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1	Alkaline magmas in zones of continental convergence:
2	The Tezhsar volcano-intrusive ring complex, Armenia
3	
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### 24 Abstract

25 Alkaline igneous rocks are relatively rare in settings of tectonic convergence and little 26 is known about their petrogenesis in these settings. This study aims to contribute to a 27 better understanding of the formation of alkaline igneous rocks by an investigation of the 28 Tezhsar volcano-intrusive alkaline ring complex (TAC) in the Armenian Lesser Caucasus, 29 which is located between the converging Eurasian and Arabian plates. We present new 30 petrological, geochemical and Sr-Nd isotope data for the TAC to constrain magma genesis and magma source characteristics. Moreover, we provide a new  $^{40}$ Ar/ $^{39}$ Ar age of 41.0±0.5 31 32 Ma on amphibole from a nepheline syenite that is integrated into the regional context of 33 ongoing regional convergence and widespread magmatism.

The TAC is spatially concentric and measures ~10 km in diameter representing the relatively shallow plumbing system of a major stratovolcano juxtaposed by ring faulting with its extrusive products. The plutonic units comprise syenites and nepheline syenites, whereas the extrusive units are dominated by trachytic-phonolitic rocks. The characteristic feature of the TAC is the development of pseudomorphs after leucite in all types of the volcanic, subvolcanic and intrusive alkaline rocks.

40 Whole-rock major element data show a metaluminous (Alkalinity Index = 0-0.1), alkalic and silica-undersaturated (Feldspathoid Silica-Saturation Index <0) character of 41 42 the TAC. The general trace element enrichment and strong fractionation of REEs (La<sub>N</sub>/Yb<sub>N</sub> 43 up to 70) indicate a relatively enriched magma source and small degrees of partial melting. All TAC rocks show a negative Nb-Ta anomalies typical of subduction zone settings. The 44 initial <sup>87</sup>Sr/<sup>86</sup>Sr ratios (0.704-0.705) and positive ɛNd values (+3 to +5) indicate an 45 isotopically depleted upper mantle and lack of significant crustal influence, which in turn 46 47 suggests the TAC magma has formed via differentiation from lithospheric mantle melts.

Regionally, the age of ~41 Ma places the TAC amid a Lesser Caucasian Eocene period of dominantly calc-alkaline magmatism. The TAC's arc-like geochemical signatures are interpreted to result from prior subduction of the Tethyan slab beneath the Eurasian continental margin. The alkaline character, distinct from regional trends, is attributed to Neotethyan slab rollback causing extension and inducing small degrees of decompression melting of metasomatised lithospheric mantle.

- 54
- 55 **Keywords:** Alkaline igneous rocks, ring complex, Armenia, geochemistry, <sup>40</sup>Ar/<sup>39</sup>Ar
- 56 dating, pseudoleucite

## 57 **1. Introduction**

58 Studies of alkaline magmatism on the global scale have become a point of focus due 59 to the significant role of alkaline magmatic rocks for ore exploration, in particular 60 regarding prospecting for rare earth elements (REEs), niobium (Nb), tantalum (Ta) and 61 zirconium (Zr) (e.g. Chakhmouradian and Zaitsev 2012). Many alkaline igneous rocks are 62 found in rift-related intraplate settings (e.g. Gardar Province/Greenland, Upton et al. 63 2003; Kola Alkaline Province/Russia, Downes et al. 2005; East African Rift, Woolley 64 2001), but they also occur, albeit less frequently, in settings of plate convergence (Burke 65 and Khan, 2006; Hou et al. 2006). Plate convergence includes collisional events that cause the welding of terranes into continental land and subsequent post-collisional episodes in 66 67 which convergence continues (Bonin et al. 1998). The occurrence of magmas with alkaline 68 affinities becomes more common only when the geodynamic context becomes entirely 69 intraplate in a post-orogenic episode (Bonin et al. 1998). In complex collisional and post-70 collisional settings, the timing of specific types of magmatism depends on the geotectonic 71 geometries and the relative rates of crustal thickening and subsidiary subduction (Harris 72 et al. 1986). Importantly, convergent movement between colliding plates will continue for 73 30-50 Ma after the initial collision (Harris et al. 1986). On a global scale, deformed alkaline 74 rocks and carbonatites (DARCs) may be used as indicators of where ancient oceans have 75 opened and closed, and the presence of a variety of syenites, carbonatites and other 76 alkaline igneous rocks found in proximity to older DARCs indicate the recycling of 77 material from the underlying lithosphere based on the Wilson Cycle-type model (Burke 78 and Khan 2006). Thus, investigating alkaline magmatism in convergent settings, e.g. in 79 Tibet (Williams et al. 2004; Hou et al. 2006) and the Anatolian-Armenian-Iranian plateau 80 (Jackson et al. 1995; Neill et al. 2015), has become as important as studies of rift-related 81 settings to understand alkaline magma genesis.

82	The exact mechanisms responsible for magma generation in collisional tectonic
83	settings remain enigmatic. Models include slab break-off (Keskin 2003; van Hunen and
84	Allen 2011; Neill et al. 2015), large-scale delamination or thinning of the lithospheric
85	mantle (Innocenti et al. 1982; Pearce et al. 1990) and small-scale lithospheric detachment
86	driven by convection cells (Kaislaniemi et al. 2014; Neill et al. 2015). Moreover, the source
87	of magmas in compressional regimes and their chemical impact on the crust remains
88	disputed. Processes to generate primary magmas in collision zones may involve melting
89	of thickened lithosphere due to breakdown of hydrous phases at the continental suture
90	(Allen et al. 2013) and melting of deeply-subducted continental crust (Zhao et al. 2013).
91	To explain the alkaline character of the erupted or plutonic igneous rocks, several genetic
92	models and processes have been proposed:
93	1. Low degrees of partial melting of metasomatized upper mantle (Bodeving et al.
94	2017; Dawson 1987; Marks et al. 2008).
95	2. Melting of crustal sources, which could be located in the lower crust and mafic in
96	composition (Smith et al. 1988) or in the middle to upper crust and felsic in
97	composition (Downes 1987; Fitton 1987).
98	3. Fractional crystallization from alkali basalt parental magmas (Delong et al. 1975;
99	Trumbull et al. 2003), with variable degrees of crustal assimilation (Fitton 1987;
100	Jung et al. 2007; Lan et al. 2011).
101	4. Fenitisation – a high temperature metasomatic alteration driven by alkali-rich
102	fluids incrementally expelled from alkaline or carbonatitic melts (Sindern and
103	Kramm 2000; Suikkanen and Rämö 2017).
104	
105	Armenia, landlocked between the Black Sea and the Caspian Sea, forms part of the

106 Anatolian-Armenian-Iranian Plateau and is characterised by widespread Cenozoic

107 volcano-magmatic activity, starting in the Eocene at  $\sim$ 50 Ma and intermittently lasting 108 into the Holocene and historical times (Karakhanian et al. 2002; Moritz et al. 2016; Fig. 109 1a). Several studies focused on Quaternary volcanic cones on the Anatolian-Armenian-110 Iranian plateau (Innocenti et al. 1982; Pearce et al. 1990; Keskin et al. 1998), including in 111 the Armenian segments of the Lesser Caucasus mountain range (Karapetian et al. 2001; 112 Karakhanian et al. 2002), and the Miocene/Pliocene magmatic evolution of the region 113 (Dilek et al. 2010; Neill et al. 2013; Kheirkhah et al. 2015). However, investigating the 114 much less studied Paleogene igneous rocks is important to gain a more complete 115 understanding of the long-term magmatic and geodynamic evolution in this setting of 116 continuing convergence and to improve our understanding of collision-driven continental 117 magmatism and mantle dynamics (Dilek et al. 2010; van Hunen and Allen 2011; Moritz et 118 al. 2016).

119 In this study, we use a range of petrological and geochemical methods to describe and 120 interpret the lithological variations of the Tezhsar volcano-intrusive alkaline ring 121 complex (or Tezhsar Alkaline Complex - TAC) in Armenia. We provide a new <sup>40</sup>Ar/<sup>39</sup>Ar 122 age and expand on previous petrological and geochemical studies (Abovyan et al. 1981; 123 Kogarko et al. 1995; Meliksetian 1971, 1989) with the aim to achieve a better 124 understanding of TAC petrogenesis and to integrate that into a model of alkaline magma 125 genesis within a setting of continuing plate convergence. We also highlight and discuss 126 the occurrence of cm-sized pseudoleucites in the TAC.

127

128 2. Geological history

### 129 2.1 Regional tectonic setting

130 The TAC, located about 55 km north of Yerevan in the Lesser Caucasus, has formed in131 the Eocene in a setting of general convergence between the Eurasian and Arabian plates

(Fig. 1a). This region was affected by two distinct collisional events and the emplacement
of the alkaline magmas of the TAC is crucial to the understanding of the tectono-magmatic
evolution of the region.

135 The TAC is located on basement of the South Armenian Block (SAB), which is a 136 microplate of Gondwanaland origin (Knipper and Khain 1980; Rolland 2017; Sosson et al. 137 2010). Proterozoic metamorphic basement of the SAB is exposed in the Tsakhkunyats 138 massif (Belov 1968; Aghamalyan 1998). Platform sedimentary cover of the SAB is 139 presented by folded Late Devonian to the Late Triassic sedimentary formations 140 (Arakelyan 1964; Aslanyan 1958). Ophiolites representing Jurassic oceanic crust were 141 obducted onto the northern margin of the SAB in the Late Cretaceous (90-84 Ma; 142 Rolland 2017). In the late Cretaceous to early Palaeogene (70-60 Ma), the SAB was 143 welded to the southern margin of Eurasia as a result of the closure of the northern branch 144 of the Neotethys and the termination of subduction (Rolland et al. 2009a, b; Moritz et al. 145 2016). The collision is marked by the Sevan-Akera suture zone, which is part of the 146 regional northern Neotethys suture (Hässig et al. 2013; Sosson et al. 2010). The closure 147 of the northern Neotethys branch caused a subduction jump towards the south and the 148 accretion of the SAB to the Eurasian margin resulted in formation of a Cretaceous-Eocene 149 flysch basin that overlies the ophiolites (Rolland 2017). At present, the Sevan-Akera 150 suture separates two tectonostratigraphic units, the Southern and Northern Tethyan 151 Provinces, which outline the continental provinces pre-dating the closure of the Tethys 152 Ocean (Fig. 1b; Adamia et al. 2011). The Sevan-Akera suture is located ~6 km northward 153 of the TAC. The second stage of accretion involving collision of the Arabian margin to the 154 SAB and the Tauride-Anatolian block caused the closure of the South Neotethys ocean 155 along the Bitlis-Zagros suture. This closure occurred in late Eocene to early Oligocene 156 times (40-25 Ma) based on geochronological and structural evidence (Agard et al. 2005; 157 Allen and Armstrong 2008; Rolland 2017). The convergence and collision between Arabia 158 and Eurasia induced regional compression and shortening in the overriding (SAB-159 Eurasia) continental lithosphere (Agard et al. 2011), the formation of the Anatolian-160 Armenian-Iranian orogenic plateau (Sheth et al. 2015) and lateral ejection of the 161 Anatolian and Iranian blocks, with the Armenian Highland (Lesser Caucasus and Eastern 162 Anatolia) in the centre (Phillip et al. 1989). Protracted Cenozoic magmatism lasted from 163 ~49 Ma to ~21 Ma and marked the final stages of the Neothethyan subduction, the main 164 Arabia-Eurasia collisions and subsequent post-collisional events, including emplacement 165 of the syn-collisional granite-leucogranite plutons of the Lesser Caucasus (Meliksetian 166 1989; Rezeau et al. 2017).

