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1 **Simulation, energy and exergy analysis of compressed air energy storage**
2 **integrated with Organic Rankine Cycle and single effect absorption refrigeration**
3 **for trigeneration application**

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10
11 Abstract

12 Compressed air energy storage (CAES) is increasingly investigated as a viable
13 technology for balancing electricity supply and demand. The main purpose of CAES is
14 to overcome the intermittent problem when renewable energy is introduced. However,
15 the round-trip efficiency (RTE) of the CAES system commercially developed is still
16 low (around 54%) and requires further improvement. This study proposed a novel
17 combined cooling, heating and power (CCHP) system, which improves the RTE in two
18 ways: (i) Organic Rankine Cycle (ORC), which recovers the waste heat and produces
19 extra power; (ii) single effect absorption refrigeration, which delivers cooling capacity
20 with further recovery of waste heat from the flue gas. Steady-state process simulation
21 of the CAES system, ORC and single effect absorption refrigeration system was
22 developed in Aspen Plus® software. The proposed system was evaluated through
23 process, energy, and exergy analysis. The effect of various working parameters on the
24 performance of the CCHP system was also analysed. The results indicated that under

25 design condition, the proposed CCHP system can produce about 206 MW electrical
26 energy, 28 MW heating and 0.2 MW cooling capacity. The RTE of the proposed system
27 (about 66%) showed an improvement of approximately 12% when compared with the
28 CAES system commercially deployed. The overall exergy efficiency is about 51% and
29 the total exergy destruction of the components of the system is 477 MW. The
30 combustion chamber is responsible for more than half of the exergy destruction.

31

32 Keywords

33 Compressed air energy storage (CAES); Organic Rankine Cycle (ORC);

34 Trigenation; Combined cooling, heating and power (CCHP); Process simulation;

35 Exergy analysis

36

37 Nomenclature

Q	Heat transfer (kW)
W	Work (kW)
T	temperature (°C)
h	Specific enthalpy (kJ/kg)
s	Specific entropy (kJ/kg K)
Com	Compressor
Tur	Turbine
Cave	Cavern

HX	Heat exchanger
Val	Valve
Cc	Combustion chamber
pum	Pump
OT	Organic turbine
Des	Desorber
orc-c	ORC in charging process
orc-d	ORC in discharging process
pum-c	pump in charging process
pum-d	ORC pump in discharging process
Evap	Evaporator
Abbreviations	
CAES	Compressed air energy storage
RTE	Round trip efficiency
ORC	Organic Rankine Cycle
GHG	Greenhouse gas
IEA	International Energy Agency
PHS	Pumped hydro storage
SMES	Superconducting magnetic energy storage

CCHP	Combined cooling, heating and power
D-CAES	Diabetic compressed air energy storage
KC	Kalina Cycle
LiBr	Lithium bromide
PENG-ROB	Standard Peng-Robinson cubic equation of state
COP	Coefficient of performance
Abs	Absorber
Aftc	After-cooler
G	Generator
GHG	Greenhouse gas
HPC	High pressure compressor
Intc	Inter-cooler
LPC	Low pressure compressor
M	Motor
Rec	Recuperator
Reg	Regulating valve
SHX	Solution heat exchanger
RTE	Round-trip efficiency
V	Valve

38

39

40

41 1. Introduction

42 1.1. Background

43 In recent decades, the consumption of fossil fuels has caused global environmental
44 problems, due to the production of greenhouse gases (GHG) (Chai et al., 2020; Meng
45 et al., 2019). Hence, renewable energy sources are being utilised to reduce the
46 dependency on fossil fuels and cut CO₂ emissions (Aneke and Wang, 2016). Renewable
47 energy has been growing significantly over the years. According to the 2020 world
48 renewable electricity statistics, released by the International Energy Agency (IEA), the
49 share of all renewables in electricity generation reached 27% (7486 TWh) in 2020 (IEA,
50 2020). However, the main problem with renewable energy is its intermittence,
51 especially for wind power and photovoltaics (Budt et al., 2016; Luo et al., 2015).
52 Furthermore, renewable energy production methods cannot be continuous, and the
53 output depends on geographic location and climate (Meng et al., 2019; Zhao et al.,
54 2016). Researchers have suggested energy storage technologies to solve the
55 intermittence problem (Sadreddini et al., 2018; Aneke and Wang, 2016). Such
56 technologies have numerous benefits, especially in peak shaving, black start, load
57 levelling, renewable integration, power quality improvement and energy arbitrage
58 (Meng et al., 2018; Aneke and Wang, 2016; Luo et al., 2015; Sioshansi et al., 2011;
59 Chen et al., 2009).

60 Many energy storage technologies have been commercialised or are still under research.
61 These include pumped hydro storage (PHS), compressed air energy storage (CAES),
62 batteries, fuel cells, flywheels, superconducting magnetic energy storage (SMES),

63 capacitors and supercapacitors (Letcher, 2020; Meng et al., 2018; Aneke and Wang,
64 2016; Luo et al., 2015; Kousksou et al., 2014; Chen et al., 2009). Among these energy
65 storage technologies, only PHS and CAES are suitable for large grid-scale (>100 MW)
66 systems due to their high energy capacity, high power rating, and long discharge time
67 (Letcher, 2020; Meng et al., 2018). However, PHS power plants require suitable
68 geographical conditions for the construction of reservoirs and dams, with a long
69 construction period and significant initial investment (Letcher, 2020; Meng et al., 2018;
70 Aneke and Wang, 2016; Chen et al., 2009). The huge benefits of CAES technology
71 include high reliability and round-trip efficiency (RTE), providing reserve power and
72 economic feasibility (Budt et al., 2016; Chen et al., 2009).

73 However, the RTE of the CAES system still needs further improvement. Combined
74 cooling, heating and power (CCHP) systems are among the most attractive and
75 promising solutions to reduce energy consumption (Mohammadi et al., 2017;
76 Sadreddini et al., 2018; Razmi et al., 2019). Most conventional CCHP systems are based
77 on gas-fired power plants, which cannot fully realise their energy-saving advantages
78 (Gao et al., 2015; Yan et al., 2018). The integration of CAES, Organic Rankine Cycle
79 (ORC) and a refrigeration system for CCHP application has many advantages: (i) it
80 generates three different products: cooling, heating and electricity; (ii) it improves the
81 efficiency of energy use; (iii) it reduces the emission of air pollutants (Mohammadi et
82 al., 2017; Sadreddini et al., 2018; Razmi et al., 2019).

83 1.2. Literature review

84 To date, there are two large-scale CAES plants in commercial operation. Diabatic
85 compressed air energy storage (D-CAES) is the primary type of CAES plant, and these
86 two CAES plants are of this type (Zhou et al., 2019; Budt et al., 2016). The first is the
87 Huntorf CAES plant, built in Germany and in operation since 1978. The Huntorf CAES

88 plant, with the RTE of 42%, could produce 290 MW electricity for a 2~3h discharging
89 duration (Meng et al., 2018; Chen et al., 2009; Crotogino et al., 2001). The second
90 CAES plant was built in McIntosh, Alabama, USA and has been in operation since 1991.
91 The McIntosh CAES plant could produce 110 MW of electricity for a 26h discharging
92 duration (Meng et al., 2018; Aneke and Wang, 2016; Budt et al., 2016; Luo et al., 2015;
93 Chen et al., 2009). However, the Huntorf CAES plant has unique benefits as it can be
94 operated as reserve power and black-start capacity (Radgen, 2008; Budt et al., 2016).
95 Many studies have focused on the investigation of the CAES system integrated with
96 renewable energy (wind and photovoltaic power) to overcome the intermittence
97 problem (Meng et al., 2019; Zhang et al., 2017; Krupke et al., 2016; Zhang et al., 2014;
98 Cavallo, 2007; Cazzaniga et al., 2017; Simpure et al., 2016; Arabkoohsar et al., 2015).
99 In addition, some researchers have focused on the investigation of the CAES integrated
100 with a waste heat recovery system (Meng et al., 2018; Mohammadi et al., 2017). The
101 ORC and Kalina Cycle (KC) could recover waste from a gas turbine exhaust in the
102 CAES system to improve the RTE (Meng et al., 2018; Quoilin et al., 2013). Furthermore,
103 Soltani et al. (2020) analysed and compared ORC and KC combined with CAES. Their
104 findings indicated that the highest RTE is achieved by the ORC-R290 cycle, with a 2.67%
105 increase. Additionally, the ORC has been demonstrated to have a low operating cost
106 and a long plant life, which is more commercially viable (Meng et al., 2018; Sadreddini
107 et al., 2018; Pinerez et al., 2021).

108 Recently, utilising the CAES system for the CCHP system has been receiving increased
109 attention (Mohammadi et al., 2017; Sadreddini et al., 2018; Razmi et al., 2019).
110 Additionally, He et al. (2015) performed model development and thermodynamic
111 analysis for CAES combined with a CCHP system. It was found that the proposed
112 system could minimise fuel consumption (29.4%) and maximise the exergy efficiency

113 (31.8%). Furthermore, Yao et al. (2016) developed a CCHP system based on a small-
114 scale D-CAES system to investigate the dependence of the system's thermodynamic
115 performance on the design variables. The results showed that the efficiency point of 51%
116 is preferred for industrial applications. Yan et al. (2018) developed an active storing
117 strategy for a novel CCHP system integrated with CAES, and the novel algorithm C-
118 NSGA-II makes the model more accurate. The proposed methodology could be applied
119 to future research on CCHP system optimisation. The results showed that the daily cost
120 decreases by 4.45% to increase the environmental benefits. Souza et al. (2020) proposed
121 a co-generation system using the ORC and an absorption refrigeration system through
122 energy, exergy and exergoeconomic assessment. The results indicated that ORC-C
123 allowed increases of 33.6% in power production and 34.5% in energy efficiency.

124 In addition, very few studies have focused on the CCHP system based on CAES, for a
125 waste heat recovery system and cooling system. Mohammadi et al. (2017) proposed a
126 CCHP system composed of CAES, ORC and an ammonia-water absorption
127 refrigeration system. It was found that the CCHP system could achieve a RTE of 53.94%
128 under the design condition. Sadreddini et al. (2018) developed a new cogeneration
129 system composed of CAES, ORC and an ejector refrigeration system. The study used
130 an optimisation process (genetic algorithm) with the aim being to find the optimal
131 solution. The results showed that the RTE of the system increased by 5.7%, from 68%
132 to 71.88%.

