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1 **Sedimentary processes dominate nitrous oxide production and emission in**
2 **the hypoxic zone off the Changjiang River estuary**

3

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25

26 **Abstract**

27 Coastal oceans, known as the major nitrous oxide (N₂O) source to the atmosphere, are
28 increasingly subject to eutrophication and concurrent near-bottom hypoxia. The natural
29 nitrogen cycle is likely to be altered markedly in hypoxic coastal oceans. However, the
30 processes responsible for N₂O production and emission remain elusive because of lacking
31 field rate measurements simultaneously conducted in the water column and sediment. Here,
32 we quantified N₂O production rates using a ¹⁵N-labeled technique in the water-column and
33 surface sediments off the Changjiang (Yangtze) River estuary, the largest hypoxic zone in the
34 Pacific margins. Our results showed that the estuarine surface sediments were the major
35 source for N₂O production, accounting for approximately 90% of the total water-column
36 accumulation and consequent efflux of N₂O in the hypoxic zone, whereas the water-column
37 nitrification and denitrification combined only contributed <10%. More importantly, the
38 coupling of nitrification and denitrification at the presence of abundant supply and
39 remineralization of labile organic matter was the main driver of the N₂O release from the
40 sediment-water interface in this region. This study highlights the dominant role of benthic
41 processes occurring at the sediment-water interface controlling the coastal N₂O budget, as the
42 anthropogenic eutrophication and hypoxia are expanding in coastal oceans.

43

44 **Keywords:** coupled nitrification-denitrification; denitrification; nitrification; ¹⁵N-labeled
45 techniques; N₂O production rate; sediment-water interface

46 **1. Introduction**

47 Nitrous oxide (N₂O) is a powerful greenhouse gas that is ~300 times more potent in
48 warming potential than CO₂ and is known to destroy the stratospheric ozone (Ravishankara et
49 al., 2009), which contributes to global warming by altering radiative forcing. The
50 concentration of atmospheric N₂O has increased by 20% during Anthropocene, largely as a
51 result of human activities (Davidson, 2009). Identifying the dominant processes that
52 contribute to N₂O production and the environmental factors that affect N₂O production is thus
53 critical, but it remains poorly characterized (Bange et al., 2019; Kuypers et al., 2018; Prosser
54 et al., 2020).

55 The marine ecosystem is the second largest source of atmospheric N₂O (Seitzinger et
56 al., 2000). N₂O is known to be produced by nitrification and denitrification, two microbially-
57 mediated pathways, which extensively occur in the water column and sediments of marine
58 environments. Under the oxygenated condition, N₂O is produced as a by-product via
59 ammonium oxidation (i.e., the first step of nitrification). The N₂O production rates via
60 nitrification are elevated as oxygen level decreases (Ji et al., 2015; Löscher et al., 2012). By
61 contrast, denitrification (i.e., the stepwise reduction of nitrate or nitrite to N₂), occurring in
62 the suboxic/anoxic environments, forms N₂O as an intermediate product. Because of the O₂
63 inhibition of N₂O reduction to N₂, the hypoxic condition could be especially conducive to
64 promoting the N₂O production via denitrification (Bourbonnais et al., 2017; Ji et al., 2015).
65 Hence, the N₂O production via nitrification and denitrification are sensitive to ambient
66 oxygen levels.

67 Among various marine environments, diverse processes in coastal oceans including
68 estuaries and near-shore regions complicate marine nitrogen cycle. These regions are
69 identified as hotspots for N₂O production, accounting for up to 60% of the global oceanic
70 N₂O emissions (Bange et al., 1996). Coastal oceans are subject to severe human perturbation

71 for decades. The riverine nutrient loads have more than doubled over the last century due to
72 anthropogenic inputs (Beusen et al., 2016; Lee et al., 2011), leading to global expansion of
73 hypoxic zones in estuaries and near-shore regions (Breitburg et al., 2018). The eutrophication
74 and seasonal hypoxia have accelerated, particularly in some large river-dominated ocean
75 margins (RiOMars) in the vicinity of populated continents (Rabouille et al., 2008). In these
76 large RiOMars, the input of excess nitrogenous nutrients and resultant hypoxia are altering
77 the natural nitrogen cycling, and subsequently, the production and emission of N₂O
78 (Seitzinger and Kroeze, 1998). For instance, the N₂O emission off the Pearl River estuary
79 (severely perturbed by anthropogenic activities) is comparable in magnitude (1.67×10^9 g
80 year⁻¹) with the total amount from 19 European inner estuaries (Lin et al., 2016). Field studies
81 in the hypoxic RiOMars are thus crucial to validate the modeled marine N₂O fluxes and
82 evaluate its potential evolution in response to the ongoing changes of the oceans (Codispoti,
83 2010; Landolfi et al., 2017; Tan et al., 2020). However, previous studies on the pathways of
84 N₂O production are conducted in either the water column or the sediments, simultaneous
85 measurements of N₂O production in the water column and sediments are limited.

86 Compared to commonly used methods of natural stable isotopes and inhibitors, the
87 ¹⁵N-labeled techniques have advantages in quantifying the rates of N-cycling processes in
88 aquatic environments with dynamic N pools (Groffman et al., 2006). Moreover, the ¹⁵N-
89 labeled approaches have additional potentials in understanding and explaining complex N-
90 cycling processes in complicated estuarine environments (Marchant et al., 2016).

91 In this study we carried out shipboard ¹⁵N-labeled incubations to measure N₂O
92 production rates via various pathways in the water column and sediments during summer off
93 the Changjiang (Yangtze) River (CJR) estuary on the East China Sea (ECS) inner shelf,
94 which is the largest seasonal hypoxic RiOMars in the Pacific Ocean. The surrounding waters
95 off the CJR estuary are generally oversaturated in N₂O and thus found to release N₂O to the

96 atmosphere. The sea-to-air N₂O fluxes are significantly higher during late spring and summer
97 when hypoxia occurs on the inner shelf, accounting for more than 90% of annual fluxes in
98 this region (Chen et al., 2021). In addition, the bottom-water hypoxia is assumed to enhance
99 the N₂O emission in this region (Wang et al., 2016b). Therefore, the aim of the present study
100 was to gain insights into the major pathways of N₂O production and their regulators in the
101 hypoxic zone adjacent to the CJR estuary and improve our understanding on the present and
102 future role of such coastal systems in budgeting global atmospheric N₂O.

103

104 **2. Methods**

105 2.1. Study area

106 The CJR is the third largest river in the world with freshwater discharge as high as 9.2
107 $\times 10^{11}$ m³ year⁻¹. Inputs of anthropogenic nutrients have increased steadily over decades,
108 most of which are delivered into the coastal waters of the ECS (Dai et al., 2011; Kim et al.,
109 2011; Yan et al., 2003). The present annual load of dissolved inorganic nitrogen (DIN) is
110 eight times higher than that observed in 1960s, leading to frequent events of serious
111 eutrophication and harmful algal bloom off the CJR estuary (Chai et al., 2009).

112 Accompanying the elevated nutrient load, summer hypoxia occurs more frequently in the
113 bottom waters due to strong stratification of water column and decomposition of organic
114 matter (Wang et al., 2016a). The hypoxic area (DO < 62–94 $\mu\text{mol L}^{-1}$) off the CJR estuary
115 has been increasing, extending up to more than 20,000 km² during the early 2000s, and this
116 region has become one of the largest low-oxygen zones in global coastal waters (Chen et al.,
117 2007).

118

119 2.2. Sample collection and chemical analysis

120 The cruise was conducted using the R/V *Runjiang I* on August 15–24, 2011 on the ECS
121 inner shelf off the CJR and Qiantang River (QTR) estuaries (Fig. 1). Water samples for the
122 analyses of dissolved oxygen (DO), N₂O and nutrients were collected using 12 L Niskin
123 bottles attached to a conductivity-temperature-depth (CTD, SBE 911 SeaBird) rosette
124 sampler. DO and nutrients were analyzed aboard, while samples for N₂O were poisoned with
125 100 μL saturated HgCl₂ solution and kept at 4°C until shore-based analysis.

126 DO concentrations were analyzed using the Winkler method. Ammonium was
127 determined by the indophenol blue spectrophotometric method, and nitrite and nitrate were
128 measured using the Autoanalyzer III system (Hsiao et al., 2014). The detection limits for
129 ammonium, nitrite and nitrate were 0.16, 0.02 and 0.07 μmol L⁻¹, respectively (Zhang et al.,
130 2001). Water-column N₂O concentrations were analyzed by a purge and trap system (Tekmar
131 Velocity XPT) coupled with a gas chromatograph (see details in Text S1; Lin et al., 2016).
132 The detection limit and precision of N₂O measurement were 30 ppb and < ±5%, respectively.

133 Based on the geographical setting and surface-water salinity (Fig. 1 and 2a), the
134 sampling sites were classified as the river-mouth (sites Y0–Y0e and N1–N5 with salinity
135 lower than 20), the inner-plume (sites in the trapezoid with salinity of 20–29), and the high-
136 salinity (sites Y4–Y7, Y12 and Y19 with salinity higher than 29) zones. Physical and
137 chemical parameters measured at these sites provided the hydrographical background of the
138 study area, which showed the extent of the hypoxic zone and the influences of riverine inputs
139 over the study period (Fig. 2). N₂O production rates were determined in the inner-plume
140 zone, which was the previously reported hypoxic zone.

141 In the inner-plume zone (i.e., the hypoxic zone) the ¹⁵N-labeled incubation
142 experiments for the water and sediment samples were performed at twelve and nine stations,
143 respectively (Fig. 1). Water samples were collected at 2–5 layers from the surface to the
144 bottom and filled in 100 mL narrow-necked gas-tight glass bottles (Wheaton). Each bottle

145 was flushed twice and overflowed before being filled without any headspace to avoid air
146 contamination, and then sealed with a butyl rubber stopper and an open-top aluminum crimp
147 cap (CNW) (Lin et al., 2016). Twenty sample-filled bottles were recovered at each depth and
148 stored at room temperature for less than 1 h before the ^{15}N -labeled experiments.

149 Sediment samples were taken using a Soutar-type box corer and then subsampled by
150 the Plexiglas tubes (30 cm long, 4.5 cm diameter) on deck. All intact sediment cores were
151 carefully adjusted to 22 cm long with 8 cm of overlying water. These cores were
152 subsequently sealed with rubber stoppers, and immediately equilibrated in a barrel filled with
153 bottom waters at 25°C overnight (Tan et al., 2019). Four intact sediment cores were kept in
154 the dark at in situ temperature for 4 h. The total oxygen utilization (TOU) of the surface
155 sediments was determined based on the DO changes in the overlying waters (Rysgaard et al.,
156 2004). Approximately 0.1 g of dry surface sediment was acidified with 1 mL of 1 N HCl to
157 remove inorganic carbon. The decarbonated samples were used to measure sedimentary
158 organic carbon content (SOC%) by Carbon Analyzer (Horiba EMIA).

159

160 2.3. Measurements of N_2O production rates via nitrification and denitrification in the water
161 column

162 To determine the N_2O production rates via the water-column nitrification (rN_2O -
163 WCN), ^{15}N -labeled NH_4Cl (98 ^{15}N atom%, Sigma-Aldrich) were added to ten sample-filled
164 bottles using 100 μL gas-tight syringes (Hamilton, USA) to reach final concentrations of 10
165 $\mu\text{mol } ^{15}\text{N L}^{-1}$ inside. It is worth noting that, because of very low N_2O yield via nitrification,
166 higher concentrations of ^{15}N -labeled NH_4^+ relative to the ambient NH_4^+ concentrations were
167 added to ensure the produced N_2O measurable (Punshon and Moore, 2004).

