋₇₉). We also obtained temperatures from 1104 to 1122 °C (± 42 °C) with an average of
482 1113 °C ($\sigma = 13$ °C) for Group 2 olivine phenocrysts (Fo₆₉₋₇₃).

483 *Olivine-spinel oxybarometry*

484 We used the olivine-spinel oxybarometer of Ballhaus et al. (1991) in spinels inclusions in
485 olivine phenocryst cores, which yielded an oxygen fugacity buffer of $\Delta\text{QFM} + 2.3$ (± 0.7)
486 for Group 1 olivine phenocrysts (Fo₇₆₋₇₉) and $\Delta\text{QFM} + 2.7$ (± 0.7) for Group 2 olivine
487 phenocrysts (Fo₆₉₋₇₃). We calculated the values of oxygen fugacity buffer considering the
488 temperatures obtained using the olivine-spinel thermometry (see Olivine-spinel thermometry
489 section) and a constant pressure of 2 kbar (results do not show significant variations with
490 pressure). The total uncertainty considers uncertainties of pressure, temperature, EPMA
491 measurements, and inherent errors of the method.

492 *Spinel-melt oxybarometry in melt inclusions*

493 For these calculations, we considered the same pairs of spinel inclusions and olivine
494 phenocrysts that we used for the determination of equilibrium olivine-spinel (following the
495 procedure of Kamenetsky et al. (2001)). The equilibrium test of olivine-spinel pairs shows a
496 positive correlation of Mg# of spinel inclusions and the hosting forsterite olivine phenocryst
497 (details in Supplementary Data 3). Melt inclusion-olivine host equilibrium is determined as
498 explained in the Thermodynamic Equilibrium section. When the olivine host phenocryst is
499 in equilibrium with both the hosted melt inclusion and the hosted spinel inclusion, we use the
500 Maurel and Maurel (1982) method to determine the Fe²⁺/Fe³⁺ ratio of olivine-hosted melt

501 inclusions considering the Fe^{2+}/Fe^{3+} of the olivine-hosted spinel composition (see Mineral
 502 Chemistry section), which we assume equilibrated by transitivity. Finally, we applied the
 503 Borisov et al. (2018) equation to calculate oxygen fugacity values recorded in olivine-hosted
 504 melt inclusions:

$$505 \quad \log\left(\frac{FeO_{1.5}}{FeO}\right) = k \cdot \log(f_{O_2}) + \frac{h}{T} + c + \sum d_i \cdot X_i + d_{SiAl} \cdot X_{SiO_2} \cdot X_{Al_2O_3} + d_{SiMg} \cdot X_{SiO_2} \\ 506 \quad \cdot X_{MgO} \quad \text{equation 3}$$

507 where T is temperature (in K), X_i is the anhydrous mole fraction of melt components, and c,
 508 h, k, d_i , and d_{X_i} are regression coefficients (constant values as supplied in Supplementary
 509 Data 5).

510 The calculations yield an oxygen fugacity buffer range of $\Delta QFM + 1.1 (\pm 0.5)$ for Group 1
 511 olivine phenocrysts and $\Delta QFM + 1.4 (\pm 1)$ for Group 2 olivine phenocrysts. We performed
 512 these calculations considering a temperature of 1060 °C and 1113 °C, respectively (± 42 °C;
 513 see Olivine-spinel thermometry). The uncertainty of temperature depends on the uncertainty
 514 of both the Fe^{2+}/Fe^{3+} ratio in olivine-hosted spinel inclusions and the method of Borisov et
 515 al. (2018; Equation 3). These oxygen fugacity values are also consistent with those yielded
 516 by the method of Jayasuriya et al. (2004): $\Delta QFM + 0.7 (\pm 0.3)$ and $\Delta QFM + 1.4 (\pm 0.3)$ for
 517 Group 1 and Group 2 of olivine phenocrysts, respectively.

518 *Water dissolved in melt inclusions*

519 We calculated the water dissolved in magma in the olivine-hosted melt inclusions
 520 (representing an early stage of magmatic evolution) using the Ca-in-olivine hygrometer of
 521 Gavrilenko et al. (2016). The calculations yielded water contents from 3.4 to 4.8 wt. % (\pm

522 0.8 wt. %), with an average of 4.3 wt. % ($\sigma = 0.5$ wt. %) for Group 1 olivine phenocrysts and
523 from 3.4 to 5 wt. % (± 0.8 wt. %) H₂O, with an average of 4.2 wt. % ($\sigma = 1.1$ wt. %) for
524 Group 2 olivine phenocrysts. We applied the Gavrilenko et al. (2016) method only in those
525 cases when the melt inclusions are in equilibrium with the host olivine phenocryst (see Fig.
526 12). These results are consistent with those obtained using the Kelley and Cottrell (2009)
527 method for our olivine-hosted melt inclusions to determine water solubility in primitive
528 melts. Although Ca is sensitive to changes in olivine growth (e.g., Simkin and Smith, 1970;
529 Shejwalkar and Coogan, 2013), this effect is negligible for the studied olivine-melt inclusion
530 system because the timescales in which post-entrapment crystallisation could affect the Ca
531 in olivine composition would be relatively long, in contrast with the very short timescales
532 calculated for post-entrapment crystallisation in our samples. In consequence, the Ca in
533 olivine is not significantly affected by dissolution, reprecipitation and other processes. Our
534 calculated ranges of water dissolved in magmas are consistent with other values measured
535 for olivine-hosted melt inclusions from Llaima Volcano, in the SVZ (up to 4.4 wt.%; Bouvet
536 de Maisonneuve et al., 2012; Ruth et al., 2016). Also, our calculated ranges and averages of
537 water dissolved in magmas are consistent with those measured and reported for olivine-
538 hosted melt inclusions from other arc geological settings, such as Marianas (e.g., Shaw et al.,
539 2008; Kelley et al., 2010; Brounce et al., 2014), Central America (e.g., Roggensack et al.,
540 1997; Sadofsky et al., 2008; Portnyagin et al., 2014), and Kamchatka (e.g., Auer et al., 2009;
541 Tobelko et al., 2019).

542 We also used the Putirka (2005; 2008) and Zhang et al. (2007) equations to determine H₂O
543 wt. % dissolved in melts and pressure iteratively, considering the plagioclase phenocrysts
544 composition (An₇₅, Zone 2) and the corresponding plagioclase-hosted melt inclusion

545 **compositions.** The calculations yielded water dissolved concentration between 2.1 and 4.2
546 wt. % and pressures up to 2.6 kbar (details in Supplementary Data 6). Those values are
547 consistent with our other water dissolved calculations. In addition, the “accurate volatiles by
548 difference” method of Hughes et al. (2019) yielded water dissolved contents from 0 to 3 wt.
549 % in olivine-hosted melt inclusions. Due to the slow diffusivity of Ca in olivine (Coogan et
550 al., 2005) compared to the high diffusivity of volatiles in melts (e.g., Zhang and Ni, 2010;
551 Zhang et al., 2010; Zhang and Gan, 2022), we infer that the method of Gavrilenko et al.
552 (2016) would yield water content representing an initial stage after melt inclusion
553 entrapment. Although CaO is very sensitive to olivine growth rate, the relatively constant
554 core CaO concentration in olivine phenocrysts suggest they did not change significantly in
555 time. In contrast, the procedure of Hughes et al. (2019) would likely give dissolved water
556 content in melts after diffusive hydrogen (and other volatiles) loss, which occurs rapidly in
557 response to changes in intensive conditions (e.g., Hauri, 2002; Chen et al., 2011; 2013;
558 Gaetani et al., 2012).

559 *Fe-Ti oxides thermometry and oxybarometry in xenoliths*

560 We used the Ghiorso and Evans (2008) thermometer and oxybarometer in ilmenite-
561 titanomagnetite grain pairs from the volcanic xenoliths within the 1835 AD Osorno volcano
562 products. The cores yielded temperatures from 751 to 841 °C (± 15 °C), with an average of
563 782 °C ($\sigma = 33$ °C) and oxygen fugacity buffer from $\Delta\text{QFM}-0.1$ to $\Delta\text{QFM}+0.7$ (± 0.1), with
564 an average of $\Delta\text{QFM}+0.2$ ($\sigma = 0.3$). The rims yielded higher temperatures from 751 to 983
565 °C (± 15 °C), with an average of 843 °C ($\sigma = 90$ °C), and an oxygen fugacity buffer ranging
566 from $\Delta\text{QFM}-0.4$ to $\Delta\text{QFM}+1.5$ (± 0.1), with an average of $\Delta\text{QFM}+0.4$ ($\sigma = 0.8$).

567 **Thermodynamic modelling**

568 We performed thermodynamic modelling using the freeware Petrolog3 (Danyushevsky and
569 Plechov, 2011) to obtain physical magmatic intensive conditions of two melt compositions:
570 olivine-hosted melt inclusions (both representing an early stage of magmatic evolution) in
571 Group 1 olivine phenocrysts (Fo₇₆₋₇₉) and Group 2 olivine phenocrysts (Fo₆₉₋₇₃). Via
572 Petrolog3, we determined the equilibrium between melt inclusions and their host olivine
573 phenocrysts by crystallisation modelling (for details see Supplementary Data 7). We
574 modelled the equilibrium at different oxygen fugacity buffer (from $\Delta\text{QFM}+0.4$ to
575 $\Delta\text{QFM}+2.4$) and pressure conditions for water-saturated systems. Our objective was to
576 construct a model that was consistent with the calculated water content dissolved in the melt
577 inclusions. Melt inclusions hosted by Group 1 olivine phenocrysts (Fo₇₆₋₇₉) reached
578 equilibrium with up to 1 wt. % crystallisation (see Supplementary Data 7) at pressures of up
579 to 2.5 kbar (equivalent to 7 km depth). Melt inclusions hosted by Group 2 olivine phenocrysts
580 (Fo₆₉₋₇₃) reached equilibrium with up to 4 wt. % crystallisation (see Supplementary Data 7)
581 at the same pressure conditions as inclusions hosted in Group 1 olivine phenocrysts (up to
582 2.5 kbar). Consequently, the melt inclusions were close to equilibrium with the host mineral.
583 These results confirm the validity of the chemical filter to determine the melt-olivine
584 equilibrium and suggest that the early stage of magmatic evolution occurred in the upper
585 crust, at depths up to 7 km depth (i.e., at pressures up to 2.5 kbar). These pressure conditions
586 yielded by thermodynamic modelling are consistent with those calculated, considering
587 measured dissolved water concentrations, in olivine-hosted melt inclusions from arc volcanic
588 products (e.g., Shaw et al., 2008; Auer et al., 2009; Werner et al., 2020; Rasmussen et al.,

589 2022) and those yielded by thermodynamic modelling for Osorno products (Rhyolite-
590 MELTS; Bechon et al., 2022).

591 **A shallow reservoir beneath Osorno volcano**

592 We calculated the pre-eruptive temperature using several complementary methods.
593 Calculations via Fe-Mg exchange geothermometer (Loucks, 1996) yielded temperatures of
594 1114 °C and 1125 °C (± 15 °C) for olivine phenocrysts of composition in the range of Group
595 1 (Fo₇₆₋₇₉) and Group 2 (Fo₆₉₋₇₃), respectively. Ca-in-olivine thermometry (Shejwalkar and
596 Coogan, 2013) yielded slightly different temperatures of 1096 and 1129 (± 22 °C) for Group
597 1 and Group 2 olivine phenocrysts, respectively. Olivine-spinel thermometry (Coogan et al.,
598 2014) yielded lower temperatures of 1061 °C and 1113 °C (± 42 °C) for Group 1 (Fo₇₆₋₇₉) and
599 Group 2 (Fo₆₉₋₇₃) olivine phenocrysts, respectively. Experiments by Müller et al. (2013)
600 suggest diffusion in clinopyroxene is limited. This suggests that olivine-augite thermometer
601 records temperatures that would be representative of the first stage of early magmatic
602 evolution for systems with up to 2 wt. % of water dissolved in melt (Loucks, 1996), which is
603 not the case for Osorno volcanic products. However, experiments performed in andesitic and
604 basaltic rocks from Santorini volcano (Andújar et al. 2015; 2016) yielded pressure (2-4 kbar),
605 temperature (1000-1040 °C) and dissolved water content in the melt (3-5 wt.%), which are
606 consistent with those we obtained from samples of the 1835 AD Osorno eruption.

