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52 **Abstract:**

53 The lacustrine Dameigou organic-rich shale of the northern Qaidam Basin is a
54 particularly promising play for recent unconventional oil and gas exploration in China.
55 Deposition was associated with arid intervals during the otherwise humid Middle
56 Jurassic, but the depositional mechanism and organic matter enrichment processes are
57 poorly understood. This study integrates high-resolution organic and inorganic
58 geochemical proxies with detailed sedimentary observations for samples from the
59 Chaiye (CY1) borehole, located in the Yuqia Depression, to investigate the depositional
60 environment and development of the Dameigou shales. Four major lithofacies,
61 including sandy mudstones, oil shale, organic-rich mudstone, and argillaceous
62 mudstone, were identified through detailed microscopic observation and mineralogical
63 analyses. Geochemical paleoclimate proxies (CIA, Fe/Mn and Mg/Ca) indicate that
64 humidity fluctuations primarily drove the lithofacies variation. Other geochemical
65 proxies, including terrigenous supply (Al_2O_3 , Ti_2O and regular sterane), lake salinity
66 (Sr/Ba, S/TOC and Ga), redox conditions (V/Cr, U/Th and Pr/Ph), and primary
67 productivity (Ba_{xs} , P_{xs}) indicate that climatic variability exerted a first-order control on
68 the chemical evolution of the lake and the development of organic-rich shale in the
69 Yuqia area. During arid intervals, fresh water supply through precipitation and river

70 runoff weakened, leading to an increase in salinity and water column stratification. As
71 a result, surface water became brackish and oxygen-rich, promoting growth of algae,
72 but density stratification drove the bottom waters to anoxia, providing favorable
73 conditions for organic matter preservation. Recycling of nutrients such as phosphorus
74 under anoxic conditions further promoted eutrophication and high productivity in the
75 surface water, which ultimately promoted the precipitation of carbonate minerals.

76 **Keywords:**

77 Organic-rich shale; Paleoclimate; Redox history; Saline lake; Water column
78 stratification

79 **1. Introduction**

80 Lacustrine organic-rich shale has long been a significant source rock for
81 conventional petroleum, providing more than 20% of the conventional global oil
82 resource (Bohacs et al., 2000). Many of these deposits, including the Green River
83 Formation in America, the Brown Shale in southeastern Asia, and the Aptian-Albian
84 sequences in Brazil, are also significant sources of unconventional oil and gas
85 (Neumann et al., 2003; Katz and Lin, 2014; Burton et al., 2014; Rodriguez and Philp,
86 2015; Smith et al., 2015). Continental basins are widespread in China, providing
87 extensive lacustrine shales (Hao et al., 2013; Yang et al., 2019) that account for ~85%
88 of conventional reserves (Katz, 1990; Li et al., 1995; Katz et al., 1998). These lacustrine

89 organic-rich shales have become significant exploration targets for shale oil and gas in
90 China (Li et al., 2014; Ma et al., 2016; Hu et al., 2018; Song et al., 2019).

91 The formation of lacustrine organic-rich shales is particularly sensitive to
92 paleoclimatic conditions (Anderson and Dean, 1988; Wang et al., 2005; Macquaker et
93 al., 2007; Smith et al., 2008). Climate change affects the hydrological regime and alters
94 the relative significance of precipitation and evaporation (Carroll and Bohacs, 1999;
95 Jiang et al., 2007; Kent-Corson et al., 2010; Chamberlain et al., 2013). Under different
96 climatic conditions, lake water chemistry, primary productivity and terrigenous input
97 may all evolve, which further controls the nature of sedimentation and organic matter
98 enrichment in lacustrine sediments (Carroll and Bohacs, 2001; Doebbert et al., 2010;
99 Hao et al., 2011).

100 The Qaidam Basin is an important petroleum basin in northwest China (Fig. 1A).
101 Petroleum exploration started in the mid-1950s, and many oil and gas fields were found
102 in the following 60 years (Li et al., 2014). These petroleum resources mainly occur in
103 the seventh member of the Middle Jurassic Dameigou Formation (J_2d^7), which has a
104 high organic matter content and was deposited in a lacustrine setting (Yang et al., 2004;
105 Liu et al., 2008; Li et al., 2016; Qin et al., 2018). The Chaiye 1 (CY1) well, located in
106 the Yuqia Sag at an altitude of about 3050 m, was the first lacustrine shale gas parameter
107 well drilled by the China Geological Survey in 2013 (Fig. 1C). Shale adsorption
108 isotherms show that the J_2d^7 has a Langmuir adsorption capacity ranging from 0.55-
109 5.93 m^3/t , indicating a significant shale gas exploration potential (Wang et al., 2016;

110 Guo et al., 2018). The sequence stratigraphic framework of the Middle Jurassic
111 Dameigou Formation in Yuqia Sag is well-established (Li et al., 2014; Shao et al., 2015;
112 Shang et al., 2018; Qian et al., 2018), and organic geochemical analyses have indicated
113 large-scale paleoclimatic change during its deposition (Ritts et al., 1999; Wang et al.,
114 2015; Meng et al., 2018; Hu et al., 2019). However, a detailed understanding of climate
115 variability during deposition, particularly with regard to the changing hydrological
116 conditions, remains unclear. Furthermore, links between climate change and deposition
117 of the organic-rich oil shale are poorly understood.

118 This study presents high-resolution geochemical analyses, combined with a
119 detailed sedimentological description, of the lacustrine shales in the CY1 well of the
120 J₂d⁷ Formation in the Yuqia Sag. Our aims are to (1) reconstruct the paleoclimatic
121 evolution of the north Qaidam Basin during the late Middle Jurassic; (2) identify
122 variability in hydrological conditions (including paleosalinity and redox condition)
123 linked to regional climate change; and (3) develop a sedimentary model for organic-
124 rich shale development in the Dameigou Formation.

125 **2. Geological setting**

126 The Qaidam Basin is located in the northeastern Qinghai-Tibetan Plateau, NW
127 China (Fig. 1A), which is a Mesozoic and Cenozoic intracontinental basin that
128 developed on Precambrian crystalline basement (Ritts and Biffi, 2001; Wang et al.,
129 2006; Ren et al., 2017). It is bounded by the northeast Qilian Mountains, northwest

130 Altyn Mountains and the southwest Kunlun Mountains (Fig. 1B). The northern Qaidam
131 Basin, covering an area of about 3.4×10^4 km², is one of the most important areas for
132 petroleum resources in the basin (Shao et al., 2014). The Yuqia Sag is situated in the
133 Middle of the northern Qaidam Basin (Fig. 1C) and has significant energy potential due
134 to abundant coal and coalbed methane resources, as well as shale gas exploration
135 potential (Li et al., 2016; Qin et al., 2018). The studied Chaiye 1 (CY1) well (97° 2' E,
136 36° 55' N) is within the Yuqia Sag (Fig. 1C).

137 The terminal Paleozoic marked an end to a period of transgression in the northern
138 Qaidam Basin. Due to the Indosinian movement, the northern Qaidam Basin was in a
139 steady stage of uplift, resulting in no sediment deposition at this time (Li et al., 2016;
140 Shang et al., 2018; Fig. 1C). At the beginning of the Jurassic, the crust began to subside,
141 with sedimentation due to the effects of regional extension and stress relaxation (Wang
142 et al., 2006). The depocenters gradually moved to the southwestern and eastern parts of
143 the northern Qaidam Basin. Owing to the large-scale eastward migration of the Tarim
144 plate relative to the Qaidam microplate at the end of the Early Jurassic, the regional
145 extensional environment transformed into a compressive environment, and the southern
146 part of the northern Qaidam Basin began to uplift (Chen et al., 2012). The depocenter
147 at this time moved eastward to the Yuqia Sag (Fig. 1C).

148 Paleogene-Neogene reservoirs provide the main host for oil and gas in the northern
149 Qaidam Basin, and are mostly derived from Middle Jurassic coal-bearing mudstone
150 sequences of the Dameigou Formation (Li et al., 2014; Wang et al., 2015). The

151 Dameigou Formation is further divided into seven members based on lithology and
152 depositional environment. The first to third members of the Dameigou Formation
153 ($J_2d^1 \sim J_2d^3$) were formed during the Early Jurassic, and comprise oil shale and
154 carbonaceous mudstone in the first member, and organic-lean mudstones and
155 sandstones in the second and third members (Fig. 2). The fourth to seventh members of
156 the Dameigou Formation ($J_2d^4 \sim J_2d^7$) were formed during the Middle Jurassic. The J_2d^4
157 member is dominated by organic-lean coarse to fine-grained sandstones that deposited
158 in braided river deltas. The J_2d^5 member is dominated by carbonaceous mudstones and
159 coal seams that deposited in a meandering river delta with significant terrestrial higher
160 plant input. The J_2d^6 member comprises mudstones and interbedded sandstones that
161 deposited in a shallow lake and meandering river delta. The J_2d^7 member, which was
162 formed in a lacustrine setting, is composed of sandstones at the bottom, coal seams and
163 carbonaceous mudstones in the middle, and oil shale at the top. This study focuses on
164 the middle and upper part of the J_2d^7 member (Fig. 2). We note that some studies (Shao
165 et al., 2014, Meng et al., 2018) define the J_2d^6 and J_2d^7 members as the Shimengou
166 Formation in the Yuqia area, but here we use the Dameigou Formation terminology to
167 represent the seven members.

168 **3. Methods**

169 Core samples (83 in total) from the J_2d^7 member were obtained from the CY1 well
170 (Fig. 3), with a core depth ranging from 1900 to 2038 m. Samples were freshly cut after
171 removing weathered surfaces, crushed and powdered to $<74 \mu\text{m}$ in an agate mortar.

172 **3.1 TOC and Rock-Eval pyrolysis**

173 Samples were subjected to total organic carbon (TOC) content determination and
174 Rock-Eval pyrolysis. TOC content was determined, after pre-treatment with 10% HCl
175 at 60°C, using a LECO CS 400 carbon-sulfur analyzer at Beijing Research Institute of
176 Uranium Geology (BRIUG), Beijing, China. The analyzing process followed the
177 Chinese Oil and Gas Industry Standard (G/T) 19145-2003 and the analytical precision
178 was <0.5%. The same sample split was subjected to Rock-Eval pyrolysis using a Rock-
179 Eval II instrument. S1 represents hydrocarbon concentration released after heating to
180 300°C. S2 represents the hydrocarbon concentration generated in samples heated from
181 300-600°C. The hydrogen index (HI) was calculated according to $100 \times S2/TOC$ (Behar
182 et al., 2001).

