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1 **NITROGEN CYCLING IN THE LATE PALEOPROTEROZOIC REDOX STRATIFIED**
2 **OCEAN**

3

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17 **ABSTRACT**

18 Nitrogen cycling has been evaluated across a depth transect in the late Paleoproterozoic
19 Animikie Basin, spanning the end of Earth's final period of global iron precipitation and a major
20 transition to euxinic conditions in areas of high productivity. Sediments from near shore where
21 productivity was highest have $\delta^{15}\text{N}$ compositions up to ~3‰ higher than at more distal sites.
22 This suggests that as NH_4^+ mixed vertically upwards into the photic zone, it was either
23 assimilated by organisms or oxidized. Subsequent enhanced production of N_2 by denitrification
24 or anammox led to the observed increase in $\delta^{15}\text{N}$ close to shore. Any deficit in biologically
25 available N was overcome by N_2 -fixing organisms, but the input of N with low $\delta^{15}\text{N}$ from this
26 process did not overwhelm the increase in $\delta^{15}\text{N}$ from denitrification. Since there is no evidence
27 for conditions of severe N-stress arising from trace metal limitation (particularly Mo) of N
28 fixation during the transition to euxinic conditions, losses of N were either very small
29 (potentially because low O_2 levels limited NH_4^+ oxidation), or alternative pathways which
30 retained N were important. The fact that Mo appears to have remained bioavailable for N
31 fixation, either suggests that the extent or severity of sulfidic water column conditions was not
32 sufficient to quantitatively sequester Mo on a global scale, or that rivers directly delivered Mo to
33 surface waters on the inner shelf. The effects of N_2 fixation on $\delta^{15}\text{N}$ increased to more distal
34 parts of the shelf, consistent with models invoked for modern upwelling zones over broad
35 continental margins.

36 **INTRODUCTION AND SETTING**

37 Around 1840 Myr ago, a widespread transition to euxinia occurred along productive
38 continental margins (Canfield, 1998; Poulton et al., 2004; 2010; Kendall et al., 2011). These
39 anoxic sulfidic conditions extended outwards from near shore to the mid-shelf, and were overlain

40 by oxic surface waters, with deeper waters that were anoxic and Fe-rich (ferruginous) (Poulton et
41 al., 2010; Kendall et al., 2011; Planavsky et al., 2011). Widespread sulfidic conditions
42 throughout the mid-Proterozoic have been implicated in the protracted oxygenation of the
43 atmosphere and slow rates of eukaryotic evolution (Anbar and Knoll, 2002). However, the
44 continuation of ferruginous deep water conditions through the Mesoproterozoic, with sulfidic
45 conditions limited to areas of high organic C production (Scott et al., 2008; Kendall et al., 2009,
46 2001; Poulton et al., 2010; Poulton and Canfield, 2011; Planavsky et al., 2011), highlights that
47 the controls on productivity, and oxygenic productivity in particular, remain poorly understood.

48 One major question concerns feedbacks associated with the N cycle, and is the focus of
49 this study. The classical view of the N cycle is that as anoxic, NH_4^+ bearing deep water vertically
50 mixed into oxic surface water, nitrification-denitrification and anammox reactions would have
51 severely depleted the ocean inventory of dissolved fixed N (Fennel et al., 2005). Furthermore, it
52 has also been suggested that expansive euxinia may have placed additional stress on the N cycle
53 by the precipitation of chalcophiles such as Mo and Fe, which are essential to N_2 fixation (Anbar
54 and Knoll, 2002). However, the ocean inventory of these metals had to become extremely
55 depleted in order to restrict N-fixation (e.g., Zerkle et al., 2006). Recent evidence for the
56 persistence of ferruginous deep ocean conditions through the mid-Proterozoic (Poulton et al.,
57 2010; Planavsky et al., 2011), coupled with the likelihood that Fe was present in solution even
58 under euxinic conditions (potentially as $\text{FeS}_{(\text{aq})}$ clusters; Luther et al., 1996), suggests that Fe was
59 unlikely to be biolimiting at this time. In terms of Mo, estimates based on $\delta^{98}\text{Mo}$ suggest that the
60 extent of sulfidic water column conditions may have been an order of magnitude higher than at
61 present (Kendall et al., 2011), which may not have been sufficiently depleted to restrict N_2
62 fixation on a global scale. Furthermore, while sediment Mo and reactive Fe measurements

63 address overlying water, they do not account for the whole of water column, including the upper
64 ocean where the N cycle is most active. Consequently a direct evaluation of the operation of the
65 N cycle is needed to determine its state during this major transition in ocean chemistry.