607 According to Coogan et al. (2014), it seems likely that most of the Al in olivine phenocrysts
608 is due to reaction with the spinel Al component, and therefore olivine-spinel thermometry
609 would preserve a lower temperature more representative of the coexistence of olivine and
610 spinel rather than a liquidus temperature.

611 The oxygen fugacity of the early stage of evolution of the magmatic systems is represented
612 by the compositions of olivine-hosted melt inclusions. The melt inclusion oxybarometry
613 yielded an oxygen fugacity buffer from $\Delta\text{QFM}+1.1$ to $\Delta\text{QFM}+1.4$, suggesting that the
614 oxygen fugacity buffer remained relatively constant during the crystallisation of the olivine
615 phenocrysts from Group 1 and Group 2 of 1835 AD Osorno samples. The absence of
616 orthopyroxene in our samples and the calibration for spinel-bearing mantelic rocks of the
617 Ballhaus et al. (1991) olivine-orthopyroxene-spinel oxybarometer lead us to favour the
618 oxygen fugacity buffer values yielded by other methods.

619 In the cognate xenoliths, the core of the Fe-Ti oxide grains yielded a temperature range of
620 751 to 841 °C and oxygen fugacity conditions up to $\Delta\text{QFM}+1.5$, which correspond to
621 subsolidus conditions, according to Rhyolite-MELTS modelling (Ghiorso and Sack, 1995;
622 Gualda et al., 2012). This suggests that Fe-Ti oxide crystals recorded a compositional
623 equilibrium after a quenching event, which does not represent pre-eruptive magmatic
624 conditions. This re-equilibrium at low temperature occurs because elements in Fe-Ti oxide
625 grains show much higher interdiffusivity than in silicate phases (Aragon et al. 1984;
626 Aggarwal and Dieckmann 2002a, b). Among the six measured traverses in ilmenite-
627 titanomagnetite pairs, we recognised a heating event in only three ilmenite-titanomagnetite
628 pairs (details in Supplementary Data 8). This heating recorded in the grain boundaries
629 suggests the xenoliths were incorporated during the magma ascent or shortly before it. We
630 could not determine the timescales from heating to eruption with a spacing of $\sim 2\ \mu\text{m}$ between
631 single measurements by EPMA.

632 Petrolog3 modelling of olivine-hosted melt inclusion compositions suggests that the early
633 stage of magmatic evolution occurred in the upper crust, which is consistent with the shallow

634 magmatic reservoir beneath Osorno volcano inferred by Díaz et al. (2020) at 4-8 km depth
635 and 2 km to the E of the summit of Osorno volcano. This is also corroborated by
636 SERNAGEOMIN–OVDAS (2017b; 2018a, b; 2019), who reported recent volcano-tectonic
637 seismic activity, in which the associated sources correspond to 2.6–3.9 km depth. In
638 conclusion, all approaches that estimate magmatic pressure conditions suggest the existence
639 of a magmatic reservoir in the upper crust, which seems to be a common feature in the
640 plumbing systems of the SVZ of the Chilean Andes (e.g., Calbuco volcano, Nikkhoo et al.,
641 2016; Delgado et al., 2017; Morgado et al., 2019a; Arzilli et al., 2019; Llaima volcano,
642 Bouvet de Maisonneuve et al., 2012; Villarrica volcano, Lohmar et al., 2012, Morgado et al.,
643 2015; Quetrupillán, Brahm et al., 2018). Moreover, Díaz et al. (2020) also reported (based
644 on magnetotellurics) the presence of a deep (7-18 km depth) magmatic reservoir in the
645 Osorno volcano system.

646 Several prior studies have suggested that relatively reducing magmatic conditions produce
647 tholeiitic magmas, whereas oxidizing magmatic conditions produce calc-alkaline magmas
648 (e.g., Druitt et al., 1999; Andújar et al., 2015; 2016). However, Tatsumi et al. (2008) reported
649 that calc-alkaline and tholeiitic magmas could coexist in temporal and spatial proximity and
650 inferred the existence of both types of basaltic magmas beneath Zoa volcano (Japan).
651 Moreover, Elburg (2010) show that oxybarometers can yield lower oxygen fugacity
652 conditions for calc-alkaline compared to tholeiitic magmas, suggesting fO_2 is not the only
653 variable controlling the differentiation trend. Tholeiitic trends in volcanic products are
654 usually associated with relatively lower concentrations of water dissolved in melts (< 4 wt.
655 %) than calc-alkaline, which would play a role in the tholeiitic trend (e.g., Zimmer et al.,
656 2010; Parman et al., 2011; Hamada et al., 2014). However, recent experiments (Firth et al.,

657 2019) have shown that other reasons also could be invoked to explain both calc-alkaline and
658 tholeiitic trends in volcanic products: contrasting conditions for magmatic fractionation or
659 differences in the parental magma. We suggest the tholeiitic trend recognised in whole-rock
660 and melt inclusion compositions (considering an enrichment of total Fe from melt inclusions
661 to groundmass composition) from the 1835 AD Osorno eruption could initially evolve in the
662 lower crust, in a deep reservoir beneath Osorno volcano (7-15 km depth; Díaz et al., 2021),
663 generating a tholeiitic trend (as in Santorini volcanic products; Andújar et al., 2016). This
664 magma did not subsequently deviate from the tholeiitic trend during subsequent evolution in
665 the shallow magmatic reservoir (Bechon et al., 2022) because the composition of the melt is
666 not controlled solely by fO_2 , but also by dissolved water content, fractionation, and mixing
667 (Elburg, 2010).

668 **Volatile exsolution between early and late evolution stages**

669 We determined the water dissolved in magmas considering olivine-hosted melt inclusions
670 composition representing an early stage of magmatic evolution, which yielded values of up
671 to 5.6 wt. % (following the method of Kelley and Cottrell, 2009) and ~4.2 wt. % average
672 (following the method of Gavrilenko et al., 2016). Other methods to determine water
673 dissolved concentration are related to higher uncertainties: the iterative use of Putirka (2005;
674 2008) and Zhang et al. (2007) equations (which yielded concentrations between 2.1 and 4.2
675 wt. %) and the Hughes et al. (2019) method (up to 3 wt. % concentration), suggest volatile
676 exsolution after the early stage of magmatic evolution before the 1835 AD Osorno eruption.
677 This likely volatile exsolution process could be associated with an accumulation of gas
678 phases in the magma reservoir before the eruption (presumably at the top of the system), the

679 release of which may have been described by Darwin (1839a) as the “smoke” emitted around
680 two months before the onset of the 1835 AD eruption of Osorno.

681 **Crystal-mush within the magma reservoir and crystallisation events**

682 The existence of a crystal mush in the magmatic reservoir can be inferred using several lines
683 of evidence: (i) glomerocrysts, (ii) interstitial glass between grains of crystal clots with
684 different composition from the outer glass (Fig. 6, Table 7), (iii) disequilibrium textures
685 (zoning and resorption) in isolated crystals as well as crystals from clots, and (iv) high
686 degrees of crystallinity (25 to 45 vol. %, Table 5). Based on textural features and the narrow
687 temperature crystallisation range (considering overlapping uncertainties), we infer that the
688 magma was multiply-saturated with the minerals observed throughout the crystallisation
689 sequence. However, the zoning patterns preserved in plagioclase phenocrysts suggest a
690 complex crystallisation history for that phase. The resorption observed in several plagioclase
691 phenocrysts, affecting plagioclase Zones 1 and 2, suggests the occurrence of plagioclase
692 destabilisation after initial crystallisation events. We interpret this destabilisation as relating
693 to processes that caused mush disaggregation. Several authors have given different
694 interpretations of resorption and sieve textures – typical features of disequilibrium between
695 plagioclase and melt – such as fluid-absent rapid decompression (Nelson and Montana, 1992;
696 Annen et al., 2006) or magma mixing (thermal and compositional mixing) usually related to
697 heating (Johannes et al., 1994; Ginibre and Wörner, 2007; Pietranik et al., 2006; Lohmar et
698 al., 2012). Additionally, the existence of patchy zoning with the same composition as
699 plagioclase phenocryst rims and microphenocrysts, equivalent to Zone 3 composition, is
700 consistent with a late pre-eruptive crystallisation episode, after mush disaggregation (Fig.
701 13). Tagiri et al. (1993) inferred that the porphyritic products from Osorno volcano, similar

702 to the volcanic products studied in this work, resulted from the addition of calcic plagioclase
703 to the aphyric products. The most reasonable explanation for this is melting of ~ 20 wt. % of
704 $X_{An70-80}$, similar to Zone 2 plagioclase compositions. Compositions of aphyric materials
705 (Tagiri et al., 1993) are similar to Os-144* (groundmass composition), suggesting that
706 aphyric products from Units 1 and 3 from Osorno volcano also could be explained as
707 interstitial melt extracted from a crystal-mush magma reservoir.

708 Group 1 and Group 2 olivine phenocrysts exhibit fairly constant (unzoned) compositions
709 with only syn-eruptive zoning patterns in the crystal rims. This absence of pre-eruptive
710 zoning patterns suggests that there were either at least two crystal mushes or a zoned crystal
711 mush, where Group 1 and Group 2 olivine phenocrysts grew independently. For this, we
712 conclude that there could not have been any protracted interaction between Groups 1 and 2
713 olivine phenocrysts with a common melt. Moreover, the difference in compositions of
714 pyroxene coupled with Group 1 and those coupled with Group 2 in crystal clots, also suggests
715 the olivine phenocrysts Groups 1 and 2 grew independently of each other. Recent studies
716 (e.g., Bergantz et al., 2015; 2017; Schleicher et al., 2016; Morgado et al., 2019b) have shown
717 that crystal mushes within a magma reservoir interact with a magma intrusion in a “mixing
718 bowl”, above the location of the intrusion itself. This “mixing bowl” corresponds to the lower
719 parts of the crystal mush where thermo-mechanical effects are at a maximum, whereas those
720 places located relatively far from it could form and maintain the distinct identities of the
721 unzoned Group 1 and Group 2 olivine crystals observed in this study of a magma reservoir
722 beneath Osorno volcano (e.g., Morgado et al., 2019b; Cheng et al., 2020). Although we
723 favour the "zoned crystal mush" alternative, Díaz et al. (2020) inferred at least two magma
724 reservoirs below Osorno volcano; one at upper crustal depths (between 4 and 8 km depth)

725 and a second at lower in the crust (between 7 and 15 km). The existence of those reservoirs
726 could explain two crystal mushes that evolved separately and interacted just before the
727 eruption, rather than one single “zoned crystal mush”. The Cr-Al-Fe³⁺ composition of spinel
728 inclusion in olivine phenocryst follow the oxidation trend in cumulus: increasing Fe³⁺ at
729 constant Cr composition (e.g., Simakin et al., 2021; details in Supplementary Data 3). The
730 composition of those spinels would suggest that the interstitial melt is reduced during olivine
731 oxidation (e.g., Knafelc et al., 2019; Simakin et al., 2021). Moreover, embayments in some
732 olivine phenocrysts suggest a disequilibrium event (presumably a heating event), which
733 could be related to crystal mush disaggregation with interstitial melt close to water saturation
734 (Alidibirov and Dingwell, 1996) prior the 1835 AD Osorno eruption, but it did not generate
735 zoning patterns at the upper levels of the zoned crystal mush. The very thin rims of olivine
736 phenocrysts would represent a syn-eruptive quenching overgrowth (e.g., Pankhurst et al.,
737 2018; Couperthwaite et al., 2020).

