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Figure 1: X65 carbon steel microstructure



Figure 2: Schematic of autoclave set-up for evaluation of X65 carbon steel corrosion in CO₂-containing environments at elevated temperature



Figure 3: (a) Mass loss measurements and the corresponding calculated corrosion rates in mm/year and (b) corrosion product mass determined from mass gain measurements as a function of time for X65 carbon steel exposed to a CO₂-saturated 3 wt.% NaCl solution at 80 and 150°C



(a)

(b)



(c)





(e)

Figure 4: Top view SEM images of X65 carbon steel after exposure to a CO₂-saturated 3 wt.% NaCl solution after (a) 6 h at 80°C; (b) 96 h at 80°C; (c) 96 h at 80°C at higher magnification; (d) 6 h at 150°C and (e) 96 h at 150°C









(a)

(b)

Figure 6: XRD patterns collected from X65 carbon steel surfaces exposed to a CO₂saturated 3 wt.% NaCl solution for different time periods between 6 h and 96 h at (a) 80°C and (b) 150°C



Figure 7: (a) Mass loss measurements and the corresponding calculated corrosion rates in mm year⁻¹ and (b) corrosion product mass determined from mass gain measurements as a function of time for X65 carbon steel exposed to a CO₂-saturated 1.54 wt.% NaCl solution with 1.83 wt.% CaCl₂·2H₂O at 80 and 150°C





(b)



(c)





(e)



(f)

(g)

Figure 8: Top view SEM images of X65 carbon steel after exposure to a CO₂-saturated 1.54 wt.% NaCl solution with 1.83 wt.% CaCl₂·2H₂O after (a) 6 h at 80°C; (b) 96 h at 80°C; (c) 96 h at 80°C at higher magnification (d) FIB-SEM 96 h at 80°C; (e) FIB 96 h at 80°C; (f) 6 h at 150°C and (g) 96 h at 150°C





Figure 9: Cross-section SEM images of X65 carbon steel after exposure to a CO₂-saturated 1.54 wt.% NaCl solution with 1.83 wt.% CaCl₂·2H₂O after 96 h at (a) 80°C and (b) 150°C

Figure 10: XRD patterns collected from X65 carbon steel surfaces exposed to a CO₂saturated 1.54 wt.% NaCl solution with 1.83 wt.% CaCl₂·2H₂O for different time periods between 6 h and 96 h at (a) 80°C and (b) 150°C



Figure 11: XRD patterns for X65 steel samples exposed to a CO₂-saturated solution containing either 3 wt.% NaCl or 1.54 wt.% NaCl with 1.83 wt.% CaCl₂·2H₂O at 80°C after 24, 48 and 96 h of exposure



Figure 12: FIB section and EDX line scan of cross-section of X65 carbon steel after exposure to a CO₂-saturated 1.54 wt. % NaCl solution with 1.83 wt.% CaCl₂·2H₂O after (a) 96 h at 80°C and (b) 96 h at 150°C

No Ca2+

With Ca2+



Figure 13: Uniform thickness loss (calculated from mass loss measurements), average of the top 10 pit depths relative to the corroded steel surface (from profilometry analysis) and total penetration depth (summation of the two aforementioned measurements) on X65 carbon steel as a function of time for a CO₂-saturated solution containing (a) 3 wt.% NaCl at 80°C; (b) 1.54 wt.% NaCl and 1.83 wt.% CaCl₂·2H₂O at 80°C; (c) 3 wt.% NaCl at 150°C; (d) 1.54 wt.% NaCl and 1.83 wt.% CaCl₂·2H₂O at 150°C



Figure 14: Surface profilometry images of carbon steel surfaces after exposure to a CO₂saturated solution for 96 h at 80°C containing (a) 3 wt.% NaCl and (b) 1.54 wt.% NaCl and 1.83 wt.% CaCl₂·2H₂O. Images are acquired after removal of the corrosion product layer



Figure 15: Surface profilometry images of carbon steel surfaces after exposure to a CO₂saturated solution for 96 h at 150°C containing (a) 3 wt.% NaCl and (b) 1.54 wt.% NaCl and 1.83 wt.% CaCl₂·2H₂O. Images are acquired after removal of the corrosion product layer