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1 **Geochemical insight during archaeological geophysical exploration**
2 **through in situ X-ray fluorescence spectrometry**

3 (Short title: In situ archaeological XRF exploration)

4

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14

15 **Abstract**

16 Geophysical techniques are widely applied in archaeological exploration, providing
17 rapid and non-invasive site appraisal. Geochemical analyses contribute significantly
18 in archaeometry, but conventional laboratory apparatus requires that samples are
19 removed from their in situ context. Recent advances in field-portable apparatus
20 facilitate in situ geochemical analysis, and this apparatus is deployed in this paper
21 alongside conventional geophysical analysis to characterise the archaeological
22 prospectivity of a site. The target is subsurface debris at the crash site of a Second
23 World War Mosquito aircraft.

24

25 A 100 m long transect of magnetic, electromagnetic (EM) and in situ XRF
26 measurements was acquired in November 2014, with soil samples also collected for
27 laboratory validation. A subset of in situ XRF measurements was repeated in August
28 2015 alongside a targeted grid, 900 m² in area, of magnetic gradiometry profiles. Built
29 chiefly from wood, the Mosquito responds weakly in magnetic and EM data; magnetic
30 gradient anomalies of ± 10 nT/m are instead attributed to thermoremanence in a burnt
31 layer at 0.2-0.4 m depth, produced by the impact fire following the crash. XRF
32 spectrometry reveals co-located enrichments in copper and zinc ions (400% and
33 200%, respectively, above background). These metals are alloyed into brass, present
34 in abundance in the ammunition on board the Mosquito.

35

36 Records from the in situ XRF sampling compare well with laboratory validated data,
37 although a bespoke calibration for the local soil type would improve the reliability of
38 absolute geochemical concentrations. XRF responses vary significantly with ground
39 conditions: the November 2014 acquisition was performed soon after ploughing at the
40 site, potentially providing a fresh charge of metallic contaminants to the ground
41 surface. Where the chemistry of a target is anomalous with respect to host soil and a
42 source-to-surface transport mechanism is present, in situ XRF analysis offers
43 improved understanding of a target compared to geophysical interpretation alone.

44

45 **Keywords:** XRF spectrometry, magnetometry, geophysics, geochemistry, air crash,
46 conflict archaeology

47

48 **Introduction**

49 The detection and delineation of archaeological targets is often based on contrasts in
50 the subsurface distribution of physical properties (density, electrical conductivity,
51 magnetic susceptibility, etc.). The chemical composition of the target is typically of
52 secondary importance, and it usually suffices to observe a response consistent with
53 (e.g.) a metallic target rather than identifying the specific metallic elements involved.
54 However, an understanding of geochemistry could add significant archaeological
55 value, particularly where a given practice is linked to a specific chemical element (e.g.,
56 industrial processes that leave a diagnostic abundance of marker elements; Millard,
57 1999; MacKenzie and Pulford, 2002). While geochemical analyses are routine in
58 archaeometry, applied to study the composition of (e.g.) glass (Aidona et al., 2001;
59 Falcone et al., 2008), paint (Bonizzoni et al., 2011) and ceramic (Aidona et al., 2001;
60 Cultrone et al., 2011), their use in archaeological exploration has not been widely
61 reported.

62

63 Among the advantages of most geophysical methods is the in situ and non-destructive
64 nature of survey. By contrast, geochemical analyses are usually conducted in the
65 laboratory, on prepared (often destroyed) samples of material extracted from a site
66 (e.g., Wilson et al., 2008; Cook et al., 2010; Dirix et al., 2013; Vittori Antisari et al.,
67 2013; Carey et al., 2014; Scott et al., 2016). In situ geochemical analysis using X-ray
68 fluorescence (XRF) spectrometry has become possible, however, with the
69 development of handheld, field-portable XRF analysers (e.g., Lu et al., 2010;
70 Schneider et al., 2016). Such in-field equipment is particularly valuable in forensic
71 cases (e.g., Bergslien, 2013) since the material under inquiry must be uncontaminated

72 by any sampling process. The use of field-portable XRF spectrometers is reported for
73 archaeological applications (e.g., Shugar, 2013; Fernandes et al., 2013; Del Solar
74 Velarde et al., 2016; Sepúlveda et al., 2015) but seldom using a sampling strategy
75 consistent with standard geophysical survey (i.e., spot samples are considered, rather
76 than systematically-defined areas and/or transects).