738 **Regional geochemical heterogeneity: Osorno in context**

739 In volcanic rocks of relatively primitive composition, whole-rock chemistry can give insights
740 into other pre-eruptive processes, such as mantle source and crustal contamination, which
741 complement the information yielded by mineral chemistry and allow comparison with
742 different nearby volcanic systems. The composition of the 1835 AD Osorno products shows
743 similar major element chemistry and trace-element patterns to those reported by Tagiri et al.
744 (1993), Moreno et al. (2010), and Bechon et al. (2022) for Units 4 from the Osorno volcano
745 (Figure 14). Therefore, we consider the volcanic products of the 1835 AD Osorno eruption
746 as representatives of Holocene Osorno volcanic products in general.

747 Volcanic products from Calbuco, Osorno, and La Picada – a stratovolcano on the same NE-
748 SW lineament as Osorno – possess higher whole-rock fluid-mobile to fluid-immobile trace
749 elemental ratios (e.g., Ba/La vs Rb/Nd, Fig. 14) than La Viguería, a nearby small eruptive
750 centre built over the main trace of the LOFZ. We propose that the higher fluid-mobile to
751 fluid-immobile element ratios can be linked with a higher slab-derived fluid component in
752 the mantle source beneath Calbuco, Osorno and La Picada than La Viguería. Moreover,
753 Calbuco, Osorno, and La Picada volcanic products also have lower La/Yb ratios than La
754 Viguería volcanic material, with the lower La/Yb ratios being associated with either a higher
755 degree of partial melting (Fig. 14c), or melting of a previously melt-depleted source.
756 Although La Viguería samples are the most primitive of the studied region (lowest SiO₂ wt.
757 % and highest Mg#, and MgO wt. %; see Fig. 8 and Fig. 14), they also contain the highest
758 K₂O concentrations. Higher K₂O concentrations and a decrease in H₂O content in the
759 volcanic products from La Viguería could be derived from slab sediment melting, resulting
760 from a temperature increase of the slab surface (Hermann and Spandler, 2008). Temperature
761 increases at higher slab depth (Molnar and England, 1995) yielding water-poor sub-arc
762 mantle melts, lower H₂O supply to the mantle wedge and, in consequence, lower degrees of
763 partial melting recorded in volcanic products eastward from the Peru–Chile trench. This is
764 consistent with the conclusions of McGee et al. (2017) and Hickey-Vargas et al. (1989) for
765 the volcanic material from Caburgua-Huelemolle small eruptive centres, located eastward
766 from the nearby Villarrica volcano (39.3°S, Southern Andes), which is, like La Viguería,
767 built over the main trace of the LOFZ. Watt et al. (2013) also came to similar conclusions in
768 their study of the nearby Apagado and Hornopirén volcanoes at the latitude of 42°S.

769 A comparison of the REE systematics of the volcanic products from Calbuco, Osorno, La
770 Picada, and La Viguera (Fig. 9) suggests all these volcanic systems have a similar mantle
771 source. La Viguera cone samples (particularly the sample LV-152) show high Ni, Cr, and
772 low Rb values (Fig. 8 and 14), also suggesting their products are the most primitive of the
773 region.

774 **Sources of volcanic products composition at latitude ~41°06'S**

775 $^{87}\text{Sr}/^{86}\text{Sr}$ and $^{143}\text{Nd}/^{144}\text{Nd}$ isotope ratios can be used to identify the possible involvement of
776 subduction-related fluids and crustal contamination in the generation of the volcanic products
777 from Osorno, Calbuco, and La Viguera (we do not possess isotopic data from the other
778 nearby volcanic centres). We used the compositions of several representative end-member
779 rocks, whose chemistry could influence the Calbuco and Osorno whole-rock chemistry,
780 should the original magma interact with them in any significant way. The mantle source for
781 magmas supplying these volcanoes lies close to a mixing line in $^{87}\text{Sr}/^{86}\text{Sr}$ vs. $^{143}\text{Nd}/^{144}\text{Nd}$
782 space between a representative depleted MORB mantle (DMM; Workman and Hart, 2005)
783 and a representative enriched mantle end member EM-I (enriched mantle, Zindler and Hart,
784 1986; Garapić et al., 2015; Fig. 15). The most primitive samples from La Viguera plot on
785 this mixing line, whereas those from Osorno and Calbuco have, by comparison, elevated
786 $^{87}\text{Sr}/^{86}\text{Sr}$ for a given $^{143}\text{Nd}/^{144}\text{Nd}$ (Fig. 15). This primitive composition of La Viguera
787 volcanic products could be related to the LOFZ, which would facilitate a more rapid
788 magmatic ascent, minimising the interaction with the crust (e.g., López-Escobar et al., 1995a;
789 Cembrano and Lara, 2009; Segovia et al., 2021). Figure 15 also shows a mixing line between
790 altered oceanic crust (AOC; Jacques et al., 2014, via Arc Basalt Simulator; Kimura et al.,
791 2009) and CSVZ trench sediments (Kilian and Behrmann, 2003); both are likely potential

792 contaminants during the production of magma in this region. Possible sources of crustal
793 contamination are also shown on Figure 15 (details on Table 3). These include lower crustal
794 granulites, gabbros (Hickey-Vargas et al., 1995), and upper crustal granodiorites (Aragón et
795 al., 2011), all of which have been identified in the basement of this region.

796 In terms of potential fluid inputs, Calbuco, Osorno, and La Viguería volcanic products have
797 lower $^{87}\text{Sr}/^{86}\text{Sr}$ than AOC and sediments. Also, in terms of $^{143}\text{Nd}/^{144}\text{Nd}$, the studied samples
798 of volcanic products plot in between AOC (higher ratio) and sediments (lower ratio). Using
799 a simple mixing model (Fig. 15; details in Supplementary Data 9), the composition of a fluid
800 that is likely to have influenced the composition of the magmas investigated in this study
801 likely consists of contributions from both AOC and trench sediments. Using the composition
802 of La Viguería volcanic products as the end-member best representing the mantle source, in
803 order to reach the composition of Calbuco and Osorno volcanic products, the contributions
804 of the fluid component should have sediment:AOC ratios ranging from 0.15 to 9 (details of
805 calculations are available as Supplementary Data 9). These significant variations probably
806 occur because the sediments from the trench of CSVZ (Kilian and Behrmann, 2003) can be
807 extremely heterogeneous in terms of its $^{87}\text{Sr}/^{86}\text{Sr}$ and $^{143}\text{Nd}/^{144}\text{Nd}$ signature (Kilian and
808 Behrmann, 2003; Fig. 15). Although it is reasonable to consider some contributions from
809 sediments in the slab-derived fluid, calculations of the fractions are unlikely to be conclusive.
810 The deviation of the Calbuco and Osorno products from the simple mixing line between the
811 representatives DMM and EM1 values likely reflects the influence of these slab-derived
812 fluids. This model does not consider mantle heterogeneity (as suggested for the CSVZ, at
813 $\sim 39^{\circ}30'\text{S}$ latitude by McGee et al., 2017), primarily because the similarities in the non fluid
814 mobile elements systematics of these three volcanoes preclude significant heterogeneity in

815 the mantle source. Chemical variations of slab-derived fluid input have also been suggested
816 for the CSVZ, at ~39°30'S latitude by Hickey-Vargas et al., 1989), but are not accounted for
817 in this first-order model. While chemical heterogeneity has been suggested, we are unable to
818 determine the effect, if any, that different elemental concentrations would likely have on the
819 final isotopic composition of the slab fluids, so we are reluctant to speculate and complicate
820 our model unnecessarily.

821 In terms of assimilation, the presence of both granulite and gabbro xenoliths (Hickey-Vargas
822 et al., 1995) in the volcanic products of the 1961 eruption of Calbuco and the presence of the
823 North-Patagonian batholith in the region (Adriasola et al., 2006; Aragón et al., 2011), infer
824 the possibility of interaction between those crustal rocks and the ascending magma.
825 Following the mixing lines described above for fluid input, we modelled granodiorite,
826 granulite, and gabbro assimilation to assess the $^{143}\text{Nd}/^{144}\text{Nd}$ variations observed in the
827 Calbuco and Osorno products (details in Supplementary Data 9). In order to shift the Osorno
828 volcanic products from the mixing line between DMM and EM1, the assimilation of 40% of
829 granulite would be necessary to reproduce the $^{143}\text{Nd}/^{144}\text{Nd}$ range observed, but this value is
830 unrealistic (Taylor Jr., 1980; DePaolo, 1981). In contrast, less than 10 % assimilation of
831 gabbro and granodiorite is required to reproduce the $^{87}\text{Sr}/^{86}\text{Sr}$ and $^{143}\text{Nd}/^{144}\text{Nd}$ ranges
832 observed. For Calbuco samples, granulite, gabbro, and granodiorite mixing models cannot
833 reproduce the $^{143}\text{Nd}/^{144}\text{Nd}$ range observed and they are not considered as significant
834 contaminants.

835 These models suggest the most important factors controlling the Sr-Nd isotopic ratios of the
836 studied volcanic products (at ~41°S) are the nature of the mantle source and fluid input from
837 the subduction zone rather than crustal assimilation. This is consistent with the conclusions

838 of Jacques et al. (2014), who investigated the compositions of volcanic products from the
839 whole CSVZ. Similarly, the $^{87}\text{Sr}/^{86}\text{Sr}$ of the volcanic material interpreted as xenoliths in 1835
840 Osorno volcanic products from the volcanic vent would not modify considerably the $^{87}\text{Sr}/^{86}\text{Sr}$
841 ratios because the volcanic units 1, 3 and 4 of Osorno (unit 2 data are not reported) exhibit
842 similar values to those we present in this study (Tagiri et al., 1993; Fig. 10).

843 CONCLUSIONS

844 The last eruption of Osorno volcano occurred in 1835-1836 in two eruptive stages, with
845 similar erupted products of basaltic andesite lavas and fall deposits, with plagioclase, olivine,
846 and clinopyroxene as phenocrysts and spinels as inclusions in olivine phenocrysts. We used
847 the composition of mineral phases, together with olivine-hosted melt inclusions to infer the
848 following pre-eruptive features for the Osorno 1835 AD eruption (representative of the
849 Osorno Unit 4):

- 850 - There is at least one upper-crustal magma reservoir beneath Osorno volcano at depths
851 up to ~7 km (equivalent to 2.6 kbar, based on thermodynamic modelling and the use
852 of the plagioclase-melt hygrometer-barometer equations iteratively), consistent with
853 geophysical constraints. Calculated pre-eruptive magma temperatures derived from
854 olivine-hosted spinel inclusions range from 1060 °C to 1113 °C with oxygen fugacity
855 buffer of $\sim\Delta\text{QFM} + 1.1$. Dissolved water concentrations in the olivine-hosted melt
856 inclusions are estimated to have been up to 5.6 wt. %, with an average of ~4.2 wt. %.
- 857 - Disequilibrium textures in combination with evidence of local crystal accumulations
858 make it likely that there was interaction of the erupted magmas with a pre-existing
859 crystal mush. in erupted products (Fig. 16). This material underwent disaggregation
860 due to either increased temperatures or volatile concentrations (Fig.16).

