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Bondar, D, Basheer, M orcid.org/0000-0002-0835-8029 and Nanukuttan, S (2019) Suitability of alkali activated slag/fly ash (AA-GGBS/FA) concretes for chloride environments: Characterisation based on mix design and compliance testing. *Construction and Building Materials*, 216. pp. 612-621. ISSN 0950-0618

<https://doi.org/10.1016/j.conbuildmat.2019.05.043>

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1 **Suitability of alkali activated slag/fly ash (AA-GGBS/FA) concretes for chloride environments:**
2 **Characterisation based on mix design and compliance testing**

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8 **Abstract**

9 Alkali Activated Materials (AAM's) exhibit superior performance in terms of abrasion and acid
10 resistance and fire protection. Ground granulated blast furnace slag (GGBS) and fly ash are the two
11 most common industrial by-products used as precursors in alkali activated material. Alkali-activated
12 systems based on fly ash often require elevated temperature curing, while GGBS alone can lead to
13 flash setting, low workability and high degree of drying shrinkage. This article presents data from a
14 range of GGBS/fly ash blends designed to avoid elevated temperature curing, with desirable fresh
15 properties. Further the suitability of such blended concrete for their use in chloride environments is
16 outlined with the help of diffusivity testing and binding assessment. The results show that while
17 attractive engineering properties can be achieved, the chloride ingress in the AA-GGBS/FA concretes
18 studied here with $\geq 20\%$ fly ash is moderate to high and this will make them unsuitable for protecting
19 steel reinforced in concrete exposed to chloride environments.

20 **Keywords:** alkali activated slag/fly ash (AA-GGBS/FA) concretes, workability, compressive
21 strength, bulk resistivity, chloride diffusion coefficient, chloride binding

22 **1. Introduction**

23 Alkali Activated Materials (AAM's) have been under consideration as an alternative binder system
24 since 1895 [1]. The main limitations facing the commercial adoption of these binders are the
25 challenges in controlling engineering properties when using intrinsically variable waste-derived
26 precursors, and the lack of field performance data, especially in extreme environments. GGBS and fly
27 ash are the major by-products used as precursors for making these types of binders. Previous
28 investigations show that alkali-activated high calcium systems such as AAS concretes usually set
29 rapidly, can have low workability, and sometimes also a high degree of drying shrinkage compared to
30 Portland cement (PC) based systems, and all of these are known to affect durability properties [2-6].
31 Low calcium alkali activated systems, on the other hand, can exhibit high workability and a lower
32 shrinkage than PC systems [7], but need elevated temperatures for curing and hardening. Chi and
33 Huang [8] studied the behaviour of the binder products and properties of AA-GGBS/FA blended
34 mortars and concluded that better properties, compared to PC, were obtained in terms of compressive
35 strength, flexural strength and water absorption. Abdalqader et al. [9] found that the highest strength
36 of AA-GGBS/FA obtained was with a paste mix consisting of GGBS: fly ash in a 3:1 ratio. Gao et al.
37 reported that concrete with a higher GGBS/fly ash ratio exhibit lower porosity [9-10]. However,
38 despite having been aware of this materials potential for over a century, there is still relatively little
39 known about the durability of AAM's, especially the resistance to chloride ingress and the binding
40 behaviour. It was reported by Roy et al. that the advantageous properties of AA-GGBS/FA are their
41 high early strength, rapid hardening and high ultimate strength, low heat of hydration, and resistance
42 to effects resulting from chemical attack. Diffusion coefficients of 50×10^{-15} and 5×10^{-15} m²/s were
43 reported for the blended OPC and GGBS paste activated with NaOH, with 0 and 100% GGBS
44 replacement [11].

45 In this investigation, blended binder systems were used as they provide control of properties such as
46 setting time, workability, shrinkage, strength and durability compared to those based on 100% fly ash
47 or GGBS. An experimental programme was developed to study the workability, strength development
48 and resistance to chloride ingress for a range of AA-GGBS/FA concrete mixes that do not require
49 high temperature curing. Fresh properties, compressive strength at different ages, bulk resistivity, long
50 term chloride diffusion and binding behaviour were determined and reported for various AA-
51 GGBS/FA concretes.

52 **2. Experimental details**

53 **2.1 Materials**

54 The primary raw materials used in this study were GGBS and low-CaO fly ash, which were provided
55 by ECOCEM – Ireland and Power Minerals Ltd, UK, respectively. The chemical and physical
56 properties are presented in Tables 1 and 2. GGBS and fly ash were blended in 80/20, 60/40, 40/60 and
57 20/80 mass proportions, to be activated and used as binder.

58 Sodium hydroxide (NaOH) powder or pellets were dissolved in water to produce the alkaline
59 solutions. The chemical composition of the sodium silicate solution was 15.5% sodium oxide (Na₂O),
60 30.5% silicon oxide (SiO₂) and 54% water. In this study, the percentage Na₂O% in the paste (Na₂O%)
61 was selected to be 6 and 8% by powder (GGBS and fly ash) weight, and the silicate modulus (molar
62 ratio SiO₂/Na₂O) was varied from 0.45, 1.0 and 1.25, by blending sodium silicate and sodium
63 hydroxide.

64 The aggregates used in this study were crushed basalt from local sources in Northern Ireland and
65 comprised of 16.5 mm and 10 mm crushed coarse aggregates, and 4 mm sand. These were combined
66 in a ratio of 48:12:40 to get the maximum packing density in (AA-GGBS/FA) concrete mixes. The
67 bulk specific gravity and water absorption of these materials were measured based on BS EN 1097-1
68 and are presented in Table 3. Potable tap water (i.e. drinking water quality) was used to make the
69 concrete mixes.

70 **2.2 Mix details, casting procedure and test specimens**

71 Sixteen (AA-GGBS/FA) concrete mixes were studied with different GGBS/fly ash content, water-to-
72 binder, percentage of alkali and the SiO₂/Na₂O ratio (silica modulus, Ms). The details of the different
73 mixes and their initial properties are presented in Table 4. The total binder content, which is the sum
74 of GGBS, fly ash and solid component of the WG, was kept constant at 425 kg/m³ for all mixes and
75 the water content in the WG was taken into account while determining the mixing water. For
76 increasing proportions of the fly ash, the W/B for mixtures was reduced while the paste content of the
77 mixtures was kept constant.

78 **Mixing**

79 Blended cements were prepared by dry-mixing GGBS and fly ash in required proportions in a bench
80 top food mixer to get a homogeneous blended powder. Crushed basalt aggregates and sand were dry
81 mixed together in a laboratory pan-mixer for one minute. After adding the blended GGBS and fly ash
82 powder, mixing continued for another 2 minutes. The sodium hydroxide solution was then added, and
83 after 2 minutes of further mixing, sodium silicate solution was added and mixing continued for a
84 further minute. The details of the different mixes and their essential properties are presented in Table
85 4.

86 **Casting and curing of the specimens**

87 Fresh properties of concrete were measured according to BS EN 12350 [12]. From each concrete mix,
88 nine 100 mm cubes, and three 100x200 mm cylinders, were cast for the determination of compressive

89 strength in accordance with BS EN 12390 [13], bulk resistivity and chloride diffusion coefficient
90 according to Nordtest NT Build 443 [14]. The concrete specimens were cast in three layers and
91 compacted on a vibrating table. After casting, all the specimens (still in the mould) were covered with
92 plastic sheets and left in the casting room for 24 h. They were demoulded and kept in sealed plastic
93 zip bags until the test date. The storage room was maintained at 23°C & 65% RH.

94 **2.3 Testing procedures**

95 Chloride transport through AA-GGBS/FA concretes was assessed using a non-steady state chloride
96 diffusion test, Nordtest NT Build 443 [14]. One day before the test age of 91 days, the 100x200 mm
97 concrete cylinders were cut to four slices of diameter 100 mm with a thickness of 50 mm per mix. A
98 slice with a thickness of 50 mm from the cast surface (trowel finished face) was considered for
99 measuring bulk resistivity, and the rest were kept for carrying out the chloride diffusion test. The
100 vacuum saturation regime specified in the standard was used to precondition the slices so that the
101 chloride flow is predominantly diffusive, and initial sorption or capillary forces are negligible. The
102 vacuum was applied to remove air for three hours and released afterwards. Samples were wrapped in
103 hessian saturated in deionised water to prevent leaching of ions, and placed in the container. The
104 weight of the sample was noted after an hour (W_1) and then vacuum was applied, followed by further
105 saturation. Weight was checked again (W_2). Usually after 6 hours, when $W_i - W_{(i-1)}$ was less than 0.1%,
106 the samples were considered fully saturated; if not, saturation was continued until this criterion was
107 met. After conditioning to a surface-dry condition, an epoxy resin was applied onto the surfaces of the
108 specimens in three layers except the exposure face (saw cut face). When the epoxy coating was dry,
109 the cores were immersed in a NaCl solution of concentration 165 g/L (~2.82 M) for six months. After
110 immersion, two cores for each mix were profile ground to obtain concrete dust from different depths
111 up to a depth was determined for each sample by spraying a 0.1 N AgNO₃ solution on the third
112 freshly broken concrete sample. These depths were measured from the exposed surface. The total
113 chloride content of the dust samples was determined in accordance with the recommendations of
114 RILEM TC 178-TMC [15] using a pre-calibrated potentiometric titration method. The concrete dust
115 was dissolved in deionised water in accordance with RILEM TC 178-TMC recommendations,
116 considering 3-5 min digestion [16] to measure the pH value of the suspension, and for the
117 determination of water soluble chlorides. Chloride diffusivity and the surface chloride content were
118 determined by using curve fitting to the error function solution of Fick's second law of diffusion, as
119 described in NT BUILD 443 [14].