77

78 Here, in situ XRF spectrometry is applied as part of a conventional deployment of
79 magnetic and electromagnetic methods to characterise a potential archaeological site,
80 specifically the crash-site of a Second World War aircraft. The additional geochemical
81 insight reduces the ambiguity in the interpretation of the geophysical data: geophysical
82 anomalies are co-located with enriched concentrations of copper and zinc ions,
83 associated with brass (copper-zinc) alloy in the aircraft's ammunition. The in situ data
84 compare favourably to XRF and mass spectrometry applied under laboratory
85 conditions, but the same survey locations show variability given the changing supply
86 of chemical elements to the ground surface. In situ XRF spectrometry can offer a
87 valuable complement to a campaign of exploratory field geophysics, but only under
88 certain site conditions as considered in discussion.

89

90 **X-Ray Fluorescence Spectroscopy – Fundamental Theory**

91 X-ray fluorescence spectroscopy determines the elemental composition of a sample
92 material using high-energy, short-wavelength (X-ray) radiation (note: spectroscopy
93 and spectrometry are distinct; the former is a technique, whereas the latter is the
94 quantitative analysis of data). When bombarded with X-ray radiation, different

95 elements can be identified by the characteristic ‘fluorescent’ energy that they emit
96 (Weltje and Tjallingii, 2008).

97

98 XRF responses are adversely affected by several factors, including matrix composition
99 (Hall et al., 2014; Quye-Sawyer et al., 2015), surface morphology (Potts et al., 1997;
100 Forster et al., 2011; Shugar, 2013) and instrumental sensitivity (Weltje and Tjallingii,
101 2008). Matrix composition effects are mitigated using manufacturers’ calibrations for
102 representative materials (e.g., mudrock, glass, alloys, etc.). Although challenging to
103 define, bespoke calibrations can be made (Quye-Sawyer et al., 2015; Scott et al.,
104 2016) and allow the XRF data are to be used as an absolute rather than relative
105 indicator of composition (Środoń et al., 2001). Laboratory XRF practice mitigates the
106 effects of surface morphology by (destructively) grinding samples into a fine powder.
107 Equivalent sample preparation is impractical for in situ XRF spectrometry hence field-
108 portable XRF instruments have faced scepticism in the geochemical community
109 (Frahm, 2013). However, recent research (e.g., Schneider et al., 2016) has reported
110 similar accuracy and precision between field- and laboratory-based observations.

111

112 The instrument deployed here is a hand-held Bruker Tracer IV-SD spectrometer
113 (Figure 1), an energy-dispersive instrument with a rhodium target. The detection of
114 elements lighter than calcium can be challenging since these have a low ‘fluorescence
115 yield’ (i.e., their energy emissions are weak; Krause, 1979; Berlin, 2011), but this is
116 overcome here with the use of a silicon drift detector (Speakman et al., 2011).
117 Sensitivity is further improved by including a Bruker 3V Vacuum Pump (Figure 1) to
118 inhibit the attenuation of fluorescent energy by air in the spectrometer’s analysis

119 chamber. The presence of water also impedes XRF analysis, since water scatters
120 the X-ray radiation; therefore, in situ XRF surveys may always be vulnerable to the
121 presence of groundwater (e.g., Tjallingil et al., 2007), especially for low-yield elements.

122

123 The sample area (spot size) of an XRF measurement is typically 1 cm in diameter.
124 However, the depth penetration of XRF energy in soil is on the millimetre-to-centimetre
125 scale, hence in situ XRF measures only the surface chemistry of host soil. While it
126 may be detectable with geophysical methods, a target would therefore be invisible to
127 XRF sampling unless the ground surface is enriched in relevant marker elements via
128 some source-to-surface transport mechanism (e.g., ploughing, groundwater
129 circulation; Hedges and Millard, 1995; Campana, 2009). Even then, such transport
130 may not only be in a vertical direction hence the strongest concentrations of ions may
131 not be observed directly above the source. As such, in situ XRF prospection will likely
132 always benefit from the constraint provided by conventional geophysical survey.