167 To explain the Palaeogene magmatism of the entire region, Dilek et al. (2010) proposed the opening of an asthenospheric window beneath the arc mantle wedge and 168 169 the collision zone. The presence of adakites of Early Eocene age in the Pontides 170 interpreted as a result of slab window formation (Eyuboglu et al. 2011) supports this 171 hypothesis. Lordkipanidze et al. (1989) and Sahakyan et al. (2016) consider a subduction-172 modified upper mantle source for Lower-Middle Eocene volcanism and an increase of 173 crustal input within the Late Eocene-Early Oligocene magmatic series of the Lesser 174 Caucasus.

Considering the age and location of the TAC (<sup>40</sup>Ar/<sup>39</sup>Ar of 41.0±0.5 Ma, this study; 36.3-37.5 Ma, K-Ar, Baghdasaryan and Ghukasyan 1985; 36-39 Ma, K-Ar, Meliksetian 177 1989), it formed in a plate convergence setting, in between two major collisional events that occurred in region – first at the northern edge of the SAB in the Late Cretaceous to Early Paleogene, and subsequently to the south of the SAB in the Late Eocene to Early Oligocene. The TAC can thus be described as post-collisional relative to the initial collisional event between the SAB and the Eurasian plate. 182

## 183 **2.2 Geological setting of the Tezhsar Alkaline Complex**

184 The TAC is located on the Pambak ridge at the northern edge of the SAB within the 185 Sevan-Shirak basin. To the south, the TAC is in contact with the Proterozoic metamorphic 186 basement of the SAB across the Marmarik Fault. Presence of abundant xenoliths from the 187 Tsakhkunyats basement, such as mica schists, confirms the affinity of the TAC to the SAB 188 continental terrane. To the north, the TAC borders the Margahovit intrusion comprising porphyritic granosyenites. Country rocks exposed to the W-NW of the TAC comprise 189 190 Upper Cretaceous clastic and carbonate strata and Mid-to-Late Eocene extrusive igneous rocks, which also outcrop to the E-SE. The Ulashik Fault cuts the TAC in SW-NE direction 191 192 with horizontal left-lateral displacement of intrusive and volcanic units reaching 700 m.

The TAC represents a ring complex that can be subdivided into several concentric units of both volcanic and plutonic rocks. Such classical ring complexes are quite rare (Johnson et al., 1999) and are of special interest considering their structural and volcanological evolution as well as petrological aspects. According to Meliksetian (1971), the TAC includes the following major units (Fig. 2):

198 1. Outer cone sheets characterized by inward-dipping contacts

Ring unit of volcanic alkaline rocks with a thickness up to 600 m characterised by
 its concentric structure and inward-dipping contacts (Outer Volcanic Unit, OVU)

- 201 3. Central intrusive unit comprising syenites and nepheline syenites (Syenitic Unit,
  202 SYU)
- 203 4. Ring dykes, circular bodies with sub-vertical contacts cutting both the volcanic and
  204 central intrusive units
- 205 5. Resurgent volcanic unit, inside the central intrusive unit, formed by volcanic
  206 breccias, dykes and subvolcanic rocks (Central Volcanic Unit, CVU).

207 For the purpose of the geochemical investigation in this study, we use a simplified 208 subdivision into Outer Volcanic Unit, Syenitic Unit and Central Volcanic Unit (Fig. 2). 209 Based on a structural analysis including bedding attitudes of units and relationships 210 between volcanic ring, cone sheets and central pluton, the presence of circular dykes and 211 remains of a volcanic centre, most researchers, namely Kotlyar (1958), Bagdasaryan 212 (1966) and Meliksetian (1971) concluded that the TAC formed via a caldera collapse and 213 the volcanic ring was emplaced through collapse along concentric faults. The 214 exceptionally large elliptical palaeocaldera structure of the TAC is ~13.6×11.5 km in size 215 and has an area of  $\sim$ 131 km<sup>2</sup>, comparable in dimensions to the Santorini caldera in the 216 Aegean Sea. Such a ring morphology provides a unique insight into the roots of an 217 alkaline volcano-plutonic complex.

218 Beyond the petrogenetic and structural significance of the TAC, there is also a 219 characteristic widespread development of pseudomorphs after leucite, which have been 220 studied in detail by B. Meliksetian (1970, 1971, 1979, 1989) and Yagi and Gupta (1978). 221 They feature in volcanic, subvolcanic and intrusive alkaline rocks and the largest crystals, 222 reaching up to 8 cm in size (Fig. 3) are found in porphyry tinguaite dykes (Meliksetian 223 1978; Yagi and Gupta 1978). Their crystallographic habit is either icositetrahedral (in 224 volcanic rocks and dykes) or triakis octahedral (in intrusive syenites). In the Soviet 225 petrological literature according to Zavaricky (1934), pseudomorphism after leucite is 226 divided into two mineralogical and genetic types: "Pseudoleucites" referring to leucite 227 breakdown into nepheline and orthoclase, and "epileucites" describing pseudomorphism 228 after leucite composed of agglomerated orthoclase, muscovite, analcime, chlorite, calcite 229 and zeolites. In the Western petrological literature, usually both types are referred to as 230 pseudoleucites, and both types have been described in the TAC.

231

## 232 3. Field observations

233 Field campaigns in the TAC were carried out in 2008, 2012 and 2015 in order to 234 achieve two major aims: i) Help the completion of geological map (incl. GIS database) of 235 the complex; ii) sampling the various lithologies of the TAC for petrological and 236 geochemical investigations (Fig. 2). Sampling was focused on the three major units 237 generalised for the purposes of this study: The Outer Volcanic Unit (OVU), the inner 238 Syenitic Unit (SYU) and the Central Volcanic Unit (CVU) (Fig. 2), which have been 239 juxtaposed by ring faulting. In total, 46 samples were collected and analysed, and one of 240 those (sample 6-8-12 from the SYU) was used for <sup>40</sup>Ar/<sup>39</sup>Ar age determination. Field 241 relations demonstrate that the syenitic magmas of the SYU intruded into the OVU (Fig. 242 3a). More localized and subordinate lithologies of the complex include syenitic pegmatites 243 (Fig. 3b) and pseudoleucite-bearing phonolites (Fig. 3c-f).

244

## 245 4. Petrography

The pioneering works of Meliksetian (1989) identified >50 different mineral species in rocks of the TAC, including a variety of rare earth element (REE) and high field strength element (HFSE) bearing phases. In our study, we focus on the major rock-forming minerals in the three major rock units of the complex to provide a general overview of the lithologies.

The volcanic rocks of the Outer Volcanic Unit (OVU) are typically porphyritic with an aphanitic groundmass. Major minerals are plagioclase + clinopyroxene + amphibole + biotite + alkali feldspar + Fe-Ti oxides ± nepheline, and apatite and titanite are present as accessory phases. Plagioclase is euhedral to subhedral, weakly zoned and often shows sieve textures (Fig. 4a). Euhedral clinopyroxene phenocrysts are up to 2 mm in size and typically poikilitic. Volcanic breccias are observed occasionally, containing angular fragments and xenoliths, the latter partly rich in quartz. Volumetrically small occurrences of altered pseudoleucite phonolites are present, where we found pseudomorphed leucite up to several cm in diameter. The deltoidal icositetrahedral crystal habit of the primary leucite is well preserved, but leucite has been completely replaced by secondary minerals. These are dominated by alkali feldspar and cancrinite-group minerals and comprise minor amounts of analcime. Other phases found in the pseudoleucite are clinopyroxene, biotite, apatite and calcite.

The volcanic rocks of the Central Volcanic Unit (CVU) are generally porphyritic with a fine-grained matrix. They contain euhedral plagioclase + alkali feldspar + clinopyroxene + amphibole + biotite + Fe-Ti oxides as major mineral phases. Some samples contain amphibole glomerocrysts and clinopyroxene overgrowing biotite (Fig. 4b). Rare pseudoleucite phonolites occur in this unit as well. The samples of the CVU are often intensely altered.

270 The Syenitic Unit (SYU) comprises equigranular, phaneritic, medium to coarse-271 grained syenites and nepheline syenites (Fig. 4c-h). Several samples show a trachytoidal 272 preferential alignment of feldspars. Major mineral phases are alkali feldspar + amphibole 273 + biotite + clinopyroxene + Fe-Ti oxides ± nepheline ± plagioclase. Garnet is rare but very 274 prominent in the coarse grained (pegmatitic) rock varieties, where euhedral to subhedral 275 brown garnet forms clusters with euhedral, black todark green amphibole. Accessory 276 phases observed include zircon, titanite, fluorite, muscovite, apatite, calcite, sodalite and 277 cancrinite. Subhedral alkali feldspar is typically the most abundant phase, frequently 278 exhibiting significant alteration. Primary clinopyroxene commonly shows signs of 279 incipient alteration to green amphibole.

280

# 281 5. Analytical methods

Major and trace elements were analysed by standard X-ray fluorescence (XRF), inductively coupled plasma atomic emission spectrometry (ICP-AES) and inductively coupled plasma mass spectrometry (ICP-MS) methods. Detailed information about the analytical methods used is provided in the supplementary material. Systematic differences between analyses from different laboratories are not observed. If they exist, they are likely to be small relative to the compositional effects of the magmatic processes operating, and considered negligible for the overall interpretation of the dataset.

289 Strontium (Sr) and neodymium (Nd) isotope analyses were performed at the School 290 of Earth and Environment, University of Leeds. Conventional ion-exchange chromatographic techniques were applied and samples were analyzed on a Thermo 291 292 Finnigan Triton multicollector mass spectrometer (see Halama et al. 2013 for details of 293 the analytical protocol). Information about reference materials analysed as well as 294 normalization and correction procedures applied is given in the supplementary material. 295  $^{40}$ Ar/ $^{39}$ Ar dating of amphibole from syenite sample 6-8-12 was performed using a CO<sub>2</sub> 296 laser stepwise heating technique at the Institute of Earth and Environmental Science, 297 Universität Potsdam. The analytical protocol follows established procedures and a brief 298 summary about procedural aspects, standards used and corrections applied is provided 299 in the supplementary material. Calculation of ages and errors was performed following 300 Uto et al. (1997) using the total  ${}^{40}$ K decay constant of 5.543 x 10<sup>-10</sup> a<sup>-1</sup>.

- 301
- 302 6. Results

## 303 6. 1. Rock classification and major element geochemistry

The Total Alkali versus Silica (TAS) diagram was used to classify the volcanic rocks from the OVU and CVU (Fig. 5a). For the intrusive rocks of the SYU, we used the classification diagram of De La Roche et al. (1980; Fig. 5b). Whole rock geochemical analyses are presented in Table 1. All volcanic rocks of the TAC are classified as alkaline
in the TAS diagram (Fig. 5a). Rocks of the OVU cover a wide compositional range from
basaltic trachyandesite to phonotephrite, tephriphonolite and phonolite. The
compositional range of the CVU rocks is more restricted, comprising trachyandesites and
trachytes. The plutonic rocks of the SYU are classified as nepheline syenites and syenites
based on the R1 and R2 parameters (Fig. 5b), which generally agrees with the
petrographic observations.