168 These bottles were gently shaken to ensure the tracer well mixed. Two bottles were
169 immediately fixed by adding 0.2 mL of saturated HgCl_2 as initials. The remaining eight

170 bottles were incubated in a water bath at in situ temperature in the dark, and duplicate
171 samples were stopped by adding HgCl₂ at 6-h intervals over a 24-hour incubation. After
172 incubation, all the fixed samples were kept upside down for future isotopic analysis. The
173 same procedure and condition were used to determine the N₂O production rates via the water-
174 column denitrification (rN₂O-WCD), except for adding ¹⁵N-labeled NaNO₃ (98 ¹⁵N atom%,
175 Sigma-Aldrich) to reach final concentrations of 10 μmol ¹⁵N L⁻¹ in the bottles.

176 Isotopic signals of N₂O in the bottles were measured by isotope ratio mass
177 spectrometer (IRMS) mounted with a modified dilution pre-concentration system that
178 allowed measurements of the isotopic signals of N₂ at the same time (Hsu and Kao, 2013).
179 The rN₂O-WCN and rN₂O-WCD at each sampling depth were quantified based on the
180 increases of ⁴⁵N₂O and ⁴⁶N₂O over time (Δ⁴⁵N₂O and Δ⁴⁶N₂O):

$$181 \quad r_{\text{N}_2\text{O-WCN}} \text{ or } r_{\text{N}_2\text{O-WCD}} = (\Delta^{45}\text{N}_2\text{O} + 2 \times \Delta^{46}\text{N}_2\text{O}) / (F \times V) \quad (1)$$

182 where F represents the ¹⁵N fraction in the bottle at the beginning of incubation, V is the bottle
183 volume. The detection limit of the N₂O production rate was 0.005 nmol L⁻¹ d⁻¹.

184

185 2.4. Measurement of N₂O production rates from intact sediment cores

186 Twenty intact sediment cores at each site were used to perform a concentration-series
187 experiment to determine the rates of in situ N₂O production from sediments (Hsu and Kao,
188 2013; Trimmer et al., 2006). Briefly, Na¹⁵NO₃ (98 ¹⁵N atom%, Sigma-Aldrich) were added
189 into the in-site overlying waters of intact sediment cores to reach a concentration gradient in
190 range of 10–100 μmol ¹⁵N L⁻¹ with an increasing interval of 10 μmol ¹⁵N L⁻¹ following Tan
191 et al. (2019). The overlying water of intact sediment core was then carefully stirred by a small
192 stir bar ensuring that the sediment surface remained undisturbed throughout the pre-
193 incubation period. The pre-incubation time was 15–30 mins to achieve a stable ¹⁵N-labeled
194 signal reaching the sedimentary denitrification zone (Hsu and Kao, 2013). All cores were

195 sealed and incubated in the dark at in situ temperature. Ten intact sediment cores were
 196 sacrificed immediately after the pre-incubation as initials. The remaining cores were stopped
 197 after 3 h incubation. Because the produced amounts of N₂O and N₂ in the surface sediments
 198 were small during the incubation, we suggested that they were overwhelmingly dissolved in
 199 the pore water and the overlying water.

200 Generally, sedimentary nitrogen loss occurs within the top 2 cm of sediments off the
 201 CJR estuary, especially in summer (e.g., Wei et al., 2022). To stop the incubation, the top 2
 202 cm of sediments were mixed gently with the overlying waters using a plexiglass rod with a 2
 203 cm mark (Dalsgaard et al., 2000). Subsequently, 4 mL of mixed slurry was pipetted into a 12-
 204 mL gas-tight vial (Exetainer) filling with 100 µL of formaldehyde solution (38% w/v) and a
 205 glass bead (5 mm diameter) for mixing. After capping all vials, the vial headspace was
 206 purged with helium for 1 min to remove the remaining air inside for eliminating its possible
 207 effect on the measurement of the produced N₂ (Tan et al., 2019). All vials were kept upside
 208 down at room temperature before isotopic measurements. Isotopic signals of N₂O (⁴⁴N₂O,
 209 ⁴⁵N₂O, ⁴⁶N₂O) and N₂ (²⁸N₂, ²⁹N₂, ³⁰N₂) in the vials were measured using the same IRMS
 210 system as the water-sample measurements.

211 According to the revised ¹⁵N isotope pairing technique (IPT) proposed by Hsu and
 212 Kao (2013), the N₂O production rates via canonical denitrification (pN₂O-den) and coupled
 213 nitrification-denitrification (pN₂O-cnd) in sediments were quantified based on the production
 214 rates of ⁴⁵N₂O, ⁴⁶N₂O, ²⁹N₂ and ³⁰N₂ (P_{45} , P_{46} , P_{29} and P_{30}) and the genuine ratio between
 215 ¹⁴NO₃⁻ and ¹⁵NO₃⁻ undergoing nitrate reduction (r_{14}), using the following equations:

$$216 \quad \text{pN}_2\text{O-den} = 2 \times r_{14} \times (r_{14} + 1) \times P_{46} \quad (2)$$

$$217 \quad \text{pN}_2\text{O-cnd} = 2 \times r_{14} \times (P_{45} - P_{46} \times 2 \times r_{14}). \quad (3)$$

$$218 \quad \text{pN}_2\text{O-SED} = \text{pN}_2\text{O-den} + \text{pN}_2\text{O-cnd} = 2 \times r_{14} \times (P_{45} + (1 - r_{14}) \times P_{46}) \quad (4)$$

219 where pN_2O -SED denotes the gross sedimentary N_2O production rate. The r_{14} is related to the
 220 ^{15}N proportion in the total produced N_2O pool (qN_2O) or that in the total produced N_2 pool
 221 (qN_2) (Trimmer et al., 2006). Theoretically if the canonical denitrification is the sole process
 222 to reduce $^{15}NO_3^-$ in sediments, the qN_2O would be equal to the qN_2 . In other words, the slope
 223 of qN_2O vs. qN_2 (S_{N_2O/N_2}) would be 1. Anammox could form hybrid N_2 with one nitrogen
 224 atom from unlabeled nitrogen pool and the other from the ^{15}N -labeled pool, while the coupled
 225 nitrification-denitrification could form hybrid N_2O . Therefore, when anammox or the coupled
 226 nitrification-denitrification occurring in sediments contributes substantially to N_2 or N_2O
 227 production, the S_{N_2O/N_2} is expected to be larger or lower than one. In this case, the r_{14} may be
 228 biased at the presence of anammox or the coupled nitrification-denitrification. We thus
 229 estimated the relative contribution of the pN_2O -cnd to the pN_2O -SED (R_{cnd} , %), using the
 230 following equation that is related to the S_{N_2O/N_2} (Trimmer et al., 2006):

$$231 \quad R_{cnd} = (2 - 2 \times S_{N_2O/N_2}) / (2 - S_{N_2O/N_2}) \quad (5).$$

232 By combining the equations 2–5, the pN_2O -den, pN_2O -cnd, pN_2O -SED and r_{14} can be
 233 calculated, respectively. In addition, we also calculated the N_2 production rates in the surface
 234 sediments (pN_2 -SED) using the following equation (Hsu and Kao, 2013):

$$235 \quad pN_2\text{-SED} = 2 \times (r_{14} + 1) \times r_{14} \times P_{30} \quad (6).$$

236

237 2.5. Sea-to-air N_2O flux

238 Sea-to-air N_2O flux (F_{N_2O} , $\mu\text{mol m}^{-2} \text{d}^{-1}$) was estimated using equation 6:

$$239 \quad F_{N_2O} = k \times (C_{\text{obs}} - C_{\text{eq}}) = k \times \Delta N_2O \quad (7)$$

240 where C_{obs} is the observed concentration of dissolved N_2O in seawater; and C_{eq} is the
 241 dissolved N_2O concentration at in situ temperature and salinity that is equilibrated with its
 242 atmospheric concentration (Weiss and Price, 1980). The global mean atmospheric N_2O
 243 concentration of 324.2 ppb was used for the calculation in this study. The calculated C_{eq}

244 values in the surface waters of the sampling sites ranged from 6.5 to 7.2 nmol L⁻¹ (Text S2
245 and Table S1). ΔN_2O means the discrepancy between C_{obs} and C_{eq} , indicating the excess N_2O .
246 k (cm h⁻¹) denotes the gas transfer velocity, which is expressed as a function of the wind
247 speed and the Schmidt Number (Sc) derived from temperature. In this study, k was calculated
248 using the following equation given by Wanninkhof (1992):

$$249 \quad k = 0.39 \times u_{10}^2 \times (Sc / 660)^{-0.5} \quad (8)$$

250 where u_{10} is the wind speed at the height of 10 m, with an average value of 5.2 ± 2.6 m s⁻¹
251 observed during the entire cruise. In addition, N_2O saturation (R_{N_2O} , %) was estimated using
252 equation 8:

$$253 \quad R_{N_2O} = (C_{obs} / C_{eq}) \times 100 \quad (9)$$

254

255 2.6. Statistical analysis

256 Correlations of concentrations and production rates of N_2O with environmental factors
257 were tested using Pearson's correlation. A one-way analysis of variance (ANOVA) was used
258 to determine the significant differences in N_2O production rates via water-column
259 nitrification and denitrification. The statistical analyses were performed using SPSS at a 0.05
260 significance level unless otherwise indicated.

261

262 3. Results

263 3.1. Hydrochemistry in the water column

264 Both temperature and salinity in the surface and bottom waters showed zonal
265 distributions across the study area (Fig. 2 and Fig. S1). The lowest salinity (~ 0.2) and highest
266 temperature ($> 29^\circ C$) were found in the surface water at site Y0 off the CJR mouth. The low-
267 salinity waters observed in the study area were primarily related to the CJR plume, because
268 the CJR discharge during the study period was more than 80 times higher than the QTR

269 discharge (MWR of China, 2012). As a result of mixing with ambient open ocean waters,
270 salinity increased dramatically up to 34.4 in the bottom water at site Y4. The temperature-
271 salinity (T-S) properties in the study area were controlled by three water masses, the CJR
272 freshwater, the northward Nearshore Kuroshio Branch Current (NKBC) and the westward
273 East China Sea Surface Water (ECSSW) (Fig. 1 and 3a; Yan et al., 2017).

274 Concentrations of dissolved inorganic nitrogen (DIN, in which nitrate comprised more
275 than 98%) in the surface and bottom waters were higher than $130 \mu\text{mol L}^{-1}$ near the CJR and
276 QTR mouths but decreased to $< 4 \mu\text{mol L}^{-1}$ at the offshore sites (Fig. 2). Our measurements
277 showed that the distribution of DIN was largely controlled by the mixing of the three water
278 masses indicated above (Fig. 3b). The freshwater with high level of DIN was diluted by the
279 N-depleted ECSSW and low- NO_3^- NKBC. However, the surface DIN at some high-salinity
280 sites showed non-conservative behavior, as evidenced by some data points deviated from the
281 conservative mixing lines of the three water masses (Fig. 3b). This was likely due to nitrate
282 assimilation by phytoplankton (Yan et al., 2017). In addition, NH_4^+ concentrations were
283 generally low in the study area, ranging from undetectable to $2.0 \mu\text{mol L}^{-1}$ (Fig. S1).

284 DO concentrations were high in the river-mouth ($151\text{--}194 \mu\text{mol L}^{-1}$) and high-
285 salinity zones ($140\text{--}249 \mu\text{mol L}^{-1}$), but low ($62\text{--}183 \mu\text{mol L}^{-1}$) in the inner-plume zone (Fig.
286 2e and f), indicating that the DO distribution was not mainly regulated by the water mixing.
287 Low levels of DO down to $\sim 62.0 \mu\text{mol L}^{-1}$ (i.e., hypoxia) were detected in both the surface
288 and bottom waters of the inner-plume zone. Such low concentrations of DO during summer
289 were consistent to the previous report in this region (Zhu et al., 2011).

290

291 3.2. Surface sediment characteristics

292 The SOC% varied from 0.19% to 0.64%, with the lowest values at the deeper sites Y8
293 and Y4 where the bottom-water DO concentrations were low ($81\text{--}108 \mu\text{mol L}^{-1}$; Table 1).

