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29 **Abstract**

30

31 The effects of reaction kinetics of bridgmanite and ferropericlae transforming to ringwoodite  
32 on elastic properties in upwelling mantle are investigated using data of kinetic experiments and  
33 internally self-consistent thermodynamic modelling of density and seismic velocities. The  
34 kinetic experiments show inhibited grain growth of ringwoodite. At the initiation of ringwoodite  
35 growth, bridgmanite completely transforms to a metastable pyrope-bearing garnet.  
36 Ringwoodite then gradually grows from the metastable assemblage of ferropericlae and garnet.  
37 The changes in mineralogy result in a low-velocity zone directly above the 660 km seismic  
38 discontinuity due to the lower seismic velocities and densities of ferropericlae and garnet  
39 compared to ringwoodite and bridgmanite. The modelling of the effects of reaction kinetics and  
40 its effect on seismic structure at ~660 km depth shows more sensitivity to grain size than to  
41 temperature and upwelling rate. Modelling 1-D synthetic seismograms of PP (SS) underside  
42 reflections off the kinetically inhibited backward reaction to ringwoodite shows advanced travel  
43 times of underside reflections off ~660 km depth of 0.2 - 0.8 s (1.2 - 1.6 s) for upwelling rates  
44 of 50 cm/yr and initial grain sizes between 5 and 20 cm due to the low-velocity zone above the  
45 660 km discontinuity. The finite width of the low-velocity layer results in frequency-dependent  
46 behaviour of PP and SS underside reflection amplitudes, with higher amplitudes towards shorter  
47 periods. The effect on the travel times of P-to-s conversions used for receiver function is small  
48 (< 0.7 s) but the inhibited backward reaction leads to strong amplitude changes and significant  
49 waveform variations. The effects of reaction kinetics in mantle upwellings might serve as an  
50 additional means to map regions of large-scale upwellings and to constrain grain size in the  
51 lower mantle.

52

53 **Key words:** Phase transitions, Composition and structure of the mantle, High-pressure  
54 behaviour, Body waves

55

## 56 **1 Introduction**

57

58 The seismic structure of the Earth's mantle transition zone between upper and lower mantle is  
59 dominated by two seismic discontinuities at which the seismic velocities and density increase  
60 over short depth intervals. These discontinuities are found in most spherically averaged 1-D  
61 Earth reference models, e.g., ak135 (Kennett et al., 1995), and are considered as the seismic  
62 expressions of solid-solid phase transitions and mineral reactions (both hereafter called phase  
63 transformations) of the olivine system (e.g., Ringwood, 1969; Helffrich, 2000; Shearer, 2000;  
64 Weidner & Wang, 2000; Deuss, 2015). Thermodynamic calculations of mantle mineral phase  
65 relations are frequently used to interpret observations of seismic velocities and seismic  
66 discontinuity structure and help to identify and quantify their thermal and compositional origins  
67 (e.g., Cobden et al., 2008; Ritsema et al., 2009; Saki et al., 2019; Vilella et al., 2021; Waszek et  
68 al., 2021).

69

70 For thermodynamic modelling of phase equilibria and their seismic properties, the Earth's  
71 mantle is usually assumed as vigorously convecting, homogeneous and well mixed. However,  
72 geodynamic studies have demonstrated the long-term existence of thermal and compositional  
73 heterogeneities throughout the mantle (e.g., Allègre & Turcotte, 1986; Kellogg et al., 2002;  
74 Tackley et al., 2005; Ballmer et al., 2017). Small diffusivities of the solid mantle (e.g., Hofmann  
75 & Hart, 1978; Farber et al., 1994) lead to the question: to what degree have the mineral  
76 assemblages reached thermodynamic equilibrium and mineral transformations been  
77 completed? The mechanisms of mineral transformations, i.e., their reaction kinetics, might  
78 therefore affect the dynamics and seismic structure of the Earth's mantle (e.g., Kubo et al., 2002;  
79 Kubo et al., 2008). Therefore, taking the reaction kinetics into account might be essential for  
80 our interpretation of the seismic signals in terms of mantle composition and temperature.

81

82 The influence of reaction kinetics on mineral transformations was first investigated in detail  
83 when seismologists looked for mechanisms of deep earthquakes in subducting slabs (e.g., Sung  
84 & Burns, 1976; see Kirby, 1996, for a review). Motivated by observations of apparently  
85 stagnant slabs at ~660 km depth in seismic tomography models (e.g., van der Hilst, 1991; Fukao  
86 et al., 2001; Fukao et al., 2009), Kubo et al. (2002) and Kubo et al. (2008) investigated the  
87 reaction kinetics of phase transformations of ringwoodite to ferropericlase and bridgmanite and  
88 of garnet to bridgmanite at ~660 km depth and temperature conditions typical for subducting  
89 slabs. The kinetic experiments by Kubo et al. (2002) and Kubo et al. (2008) show that low  
90 temperatures within the slab decrease the mobility of atoms in the crystal lattice and hence

91 could reduce reaction rates of mineral transformations. The kinetic inhibition of the phase  
92 transformations of ringwoodite to ferropericlase and bridgmanite and of garnet to bridgmanite  
93 result in neutral to positive buoyancy which may lead to stagnant slabs. Recently it has also  
94 been shown that kinetics of the bridgmanite to post-perovskite phase transition can affect  
95 visibility of the D" reflector (Langrand et al., 2019).

96  
97 Reaction mechanisms of mineral transformations can also change for higher temperatures.  
98 Studies by Gasparik (1996a), Gasparik (1996b), Weidner & Wang (1998), Irifune et al. (1998)  
99 and Hirose (2002) show that the reaction mechanism of ringwoodite to ferropericlase and  
100 bridgmanite changes for mantle temperatures larger than 2100 K. This temperature is  
101 considered as a reasonable estimate for the centre of hot mantle upwellings (e.g., Sleep, 1990;  
102 Schilling, 1991; Weidner & Wang, 1998). For aluminium contents of approximately 5 wt% and  
103 temperatures of at least 2100 K, ringwoodite transforms to bridgmanite and ferropericlase via  
104 the dissociation of ringwoodite to ferropericlase and majorite garnet, and finally to  
105 ferropericlase and bridgmanite (Gasparik, 1996a; Gasparik, 1996b; Weidner & Wang, 1998;  
106 Irifune et al., 1998; Hirose, 2002).