314 A further geochemical classification was carried out using various geochemical 315 indices (Table 1) that allow an evaluation of petrogenetic relationships (Shand 1947; 316 Frost et al. 2001; Frost and Frost 2008). The majority of the Tezhsar rocks are ferroan, 317 alkalic, metaluminous and silica-undersaturated. The Alkalinity Index (AI; AI = Al-(K+Na) 318 on a molecular basis) typically varies between 0 and 0.1, indicating that peralkaline rocks 319 (AI<0) are largely absent at the TAC. Values for the feldspathoid silica-saturation index 320 (FSSI; normative Q-[Lc+2(Ne+Kp)]/100, where Q = quartz, Lc = leucite, Ne = nepheline 321 and Kp = Kaliophilite) mostly range from -0.6 to 0. The negative FSSI values demonstrate 322 that the rocks are generally silica-undersaturated. Diagrams using the aluminium-323 saturation index (ASI; molecular Al/(Ca-1.67P+Na+K) and the modified alkali-lime index 324 (MALI; Na<sub>2</sub>O+K<sub>2</sub>O-CaO) classification demonstrate the predominantly metaluminous and 325 alkalic nature of the TAC rocks (Fig. 5c, d). Peraluminous compositions (ASI>1) are very 326 rare. Compared to the restricted compositions of SYU and CVU, the OVU shows the largest 327 variations in A/NK ratios.

Harker diagrams show a relatively smooth decrease of MgO, total FeO (FeO<sub>T</sub>) and CaO with increasing SiO<sub>2</sub> contents (Fig. 6a-c). MgO contents are below 3 wt% for OVU rocks and SYU and CVU rocks have less than 1wt% MgO, demonstrating their highly evolved

331 character and suggesting substantial fractionation of mafic minerals prior to332 crystallisation.

333

### 334 6.2. Trace element geochemistry

335 Whole-rock trace element concentrations in the TAC are variable and show some 336 significant enrichment in Sr (up to ~5000 ppm), Ba (up to ~4000 ppm), Zr (up to ~1000 337 ppm) and  $\Sigma REE$  (up to ~1200 ppm), which is typical for alkaline igneous rocks 338 (Chakhmouradian and Zaitsev 2012). Incompatible trace elements such as Th and Zr 339 show pronounced enrichment with increasing silica, in particular evident for SYU and 340 CVU rocks (Fig. 6d, e). In contrast, Sr contents remain relatively constant for intermediate 341 rocks with <58 wt% SiO<sub>2</sub> and diminishing at higher silica contents (Fig. 6f). A chondrite-342 normalised REE diagram (Fig. 7a) shows that both the volcanic and plutonic rocks of the 343 TAC are characterised by a strong fractionation between LREE and HREE with  $La_{(N)}/Yb_{(N)}$ 344 ratios predominantly around 10-40 but reaching values as high as 70. Absolute amounts 345 of LREE are generally higher in the SYU (~200-1000 x chondrite) compared to the OVU and CVU (~40-500 x chondrite). Europium anomalies, defined as Eu/Eu\*= $\frac{Eu_N}{\sqrt{(Sm_N \times Gd_N)}}$ , are 346 347 moderately negative in the volcanic units OVU (0.80 - 1.08) and CVU (0.68 - 0.91). The 348 majority of the SYU rocks have more pronounced negative Eu anomalies with Eu/Eu\* 349 values between 0.44 and 0.97 (Fig. 7a). On primitive mantle-normalised trace element 350 diagrams (Fig. 7b-d), negative anomalies for Nb, Ta and Ti are the most prominent 351 features in all three units. In contrast, a strong relative enrichment of Th and U compared 352 to Rb and Ba is only significant in the SYU and CVU, but not discernible in the OVU.

353

#### 354 **6.3 Sr and Nd isotopes**

355

5 Initial Sr and Nd isotope ratios of volcanic and plutonic rocks from the TAC,

356 recalculated to an age of 41 Ma, range from 0.7040 to 0.7052 and 0.51274 to 0.51283, 357 respectively (Table 1). 19 of 20 samples fall within the range 0.7040 to 0.7044 for the 358 initial  ${}^{87}$ Sr/ ${}^{86}$ Sr ratio. The Nd isotopic compositions correspond to positive  $\varepsilon$ Nd values 359 between +3.0 and +4.8 (Table 1).

360

## 361 **6.4** <sup>40</sup>Ar/<sup>39</sup>Ar geochronology

362 One syenite sample (sample number 6-8-12) was dated by <sup>40</sup>Ar/<sup>39</sup>Ar step heating. The 363 total gas age is 42.1 ± 0.5 Ma (Fig. 8; Table 2). We use the following criteria outlined by 364 Fleck et al. (1977) for defining a plateau age: (1) The plateau includes at least 50% of the 365 total <sup>39</sup>Ar released, (2) the ages of two contiguous steps in the plateau agree within 2s 366 error, excluding the J value error, (3) the plateau consists of three steps or more, and (4) 367 each degassing step contributing to the plateau contains >3% of the total <sup>39</sup>Ar released. 368 For the syenite sample 6-8-12, five plateau steps constituting 98.88% of the total <sup>39</sup>Ar 369 released can thus be used to define a plateau age of  $41.0 \pm 0.5$  Ma (Fig. 8; Table 2). Using 370 the plateau steps only, a normal isochron age of  $41.3 \pm 2.5$  Ma with a  $^{40}$ Ar/ $^{36}$ Ar intercept 371 at  $281 \pm 58$  is obtained. The corresponding inverse isochron yields an age of  $41.2 \pm 2.1$  Ma 372 with  $({}^{40}\text{Ar}/{}^{36}\text{Ar})_i = 289 \pm 57$ . The good agreement between the three ages underlines the 373 reliability of the age determination, with the plateau age of  $41.0 \pm 0.5$  Ma representing the 374 most precise and hence preferred age.

375

## 376 **7. Discussion**

## 377 **7.1 Comparison with regional magmatic signatures**

The alkaline and highly evolved nature of the TAC rocks makes them distinct from volcanic rocks outcropping in Armenia, which are typically transitional between alkaline and subalkaline. This includes the trachybasaltic to trachyandesitic Pliocene-Quaternary

381 rocks from northern Armenia (Neill et al. 2013, 2015), as well as rocks from the large 382 polygenetic Aragats volcano (Connor et al., 2011) and from the Gegham, Vardenis and 383 Syunik Volcanic Highlands in South Armenia (Karapetian et al. 2001; Sugden et al. 384 submitted). A comparison with data for regionally related Miocene to Quaternary 385 Armenian igneous rocks from the Yerevan and Shirak regions (Neill et al. 2015) reveals a 386 general enrichment of the TAC rocks in almost all moderately to highly incompatible trace 387 elements (Fig. 7b-d). Key features, such as negative Nb-Ta and Ti anomalies and a relative 388 enrichment of LREE compared to HREE, are similar. Isotopically, the TAC rocks, which 389 plot on the Sr-Nd mantle array, overlap with plutonic rocks from the Meghri-Ordubad 390 pluton and with other Miocene to Quaternary volcanic rocks from Armenia (Fig. 9). 391 Quaternary volcanic rocks from Aragats (Lebedev et al. 2007; Connor et al. 2011) and the 392 Gegham Ridge (Lebedev et al. 2013) also overlap in their Sr-Nd isotopic compositions. 393 This comparison reveals that there is a broad Sr-Nd isotopic homogeneity across a large 394 area of the Armenian highlands from the Eocene to the Quaternary, indicating that similar 395 source regions are involved in magma genesis. Broadly contemporary (47-40 Ma) post-396 collisional magmatic rocks from the Eastern Pontides (NE Turkey), which are 397 characterized by tholeiitic/calc-alkaline affinities enriched in LILE with pronounced 398 depletions in HFSE, also overlap in their isotopic composition (Aydınçakır & Şen 2013). 399 In contrast, extending the comparison to Eocene magmatic rocks in NW Iran reveals that 400 post-collisional granites and syenites from the Sanandaj-Sirjan Zone and granitoids from 401 the Urumieh-Dokhtar magmatic arc extend to significantly more radiogenic Sr-Nd isotope 402 compositions (Fig. 9).

403

### 404 **7.2. Magma differentiation and magma source geochemistry**

405 Both volcanic and plutonic rocks of the TAC are evolved and only a few samples are

406 of intermediate composition (Fig. 5a, b). The influence of mixing, fractional crystallization 407 and batch partial melting on the bulk geochemical composition of the rocks can be 408 evaluated using incompatible trace elements with different bulk solid/liquid partition 409 coefficients (Schiano et al. 2010). In the Rb/Nd vs. Rb diagram (Fig. 10a), the near-410 horizontal trend for the majority of data points emphasizes the dominant role of fractional 411 crystallization, whereas mixing and differences in batch partial melting would yield 412 positive correlations (Schiano et al. 2010). This interpretation is supported by a curved 413 overall trend in the Rb vs. Rb/V diagram (Fig. 10b), which is consistent with fractional 414 crystallisation or mixing, but not with different degrees of partial melting (Schiano et al. 415 2010). Moreover, the coherent trend of the Rb/Ba vs. Ba diagram (Fig. 10c) reflects 416 feldspar fractionation and does not indicate any significant effects of hydrothermal 417 alteration. A major role of role of crustal contamination processes can also be excluded 418 based on the unradiogenic initial <sup>87</sup>Sr/<sup>86</sup>Sr isotope ratios that remain relatively constant 419 with increasing silica (Fig. 10d). Crustal contamination typically leads to an coupled 420 increase in (<sup>87</sup>Sr/<sup>86</sup>Sr)<sub>i</sub> and SiO<sub>2</sub>, which is not observed for the TAC. The only sample with 421 an elevated (<sup>87</sup>Sr/<sup>86</sup>Sr)<sub>i</sub> ratio (2-7-09) has a high Rb/Sr ratio of ~8 and might be affected 422 by a larger uncertainty in recalculation of the initial value and/or post-magmatic Rb or Sr 423 mobilization. There is also no indication of limestone assimilation, which would lead to a 424 significant enrichment in CaO (Fig. 6c).

Olivine is conspicuously absent in all TAC rocks, but the low MgO contents and the highly evolved character point to preceding fractionation of mafic minerals (Fig. 6). The decrease in CaO/Al<sub>2</sub>O<sub>3</sub> coupled with increasing FeO<sub>t</sub>/MgO ratios, Eu anomalies that become more negative with increasing degree of differentiation, and decreasing Sr contents and Dy/Yb ratios with increasing SiO<sub>2</sub> suggest a significant role of amphibole/clinopyroxene and plagioclase fractionation whereas garnet fractionation

was insignificant (Fig. 11a-c). Typically, the OVU rocks are more primitive than both CVU
and SYU rocks. OVU rocks even retain Eu/Eu\* values close to 1, pointing to lack of
significant plagioclase fractionation (Fig. 11b).