294 The TOU in the surface sediments ranged from 11.0 to 53.6 mmol m⁻² d⁻¹ (Table 1). The
295 lowest TOU was found at the shallower sites N0 and Y1 near the river-mouth zone. At site
296 Y8 near the hypoxic center, we found the highest TOU in the surface sediments.

297

298 3.3. Distribution of N₂O concentration and its sea-to-air flux in the hypoxic zone

299 N₂O concentrations at most sites were high below the surface waters (Fig. 2 and S2).
300 This vertical pattern is commonly observed in this region (Wang et al., 2016b; Zhang et al.,
301 2010), and in other coastal waters (e.g., the northern South China Sea and Bay of Bengal)
302 (Han et al., 2013; Rao et al., 2013). Spatially, N₂O concentrations showed similar
303 distributions in the surface and bottom waters. A substantially higher N₂O concentration, up
304 to 17.8 nmol L⁻¹, was observed near the CJR mouth. The values decreased markedly to less
305 than 10.0 nmol L⁻¹ away from the river mouth at site Y0e. The N₂O concentrations off the
306 QTR mouth, ranging from 7.0 to 9.0 nmol L⁻¹, were considerably lower (Fig. 2g). In the
307 inner- and high-salinity zones we observed a patch of elevated N₂O concentrations below the
308 surface waters (Fig. 2g and h), which was spatially associated with the hypoxic core.
309 According to the mixing lines of three water masses we found that the N₂O was mainly
310 accumulated in the intermediate and bottom waters, suggesting its non-conservative nature
311 (Fig. 3c).

312 In the hypoxic zone the average concentration of surface N₂O was 9.4±1.6 nmol L⁻¹,
313 higher than the average surface-water N₂O concentration in equilibrium with the atmosphere
314 (6.4±0.2 nmol L⁻¹). The surface N₂O saturation ranged from 111% to 175% with an average
315 value of 144±21% (Table 2). An average bottom-water N₂O concentration was 10.6±2.2
316 nmol L⁻¹ and was also oversaturated (111–194%, Table 2). The sea-to-air N₂O fluxes in the
317 hypoxic zone were 3.6–15.0 μmol m⁻² d⁻¹ with a mean value of 8.9±4.1 μmol m⁻² d⁻¹ during
318 the study period (Fig. 4a and Table 2). We also found that the sea-to-air N₂O flux was highest

319 in the vicinity of the low-oxygen center. Our values were in good agreement with the
320 previous observations made off the CJR estuary during summer (Chen et al., 2021). A
321 hotspot for N₂O release is commonly present during summer in the hypoxic zone off the CJR
322 estuary.

323

324 3.4. Rates of N₂O production in the water column

325 In the hypoxic zone off the CJR estuary, the rN₂O-WCD rates were found to be
326 extremely low, ranging from 0 to 0.11 nmol L⁻¹ d⁻¹ with an average of 0.01±0.02 nmol L⁻¹ d⁻¹
327 (Table S2). The rN₂O-WCD rates were below the detection limit at most sites. The highest
328 rN₂O-WCD rate was observed in the bottom water at site Y9a with the lowest DO
329 concentration. By contrast, the rN₂O-WCN rates varied between 0–0.24 nmol L⁻¹ d⁻¹ with a
330 mean value of 0.04±0.05 nmol L⁻¹ d⁻¹ (Table S2). The rN₂O-WCN rates were generally
331 higher below the surface waters. Similarly, the maximum rN₂O-WCN rate was found in the
332 bottom water at site Y8 close to the hypoxic center.

333 With the rN₂O-WCD and rN₂O-WCN at the sampling depth, we calculated the depth-
334 integrated N₂O production via water-column denitrification and nitrification (pN₂O-WCD
335 and pN₂O-WCN), respectively (Table 2). The pN₂O-WCD rates were generally at very low
336 levels (0–1.37 μmol m⁻² d⁻¹) and spatially homogeneous, whereas the pN₂O-WCN rates
337 ranged between 0.11–4.74 μmol m⁻² d⁻¹ with higher values found at the low-oxygen center
338 (Fig. 4). The average pN₂O-WCD and pN₂O-WCN rates in the hypoxic zone were estimated
339 to be 0.26±0.50 μmol m⁻² d⁻¹ and 1.39±1.48 μmol m⁻² d⁻¹, respectively (Table 2). In
340 summary, the N₂O production rates via the water-column nitrification were considerably
341 higher than those via the water-column denitrification (ANOVA, *p* < 0.05).

342

343 3.5. Rates of N₂O and N₂ production in the surface sediments

344 From the concentration-series experiments of the intact sediment cores the S_{N_2O/N_2}
345 values were found to vary between 0.25 and 0.89 across the study area (Table 2), indicating
346 that the q_{N_2O} were lower than the q_{N_2} at all sampling sites. The potential anammox activities
347 were not determined from the anoxic slurry incubation experiments (Texts S3 and S4; Fig.
348 S3). Similarly, only a minor contribution of anammox to total sedimentary nitrogen removal
349 was observed off the CJR estuary (Liu et al., 2019). These findings suggested that the q_{N_2} in
350 the study area were not significantly diluted by the effect of the anammox process. By
351 contrast, the q_{N_2O} in the surface sediments were diluted by the hybrid N_2O production via
352 the coupled nitrification-denitrification that incorporated unlabeled nitrogen from NH_4^+ or
353 organics (see below). The R_{cnd} was in the range of 19.9–85.9% (average of $58.6 \pm 22.7\%$) with
354 values greater than 50% at seven of nine sites (Table 2).

355 The p_{N_2O} -den ranged from 1.6 to $9.0 \mu\text{mol m}^{-2} \text{d}^{-1}$ with the lowest rate found at site
356 Y1 and the highest rate at site Y3. The p_{N_2O} -cnd varied by over an order of magnitude, from
357 2.0 to $31.0 \mu\text{mol m}^{-2} \text{d}^{-1}$. Unlike to the p_{N_2O} -den showing a patchy distribution, the p_{N_2O} -
358 cnd increased seaward from the vicinity of the river mouth to the low-oxygen zone. The
359 highest p_{N_2O} -cnd was found at site Y8 where was close to the hypoxic center and the TOU
360 in the surface sediments was highest. By contrast, the lowest p_{N_2O} -cnd was found at sites Y1
361 and N0 near the river mouth where the bottom-water DO was high and the sedimentary TOU
362 was relatively low (Tables 1 and 2). In summary, the p_{N_2O} -SED ranged from 4.4 to 36.0
363 $\mu\text{mol m}^{-2} \text{d}^{-1}$ (mean of $15.0 \pm 9.4 \mu\text{mol m}^{-2} \text{d}^{-1}$), showing a distribution pattern similar to the
364 p_{N_2O} -cnd (Fig. 4).

365 The p_{N_2} -SED ranged from 0.3 to $6.3 \text{mmol m}^{-2} \text{d}^{-1}$ off the CJR estuary, with higher
366 rates near the river mouth and lower rates close to the hypoxic center (Fig. 4g and Table S3).
367 The distribution of the p_{N_2} -SED was roughly contrasted with that of the p_{N_2O} -SED. The
368 highest $N_2O:N_2$ production ratio (11.80%) in sediments was found at site Y8 where the p_{N_2} -

369 SED was lowest and the pN₂O-SED was highest (Fig. 4h). The sedimentary N₂O:N₂
370 production ratios at other sites in the hypoxic zone off the CJR estuary were relatively low
371 and varied within a narrow range (0.09–1.18%; Table S3).

372

373 **4. Discussion**

374 This is, to our knowledge, the first report of simultaneous measurements of the N₂O
375 production rates in the water column and sediments in the hypoxic zone off the CJR estuary.
376 The results allow us to understand how the environmental factors affect the water-column
377 and sedimentary N₂O production and offer clear evidence for the main driver of N₂O
378 production and emission in this hypoxic zone.

379 4.1. Key environmental factors affecting the water-column N₂O production

380 The rN₂O-WCD rates were undetectable at most sites (Fig. 4b), indicating that the
381 water-column denitrification was suppressed in this hypoxic zone over the study period. This
382 is likely because the water-column denitrification can only efficiently produce N₂O at the
383 oxic-anoxic interface (DO < 5 μmol L⁻¹; Codispoti et al., 2001). Although the extent of
384 hypoxia off the CJR estuary has been increasing since 1950s, the bottom-water DO minimum
385 varies between 10 and 60 μmol L⁻¹ all the time (Zhu et al., 2011). Hence, the contribution of
386 the water-column denitrification to the water-column N₂O production is very limited off the
387 CJR estuary.

388 The rN₂O-WCN rates were significantly higher than the rN₂O-WCD rates in this
389 hypoxic zone (n = 42, p < 0.01), suggesting that nitrifying microorganisms were more active
390 relative to denitrifiers. This finding was supported by the evidence of microbial communities
391 and geochemical indicators observed during this cruise. For example, Zhang et al. (2014)
392 reported that the copy numbers of the ammonia monooxygenase gene (including bacterial and
393 archaeal *amoA*) were significantly higher than those of the dissimilatory nitrite reductase

394 gene *nirS* in this region. Dual isotope measurements revealed that the nitrate dynamics in this
395 summer was predominated by the water-column nitrification, rather than the water-column
396 denitrification (Yan et al., 2017). On the other hand, increasing rates of the water-column
397 N₂O production were strongly associated with decreasing DO concentrations in the hypoxic
398 zone (Fig. S4a). Neither DIN nor particle concentrations in the water column were
399 significantly correlated with the water-column N₂O production rates (Fig. S5). Therefore,
400 these results suggested that ambient DO concentration was the important factor regulating the
401 N₂O formation in the water column and that low-oxygen condition enhanced the N₂O
402 production.

403 Notably, the total water-column N₂O production rate of $1.7 \pm 1.6 \mu\text{mol m}^{-2} \text{d}^{-1}$ in the
404 hypoxic zone off the CJR estuary is at the low limit of the reported N₂O production rates in
405 other estuarine waters (Murray et al., 2015). Considering that the water-column nitrification
406 rates were 100–3200 nmol L⁻¹ d⁻¹ during this cruise (Hsiao et al., 2014), we found that the
407 average N₂O yield via nitrification was only 0.007%. This value is lower than that found in
408 other coastal waters (0.01–0.42%; de Wilde and de Bie, 2000; Punshon and Moore, 2004).
409 The optimum condition for nitrifiers to produce N₂O is expected to be 10–30 $\mu\text{mol O}_2 \text{L}^{-1}$ or
410 O₂ saturation of 2–15% (de Bie et al., 2002; Punshon and Moore, 2004). We thus speculated
411 that the nitrifiers were not efficiently conducting N₂O production, since the water-column DO
412 concentrations were not low enough over the study period. Caution is needed on the declining
413 oxygen in estuaries and coastal waters (Breitburg et al., 2018), which may result in an
414 enhancement of the water-column N₂O production.

415

416 4.2. Key environmental factors affecting the sedimentary N₂O production

417 The incubation results showed that the surface sediments were a net N₂O source in the
418 hypoxic zone off the CJR estuary. The measured in situ sedimentary N₂O production rates in

419 this study ($4.4\text{--}36.0 \mu\text{mol m}^{-2} \text{d}^{-1}$) fall into the reported ranges of the sedimentary N_2O fluxes
420 ($2.4\text{--}240 \mu\text{mol m}^{-2} \text{d}^{-1}$) observed in other eutrophic RiOMars (Murray et al., 2015; Sun et al.,
421 2014; Tan et al., 2019). Previous study has reported that the potential sedimentary N_2O
422 production rates near the river-mouth zone are higher than those in the hypoxic zone based on
423 the slurry experiments (Lin et al., 2017). However, the in situ sedimentary N_2O production
424 rates are comparable between the two zones off the CJR estuary based on the intact sediment
425 experiments ($0.1\text{--}36.0 \mu\text{mol m}^{-2} \text{d}^{-1}$ observed near the river-mouth zone; Chen et al., 2021;
426 Chen et al., 2022; Wang et al., 2007; Zhang et al., 2010). Thus, the sediment N_2O release in
427 the hypoxic zone may play an important role in contributing to the regional N_2O budget.

