



UNIVERSITY OF LEEDS

This is a repository copy of *Progressive development of ocean anoxia in the end-Permian pelagic Panthalassa*.

White Rose Research Online URL for this paper:

<https://eprints.whiterose.ac.uk/178489/>

Version: Accepted Version

Article:

Takahashi, S, Hori, RS, Yamakita, S et al. (9 more authors) (2021) Progressive development of ocean anoxia in the end-Permian pelagic Panthalassa. *Global and Planetary Change*, 207. 103650. ISSN 0921-8181

<https://doi.org/10.1016/j.gloplacha.2021.103650>

© 2021 Published by Elsevier B.V. This is an author produced version of an article published in *Global and Planetary Change*. Uploaded in accordance with the publisher's self-archiving policy. This manuscript version is made available under the CC-BY-NC-ND 4.0 license <http://creativecommons.org/licenses/by-nc-nd/4.0/>.

Reuse

This article is distributed under the terms of the Creative Commons Attribution-NonCommercial-NoDerivs (CC BY-NC-ND) licence. This licence only allows you to download this work and share it with others as long as you credit the authors, but you can't change the article in any way or use it commercially. More information and the full terms of the licence here: <https://creativecommons.org/licenses/>

Takedown

If you consider content in White Rose Research Online to be in breach of UK law, please notify us by emailing eprints@whiterose.ac.uk including the URL of the record and the reason for the withdrawal request.



eprints@whiterose.ac.uk
<https://eprints.whiterose.ac.uk/>

1 Title: Progressive development of ocean anoxia in the end-Permian pelagic Panthalassa

2

3 Authors: Satoshi Takahashi¹, Rie S. Hori², Satoshi Yamakita³, Yoshiaki Aita⁴, Atsushi
4 Takemura⁵, Minoru Ikehara⁶, Yijun Xiong⁷, Simon W. Poulton⁷, Paul B. Wignall⁷, Takaaki
5 Itai¹, Hamish J. Campbell⁸, Bernard K. Spörli⁹

6

7 1. Department of Earth and Planetary Science, University of Tokyo, Japan.

8 2. Department of Earth Sciences, Graduate School of Science and Engineering, Ehime

9 University, Japan.

10 3. Department of Earth Sciences, Faculty of Culture, Miyazaki University, Japan.

11 4. Department of Geology, Faculty of Agriculture, Utsunomiya University, Japan.

12 5. Geosciences Institute, Hyogo University of Teacher Education, Japan.

13 6. Centre for Advanced Marine Core Research, Kochi University, Japan.

14 7. School of Earth and Environment, University of Leeds, Leeds LS2 9JT, United Kingdom.

15 8. GNS Science, New Zealand.

16 9. School of Environment, The University of Auckland, New Zealand.

17

18 Keywords: carbon isotope, iron speciation, redox-sensitive metals, mass extinction, ocean
19 anoxia, pelagic deep sea, Permian-Triassic boundary

20 Abstract

21 The end-Permian mass extinction (EPME) has been linked with the widespread
22 development of oxygen-poor oceanic conditions. However, information on the spatial extent
23 of anoxia in the Panthalassa super-ocean has been limited. This study reports oceanic redox
24 records from a deep-sea chert succession (the Waiheke 1 section, WHK 1, New Zealand) that
25 was located in southern mid-latitudes of Panthalassa. High-resolution carbon isotope ($\delta^{13}\text{C}$)
26 correlation between Waiheke and the Permian-Triassic boundary (PTB) type section indicates
27 that the EPME is recorded in a thin black claystone interbedded between siliceous mudstone
28 beds at WHK 1. Pyrite-dominated enrichment in highly reactive iron, coupled with elevated
29 U/Al and Mo/Al ratios, are prevalent through this black claystone bed and the overlying
30 Permo-Triassic transition strata, suggesting the development of euxinic water column
31 conditions. Similar redox variations across the EPME horizon have been reported from other
32 Panthalassic deep-sea PTB sections. Comparison with these PTB sections indicates that
33 euxinic conditions were widespread in low-latitude regions of the Panthalassan ocean, and

34 such conditions developed earlier than in mid-latitude settings, up to 100,000 years before the
35 EPME. This suggests there was a gradual expansion of ocean anoxia from low to middle-high
36 latitude regions during the Permo-Triassic transition. The extent of ocean anoxia resulted in a
37 decrease in the seawater inventory of redox sensitive trace metals (e.g., Mo), which is evident
38 in the earliest Triassic strata of the studied section and other PTB sections. Panthalassic
39 anoxia during the EPME coincides with extreme climate warming and the associated effects
40 (e.g., changes in ocean circulation, marine eutrophication intensified by terrestrial
41 weathering) were likely critical triggers for ocean deoxygenation.

42

43 1. Introduction

44 The most significant biotic catastrophe of the Phanerozoic, the end-Permian mass extinction
45 event (EPME), was associated with the widespread development of oceanic anoxia (e.g.,
46 Wignall and Twitchett, 1996; Isozaki, 1997). Previous research has demonstrated that many
47 shallow water and deep oceanic regions of Late Permian and Early Triassic age experienced
48 anoxic conditions (e.g., Wignall and Twitchett, 2002; Kaiho et al., 2016). Organic molecular
49 fossils (biomarkers) of anaerobic green sulphur bacteria have shown that euxinia (anoxic,
50 sulphidic conditions) extended into the photic zone in the paleo-Tethys (e.g., Grice et al.,
51 2005, locality 3 in Figure 1A) and Panthalassan oceans (Hays et al., 2007, locality 5 in Figure
52 1A). Enrichments in pyrite and redox-sensitive trace elements have also been reported from
53 PTB sections of paleo-Tethys (e.g., Lei et al., 2017, Xiang et al., 2020, Shen et al. 2016,
54 Clarkson et al., 2016), pelagic Panthalassa (locality 2 in Figure 1A; Algeo et al., 2011;
55 Takahashi et al., 2014; Onoue et al., 2021), and the North Pangean margin (e.g., locality 4 in
56 Figure 1A; Grasby et al., 2009; Schobben et al., 2020), indicating the widespread
57 development of anoxia/euxinia. The extent of oxygen-poor conditions may have caused a
58 decrease in both seawater sulphate concentrations (Song et al., 2014; Schobben et al., 2017)
59 and trace elements such as Mo (e.g., Takahashi et al., 2014; Xiang et al., 2020), due to
60 enhanced pyrite burial and trace metal drawdown (Takahashi et al., 2014; Schobben et al.,
61 2015). The extent of ocean anoxia also changed the U isotope ratio of seawater, as recorded
62 in marine carbonate ($\delta^{238}\text{U}_{\text{carbonate}}$) (Brennecka et al., 2011, Lau et al., 2016; Zhang et al.,
63 2018ab, 2020).

64 While anoxia was clearly widespread, there remains a significant absence of
65 environmental records from the Panthalassa (Fig. 1A). Since most of the sedimentary record
66 of Permian-Triassic Panthalassic seafloor has been subducted, limited material is available to

67 allow reconstruction of redox conditions, particularly in terms of revealing temporal and
68 spatial variability in oceanic redox during the EMPE. Previous research has, however,
69 focussed on Panthalassic PTB sections from Japanese accretionary complexes. These formed
70 in low latitude central Panthalassa (locality 2 in Figure 1A), based on measured paleo-
71 latitudes of Middle Triassic chert from the Mino Belt of central Japan (cf. paleomagnetic
72 studies by Ando et al., 2001 and Oda and Suzuki, 2000). It was at least one hundred million
73 years before these sediments were accreted in the Middle to Late Jurassic (Mastuda and
74 Isozaki, 1991; Takahashi et al., 2014). Since oceanic plates migrate at a rate of several cm/yr,
75 this indicates that the Japanese PTB sections were originally thousands of kilometres from
76 land.

77 According to reports from Japan (e.g., Isozaki, 2009, Muto et al., 2020), the EPME
78 horizon occurs at the transition from Permian radiolarian chert to siliceous claystone and
79 black claystone layers. Based on the concentrations of redox-sensitive trace elements (Mo, U
80 and V) and framboidal pyrite populations in the PTB sections, a shift to oxygen-poor
81 conditions during the EPME has been suggested, with the specific development of either
82 anoxic through to euxinic bottom waters, or an oxygen minimum zone setting (Wignall et al.,
83 2010; Algeo et al., 2011; Takahashi et al., 2014; Onoue et al., 2021).

84 Southern Panthalassa PTB sections have also been documented in the accretionary
85 complex on the North Island of New Zealand (Spörli et al., 2007a; Hori et al., 2011; Grasby
86 et al., 2021), where data from Lower Triassic chert suggests a paleolatitude of ca. 34°S
87 (Kodama et al., 2007). The Permian-Triassic deposits accreted during the Late Triassic,
88 suggesting the original depositional location was close to Gondwana (Spörli et al, 2007b;
89 locality 1 in Figure 1A). Two sections from Arrow Rocks and Waiheke Island (Fig.1B)
90 indicate that black, bedded chert and claystone occur around the EPME, similar to their
91 occurrence in the Japanese sequences. Hori et al. (2007) measured trace element enrichments
92 (Mo and U) in the Griesbachian black chert layers of the Arrow Rocks (Unit 2) and identified
93 an anoxic water column event they termed “OAE α ” (they described an “OAE β ” in the
94 Dienerian). However, this PTB section has minor faults around the basal part of the black
95 chert layers, making the continuous redox history in relation to the EPME and PTB unclear.

96 The PTB section on Waiheke Island (Waiheke section) has a continuous succession across
97 the PTB, which has been dated using conodonts (Hori et al., 2011). Organic carbon isotope,
98 sulphide sulphur isotope, and trace element profiles through this section suggest the
99 development of anoxic-sulphidic water column conditions during the EPME (Hori et al.,
100 2007; Takahashi et al., 2013; Grasby et al., 2021). Here, we present new, high-resolution

101 paleoenvironmental records for the Waiheke section, incorporating multiple paleoredox
102 proxies, including iron speciation and redox-sensitive trace element systematics. We aim to
103 reveal how ocean anoxia developed prior to and after the EPME, as well as the associated
104 changes in seawater chemistry. In addition, we present a compilation of pelagic Panthalassic
105 PTB records to investigate the spatial and temporal development of redox conditions in the
106 Permian-Triassic Panthalassa.

107

108 2. Geological setting

109 The Waiheke PTB section is an intertidal platform section in Island Bay, north-western
110 Waiheke Island, North Island, New Zealand (Fig. 1B). It belongs to the Kiripaka Formation
111 of the Waipapa composite terrain (Schofield, 1974, 1979), which is the accretionary complex
112 forming part of the basement of New Zealand. The sedimentary sequence comprising the
113 Waiheke section is approximately 10 m thick and divided into six litho-units (Unit A to F;
114 Hori et al., 2011). This study focuses on the lowest part of these units (Unit A), in the
115 Waiheke 1 section (abbreviated as “WHK 1” hereafter). It mainly consists of pale green
116 tuffaceous claystone and mudstone towards the base, followed by black and yellowish-black
117 chert with black claystone (Fig. 2).