434 The pronounced depletions in HFSE (Nb, Ta, and Ti) in all TAC rocks emphasizes the 435 influence of subduction processes on the mantle source (Fig. 7). Similar negative HFSE 436 anomalies have been observed in alkaline rocks of the Longbaoshan Complex, North China 437 Craton (Lan et al. 2011) and carbonatites from east Tibet in the Himalayan collision zone 438 (Hou et al. 2006) and were attributed to subduction processes influencing the magma 439 source regions prior to continental collision. In addition, various trace element indicators 440 for source enrichment processes support the notion that the OVU and the CVU are geochemically distinct (Fig. 11d-f). The OVU shows elevated Sm/Yb and Ba/La ratios, as 441 442 well as relatively low La/Sm and Th/Yb ratios compared to the CVU (Fig. 11d-f). 443 Collectively, these geochemical features of the OVU are interpreted as a signature of 444 moderate fluid enrichment via slab dehydration inherited from earlier subduction events. 445 Both CVU and OVU rocks share high Ba/Nb ratios, similar to arc volcanic rocks in general 446 (Fig. 12a). There is little overlap between the two groups as OVU rocks are additionally 447 characterized by, on average, higher La/Nb ratios and Ba/Nb ratios >100, suggesting a 448 temporal evolution towards a decreasing subduction influence from the early OVU to the 449 late stage CVU. The more scattered trend towards lower Ba/Nb ratios in the syenites and 450 nepheline syenites is likely a result of progressive alkali feldspar fractionation and should 451 not be considered as the parental magma signature. The felsic plutonic rocks from the SYU 452 tend to exhibit a larger geochemical variability when compared with the volcanic rocks, 453 which is likely related to the fact that some show cumulate textures and may not 454 represent melt compositions. The CVU, in contrast, has compositions that are more tightly 455 clustered, with a faint indication of source enrichment from subducted sediments. Mechanisms of enrichment of the mantle source can be distinguished using [Hf/Sm]<sub>N</sub> and
[Ta/La]<sub>N</sub> ratios (Fig. 12b), where TAC rocks are characterized by a subduction
metasomatism signature, clearly distinct from carbonatitic metasomatism.

459 Sr-Nd isotopic compositions are broadly overlapping with Eocene to Pliocene 460 magmas from the Meghri-Ordubad pluton and Pliocene to Quaternary volcanism in 461 central and northern Armenia, pointing to only minor spatial variations in the respective 462 mantle source regions (Fig. 10). The source of the TAC magmas is dominated by a depleted 463 mantle component and crustal contamination is essentially absent, as all of the possible 464 crustal contaminants would greatly enhance the radiogenic isotope signatures of the 465 magmas, which is not the case. Silica-undersaturated alkaline rocks commonly have 466 isotopic compositions that suggest a magma source in the mantle (Dunworth and Bell 467 2001; Kramm and Kogarko 1994). For instance, nepheline syenites from the Gardar 468 Province (Greenland) show Nd isotopic compositions typical for mantle-derived rocks 469 without any significant crustal assimilation (Halama et al. 2005; Marks et al. 2004). 470 Therefore, evolved silica-undersaturated rocks are interpreted as products of 471 differentiation from more primitive nephelinitic, basanitic or alkali basaltic magmas 472 derived from the upper mantle (Kramm and Kogarko 1994; Trumbull et al. 2003). 473 Basanitic volcanism is common to the south of TAC in the Syunik Volcanic Highland 474 (Sugden et al. submitted) near the Armenia-Azerbaijan-Iran border region.

The trace element evidence for a subduction modifications and the Sr-Nd isotopic evidence for previous melt extraction suggest that the TAC magmas are predominantly derived by low degrees of partial melting from a lithospheric mantle source which has been affected by pre-Eocene subduction i.e., prior to post-collisional melt generation. This magma generation model is also the preferred model for volcanism in East Anatolia (Keskin 2003), and similar geochemical features in volcanic rocks from the Eastern

481 Pontides (Artvin Province) contemporary (47-40 Ma) to emplacement of the TAC were 482 also interpreted to be derived from a mantle source that had experienced metasomatism 483 by slab-derived fluids (Aydınçakır & Şen 2013). Post-collisional magmatic processes are 484 commonly affected by prior subduction processes and LILE-enriched mantle sources are 485 characteristic for these rocks (Bonin et al. 1998), typically resulting in calc-alkaline 486 magmatic suites (Harris et al. 1986). The TAC represents an unusual case insofar as the 487 post-collisional magmatic rocks are alkaline in character but also derive from a 488 subduction-modified mantle source.

489

## 490 **7.3. The age of the Tezhsar Alkaline Complex in a regional context**

491 The mid-Eocene age of  $41.0 \pm 0.5$  Ma falls into a time of widespread magmatism in the 492 Lesser Caucasus region, which lasted from ~49 to ~38 Ma and comprised the 493 emplacement of alkaline and nepheline-bearing gabbros, monzonites and syenites as well 494 as gabbro-diorite-granodiorite-syenogranite complexes and granites (Ghukasyan et al. 495 2006; Melkonyan et al. 2008; Moritz et al. 2016). The magmatic activity was accompanied 496 by porphyry-type Cu-Mo mineralization that was dated at 44-40 Ma by Re-Os analyses of 497 molybdenite (Moritz et al. 2016). Slightly younger alkaline magmatism is represented by 498 the Bunduk alkaline complex (38-32 Ma) located ~15 km northeast of the TAC (Abovyan 499 et al., 1981; Meliksetian, 1989). This pluton intrudes the Middle-Late Eocene volcanic 500 suite of the Bazum ridge and the Bazum gabbro-granitoid intrusive complex, exhibiting 501 an elongate morphology, parallel to the segment of Pambak-Sevan fault.

Regionally, broadly contemporaneous magmatic activity is also recorded in the Talysh mountain range (Azerbaijan/Iran) at around 41-38 Ma (Vincent et al. 2005), in the Eastern Pontides (Turkey) at ~46-40 Ma (Aydınçakır and Şen 2013) and in western Georgia at ~47-41 Ma (Lebedev et al. 2009). Further to the SE in the Zagros orogen, ~41

506 Ma old granites and syenites occur in the Piranshahr massif (Mazhari et al. 2009) and ~40 507 Ma granitoids in the Urumieh-Dokhtar arc (Kazemi et al. 2018). The peak of subduction-508 related magmatism in Iran is also close to 40 Ma (Allen and Armstrong 2008), and a 509 magmatic flare-up lasting ~18 million years from 55 to 37 Ma has been postulated in the 510 Urumieh-Dokhtar belt and the Alborz Mountains in Iran (Verdel et al. 2011). Throughout 511 the Eocene, the plate convergence between the Arabian and Eurasian plates was 512 proceeding at rates of 2-3 cm/year (McQuarrie et al. 2003). Following the initiation of the 513 Arabia-Eurasia collision, arc magmatism declined in the Late Eocene (Allen and 514 Armstrong 2008). However, convergence was relatively rapid throughout Eocene-515 Oligocene time, and only slowed since Early Miocene (Rosenbaum et al. 2002).

516 The age of the TAC falls within this period of extensive magmatism during 517 convergence between the Arabian and Eurasian plates, and its geochemical characteristics demonstrate a subduction-related origin. This subduction signature is 518 519 inherited from prior northward subduction of the Neotethys ocean underneath the 520 Eurasian margin, leading to a preconditioning of the mantle (Verdel et al. 2011). Typical 521 calc-alkaline, subduction-related Eocene magmatism typical for active arc environments 522 is preserved in the oldest granitoids (49-44 Ma) of the Meghri-Ordubad pluton (Moritz et 523 al. 2016). The Lesser Caucasus experienced extension and crustal thinning at around 40 524 Ma causing decompression melting of the hydrated, subduction-influenced lithospheric 525 mantle (Verdel et al. 2011), which imparted its geochemical signature onto the TAC 526 magmas. Middle Eocene (ca. 49-40 Ma) extension, accompanied by magmatism, also 527 occurred in Iran (Ballato et al. 2011). The extension-related magmatism in an overall 528 setting of convergence (Rosenbaum et al. 2002) is caused by the rollback of the Neotethys 529 slab (Vincent et al. 2005; Verdel et al. 2011).

530 The oldest rocks at TAC in the OVU show some geochemical characteristics 531 reminiscent of a dehydration fluid signature in arc magmatic rocks (high Ba/Nb, Ba/La ratios; Figs. 7 and 11e, f). A clear arc signature, most evident in the pronounced negative 532 533 Nb-Ta anomalies, is present in all of the TAC rocks, similar to the Meghri-Ordubad pluton 534 at the Armenia-Iran border. However, the TAC rock compositions are distinct as they are 535 not calc-alkaline but alkaline (Fig. 5) with a pronounced enrichment in incompatible trace 536 elements (e.g. up to 5000 ppm Sr and typically 100-500 ppm Rb compared to <1000 ppm 537 Sr and 10-200 ppm Rb in rocks from Meghri). This geochemical character is not due to 538 long-lived differences in the mantle source compared to Meghri-Ordubad pluton since the 539 Sr-Nd isotopic characteristics are similar (Fig. 9). Instead, smaller degrees of melting 540 and/or a metasomatic enrichment episode(s) immediately prior to magma generation 541 have to be invoked. The very pronounced subduction signature in the TAC supports the 542 predominant melting of hydrated and HFSE-depleted lithospheric mantle, with 543 subordinate contributions from upwelling astenospheric mantle (Verdel et al. 2011). The 544 occurrence of these alkaline rocks in a general setting of convergence is unusual, but can 545 be attributed to periods of localized extension in the Lesser Caucasus. The overall 546 convergence throughout Eocene and Oligocene is well established based on kinematic 547 data and modelling (Rosenbaum et al. 2002), but if the lithospheric structures allowed 548 ascent of mantle-derived magmas via localized faulting and/or rift tectonics alkaline 549 magmatism can develop even in collision zones (Harris et al. 1986). Development of an 550 extensional regime along this sector of Lesser Caucasus was previously suggested to 551 explain the alkaline character of Paleogene magmatic rocks, particularly those within 552 Armenia (Kogarko et al., 1995).

553

## 554 **7.4. Petrogenesis of pseudoleucite phonolites**

555 Based on optical microscopy and geochemical analyses including XRD, 6 types of 556 "epileucites" and 5 types of pseudoleucites were distinguished by their mineral 557 associations, host rocks and crystallographic habit (Meliksetian 1971, 1978). "Epileucites" 558 are considered to be a result of post-magmatic hydrothermal alterations, whereas 559 pseudoleucites are considered to be a result of disintegration of metastable K-Na leucite 560 into mixture of orthoclase and nepheline under subsolidus conditions (T=600°C) in late 561 magmatic stage (Meliksetian 1978; Gittins et al. 1980). Yagi and Gupta (1978) mention 562 that the K<sub>2</sub>O/Na<sub>2</sub>O ratio of 4.3 in pseudoleucites of porphyry tinguaite dykes of TAC is the 563 highest among those studied worldwide highlighting the importance of resolving 564 complex's evolutionary story to better understand the conditions of pseudoleucite 565 paragenesis.

