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Supporting Information for:

H₂O₂ enables convenient removal of RAFT end-groups from block copolymer nano-objects prepared via polymerization-induced self-assembly in water

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Figure S1. (a) Integrated ¹H NMR spectra and (b) DMF GPC chromatograms for G_{52} -TTC, G_{52} -DB, G_{61} -DB and G_{104} -DB macro-CTAs



Figure S2. (a) Integrated ¹H NMR spectra and (b) DMF GPC chromatograms for G_{104} -H_X (X = 300, 600, 900) diblock copolymer spheres



Figure S3. (a) Integrated ¹H NMR spectra and (b) DMF GPC chromatograms for G_{52} -H₁₃₅-TTC and G_{52} -H₁₃₅-DB worms, G_{61} -B₁₀₀ spheres and G_{52} -H₄₀₀ vesicles



Figure S4. DMF GPC traces recorded for G_{52} - H_{135} -DB before (black) and after (red) H_2O_2 treatment. Conditions: H_2O_2 /dithiobenzoate molar ratio = 20 for 3 h at 70 °C.



Figure S5. Gel storage modulus (G', closed symbols) and loss modulus (G'', open symbols) vs. temperature plots obtained for a G_{52} - H_{135} -DB worm gel before (black) and after (red) treatment with H_2O_2 . Conditions: $[H_2O_2]/[DB] = 20$ for 3 h at 70 °C. Note that a weaker worm gel is obtained after H_2O_2 treatment (G' = 71 Pa, vs. G' = 96 Pa originally) and the critical gelation temperature (CGT) is raised from 19 °C to 21 °C.



Figure S6. DMF GPC chromatograms (UV detector) of G_{104} -H_X-DB spheres before end-group removal and after H₂O₂ treatment for 24 h (see arrows) using a H₂O₂/dithiobenzoate molar ratio of 5.0 at 70 °C. In each case at least 98 % of the original end-groups are removed.



Figure S7. GPC chromatograms recorded for the G₅₂-DB macro-CTA before (blue traces) and after (red traces) end-group removal via H₂O₂ treatment using a H₂O₂/dithiobenzoate molar ratio of 5.0 at 70 °C: (a) minimal change in the molecular weight distribution as judged using a refractive index detector and (b) 97 % disappearance in the 309 nm signal associated with the RAFT end-group using the UV detector.



Figure S8. DMF GPC chromatograms (refractive index detector) of G_{104} -H_X-DB spheres before end-group removal and after H₂O₂ treatment for 7 h using a H₂O₂/dithiobenzoate molar ratio of 5.0 at 70 °C. Note that there is minimal change in the molecular weight distributions under these optimized end-group removal conditions.