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Crowding induced morphological changes in synthetic lipid vesicles determined using smFRET

Steven D. Quinn^{1,2}, Lara Dresser¹, Sarah Graham¹, Donato Conteduca¹, Jack Shepherd^{1,3} & Mark C. Leake^{1,2,3}

¹School of Physics, Engineering and Technology, University of York, York, UK. YO10 5DD.

²York Biomedical Research Institute, University of York, York, UK. YO10 5DD.

³Department of Biology, University of York, York, UK. YO10 5DD.

Correspondence to: mark.leake@york.ac.uk

Abstract:

Lipid vesicles are valuable mesoscale molecular confinement vessels for studying membrane mechanics and lipid-protein interactions, and they have found utility among bio-inspired technologies including drug delivery vehicles. While vesicle morphology can be modified by changing the lipid composition and introducing fusion or pore-forming proteins and detergents, the influence of extramembrane crowding on vesicle morphology has remained under explored owing to a lack of experimental tools capable of capturing morphological changes on the nanoscale. Here, we use biocompatible polymers to simulate molecular crowding *in vitro*, and through combinations of FRET spectroscopy, lifetime analysis, dynamic light scattering and single-vesicle imaging, we characterize how crowding regulates vesicle morphology. We show that both freely-diffusing and surface-tethered vesicles fluorescently tagged with the DiI and DiD FRET pair undergo compaction in response to modest concentrations of sorbitol, polyethylene glycol and Ficoll. A striking observation is that sorbitol results in irreversible compaction, whereas the influence of high molecular weight PEG-based crowders was found to be reversible. Regulation of molecular crowding allows for precise control of vesicle architecture *in vitro*, with vast implications for drug delivery and vesicle trafficking systems. Furthermore, our observations of vesicle compaction may also serve to act as a mechanosensitive readout of extramembrane crowding.

Keywords:

Single-molecule, TIRF, FRET, membrane mechanics, lipid vesicle, molecular crowding.

Introduction

Native biological membranes are highly complex and heterogeneous in both size and composition, motivating the development of controllable model-membrane systems whose physical and chemical parameters can be easily tuned¹. A particularly valuable class of such systems are spherical synthetic vesicles, which comprise a phospholipid bilayer surface whose radius of curvature can be carefully controlled. Small unilamellar vesicles (SUVs), for example, range from *ca.* 10-100 nm in size, whereas large unilamellar vesicles (LUVs) and giant unilamellar vesicles (GUVs) are *ca.* 100 - 1000 nm and > 1 μm in diameter, respectively^{1, 2}. Importantly, their phospholipid compositions can be tailored to enable variations in charge, localized membrane roughness, phase separation behaviour across the bilayer and membrane fluidity¹.

Synthetic vesicles have enabled new insights into membrane mechanics³, interactions between lipids and proteins^{4,5} fusion dynamics⁶, and inspired the development of drug delivery vehicles due to their

low toxicity, high loading capacity and controllable release kinetics²⁻³. LUVs have found particular utility as nanoscale containers for constraining biomolecules for single-molecule analysis, where inducing porosity into their membrane facilitates buffer exchange without removing the biomolecule under interrogation⁴. Additionally, vesicles can be biochemically programmed to interact, enabling lipid mixing and content exchange⁵⁻⁷, and the use of perturbative detergents, which alter their morphology⁸⁻⁹, are important in the context of lysis and for triggering release of encapsulated molecules¹⁰. It is also now clear that the structure and dynamics of synthetic vesicles are influenced by factors such as molecular crowding but new experimental approaches are required to explore these interactions further.

The living cell's interior is a densely crowded environment, with up to 40 % of the cytoplasm occupied by solubilised macromolecules¹¹. In this tightly filled space, excluded volumes give rise to steric repulsions, depletion attractions, reduced translational degrees of freedom, biomolecular shape changes, and diffusional effects, all of which contribute to the cell's overall function¹². In this context, the effect of macromolecular crowding on protein¹³⁻¹⁴ and nucleic acid¹⁵⁻¹⁷ structures have been heavily studied *in vitro*. For example, molecular crowding influences protein stability¹⁸⁻¹⁹, interactions, kinetics, diffusion²⁰ and liquid-liquid phase separation events²¹. These results not only lend support to the idea that molecular crowding is a regulator of biomolecular activity, but that living cells may actively regulate crowding to enhance or adjust key processes²². Several different crowding mechanisms have been proposed depending on the biomolecule and co-solute, but recent theoretical studies and energy transfer experiments point towards an intriguing size dependence, where smaller molecular weight crowders are more effective than larger polymers^{17,23-24}. Despite such developments however, the question of how macromolecular crowding influences the global structure of single lipid vesicles remains largely under explored.

Initially, experiments assessing the impact of molecular crowding on the membrane involved the use of vesicles *in vitro* under relatively dilute concentrations of ions or other chemical factors. For example, crowding was observed to give rise to concentration- and polymer-dependent osmotic pressures and non-specific depletions resulting from effects assigned to excluded volumes²⁵⁻²⁶. When high molecular weight crowding agents were introduced, the preferential exclusion of macromolecules from the membrane bilayer introduced an osmotic imbalance which in turn altered the global membrane conformation and promoted membrane fusion²⁷⁻²⁸. Recent studies have shown that encapsulation of crowding agents within single vesicles leads to depletion forces that results in variations in membrane topology and changes to the surface area^{29,30}. When highly hygroscopic crowders such as polyethylene glycol (PEG) were encapsulated, the membrane dehydrated, effectively leading to vesicle compression³¹. Optical microscopy experiments on 20-60 μm sized GUVs have also demonstrated that the encapsulation of high molecular weight PEGs induce membrane stress, oscillations in vesicle size, changes to the membrane tension, permeabilization and variations in the spatial orientation of membrane-bound molecules³². Furthermore, recent investigations on GUV crowding by nucleic acids point towards the formation of local elastic deformations and transient instabilities in the membrane³³. These results support a model in which molecular crowding influences the architecture, dynamics and integrity of biological membranes. However, optical microscopy experiments on GUVs, such as those reported, only reveal macroscopic changes taking place within a 2-dimensional image plane, without quantitatively reporting on molecular level changes across the entire volume. Moreover, it is currently unclear what the influence of low molecular weight crowders are. These challenges, combined with the need to assess the structural integrity of vesicles at the opposite end of the membrane-curvature space, motivated us to extend our single-molecule Förster resonance energy transfer (smFRET) toolbox to quantitatively assess conformational changes taking place within sub-micron sized LUVs in response to a range of molecular crowders *in vitro*.

Here, we employed a single-vesicle assay based on measuring the extent of smFRET between lipophilic fluorophores integrated into the membrane of LUVs to evaluate their conformation in response to crowding agents in the extravescicle space. We first integrated the probes into the LUV bilayer and observed conformational changes in response to crowding via ensemble fluorescence spectroscopy and time-correlated single photon counting. Electron microscopy and dynamic light scattering approaches were then used to quantify structural variations, before wide-field total internal reflection fluorescence microscopy was used to capture the FRET response from single vesicles. By monitoring changes to the FRET efficiency within freely-diffusing and surface-immobilized vesicles, we correlate crowding-induced changes in fluorescence signals to morphological changes within single vesicles. In particular, we found that both sorbitol, a model sugar based cosolvent for low molecular weight crowding³⁴, and high molecular weight crowders such as PEG400, Ficoll400 and PEG8000 could induce enhancements to the observable FRET efficiency. Compaction induced by the high molecular weight crowders was found to be reversible, whereas sorbitol-induced vesicle compaction was permanent. We expect the presented tools will be widely applicable beyond the interactions studied here and we discuss the implications of our findings for enabling control over vesicle morphology, for drug delivery, vesicle trafficking and the regulation of vesicle curvature.

Materials and Methods

Lipid vesicle preparation

1-palmitoyl-2-oleoyl-glycero-3-phosphocholine (POPC) and 1-palmitoyl-2-oleoyl-sn-glycero-3-phospho-L-serine (POPS) lipids in chloroform were purchased from Avanti Polar Lipids and used without any additional purification. 1,1'-Dioctadecyl-3,3,3',3' Tetramethylindocarbocyanine Perchlorate (DiI) and 1,1'-Dioctadecyl-3,3,3,3-tetramethylindodicarbocyanine (DiD) were obtained from ThermoFisher Scientific. Synthetic vesicles were prepared via the extrusion method as previously described⁸⁻⁹. Briefly, mixtures of lipids and membrane stains were mixed in chloroform at final lipid concentrations of 10 mg lipid/ml. The solvent was then evaporated by nitrogen flow to create a dry lipid film, subsequently hydrated in 50 mM Tris buffer (pH 8.0) and mixed by vortex. The resuspended solution was then extruded through a polycarbonate membrane filter to produce vesicles of appropriate diameter.

Ensemble Fluorescence Spectroscopy

Fluorescence emission spectra obtained from DiI and DiD loaded vesicles in 50 mM Tris buffer (pH 8) were recorded using a HORIBA Fluoromax-4 spectrophotometer with $\lambda_{\text{ex}} = 532$ nm. All experiments were performed using a final lipid concentration of 25 μM . Apparent FRET efficiencies, E_{FRET} , were estimated via $E_{\text{FRET}} = (I_A / [I_A + I_D])$, where I_A and I_D are the integrated, background-corrected fluorescence emission intensities of the donor, DiI, and acceptor, DiD, respectively. Data points plotted represent the mean and standard deviation obtained from three separated experimental runs.

Time Correlated Single Photon Counting

Time-resolved fluorescence decays were collected using a FluoTime300 time-correlated single photon counting spectrophotometer equipped with a hybrid PMT detector (Picoquant, Germany). Decays were measured under magic angle conditions using pulsed excitation of 532 nm at 80 MHz, and emission of 565 nm for DiI-DiD loaded vesicles. Excitation of 485 nm (20 MHz) and emission of 600 nm was used for vesicles incorporating the tension reporter FliptR³⁵. All experiments were performed using a final lipid concentration of 25 μM in 50 mM Tris buffer (pH 8). Decays were acquired until 10^4 counts at the decay maximum were observed and fitted by iterative re-convolution of the instrument

response function and the observed fluorescence decay using a multi-exponential decay function of the form $I_t = \sum_{i=1}^n a_i e^{-\frac{t}{\tau_i}}$, where I_t is the intensity at time, t , and t_i and a_i represent the fluorescence lifetime and fractional amplitude of the i^{th} decay component.

Dynamic Light Scattering (DLS)

The hydrodynamic diameter, d_H , of freely diffusing vesicles was estimated using a Zetasizer μ V system equipped with a $\lambda_o = 633$ nm wavelength line. Briefly, Brownian motion of LUVs in solution gives rise to fluctuations in the intensity of back scattered light at $\theta = 178^\circ$. This was used to produce a correlation function, $G(\tau)$, via $G^2(\tau) = \langle I(t)I(t+\tau) \rangle / \langle I(t) \rangle^2$ where τ is the lag time. Correlation curves were fitted to a model of the form $G(\tau) = A[1 + B e^{-2Dq^2\tau}]$ where D is the vesicle diffusion coefficient, A and B are positive constants and $q = \frac{4\pi n}{\lambda_o} \sin\left(\frac{\theta}{2}\right)$, where n is the refractive index of the solution ($n=1.33$). Hydrodynamic diameters were then calculated according to the Stokes-Einstein relationship³⁶ and reported as the mean and standard deviation from three separated experimental runs.

Scanning Electron Microscopy (SEM)

SEM micrographs of vesicles non-specifically bound to a silicon substrate were acquired using a JEOL JSM-7800F system operating at 5 kV^{5,37}. Vesicles were prepared in 50 mM Tris buffer (pH 8) containing molecular crowders at concentration specified in the main text, diluted $\sim 10\times$ in deionized water and vortexed. The vesicles were then applied to the silicon substrate and the solution evaporated. The substrate was then sputtered with a 5 nm thick Pt/Pd layer to avoid charging effects and possible damage of the vesicles during the micrographs acquisition. Vesicle diameters were then determined using ImageJ, and histograms produced using 25 nm bin widths.