133 In this present study, a CCHP system comprising of CAES, ORC and a single-effect
134 LiBr/H₂O absorption system is proposed. A detailed process description of the CCHP
135 system is given in Section 2. Furthermore, the CCHP system can be integrated with the
136 renewable energy to overcome the intermittence problem. There are three points to
137 increase the RTE of the CCHP system: (1) the use of a recuperator in the proposed

138 CCHP system, (2) the use of a multi-stage centrifugal compressor with six intercooled
139 stages, (3) the use of ORC integrated with the CAES system.

140 1.3. Motivation

141 The new cogeneration system is composed of a CAES system, ORC and a single effect
142 absorption refrigeration system. The CAES system is the potential technology for large-
143 scale energy storage. The ORC can be integrated with the CAES system to recover the
144 waste heat and improve the RTE during charging and discharging time. The hybrid
145 system has all the advantages of energy storage technology and can produce three
146 different commodities, namely power, cooling, and heating. The hybrid system is
147 investigated and analysed through simulation and thermodynamic analysis to improve
148 system performance in this work.

149 1.4. Aims and novel contributions

150 In this study, a high-efficiency cogeneration system, comprising a CAES system, ORC
151 and a single effect absorption refrigeration system, is investigated. The aim of this study
152 is to improve the RTE of the hybrid system. During the charging process, the ORC
153 could recover waste heat from the multi-stage compressors. The heat source of the
154 single-effect absorption refrigeration system is flue gas recovered from the low-
155 pressure turbine. The proposed CCHP system has some advantages. Firstly, it can
156 provide cooling, heating and power simultaneously, with high RTE. Secondly, it can be
157 utilised as peak shaving. To achieve this aim, the following objectives have been
158 identified:

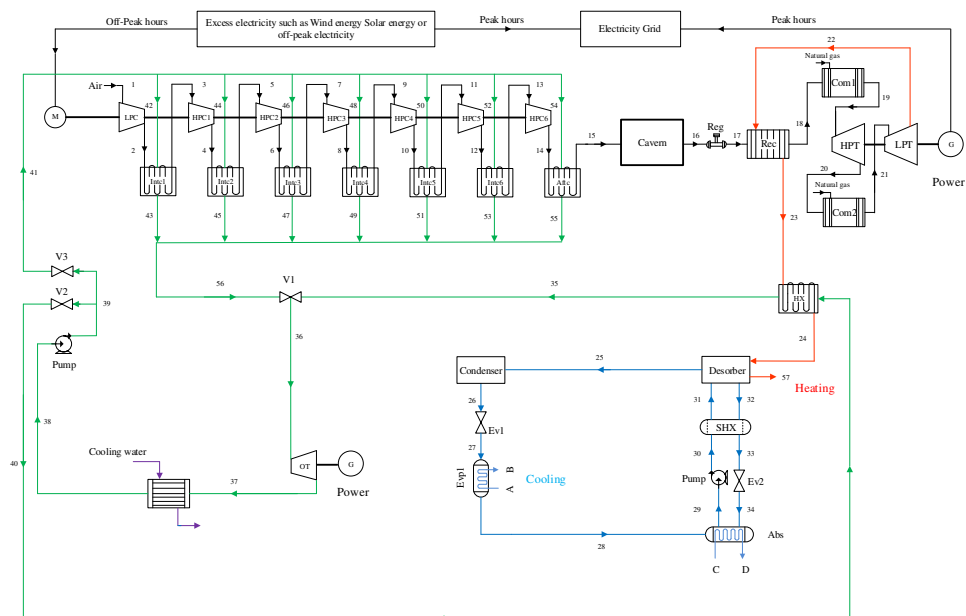
- 159 • To develop steady-state models of CAES, ORC and single effect absorption
160 refrigeration system in Aspen Plus®.
- 161 • Model validation of CAES, ORC, and single effect absorption refrigeration
162 system.

- 163 • Energetic and exergetic performance evaluation of the integrated system.
- 164 • Sensitivity analysis of the system performance to study the effect of
- 165 compressor inlet temperature, inter-cooler temperatures, ORC turbine inlet
- 166 pressure and vapour generator temperature.

167 The novel contributions of this study include:

- 168 • A new CCHP system composed of CAES, ORC and a single effect absorption
- 169 refrigeration system is proposed.
- 170 • A new scheme to improve the RTE is proposed for the CAES system.
- 171 • The process performance is analysed through energy and exergy analysis.
- 172 • Sensitivity analysis of different important parameters is undertaken.

173 2. System description



174
175 **Figure 1.** Schematic diagram of the proposed system

176 A schematic diagram of the newly proposed CCHP system is shown in Fig 1, consisting
177 of a CAES system, an ORC and a single effect absorption refrigeration system. The

178 CCHP system itself is composed of two subsections, including charging and
179 discharging subsystems.

180 In the charging process, the multi-stage compressors compress air to a high pressure
181 and store it in the cavern, converting electrical energy into the internal energy of the air,
182 for storage during off-peak power consumption periods. The advantages of using multi-
183 stage compressors are improved efficiency, less moisture build-up and a smaller
184 footprint (Meng et al., 2018; McNevin and Harrison, 2017). An ORC is used to recover
185 low-grade waste heat from the multi-stage compressors, using inter-coolers and after-
186 coolers to improve the RTE (Meng et al., 2018; Quoilin et al., 2013). The ORC is
187 comprised of an evaporator, expander, condenser, pump and heat exchangers. Then, the
188 organic working fluid will be pumped into the evaporator to be heated and transformed
189 into evaporated liquid. Finally, the evaporated fluid passes into the expander to generate
190 more power (Meng et al., 2018; Wang et al., 2013).

191 In the discharging process, the high-pressure air is released from the cavern and heated
192 in the recuperator, by recovering the waste heat from the exhaust of the low-pressure
193 turbine. The fuel is then burnt in the combustion chamber by mixing with the preheated
194 air. The high-temperature and high-pressure gas enters the turbine where it is expanded
195 to generate electricity during peak electricity consumption (Budt et al., 2016; Zhou et
196 al., 2019). However, the low-pressure gas turbine still has some waste heat, one part is
197 used to drive the ORC turbine and the other to drive the desorber. The recovered waste
198 heat can further improve the round-trip efficiency of the CCHP system and generate
199 cooling capacity.

200 Single effect absorption refrigeration systems can be driven by waste flue gas from a
201 low-pressure gas turbine. The advantages of using a single effect absorption
202 refrigeration system integrated with an ORC include high economic efficiency and

203 being environmentally friendly (Misra et al., 2003; Gomri, 2009). The single-effect
204 absorption refrigeration system consists of the desorber, condenser, evaporator,
205 absorber, and solution heat exchanger. In this study, Lithium bromide (LiBr) is used as
206 the absorbent and water is considered the refrigerant of the refrigeration cycle. The
207 reason is that the refrigeration system has the characteristics of high economic
208 efficiency, safety reliability, non-toxicity and the odourless LiBr solution (Chen et al.,
209 2017; Misra et al., 2003). The working principle of a single effect absorption
210 refrigeration system can be divided into two parts: One part is separated from the
211 solution as cold agent water vapour, and the other becomes a concentrated lithium
212 bromide solution. The separated water vapour is condensed into water by a condenser
213 and then evaporated by an evaporator to absorb heat and produce cold before entering
214 the absorber and mixing with the concentrated lithium bromide solution to form a dilute
215 lithium bromide solution. Conversely, the concentrated lithium bromide solution in the
216 generator, after heat exchanging with the dilute lithium bromide solution, enters the
217 absorber through the throttle valve, where it combines with the refrigerant vapour and
218 dilutes into a dilute solution. This solution passes through the solution pump and the
219 solution heat exchanger before entering the generator. Through the working process, a
220 cooling capacity is generated.

221 3. Model development and validation

222 Aspen Plus[®] is a general-purpose steady-state process simulation software. In this study,
223 the CAES system, the single effect absorption refrigeration system and the ORC are
224 developed and simulated by Aspen Plus[®] V12. Each subsystem has been validated
225 separately because the proposed system is yet to be built.

226 3.1. CAES model development and validation

227 CAES can be divided into two parts: charging and discharging. To decide the round-
228 trip performance, some system and component parameters need to be specified (Meng,
229 2019; Canepa et al., 2013). The following is a list of the steady-state simulation
230 assumptions (Meng et al., 2018; Zhao, Dai and Wang, 2015; Mohammadi et al., 2017;
231 Sadreddini et al., 2018):

- 232 1. The flow remains in a steady state condition over the entire length of the
233 streamlines.
- 234 2. The assumed air content is 78 vol% nitrogen, 21 vol% oxygen, and 1 vol% Ar.
- 235 3. The isentropic efficiency of the compressors is 75%.
- 236 4. The isentropic efficiency of the turbines is 93%.
- 237 5. The mechanical efficiency of compressors and turbines is 100%.
- 238 6. Exergy ambient reference condition of 25°C and 1.01325 kPa.
- 239 7. The pressure in the combustion chamber is the same as the cavern outlet
240 pressure (4.6 MPa).
- 241 8. The fuel of choice in combustion is natural gas, and its temperature is set to
242 32°C.
- 243 9. Pump mechanical efficiency of 80%.

244 The physical properties were calculated using the PENG-ROB (Standard Peng-
245 Robinson cubic equation of state) method. The PENG-ROB method was developed by
246 Peng and Robinson in 1976 (Meng et al., 2019; Liu et al., 2014), and was selected for
247 the property calculation for the CAES model.

248 The different critical components of the CAES system were simulated with different
249 blocks in Aspen Plus[®], as summarised in Table 1.