120 **3. Results and discussions**

121 The following sections discuss the slump, flow and air content, compressive strength, chloride
122 diffusivity, and chloride binding capacity of AA-GGBS/FA concretes. The histograms categorize the
123 results for groups of concretes with different ratios of GGBS/FA and the contours graph are a good
124 map for designing AA-GGBS/FA concretes mixes based on GGBS/FA ratio and activator parameters
125 ($\text{Na}_2\text{O} \% * \text{Ms}$).

126 **3.1 Slump, flow and air content**

127 The purpose of this testing programme was to demonstrate the range of slump values that AA-
128 GGBS/FA concretes is capable of producing, and the changes to the governing variables that are
129 necessary to achieve high slump with low air content of the GGBS/FA mixes to have better mixes for
130 durability. The slump and flow results of the concretes are presented in Figures 1 and 2. The slump
131 values of the concretes show that all the mixes had a slump value greater than 120 mm. It can be
132 observed that although W/B was adjusted based on the proportion of the fly ash in the system,
133 increasing the fly ash in the blend increased the workability, and the maximum workability was
134 achieved for mix with the maximum fly ash content due to its bubble structure which increase the
135 lubricity. The literature shows that increasing the sodium oxide content results in increased

136 workability, reduced setting times and higher compressive strength. However, high concentrations
137 resulted in loose structure in sodium hydroxide activated samples because of the excess water is being
138 retained in the system before forming a monolithic hydrated structure [18, 19]. For mixes with the
139 same GGBS to fly ash content and water to binder ratios, comparing mixes 1, 5, 9 and 13 and mixes
140 4, 8, 12 and 16 respectively, the slump values of the AA-GGBS/FA concretes with the same Ms ratio
141 increased with an increase of Na₂O%. The slump values also increased with the increase of Ms which
142 can be seen by comparing odd and even mixes (e.g. mix no. 1 against 2, 3 against 4) in the two groups
143 with two different Na₂O% and the same ratios of GGBS to fly ash.

144 As is evident from Fig. 1, the AA-GGBS/FA concretes can be designed for S3, S4 and S5 slump
145 ranges. Thus, all the mixes meeting the minimum slump requirement for their use in chloride
146 environments, such as S2 specified in BS8500-1:2015 for different exposure of marine environments
147 [17].

148 The contour map graphs in Figure 3 (a) and (b) shows the results of slump and flow test against
149 GGBS/FA ratio and a function of activator parameters or (Na₂O%*Ms) values for different mixes
150 with same paste content. (Na₂O%*Ms) is the ratio of the SiO₂ content of the activator to the total
151 cementitious materials content for mixes. It is depicted from both figures that increasing the
152 GGBS/FA ratio, decreases the workability of mixes for GGBS/FA<0.67 and GGBS/FA>1.5 while
153 increases that for 0.67<GGBS/FA<1.5. The maximum workability of mixes happens when
154 1.5<GGBS/FA<2.5 when (Na₂O%*Ms) values is less than 7.5. It is also evident from these figures
155 that workability increased when (Na₂O%*Ms) was increased which can be due to disparity of silica in
156 solution form.

157 The air content of the GGBS/FA mixes are presented in Figure 4 and Figure 5 and they show the
158 contour graph for air content in fresh mixes of AA-GGBS/FA concretes against GGBS/FA ratio and
159 (Na₂O%*Ms) values. The lowest air content was measured for mixes with a GGBS/fly ash ratio of
160 60/40 while the highest was for 20/80 (except mix 15) while considering the map analysis there is two
161 ideal island which shows the minimum air content in the fresh AA-GGBS/FA concretes. These two
162 islands happen when 0.6<GGBS/FA<1.0 or 1.0<GGBS/FA<2.25 and the (Na₂O%*Ms) for mixes is
163 more than 6.75% and less than 4.25%, respectively which are recommendable to have workable AA-
164 GGBS/FA concrete with low air content. Figure 4 also shows in the mixes made with a higher alkali
165 percentage, the air content is lower (comparing Mixes 1, 5 and 13 to Mixes 4, 8 and 16, respectively)
166 and higher Ms for mixes has had no major effect, especially for mixes with higher alkaline content
167 (comparing Mixes 3 to 4 or 7 to 8). However higher silica content in the activator was more effective
168 when the fly ash proportion increased in the blend, and this causes higher air content when using
169 higher silica content (comparing Mixes 15 to 16) which means silica do not let air to be pushed out
170 from fresh concrete.

171 3.2 Compressive strength

172 Figure 6 shows the compressive strengths of AA-GGBS/FA concrete mixes at 2, 28 and 90 days.
173 From the results, it can be stated that: (1) the first three groups had more than 20 MPa after 2 days and
174 more than 50 MPa after 28 days curing and 28 day strength is 80–90% of the 91 day strength, offering
175 insight into the short and long term microstructural development in such binders, (2) as is expected,
176 the more fly ash in the system, the lower is the early age and long term strength, (3) an increase in
177 Na₂O% and Ms generally increases the compressive strength of AA-GGBS/FA concretes, which
178 agrees with the results reported by others. The last one is due to more N-A-S-H (sodium alumina
179 silicate hydrate) reaction product being generated, based on the increase of Na₂O%, and more C-A-S-
180 H (calcium alumina silicate hydrate) reaction product being produced, based on the increase of SiO₂
181 content. Silica modulus of 0.45 will drop down the strength for Na₂O% of 8% and this can be because
182 of low silica content in the activator to make the gel formation. The activated GGBS/fly ash concretes
183 with Ms of 1.0 up to 1.25 generally obtained the highest compressive strength (see Figure 6), which

184 also has a good agreement with the literature and shows that the activated GGBS/fly ash with Ms of
185 around 1.0 obtains the best hydration. From the 28day compressive strengths, Table 4 and Figure 6
186 both show that except mixes no. 7(1.5-8%-0.45), 11(0.67-8%-0.45) and 13 to 16 with
187 GGBS/FA=0.25, all achieved the required strength for the EN206 exposure classes XS3 and XD3
188 which is equivalent to an average strength ≥ 56.99 MPa (calculated as $45 + 1.48(8.1)$ MPa; where 8.1
189 is the standard deviation of the test results). However, mixes 7(1.5-8%-0.45), 11(0.67-8%-0.45) and
190 16(0.25-8%-1.0) met the strength requirement of $48.99 (=37+1.48(8.1))$ MPa for the exposure classes
191 XS1, XD1 and XD2. Although it was evident that the strength requirement and w/b will not comply
192 with BS 8500-1:2015 norms, the intention of including the mixes were to assess their performance
193 against Cl⁻ ingress.

194 Figure 7 (a) to 7 (c) presents the contour map of 2, 28 and 90 days compressive strength against
195 GGBS/FA ratio and ($\text{Na}_2\text{O}\% * \text{Ms}$) values. In general, there is an increase in the 2, 28-days
196 compressive strength values with increase in the SiO_2 content of the activator to the total cementitious
197 materials content. However, when the SiO_2 content of the activator reach to 6.5% and 7.5% of the
198 total cementitious materials content respectively the reverse trend is started. 90-days compressive
199 strength does not change significantly for GGBS/FA ratio >1.5 and ($\text{Na}_2\text{O}\% * \text{Ms}$) >6 .