133

134 **Field Survey**

135 This study is located within Nuthampstead Airfield (Hertfordshire, UK; National Grid
136 Reference TL419347), over the suspected crash site of a Second World War aircraft
137 (Figure 2). A local borehole, BGS TL43NW39 (Figure 2a, UK National Grid Reference
138 TL 41404 35166), shows surface soils are clay-rich, beneath which is stiff clay to a
139 depth of at least 40 m with occasional flint and chalk cobbles (BGS, 2014). The
140 underlying solid geology – the Lewes and Seaford Chalk formations – is observed at
141 ~ 70 m depth. Surveys over the likely crash site were undertaken in November 2014
142 and August 2015, in support of investigations conducted at the site by Nuthampstead
143 Airfield Museum.

144

145 The specific aircraft believed to have crashed at the site is de Havilland DH.98
146 Mosquito Mark IV 'LR343'. In February 1944, at the time of the crash, Nuthampstead
147 Airfield was operated by the United States Army Air Force; it returned to British control
148 in 1945, and was decommissioned and returned to agricultural use in 1959. The
149 Mosquito crashed in the grounds of Nuthampstead shortly after its take-off from RAF
150 Hunsdon (also in Hertfordshire). Records suggest that the port engine detached from
151 the aircraft, causing it to invert and impact the ground at near-vertical angle. The crash
152 caused an intense fire, and claimed the lives of the two crewmen (members of 487
153 Squadron Royal New Zealand Air Force). Their bodies were recovered from the site,
154 along with some wreckage, but it is doubtful that all debris was cleared from the site
155 and some components (including armaments and the starboard engine) may remain
156 present today.

157

158 The airfield has been extensively ploughed, but runways still remain and evidence of
159 military infrastructure remains present in as cropmarks. The likely crash site has been
160 identified by Nuthampstead Airfield Museum using contemporary photographs of the
161 impact (e.g. Figure 3a) plus a local concentration of surface finds identifiable as
162 Mosquito wreckage (Figure 3b). An exploratory geophysical survey was therefore
163 conducted, in November 2014, to investigate this hypothesis (Figure 2c): magnetic
164 gradiometer and electromagnetic (EM) methods were deployed along a transect over
165 the hypothesised crash site. The opportunity also arose to deploy in situ XRF analysis
166 alongside the geophysical methods. Promising initial results from the XRF survey
167 motivated a second acquisition in August 2015, in which a repeat set of XRF
168 measurements and a larger grid of magnetic data was acquired (Figure 2c).

169

170 Before describing these surveys in more detail, the detectability of the Mosquito
171 aircraft is considered; first by geophysical survey, then through geochemical analysis.

172

173 **Geophysical Detection of the Target**

174 The wingspan of a Mark IV Mosquito is 16.5 m, and it is 12.5 m nose-to-tail. In
175 horizontal flight, the tip of its fin and rudder is 3.8 m above the base of its belly (Figure
176 4). The speed and steep angle of impact into soft clay soil suggests that any remaining
177 components of the Mosquito could be buried at up to 6 m depth (Masters and Osgood,
178 in press).

179

180 Most surveys for aircraft wreckage can exploit the presence of aluminium and/or steel
181 in the ground (i.e., relying on contrasts in electrical and/or magnetic properties; e.g.,
182 Osgood, 2014), but the Mosquito was one of the few Second World War aircraft to be
183 made chiefly of wood. Aluminium is only used in the rudder and elevator and, at this
184 site, the steel engine and armaments may not be present. Therefore, in addition to
185 any remaining aircraft components, it was assumed that magnetic surveying could
186 respond to any thermoremanent magnetic signature of the impact fire. Despite ~ 55
187 years of ploughing at the site, trial excavations suggested that a layer of intense
188 burning remains present at 0.2-0.4 m depth (Figure 3b). Melted aluminium fragments
189 are among the debris observed at the site hence the temperature of the fire must have
190 exceeded 660°C, the melting point of aluminium. This exceeds the temperature at
191 which iron minerals in the soil transform into ferrimagnetic iron oxides (200-400°C;
192 Schwertmann and Taylor, 1989; Hanesch et al., 2006), and then approaches the Curie

193 temperature for haematite (675°C; Herz and Garrison, 1998). Magnetic survey
194 methods were therefore prioritised, with additional electromagnetic surveys in case of
195 remaining metallic debris.