Cryo-Transmission Electron Microscopy (Cryo-TEM)

Cryo-TEM was used for direct visualization of vesicle bilayers. To prepare samples for cryo-TEM analysis, Quantifoil copper R 1.2/1.3 200 Mesh grids (Electron Microscopy Sciences) were prepared by glow discharging at 20 mA and 0.26 millibar for 1 minute in a Pelco easiGlow system. Small volumes (2 μ L) of the vesicle sample was applied to the carbon side of the EM grid in 90% humidity, then liquid was blotted off for 0.5 s and the grids plunge-frozen into precooled liquid ethane using a Vitrobot system (Thermo Scientific). This process enabled single vesicles to be embedded within a thin layer of amorphous ice, preserving them in their native state. The samples were then evaluated using a Thermo Scientific Glacios Cryo-TEM electron microscope. TEM images were acquired using an accelerating voltage of 200 kV, and 120,000x magnification. Vesicle sizes were then measured using ImageJ.

Total Internal Reflection Fluorescence (TIRF) microscopy

Microfluidic flow cells were constructed as described previously³⁸ and sequentially incubated with 0.1 mg/mL BSA-Biotin, 1 mg/mL BSA and 0.2 mg/mL NeutrAvidin. After each incubation step the flow cells were rinsed with buffer (50 mM Tris, pH 8) to remove unbound material. Biotinylated vesicles containing 0.1 mol % Dil and 0.1 mol% DiD were then added to the surface using a final concentration of 70 μ g/mL lipids in imaging buffer (50 mM Tris, 6 % (W/V) D-(+)-glucose containing 1 mM Trolox, 6.25 μ M glucose oxidase and 0.2 μ M catalase) and incubated for 15 minutes at room temperature to achieve a surface coverage of ~ 150 -200 vesicles per 50 x 50 μ m field of view. After incubation, the

flow cells were rinsed with imaging buffer. TIRF microscopy was then performed on a custom-modified inverted microscope (Nikon Eclipse Ti) containing a CFI Apo TIRF 100 x NA 1.49 oil-immersion objective lens (Nikon). TIRF illumination was provided by a TEM₀₀ 532 nm wavelength line (Obis, Coherent) at < 8.2 mW cm⁻². Emission was separated from the excitation line via a dichroic and emission filter mounted beneath the lens turret (Chroma 59907-ET-532/640). Dil and DiD emission was then spatially separated using a DualView image splitter (Photometrics) containing a dichroic filter (T640LPXR, Chroma) and band pass filters (ET585/65M and ET700/75M, Chroma) and imaged in parallel on a back-illuminated Prime 95B CMOS camera cooled to -30°C (Photometrics). After each addition of crowder in imaging buffer, 500 frame movies were acquired with 50 ms exposure time. Recorded images were then analysed in MATLAB (R2019a) using iSMS single-molecule FRET microscopy software³⁹. Briefly, the donor and acceptor emission channels were aligned, and background-corrected Dil and DiD integrated emission trajectories were obtained within the excitation field. Apparent FRET efficiencies across the trajectories were then calculated using $I_A/(I_A+I_D)$ as previously described, and related to the mean distance between probes, R , via $E = \frac{R_0^6}{R_0^6+R^6}$, where R_0 is the Förster radius (5.3 nm)⁸. E_{FRET} histograms from $N > 2000$ vesicles were then produced using bin widths of 0.01.

Results and Discussion

Sorbitol induces conformational changes in freely diffusing lipid vesicles.

We first used smFRET between lipophilic membrane stains to explore the structure of LUVs composed of POPC lipids in the presence of synthetic molecular crowding agents. Previous studies evaluating membrane deformation have largely relied on encapsulating molecular crowders or introducing pegylated lipids into GUV bilayers, with single-colour fluorescence imaging used to evaluate macroscopic changes⁴⁰. However, such approaches do not provide detailed information at the molecular level, nor do they report on vesicles smaller than the optical diffraction limit⁸. We therefore applied smFRET to quantify nanoscale structural variations in LUVs of ~200 nm in size in response to dilute crowding conditions.

We first optimized the amount of donors (Dil) and acceptors (DiD) per LUV (1:1 ratio, 0.1 mol % of each dye), such that the average FRET efficiency (E_{FRET}) per vesicle was initially ~0.5. This corresponds to an average Dil-DiD separation close to their Förster radius, and allows for nanoscale changes because of vesicle compaction or swelling, to be quantified by an observable increase or decrease in E_{FRET} . To mimic low molecular weight crowding, we used the polyol osmolyte sorbitol, an established low molecular weight crowding agent which has previously been applied to regulate protein clustering⁴¹⁻⁴³, induce nuclear organization and compact chromatin⁴⁴.

Dynamic light scattering first confirmed the formation of labelled vesicles with a log-normal hydrodynamic diameter centred on 227 nm (**Figure S1**). When vesicles were prepared without labels, a similar distribution was observed, providing confidence that the labelling process does not perturb morphology (**Figure S1**). We then recorded the ensemble fluorescence spectra obtained from the labelled vesicles in solution. As shown in **Figure 1A**, Dil and DiD loaded LUVs displayed a progressive increase in sensitized acceptor emission and E_{FRET} as the sorbitol concentration progressively increased, translating to a reduction in the mean Dil-DiD separation distance from 5.02 ± 0.02 nm in the absence of crowding to 4.76 ± 0.01 nm in the presence of 3M sorbitol (**Figure 1B**). To further confirm the presence of an energy transfer mechanism, we measured the amplitude weighted fluorescence lifetime, τ_{av} , of Dil in the presence of DiD via time-correlated single photon counting. Here, τ_{av} decreased with increasing sorbitol, consistent with a progressive quenching of Dil and corresponding increase in E_{FRET} (**Figure 1C**). The decays fitted well to a bi-exponential decay after

reconvolution with the instrumental response function, which we assigned to energy transfer between dyes on the inner and outer leaflets. Indeed, a bootstrap analysis revealed that both lifetime components, termed τ_1 and τ_2 , progressively decreased with increasing sorbitol (**Figure S2**), suggestive of global conformational changes on both sides of the membrane. Taking these measurements together, the data pointed towards fluorophore packing in the ensemble and were suggestive of vesicle compaction.

To test whether sorbitol induced variations in membrane tension, we next evaluated changes in the fluorescence lifetime of 1 mol % Fluorescent LIPid Tension Reporter (FliptR) incorporated within the LUV bilayer. The FliptR lifetime correlates well with membrane tension³⁵ and in our case, we found that decays fitted well to a biexponential model. In line with previous observations, the longer lifetime component, τ_2 , only accounted for a small fraction of the overall signal. In the absence of sorbitol the ensemble FliptR lifetime was found to be 3.14 ± 0.01 ns, corresponding to a situation where the molecules reside in a planar conformation. However, as sorbitol was progressively titrated, we observed a reduction in lifetime to $\sim 90\%$ of its original value (**Figure 1D, Figure S3**), which we hypothesise is due to a fraction of FliptR molecules twisting into a conformation that is sterically more favourable because of changes to the lipid packing density and decreased membrane tension. Similar to observations made with single vesicles in response to surfactants⁴⁵, we speculate that phospholipid bilayers mixed well with sorbitol leads to a situation where bilayer components are forced by entropy resulting in mixed sorbitol-lipid aggregates, local membrane undulations, and reduced membrane tension.

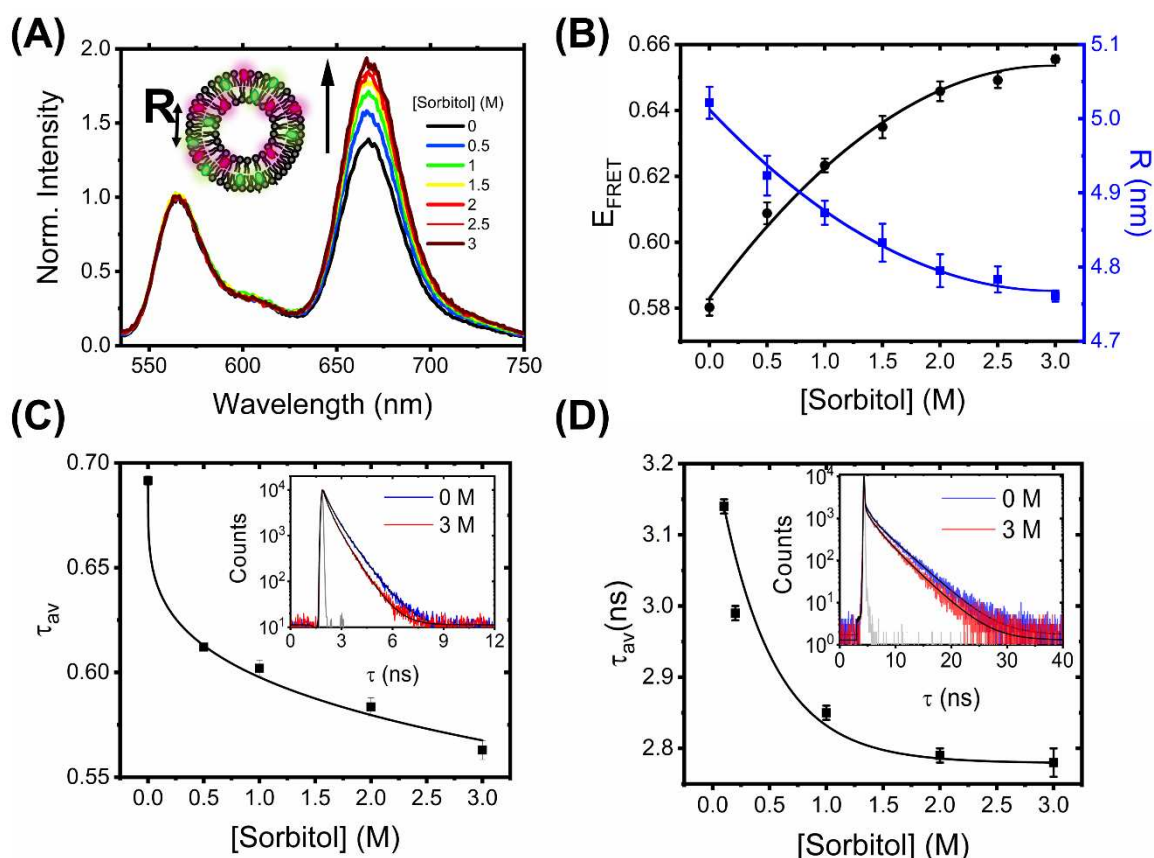


Figure 1. Sorbitol induces conformational changes in freely diffusing vesicles. (A) Normalized fluorescence emission spectra of Dil-DiD LUVs in the absence and presence of sorbitol ($\lambda_{\text{ex}} = 532$ nm). Inset: schematic illustration of the vesicles, where R corresponds to the mean dye-dye separation

distance. (B) Corresponding variations in E_{FRET} , and R . Heuristic fits shown are quadratic (solid black lines), determined with Python 3 using NumPy's polyfit routine. (C) The amplitude weighted average lifetime of DiI and (D) FliptR as a function of sorbitol. Insets correspond to the time-resolved fluorescence decays in the absence (blue) and presence of 3M sorbitol (red). Solid black lines represent biexponential fits to the raw data and the solid gray lines represent the instrument response functions.