200 3.3 Bulk electrical resistivity

201 Electrical resistivity can evaluate microstructure of concrete and be related to the pore structure and
202 conductivity of the pore solution to predict the diffusion coefficients of chloride ions. The test set-up
203 for measuring bulk resistivity is shown in Figure 8. Figure 9 depicts the average bulk electrical
204 resistivity for different AA-GGBS/FA concrete mixes. Higher resistivity could be attributed to denser
205 structure with low connected porosity and/or less conductivity of the binder matrix and such mixes
206 will be expected to be superior in resisting ionic (chloride) flow. Higher GGBS content in AA-
207 GGBS/FA concretes resulted in higher bulk resistivity, while mixes with higher alkali content show
208 higher bulk resistivity when GGBS dominates the blend. Furthermore, bulk electrical resistivity
209 decreases for mixes made with higher silica modulus (except Mixes 3 and 4), whereas the bulk
210 resistivity seems to be a maximum at $\text{Na}_2\text{O}=8\%$ and $\text{Ms}=1.0$ when GGBS/FA=80/20. This may be
211 related to the alkali content which is required to produce maximum reaction products and gives the
212 confidence for designing the concretes for strength and durability.

213 Figure 10 shows contour graph of bulk electrical resistivity for different mixes against GGBS/FA
214 ratio and ($\text{Na}_2\text{O}\% * \text{Ms}$) values. It is depicted from Figure 10 that for mixes with GGBS/FA < 0.67 the
215 SiO_2 content of the activator has minimal influence on bulk resistivity and the resistivity does not
216 change for ($\text{Na}_2\text{O}\% * \text{Ms}$) $< 6.25\%$. This can show that for GGBS to fly ash less than 0.67 and
217 ($\text{Na}_2\text{O}\% * \text{Ms}$) $< 6.25\%$ pore structure dominates the conductivity of the system which is not
218 dependent to the activator specification. Whereas for ($\text{Na}_2\text{O}\% * \text{Ms}$) $> 6.25\%$ the resistivity decreases
219 with increasing of the ($\text{Na}_2\text{O}\% * \text{Ms}$) which shows that pore solution domination and higher
220 conductivity because of free ions in the binder matrix reduce the resistivity. For mixes with same
221 ($\text{Na}_2\text{O}\% * \text{Ms}$) the bulk resistivity increases with increasing of GGBS to fly ash ratio for GGBS/FA
222 < 0.67 and GGBS/FA > 1.5 . This shows the effect having denser structure and is because of higher
223 GGBS content in the binder matrix of mixes. While for $0.67 < \text{GGBS/FA} < 1.5$ and ($\text{Na}_2\text{O}\% * \text{Ms}$) < 7.5
224 pore solution domination causes the reduction of resistivity and higher conductivity because of free
225 ions in the pore solution. This can show the range of ($\text{Na}_2\text{O}\% * \text{Ms}$) which is suitable for activation
226 corresponding to different GGBS to fly ash ratio and should be more than 7.5 for
227 $0.67 < \text{GGBS/FA} < 1.5$.

228 3.5 Chloride diffusion

229 Concrete sample disks with all surface painted except the top face, placed in air tight container with
230 sodium chloride solution for diffusion test as per NT Build 443 are shown in Figure 11.

231 Figure 12 presents average chloride diffusion depth and D_{nssd} for mixes. As shown in Figure 12 the
232 chloride diffusion depth of AA-GGBS/FA concretes was from 13 mm for mixes 2(6%-1.25), 3(8%-
233 0.45), 4(8%-1.0) with GGBS/FA=80/20 to 40 mm for mixes 13(6%-1.0) and 14(6%-1.25) with
234 GGBS/FA=20/80.

235 The non-steady state diffusion coefficient with AA-GGBS/FA concretes was resulted between
236 $1.05 \cdot 10^{-12} \text{m}^2/\text{s}$ [mix3 (8%-0.45)] with GGBS/FA=80/20 and $126.51 \cdot 10^{-12} \text{m}^2/\text{s}$ [mix14 (6%-1.25)]
237 with GGBS/FA=20/80. It can be seen that all the D_{nssd} for AA-GGBS/FA concretes was moderate to
238 high except for the mixes with GGBS to fly ash ratio of 80/20. This is in line with Mundra et al.
239 findings that states for fly ash blended AAMs concretes service life influenced by Cl^- transport. The
240 non-steady diffusion coefficient, D_{nssd} , of the AA-GGBS/FA concretes was found to be influenced by
241 the fly ash content, $\text{Na}_2\text{O}\%$ and M_s (at mixes with low alkali content). All the D_{nssd} values for AA-
242 GGBS/FA concretes were higher for higher fly ash content which is due to the influence of larger
243 pore size and more pore connectivity of the concrete.

244 The third group of data, with average X_d closer to 35 mm, the average D_{nssd} is 13.56. This is three
245 times bigger than second group which has an X_d close to 30. This can be because of more chloride
246 absorption in third group which has more fly ash in the system. Activated fly ash has been stated to
247 have zeolite formation with better initial surface absorption (physical chloride binding) than activated
248 slag [21]. Surface absorption is stated the main responsible for around 90% of the total chloride
249 uptake [22].

250 Figure 13 shows the contour map of D_{nssd} of different AA-GGBS/FA concrete mixes against
251 GGBS/FA ratio and $(\text{Na}_2\text{O}\% * M_s)$ values. D_{nssd} of AA-GGBS/FA concrete mixes does not change
252 significantly with increase of the ratio of the SiO_2 content of the activator to the total cementitious
253 materials content for GGBS/FA ratio < 1.0 and $(\text{Na}_2\text{O}\% * M_s) < 6.0$. Furthermore $(\text{Na}_2\text{O}\% * M_s) = 6.0$ is
254 an extreme point and at which D_{nssd} of AA-GGBS/FA concrete mixes for values less than 6.0 is
255 decreasing while for values more than 6.0 is increasing. The plaid area in Figure 13 shows the limits
256 of designing AA-GGBS/FA concrete mixes suitable for chloride exposure environments, if
257 reinforcing steel is preferred.

258 3.4 Chloride binding in AA-GGBS/FA concrete

259 There is limited information available on chloride binding of AA-GGBS/FA concrete in the literature.
260 The available CSH and aluminate phases (C-(N)-A-S-H or two layered double hydroxides) in
261 activated GGBS contribute to the physical and chemical binding reactions [22, 23] while there is no
262 information on the conformed phases in activated fly ash systems to show contribution of that to
263 chloride binding reactions. Binding behavior of different alkali activated slag (AAS) concrete was
264 reported in previous article by the authors [24]. Following the same procedure, the total and water-
265 soluble chloride concentration of the concrete dust removed for chloride diffusion test in section 3.3
266 were determined for mixes 13-16. This data is presented in Figures 14 a-d along with the
267 corresponding pH values. It is evident from Figure 14 that for the mixes studied, the total chloride
268 concentration is almost entirely represented by water-soluble chlorides. It is known that pH reduction
269 can release the Cl^- otherwise bound to Friedel's salt [25] in PC based systems. In a typical PC based
270 system, the total Cl^- will be composed of both bound and free chlorides, and free chloride is often
271 represented by the water-soluble fraction which may also contain a proportion of the adhered Cl^-
272 physically bound to the cement phases. Two aspects are noticeable from the figure, (1) the pH values
273 are ≤ 11 for most depth (or all depth), (2) the surface region is undergoing leaching-induced changes
274 during Cl^- transport that result in a reduction of pH; there is also a skin effect which causes a near-
275 surface dip in the total Cl^- content. While as resulted in previous study, for alkali activated slag (AAS)
276 concrete the pH values are > 11 for most depth and the surface region occurs within the depth of 5-12
277 mm [24].

278 The main reaction product in AA-GGBS/FA concrete is a Na-Al silicate hydrate (N-A-S-H) gel which
279 it seems have no tendency to bind chlorides in the alkalinity range observed in mixes 13-16. The
280 lower alkalinity could have affected the stability of the reaction products that bind chlorides. The least
281 fluctuation in pH reduction is for the mix 13, that has 6% Na₂O and Ms = 1.

282 The results presented in this article shows that AA-GGBS/FA with GGBS/FA ratio less than 2.5 are
283 not suitable for chloride exposure environments, if reinforcing steel is preferred. If the GGBS/FA
284 ratio is between 2.5 and 4, silica content in the activators should be selected based on Figure 13 to
285 have concrete with low diffusivity.