428 The sedimentary N_2O production was predominated by coupled nitrification-
429 denitrification especially at sites with relatively low DO concentrations (Fig. 5). As a result,
430 the $p\text{N}_2\text{O}\text{-cnd}$ were closely correlated with the $p\text{N}_2\text{O}\text{-SED}$ and bottom-water DO (Table 3).
431 The importance of nitrification-related pathway to sedimentary N_2O production may be
432 related to the abundance of autotrophic and heterotrophic nitrifiers that simultaneously
433 consume organic carbon and ammonium in the surface sediments off the CJR estuary (Jin et
434 al., 2017; Zheng et al., 2014). We also noted that the $p\text{N}_2\text{O}\text{-den}$ and $p\text{N}_2\text{-SED}$ had no
435 significant relationship with bottom-water NO_3^- , suggesting that the major substrate for
436 denitrification did not come from the nitrate diffused from the overlying waters into
437 sediments. Instead, the nitrification-derived nitrate may primarily fuel the sedimentary
438 denitrification. A tight interaction between denitrification and nitrification generally occurs at
439 the oxic-anoxic interface of estuarine sediments and is conducive to sedimentary N_2O
440 production (Liu et al., 2019; Tan et al., 2019). The $p\text{N}_2\text{O}\text{-cnd}$ were positively correlated with
441 the TOU in the surface sediments and the R_{cnd} were negatively correlated with the SOC%
442 (Table 3). In addition, the positive relationship was observed between the sedimentary
443 $\text{N}_2\text{O}:\text{N}_2$ production ratio and the TOU. Remineralization of organic matter in sediments

444 provides the substrate (i.e., NH_4^+) for the nitrification and creates anaerobic condition for
445 denitrification. Recent studies have indicated that the quality of sediment organic matter
446 determines the N_2O fluxes in coastal sediments, showing that marine organic matter
447 stimulates N_2O production (Chen et al., 2022; Lin et al., 2017). During this cruise particulate
448 organic matter was mainly of marine source produced in situ, and its remineralization
449 dominantly contributed to oxygen consumption in this region (Wang et al., 2016a). Thus,
450 increasing remineralization of labile sedimentary organic matter may lead to increases in the
451 N_2O production via coupled nitrification-denitrification in the surface sediments.

452 Active nitrification occurs at the sediment-water interface, in contrast to
453 denitrification occurring in the deeper anoxic sediment layers (Hou et al., 2007; Wankel et
454 al., 2017). The N_2O produced by sedimentary nitrification would easily diffuse into the
455 overlying water and subsequently be released to the atmosphere (Meyer et al., 2008). Results
456 from this study highlight a substantial role of sedimentary nitrification in stimulating N_2O
457 production and emission in the hypoxic RiOMars.

458

459 4.3. The N_2O budget in the hypoxic zone off the Changjiang estuary

460 The hypoxic zone off the CJR estuary acted as a net source of N_2O to the atmosphere,
461 with higher fluxes in the low-oxygen center (Fig. 4a). On the other hand, the positive $\Delta\text{N}_2\text{O}$
462 values and its inventory suggested the N_2O accumulation in the water column of the hypoxic
463 zone (Fig. S6). The $\Delta\text{N}_2\text{O}$ values were positively correlated with the apparent oxygen
464 utilization (AOU; Pearson correlation, $R^2 = 0.35$, $p < 0.001$) and negatively correlated with
465 DO (Pearson correlation, $R^2 = 0.41$, $p < 0.001$) in this region (Fig. 3d and S4b), indicating
466 that the accumulation of N_2O was enhanced under the low-oxygen condition. These findings
467 were consistent with higher rates of in situ N_2O production observed at low-oxygen sites
468 (Fig. 4).

469 To further evaluate the relative contributions of the water-column and sedimentary
470 processes to the N₂O budget in the hypoxic zone, we considered that the exchange of waters
471 in the hypoxic zone with the surrounding waters were limited over the study period. This
472 assumption was supported by the results that strong stratification of the water column and
473 gradual formation of subsurface hypoxia were observed over the study period (Fig. S7; Wang
474 et al., 2016a). External pathways had limited influence on the N₂O budget in the hypoxic
475 zone during the study period. Explicitly, the high levels of N₂O from river input were
476 confined to the areas near the river mouth only and the N₂O concentrations rapidly decreased
477 away from the river mouth (Fig. 2g and 3c). This finding suggested that most of N₂O in the
478 river-mouth zone were released to the atmosphere because of the high N₂O supersaturation
479 (up to 220%) in the water column. Thus, the N₂O production via the water-column and
480 sedimentary processes was the major source of N₂O and the sea-to-air emission was the main
481 sink of N₂O (Fig. 6).

482 In a steady state, the gross N₂O production within the system was balanced by the
483 sum of the sea-to-air N₂O flux and the water-column N₂O accumulation taking into account
484 the water residence time (τ). The amount of the water-column N₂O accumulation, determined
485 by the Δ N₂O inventory, was $134.8 \pm 101.5 \mu\text{mol m}^{-2}$ (Table 2). By dividing the Δ N₂O
486 inventory by the difference between the gross rate of the water-column and sedimentary N₂O
487 production ($16.7 \pm 9.5 \mu\text{mol m}^{-2} \text{d}^{-1}$) and the sea-to-air N₂O flux ($8.9 \pm 4.1 \mu\text{mol m}^{-2} \text{d}^{-1}$; Table
488 2), the τ was calculated to be 17.3 ± 1.5 days. The estimated τ was in agreement with a
489 previous report of the water exchange time (~ 16 days) off the CJR estuary based on the water
490 and salt budgets (Li et al., 2011), strengthening the plausibility of this estimate. Our
491 observation showed that in the hypoxic zone off the CJR estuary the sedimentary N₂O
492 production overwhelmingly contributed to the water-column N₂O accumulation and the N₂O
493 emission to the atmosphere ($\sim 90\%$) relative to a minor contribution ($<10\%$) from the water-

494 column N₂O production. As a result, the distribution of the pN₂O-SED was spatially
495 coincident with those of the sea-to-air N₂O fluxes and the ΔN₂O inventory (Figs. 4 and S6).
496 Taken together, our results suggest the predominant role of sedimentary processes in
497 regulating the N₂O budget of the hypoxic RioMars.

498

499 **5. Concluding remarks**

500 Results from this study showed that the low-oxygen levels may increase the water-
501 column and sedimentary N₂O production rates in the hypoxic zone off the CJR estuary. We
502 concluded that the benthic processes played the major role in the accumulation of water-
503 column N₂O and its emission to the atmosphere. Such high rates of sedimentary N₂O
504 production were mainly supported by coupled nitrification-denitrification, which was likely
505 related to the input and remineralization of labile sedimentary organic matter. These
506 quantitative results provide evidence of the main mechanisms which result in the hypoxic
507 zone of this human-perturbed margin as a significant source of N₂O.

508 As severe surface eutrophication and near-bottom hypoxia in the estuarine and coastal
509 systems are predicted to increase, amounts of fresh labile organic matter deposited onto
510 surface sediments will increase. This may lead to an increase in N₂O production from the
511 sediment-water interface and subsequent N₂O emission to the atmosphere. Such effects could
512 be expected in this area (Lin et al., 2017) and other renowned human-perturbed hypoxic
513 coastal oceans, such as the Indian coast, the Gulf of Mexico and the Baltic Sea (Conley et al.,
514 2011; Kim et al., 2013; Naqvi et al., 2000). Therefore, our study highlights that as areas of
515 coastal hypoxia increase, it is likely that coastal sediments will result in increasing N₂O
516 fluxes to the atmosphere and hence enhance the global greenhouse effect. This study thus has
517 implications for decision-making that seeks to mitigate N₂O emission and its impact on
518 climate change.

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529 **References**

- 530 Bange, H.W., Arévalo-Martínez, D.L., de la Paz, M., Farías, L., Kaiser, J., Kock, A., et al.,
531 2019. A Harmonized Nitrous Oxide (N₂O) Ocean Observation Network for the 21st
532 Century. *Front. Mar. Sci.* 6, 157. doi:10.3389/fmars.2019.00157
- 533 Bange, H.W., Rapsomanikis, S., Andreae, M.O., 1996. Nitrous oxide in coastal waters.
534 *Global Biogeochem. Cycles* 10, 197-207. doi:10.1029/95GB03834
- 535 Beusen, A.H.W., Bouwman, A.F., Van Beek, L.P.H., Mogollón, J.M., Middelburg, J.J., 2016.
536 Global riverine N and P transport to ocean increased during the 20th century despite
537 increased retention along the aquatic continuum. *Biogeosciences* 13, 2441-2451.
538 doi:10.5194/bg-13-2441-2016
- 539 Bourbonnais, A., Letscher, R.T., Bange, H.W., Échevin, V., Larkum, J., Mohn, J., et al.,
540 2017. N₂O production and consumption from stable isotopic and concentration data in the
541 Peruvian coastal upwelling system. *Global Biogeochem. Cycles* 31, 678-698.
542 doi:10.1002/2016GB005567
- 543 Breitburg, D., Levin, L.A., Oschlies, A., Grégoire, M., Chavez, F.P., Conley, D.J., et al.,
544 2018. Declining oxygen in the global ocean and coastal waters. *Science* 359, eaam7240.
545 doi:10.1126/science.aam7240
- 546 Chai C., Yu, Z., Shen, Z., Song, X., Cao, X, Yao, Y., 2009. Nutrient characteristics in the
547 Yangtze River Estuary and the adjacent East China Sea before and after impoundment of
548 the Three Gorges Dam, *Sci. Total Environ.* 407, 4687-4695.
549 doi:10.1016/j.scitotenv.2009.05.011
- 550 Chen, C.-C., Gong, G.-C., Shiah, F.-K., 2007. Hypoxia in the East China Sea: One of the
551 largest coastal low-oxygen areas in the world. *Mar. Environmen. Res.* 64, 399-408.
552 doi:10.1016/j.marenvres.2007.01.007

553 Chen, X., Ma, X., Gu, X., Liu, S., Song, G., Jin, H., et al., 2021. Seasonal and spatial
554 variations of N₂O distribution and emission in the East China Sea and South Yellow Sea.
555 *Sci. Total Environ.* 775, 145715. doi:10.1016/j.scitotenv.2021.145715

556 Chen, S., Wang, D., Yu, Z., Nie, J., Chen, J., Li, Y., et al., 2022. Quality of sediment organic
557 matter determines the intertidal N₂O response to global warming. *J. Geophys. Res. -*
558 *Biogeosciences*. doi:10.1029/2021JG006572

559 Codispoti, L.A., 2010. Interesting times for marine N₂O. *Science* 327, 1339-1340.

560 Codispoti, L.A., Brandes, J.A., Christensen, J.P., Devol, A.H., Naqvi, S.W.A., Paerl, H.W., et
561 al., 2001. The oceanic fixed nitrogen and nitrous oxide budgets: moving targets as we
562 enter the anthropocene? *Sci. Mar.* 65, 85-105. doi:10.3989/scimar.2001.65s285

563 Conley D.J., Carstensen, J., Aigars, J., Axe, P., Bonsdorff, E., Eremina, T., et al., 2011.
564 Hypoxia is increasing in the coastal zone of the Baltic Sea. *Environmen. Sci. Technol.* 45,
565 6777-6783. doi:10.1021/es201212r

566 Dai, Z., Du, J., Zhang, X., Su, N. and Li, J., 2011. Variation of riverine material loads and
567 environmental consequences on the Changjiang (Yangtze) Estuary in Recent Decades
568 (1955–2008). *Environmen. Sci. Technol.* 45, 223-227. doi:10.1021/es103026a

569 Dalsgaard, T., Nielsen, L.P., Brotas, V., Viaroli, P., Underwood, G.J.C., Nedwell, D.B., et al.,
570 2000. Protocol Handbook for NICE - Nitrogen Cycling in Estuaries: A Project under the
571 EU Research Programme : Marine Science and Technology (MAST III), National
572 Environmental Research Institute, Denmark.