118 The pale green siliceous and tuffaceous claystone and mudstone in the lower half of Unit
119 A are composed of mm- to cm-scale alternations of siliceous microfossil enriched layers
120 (radiolarians and sponges) and muddy layers (Fig. 3A). The black claystone consists of fine-
121 grained clay materials and organic matter (Fig. 3B). Although siliceous microfossils are
122 observed in the black claystone, fossils are rare and generally small (up to several μm in
123 diameter). Black and black-yellow cherts in the upper part of WHK 1 also show mm- to cm-
124 scale microfossil enriched layers and muddy layers (Figs 3C, 3D). The size of visible
125 microfossils are similar to those from the pale green beds in the lower part of the study
126 section. 88 single beds have been identified and labelled as Beds 1 to 88 (Hori et al., 2011;
127 Takahashi et al., 2013). The conodont occurrences indicate that the section ranges from the
128 uppermost Permian to the lowermost Triassic (Figs. 2 and 4; Hori et al., 2011). *Hindeodus* sp.
129 and *Neogondolella carinata* were obtained from Bed 29 of the black-yellow chert. Above this
130 horizon, *Ng. carinata* and *Sweetospathodus* cf. *kummeli* occur from Bed 40. These index
131 fossils indicate that Bed 29 is Induan, and Bed 40 extends into the lower to upper part of
132 Induan (uppermost Griesbachian or even the Dienerian). Ten previously published organic
133 carbon isotope ($\delta^{13}\text{C}_{\text{org}}$) analyses demonstrate a negative excursion from Bed 14 to Bed 22,

134 indicating the uppermost Permian carbon isotopic signal corresponding to the end-Permian
135 mass extinction event (Hori et al., 2011). Based on this information, we place the Permian-
136 Triassic transitional zone at this level. Here, we provide further $\delta^{13}\text{C}_{\text{org}}$ data and discuss the
137 timescale issue in more detail.

138 3. Methods

139 *3.1. Organic carbon isotope analysis*

140 We examined organic carbon isotope ratios of bulk chert and claystone samples from 28
141 horizons of WHK 1, using samples with no visible veinlets or staining. After washing with
142 Ellix® water, all samples were ground into fine powder using an agate mill. The powdered
143 samples were decarbonated with 10% HCl. Organic carbon isotopic compositions and
144 organic carbon contents were measured using an elemental analyser (Flash EA 1112) coupled
145 with a Thermo-Finnigan Delta plus Advantage isotope ratio mass spectrometer at the Center
146 for Advanced Marine Core Research, Kochi University, Japan. Isotopic measurements were
147 repeated up to four times for each horizon to check reproducibility. The carbon contents were
148 calculated using standard sulfanilamide, and the precision of organic carbon isotope ($\delta^{13}\text{C}_{\text{org}}$)
149 analyses was better than 0.1‰. Organic carbon isotope measurements were calibrated using
150 histidine and alanine obtained from Sugito Ltd. Lab. All isotopic results are reported in
151 conventional delta (δ) notation, defined as per mil (‰) deviation from the Pee Dee belemnite
152 (PDB) standard value. Here we use the average values calculated from repeat (up to 4 times)
153 measurements of the same sample.

154

155 *3.2. Major and trace element analysis*

156 Major and trace elements were measured by inductively coupled plasma atomic emission
157 spectrometry (ICP-AES; SPS3500, SII nanotechnology) and inductively coupled plasma
158 mass spectrometry (ICP-MS; iCAPQc, Thermo scientific) at the Department of Earth and
159 Planetary Science, University of Tokyo. For these analyses, 0.5 g of sample powder was
160 treated with 4 mL of $\text{HClO}_3 + \text{HNO}_3$ (1:1 mixture) in PTFE beakers for 45 min at 120°C,
161 followed by two treatments with 4 mL of $\text{HClO}_3 + \text{HF}$ (1:1 mixture) for 30 min at 120°C.
162 The residues from these acid treatments were re-dissolved with 5 mL HNO_3 and diluted to
163 1:2000 using ultra pure water. Indium (In) and bismuth (Bi) were used as internal standards
164 for the ICP-MS analyses. Working standards were prepared from a series of SPEX Multi-
165 Element Plasma Standards supplied by SPEX Industries (SPEX). Repeated analysis on the
166 reference material (GSJ-JR-1; rhyolite reference material supplied by Geological Survey of

167 Japan) confirms that values are reproducible to within 1.8% for Al, 0.8% for Fe, 12% for U,
 168 and 9% for Mo, respectively (as relative standard deviation: RSD).

169

170 3.3. Normalization

171 To assess the enrichment or depletion of redox sensitive elements relative to typical
 172 detrital material, enrichment factors (X_{EF}) were calculated as follows (e.g., Tribovillard et al.,
 173 2006):

$$174 \quad X/Al = X_{\text{sample}} \text{ (ppm)} / Al_{\text{sample}} \text{ (wt \%)} \quad (1)$$

$$175 \quad X_{EF} = (X_{\text{sample}} / Al_{\text{sample}}) / (X_{PAAS} / Al_{PAAS}) \quad (2)$$

176 Here, X_{sample} , Al_{sample} , X_{PAAS} and Al_{PAAS} are the weight concentrations of element X or Al
 177 in the sample or in PAAS (the post-Archean average shale; Tayler and McLennan, 1985). To
 178 compare our geochemical data with other PTB sections, Fe/Al, U/Al and Mo/Al ratios were
 179 also calculated and compared with literature data (Algeo et al., 2011; Takahashi et al., 2014).

180

181 3.4. Iron speciation

182 A sequential extraction targeting different operationally defined Fe pools, including
 183 carbonate ($Fe_{\text{carbonate}}$), ferric (oxyhydr)oxides (Fe_{OX}) and magnetite ($Fe_{\text{magnetite}}$), was
 184 performed on a split of 0.07–0.09 g of powdered sample (Poulton and Canfield, 2005;
 185 Poulton, 2021). $Fe_{\text{carbonate}}$ was first extracted using Na-acetate at pH 4.5 and 50 °C for 48 h,
 186 followed by Fe_{OX} using Na-dithionite for 2 hours at room temperature, and finally $Fe_{\text{magnetite}}$
 187 was determined with an ammonium oxalate extraction for 6 h. Dissolved iron concentrations
 188 (for $Fe_{\text{carbonate}}$, Fe_{OX} and $Fe_{\text{magnetite}}$ pools) were measured by atomic absorption spectrometry
 189 (AAS; Thermo ice 3000 at the School of Earth and Environment, University of Leeds).
 190 Replicate extractions of an international reference material (WHIT; Alcott et al., 2021) gave a
 191 RSD of <5% for each Fe pool.

192 Pyrite iron (Fe_{pyrite}) was determined stoichiometrically based on Ag_2S precipitates formed
 193 after a boiling HCl and chromous chloride distillation (Canfield et al., 1986). All samples
 194 were tested for the presence of acid volatile sulphide (AVS; predominantly FeS) via an initial
 195 6 M HCl distillation, but in all cases AVS was below detection (<0.001 wt %). Replicate
 196 pyrite extractions gave a RSD of <5%. The sum of these measured Fe pools defines the total
 197 concentration of highly reactive Fe (Fe_{HR}):

$$198 \quad Fe_{HR} = Fe_{\text{carbonate}} + Fe_{OX} + Fe_{\text{magnetite}} + Fe_{\text{pyrite}} \quad (1)$$

199

200 A 1 h 0.5 N HCl extraction was also conducted on ~0.5_g splits of powdered samples to
 201 assess for potential oxidation during weathering (see Matthews et al., 2017). This treatment
 202 targets poorly crystalline ferric oxide minerals (Fe(III)_{HCl}), such as ferrihydrite, in addition to
 203 reduced Fe phases such as Fe_{AVS} and potentially some ferrous carbonate/phosphate phases.
 204 Since Fe_{AVS} was not present in our samples (see above), this phase can be discounted. Thus,
 205 this extraction targets the remaining non-sulphidized Fe(II) minerals (termed Fe(II)_{HCl}), as
 206 well as Fe(III)_{HCl}. Poorly crystalline Fe(III)_{HCl} minerals such as ferrihydrite are not stable on
 207 geological timescales, and hence the Fe(III)_{HCl} pool can be considered to dominantly result
 208 from more recent weathering of reduced minerals (particularly pyrite). Thus, quantification of
 209 this pool provides a maximum estimate of the extent of post-depositional pyrite oxidation
 210 (e.g., Matthews et al., 2017).

211 The Fe(II)_{HCl} pool was determined immediately by spectrophotometer using the ferrozine
 212 assay (Stookey, 1970). The total amount of Fe dissolved by the 0.5 N HCl extraction was
 213 determined following reduction of Fe(III) to Fe(II) using hydroxylamine hydrochloride,
 214 followed by analysis by spectrophotometer using the ferrozine assay. Subtraction of Fe(II)_{HCl}
 215 from this Fe pool gives Fe(III)_{HCl}. Corrected Fe pools were calculated as:

$$216 \quad \text{Corrected Fe}_{\text{carbonate}} (\text{Fe}^*_{\text{carbonate}}) = \text{Fe}_{\text{carbonate}} - \text{Fe(III)}_{\text{HCl}} \quad (2)$$

$$217 \quad \text{Corrected Fe}_{\text{pyrite}} (\text{Fe}^*_{\text{pyrite}}) = \text{Fe}_{\text{pyrite}} + \text{Fe(III)}_{\text{HCl}} \quad (3)$$

218

219 4. Results

220 All geochemical data are shown in Table S1 and Figure 2.

221 4.1. Organic carbon isotopes

222 Organic carbon isotope ratios in WHK1 vary between -32.8‰ and -25.4‰, but with
 223 distinct trends up-section. Specifically, $\delta^{13}\text{C}_{\text{org}}$ values in Beds 1-7 vary between -27.5‰ and
 224 -25.4‰ (Fig. 2). Values then decrease to -28.6‰ in Bed 8, followed by an increase to-
 225 26.1‰ in Bed 14. The most pronounced decrease in $\delta^{13}\text{C}_{\text{org}}$ then occurs between beds 14 and
 226 17, reaching the minimum value of -32.8 ‰. Low $\delta^{13}\text{C}_{\text{org}}$ values persist up to Bed 28, before
 227 a gradually increasing trend, interrupted by a low value in Bed 48 (-29.2‰), is established up
 228 to bed 50. Above Bed 50, $\delta^{13}\text{C}_{\text{org}}$ values are relatively stable around -26.9‰, with variability
 229 within 0.6‰.