- 861 - Major and trace element concentrations in volcanic products from the latitude of
862 Osorno volcano (41°06'S) indicate a higher fluid contribution to their mantle sources
863 combined with higher degrees of source melting beneath the composite volcanoes
864 (Calbuco, Osorno, and La Picada) than for the smaller calc-alkaline eruptive centres
865 (represented by La Viguería cone), which are built over the LOFZ at the same
866 latitude, but several km eastward of the main volcanic arc.
- 867 - Mixing models suggest that the most dominant factors controlling the composition of
868 volcanic products at 41°06'S latitude are mantle source (depleted peridotite) and slab-
869 derived fluid (with its sediment contribution). The contribution of crustal assimilates
870 (represented by granites and gabbros) are restricted to < 10 wt.%. Our results have
871 implications for the prediction of eruption characteristics in future eruptions of
872 Osorno and neighbouring volcanoes.
- 873 - The approach we employ here, using olivine-hosted melt inclusions composition
874 coupled to diffusion chronometry is adequate to study pre-eruptive and could be used
875 for future studies, together with whole-rock chemistry and geophysical techniques to
876 assess magmatic and volcanic systems holistically.

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886 **Supplementary Data**

887 Supplementary Data are available at Journal of Petrology online.

888 **REFERENCES**

889 Adriasola, A.C., Thomson, S.N., Brix, M.R., Hervé, F. & Stöckhert, B. (2006). Postmagmatic
890 cooling and late Cenozoic denudation of the North Patagonian Batholith in the Los Lagos
891 region of Chile, 41– 42 15' S. *International Journal of Earth Sciences* 95(3), 504-528.

892 Aggarwal, S., & Dieckmann, R. (2002a). Point defects and cation tracer diffusion in
893 $(\text{Ti}_x\text{Fe}_{1-x})_{3-\delta}\text{O}_4$ I. Non-stoichiometry and point defects. *Physics and Chemistry of Minerals*,
894 29(10), 695-706.

895 Aggarwal, S., & Dieckmann, R. (2002b). Point defects and cation tracer diffusion in
896 $(\text{Ti}_x\text{Fe}_{1-x})_{3-\delta}\text{O}_4$. II. Cation tracer diffusion. *Physics and Chemistry of Minerals*, 29(10), 707-
897 718.

898 Alidibirov, M., & Dingwell, D. B. (1996). Magma fragmentation by rapid decompression.
899 *Nature*, 380(6570), 146-148.

900 Andújar, J., Scaillet, B., Pichavant, M., & Druitt, T. H. (2015). Differentiation conditions of
901 a basaltic magma from Santorini, and its bearing on the production of andesite in arc settings.
902 *Journal of Petrology*, 56(4), 765-794.

903 Andújar, J., Scaillet, B., Pichavant, M., & Druitt, T. H. (2016). Generation conditions of
904 dacite and rhyodacite via the crystallization of an andesitic magma. Implications for the
905 plumbing system at Santorini (Greece) and the origin of tholeiitic or calc-alkaline
906 differentiation trends in arc magmas. *Journal of Petrology*, 57(10), 1887-1920.

907 Angermann, D., Klotz, J. & Reigber, C. (1999). Space-geodetic estimation of the Nazca-
908 South America Euler vector. *Earth and Planetary Science Letters*, 171(3), 329-334.

909 Annen, C., Blundy, J.D. & Sparks, R.S.J. (2006). The genesis of intermediate and silicic
910 magmas in deep crustal hot zones. *Journal of Petrology* 47(3), 505-539.

- 911 Aragón, E., Castro, A., Díaz-Alvarado, J. & Liu, D-Y. (2011). The North Patagonian
912 batholith at Paso Puyehue (Argentina-Chile). SHRIMP ages and compositional features.
913 *Journal of South American Earth Sciences* 32, 547-554.
- 914 Aragon, R., McCallister, R. H., & Harrison, H. R. (1984). Cation diffusion in
915 titanomagnetites. *Contributions to Mineralogy and Petrology*, 85(2), 174-185.
- 916 Arzilli, F., Morgavi, D., Petrelli., M, Polacci., M, Burton, M., Di Genova, D., Spina, L., La
917 Spina, G., Hartley, M.E., Romero, J.E. & Fellowes, J. (2019). The unexpected explosive sub-
918 Plinian eruption of Calbuco volcano (22–23 April 2015; southern Chile): Triggering
919 mechanism implications. *Journal of Volcanology and Geothermal Research* 378, 35-50.
- 920 Auer, S., Bindeman, I., Wallace, P., Ponomareva, V., & Portnyagin, M. (2009). The origin
921 of hydrous, high- $\delta^{18}\text{O}$ voluminous volcanism: diverse oxygen isotope values and high
922 magmatic water contents within the volcanic record of Klyuchevskoy volcano, Kamchatka,
923 Russia. *Contributions to Mineralogy and Petrology*, 157(2), 209-230.
- 924 Ballhaus, C., Berry, R. F., & Green, D. H. (1991). High pressure experimental calibration of
925 the olivine-orthopyroxene-spinel oxygen geobarometer: implications for the oxidation state
926 of the upper mantle. *Contributions to Mineralogy and Petrology*, 107(1), 27-40.
- 927 Bechon, T., Billon, M., Namur, O., Bolle, O., Fugmann, P., Foucart, H., Devidal, J-L.,
928 Delmelle, N., & Vander Auwera, J. (2022). Petrology of the magmatic system beneath
929 Osorno volcano (Central Southern Volcanic Zone, Chile). *Lithos*, 426, 106777.
- 930 Bergantz, G. W., Schleicher, J. M., & Burgisser, A. (2015). Open-system dynamics and
931 mixing in magma mushes. *Nature Geoscience*, 8(10), 793-796.
- 932 Bergantz, G. W., Schleicher, J. M., & Burgisser, A. (2017). On the kinematics and dynamics
933 of crystal- rich systems. *Journal of Geophysical Research: Solid Earth*, 122(8), 6131-6159.
- 934 Borisov, A., Behrens, H., & Holtz, F. (2018). Ferric/ferrous ratio in silicate melts: a new
935 model for 1 atm data with special emphasis on the effects of melt composition. *Contributions*
936 *to Mineralogy and Petrology*, 173(12), 1-15.
- 937 Bouvet de Maisonneuve, C., Dungan, M.A., Bachmann, O. & Burgisser, A. (2012). Insights
938 into shallow magma storage and crystallization at Volcán Llaima (Andean Southern Volcanic
939 Zone, Chile). *Journal of Volcanology and Geothermal Research* 211, 76-91.
- 940 Brahm, R., Parada, M.Á., Morgado, E., Contreras, C. & McGee, L.E. (2018) Origin of
941 Holocene trachyte lavas of the Quetrupillán Volcanic Complex, Chile (39°30'S): Examples
942 of residual melts in a rejuvenated crystalline mush reservoir. *Journal of Volcanology and*
943 *Geothermal Research* 357, 163-176.

- 944 Brounce, M. N., Kelley, K. A., & Cottrell, E. (2014). Variations in $\text{Fe}^{3+}/\Sigma \text{Fe}$ of Mariana Arc
945 basalts and mantle wedge $f\text{O}_2$. *Journal of Petrology*, 55(12), 2513-2536.
- 946 Castruccio, A., Clavero, J., Segura, A., Samaniego, P., Roche, O., Le Penec, J. L., &
947 Droguett, B. (2016). Eruptive parameters and dynamics of the April 2015 sub-Plinian
948 eruptions of Calbuco volcano (southern Chile). *Bulletin of Volcanology*, 78(9), 1-19.
- 949 Cembrano, J., Hervé, F. & Lavenu, A. (1996). The Liquiñe Ofqui fault zone: a long-lived
950 intra-arc fault system in southern Chile. *Tectonophysics* 259(1-3), 55-66.
- 951 Cembrano, J. & Lara, L. (2009). The link between volcanism and tectonics in the southern
952 volcanic zone of the Chilean Andes: A review. *Tectonophysics* 471, 96-113.
- 953 Chen, Y., & Zhang, Y. (2008). Olivine dissolution in basaltic melt. *Geochimica et*
954 *Cosmochimica Acta*, 72(19), 4756-4777.
- 955 Chen, Y., Provost, A., Schiano, P., & Cluzel, N. (2011). The rate of water loss from olivine-
956 hosted melt inclusions. *Contributions to Mineralogy and Petrology*, 162(3), 625-636.
- 957 Chen, Y., Provost, A., Schiano, P., & Cluzel, N. (2013). Magma ascent rate and initial water
958 concentration inferred from diffusive water loss from olivine-hosted melt inclusions.
959 *Contributions to Mineralogy and Petrology*, 165(3), 525-541.
- 960 Cheng, L., Costa, F., & Bergantz, G. (2020). Linking fluid dynamics and olivine crystal scale
961 zoning during simulated magma intrusion. *Contributions to Mineralogy and Petrology*, 175,
962 1-14.
- 963 Coogan, L. A., Hain, A., Stahl, S., & Chakraborty, S. (2005). Experimental determination of
964 the diffusion coefficient for calcium in olivine between 900 C and 1500 C. *Geochimica et*
965 *Cosmochimica Acta*, 69(14), 3683-3694.
- 966 Coogan, L. A., Saunders, A. D., & Wilson, R. N. (2014). Aluminum-in-olivine thermometry
967 of primitive basalts: Evidence of an anomalously hot mantle source for large igneous
968 provinces. *Chemical Geology*, 368, 1-10.
- 969 Couperthwaite, F. K., Thordarson, T., Morgan, D. J., Harvey, J., & Wilson, M. (2020).
970 Diffusion timescales of magmatic processes in the Moinui lava eruption at Mauna Loa,
971 Hawaii, as inferred from bimodal olivine populations. *Journal of Petrology*, 61(7), ega058.
- 972 Crank, J. (1975). *The mathematics of diffusion*. Oxford University Press.
- 973 Danyushevsky, L. V., Della-Pasqua, F. N., & Sokolov, S. (2000). Re-equilibration of melt
974 inclusions trapped by magnesian olivine phenocrysts from subduction-related magmas:
975 petrological implications. *Contributions to Mineralogy and Petrology*, 138(1), 68-83.