183 **3.2 X-ray diffraction**

184 The whole rock and clay mineral composition of 77 samples was determined on a
185 Panalytical X'PertPRO MPD X-ray diffractometer (XRD) at the Beijing Research
186 Institute of Uranium Geology (BRIUG), Beijing, China. These shale samples were
187 analyzed at 40 kV and 40 mA with Cu K α radiation, measured at a scanning rate of
188 2°/min, with the testing angle ranging from 5° to 90°. The relative mineral percentages
189 were calculated using the area under the curve for the major peaks of each mineral, with
190 correction for Lorentz polarisation (Chalmers and Bustin, 2008).

191 **3.3 Major and trace elements**

192 Samples were measured for major and trace element concentrations at the State
193 Key Laboratory of Biogeology and Environmental Geology at the China University of
194 Geosciences (Wuhan). Major elements were analyzed via a Philips PW2404 X-ray
195 fluorescence spectrometer, using fused glass discs consisting of a mixture of the
196 powdered sample heated to a temperature of 1000°C with flux ($\text{Li}_2\text{B}_4\text{O}_7$) at a proportion
197 of 1:8, with a weight of 5.0 g. Precision and accuracy were better than 5% for all major
198 elements (reported as oxides).

199 Trace element concentrations were determined on a PerkinElmer Elan DCR-e
200 standard inductively coupled plasma-mass spectrometer (ICP-MS). The samples were
201 placed in an oven and dried at 105°C for 12 h. Samples were then accurately weighed
202 (50 ± 1 mg) and placed in a Teflon crucible. 1.5 mL of high purity nitric acid (HNO_3)
203 and 1.5 mL of high purity hydrofluoric acid (HF) were then added to the sample, and
204 the closed crucibles were heated for 48 h at 190° C. After cooling, the samples were
205 heated to dryness and then 1 mL HNO_3 was added and evaporated to dryness. Finally,
206 3ml 30% HNO_3 was added, and the sealed samples were heated for 48 h at 190°C.
207 Analytical precision for trace element concentrations was better than $\pm 5\%$.

208 **4. Results**

209 **4.1 Lithofacies description**

210 Four major lithofacies of the Dameigou Formation (J₂d⁷) mudstone in CY1 are
211 shown in Figs. 3 and 4, including the sandy mudstones, oil shale, organic-rich mudstone,
212 and argillaceous mudstone. An obvious variation in lithofacies assemblages is observed
213 (Fig. 3), with sandy mudstone mainly occurring at the top and bottom of the study
214 section, and this comprises different types of bedding and bioturbation (Fig. 3). The
215 sandy mudstone is mainly composed of coarse-grained detrital minerals, including
216 euhedral quartz and feldspar (Fig. 4A).

217 The thick-bedded organic-rich mudstone, occurring in the middle and lower
218 section, is the major type of shale lithofacies (Fig 3). Horizontal and small-scale cross
219 bedding is apparent. Under the microscope, this mudstone is characterized by high
220 organic matter content with abundant fine-grained feldspar, quartz and clay minerals
221 (Fig 4.B). The oil shale bed, which is ~8 m thick (Fig 3), occurs in the upper part of the
222 section. This unit has a notable growth of carbonate minerals, such as calcite and
223 dolomite. However, the detrital mineral contents, including quartz, feldspar and clay
224 minerals is relatively lower (Fig. 4C, D). Horizontal laminae are prevalent in the oil
225 shale, comprising fine-grained clastic layers such as organic-rich mudstones and
226 cryptocrystalline carbonate layers (Fig. 3). The cryptocrystalline carbonate layers have
227 distinct boundaries which are partly wavy at the micro-scale (Fig. 4D). Microscopic

228 fossils, including algal aggregates and plankton residue are also frequently observed,
229 and are preserved parallel to the bedding or stacked as lenticles (Fig. 4E, F, G). The
230 argillaceous mudstones are relatively concentrated in the upper part of the study section.
231 This facies is organic lean and mainly composed of clay minerals, with a small
232 percentage of quartz and feldspar (Fig. 4H).

233 **4.2 Mineralogical characteristics**

234 The XRD results are plotted in Figure 5 and tabulated in Table 1. The sandy
235 mudstone is characterized by a high proportion of silicate minerals (quartz + K-feldspar
236 + plagioclase), ranging from 20.9 to 75.8 wt.%, with an average of 36.2 wt.%; low
237 carbonate content (calcite + dolomite + aragonite + siderite), ranging from 0 to 31.6
238 wt.%, with an average of 6.0 wt.%; and a high amount of clay minerals, ranging from
239 38.0 to 74.8 wt.%, with an average of 57.8 wt.% (Fig. 5; Table 1). Quartz is the
240 dominant mineral in the silicates, ranging from 17.2 to 50.3 wt.% (avg. 32.7 wt.%). K-
241 feldspar and plagioclase comprise less than 3.0 wt.% (avg. 2.3 wt.% and 1.2 wt.%,
242 respectively). The average proportion of calcite and siderite is 1.1 wt.% and 4.9 wt.%,
243 respectively. Dolomite is rarely found in this sequence.

244 In the organic-rich mudstone, the amount of silicate minerals decreases to 30.6 wt.%
245 on average, with a range from 16.1 to 77.6 wt.%; the amount of carbonate increases to
246 8.1 wt.% on average, with a range of 0 to 47.3 wt.%; and the average amount of clay
247 minerals slightly increases to 61.3 wt.%, with a range of 14.7 to 76.6 wt.%. Compared

248 to the sandy-mudstone, the average amount of feldspar in the organic-rich mudstone
249 shows no apparent change. The average content of K-feldspar slightly decreases to 1.6
250 wt.%, while plagioclase slightly increases to 2.5 wt.%. The quartz content also
251 decreases, and ranges from 11.1 to 77.3 wt.%, with an average of 26.4 wt.%. Calcite
252 and dolomite are very minor in this lithofacies. The siderite ranges from 0 to 47.3 wt.%
253 (avg. 7.8 wt.%).

254 For the oil shale, the silicate fraction drops to 26.5 wt.% on average, with a range
255 of 8.8 to 30.2 wt.%; the clay mineral fraction decreases to 38.5 wt.% on average, with
256 a range of 10.0 to 47.8 wt.%; the amount of carbonate increases up to 35.0 wt.% on
257 average (Fig. 5; Table 1). Carbonate minerals, including calcite, dolomite and aragonite,
258 with average contents of 12.6 wt.%, 3.1 wt.% and 17.8 wt.%, respectively, dominate in
259 the oil shale. The mineral composition of the argillaceous mudstone is similar to the
260 organic-rich mudstone, with a medium silicate content (avg. 31.3 wt.%) and low
261 carbonate content (avg. 9.2 wt.%). Clay becomes a dominant mineral fraction, ranging
262 from 22.1 wt.% to 74.5 wt.%, with an average of 59.5 wt.%.

263 **4.3 Organic matter characteristics**

264 The results of TOC analyses and Rock-Eval pyrolysis are plotted in Fig. 6 and Fig.
265 7. We divide the study section into three units based on TOC content, including two
266 low-TOC units (unit 1 and unit 3), and a high-TOC unit (unit 2). The sandy mudstone
267 has the lowest TOC content, between 0.3 and 2.9 wt.% (avg. 0.8 wt.%). It also has

268 S1+S2 and HI indices between 0.2 and 6.3 mg/g (avg. 1.23 mg/g), and 82 and 471 mg/g
269 (avg. 191 mg/g), respectively, suggesting low organic matter abundance and primarily
270 gas generation potential. The organic-rich mudstones are characterized by TOC
271 contents between 0.65 and 6.5 wt.% (avg. 3.6 wt.%), S1+S2 between 1.0 and 13.9 mg/g
272 (avg. 7.4 mg/g), and HI between 146 and 508 mg/g (avg. 240 mg/g), indicating high
273 organic matter richness and good gas generation potential.

274 Oil shale samples collected from the upper part of unit 2 have the highest organic
275 matter contents and HI values. The TOC contents range between 3.9 and 6.7 wt.% (avg.
276 4.3 wt.%), and S1+S2 values are between 14.8 and 22.9 mg/g (avg. 18.4 mg/g). The HI
277 is between 398 and 426 mg/g (avg. 413 mg/g), indicating type II and II/III organic
278 matter, with partial oil generation potential. Most of the argillaceous mudstones of unit
279 3 have low organic matter abundance and HI values. The TOC contents range from 0.3
280 to 5.0 wt.% (avg. 1.6 wt.%) and S1+S2 ranges from 0.02 to 10.6 mg/g (avg. 3.39 mg/g).
281 From the bottom of unit 1 to the top of unit 3, the TOC content, hydrocarbon generation
282 potential (S1+S2), and hydrogen index (HI) of shale samples show a similar trend (Fig.
283 6A, B, C). The trends in TOC vary within different units, with no obvious change in
284 TOC in unit 1. However, an upwards increase is evident in unit 2, climbing from ~3 to
285 ~6 wt.%. In unit 3, the TOC content rapidly drops to ~0.5 wt.%. Both S1+S2 and HI
286 follow a similar pattern.

287 **4.4 Major element geochemistry**

288 Geochemical parameters are shown in Table 2. Major element analyses show that
289 CaO dominates in the oil shale samples, ranging from 17.6 wt.% to 47.9 wt.% (avg.
290 29.3 wt.%). However, CaO is less than 1 wt.% in the other three lithofacies (sandy
291 mudstone, organic-rich mudstone, argillaceous mudstone). SiO₂, Al₂O₃ and Fe₂O₃ are
292 major constituents in most of the samples, with average concentrations of 49.1, 20.1,
293 and 7.3 wt.%, respectively. The average concentrations of K₂O and MgO are 2.3 and
294 1.8 wt.%, respectively. Other elements, including MnO, Na₂O, TiO₂ and P₂O₅, have
295 average concentrations below 1 wt.%. Fe/Mn and Mg/Ca ratios remain relatively high
296 in unit 1 (Fig. 6D, E). Then, the Fe/Mn ratios decrease modestly in unit 2 and re-
297 climb in unit 3 (Fig. 6D), whereas Mg/Ca ratios remain high in the organic-rich shale, then
298 drop in the oil shale in unit 2 and remain low in unit 3 (Fig. 6E). The terrigenous
299 indicators, Al₂O₃ and Ti₂O, show similar variability, and remain relatively high through
300 most of the study section, with the exception of some very low contents, for example
301 in the oil shale (Fig. 6F, G).