573 Davidson, E.A. 2009. The contribution of manure and fertilizer nitrogen to atmospheric
574 nitrous oxide since 1860. *Nat. Geosci.* 2, 659-662. doi:10.1038/NGEO608

575 de Bie, M.J.M., Middelburg, J.J., Starink, M., Laanbroek, H.J., 2002. Factors controlling
576 nitrous oxide at the microbial community and estuarine scale. *Mar. Ecol. Progr. Ser.* 240,
577 1-9. doi:10.3354/meps240001

578 de Wilde, H.P.J., de Bie, M.J.M., 2000. Nitrous oxide in the Schelde estuary: production by
579 nitrification and emission to the atmosphere. *Mar. Chem.* 69, 203-216.
580 doi:10.1016/S0304-4203(99)00106-1

581 Han, Y., Zhang, G.-L., Zhao, Y.-C., Liu, S.-M., 2013. Distributions and sea-to-air fluxes of
582 nitrous oxide in the coastal and shelf waters of the northwestern South China Sea. *Estuar.
583 Coast. Shelf Sci.* 133, 32-44. doi:10.1016/j.ecss.2013.08.001

584 Hou, L.J., Liu, M., Xu, S.Y., Ou, D.N., Yu, J., Cheng, S.B., et al., 2007. The effects of semi-
585 lunar spring and neap tidal change on nitrification, denitrification and N₂O vertical
586 distribution in the intertidal sediments of the Yangtze estuary, China. *Estuar. Coast. Shelf
587 Sci.* 73, 607-616. doi:10.1016/j.ecss.2007.03.002

588 Hsiao, S.S.-Y., Hsu, T.-C., Liu, J.-W., Xie, X., Zhang, Y., Lin, J., et al., 2014. Nitrification
589 and its oxygen consumption along the turbid Chang Jiang River plume. *Biogeosciences*
590 11, 2083-2098. doi:10.5194/bg-11-2083-2014

591 Hsu, T.C., Kao, S.J., 2013. Technical Note: Simultaneous measurement of sedimentary N₂
592 and N₂O production and a modified ¹⁵N isotope pairing technique. *Biogeosciences* 10,
593 7847-7862. doi:10.5194/bg-10-7847-2013

594 Ji, Q., Babbin, A.R., Jayakumar, A., Oleynik, S., Ward, B.B., 2015. Nitrous oxide production
595 by nitrification and denitrification in the Eastern Tropical South Pacific oxygen minimum
596 zone. *Geophys. Res. Lett.* 42, 10755-10764. doi:10.1002/2015GL066853

597 Jin, Q., Lu, J., Wu, J., Luo, Y., 2017. Simultaneous removal of organic carbon and nitrogen
598 pollutants in the Yangtze estuarine sediment: The role of heterotrophic nitrifiers. *Estuar.
599 Coast. Shelf Sci.* 191, 150-156. doi:10.1016/j.ecss.2017.04.019

600 Kim, I.N., Lee, K., Bange, H.W., Macdonald, A.M., 2013. Interannual variation in summer
601 N₂O concentration in the hypoxic region of the northern Gulf of Mexico, 1985–2007.
602 *Biogeosciences* 10, 6783-6792. doi:10.5194/bg-10-6783-2013

603 Kim, T.-W., Lee, K., Najjar, R.G., Jeong, H.-D., Jeong, H.-J., 2011. Increasing N abundance
604 in the northwestern Pacific Ocean due to atmospheric nitrogen deposition. *Science* 334,
605 505-509. doi:10.1126/science.1206583

606 Kuypers, M.M.M., Marchant, H.K., Kartal, B., 2018. The microbial nitrogen-cycling
607 network. *Nat. Rev. Microbiol.* 16, 263-276. doi:10.1038/nrmicro.2018.9

608 Landolfi, A., Somes, C.J., Koeve, W., Zamora, L.M., Oschlies, A., 2017. Oceanic nitrogen
609 cycling and N₂O flux perturbations in the Anthropocene. *Global Biogeochem. Cycles* 31,
610 1236-1255. doi:10.1002/2017GB005633

611 Lee, K., Chris, S., Tanhua, T., Kim, T.-W., Feely, R., Kim, H.-C., 2011. Roles of marginal
612 seas in absorbing and storing fossil fuel CO₂. *Energ. Environ. Sci.* 4, 1133-1146.
613 doi:10.1039/C0EE00663G

614 Lin, H., Dai, M., Kao, S.J., Wang, L., Roberts, E., Yang, J.Y.T., et al., 2016. Spatiotemporal
615 variability of nitrous oxide in a large eutrophic estuarine system: The Pearl River Estuary,
616 China. *Mar. Chem.* 182, 14-24. doi:10.1016/j.marchem.2016.03.005

617 Lin, X., Liu, M., Hou, L., Gao, D., Li, X., Lu, K., et al., 2017. Nitrogen losses in sediments of
618 the East China Sea: Spatiotemporal variations, controlling factors, and environmental
619 implications. *J. Geophys. Res. - Biogeosciences* 122, 2699-2715.
620 doi:10.1002/2017JG004036

621 Liu, C., Hou, L., Liu, M., Zheng, Y., Yin, G., Han, P., et al., 2019. Coupling of denitrification
622 and anaerobic ammonium oxidation with nitrification in sediments of the Yangtze Estuary:
623 Importance and controlling factors. *Estuar. Coastal Shelf Sci.* 220, 64-72.
624 doi:10.1016/j.ecss.2019.02.043

625 Löscher, C.R., Kock, A., Könneke, M., LaRoche, J., Bange, H.W., Schmitz, R.A., 2012.
626 Production of oceanic nitrous oxide by ammonia-oxidizing archaea. *Biogeosciences* 9,
627 2419-2429. doi:10.5194/bg-9-2419-2012

628 Meyer, R.L., Allen, D.E., Schmidt, S., 2008. Nitrification and denitrification as sources of
629 sediment nitrous oxide production: A microsensors approach. *Mar. Chem.* 110, 68-76.
630 doi:10.1016/j.marchem.2008.02.004

631 Ministry of Water Resources of the People's Republic of China (MWR of China), 2012.
632 River sediment bulletin of China (2011), China Water & Power Press, Beijing.

633 Murray, R.H., Erler, D.V., Eyre, B.D., 2015. Nitrous oxide fluxes in estuarine environments:
634 response to global change. *Global Change Biol.* 21, 3219-3245. doi:10.1111/gcb.12923

635 Naqvi, S.W.A., Jayakumar, D.A., Narvekar, P.V., Naik, H., Sarma, V.V.S.S., D'Souza, W., et
636 al., 2000. Increased marine production of N₂O due to intensifying anoxia on the Indian
637 continental shelf. *Nature* 408, 346-349. doi:10.1038/35042551

638 Prosser, J.I., Hink, L., Gubry-Rangin, C., Nicol, G.W., 2019. Nitrous oxide production by
639 ammonia oxidizers: Physiological diversity, niche differentiation and potential mitigation
640 strategies. *Global Change Biol.* 26, 103-118. doi:10.1111/gcb.14877

641 Punshon, S., Moore, R., 2004. Nitrous oxide production and consumption in a eutrophic
642 coastal embayment. *Mar. Chem.* 91, 37-51. doi: 10.1016/j.marchem.2004.04.003

643 Rabouille, C., Conley, D.J., Dai, M.H., Cai, W.J., Chen, C.T.A., Lansard, B., et al., 2008.
644 Comparison of hypoxia among four river-dominated ocean margins: The Changjiang
645 (Yangtze), Mississippi, Pearl, and Rhône rivers. *Continental Shelf Res.* 28, 1527-1537.
646 doi:10.1016/j.csr.2008.01.020

647 Rao, G.D., Rao, V.D., Sarma, V.V.S.S., 2013. Distribution and air-sea exchange of nitrous
648 oxide in the coastal Bay of Bengal during peak discharge period (southwest monsoon).
649 *Mar. Chem.* 155, 1-9. doi:10.1016/j.marchem.2013.04.014

650 Ravishankara, A.R., Daniel, J.S., Portmann, R.W., 2009. Nitrous Oxide (N₂O): The dominant
651 ozone-depleting substance emitted in the 21st century. *Science* 326, 123-125.
652 doi:10.1126/science.1176985

653 Rysgaard, S., Glud, R.N., Risgaard-Petersen, N., Dalsgaard, T., 2004. Denitrification and
654 anammox activity in Arctic marine sediments. *Limnol. Oceanogr.* 49, 1493-1502.
655 doi:10.4319/lo.2004.49.5.1493

656 Seitzinger, S.P., Kroeze, C., 1998. Global distribution of nitrous oxide production and N
657 inputs in freshwater and coastal marine ecosystems. *Global Biogeochem. Cycles* 12, 93-
658 113. doi:10.1029/97GB03657

659 Seitzinger, S.P., Kroeze, C., Styles, R.V., 2000. Global distribution of N₂O emissions from
660 aquatic systems: natural emissions and anthropogenic effects. *Chemosphere - Global
661 Change Sci.* 2, 267-279. doi:10.1016/S1465-9972(00)00015-5

662 Sun, Z., Wang, L., Mou, X., Jiang, H., Sun, W., 2014. Spatial and temporal variations of
663 nitrous oxide flux between coastal marsh and the atmosphere in the Yellow River estuary
664 of China. *Environmen. Sci. Pollut. Res.* 21, 419-433. doi:10.1007/s11356-013-1885-5

665 Tan, E., Zou, W., Jiang, X., Wan, X., Hsu, T.-C., Zheng, Z., et al., 2019. Organic matter
666 decomposition sustains sedimentary nitrogen loss in the Pearl River Estuary, China. *Sci.
667 Total Environ.* 648, 508-517. doi:10.1016/j.scitotenv.2018.08.109

668 Tan, E., Zou, W., Zheng, Z., Yan, X., Du, M., Hsu, T.-C., et al., 2020. Warming stimulates
669 sediment denitrification at the expense of anaerobic ammonium oxidation. *Nat. Clim.
670 Change* 10, 349–355. doi:10.1038/s41558-020-0723-2

671 Trimmer, M., Risgaard-Petersen, N., Nicholls, J.C., Engström, P., 2006. Direct measurement
672 of anaerobic ammonium oxidation (anammox) and denitrification in intact sediment cores.
673 *Mar. Ecol. Progr. Ser.* 326, 37-47. doi:10.3354/meps326037

674 Wang, D., Chen, Z., Wang, J., Xu, S., Yang, H., Chen, H., et al., 2007. Summer-time
675 denitrification and nitrous oxide exchange in the intertidal zone of the Yangtze Estuary.
676 *Estuar. Coast. Shelf Sci.* 73, 43-53. doi:10.1016/j.ecss.2006.11.002

677 Wang, H., Dai, M., Liu, J., Kao, S.-J., Zhang, C., Cai, W.-J., et al., 2016a. Eutrophication-
678 driven hypoxia in the East China Sea off the Changjiang Estuary. *Environmen. Sci.*
679 *Technol.* 50, 2255-2263. doi:10.1021/acs.est.5b06211

680 Wang, L., Zhang, G., Zhu, Z., Li, J., Liu, S., Ye, W., et al., 2016b. Distribution and sea-to-air
681 flux of nitrous oxide in the East China Sea during the summer of 2013. *Continent. Shelf*
682 *Res.* 123, 99-110. doi:10.1016/j.csr.2016.05.001