The ensemble FRET, lifetime and FliptR analysis broadly supports conformational rearrangements taking place in LUVs in response to sorbitol. However, to understand the parameters that affect this process, we next moved to interrogate the impact of vesicle size, composition and phase. When vesicles of various sizes (100, 400 and 1000 nm), as confirmed by DLS (**Figure S4**), interacted with sorbitol, the FRET efficiencies in all cases increased with crowder concentration, signifying similar dye-dye distance changes (**Figure S5**). Indeed, the relative magnitude of the FRET enhancement compared to those observed from 200 nm diameter vesicles was similar across all conditions tested.

To assess the impact of lipid composition and phase, we next probed the interaction between sorbitol and 100 nm diameter vesicles containing POPS lipids as a function of temperature and compared the relative change in FRET efficiency to similarly sized vesicles composed of POPC (**Figures S5**). POPC has a gel-to-liquid phase transition temperature, T_m , of -2°C , and is therefore in the liquid phase at temperatures $> 4^\circ\text{C}$, whereas $T_m = 14^\circ\text{C}$ for POPS, below which the vesicles are in the gel phase. We observed that the initial FRET efficiency magnitude was similar for both sets of vesicles at 4°C , 21°C and 37°C , however the relative change in E_{FRET} for each sorbitol condition was generally larger in the case of POPC vesicles. One possible explanation for the observed difference could rest in the hydration of the lipid carbonyls. POPS lipids are less mobile and give rise to dehydrated vesicle forms⁴⁶, suggesting uptake of sorbitol into the bilayer may be key for conferring the observed changes.

Morphological Characterization of Single Vesicles in Response to Sorbitol

To explore whether the observed FRET changes were coupled with changes in vesicle morphology, we next investigated the vesicle sizes using scanning electron microscopy (SEM). As shown in **Figure 2A**, vesicles composed of POPC lipids in the absence of sorbitol were mostly spherical (circularity = 0.87 ± 0.16), in line with previous observations³⁷, with a size distribution centred on 132 ± 3 nm (FWHM = 125 ± 8 nm). However, with the addition of sorbitol, the peak shifted to 105 ± 5 nm (FWHM = 155 ± 15 nm) (**Figure 2B**) and the circularity reduced to 0.55 ± 0.12 , consistent with a model in which sorbitol leads to local undulations in membrane morphology and an overall compaction. This analysis was further supported by variations in the vesicle hydrodynamic diameter (d_H) reported by DLS, where the size distributions in solution progressively decreased upon sorbitol addition (**Figure S6**). We note that the vesicle sizes reported by DLS are generally larger than those reported by SEM, likely due to vesicles being dehydrated and fixed under vacuum for SEM imaging. Nevertheless, both measurements pointed towards sorbitol-induced structural changes taking place within single vesicles.

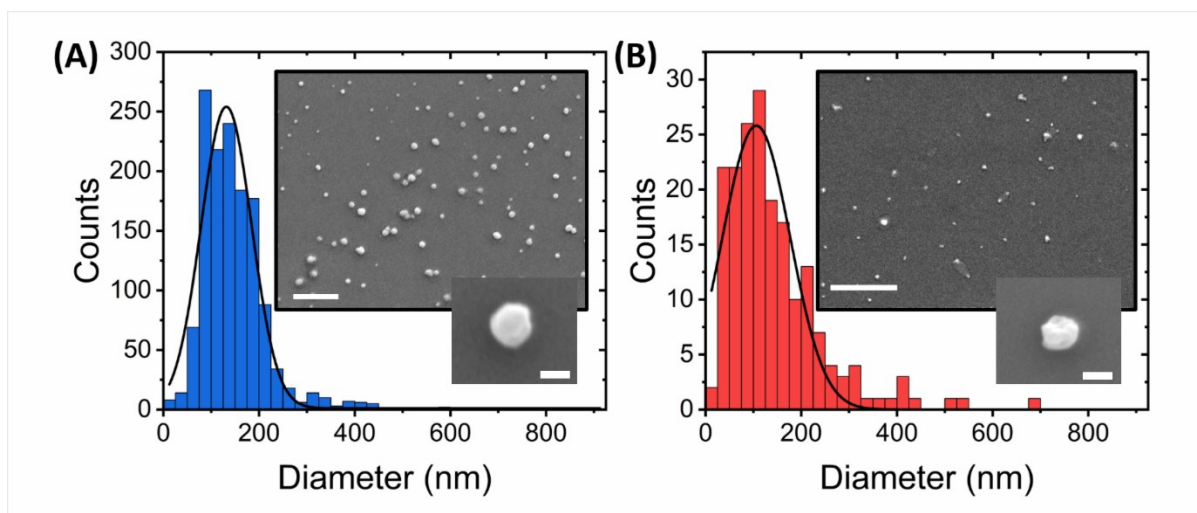


Figure 2. Sorbitol induces compaction and undulations in single lipid vesicles. Quantitative comparison of diameter distributions of POPC vesicles in (A) the absence and (B) presence of 3 M sorbitol. Insets: representative SEM images of immobilized vesicles under the respective conditions. Scale bars = 1 μ m and 100 nm in the larger and smaller insets, respectively.

To further characterize the morphology of the vesicles, we also performed cryo-transmission electron microscopy (cryo-TEM). This enabled us to visualize the internal structures of hydrated vesicles in a frozen state, bypassing the requirement for dehydration. In the absence of crowding agents, the majority of vesicles were classified as unilamellar, exhibiting only a single bilayer (**Figure 3A-D**), with lower fractions (< 20 %) containing double (**Figure 3E,F**) or multiple layers (**Figure 3G,H**). All of the vesicles were intact, lacked any evidence of pore formation and in line with our SEM analysis, they were spherical in nature with diameters in the range 60-315 nm. Among the observed variants, we also observed larger vesicles containing encapsulated smaller vesicles (**Figure 3I**) unusually shaped structures in the form of a bowling pin (**Figure 3J, K**) and large conglomerates (**Figure 3L**). Interestingly, the structures produced from our synthetic vesicle preparation have a striking resemblance to those observed from extracellular vesicles isolated from cerebrospinal fluid⁴⁷, further validating our systems as excellent membrane mimetics. Of those vesicles classified as unilamellar, we observed a mean particle size of 157 ± 5 nm, with a membrane thickness of 6 ± 0.7 nm ($N = 84$) (**Figure S7**). In the presence of 0.5 M sorbitol, the observed species were structurally similar (**Figure 3 M, N, O**) with some vesicles displaying evidence of irregular membrane undulations and bulging (**Figure 3P**). However, in line with our SEM and smFRET analysis, the vesicle size distribution shifted to smaller values and the unilamellar vesicle size reduced by $\sim 30\%$ to 115 ± 5 nm ($N = 138$). When 1 M sorbitol was introduced, most vesicles were intact (**Figure 3Q,R**), however, others displayed signs of damaged, porous membranes, as indicated by the lack of a visible bilayer and lack of electron-dense material spaced across the projection of the vesicle surface (**Figure 3S**), and the particle size reduced further to 82 ± 5 nm ($N = 24$). Based on this evidence, we suggest that the synthetic lipid vesicles studied in this work undergo global morphological changes upon sorbitol addition that broadly involves vesicle compaction coupled with membrane undulations, direct membrane damage, and the formation of pores.

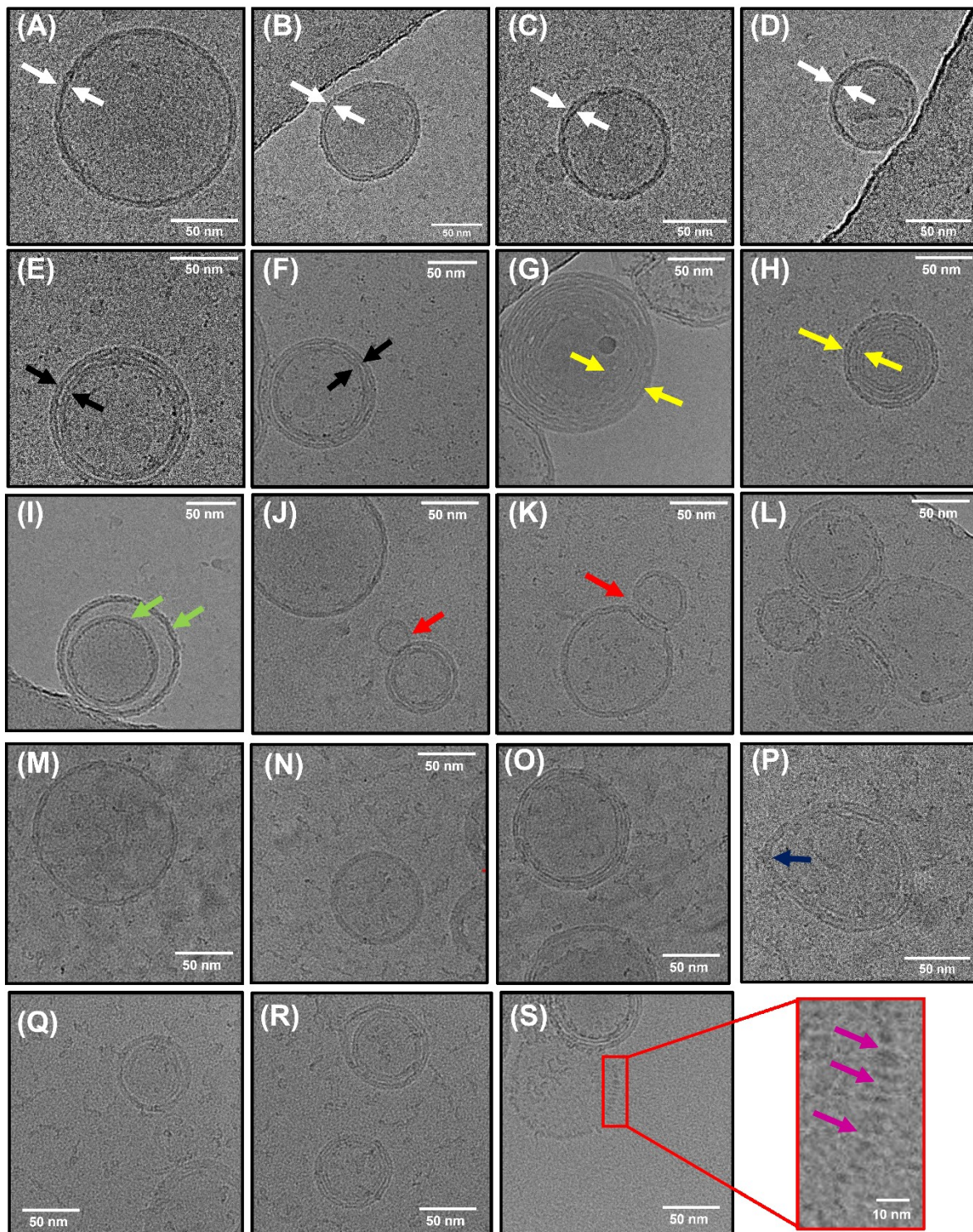


Figure 3. Morphological characterization of lipid vesicles by Cryo-TEM. Representative examples of (A-D) unilamellar (white arrows), (E-F) bilamellar (black arrows) and (G-H) multilamellar (yellow arrows) vesicles in the absence of sorbitol. Also shown are examples of (I) encapsulated vesicles (green arrows), (J-K) bowling pin shaped vesicles and (L) large conglomerates. (M-O) Representative examples of vesicles in the presence of 0.5 M sorbitol, and (P) example of vesicle displaying irregular bulging. (Q-R) Representative examples of vesicles in the presence of 1 M sorbitol with (S) example of broken membrane indicative of regions of broken membrane (purple arrows).