302 Regional weathering conditions and paleoclimate may be evaluated by the
303 chemical index of alteration (CIA) weathering proxy (Nesbitt et al., 1982; Young and
304 Nesbitt, 1999; Cullers and Podkovyrov, 2002), which is calculated by $CIA =$
305 $Al_2O_3 / (Al_2O_3 + CaO^* + Na_2O + K_2O) \times 100$ (where CaO* represents CaO incorporated
306 into silicate minerals). Therefore, it is necessary to subtract the CaO incorporated into
307 carbonates (calcite, dolomite) and phosphates (assuming all the P₂O₅ is present as

308 apatite) (Fedó et al., 1995). As Ca is preferentially removed relative to Na during
309 chemical weathering (McLennan, 1993), the remaining number of moles is adopted if it
310 is less than that of Na₂O; otherwise, CaO* is assumed to be equivalent to Na₂O. The
311 CIA indicator shows low values in the middle of unit 1 and persistently high values in
312 the bottom to middle of unit 2 (Fig. 6H). Then CIA values show a slight upwards
313 decrease in the remaining organic-rich shale, followed by a marked decrease in the oil
314 shale. In unit 3, the CIA indicator rises rapidly to higher values.

315 **4.5 Trace element geochemistry**

316 We use a variety of geochemical indicators to reconstruct paleoenvironmental
317 change in the basin (Fig 7; Table 2). Sr/Ba ratios exhibit an extremely high peak in oil
318 shale samples, whereas very low ratios occur in other lithofacies (Fig. 7D). Ga contents
319 show similar trends (Fig. 7E), with high and relatively stable contents in unit 3, a slight
320 decline in organic-rich shale samples, a significant drop in the oil shale samples of unit
321 2, and a rapid increase in unit 3.

322 Elemental enrichment factors (EF) were calculated as:
323 $X_{EF} = (X/Al)_{\text{sample}} / (X/Al)_{\text{PAAS}}$. The V_{EF} , Cr_{EF} and U_{EF} ratios demonstrate the same pattern,
324 with low values in units 1 and 3, but much higher values in unit 2, particularly in oil
325 shale samples (Fig. 7 F, G, H). The concentrations of Ba and Si in “excess” of normal
326 terrigenous content (Ba_{xs} , P_{xs}) are considered to represent the biogenic fraction for these
327 elements, and are calculated by $X_{xs} = X_{\text{total}} - (Al \times [X/Al]_{\text{PAAS}})$, where X is the

328 concentration of the element of interest, and PAAS represents post-Archean average
329 shale (PAAS). Both Ba_{xs} and P_{xs} contents are very high in the oil shale samples.
330 However, in other samples, Ba_{xs} and P_{xs} are close to zero, although there are some
331 isolated peaks in the middle part of unit 2.

332 **5. Discussion**

333 **5.1 Paleoclimatic reconstruction**

334 Generally, CIA values of between 50 to 65 represent an arid climate with low
335 chemical weathering intensity, values of 65-85 indicate a temperate climate with a
336 moderate chemical weathering rate, and values > 85 reflect a hot and humid climate
337 with strong chemical weathering intensity (Nesbitt and Young, 1982). The CIA values
338 are generally high (Fig. 6H), with stable high values in the upper part of unit 1 and the
339 lower part of unit 2, suggesting a long-term interval of intense chemical weathering rate.
340 However, CIA values exhibit a slow decline through the uppermost organic-rich shale,
341 with low values in the oil shale at the top of unit 2, as well as in the middle of unit 1
342 (Fig. 6H). This shows that the paleoclimatic regime was periodically altered, with
343 transitions to more arid conditions.

344 Fe/Mn and Mg/Ca ratios may also be used as proxies to characterize paleoclimatic
345 conditions (Reheis, 1990), although Fe/Mn ratios may also be affected by water column
346 redox conditions. Manganese tends to remain high in sediments deposited under arid
347 conditions, but may be lower under relatively humid conditions. By contrast, with more

348 intense chemical weathering under humid conditions, Fe is retained in the weathered
349 sediment (e.g., Poulton and Raiswell, 2002). Therefore, high Fe/Mn ratios tend to
350 represent hot-humid environments, whereas low ratios commonly occur in a cold-dry
351 climate (Reheis, 1990). In terms of Mg/Ca, high ratios in alkaline lakes may indicate a
352 warm-humid climate (Lermen et al., 1995). Thus, high Fe/Mn and Mg/Ca ratios support
353 the suggestion, based on the CIA proxy, of a dominantly warm and humid climate
354 during deposition of the 7th member of the Dameigou Formation. However, low values
355 for both Fe/Mn and Mg/Ca support short-term and high-intensity arid climate
356 disturbance during deposition of the oil shale section in the upper part of unit 2 (Fig.
357 6D, E).

358 **5.2 Terrigenous supply**

359 **5.2.1 Clastic input**

360 The intensity and rate of clastic influx directly affects the mineral composition and
361 organic matter characteristics of lacustrine black shale (Johnson Ibach, 1982; Rimmer,
362 2004). A moderate rate of deposition is beneficial for the burial efficiency of organic
363 matter, but fast depositional rates dilute the organic matter (Canfield, 1994; Wang et
364 al., 2019). The aluminum (Al) and titanium (Ti) content of sediments may be used to
365 evaluate the relative clastic influx rate (Rimmer, 2004; Wang et al., 2020). Aluminium
366 is principally present in clays, feldspars and other aluminum silicate minerals, while
367 titanium is usually related to clays and heavy minerals (e.g., ilmenite and rutile).

368 In the study section, Al and Ti show a similar pattern: persistent high values
369 through the humid climate zones, with two short-term transitions to lower values in the
370 arid climate zones (Fig. 6F, G). This indicates that the nature of the clastic sediment
371 influx was controlled by paleoclimate. Unit 1 was fluvially-influenced, comprising
372 lacustrine-swamp and meandering fluvial facies (Fig. 3), and under the humid climatic
373 conditions, chemical weathering was intense. Clastic sediment was transported into the
374 sedimentary basin, forming the sandy mudstone and minor intervals of organic-rich
375 mudstone. However, when the climate became more arid, the size of the sedimentary
376 basin diminished, resulting in deposition of coarse-grained siltstone-sandstone, and
377 even conglomeratic sandstone in the middle of unit 1.

378 Units 2 and 3 developed in a lake delta and deep lake sedimentary system. Under
379 humid climatic conditions, high intensity runoff maintained the deep-water lacustrine
380 environment. The detrital material delivered to the basin by rivers mainly comprised
381 fine-grained minerals, such as feldspar and quartz. During arid intervals, due to the
382 simultaneous reduction of weathering and fluvial intensity, less detrital material
383 deposited in the deep-water lakes. Therefore, authigenic minerals such as calcite,
384 dolomite and aragonite dominated during deposition of the oil shale intervals.

385 **5.2.2 Organic matter sources**

386 The nature of the paleoenvironment controls the source of organic matter in
387 lacustrine sediments, and different types of organic matter directly determine the
388 quality of the source rocks (Tissot and Welte, 1984; Hedges et al., 1994; Zakir Hossain

389 et al., 2009). The source of lacustrine organic matter may be divided into intrabasinal
390 organic matter and extrabasinal organic matter, of which the intrabasinal organic matter
391 represents productivity via aquatic organisms in the lake itself, whereas extrabasinal
392 organic matter is mainly derived from terrestrial sources transported by rivers and wind.
393 The organic matter classification is defined by a cross-plot of HI vs. T_{\max} (Qin et al.,
394 2018), and the data fall into four groups based on lithology (Fig. 8A). Sandy mudstones
395 and organic-rich mudstones (including coal beds) mainly comprise types III or type II₂
396 kerogen, representing a dominant source from higher terrestrial plants. However, the
397 kerogen type changes to type II₁ for most organic-rich shale samples, and type I for oil
398 shale samples, which documents a primary source from within the basin.

399 The organic matter composition provides further information on the source of
400 organic matter. Different organic substances have different C₂₇, C₂₈, and C₂₉ sterol
401 contents (Peters et al., 2005), however, lower aquatic organisms are abundant in C₂₇
402 sterols and higher plants tend to be rich in C₂₉ sterols (Ghassal et al., 2018; Waples and
403 Machihara, 1990). From the sandy mudstone in unit 1 to the oil shale in unit 2, the type
404 of organic matter evolves: the significance of higher plants decreases while plankton
405 progressively increases through the section (Fig. 8B). During the humid climate period,
406 surface runoff strengthened, and larger organic matter particles such as higher plants,
407 were readily transported to lakes (or swamps) to deposit. By contrast, under arid
408 conditions, transportation via rivers was greatly weakened, thus decreasing the organic

409 matter input from higher plants, leading to an increased contribution from lower
410 plankton or algae.

411 **5.3 Water column chemistry**

412 **5.3.1 Paleosalinity**

413 Trace element proxies such as strontium/barium (Sr/Ba) and gallium (Ga)
414 concentration may be used to infer paleo-salinity in lacustrine basins (Pais and Jones,
415 1997; Wei and Algeo, 2019). The chemical properties of Sr and Ba are similar, but they
416 respond differently to increasing salinity (Lan et al., 1987; Zheng and Liu, 1999). In
417 terrestrial sediments, strontium is liberated more quickly than barium during
418 weathering, causing low Sr/Ba ratios (~1.4) in freshwaters (Yang et al., 2004a; Yang et
419 al., 2004b). By contrast, there are much higher Sr/Ba ratios (~39) in seawater because
420 of their different concentrations (Sr is 8.1 mg/l and Ba is 0.021 mg/l) (Krauskopf, 1956).
421 These fundamental differences make the Sr/Ba ratio a proxy for the salinity of an
422 ancient water body (Deng and Qian, 1993; Zheng and Liu, 1999; Wei et al., 2018; Wei
423 and Algeo, 2019). Generally, Sr/Ba ratios of <0.2, 0.2-0.5, and >0.5 are thresholds for
424 sediments deposited under freshwater, brackish and seawater conditions (Wei and
425 Algeo, 2019). Gallium is primarily derived from quartzose and feldspathic silicate rocks
426 and is often adsorbed onto clay minerals (Couch, 1971; Chen et al., 1997). This process
427 leads to higher Ga contents in freshwater sediments relative to marine sediments
428 (Bowen, 1982; Salminen et al., 2005).