66 We report new N and C isotope data for sediments from six drill cores that provide a 350
67 km long oblique transect to the paleo-coastline of the Animikie Basin on the margin of Superior
68 Province (Figure 1; Poulton et al., 2010). The Gunflint and Biwabik iron formations, which
69 accumulated in a back-arc basin setting 1878±1.3 Myr ago (Fralick et al., 2002), are overlain by
70 ejecta material from the Sudbury impact event, deposited sub-aerially during regression of the
71 ocean at ~1850±1 Myr ago (Krogh et al., 1984). Later flooding from the south led to deposition
72 of fine grained siliciclastic material of the Rove and Virginia Formations ~1836±5 Myr ago
73 (Addison et al., 2005). The lower 100 m of these formations is dominated by black shales, with
74 turbidites, derived from the north, gaining importance above this zone. Estimated water depths
75 are between ~100 to 400 m, with good connectivity to the world ocean (Rasmussen et al., 2012).
76 Iron speciation measurements suggest that a transition to persistently sulfidic mid-depth water in
77 near shore environments occurred in the Rove and Virginia Formations, while deeper waters
78 remained ferruginous (Poulton et al., 2010), providing an ideal setting in which to study the N
79 cycle during this important transition in ocean chemistry.

80

81 **RESULTS AND DISCUSSION**

82 Analytical methods, data and discussion about the preservation of primary $\delta^{15}\text{N}$ can be found in
83 the on-line supplement.

84 **Spatial variations in organic carbon and $\delta^{13}\text{C}_{\text{org}}$ across the Animikie Basin shelf**

85 Concentrations of C_{org} are similar across the shelf during iron formation deposition, only
86 increasing in a few thin tuffaceous layers. In the siliciclastic Rove and Virginia Formations there
87 is a tripling in C_{org} concentration towards the shore. If sediment accumulation rates were higher
88 near shore than at the distal sites, as is typical, and indicated by the up-core increase in grain size
89 in cores MC-1 and PR9-1, the C_{org} distribution is consistent with an increase in C_{org}
90 accumulation due to higher productivity associated with mixing through coastal upwelling. The
91 development of sulfidic water overlying deeper ferruginous water is analogous to modern
92 oxygen minimum zones associated with upwelling and high OM fluxes, but because of limited
93 O_2 and NO_3^- , SO_4^{2-} was respired at a rate higher than that at which deeper water Fe(II) was
94 supplied (Poulton et al., 2010; Planavsky et al., 2011). Organic $\delta^{13}C$ varies between -27 and -
95 34‰ in both iron formation and siliciclastic sediments (Figs 1B and 2), typical of values for
96 Proterozoic sediments (Karhu and Holland, 1996; Kaufman et al., 1997; Bekker and Kaufman,
97 2007). While $\delta^{13}C_{org}$ is evenly distributed across the Gunflint-Biwabik shelf, there is a trend of
98 decreasing $\delta^{13}C_{org}$ shoreward in association with increasing C_{org} during deposition of the Rove-
99 Virginia Formations (Fig. 1). Lower $\delta^{13}C_{org}$ and higher C_{org} suggests that CH_4 fluxes resulting
100 from high productivity near shore were oxidized aerobically or anaerobically with SO_4^{2-} .

101

102 **Spatial variations in $\delta^{15}N$ and a nitrogen isotope model for the Animikie Basin**

103 Unlike C, N is a limiting nutrient and when productivity consumes all available NO_3^- ,
104 NH_4^+ or NO_2^- (cumulatively dissolved inorganic N; DIN), $\delta^{15}N_{OM}$ will be similar to the isotope
105 composition of DIN available at a particular location. $\delta^{15}N_{OM}$ can be lower than $\delta^{15}N_{DIN}$ if there
106 are other limitations on production and DIN is not fully consumed, or if there are diazotrophs
107 present. Except for N_2 -fixation and ammonification of OM, each transfer step of N between

108 different phases is accompanied by strong isotope discrimination which favors the light isotope
109 in all but one step, NO_2^- oxidation (Casciotti, 2009, and references therein). The N cycle is
110 sensitive to the availability of O_2 and, when the N and O cycles interact, the isotope composition
111 of DIN as well as its N inventory can be influenced. Organic matter sinking from the surface
112 ocean decays and forms NH_4^+ . In an anoxic ocean, NH_4^+ is returned to the surface ocean where it
113 is consumed, while in a fully oxic ocean, nitrifying bacteria immediately oxidize this NH_4^+ to
114 NO_3^- , and NO_3^- is returned to the surface ocean. Where redox gradients exist, either due to a
115 weakly oxygenated atmosphere or because O_2 is drawn down by respiration, there is potential for
116 substantial losses of N by autotrophic anammox or classical denitrification, which will increase
117 $\delta^{15}\text{N}_{\text{DIN}}$. However, the geological $\delta^{15}\text{N}$ record, $\delta^{15}\text{N}_{\text{SED}}$, is derived from export production, and
118 reflects the biological response to N availability, and thus will be affected by N_2 -fixation
119 responding to DIN deficits (Deutsch et al., 2004).