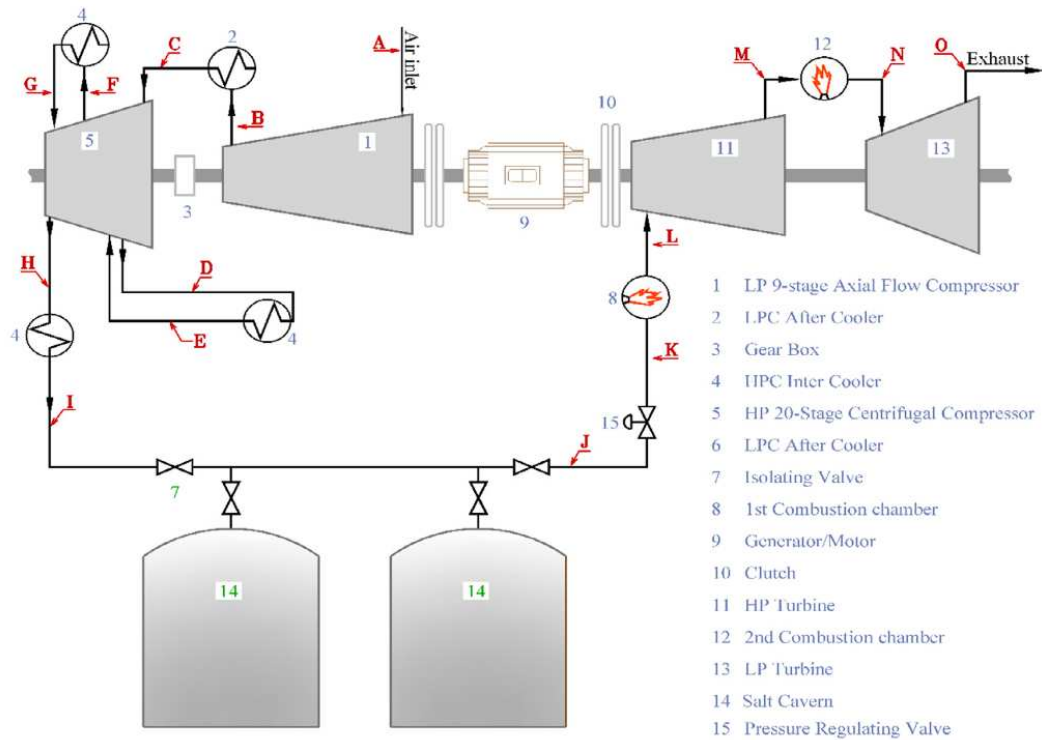
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Table 1. The different CAES components simulated in Aspen Plus®.

Components	Blocks
Compressors / Turbines	<i>Compr</i>
Inter-cooler / After-cooler	<i>Heater</i>
Pressure regulating valve	<i>Valve</i>
Recuperator	<i>HeatX</i>
Cavern	<i>Tank</i>
Combustors	<i>RGibbs</i>
Water pump	<i>Pump</i>

251

252 The Huntorf CAES plant data were collected from Meng (2019), Budt et al. (2016) and
253 Hoffein (1994) to validate the models. Figure 2 shows the flowsheet of the Huntorf
254 plant. Table 2 presents the key input process conditions and parameter values for
255 thermodynamic simulation in Aspen Plus®. Table 3 shows the results of the steady-state
256 simulation against the Huntorf CAES plant. The results show that the relative error is
257 less than 0.38%.



258

259 **Figure 2.** Schematic diagram of Huntorf CAES plant (Jafarizadeh et al., 2020)

260 **Table 2.** The critical operational parameters of Huntorf plant

Main Streams	Process	Parameters	Value
1	Charging	Rated air mass flow rate	108 kg/s
2	Charging	Rated power of compressor	60 MW
3	Charging	Inter-cooler outlet temperature	50°C
4	Charging	Inter-cooler outlet pressure	4.6-7.2 MPa
5	Charging	Operation time	8 h
6	Charging	Compressor stage	2
7	Charging	High pressure ratio	8
8	Charging	Low pressure ratio	6
9	Discharging	Rated power of turbine	290 MW
10	Discharging	Rated air mass flow rate	417 kg/s

11	Discharging	Inlet pressure of 1-stage turbine	4.2 MPa
12	Discharging	Inlet temperature of 1-stage turbine	550°C
13	Discharging	Inlet pressure of 2-stage turbine	1.13 MPa
14	Discharging	Inlet temperature of 2-stage turbine	825°C
15	Discharging	Operation time	2 h

261

262

Table 3. Model validation results (with Huntorf CAES plant)

Parameters	Huntorf data	Simulation results	Relative errors (%)
Compressors consumption (MW)	60	60.11	0.18
Rated power of turbine (MW)	290	291.11	0.38

263

264

As the data from the Huntorf plant is not sufficiently detailed, an additional model

265

comparison with Kakodkar (2018) was carried out. Table 4 shows the results of the

266

steady-state simulation compared with literature data from Kakodkar (2018).

267

Table 4. The critical operational parameters of Kakodkar (2018)

Parameters	Literature Data	Simulation Results	Error (%)
Inlet temperature of cavern (°C)	35	35	0
Outlet temperature of recuperator (°C)	642.6	642.35	0.03
Outlet temperature of combustor (°C)	712.67	712.53	0.01
Outlet pressure of turbine (MPa)	15	15	0
Outlet temperature of turbine (°C)	1562.56	1563.12	0.03

Outlet mass flow rate of cavern (kg/s)	400	400.85	0.21
Round-trip efficiency (%)	52.45	53	1.04
Compressors consumption (MW)	50	50.2	0.4
Rated power of turbine (MW)	167	167.78	0.46

268

269 3.2. ORC model development and validation

270 Figure 3 shows the main components of the ORC. The main working process comprises

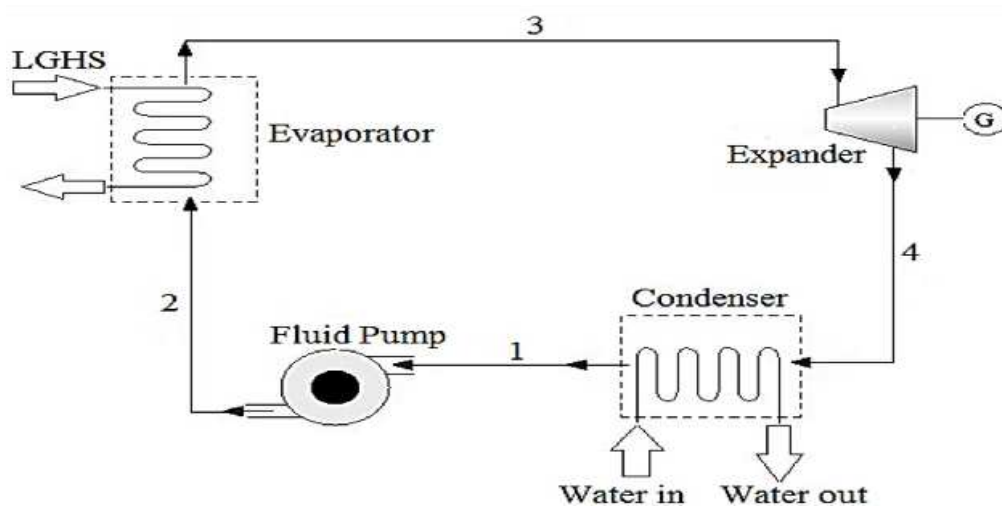
271 four components: pump, evaporator, expander and condenser.

272 The pump is used to increase liquid pressure. The high-pressure liquid then enters the

273 evaporator where it is evaporated to form high pressure saturated vapour. The saturated

274 vapour then passes through the expander where it expands and drives a turbine to

275 generate electricity.



276

277 **Figure 3.** Schematic diagram of simple ORC (Kheiri and Ghaebi, 2016).

278 The main components of the ORC are simulated with different blocks in Aspen Plus[®],

279 as summarised in Table 5.

280 **Table 5.** The different ORC components simulated in Aspen Plus[®].

Components	Blocks
Evaporator/ Condenser	<i>HeatX</i>
Recuperator	<i>HeatX</i>
Pump	<i>Pump</i>
Expander	<i>Compr</i>
Pressure-regulating valve	<i>Value</i>

281

282 The validation data for the ORC model were obtained from Kanoglu and Bolatturk,
 283 (2008) for the geothermal power plant in Reno, NV, USA. The organic working fluid
 284 used is R600a (Iso-Butane), which is a widely used refrigerant. The parameters in Table
 285 6 are the input process conditions and main operational parameters used in the ORC
 286 model. Table 7 shows the simulation results, which are compared to the reference plant
 287 data, for model validation. The relative errors are all less than 5%, and the simulation
 288 results agreed with the reference plant data.

289 **Table 6.** The different ORC components simulated in Aspen Plus®

Main streams	Parameters	Value
1	Hot steam inlet flowrate (kg/s)	555.9
2	Hot steam inlet temperature (°C)	158
3	Working fluid flowrate (kg/s)	305.6
4	Cooling water inlet temperature(°C)	3
5	Cooling water inlet mass flowrate (kg/s)	1695.6
6	Turbine inlet / outlet pressure (bar)	32.5/4.1
6	Turbine isentropic efficiency	85%

7	Pump efficiency	92.5%
8	Recuperator efficiency	95%
9	Expander efficiency	80%

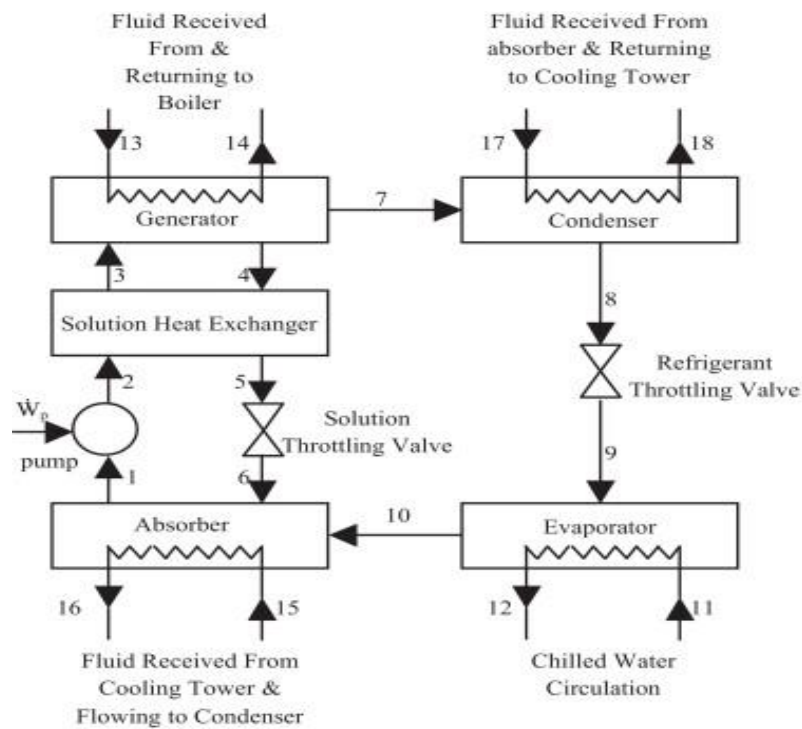
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291 **Table 7.** Model validation results (in comparison with Reno geothermal power plant)

Parameters	Reference plant data	Simulation results	Relative errors (%)
Hot steam outlet temperature (°C)	128	128.7	0.54
Cooling water outlet temperature(°C)	11.7	11.4	2.56
Turbine heat transfer power (kW)	21,744	21,366	1.74
Condenser heat transfer power (kW)	141,271	137,484	2.68
Evaporator heat transfer power (kW)	160,929	156,746	2.60
Pump heat transfer power (kW)	2087	2049	1.82
Output power of Expander (kW)	16,396	15,999	2.42

292

293 3.3. Single effect absorption refrigeration system model development and validation
 294 Figure 4 shows the main components of a single-effect absorption refrigeration
 295 system. It uses thermal energy as the driving heat source to drive the entire refrigeration
 296 cycle and thus produce the cooling capacity. In this single effect absorption
 297 refrigeration system, water is the refrigerant, and the lithium bromide solution is the
 298 absorbent. After continuous heating of the lithium bromide solution in the generator,
 299 part of it is separated from the solution as cold agent water vapour, while the other part
 300 becomes a concentrated lithium bromide solution. The separated water vapour is
 301 condensed into water by the condenser and then evaporated by the evaporator to absorb
 302 heat to produce the cooling capacity.