Sorbitol-induced compaction of surface-tethered vesicles is irreversible

To further visualize the observed compaction events, we next evaluated the mean FRET response from single surface-tethered vesicles under crowding conditions via a custom-modified wide-field objective-type total internal reflection fluorescence microscope⁴⁷. Prior to extrusion, the lipid suspension contained 1 mol % of biotinylated lipids (Biotin-PE), allowing formed vesicles to be tethered to a glass substrate via NeutrAvidin⁹. The application of picomolar vesicle concentrations to a NeutrAvidin coated surface led to the detection of 195 ± 14 FRET-active vesicles per $25 \times 50 \mu\text{m}$, with a mean nearest-neighbour vesicle separation distance of $\sim 1 \mu\text{m}$ (**Figure 4A**). In the absence of sorbitol, the vesicles displayed fluorescence across donor and acceptor emission channels, indicative of substantial FRET due to their close proximity. Dil and DiD intensity distributions obtained from $N = 1,566$ foci displayed lognormal behaviour (**Figure 4B**), which likely represents a distribution of vesicle sizes on the surface, consistent with our EM analyses^{48,49}. As sorbitol was progressively added, we then recorded variations in the FRET efficiency per vesicle via changes to the Dil and DiD emission. Specifically, we observed a progressive increase in sensitized acceptor emission as crowding was increased because of enhanced FRET between the dyes (**Figure 4C**). During the titration, the number of foci per field of view and the mean total fluorescence intensity per vesicle defined as the sum of Dil and DiD emission intensities remained largely unchanged, providing confidence that sorbitol addition left vesicles intact on the surface. In the absence of sorbitol, the FRET efficiency distribution displayed Gaussian behaviour and was centred on 0.45, corresponding to a mean Dil-DiD separation distance of 5.4 nm. With increasing levels of sorbitol, the peak of the distribution then shifted by 15 % to 0.53 at 2.5 M, corresponding to a reduction in the mean Dil-DiD separation (**Figure 4D**). Given the total intensity of the foci remained largely invariant upon sorbitol addition, the positive shift in the FRET efficiency population distributions observed as sorbitol was added (**Figure 4E**) could not therefore be attributed to lipid loss or photophysical artefacts, but rather to structural alterations within single vesicles where the mean donor-acceptor separation distance, $\langle d \rangle$ progressively reduced. The crowding-induced morphological changes, observed here are thus consistent with compaction and the data is broadly complementary to previous observations where encapsulated molecular crowders led to vesicle bulging³². In our case the progressive addition of sorbitol led to instantaneous morphological changes on our measurement timescale. Furthermore, we observed that the compaction by sorbitol was irreversible. As shown in the top panel of **Figure 4F**, the vesicles were initially prepared in the absence of sorbitol and had a FRET distribution centred on 0.42. After incubation with 2.5 M sorbitol, a positive shift in the FRET distribution was observed (**Figure 4F, middle panel**), however, after rinsing the flow cell with buffer, the distribution post-sorbitol addition remained unchanged (**Figure 4F, bottom panel**). This observation rules out an excluded volume effect as the underlying cause of the compaction since freely-diffusing sorbitol was removed from solution. Instead, this observation points towards a situation where after interaction of the membrane with sorbitol, the phase transition temperature is irreversibly depressed in a dose-dependent manner via an interaction that we speculate could involve sorbitol interactions with the functional phosphate groups on the lipids.

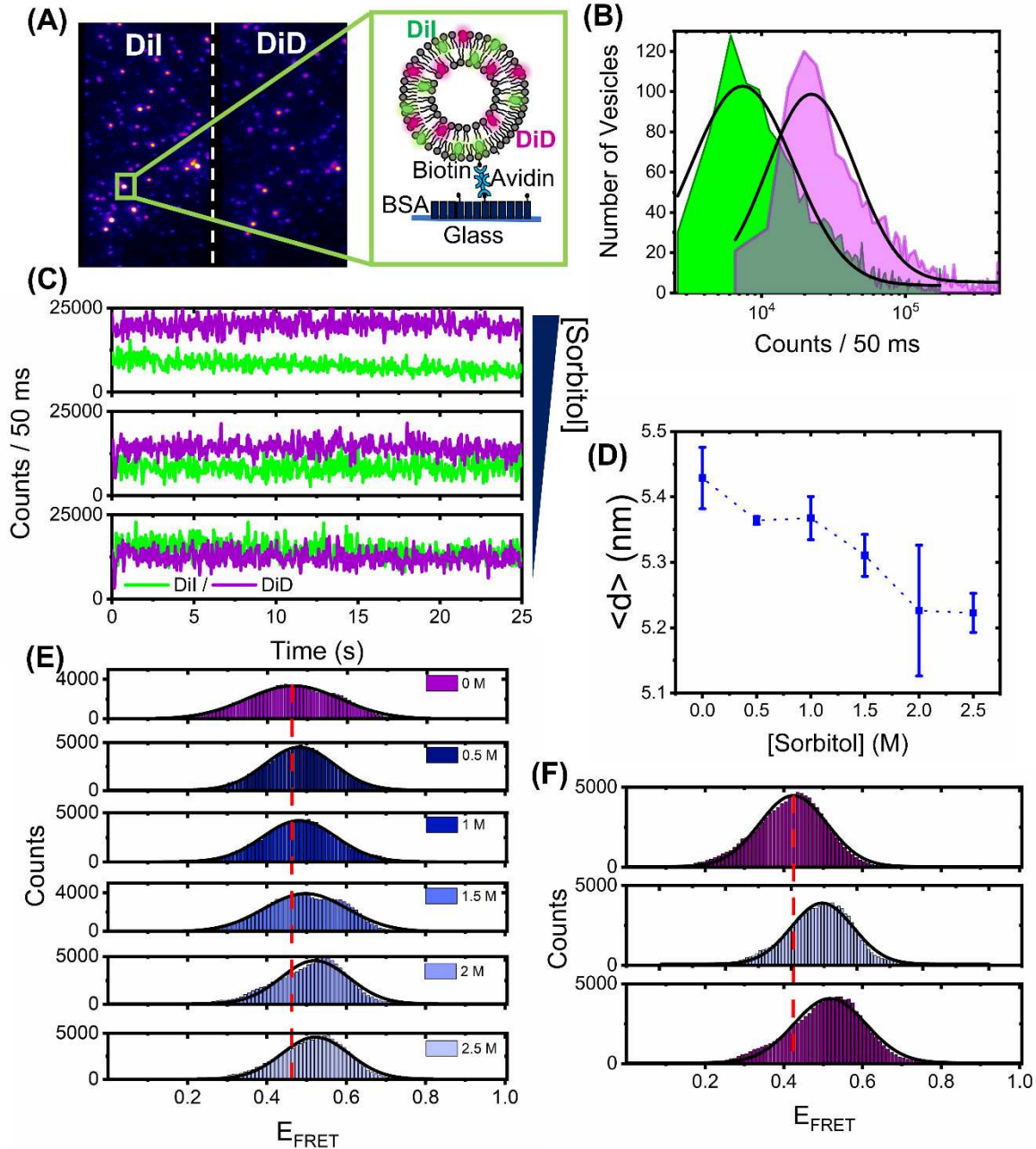


Figure 4. Sorbitol induces irreversible structural changes in single surface-tethered vesicles. (A) Representative wide-field TIRF image of surface tethered vesicles composed of Dil and DiD. Donor and acceptor emission channels are shown on the left- and right-hand side of the dashed line, respectively. Inset: surface immobilization scheme. Single vesicles containing biotinylated lipids are immobilized onto a BSA-Biotin coated glass coverslip via NeutrAvidin. (B) Fluorescence intensity population histograms of Dil (green) and DiD (magenta) obtained from surface tethered vesicles in the absence of sorbitol. (C) Representative time traces of Dil (green) and DiD (magenta) obtained from single surface tethered vesicles with 0 mM, 1 M and 3 M sorbitol. (D) Representative variation in the peak probe separation distance, $\langle d \rangle$, as a function of sorbitol concentration. Also shown are fitting errors associated with the application of single Gaussian distributions to histograms of the probe-separation distance. (E) Corresponding variations in the FRET efficiency histograms obtained for $N > 2,000$ vesicles. (F) FRET efficiency histograms obtained from $N > 2,000$ vesicles at 0 mM sorbitol, after a 3 M sorbitol rinse step (middle panel), and after vigorous washing of the sample with imaging buffer (lower

panel). The dashed red lines in (E) and (F) correspond to the peak positions of the FRET efficiency histograms obtained in the absence of sorbitol. Solid black lines represent single Gaussian fits.

Having established the DiI-DiD FRET pair as a sensitive indicator of vesicle compaction, we next explored the effect of higher molecular weight crowding agents known to induce excluded volume effects on biomolecular morphology. Here, we evaluated the fluorescence response of surface-immobilized vesicles labelled with DiI and DiD in response to variations of PEG and Ficoll. As reported previously, macromolecular crowding by PEG and Ficoll, even at modest concentrations in solution, can trigger substantial excluded volume effects, regulate mesoscale biological functions and impact the conformations of single biomolecules^{15, 50-53}. We therefore hypothesised that high molecular weight crowders in the extravascular space could influence the structural integrity of single lipid vesicles. By using the same immobilization strategy and conditions as described previously we imaged the vesicles under low-excitation TIRF conditions with 50 ms time integration before and after addition of molecular crowders. In the absence of crowding agents, the FRET efficiency distributions obtained from $N > 2000$ vesicles displayed Gaussian behaviour centred on 0.4 corresponding to $\langle d \rangle = 5.7$ nm (**Figure 5 A-D**). When 5 – 20 % (w/w) of the low molecular weight crowder PEG 200 (a 200 Da grade of PEG) was injected, the distributions remained largely invariant, indicative of little-to-no effect on vesicle morphology (**Figure 5A**). However, when similar experiments were performed with the highly branched Ficoll 400 and linear PEG 400, both of which represent 400 Da crowding agents, we observed positive shifts in the population histograms, with $\langle d \rangle$ decreasing to 5.4 nm at 10-15 % (w/w) crowder (**Figure 5B, C**). When PEG 8000 was introduced, the FRET distribution shifted further, peaking 0.6 at 20 % (w/w), with $\langle d \rangle = 4.9$ nm (**Figure 5D**). On the basis of spherical vesicle morphologies, the change in inter-dye distance observed from the vesicles in the presence of Ficoll 400, PEG 400 and PEG 8000 scales directly with the vesicle radius, and the change is thus supportive of compaction. The maximum FRET efficiency shift observed across the titration displayed a dependence on crowder size in the order PEG 8000 > PEG 400 > Ficoll 400 > PEG 200 (**Figure 5E**), suggesting that high molecular weight polymers are more effective at inducing this morphological change. Although the exact nature of these variations requires further investigation, our data suggests that the DiI-DiD FRET pair is sensitive to the three-dimensional structure of the vesicle, and that changes to the FRET signal may arise from an excluded volume effect, with the higher molecular weight polymers leading to more pronounced changes in morphology. A particularly striking observation was that in the cases of PEG 400 and PEG 8000, the FRET efficiency distributions recovered to their original state after the vesicles were thoroughly washed with buffer solution (**Figure 5F**), indicating reversibility. Unlike in the case of sorbitol where we speculate that direct sorbitol-lipid interactions lead to membrane dehydration and irreversible compaction, here we speculate crowding arises primarily through an excluded volume effect. By removing the crowding agents from solution, we effectively remove this influence, and thereby allow the vesicles to recover to their original form, as schematically shown in **Figure 5G**.

