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1 Supporting Information For

3	Arsenite and Arsenate Binding to Ferrihydrite Organo-Mineral
4	Coprecipitate: Implications for Arsenic Mobility and Fate in Natural
5	Environments
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22	This file includes 5 Figures
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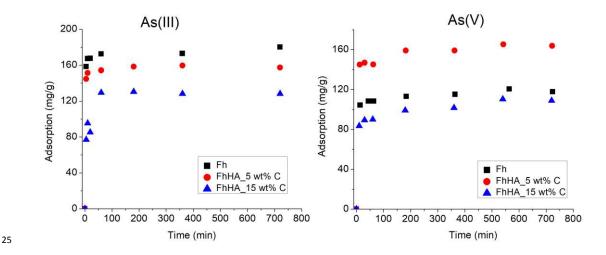


Figure S1. Arsenite and arsenate adsorption kinetics on ferrihydrite and ferrihydriteHA coprecipitates at pH 7. Experimental solution contains 0.1 g/L adsorbent and ~0.4
mmol/L As(III) or As(V) in 0.01 M NaCl electrolyte.



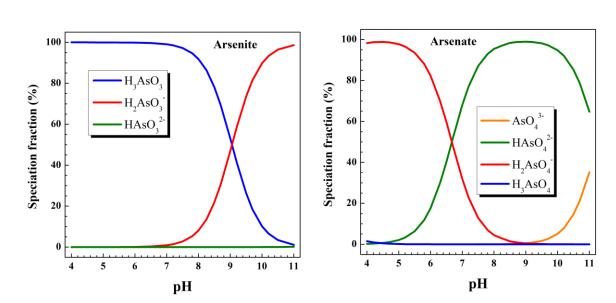


Figure S2. Arsenite and arsenate speciation in aqueous solution calculated using the
 software Visual Minteq (ver. 3.0). Experimental solution contains ~0.4 mmol/L As(III)
 or As(V) in 0.01 M NaCl electrolyte.

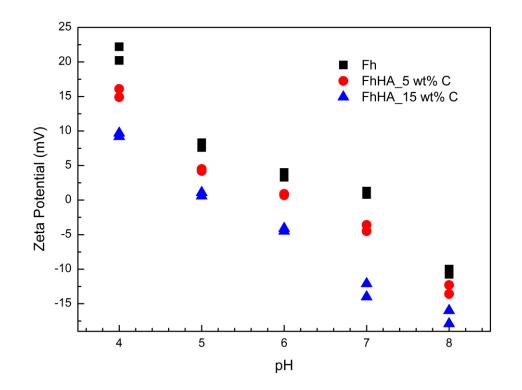


Figure S3. Zeta potentials of ferrihydrite and ferrihydrite-HA coprecipitates in the
 experimental adsorption solution.

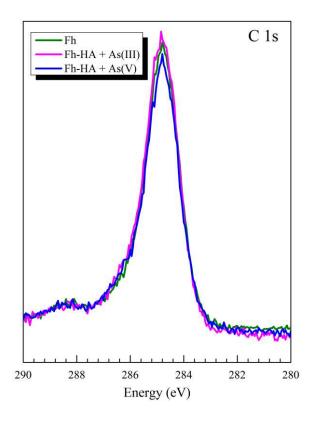
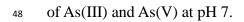


Figure S4. C 1s XPS spectra of ferrihydrite-HA composite before and after the binding



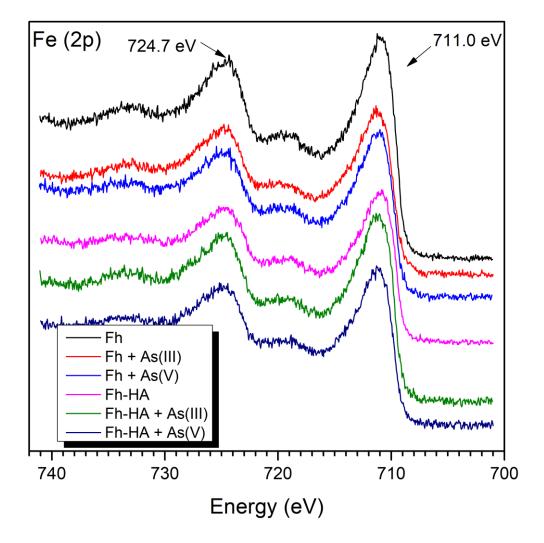


Figure S5. Fe 2p XPS spectra of ferrihydrite and ferrihydrite-HA composite before and

s1 after the binding of As(III) and As(V) at pH 7.

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