303

304 **Figure 4.** Schematic diagram of a single-effect LiBr/H₂O absorption system

305

(Avanessian and Ameri, 2014).

306 Due to the LiBr solution being a strong electrolyte, the model physical property method
 307 is calculated using the ELECNRTL property method. For the absorption refrigeration

308 system in this paper, the cooling water is connected in series, entering the system from
309 the absorber inlet, and leaving the system at the condenser outlet. The steady-state
310 simulation assumptions for the single-effect LiBr/H₂O absorption system are given
311 below:

312 (1) The generator's pressure is the same as the condenser, and the pressure of the
313 absorber is the same as the evaporator.

314 (2) The thick solution of LiBr at the outlet of the generator and the diluted solution of
315 LiBr at the outlet of the absorber are both saturated solutions.

316 (3) The pressure drops and heating loss of the piping and components are not considered.

317 (4) The amount of heat diffusion in the flow direction is ignored.

318 The components of a single-effect LiBr/H₂O absorption system are simulated with
319 different blocks in Aspen Plus[®], as summarised in Table 8 (Somers et al., 2011). Table
320 9 presents the key input process conditions and their values of a single-effect LiBr/H₂O
321 absorption system for thermodynamic simulation in Aspen Plus[®]. Table 10 shows the
322 simulation results compared with validation data from Herold et al., (2016).

323 **Table 8.** The different single-effect LiBr/H₂O absorption components were simulated
324 in Aspen Plus[®]

Components	Blocks
Desorber	<i>Flash2, HeatX</i>
Solution heat exchanger	<i>HeatX</i>
Evaporator/ Condenser	<i>HeatX</i>
Absorber	<i>HeatX</i>

Pump	<i>Pump</i>
Throttling valve	<i>Valve2</i>

325 **Table 9.** The different single-effect LiBr/H₂O absorption components were simulated
326 in Aspen Plus[®]

Main streams	Parameters	Value
1	Desorber inlet temperature (°C)	100
2	Desorber inlet pressure (kPa)	7.406
3	Evaporator inlet pressure (kPa)	0.676
4	Evaporator cooling water inlet temperature (°C)	10
5	Condenser cooling water inlet temperature (°C)	25
6	Condenser inlet pressure (kPa)	7.406
7	Evaporator cooling water inlet flowrate (kg/s)	0.4
8	Absorber cooling water inlet flowrate (kg/s)	0.28
9	Absorber inlet flowrate (kg/s)	27.790
10	Evaporator inlet flowrate (kg/s)	23.480
11	Dilute solution mass fraction	57.5%
12	Thick solution mass fraction	61.5%

327

328 **Table 10.** Simulation results compared with literature data from Herold et al. (2016)

Parameters	Reference plant data	Simulation results	Relative errors (%)
Coefficient of performance	0.815	0.811	0.49

Concentration of dilute solution (%)	57.5	57.2	0.52
Concentration of thick solution (%)	61.5	61.37	0.21
Generator-Condenser heat transfer power (kW)	604.9	604.9	0
Absorber heat transfer power (kW)	583.6	583.6	0
Vapour compressor heat transfer power (kW)	514.3	514.3	0
Evaporator heat transfer power (kW)	493.1	493.1	0
Solution heat exchanger	66.72	66.59	0.19
Cooling capacity (kW)	10.00	10.00	0

329 3.4. Model development of CCHP system

330 The proposed CCHP system is comprised of three parts, namely the CAES, ORC and
331 single-effect LiBr/H₂O absorption system. To analyse the CCHP system, all models
332 have been developed and integrated successfully within a single flowsheet in Aspen
333 Plus[®] V12. For the CAES, the design data comes from the Columbia Hills CAES system
334 (McGrail et al., 2013). For the ORC and absorption system, the reasonable assumptions
335 were based on the output of the CAES system charging part, which ultimately ensured
336 that the model of the CCHP system would work well. The assumed input parameters'
337 design condition of the CCHP system is shown in Table 11.

338 **Table 11.** Design conditions of CCHP system (charging and discharging process)

Charging Process		
Parameter	Value	Unit
Ambient temperature	25	°C
Ambient pressure	1.013	bar
Pressure ratio of compressor	1.96177	-
Mass flow rate of compressor	353	kg/s
Compressor isentropic efficiency	75	%
Charging time	3	hour
Cavern inlet pressure	115.65	kg/s
Cavern outlet pressure	35.78	bar
Cavern inlet temperature	40.56	°C
ORC turbine inlet pressure	4	bar
Condenser pressure	4	bar
ORC turbine isentropic efficiency	80	%
ORC pump isentropic efficiency	80	%
Discharging Process		
Discharging time	6	hour
Inlet pressure of high-pressure turbine	34.40	bar
Turbine isentropic efficiency	93	%
Inlet pressure of low-pressure turbine	17.93	bar

Fuel inlet pressure of combustor 1	44.82	bar
Fuel inlet pressure of combustor 2	24.13	bar
Fuel inlet temperature of combustor 1	32.22	°C
Fuel inlet temperature of combustor 2	32.22	°C
Heat duty of recuperator	105.51	kW
ORC turbine inlet pressure	19.85	bar
Condenser pressure	4	bar
ORC turbine isentropic efficiency	80	%
ORC pump isentropic efficiency	80	%
Outlet temperature of Desorber	40.2	°C
Cooling system pump isentropic efficiency	85	%
Outlet temperature of Solution heat exchange	53.5	°C
Outlet pressure of Solution heat exchange	0.07461	bar

339

340 4. Thermodynamic analysis

341 4.1. Exergy analysis

342 Exergy is defined as the maximum shaft work that a system can perform in a specified
343 reference environment. Exergy analysis is a thermodynamic analysis technique resulted
344 from the second law of thermodynamics, which indicates exergy destruction.

345 The general exergy balance of a quantity can be calculated by:

346
$$\text{Exergy Input} + \text{Exergy Generation} - \text{Exergy Output} - \text{Exergy Consumption} = \text{Exergy Accumulation} \quad (1)$$

347 The exergy of work and heat can be expressed as (Mohammadi et al., 2017;
348 Sadreddini et al., 2018):

349
$$\dot{E}_x^Q = \left(1 - \frac{T_0}{T_i}\right) \dot{Q}_i \quad (2)$$

350
$$\dot{E}_x^W = \dot{W} \quad (3)$$

351 Where \dot{E}_x^Q and \dot{E}_x^W are exergy associated with heat and work; T_0 is the ambient
352 temperature and T_i is the temperature of the heat transfer. The exergy of the stream can
353 be divided into two parts, namely physical and chemical. The physical exergy is defined
354 as:

355
$$ex_{ph} = (h_i - h_0) - T_0(s_i - s_0) \quad (4)$$

356 Where, h_i is enthalpy, h_0 is enthalpy at the reference state, s_i and s_0 are specific
357 entropy, which related to different streams. The entropy term $(s_i - s_0)$ is defined as:

358
$$s_i - s_0 = C_{avg} \ln \frac{T_i}{T_0} \quad (5)$$

359 Where C_{avg} is the average specific heat capacity of the substance.

360 The chemical exergy is defined as:

361
$$ex_{ch} = \sum_{i=1}^N y_i ex_i^{ch} + RT_0 \left(\sum_{i=1}^N y_i \ln(y_i) \right) \quad (6)$$

362 The ex is the sum of physical and chemical exergy, which can be calculated by the
363 following equation:

364
$$ex = ex_{ph} + ex_{ch} \quad (7)$$

365 Based on the above equations, exergy balance could be calculated as:

$$366 \quad \dot{E}_x^Q + \sum \dot{E}x_{in} = \sum \dot{E}x_{out} + \dot{E}_x^W + \dot{E}_x^D \quad (8)$$

367 Where the \dot{E}_x^D is the exergy destruction of the component. Total exergy destruction
368 can be defined as:

$$369 \quad \dot{E}x_{total}^D = \text{sum}(\dot{E}_x^D) \quad (9)$$

370 Table 12 shows the exergy destruction and exergy efficiency equations used for the
371 calculation of exergy destruction and exergy efficiency of each key component.