107  
108 Kinetic studies by Shimojuku et al. (2014) and Dobson & Mariani (2014) have investigated the  
109 reaction kinetics for the reverse reaction of bridgmanite and ferropericlase to ringwoodite in  
110 upwelling mantle. The kinetic experiments show that the growth of ringwoodite is kinetically  
111 inhibited, and that bridgmanite transforms to pyrope-bearing garnet. The reverse reaction from  
112 bridgmanite and ferropericlase to ringwoodite and garnet (in mantle upwellings) requires  
113 diffusion on a much larger length-scale than the forward reaction because macroscopic grains  
114 of bridgmanite and ferropericlase must recombine to form ringwoodite (Dobson & Mariani,  
115 2014). In the case of aluminous garnets, the interface between ringwoodite and garnet develops  
116 a fingering instability resulting in a complex intergrowth (Figure 1). Garnet incorporates the  
117 aluminium oxide of the bridgmanite into its crystal lattice and has the same bulk chemistry as  
118 the bridgmanite before the transformation, and hence this isochemical transformation is fast  
119 compared to the diffusion-controlled reaction between bridgmanite and ferropericlase. Over  
120 time, this chemically metastable garnet reacts with ferropericlase, and the stable phase,  
121 ringwoodite, starts to nucleate and grow at the interface between ferropericlase and garnet.

122  
123 The growth of ringwoodite on the ferropericlase grains is most likely controlled by the slow  
124 diffusion of SiO<sub>2</sub> from garnet across the reaction layer into the grains of ferropericlase. The  
125 ringwoodite growth rate is sensitive to the initial grain size of lower mantle minerals (Dobson

126 & Mariani, 2014). The initial ferropericlasite grain size is the typical length scale of diffusion.  
127 The amount of ringwoodite formed depends on bulk composition and ambient temperature. The  
128 bulk composition affects the individual molar amounts of the mineral solid solutions. The  
129 ambient mantle temperature influences the equilibrium phase relations and mobility of atoms,  
130 determining the diffusion rate of atoms across interfaces and grain boundaries (e.g., Putnis,  
131 1992). The initial grain size controls the relative proportions of reactants and products for a  
132 given amount of diffusion across the reacting interface.

133

134 Our study investigates the effects of reaction kinetics for the recombination of bridgmanite and  
135 ferropericlasite to ringwoodite in mantle upwellings on density and seismic wave velocities to  
136 assess the possibility to observe the effects of the kinematically inhibited back reaction  
137 seismically. The kinetically inhibited growth of ringwoodite is determined for a range of mantle  
138 temperatures at 660 km depth, varying grain sizes and mantle upwelling rates. The changed  
139 volume fractions of the minerals as a function of pressure are used to calculate profiles of  
140 density, P- and S-wave velocities for kinetically inhibited and equilibrated phase assemblages  
141 using internally self-consistent thermodynamic modelling. These models are used to compute  
142 1-D synthetic seismograms for PP and SS underside reflections off the resulting discontinuity  
143 and for P-to-s conversions at the discontinuity (known as P receiver functions), two probes often  
144 used to study upper mantle discontinuities, to assess the effects on travel times and waveforms.  
145 Our aim is not to carry out a seismic study at this point to search for such effects in observed  
146 seismic data. This work presents a previously unstudied effect of mantle phase transitions on  
147 seismic data that should be kept in mind in future seismic studies.

148

## 149 **2 Methods**

150

151 The kinetically inhibited growth of ringwoodite and its effects on phase assemblages and  
152 seismic properties are computed in five steps:

- 153 1. calculation of phase equilibria,
- 154 2. calculation of the thickness of the ringwoodite layer produced for a given temperature,  
155 grain size and mantle upwelling rate,
- 156 3. mass balancing consumption of metastable ferropericlasite and garnet,
- 157 4. calculation of density and seismic wave velocities.
- 158 5. computation of 1-D synthetic seismograms

159

### 160 **2.1 Calculation of phase equilibria**

161

162 Phase equilibria for the CFMAS (calcium oxide, iron(II) oxide, magnesium oxide, aluminium  
163 oxide and silica) system are calculated for a pyrolitic mantle composition (Table 1) with the  
164 `Perple_X` program package (Connolly, 2005) that utilises the free-energy minimization  
165 algorithm by Ita & Stixrude (1992) and Stixrude & Lithgow-Bertelloni (2005), and the elastic  
166 and thermodynamic properties and the solid solution model of Stixrude & Lithgow-Bertelloni  
167 (2011). From the calculated phase equilibria, profiles of P-wave velocity, S-wave velocity and  
168 density are extracted along adiabatic mantle geotherms (e.g., Cobden et al., 2008). The values  
169 of P-wave velocity, S-wave velocity, and density of the phase assemblage at a specific P-T-  
170 point are calculated as the Hill average of Voigt and Reuss bounds of the constituent minerals  
171 (Voigt, 1928; Reuss, 1929; Hill, 1965).

172

## 173 **2.2 Kinetically inhibited growth of ringwoodite and mineral reactions**

174

175 The seismic structures of the kinetically inhibited phase assemblages are calculated from the  
176 grain growth of ringwoodite for fixed temperature, grain size and upwelling rate (Figure 2).  
177 Before we describe the calculations of kinetically inhibited growth of ringwoodite and its  
178 impact on seismic properties, we compare the mineral reactions at thermodynamic equilibrium  
179 with those observed by Dobson & Mariani (2014).

180

181 The recombination reaction at thermodynamic equilibrium is described as

182



ferropericlaase + bridgmanite  $\leftrightarrow$  ringwoodite

183

184 Dobson & Mariani (2014) observe in the kinetic experiments that bridgmanite entirely  
185 transforms to garnet, where the aluminium-bearing bridgmanite transform to pyrope garnet:

186



aluminous bridgmanite  $\leftrightarrow$  majorite garnet + pyrope garnet

187

188 The consumption of bridgmanite to ringwoodite comprises of several reactions. At the onset of  
189 ringwoodite growth, the entire amount of bridgmanite transforms to garnet,

190



bridgmanite  $\rightarrow$  majorite garnet

191

192 creating an excess of garnet compared to conditions at thermodynamic equilibrium.