429 The Sr/Ba ratios are extremely high (avg. 1.35) in oil shale samples and adjacent
430 organic-rich shale samples, but are very low in other samples (avg. 0.19) (Fig. 7D).
431 This indicates that during organic-rich shale deposition, the lake maintained a relatively
432 stable freshwater environment. By contrast, in the oil shale section, the salinity of the
433 lake increased sharply and then quickly desalinated and returned to the previous level.
434 Gallium concentrations show the same trend as Sr/Ba ratios and thus support this
435 inference (Fig. 7E). Thus, this lacustrine setting began to salinize in the middle part of
436 the organic-rich shale in unit 2, and then in the oil shale section, both the rate and
437 intensity of lake salinization increased sharply.

438 The ratio S/TOC is an additional, frequently used paleosalinity indicator (Berner
439 and Raiswell, 1984; Wei and Algeo, 2019). When the TOC content of samples is > 1
440 wt.%, the S/TOC thresholds are > 0.5 for seawater conditions, and < 0.5 for freshwater
441 conditions (Wei et al., 2018). In the study section, samples from the oil shale in unit 2
442 and the organic-rich shale in unit 1 have high S/TOC ratios, which supports salinization
443 during deposition of these lithologies (Fig. 7C). These combined results for Sr/Ba, Ga
444 and S/TOC suggest that salinity was mainly controlled by climate, since there is no
445 obvious evidence of transgression in the study area. Thus, under a humid climate, the
446 lake had sufficient water supply to maintain freshwater conditions, whereas during
447 deposition of the oil shale, the rapid development of arid conditions resulted in
448 salinization.

449 **5.3.2 Redox conditions**

450 Redox conditions in the water column can be evaluated by the degree of
451 enrichment of redox-sensitive elements such as V, Cr and U (Calvert and Pedersen,
452 1993; Dean et al., 1997; Tribovillard et al., 2006; Wang et al., 2019; Algeo and Li, 2020;
453 Algeo and Liu, 2020). Both V and Cr are soluble under oxic conditions, but may be
454 enriched under reducing conditions (Morford and Emerson, 1999). However, the
455 removal of V occurs coincident with the beginning of denitrification, whereas the
456 removal of Cr occurs later as conditions become more reducing towards the end of
457 denitrification (Pattan et al., 2005). These behaviours are widely used to evaluate
458 paleoredox conditions in the column (Tribovillard et al., 2006).

459 Uranium exists in the form of $\text{UO}_2(\text{CO}_3)_3^{4-}$ in oxic seawater, where it is highly
460 soluble. However, in reducing environments, soluble U(VI) is reduced to U(IV)
461 coincident with the reduction of Fe(III) to Fe(II), causing U enrichment in the sediments
462 (Chaillou et al., 2002; McManus et al., 2005). This removal process of U from the water
463 column to the sediments is considered to be accelerated by organometallic ligands in
464 humic acids (Zheng et al., 2002a,b; McManus et al., 2005). Hence, U abundance in
465 sediments is commonly used to indicate redox conditions and organic matter
466 preservation (Algeo and Maynard, 2004). Based on these systematics, all of the redox
467 proxies we employ (V_{EF} , Cr_{EF} and U_{EF}) suggest oxic conditions throughout deposition
468 of most of the 7th member of the Dameigou Formation (Fig. 7F, G, H). However, during
469 deposition of the oil shale, the water column was dominantly anoxic (Fig. 7).

470 Reactive organic matter produced in the photic zone is more enriched in hydrogen
471 than vascular land plant and thus more sensitive to redox conditions (Tissot and Welte,
472 1984). The HI can therefore also serve as an indicator of redox conditions. The HI is
473 higher in the oil shale (avg. 305 mg/g TOC) relative to the organic-rich mudstone,
474 argillaceous mudstone and sandy mudstone (avg. 257, 185 and 129 mg/g TOC,
475 respectively), indicating more reducing bottom water conditions during deposition of
476 the oil shale (Fig. 6). The pristane/phytane (Pr/Ph) ratio is also widely used as an index
477 to interpret redox conditions (Didyk et al., 1978; Hughes et al., 1995). Based on the
478 analyses of about 100 crude oil samples from major petroliferous basins in China, Mei
479 and Liu (1980) suggest that Pr/Ph ratios <0.8 indicate anoxic environments, Pr/Ph ratios
480 between 0.8 and 2.8 indicate suboxic conditions, and Pr/Ph ratios >2.8 indicate dysoxic
481 to oxic settings. The organic-rich mudstone and argillaceous mudstone have Pr/Ph
482 ratios between 2.0 to 2.2, suggesting suboxic bottom water conditions. By contrast,
483 Pr/Ph ratios are much lower in the oil shale at 0.8, supporting anoxic bottom water
484 conditions (Fig. 8D). A plot of Pr/nC₁₇ versus Pr/nC₁₈ also indicates that the oil shale
485 deposited in a more reducing environment relative to the other lithofacies (Fig. 8C).

486 Combining the evidence from organic and inorganic geochemical data thus
487 strongly suggests that the lake was anoxic at times of arid climate, and oxic under humid
488 climatic conditions. This may also partly relate to the tectonic regime, whereby the
489 Yuka Sag coincided with a tectonic subsidence stage during the middle Jurassic dry
490 climate period (Shao et al., 2014; Li et al., 2014). Hence, because the rate of tectonic

491 subsidence was higher than the rate of lake-level decline due to climate drying, the
492 system remained relatively deep-water for a considerable time, allowing deposition of
493 the oil shale sections.

494 **5.4 Paleoproductivity**

495 Excess Phosphorus (P_{xs}) and Barium (Ba_{xs}) are commonly used as proxies to
496 evaluate relative levels of productivity (Nameroff et al., 2002; Piper and Perkins, 2004;
497 Algeo and Rowe, 2012; Schoepfer et al., 2015; Wang et al., 2020). In the study section,
498 both Ba_{xs} and P_{xs} are high in the oil shale and adjacent organic-rich shale samples, but
499 are low throughout the rest of the succession (Fig.7 H, I), suggesting elevated primary
500 productivity during oil shale sedimentation. Evidence from organic geochemistry also
501 supports this inference. The plots of Pr/nC17 versus Pr/nC18 plot, and GI versus Pr/Ph,
502 suggest that the organic matter in the oil shale was deposited under enhanced salinity
503 (Fig. 8C), and derives from algae and microorganisms. This is further supported by
504 large quantities of algal material in the oil shale samples, suggesting algal blooms at
505 this time.

506 These observations suggest that, during deposition of the oil shale, the arid climate
507 led to a decrease in the size of the lake, with strong salinization of the water. This saline
508 environment was suitable for the survival of algae and halophilic bacteria, leading to
509 an intense algal bloom during this period. Dissolved oxygen in the lake water was
510 rapidly consumed by microbial respiration, and the deeper lake waters became anoxic.

511 At the same time, algal growth in surface waters may have decreased gas exchange
512 between water and air, further promoting anoxia.

513 Recycling of major nutrients (particularly P) under anoxic conditions would also
514 have exerted a positive feedback on productivity (Benitez-Nelson, 2000; Sageman et
515 al., 2003). In particular, P is preferentially released from organic matter during
516 microbial remineralization, while reductive dissolution of iron (oxyhydr)oxide minerals
517 also releases adsorbed phosphate to solution (e.g., Krom and Berner, 1981; Froelich et
518 al., 1988; Slomp et al., 1996a,b; Anschutz et al., 1998; Xiong et al., 2019). Some of the
519 released P may be fixed in the sediment, via sink-switching to other phases such as
520 carbonate fluorapatite or vivianite (e.g., Van Cappellen and Ingall, 1994; Slomp et al.,
521 1996a,b; Xiong et al., 2019), but under anoxic water column conditions, a significant
522 proportion of the P may be recycled back to the water column, thus potentially
523 stimulating further productivity (Ingall and Jahnke, 1994, 1997; Slomp et al., 2002,
524 2004). This is exemplified by Lake Kivu, which is a salinity stratified lake in East
525 Africa. The lake has a phosphorus concentration of ~70 mg/L in bottom waters, which
526 results in substantial algal blooms and an organic carbon content in the lake sediments
527 of up to ~15 wt.% (Anadon et al., 1991). For ancient lakes, salinized lacustrine oil shale
528 with typical carbonate lamina also occurs in the Shahejie Formation of the Bohai Basin,
529 East China, which has been suggested to have been influenced by this recycling of
530 phosphorus (Wang and Zhong, 2004, Zhu et al., 2004).

531 **5.5 Mechanisms of lacustrine oil shale development under** 532 **arid climatic conditions**

533 Previous studies have shown that lacustrine shale deposition is controlled by
534 multiple factors, including climate, salinity and water depth (Anderson and Dean, 1988;
535 Dean et al., 1999; Wang and Zhong, 2004). The oil shale of the 7th member of the
536 Dameigou Formation in the Qaidam Basin exhibits typical carbonate lamina structure,
537 which deposited under high-productivity, high-salinity, reducing conditions. Various
538 studies have shown that deposition of laminated oil shale is closely related to the
539 stratification of the lake water (Dean et al., 1999; Zhu et al., 2005; Wang et al., 2012).
540 Due to the development of arid climatic conditions (Hu et al., 2017), freshwater supply
541 to the Yuqia Lake Basin rapidly decreased, leading to the gradual salinization of the
542 original freshwater lake. Due to the difference in density, the lake water formed a stable
543 saline-stratified environment, with oxic, lower salinity surface waters (Fig. 9A). These
544 conditions were conducive to algal organisms, generating very high rates of primary
545 productivity in surface waters.

546 The salinity of mid-depth water likely varied in relation to the local paleoclimate,
547 with consequent shifts in the depth of the halocline (Jin and zhu, 2006; Liu et al., 2015).
548 Although most plankton struggle to survive under saline conditions, halophilic bacteria
549 may breed in large numbers in this layer (Kemp, 1996). Bottom waters were highly
550 saline, and dissolved oxygen was rapidly consumed, producing an anoxic environment.
551 The lake stratification was stagnant, ensuring a long-period of anoxic and saline

552 conditions in the bottom water, thus providing good preservation potential for organic
553 matter. Furthermore, due to the lack of oxygen in the bottom water, nutrients such as
554 phosphorus were effectively recycled back to the water column, enhancing
555 eutrophication. Thus, both elevated primary productivity and the enhanced preservation
556 potential were crucial for the enrichment of organic matter in the oil shale.