196

197 **Geochemical Detection of the Target**

198 With little precedent for similar XRF practice, it was initially unclear which elements
199 could diagnose the crash site. While aluminium (Al) enrichment might ordinarily be
200 consistent with buried aircraft wreckage, this is unlikely to be significant for the wooden
201 Mosquito. Additionally, any small Al anomaly may be masked by the high background
202 aluminium content in Nuthampstead's clay soil and, furthermore, attenuated by
203 groundwater. To identify alternative geochemical targets, the XRF characteristics of
204 surface debris from the putative crash site were considered, including:

- 205 1) brass ammunition cartridges: British cartridge brass from the Second World
206 War period, used in .303 ammunition, is an alloy of 70% copper (Cu) and
207 30% zinc (Zn), occasionally containing small quantities of lead (Pb).
208 Cartridges may also have jacket of cupronickel alloy. None of the cartridges
209 recovered show signs of melting (the melting point of most brass alloys
210 exceeds 900°C), but all had exploded.
- 211 2) cannon rounds: this ammunition is made principally from steel, possibly
212 alloyed with a nickel-chromium-molybdenum (Ni-Cr-Mo) blend. British
213 aircraft carried several variants: armour-piercing ammunition may be tipped
214 with a tungsten (W) carbide alloy, whereas explosive and incendiary
215 variants have TNT and phosphorous (P) cores, respectively.

216 3) burnt wood: although dominated by light elements (e.g., carbon, oxygen),
217 traces of heavier elements, such as Pb, could be present in any paint
218 residue.

219 In addition to these fragments, a sample of burnt soil was tested to monitor any
220 chemical alteration caused by the impact fire.

221

222 Figure 5 shows the concentrations of elements in the debris fragments, expressed in
223 parts per million (on a log scale due to the variability between elements). All XRF
224 analyses use a “mudrock” calibration for which the spectrometer operates at 40 kV.
225 This manufacturer-defined setting was the most appropriate for Nuthampstead’s clay
226 rich soil, though this implies that the measured concentrations are relative rather than
227 absolute indicators. Elements lighter than calcium and those too scarce to be detected
228 (e.g., molybdenum, tin, antimony), are absent from this plot. Each concentration is
229 compared to a background value (orange bars, Figure 5), with error bars spanning the
230 observed concentration range. This background value is defined for each element as
231 the mean concentration recorded along the transect in a subset of locations outside
232 the magnetic survey grid (Figure 2c, omitting transect positions at 45-75 m). While it
233 is possible that background concentrations are influenced both by any unknown land
234 use at the site and the crash itself, this subset of samples is at least outside of the
235 area of surface-scattered fragments. As such, these concentrations define a local
236 background to which observed geochemical anomalies can be compared.

237

238 The brass sample (green, Figure 5) is dominated by Cu, with concentration exceeding
239 10^5 ppm. A high Zn fraction is also recorded (~80,000 ppm), with arsenic (As) and

240 nickel (Ni) also increased in abundance. The steel sample is Fe-enriched, although
241 with a surprisingly low concentration of ~250,000 ppm. The low value could again
242 indicate a calibration issue, or non-ideal conditions of the sample surface caused by
243 corrosion (Scott et al., 2016). Pb is somewhat enriched in both metallic samples, but
244 in very low concentrations which may approach the limit of instrumental sensitivity.
245 The burnt wood sample is generally depleted in metallic elements although no element
246 is obviously enriched against the background trend. The burnt soil samples shows
247 little significant alteration with respect to background.

248

249 Despite the vulnerability to calibration effects, any geochemical anomaly presented by
250 the Mosquito would likely be in elements associated with brass, specifically Cu and
251 Zn. In addition to ammunition, the Mosquito was held together with ~50,000 brass
252 screws, therefore brass may be highly abundant in the ground. While Fe could also
253 have been an attractive target, the concentrations of Cu and Zn are more significant
254 above the background geochemistry, and its associated variability, in our observations
255 at Nuthampstead.