193 The difference between the decrease of ferropericlasite with kinetic inhibition effects and the

194 phase fraction of ferropericlasite is later used to infer the decrease of garnet which recombines

195 with ferropericlasite to give ringwoodite via the reaction

196



ferropericlasite + majorite garnet  $\leftrightarrow$  ringwoodite

197

198 The kinetically inhibited growth of ringwoodite and the dependent mineral transformations of

199 bridgmanite and garnet are assumed to leave calcium-bearing garnets and the exsolution of

200 calcium perovskite unaffected. Therefore, amounts of calcium-bearing garnets and calcium

201 perovskite are extracted from the thermodynamic calculations.

202

203 We now derive how we calculate the growth kinetics of ringwoodite and account for the mineral

204 reactions mentioned above. The growth kinetics of ringwoodite are controlled by two

205 parameters which are determined in experiments: its enthalpy  $\Delta H_a$  and its reaction rate constant

206  $k_0$  which is closely linked to the reaction entropy  $\Delta S_a$ .

207

208 For a mineral transformation from mineral assemblages A and B to a mineral assemblage C,

209 the reaction rate is proportional to the molar concentrations of mineral phases A and B,  $C_A$  and

210  $C_B$  (Putnis, 1992):

211

$$\text{Reaction rate} \propto K^* C_A C_B \quad (5)$$

212

213 In our case,  $C_A$  and  $C_B$  are the concentration of ferropericlasite and majorite garnet.

214 The equilibrium constant of the reaction  $K^*$  is a function of the change of reaction Gibbs free

215 energy,  $\Delta G_a$ ,

216

$$K^* = \exp(-\Delta G_a/(R T)) = \exp(\Delta S_a/R) \exp(-\Delta H_a/(R T)) \quad (6)$$

217

218 using the definition of the reaction Gibbs free energy  $\Delta G_a = \Delta H_a - T\Delta S_a$  with the enthalpy of  
 219 the reaction  $\Delta H_a$  and the entropy of the reaction  $\Delta S_a$ .  $R$  denotes the gas constant and  $T$  is  
 220 temperature. The reaction rate can be rewritten as

$$\text{Reaction rate} = c \exp(\Delta S_a/R) \exp(-\Delta H_a/(R T)) C_A C_B \quad (7)$$

222  
 223 where  $c$  is a constant. As the concentration of the reactants changes during the reaction, the  
 224 reaction rate will change accordingly. The reaction rate thus becomes a function of the phase  
 225 fraction  $C$

$$\text{Reaction rate} = k f(C) \quad (8)$$

227  
 228 where the reaction rate parameter  $k$  describes how the reaction rate depends on the concentration  
 229 of the reactants. Therefore, the rate constant is

$$k = k(T) = c \exp(\Delta S_a/R) \exp(-\Delta H_a/(R T)) = k_0 \exp(-\Delta H_a/(R T)) \quad (9)$$

231  
 232 with the reaction rate constant  $k_0 = c \exp(\Delta S_a/R)$  where  $c$  is an experimentally determined factor.

233  
 234 Dobson & Mariani (2014) show that the growth kinetics of ringwoodite are controlled by  
 235 diffusion kinetics. In this case, the rate of growth of the interface is linear with the square-root  
 236 of time and the thickness of reaction rims measured at each temperature can be converted into  
 237 apparent rate constants (Watson & Price, 2002). Following Tammann (1920) and Watson &  
 238 Price (2002), the temperature dependent rate constant  $k(T)$  is given by

$$k(T) = \frac{1}{2} \frac{d}{dt} x^2 \quad (10)$$

240  
 241 where  $x$  is the thickness of the growing interface and  $t$  is time. Integration of (10) gives the  
 242 interface thickness evolution as a function of time and temperature:

$$x(t, T) = ((2 k(T) t)^{1/2}). \quad (11)$$

244  
 245 To model the kinetic inhibition of ringwoodite growth, the interface grows until reaching the  
 246 assumed initial grain size  $d$ . In this study, radially isotropic growth of the ringwoodite interface

247 is assumed. We therefore normalise the ringwoodite interface thickness evolution (11) with half  
248 of the desired grain size ( $d$ ) which results in a relative grain size: :

249

$$x_{\text{rel}}(t, T) = ((2 k(T) t)^{1/2}) / (0.5 d). \quad (12)$$

250

251 Since temperatures are approximately constant along the chosen adiabats, we assume a constant  
252 temperature  $T_0$  during the reaction:

253

$$x_{\text{rel}}(t) = ((2 k(T_0) t)^{1/2}) / (0.5 d). \quad (13)$$

254

255 Due to its inverse dependency on grain size  $d$ , eq. (13) indicates that growth of ringwoodite  
256 takes more time in case of large grain size.

257

258 Assuming an upwelling rate  $v$  and utilizing the depth-to-pressure scale used by Cobden et al.  
259 (2008), the time of grain growth is converted to a grain growth curve as a function of pressure,  
260  $x_{\text{rel}}(P)$ . In our calculations, upwelling rates in the mantle are estimated to range between 1 and  
261 100 cm/yr considering current heat flux estimates in upwelling mantle (e.g., Sleep, 1990;  
262 Courtillot et al., 2003) and geodynamic modelling experiments (Farnetani & Hofmann, 2010).  
263 The average grain size in the lower mantle is estimated to range between 1  $\mu\text{m}$  (Yamazaki et  
264 al., 1996) and 10 cm (Solomatov, 1996; Korenaga, 2005) based on varying estimates of grain  
265 growth kinetics of mineral transformations in downwelling, or primordial, mantle. In the  
266 interior of a subducting slab, Yamazaki et al. (1996) and Yamazaki et al. (2005) find very slow  
267 grain growth rates for ferropericlase and bridgmanite, with grain sizes ranging between 10  $\mu\text{m}$   
268 to 100  $\mu\text{m}$  for temperatures typical for subduction zones and for 10 million years after grain  
269 nucleation, but Fei et al. (2021) find a much smaller grain size exponent compared to Yamazaki  
270 et al. (1996). Their estimate for grain size in subducted slabs is on the order of 30-45  $\mu\text{m}$ .  
271 Solomatov et al. (2002) question estimates of grain size for the bulk mantle as they cannot be  
272 reconciled with viscosity models or laboratory data for other materials. Solomatov et al. (2002)  
273 argue that the experiments have not been run sufficiently long to evaluate grain growth kinetics  
274 beyond transient effects. Solomatov (1996) and Korenaga (2005) have additionally suggested  
275 that deep-mantle plumes might have a significantly larger grain size than the ambient lower  
276 mantle due to (i) the temperature of the source region of the mantle upwelling, (ii) the kinetics  
277 of grain growth from an early magma ocean, and (iii) seismic tomographic imaging of wide  
278 slow structures which have been interpreted as wide plume roots (French & Romanowicz,  
279 2015).

