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Thiolene and polycaprolactone methacrylate-based polymerised high internal phase emulsion (PolyHIPE) scaffolds for tissue engineering

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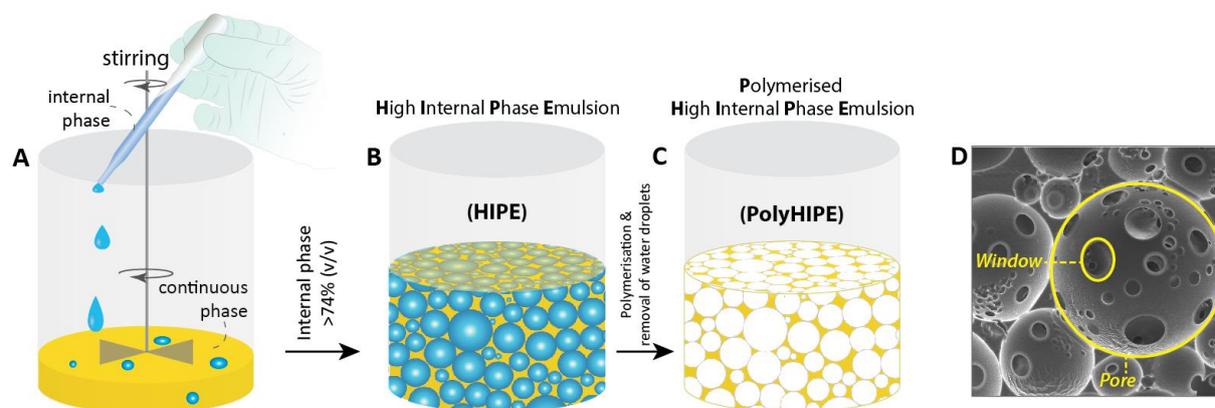
Abstract:

Highly porous emulsion templated polymers (PolyHIPEs) provide a number of potential advantages in the fabrication of scaffolds for tissue engineering and regenerative medicine. Porosity enables cell ingrowth and nutrient diffusion within - as well as waste removal from - the scaffold. The properties offered by emulsion templating alone include the provision of high interconnected porosity, and - in combination with additive manufacturing - the opportunity to introduce controlled multi-scale porosity to complex or custom structures. However, the majority of monomer systems reported for PolyHIPE preparation are unsuitable for clinical applications as they are non-degradable. Thiol-ene chemistry is a promising route to produce biodegradable photocurable PolyHIPEs for the fabrication of scaffolds using conventional or additive manufacturing methods, however, relatively little research has been reported on this approach. This study reports the groundwork to fabricate a thiol and polycaprolactone (PCL) based PolyHIPE materials via a photoinitiated thiolene click reaction. Two different formulations, either 3-arm PCL methacrylate (3PCLMA) or 4-arm PCL methacrylate (4PCLMA) moieties were used in the PolyHIPE formulation. Biocompatibility of the PolyHIPEs was investigated using human dermal fibroblasts (HDFs) and human osteosarcoma cell line (MG-63) by DNA quantification assay, and developed PolyHIPEs were shown to be capable of supporting cell attachment and viability.

Keywords: *emulsion templating, PolyHIPE, polycaprolactone, thiol-ene, tissue engineering, biomaterials, photopolymerisation, porosity*

34 1. Introduction

35 A key aspect when producing scaffolds for biomedical applications is the inclusion of interconnected
36 porosity within the construct; this enables cell ingrowth, nutrient diffusion and waste removal from
37 the implanted scaffold. Various methods reported to fabricate porous substrates for biomedical
38 applications include electrospinning¹⁻³, 3D printing⁴, and porogen leaching^{5,6}. Emulsion templating is
39 an alternative fabrication route that has gained attention due to its advantages of being (i) tunable,
40 (ii) providing high porosity (up to 99%) and (iii) interconnectivity (open cellular morphology)⁷. While
41 an interconnected porous scaffold is important for cell infiltration, tunable chemical, mechanical and
42 morphological cues enable precise engineering of the substrates for specific biomedical applications.
43 The principle of emulsion templating is based on creating biphasic emulsions and polymerisation of
44 the continuous phase (Figure 1). Solidifying of a monomeric continuous phase and subsequent
45 removal of the internal droplet phase leaves behind a porous structure that is a 3D replica of the initial
46 emulsion. Emulsions with an internal volume higher than 74% are defined as **High Internal Phase**
47 **Emulsions (HIPEs)**, and substrates obtained by their polymerisation are called **Polymerised HIPEs**
48 **(PolyHIPEs)**. To date, there has been a number of significant reviews from the pioneers of the field of
49 emulsion templating in the literature⁸⁻¹³. Also, recently, comprehensive reviews on the development
50 and use of PolyHIPEs specifically in biomedical applications has been reported^{7,14}.



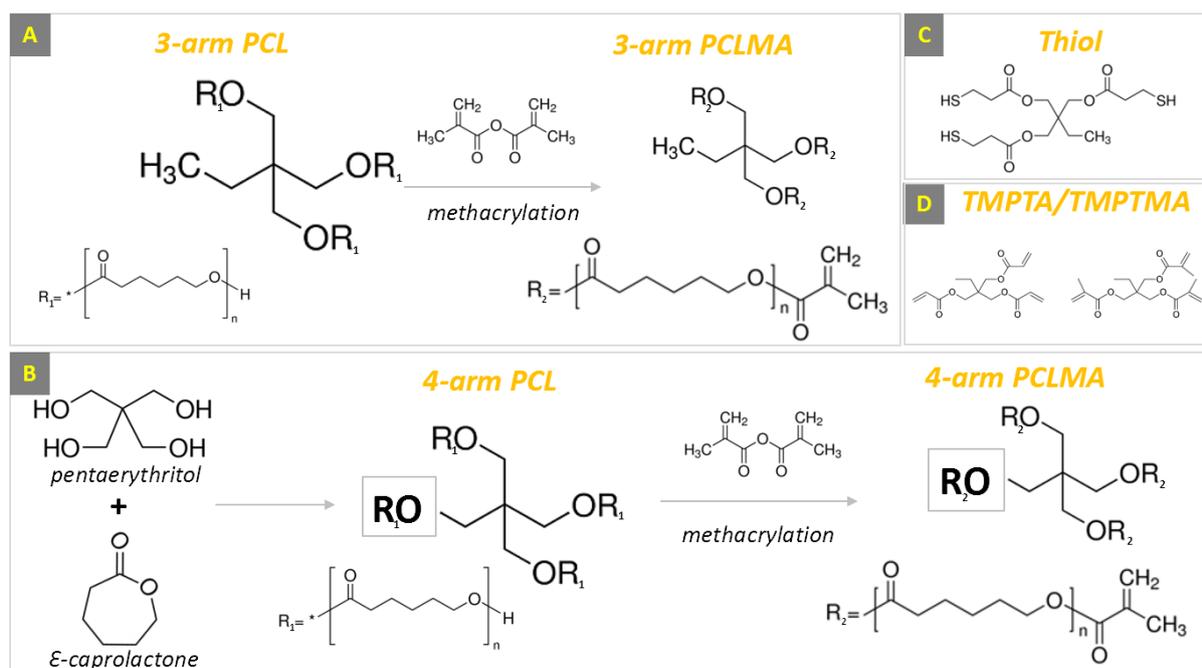
52 **Figure 1:** Fabrication steps of the Polymerised High Internal Phase Emulsion (PolyHIPE). (A, B) The gradual
53 addition of the internal phase into the continuous phase while the system is mixed, (C) polymerisation of the high
54 internal phase emulsion (HIPE), (D) scanning electron microscope image of the PolyHIPE (Adapted from⁷).