230

231 *4.2. Major and trace elements*

232 Concentrations of Al and Fe range between 0.74–5.2 wt % and 0.26–6.4 wt %, respectively
233 (Table S1). Beds 35 and 38 have low values of 0.74 and 0.75 wt % for Al, and 0.42 and 0.40
234 wt % for Fe, respectively. The Al content is relatively high in the tuffaceous claystone and
235 mudstone (3.3–5.2 wt %) and the black claystone (1.7–4.2 wt %), but is low in chert beds (0.9–
236 2.7 wt %). Iron is high in the black claystone (more than 4 wt %) and the black chert of Beds
237 29 (6.4 wt %).

238 Concentrations of U and Mo vary between 1.0–39.1 ppm and 0.8–194.3 ppm, respectively.
239 Normalized U/Al ratios (Fig. 2) and U_{EF} (Table S1) values range between 0.84–13.5 and 2.7–
240 43.6, respectively. Notably, high U/Al ratios (>5) occur in Beds 17–35 and Bed 44. Mo/Al (Fig.
241 2) and Mo_{EF} (Table S1) vary between 0.3–61.1 and 2.4–611.0, respectively, with particularly
242 high Mo/Al ratios (>5) occurring in Beds 17–24 and Bed 29.

243

244 *4.3. Iron speciation*

245 Fe speciation profiles are shown in Figure 2. Relatively high concentrations of $Fe(III)_{HCl}$
246 (more than 0.7 wt %) were detected in the black claystone samples of Beds 22, 24 and 48, as
247 well as the black chert of Beds 23 and 29, which for these samples has a significant effect on
248 corrected $Fe_{carbonate}$ and Fe_{pyrite} concentrations. However, this correction for oxidative
249 weathering of pyrite brings the Fe speciation data more broadly in line with the trends evident
250 in Mo/Al ratios (Fig. 2). Indeed, both Fe_{pyrite}/Fe_{HR} and Mo/Al reflect the availability of sulphide
251 in the water column and during early diagenesis, and as such, the general similarity in the
252 position of peaks in Fe^*_{pyrite}/Fe_{HR} and Mo/Al (Fig. 2) provides strong support that our approach
253 is robust.

254 We distinguish the Fe speciation results for Beds 31, 35 and 50 (open circles in Figure 2)
255 from the remainder of the data, as these samples contain less than 0.5 wt % total Fe, and as
256 such should be viewed with caution when interpreting Fe speciation data (Clarkson et al., 2016).
257 We thus do not discuss the Fe speciation characteristics of these samples further. For the
258 remaining samples, Fe_{HR}/Fe_{total} ratios range from 0.15 to 1.0. Fe_{HR}/Fe_{total} ratios are higher than
259 0.38, which represents the conventional boundary for distinguishing samples deposited from
260 an oxic or anoxic water column (Raiswell and Canfield, 1998; Poulton and Canfield, 2011),
261 across most horizons, with the exception of Beds 46 and 58. Fe^*_{pyrite}/Fe_{HR} ratios range from
262 0.11 to 0.88, with low values (less than 0.22) in the siliceous siltstone and claystone beds of
263 the lower part of the section (Beds 1, 6, 7, 8, and 13). Fe^*_{pyrite}/Fe_{HR} ratios exceed 0.6 in the

264 black claystone and chert beds in the middle and upper parts of the section, except for Beds 20
265 and 28 which show moderate values (0.48 and 0.42, respectively).

266

267 5. Discussion

268 5.1. Carbon isotope correlation

269 The Upper Permian strata of WHK1 exhibits a two-step decrease in $\delta^{13}\text{C}_{\text{org}}$ values,
270 resulting in an overall decrease of ca. 8‰. The first decrease occurs in Beds 7-8, followed by
271 a larger decrease in Beds 14-17 (Figs. 2 and 4). The minimum $\delta^{13}\text{C}_{\text{org}}$ value occurs in the thin
272 black claystone bed (Bed 17) sandwiched between siliceous claystone beds. Then, $\delta^{13}\text{C}_{\text{org}}$
273 shows an increasing trend towards the overlying Lower Triassic black chert beds.

274 Inorganic carbonate carbon isotope ($\delta^{13}\text{C}_{\text{carb}}$) values from low-latitude carbonate PTB
275 sections show similar trends. One of the best documented examples is from the Meishan
276 section, the type section for the Permian-Triassic boundary (Fig. 4; GSSP: Global strata-type
277 section and point; Yin et al., 2001). The Meishan section also displays a two-step decrease in
278 $\delta^{13}\text{C}_{\text{carb}}$ in the uppermost Permian strata below the base of the Triassic (Beds 23-25 in Fig 4;
279 Jin et al., 2000; Cao et al., 2002; Kaiho et al., 2009). We label these two decreases in $\delta^{13}\text{C}$ as
280 “the first $\delta^{13}\text{C}$ decrease” and “the second $\delta^{13}\text{C}$ decrease”, respectively (marked by green and
281 blue horizontal bars in Figure 4).

282 Based on U-Pb ages from zircon minerals in volcanic ash beds at the GSSP, these two
283 isotopic steps are estimated to have occurred at 251.999 ± 0.039 Ma and 251.941 ± 0.037 Ma,
284 respectively (Burgess et al., 2014), suggesting a 58 kyr separation. The first and second
285 decreases are correlated with the two negative drops in $\delta^{13}\text{C}_{\text{org}}$ observed at WHK 1 (Fig. 4).
286 The second decrease in $\delta^{13}\text{C}_{\text{carb}}$ at Meishan corresponds to the EPME horizon, and is
287 associated with the loss of many marine taxa (Jin et al., 2000; Song et al., 2012). The
288 Meishan section also shows a gradually increasing trend in $\delta^{13}\text{C}_{\text{carb}}$ above the EPME horizon
289 up to the Permian-Triassic boundary and into the overlying lowest Triassic strata, similar to
290 the increasing trend seen in the strata of Beds 17-28 below the first Triassic conodont fossil
291 occurrence from Bed 29 in WHK 1 (Figs. 2 and 4). This increasing $\delta^{13}\text{C}$ trend above the
292 EPME horizon provides a further correlative interval between PTB sections (marked by the
293 pink horizontal bar in Figure 4). Therefore, based on this carbon isotope correlation within
294 the biostratigraphic framework for the Permian-Triassic transition, defined at the GSSP
295 Meishan section, we place the EPME horizon at the base of Bed 17 in WHK 1, which is a
296 thin black claystone bed within the uppermost pale-green siliceous mudstone beds (Figs. 2

297 and 4). Another deep-sea PTB section from New Zealand (Arrow Rocks) also has a published
 298 $\delta^{13}\text{C}_{\text{org}}$ profile across the PTB (Fig. 4; Hori et al., 2007). This PTB section also shows
 299 decreasing trends in $\delta^{13}\text{C}_{\text{org}}$ between a chert bed with the Changhsingian *Neogondolella*
 300 *prechangxingensis* in the lower part of the section (at -7 m, not shown in the Figure 4) and a
 301 Triassic black chert beds which yield conodonts of *Neogondolella carinata* and *Hindeodus*
 302 sp. of Triassic affinity (Yamakita et al., 2007). As the number of $\delta^{13}\text{C}$ data from the lower half
 303 of the section are limited, the exact positions of the first and second $\delta^{13}\text{C}$ decreases and their
 304 minimum troughs are uncertain. However, the $\delta^{13}\text{C}_{\text{org}}$ profile in the Permian chert and
 305 siliceous mudstone beds shows a decrease at -0.6 to -0.3 m, and then a further decrease across
 306 the fault dividing a siliceous claystone unit (Unit 2a; Hori et al., 2007) and overlying black
 307 chert unit (Unit 2b; Hori et al., 2007). These $\delta^{13}\text{C}$ trends are interpreted to correspond to the
 308 first $\delta^{13}\text{C}$ decrease and the negative trough above the second $\delta^{13}\text{C}$ decrease observed in the
 309 WHK 1 section (Fig. 4).

310 The two-step $\delta^{13}\text{C}_{\text{org}}$ decrease during the Permian-Triassic transition, followed by a
 311 gradual increase toward the PTB, are also observed in the low latitude, deep-sea PTB
 312 sections of the Canadian Ursula Creek section (Wang et al., 1994), as well as the Japanese
 313 Akkamori-2 (Takahashi et al., 2010; Fig. 4), Ubara (Kaiho et al., 2012, Fig. 4) and NF1212C
 314 (Sano et al., 2012) sections. In the Japanese PTB sections, located in central Panthalassa,
 315 black claystone commonly occurs instead of siliceous claystone and bedded chert during the
 316 second $\delta^{13}\text{C}$ decrease. This lithologic change reflects a significant decrease in siliceous
 317 microfossils (Takahashi et al., 2009). The black claystone facies in the Japanese PTB section
 318 continues during the Olenekian period, over a million years later (Takahashi et al., 2009). By
 319 contrast, WHK 1 has bedded chert beds just above the EPME horizon. The dominant black
 320 claystone facies of WHK 1 appears above these chert beds, with a return to chert beds below
 321 the PTB. This discrepancy in the timing of bedded chert deposition indicates a relatively
 322 short duration for the radiolarian chert gap in the southern mid-latitudes of Panthalassa.

323

324 5.2. Redox evaluation inferred from Fe speciation and trace elements in the Waiheke 1 325 section

326 5.2.1. Fe speciation systematics

327 Fe-speciation is a widely used proxy for reconstructing regional water column redox
 328 conditions (Poulton and Canfield, 2011; Poulton, 2021). Enrichments in Fe_{HR} ($\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}} >$
 329 0.38) commonly characterise deposition under anoxic water column conditions, and for

330 samples that show such enrichment, the extent of pyritization of the Fe_{HR} pool provides an
331 indication of whether the water column was ferruginous ($\text{Fe}_{\text{pyrite}}/\text{Fe}_{\text{HR}} < 0.6-0.8$) or euxinic
332 ($\text{Fe}_{\text{pyrite}}/\text{Fe}_{\text{HR}} > 0.6-0.8$). The samples that record a euxinic water column obtain syngenetic
333 pyrite from the water column, in addition to possible continued pyritization of Fe_{HR} during
334 diagenesis. The threshold values for $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}}$ and $\text{Fe}_{\text{pyrite}}/\text{Fe}_{\text{HR}}$ are not absolute, and may be
335 influenced by high sedimentation rates (particularly during turbidite deposition; e.g., Canfield
336 et al., 1996) or transfer of Fe_{HR} to clay minerals (such as glauconite) during diagenesis, the
337 latter of which may be particularly prevalent in organic-rich, non-sulphidic sediments (e.g.,
338 Poulton and Raiswell, 2002; Poulton et al., 2010). However, we found no evidence for
339 glauconite in the samples from WHK 1, based on thin section observations, which would be
340 consistent with the relatively high pyrite contents of many of the samples. Regarding
341 sedimentation rates, deep sea cherts in general have slow sedimentation rates (order of
342 mm/kyr; e.g., Mastuda and Isozaki, 1991). Indeed, the calculated sedimentation rate for
343 WHK 1 (3.8 mm/kyr) roughly corresponds to $0.001 \text{ g cm}^{-2}/\text{yr}$ ($= 0.38 \text{ cm/kyr} \times 2.7 \text{ g/cm}^3$),
344 based on dating for Beds 7 and 17 (see above), which would not be expected to impact Fe
345 speciation systematics. However, the tuffaceous sedimentary layers in this interval may
346 potentially have masked Fe_{HR} enrichments due to rapid sedimentation (see Poulton et al.,
347 2004).