280

### 281 **2.3 Mass balancing consumption of metastable ferropericlase and garnet**

282

283 We follow the workflow in Figure 2 to calculate volume fractions of the mineral phases affected  
284 by the kinetically inhibited growth of ringwoodite for given grain size  $d$ , upwelling rate  $v$  and  
285 constant temperature  $T_0$ . Ringwoodite starts to grow at pressure  $P_{\text{start}}$  where the molar amount  
286 of ringwoodite is non-zero in the equilibrium phase assemblage. The kinetically inhibited  
287 growth of ringwoodite is calculated as the convolution of the grain growth curve and the  
288 derivative of the molar amounts of ringwoodite at thermodynamic equilibrium as a function of  
289 depth. The growth curve  $x_{\text{rel}}$  is truncated where the relative growth of ringwoodite equals 1. At  
290 this pressure, hereafter called  $P_{\text{end}}$ , the molar amount of ringwoodite with kinetically inhibited  
291 growth conditions is considered to be equal to the amount of ringwoodite at thermodynamic  
292 equilibrium. Using the molar amount of ringwoodite at  $P_{\text{end}}$ ,  $n_{\text{Rw,end}}$ , we convert the grain growth  
293 curve (12) into molar amount of ringwoodite:

294

$$n_{\text{Rw}}(P) = n_{\text{Rw,end}} x_{\text{rel}}(P) \quad (13)$$

295

296 We calculate the mass balances of ferropericlase, majorite garnet and pyrope garnet, following  
297 the mineral reactions in (2) – (4). Using molar volumes of the individual mineral phases from  
298 our thermodynamic calculations in section 2.1, we convert molar amounts to volume fractions  
299 within the pressure interval from  $P_{\text{end}}$  to  $P_{\text{start}}$ .

300

### 301 **2.4 Calculation of density and seismic wave velocities**

302

303 Utilizing their volume fractions calculated as described in section 2.3, we calculate the seismic  
304 properties of the kinetically inhibited phase assemblages, for the Hill average of Voigt and  
305 Reuss bounds of the constituent minerals (Voigt, 1928; Reuss, 1929; Hill, 1965). For each  
306 mineral, the density and seismic velocity are determined at the given pressure and temperature.  
307 The seismic properties of the kinetically inhibited phase assemblages are fit to the seismic  
308 profiles of phase assemblages at thermodynamic equilibrium for the corresponding adiabatic  
309 mantle geotherm.

310

### 311 **2.5 Computation of 1-D synthetic seismograms**

312

313 The mineral physical results can be used to test if the effects of the kinetically inhibited

314 transformation of bridgmanite and ferropericlase to ringwoodite can be observed in  
315 seismological data. We focus on effects of PP and SS underside reflections and P-to-s  
316 conversions as they are commonly used probes to study upper mantle discontinuities. For this,  
317 the computed profiles of density, P wave and S wave velocity for equilibrated and kinetically  
318 inhibited phase assemblages are used to calculate 1-D reflectivity synthetic seismograms (Fuchs  
319 & Müller, 1971; Müller, 1985) for a range of dominant periods. Attenuation values are taken  
320 from PREM (Dziewonski & Anderson, 1981). For comparison, 1-D reflectivity synthetic  
321 seismograms using the 1-D velocity model ak135 (Kennett et al., 1995) and attenuation of  
322 PREM (Dziewonski & Anderson, 1981) are computed. Dominant periods of PP underside  
323 reflections are 1, 3, 5, 8 and 10 s. A source at 50 km depth is chosen and the data are analysed  
324 on the vertical component. For SS underside reflections, dominant periods are 5, 8, 10, 15 and  
325 20 s. The source is at 50 km depth and the data are analysed on the transverse component. 21  
326 receivers are distributed as a linear array around  $120^\circ$  epicentral distance with an aperture of  
327  $5^\circ$ . An epicentral distance of  $120^\circ$  is within typical distances used for PP and SS precursor  
328 studies (e.g., Shearer, 1991; Thomas & Billen, 2009; Deuss, 2009). The seismograms are  
329 aligned on the PP/SS arrival and are normalised to unit amplitude.

330

331 For the P-to-s conversions we use dominant periods of 0.5 s, 1 s and 2 s with a source depth of  
332 2 km to avoid interference of the P660s (and P410s for reference) arrivals with depth phases.  
333 We use a linear array with  $1^\circ$  receiver spacing between  $40^\circ$  and  $80^\circ$ . We calculate three-  
334 component seismograms (Z, R, T) but only analyse the radial component. Due to the simple  
335 source time function and the consistent waveforms, we abstain from deconvolving the vertical  
336 component P-wave and interpret the relevant waveform variation in the Results section.

337

### 338 **3 Results**

339

#### 340 **3.1 Effects of kinetic inhibition on seismic fine structure**

341

342 The effects of kinetic inhibition on phase assemblages are compared with phase assemblages at  
343 thermodynamic equilibrium with regard to the mineral proportions and seismic properties. The  
344 kinetically inhibited growth of ringwoodite and the metastable assemblage of ferropericlase and  
345 pyrope-bearing garnet change the phase proportions above the onset depth of the phase  
346 transformation (Figure 3). For equilibrated phase assemblages at a temperature of  $\sim 1850$  K  
347 (following the adiabat with  $1300^\circ\text{C}$  potential surface temperature, abbreviated hereafter as  
348 “ $1300^\circ\text{C}$  adiabat”; please see Table 2 and Cobden et al., 2008 for details), ringwoodite

349 dissociates to ferropericlasite and bridgmanite within a pressure interval of ~0.1-0.2 GPa (i.e.,  
350 ~2 km) at ~23.4 GPa (i.e., at ~659 km depth) (Figure 3a). The amount of garnet decreases due  
351 to the exsolution of bridgmanite, starting at ~22.7 GPa (~642 km).

352

353 The thickness of inhibited growth of ringwoodite is sensitive to temperature, grain size and  
354 mantle upwelling rate. For 5 cm grain size and 50 cm/yr upwelling rate, the kinetic inhibition  
355 leads to a growth of ringwoodite over a pressure interval of 1 GPa (~25 km) (Figure 3b). The  
356 gradual growth of ringwoodite results in a gradual consumption of ferropericlasite. At the  
357 pressure where ringwoodite starts to grow, the entire amount of bridgmanite reacts to garnet,  
358 thus its volume fraction increases to 70%. The recombination reaction of garnet and  
359 ferropericlasite to ringwoodite gradually decreases the amount of garnet towards lower pressures.