286

287 **4. Conclusion**

- 288 1. Results show that it is possible to achieve desirable fresh properties and compressive strength
289 for AA-GGBS/FA concrete when cured at 20⁰C. This means such concretes do not require
290 elevated temperature curing for strength gain.
- 291 2. A higher proportion of GGBS results in high early and long-term strength.
- 292 3. AA-GGBS/FA concretes with GGBS to fly ash ratio less than 4 have moderate to high
293 chloride migration coefficients, making them non-viable for protection of reinforcing steel in
294 chloride environments.
- 295 4. The blends can be categorised based on the GGBS to fly ash ratio (GGBS/FA) as low
296 diffusivity (GGBS/FA = 80/20), high diffusivity (GGBS/FA = 60/20), and very high
297 diffusivity (GGBS/FA = 40/60 & 20/80).
- 298 5. No bound chloride in the alumina silicate-dominated alkali-activated concretes, characteristic
299 of a lack chemical binding in the fly ash dominated AA-GGBS/FA concretes and as a result,
300 the chloride diffusivity in these binders depends on the role of the pore structure in physical
301 chloride binding and encapsulation which highlights the importance of the pore structure
302 studies in AA-GGBS/FA concretes.
- 303 6. Typically, workable AA-GGBS/FA concretes require binder content of 425 kg/m³
304 (GGBS/FA=4) and a water/binder of 0.47, which is close to the maximum acceptable limit for
305 XD3 and XS3 chloride exposure classes as per BS 8500.

306 **Acknowledgements**

307 The authors gratefully acknowledge the Engineering and Physical Sciences Research Council for
308 providing the financial support for conducting this work [EP/M003272/1]. The authors would like to
309 acknowledge the technical support provided by Dr. John Provis at The University of Sheffield as PI of
310 the project.

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Table 1: Oxide composition of GGBS & Fly ash

precursor	Component (mass% as oxide)						
	SiO ₂	Al ₂ O ₃	CaO	Fe ₂ O ₃	MgO	others	LOI
GGBS	35.7	11.2	43.9	0.3	6.5	2.09	0.31
Fly ash	46.8	22.5	2.2	9.1	1.3	14.5	3.6

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Table 2: Physical properties of GGBS & Fly ash

Material	GGBS	Fly ash
Fineness $\geq 45 \mu\text{m}$	7.74%	18.39%
Particle density	2.86	2.21
Water absorption	35.14%	27%

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Table 3: Physical properties of aggregates

Aggregates	Bulk specific gravity	Bulk SSD Specific gravity	Water Absorption (%)
Sand (0-4mm)	2.72	2.73	0.75
Fine Crushed Agg. (5-10mm)	2.67	2.75	3.14
Coarse Crushed Agg. (10-16mm)	2.60	2.67	2.60

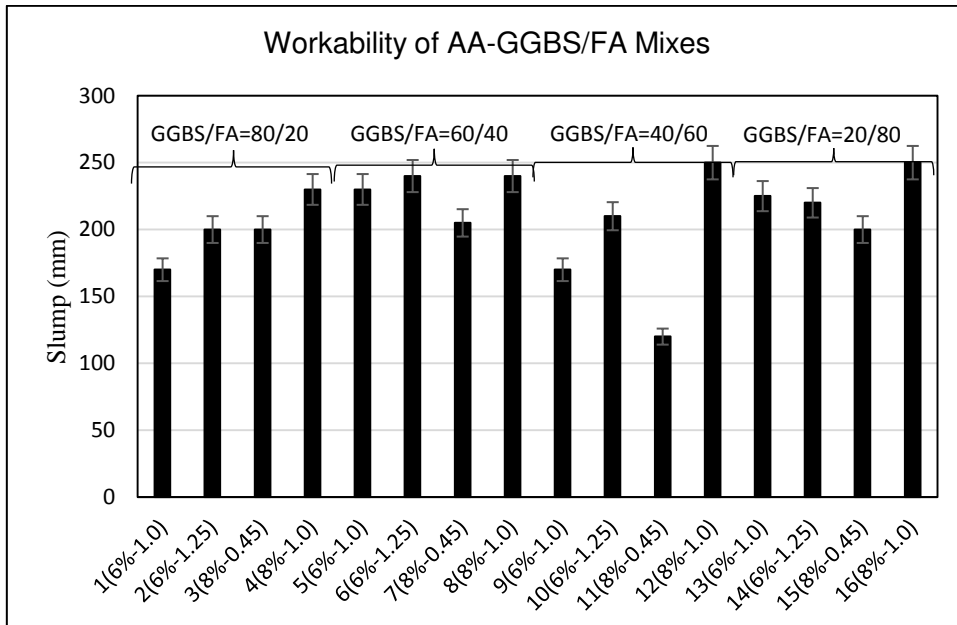
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Table 4: The details of the different mixes and their properties

Mix No.	GGBS/FA (kg/m ³)	GGBS/FA ratio	M _s (=SiO ₂ /Na ₂ O)	Na ₂ O (%)	W/B	Slump (mm)	2 days Comp. St. (MPa)	28 days Comp. St. (MPa)	Concrete Grades
1	340/85	4	1.0	6	0.47	170	54.6	81.1	C65/80
2	340/85	4	1.25	6	0.47	200	51.2	84.8	C65/80
3	340/85	4	0.45	8	0.47	200	35.2	60.2	C49/60
4	340/85	4	1.0	8	0.47	230	50.9	80.6	C65/80
5	255/170	1.5	1.0	6	0.46	230	42.1	73.4	C56/70
6	255/170	1.5	1.25	6	0.46	240	40.3	78.5	C60/75
7	255/170	1.5	0.45	8	0.46	205	28.8	52.4	C40/50
8	255/170	1.5	1.0	8	0.46	240	31.9	75.7	C60/75
9	170/255	0.67	1.0	6	0.44	170	28.2	64.8	C49/60
10	170/255	0.67	1.25	6	0.44	210	22.3	69.4	C52/65
11	170/255	0.67	0.45	8	0.44	120	22.2	54.0	C40/50
12	170/255	0.67	1.0	8	0.44	250	23.0	61.2	C49/60
13	85/340	0.25	1.0	6	0.43	225	13.8	42.6	C32/40
14	85/340	0.25	1.25	6	0.43	220	9.5	47.1	C35/45
15	85/340	0.25	0.45	8	0.43	200	9.7	28.4	C20/25
16	85/340	0.25	1.0	8	0.43	250	10.4	49.2	C35/45

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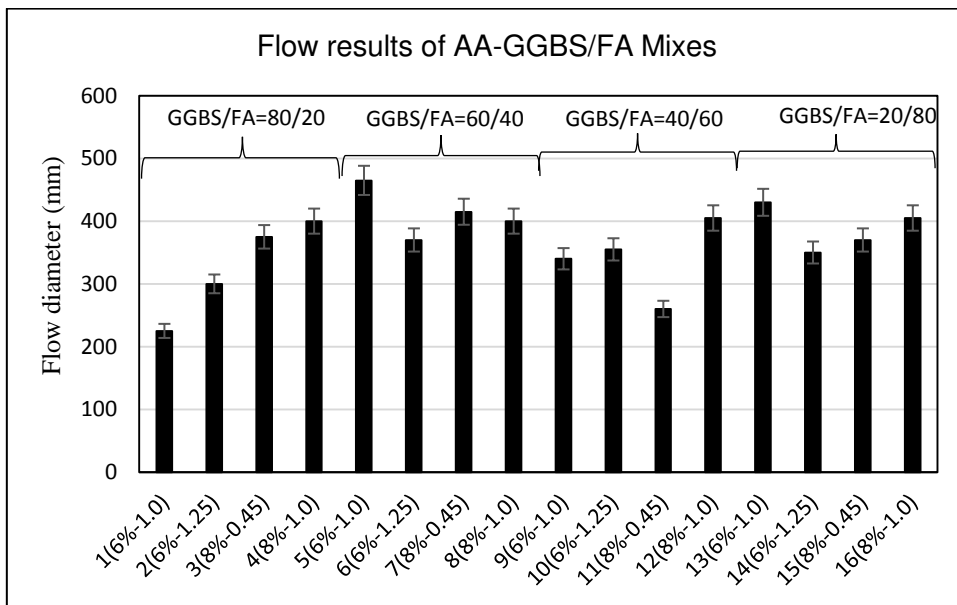


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Fig. 1-Slump results of AA-GGBS/FA concretes