- 976 Danyushevsky, L. V., McNeill, A. W., & Sobolev, A. V. (2002a). Experimental and
977 petrological studies of melt inclusions in phenocrysts from mantle-derived magmas: an
978 overview of techniques, advantages and complications. *Chemical Geology*, 183(1-4), 5-24.
- 979 Danyushevsky, L. V., & Plechov, P. (2011). Petrolog3: Integrated software for modeling
980 crystallization processes. *Geochemistry, Geophysics, Geosystems*, 12(7).
- 981 Danyushevsky, L. V., Sokolov, S., & Falloon, T. J. (2002b). Melt inclusions in olivine
982 phenocrysts: using diffusive re-equilibration to determine the cooling history of a crystal,
983 with implications for the origin of olivine-phyric volcanic rocks. *Journal of Petrology*, 43(9),
984 1651-1671.
- 985 Darwin, C.R. (1839a). Narrative of the surveying voyages of His Majesty's Ships Adventure
986 and Beagle between the years 1826 and 1836, describing their examination of the southern
987 shores of South America, and the Beagle's circumnavigation of the globe. *Journal and*
988 *remarks*. London: Henry Colburn.
- 989 Darwin, C.R. (1839b). *Journal of Researches into the Geology and Natural History of the*
990 *various countries visited by H.H.S. Beagle*. London: Henry Colburn.
- 991 Darwin, C.R. (1840). On the connexion of certain volcanic phenomena in South America;
992 and on the formation of mountain chains and volcanos, as the effect of the same powers by
993 which continents are elevated. *Transactions of the Geological Society of London* 5(3), 601-
994 631.
- 995 Delgado, F., Pritchard, M.E., Ebmeier, S., González, P. & Lara, L. (2017). Recent unrest
996 (2002-2015) imaged by space geodesy at the highest risk Chilean volcanoes: Villarrica,
997 Llaima, and Calbuco (Southern Andes). *Journal of Volcanology and Geothermal Research*
998 344, 270-288.
- 999 DeMets, C., Gordon, R. G., Argus, D. F., & Stein, S. (1994). Effect of recent revisions to the
1000 geomagnetic reversal time scale on estimates of current plate motions. *Geophysical Research*
1001 *Letters*, 21(20), 2191-2194.
- 1002 DePaolo, D.J. (1981). Trace element and isotopic effects of combined wallrock assimilation
1003 and fractional crystallization. *Earth and Planetary Science Letters*, 53(2), 189-202.
- 1004 Dewey, J. F., & Lamb, S. H. (1992). Active tectonics of the Andes. *Tectonophysics*, 205(1-
1005 3), 79-95.
- 1006 Díaz, D., Zuñiga, F., & Castruccio A. (2020). The interaction between active crustal faults
1007 and volcanism: A case study of the Liquiñe-Ofqui Fault Zone and Osorno volcano, Southern

- 1008 Andes, using magnetotellurics. *Journal of Volcanology and Geothermal Research*, 393,
1009 106806.
- 1010 Droop, G.T.R. (1987). A general equation for estimating Fe³⁺ concentrations in
1011 ferromagnesian silicates and oxides from microprobe analyses, using stoichiometric
1012 criteria. *Mineralogical Magazine* 51(361), 431-435.
- 1013 Druitt, T. H., Edwards, L., Mellors, R. M., Pyle, D. M., Sparks, R. S. J., Lanphere, M.,
1014 Davies, M. & Barreirio, B. (1999). Santorini volcano. *Geological Society Memoir*, 19.
- 1015 Elburg, M. A. (2010). Sources and processes in arc magmatism: The crucial role of water
1016 (An inaugural lecture to the Society). *Geologica Belgica*.
- 1017 Firth, C., Adam, J., Turner, S., Rushmer, T., Brens, R., Green, T. H., Erdmann, S., & O'Neill,
1018 H. (2019). Experimental constraints on the differentiation of low-alkali magmas beneath the
1019 Tonga arc: Implications for the origin of arc tholeiites. *Lithos*, 344, 440-451.
- 1020 Gaetani, G. A., O'Leary, J. A., Shimizu, N., Bucholz, C. E., & Newville, M. (2012). Rapid
1021 reequilibration of H₂O and oxygen fugacity in olivine-hosted melt inclusions. *Geology*,
1022 40(10), 915-918.
- 1023 Garapić, G., Jackson, M.G., Hauri, E.H., Hart, S.R., Farley, K.A., Blusztajn, J.S. &
1024 Woodhead, J.D. (2015). A radiogenic isotopic (He-Sr-Nd-Pb-Os) study of lavas from the
1025 Pitcairn hotspot: Implications for the origin of EM-1 (enriched mantle 1). *Lithos* 228, 1-11.
- 1026 Gavrilenko, M., Herzberg, C., Vidito, C., Carr, M. J., Tenner, T., & Ozerov, A. (2016). A
1027 calcium-in-olivine geothermometer and its application to subduction zone magmatism.
1028 *Journal of Petrology*, 57(9), 1811-1832.
- 1029 Ghiorso, M. S., & Evans, B. W. (2008). Thermodynamics of rhombohedral oxide solid
1030 solutions and a revision of the Fe-Ti two-oxide geothermometer and oxygen-barometer.
1031 *American Journal of Science*, 308(9), 957-1039.
- 1032 Ghiorso, M.S. & Sack, R.O. (1995). Chemical mass transfer in magmatic processes IV. A
1033 revised and internally consistent thermodynamic model for the interpolation and
1034 extrapolation of liquid-solid equilibria in magmatic systems at elevated temperatures and
1035 pressures. *Contributions to Mineralogy and Petrology* 119(2-3), 197-212.
- 1036 Gillis, J.M. (1855). The US Naval Astronomical Expedition to the Southern Hemisphere
1037 during the Years 1849-'50-'51-'52: Chile: its geography, climate, earthquakes, government,
1038 social condition, mineral and agricultural resources, commerce 79. Nicholson.

- 1039 Ginibre, C. & Wörner, G. (2007). Variable parent magmas and recharge regimes of the
1040 Parinacota magma system (N. Chile) revealed by Fe, Mg and Sr zoning in plagioclase. *Lithos*
1041 98(1-4), 118-140.
- 1042 Grove, T.L., Donnelly-Nolan, J.M. & Housh, T. (1997). Magmatic processes that generated
1043 the rhyolite of Glass Mountain, Medicine Lake volcano, N. California. *Contributions to*
1044 *Mineralogy and Petrology* 127(3), 205-223.
- 1045 Gualda, G. A., Ghiorso, M. S., Lemons, R. V., & Carley, T. L. (2012). Rhyolite-MELTS: a
1046 modified calibration of MELTS optimized for silica-rich, fluid-bearing magmatic systems.
1047 *Journal of Petrology*, 53(5), 875-890.
- 1048 Hamada, M., Okayama, Y., Kaneko, T., Yasuda, A., & Fujii, T. (2014). Polybaric
1049 crystallization differentiation of H₂O-saturated island arc low-K tholeiite magmas: a case
1050 study of the Izu-Oshima volcano in the Izu arc. *Earth, Planets and Space*, 66(1), 1-10.
- 1051 Hauri, E. (2002). SIMS analysis of volatiles in silicate glasses, 2: isotopes and abundances
1052 in Hawaiian melt inclusions. *Chemical Geology*, 183(1-4), 115-141.
- 1053 Hickey-Vargas, R., Abdollahi, M.J., Parada, M.Á., López-Escobar, L. & Frey, F.A. (1995)
1054 Crustal xenoliths from Calbuco Volcano, Andean Southern Volcanic Zone: implications for
1055 crustal composition and magma-crust interaction. *Contributions to Mineralogy and Petrology*
1056 119(4), 331–344.
- 1057 Hickey-Vargas, R., Holbik, S., Tormey, D., Frey, F.A. & Moreno-Roa, H. (2016). Basaltic
1058 rocks from the Andean Southern Volcanic Zone: Insights from the comparison of along-
1059 strike and small-scale geochemical variations and their sources. *Lithos* 258, 115-132.
- 1060 Hickey-Vargas, R., Moreno-Roa, H., López-Escobar, L. & Frey, F.A. (1989). Geochemical
1061 variations in Andean basaltic and silicic lavas from the Villarrica-Lanin volcanic chain (39.5
1062 S): an evaluation of source heterogeneity, fractional crystallization and crustal assimilation.
1063 *Contributions to Mineralogy and Petrology* 103(3), 361-386.
- 1064 Hickey-Vargas, R., Sun, M., López-Escobar, L., Moreno-Roa, H., Reagan, M.K., Morris,
1065 J.D. & Ryan, J.G. (2002). Multiple subduction components in the mantle wedge: evidence
1066 from eruptive centers in the Central Southern volcanic zone, Chile. *Geology* 30(3), 199-202.
- 1067 Hughes, E. C., Buse, B., Kearns, S. L., Blundy, J. D., Kilgour, G., & Mader, H. M. (2019).
1068 Low analytical totals in EPMA of hydrous silicate glass due to sub-surface charging:
1069 Obtaining accurate volatiles by difference. *Chemical Geology*, 505, 48-56.

- 1070 Jacques, G., Hoernle, K., Gill, J., Wehrmann, H., Bindeman, I. & Lara, L.E. (2014).
1071 Geochemical variations in the Central Southern Volcanic Zone, Chile (38–43 S): the role of
1072 fluids in generating arc magmas. *Chemical Geology* 371, 27-45.
- 1073 Jarrard, R. D. (1986). Relations among subduction parameters. *Reviews of Geophysics*,
1074 24(2), 217-284.
- 1075 Jayasuriya, K. D., O'Neill, H. S. C., Berry, A. J., & Campbell, S. J. (2004). A Mossbauer
1076 study of the oxidation state of Fe in silicate melts. *American Mineralogist*, 89(11-12), 1597-
1077 1609.
- 1078 Johannes, W., Koepke, J. & Behrens, H. (1994). Partial melting reactions of plagioclases and
1079 plagioclase-bearing systems. In *Feldspars and their Reactions* (pp. 161-194). Springer,
1080 Dordrecht.
- 1081 Kamenetsky, V.S., Crawford, A.J. & Meffre, S. (2001). Factors controlling chemistry of
1082 magmatic spinel: an empirical study of associated olivine, Cr-spinel and melt inclusions from
1083 primitive rocks. *Journal of Petrology* 42(4), 655-671.
- 1084 Kelley, K.A. & Cottrell, E. (2009). Water and the oxidation state of subduction zone magmas.
1085 *Science* 325(5940), 605-607.
- 1086 Kelley, K. A., Plank, T., Newman, S., Stolper, E. M., Grove, T. L., Parman, S., & Hauri, E.
1087 H. (2010). Mantle melting as a function of water content beneath the Mariana Arc. *Journal*
1088 *of Petrology*, 51(8), 1711-1738.
- 1089 Kent, A.J. (2008). Melt inclusions in basaltic and related volcanic rocks. *Reviews in*
1090 *Mineralogy and Geochemistry* 69(1), 273-331.
- 1091 Kilian, R. & Behrmann, J.H. (2003). Geochemical constraints on the sources of Southern
1092 Chile Trench sediments and their recycling in arc magmas of the Southern Andes. *Journal of*
1093 *the Geological Society* 160(1), 57-70.
- 1094 Kimura, J. I., Hacker, B. R., van Keken, P. E., Kawabata, H., Yoshida, T., & Stern, R. J.
1095 (2009). Arc Basalt Simulator version 2, a simulation for slab dehydration and fluid- fluxed
1096 mantle melting for arc basalts: Modeling scheme and application. *Geochemistry,*
1097 *Geophysics, Geosystems*, 10(9).
- 1098 Knafelc, J., Filiberto, J., Ferré, E. C., Conder, J. A., Costello, L., Crandall, J. R., Dyar, M.
1099 D., Friedman, S. A., Hummer, D. R., & Schwenzer, S. P. (2019). The effect of oxidation on
1100 the mineralogy and magnetic properties of olivine. *American Mineralogist*, 104(5), 694-702.