256

257 **Survey Procedure**

258 Surveys in November 2014 were conducted along a transect, 100 m in length, which
259 intersected the suspected crash site at a distance of 50-60 m (Figure 2c). Geophysical
260 surveys were conducted with a Geonics G-858 caesium vapour magnetic gradiometer
261 and a Geometrics EM31 electrical conductivity meter. Both instruments were used in
262 a 'continuous' recording mode, resulting in an along-transect sampling interval of ~ 0.1
263 m. The two sensors of the G-858 were mounted at 0.4 m and 1.2 m above the ground.

264 The EM31 was carried at a height of 1.1 m, with the antenna boom orientated parallel
265 to the transect; electrical conductivity and in-phase components of the EM31 response
266 were recorded, since in-phase anomalies are particularly indicative of buried metal
267 (McNeill, 1983). XRF measurements were conducted with the Bruker spectrometer
268 along the transect at intervals ≤ 2.5 m, with each position irradiated for ~ 1 minute.

269

270 Soil samples were also taken from each XRF survey position, for laboratory validation.
271 Laboratory XRF analysis was conducted with the Bruker spectrometer on soil samples
272 that were kiln-dried for several days, at 60°C, then ground with a pestle and mortar.
273 Selected samples (17 in total) were also analysed by inductively coupled plasma mass
274 spectrometry (ICP-MS). ICP-MS is regarded as a more precise means of
275 quantitatively measuring elemental composition than XRF (Pye and Croft, 2007), and
276 is less vulnerable to calibration issues, but requires more extensive preparation of
277 samples. Aliquots of 100 mg of dried-and-ground soil were dissolved in 5 ml of hot
278 Aqua Regia (37% hydrochloric acid and 68% nitric acid, in a molar ratio of 3:1) at
279 140°C for 1 hour. A dilution series of 1:100 was made in 2% nitric acid and analysed
280 for elemental concentrations on an Agilent ICP-MS instrument. The soil samples
281 completely dissolved in Aqua Regia, and hence, the ICP-MS results are representative
282 of the chemical composition of the whole sample, similar to XRF results. It is however
283 known that components such as quartz will not dissolve in Aqua Regia, and could
284 result in small differences in the measured chemical composition determined by ICP-
285 MS of the dissolved material of the sample versus XRF on the whole material of the
286 sample. For the elements of consideration in this study, comparisons were made of
287 reported XRF versus Aqua Regia digestion ICP-MS measurements of standard soil

288 samples to confirm that those elements did not show significant differences between
289 the two methods.

290

291 During August 2015, a grid of dimensions 30 m x 30 m (Figure 2c, British National Grid
292 coordinate [542379 mE, 235690 mN]) was acquired with a Bartington Grad601 vertical
293 component fluxgate gradiometer. Grid profiles were separated by 1 m, orientated
294 along bearing 341°161° with an along-profile sampling interval of ~ 0.1 m. The
295 transect was resurveyed with the XRF spectrometer between distances of 40-80 m
296 (i.e., the span of the grid along the transect) at a sample resolution of 1 m.

297

298 Ground conditions during the two surveys were markedly different. In November 2014,
299 the site had recently been ploughed and sown with a crop of winter barley. The survey
300 followed a period of heavy rain and the soil was waterlogged. By contrast, the August
301 2015 survey took place after prolonged warm weather: the field was dry and dusty,
302 and covered with the stubble of a barley crop (Figure 1).

303

304 **Results**

305 **Geophysical Surveys**

306 Figure 6 shows a compilation of geophysical observations from the two surveys. A
307 magnetic anomaly is detected with the Bartington Grad601, specifically in the eastern
308 half of the survey grid close to the hypothesised crash site. The anomaly appears to
309 be elongate in an east-west orientation, extending 16 m across the profiles of the grid.
310 The typical magnetic gradient anomaly in this region has a magnitude of ± 10 nT/m

311 (against a mean background of approximately -1 nT/m) but exceeds ± 100 nT/m in
312 certain profiles (e.g., at 19 m).

313

314 Two prominent anomalies are also detected with the G-858 magnetometer. A total
315 field anomaly of ~ 4 nT is observed in both of the instrument's sensors between 55-60
316 m along the transect, and in the lower sensor at 70 m. The response from the lower
317 sensor contains more short-wavelength responses, likely arising from the increased
318 proximity to the surface scatter of debris. However, both anomalies are consistent
319 with responses in the Grad601 grid. The position of the broader anomaly conforms
320 with the main anomaly in Figure 6a, and a local 'blip' in the Grad601 grid is also
321 observed 70 m along the transect.