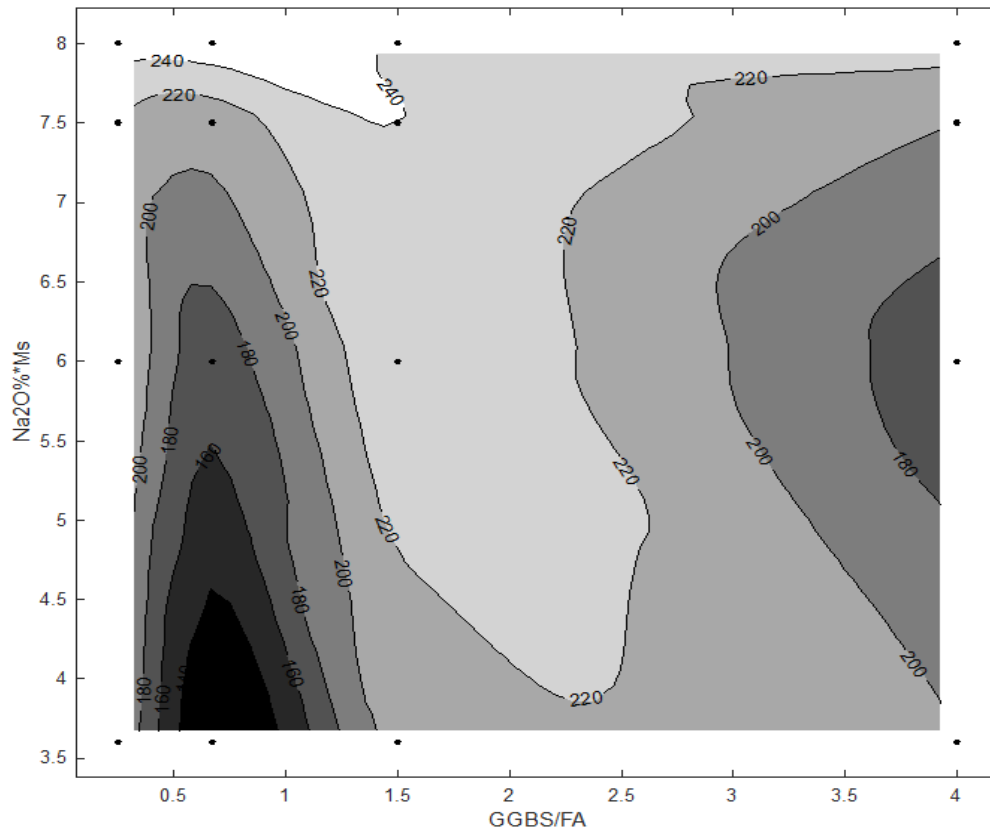
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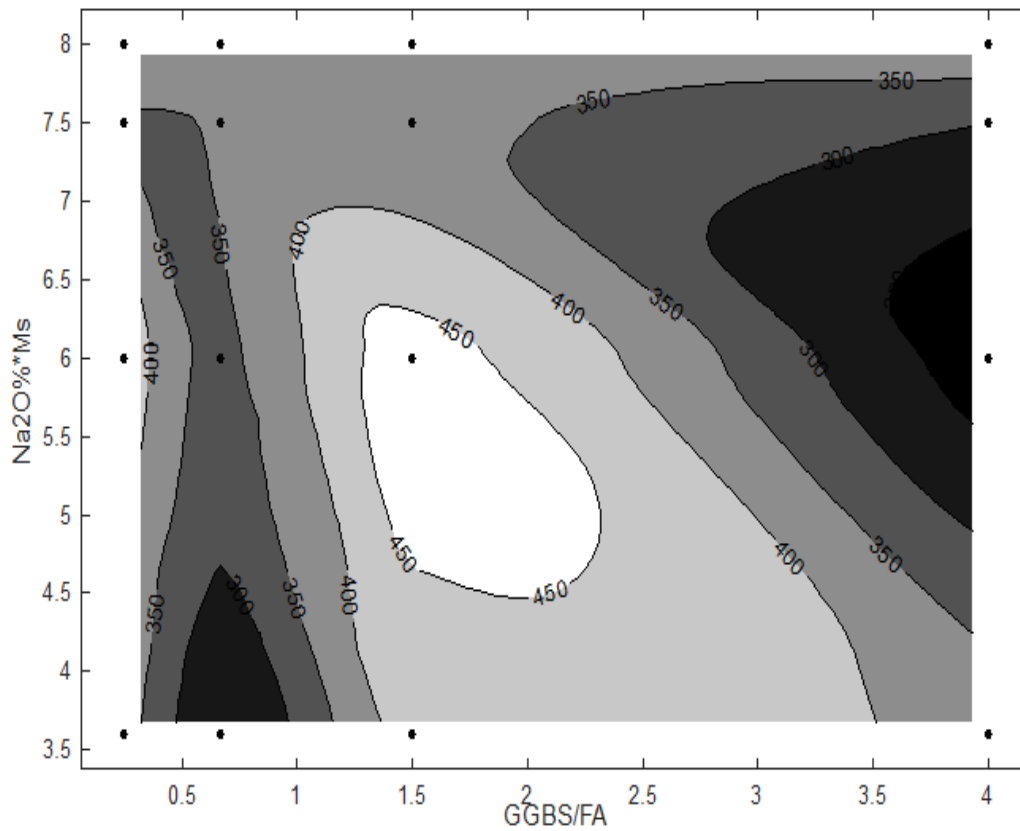
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Fig. 2-Flow results of AA-GGBS/FA concretes



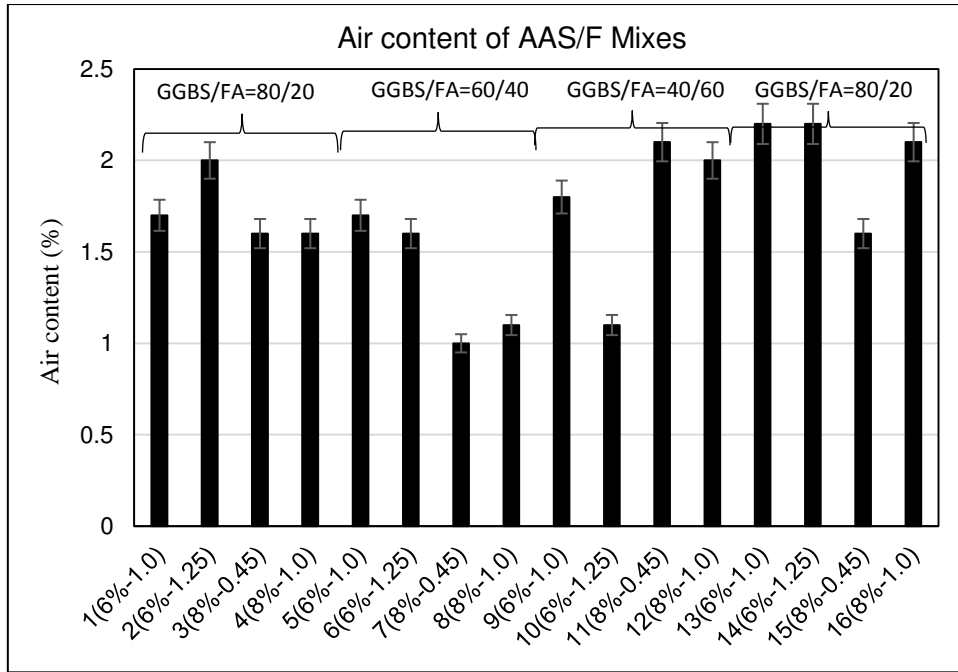
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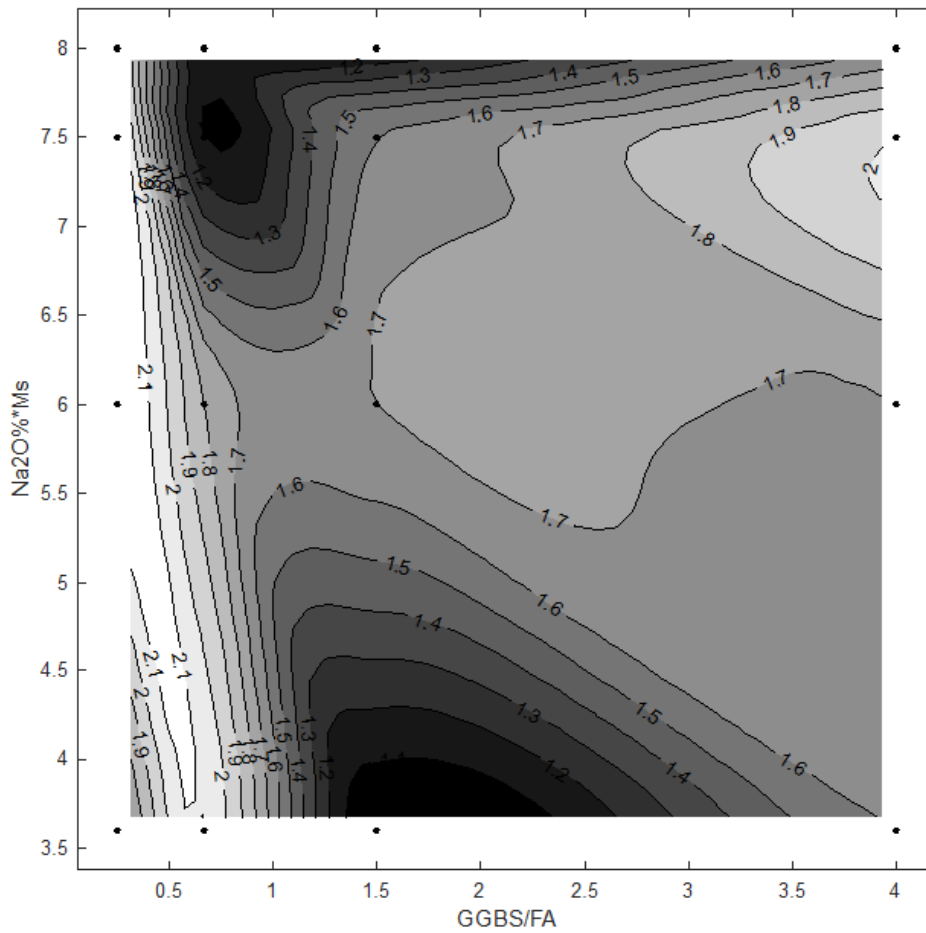
Fig. 3-Contour graph for workability of AA-GGBS/FA concretes - (a) Slump test, (b) Flow test