557 In saline lacustrine environments, deposition of carbonate minerals is usually
558 controlled by CO₂ in the lake water (Lallier et al., 1996; Scholle et al., 1983). This
559 depends on a dynamic equilibrium between CO₂ consumption by plants and
560 microorganisms during photosynthesis, and CO₂ release by respiration and organic
561 matter degradation. During periods of algal blooming, dissolved CO₂ in the lake water
562 would be rapidly consumed. Indeed, Scholle et al. (1983) found that in highly eutrophic
563 lakes, the rate of CO₂ consumption can reach 1g/m²d, causing the pH of lake water to
564 rise above 9.0. Under dry climatic conditions, there was little surface runoff and
565 precipitation, which maintained stratification, and at the same time, input of terrigenous
566 minerals and organic matter was maintained at a low level. Enhanced photosynthesis
567 consumed CO₂ in the upper water of the lake, causing CaCO₃ to become oversaturated,
568 thereby precipitating carbonate minerals as laminae within the oil shale (Fig 3D).

569 However, under humid conditions, freshwater was continuously introduced by
570 precipitation and surface runoff, and the lake water was well mixed (Fig 9B). At this
571 time, productivity in the water column was at a normal level, resulting in only moderate
572 formation and preservation of organic matter. At the same time, more organic matter

573 and detrital minerals were imported from land sources. The input of terrigenous
574 material, including detrital quartz, feldspar and organic matter from higher plants
575 increased significantly, forming the organic-rich shale horizons in the 7th member of
576 the Dameigou Formation.

577 **6. Conclusions**

578 Our combined study of the geochemistry and sedimentology of the 7th member of the
579 late Middle Jurassic Dameigou Formation of northern Qaidam, provides new insight
580 into the mechanisms responsible for the deposition of organic-rich sediments in this
581 setting. The environment was prone to climate change, with humid periods interspersed
582 with intervals of high-intensity arid conditions. This paleoclimatic evolution had a
583 strong influence on the depositional setting, which controlled the development of
584 organic-rich shales in this lacustrine setting. Climate change altered surface runoff,
585 which gave rise to the deposition of different shale lithofacies. In particular, the
586 changing climate regime influenced the salinity and redox evolution of the lake water.
587 During humid periods, the water column was well-mixed with significant freshwater
588 supply.

589 Under arid conditions, however, insufficient freshwater supply led to lake
590 salinization and stratification. Vertical water circulation was inhibited, causing the
591 development of anoxic and saline conditions in the bottom water. This ultimately led
592 to eutrophication of the lake water, enhancing rates of primary productivity. The

593 remineralization of sinking organic matter consumed dissolved oxygen in the surface
594 lake water, and oxygen levels may also have been hindered by limited gas exchange
595 between the water and the air. The subsequent development of persistent anoxia
596 induced the recycling of nutrients such as P, which further promoted and maintained
597 high productivity in the lake.

598 These processes were responsible for oil shale development and organic matter
599 enrichment under arid climatic conditions. On the one hand, the blooms of algae and
600 plankton caused high productivity in the surface water, while due to the stable anoxic
601 conditions, the bottom water provided ideal conditions for the preservation of organic
602 matter. On the other hand, under arid climatic conditions, the input of freshwater,
603 detrital minerals and organic matter was significantly reduced, which facilitated water
604 column stratification and contributed to the high burial rate of organic matter. In
605 addition, consumption of CO₂ by the algal blooms caused carbonate precipitation,
606 forming the typical carbonate laminae that are associated with the oil shale.

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Figure captions

952

953 Figure 1. (A) Location map showing the position of Qaidam Basin in China; (B) Map showing the
954 structural divisions of the Qaidam Basin (modified from Guo et al., 2018); (C) Variations in
955 the thickness and facies within the J_2d^7 shales, northern Qaidam Basin (modified from Kang et
956 al., 2015). The location of Chaiye 1 (CY1) is well marked in Fig 1C. Cross-section showing
957 the structural framework of the Yuqia sag (modified from Li et al., 2014).

958 Figure 2. Generalized stratigraphy of the Middle Jurassic Dameigou Formation in Yuqia area of the
959 northern Qaidam Basin (modified from Li et al., 2015). The target stratum is highlighted.
960 MRS=maximum regressive surface; MFS=maximum flooding surface; SU=subaerial
961 unconformity; HST=highstand systems tract; TST=transgressive systems tract; LST=lowstand
962 systems tract.

963 Figure 3. Stratigraphic plot showing the lithology and sedimentary structures of the middle Jurassic
964 Dameigou Formation in the CY1 well. Black arrows show the positions of the thin sections,
965 XRD and TOC analyses. Red arrows show the sample positions for major and trace element
966 analyses. Green stars represent locations of core samples for the organic geochemical analyses
967 shown in Fig. 8.

968 Figure 4. Representative thin sections from the J_2d^7 formation in the CY1 well. (A) Sandy mudstone,
969 1917.5 m, mainly composed of coarse-grained detrital minerals, such as quartz and feldspar;
970 (B) Organic-rich mudstone, 1950.2 m, high organic matter content with abundant fine-grained
971 feldspar, quartz and clay minerals; (C) Oil shale, 1924.2 m, black colored oil shale; (D) Oil
972 shale, 1925.5 m, finely laminated black shale, supporting deposition from anoxic bottom
973 waters; (E) Oil shale, 1926.7 m, fossils of algal residue paralleled distributed with beddings; (F)

974 Oil shale, 1926.7 m, fossils of algal residue stacked as lenticles; (G) Oil shale, 1927.5m, single
975 algal fossil; (H) Argillaceous mudstone, 1910.2 m, gray and black massive shale, with a high
976 clay content, low organic matter content, and a small proportion of quartz and feldspar.

977 Figure 5. Ternary diagram showing the mineral composition of four major lithofacies in the CY1
978 well through the J₂d⁷ formation.

979 Figure 6. Stratigraphic distribution of lithofacies, organic matter features, and paleoclimate data for
980 samples from the J₂d⁷ formation in the CY1 well.

981 Figure 7. Stratigraphic distribution of palaeosalinity, redox conditions, and primary productivity
982 data for the CY1 well samples in the J₂d⁷ formation.

983 Figure 8. Organic geochemical data for the J₂d⁷ formation in the Yuqia area. Colored sample points
984 were collected from the CY1 well (data from Qin et al., 2018). Gray samples were collected
985 from other wells in the Yuqia area (data from Ritt et al., 1999; Guo et al., 2018; Chen et al.,
986 2018; Meng et al., 2018; Meng et al., 2019; Zeng et al., 2019; Bai 2020). (A) Cross-plot of HI
987 and T_{mas} for shales of the J₂d⁷ formation in the Yuqia area; (B) Ternary diagram of C₂₇, C₂₈ and
988 C₂₉ steranes based on their responses on GC-MS m/z 217 mass chromatograms; (C) Cross-
989 plot of Pr/n-C17 and Ph/n-C18; (D) Cross-plot of GI and Pr/Ph ratios for different shales.

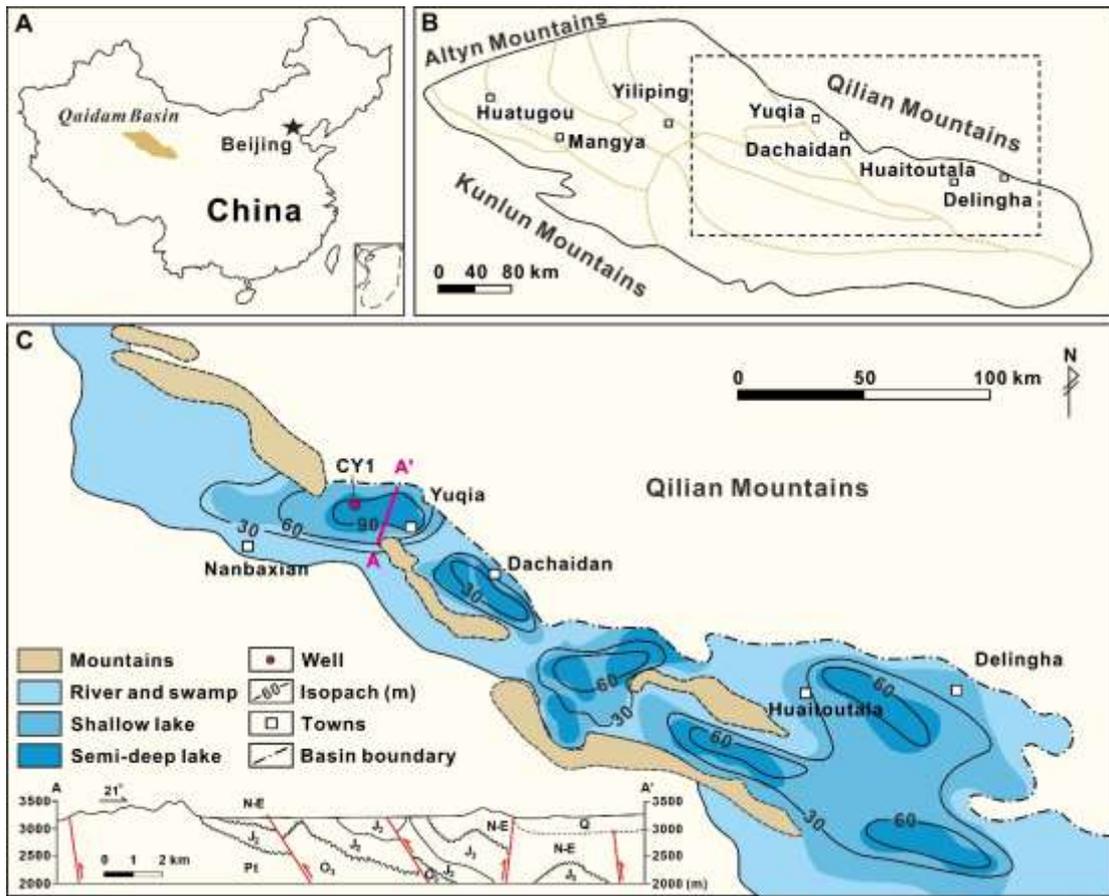
990 Figure 9. Schematic illustration showing factors controlling oil shale development and the
991 accumulation of organic matter in a lacustrine setting under an arid (A) and humid (B) climate.

992 **Table Captions**

993 Table 1. Mineral composition (wt.%) and total organic carbon concentrations (TOC wt.%) for the
994 four lithofacies in the J₂d⁷ formation.