360

361 The excess of garnet changes the seismic structure around 660 km depth (Figure 3c). Since  
362 garnet has lower P- and S-wave velocities (garnet:  $v_p \approx 9.8$  km/s,  $v_s \approx 5.2$  km/s at 23.4 GPa and  
363 1850 K) than ringwoodite and bridgmanite (ringwoodite:  $v_p \approx 10.2$  km/s,  $v_s \approx 5.6$  km/s;  
364 bridgmanite:  $v_p \approx 11.2$  km/s,  $v_s \approx 6.3$  km/s at 23.4 GPa and 1850 K), the kinetically inhibited  
365 growth of ringwoodite and the abundance of garnet results in a low-velocity layer above the  
366 660 km discontinuity of  $\Delta v_p \approx -5\%$ ,  $\Delta v_s \approx -7\%$  (Figure 45). The low-velocity layer above the  
367 lower boundary of the ringwoodite-forming reaction thus increases the P- and S-wave  
368 impedance contrast. When the growth of ringwoodite has reached equilibrium conditions, the  
369 profiles of density and seismic wave speeds intersect the profiles of the phase assemblages at  
370 thermodynamic equilibrium.

371

372 The effects of initial grain size  $d$  and mantle upwelling rate  $v$  can be described by a kinetic  
373 parameter  $D$  which is proportional to the upwelling rate and the squared grain size (Dobson &  
374 Mariani, 2014):

375

$$D \sim v d^2.$$

377

378 Scenarios of kinetic inhibition with different initial grain sizes and upwelling rates give the  
379 same results if they can be described by the same kinetic parameter. Therefore, the kinetic  
380 parameter is an appropriate entity to compare scenarios of kinetically inhibited phase  
381 assemblages. Since the kinetic parameter scales with the square of initial grain size  $d$ , the kinetic  
382 inhibition of ringwoodite is more sensitive to grain size than to upwelling rate. In this study, we

383 set the proportional factor to unity and use

384

$$385 \quad D = v d^2.$$

386

387 in order to compare scenarios of initial grain size  $d$  and mantle upwelling rate  $v$ .

388

389 The thickness of the low-velocity layer, i.e., the depth interval between the onset and the  
390 completion of ringwoodite growth, increases exponentially with increasing kinetic parameter  
391 of the kinetic parameter (Figure 5). The exponential behaviour can be explained by the grain  
392 growth equation which includes the Arrhenius relation for the reaction rate parameter  $k(T)$  (cf.  
393 Eq. (9)). For kinetic parameters  $D$  between 10 and 100 cm<sup>3</sup>/yr, kinetic inhibition occurs over a  
394 pressure interval of ~0.1 GPa which is as narrow as the transformation pressure interval of the  
395 recombination of ringwoodite from ferropericlase and bridgmanite. A transformation pressure  
396 interval of ~0.1 GPa is equivalent to a low-velocity layer of thickness ~2 km which is likely  
397 below the resolution threshold even of short period PP waves and at the resolution level for  
398 receiver function (Ps). Seismic waves therefore sample the reflector and low-velocity layer  
399 above in the same way as the reflector for phase assemblages at thermodynamic equilibrium.  
400 The onset of seismically detectable kinetic inhibition in terms of the kinetic parameter  $D$  is  $D$   
401  $> 100$  cm<sup>3</sup>/yr for 1850 K,  $D > 1000$  cm<sup>3</sup>/yr for 1960 K and  $D > 2500$  cm<sup>3</sup>/yr for 2050 K. The  
402 temperature dependence can be explained by the increased diffusion of ions at higher  
403 temperatures.

404

## 405 **3.2 Synthetic seismograms**

406

407 We have calculated 1-D synthetic seismograms from the synthetic velocity and density profiles  
408 using the reflectivity methods (Fuchs & Müller, 1971; Müller, 1985) as described in section  
409 2.4. Here we discuss the effects of the kinematically inhibited velocity structures on PP/SS  
410 underside reflections and P-to-s conversions, two probes regularly used for the study of upper  
411 mantle discontinuities.

412

### 413 **3.2.1 PP and SS Precursors**

414

415 The seismograms calculated for the velocity models with kinetically inhibited recombination  
416 reaction of ringwoodite from bridgmanite and ferropericlase are compared to seismograms for  
417 velocity models with equilibrated phase assemblages. Using linear vespagrams (Davies et al.,

418 1971; Muirhead et al., 1976), PP/SS and P660P/S660S waveforms are extracted for travel times  
419 predicted by ak135 (Kennett et al., 1995) (Figure 6).

420

421 For velocity models where the kinetic inhibition of the ringwoodite growth results in a thicker  
422 low-velocity layer of the order of  $\sim 25$  km, the PP/SS precursors off the reflector caused by  
423 kinetically inhibited recombination of ringwoodite are noticeably advanced for reaction  
424 intervals wider than 25 km compared to PP/SS precursors reflected off discontinuities of  
425 equilibrated phase transformations (Figure 3b). The travel time advances also affect secondary  
426 phases of the underside reflections such as depth phases, e.g., pP660P (Figure 6). For the kinetic  
427 parameters tested here, the maximum travel time advances for PP underside reflections are  $\sim 0.8$   
428 s, whereas maximum travel time advances for SS underside reflections are  $\sim 1.3$  s (Figure 7).  
429 These travel time residuals correspond to apparent maximum shifts of the discontinuity depth  
430 on the order of 5 km.

431

432 Amplitudes for PP precursors with a dominant period of 1 s reflected off the kinetically  
433 inhibited formation of ringwoodite vary between 120% and 160% relative to the equilibrium  
434 model at 1850 K. For the same temperature, amplitudes of SS precursors with a dominant period  
435 of 5 s range between 120% and 160% relative to the equilibrium model. The precursor  
436 amplitudes for the kinetically inhibited phase assemblages decrease with increasing periods  
437 (Figure 8). The transition from enhanced to apparently unaltered precursor amplitudes is at a  
438 dominant period of 5 s for PP and 10 s for SS. Moreover, the PP and SS precursor amplitudes  
439 increase with increasing kinetic parameters and increasing thickness of the low-velocity layer.  
440 The PP/SS precursors off the lower boundary of the phase transformation are less affected by  
441 interference with PP/SS precursors off the gradient above the low-velocity layer.