120 During the deposition of the iron formation, $\delta^{15}\text{N}$ deposited on the outer shelf was about
121 1‰ lower than on the inner shelf, excluding tuffaceous layers which have variable $\delta^{15}\text{N}$ values.
122 After euxinic conditions developed on the inner shelf, sediment $\delta^{15}\text{N}$ at the inner shelf sites
123 remained at ~5.3‰ with outer shelf sites ~1.4‰ lower. A one-way ANOVA was used to confirm
124 that $\delta^{15}\text{N}_{\text{SED}}$ differed between the inner and outer shelf, $F(4, 174) = 39.6$, $p < .001$. While
125 changes in the distribution of C_{org} concentrations and $\delta^{13}\text{C}_{\text{org}}$ across the shelf reflect enhanced
126 coastal mixing and productivity, the distribution of $\delta^{15}\text{N}_{\text{SED}}$ across the shelf was largely
127 unchanged, suggesting little change in the operation of the N cycle during a time when profound
128 changes in ocean chemistry took place.

129 Ammonium would have been the stable form of N in ferruginous and euxinic water, with
130 NO_3^- stable in oxic surface water. The base of the mixed layer would have coincided with the

131 redox gradient between oxic surface water and either ferruginous deep water or euxinic mid-
132 depth water. NH_4^+ mixing upwards was either biologically assimilated or oxidized to NO_2^- or
133 NO_3^- , the process of nitrification. The oxidized forms of DIN could have been assimilated or
134 reduced to N_2 in order to respire C_{org} or oxidize reduced metal or S species (Fig. 2 node A).
135 Which of these processes dominated the NH_4^+ flux was probably determined by upwelling rates,
136 surface water O_2 concentration and the organisms that were present, but the potential for DIN
137 losses were large.

138 If N and associated metal co-factors were the main controls on productivity, for
139 productivity to be highest on the inner shelf, either the activity of anoxygenic phototrophs kept
140 O_2 levels low (which limited nitrification; Johnston et al, 2009), organisms were actively fixing
141 N, or there were mechanisms which circumvented denitrification and anammox. Nitrification,
142 denitrification and DIN assimilation select for ^{14}N . In a redox layered ocean, oxidation and
143 assimilation of upwelling NH_4^+ both occur at the chemocline and may consume all NH_4^+
144 entering the euphotic zone. If nitrification and assimilation selected equally for ^{14}N , biomass
145 grown from NH_4^+ and $\text{NO}_2^-/\text{NO}_3^-$ would have had similar $\delta^{15}\text{N}$, and equal to the deep source of
146 NH_4^+ . If all the $\text{NO}_2^-/\text{NO}_3^-$ was then denitrified and lost to the atmosphere, there would be no net
147 effect on $\delta^{15}\text{N}$ by denitrification and the deep NH_4^+ reservoir, biomass and sediment $\delta^{15}\text{N}$ would
148 have been the same as N_2 fixed by diazotrophs, $\sim 0\%$. The very negative $\delta^{15}\text{N}_{\text{SED}}$ of the
149 Mesozoic ocean anoxic events has been explained by large vertical fluxes of NH_4^+ , also
150 predicted to be the dominant N species at depth, which fueled productivity and due to its high
151 concentration, exhibited greater isotope discrimination than nitrification (Higgins et al., 2012).
152 The positive values of $\delta^{15}\text{N}_{\text{SED}}$ on the Animikie margin suggest that mixing did not lead to high
153 NH_4^+ concentrations where it was assimilated which reduced the ability of assimilation to

154 discriminate against ^{15}N (Hoch et al., 1994), or that subsequent reduction of $\text{NO}_2^-/\text{NO}_3^-$ was not
155 quantitative. If ^{14}N was partitioned into $\text{NO}_2^-/\text{NO}_3^-$ because of low deep NH_4^+ concentrations,
156 quantitative denitrification would have increased remaining DIN, requiring high rates of N_2 -
157 fixation to sustain productivity and moderate $\delta^{15}\text{N}_{\text{SED}}$. For N-fixation rates to be high, this
158 implies that Mo could not have been limiting, either because euxinic conditions were not severe
159 enough or that Mo was supplied directly into the coastal system by nearby river systems (Maric
160 and Fralick, 2005). However, the stability of NH_4^+ in ferruginous and sulfidic water suggests that
161 deep NH_4^+ concentrations were unlikely to be particularly low, so in order for $\delta^{15}\text{N}_{\text{SED}}$ to be
162 positive, denitrification could not have been quantitative. The fate of NO_3^- , either direct
163 assimilation or conservation through dissimilatory nitrite reduction to ammonia (DNRA; or
164 respiratory ammonification) or denitrification (Fig 2, node B) determined how ‘leaky’ the system
165 was toward N, and how much N fixation was required to make up the deficit. While DNRA was
166 once thought to be restricted to euxinic systems (Sørensen, 1978; Simon, 2002; Senga et al.,
167 2006; Tugtas and Pavlostathis, 2007), its occurrence in the Benguela and Peru upwelling systems
168 (Kartal et al., 2007; Lam et al., 2009) suggests that the presence of sulfide is not a pre-requisite,
169 and it occurs over a range of O_2 concentrations (Kraft et al., 2011). The strong redox gradient
170 suggests that N losses and $\delta^{15}\text{N}$ could have been high above euxinic areas of the margin. The
171 inference from C_{org} is that euxinic conditions did not limit productivity, and hence either DNRA
172 was effective at retaining ^{14}N -rich DIN in the coastal zone, or the high productivity was fueled
173 by N-fixation.