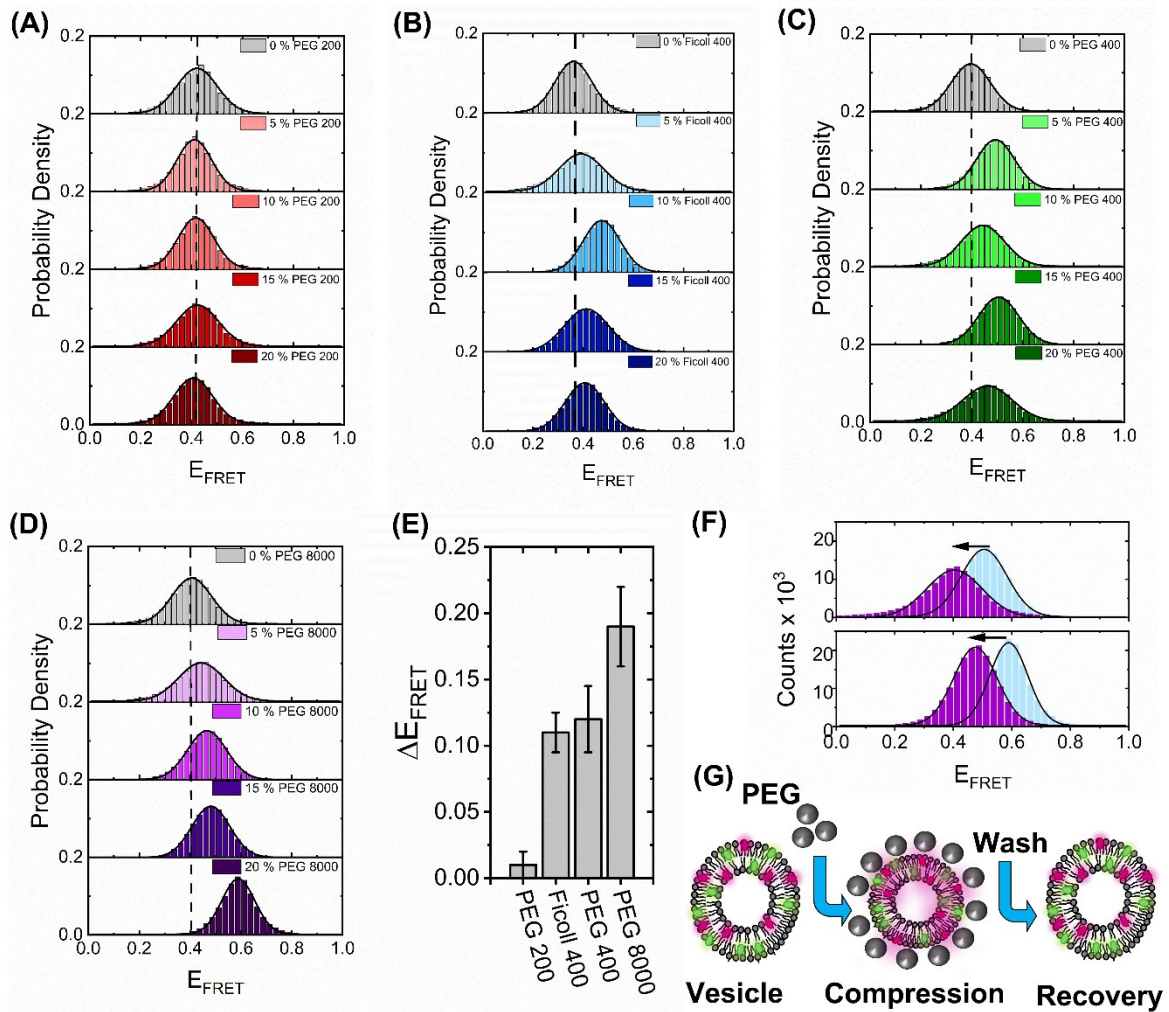


Figure 5. High molecular weight crowders induce reversible vesicle compaction. Representative variations in the FRET efficiency histograms obtained for $N > 2,000$ vesicles in the absence and presence of (A) PEG 200, (B) Ficoll 400 and (C) PEG 400 and (D) PEG 8000 at 5 %, 10 %, 15 % and 20 % (w/w) in 50 mM Tris buffer (pH 8). Dashed lines correspond to the peak positions of the FRET efficiency histograms obtained in the absence of crowder. Solid black lines represent single Gaussian fits. (E) Comparative bar plot summarizing the maximum variation in FRET efficiency observed under different crowding conditions. (F) Representative FRET efficiency histograms obtained from $N > 2,000$ vesicles in the presence of PEG400 (blue, top panel) and PEG8000 (blue, lower panel) and after vigorous washing with 50 mM Tris (pH 8) buffer (purple). (G) Injection of high molecular weight PEG crowders leads to reversible vesicle compaction.

A direct quantitative comparison between the effects of molecular crowding seen here on highly curved LUVs and previous work on GUVs is not entirely straightforward. For instance, the optical imaging of GUVs encapsulating molecular crowders as performed previously³² only reports on macroscopic changes from a cross-sectional slice across the vesicles. In contrast, the smFRET approach described here provides access to the mean dye separation distance across the entire three-dimensional volume of the vesicle. Nevertheless, both sets of vesicles undergo substantial morphological changes in response to molecular crowders at similar concentrations, and thus our smFRET work on sub-micron sized vesicles is complementary to the optical imaging of GUVs, with both datasets suggesting membrane conformation is strongly influenced by molecular crowding. While the smFRET approach provides details of vesicle perturbation on the nanoscale, a limitation is that it does

not quantitatively report on pore formation or solution exchange across the bilayer. Recent studies on GUVs have provided evidence to support the transient formation of nanoscale pores induced by osmotic pressure and variations in membrane tension⁵⁴⁻⁵⁶. While we obtained limited evidence for the formation of pores induced by sorbitol (**Figure 3S**), whether this is also true for the sub-micron sized vesicles in response to PEG-based crowders remains an open question. We expect the smFRET approach to play a key role in identifying conformational changes within single vesicles in response to a wide variety of complex crowding conditions, including various solvents, solutes of varying molecular weight, composition, concentration, temperature, pH and viscosity.

Conclusions

In this work, we characterized the response of synthetic sub-micron sized vesicles to environmental crowding conditions and found that both sorbitol, and polymers of PEG and Ficoll influence their lipid packing density and size, even at modest concentrations of crowder. We conclude that molecular crowding leads to a global compaction of the intact vesicle structure, with our hypothesis supported by the direct observation of single vesicles by electron microscopy. Given the variability of molecular crowding *in vivo*, we subjected the vesicles to a range of molecular weight crowders and unlike the effects of crowding on nucleic acids, which indicate smaller molecules induce more pronounced crowding effects^{57,58}, our observations point towards an intriguing size dependence. On the one hand, we observed that crowding by the low-molecular weight sorbitol leads to permanent vesicle compaction via a biophysical mechanism that likely involves integration of the crowder into the vesicle bilayer and membrane dehydration. In contrast, crowding by longer polymers of PEG and Ficoll induced reversible vesicle compaction, likely via an excluded volume effect^{23,59-61}, with the observed structural changes found to be more pronounced with higher molecular weight crowding agents. It should, however, be noted that these observations were made exclusively on phosphocholine-based vesicles, and future work is required to disentangle the influence of lipid composition and cholesterol content. Whether the presented FRET sensor can also be adapted to sense morphological changes to extracellular vesicles *in vivo* has yet to be explored, but we note that morphological compaction has been observed here in synthetic vesicles with similar compositions, sizes and structures.

Understanding the structural stability of sub-micron sized vesicles, especially those with high curvature, and how they dynamically respond to molecular crowders has a number of implications. First, important trafficking pathways between the endoplasmic reticulum rely on the formation of highly curved vesicles, and regulation of extramembrane crowding may be one mechanism through which vesicles alter morphology for vital processes including endocytosis and vesicle budding⁶²⁻⁶⁴. More generally, crowding-induced changes to local membrane curvature could also play a role in the shaping of organelles, and the triggering of adaptive cellular responses such as gene expression⁶⁵. Second, our measurements indicate that extramembrane crowding leads to variations in the lipid packing density which does not always guarantee stability⁶⁶. In the context of vesicles as drug delivery vehicles, it is therefore possible that molecular crowding influences the efficiency at which encapsulated drug molecules are released and therefore the fate of the targeted cell. Finally, the ability to precisely control vesicle morphology is highly desirable for a number of biotechnological applications⁶⁷. For instance, we foresee that the ability to alter vesicle curvature *in vitro* could open a platform for investigating crowding effects on membrane protein signalling, and for evaluating encapsulation efficiencies in more physiologically relevant conditions.

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Data accessibility

Data associated with this study is freely available from DOI:10.5281/zenodo.7081992.

References

1. Chan, Y. H. M.; Boxer, S. G. Model Membrane Systems and Their Applications. *Current Opinion in Chemical Biology* **2007**, *11* (6), 581-587.
2. Zong, W.; Shao, X. T.; Chai, Y. H.; Wang, X. W.; Han, S.; Chu, H. T.; Zhu, C. T.; Zhang, X. A. Liposomes Encapsulating Artificial Cytosol as Drug Delivery System. *Biophys Chem* **2022**, *281*.
3. Torchilin, V. P. Recent Advances with Liposomes as Pharmaceutical Carriers. *Nat Rev Drug Discov* **2005**, *4* (2), 145-160.
4. Okumus, B.; Arslan, S.; Fengler, S. M.; Myong, S.; Ha, T. Single Molecule Nanocontainers Made Porous Using a Bacterial Toxin. *J Am Chem Soc* **2009**, *131* (41), 14844-14849.
5. Stengel, G.; Zahn, R.; Hook, F. DNA-Induced Programmable Fusion of Phospholipid Vesicles (Vol 129, Pg 9584, 2007). *J Am Chem Soc* **2008**, *130* (7), 2372-2372.
6. Mora, N. L.; Boyle, A. L.; van Kolck, B. J.; Rossen, A.; Pokorna, S.; Koukalova, A.; Sachl, R.; Hof, M.; Kros, A. Controlled Peptide-Mediated Vesicle Fusion Assessed by Simultaneous Dual-Colour Time-Lapsed Fluorescence Microscopy. *Sci Rep-Uk* **2020**, *10* (1).
7. Diao, J. J.; Ishitsuka, Y.; Lee, H.; Joo, C.; Su, Z. L.; Syed, S.; Shin, Y. K.; Yoon, T. Y.; Ha, T. A Single Vesicle-Vesicle Fusion Assay for in Vitro Studies of Snares and Accessory Proteins. *Nat Protoc* **2012**, *7* (5), 921-934.
8. Dalgarno, P. A.; Juan-Colas, J.; Hedley, G. J.; Pineiro, L.; Novo, M.; Perez-Gonzalez, C.; Samuel, I. D. W.; Leake, M. C.; Johnson, S.; Al-Soufi, W.; Penedo, J. C.; Quinn, S. D. Unveiling the Multi-Step Solubilization Mechanism of Sub-Micron Size Vesicles by Detergents. *Sci Rep-Uk* **2019**, *9*.
9. Juan-Colas, J.; Dresser, L.; Morris, K.; Lagadou, H.; Ward, R. H.; Burns, A.; Tear, S.; Johnson, S.; Leake, M. C.; Quinn, S. D. The Mechanism of Vesicle Solubilization by the Detergent Sodium Dodecyl Sulfate. *Langmuir* **2020**, *36* (39), 11499-11507.
10. Anandan, A.; Vrieling, A. Detergents in Membrane Protein Purification and Crystallisation. *Adv Exp Med Biol* **2016**, *922*, 13-28.
11. Model, M. A.; Hollembeak, J. E.; Kurokawa, M. Macromolecular Crowding: A Hidden Link between Cell Volume and Everything Else. *Cell Physiol Biochem* **2021**, *55* (S1), 25-40.
12. Zimmerman, S. B.; Minton, A. P. Macromolecular Crowding - Biochemical, Biophysical, and Physiological Consequences. *Annu Rev Bioph Biom* **1993**, *22*, 27-65.
13. Neuweiler, H.; Lollmann, M.; Doose, S.; Sauer, M. Dynamics of Unfolded Polypeptide Chains in Crowded Environment Studied by Fluorescence Correlation Spectroscopy. *J Mol Biol* **2007**, *365* (3), 856-869.
14. Soranno, A.; Koenig, I.; Borgia, M. B.; Hofmann, H.; Zosel, F.; Nettels, D.; Schuler, B. Single-Molecule Spectroscopy Reveals Polymer Effects of Disordered Proteins in Crowded Environments. *P Natl Acad Sci USA* **2014**, *111* (13), 4874-4879.
15. Baltierra-Jasso, L. E.; Morten, M. J.; Laflor, L.; Quinn, S. D.; Magennis, S. W. Crowding-Induced Hybridization of Single DNA Hairpins. *J Am Chem Soc* **2015**, *137* (51), 16020-16023.
16. Dupuis, N. F.; Holmstrom, E. D.; Nesbitt, D. J. Molecular-Crowding Effects on Single-Molecule Rna Folding/Unfolding Thermodynamics and Kinetics. *P Natl Acad Sci USA* **2014**, *111* (23), 8464-8469.
17. Holmstrom, E. D.; Dupuis, N. F.; Nesbitt, D. J. Kinetic and Thermodynamic Origins of Osmolyte-Influenced Nucleic Acid Folding. *J Phys Chem B* **2015**, *119* (9), 3687-3696.