55 Common monomers reported in PolyHIPE preparations are styrene, its derivatives such as divinyl
56 benzene (DVB) and acrylate-based monomers, including 2-ethyl-hexyl acrylate (EHA) and isobornyl
57 acrylate (IBOA)¹⁵⁻²⁰. The aforementioned conventional monomers have limited applications for tissue
58 engineering as they do not degrade within the body.

59 To create a PolyHIPE scaffold often free radical initiated thermal polymerisation is used to set the
60 monomeric continuous phase into a solid. However, the number of researchers using
61 photopolymerisation to fabricate PolyHIPE scaffolds has increased ^{4,15-17,21-28}. Photocuring to produce
62 PolyHIPEs is a fast and versatile approach; it allows a wide choice of monomers and permits the curing
63 of less stable emulsions which would otherwise destabilise over the long thermal cure process ²².

64 There has been an emerging research effort exploring thiol-ene chemistry to produce biodegradable
65 photocurable PolyHIPEs for tissue engineering applications ^{26,29,30}. The resulting PolyHIPE material
66 formed via this chemistry is intended to be fully biodegradable as they are aliphatic polyesters.
67 Caldwell et al. used a 1:1 trimethylolpropane tris (3-mercaptopropionate) (tri-thiol) and
68 trimethylolpropane triacrylate (TMPTA) mixture to produce the degradable material ²⁶. Johnson et al.
69 showed successful preparation of thiol and triacrylate polycaprolactone (PCL) based PolyHIPE by thiol-
70 ene click chemistry, which resulted in a fully degradable PolyHIPE material (90% and 95% nominal
71 porosity) with 60 μm void diameter ³⁰. Additionally, Susec et al. reported a biocompatible divinyl
72 adipate and pentaerythritol tetrakis (3-mercaptopropionate) based PolyHIPEs, which they used in a
73 follow-up study as scaffolds for cartilage repair ^{21,31}. Recently, Whitely et al. reported on the use of
74 thiol additives (5-10%) to reduce the oxygen quenching of the radical initiated polymerisation with
75 negligible change in the properties of the material and biocompatibility of their PolyHIPEs ³².
76 Additionally, Langford et al. have shown that thiol-ene based polyHIPEs can be consecutively
77 functionalised via using the pendant sulphur groups as binding sites via click-chemistry ³³.

78 To summarise, PolyHIPEs have tremendous potential as a scaffold fabrication approach to produce
79 highly porous biomaterials, where controlled porosity may influence both mass transfer and cell
80 attachment/migration. However, relatively little progress has been reported in the development of
81 methods that are ideally suited to the preparation of biocompatible, biodegradable PolyHIPEs with
82 properties tailored to tissue engineering. PCL has been widely reported as ideally suited for these
83 applications on account of its excellent biocompatibility and favourable cell response ^{24,34,35}, and the
84 aim of this research was, therefore, to investigate a method to prepare thiolene and PCL methacrylate-
85 based PolyHIPE scaffolds, and evaluate their *in vitro* biocompatibility.



86

87 **Figure 2:** Monomers used in the composition of Thiol-PCL PolyHIPEs. (A) Methacrylate functionalisation of PCL
88 triol to obtain 3PCLMA, (B) Synthesis of 4PCL from ring-opening polymerisation of pentaerythritol and
89 caprolactone and methacrylate functionalisation. Chemical structures of (C) tri-trithiol and (D) crosslinkers;
90 TMPTA/TMPTMA.

91 2. Materials and Methods

92 2.1. Materials

93 ϵ -Caprolactone, pentaerythritol, methacrylic anhydride (MAAn), chloroform, toluene, tin 2-
94 ethylhexanoate (SnOct₂), dichloromethane (DCM), triethylamine (TEA) isopropanol,
95 trimethylolpropane tris (3-mercaptopropionate) (thiol), trimethylolpropane triacrylate (TMPTA),
96 trimethylolpropane trimethacrylate (TMPTMA), diphenyl (2,4,6-trimethyl benzoyl) phosphine
97 oxide/2-hydroxy-2-methylpropiophenone blend (photoinitiator), 1,2-Dichloroethane (DCE),
98 polycaprolactone triol (M_n 900 g/mol), Dulbecco's Modified Eagle Media (DMEM), amphotericin B,
99 fetal calf serum (FCS), penicillin/streptomycin (PS), L-glutamine, trypsin, paraformaldehyde, resazurin
100 sodium salt, MTT, ethanol, DAPI and Phalloidin were all purchased from Sigma Aldrich. Hypermer B246
101 (surfactant) was donated by Croda Ltd.

102 2.2. Synthesis of 3PCLMA

103 The PCL triol was methacrylate functionalised to obtain 3PCLMA (Figure 2A). First, PCL triol
104 (M_n 900 g/mol, 1 molar equivalent) was dissolved in DCM. TEA (6 molar equivalent) was dissolved in
105 50 mL DCM was added to the solution. The mixture was cooled by submerging in a salted ice bath for

106 30 minutes. MAAn (6 molar equivalent) was dissolved in 50 mL DCM and was added dropwise using a
107 dropping funnel. When MAAn was completely dispensed, the solution was allowed to warm up slowly
108 to room temperature (RT) and was left to react for 24 hours while covered in foil. Almost all solvent
109 was removed using a rotary evaporator, and the polymer was purified three times by precipitation
110 from methanol at -80°C.

111 2.3. Synthesis of 4PCLMA

112 The synthesis of 4PCLMA was described in detail in other studies ^{2,23,24,36,37}. Briefly, 4PCL was
113 synthesised via ring-opening polymerisation of ϵ -caprolactone and multifunctional alcohol initiator
114 pentaerythritol and then methacrylate functionalised. 4PCLMA and pentaerythritol were then added
115 to a round bottom flask in the presence of toluene stirred using a magnetic stirrer under a nitrogen
116 atmosphere. The resulting reaction mixture was heated to 130°C, and SnOct₂ was added.
117 Furthermore, the reaction was stirred continuously for 6 hours at RT prior to solvent removal via rotary
118 evaporation. For the methacrylation, hydroxyl-terminated PCL was dissolved in DCM, then the
119 solution was cooled in an ice bath before the addition of TEA (2 molar equivalents). MAAn (2 molar
120 equivalents) was added dropwise whilst maintaining a low temperature. The reaction was allowed to
121 stir for 24 hours at RT in the absence of light under nitrogen gas. Finally, the solvent was removed by
122 rotary evaporation, and the polymer was purified three times by precipitation from methanol at -80°C.