- 1101 Lara, L.E., Orozco, G. & Piña-Gauthier, M. (2012). The 1835 AD fissure eruption at Osorno
1102 volcano, Southern Andes: Tectonic control by the intraarc stress field instead of remote
1103 megathrust-related dynamic strain. *Tectonophysics* 530, 102-110.
- 1104 Le Bas, M.L., Maitre, R.L., Streckeisen, A., Zanettin, B. & IUGS Subcommission on the
1105 Systematics of Igneous Rocks (1986). A chemical classification of volcanic rocks based on
1106 the total alkali-silica diagram. *Journal of Petrology* 27(3), 745-750.
- 1107 Lohmar, S., Parada, M., Gutiérrez, F., Robin, C. & Gerbe, M.C. (2012). Mineralogical and
1108 numerical approaches to establish the pre-eruptive conditions of the mafic Lican Ignimbrite,
1109 Villarrica Volcano (Chilean Southern Andes). *Journal of Volcanology and Geothermal*
1110 *Research* 235, 55-69.
- 1111 López-Escobar, L., Cembrano, J. & Moreno, H. (1995a). Geochemistry and tectonics of the
1112 Chilean Southern Andes basaltic Quaternary volcanism (37-46 S). *Andean Geology* 22(2),
1113 219-234.
- 1114 López-Escobar, L., Parada, M.Á., Hickey-Vargas, R., Frey, F.A., Kempton, P.D. & Moreno,
1115 H. (1995b). Calbuco Volcano and minor eruptive centers distributed along the Liquiñe-Ofqui
1116 Fault Zone, Chile (41–42 S): contrasting origin of andesitic and basaltic magma in the
1117 Southern Volcanic Zone of the Andes. *Contributions to Mineralogy and Petrology* 119(4),
1118 345-361.
- 1119 López-Escobar, L., Parada, M.Á., Moreno, H., Frey, F.A. & Hickey-Vargas, R. (1992). A
1120 contribution to the petrogenesis of Osorno and Calbuco volcanoes, Southern Andes (41°00'
1121 – 41°30'S): comparative study. *Revista Geológica de Chile* 19 (2), 211-226.
- 1122 Loucks, R.R. (1996). A precise olivine-augite Mg-Fe-exchange geothermometer.
1123 *Contributions to Mineralogy and Petrology* 125(2-3), 140-150.
- 1124 Martin, C. (1901). Los volcanes activos de Chile. *Revista Chilena de Historia Natural*.
1125 Imprenta Gillet. Valparaíso. Año V, 242-250.
- 1126 Maurel, C. & Maurel, P. (1982). Etude expérimentale de l'équilibre Fe^{2+} - Fe^{3+} dans les
1127 spinelles chromifères et les liquides silicatés basiques coexistants a 1 atm. *Comptes Rendus*
1128 *de l'Académie des Sciences* 285, 209-215.
- 1129 McGee, L.E., Brahm, R., Rowe, M.C., Handley, H.K., Morgado, E., Lara, L.E., Turner, M.B.,
1130 Vinet, N., Parada, M.Á. & Valdivia, P. (2017). A geochemical approach to distinguishing
1131 competing tectono-magmatic processes preserved in small eruptive centres. *Contributions to*
1132 *Mineralogy and Petrology* 172(6), 44.

- 1133 McGee, L.E. & Smith, I.E. (2016). Interpreting chemical compositions of small scale basaltic
1134 systems: a review. *Journal of Volcanology and Geothermal Research* 325, 45-60.
- 1135 Mollo, S., Scarlato, P., Lanzafame, G. & Ferlito, C. (2013). Deciphering lava flow post-
1136 eruption differentiation processes by means of geochemical and isotopic variations: A case
1137 study from Mt. Etna volcano. *Lithos* 162, 115-127.
- 1138 Moraleda y Montero, J. (1888). *Exploraciones geográficas e Hidrográficas, 1789-1793.*
1139 *Introducción de Diego Barros Arana.* Imprenta Nacional, 533.
- 1140 Moreno, H., Lara, L. & Orozco, G. (2010). *Geología del volcán Osorno.* Servicio Nacional
1141 de Geología y Minería, Carta Geológica de Chile, Serie Geológica Básica, No. 126, Mapa
1142 escala 1:50000.
- 1143 Morgado, E., Parada, M.Á., Contreras, C., Castruccio, A., Gutiérrez, F. & McGee, L.E.
1144 (2015). Contrasting records from mantle to surface of Holocene lavas of two nearby arc
1145 volcanic complexes: Caburgua-Huelemolle Small Eruptive Centers and Villarrica Volcano,
1146 Southern Chile. *Journal of Volcanology and Geothermal Research* 306, 1-16.
- 1147 Morgado, E., Parada, M.Á., Morgan, D.J., Gutiérrez, F., Castruccio, A. & Contreras, C.
1148 (2017). Transient shallow reservoirs beneath small eruptive centres: Constraints from Mg-Fe
1149 interdiffusion in olivine. *Journal of Volcanology and Geothermal Research* 347, 327-336.
- 1150 Morgado, E., Morgan, D.J., Castruccio, A., Ebmeier, S.K., Parada, M.Á., Brahm, R., Harvey,
1151 J., Gutiérrez, F. & Walshaw, R. (2019b). Old magma and a new, intrusive trigger: using
1152 diffusion chronometry to understand the rapid-onset Calbuco eruption, April 2015 (Southern
1153 Chile): *Contributions to Mineralogy and Petrology* 174(61).
- 1154 Morgado, E., Morgan D.J., Harvey, J., Parada, M.Á., Castruccio, A., Brahm, R., Gutiérrez,
1155 F., Georgiev, B. & Hammond, S.J. (2019a). Localised heating and intensive magmatic
1156 conditions prior the 22-23 April 2015 Calbuco volcano eruption (Southern Chile). *Bulletin*
1157 *of Volcanology* 81(4).
- 1158 Müller, T., Dohmen, R., Becker, H. W., Ter Heege, J. H., & Chakraborty, S. (2013). Fe–Mg
1159 interdiffusion rates in clinopyroxene: experimental data and implications for Fe–Mg
1160 exchange geothermometers. *Contributions to Mineralogy and Petrology*, 166(6), 1563-1576.
- 1161 Munizaga, F., Hervé, F., Drake, R., Pankhurst, R.J., Brook, M. & Snelling, N. (1988).
1162 Geochronology of the Lake Region of south-central Chile (39–42 S): Preliminary results.
1163 *Journal of South American Earth Sciences* 1(3), 309-316.
- 1164 Nakamura, K. (1977). Volcanoes as possible indicators of tectonic stress orientation—
1165 principle and proposal. *Journal of Volcanology and Geothermal Research* 2(1), 1-16.

- 1166 Namur, O., Montalbano, S., Bolle, O., & Vander Auwera, J. (2020). Petrology of the April
1167 2015 eruption of Calbuco volcano, southern Chile. *Journal of Petrology*, 61(8), ega084.
- 1168 Nelson, S.T. & Montana, A. (1992). Sieve-textured plagioclase in volcanic rocks produced
1169 by rapid decompression. *American Mineralogist* 77(11-12), 1242-1249.
- 1170 Newcombe, M. E., Fabbrizio, A., Zhang, Y., Ma, C., Le Voyer, M., Guan, Y., Eiler, J. M.,
1171 Saal, A. E. & Stolper, E. M. (2014). Chemical zonation in olivine-hosted melt inclusions.
1172 *Contributions to Mineralogy and Petrology*, 168(1), 1030.
- 1173 Nikkhoo, M., Walter, T.R., Lundgren, P.R. & Prats-Iraola, P. (2016). Compound dislocation
1174 models (CDMs) for volcano deformation analyses. *Geophysical Journal International* 208,
1175 877-894.
- 1176 Pankhurst, M. J., Morgan, D. J., Thordarson, T., & Loughlin, S. C. (2018). Magmatic crystal
1177 records in time, space, and process, causatively linked with volcanic unrest. *Earth and*
1178 *Planetary Science Letters*, 493, 231-241.
- 1179 Parman, S. W., Grove, T. L., Kelley, K. A., & Plank, T. (2011). Along-arc variations in the
1180 pre-eruptive H₂O contents of Mariana arc magmas inferred from fractionation paths. *Journal*
1181 *of Petrology*, 52(2), 257-278.
- 1182 Pérez-Flores, P., Cembrano, J., Sánchez-Alfaro, P., Veloso, E., Arancibia, G., & Roquer, T.
1183 (2016). Tectonics, magmatism and paleo-fluid distribution in a strike-slip setting: Insights
1184 from the northern termination of the Liquiñe–Ofqui fault System, Chile. *Tectonophysics*,
1185 680, 192-210.
- 1186 Petit-Breuilh, M.E. (1999). Cronología eruptiva histórica de los volcanes Osorno y Calbuco,
1187 Andes del Sur (41°-41°30'S). Servicio Nacional de Geología y Minería (SERNAGEOMIN),
1188 Chile. Boletín (n.53): 46 p.
- 1189 Pietranik, A., Koepke, J. & Puziewicz, J. (2006). Crystallization and resorption in plutonic
1190 plagioclase: implications on the evolution of granodiorite magma (Gęsiniec granodiorite,
1191 Strzelin Crystalline Massif, SW Poland). *Lithos* 86(3-4), 260-280.
- 1192 Piochi, M., Bruno, P.P., & De Astis, G. (2005). Relative roles of rifting tectonics and magma
1193 ascent processes: Inferences from geophysical, structural, volcanological, and geochemical
1194 data for the Neapolitan volcanic region (southern Italy). *Geochemistry Geophysics*
1195 *Geosystems* 6(7).
- 1196 Portnyagin, M. V., Hoernle, K., & Mironov, N. L. (2014). Contrasting compositional trends
1197 of rocks and olivine-hosted melt inclusions from Cerro Negro volcano (Central America):

- 1198 implications for decompression-driven fractionation of hydrous magmas. *International*
1199 *Journal of Earth Sciences*, 103(7), 1963-1982.
- 1200 Prior, D.J., Boyle, A.P., Brenker, F., Cheadle, M.C., Day, A., López, G., Peruzzo, L., Potts,
1201 G.J., Reddy, S., Spiess, R., Timms, N.E., Trimby P., Wheeler, J. & Zetterström, L. (1999).
1202 The application of electron backscatter diffraction and orientation contrast imaging in the
1203 SEM to textural problems in rocks. *American Mineralogist* 84(11-12), 1741-1759.
- 1204 Putirka, K. D. (2005). Igneous thermometers and barometers based on plagioclase + liquid
1205 equilibria: Tests of some existing models and new calibrations. *American Mineralogist*, 90(2-
1206 3), 336-346.
- 1207 Putirka, K. D. (2008). Thermometers and barometers for volcanic systems. *Reviews in*
1208 *mineralogy and geochemistry*, 69(1), 61-120.
- 1209 Pyle, D.M. (2000). Sizes of volcanic eruptions. In: *Encyclopedia of Volcanoes*. Sigurdsson,
1210 H., Houghton, B., McNutt, S.R., Rymer, H., Stix, J. Academic Press, San Diego.
- 1211 Rasmussen, D. J., Plank, T. A., Roman, D. C., & Zimmer, M. M. (2022). Magmatic water
1212 content controls the pre-eruptive depth of arc magmas. *Science*, 375(6585), 1169-1172.
- 1213 Rawson, H., Pyle, D. M., Mather, T. A., Smith, V. C., Fontijn, K., Lachowycz, S. M., &
1214 Naranjo, J. A. (2016). The magmatic and eruptive response of arc volcanoes to deglaciation:
1215 Insights from southern Chile. *Geology*, 44(4), 251-254.
- 1216 Roduit, N. (2005). Two complementary efficient methods to quantify porosity types in digital
1217 images of thin sections with the software JMicroVision. In Abstract, *Scenic Sedimentology*,
1218 24th IAS Meeting.
- 1219 Roedder, E. (1979). Origin and significance of magmatic inclusions. *Bulletin de Mineralogie*,
1220 102(5), 487-510.
- 1221 Roeder, P. L., & Emslie, R. (1970). Olivine-liquid equilibrium. *Contributions to Mineralogy*
1222 *and Petrology*, 29(4), 275-289.
- 1223 Roggensack, K., Hervig, R. L., McKnight, S. B., & Williams, S. N. (1997). Explosive basaltic
1224 volcanism from Cerro Negro volcano: influence of volatiles on eruptive style. *Science*,
1225 277(5332), 1639-1642.
- 1226 Ruth, D. C., Costa, F., De Maisonneuve, C. B., Franco, L., Cortés, J. A., & Calder, E. S.
1227 (2018). Crystal and melt inclusion timescales reveal the evolution of magma migration before
1228 eruption. *Nature Communications*, 9(1), 1-9.