683 Wankel, S.D., Ziebis, W., Buchwald, C., Charoenpong, C., de Beer, D., Dentinger, J., et al.,
684 2017. Evidence for fungal and chemodenitrification based N₂O flux from nitrogen
685 impacted coastal sediments. *Nat. Commun.* 8, 15595. doi:10.1038/ncomms15595

686 Wanninkhof, R., 1992. Relationship between wind speed and gas exchange over the ocean. *J.*
687 *Geophys. Res. - Oceans* 97, 7373-7382. doi:10.1029/92JC00188

688 Wei, L., Cai, P., Shi, X., Cai, W.-J., Liu, W., Hong, Q., et al., 2022. Winter mixing
689 accelerates decomposition of sedimentary organic carbon in seasonally hypoxic coastal
690 seas. *Geochim. Cosmochim. Acta* 317, 457-471. doi:10.1016/j.gca.2021.11.003

691 Weiss, R.F., Price, B.A., 1980. Nitrous oxide solubility in water and seawater. *Mar. Chem.* 8,
692 347-359. doi:10.1016/0304-4203(80)90024-9

693 Yan, X., Xu, M.N., Wan, X.S., Yang, J.-Y.T., Trull, T.W., Dai, M., et al., 2017. Dual isotope
694 measurements reveal zoning of nitrate processing in the summer Changjiang (Yangtze)
695 River plume. *Geophys. Res. Lett.* 44, 12289-12297. doi:10.1002/2017GL075951

696 Yan, W., Zhang, S. Sun, P., Seitzinger, S.P., 2003. How do nitrogen inputs to the Changjiang
697 basin impact the Changjiang River nitrate: A temporal analysis for 1968–1997. *Global*
698 *Biogeochem. Cycles* 17, 1091. doi:10.1029.2002GB002029

699 Zhang, G.L., Zhang, J., Liu, S.M., Ren, J.L., Zhao, Y.C., 2010. Nitrous oxide in the
700 Changjiang (Yangtze River) Estuary and its adjacent marine area: Riverine input,

701 sediment release and atmospheric fluxes. Biogeosciences 7, 3505-3516. doi:10.5194/bg-7-
702 3505-2010

703 Zhang, J.-Z., Wanninkhof, R., Lee, K., 2001. Enhanced new production observed from the
704 diurnal cycle of nitrate in an oligotrophic anticyclonic eddy. Geophys. Res. Lett. 28, 1579-
705 1582. doi:10.1029/2000GL012065

706 Zhang, Y., Xie, X., Jiao, N., Hsiao, S.S.Y., Kao, S.J., 2014. Diversity and distribution of
707 *amoA*-type nitrifying and *nirS*-type denitrifying microbial communities in the Yangtze
708 River estuary. Biogeosciences 11, 2131-2145. doi:10.5194/bg-11-2131-2014

709 Zheng, Y., Hou, L., Newell, S., Liu, M., Zhou, J., Zhao, H., et al., 2014. Community
710 dynamics and activity of ammonia-oxidizing prokaryotes in intertidal sediments of the
711 Yangtze Estuary. Appl. Environmen Microbiol. 80, 408-419. doi:10.1128/AEM.03035-13

712 Zhu, Z.Y., Zhang, J., Wu, Y., Zhang, Y.Y., Lin, J., Liu, S.M., 2011. Hypoxia off the
713 Changjiang (Yangtze River) Estuary: Oxygen depletion and organic matter
714 decomposition. Mar. Chem. 125, 108-116. doi:10.1016/j.marchem.2011.03.005

715 **Table 1.** Sampling information of sediment in the hypoxic zone off the Changjiang River estuary. Salinity, concentrations of dissolved oxygen
 716 (DO) and nitrate (NO_3^-) in the bottom waters at each site are also shown. SOC% and TOU indicate sedimentary organic carbon content and total
 717 oxygen utilization, respectively.

718

| Station | Sampling date | Bottom Depth (m) | Salinity | Bottom DO ($\mu\text{mol L}^{-1}$) | Bottom NO_3^- ($\mu\text{mol L}^{-1}$) | SOC% | TOU ($\text{mmol m}^{-2} \text{d}^{-1}$) |
|---------|---------------|------------------|----------|--------------------------------------|---|------|--|
| N0 | August 18 | 26 | 26.8 | 164 | 40.4 | 0.50 | 11.0 |
| Y1 | August 17 | 12 | 23.1 | 183 | 52.4 | 0.41 | 11.4 |
| Y2 | August 17 | 20 | 29.3 | 143 | 28.7 | 0.52 | 15.1 |
| Y3 | August 19 | 23 | 32.2 | 133 | 24.5 | 0.64 | 26.3 |
| Y4 | August 20 | 56 | 34.4 | 108 | 12.8 | 0.19 | 25.9 |
| Y8 | August 21 | 63 | 33.8 | 81 | 16.3 | 0.25 | 53.6 |
| Y10 | August 22 | 48 | 32.8 | 81 | 15.1 | 0.50 | 16.3 |
| Y17 | August 16 | 23 | 30.5 | 167 | 11.9 | 0.57 | 19.4 |
| Y18 | August 22 | 42 | 28.7 | 88 | 19.6 | 0.56 | 31.8 |

719 **Table 2.** Concentrations (C_{N_2O} , nmol L^{-1}) and saturations of N_2O (R_{N_2O} , %) in the surface and bottom waters, the sea-to-air N_2O flux (F_{N_2O} ,
720 $\mu\text{mol m}^{-2} \text{d}^{-1}$) and the water-column inventory of excess N_2O ($I_{\Delta N_2O}$, $\mu\text{mol m}^{-2}$) in the hypoxic zone off the Changjiang River estuary. We
721 compare the N_2O production rates ($\mu\text{mol m}^{-2} \text{d}^{-1}$) via the water-column nitrification (p N_2O -WCN), denitrification (p N_2O -WCD) and
722 sedimentary processes (p N_2O -SED) at sampling sites. The slope of the ^{15}N proportion in the total produced N_2O pool vs. that in total produced
723 N_2 pool (S_{N_2O/N_2}) and the relative contribution of N_2O production via coupled nitrification-denitrification in sediment to the p N_2O -SED (R_{cnd} , %)
724 are also shown.

| Station | Surface waters | | Bottom waters | | F_{N_2O} | $I_{\Delta N_2O}$ | p N_2O -WCD* | p N_2O -WCN | p N_2O -SED* | S_{N_2O/N_2} * | R_{cnd} * |
|---------|----------------|------------|---------------|------------|------------|-------------------|----------------|---------------|----------------|------------------|--------------------|
| | C_{N_2O} | R_{N_2O} | C_{N_2O} | R_{N_2O} | | | | | | | |
| N0 | 9.0 | 135 | 7.4 | 111 | 7.1 | 29.0 | 0.02 | 0.60 | 10.01 | 0.89 | 19.9 |
| Y1 | 10.6 | 156 | 10.5 | 157 | 11.5 | 45.6 | 0.01 | 0.24 | 4.38 | 0.52 | 64.5 |
| Y2 | 8.0 | 118 | 8.3 | 121 | 3.6 | 25.0 | 0.09 | 0.11 | 9.27 | 0.54 | 62.8 |
| Y3 | 10.4 | 156 | 13.7 | 194 | 11.2 | 112.0 | 0.02 | 0.60 | 12.02 | 0.85 | 25.5 |
| Y4 | 7.2 | 111 | 9.6 | 127 | 2.2 | 91.0 | 0.25 | 0.34 | 16.63 | 0.30 | 82.1 |
| Y8 | 12.5 | 175 | 12.2 | 165 | 15.0 | 305.1 | 0.11 | 4.74 | 36.03 | 0.25 | 85.9 |
| Y9 | 10.0 | 153 | 10.7 | 148 | 10.7 | 118.4 | 0.02 | 2.36 | n.s. | n.s. | n.s. |
| Y9a | 10.2 | 146 | 13.2 | 178 | 9.1 | 285.9 | 1.28 | 3.54 | n.s. | n.s. | n.s. |
| Y10 | 12.0 | 173 | 13.7 | 188 | 14.4 | 282.8 | 1.37 | 1.86 | 18.58 | 0.52 | 64.6 |
| Y11 | 7.7 | 118 | 8.3 | 118 | 3.6 | 111.9 | n.d. | 0.23 | n.s. | n.s. | n.s. |
| Y17 | 9.3 | 140 | 9.1 | 131 | 7.9 | 63.4 | 0.01 | 0.65 | 8.55 | 0.64 | 52.8 |
| Y18 | 10.0 | 151 | 11.0 | 150 | 10.1 | 148.0 | n.d. | 1.42 | 19.64 | 0.47 | 69.2 |
| Mean±SD | 9.4±1.6 | 144±21 | 10.6±2.2 | 149±28 | 8.9±4.1 | 134.8±101.5 | 0.26±0.50 | 1.39±1.48 | 15.01±9.36 | 0.55±0.22 | 58.6±22.7 |

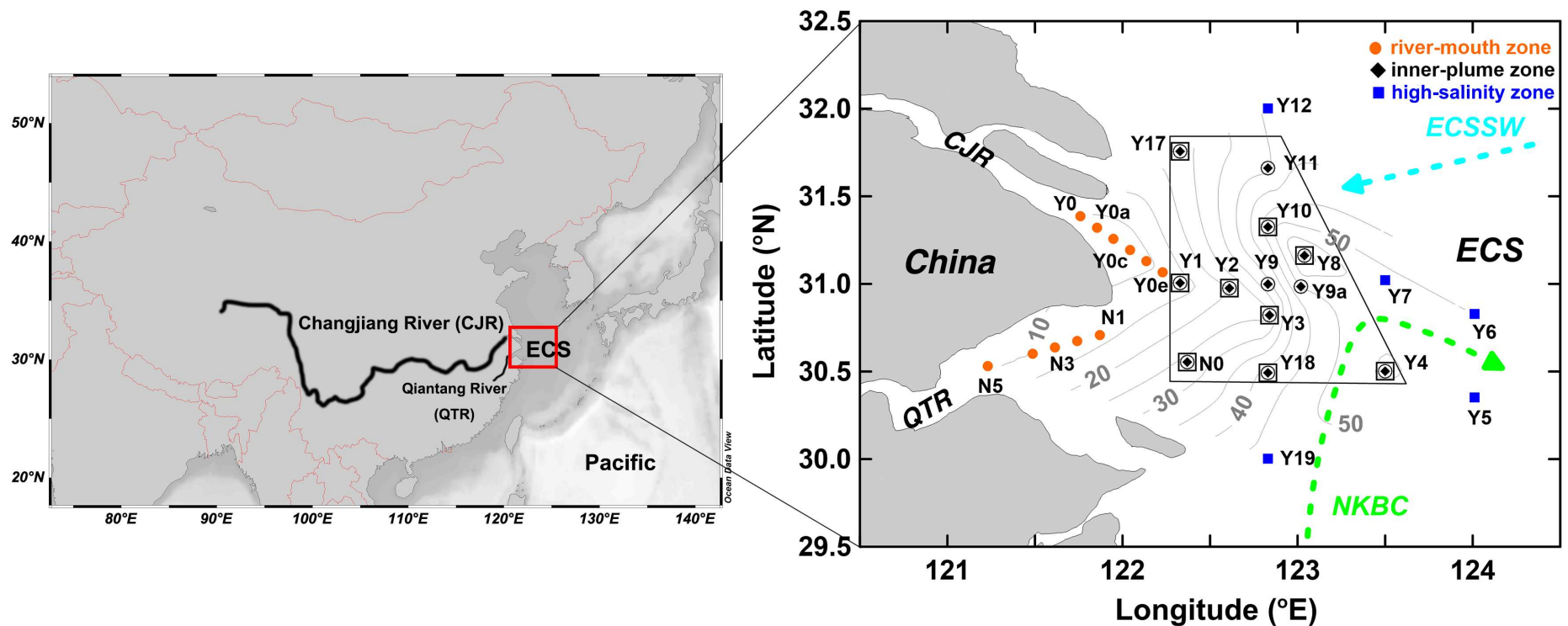
725 * n.d. and n.s. denote not detected and not sampled, respectively.