566 The investigated leucite pseudomorphs occur in a phonolite (Fig. 3c-f). Relicts of 567 primary leucite are lacking, and they are generally rarely observed in leucite 568 pseudomorphs. The leucite pseudomorphs mainly consist of alkali feldspar but do not 569 contain nepheline, instead comprising abundant cancrinite (Fig. 13b-c). Different theories 570 about the genesis of leucite pseudomorphs were put forward (see Edgar, 1984, and 571 references therein), including (1) subsolidus breakdown of leucite to orthoclase and 572 nepheline, (2) reaction of leucite with a Na-rich liquid and (3) alkali ion exchange 573 reactions between leucites and Na-rich glass or fluid. We will briefly discuss these 574 theories in relation to the leucite pseudomorphs in the phonolite.

575 Subsolidus breakdown of common K-rich leucite would produce alkali feldspar and 576 kalsilite, hence a process to cause relative enrichment of Na is required to explain the 577 occurrence of Na-bearing phases in pseudoleucites. Leucite solid solutions with up to 40 578 wt.% NaAlSi<sub>2</sub>O<sub>6</sub> were produced experimentally, and these experienced subsequent 579 breakdown into nepheline and alkali feldspar (Fudali 1963). However, natural leucite

does not contain excess amount of sodium to form this type of intergrowth on decomposition (Viladkar 2010). The mineralogy of the leucite pseudomorph, comprising abundant Na-bearing phases such as cancrinite and analcime (Fig. 13b-c), suggest that they are derived from a Na-rich precursor phase. Hence, subsolidus breakdown of natural K-rich leucite alone cannot explain their occurrence, but formation of a metastable Narich leucite before breakdown might be possible (Taylor and MacKenzie 1975).

586 The pseudoleucite reaction is a reaction of leucite with a Na-rich magma to form alkali 587 feldspar and nepheline in the system NaAlSiO<sub>4</sub> – KAlSiO<sub>4</sub> – SiO<sub>2</sub> (Bowen and Ellestad 1937; 588 Edgar 1984). This reaction terminates the leucite stability field and leucite disappears by reaction with the magma (Bowen and Ellestad 1937). The TAC leucite pseudomorphs, 589 590 however, are characterized by a well-preserved deltoidal icositetrahedral crystal habit, 591 reflecting the external shape of the precursor phase. It is difficult to envisage this reaction 592 to fully replace primary leucite without modifying the morphology of the leucites (Taylor 593 and MacKenzie 1975), which is so beautifully preserved (Fig. 3). Moreover, various minor 594 mineral phases that contain additional elements occur within the pseudomorphs. Some 595 of these (e.g. clinopyroxene, apatite) may be explained as primary magmatic inclusions, 596 but others (analcime, calcite) texturally appear as secondary phases (Fig. 13b-c). This 597 suggests that explaining the genesis of the leucite pseudomorphs based on the phase 598 relations in this petrogenetic system is an oversimplification (Edgar 1984).

Alkali ion exchange reactions between leucites and Na-rich glass or fluid was proposed as mechanism to produce pseudomorphs after leucite that are similar in composition to natural pseudoleucites (Taylor and MacKenzie 1975). Fluid-induced reactions would facilitate the increase in Na content and formation of Na-dominated phases, such as cancrinite and analcime in TAC. Cancrinite is assumed to replace nepheline due to a reaction between nepheline and volatile-rich melts or fluids, a common

605 late magmatic- hydrothermal process (Martins et al. 2017). A reaction with fluids was 606 also used to explain pseudoleucite with intergrowth of alkali feldspar, sericite and 607 cancrinite from the Gardar Province, Greenland (Hesselbo 1986) and the replacement of 608 nepheline by analcime, cancrinite, sodalite and muscovite in pseudoleucite from India 609 (Viladkar 2010). Cancrinite is also an important constituent of the pseudoleucite 610 phenocrysts from Spotted Fawn Creek (Yukon Territory, Canada), where also garnet, 611 biotite, calcite, muscovite and plagioclase occur as inclusions within pseudoleucite 612 (Tempelman-Kluit 1969). Removal of K, addition of Na and water was attributed to the 613 entry of a fluid phase to permit the chemical exchange (Tempelman-Kluit 1969). The 614 presence of cancrinite in the TAC leucite pseudomorphs bears evidence for interaction 615 with a H<sub>2</sub>O-CO<sub>2</sub>-bearing fluid, possibly with minor amounts of S and Cl, as the general 616 formula for cancrinite is (Na,Ca,K)<sub>6-8</sub>Al<sub>6-x</sub>Si<sub>6+x</sub>O<sub>24</sub>(CO<sub>3</sub>,SO<sub>4</sub>,Cl,OH)<sub>1-2</sub>·nH<sub>2</sub>O with x << 1 and 617 n = 1-5 (Martins et al. 2017) illustrates. Given the scarcity of analcime in the TAC 618 pseudoleucites, a conversion of primary leucites into analcime via reaction with Na-rich 619 fluids as proposed for pseudoleucites from a phonolite dyke in Bohemia (Pivec et al. 2004) 620 seems unlikely. The texture of the TAC leucite pseudomorphs pseudoleucites has 621 resemblance to a "palisade texture", in which orthoclase laths near the margins of the 622 pseudomorphs are oriented at right angles to the crystal boundaries (Tempelman-Kluit 623 1969). These textures can be interpreted to form by subsolidus reactions in response to 624 increasing fluid pressure when pervasive fluids come in contact with the leucite (Hesselbo 625 1986). All these lines of evidence point to a late/post-magmatic hydrothermal alteration 626 for the formation of the leucite pseudomorphs in the investigated phonolite, and they can 627 be referred to as "epileucites". Complementary evidence for fluid-rich conditions during 628 the late to post-magmatic evolution of the TAC are the presence of pegmatites and the 629 widespread alteration in the CVU rocks.

630

# 631 8. Conclusions

- A combination of small degrees of partial melting and pre-conditioning of the
   mantle source by slab dehydration and subsequent metasomatic processes can
   explain the alkaline, subduction-influenced geochemical character of the TAC.
- The Sr-Nd isotopic data demonstrate a mantle source with negligible crustal
   influence. There is a broad isotopic overlap with Eocene to Quaternary magmatism
   in other regions of Armenia, suggesting the regional presence of isotopically
   similar mantle source regions.
- The emplacement of the syenitic units of the TAC was dated by <sup>40</sup>Ar/<sup>39</sup>Ar at 41.0 ±
   0.5 Ma. The emplacement of the TAC can thus be linked to a previously proposed
   model of Eocene Neotethyan slab rollback driving decompression melting and
   extension-related magmatism in Iran and Azerbaijan within a tectonic setting of
   general convergence between the Arabian and Eurasian plates.
- 644 The formation of leucite pseudomorphs is related to initial leucite crystallization ٠ 645 from an evolved, silica-undersaturated magma followed by subsolidus breakdown 646 and interaction with a late to post-magmatic fluid. The magmatic-hydrothermal 647 fluid percolating through the rocks caused alteration of nepheline into cancrinite 648 and amphibolitisation of clinopyroxenes. This fluid overprint may be responsible 649 for the plethora of REE-bearing phases described previously within the TAC and 650 hence be a crucial factor in the (re)distribution of rare elements in alkaline igneous 651 rocks.
- 652

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662

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1161 **Figure captions** 

1162

Figure 1 - (a) Geotectonic framework of the Caucasus region showing major 1163 1164 tectonostratigraphic provinces, associated terranes and the location of Tezhsar Alkaline 1165 Complex (star) about 50 km north of Yerevan (modified after Adamia et al., 2011, and 1166 Rezeau et al., 2017). (b) Palaeogeographical reconstruction of the Eurasian-Arabian 1167 collision in the Ypresian (52Ma) just before formation of TAC (modified after Mederer et 1168 al., 2013). SAB – South Armenian Block, SAS – Sevan-Akera Suture, BZS – Bitlis-Zagros 1169 Suture, TAB – Tauride-Anatolian Block, NTP – Northern Tethyan Province, STP – Southern 1170 Tethyan Province.

1171

Figure 2 – Geological map of the Tezhsar Alkaline Complex. The inset show a simplified
subdivision of the TAC which is used for the geochemical diagrams of this study.

1174

1175 **Figure 3** – Field relations (a, b) and hand specimen photographs (c, d) of the TAC. (a) 1176 Light coloured syenite intruding into dark grey volcanic rocks of the Outer Volcanic Unit. 1177 (b) Coarse-grained nepheline syenite pegmatite comprising dark patches of garnet and 1178 amphibole. (c) Phonolite handspecimen with idiomorphic leucite pseudomorphs 1179 reaching up to 2 cm in diameter. (d) Polished surface of a pseudoleucite phonolite. (e, f) 1180 Hand specimen of pseudoleucite megacrysts (up to ~8cm in diameter) as deltoidal icositetrahedra found in TAC phonolites. Samples are from old collections of B. 1181 1182 Meliksetian and Z. Chibukhcyan.

1183

Figure 4 – Photomicrographs illustrating characteristic features of rocks from the TAC in
plane polarized (PPL) and cross-polarized (XPL) light. (a) Plagioclase (Pl) phenocrysts in

feldspathic matrix of a basaltic trachyandesite, OVU (XPL). (b) Biotite (Bt) surrounded by
clinopyroxene (Cpx) with accessory apatite (Ap) in trachyte, CVU (PPL). (c)
Clinopyroxene and titanite (Ttn) in nepheline syenite, SYU (PPL). (d) Amphibole (Amp) in
syenite, SYU (PPL). (e) Sodalite (Sdl) and nepheline (Nph) in nepheline syenite, SYU (PPL).
(f) Amphibolitization of clinopyroxene in syenite, SYU (PPL) (g) Garnet (Grt) in syenite
with inclusions of alkali feldspar, SYU (PPL) (h) Garnet-amphibole cluster with alkali
feldspar and nepheline in pegmatitic nepheline syenite, SYU (PPL).

1193

1194 Figure 5 – (a) Total Alkali-Silica (TAS) classification diagram of the volcanic units (OVU 1195 and CVU) of the TAC. Alkaline-subalkaline division from Irvine & Baragar (1971). (b) R1-1196 R2 classification diagram (from De La Roche et al., 1980) of the intrusive SYU unit of the 1197 TAC. (c) A/NK vs A/CNK diagram (after Shand, 1947) based on the molecular proportions 1198 of Al (A), Na (N), K (K) and Ca (C), showing that the rocks of the TAC can largely be 1199 classified as metaluminous. (d) Modified Alkali-Lime Index (MALI, after Frost and Frost, 2008) plotted as a function of SiO<sub>2</sub> content for the TAC rocks that are generally alkalic in 1200 1201 composition. Comparative data for Eocene magmatic rocks from the Talysh mountains, 1202 Azerbaijan (Vincent et al., 2005 – pink diamonds) and Pliocene-Quaternary volcanic rocks 1203 from central and northern Armenia (Neill et al., 2013, 2015 – orange field).

1204

Figure 6 – Harker diagrams of the TAC samples for selected major (a-c) and trace (d-f)
elements. The limestone assimilation trend in (c) was calculated after Costa et al. (2013)
using limestone composition WGZ-3 from Zhang et al. (2017). All symbols as in Fig. 5.

Figure 7 – (a) Chondrite-normalised REE diagram highlighting more pronounced LREE
fractionation and negative Eu anomalies within the SYU relative to the volcanic units of

the TAC. Normalisation values from Boynton et al. (1984). (b-d) Mantle-normalised trace
element diagrams of rocks from the TAC; (b) – OVU, (c) – SYU, (d) – CVU. Normalisation
values after McDonough & Sun, 1995. Comparative data from Neill et al. (2015) for
Pliocene-Quaternary volcanic rocks from central and northern Armenia.