442

443 While it appears that underside reflections of PP waves at the ringwoodite to bridgmanite plus  
444 ferropericlase transition are more difficult to detect than the corresponding SS underside  
445 reflection (Estabrook & Kind, 1996), there is evidence for P660P waves in some regional  
446 studies (e.g., Deuss et al., 2006; Thomas & Billen, 2009; Schmerr & Thomas, 2011, Saki et al.,  
447 2014) and usually the amplitudes of observed P660P waves are smaller than predicted by PREM  
448 or a pyrolytic mantle. There is also a noted discrepancy between the detection of P660P and its  
449 equivalent underside reflection of PKPPKP (P'660P') (Day & Deuss, 2013). The scarcity of  
450 P660P detections in recorded data can likely be explained by compositional and thermal effects  
451 that are not captured in our modelling and the assumed pyrolytic starting model (Lessing et al.,  
452 2015; Guo & Zhou et al., 2020; Waszek et al., 2021).

453 Underside reflection of PP from the 660 km phase transition are rarely detected in recorded data  
454 although such detections exist (Deuss et al., 2006; Day & Deuss, 2013).

455

456 The finite width of the low-velocity layer (Figure 3a) and the vertical resolution of the seismic  
457 waves lead to a frequency-dependent behaviour of P660P/S660S precursor amplitudes (Figure  
458 9). PP waves with 1 s dominant period appear to resolve the increased P wave impedance  
459 contrast off the reflector caused by the kinetically inhibited recombination of ringwoodite. The  
460 increased P wave impedance contrast results in increased PP precursor amplitudes compared to  
461 the reflections for equilibrated phase assemblages. If the dominant period is increased, PP  
462 precursor amplitudes off the reflector with kinetic inhibition decrease and have similar PP  
463 precursor amplitudes off the reflector as those without kinetic inhibition.

464

### 465 **3.2.2 P-to-s conversions (P receiver functions)**

466

467 Examples of the synthetic waveforms for the kinetically inhibited models and pyrolitic models  
468 at equilibrium conditions are shown in Figure 10. For varying temperatures at equilibrium, we  
469 observe strong conversions from the 660 km discontinuity and other discontinuities (e.g., 410-  
470 km as well as shallower) with minor waveform and amplitude variations. We calculated  
471 synthetics for dominant periods of 0.5 s, 1 s, and 2 s and observe very minor variability of the  
472 results. While absolute travel times for P- and P-coda arrivals vary, differential travel time  
473 variations are minor, with 0.2 s and 0.6 s for the 1960 K and 2050 K simulations, respectively,  
474 compared to the 1850 K simulations. Waveform variations are limited to larger moveout for a  
475 deeper transition, arriving 5.1 to 8.9 s after P660s.

476

477 For the kinetically inhibited transformations we observe strong waveform variations with  
478 temperature, where the waveforms of the high temperature models show a dominant precursor  
479 about 7 s preceding the arrival from P660s (Figure 10b and c). The precursor due to the low  
480 velocity zone will lead to a likely detectable negative polarity precursor in P-wave receiver  
481 functions (see Figure S1 for an additional velocity profile at 2050 K). Those detectable negative  
482 polarity precursors have been reported in some up-flowing regions of the mantle (e.g., Negi et  
483 al., 2022). We observe amplitude variations dependent on grain size, upwelling rate, and  
484 temperature (Figure 10d). Amplitudes vary between 52% and 122% relative to the equilibrium  
485 model at 1960 K with the smallest amplitudes observed for the highest temperatures.  
486 Interference effects lead to apparent inverted waveforms for the P-to-s conversion relative to P.  
487 Both amplitude and waveform variations for these models would be large enough to be

488 observed in receiver function analyses.

489

490 Travel times relative to P vary slightly for the different models. We observe travel time  
491 variations of  $\pm 0.3$  s relative to the equilibrium model at 1960 K. While detectable, these might  
492 be difficult to detect for recorded data due to the 3-D velocity variations known to exist in upper  
493 mantle.

494

## 495 **4 Discussion**

496

### 497 **4.1 Estimates of uncertainties**

498

499 The effects of initial grain size and upwelling rate on the seismic properties of kinetically  
500 inhibited phase assemblages have been investigated in section 3.1. Among the kinetic input  
501 parameters, i.e., kinetic rate parameter  $k_0$ , reaction enthalpy  $\Delta H_a$  and temperature  $T$ ),  
502 uncertainties of the reaction enthalpy predominantly affect the seismic properties of kinetically  
503 inhibited phase assemblages (Figure 11a and b). Dobson & Mariani (2014) measure the  
504 enthalpy of the reaction with an uncertainty of  $\pm 40$  kJ mol<sup>-1</sup> for the ringwoodite layer (Table 1).  
505 Varying the enthalpy of the reaction with this value while keeping remaining parameters  
506 constant in eq. (9) results in variations of the growth time of ringwoodite until equilibrium  
507 conditions by one order of magnitude (Figure 11b).

508

509 Converting growth time into the thickness of the reaction interval using the upwelling rate gives  
510 uncertainties of the reaction interval of one order of magnitude. Given the uncertainty of the  
511 enthalpy of the reaction for the ringwoodite layer, the reaction interval could therefore range  
512 from 1 km to  $\sim 250$  km for kinetic parameters ranging from 1 to 10,000 cm<sup>3</sup>/yr. On the one hand,  
513 reaction intervals with a width of 1-10 km are close to the vertical resolution of PP and SS  
514 precursors (e.g., Thomas & Billen, 2009; Schmerr & Thomas, 2011) and it might be challenging  
515 to discriminate kinetic effects from other influences on reaction intervals. On the other hand,  
516 reaction intervals with widths of 100-200 km would give increased PP and SS precursor  
517 amplitudes due to the increased P wave and S wave impedance contrast and due to the more  
518 gradual change of seismic properties from the low-velocity region relative to seismic properties  
519 at thermodynamic equilibrium. P-to-s conversions at the discontinuities beneath the receiver  
520 typically show higher resolution and are likely able to detect even the thinner layers inferred  
521 for the smaller kinetic parameters. Thicker reaction intervals would result in strongly reduced  
522 amplitudes of Ps. Reaction intervals of 100-200 km would also affect phase transformations at

523 smaller pressures/shallower depths, e.g., the wadsleyite to ringwoodite transition, as well as the  
524 phase transformations from pyroxene to garnet and from garnet to calcium silicate perovskite.  
525 The kinetics of the olivine to wadsleyite phase transition (Poirier, 1982; Brearley et al., 1992;  
526 Kirby et al., 1996) and of pyroxene to garnet (Hogrefe et al., 1994) have been studied for  
527 harzburgitic mantle composition so far. However, kinetic data of other phase transformations in  
528 the transition zone are necessary to allow for consistent treatment of kinetic effects on seismic  
529 properties within and above the transition zone.