322 No anomalies are observed in either component of the EM31 record at a location
323 consistent with the magnetic anomalies. The stability of conductivity and in-phase
324 responses (~ 50 mS/m and 4 ppt, respectively) in the vicinity of the magnetic anomaly
325 suggests that there are no large metallic fragments buried along the transect (at least
326 within the few metres of EM31 depth sensitivity).

327

328 **In situ XRF Spectrometry**

329 Figure 7a shows the variation in XRF responses observed along the transect in the
330 November 2014 survey (for selected elements; the full suite is available in
331 supplementary material). The lines in each plot are a 3-point moving median trend;
332 light and dark shading corresponds respectively to the span of the Grad601 grid and
333 the extent of the magnetic anomaly (56-62 m along the transect). An enrichment
334 anomaly is observed for Cu ~ 60 m along the transect, with peak values 400% above

335 the 50 ppm background concentration. A moderate Zn anomaly is perceived at ~60
336 m with a peak approaching 200 ppm (< 200% above the background of ~100 ppm).
337 Weaker evidence of an anomaly in As concentrations is also observed. While other
338 elements show no consistent trend other than a consistent distribution across the
339 transect, Fe appears to become more scattered in the most northerly 30 m of the
340 transect, with greater variation (exceeding 100,000 ppm) about the background. For
341 the repeat acquisition in August 2015 (Figure 7b), background concentrations show a
342 comparable magnitude to the earlier archive, but Cu, Zn and As in particular show
343 greater scatter about the median trend; yet Cu appears to be consistently enriched
344 ~60 m along the profile. However, changes in the observed concentrations suggest a
345 fundamental control of ground conditions which is revisited in the discussion section.

346

347 If the geochemical anomalies share a common source, it may be expected their
348 concentrations are correlated at positions along the transect. Following Bergslien
349 (2013), correlation was classified using Spearman's rank correlation coefficient (r_s).
350 Figure 8 shows the correlation between different elements, with symbols coloured
351 according to their distance along the transect. The frames in each plot are coloured
352 according to the strength of correlation: green defines a strong correlation ($r_s > 0.65$),
353 red a moderate correlation ($0.45 < r_s < 0.65$) and black a weak correlation ($r_s < 0.45$)
354 as no correlation. For clarity, only correlations between Cu, Zn and Pb are shown
355 (others are included in supplementary material). For the 2014 archive (upper-right
356 quadrant), strong correlations are observed between Cu and Zn; the strongest
357 correlations are observed 50-70 m along the transect (magenta and red symbols).
358 Correlations are all reduced in the 2015 dataset (lower-left quadrant), attributable to
359 the higher degree of scatter in the observed concentrations. Nonetheless, a moderate

360 correlation is still perceived between the concentrations of Cu and Zn, in the 50-70 m
361 division of the plots.

362

363

364 **Laboratory XRF and ICP-MS Spectrometry**

365 Figure 9 shows results from laboratory XRF and mass spectrometry analysis of soil
366 samples, recovered along the transect during the November 2014 survey. ICP-MS
367 analysis can be extended to aluminium, hence Figure 9b includes an Al response.

368

369 Laboratory XRF data (Figure 9a) are broadly consistent with results from the in-field
370 survey, with anomalies identifiable in Cu, Zn and As. Cu and Zn show anomalies
371 approaching 150 ppm above background values (~50 ppm and ~130 ppm
372 respectively); the weaker As anomaly reaches ~30 ppm above a background of 25
373 ppm. However, concentrations recorded for all elements show considerably less
374 variation about their median trend than for either archive or in situ data. For example,
375 the root-mean-square variability of in situ Cu observations about their median trend is
376 25 ppm, but this reduces to 7 ppm for the laboratory analysis.