372 **Table 12.** Exergy destruction of each component (Zhao et al., 2015; Mohammadi et
373 al., 2017; Sadreddini et al., 2018)

System component	Exergy destruction	Exergy efficiency
Compressor	$\dot{E}x_{Com}^D = \dot{E}x_{in} - \dot{E}x_{out} + \dot{W}_{Com}$	$\Psi_{Com} = \frac{\dot{E}x_{out} - \dot{E}x_{in}}{\dot{W}_{Com}} \times 100$
Turbine	$\dot{E}x_{Tur}^D = \dot{E}x_{in} - \dot{E}x_{out} - \dot{W}_{Tur}$	$\Psi_{Tur} = \frac{\dot{W}_{Tur}}{\dot{E}x_{in} - \dot{E}x_{out}} \times 100$
Cavern	$\dot{E}x_{Cave}^D = \dot{E}x_{in} - \dot{E}x_{out}$	$\Psi_{Cave} = 1 - \frac{\dot{E}x_{Cave}^D}{\dot{E}x_{in}} \times 100$
Heat exchanger (Condenser, Evaporator, Intercoolers, Aftercooler and Recuperator)	$\dot{E}x_{HX}^D = \sum \dot{E}x_{in} - \sum \dot{E}x_{out}$	$\Psi_{HX} = \frac{(\dot{E}x_{out} - \dot{E}x_{in})_{cold}}{(\dot{E}x_{in} - \dot{E}x_{out})_{hot}} \times 100$

Value	$\dot{E}x_{Val}^D = \dot{E}x_{in} - \dot{E}x_{out}$	$\Psi_{Valve} = \frac{\dot{E}x_{in}}{\dot{E}x_{out}} \times 100$
Combustion chamber	$\dot{E}x_{Cc}^D = \dot{E}x_{in} - \dot{E}x_{fuel} - \dot{E}x_{out}$	$\Psi_{Cc} = \frac{\dot{E}x_{out} - \dot{E}x_{in}}{\dot{E}x_{fuel}} \times 100$
Pump	$\dot{E}x_{pum}^D = \dot{E}x_{in} - \dot{E}x_{out} + \dot{W}_{pum}$	$\Psi_{pum} = \frac{\dot{E}x_{out} - \dot{E}x_{in}}{\dot{W}_{com}} \times 100$
ORC turbine	$\dot{E}x_{OT}^D = \dot{E}x_{in} - \dot{E}x_{out} - \dot{W}_{OT}$	$\Psi_{OT} = \frac{\dot{W}_{OT}}{\dot{E}x_{in} - \dot{E}x_{out}} \times 100$
Desorber	$\dot{E}x_{Des}^D = \dot{E}x_{in} - \dot{E}x_{out} + \dot{W}_{VG}$	$\Psi_{Des} = \frac{\dot{E}x_{out} - \dot{E}x_{in}}{\dot{W}_{com}} \times 100$

374

375 4.2. Performance criteria

376 The performance criteria of D-CAES systems are different from those of conventional
377 power plants due to the charging process and the discharging process at different time
378 (as explained in Section 2). Another reason is that the first law efficiency cannot be
379 used to describe the CAES system. During the charging process, only the electrical
380 energy is used for the compressors work. The natural gas is combusted to heat
381 compressed air driving the turbine during the discharging process. Some waste heat in
382 the gas turbine is recovered by the ORC and single-effect LiBr/H₂O absorption system.
383 RTE is defined as the ratio of total energy output (comprising generated electricity of
384 the gas turbine and cooling energy) to the total energy input to the CAES system (Meng
385 et al., 2018; Zhao et al., 2015):

$$386 \quad RTE_{CAES} = \frac{W_{Tur}}{E_{fuel} + W_{Com}} \quad (10)$$

387 Where W_{Tur} is the output power of Turbine (kWh); W_{Com} is the electrical energy taken
 388 from the grid for driving the compressors (kWh); E_{fuel} is the thermal energy of fuel
 389 consumed (kWh). Based on Eq. (10), the RTE of the CCHP system could be described
 390 as:

$$391 \quad RTE_{CCHP} = \frac{W_{Tur} + W_{orc-c} + W_{orc-d} + W_{cooling}}{E_{fuel} + W_{Com} + W_{ORC,pum-c} + W_{ORC,pum-d} + W_{LiBr,pum}} \quad (11)$$

392 Where W_{orc-c} is the output power of the ORC in the charging process (kWh); W_{orc-d}
 393 is the output power of the ORC in the discharging process (kWh); $W_{cooling}$ is the output
 394 power of the single-effect LiBr/H₂O absorption system. $W_{ORC,pum-c}$ is the power
 395 consumption of the ORC pump in the charging process; $W_{ORC,pum-d}$ is the power
 396 consumption of the ORC pump in the discharging process; $W_{LiBr,pum}$ is the power
 397 consumption of the pump in the single-effect LiBr/H₂O absorption system.

398 For the CCHP system, the exergy destruction can be described as the difference
 399 between the total exergy flows into and out of the system minus the exergy
 400 accumulation in the system, which can be defined as:

$$401 \quad \dot{E}x_{total}^D = \text{sum}(\dot{E}x^D) \quad (12)$$

402 Furthermore, the coefficient of performance (COP) is a ratio of useful cooling provided
 403 to work (energy) required that is used for the single-effect LiBr/H₂O absorption system,
 404 which can be defined as:

$$405 \quad COP = \frac{Q}{W} = \frac{\dot{Q}_{Evap}}{\dot{W}_{Cooling} + \dot{Q}_{Des}} \quad (13)$$

406 5. Results and discussion

407 The results of the proposed CCHP system analysis are given in this section.

408 5.1. Thermodynamic analysis

409 Table 13 lists the thermodynamic properties of each stream of the proposed CCHP
 410 system. The following assumptions are used for the analysis of the proposed CCHP
 411 system (Meng et al., 2018):

- 412 • The ambient reference temperature and pressure are 25°C and 1.01325 kPa.
- 413 • All the cycles are at steady-state.
- 414 • The pressure drop, kinematic and potential energy of all components are ignored.
- 415 • Pure methane is used as the fuel in the combustion of the CAES discharging
 416 process.
- 417 • The outlet streams of the condenser are assumed to be saturated liquid.

418 **Table 13.** Thermodynamic properties of proposed CCHP system (charging and
 419 discharging process)

Charging process						
Steam Number	Working Fluid	P [bar]	T [°C]	h [kJ/kg]	m [kg/s]	s [kJ/kg K]
1	Air	1.013	25.00	-0.235	353.00	0.162
2	Air	1.987	109.06	84.504	353.00	0.219
3	Air	1.987	40.12	14.791	353.00	0.018
4	Air	3.898	129.01	103.819	353.00	0.075
5	Air	3.898	39.65	13.917	353.00	-0.178
6	Air	7.648	127.73	102.791	353.00	-0.121
7	Air	7.648	40.43	13.936	353.00	-0.371
8	Air	15.003	128.75	103.004	353.00	-0.314
9	Air	15.003	40.03	12.023	353.00	-0.569

10	Air	29.434	128.27	100.937	353.00	-0.513
11	Air	29.434	39.90	9.010	353.00	-0.771
12	Air	57.742	128.08	97.896	353.00	-0.715
13	Air	57.742	39.93	3.724	353.00	-0.980
14	Air	113.278	127.86	92.972	353.00	-0.923
15	Air	113.278	39.76	-5.453	353.00	-1.200
36	R600a	19.853	100.00	-2225.935	528.00	-6.666
37	R600a	4.000	52.16	-2278.147	528.00	-6.625
38	R600a	4.000	29.33	-2645.5457	528.00	-7.835
39	R600a	10.860	29.81	-2643.972	528.00	-7.834
41	R600a	19.853	29.81	-2643.972	528.00	-7.834
42	R600a	19.853	30.00	-2643.032	59.00	-7.837
43	R600a	19.853	100.00	-2225.935	59.00	-6.666
44	R600a	19.853	29.81	-2643.510	76.00	-7.838
45	R600a	19.853	100.00	-2225.935	76.00	-6.666
46	R600a	19.853	30.000	-2643.032	75.20	-7.837
47	R600a	19.853	100.00	-2225.935	75.20	-6.666
48	R600a	19.853	30.00	-2643.032	77.00	-7.837
49	R600a	19.853	100.00	-2225.935	77.00	-6.666
50	R600a	19.853	30.00	-2643.031	77.80	-7.836
51	R600a	19.853	100.00	-2225.935	77.80	-6.666
52	R600a	19.853	30.00	-2643.031	79.70	-7.836
53	R600a	19.853	100.00	-2225.935	79.70	-6.666
54	R600a	19.853	30.00	-2643.031	83.30	-7.836

55	R600a	19.853	100.00	-2225.935	83.30	-6.666
56	R600a	19.853	100.00	-2225.935	528.00	-6.666
57	Flue gas	1.03	94.21	-1237.250	193.63	0.318
Discharging process						
Steam Number	Working Fluid	P [bar]	T [°C]	h [kJ/kg]	m [kg/s]	s [kJ/kg K]
17	Air	35.780	40.56	8.605	189.00	-0.853
18	Air	35.780	561.24	567.377	189.00	0.191
19	Flue gas	34.400	676.67	689.262	189.53	0.349
20	Flue gas	18.270	545.07	538.902	189.53	0.363
21	Flue gas	17.930	1331.67	1444.571	193.64	1.166
22	Flue gas	1.030	600.61	528.478	193.64	1.247
23	Flue gas	1.030	115.56	-16.910	193.64	0.344
24	Flue gas	1.030	95.68	-37.909	193.64	0.289
25	Water	0.075	78.40	-13332.897	0.17	-0.950
26	Water	0.075	40.20	-15812.315	0.17	-8.849
27	Water	0.007	1.23	-15812.315	0.17	-8.810
28	Water	0.007	1.30	-13477.120	0.17	-0.305
29	LiBr- H ₂ O	0.007	32.70	-9919.965	1.00	-6.925
30	LiBr- H ₂ O	0.075	358.44	-9919.939	1.00	-6.358
31	LiBr- H ₂ O	0.075	50.48	-10251.041	1.00	-8.009

32	LiBr- H ₂ O	0.075	47.42	-9888.558	0.83	-10.280
33	LiBr- H ₂ O	0.075	53.30	-9491.444	0.83	-8.899
34	LiBr- H ₂ O	0.007	12.84	-9491.444	0.83	-9.815
35	R600a	19.853	100.00	-2225.935	10.00	-6.666
36	R600a	19.853	100.00	-2225.935	10.00	-6.666
37	R600a	4.000	52.16	-2278.147	10.00	-6.626
38	R600a	4.000	29.33	-2634.455	10.00	-7.811
39	R600a	19.853	30.62	-2630.819	10.00	-7.800
40	R600a	19.853	30.62	-2630.819	10.00	-7.800

420

421 Table 14 shows the main results of the CCHP system based on the equations and the
422 data in Table 12. From Table 14, the charging time requires 8 hours to fill the cavern,
423 and the discharging time requires 6 hours. This is because the discharging process of
424 the CAES system needs to be performed during peak hours, which aims to maximise
425 the revenue. During the charging process, 218421.63 kW power of electricity is
426 consumed by the multi-stage compressors. The inter-cooler and after-cooler captured
427 220263.22 kW of heat. During the charging process, this captured heat is used to drive
428 the ORC turbine. During the discharging process, the flue gas from the turbine still has
429 some waste heat, which can be recovered by the ORC turbine to drive the desorber for
430 the cooling system. The ORC and cooling system can recover low-grade heat and
431 improve the RTE of the CCHP system. Moreover, the (high-pressure and low-pressure)
432 turbines and ORC turbine each have a capacity of 233977.36 kW. The compressed air

433 reaches a pressure of 113.28 bar before entering the cavern, with an inlet mass flow rate
 434 of 353 kg/s. Since the charging process takes place during off-peak hours when
 435 electricity prices are low, this will vastly improve the system's economy value. The
 436 single-effect absorption refrigeration system can provide 245.45 kW of cooling
 437 capacity and the COP of the cooling system is 0.74. The COP of refrigeration systems
 438 is affected by various factors, including the inlet temperature of the heating source, the
 439 inlet temperature of the cooling water, and the outlet temperature of the refrigerant
 440 water (to be discussed in the following section). The addition of the ORC and
 441 refrigeration system allows the CCHP system to achieve a RTE of 66.35%,
 442 approximately a 23% improvement over the D-CAES system alone.