348

349 *5.2.2. Uranium and molybdenum systematics*

350 Uranium and Mo exist as soluble oxyanions in oxic seawater (e.g., Tribovillard et al.,
351 2006), with a relatively homogenous distribution because of their longer residence times
352 relative to the mixing time of the ocean. Under reducing conditions, U transforms to
353 tetravalent U(IV), whereas in the presence of dissolved sulphide, Mo converts to
354 thiomolybdate. This results in enrichments in these elements, with Mo removal primarily
355 occurring under euxinic conditions (but a degree of drawdown can also occur in open system
356 sulphidic porewaters), whereas U uptake occurs within anoxic sediments (McManus et al.,
357 2006). Based on observations from modern environments, Mo enrichments may begin to
358 occur when transformation to thiomolybdates occurs ($>11 \mu\text{M H}_2\text{S}$; Helz et al., 2011). Scott
359 and Lyons (2012) suggest that sediment Mo concentrations >25 ppm commonly indicate at
360 least intermittent water column euxinia, whereas concentrations >100 ppm likely indicate a
361 persistent, strongly euxinic water column. Bennett and Canfield (2020) recommend using
362 U/Al and Mo/Al ratios as the most objective normalization approach for minimising

363 influence from detrital inputs. Furthermore, based on a compilation of modern settings,
364 Bennett and Canfield (2020) suggest that euxinic conditions are indicated when U/Al and
365 Mo/Al ratios exceed ~ 5 ($\mu\text{g/g}$), and oxygen-poor sea floor sediments ($\text{O}_2 \geq 10 \mu\text{M}$) are
366 indicated when U/Al ratios are >1 ($\mu\text{g/g}$).

367 Consistent with the emerging perspective that local baseline values for redox proxy
368 thresholds should be assigned wherever possible (Algeo and Li, 2020; Poulton, 2021), we
369 apply different threshold values to the very different lithologies that comprise the upper
370 cherty (partly muddy) strata (Beds 19-88) and the lower tuffaceous strata (Beds 1-18) (see
371 Figure 2). For the upper cherty sediments, two samples (Beds 46 and 58) that give a clear
372 oxic signal based on all of our applied redox proxies (with $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}} < 0.22$, low U/Al (0.90
373 and 0.84; average = 0.87) and low Mo/Al (0.93 and 0.77; average = 0.83), can be used to
374 define the oxic baseline for these parameters. By contrast, the tuffaceous strata in the lower
375 part of WHK 1 would have a different baseline value, due to dilution by tuffaceous materials
376 with a very different chemical composition, as well as the potential for masking of
377 enrichments due to rapid sedimentation (see above). Minimum U/Al and Mo/Al ratios are
378 observed in Beds 1 and 6, respectively. We use these beds to define oxic baseline ratios of
379 0.84 for U/Al, and 0.30 for Mo/Al. Although the elemental composition of volcanic ash may
380 depend on multiple factors (including magma source and crystallization differentiation), U is
381 commonly enriched in tuff, whereas Mo is relatively depleted (Zielinski, 1983; Pichavant et
382 al., 1987). For example, the Macusani tuff in southeast Peru has a U/Al ratio of at least 0.79,
383 and a Mo/Al ratio of 0.03 (Pichavant et al., 1987). Furthermore, Bennet and Canfield (2020)
384 show that oxic marine sediments have U/Al ratios in the range of 0.2-1.0, while Mo/Al ratios
385 are in the range of 0.08-0.4. Thus, our defined oxic baseline for U/Al is comparable to these
386 values, whereas our Mo/Al oxic baseline value is only slightly higher than these values. In
387 addition, our approach to defining local redox thresholds provides a high level of consistency
388 when interpreting the different redox proxies we apply. Thus, taken together, these
389 observations provide strong support that our approach is robust.

390

391 5.2.3. Redox evaluation of the Waiheke 1 section

392 The lower part of WHK 1 (Beds 1-15) shows slightly elevated $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}}$ and U/Al
393 ratios, both of which show an overall progressive increase through the Upper Permian strata
394 (Fig. 2). This consistent behaviour between two independent proxies suggests a progressive
395 decline of bottom-water oxygen levels through this interval. Low $\text{Fe}^*_{\text{pyrite}}/\text{Fe}_{\text{HR}}$ and Mo/Al

396 ratios indicate that the water column did not become euxinic at this time. However, during
397 deposition of the thin black claystone bed at the EPME horizon (Bed 17), enrichments in
398 Fe_{HR}/Fe_{total} and U/Al, combined with high Fe^*_{pyrite}/Fe_{HR} and Mo/Al (Fig. 2), strongly support
399 the onset of euxinia. This redox state was dominant until deposition of the earliest Triassic
400 bedded cherts, which contain the first occurrence of Triassic conodont fossils (Bed 29),
401 although two intervals of low Fe^*_{pyrite}/Fe_{HR} and Mo/Al in Beds 19 and 28, combined with
402 elevated Fe_{HR}/Fe_{total} and U/Al, suggest intermittent development of ferruginous conditions
403 (Fig. 2).

404 Above these strata, up until around Bed 44, Fe_{HR}/Fe_{total} and U/Al ratios remain high
405 (although U/Al ratios are somewhat lower in this interval, relative to the underlying euxinic
406 sediments; Fig. 2), suggesting persistent anoxia (as discussed above, the low Fe_{total} data
407 shown by open circles are not considered reliable). The Fe^*_{pyrite}/Fe_{HR} ratios are elevated,
408 suggesting water column euxinia, while Mo/Al ratios are also slightly elevated, but at a lower
409 level than in the underlying euxinic strata. On a local scale, a reduction in the extent of Mo
410 drawdown could simply be a consequence of less intense euxinia in the water column (Nägler
411 et al., 2011). However, since U also shows a reduction in the extent of drawdown, and since
412 U responds to anoxia rather than specifically euxinia, less intense euxinia does not appear to
413 provide a valid interpretation for the muted trace metal enrichments we observe.
414 Alternatively, however, if euxinia was widespread, then the muted enrichments we observe
415 under euxinic conditions would more likely reflect a significant decrease in the seawater trace
416 metal inventory (e.g., Algeo, 2004; Takahashi et al., 2014; Goldberg et al., 2016), which we
417 consider in more detail below.

418 Above Bed 44, the limited available data show evidence for fluctuating oxic and anoxic
419 redox conditions, with the possible development of euxinic conditions in some horizons (e.g.,
420 Bed 71). When the redox state appears to have been euxinic (i.e., Bed 71), low trace metal
421 drawdown (Fig. 2) may indicate prolonged seawater trace metal limitation through the
422 earliest Triassic.

423

424 *5.3. Temporal relationship of end-Permian oceanic anoxia in the low latitude and south mid-* 425 *latitude Panthalassa*

426 To investigate the development of anoxia across the Panthalassic ocean, we have
427 compiled redox proxy records for deep-sea PTB sections across the Permian to Triassic
428 transition (Fig. 5). Available conodont-dated, Panthalassic deep-sea PTB sections carbon

429 isotope records and redox sensitive trace element data include Arrow Rocks in New Zealand
430 (Hori et al., 2007), and the Akkamori 2 (Takahashi et al., 2014) and Ubara sections (Algeo et
431 al., 2011) in Japan. In Figure 5, U/Al and Mo/Al ratios for WHK 1 and these additional PTB
432 sections are compared based on $\delta^{13}\text{C}$ chemostratigraphic correlation (Fig. 4).

433 WHK 1 and Arrow Rocks, which were deposited in the south mid-latitude of the
434 Panthalassa Ocean, might be expected to have similar redox characteristics. During the first
435 $\delta^{13}\text{C}$ decrease in the late Changhsingian, WHK 1 consists of pale green-coloured tuffaceous
436 sedimentary rocks. Arrow Rocks also consist of similar tuffaceous strata in the late
437 Changhsingian (Fig. 5). Given this lithologic similarity, we use the baseline values for U/Al
438 and Mo/Al defined for WHK 1 to guide interpretation of the Arrow Rocks redox data.
439 However, U concentrations were below detection (<1.5 ppm) in the lower part of the Arrow
440 Rocks succession (Hori et al., 2007), preventing a direct redox interpretation. Furthermore,
441 while Mo/Al ratios are relatively high, total sulphur concentrations are below detection (<1.8
442 ppm), and thus it is difficult to provide a robust redox interpretation for this part of the
443 succession. However, in the overlying Permian-Triassic transitional section at Arrow Rocks,
444 enrichment of U/Al is evident, alongside very high peaks in Mo/Al (up to 670). This
445 coincides with U and Mo enrichment in WHK 1 (Fig. 5), supporting the notion of widespread
446 development of euxinic water column conditions in the south mid-latitude Panthalassa, at
447 least from the EPME to the PTB.

448 Geochemical records from low latitude Panthalassa, represented by the Akkamori 2 and
449 Ubara sections, provide further insight into temporal and spatial redox dynamics. To evaluate
450 the redox records for these PTB sections, we again use the minimum measured U/Al
451 (Akkamori = 0.55, Ubara = 0.11) and Mo/Al (Akkamori = 1.13, Ubara = 1.01) ratios to
452 define oxic baseline values. Both sites show evidence for generally persistent anoxia
453 (elevated U/Al) throughout the entire section, while elevated Mo/Al ratios suggest that the
454 water column may have become euxinic prior to the first $\delta^{13}\text{C}$ decrease (Fig. 5). Furthermore,
455 while there was some fluctuation in the precise chemistry of the anoxic water column after
456 the EPME, euxinia was a prevalent feature immediately after the EPME at the Akkamori site,
457 and potentially also at the Ubara site.