174 The $\delta^{15}\text{N}$ distribution indicates that as surface water moved away from the coastline, it
175 likely carried a N deficit (Fig. 2 inset). Assuming the lower $\delta^{15}\text{N}_{\text{SED}}$ at distal sites reflects
176 increased diazotrophy, then as discussed above, the Fe(II)-bearing deep ocean was able to supply

177 labile Fe to the surface ocean, while Mo was also not limiting, either due to non-quantitative
178 removal under euxinic conditions and lateral mixing from the inner shelf, or by vertical mixing
179 of ferruginous water in from which Mo had not been sequestered.

180 **Implications for the Proterozoic N cycle.**

181 It has been predicted that sulfidic conditions in the mid-Proterozoic redox-stratified water
182 column would have led to severe N-limitation, due to N losses and sequestration of the
183 chalcophilic metals (particularly Mo and Fe) used in cofactors of various enzymes employed in
184 the N cycle, especially those related to N-fixation (Anbar and Knoll, 2002; Glass et al., 2009).
185 The lack of change in $\delta^{15}\text{N}$ as free HS^- appeared in the Animikie Basin, combined with the
186 highest productivity coinciding with euxinic conditions, do not support this model of extreme N
187 stress. Either euxinia was not severe enough to lower Mo concentrations to levels which affected
188 N-fixation, or continental run-off delivered Mo directly to coastal systems. Furthermore,
189 pathways retaining N such as DNRA, could have mitigated DIN loss and its associated isotope
190 effect because the ^{14}N -rich NO_2^- was converted back to NH_4^+ and not lost to the atmosphere as
191 N_2 . Consideration of the N cycle in the geological past should include all potential pathways,
192 especially those of great antiquity that developed under regimes of extreme N limitation very
193 early in the development of N metabolism (Mancinelli and McKay, 1988; Klotz and Stein,
194 2008).

195
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201

202

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293

294 **FIGURE CAPTIONS**

295

296 Figure 1. Six cores from the margin of the Superior Province (A) constitute a transect of samples
297 across the Animikie Basin margin (B) and consist of iron formation and siliciclastic material
298 interspersed with occasional other lithologies(C). Ejecta from the Sudbury meteorite impact
299 provides a correlative layer across the basin from shallow water at MC-1 to distal shelf at MGS8.
300 Ferruginous versus euxinic conditions are from Poulton et al., (2010). $\delta^{13}\text{C}_{\text{org}}$ (D) and $\delta^{15}\text{N}$ (E)
301 exhibit persistent gradients across the margin.

302

303 Figure 2. The N cycle in the Paleoproterozoic is strongly influenced by the redox gradient which
304 separated oxygenated surface ocean from deep ferruginous water or euxinic mid- depth water
305 which developed in productive near shore environments. Deep NH_4^+ upwells at the coast and
306 crosses the redox boundary where it can be nitrified to $\text{NO}_2^-/\text{NO}_3^-$ or assimilated (node A), the
307 pathway taken may be influenced by O_2 availability and upwelling rate. $\text{NO}_2^-/\text{NO}_3^-$ can be
308 respired to N_2 or reduced directly to NH_4^+ by DNRA (node B). The proportion of $\text{NO}_2^-/\text{NO}_3^-$
309 which is converted to N_2 determines the loss of N from the coastal system and the enrichment in
310 remaining dissolved fixed N in ^{15}N and DIN deficit. The inset diagram depicts changes in N/P
311 and $\delta^{15}\text{N}$ as water flows through this system. The deficit in DIN may promote N-fixation which
312 introduces N with low $\delta^{15}\text{N}$, but it cannot overcome the strong preference for ^{14}N shown by
313 denitrification unless all upwelling NH_4^+ is lost. R is the Redfield ratio, although it may deviate
314 from R according to the ocean average biological requirements.