443 **Table 14.** Performance indicators of the CCHP system (Charging and discharging
 444 processes)

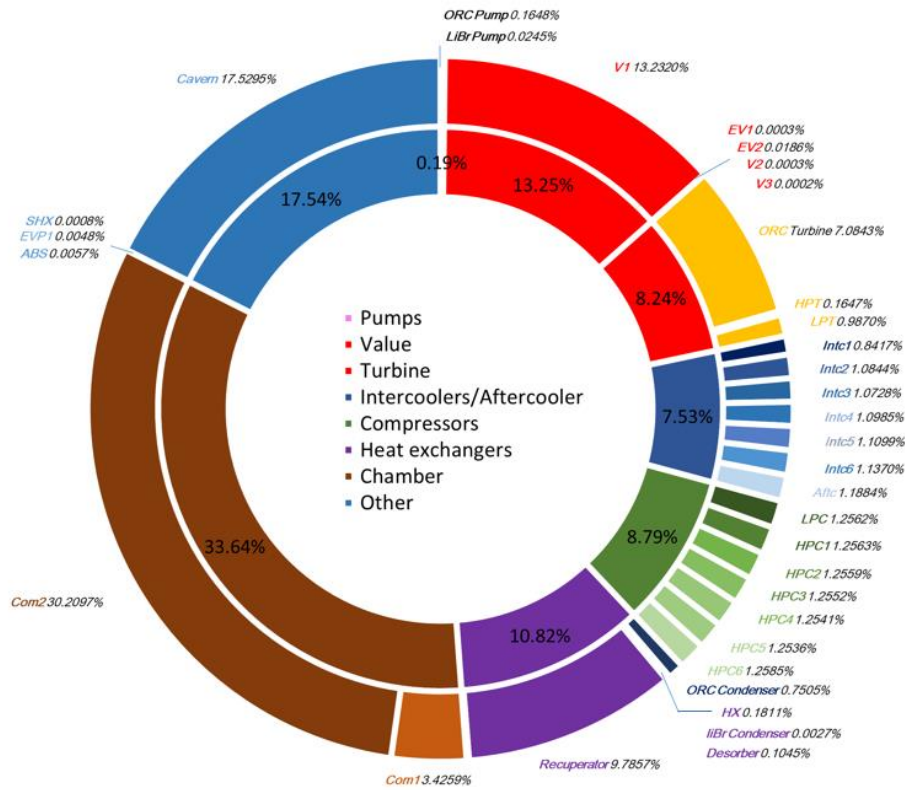
Charging process		
Parameter	Value	Unit
Charging time	3	Hour
\dot{W}_{Com1}	29913.29	kW
\dot{W}_{Com2}	31426.87	kW
\dot{W}_{Com3}	31372.62	kW
\dot{W}_{Com4}	31441.04	kW
\dot{W}_{Com5}	31386.49	kW
\dot{W}_{Com6}	31376.67	kW
\dot{W}_{Com7}	31504.65	kW
$\dot{W}_{comp,total}$	218421.63	kW

$\dot{E}_{\text{comp,total}}$	1747373.04	kWh
\dot{Q}_{Intc1}	24608.68	kW
\dot{Q}_{Intc2}	31735.71	kW
\dot{Q}_{Intc3}	31365.64	kW
\dot{Q}_{Intc4}	32116.41	kW
\dot{Q}_{Intc5}	32450.09	kW
\dot{Q}_{Intc6}	33242.57	kW
\dot{Q}_{Aftc}	34744.119	kW
\dot{W}_{OT}	27567.82	kW
$\dot{Q}_{\text{Condenser}}$	193986.41	kW
$\dot{W}_{\text{ORC,pump}}$	830.79	kW
Discharging process		
Parameter	Value	Unit
Discharging time	6	Hour
\dot{W}_{HPT}	28497.17	kW
\dot{W}_{LPT}	177390.27	kW
\dot{Q}_{Com1}	25874.17	kW
\dot{Q}_{Com2}	196809.55	kW
\dot{Q}_{Rec}	105607.815	kW
\dot{Q}_{HX}	4066.25	kW
\dot{W}_{OT}	522.12	kW
$\dot{Q}_{\text{Condenser}}$	3563.08	kW
$\dot{W}_{\text{ORC,pump}}$	36.36	kW

$\dot{Q}_{\text{Desorber}}$	330.72	kW
$\dot{Q}_{\text{Condenser}}$	260.63	kW
\dot{Q}_{Evap}	245.47	kW
\dot{Q}_{Absorb}	274.20	kW
$\dot{W}_{\text{Cooling,pump}}$	0.005	kW
\dot{Q}_{SHX}	134.91	kW
For whole process		
COP	0.74	—
$E_{x-\text{CCHP}}$	51.21	%
RTE	66.35	%

445

446 Figure 5 shows the results of the exergy analysis for the proposed CCHP system, based
447 on the design conditions. From the results of exergy analysis, it can be seen that the
448 highest exergy destruction (160635 kW) belongs to the combustion chamber. This is
449 because the combustion process occurs in the combustor, and the combustor's
450 irreversibility results in the highest level of exergy destruction. The use of fuels with a
451 simple molecular structure containing oxygen molecules will reduce exergy destruction.
452 Additionally, a recuperator can be used to preheat the air and fuel before introducing
453 them into the combustion chamber, which is another effective way to prevent fire
454 damage in the chamber. The second highest exergy destruction occurred at cavern
455 83716 kW, the third and fourth at the valve and recuperator, respectively. This is due
456 to the significant temperature difference between the hot and cold flow entering these
457 components.



458

459

Figure 5. Exergy destruction of all components

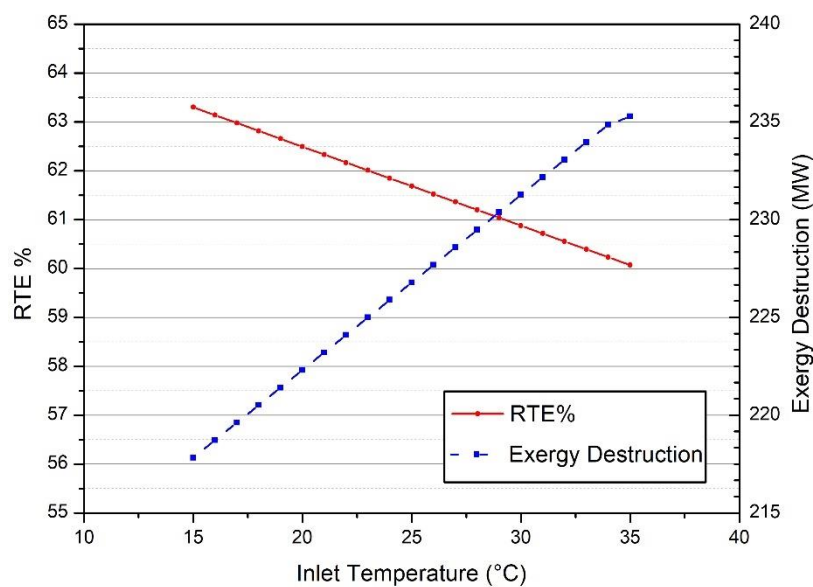
460

461 5.2. Parametric analysis

462 The effect of the system's most critical parameters on performance is investigated in
 463 this section by altering specific parameters while maintaining others constantly. The
 464 effect of individual change in the inlet temperature of the compressor, the inlet pressure
 465 of the ORC turbine, different working fluids for the ORC, the inlet temperature of
 466 combustion chamber one and the inlet mass flow rate of the pump in the cooling system
 467 on the performance of the proposed CCHP system are conducted.

468 5.2.1 Inlet temperature of the compressor

469 Figure 6 illustrates the effect of the compressor's inlet temperature variation on the
470 CCHP system's performance. The higher inlet temperature of the compressor means a
471 higher power output of multi-stage compressors. As the compressor's inlet air
472 temperature increases, the exergy destruction of the CCHP system gradually decreases.
473 When the inlet air temperature entering the compressor rises, the compressor needs less
474 power output to work, which means a drop in overall compressor output. Based on Eq.
475 (15), it will improve the RTE of CCHP system. It should be noted that the higher inlet
476 temperature of the compressor leads to the higher exergy destruction. The main reason
477 is that the variation of inlet temperature of the compressor only affects the charging
478 process, whilst the chemical dissipation of the fuel and the output work of the gas
479 turbine remains constant.