458 Our interpretation of the redox dynamics at the Ubara site based on Al-normalized values
459 of U and Mo contrast with those of Algeo et al. (2011), who considered suboxic conditions to
460 have been prevalent. In their study, the elevated bulk concentrations of U and Mo at Ubara
461 site are interpreted as being due to the slower sedimentation rate of Permian-Triassic black
462 claystone relative to the Permian chert beds resulting in enrichment. There thus remains some

463 uncertainty in the precise redox state of the Ubara site. However, we note that our
464 interpretation is more consistent with the redox conditions that are much more compellingly
465 demonstrated by highly elevated enrichments in Mo/Al at the Akkamori site. Thus, taken
466 together, it appears likely that the ocean oxygenation state of the Akkamori and Ubara sites in
467 the low latitude Panthalassa deteriorated earlier, relative to the Waiheke site in the south mid-
468 latitude Panthalassa.

469

470 *5.4. Covariation between U and Mo, and Mo drawdown after the Permian-Triassic boundary*

471 The bulk concentrations of Mo and U in the PTB sections are recast as enrichment factors
472 (M_{OEF} and U_{EF}) and shown as a cross plot in Figure 6. Algeo and Tribovillard (2009) and
473 Tribovillard et al. (2011) provide a framework for utilizing M_{OEF} and U_{EF} as a means to
474 identify three enrichment patterns. Firstly, co-increases in M_{OEF} and U_{EF} may occur following
475 a deterioration in redox conditions from suboxic to euxinic in open marine settings (Fig. 6).
476 The second case occurs where the M_{OEF} is significantly elevated relative to the U_{EF} , which
477 may be a consequence of active Mo transportation from the water column to the sediment, for
478 example via Fe-Mn (oxyhydr)oxide shuttling. The third pattern occurs when smaller and
479 relatively constant M_{OEF} are combined with high U_{EF} . This scenario is considered to represent
480 the case where extensive Mo-depleted seawater arises because of a persistent sulphidic deep
481 water mass, as may occur in a relatively restricted setting.

482 In the PTB sections of the low latitude Panthalassa during the Late Permian
483 (Changhsingian), before the $\delta^{13}\text{C}$ decrease, M_{OEF} values are high relative to U_{EF} values (Fig.
484 6-1). This supports the redox observations based on U/Al and Mo/Al ratios (Fig. 5), and
485 suggests anoxic, non-sulphidic water column conditions across this interval, with elevated
486 Mo drawdown due to uptake and sequestration by Fe-Mn (oxyhydr)oxide minerals. At the
487 WHK 1 site, the M_{OEF} and U_{EF} data (Fig. 6-1) document the slower development of anoxia in
488 the south mid-latitude Panthalassa, with evidence for the very beginning of water column de-
489 oxygenation, consistent with the Fe speciation, U/Al and Mo/Al (Fig. 2).

490 Between the two step decreases of $\delta^{13}\text{C}$, highly elevated M_{OEF} , and to a lesser extent U_{EF} ,
491 are evident for the low latitude Panthalassa localities (Fig. 6-2). These values support the
492 continued deterioration of water column redox conditions, with sporadic development of
493 euxinic conditions (as documented in Figure 5) further enhancing trace metal drawdown
494 (particularly Mo). In contrast, at the WHK 1 site in the south mid-latitude Panthalassa, U_{EF}
495 and M_{OEF} values document a later, progressive increase (Fig. 6-2). Across this interval, the

496 gradual co-increase in U_{EF} and Mo_{EF} values at the WHK 1 site suggests a classic redox
497 transition from suboxic, through anoxic, to euxinic water column conditions (Fig. 6; Algeo et
498 al., 2009; Tribouillard et al., 2012). This is generally consistent with the trends evident in the
499 Fe speciation data (Fig. 2), supporting a progressive deterioration in the redox state of the
500 water column and enhanced drawdown of U and Mo into the sediments.

501 During the interval from the EPME to the PTB, high U_{EF} and Mo_{EF} values are recorded in
502 both the south middle latitude and low latitude Panthalassa (Fig. 6-3). However, data from
503 the WHK 1 site plot closer to the typical trend for euxinic settings, which supports the
504 suggestion from trends in U/Al and Mo/Al (Figs. 2 and 5), that while euxinia was likely
505 widespread, this redox state may have become more prevalent and persistent in south middle
506 latitude regions, relative to the low latitude Panthalassa.

507 After the peaks across the EPME-PTB interval, U_{EF} and Mo_{EF} shift to generally lower
508 values during the early Triassic (Fig. 6-4). Indeed, enrichment patterns do not return to the
509 open-marine redox evolution trend. Instead, while the data show a wide range, Mo_{EF} values
510 in particular are generally lower than during the EPME-PTB interval. Thus, we suggest that
511 the development of widespread anoxia, with the common occurrence of euxinia, during the
512 EPME-PTB interval subsequently resulted in limited Mo sequestration in the sediments due
513 to a significant decrease in the seawater Mo inventory caused by intense Mo-bearing sulphide
514 deposition (e.g., Algeo, 2004; Takahashi et al., 2014; Goldberg et al., 2016). Although euxinic
515 water condition enhances uptake of both Mo and U into the sediment, relatively stronger
516 accumulation of Mo results in significantly decreased residence time of Mo in the seawater
517 and depletion of Mo relative to U. Alternatively, such a signal could be due to lower sulphide
518 availability in the water column, which might result from seawater sulphate drawdown after
519 EPME anoxia (Song et al., 2014; Schobben et al., 2017). However, the fact that several
520 intervals have Fe_{HR}/Fe_{total} and Fe^*_{pyrite}/Fe_{HR} ratios suggestive of euxinia, implies that Mo
521 depletion from the water column, rather than low sulphate availability, is a more likely
522 explanation for the relatively subdued Mo enrichments evident in the earliest Triassic.

523

524 5.5. Possible drivers of Panthalassic anoxia

525 This study indicates that the pelagic Panthalassa became anoxic, and was often euxinic, in
526 south middle latitude and low latitude regions during the EPME. However, the development
527 of anoxia occurred earlier at low latitudes. The progressive development of ocean anoxia
528 recorded by our data is in agreement with a decreasing trend in $\delta^{238}U_{carbonate}$ recorded in

529 multiple carbonate PTB sections (Fig. 7; Zhang et al., 2020). The $\delta^{238}\text{U}_{\text{carbonate}}$ signal begins to
530 decrease during the first $\delta^{13}\text{C}$ decrease of the Changhsingian, which continues across the
531 EPME, resulting in a negative trough in the earliest Triassic. This $\delta^{238}\text{U}_{\text{carbonate}}$ variability is
532 consistent with drawdown of U into Panthalassic anoxic deep-sea sediments through the Late
533 Permian and Early Triassic.

534 In Figure 7, data that can promote marine anoxia are compiled for the EPME. The
535 Siberian Traps were likely the main trigger for environmental change across the Permo-
536 Triassic boundary (e.g., Wignall., 2001). A mercury (Hg) anomaly, taken as an indicator of
537 enhanced volcanic activity, has been detected in late Changhsingian horizons in the Akkamori
538 2 and Ubara sections in the low latitude Panthalassa (Fig. 7; Shen et al., 2019), and at WHK 1
539 (Grasby et al., 2021), demonstrating a coincidence of major volcanism with the onset of
540 euxinia. Identified Hg anomalies appear below the EPME horizon at the low latitude sites,
541 but just above the EPME horizon at the south mid-latitude WHK 1 site (Fig. 7). One possible
542 reason for this discrepancy is the south mid-latitude location was too far from the Siberian
543 Traps to receive enough Hg before the EPME. The low sampling resolution across the WHK
544 1 section could be another possibility for this discrepancy. Grasby et al. (2021) did not
545 measure Hg in the thin black claystone bed at the EPME (Bed 17 of this study) and adjacent
546 layers.

547 The development of ocean anoxia prior to the EPME and persistence into the earliest
548 Triassic coincides with extreme climate warming (Fig. 7; Joachimski et al., 2012, 2019). This
549 warming has been linked to massive release of CO_2 from heated sediments and upper crusts
550 by Siberian Traps sills (Svensen et al., 2009). Subsequently, seawater warming would
551 contribute to marine deoxygenation in these oceanic regions through the decreasing solubility
552 of dissolved O_2 . Furthermore, thermohaline circulation driven by subduction of warm saline
553 surface water formed in mid and low latitudes would potentially contribute to the
554 development of a stratified ocean structure with anoxic deep waters (Kidder and Worsley,
555 2004).

556 The development of anoxia during the EPME coincides the cessation of radiolarian chert
557 deposition (Fig. 7). Our $\delta^{13}\text{C}$ correlation of PTB sections reveals that this phenomenon
558 occurred in the south mid-latitude Panthalassa around ten kiloyears later than in the low
559 latitude Panthalassa (Fig. 7). This may be due to a decreased biogenic silica burial flux
560 (declining radiolarian population and/or degraded preservation of siliceous tests) and dilution
561 of biogenic silica by increased clastic material (Takahashi et al., 2009; Muto et al., 2018,
562 2020). The environmental conditions during the EPME, including ocean anoxia and seawater

563 warming may have decreased the siliceous faunal population and increased sedimentation of
564 clay-rich material relative to biotic silica. If warming is a major factor limiting deposition of
565 biotic silica, then the different timing for the disappearance of radiolarian chert in the low
566 latitude and south mid-latitude ocean is noteworthy. For example, the higher latitude ocean
567 would warm more slowly, thus resulting in the observed delayed onset of both ocean anoxia
568 and the decrease in biotic silica deposition. The subsequent return to lower temperatures
569 favourable for the radiolarian chert deposition would also arrive earlier in mid-latitude
570 regions.

571

572 6. Conclusions

573 This study provides new geochemical records across the end-Permian mass extinction
574 event in the south middle latitude pelagic Panthalassa. Fe-speciation and redox-sensitive
575 elements (U and Mo) show the progressive development of ocean anoxia, with euxinia
576 occurring during the EPME. Comparison with previously reported paleo redox records from
577 the low latitude Panthalassa indicates that anoxia developed earlier in low latitude regions.
578 The pelagic Panthalassa PTB sections show Mo drawdown in the earliest Triassic, after the
579 EPME likely attributable to high Mo burial flux into anoxic marine sediments. The Late
580 Permian anoxia that occurred in the low latitude Panthalassa, and the progressive
581 development of anoxia during the run-up to the EPME in the south middle latitude
582 Panthalassa, coincides with Siberian Traps volcanic activity and de-vegetation trends on land.
583 Furthermore, the persistent anoxia that developed in both latitudinal regions of Panthalassa
584 coincide with severe seawater warming.

585 The results of this study shed light on spatiotemporal changes in open ocean redox
586 conditions across the Permian-Triassic boundary and show that the redox history of this
587 super-ocean varied according to latitude. Conditions for radiolarians were persistent and
588 harsh in equatorial latitudes relative to the mid-latitude site discussed here, indicating that
589 high temperatures were a major factor controlling their absence. Environmental records for
590 mid-high latitudes and pelagic regions of the ocean remain highly limited, and this should be
591 a prime target for further investigation.