377

378 Concentrations determined through ICP-MS analysis (Figure 9b) are of the same order
379 of magnitude as the equivalent XRF data, but differences in base-levels (evident for
380 Ni, Cu, Zn and As) are evident. These are attributed to the inappropriate calibration
381 of the XRF survey, implying that these in situ surveys should be considered relative
382 rather than absolute indicators of concentration. Nonetheless, anomalies in Cu and
383 Zn remain well-defined, ~60 m along the transect, but trends in As and Pb are

384 inconsistent. A Pb anomaly is distinct in the ICP-MS record, approaching ~20 ppm
385 above background. The XRF energies for As and Pb are very similar: 10.543 keV for
386 $K\alpha$ for As and 10.551 keV for $L\alpha$ for Pb. Therefore, the spectral interference between
387 these elements makes it challenging for XRF to distinguish between As and Pb,
388 particularly at low concentrations. As such, the XRF anomaly in As is likely a false
389 positive. Pb is feasibly associated with the crash, since Second World War aircraft
390 were balanced using lead weights.

391

392 It is worth noting that ICP-MS gives evidence of an aluminium anomaly. While the
393 variability of the observed concentrations impedes its definition, Al concentrations
394 appear consistently high 60-70 m along the transect, approaching 10,000 ppm (~5%)
395 above background.

396

397

398 **Interpretation and Discussion**

399 **Crash Site Prognosis**

400 Significant geophysical and geochemical anomalies (Figures 6 and 7) are observed at
401 the study site, which appear consistent with an aircraft crash at this location.
402 Specifically, these are a widespread magnetic anomaly and enriched concentrations
403 of elements associated with brass alloy.

404

405 The low-amplitude magnetic anomalies observed in both the Grad601 grid and the G-
406 858 transect are interpreted as the response to the thermoremanence in burnt clay.

407 Assuming a near-vertical impact, the area of this response is not inconsistent with the
408 footprint of the Mosquito (~16 x 4 m), which would have been affected by the impact
409 fire. Additionally, the power spectrum of the G-858 response indicates that the
410 magnetic source is located within 1 m of the ground surface, based on modelling the
411 burnt layer as a thin layer with random magnetisation. Spector and Grant (1971) show
412 that for a vertically-extended random magnetic layer, the slope of linear sections of a
413 power spectrum of log-power vs. wavenumber (= 1/wavelength) is a factor of 4π times
414 the source depth. Figure 10 shows the spectrum for the upper sensor of the G-858
415 (Figure 6b), mounted at ~1.2 m from the ground; this power spectrum has been
416 modified for thin layer sources. The spectrum contains three linear sections, the first
417 of these (i, wavenumber < 0.6 m^{-1}) has a gradient of $-24.7 \pm 1.4 \text{ m}$, implying that the
418 causative body is $2.0 \pm 0.1 \text{ m}$ away from the sensor, i.e., at a depth of $\sim 0.8 \pm 0.1 \text{ m}$.
419 To a first-order, this is comparable with the depth extent of the burnt material observed
420 during small excavations at the site (Figure 3b). The second linear section (ii)
421 suggests a body 0.2 m from the sensor, but this likely relates to the 'wobble' in the
422 position of the sensor during continuous data acquisition. The third section (iii) has a
423 very low gradient, and most-likely corresponds to ambient magnetic noise.

424

425 The higher amplitude magnetic anomalies ($> \pm 100 \text{ nT/m}$) observed in the Grad601
426 grid could be responses from larger fragments of ferrous wreckage, but further survey
427 would be required to evaluate the size and/or depth of these potential targets.

428

429 This interpretation is greatly strengthened by the XRF spectrometry. Co-located with
430 the magnetic anomalies are local geochemical anomalies, particularly evident for

431 elements (Cu and Zn) associated with brass. Besides iron and aluminium, brass is
432 the most significant metallic component of the fully-armed Mosquito aircraft. The
433 geochemical evidence is particularly compelling since, in the absence of other
434 information, the air-crash is the most plausible means of introducing these elements
435 into the ground at this location; by contrast, the burnt layer alone could be more simply
436 explained by (e.g.) disposal at some point in the recent history of the site. The full
437 suite of geophysical and geochemical observations is therefore consistent with an air
438 crash at the site identified within Nuthampstead Airfield.