- 1229 Ruth, D. C., Cottrell, E., Cortés, J. A., Kelley, K. A., & Calder, E. S. (2016). From passive
1230 degassing to violent Strombolian eruption: the case of the 2008 eruption of Llaima volcano,
1231 Chile. *Journal of Petrology*, 57(9), 1833-1864.
- 1232 Sadofsky, S. J., Portnyagin, M., Hoernle, K., & van den Bogaard, P. (2008). Subduction
1233 cycling of volatiles and trace elements through the Central American volcanic arc: evidence
1234 from melt inclusions. *Contributions to Mineralogy and Petrology*, 155(4), 433-456.
- 1235 Sánchez, P., Pérez-Flores, P., Arancibia, G., Cembrano, J. & Reich, M. (2013). Crustal
1236 deformation effects on the chemical evolution of geothermal systems: the intra-arc Liquiñe–
1237 Ofqui fault system, Southern Andes. *International Geology Review* 55(11), 1384-1400.
- 1238 Schleicher, J. M., Bergantz, G. W., Breidenthal, R. E., & Burgisser, A. (2016). Time scales
1239 of crystal mixing in magma mushes. *Geophysical Research Letters*, 43(4), 1543-1550.
- 1240 Segovia, M. J., Diaz, D., Slezak, K., & Zuñiga, F. (2021). Magnetotelluric study in the Los
1241 Lagos Region (Chile) to investigate volcano-tectonic processes in the Southern Andes. *Earth,
1242 Planets and Space*, 73(1), 1-14.
- 1243 Sellés, D., Rodríguez, A.C., Dungan, M. A., Naranjo, J. A., & Gardeweg, M. (2004).
1244 Geochemistry of Nevado de Longaví Volcano (36.2 S): a compositionally atypical arc
1245 volcano in the Southern Volcanic Zone of the Andes. *Revista Geológica de Chile*, 31(2), 293-
1246 315.
- 1247 Sellés, D., & Moreno, H. (2011) *Geología del Volcán Calbuco, Región de Los Lagos. Carta
1248 Geológica de Chile. Serie Geológica Básica, No. 130. Escala 1:50.000.*
- 1249 [SERNAGEOMIN–OVDAS \(2017a\). Reporte Especial de Actividad Volcánica \(REAV\) —
1250 Región de los Lagos, volcán Osorno 2017, diciembre 19 —18:30 \(local time\).](#)
- 1251 SERNAGEOMIN–OVDAS (2017b). Red Nacional de Vigilancia Volcánica, Volcán
1252 Calbuco. <https://rnvv.sernageomin.cl/volcan-calbuco/>. Accessed 26 September 2022.
- 1253 [SERNAGEOMIN–OVDAS \(2018a\). Reporte especial de actividad volcánica \(REAV\) —
1254 Región de los Lagos, volcán Osorno 2018, junio 23 —11:00 \(local time\).](#)
- 1255 [SERNAGEOMIN–OVDAS \(2018b\). Reporte especial de actividad volcánica \(REAV\) —
1256 Región de los Lagos, volcán Osorno 2018, agosto 5 —11:47 \(local time\).](#)
- 1257 [SERNAGEOMIN–OVDAS \(2018c\). Reporte Especial de Actividad Volcánica \(REAV\) —
1258 Región de los Lagos, volcán Osorno 2018, abril 14 —14:30 \(local time\).](#)
- 1259 [SERNAGEOMIN–OVDAS \(2019\). Reporte de actividad volcánica \(RAV\) — Región de los
1260 Lagos, volcán Osorno, febrero 2019.](#)

- 1261 [SERNAGEOMIN–OVDAS \(2022\). Reporte Especial de Actividad Volcánica \(RAV\) —](#)
1262 [Región de los Lagos, volcán Osorno, septiembre 2022.](#)
- 1263 Shaw, A. M., Hauri, E. H., Fischer, T. P., Hilton, D. R., & Kelley, K. A. (2008). Hydrogen
1264 isotopes in Mariana arc melt inclusions: Implications for subduction dehydration and the
1265 deep-Earth water cycle. *Earth and Planetary Science Letters*, 275(1-2), 138-145.
- 1266 Shejwalkar, A. & Coogan, L.A. (2013). Experimental calibration of the roles of temperature
1267 and composition in the Ca-in-olivine geothermometer at 0.1 MPa. *Lithos* 177, 54-60.
- 1268 Simakin, A. G., Salova, T. P., Shaposhnikova, O. Y., Isaenko, S. I., & Nekrasov, A. N.
1269 (2021). Experimental Study of Interaction of Carbonic Fluid with Cumulus Minerals of
1270 Ultrabasic Intrusions at 950 °C and 200 MPa. *Petrology*, 29(4), 371-385.
- 1271 Simkin, T., & Smith, J. V. (1970). Minor-element distribution in olivine. *The Journal of*
1272 *Geology*, 78(3), 304-325.
- 1273 Sinton, C. W., Christie, D. M., Coombs, V. L., Nielsen, R. L., & Fisk, M. R. (1993). Near-
1274 primary melt inclusions in anorthite phenocrysts from the Galapagos Platform. *Earth and*
1275 *Planetary Science Letters*, 119(4), 527-537.
- 1276 Smith, I.E.M. & Németh, K. (2017). Source to surface model of monogenetic volcanism: a
1277 critical review. Geological Society, London, Special Publications 446(1), 1-28.
- 1278 Stern, C.R. (2004). Active Andean volcanism: its geologic and tectonic setting. *Revista*
1279 *Geológica de Chile* 31(2), 161-206.
- 1280 Stern, C.R., Moreno, H., López-Escobar, L., Clavero, J.E., Lara, L.E., Naranjo, J.A., Parada,
1281 M.Á. & Skewes, M.A. (2007). Chilean Volcanoes. In: Moreno, T., Gibbons, W. (eds) *The*
1282 *Geology of Chile*, Geological Society of London, London, 149-180.
- 1283 Stormer, J.C. (1983). The effects of recalculation on estimates of temperature and oxygen
1284 fugacity from analyses of multicomponent iron–titanium oxides. *American Mineralogist* 68,
1285 586–594.
- 1286 Sun, S.S. & McDonough, W.S. (1989). Chemical and isotopic systematics of oceanic basalts:
1287 implications for mantle composition and processes. Geological Society, London, Special
1288 Publications 42(1), 313-345.
- 1289 Tagiri, M., Moreno, H., López-Escobar, L. & Notsu, K. (1993). Two magma types of the
1290 high-alumina basalt series of Osorno Volcano, Southern Andes (41°06'S) – plagioclase
1291 dilution effect. *Journal of Mineralogy, Petrology, and Economic Geology* 88, 359-371.

- 1292 Takada, A. (1994). The influence of regional stress and magmatic input on styles of
1293 monogenetic and polygenetic volcanism. *Journal of Geophysical Research: Solid Earth*
1294 99(B7), 13563-13573.
- 1295 Tassara, S., Reich, M., Cannatelli, C., Konecke, B. A., Kausel, D., Morata, D., ... & Leisen,
1296 M. (2020). Post-melting oxidation of highly primitive basalts from the southern Andes.
1297 *Geochimica et Cosmochimica Acta*, 273, 291-312.
- 1298 Tatsumi, Y., Takahashi, T., Hirahara, Y., Chang, Q., Miyazaki, T., Kimura, J. I., Ban, M. &
1299 Sakayori, A. (2008). New insights into andesite genesis: the role of mantle-derived calc-
1300 alkalic and crust-derived tholeiitic melts in magma differentiation beneath Zao Volcano, NE
1301 Japan. *Journal of Petrology*, 49(11), 1971-2008.
- 1302 Taylor Jr, H. P. (1980). The effects of assimilation of country rocks by magmas on $^{18}\text{O}/^{16}\text{O}$
1303 and $^{87}\text{Sr}/^{86}\text{Sr}$ systematics in igneous rocks. *Earth and Planetary Science Letters*, 47(2), 243-
1304 254.
- 1305 Tilling, R. I. (2008). The critical role of volcano monitoring in risk reduction. *Advances in*
1306 *Geosciences* 14, 3-11.
- 1307 Tobelko, D. P., Portnyagin, M. V., Krasheninnikov, S. P., Grib, E. N., & Plechov, P. Y.
1308 (2019). Compositions and formation conditions of primitive magmas of the Karymsky
1309 volcanic center, Kamchatka: evidence from melt inclusions and trace-element
1310 thermobarometry. *Petrology*, 27(3), 243-264.
- 1311 Toplis, M. J. (2005). The thermodynamics of iron and magnesium partitioning between
1312 olivine and liquid: criteria for assessing and predicting equilibrium in natural and
1313 experimental systems. *Contributions to Mineralogy and Petrology*, 149(1), 22-39.
- 1314 Tormey, D.R., Hickey-Vargas, R., Frey, F.A. & López-Escobar, L. (1991). Recent lavas from
1315 the Andean volcanic front (33 to 42 S); interpretations of along-arc compositional
1316 variations. *Andean magmatism and its tectonic setting: Geological Society of America*
1317 *Special Paper* 265, 57-77.
- 1318 Vander Auwera, J., Namur, O., Dutrieux, A., Wilkinson, C.M., Ganerød, M., Coumont, V.
1319 & Bolle, O. (2019). Mantle melting and magmatic processes under La Picada stratovolcano
1320 (CSVZ, Chile). *Journal of Petrology* 60(5), 907-944.
- 1321 Völker, D., Kutterolf, S., & Wehrmann, H. (2011). Comparative mass balance of volcanic
1322 edifices at the southern volcanic zone of the Andes between 33 S and 46 S. *Journal of*
1323 *Volcanology and Geothermal Research*, 205(3-4), 114-129.

- 1324 Wan, Z., Coogan, L. A., & Canil, D. (2008). Experimental calibration of aluminum
1325 partitioning between olivine and spinel as a geothermometer. *American Mineralogist*, 93(7),
1326 1142-1147.
- 1327 Watson, E.B. & Müller, T. (2009). Non-equilibrium isotopic and elemental fractionation
1328 during diffusion-controlled crystal growth under static and dynamic conditions. *Chemical*
1329 *Geology* 267(3-4), 111-124.
- 1330 Watt, S.F., Pyle, D.M. & Mather, T.A. (2009). The influence of great earthquakes on volcanic
1331 eruption rate along the Chilean subduction zone. *Earth and Planetary Science Letters* 277(3-
1332 4), 399-407.
- 1333 Watt, S. F., Pyle, D. M., Mather, T. A., & Naranjo, J. A. (2013). Arc magma compositions
1334 controlled by linked thermal and chemical gradients above the subducting slab. *Geophysical*
1335 *Research Letters*, 40(11), 2550-2556.
- 1336 Weis, D., Kieffer, B., Maerschalk, C., Pretorius, W., & Barling, J. (2005). High- precision
1337 Pb- Sr- Nd- Hf isotopic characterization of USGS BHVO- 1 and BHVO- 2 reference
1338 materials. *Geochemistry, Geophysics, Geosystems*, 6(2).
- 1339 Weller, D. J., & Stern, C. R. (2018). Along-strike variability of primitive magmas (major and
1340 volatile elements) inferred from olivine-hosted melt inclusions, southernmost Andean
1341 Southern Volcanic Zone, Chile. *Lithos*, 296, 233-244.
- 1342 Werner, C., Rasmussen, D. J., Plank, T., Kelly, P. J., Kern, C., Lopez, T., Gliss, J., Power,
1343 J.A., Roman, D. C., Izbekov, P., & Lyons, J. (2020). Linking subsurface to surface using gas
1344 emission and melt inclusion data at Mount Cleveland volcano, Alaska. *Geochemistry,*
1345 *Geophysics, Geosystems*, 21(7), e2019GC008882.
- 1346 Workman, R.K. & Hart, S.R. (2005). Major and trace element composition of the depleted
1347 MORB mantle (DMM). *Earth and Planetary Science Letters* 231(1-2), 53-72.
- 1348 Zindler, A. & Hart, S. (1986). Chemical geodynamics. *Annual Review of Earth and Planetary*
1349 *Sciences* 14(1), 493-571.
- 1350 Zimmer, M. M., Plank, T., Hauri, E. H., Yogodzinski, G. M., Stelling, P., Larsen, J., Singer,
1351 B., Jicha, B., Mandeville, C., & Nye, C. J. (2010). The role of water in generating the calc-
1352 alkaline trend: new volatile data for Aleutian magmas and a new tholeiitic index. *Journal of*
1353 *Petrology*, 51(12), 2411-2444.
- 1354 Zhang, Y., & Gan, T. (2022). Diffusion in melts and magmas. *Reviews in Mineralogy and*
1355 *Geochemistry*, 87(1), 283-337.