18. Minton, A. P. Models for Excluded Volume Interaction between an Unfolded Protein and Rigid Macromolecular Cosolutes: Macromolecular Crowding and Protein Stability Revisited. *Biophys J* **2005**, *88* (2), 971-985.
19. Sarkar, M.; Smith, A. E.; Pielak, G. J. Impact of Reconstituted Cytosol on Protein Stability. *P Natl Acad Sci USA* **2013**, *110* (48), 19342-19347.
20. Dix, J. A.; Verkman, A. S. Crowding Effects on Diffusion in Solutions and Cells. *Annu Rev Biophys* **2008**, *37*, 247-263.
21. Keating, C. D. Aqueous Phase Separation as a Possible Route to Compartmentalization of Biological Molecules. *Accounts Chem Res* **2012**, *45* (12), 2114-2124.
22. van den Berg, J.; Boersma, A. J.; Poolman, B. Microorganisms Maintain Crowding Homeostasis. *Nat Rev Microbiol* **2017**, *15* (5), 309-318.
23. Sung, H. L.; Sengupta, A.; Nesbitt, D. Smaller Molecules Crowd Better: Crowder Size Dependence Revealed by Single-Molecule FRET Studies and Depletion Force Modeling Analysis. *J Chem Phys* **2021**, *154* (15).
24. Culik, R. M.; Abaskharon, R. M.; Pazos, I. M.; Gai, F. Experimental Validation of the Role of Trifluoroethanol as a Nanocrowder. *J Phys Chem B* **2014**, *118* (39), 11455-11461.
25. Money, N. P. Osmotic Pressure of Aqueous Polyethylene Glycols : Relationship between Molecular Weight and Vapor Pressure Deficit. *Plant Physiol* **1989**, *91* (2), 766-9.
26. Parsegian, V. A.; Rand, R. P.; Rau, D. C. Osmotic Stress, Crowding, Preferential Hydration, and Binding: A Comparison of Perspectives. *P Natl Acad Sci USA* **2000**, *97* (8), 3987-3992.
27. Lentz, B. R. Polymer-Induced Membrane-Fusion - Potential Mechanism and Relation to Cell-Fusion Events. *Chem Phys Lipids* **1994**, *73* (1-2), 91-106.
28. Kuhl, T.; Guo, Y. Q.; Alderfer, J. L.; Berman, A. D.; Leckband, D.; Israelachvili, J.; Hui, S. W. Direct Measurement of Polyethylene Glycol Induced Depletion Attraction between Lipid Bilayers. *Langmuir* **1996**, *12* (12), 3003-3014.
29. Chen, I. A.; Roberts, R. W.; Szostak, J. W. The Emergence of Competition between Model Protocells. *Science* **2004**, *305* (5689), 1474-1476.
30. Terasawa, H.; Nishimura, K.; Suzuki, H.; Matsuura, T.; Yomo, T. Coupling of the Fusion and Budding of Giant Phospholipid Vesicles Containing Macromolecules. *P Natl Acad Sci USA* **2012**, *109* (16), 5942-5947.
31. Lehtonen, J. Y. A.; Kinnunen, P. K. J. Poly(Ethylene Glycol)-Induced and Temperature-Dependent Phase-Separation in Fluid Binary Phospholipid-Membranes. *Biophys J* **1995**, *68* (2), 525-535.
32. Su, W. C.; Gettel, D. L.; Chabanon, M.; Rangamani, P.; Parikh, A. N. Pulsatile Gating of Giant Vesicles Containing Macromolecular Crowding Agents Induced by Colligative Nonideality. *J Am Chem Soc* **2018**, *140* (2), 691-699.
33. Cherstvy, A. G.; Petrov, E. P. Modeling DNA condensation on freestanding cationic lipid membranes. *Phys. Chem. Chem. Phys.* **2013**, *16*, 2020-2037.
34. Yadav, J. K. Macromolecular Crowding Enhances Catalytic Efficiency and Stability of Alpha-Amylase. *ISRN Biotechnol* **2013**, *2013*, 737805.
35. Colom, A.; Derivery, E.; Soleimanpour, S.; Tomba, C.; Molin, M. D.; Sakai, N.; Gonzalez-Gaitan, M.; Matile, S.; Roux, A. A Fluorescent Membrane Tension Probe. *Nat Chem* **2018**, *10* (11), 1118-1125.
36. Stetefeld, J.; McKenna, S. A.; Patel, T. R. Dynamic Light Scattering: A Practical Guide and Applications in Biomedical Sciences. *Biophys Rev* **2016**, *8* (4), 409-427.
37. Conteduca, D.; Quinn, S. D.; Krauss, T. F. Dielectric Metasurface for High-Precision Detection of Large Unilamellar Vesicles. *J Optics-Uk* **2021**, *23* (11).
38. Dresser, L.; Hunter, P.; Yendybayeva, F.; Hargreaves, A. L.; Howard, J. A. L.; Evans, G. J. O.; Leake, M. C.; Quinn, S. D. Amyloid-Beta Oligomerization Monitored by Single-Molecule Stepwise Photobleaching. *Methods* **2021**, *193*, 80-95.
39. Preus, S.; Noer, S. L.; Hildebrandt, L. L.; Gudnason, D.; Birkedal, V. Isms: Single-Molecule FRET Microscopy Software. *Nat Methods* **2015**, *12* (7), 593-4.

40. Garenne, D.; Libchaber, A.; Noireaux, V. Membrane Molecular Crowding Enhances MreB Polymerization to Shape Synthetic Cells from Spheres to Rods. *P Natl Acad Sci USA* **2020**, *117* (4), 1902-1909.
41. Khodadadi, S.; Clark, N. J.; McAuley, A.; Cristiglio, V.; Curtis, J. E.; Shalaev, E. Y.; Krueger, S. Influence of Sorbitol on Protein Crowding in Solution and Freeze-Concentrated Phases. *Soft Matter* **2014**, *10* (23), 4056-4060.
42. Sukenik, S.; Sapir, L.; Harries, D. Osmolyte Induced Changes in Peptide Conformational Ensemble Correlate with Slower Amyloid Aggregation: A Coarse-Grained Simulation Study. *J Chem Theory Comput* **2015**, *11* (12), 5918-5928.
43. Sukenik, S.; Politi, R.; Ziserman, L.; Danino, D.; Friedler, A.; Harries, D. Crowding Alone Cannot Account for Cosolute Effect on Amyloid Aggregation. *Plos One* **2011**, *6* (1).
44. Richter, K.; Nessling, M.; Lichter, P. Experimental Evidence for the Influence of Molecular Crowding on Nuclear Architecture. *J Cell Sci* **2007**, *120* (9), 1673-1680.
45. Drab, M.; Pandur, Z.; Penic, S.; Iglic, A.; Kralj-Iglic, V.; Stopar, D. A Monte Carlo Study of Giant Vesicle Morphologies in Nonequilibrium Environments. *Biophys J* **2021**, *120* (20), 4418-4428.
46. Jurkiewicz, P.; Cwiklik, L.; Vojtiskova, A.; Jungwirth, P.; Hof, M. Structure, Dynamics, and Hydration of Popc/Pops Bilayers Suspended in Nacl, Kcl, and Cscl Solutions. *Bba-Biomembranes* **2012**, *1818* (3), 609-616.
47. Emelyanov, A.; Shtam, T.; Kamyshinsky R.; Garaeva, L.; Verlov, N.; Miliukhina, I.; Kudrevatykh, A.; Gavrillov, G.; Zbrodskaya, Y.; Pchelina, S.; Konevega, A. Cryo-electron microscopy of extracellular vesicles from cerebrospinal fluid. *PLOS One*. **2020**, *15*, e0227949.
48. Kunding, A. H.; Mortensen, M. W.; Christensen, S. M.; Stamou, D. A Fluorescence-Based Technique to Construct Size Distributions from Single-Object Measurements: Application to the Extrusion of Lipid Vesicles. *Biophys J* **2008**, *95* (3), 1176-1188.
49. Balomenos, A. D.; Stefanou, V.; Manolagos, E. S. Analytics and Visualization Tools to Characterize Single-Cell Stochasticity Using Bacterial Single-Cell Movie Cytometry Data. *Bmc Bioinformatics* **2021**, *22* (1).
50. Kaur, T.; Alshareedah, I.; Wang, W.; Ngo, J.; Moosa, M. M.; Banerjee, P. R. Molecular Crowding Tunes Material States of Ribonucleoprotein Condensates. *Biomolecules* **2019**, *9* (2).
51. Paudel, B. P.; Fiorini, E.; Borner, R.; Sigel, R. K. O.; Rueda, D. S. Optimal Molecular Crowding Accelerates Group II Intron Folding and Maximizes Catalysis. *P Natl Acad Sci USA* **2018**, *115* (47), 11917-11922.
52. Banerjee, P. R.; Moosa, M. M.; Deniz, A. A. Two-Dimensional Crowding Uncovers a Hidden Conformation of Alpha-Synuclein. *Angew Chem Int Edit* **2016**, *55* (41), 12789-12792.
53. Junker, N. O.; Vaghefikia, F.; Albarghash, A.; Hofig, H.; Kempe, D.; Walter, J.; Otten, J.; Pohl, M.; Katranidis, A.; Wiegand, S.; Fitter, J. Impact of Molecular Crowding on Translational Mobility and Conformational Properties of Biological Macromolecules. *J Phys Chem B* **2019**, *123* (21), 4477-4486.
54. Billah, M.; Saha, S. K.; Rashid, M. O.; Hossain, F.; Yamazaki, M. Effect of osmotic pressure on pore formation in lipid bilayers by the antimicrobial peptide magainin 2. *Phys. Chem. Chem. Phys.* **2022**, *24*, 6716-6731.
55. Shibly, S. U. A.; Ghatak, C.; Karal, M. A. S.; Moniruzzaman, M.; Yamazaki, M. Experimental estimation of membrane tension induced by osmotic pressure. *Biophys. J.* **2016**, *111*, 2190-2201.
56. Malik, V. K.; Pak, O. S.; Feng, J. Pore dynamics of lipid vesicles under light-induced osmotic stress. *Phys. Rev. Applied.* **2022**, *17*, 024032.
57. Sung, H.; Sengupta, A.; Nesbitt, D. Smaller molecules crowd better: crowder size dependence revealed by single-molecule FRET studies and depletion force modelling analysis. *J. Chem. Phys.* **2021**, 155101.
58. Sharp, K. A. Analysis of the size dependence of macromolecular crowding shows that smaller is better. *Proc. Natl. Acad. Sci.* **2015**, *112*, 7990-7995.
59. Kilburn, D.; Roh, J. H.; Guo, L.; Briber, R. M.; Woodson, S. A. Molecular Crowding Stabilizes Folded Rna Structure by the Excluded Volume Effect. *J Am Chem Soc* **2010**, *132* (25), 8690-8696.