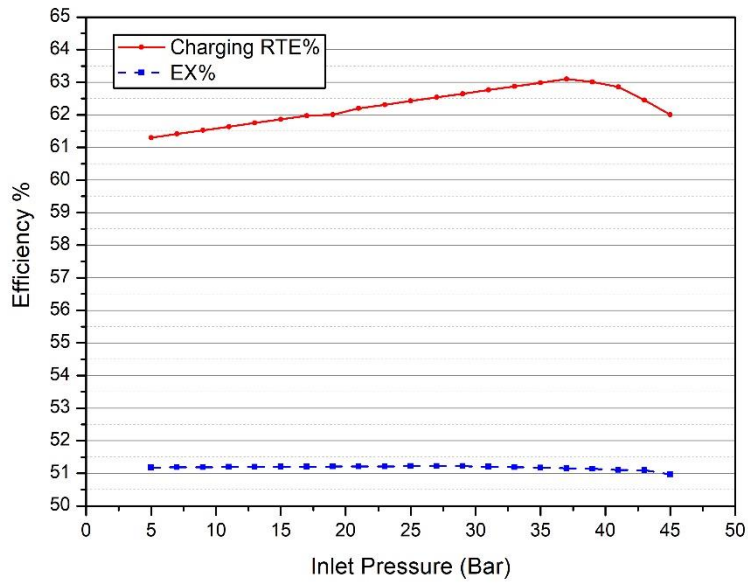
480

481 **Figure 6.** Effect of compressor inlet temperature variation on CCHP?? system

482

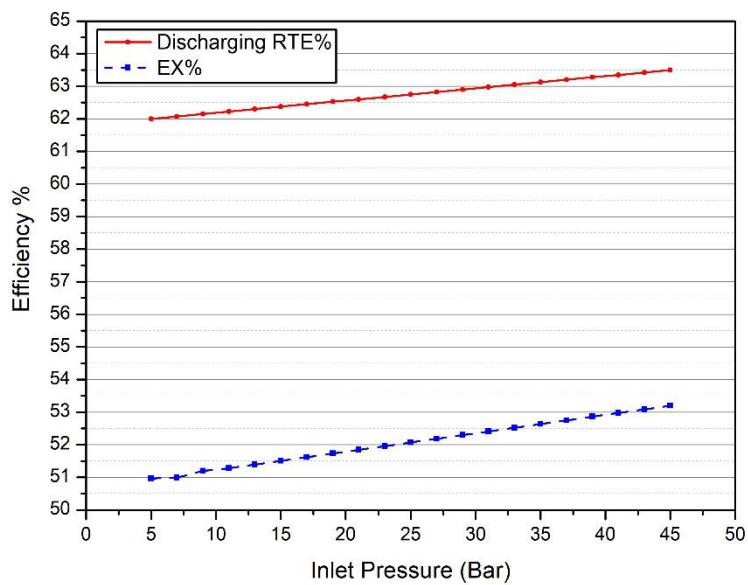
performance

483 5.2.2 Inlet pressure of the ORC turbine



484

485 **Figure 7 (a).** Effect of inlet pressure of the ORC turbine on CCHP system
486 performance (charging process)



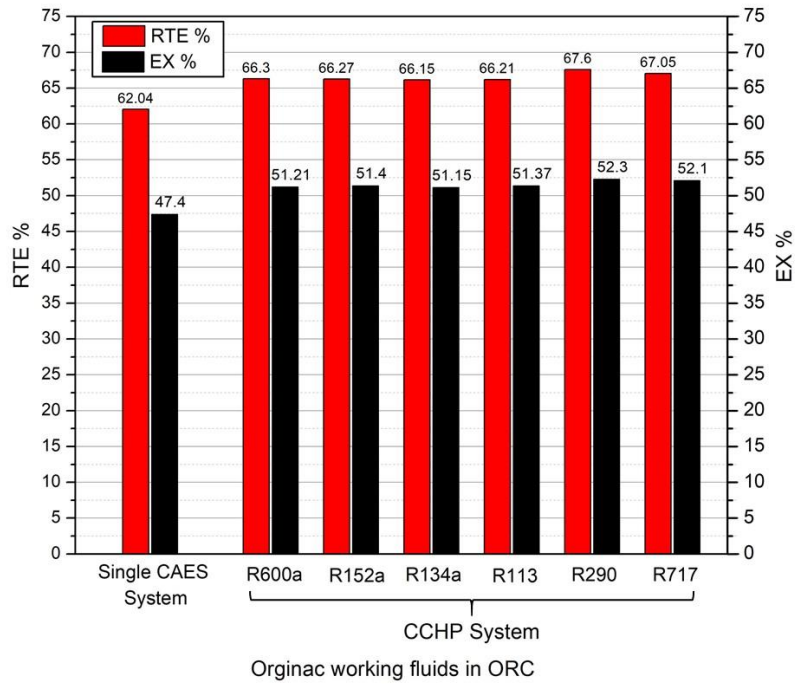
487

488 **Figure 7 (b).** Effect of inlet pressure of the ORC turbine on system performance
489 (discharging process)

490 Figure 7 shows the effect of the inlet pressure of the ORC turbine on total exergy
491 destruction and the RTE during (a) charging and (b) discharging processes. An increase
492 in the inlet pressure leads to the increase of the ORC power output. The results show
493 that an optimum inlet pressure exists for the ORC turbine, at which the power generated,
494 and the subsequent cooling and heating capacity are maximised. The results show that
495 the optimal operating point of inlet pressure during the charging process is around 35
496 bar. Additionally, both the RTE and the exergy efficiency of the system continue to
497 grow with the growth of the inlet pressure during the discharging process. When the
498 inlet pressure of the ORC turbine is at 35 bar, the proposed CCHP system could achieve
499 the highest RTE (66.3%).

500 5.2.3 Different working fluids for ORC

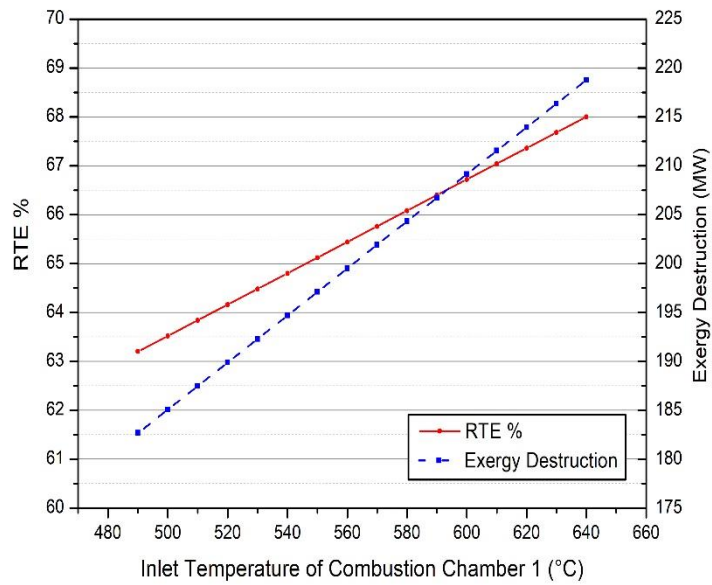
501 The selection of organic working fluids for the ORC to achieve the maximum utilisation
502 of waste heat is important. Therefore, the performance of different working fluids
503 operating in specific regions is analysed. Figure 8 shows that using different working
504 fluids has positive effects on RTE of the integrated system. In this proposed CAES
505 system, due to the use of multi-stage compressors and setting the 3 hours charging time
506 and 6 hours discharging time, the RTE of CAES only can reach up to 62.04%. This is
507 a very significant improvement compared to the RTE (42%) of the Huntorf CAES plant.
508 The results show that the RTE of the CCHP system increases by 4.11–5.56%.



509

510 **Figure 8.** Effect of different working fluids of the ORC on system performance

511 5.2.4 Inlet temperature of combustion chamber 1



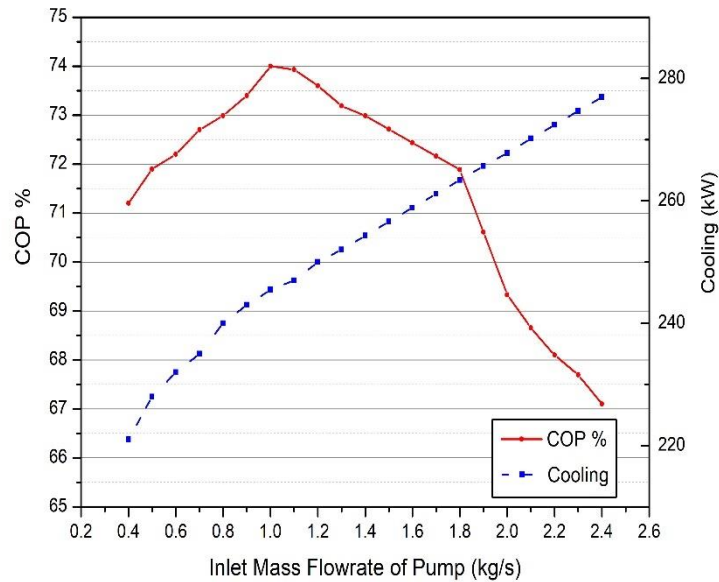
512

513 **Figure 9.** Effect of inlet temperature of combustion chamber 1 on system
 514 performance

515

516 Figure 9 shows the effect of combustion chamber 1's inlet temperature on the system
 517 performance. The results show that a higher inlet temperature implies higher gas turbine
 518 power output. At the same time, it leads to a higher turbine outlet temperature, which
 519 increases the power output generated in the ORC turbine as well. The cooling capacity
 520 produced in the single effect absorption refrigeration system also increases due to the
 521 higher generated heating from the flue gas.

522 5.2.5 Inlet mass flow rate of pump in single effect absorption refrigeration system



523

524 **Figure 10.** Effect of inlet mass flow rate of pump on system performance

525 Figure 10 shows the effect of the inlet mass flow rate of the pump on system
 526 performance. Since the single effect absorption refrigeration system is located

527 downstream of the ORC, any change in its parameters does not have an effect on the
528 ORC. The overall cooling capacity of the single effect absorption refrigeration system
529 is not very large (245.45 kW) because the heating source comes from flue gas. From
530 the results, it can be seen that the COP efficiency reaches a maximum of 74% at a mass
531 flow rate of pump of 1 kg/s.

532 6. Conclusion

533 This study has proposed a new CCHP system comprised of a D-CAES system, ORC
534 and a single effect absorption refrigeration system. The proposed CCHP system can
535 produce power, heating and cooling. A comprehensive thermodynamic analysis of a
536 CCHP system was applied to the system to analyse its performance. The main findings
537 are summarised as follows: (a) The RTE of the proposed CCHP system has been
538 improved by approximately 12% when compared with the McIntosh CAES plant (54%).
539 Under the design condition, the RTE and overall exergy efficiency of the system are
540 66.35% and 51.21%, respectively. (b) At design conditions, the proposed CCHP system
541 can produce 206 MW electrical energy, 28 MW heating and 0.2 MW cooling capacity,
542 and the COP of the single effect absorption refrigeration system is 0.74. (c) Total exergy
543 destruction of the CCHP system is equal to 478 MW, in which the combustion chamber
544 is responsible for more than half of it. After the combustion chamber, the cavern, and
545 recuperator have the highest value of exergy destruction. (d) R290 (Propane) as the
546 ORC working fluid is the most suitable working fluid with the best performance. (e) Of
547 all the parameters considered, the inlet temperature of the compressor, the inlet
548 temperature of the combustion chamber and the inlet pressure of the ORC turbine are
549 the most critical parameters, which can significantly improve the RTE and overall
550 exergy efficiency. However, the proposed CCHP system has not been commercially

551 deployed. Further opportunities are emphasised: (1) Economic evaluation (2)
552 Optimization to find the optimal operating conditions that maximize RTE. This
553 research is expected to provide guidance for commercial deployment of CAES.