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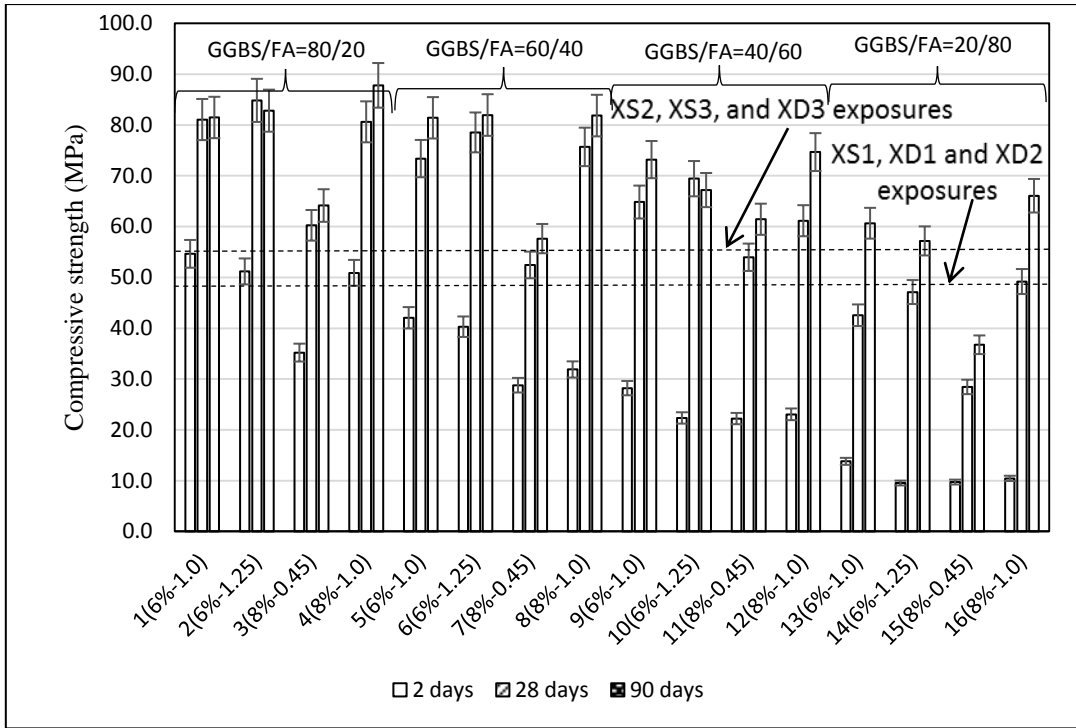
Fig. 4-Air content in fresh mixes of AA-GGBS/FA concretes



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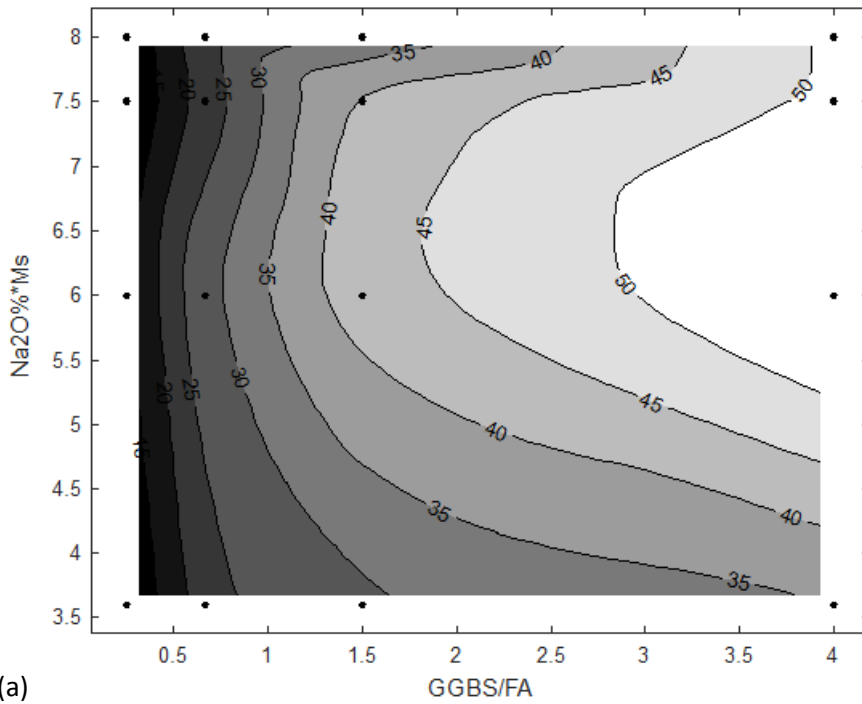
Fig. 5-Contour graph for air content in fresh mixes of AA-GGBS/FA concretes



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Fig. 6-Compressive strength of AA-GGBS/FA concretes mixes at 2, 28 and 90days