60. Minton, A. P. Excluded Volume as a Determinant of Macromolecular Structure and Reactivity. *Biopolymers* **1981**, 20 (10), 2093-2120.
61. Minton, A. P. The Influence of Macromolecular Crowding and Macromolecular Confinement on Biochemical Reactions in Physiological Media. *J Biol Chem* **2001**, 276 (14), 10577-10580.
62. Shibata Y.; Hu, J.; Kozlov, M. M.; Rapoport, T. A. Mechanisms shaping the membranes of cellular organelles. *Annu. Rev. Cell Dev. Biol.* **2009**, 25, 329-354.
63. Kozlov, M. M.; Campelo, F.; Liska, N.; Chernomordik, L. V.; Marrink, S. J.; McMahon, H. T. Mechanisms shaping cell membranes. *Curr. Opin. Cell Biol.* **2014**, 29, 53-60.
64. McMahon, H. T.; Boucrot, E. Membrane curvature at a glance. *J. Cell Sci.* **2015**, 128, 1065-1070.
65. Pittman, M.; Lu, E.; Li, K.; Wang, M.; Chen, J.; Taneja, N.; Jo, M. H.; Park, S.; Jung, W.; Liang, L.; Barman, I.; Ha, T.; Gaitanaros, S.; Liu, J.; Burnette, D.; Plotnikov, S.; Chen, Y. Membrane ruffling is a mechanosensory of extracellular fluid viscosity. *Nat. Phys.* **2022**, In press.
66. van der Koog, L.; Gandek, T. B.; Nagelkerke, A. Liposomes and extracellular vesicles as drug delivery systems: a comparison of composition, pharmacokinetics, and functionalization. *Adv. Healthc. Mater.* **2021**, 11, 2100639.
67. Guerzoni, L. P. B.; Goes, A. V. C.; Kalacheva, M.; Haduła, J.; Mork, M.; De Laporte, L.; Boersma, A. J. High macromolecular crowding in liposomes from microfluidics. *Adv. Sci.* **2022**, 2201169.

Crowding induced morphological changes of synthetic lipid vesicles determined using smFRET

Steven D. Quinn^{1,2}, Lara Dresser¹, Sarah Graham¹, Donato Conteduca¹, Jack Shepherd^{1,3} & Mark C. Leake^{1,2,3}

¹School of Physics, Engineering and Technology, University of York, York, UK. YO10 5DD.

²York Biomedical Research Institute, University of York, York, UK. YO10 5DD.

³Department of Biology, University of York, York, UK. YO10 5DD.

Correspondence to: mark.leake@york.ac.uk

SUPPORTING INFORMATION

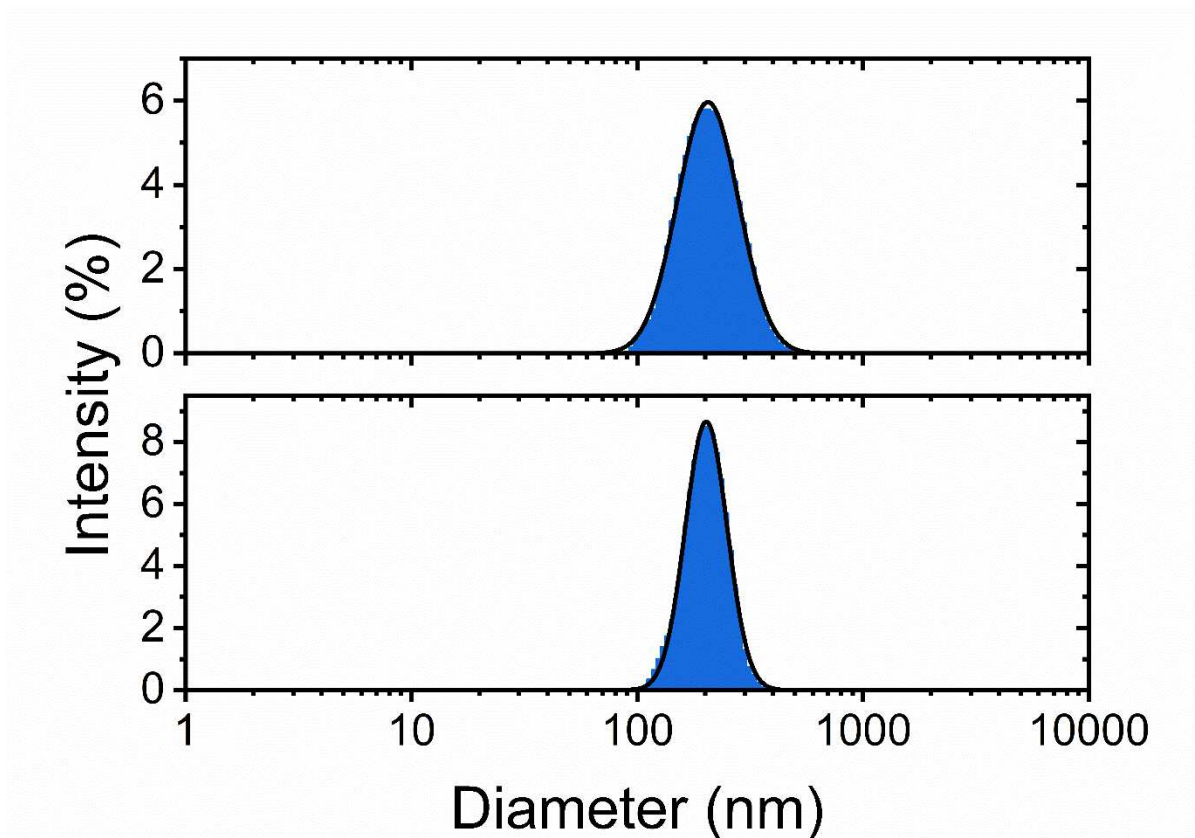


Figure S1. Representative DLS diameter distributions obtained from DiI and DiD loaded POPC vesicles (top panel) and unlabelled vesicles (bottom panel). Solution conditions: 50 mM Tris, pH 8.

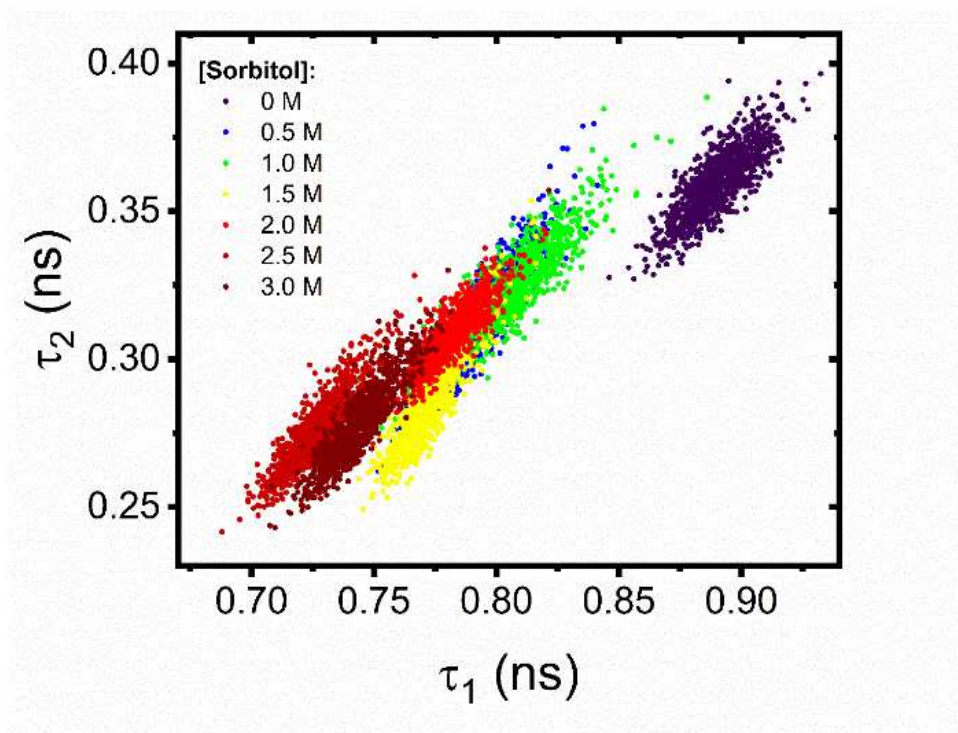


Figure S2. Representative variations in the lifetime components τ_1 and τ_2 extracted from DiI and DiD loaded POPC vesicles in the absence and presence of sorbitol.

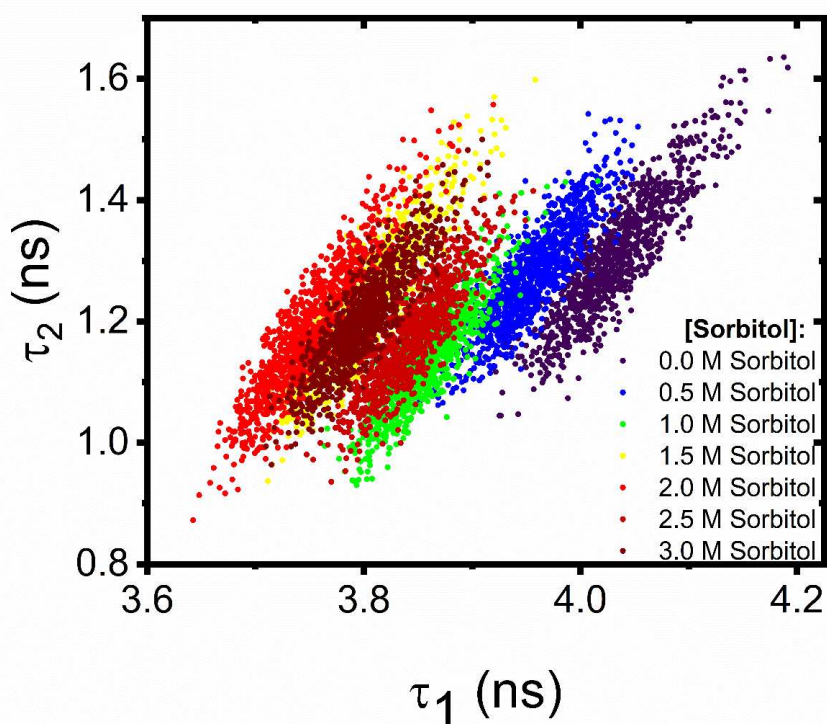


Figure S3. Representative variations in the lifetime components τ_1 and τ_2 extracted from FluptR loaded POPC vesicles in the absence and presence of sorbitol.