726 **Table 3.** Correlation matrix of variables in the bottom water and surface sediment. Note that only statistically significant correlations between
727 variables are shown. DO: dissolved oxygen; TOU: total oxygen utilization in surface sediment; SOC%: sedimentary organic carbon content;
728 pN₂O-SED: total sedimentary N₂O production rate; pN₂O-den: N₂O production rate via sedimentary denitrification; pN₂O-cnd: N₂O production
729 rate via coupled nitrification-denitrification in sediment; R_{cnd}: the relative contribution of the pN₂O-cnd to the pN₂O-SED; pN₂-SED: total
730 sedimentary N₂ production rate; N₂O:N₂: sedimentary N₂O:N₂ production ratio.

| Variable | DO | Salinity | NO ₃ ⁻ | TOU | SOC% | pN ₂ O-SED | pN ₂ O-den | pN ₂ O-cnd | R _{cnd} | pN ₂ -SED |
|---------------------------------|---------|----------|------------------------------|--------|--------|-----------------------|-----------------------|-----------------------|------------------|----------------------|
| DO | | | | | | | | | | |
| Salinity | -0.71* | | | | | | | | | |
| NO ₃ ⁻ | 0.67* | -0.88** | | | | | | | | |
| TOU | -0.67* | — | — | | | | | | | |
| SOC% | — | — | — | — | | | | | | |
| pN ₂ O-SED | -0.84** | — | — | 0.91** | — | | | | | |
| pN ₂ O-den | — | — | — | — | — | — | | | | |
| pN ₂ O-cnd | -0.79* | — | — | 0.89** | — | 0.97** | — | | | |
| R _{cnd} | — | — | — | — | -0.69* | — | — | 0.73* | | |
| pN ₂ -SED | — | — | — | — | — | — | — | — | — | |
| N ₂ O:N ₂ | — | — | — | 0.85** | — | 0.87** | — | 0.88** | — | -0.69* |

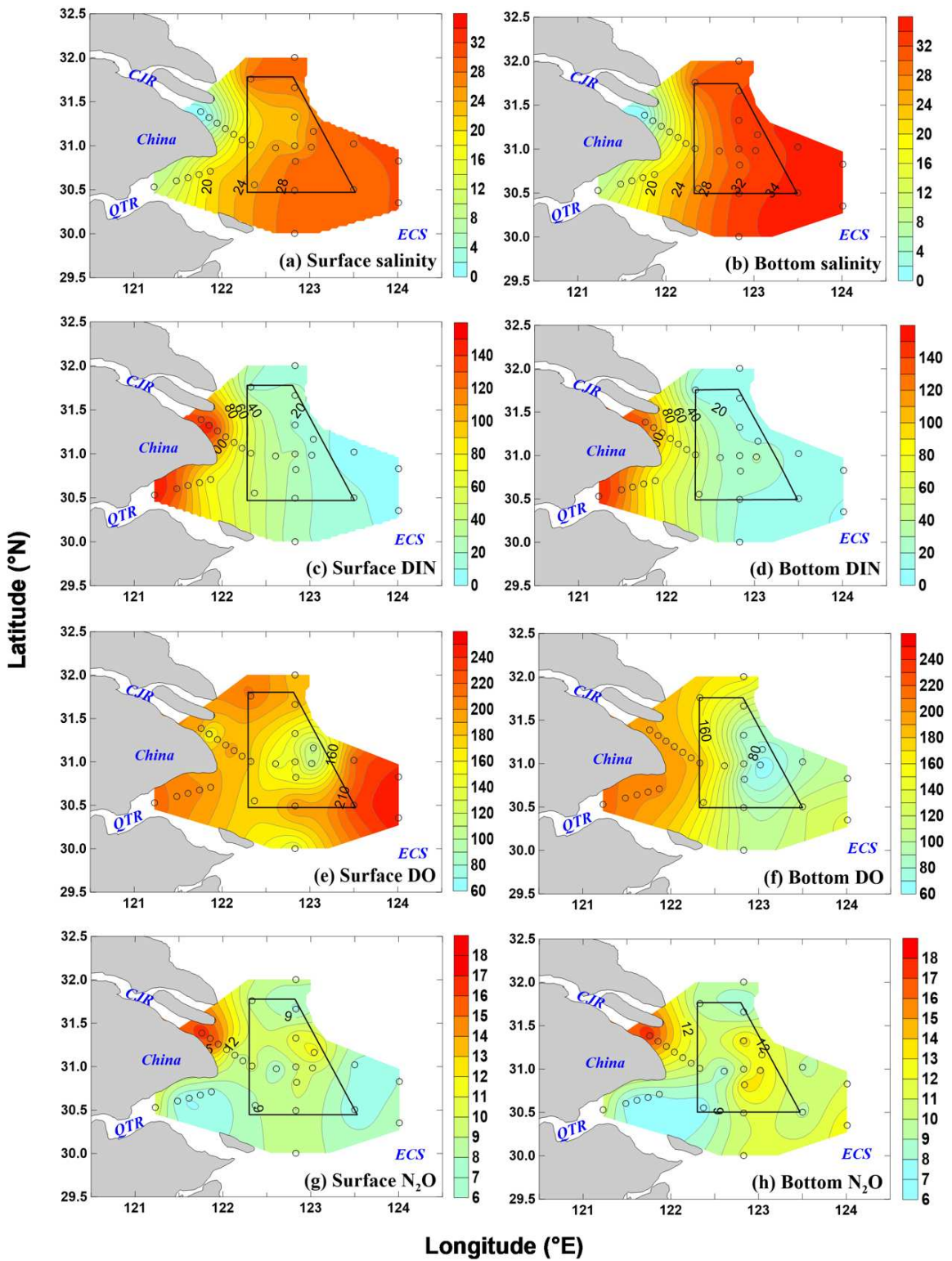
731 * Correlation is significant at the 0.05 level

732 ** Correlation is significant at 0.01 level



733

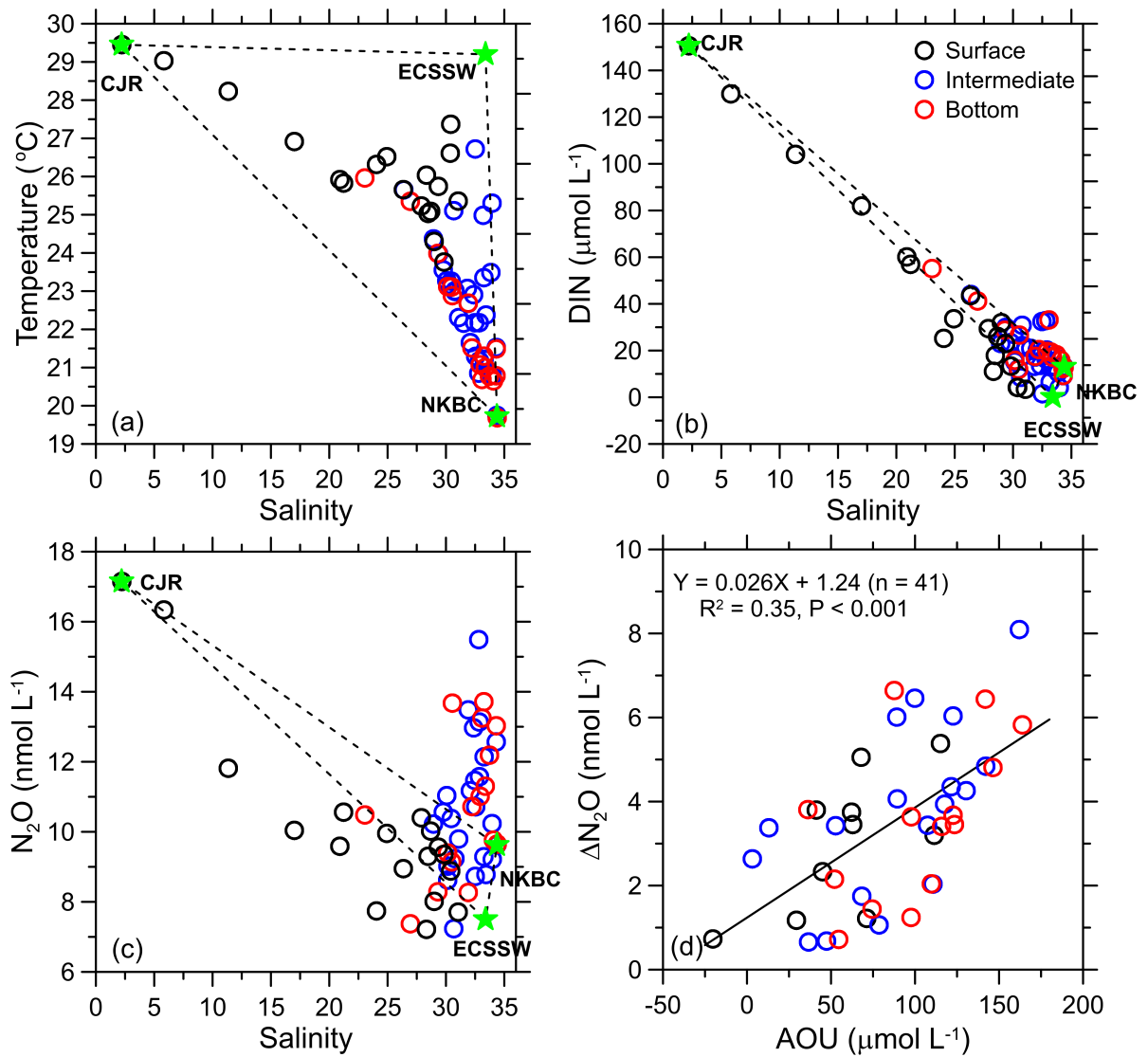
734 **Fig. 1.** Map of sampling sites off the Changjiang River (CJR) and Qiantang River (QTR) estuaries on the East China Sea (ECS) shelf. Water
 735 samples were collected at all sites. The ^{15}N -labeled incubations for N_2O production via the water-column (\odot) and sedimentary (\oplus) processes
 736 were conducted in the inner-plume zone, which was the previously reported hypoxic zone in this region (shown in the trapezoid). The green and
 737 light blue dashed arrows show the pathways of the NKBC and ECSSW from the outer shelf, respectively. The contours indicate the water depth
 738 in the study area.