530

531 The reaction rate constant  $k_0$  has a minor effect on the seismic properties of the kinetically  
532 inhibited phase assemblage (Figure 11c). P- and S-wave velocity vary by  $\pm 0.05$  km/s, and  
533 density is changed by  $\pm 0.02$  g cm<sup>-3</sup> for varying  $k_0$  to its lower and upper bounds. The thickness  
534 of the low-velocity layer above the 660 km discontinuity shifts by  $\pm 4$  km.

535

## 536 **4.2 Seismological and geodynamic implications**

537

538 Patterns of mantle flow and dynamics have been investigated using seismic wave speed  
539 anomalies in seismic tomography models (e.g., Li et al., 2008; Ritsema et al., 2011) and seismic  
540 waves sampling the upper mantle and mantle transition zone discontinuities using receiver  
541 functions (e.g., Vinnik, 1977; Farra & Vinnik, 2000; Rondenay, 2009), short period P'P'  
542 reflections (e.g., Niazi & Anderson, 1965; Vidale & Benz, 1993; Day & Deuss, 2013), and PP  
543 and SS precursors (e.g., Shearer, 1991; Deuss, 2009). Since the upper mantle and mantle  
544 transition zone discontinuities are sensitive to thermal variations due to their Clapeyron slopes  
545 (Bina & Helffrich, 1994; Helffrich, 2000; Weidner & Wang, 2000), variations of discontinuity  
546 depth and of transition zone thickness (i.e., the depth between the 410 and 660 km discontinuity)  
547 and constraints from mineral physics, are frequently used to infer thermal variations in the  
548 mantle (e.g., Shearer, 2000, Ritsema et al. 2009).

549

550 The investigation of reaction kinetics of the reaction from ferropericlasite and bridgmanite to  
551 ringwoodite can add further constraints on mantle temperature and flow patterns. The effect of  
552 kinetic inhibition can be compared to seismic data using a two-step approach. In a first step,  
553 through forward modelling of synthetic seismograms for a range of temperatures, upwelling  
554 rates, and grain sizes. Relevant temperatures can be determined from adiabatic mantle  
555 geotherms (Brown & Shankland, 1981; Spiliopoulos, 1984; Ita & Stixrude, 1992; Cobden et  
556 al., 2008) to estimate excess temperatures (e.g., Sleep, 1990; Schilling, 1991; Herzberg et al.,  
557 2007). Heat flux estimates (e.g., Sleep, 1990; Courtillot et al., 2003) can be converted into

558 estimates for upwelling rates using convection models for vertical upwellings. A steady-state  
559 axisymmetric plume with temperature-dependent viscosity might serve as a starting point for  
560 estimates of upwelling rates as a function of heat flux and excess temperatures (Loper & Stacey,  
561 1983; Schubert et al., 2001). In a second step, the seismic recordings can be compared to the  
562 set of synthetic seismograms and those models are taken which give the best match with regard  
563 to frequency dependent travel time residuals and amplitude ratios. The set of matching models  
564 gives the range of corresponding grain sizes and upwelling rates.

565

566 Considering the effects of reaction kinetics of ferropericlase and bridgmanite to ringwoodite on  
567 the amplitudes and travel times may offer additional constraints on mapping mantle flow using  
568 P-to-s receiver functions. P-to-s conversions typically have higher frequencies (Li et al., 2000;  
569 Wölbern et al., 2006). Several studies (van der Meijde et al., 2003; Schmandt, 2012; Huckfeldt  
570 et al., 2013) show frequency dependent behaviour of P-to-s conversions at mantle transition  
571 zone discontinuities. We see little frequency dependence of P-to-s conversion in our synthetic  
572 models within the range typically used for receiver functions, but observe strong waveform  
573 variations that are likely to be observable in recorded data as well as strong amplitude variations  
574 and small travel time variations that might add additional constraints for detecting regions with  
575 mantle upstreams.

576

577 The analysis of PP and SS precursors might have some resolution limitations with respect to  
578 the scale of the mantle upwellings which might hamper the detection of the predicted  
579 waveforms. Geodynamic modelling indicates that deep thermal mantle plumes have typical  
580 diameters of  $\sim 200$  km (e.g., Sleep, 1990; Schubert et al., 2001; Steinberger & Antretter, 2006)  
581 whereas thermochemical plumes might have larger diameters ranging from 600 to 800 km  
582 (Farnetani, 2005; Farnetani & Hofmann, 2010; Stockmann et al., 2019). Lessing et al. (2015)  
583 use the 2.5-D axisymmetric finite difference technique PSVaxi (Jahnke, 2008; Thorne et al.,  
584 2013) to examine the effects of topography of the upper mantle discontinuities imaged by PP  
585 underside reflections with dominant periods of 2 s. Their models show that the PP wavefield  
586 reflected off a locally elevated 660 km with a diameter of 200 to 400 km does not differ  
587 significantly from the PP wavefield reflected off a flat reflector at 660 km depth. The size of  
588 the detailed topography of the perturbed reflector is smaller than the PP Fresnel zone at 660 km  
589 depth and thus might not be resolved. The detection of effects of the kinetically inhibited  
590 reaction from ferropericlase and bridgmanite to ringwoodite might therefore be challenging in  
591 narrow mantle upwellings. Although conduction might lead to thermal halos around narrow  
592 mantle plumes the effect might be small (Sleep, 2004) and might not influence the detectability