1215

Figure 8 - <sup>40</sup>Ar/<sup>39</sup>Ar age spectrum plot for the amphibole separate from syenite sample
6-8-12.

1218

Figure 9 – Sr-Nd isotope diagram of the TAC data (red squares) in comparison to other
Eocene-Quaternary igneous rocks in the Lesser Caucasus and adjacent regions. The
mantle array is from Lebedev et al. (2007) after DePaolo & Wasserburg (1979). Data
sources: (I) – Moritz et al. (2017); (II) – Aydınçakçır & Şen, (2013); (III) – Kazemi et al.
(2018); (IV) – Mazhari et al. (2009); (V) – Connor et al. (2011); (VI) – Kheirkhah et al.
(2009); (VII) – Neill et al. (2013; 2015). Literature data were recalculated using the <sup>87</sup>Rb
decay constant of 1.3972 x 10<sup>-11</sup> a<sup>-1</sup>.

1226

1227 Figure 10 – TAC samples plotted in various diagrams to evaluate effects of distinct 1228 magmatic processes. (a) Rb/Nd vs. Rb diagram (after Schiano et al. 2010) where 1229 horizontal trends reflect fractional crystallization and positive correlations can be caused 1230 by mixing or batch partial melting. (b) Rb vs. Rb/V diagram (after Schiano et al. 2010) 1231 where curved trends, as observed for the TAC rocks, reflect fractional crystallisation or 1232 mixing. (c) Rb/Ba vs. Ba diagram exhibiting a smooth trend indicative of feldspar fractionation. (d) Initial <sup>87</sup>Sr/<sup>86</sup>Sr isotope ratios of the TAC samples plotted against SiO<sub>2</sub> 1233 content. All samples except one plot in a very narrow range of (87Sr/86Sr)<sub>i</sub> ratios and there 1234 1235 is no clear trend with increasing SiO<sub>2</sub> content. The sample with the elevated  $(87Sr/86Sr)_i$  ratio (2-7-09) has a high Rb/Sr ratio of ~8 and might be affected by post-magmatic Rband/or Sr mobilization and a larger uncertainty in recalculation.

1238

1239 Figure 11 - Major and trace element indicators for fractionation and source enrichment 1240 processes. (a) CaO/Al<sub>2</sub>O<sub>3</sub> vs FeO<sub>t</sub>/MgO diagram showing fractionation trends for 1241 plagioclase, olivine and amphibole(am)/clinopyroxene(cpx) after Moritz et al. (2016). (b) 1242 Eu/Eu\* vs SiO<sub>2</sub> diagram depicting negative Eu anomalies in SYU and CVU samples, 1243 indicating plagioclase fractionation. (c) Dy/Yb vs SiO<sub>2</sub> diagram with fractionation trends for garnet and amphibole after Davidson et al. (2007). (d) La/Sm vs Sm/Yb diagram with 1244 approximate mineral stability thresholds of in mantle melt residues after Mamani et al. 1245 (2010). Note the distinct signatures for the two volcanic units of TAC. (e) Ba vs Nb/Y 1246 1247 diagram displaying trends for fluid enrichment due to slab dehydration and mantlederived melt enrichments after Kepehinskas et al. (1997). Slab fluid enrichment is 1248 1249 prominent in the OVU rocks. (f) Th/Yb vs Ba/La diagram with trends for enrichment from 1250 subducted slab sediments and slab fluids from Woodhead et al. (2001). Elevated Ba/La 1251 ratios in OVU rocks suggest source enrichment via slab fluids.

1252

Figure 12 – Trace element ratio diagrams of TAC rocks. (a) Ba/Nb vs La/Nb. Alkali
feldspar fractionation trend highlighted as a result of Ba depletion. Field boundaries after
Jahn et al. (1999). (b) (Ta/La)<sub>N</sub> vs (Hf/Sm)<sub>N</sub>. Influence of subduction metasomatism is
suggested by strongly decreasing (Ta/La)<sub>N</sub> ratios. Field boundaries after La Flèche et al.
(1998). Comparative data for Eocene magmatic rocks from the Talysh mountains,
Azerbaijan (Vincent et al., 2005 – pink diamonds) and Pliocene-Quaternary volcanic rocks
from central and northern Armenia (Neill et al., 2013, 2015 – orange field).

- 1261 **Figure 13** Pseudoleucite from the OVU of the Tezhsar Complex. (a) Scanned thin section
- image of a single pseudoleucite crystal. (b,c) Back-scattered electron images of (b) the
- 1263 boundary between matrix and pseudoleucite and (c) the interior of the pseudoleucite.
- 1264 Note the presence of cancrinite (Ccn) and analcime (Anl), other mineral abbreviations as
- in Figure 4.
- 1266









