739

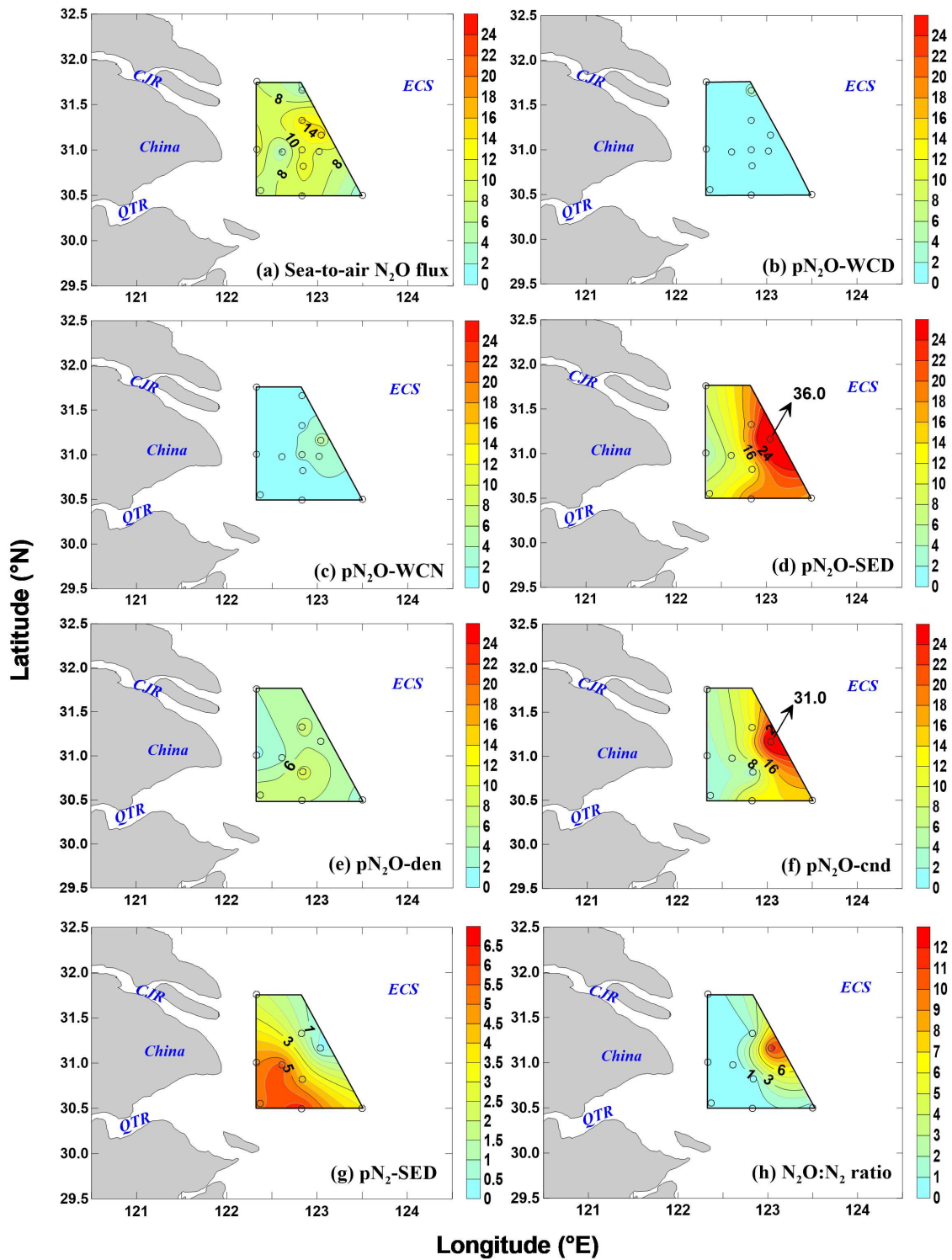
740 **Fig. 2.** Spatial distribution of salinity (a and b), concentrations of DIN (c and d, $\mu\text{mol L}^{-1}$),

741 DO (e and f, $\mu\text{mol L}^{-1}$), and N_2O (g and h, nmol L^{-1}) in the surface and bottom waters.



742

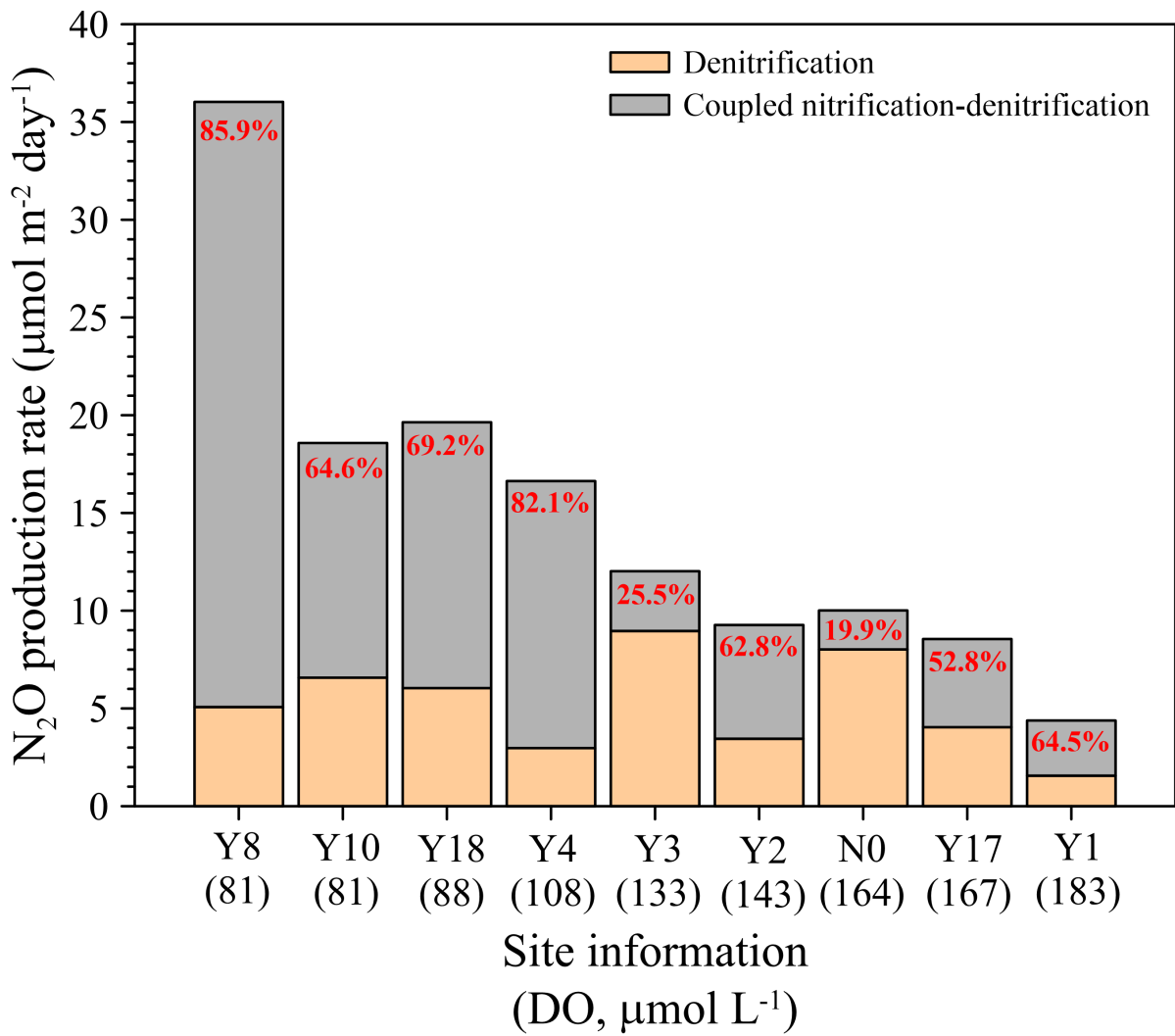
743 **Fig. 3.** Scatter plots of salinity vs. temperature (a), DIN (b), N₂O (c), and AOU vs. ΔN₂O (d)
 744 in the study area. Relevant endmembers of three water masses are shown as green stars in a–c
 745 (see text). The dotted lines in (a–c) represent conservative mixing curves of different water
 746 masses. Note that only data from the hypoxic zone are shown in (d).



747

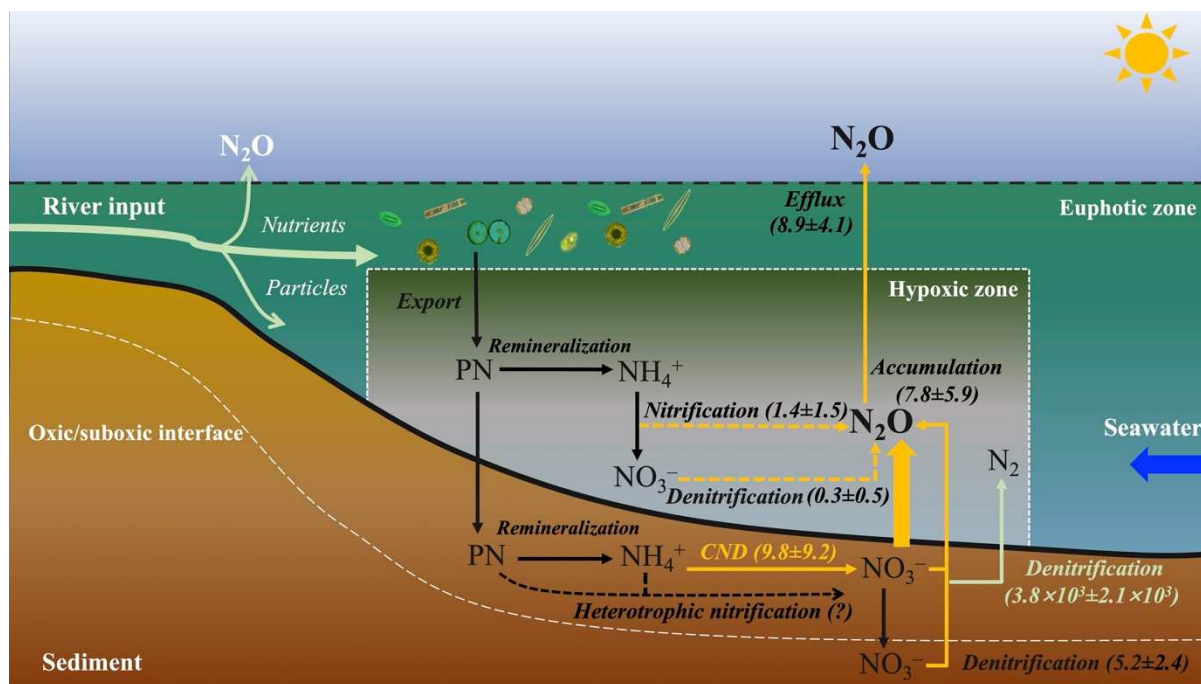
748 **Fig. 4.** Spatial distribution of (a) the sea-to-air N_2O flux ($\mu\text{mol m}^{-2} \text{day}^{-1}$), (b) N_2O
 749 production rate via water-column denitrification ($\text{pN}_2\text{O-WCD}$; $\mu\text{mol m}^{-2} \text{day}^{-1}$), (c) N_2O
 750 production rate via water-column nitrification ($\text{pN}_2\text{O-WCN}$; $\mu\text{mol m}^{-2} \text{day}^{-1}$), (d) total N_2O
 751 production rate via sedimentary processes ($\text{pN}_2\text{O-SED}$; $\mu\text{mol m}^{-2} \text{day}^{-1}$), (e) N_2O production

752 rate via sedimentary denitrification (pN_2O -den; $\mu\text{mol m}^{-2} \text{day}^{-1}$), (f) N_2O production rate via
753 sedimentary coupled nitrification-denitrification (pN_2O -cnd; $\mu\text{mol m}^{-2} \text{day}^{-1}$), (g)
754 sedimentary N_2 production rate (pN_2 -SED; $\text{mmol m}^{-2} \text{day}^{-1}$) and (h) sedimentary $N_2O:N_2$
755 production ratio ($N_2O:N_2$ ratio; %) in the hypoxic zone off the CJR estuary.



756

757 **Fig. 5.** N₂O production rates via denitrification and coupled nitrification-denitrification (cnd)
 758 in the surface sediments in the hypoxic zone off the CJR estuary. Numbers inside the cnd
 759 bars denote the relative contribution of cnd to the total sedimentary N₂O production (R_{cnd} in
 760 Table 2). Note that we present these datasets along the bottom-water DO gradient (values in
 761 units of µmol L⁻¹ in parentheses).



762

763 **Fig. 6.** Conceptual diagram of N₂O production and emission via different pathways in the
 764 hypoxic zone off the human-perturbed estuary. CND denotes coupled nitrification-
 765 denitrification in sediments. Note that values in units of $\mu\text{mol m}^{-2} \text{day}^{-1}$ in parentheses are
 766 shown as the N₂O production rates via the water-column and sedimentary processes, the N₂O
 767 accumulation rate in the water column, the sea-to-air N₂O flux and sedimentary N₂
 768 production rate in the hypoxic zone.