- 1356 Zhang, Y., & Ni, H. (2010). Diffusion of H, C, and O components in silicate melts. Reviews
1357 in Mineralogy and Geochemistry, 72(1), 171-225.
- 1358 Zhang, Y., Ni, H., & Chen, Y. (2010) Diffusion data in silicate melts. Reviews in Mineralogy
1359 and Geochemistry, 72(1), 311-408.
- 1360 Zhang, Y., Xu, Z., Zhu, M., & Wang, H. (2007). Silicate melt properties and volcanic
1361 eruptions. Reviews of Geophysics, 45(4).

1362 **Figure 1.** a) Location of the different volcanic zones of the Andes. b) Location of the four
1363 subdivisions of the Southern Volcanic Zone. c) The Central Southern Volcanic Zone and
1364 location of stratovolcanoes and the Liquiñe-Ofqui Fault Zone (LOFZ). Location of the LOFZ
1365 (solid and dashed lines) inferred from Cembrano et al. (1996) and Cembrano and Lara (2009).

1366 **Figure 2.** (a) Location of the Osorno volcano in Chile. (b) Location of Osorno, La Picada,
1367 Punttiagudo, and Calbuco volcanoes and Cordón Cenizos volcanic chain, La Viguería cone,
1368 and the Liquiñe-Ofqui Fault Zone (LOFZ, white dashed line). The alignment of Osorno, La
1369 Picada, Punttiagudo and Cordón Cenizos is represented by a yellow, dashed line. Advanced
1370 Spaceborne Thermal Emission and Reflection Radiometer (ASTER) Global Digital
1371 Elevation Model (GDEM) image was obtained via EarthExplorer, USGS
1372 (<http://earthexplorer.usgs.gov>).

1373 **Figure 3.** Osorno volcano and the deposits of the first and second stages of the 1835 eruption.
1374 Yellow stars represent locations where samples are collected from a lava flow, whereas lapilli
1375 sample locations are represented by red stars. Locations and dates of the lava flows and fall
1376 deposits are based on Moreno et al. (2010) and Lara et al. (2012). ASTER–GDEM image
1377 from EarthExplorer, USGS (<http://earthexplorer.usgs.gov>).

1378 **Figure 4.** (a) Backscatter electron (BSE) images of an isolated plagioclase phenocryst where
1379 different compositional zones are found. Brighter greyscale colours are correlated to higher
1380 density. (b) The arrow represents the profile measured by electron probe micro-analysis
1381 (EPMA) and its direction. (c) Relation and coefficient of determination (r^2) between BSE
1382 greyscale profile and the measured anorthite content ($An = 100 \times Ca / (Ca + Na + K)$; in molar
1383 proportions). “An” is a representative description of the composition of plagioclase because
1384 in all crystals the K content remains constant throughout (details in Supplementary Data 2).
1385 (d) Anorthite profile composition (An) by EPMA of the measured profile (circles) coupled
1386 with the BSE profile based on greyscale values calibrated with the composition measured by
1387 electron microprobe (solid line).

1388 **Figure 5.** (a) Olivine-hosted melt inclusion of ~110 μm diameter. (b) Resorption feature
1389 (embayment) in olivine phenocryst with glass and plagioclase microlites. The yellow dashed
1390 line represents the olivine grain boundary.

1391 **Figure 6.** (a) BSE image of a crystal clot of olivine grains. (b) Zoom of the BSE image,
1392 which shows interstitial glass between the crystals constituting the clot. (c) EBSD map
1393 represents the diversity of olivine crystal orientations, consistent with the observation that
1394 the glasses are interstitial and are not melt inclusions.

1395 **Figure 7.** Total alkali versus silica (Le Bas et al., 1986) plots of the Osorno 1835 volcanic
1396 products (grey squares), the modified composition of Os-144 (Os-144*, yellow star), melt
1397 inclusions (Group 1 of glass, red field), and glass from groundmass (Group 2 glass, green

1398 field), La Viguera cone products (pink triangles), La Picada volcanic products (Vander
1399 Auwera et al., 2019, blue field), and Calbuco 2015 volcanic products (Morgado et al., 2019a,
1400 yellow squares).

1401 **Figure 8.** (a) Variations of Ni ($\mu\text{g}\cdot\text{g}^{-1}$) versus SiO₂ (wt. %), (b) K₂O (wt. %) versus SiO₂ (wt.
1402 %), (c) FeO*/MgO versus SiO₂ (wt. %), (d) K₂O (wt. %) versus MgO (wt. %), (e) CaO (wt.
1403 %) versus MgO (wt. %), and (f) Al₂O₃ (wt. %) versus MgO (wt. %) of Holocene erupted
1404 products from Calbuco volcano (Castruccio et al., 2016; Morgado et al., 2019a; Arzilli et al.,
1405 2019; Namur et al., 2020), Osorno volcano (Tagiri et al., 1993; **Moreno et al., 2010; Bechon**
1406 **et al., 2022;** this study), and Cayutué-La Viguera field (López-Escobar et al., 1995a; this
1407 study).

1408 **Figure 9.** (a) Primitive mantle-normalised (Sun and McDonough, 1989) incompatible trace
1409 element diagram for Calbuco erupted products, taken from López-Escobar et al. (1995b)
1410 Morgado et al. (2019a). Major and trace element concentrations from Osorno and La
1411 Viguera are available in Table 2. (b) REE patterns of bulk rock samples from volcanic
1412 systems of the region (La Viguera, Osorno, Calbuco, and La Picada volcanoes).

1413 **Figure 10.** (a) ¹⁴⁴Nd/¹⁴³Nd versus ⁸⁷Sr/⁸⁶Sr of volcanic products of Osorno, Calbuco,
1414 sediments from the CSVZ trench, granulite xenoliths, gabbro xenoliths and from the
1415 basement, and granitoids from basement. (b) Comparison of ¹⁴⁴Nd/¹⁴³Nd versus ⁸⁷Sr/⁸⁶Sr
1416 with other volcanic zones from the SVZ (from Hickey-Vargas et al., 2016 and references
1417 therein).

1418 **Figure 11.** Olivine core compositions ($X_{\text{ol}}(\text{Fe}^{2+}/\text{Mg})$ vs $X_{\text{melt}}(\text{Fe}^{2+}/\text{Mg})$) for products of the
1419 1835 Osorno eruption. Most of the olivine compositions from group 1 (F₀₇₆₋₇₉) are in
1420 equilibrium with the hosted melt inclusions, whereas the two olivine crystals from group 2
1421 (F₀₆₉₋₇₃) are in equilibrium with the hosted melt inclusions. The Fe²⁺ in the melt is calculated
1422 using the olivine-hosted spinel inclusions Fe²⁺/Fe³⁺. We calculate the equilibrium lines as
1423 $X_{\text{ol}}(\text{Fe}^{2+}/\text{Mg}) = K_{\text{D}} \times X_{\text{melt}}(\text{Fe}^{2+}/\text{Mg})$, where K_{D} is calculated via the Toplis (2005) procedure.
1424 Dashed lines represent the uncertainty of the Toplis (2005) method.

1425 **Figure 12.** (a) BSE image of an olivine phenocryst and melt inclusions. (b) Zoom of the BSE
1426 image, which shows the EPMA profiles in olivine and large melt inclusion. (c) Mg#
1427 composition of olivine phenocryst traverse, which is flat (showing the absence of
1428 disequilibrium), (d) MgO (wt. %) composition profiles of the melt inclusion shown in b),
1429 which show depletion towards the rim. That depletion represents diffusion and the related
1430 timescales are a few minutes.

1431 **Figure 13.** Evolution of plagioclase phenocrysts zoning over time: first event (I) is nucleation
1432 and growth of plagioclase phenocryst cores (Zone 1 composition); after that (II), growth of
1433 plagioclase phenocryst rims and nucleation and growth of new plagioclase phenocrysts

1434 occurred (Zone 2 composition). These two first crystallization events occurred during the
1435 crystal mush formation. After the crystal mush building, another process generating
1436 plagioclase-melt disequilibrium and resorption (we suggest volatile additions or heating)
1437 occurred (III), and finally (IV), a last growth event occurred. We infer that the last growth
1438 event occurred before or during the eruption triggering.

1439 **Figure 14.** Plots of fluid mobile/immobile elements of **Holocene** volcanic samples of the
1440 region: (a) Ba/La versus Rb/La and (b) plot of Ba/Nd versus Rb/Nd. (c) plot of La/Yb versus
1441 SiO₂. (d) plot of Ni (μg·g⁻¹) versus Cr (μg·g⁻¹). (e) plot of Rb (μg·g⁻¹) versus Mg#. In a) and
1442 b) the arrows represent the direction in which the fluid component in the source is increasing.
1443 In c) the arrow represents the direction in which partial melting degree of the source is
1444 increasing. The arrows in d) and e) represent the direction in which the products are more
1445 primitive. **We calculated Mg# assuming Fe²⁺/Fe³⁺ ≈ 3 (the same ratio we reported in this**
1446 **article via titration) in those samples from studies in which only Fe₂O_{3(t)} is reported (Moreno**
1447 **et al., 2010; Bechon et al., 2022).**

1448 **Figure 15.** ¹⁴³Nd/¹⁴⁴Nd vs. ⁸⁷Sr/⁸⁶Sr of whole-rock compositions of the samples from the
1449 regional crustal rocks (granulite, gabbro, and granitoids; Hickey-Vargas et al., 1995; Aragón
1450 et al., 2011), Calbuco and Osorno stratovolcanoes, La Viguera small eruptive centre, and
1451 samples representing mantle compositions (DMM and EM-I; Workman and Hart, 2005;
1452 Garapić et al., 2015) and mixing lines showing possible fluid composition supplies
1453 considering the ranges of ⁸⁷Sr/⁸⁶Sr ratio and ¹⁴³Nd/¹⁴⁴Nd for trench sediments in Kilian and
1454 Behrmann (2003). The sediment compositional end-members are modelled in a) and b),
1455 respectively. The mixing lines are built according to the isotopic values presented in Table
1456 3.

1457 **Figure 16.** Schematic representation of the evolution of the shallow reservoir beneath Osorno
1458 volcano before the 1835 eruption. Figure (a) shows the initial crystal mush, which is
1459 perturbed (by heating and/or volatile addition), as represented in Figure (b). Figure (c) shows
1460 how crystals are disaggregated from the crystal mush and incorporated to the eruptible
1461 magma. Figure (d) shows when the eruption is triggered, it occurs in the main crater as well
1462 as the parasitic cones (image not to scale). The main crater and parasitic cones erupt lava
1463 flows and fall deposits. The sequence occurred for both eruptive events: January-February
1464 1835 and November 1835-January 1836. Volatile accumulation events during magma
1465 evolution and their subsequent release could have triggered the relatively explosive first
1466 event in January 1985.