123 2.3. Characterisation of 3PCLMA, 4PCL and 4PCLMA

124 Proton (¹H) nuclear magnetic resonance NMR spectroscopy analysis was performed on an AVANCE III
125 spectrometer at 400 MHz to confirm the structure of 3PCLMA and 4PCLMA. The spectra were
126 recorded using an 8.2 kHz acquisition window, with 64 k data points in 16 transients with a 60 s recycle
127 delay (to ensure full relaxation). Deuterated chloroform was used as a diluent (CDCl₃). Spectra were
128 analysed using MestReNova software. Chemical shifts were referenced relative to CDCl₃ at 7.27 ppm.
129 The degree of methacrylation (DM) of 3PCLMA and 4PCLMA was calculated by comparing the signal
130 intensity of the methylene groups and the signal intensities of the methacrylate groups from the NMR
131 data (Equation 1, 2).

$$DM = ((\int \text{methacrylated ends}) / (\int \text{non - methacrylated} + \text{methacrylated ends})) * 100 \quad (1)$$

$$DM = \frac{\left((\int I_{5.5} + \int I_{6.1}) / 2 \right)}{\left((\int I_{5.5} + \int I_{6.1}) / 2 \right) + \left((\int I_{3.6}) / 2 \right)} * 100 \quad (2)$$

132 Molecular weight and molecular weight distributions of 4PCLMA were determined using a Viscotek
133 GPCmax VE200 gel permeation chromatography (GPC) system with a differential refractive index
134 detector (Waters 410). Tetrahydrofuran was used as the eluting solvent at a flow rate of 1 mL/minute
135 at 40 °C, and polystyrene standards were used as the calibration sample.

136 **2.6. Mechanical Characterisation 3PCLMA and 4PCLMA**

137 Dog-bone shaped tensile test samples were fabricated as previously described²³ and mechanical test
138 was applied as described previously³⁸. Briefly, 0.5 mL 4PCLMA and 3PCLMA were pipetted into the
139 moulds and cured for 3 min on each side. Samples were tested using a uniaxial mechanical testing
140 machine (BOSE Electroforce Test Instruments, Minnesota, USA) equipped with a 22 N load cell. Grip
141 distance and extension rate were set to 10 mm and 0.1 mm/s, respectively, and the force and
142 elongation data were recorded. The stress and strain values were calculated using the cross-sectional
143 area where force was applied. Young's modulus was determined using the linear-elastic region of each
144 sample's stress-strain curves. The ultimate tensile strength (UTS) was calculated as that of the
145 maximum force applied divided by the cross-sectional area of the sample.

146 **2.4. Preparation of PolyHIPE Scaffolds**

147 Thiol-PCL PolyHIPEs were manufactured via photopolymerisation with the formulations presented in
148 Figure 5A. The continuous phase of the emulsion consisted of PCL, thiol, surfactant (Hypermer B246),
149 and crosslinker were dissolved in the solvent and stirred using an overhead stirrer (Pro40, SciQuip).
150 The water was added dropwise, and then the emulsion was left to further mix for a further 5 minutes.
151 Finally, the photoinitiator was added in the absence of light, mixed and transferred to a silicone mould
152 and cured via Light Hammer[®] 6 UV curing system (Fusion UV Systems Inc, USA) assisted with bench-
153 top conveyor (LC6E, Fusion UV Systems Inc, USA) with a power output of 200 W/cm² at 100% intensity.

154 PCL-Thiol PolyHIPE samples were washed in a Soxhlet extractor for 24 hours with ethanol. For cell
155 culture, the samples were transferred to a 70% ethanol and distilled water solution for 30 minutes.
156 The samples were transferred to sterile PBS and washed three times to remove traces of ethanol. To
157 remove all trapped air from the discs, the samples were kept in a sterile PBS container, a lid affixed
158 with a 0.2 µm pore syringe filter was used to seal the container, which was then put into a vacuum
159 oven and cycled from being under vacuum to normal atmospheric pressure to draw the trapped air
160 out of the sample. This was repeated three times until all the samples remained submerged under
161 normal atmospheric pressure.

162 2.5. Chemical and Morphological Characterisation of PolyHIPEs

163 To prepare the 4PCLMA and 3PCLMA based PolyHIPE specimens for SEM they were first freeze-dried
164 and cut to reveal their internal cross-section. These were then attached to carbon fibre pads adhered
165 to aluminium stubs and sputter-coated with gold (Emscope SC500, Philips). The morphologies of the
166 PolyHIPE disks were then imaged using a Philips XL-20 scanning electron microscope operating at 10.0
167 kV. The SEM micrographs were analysed using the software Image J 1.48 to quantify the average void
168 diameter of PolyHIPE disks, the diameters of 90 voids and 20 windows were measured, and a statistical
169 correction factor was applied to the average void diameter³⁹. Pore and window size distribution
170 histograms were created and average pore (D) and window sizes (d) were reported. The degree of
171 interconnectivity was calculated by dividing the average window size by the average pore size (d/D)
172^{23,40}, and the degree of openness was calculated by dividing open surface area by total surface area
173 for randomly selected 10 pores^{7,13,17}. Fourier Transform Infrared spectroscopy was performed on
174 Thiolene 4PCLMA PolyHIPE. Readings were taken between 500 – 4000 cm⁻¹ and resolution of 4 cm⁻¹.

175 2.5. Biological Characterisation

176 2.5.1. Fibroblast Cell Culture

177 Human dermal fibroblast cells were isolated from tissue samples obtained from consenting patients
178 undergoing either elective abdominoplasty or breast reduction surgery as described previously^{3,23,41}.
179 The collected tissue was used under the requirements stipulated by the Research Tissue Bank Licence
180 12179, and fibroblast cells were isolated (Ethical approval for the tissue acquisition was granted by
181 the local ethical approval committee of the NHS Trust, Sheffield, UK, ethics reference: 15/YH/0177).
182 The fibroblasts were cultured and expanded in T75 culture flasks until they were ~80% confluent.
183 Scaffolds were disinfected in ethanol and washed with PBS three times. Dulbecco's modified Eagle's
184 medium supplemented with 10% FCS, 0.01% L-Glutamine, 0.01% penicillin-streptomycin and 0.0025%
185 amphotericin B was used as cell culture media. Cells were trypsinised and were seeded with the
186 concentration of 75000 cells/40 µL media and were left in the incubator at 37°C, 5% CO₂ for 20 minutes
187 for cell attachment before 1 mL of media was added. The cell medium was changed every three days.