995 Table 2. Geochemical indexes in CY1 well samples from the J₂d⁷ formation.

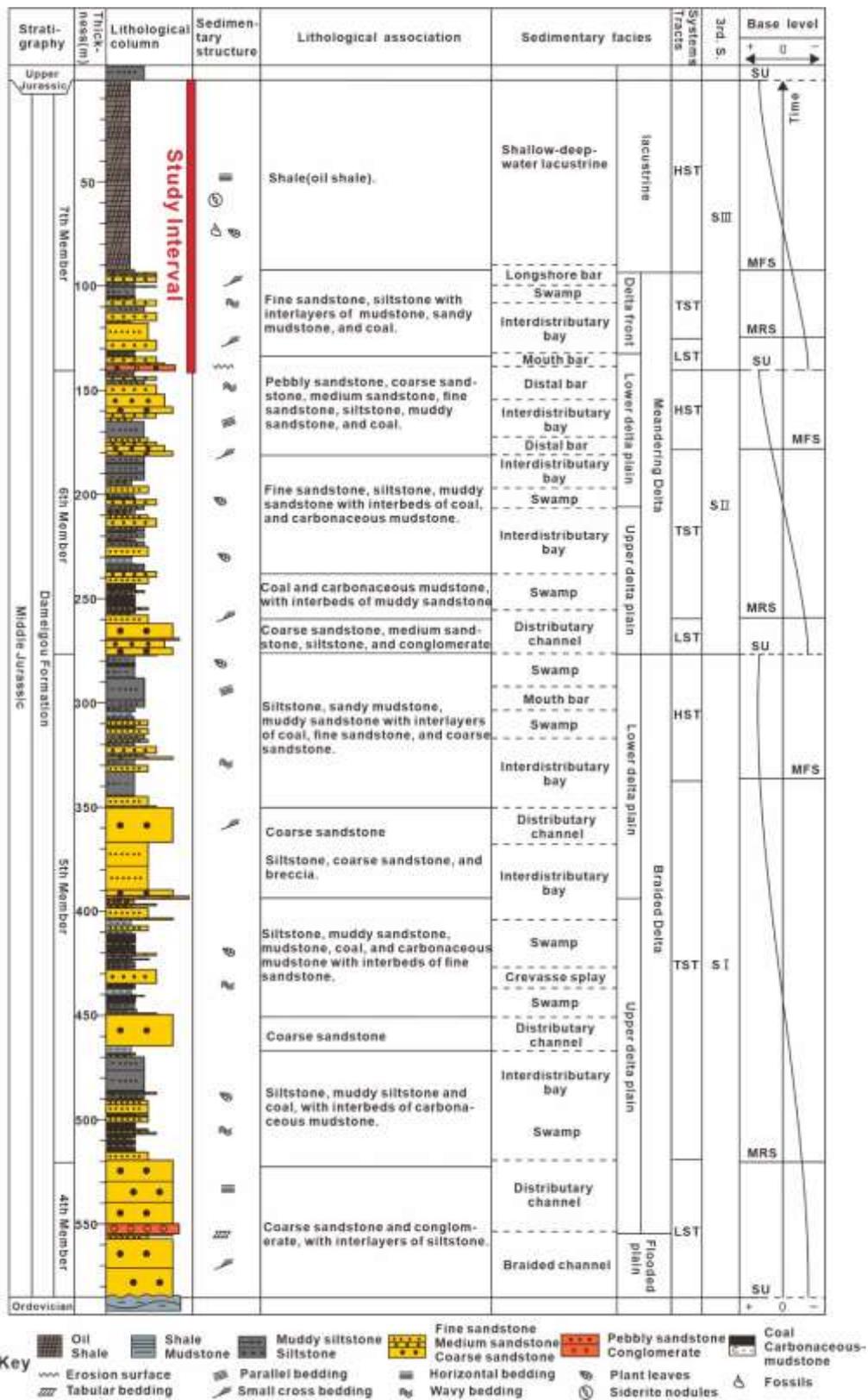
996 Figure 1



997

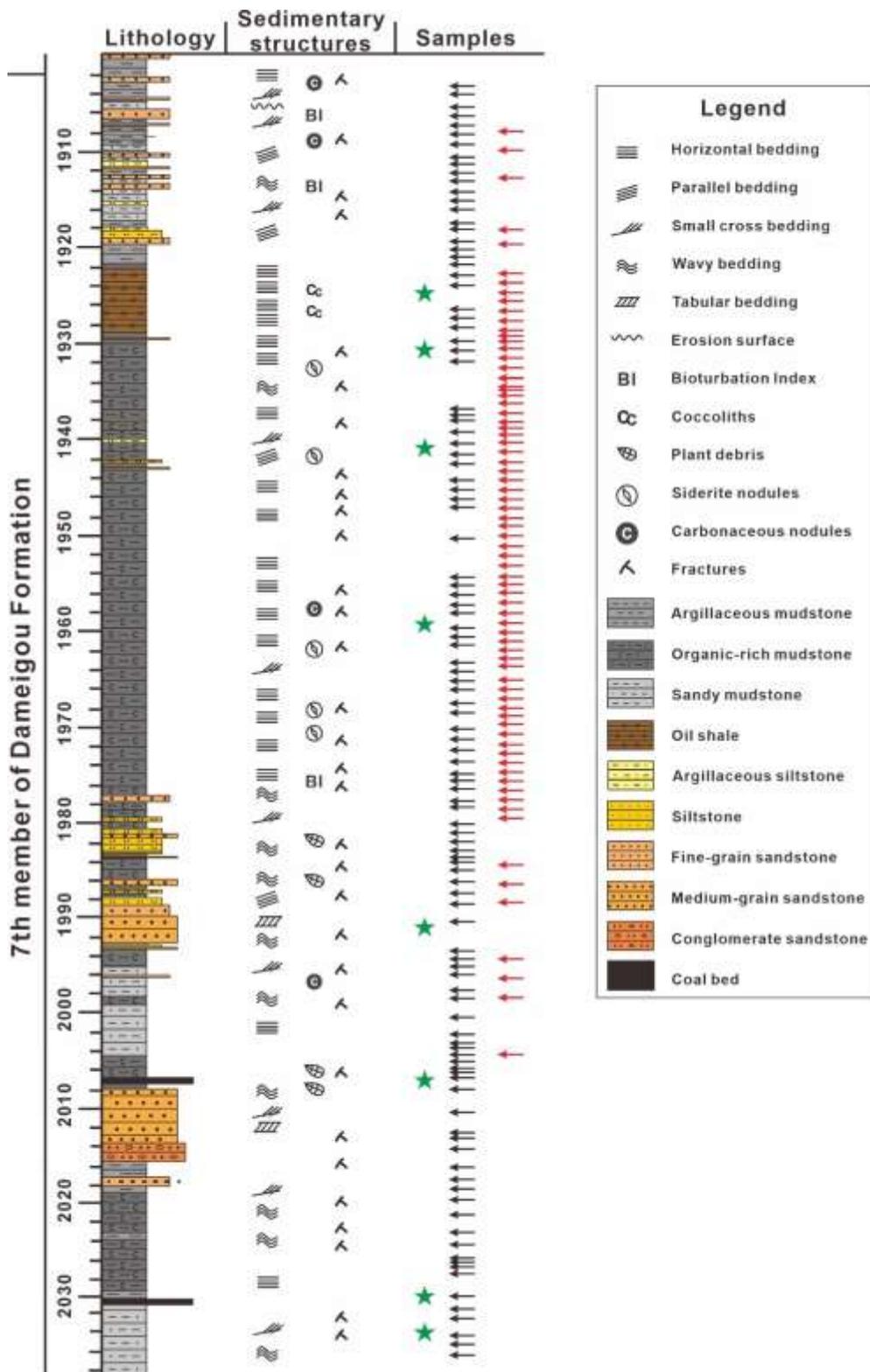
998

999 Figure 2



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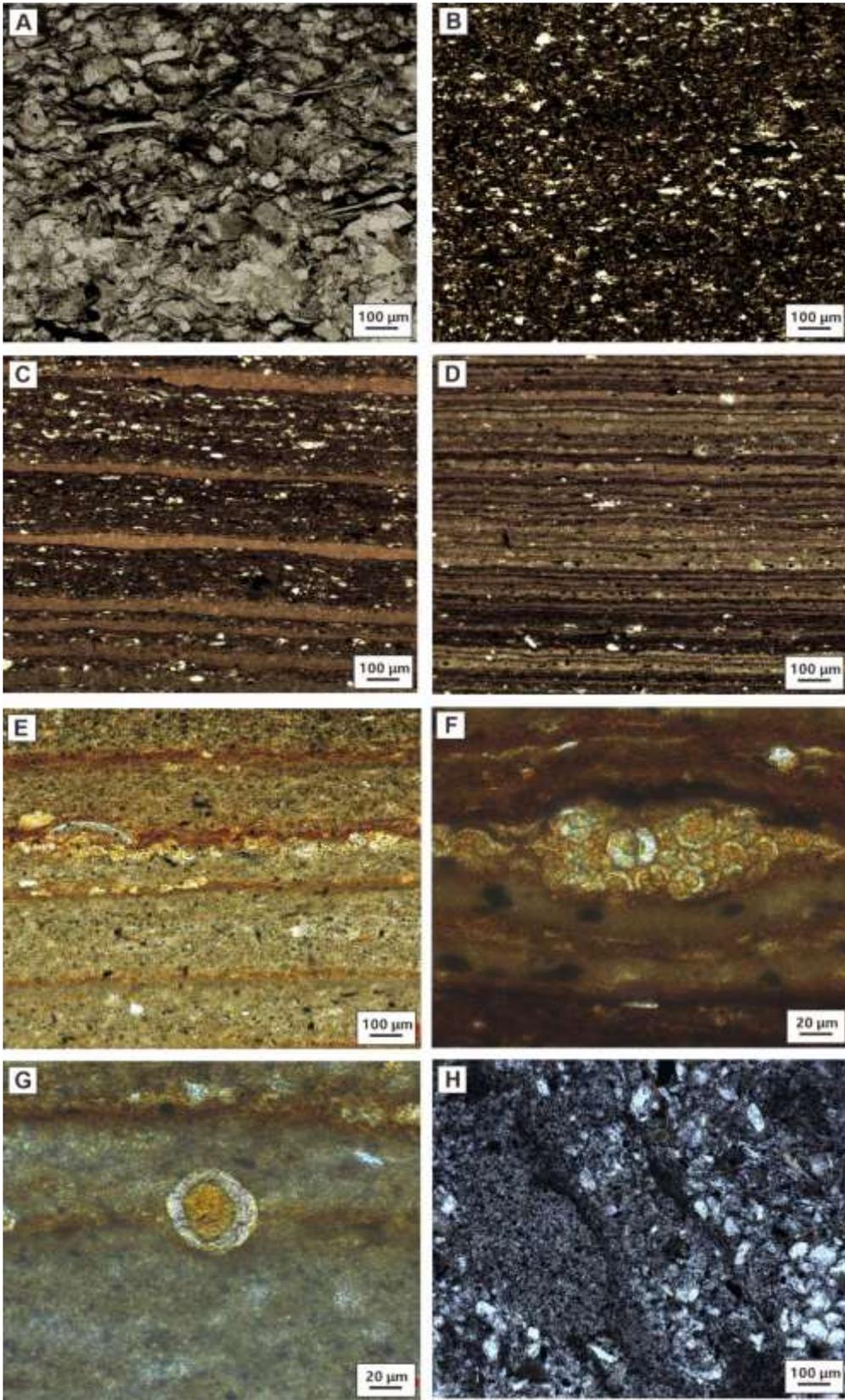
1002 Figure 3