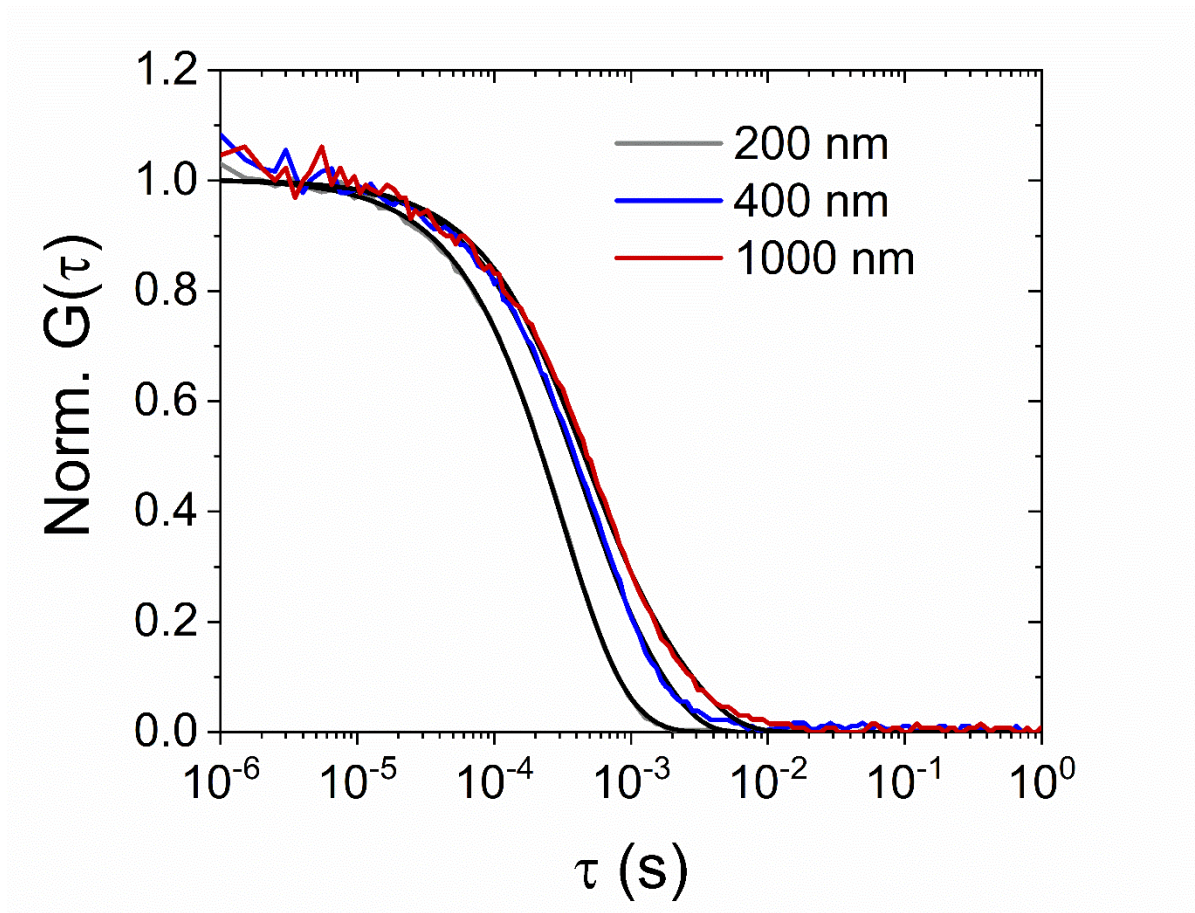


Figure S4. Representative variations in the normalized correlation curves obtained by DLS from DiI and DiD loaded POPC vesicles extruded through 200 nm (gray), 400 nm (blue) and 1000 nm (red) pore diameter polycarbonate membrane filters. Solid black lines represent fits as describe in the Methods Section. Solution conditions: 50 mM Tris, pH 8.

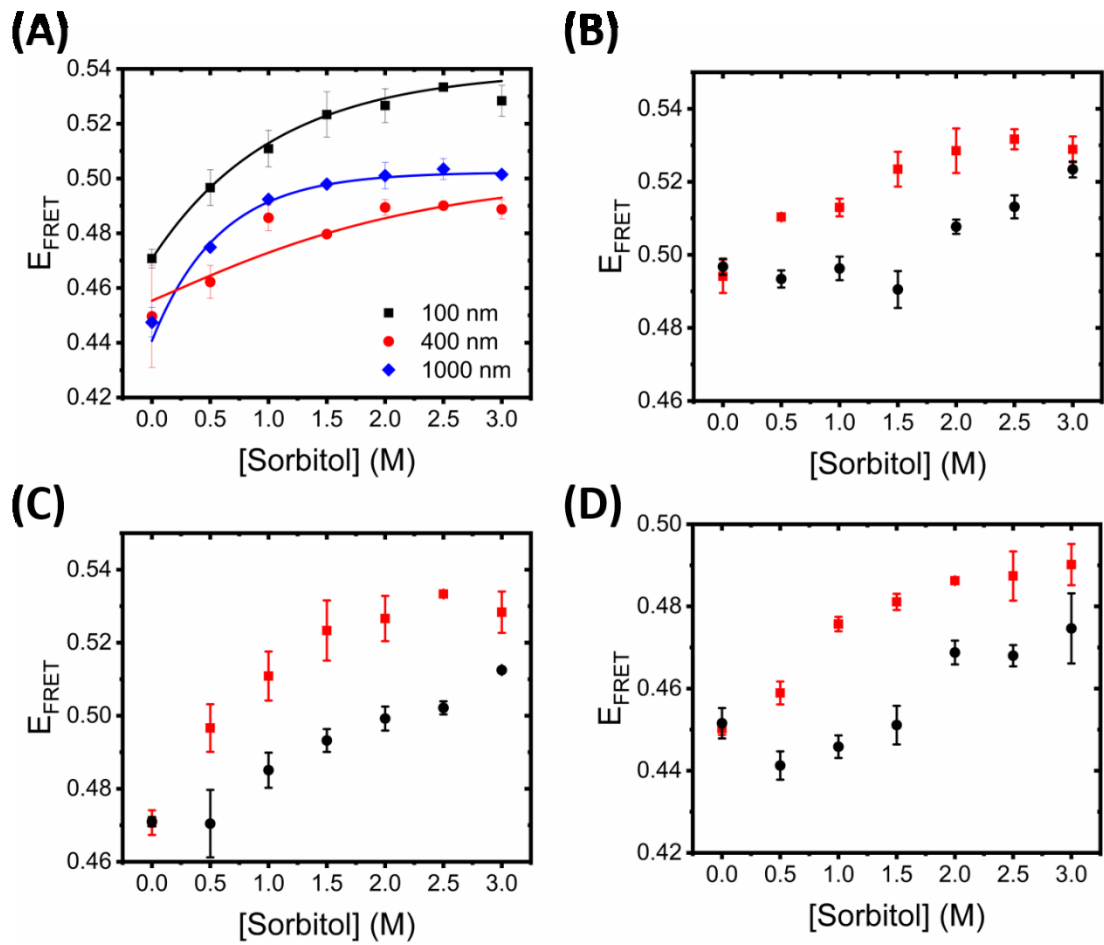


Figure S5. (A) Representative variation in E_{FRET} as a function of vesicle size. Also shown are E_{FRET} variations obtained from 100 nm diameter POPC (red) and POPS (black) vesicles at (B) 4°C, (C) 21°C and (D) 37°C.

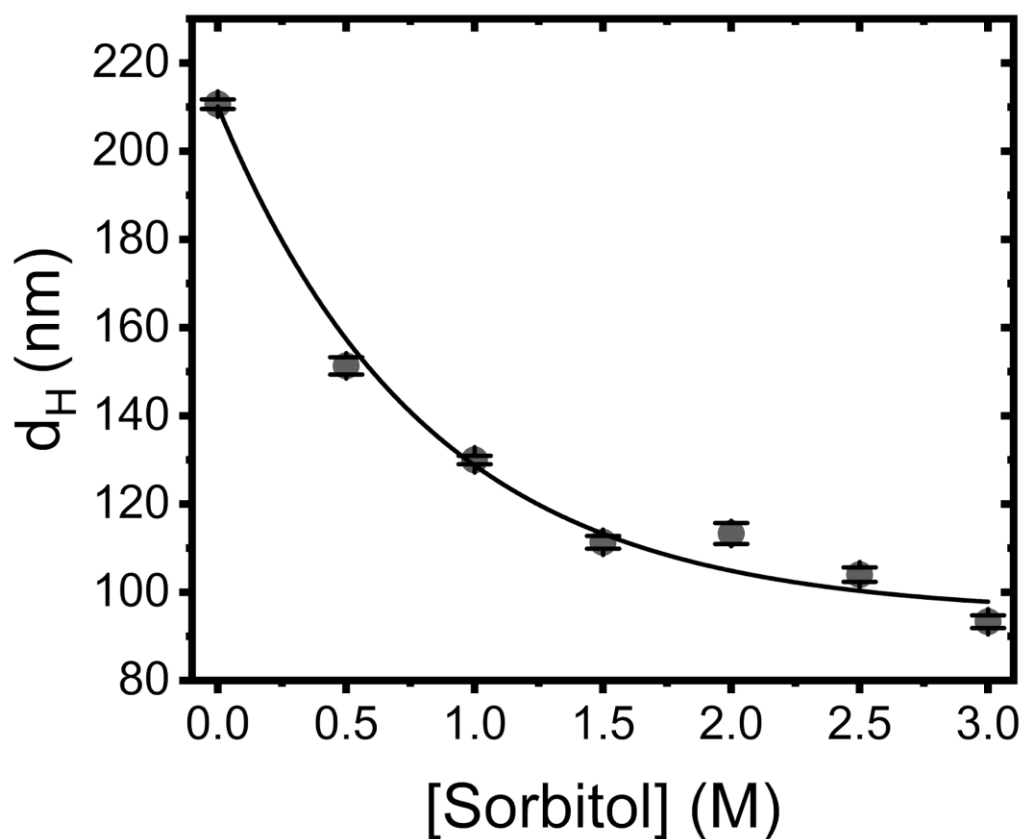


Figure S6. Variation in the mean hydrodynamic diameter of POPC-loaded LUVs under crowding conditions.

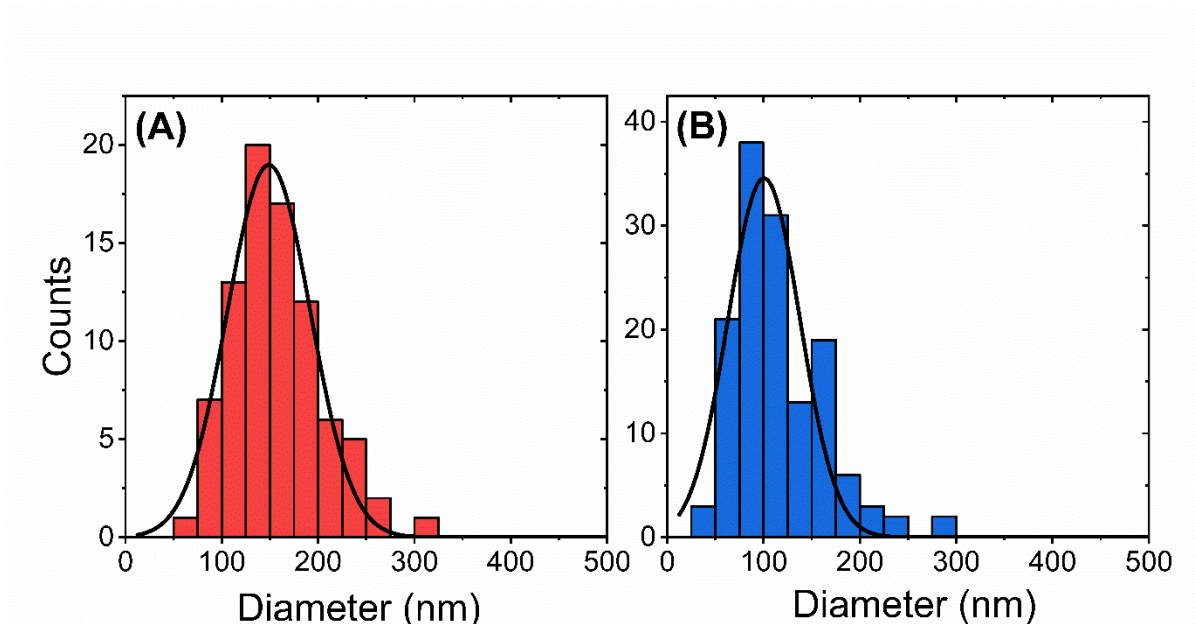


Figure S7. Comparison of vesicle diameter distributions obtained by cryo-TEM in (A) the absence ($N = 84$) and (b) the presence of 0.5 M sorbitol ($N = 138$). Solid black lines correspond to Gaussian fits.