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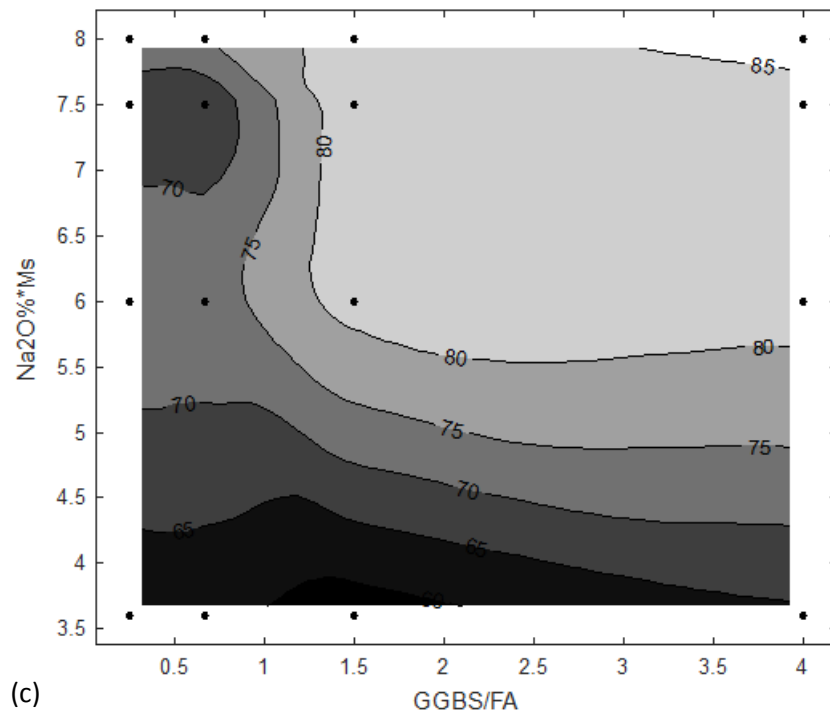
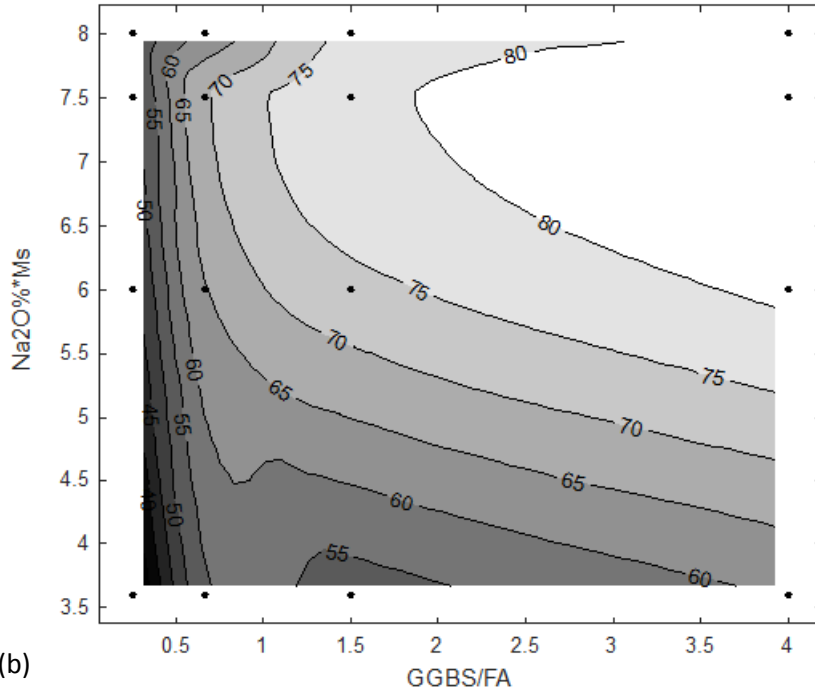
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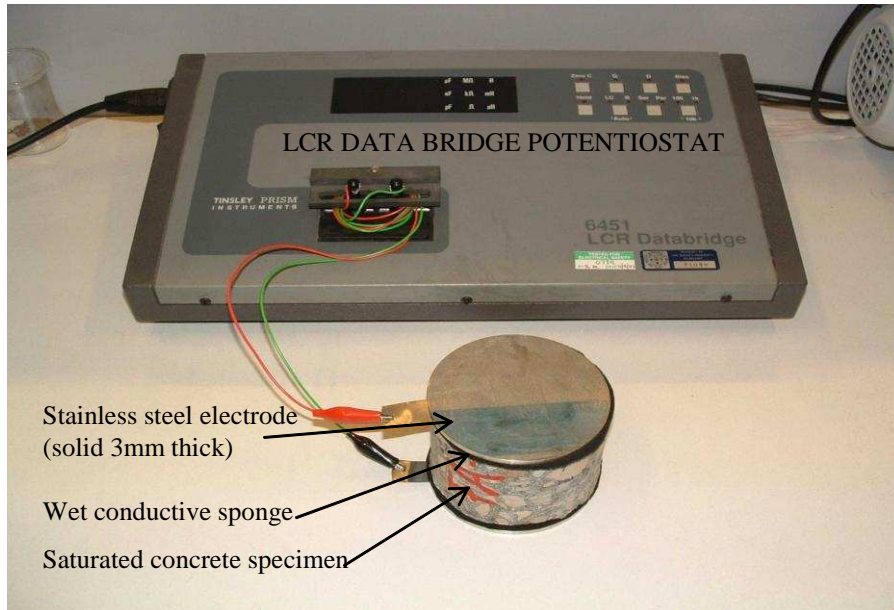
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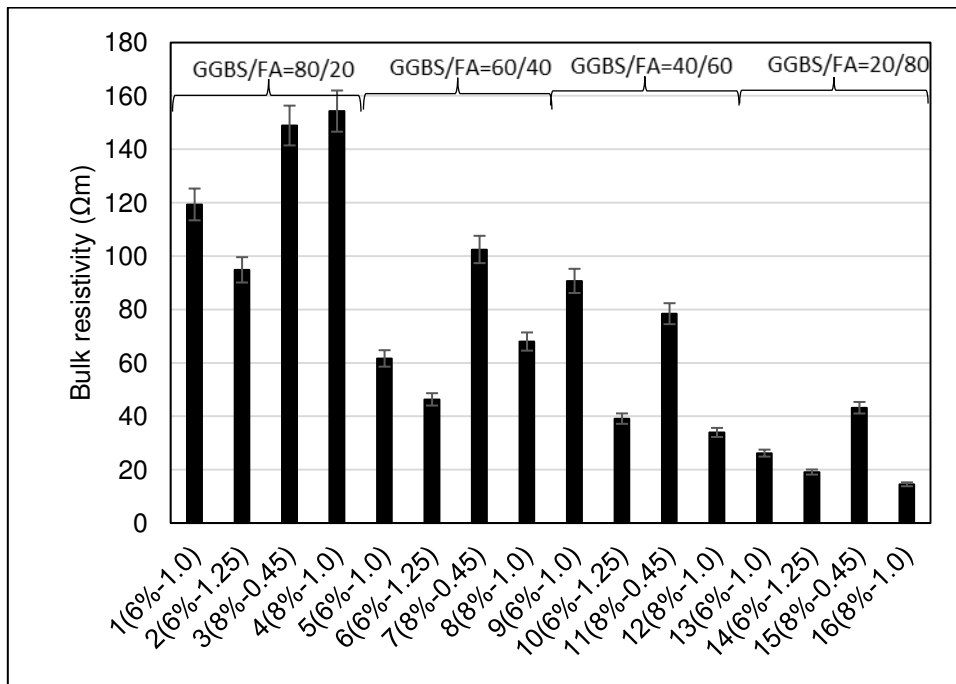
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Fig. 7-Contour graph for compressive strength of AA-GGBS/FA concretes
(a) 2days, (b) 28days, (c) 90days



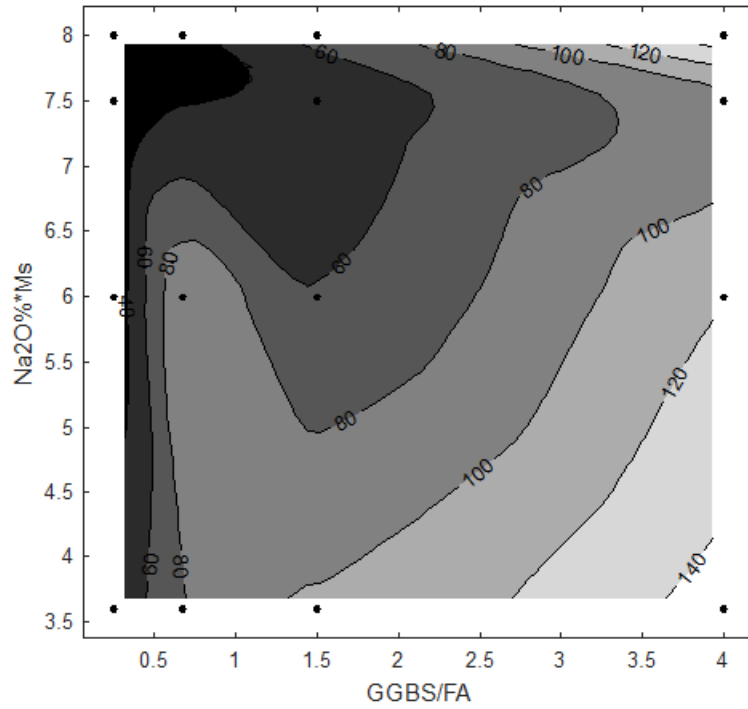
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Fig. 8-Bulk resistivity test equipment and sample



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Fig. 9-Bulk resistivity



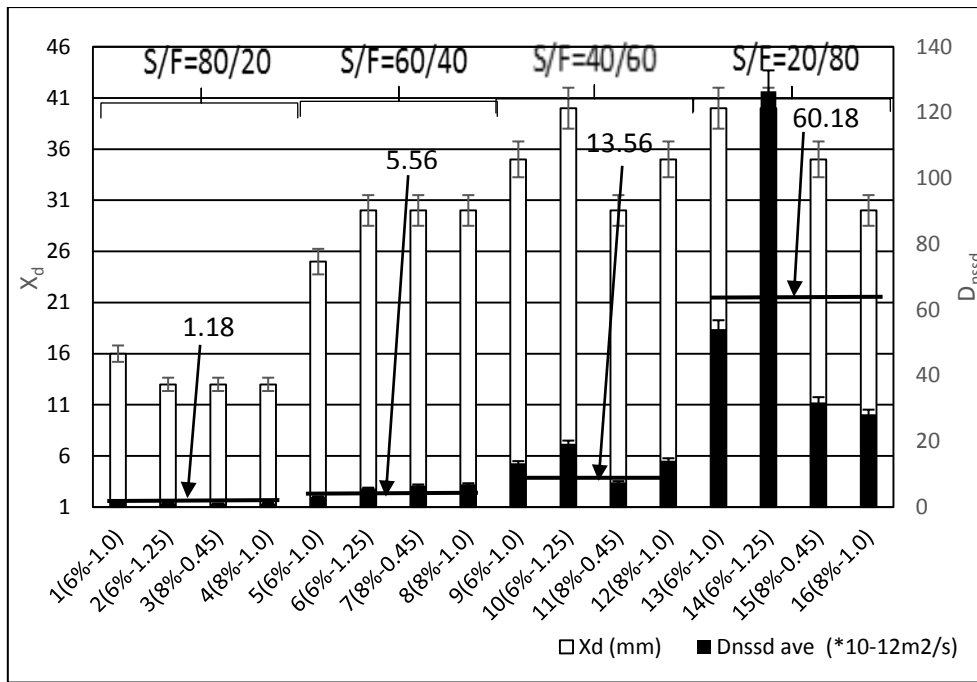
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Fig. 10-Contour graph for bulk electrical resistivity ($\Omega.m$) of AA-GGBS/FA concretes