Table 1: Whole-rock geochemical data for samples from the Tezhsar Alkaline Complex																									
Lithelesiaal Unit	OVIL	OVIL	01/11	OVIL	01/11	0001	OVIL	OVIL	01/11	01/11	OVIL	01/11	01/11	CVII	CVII	CVII	CVII	CVII	CVII	CVII	CVII	CVII	CVII	CVII	CVII
Lithological Unit	1-2-15	3-1-15	3-2-15	3-3-15	6.1.12	6.2.12	10.42.09	10.44.09	2,0,00	2.11.00	2.12.09	2.12.00	2.2.09	1-4-15	2-1-15	2-8b-15	2-11-15	3-5-15	6.2.12	5YU 6.4.12	5YU 6.5.12	6.9.12	6.0.12	5YU 6-10-12	510
Field Reference	27	42	43	44	18	19	8	9	2-0-09	5	6	2-13-09	3-2-09	29	30	37	40	46	20	21	22	24	25	26	10-43-08
Analysed at	BV	BV	BV	BV	P	P	RHL	RHL	RHL	RHL	RHL	RHL	RHL	BV	BV	BV	BV	BV	P	P	P	P	P	P	RHL
Easting	44.58448	44.57898	44.57822	44.57707	44.53700	44.53700	44.58325	44.58325	44.59320	44.59755	44.59755	44.59862	44.62817	44.58339	44.58454	44.58279	44.59697	44.57256	44.56770	44.56770	44.57538	44.59230	44.57735	44.58232	44.58325
Northing	40.71049	40.72075	40.72034	40.71685	40.66112	40.66112	40.63900	40.63900	40.63365	40.64063	40.64063	40.64185	40.65332	40.70836	40.70469	40.68375	40.69765	40.71180	40.68247	40.68247	40.68107	40.67115	40.69195	40.68042	40.63900
SiO <sub>2</sub>	52.92	52.65	51.24	55.57	55.89	56.98	53.90	56.58	56.81	54.78	57.93	59.08	52.13	59.22	60.10	63.15	57.39	56.26	60.86	62.41	59.94	56.89	64.47	59.64	58.66
TiO <sub>2</sub>	0.60	0.45	0.59	0.46	0.54	0.50	0.57	0.43	0.54	0.30	0.43	0.48	0.74	0.73	0.46	0.36	0.38	0.35	0.37	0.40	0.36	0.40	0.36	0.57	0.82
Al <sub>2</sub> O <sub>3</sub>	21.48	20.29	18.24	19.09	20.12	20.66	19.95	20.26	19.91	21.73	19.72	20.61	20.75	19.03	19.35	17.98	19.90	20.72	19.26	18.42	18.56	19.29	17.81	19.24	20.07
Fe <sub>2</sub> O <sub>3</sub> *	1.64	1.02	1.70	1.22	1.36	0.96	1.66	1.32	1.72	0.63	1.32	1.30	2.60	0.88	0.84	0.77	0.77	0.74	0.86	0.65	0.94	1.05	0.65	0.89	1.07
FeO*	3.45	2.15	3.57	2.57	2.86	2.02	3.49	2.78	3.61	1.33	2.77	2.72	5.45	1.85	1.77	1.62	1.62	1.56	1.81	1.37	1.98	2.20	1.35	1.87	2.25
Mn0 Mn0	0.17	0.22	0.17	0.22	0.14	0.11	0.19	0.20	0.15	0.13	0.11	0.19	0.23	0.22	0.16	0.16	0.15	0.14	0.16	0.12	0.22	0.17	0.07	0.14	0.19
MgO CaO	2.00	3.17	5.37	3.57	1.70	2.11	1.21	2.55	1.05	1.72	2.73	2.36	6.71	1.84	0.55	0.34	2.67	2.24	1.80	1 19	2.22	3.13	0.12	2.49	2.02
Na <sub>2</sub> O	4.32	4.03	3.93	5.34	5.30	2.01	5.26	4.41	6.07	5.88	5.36	1.90	2.90	4.42	5.66	6.49	5.28	3.59	5.25	4.21	4.86	5.15	6.47	4.76	4.68
K-0	3.38	10.00	5.46	7.06	7.88	10.73	6.23	8.32	4.65	10.16	6.49	10.52	5.28	8.80	7.83	5.97	7.02	10.67	6.55	8.82	7.10	6.55	6.33	7.39	8.82
P <sub>2</sub> O <sub>2</sub>	0.45	0.09	0.41	0.31	0.21	0.17	0.30	0.17	0.36	0.05	0.21	0.14	0.54	0.05	0.05	0.09	0.04	0.07	0.09	0.04	0.13	0.20	0.03	0.06	0.06
LOI	0.92	3.88	4.98	2.70	1.73	1.73	4.19	3.11	1.22	2.98	2.78	1.56	0.85	1.63	1.21	0.72	3.23	2.45	1.50	1.28	2.01	3.08	1.12	1.59	0.99
SUM	99.09	98.42	98.46	99.04	98.31	98.42	101.39	101.09	101.17	99.91	100.65	101.62	100.80	99.05	99.65	98.89	98.78	99.24	98.95	99.10	98.91	98.99	99.25	99.04	100.25
Ba	1060	3010	1340	1760	1570	3710	1510	991	1140	460	1110	3940	932	60	144	472	72	272	655	481	1030	891	87	150	208
Cr	30.1 bdi	17.1 bdl	24.9 bdl	Z1 bd1	na 22	na 10	9.08	6.34	9.23	3.43	6.23 2.90	7.35	18.1	13.1 bai	12.7 bai	30.2 bdl	77.2 bdl	13.1 bai	na 21	na bai	na 17	na 14	na 12	na bdl	7.4
Cs	2.1	3	0.9	2.9	na	na	6.47	5.63	1.79	3.78	1.34	1.7	5.88	4.8	3.4	4.3	4.7	9.8	na	na	na	na	na	na	2.23
Ga	18	15.7	15.3	16.4	15	17	na	na	na	na	na	na	na	18.7	20.9	21.7	19.5	14.2	19	20	19	20	23	19	na
Hf	2.9	5.1	2.8	4.5	23	10	3.01	11	2.06	3.03	3.24	3.01	2.14	11	11.7	22.4	11	3.4	21	bdl	12	14	12	bdl	6.27
Nb	6.8	18.6	6.6	11.5	12	13	10.5	12.3	8.63	19.2	10.5	11.4	6.95	75.1	61.7	49.5	45.9	16.2	41	59	32	31	45	56	55
KD Sc	83.2 bdl	Z43 bdl	102 bdl	191 bdl	204	160	3 45	2 190	5.73	0 297	2 75	193	186	233 bdl	235 bdl	261 bdl	Z4/	452 bdl	251	305	257	4.2	2.0	236	1/5
Sn	1	bdl	bdl	bdl	na	na	5.45 na	na	na	0.2 <i>51</i>	2.75 na	1.05 na	na	2	2	2	1	bdl	na	na	na	-1.2 na	na	na	1.04 na
Sr	1410	4260	1460	1380	2850	4820	2010	1830	1290	956	1230	3520	1280	885	330	384	2300	959	757	1030	1280	1710	42	1480	817
Та	0.4	0.9	0.4	0.6	na	na	0.445	3.28	0.543	0.815	0.448	0.474	0.286	3.7	3.1	2.1	2.6	0.9	na	na	na	na	na	na	3.58
Th	8.6	20.9	9	15.7	na	na	10.8	14.7	7.56	18.4	11.6	11.7	7.08	113	124	75.3	77.5	11.6	na	na	na	na	na	na	26.6
U V	2.9	202	2.7	152	na 162	na 145	3.43	5.21	4.07	52.6	4.53	3.64	3.16	9.7	16.7	19.3	15.5	2.6	na 67	na 55	na 84	na 106	na 35	79	5.71
Zr	126	271	126	201	263	344	153	189	148	204	166	180	123	494	527	1080	629	168	737	1110	547	549	471	547	304
Y	20.7	39	20.8	33	22	21	28.4	28.7	24.9	26.4	26.3	29.6	23.6	68.7	53	50.3	38	20.8	38	41	29	26	44	31	66
La	36.7	114	42.9	77.8	70	63	72.5	72.9	53.5	80.2	61.2	79.1	45.6	319	244	167	161	81	140	81	71	104	204	75	186
Ce Dr	67.8	199	76.1	138	127	120	125	124	89.9	130	104	135	79.1	558	396	247	264	135	204	224	142	175	328	174	354
Nd	31.3	78.3	35	58.1	56	54	56.2	53.1	41.1	48.5	46.1	59.5	9.38	190	117	72.6	84.4	52.4	85	103	61	69	113	70	155
Sm	6	13.5	6.5	10.7	9.3	8.8	10.3	9.7	7.92	8.67	8.49	11.3	8.19	27.2	16.6	10.7	12.5	8.1	13	18	10	10	15	11	28.1
Eu	1.9	3.9	2.1	3.1	2.4	2.8	3.1	2.74	2.33	1.97	2.32	4.47	2.26	3.8	2.2	1.4	1.9	2.3	1.5	2.2	1.9	1.9	0.2	1.6	4.77
Gd	5.2	10.9	5.9	8.8	7.3	7.1	7.58	7.16	5.79	6.48	6.05	7.94	5.82	20.2	12.8	8.2	9.7	6.2	9.5	13	7.7	7.7	11	8	19.7
Tb Du	0.7	1.4	0.7	1.1 E 0	0.94	0.7	1.17	1.19	0.934	1.05	0.988	1.26	0.959	2.6	1.7	1.2	1.3	0.8	1.3	2.1	1.1	1.1	1.8	1.1	2.97
Но	0.7	1.2	0.7	1.1	0.78	0.77	0.895	0.949	0.795	0.821	0.829	0.932	0.773	2.3	1.8	1.4	1.3	0.7	1.5	1.6	1.1	0.94	1.5	1.2	1.98
Er	2.2	3.2	1.9	3	2.2	2.1	2.37	2.43	2.22	2.21	2.35	2.41	2.08	6.1	5.1	4.8	3.7	1.9	3.9	4.2	3	2.6	4.4	3.2	4.88
Tm	0.3	0.5	0.3	0.4	0.36	0.35	0.373	0.457	0.368	0.355	0.389	0.385	0.338	0.8	0.7	0.8	0.6	0.2	0.66	0.68	0.52	0.45	0.75	0.53	0.72
Yb	2.1	2.8	1.9	2.8	2	1.8	2.38	2.56	2.35	2.21	2.52	2.37	2.08	4.3	4.5	6	3.6	1.6	4	3.5	3	2.6	4.6	3.3	4.22
Mo	0.3	0.4	0.3	0.4	U.3 pa	0.27 na	0.58	0.456	1.69	0.544	0.409	2.11	2.21	0.6	1.0	0.6	1.9	0.2	0.59 na	0.45 na	0.45 na	0.38 na	0.57 pa	0.40 na	1.01
Cu	165	2.3	117	6.1	na	na	23.8	11.9	24.3	6.77	91.1	29.6	100	7.4	41.3	8.3	45.8	15	na	na	na	na	na	na	7.98
Li	na	na	na	na	na	na	18.4	19.3	22.5	24.9	12.8	21.5	10.8	na	na	na	na	na	na	na	na	na	na	na	29.8
Pb	18.7	12	3.4	25.3	109	55	na	na	na	na	na	na	na	13.1	4.7	18.4	22.4	1.7	97	63	110	83	49	80	na
Zn	83	86	72	97	na	na	89.2	96.2	73.6	72.2	67.9	95.7	120	48	13	65	34	17	na	na	na	na	na	na	92.9
<sup>87</sup> Cr ( <sup>86</sup> Cr (managed)	1.5	0.2	10.0	0.2	0.704071	0.704002	0.704006	0.704120	0.302	0.704660	0.704470	0.704057	0.704105	0.1	0.4	0.5	0.5	0.2	na	na	0 704341	0 704205	па	0.704647	4.50
<sup>87</sup> Sr / <sup>86</sup> Sr (initial)					0.704071	0.704002	0.704096	0.704139	0.704111	0.704060	0.704470	0.704057	0.704195								0,70401	0,70401		0.70438	0.70401
<sup>143</sup> Nd/ <sup>144</sup> Nd (measured)					0.70395	0.70395	0.70397	0.70397	0.70397	0.70410	0.70429	0.70397	0.70395								0.70701	0.70101		5.7 0 100	0.512849
<sup>143</sup> Nd/ <sup>144</sup> Nd (initial)							0.512039			0.5127/0		0.512031		1											0.512820
ENd							4.8			3.0		4.6													4.6
Alkalinity Index (AI)	0.105	0.028	0.058	0.026	0.028	0.056	0.044	0.037	0.046	0.007	0.037	0.056	0.099	0.022	0.015	0.026	0.028	0.032	0.035	0.019	0.028	0.037	0.003	0.033	0.025
FSSI	-0.05	-0.38	-0.12	-0.23	-0.22	-0.02	-0.24	-0.17	-0.15	-0.51	-0.11	0.37	-0.03	-0.1	-0.16	-0.03	-0.38	-0.25	0	0.79	-0.02	-0.12	1.12	-0.05	-0.17
															ļ	ļ				ļ					
* Fe <sub>2</sub> O <sub>3</sub> and FeO contents	s were calcul	lated based o	n a Fe <sup>3+</sup> /Fe <sub>1</sub>	<sub>Fotal</sub> ratio of 0.	.3.																				
Abbreviations: BV = Bureau Veritas, P = Potsdam, RHL = Boyal Holloway London, K = Kiel- na = not analysed- bdl = -									detection lin	nit				1					. –						Т

SYU	SYU	SYU	SYU	SYU		CVU	CVU	CVU	CVU	CVU	CVU	CVU	
2-7-09	2-9-09	3-3-09	3-5-09	3-6-09		2-3-15	2-4-15	2-6-15	2-7-15	2-8a-15	3-7-09	3-8-09	
1	3	12	14	15		32	33	35	36	37	16	17	
RHL	RHL	K	RHL	K		BV	BV	BV	BV	BV	RHL	RHL	
44.59320	44.59835	44.62075	44.61660	44.61053		44.58697	44.58697	44.58893	44.58929	44.58279	44.60143	44.60703	
40.63365	40.63848	40.65353	40.66855	40.67683		40.69464	40.69464	40.69384	40.69375	40.69033	40.67552	40.67572	
56.21	51.44	55.14	58.89	60.17		59.14	57.81	55.77	63.47	61.08	60.21	61.49	
0.11	0.47	0.36	0.44	0.37		0.37	0.36	0.43	0.32	0.34	0.34	0.36	
22.61	20.47	21.01	20.04	10.07		20.12	10.40	21.71	10.61	10.54	20.00	10.63	
22.61	20.65	21.91	20.96	19.05		20.13	19.49	21./1	19.01	16.56	20.00	19.65	
0.72	1.48	0.65	0.96	0.77		0.99	0.97	0.82	0.53	0.99	1.03	0.95	
1.52	3.10	1.37	2.02	1.62		2.07	2.04	0.20	1.11	2.08	2.16	2.00	
0.28	0.18	0.22	0.12	0.17		0.18	0.17	0.20	0.03	0.13	0.17	0.10	
1 49	4.03	2.13	2.24	2.26		1.85	3.04	3.24	0.17	2 71	2 38	1.92	
9.51	4.68	7.63	4.83	5.48		4 95	5.34	7.04	5.91	6.70	6.27	6.25	
6 52	0.72	7.04	7.45	6 70		6.07	6.25	2.62	7 57	4 56	6 52	6.20	
0.32	9.73	7.94	7.43	0.79		0.97	0.33	3.03	7.37	4.30	0.55	0.79	
0.01	0.20	0.03	0.06	0.08		0.15	0.13	0.08	0.04	0.12	0.08	0.07	
1.77	2.44	1.15	2.66	1.44		1.84	2.06	4.32	0.88	0.79	1.19	100.04	
100.79	99.31	90.69	100.92	90.00		99.34	90.51	99.41	99.93	90./4	100.88	100.04	
9.31	4490	pa	170	530		909	906	787	201	1430	546	388	
1.55	7.96	па	5.61	 ра		20.9	43.1	11.3	12.8	4.6	4.81	4.3	
2.98	2.55	na	4.24	na		bdl	bdl	bdl	bdl	bdl	8.56	11.7	
5.56	2.92	na	6.38	na		5	5.8	1.3	1.1	5	7.36	6.59	
na	na	na	na	na		20.2	20.5	19	22.6	19	na	na	
10.9	4.2	na	11.1	na		11.2	11.5	4.5	16.4	11.6	13.2	10.5	
8.2	18	na	54.4	na		31.5	33.6	19.6	43.2	31	41.8	36.4	
407	227	na	226	203		227	205	169	251	183	249	232	
0.759	1.04	na	1.04	na		na 1	na	na	na 2	1	na	na	
50.6	5270	lid na	573	506		861	1990	984	515	1480	855	403	
0.00647	0.752	na	3.12	na		1.4	1.5	0.9	1.8	1400	1.71	1.67	
31.1	15.4	na	85.5	na		45.4	48.9	24.1	109	60.4	59.6	45.4	
10.4	5.73	na	12.7	na		12.1	17.6	3.6	19.9	19.1	19.7	14.6	
14.1	230	na	71.9	na		90	89	201	242	82	71.3	66.4	
856	273	na	677	na		568	612	284	810	487	723	575	
12.3	38.1	na	34.3	na		31.9	31.3	53.9	19	28.4	37.4	37.3	
169	96.7	na	191	138		137	114	181	57.3	94.1	133	121	
187	1/1	na	285	220		184	177	246	11.2	150	204	195	
26.3	74.5	lia na	25.5	68.1		19.1 61.3	55.1	29.8	36.2	14.5	10.4	10.0	
2 35	14.5	na	13.2	10.1		89	82	16.1	62	7	10.3	11.4	
0.338	5.16	na	2.14	2.09		2.1	1.9	4.5	1.1	1.7	2.05	2.22	
3.22	10.6	an	10.4	8.85		7	6.5	14	4.1	5.5	8.09	8.73	
0.295	1.7	na	1.45	1.18		0.9	0.9	1.7	0.6	0.8	1.25	1.31	
1.17	7.04	na	5.91	6.23	_	5.2	5	9.1	3.5	4.4	5.49	5.81	
0.291	1.19	na	1.06	1.17		1	0.9	1.5	0.6	0.8	1.06	1.08	
1.1	2.94	na	2.86	3.25		2.8	2.9	4.1	2.2	2.6	3.08	3.02	
0.229	0.434	na	0.454	0.488		0.4	0.5	0.6	0.4	0.4	0.561	0.529	
0.345	2.04 0.300	na	0.402	0.470		0.5	5.Z	5./ 0 E	2.9	0.1	0.93	0.592	
1.11	1.570	na	1.72	0.479		0.5	3.7	1.2	22	0.5	1.44	2.59	
2.33	20.7	na	70.4	na		32	21.1	1.2	3.2	17.9	29.5	18	
66.3	17.7	na	20.9	na		na	na	na	na	na	25	38.5	
na	na	na	na	na		20.5	25.2	4.2	20.4	13.7	na	na	
181	104	na	77.3	85	_	60	51	61	55	37	100	88.7	
0.796	2.87	na	3.33	16		2.3	0.6	0.5	0.6	0.6	0.881	1.7	
0.718557	0.704075	0.706855	0.704695	0.704873							0.704519	0.704987	
0.70520	0.70400	0.70424	0.70404	0.70414							0.70404	0.70403	
		0.512837											
		0.512813											
		4.4			_					-			
-0.003	0.021	0.005	0.047	0.025		0.044	0.038	0.061	0.017	0.026	0.023	0.017	
-0.63	-0.55	-0.55	-0.07	-0.07		-0.02	-0.11	-0.16	-0.01	-0.03	-0.16	-0.13	
						1							