439

440 **Efficacy of in situ XRF Surveying**

441 To use in situ XRF surveying as an archaeological exploration tool, some mechanism
442 must exist to transport 'exotic' (i.e., absent in the background) geochemical elements
443 from their buried source to the ground surface. No metallic fragments were observed
444 in the laboratory-powdered soil samples, suggesting that elements at the site are
445 transported in groundwater rather than being present in shards of metallic debris.

446

447 At Nuthampstead, ploughing appears to be an effective transport mechanism, and the
448 time since ploughing appears to be a key control on the clarity of the XRF anomalies.
449 The survey in November 2014 was conducted soon after a period of ploughing,
450 potentially supplying the ground surface with a 'fresh charge' of metal-rich
451 groundwater. Anomalies and their correlation coefficients were both reduced in the
452 August 2015 dataset (e.g., Figure 8) compared to November 2015. Ordinarily, it might
453 be expected that the drier ground conditions in summer would yield higher
454 geochemical concentrations (e.g., Schneider et al., 2016) but, at the time of this

455 acquisition, the ground had been undisturbed for several months. Metal ions could
456 therefore have been flushed from the site by (e.g.) rainfall, or transported back into the
457 subsurface. However, some ions must also remain adsorbed onto soil grains,
458 otherwise, XRF analysis of dry soil (including in the laboratory analyses) would have
459 detected no geochemical anomaly at all. Given that the sample size of the XRF
460 instrument is $\sim 1 \text{ cm}^2$, it is unlikely that analyses are conducted at precisely the same
461 location between different time periods; however, the changes in the XRF responses
462 are not a shift in the position of the geochemical anomalies, but in the scatter and the
463 correlation of geochemical concentrations. Separate to instrumental effects (e.g.,
464 calibration and sensitivity), the measured concentrations are therefore a function of:

- 465 i) the abundance of a given element in the source material,
- 466 ii) the groundwater solubility and adsorption potential of that given element,
467 and
- 468 iii) the efficiency of any source-to-surface transport mechanism.

469

470 Calibration issues are often unavoidable in archaeological XRF surveying (e.g., Scott
471 et al., 2016). A non-specialist should therefore consider XRF spectrometry as a
472 qualitative tool for 'anomaly spotting', rather than interpreting the absolute values of
473 the recorded concentrations. Bespoke calibrations are recommended if absolute
474 concentrations are required (e.g.) for comparative archaeometric purposes (Scott et
475 al., 2016) or where forensic analysis may lead to litigation (Sbarato and Sánchez,
476 2001; Ruffell and Wiltshire, 2004; Bergslien, 2013). Validation with laboratory analysis
477 is also advocated since XRF scattering effects are minimised in powdered samples;
478 furthermore, such samples represent a homogenised volume of material, therefore the
479 measurement is less susceptible to 'skin' anomalies.

480

481 With respect to the efficiency of acquisition, in situ XRF spectrometry compares
482 favourably with established geophysical methods. Not only is the cost of equipment
483 similar to many geophysical systems, the rate of data return (40 samples/hour, here
484 distributed across a 100 m transect) is comparable to (e.g.) surveying with electrical
485 resistivity tomography. While XRF spectrometry would likely be impractical as an
486 initial reconnaissance tool, it can contribute valuable insight to the understanding of a
487 target once that target has been identified.

488

489 **Conclusions**

490 In situ XRF spectrometry provided a valuable geochemical complement to a suite of
491 geophysical field acquisitions. Localised increases in the concentration of diagnostic
492 metallic elements improved the detectability of the crash site of a Second World War
493 aircraft, adding confidence to the interpretation of a suite of geophysical data.
494 Specifically, increases in the local abundance of Cu and Zn were identifiable as
495 originating with brass ammunition cartridges among the aircraft wreckage. The
496 applicability of in situ XRF at a given site requires not only that anomalous elements
497 are present in detectable abundance, but that some source-to-surface transport
498 mechanism (e.g., ploughing) is active. While in situ XRF responses should be
499 validated under laboratory conditions, the portable XRF spectrometer offers a useful
500 complement to a programme of field geophysical survey.

501

502

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510

511 **Supplementary Figure Caption**

512

513 Supplementary Figure 1. Extended version of Figure 8 in the main text. Scatterplots
514 of paired elemental data from in situ XRF surveys in (upper-right quadrant) November
515 2014 and (lower-left quadrant) August 2015.

516

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