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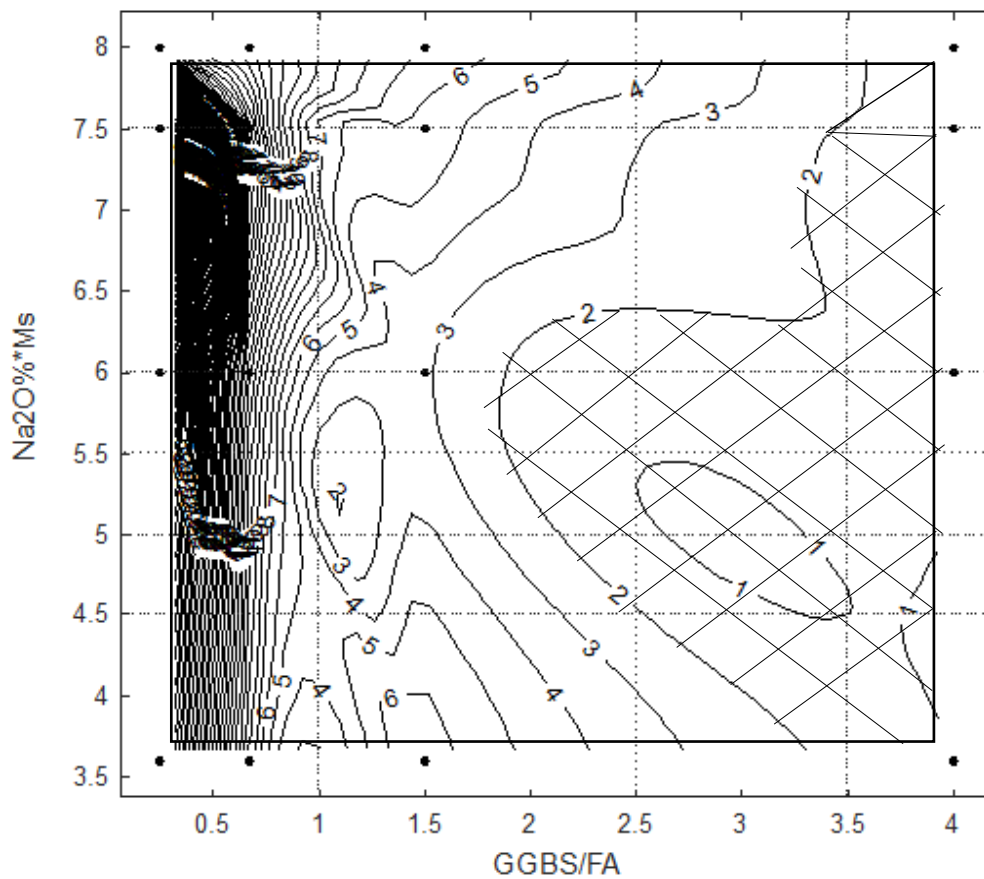
Fig. 11-Concrete sample disks with all surface painted except the top face, placed in air tight container with sodium chloride solution for diffusion test as per NT Build 443



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Fig. 12-Chloride penetration depths (X_d) and D_{nssd} of AA-GGBS/FA concretes

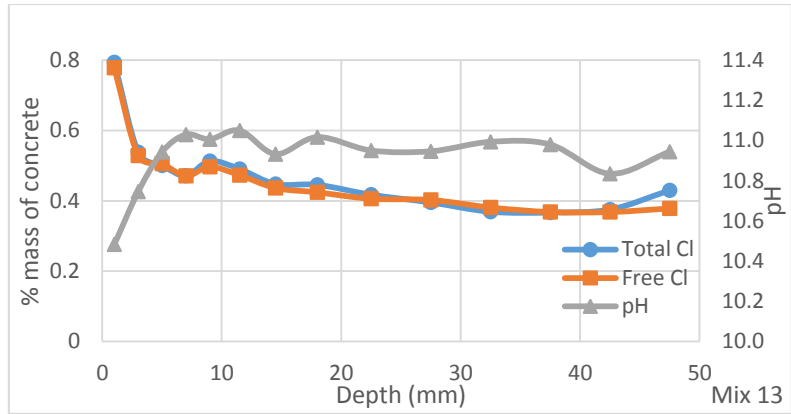


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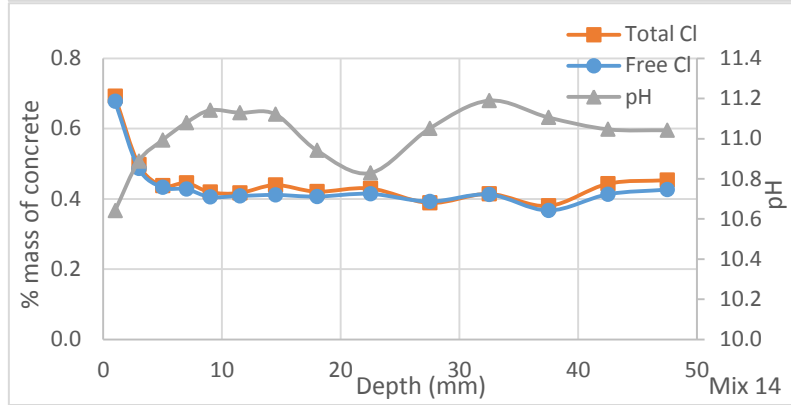
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Fig. 13-Contour graph for D_{nssd} ($\times 10^{-12} m^2/s$) of AA-GGBS/FA concretes

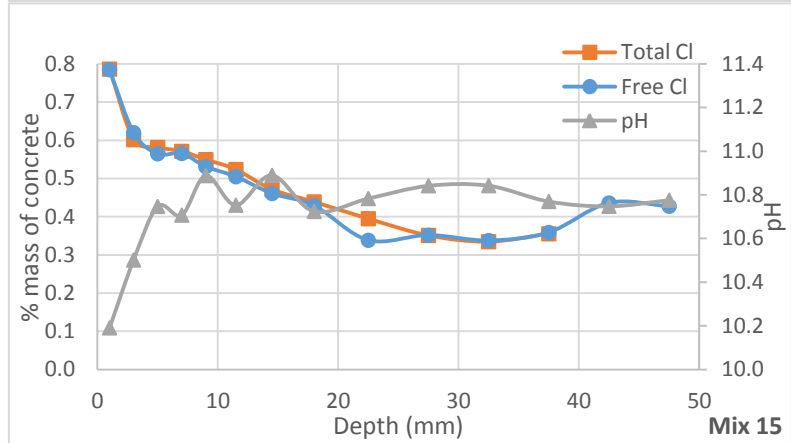
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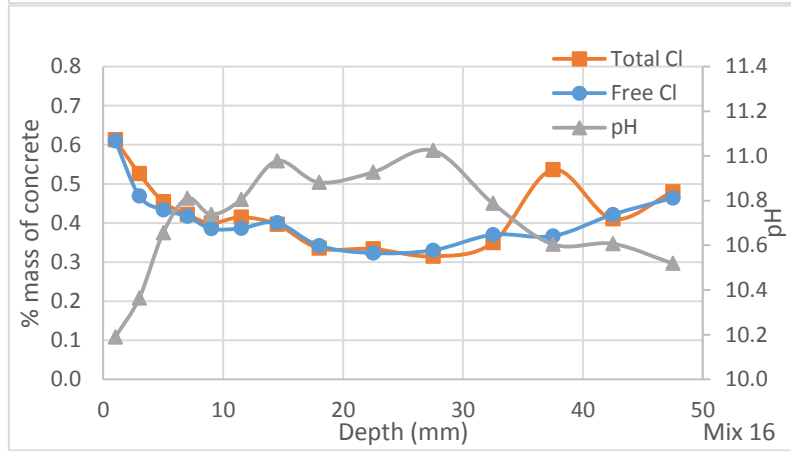
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Figure 14 Total and water-soluble chloride concentration and pH for concrete dust samples collected for NT Build 443 test for mixes 13, 14, 15 and 16

Highlights

- AA-GGBS/FA concrete cured at 20°C results in desirable fresh properties and strength.
- A higher proportion of GGBS results in high early and long-term strength.
- AA-GGBS/FA concretes with GGBS/FA < 4 are non-viable for protection of steel in chloride environments.
- There is no chloride binding capacity in the fly ash dominated AA-GGBS/FA concretes.
- The chloride diffusivity in AA-GGBS/FA concretes depends mostly on the pore structure.

Supplementary MATLAB .fig files

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