593 of the plume using underside reflection. Nonetheless, recent tomographic images show that  
594 plumes might be broad upstreams (French and Romanowicz, 2015), very different to early  
595 numerical and laboratory plume experiments which might make the upstream detectable  
596 seismically. The pooling of plume material beneath the 660 km discontinuity due to the  
597 endothermic phase transition might lead to larger scale temperature anomalies in the region of  
598 plume-like mantle upstreams (Tackley et al., 1994, Schubert et al., 1995, Stockmann et al.,  
599 2019). Possible candidates for upwellings with large, detectable lateral dimensions are large  
600 transient domes near the mantle transition zone (Courtillot et al., 2003, Davaille et al., 2005,  
601 Kumagai et al., 2007; Kumagai et al., 2008). The transient domes are characterised by  
602 upwelling rates slower than deep mantle plumes due to their small buoyancy ratio (ratio of  
603 chemically driven buoyancy to thermally driven buoyancy). Nonetheless, the transient domes  
604 could have excess temperatures up to 300 K in their centre and form a thermal boundary layer  
605 with the ambient mantle (Davaille et al., 2005). Upwelling mantle material might pond beneath  
606 the 660 km discontinuity when encountering the phase transformation from bridgmanite and  
607 ferropericlase to ringwoodite (e.g., Schubert & Turcotte, 1975; Schubert & Tackley, 1995;  
608 Tackley, 1995; Steinbach & Yuen, 1997; Brunet & Yuen, 2000; Davaille et al., 2005, Farnetani  
609 & Samuel, 2005). Due to the negative Clapeyron slope, the transformation pressure of the  
610 warmer material is shifted towards lower pressure, resulting in a negative buoyancy force which  
611 delays further upwelling. The material starts to pond beneath the phase transformation (Davaille  
612 et al., 2005) and spreads laterally, growing with dimensions broader than plume conduits (e.g.,  
613 Steinbach & Yuen, 1997; Davaille et al., 2005, Stockmann et al., 2019).

614

615 Furthermore, the viscosity contrast across the 660 km discontinuity (e.g., Mitrovica & Forte,  
616 2004) might support lateral mantle flow beneath the discontinuity. The geodynamic models of  
617 Tosi & Yuen (2013) show additional evidence for horizontal flow whose scale is governed by  
618 the viscosity contrast due to temperature variations. For small values of the viscosity contrast,  
619 broad and highly viscous plumes are generated that tend to pass through the transition zone  
620 relatively unperturbed. For higher values of the viscosity contrast ( $10^2 \leq \Delta\eta_T \leq 10^3$ ), the  
621 geodynamic simulations show horizontal channel flows beneath the 660 km discontinuity  
622 which might extend up to 1500 km horizontally before tilting to more vertical flow. Therefore,  
623 lateral mantle flow could change the fine-scale seismic structure at ~660 km depth in a way that  
624 could allow for the detection of the effects of kinetically inhibited growth with seismic methods  
625 sensitive to upper mantle discontinuity structure.

626

627 This study has predicted effects of kinetically inhibited growth of ringwoodite in mantle

628 upstreams and the expected effect on seismic probes for discontinuity structure. We find that  
629 the effect on PP/SS underside reflection travel times is small; amplitudes, although harder to  
630 assess in recorded data, might show a larger response to the altered discontinuity structure. For  
631 P-to-s conversions travel time effects are minor but waveforms and amplitudes of conversions  
632 are noticeably affected. Potentially P-to-s receiver function could be a suitable probe to detect  
633 kinetic effects in the phase transitions and to map out mantle flow.

634

## 635 **5 Conclusions**

636

637 The effects of reaction kinetics of bridgmanite and ferropericlase to ringwoodite in upwelling  
638 mantle on elastic properties were investigated using data of kinetic experiments and internally  
639 self-consistent thermodynamic modelling. As the kinetic experiments show, growth of  
640 ringwoodite is initiated at ferropericlase grain boundaries and is kinetically inhibited due to the  
641 necessary diffusion of chemical components. At the initiation of ringwoodite growth,  
642 bridgmanite transforms into metastable garnet. Ringwoodite grows from the metastable  
643 assemblage of ferropericlase and pyrope-bearing garnet.

644

645 The results of the analysis of kinetically inhibited transformation at ~660 km depth predict  
646 possible effects of kinetically inhibited growth of ringwoodite on seismic data which can be  
647 summarized as follows:

648 1. The increased amount of ferropericlase and garnet results in a localized low velocity  
649 zone of  $\Delta V_P \sim -5\%$ ,  $\Delta V_S \sim -7\%$  directly above the 660 km seismic discontinuity due to  
650 the lower seismic velocities and density of ferropericlase and garnet compared to  
651 ringwoodite and bridgmanite. The impedance contrasts for P and S waves are thus  
652 increased compared with seismic velocity profiles of an equilibrated pyrolite mineral  
653 assemblage.

654 2. The effects of reaction kinetics on the seismic fine structure at ~660 km depth depend  
655 on the product of the squared grain size and upwelling rate. The change of seismic  
656 properties correlates with the product of squared grain size and upwelling rate.  
657 Modelling 1-D synthetic seismograms using the seismic profiles of kinetically inhibited  
658 phase assemblages show travel time delays of 0.4-0.8 s for PP precursors and delays of  
659 1.2-1.6 s for SS precursors for kinetic inhibition of the ringwoodite-forming reaction.  
660 The finite width of the low-velocity layer above the seismic discontinuity results in  
661 frequency dependent behaviour of PP and SS underside reflection amplitudes, with  
662 higher amplitudes for 1-3 s PP waves and 5-8 s SS waves. However, high-frequency PP

663 and SS waves are rarely observed, and the predicted travel time differences are of  
664 similar magnitude as typical picking uncertainties. Hence, it might be challenging to  
665 utilize PP and SS underside reflections in order to observe kinetic effects.

- 666 3. Synthetic seismograms for P-to-s converted waves for velocity models for kinetically  
667 inhibited transformations predict strong waveform variations for higher temperature  
668 models and amplitude variations dependent on temperature, grain size and upwelling  
669 rate. Travel time variations are maximum  $\pm 0.3$  s relative to models at equilibrium for  
670 temperature of 1960 K. Receiver functions therefore might be a suitable seismic probe  
671 to look for kinetic effects on the seismic structure of the 660 km discontinuity in mantle  
672 upwellings. A future re-evaluation of P-to-s receiver functions at mantle upwellings  
673 would be necessary to study the predicted effects described in this paper.
- 674 4. Possible candidates for mantle upwellings where reaction kinetics may become  
675 important are transient domes with low buoyancy ratios and large lateral dimensions, or  
676 broader scale plume upwellings. Material stalling at the phase transformation of  
677 bridgmanite and ferropericlasite to ringwoodite might generate structure in the resolution  
678 range of P-to-s converted waves.

679

## 680 **Data Availability**

681

682 The resulting data of this work and sources code developed during this work are not publicly  
683 available. Except the results from kinetic experiments by Dobson & