Article:
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Supporting Information For

Arsenite and Arsenate Binding to Ferrihydrite Organo-Mineral Coprecipitate: Implications for Arsenic Mobility and Fate in Natural Environments

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This file includes 5 Figures

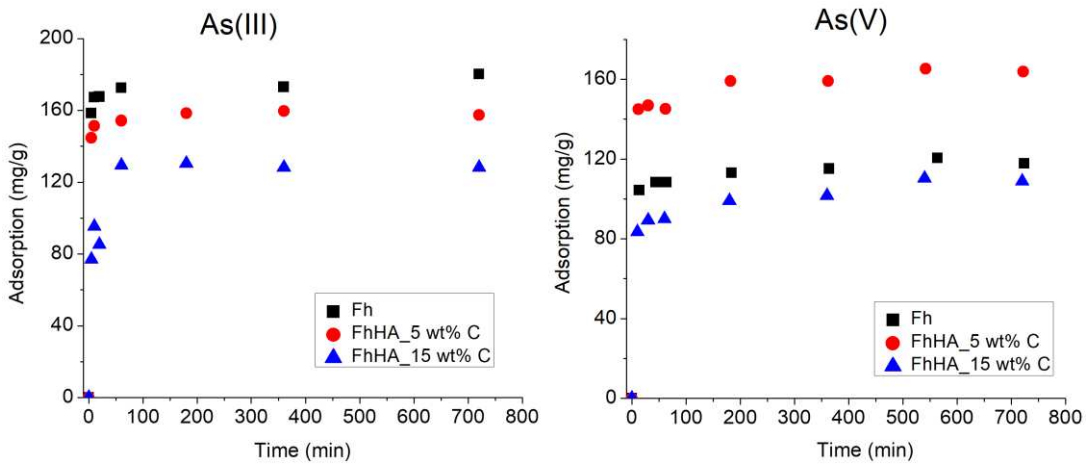


Figure S1. Arsenite and arsenate adsorption kinetics on ferrihydrite and ferrihydrite-HA coprecipitates at pH 7. Experimental solution contains 0.1 g/L adsorbent and ~0.4 mmol/L As(III) or As(V) in 0.01 M NaCl electrolyte.

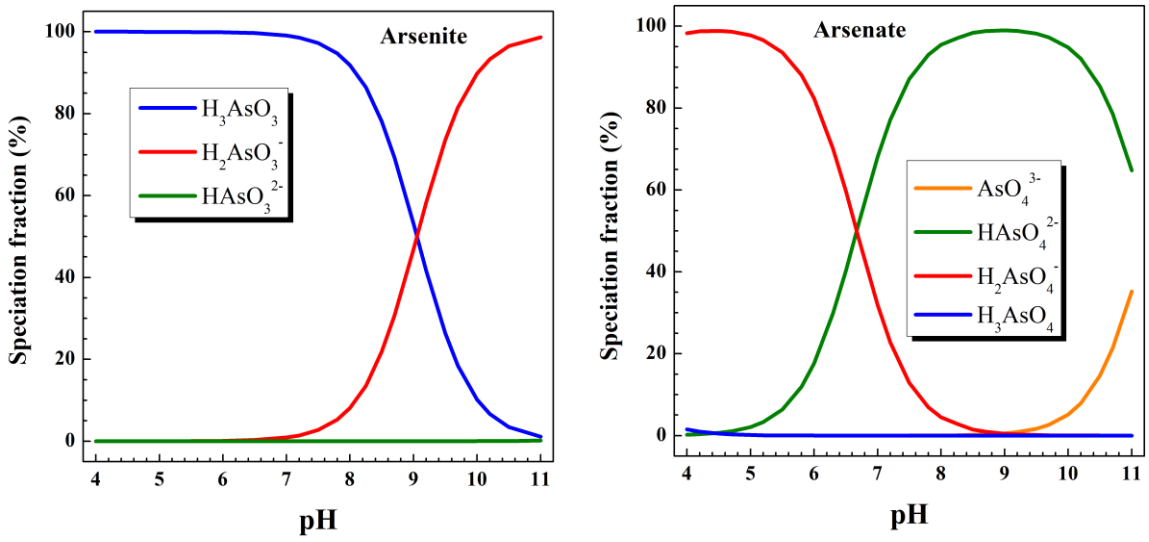


Figure S2. Arsenite and arsenate speciation in aqueous solution calculated using the software Visual Minteq (ver. 3.0). Experimental solution contains ~0.4 mmol/L As(III) or As(V) in 0.01 M NaCl electrolyte.

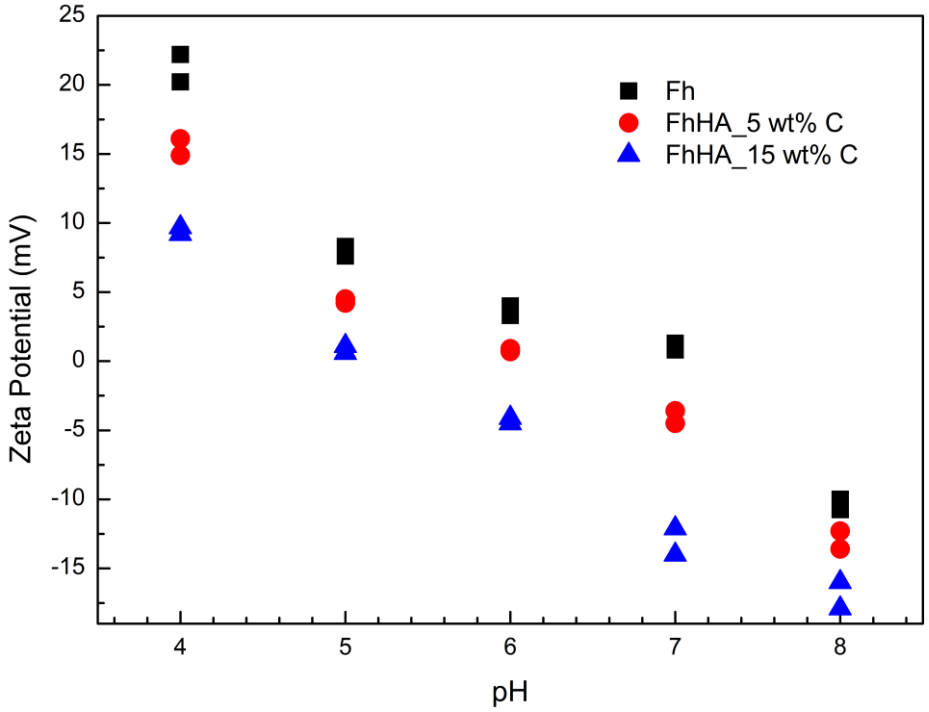
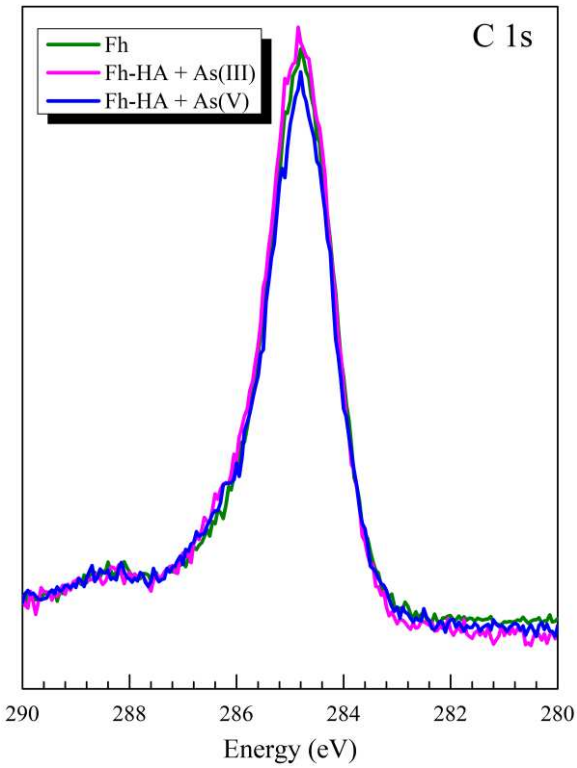


Figure S3. Zeta potentials of ferrihydrite and ferrihydrite-HA coprecipitates in the experimental adsorption solution.



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47 **Figure S4**. C 1s XPS spectra of ferrihydrite-HA composite before and after the binding

48 of As(III) and As(V) at pH 7.

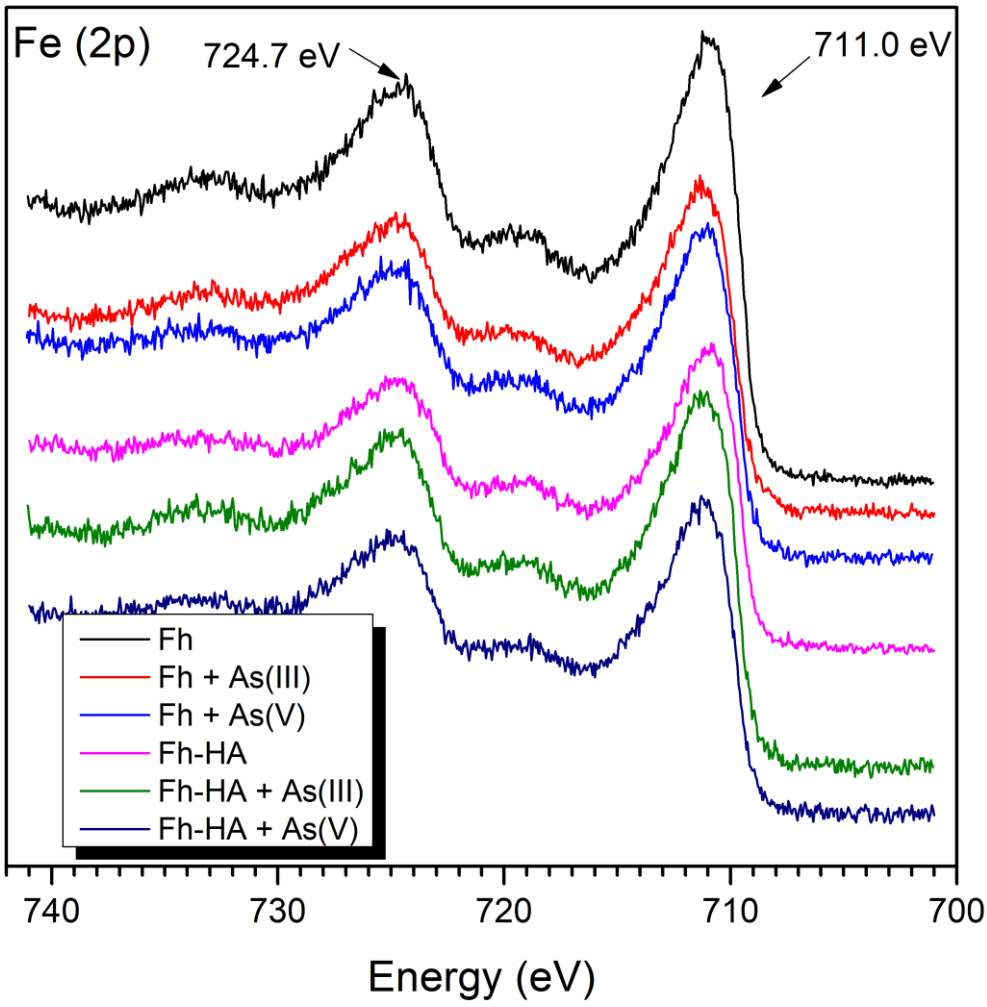

Figure S5. Fe 2p XPS spectra of ferrihydrite and ferrihydrite-HA composite before and after the binding of As(III) and As(V) at pH 7.