TZ-6-8-12	Laborator	ry ID: C	15038	Irradiati	on ID:	PO-2														
	<sup>40</sup> Ar/ <sup>39</sup> Ar		<sup>40</sup> Ar/ <sup>39</sup> Ar			<sup>37</sup> Ar/ <sup>39</sup> Ar			<sup>36</sup> Ar/ <sup>39</sup> Ar			K/Ca	<sup>40</sup> Ar*	<sup>39</sup> Ar <sub>K</sub>	<sup>40</sup> Ar*/ <sup>39</sup> Ar <sub>K</sub>			Age	±	1s
J=0.0009720							(×10	)-3			(%)	fraction (%)				(Ma)				
Laser output																				
1.8%	1397	±	251	7.4	±	137	3703	±	710.85	0.08	21.73	0.13	305	±	98	469	±	132		
2.0%	1385	±	938	141	±	566	4558	±	3110	0.00	3.54	0.03	54	±	147	93	±	245		
2.4%	352	±	60	6	±	149	919	±	203	0.10	22.92	0.14	81	±	44	137	±	71		
2.8%	77	±	3	39	±	15	141	±	33	0.01	49.59	0.82	39	±	10	67	±	17		
3.1%	37.0	±	0.4	7	±	4	42	±	5	0.09	67.55	4.85	25.1	±	1.4	44	±	2		
3.3%	29.79	±	0.16	6	±	2	20	±	3	0.10	81.79	13.29	24.5	±	0.9	42.4	±	1.5		
3.5%	29.80	±	0.14	1.6	±	1.7	18.7	±	1.7	0.37	81.82	21.93	24.4	±	0.5	42.3	±	0.9		
3.7%	30.8	±	0.2	2.4	±	0.6	26.9	±	1.0	0.25	74.79	46.26	23.1	±	0.4	40.0	±	0.6		
3.9%	31.1	±	0.4	2.4	±	1.9	27	±	3	0.25	74.67	12.56	23.3	±	0.9	40.4	±	1.5		
												Total g	as age			42.05	±	0.52		
											Plateau age (step 5-9: 98.8% of total <sup>39</sup> Ar)						±	0.46		
											No	ormal isochron	41.32	±	2.51					
											In	verse isochron	age (ste	p 5 to	9)	41.25	±	2.11		

Table 2: Argon isotopic data for an amphibole separate of syenite sample 6-8-12.

### **Supplementary material**

#### Whole rock major and trace element analyses

#### 1. Royal Holloway University, London, UK

Fourteen samples were analysed by inductively coupled plasma atomic emission spectrometry (ICP-AES) for major elements and some high abundance trace elements (Sr, Zr, Cr, Sc, Zn, Co, Li, V, Be and Ni) and by inductively coupled plasma mass spectrometry (ICP-MS) for low abundance trace elements (Rb, Nb, Y, Mo, Cs, Ba, Hf, Ta, Tl, Pb, Th, U, and all REE) using a Perkin Elmer instrument. The analytical work followed the methodology described by Walsh et al. (1981) and Garbe-Schönberg (1993), respectively. The relative standard deviation (RSD) typically was  $\leq 2\%$  for major elements and  $\leq 5\%$  for minor and trace elements.

### 2. Institute of Geosciences, Kiel University, Germany

Three samples were analysed by X-ray fluorescence (XRF) on fused glass discs using a Philips PW1480 XRF spectrometer for major elements and by ICP-MS using an Agilent 7500c instrument for trace elements. For major element oxides, the RSD is  $\leq 1.3$  % based on multiple analyses of reference material BHVO-1. The RSD for trace elements is generally  $\leq 2$  % based on multiple analyses of one sample solution. Details about sample preparation and instrument calibration are given in Garbe-Schönberg (1993) and John et al. (2008), and representative data for precision and accuracy during the course of this study are provided by Laeger et al. (2013).

#### 3. GeoForschungsZentrum (GFZ) Potsdam and Potsdam University, Germany

Nine samples were analysed for major and some trace elements (Ba, Cr, Ga, Nb, Ni, Rb, Sr, V, Y, Zn and Zr) by XRF using a Siemens SRS303-AS XRF spectrometer at the GFZ and for REE by ICP-AES using a Varian Vista MPX instrument following the methods described by Zuleger and Erzinger (1998). RSD values are in the range of 1-3% for major oxides and  $\leq$  5% for trace elements and REE (Moazzen and Oberhänsli, 2008; Hadj Zobir et al., 2014).

### 4. AcmeLabs, Bureau Veritas Minerals, Vancouver, Canada

Fifteen samples were analysed for major elements by XRF using a Panalytical Axios Max instrument and by ELAN 9000 ICP-MS for trace elements and REE. The RSD is <1.2% for major oxides based on the analyses of SY-4(D) diorite gneiss and OREAS72B VMS ore standards, while for trace elements and REEs the RSD was <3.8%.

#### Strontium (Sr) and neodymium (Nd) isotope analyses

Strontium (Sr) and neodymium (Nd) isotope analyses were performed on a Thermo Finnigan Triton multicollector mass spectrometer at the School of Earth and Environment, University of Leeds. About 30 to 60 mg of powdered whole-rock material (same was used for the major and trace element work) was dissolved in concentrated ultra-clean HF-HN03-HCl acids and Sr and Nd were extracted from the unspiked solutions by conventional ion-exchange chromatographic techniques (see Halama et al. 2013 for details of the analytical protocol). <sup>87</sup>Sr/<sup>86</sup>Sr and <sup>143</sup>Nd/<sup>144</sup>Nd ratios were normalized for mass fractionation to <sup>86</sup>Sr/<sup>88</sup>Sr = 0.1194 and <sup>146</sup>Nd/<sup>144</sup>Nd = 0.7219. The average <sup>87</sup>Sr/<sup>86</sup>Sr obtained from replicate measurements of NIST SRM-987 during this study was 0.710254 and all data were corrected for the offset from the generally accepted value 0.710250 (McArthur et al. 2000). Similarly, Nd isotope data were corrected for the offset from the LaJolla reference material (<sup>143</sup>Nd/<sup>144</sup>Nd = 0.511853; Weis et al. 2005). Initial <sup>87</sup>Sr/<sup>86</sup>Sr isotope ratios were calculated using the <sup>87</sup>Rb decay constant 1.3972 x 10<sup>-11</sup> a<sup>-1</sup> (Villa et al. 2015). For the calculations of the  $\epsilon$ Nd values, the following parameters were used: <sup>147</sup>Sm decay constant  $\lambda = 6.54 \times 10^{-12} a^{-1}$ , present-day (<sup>143</sup>Nd/<sup>144</sup>Nd)<sup>CHUR</sup> = 0.512638, (<sup>147</sup>Sm/<sup>144</sup>Nd)<sup>CHUR</sup> = 0.1966.

## <sup>40</sup>Ar/<sup>39</sup>Ar dating

About 1 mg of amphibole from syenite sample 6-8-12 was used for <sup>40</sup>Ar/<sup>39</sup>Ar analysis by the CO<sub>2</sub> laser stepwise heating technique at the Institute of Earth and Environmental Science,
Universität Potsdam. For details of the analytical protocol see Wilke et al. (2010) and Halama et al. (2014). Mineral grains were obtained by crushing, sieving and selecting the size fraction between 250-500 µm mesh size for magnetic separation and finally by hand-picking under the binocular. Separated amphiboles were cleaned ultrasonically in 10% HNO<sub>3</sub> for 15 minutes and then washed in de-ionized water and dried. Samples, the Fish Canyon Tuff sanidine age standard, prepared by the Geological Survey of Japan (27.5 Ma: Uto et al., 1997; Ishizuka, 1998) and salts of K<sub>2</sub>SO<sub>4</sub> and CaF<sub>2</sub> were irradiated at the Oregon State TRIGA Reactor for 4 hours under a neutron flux of 2.5x10<sup>13</sup> n cm<sup>-2</sup> s<sup>-1</sup>. Argon isotope ratios of the gas from the samples were analyzed by stepwise heating until total fusion using a New Wave Research DualWave laser ablation system comprising a 50W CO<sub>2</sub> continuous laser with 10.6 µm wavelength. The extracted gas is purified in the ultra-high vacuum line via SAES getter pumps and a cold trap for 10 min. The high sensitivity Micromass 5400 noble gas mass spectrometer used for Ar isotopic analysis is equipped with an electron multiplier pulse counting system for analyzing small amounts of Ar. Raw data were corrected for procedural blank contributions, mass discrimination by analysis of atmospheric Ar, interferences of Ar isotopes derived from Ca and K and decay of radiogenic <sup>37</sup>Ar and <sup>39</sup>Ar isotopes produced by irradiation. Calculation of ages and errors was performed following Uto et al. (1997) using the total <sup>40</sup>K decay constant of 5.543 x 10<sup>-10</sup> a<sup>-1</sup> (Steiger and Jäger, 1977) as well as decay constants of 1.978 x 10<sup>-2</sup> d<sup>-1</sup> for <sup>37</sup>Ar and 2.58 x 10<sup>-3</sup> a<sup>-1</sup> for <sup>39</sup>Ar.

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