592

593 Acknowledgements

594 This research was supported by the Japan Society for the Promotion of Science
595 (JSPS#24740340). The carbon isotope analyses were performed under the cooperative
596 research program (No. 08A026 and 08B023) of the Center for Advanced Marine Core
597 Research (CMCR), Kochi University, Japan. Thomas Algeo and Kunio Kaiho kindly
598 provided published data for the Ubara section. Andrew Hobson, Stephen Reid, Akihiro
599 Kobayashi and Masato Tanaka are thanked for support with geochemical analyses. We thank
600 reviewers Testuji Onoue and Jun Shen, and editor Zhong-Qiang Chen, for their constructive
601 comments.

602

603 References

604 Algeo, T.J., 2004. Can marine anoxic events draw down the trace element inventory of
605 seawater? *Geology* 32, 1057–1060.

606 Algeo, T.J., Li, C., 2020. Redox classification and calibration of redox thresholds in
607 sedimentary systems. *Geochim. Cosmochim. Acta* 287, 8–26.

608 Algeo, T.J., Henderson, C.M., Tong, J., Feng, Q., Yin, H., Tyson, R. V., 2013. Plankton and
609 productivity during the Permian-Triassic boundary crisis: An analysis of organic carbon
610 fluxes. *Glob. Planet. Change* 105, 52–67.

611 Algeo, T.J., Kuwahara, K., Sano, H., Bates, S., Lyons, T.W., Elswick, E., Hinnov, L.,
612 Ellwood, B., Moser, J., Maynard, J.B., 2011. Spatial variation in sediment fluxes,
613 redox conditions, and productivity in the Permian–Triassic Panthalassic Ocean.
614 *Palaeogeogr. Palaeoclimatol. Palaeoecol.* 308, 65–83.

615 Algeo, T.J., Tribovillard, N., 2009. Environmental analysis of paleoceanographic systems
616 based on molybdenum–uranium covariation. *Chem. Geol.* 268, 211–225.

617 Anderson, R.F., 1987. Redox behavior of uranium in an anoxic marine basin. *Uranium*
618 3, 145–164.

- 619 Ando, Akifumi, Kodama, Kazuto, Kojima, S., 2001. Low-latitude and Southern Hemisphere
620 origin of Anisian (Triassic) bedded chert in the Inuyama area, Mino terrane, central
621 Japan. *J. Geophys. Res. Solid Earth* 106, 1973–1986.
- 622 Begg, J.G., Johnston, M.R., 2000. Geology of the Wellington area 1:250,000 geological map
623 10. Geological and Nuclear Sciences. Lower Hutt, New Zealand, 64 pp.
- 624 Bennett, W.W., Canfield, D.E., 2020. Redox-sensitive trace metals as paleoredox proxies: A
625 review and analysis of data from modern sediments. *Earth-Science Rev.* 204, 103175.
- 626 Bruland, K.W., 1983. Trace elements in seawater. In: Riley, J.P., Chester, R. (Eds.), *Chemical*
627 *Oceanography*, vol.8. 2nd ed. Academic Press, New York, pp.157–220.
- 628 Burgess, S.D., Bowring, S., Shen, S., 2014. High-precision timeline for Earth's most severe
629 extinction. *Proc. Natl. Acad. Sci. U. S. A.* 111, 3316–3321.
- 630 Canfield, D.E., Raiswell, R., Westrich, J., Reaves, C., Berner, R.A., 1986. The use of
631 chromium reduction in the analysis of reduced inorganic sulfur in sediments and shales.
632 *Chem. Geol.* 54, 149–155. Canfield, D.E., Lyons, T.W., Raiswell, R., 1996. A model for
633 iron deposition to euxinic Black Sea sediments. *Am. J. Sci.* 296, 818–834.
- 634 Cao, C., Love, G.D., Hays, L.E., Wang, W., Shen, S., Summons, R.E., 2009. Biogeochemical
635 evidence for euxinic oceans and ecological disturbance presaging the end-Permian mass
636 extinction event. *Earth Planet. Sci. Lett.* 281, 188–201.
- 637 Chen, J.H., Edwards, R.L., Wasserburg, G.J., 1986. $^{238}\text{U}/^{234}\text{U}$ and ^{232}Th in seawater. *Earth*
638 *Planet. Sci. Lett.* 80, 241–251.
- 639 Clarkson, M.O., Wood, R.A., Poulton, S.W., Richoz, S., Newton, R., Kasemann, S.A., 2016.
640 Dynamic anoxic-ferruginous conditions during the end-Permian mass extinction and
641 recovery. *Nat. Commun.* 7, 1–9.
- 642 Clarkson, M.O., Poulton, S.W., Guilbaud, R., Wood, R.A., 2014. Assessing the utility of
643 Fe/Al and Fe-speciation to record water column redox conditions in carbonate-rich
644 sediments. *Chem. Geol.* 382, 111–122.
- 645 Emerson, S. R., Huested, S.S., 1991. Ocean anoxia and the concentration of molybdenum and
646 vanadium in seawater. *Mar. Chem.* 34, 177–196.

- 647 Goldberg, T., Poulton, S.W., Wagner, T., Kolonic, S.F., Rehkämper, M., 2016. Molybdenum
648 drawdown during Cretaceous Oceanic Anoxic Event 2. *Earth Planet. Sci. Lett.* 440, 81–
649 91.
- 650 Grasby, S.E., Beauchamp, B., 2009. Latest Permian to Early Triassic basin-to-shelf anoxia in
651 the Sverdrup Basin, Arctic Canada. *Chem. Geol.* 264, 232–246.
- 652 Grasby, S.E., Wignall, P.B., Yin, R., Strachan L., Takahashi, S., 2021. Transient Permian-
653 Triassic euxinia in the southern Panthalassa deep ocean. *Geology* 49, 889–893.
- 654 Grice, K., Cao, C., Love, G.D., Böttcher, M.E., Twitchett, R.J., Grosjean, E., Summons, R.E.,
655 Turgeon, S.C., Dunning, W., Jin, Y., 2005. Photic zone euxinia during the Permian-
656 triassic superanoxic event. *Science* 307, 706–709.
- 657 Hays, L., Beatty, T., Henderson, C.M., Love, G.D., Summons, R.E., 2007. Evidence for
658 photic zone euxinia through the end-Permian mass extinction in the Panthalassic Ocean
659 (Peace River Basin, Western Canada). *Palaeoworld* 16, 39–50.
- 660 Helz, G.R., Bura-Nakić, E., Mikac, N., Ciglencečki, I., 2011. New model for molybdenum
661 behavior in euxinic waters. *Chem. Geol.* 284, 323–332.
- 662 Hori, R.S., Yamakita, S., Ikehara, M., Kodama, K., Aita, Y., Sakai, T., Takemura, A.,
663 Kamata, Y., Suzuki, N., Takahashi, S., Spörli, K.B., Grant-Mackie, J., 2011. Early
664 Triassic (Induan) Radiolaria and carbon-isotope ratios of a deep-sea sequence from
665 Waiheke Island, North Island, New Zealand. *Palaeoworld* 20 168-178.
- 666 Hori, R.S., Higuchi, Y., Fujiki, T., Maeda, T., Ikehara, M., 2007. Geochemistry of the
667 Oruatemanu Formation, Arrow Rocks, Northland, New Zealand. In: Spörli, K.B.,
668 Takemura, A., Hori, R.S. (Eds.), *The Oceanic Permian/Triassic Boundary Sequence at
669 ArrowRocks (Oruatemanu) Northland, New Zealand.* GNS Science Monograph 24, pp.
670 123–156.
- 671 Isozaki, Y., 1997. Permo-Triassic boundary superanoxia and stratified superocean: Records
672 from lost deep sea. *Science* 80. 276, 235–238.
- 673 Jin, Y.G., 2000. Pattern of Marine Mass Extinction Near the Permian-Triassic Boundary in
674 South China. *Science* (80). 289, 432–436.

- 675 Joachimski, M.M., Alekseev, A.S., Grigoryan, A., Gatovsky, Y.A., 2019. Siberian Trap
676 volcanism, global warming and the Permian-Triassic mass extinction: New insights
677 from Armenian Permian-Triassic sections 1–17.
- 678 Joachimski, M.M., Lai, X., Shen, S.-Z., Jiang, H., Luo, G., Chen, B., Chen, J., Sun, Y., 2012.
679 Climate warming in the latest Permian and the Permian-Triassic mass extinction.
680 *Geology* 40, 195–198.
- 681 Kaiho, K., Chen, Z.-Q., Sawada, K., 2009. Possible causes for a negative shift in the stable
682 carbon isotope ratio before, during and after the end-Permian mass extinction in
683 Meishan, South China. *Aust. J. Earth Sci.* 56, 799–808.
- 684 Kaiho, K., Oba, M., Fukuda, Y., Ito, K., Ariyoshi, S., Gorjan, P., Riu, Y., Takahashi, S.,
685 Chen, Z.-Q., Tong, J., Yamakita, S., 2012. Changes in depth-transect redox conditions
686 spanning the end-Permian mass extinction and their impact on the marine extinction:
687 Evidence from biomarkers and sulfur isotopes. *Glob. Planet. Change* 94–95, 20–32.
- 688 Kaiho, K., Saito, R., Ito, K., Miyaji, T., Biswas, R., Tian, L., Sano, H., Shi, Z., Takahashi, S.,
689 Tong, J., Liang, L., Oba, M., Nara, F.W., Tsuchiya, N., Chen, Z.-Q., 2016. Effects of
690 soil erosion and anoxic–euxinic ocean in the Permian–Triassic marine crisis. *Heliyon*
691 2, e00137.
- 692 Kidder, D., Worsley, T., 2004. Causes and consequences of extreme Permo-Triassic warming
693 to globally equable climate and relation to the Permo-Triassic extinction and recovery.
694 *Palaeogeogr. Palaeoclimatol. Palaeoecol.* 203, 207–237.
- 695 Kodama, K., Fukuoka, M., Aita, Y., Sakai, T., Hori, S.R., Takemura, A., Campbell, H.J.,
696 Hollis, C., Grant-Mackie, J.A., Spörl, K.B., 2007. Paleomagnetic Results from Arrow
697 Rocks in the framework of Paleomagnetism in Pre- Neogene rocks from New Zealand,
698 in: *The Oceanic Permian/Triassic Boundary Sequence at Arrow Rocks (Oruatemanu)*
699 Northland, New Zealand. GNS Science Monograph. pp. 177–196.
- 700 Lau, K. V., Maher, K., Altiner, D., Kelley, B.M., Kump, L.R., Lehrmann, D.J., Silva-Tamayo,
701 J.C., Weaver, K.L., Yu, M., Payne, J.L., 2016. Marine anoxia and delayed Earth
702 system recovery after the end-Permian extinction. *Proc. Natl. Acad. Sci. U. S. A.* 113,
703 2–7.