188 2.5.2. Bone Cell Culture

189 The manufactured PolyHIPE disks had a diameter of 9 mm, which were washed in acetone and air-
190 dried prior to the sterilisation process. The PolyHIPE disks were sterilised by immersing in 70% ethanol
191 for 50 minutes following series of PBS washes (5 minutes, repeated three times). PolyHIPE disks were
192 then air-dried in a sterilised environment for 48 hours prior to DMEM media soaking for 24 hours.

193 Human osteosarcoma cell line (MG-63) was used to seed disks with the density of 20000 cells (10 μ L)
194 per sample and placed in an incubator (37°C and 5% CO₂) for 30 minutes before addition of 990 μ L of
195 cell culture medium and tissue culture plastic was used as control. The samples were left in the
196 incubator for a period of 3 and 7 days, and the medium was changed every two days.

197 2.5.3. Colourimetric Assays

198 Resazurin reduction (RR) assay was applied to measure the cellular metabolic activity and estimate
199 the cell viability on scaffolds. Resazurin solution (non-fluorescent, blue) is reduced by the cells and
200 forms resorufin (fluorescent, pink), which is detectable by a fluorescence plate reader. 1 mM resazurin
201 stock solution in dH₂O was diluted to 100 μ M in culture media to make the resazurin working solution.
202 1 mL of RR solution was added to each well, and the scaffolds were transferred into a fresh well plate
203 using sterile forceps. The well plates were protected from light and incubated for 4 hours at 37 °C.
204 From each scaffold, triplicate samples of 200 μ L of the reduced solution were added to a 96 well plate.
205 It was measured three times using a spectrofluorometer (FLX800, BIO-TEK Instruments, Inc.) at an
206 excitation wavelength of 540 nm and an emission wavelength of 630 nm. Scaffolds were washed twice
207 with PBS before adding fresh media. RR assay was performed at three time points (day 1, day 4, and
208 day 7) with fresh scaffold/cell constructs for each. The reaction of blank (cell-free) PolyHIPEs with MTT
209 reagent was also investigated. For this, first, 1 mg/mL MTT solution was prepared in PBS and filter
210 sterilised. Scaffolds were placed into a fresh 24-well plate, and 1 mL of MTT reagent was added into
211 each well, covered aluminium foil and incubated for 4 hours and 24 hours at 37°C. Colour change due
212 to formed formazan crystals was imaged.

213 2.5.3. DNA Quantification Assay

214 Quant-iT™ PicoGreen dsDNA Assay kit (Life Technologies, UK) is a DNA quantitation assay used to
215 determine the cell number. Cell culture medium was removed from samples following series of PBS
216 washes prior to treatment with 500 μ L cell assay buffer (1:10 Tris-EDTA (TE) buffer (1.5 M Tris-HCL, 1
217 mM ZnCl₂, 1 mM MgCl₂) in deionised water and 1% Triton-X100) for each time point at room
218 temperature for 30 minutes following overnight freeze at 4°C. The samples were treated through a
219 series of freeze-thaw cycles, freeze (-80°C, 10 minutes) and thaw (37°C, 15 minutes) repeated three
220 times. Following 15 seconds of vortexing and they were centrifuged at 10000 rpm for 5 minutes.
221 180 μ L of supernatant was added in 1:1 ratio to PicoGreen reagent (diluted in 1:20 TE buffer (10 mM
222 Tris-HCL, 1 mM EDTA, pH 7.5), 1:200 PicoGreen in deionised water). The samples were incubated at
223 room temperature for 10 minutes in darkness prior to fluorescence reading at λ_{ex} = 485 nm and λ_{em} =
224 528 nm.

225 2.5.2. Confocal Microscopy

226 4',6-diamidino-2-phenylindole (DAPI) and TRITC-phalloidin stained specimens were prepared by
227 diluting the stains (1:1000 in PBS) and incubated at room temperature for 30 minutes in darkness as
228 described previously^{15,17,42}. Single plane images (1024×1024 pixels) were taken via an upright confocal
229 microscope (Zeiss LSM510-META, UK) assisted with ×10 objective (W-N-Achroplan 10×/N.A. 0.3, Zeiss
230 ltd, UK). DAPI stain was excited using an 800 nm two-photon Ti-Sapphire laser set to 358 nm λ_{ex} , 461
231 nm λ_{em} . TRITC and FITC phalloidin (F-actin) was determined via single-photon laser set to λ_{ex} = 495 nm,
232 λ_{em} = 515 nm (FITC) and λ_{ex} = 545 nm, λ_{em} = 573 nm (TRITC).

233 2.6. Statistical Analysis

234 Statistical analyses were performed by using GraphPad Prism 6 using one-way and two-way analysis
235 of variance (ANOVA) for mechanical testing and cellular metabolic activity assays, respectively, and
236 plotted as mean \pm SD. A difference was deemed statistically significant if the p-value was less than
237 0.05, and the statistical differences are denoted in the figures. The total numbers of replicates (n) are
238 stated in the figure legends.

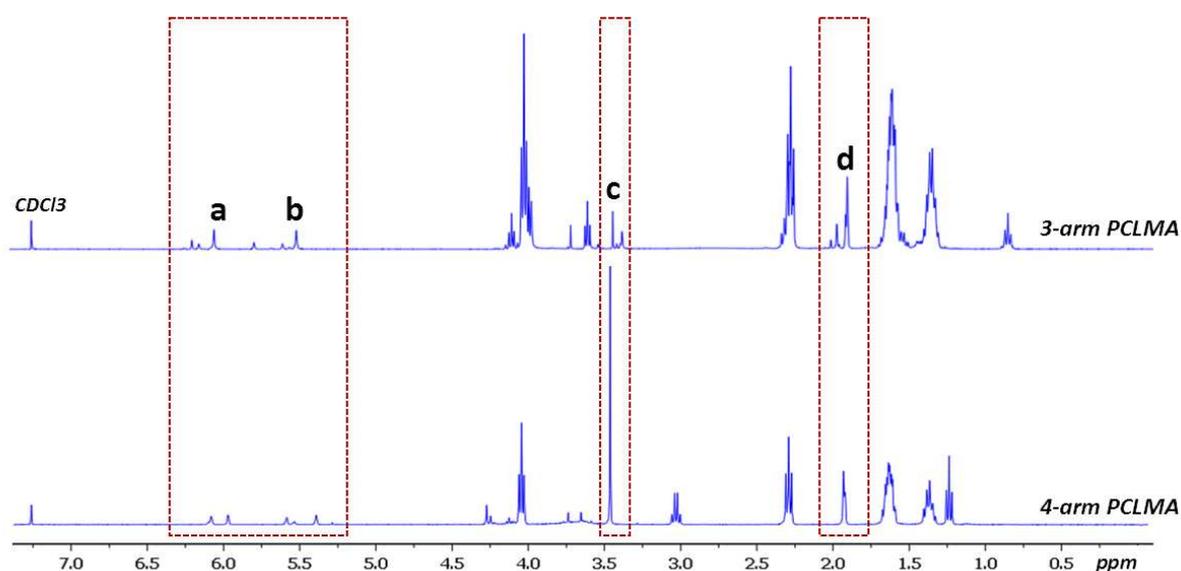
239 3. Results and Discussion

240 3.1. Synthesis and the characterisation of 3PCLMA and 4PCLMA

241 PCL is one of the most widely used synthetic polymers used to fabricate tissue engineering scaffolds.
242 Yet, its potential is limited by its form. Typically, PCL is a thermoplastic that is sold as small solid beads
243 of various high molecular weights (60.000-90.000 g/mol), these can be dissolved in a solvent and
244 electrospun into a fibrous scaffold, cast into a porous material by porogen leaching, or melted and
245 extruded into the desired shape^{35,43-45}. Photocurable monomers have gained increasing attention
246 because of their potential to rapidly polymerise (seconds to minutes depending on the sample size)
247 and their suitability to be used with a variety of fabrication techniques which include 3D printing^{4,46,47},
248 porogen leaching⁴⁸, and emulsion templating^{2,23,24}. There is a limited range of commercially available
249 degradable monomers. So, *in house* synthesis of photocurable monomers are often used.
250 Functionalised 2-arm⁴⁹⁻⁵¹, 3-arm^{30,52} and 4-arm^{2,4,23,24} PCLs are forms of PCL used for tissue
251 engineering applications. The molecular weight of the polymer⁵³ it's degree of functionalization⁴⁶ and
252 the number of monomer arms⁵³ have been shown to influence the mechanical properties of these
253 polymers. This photo-polymerisation of PCL expands the potential of this material to a range of tissue
254 engineering applications. The FDA approval of PCL made medical devices on the market is encouraging
255 for PCL to be used for other biomedical applications^{23,54}. With similar motivation, there has been a

256 number of studies in the literature on the development of PCL based PolyHIPE structures either using
257 radical polymerization or ring-opening polymerization ^{23,30,55-62}.

258 In this study, the synthesis of 4PCL and methacrylate-functionalisation of both synthesised 4PCL and
259 commercially available 3PCL is presented. 3PCL and 4PCL were successfully functionalised via
260 methacrylate functionalisation. According to ¹H NMR analysis (Figure 3), while the non-methacrylated
261 end of 3PCL and 4PCL has methylene groups adjacent to hydroxyl end groups shown with the peak at
262 3.6 ppm. On the methacrylated ends, they are converted into methacrylate groups, which are
263 indicated at peaks 1.9, 5.5 and 6.1 ppm. The degree of methacrylation of 3PCLMA and 4PCLMA was
264 calculated as 44% and 46%, respectively. The degrees of methacrylation of both 3PCLMA and 4PCLMA
265 can be controlled using various parameters such as the ratio of methacrylation agents (methacrylic
266 anhydride and triethylamine) to pre-polymer (PCL) and reaction time of methacrylation. Recently our
267 group has reported the fabrication route of PCLMA with different degrees of methacrylation and the
268 effect of methacrylation degree on mechanical properties of PCLMA ⁶³. In this study, we aimed to keep
269 the degree of methacrylation constant for both types of polymer to compare their properties. But, the
270 degree of methacrylation can be increased up to 100% for specific applications, such as bone tissue
271 engineering scaffolds, as we reported before ⁴. In this study, while PCL triol with a molecular weight
272 of 900 g/mol was used for 3PCLMA, M_w and M_n values of 4PCLMA were determined by GPC analysis
273 as 20900 g/mol and 15000 g/mol, respectively, which gives a dispersity of 1.4.



274
275 **Figure 3:** Proton NMR spectrum of 3PCLMA and 4PCLMA and the relative assignments. c: the peaks of the
276 hydroxyl group; a, b, and d are the peaks of the methacrylate group.

277 Mechanical properties of the microenvironment are reported to have a dominant impact on cell
278 behaviour. It is because cells can sense the stiffness of the material by mechanosensation and directs

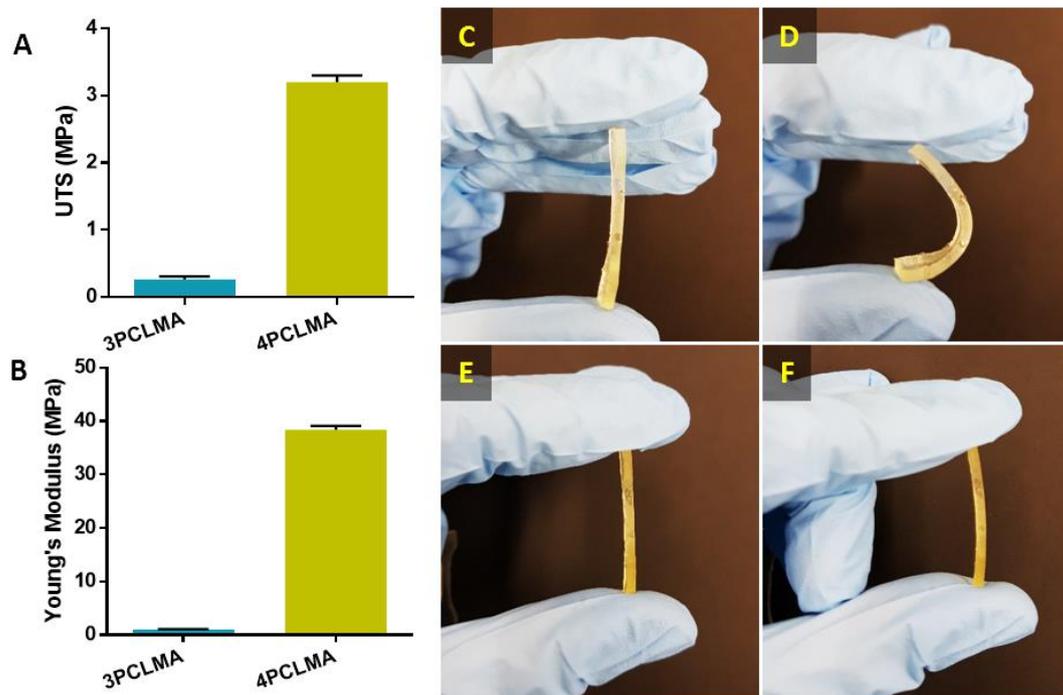
279 its morphology, gene expression, and differentiation, accordingly ⁶⁴⁻⁶⁶. Accordingly, we have
280 investigated the mechanical properties of the polymer synthesized in the scope of this study; 3PCLMA
281 and 4PCLMA. Tensile testing of 3PCLMA and 4PCLMA shows a significant difference between the
282 mechanical properties of the two polymers. The 4PCLMA had the greater Young's modulus of
283 38.37 ± 0.76 MPa and UTS of 3.20 ± 0.10 MPa compared to the lower 3PCLMA values of 1.00 ± 0.07 MPa
284 and 0.25 ± 0.04 MPa (Figure 4A, B). Digital images show the flexibility of 3PCLMA when compressed by
285 fingers while the more rigid 4PCLMA resisted being bent (Figure 4C-F).

286 Overall; there are two main differences between the structural properties of 3PCLMA and 4PCLMA
287 used in this study; (i) the number of branching (ii) the molecular weight. The main reasons for these
288 10-fold and 35-fold differences in UTS and Young's Modulus values of 3PCLMA and 4PCLMA are likely
289 to be the difference in their crosslinking density and molecular weights (20-folds). 3PCLMA and
290 4PCLMA have a similar degree of functionalisation, but 4PCLMA has an additional branch, which will
291 increase the crosslinking. Green et al. reported the UTS and tensile modulus of 79% acrylate-
292 functionalised 3PCL (3PCLA, Mw: 900 g/mol) as 0.58 ± 0.05 MPa and 4.0 ± 0.5 MPa, respectively. This is
293 also in line with the value reported by Field et al. of 3.51 ± 0.5 MPa for 77% methacrylate functionalized
294 3PCL (3PCLMA, Mw 900 g/mol) ⁶³. Around 2 to the 4-fold difference between these mechanical
295 properties is likely to be the difference in the degree of functionalisation.

296 It has been reported that an increasing number of arms enhance the mechanical properties of
297 polymers ^{53,67}. Doganci et al. also, previously reported that three-armed star-shaped PCL has
298 significantly lower elastic modulus due to high molecular chain mobility of 3-arm PCL compared to
299 linear, four- and six-armed PCLs ⁶⁸.

300 Secondly, an increase in the higher molecular weight of the polymers generally increases the
301 mechanical strength of polymers and the glass transition temperature which has an impact on the
302 storage modulus ⁶⁹. Tian et al. investigated the effects of molecular weight on properties of PCL
303 networks and revealed that the length of the molecular chains directly influences the thermal and
304 mechanical properties of the networks. Melting temperature (T_m), crystallinity, the crystallization
305 temperature of the prepared PCL networks was shown to increase with increasing molecular weight.
306 Also, polymers with higher molecular weight resulted in enhanced mechanical strength which was
307 attributed to a higher degree of crystallization ^{70,71}. Similarly, Wang et al. reported the role of
308 crystallinity on mechanical properties of photo-crosslinked poly(3-caprolactone fumarate)
309 networks ⁷².

310



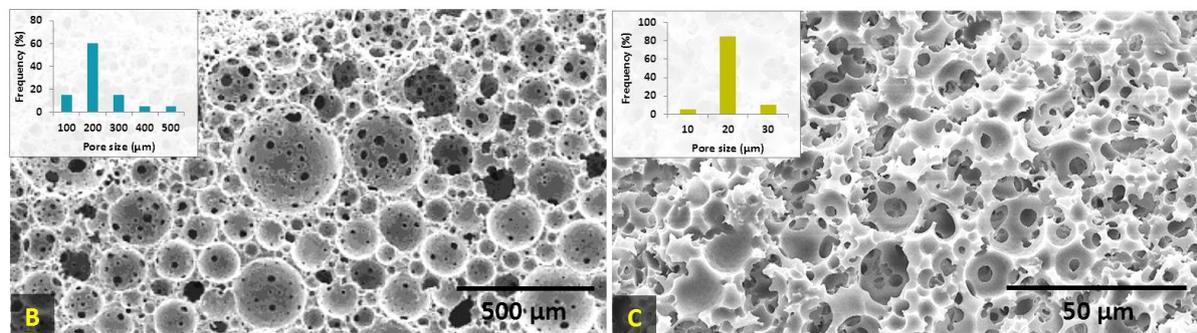
311
 312 **Figure 4:** (A) UTS, and (B) Young's Modulus values obtained from tensile testing of 3PCLMA and 4PCLMA
 313 (mean \pm STD error bars, $n = 3$). Digital photographs highlight the elastic properties of (C, D) 3PCLMA and (E, F)
 314 4PCLMA before and during finger compression, respectively.

315 3.2. Fabrication of PCLMA PolyHIPE Scaffolds

316 The emulsion ingredients and SEM pictures of the corresponding Thiol-3PCLMA and Thiol-4PCLMA
 317 PolyHIPEs are given in Figure 5A. The impact of solvent type and amount²³, water volume¹⁷, emulsion
 318 temperature⁷³, and surfactant composition⁷⁴ on the morphology of the emulsion templated scaffolds
 319 has been previously reported. We intended to produce scaffolds with different pore sizes ranging from
 320 ten to a few hundred micrometres to show the tunability of this manufacturing method. Average pore
 321 sizes of thiol-3PCLMA PolyHIPE and thiol-4PCLMA PolyHIPE were calculated as $176.4 \pm 82.2 \mu\text{m}$ and
 322 $15.7 \pm 3.0 \mu\text{m}$, respectively (Figure 5). Both of the formulations of PolyHIPE have shown open cellular
 323 morphology which is characterised by the presence of the windows between neighbouring pores
 324 (Figure 5B, C). The average window sizes were measured as $7.8 \pm 8.9 \mu\text{m}$ and $4.5 \pm 1.7 \mu\text{m}$ for the same
 325 groups, respectively. DOI and DOO values of thiol-4PCLMA PolyHIPE samples were calculated 7 and
 326 3.6 fold higher than DOI and DOO values of thiol-3PCLMA PolyHIPEs. Although the number of pores
 327 was significantly higher in thiol-3PCLMA PolyHIPEs, as their relative window diameter (to the pore
 328 size) is smaller, it resulted in lower DOI and DOO in thiol-3PCLMA PolyHIPEs compared to thiol-
 329 4PCLMA PolyHIPE.

330

Group	PCL (g)	Trithiol (g)	Surf. (g)	Crosslinker (g)	Solvent (g)	PI (g)	H ₂ O (mL)	T (°C)	D ^a (μm)	d ^b (μm)	DOI ^c	DOO ^d	Porosity ^e	Density ^f (g/cm ³)
3PCLMA	0.17	0.18	0.15	0.11 (TMPTA)	0.27 (CHCl ₃)	0.03	4	60	176.4 ± 82.2	7.8 ± 8.9	0.04	0.07	87	0.14
4PCLMA	5.00	3.22	0.15	2.50 (TMPTMA)	11.25 (DCE)	0.56	35	23	15.7 ± 3.0	4.5 ± 1.7	0.28	0.25	77	0.26



331

332 **Figure 5:** (A) Composition, (B, C) Pore size distribution histograms and SEM micrographs of thiol-3PCLMA
 333 PolyHIPE and thiol-4PCLMA PolyHIPE, respectively. (a: average pore size, b: average window size, c: degree of
 334 interconnectivity, d: degree of openness, e: nominal porosity, f: expected density calculated based on porosity).

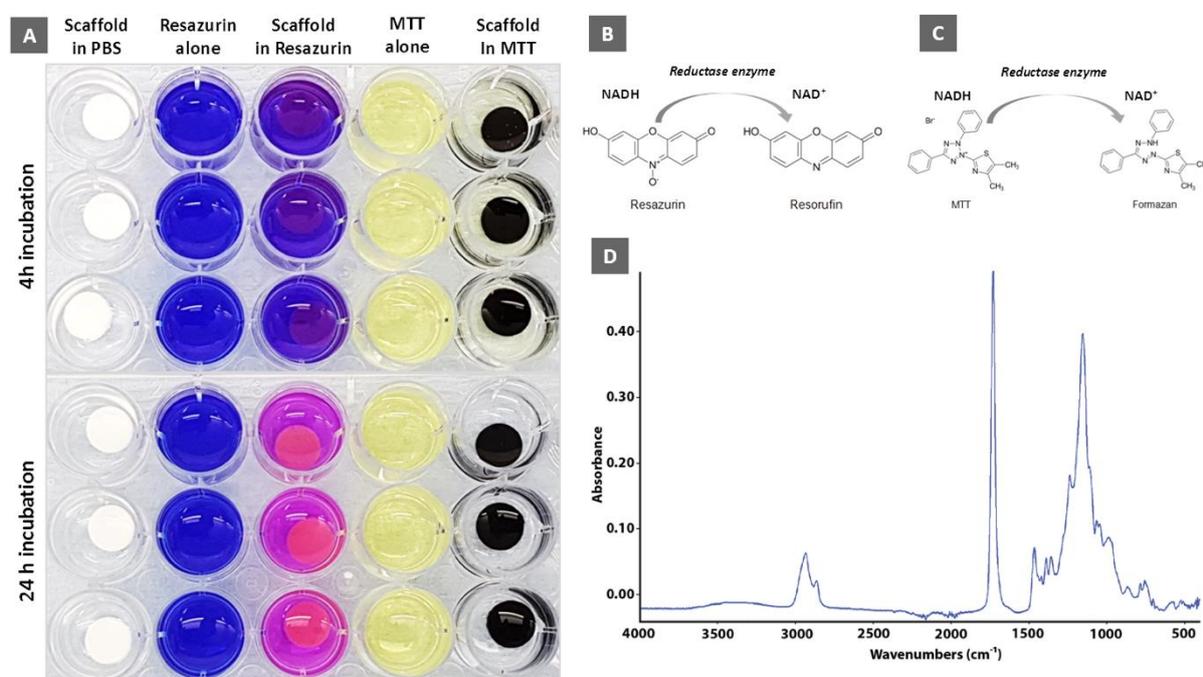
335 As higher temperature also reduced the emulsion stability, the pore size distribution of 3PCLMA based
 336 PolyHIPE compared to the 4PCLMA was comparably wider (Figure 5B, C), which results in a significantly
 337 higher standard deviation than the average pore size seen in PolyHIPE scaffolds^{26,40}. It is likely that
 338 the higher emulsion mixing temperature of 60°C is the main factor for the difference in the pore size
 339 between these samples as this reduces the emulsion stability, which causes more droplet coalescence
 340 and therefore larger pores. Also, there was a significant difference between the forms and viscosities
 341 of the polymers used in this study; while 3PCLMA was runny, liquid, 4PCLMA was solid at room
 342 temperature. This can be explained with ~20-folds molecular weight difference between 3PCLMA and
 343 4PCLMA, as the viscosity of PCL is expected to increase with increasing molecular weight^{45,75}. As the
 344 viscosity of the polymer directly affects the mixing efficiency of the two phases and emulsion
 345 formation, this may also contribute to a significant morphological difference between PolyHIPE
 346 groups. Additionally, the type of the diluting solvent used in the composition was reported to have
 347 an impact on the emulsion stability and average pore size of PolyHIPEs²³. As chloroform has a higher
 348 density (1.480 g/mL) than DCE (1.256 g/mL), its use as a diluting solvent in the oil phase increases the
 349 density difference between the oil and water phase which increases the velocity of a single droplet in
 350 the emulsion according to Stoke`s Equation⁷ and results in larger pore size.

351 3.3. Biological characterisation of PolyHIPE scaffolds

352 To assess the biocompatibility of the scaffolds, initially, an MTT assay was used. However, it was found
 353 that the scaffold alone also gave a positive signal from MTT (Figure 6). Also, Resazurin Reduction Assay

354 without any cells gave a false-positive result (Figure 6). MTT and RR assays are based on reduction of
 355 resazurin and MTT by reductase enzyme, present in mitochondria of metabolically active cells, into
 356 resorufin and formazan crystals, respectively. The initial blue and yellow colours are expected to turn
 357 to pink and black for MTT and RR, respectively.

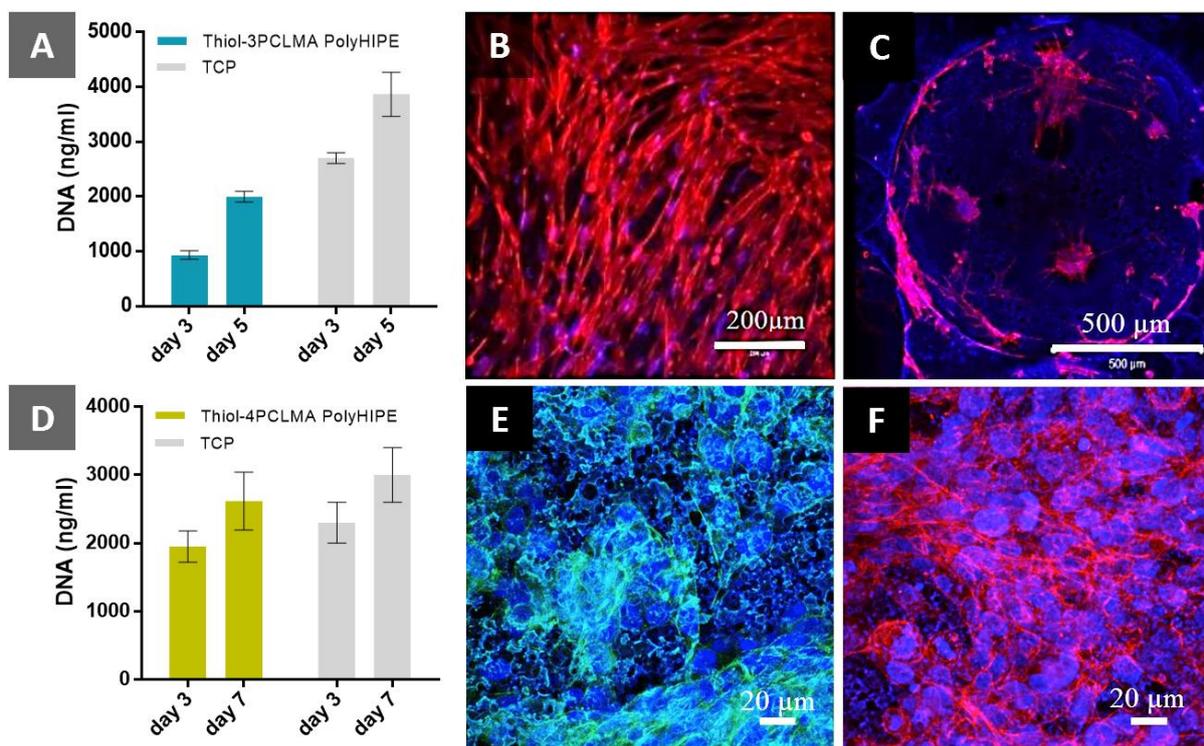
358 FTIR spectrometry of the scaffolds shown free -SH groups on the polymer, which are likely the reducing
 359 agents for the metabolic dyes (small band between 685-690 cm^{-1} representing S-H band). So, the
 360 correlation of cell growth using their metabolic activity via MTT and RR is likely to provide spurious
 361 false-positive results. The fluorescence of the blanks samples should be measured and subtracted as
 362 a background absorbance to avoid overestimation of cell viability due to this noise ⁷⁶. Similarly,
 363 Langford et al. have shown the residual thiols of thiolene (meth)acrylate PolyHIPEs using Ellman's
 364 assay, which is a colourimetric assay ³³.