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558 References

559 Aneke, M. and Wang, M., 2016. Energy storage technologies and real-life
560 applications—A state of the art review. *Applied Energy*, 179, pp.350-377.

561 Arabkoohsar, A., Machado, L., Farzaneh-Gord, M. and Koury, R.N.N., 2015.
562 Thermo-economic analysis and sizing of a PV plant equipped with a compressed air
563 energy storage system. *Renewable energy*, 83, pp.491-509.

564 Avanesian, T. and Ameri, M., 2014. Energy, exergy, and economic analysis of single
565 and double effect LiBr–H₂O absorption chillers. *Energy and Buildings*, 73, pp.26-36.

566 Budt, M., Wolf, D., Span, R. and Yan, J., 2016. A review on compressed air energy
567 storage: Basic principles, past milestones and recent developments. *Applied
568 energy*, 170, pp.250-268.

569 Cazzaniga, R., Cicu, M., Rosa-Clot, M., Rosa-Clot, P., Tina, G.M. and Ventura, C.,
570 2017. Compressed air energy storage integrated with floating photovoltaic
571 plant. *Journal of Energy Storage*, 13, pp.48-57.

572 Chai, Y., Gao, N., Wang, M. and Wu, C., 2020. H₂ production from co-
573 pyrolysis/gasification of waste plastics and biomass under novel catalyst Ni-CaO-

574 C. Chemical Engineering Journal, 382, p.122947.

575 Chen, H., Cong, T.N., Yang, W., Tan, C., Li, Y. and Ding, Y., 2009. Progress in
576 electrical energy storage system: A critical review. Progress in natural science, 19(3),
577 pp.291-312.

578 Chen, W., Shi, C., Zhang, S., Chen, H., Chong, D. and Yan, J., 2017. Theoretical
579 analysis of ejector refrigeration system performance under overall modes. Applied
580 Energy, 185, pp.2074-2084.

581 Canepa, R., Wang, M., Biliyok, C. and Satta, A., 2013. Thermodynamic analysis of
582 combined cycle gas turbine power plant with post-combustion CO₂ capture and exhaust
583 gas recirculation. Proceedings of the Institution of Mechanical Engineers, Part E:
584 Journal of Process Mechanical Engineering, 227(2), pp.89-105.

585 Cavallo, A., 2007. Controllable and affordable utility-scale electricity from
586 intermittent wind resources and compressed air energy storage (CAES). Energy, 32(2),
587 pp.120-127.

588 Crotofino, F., Mohmeyer, K.U. and Scharf, R., 2001, April. Huntorf CAES: More
589 than 20 years of successful operation. In SMRI Spring meeting (Vol. 2001).

590 Gomri, R., 2009. Second law comparison of single effect and double effect vapour
591 absorption refrigeration systems. Energy Conversion and Management, 50(5), pp.1279-
592 1287.

593 He, F., Xu, Y., Zhang, X., Liu, C. and Chen, H., 2015. Hybrid CCHP system
594 combined with compressed air energy storage. International Journal of Energy
595 Research, 39(13), pp.1807-1818.

596 Herold, K.E., Radermacher, R. and Klein, S.A., 2016. Absorption chillers and heat
597 pumps. CRC press.

598 IEA. 2020. Renewables 2020 – Analysis - IEA. [online] Available at:

599 <<https://www.iea.org/reports/renewables-2020>> [Accessed 29 November 2020].

600 Jafarizadeh, H., Soltani, M. and Nathwani, J., 2020. Assessment of the Huntorf
601 compressed air energy storage plant performance under enhanced modifications.
602 *Energy Conversion and Management*, 209, p.112662.

603 Kheiri, R. and Ghaebi, H., 2017. Thermodynamic modeling of a novel and modified
604 Organic Rankine Cycle (ORC) augmented with ejector and regenerator. *Modares*
605 *Mechanical Engineering*, 16(13), pp.23-27.

606 Kousksou, T., Bruel, P., Jamil, A., El Rhafiki, T. and Zeraouli, Y., 2014. Energy
607 storage: Applications and challenges. *Solar Energy Materials and Solar Cells*, 120,
608 pp.59-80.

609 Krupke, C., Wang, J., Clarke, J. and Luo, X., 2016. Modeling and experimental study
610 of a wind turbine system in hybrid connection with compressed air energy
611 storage. *IEEE Transactions on Energy Conversion*, 32(1), pp.137-145.

612 McNevin, C. and Harrison, S.J., 2017. Multi-stage liquid-desiccant air-conditioner:
613 Experimental performance and model development. *Building and Environment*, 114,
614 pp.45-55.

615 Meng, H., Wang, M., Aneke, M., Luo, X., Olumayegun, O. and Liu, X., 2018.
616 Technical performance analysis and economic evaluation of a compressed air energy
617 storage system integrated with an organic Rankine cycle. *Fuel*, 211, pp.318-330.

618 Meng, H., Wang, M., Olumayegun, O., Luo, X. and Liu, X., 2019. Process design,
619 operation and economic evaluation of compressed air energy storage (CAES) for wind
620 power through modelling and simulation. *Renewable energy*, 136, pp.923-936.

621 Misra, R.D., Sahoo, P.K., Sahoo, S. and Gupta, A., 2003. Thermo-economic
622 optimisation of a single effect water/LiBr vapour absorption refrigeration system.
623 *International Journal of refrigeration*, 26(2), pp.158-169.

624 Mohammadi, A., Ahmadi, M.H., Bidi, M., Joda, F., Valero, A. and Uson, S., 2017.
625 Exergy analysis of a Combined Cooling, Heating and Power system integrated with
626 wind turbine and compressed air energy storage system. *Energy Conversion and*
627 *Management*, 131, pp.69-78.

628 Letcher, T.M. ed., 2020. *Future energy: improved, sustainable and clean options for*
629 *our planet*. Elsevier.

630 Luo, X., Wang, J., Dooner, M. and Clarke, J., 2015. Overview of current
631 development in electrical energy storage technologies and the application potential in
632 power system operation. *Applied energy*, 137, pp.511-536.

633 Quoilin, S., Van Den Broek, M., Declaye, S., Dewallef, P. and Lemort, V., 2013.
634 Techno-economic survey of Organic Rankine Cycle (ORC) systems. *Renewable and*
635 *Sustainable Energy Reviews*, 22, pp.168-186.

636 Radgen, P., 2008, November. Years Compressed Air Energy Storage Plant
637 Huntorf—Experiences and Outlook. In *Proceedings of the 3rd International Renewable*
638 *Energy Storage Conference*, Berlin, Germany (pp. 24-25).

639 Sadreddini, A., Fani, M., Aghdam, M.A. and Mohammadi, A., 2018. Exergy analysis
640 and optimisation of a CCHP system composed of compressed air energy storage system
641 and ORC cycle. *Energy conversion and management*, 157, pp.111-122.

642 Sioshansi, R., Denholm, P. and Jenkin, T., 2011. A comparative analysis of the value
643 of pure and hybrid electricity storage. *Energy Economics*, 33(1), pp.56-66.

644 Simpoire S., Garde, F. David, M. Marc O. and Castaing-Lasvignottes J., 2016. Design
645 and dynamic simulation of a compressed air energy storage system (CAES) coupled
646 with a building, an electric grid and photovoltaic power plant. *CLIMA 2016*, Aalborg,
647 Denmark (2016)

648 Soltani, M., Nabat, M.H., Razmi, A.R., Dusseault, M.B. and Nathwani, J., 2020. A

649 comparative study between ORC and Kalina based waste heat recovery cycles applied
650 to a green compressed air energy storage (CAES) system. *Energy Conversion and*
651 *Management*, 222, p.113203.

652 Somers, C., Mortazavi, A., Hwang, Y., Radermacher, R., Rodgers, P. and Al-
653 Hashimi, S., 2011. Modeling water/lithium bromide absorption chillers in ASPEN Plus.
654 *Applied Energy*, 88(11), pp.4197-4205.

655 Souza, R.J., Dos Santos, C.A.C., Ochoa, A.A.V., Marques, A.S., Neto, J.L.M. and
656 Michima, P.S.A., 2020. Proposal and 3E (energy, exergy, and exergoeconomic)
657 assessment of a cogeneration system using an organic Rankine cycle and an Absorption
658 Refrigeration System in the Northeast Brazil: Thermodynamic investigation of a
659 facility case study. *Energy conversion and Management*, 217, p.113002.

660 Wang, J., Yan, Z., Wang, M., Ma, S. and Dai, Y., 2013. Thermodynamic analysis
661 and optimisation of an (organic Rankine cycle) ORC using low grade heat source.
662 *Energy*, 49, pp.356-365.

663 Yan, Y., Zhang, C., Li, K. and Wang, Z., 2018. An integrated design for hybrid
664 combined cooling, heating and power system with compressed air energy
665 storage. *Applied Energy*, 210, pp.1151-1166.

666 Yao, E., Wang, H., Wang, L., Xi, G. and Maréchal, F., 2016. Thermo-economic
667 optimisation of a combined cooling, heating and power system based on small-scale
668 compressed air energy storage. *Energy Conversion and Management*, 118, pp.377-386.

669 Zhang, Y., Yang, K., Li, X. and Xu, J., 2014. Thermodynamic analysis of energy
670 conversion and transfer in hybrid system consisting of wind turbine and advanced
671 adiabatic compressed air energy storage. *Energy*, 77, pp.460-477.

672 Zhang, Y., Xu, Y., Zhou, X., Guo, H., Zhang, X. and Chen, H., 2017. Compressed
673 air energy storage system with variable configuration for wind power

674 generation. Energy Procedia, 142, pp.3356-3362.

675 Zhao, P., Dai, Y. and Wang, J., 2015. Performance assessment and optimization of a
676 combined heat and power system based on compressed air energy storage system and
677 humid air turbine cycle. Energy Conversion and Management, 103, pp.562-572.

678 Zhao, P., Gao, L., Wang, J. and Dai, Y., 2016. Energy efficiency analysis and off-
679 design analysis of two different discharge modes for compressed air energy storage
680 system using axial turbines. Renewable energy, 85, pp.1164-1177.

681 Zhou, Q., Du, D., Lu, C., He, Q. and Liu, W., 2019. A review of thermal energy
682 storage in compressed air energy storage system. Energy, 188, p.115993.

683