- 704 Lei, L.-D., Shen, J., Li, C., Algeo, T.J., Feng, Q.-L., Cheng, M., Jin, C.-S., Huang, J.-H.,
705 2017. Controls on regional marine redox evolution during Permian-Triassic transition
706 in South China. *Palaeogeogr. Palaeoclimatol. Palaeoecol.* 486, 17–32.
- 707 Matsuda, T., Isozaki, Y., 1991. Well-documented travel history of Mesozoic pelagic chert in
708 Japan: From remote ocean to subduction zone. *Tectonics* 10, 475–499.
- 709 Matthews, A., Azrieli-Tal, I., Benkovitz, A., Bar-Matthews, M., Vance, D., Poulton, S.W.,
710 Teutsch, N., Almogi-Labin, A., Archer, C., 2017. Anoxic development of sapropel S1 in
711 the Nile Fan inferred from redox sensitive proxies, Fe speciation, Fe and Mo isotopes.
712 *Chem. Geol.* 475, 24–39.
- 713 McManus, J., Berelson, W.M., Severmann, S., Poulson, R.L., Hammond, D.E.,
714 Klinkhammer, G.P., Holm, C., 2006. Molybdenum and uranium geochemistry in
715 continental margin sediments: Paleoproxy potential. *Geochim. Cosmochim. Acta* 70,
716 4643–4662.
- 717 Muto, S., Takahashi, S., Yamakita, S., Suzuki, Noritoshi, Suzuki, Nozomi, Aita, Y., 2018.
718 High sediment input and possible oceanic anoxia in the pelagic Panthalassa during the
719 latest Olenekian and early Anisian: Insights from a new deep-sea section in Ogama,
720 Tochigi, Japan. *Palaeogeogr. Palaeoclimatol. Palaeoecol.* 490, 687–707.
- 721 Muto, S., Takahashi, S., Yamakita, S., Onoue, T., 2020. Scarcity of chert in upper Lower
722 Triassic Panthalassic deep-sea successions of Japan records elevated clastic inputs rather
723 than depressed biogenic silica burial flux following the end-Permian extinction. *Glob.*
724 *Planet. Change* 195, 103330.
- 725 Nägler, T.F., Neubert, N., Böttcher, M.E., Dellwig, O., Schnetger, B., 2011. Molybdenum
726 isotope fractionation in pelagic euxinia: Evidence from the modern Black and Baltic
727 Seas. *Chem. Geol.* 289, 1–11.
- 728 Oda, H., Suzuki, H., 2000. Paleomagnetism of Triassic and Jurassic red bedded chert of the
729 Inuyama area, central Japan. *J. Geophys. Res. Solid Earth* 105, 25743–25767.
- 730 Onoue, T., Soda, K., Isozaki, Y., 2021. Development of Deep-Sea Anoxia in Panthalassa
731 During the Lopingian (Late Permian): Insights From Redox-Sensitive Elements and
732 Multivariate Analysis. *Front. Earth Sci.* 8, 1–22.

- 733 Pichavant, M., 1987. The Macusani glasses, SE Peru: evidence of chemical fractionation in
734 peraluminous magmas. *Magmat. Process. Physicochem. Princ.* 359–373.
- 735 Poulton, S.W., Canfield, D.E., 2005. Development of a sequential extraction procedure for
736 iron: Implications for iron partitioning in continentally derived particulates. *Chem. Geol.*
737 214, 209–221.
- 738 Poulton, S.W., Raiswell, R., 2002. The low-temperature geochemical cycle of iron: From
739 continental fluxes to marine sediment deposition. *Am. J. Sci.* 302, 774–805.
- 740 Poulton, S.W., Canfield, D.E., 2011. Ferruginous conditions: A dominant feature of the ocean
741 through Earth's history. *Elements* 7, 107–112.
- 742 Poulton, S.W., Fralick, P.W., Canfield, D.E., 2010. Spatial variability in oceanic redox
743 structure 1.8 billion years ago. *Nat. Geosci.* 3, 486–490. doi:10.1038/ngeo889
- 744 Poulton, S.W., 2021. *The Iron Speciation Paleoredox Proxy*, Cambridge University Press,
745 75p.
- 746 Raiswell, R., Canfield, D.E., 1998. Sources of iron for pyrite formation in marine sediments.
747 *Am. J. Sci.* 298, 219–245.
- 748 Sano, H., Wada, T., Naraoka, H., 2012. Late Permian to Early Triassic environmental
749 changes in the Panthalassic Ocean: Record from the seamount-associated deep-marine
750 siliceous rocks, central Japan. *Palaeogeogr. Palaeoclimatol. Palaeoecol.* 363–364, 1–
751 10.
- 752 Scott, C., Lyons, T.W., 2012. Contrasting molybdenum cycling and isotopic properties in
753 euxinic versus non-euxinic sediments and sedimentary rocks: Refining the paleoproxies.
754 *Chem. Geol.* 324–325, 19–27.
- 755 Schobben, M., Stebbins, A., Algeo, T.J., Strauss, H., Leda, L., Haas, J., Struck, U., Korn, D.,
756 Korte, C., 2017. Volatile earliest Triassic sulfur cycle: A consequence of persistent
757 low seawater sulfate concentrations and a high sulfur cycle turnover rate? *Palaeogeogr.*
758 *Palaeoclimatol. Palaeoecol.* 486, 74–85.
- 759 Schobben, M., Stebbins, A., Ghaderi, A., Strauss, H., Korn, D., Korte, C., 2015. Flourishing
760 ocean drives the end-Permian marine mass extinction. *Proc. Natl. Acad. Sci. U. S. A.*
761 112, 10298–10303.

- 762 Schobben, M., Foster, W.J., Sleveland, A.R.N., Zuchuat, V., Svensen, H.H., Planke, S.,
763 Bond, D.P.G., Marcellis, F., Newton, R.J., Wignall, P.B., Poulton, S.W., 2020. A nutrient
764 control on marine anoxia during the end-Permian mass extinction. *Nat. Geosci.* 13, 640–
765 646.
- 766 Schofield, J.C., 1994. Stratigraphy, facies, structure, and setting of the waiheke and manaia
767 hill groups, East Auckland. *New Zeal. J. Geol. Geophys.* 17, 807–838.
- 768 Shen, J., Chen, J., Algeo, T.J., Yuan, S., Feng, Q., Yu, J., Zhou, L., O’Connell, B.,
769 Planavsky, N.J., 2019. Evidence for a prolonged Permian–Triassic extinction interval
770 from global marine mercury records. *Nat. Commun.* 10, 1563.
- 771 Schofield, J.C. 1974. Stratigraphy, Facies, Structure, and setting of the Waiheke and Manaia
772 Hill Groups, East Auckland. *New Zealand Journal of Geology and Geophysics*, 17,
773 807-838
- 774 Schofield, J.C., 1979. Part sheets N38, N39, N42, & N43, Waiheke. Geological map of New
775 Zealand 1:63360. Department of Scientific and Industrial Research, New Zealand
776 Geological Survey, Wellington, 16p.
- 777 Scholz, F., 2018. Identifying oxygen minimum zone-type biogeochemical cycling in Earth
778 history using inorganic geochemical proxies. *Earth-Science Rev.* 184, 29–45.
- 779 Shen, J., Chen, J., Algeo, T.J., Yuan, S., Feng, Q., Yu, J., Zhou, L., O’Connell, B.,
780 Planavsky, N.J., 2019. Evidence for a prolonged Permian–Triassic extinction interval
781 from global marine mercury records. *Nat. Commun.* 10, 1563.
- 782 Song, H., Wignall, P.B., Tong, J., Yin, H., 2012. Two pulses of extinction during the Permian
783 – Triassic crisis. *Nat. Geosci.* 6, 1–5.
- 784 Song, Huyue, Tong, J., Algeo, T.J., Song, Haijun, Qiu, H., Zhu, Y., Tian, L., Bates, S.,
785 Lyons, T.W., Luo, G., Kump, L.R., 2014. Early Triassic seawater sulfate drawdown.
786 *Geochim. Cosmochim. Acta* 128, 95–113.
- 787 Stookey, L.L., 1970. Ferrozine – A new spectrophotometric reagent for iron. *Analytical*
788 *Chemistry*, 42, 779–781.
- 789 Spörli, K.B., 1978. Mesozoic tectonics, North Island, New Zealand. *Geological Society of*
790 *America Bulletin* 89, 415–425.

- 791 Spörli, K. B., Aita, Y., Hori, R.S. Takemura, A., 2007b. Results of multidisciplinary studies
792 of the Permian/Triassic ocean floor sequence (Waipapa Terrane) at Arrow Rocks,
793 Northland, New Zealand. GNS Science Monograph, 24, 219-229.
- 794 Spörli, K.B., Takemura, A., Hori, S.R., 2007a. The oceanic Permian/Triassic boundary se-
795 quence at Arrow Rocks (Oruatemanu) Northland, New Zealand. GNS Science
796 Monograph 24, 229p.
- 797 Svensen, H., Planke, S., Polozov, a, Schmidbauer, N., Corfu, F., Podladchikov, Y., Jamtveit,
798 B., 2009. Siberian gas venting and the end-Permian environmental crisis. *Earth Planet.*
799 *Sci. Lett.* 277, 490–500.
- 800 Takahashi, S., Kaiho, K., Hori, S.R., Gorjan, P., Watanabe, T., Yamakita, S., Aita, Y.,
801 Takemura, A., Spörli, K.B., Kakegawa, T., Oba, M., 2013. Sulfur isotope profiles in the
802 pelagic Panthalassic deep sea during the Permian–Triassic transition. *Glob. Planet.*
803 *Change.* 105, 68-78.
- 804 Takahashi, S., Kaiho, K., Oba, M., Kakegawa, T., 2010. A smooth negative shift of organic
805 carbon isotope ratios at an end-Permian mass extinction horizon in central pelagic
806 Panthalassa. *Palaeogeogr. Palaeoclimatol. Palaeoecol.* 292, 532–539.
- 807 Takahashi, S., Yamakita, S., Suzuki, N., Kaiho, K., Ehiro, M., 2009. High organic carbon
808 content and a decrease in radiolarians at the end of the Permian in a newly discovered
809 continuous pelagic section: A coincidence? *Palaeogeogr. Palaeoclimatol. Palaeoecol.*
810 271, 1–12.
- 811 Takahashi, S., Yamasaki, S., Ogawa, Y., Kimura, K., Kaiho, K., Yoshida, T., Tsuchiya, N.,
812 2014. Bioessential element-depleted ocean following the euxinic maximum of the end-
813 Permian mass extinction. *Earth Planet. Sci. Lett.* 393, 94–104.
- 814 Taylor, S.R., McLennan, S.M., 1985. *The Continental Crust: Its Composition And Evolution*,
815 Blackwell Scientific Pub. 312p.
- 816 Tribouvillard, N., Algeo, T.J., Baudin, F., Riboulleau, A., 2012. Analysis of marine
817 environmental conditions based on molybdenum-uranium covariation-Applications to
818 Mesozoic paleoceanography. *Chem. Geol.* 324–325, 46–58.