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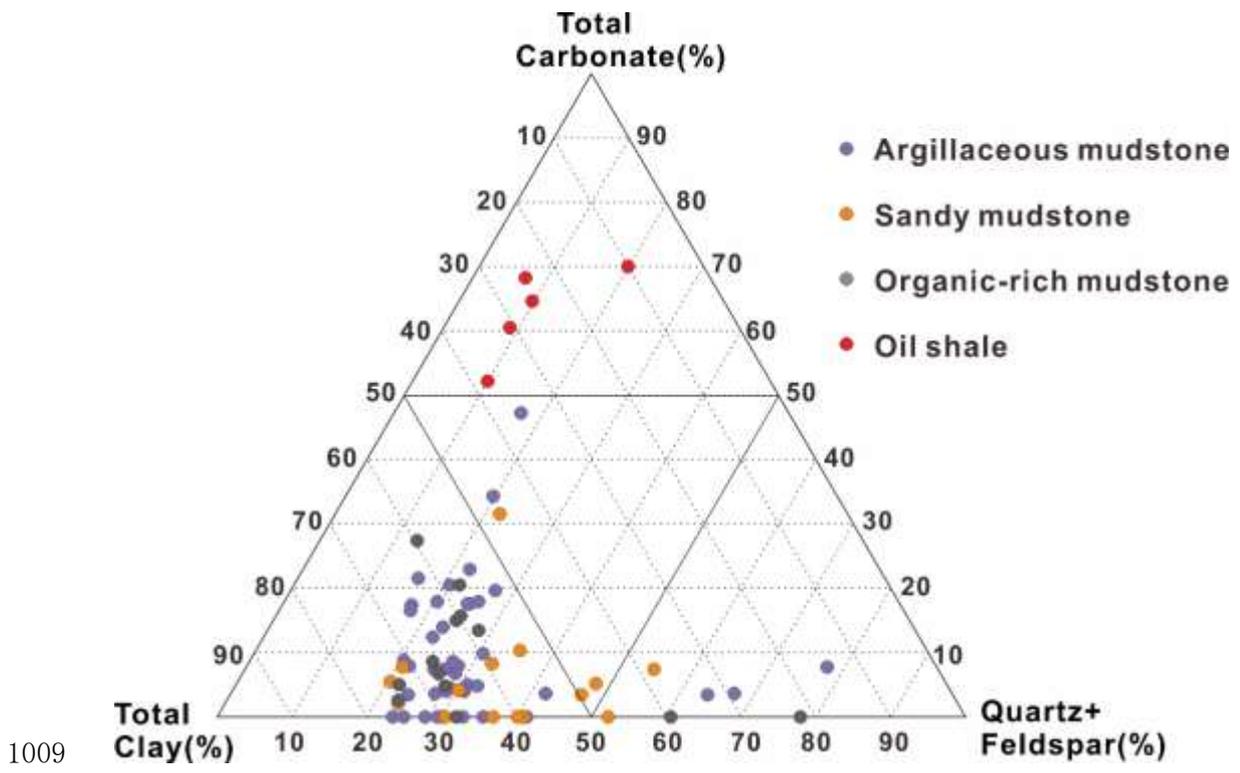
1005 Figure 4



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1008 Figure 5



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Figure 6

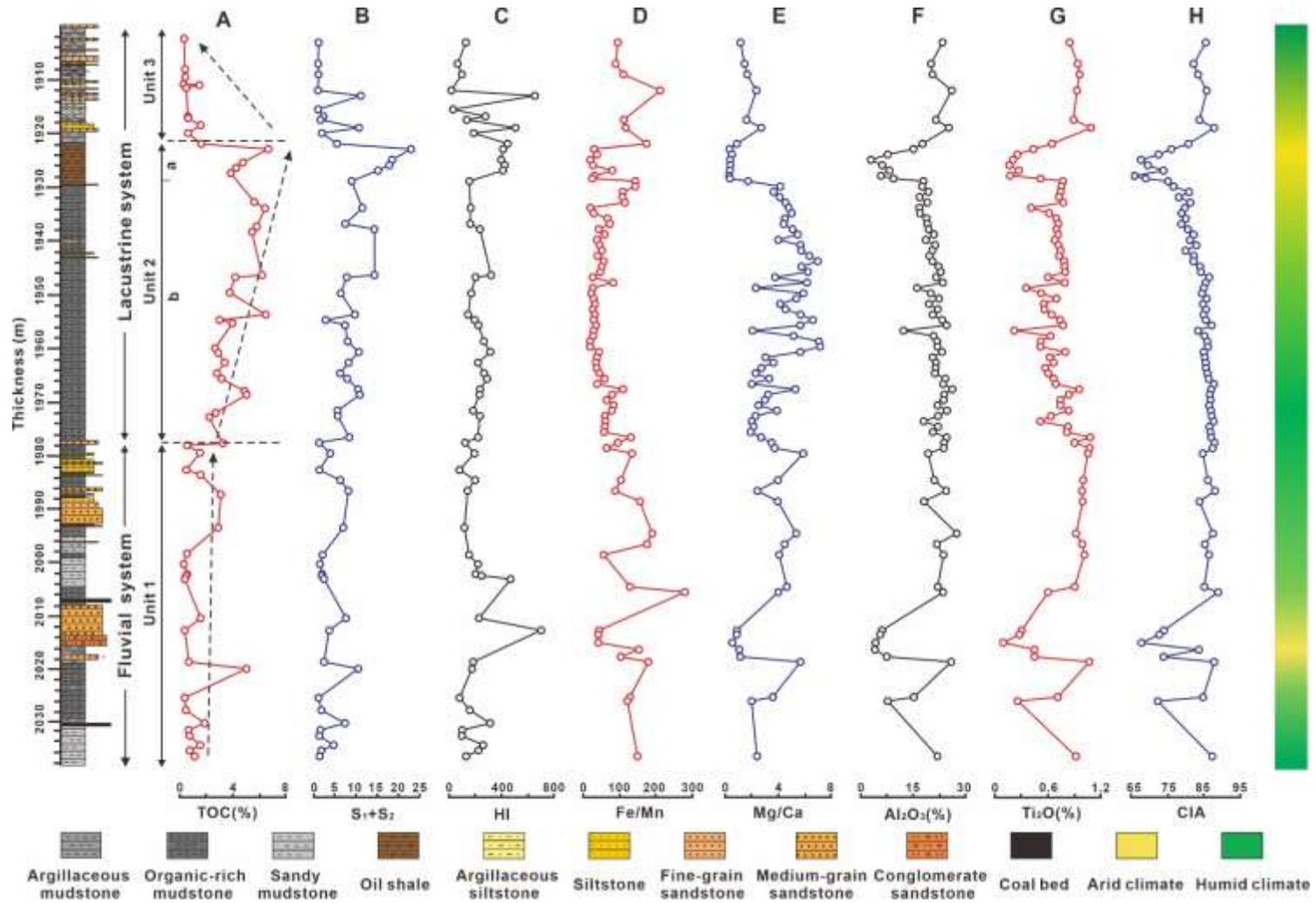


Figure 7

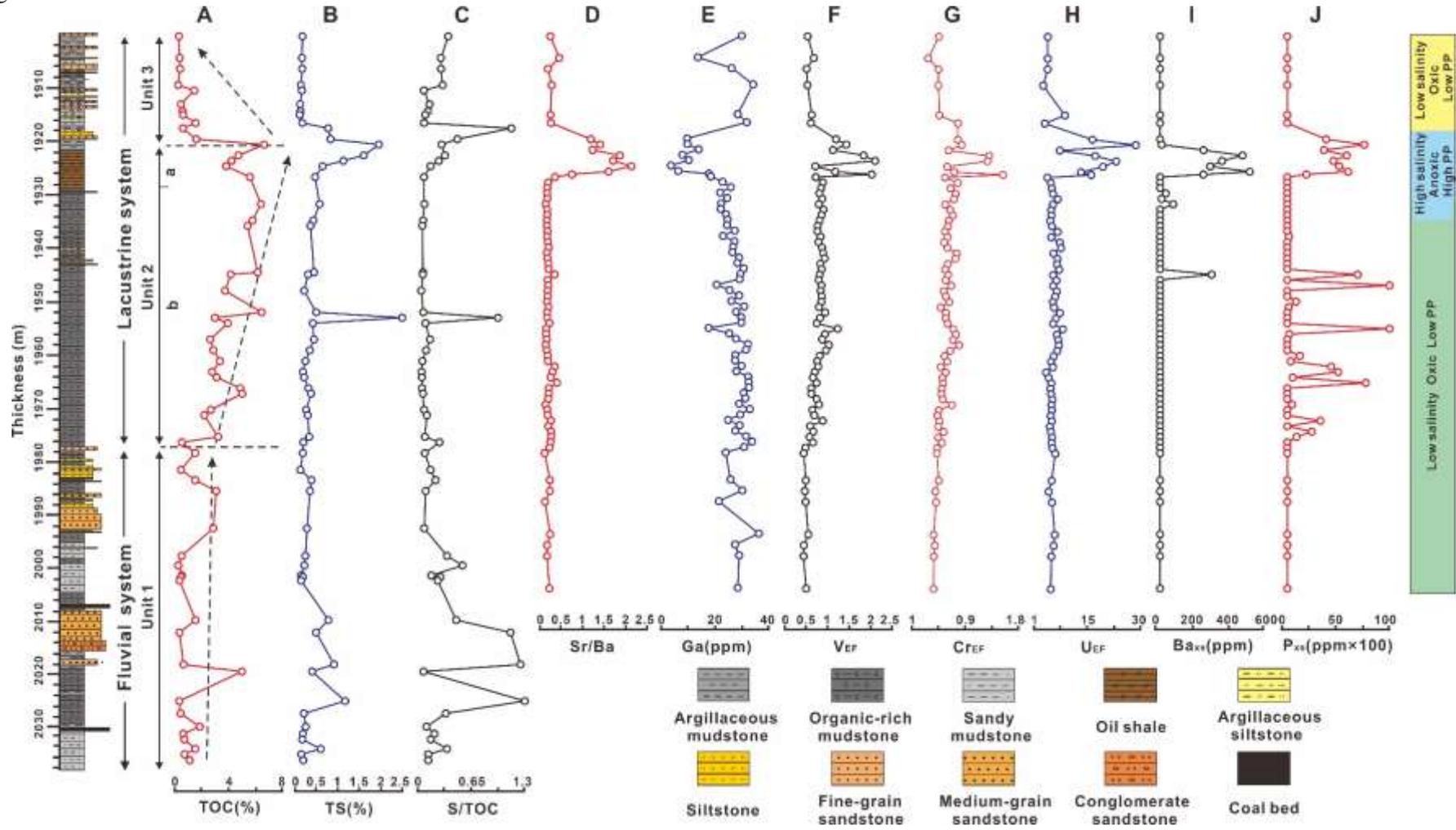


Figure 8

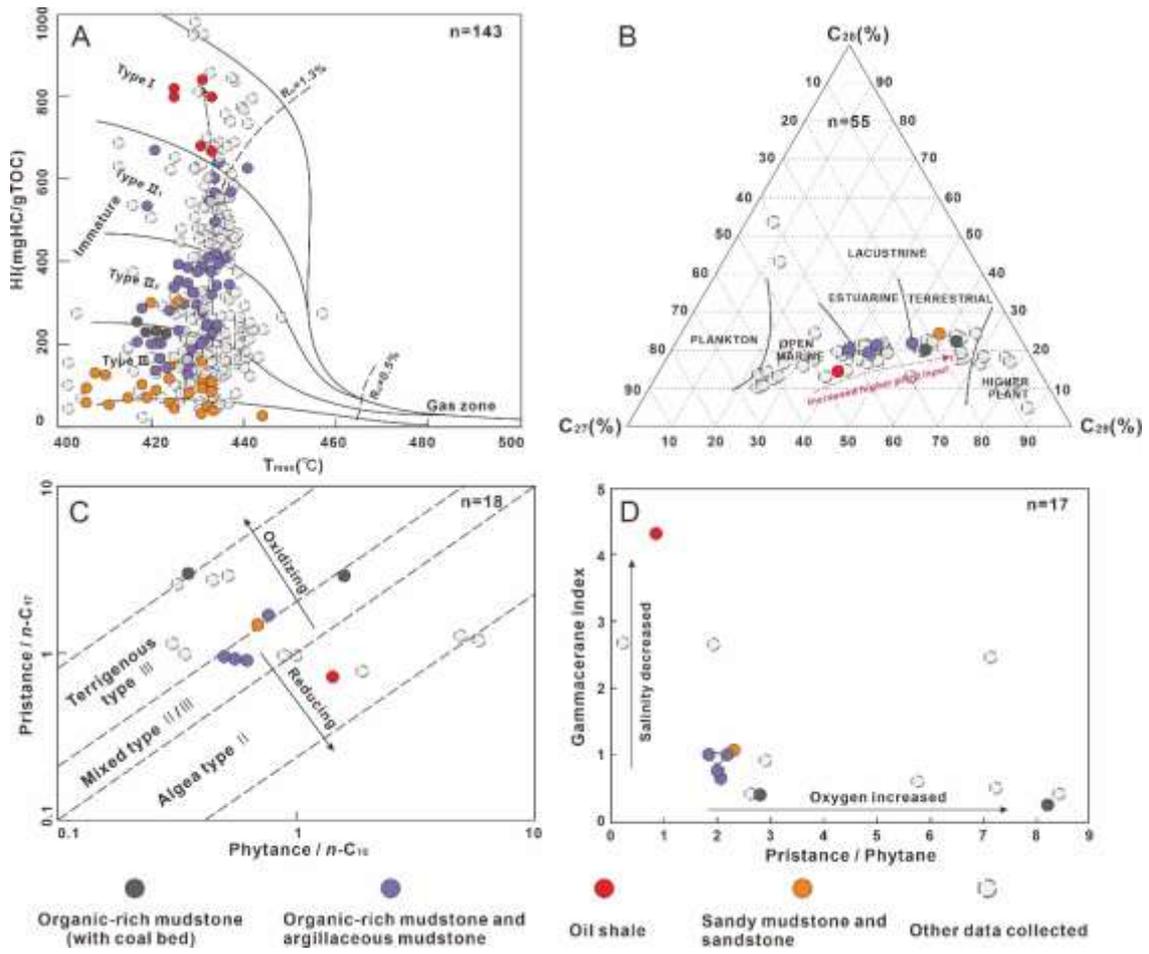


Figure 9

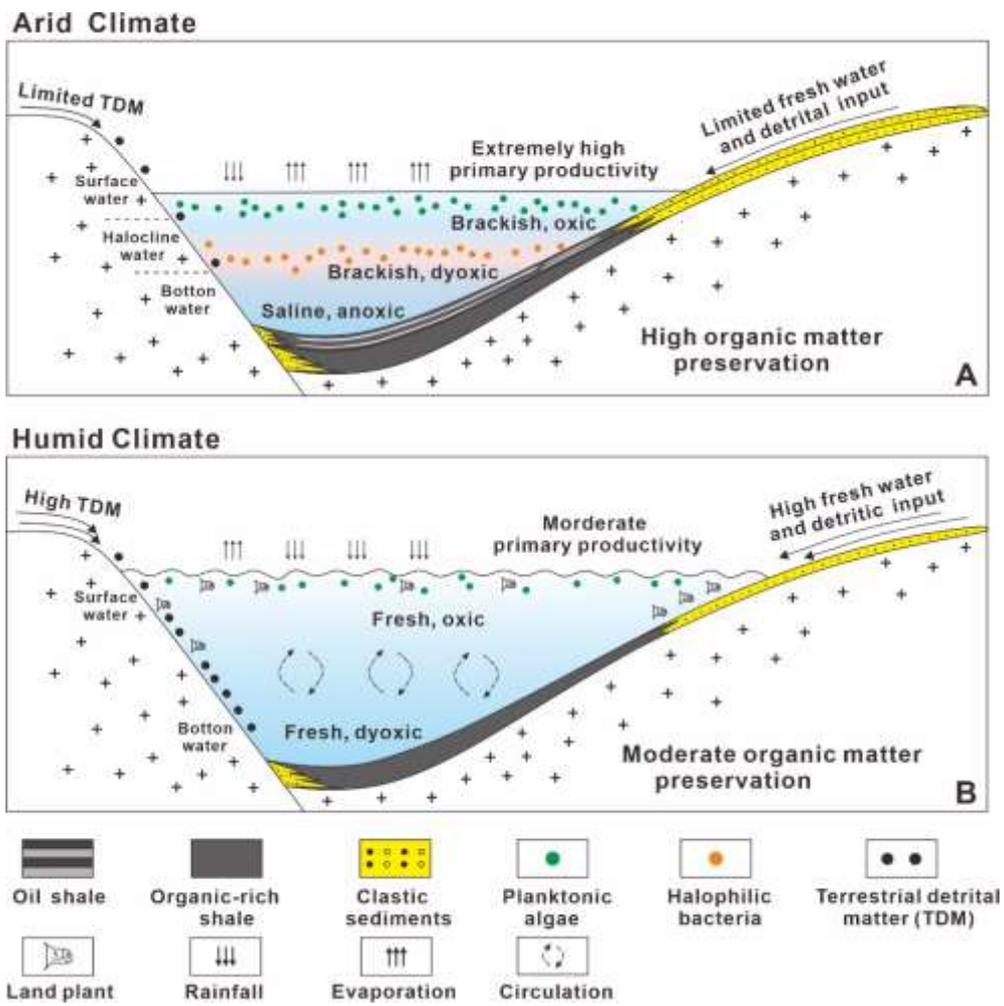


Table 1

Lithofacies	Quantities	TOC (wt%)	Average mineral content (wt%)							Rigid particle (wt%)	Carbonate (wt%)	Clay (wt%)
			Quartz	K-feldspar	plagioclase	Calcite	Dolomite	aragonite	siderite			
oil shale	6	4.3	19.8	2.1	4.7	12.6	3.1	17.8	1.6	26.5	35.0	38.5
organic-rich mudstone	53	3.6	26.4	1.6	2.5	0.3	0	0	7.8	30.6	8.1	61.3
argillaceous mudstone	13	1.6	11.3	1.7	2.1	3.0	0	0	6.2	31.3	9.2	59.5
sandy mudstone	20	0.8	32.7	2.3	1.2	1.1	0	0	4.9	36.2	6.0	57.8

Table 2.

Geochemical Proxies		Oil shale	Organic-rich Mudstone	Argillaceous Mudstone	Sandy Mudstone
Quantities		11	52	6	14
Fe/Mn	Min-max	12.5-173.4	11.7-154.4	84.9-212.4	35.5-282.9
	Average	69.8	52.2	119.7	125.2
Mg/Ca	Min-max	0.1-4.1	1.7-7.2	0.9-2.6	0.3-5.7
	Average	1.0	4.3	1.6	2.8
Al ₂ O ₃ (wt. %)	Min-max	4.0-19.8	13.0-26.3	20.5-26.2	5.1-27.5
	Average	12.3	21.4	23.1	16.0
Ti ₂ O ₃ (wt. %)	Min-max	0.2-0.8	0.2-1.9	0.9-1.1	0.1-1.1
	Average	0.5	0.7	0.9	0.6
CIA	Min-max	66.1-81.6	78.7-89.0	82.9-88.7	68.1-89.9
	Average	73.9	85.4	85.6	82.0
Sr/Ba	Min-max	0.2-2.3	0.1-0.4	0.2-0.4	0.2-0.2
	Average	1.1	0.2	0.3	0.2
Ga (ppm)	Min-max	4.3-26.1	18.0-33.8	14.1-34.2	27.6-36.3
	Average	13.8	27.8	27.5	30.4
Th/U	Min-max	1.0-4.5	2.7-6.4	2.4-8.9	3.7-5.7
	Average	2.5	4.2	5.2	4.9
V/Cr	Min-max	1.2-1.9	1.1-2.0	0.8-2.5	1.2-1.5
	Average	1.5	1.4	1.37	1.31
Ba _{xs} (ppm)	Min-max	0-523.3	0-301.2	0	0
	Average	197.8	8.1	0	0
P _{xs} (ppm*100)	Min-max	0-30.2	0-94.1	0	0
	Average	13.9	6.0	0	0