365
 366 **Figure 6:** (A) Testing the reaction of the scaffold (without any cells) itself with resazurin and MTT solutions. C/PBS:
 367 scaffold in PBS (control of scaffold), C/RR: resazurin solution alone (control of resazurin solution (RS)), S/RR:
 368 scaffold in RS (to test the reaction of the scaffold with RS), C/MTT: MTT solution alone (control of MTT solution),
 369 S/MTT: scaffold in MTT solution (to test the reaction of the scaffold with MTT solution), Incubation PCL Thiol
 370 PolyHIPE Discs in Resazurin Reduction and MTT solution. Working principles of (B) Resazurin Reduction assay and
 371 (C) MTT assay. (D) FTIR spectrum of PolyHIPE disks.

372 In order to overcome the problem associated with MTT and RR assays being reduced by the polymer,
 373 PicoGreen DNA quantification assay was used to assess the DNA content of the samples to determine
 374 the biocompatibility of PolyHIPE scaffolds. According to PicoGreen data, the cell viability of fibroblasts

375 shown an increasing trend from day three to day five on thiol-PCLMA PolyHIPEs (Figure 7A). A confocal
 376 image of seven days culture of fibroblast on flat sheets of thiol-3PCLMA polymer (not PolyHIPE) shows
 377 that cells attached, proliferated, and elongated on the polymer surface. Similarly, fibroblasts sit on the
 378 pores and spread over the walls of the PolyHIPE scaffold. Although polymer gives a blue background,
 379 Phalloiding TRITC staining clearly shown the cell cytoskeleton. Blue autofluorescence of thiol-PCL
 380 based PolyHIPEs also has been reported ³⁰. Similarly, the cell viability of MG63s on Thiol-4PCLMA
 381 PolyHIPE showed an increasing trend from day 3 to day 7. Seven days culture of MG63s were stained
 382 with both Phalloidin FITC and Phalloidin TRITC. Confocal images of both stainings show that cells
 383 elongated and spread over the pores of the PolyHIPE scaffold.



384

385 **Figure 7:** (A) Cell viability of fibroblasts on thiol-3PCLMA PolyHIPE scaffolds (n=3). (B) Confocal
 386 image of 7-days culture of fibroblasts growing on thiol-3PCLMA. (C) Fluorescent confocal images of human dermal fibroblast cells
 387 growing on the PCL/Thiol PolyHIPE. (D) Cell viability of MG63s on thiol-4PCLMA PolyHIPE scaffolds (n=3). (E-F)
 388 Confocal image of 7-days culture of MG63s growing on thiol-4PCLMA. DAPI: blue, Phalloidin FITC: green, and
 389 Phalloidin TRITC: red.

390 Previously Johnson et al. also reported the cell viability of L929 fibroblast on thiol-triacrylate
 391 functionalised PCL PolyHIPEs ³⁰. In this study, similar to our previous finding, cells tend to locate in the
 392 pores of PolyHIPEs with pore size larger than 20 μm , while they spread over the pores that have a
 393 diameter less than 20 μm ²³. Both 3PCLMA and 4PCLMA based thiolene PolyHIPEs have been shown
 394 to support cell attachment and cell viability.

395 4. Conclusion

396 In this study, thiol based PolyHIPEs using either 3PCLMA or 4PCLMA were fabricated successfully, and
397 a tunability porosity was demonstrated. *In vitro* biocompatibility of PolyHIPEs was investigated using
398 both HDFs and MG63 cells, and thiolene/acrylate-based PolyHIPEs were shown to provide a
399 biocompatible substrate in terms of cell attachment and viability. In conclusion, this research
400 demonstrated a route to synthesise novel Poly-HIPE PCL-based biomaterial that could be suitable for
401 tissue engineering and regenerative medicine applications.

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413 6. Author contributions

414 BAD contributed to the experimental design, analysis, data acquisition, and interpretation of data,
415 statistical analysis, and drafting of the manuscript. AM contributed to the experimental design,
416 analysis, data acquisition, interpretation of data and drafting of the manuscript. CS, SD, TP, and LD
417 contributed to the experimental design, analysis, and data acquisition. CL and NC provided technical
418 knowledge and equipment training. PH, IO, SM, NC, and FC contributed with their supervision and
419 critical revision and editing of the manuscript.

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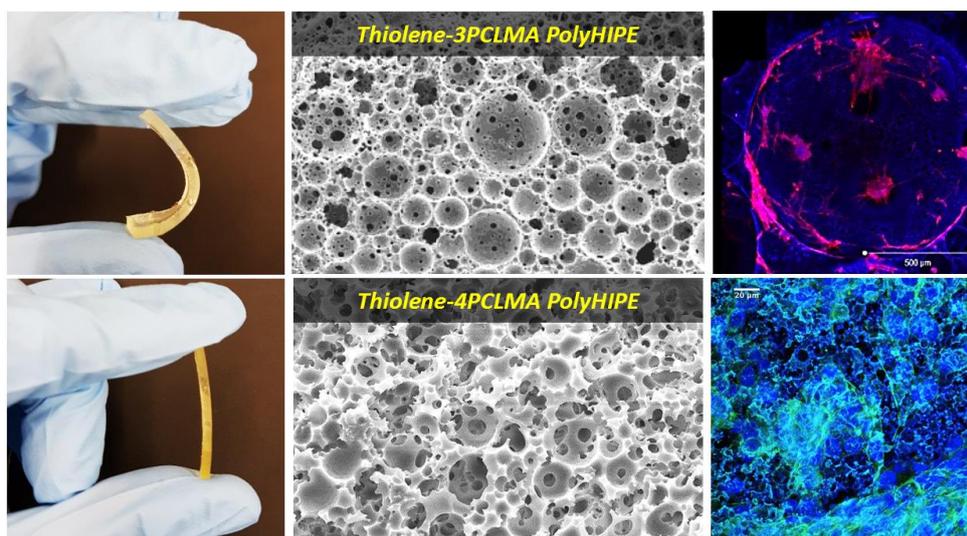
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