- 819 Tribovillard, N., Algeo, T.J., Lyons, T.W., Riboulleau, A., 2006. Trace metals as paleoredox
820 and paleoproductivity proxies: An update. *Chem. Geol.* 232, 12–32.
- 821 Wang, K., Geldsetzer, H.H.J., Krouse, H.R., 1994. Permian-Triassic extinction: Organic $\delta^{13}\text{C}$
822 evidence from British Columbia, Canada. *Geology* 22, 580–584.
- 823 Wignall, P.B., 2001. Large igneous provinces and mass extinctions. *Earth-Science Reviews*
824 53, 1–33.
- 825 Wignall, P.B., Twitchett, R.J., 1996. Oceanic Anoxia and the End Permian Mass Extinction.
826 *Science* (80). 272, 1155–1158.
- 827 Wignall, P.B., Twitchett, R.J., 2002. Extent, duration, and nature of the Permian-Triassic
828 super anoxic event. *Spec. Pap. Geol. Soc. Am.* 356, 395–413.
- 829 Wignall, P.B., Bond, D.P.G., Kuwahara, K., Kakuwa, Y., Newton, R.J., Poulton, S.W., 2010.
830 An 80 million year oceanic redox history from Permian to Jurassic pelagic sediments of
831 the Mino-Tamba terrane, SW Japan, and the origin of four mass extinctions. *Glob.*
832 *Planet. Change* 71, 109–123.
- 833 Xiang, L., Zhang, H., Schoepfer, S.D., Cao, C., Zheng, Q., Yuan, D., Cai, Y., Shen, S., 2020.
834 Oceanic redox evolution around the end-Permian mass extinction at Meishan, South
835 China. *Palaeogeogr. Palaeoclimatol. Palaeoecol.* 544, 109626.
- 836 Yin, H., Zhang, K., Tong, J., Yang, Z., Wu, S., 2001. The global stratotype section and point
837 (GSSP) of the Permian-Triassic boundary. *Episodes* 24, 102–114.
- 838 Zhang, F., Algeo, T.J., Romaniello, S.J., Cui, Y., Zhao, L., Chen, Z.-Q., Anbar, A.D., 2018a.
839 Congruent Permian-Triassic $\delta^{238}\text{U}$ records at Panthalassic and Tethyan sites:
840 Confirmation of global-oceanic anoxia and validation of the U-isotope paleoredox
841 proxy. *Geology* 46, 327–330.
- 842 Zhang, F., Romaniello, S.J., Algeo, T.J., Lau, K. V., Clapham, M.E., Richoz, S., Herrmann,
843 A.D., Smith, H., Horacek, M., Anbar, A.D., 2018b. Multiple episodes of extensive
844 marine anoxia linked to global warming and continental weathering following the
845 latest Permian mass extinction. *Sci. Adv.* 4, e1602921.
- 846 Zhang, F., Shen, S., Cui, Y., Lenton, T.M., 2020. Two distinct episodes of marine anoxia
847 during the Permian-Triassic crisis evidenced by uranium isotopes in marine
848 dolostones. *Geochim. Cosmochim. Acta.* 287, 165-179.

849 Zielinski, R.A., 1983. Tuffaceous sediments as source rocks for uranium: A case study of the
850 White River Formation, Wyoming. *J. Geochemical Explor.* 18, 285–306.

851

852

853

854 Figure captions

855

856 Figure 1. Location map of the studied Permian-Triassic section (Waiheke 1 section, New
857 Zealand). (A) Paleo-geographical map of the Permian-Triassic showing the depositional
858 position of pelagic deep-sea Permian-Triassic boundary sections. Labeled numbers (1-7)
859 indicate the following localities: (1) pelagic sections from New Zealand accretionary
860 complexes, such as the Waiheke section (Spörli et al., 2007); (2) pelagic sections from
861 Japanese accretionary complexes; (3) shallow marine sections such as Meishan, South China;
862 (4) sections in Sverdrup Basin, arctic Canada (Grasby and Beauchamp, 2011); (5) sections in
863 Peace River Basin, western Canada (Hays et al., 2007). (B) Locality of the Waiheke-1 section
864 (WHK1) in the North Island of New Zealand. Distribution of the Triassic-Jurassic
865 accretionary complex is based on Black (1994) and Begg and Johnson (2000). Locality of
866 Arrow Rocks is also shown.

867 Figure 2. Lithologic column of the Waiheke 1 section, including organic carbon isotopes
868 ($\delta^{13}\text{C}_{\text{org}}$), iron speciation pools (Fe_{pools}), highly reactive iron normalised to total iron
869 ($\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}}$), corrected pyrite iron as a function of highly reactive iron ($\text{Fe}^*_{\text{pyrite}}/\text{Fe}_{\text{HR}}$), U,
870 U/Al and Mo/Al. The error ranges on the organic carbon isotope values ($\delta^{13}\text{C}_{\text{org}}$) are based on
871 repetitions of carbon isotope measurements (Table S1). Open circles in the plots of Fe_{pools} ,
872 $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}}$ and $\text{Fe}^*_{\text{pyrite}}/\text{Fe}_{\text{HR}}$ mean the samples with low Fe content (less than 0.5 wt %
873 Fe_{total}). We did not use these samples for paleoenvironmental reconstruction. Local
874 background values of U/Al and Mo/Al for tuffaceous strata (Beds 1-18) and cherty strata
875 (Beds 19-88) are shown as vertical dashed lines with “baseline” notation. The values are
876 based on the minimum values in the tuffaceous strata and the values in the cherty strata with
877 oxic $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}}$ signals.

878 Figure 3. Thin section photographs from the Waiheke 1 section. (A) tuffaceous siliceous
879 claystone of Bed 8; (B) black claystone of Bed 22; (C) bedded chert of Bed 50; (D) bedded
880 chert from around Bed 60.

881

882 Figure 4. Stratigraphic correlation of the Permian-Triassic boundary sections of the Meishan
883 section, Waiheke 1 section, Arrow Rocks section, Akkamori 2 section and Ubara section.
884 Compiled data of the Meishan section are based on Jin et al. (2000), Cao et al. (2002), Kaiho
885 et al. (2009) and Burgess et al. (2014). Data for Arrow Rocks, Akkamori and Ubara sections
886 are modified from Hori et al. (2007), Takahashi et al. (2010) and Kaiho et al. (2012),
887 respectively. The horizontal bars indicate correlative stratigraphic zones around the Permian-
888 Triassic boundary. Green: the first decrease in carbon isotope ratio ($\delta^{13}\text{C}$) in the latest
889 Changhsingian, Blue: the second decrease in $\delta^{13}\text{C}$ in the latest Changhsingian coinciding with
890 the EPME, Pink: the interval from the $\delta^{13}\text{C}$ minimum during the EPME to the $\delta^{13}\text{C}$ increase
891 toward the PTB.

892

893 Figure 5. Vertical plots of U/Al and (Mo/Al) for the Waiheke 1 and other pelagic deep-sea
894 sections. The data for Arrow Rocks, Akkamori 2 and Ubara are modified from Hori et al.
895 (2007), Takahashi et al. (2014) and Algeo et al. (2011), respectively. The local background
896 value (baseline) for U/Al and Mo/Al are based on the WHK 1 baseline values for Arrow
897 Rocks, and are based on the minimum values for the Akkamori and Ubara sections.

898

899 Figure 6. Cross plots of molybdenum (Mo) and uranium (U) enrichment factors (Mo_{EF} and
900 U_{EF} , respectively) for Panthalassic PTB sections from Waiheke 1, Arrow Rocks, Akkamori
901 and Ubara. The data for Arrow Rocks, Akkamori 2 and Ubara are modified from Hori et al.
902 (2007), Takahashi et al. (2014) and Algeo et al. (2011), respectively. The data are plotted in
903 four different panels representing the time intervals based on the carbon isotope correlation
904 demonstrated in Figures 4 and 5. 1: Late Permian (Changhsingian) before the $\delta^{13}\text{C}$ decrease;
905 2: between the first and second steps decreases of $\delta^{13}\text{C}$ in the latest Permian; 3: the interval
906 from the $\delta^{13}\text{C}$ minimum at the EPME to the $\delta^{13}\text{C}$ increase toward the PTB; 4: Earliest
907 Triassic (Induan) after the PTB. Superimposed grey-coloured areas representing modern
908 oceanic sediments are from Algeo and Tribovillard (2009) and Tribovillard et al. (2012). The
909 U/Mosw is the weight ratio in modern seawater (SW), which ranges from 3.0 to 3.1 (modern
910 Pacific and Atlantic, respectively; Anderson, 1987; Bruland, 1983; Chen et al., 1986;
911 Emerson and Husted, 1991).

912

913 Figure 7. Compilation of oceanic environmental records from Panthalassa, seawater
914 temperature for the paleo-Tethys reconstructed by oxygen isotope ratios in conodont apatite
915 ($\delta^{18}\text{O}_{\text{apatite}}$; Joachimski et al., 2009, 2019), global anoxic water extent referred from uranium
916 isotopes in carbonate ($\delta^{238}\text{U}_{\text{carbonate}}$; Zhang et al., 2020). Time scale through Permian and
917 Triassic is from Burgess et al. (2014). Information on oceanic environmental records for
918 Panthalassa is based on litho- and chemo-stratigraphy by this study, Hori et al. (2007, 2011),
919 Algeo et al., (2011), Takahashi et al. (2014), Kaiho et al. (2016), Shen et al., (2019) and
920 Grasby et al., (2021). Thin lines indicate the range of each Permian-Triassic boundary
921 section. The bars coloured black, orange, purple and grey refer to Hg anomaly ($\text{Hg}/\text{TOC} >$
922 100), radiolarian chert occurrence, geochemical signals of euxinia (high $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}}$,
923 $\text{Fe}_{\text{pyrite}}/\text{Fe}_{\text{HR}}$ and/or Mo/Al and U/Al), and seawater Mo drawdown (lower Mo/Al and Mo_{EF}
924 despite high $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}}$ and $\text{Fe}_{\text{pyrite}}/\text{Fe}_{\text{HR}}$ and/or U/Al and U_{EF}), respectively.
925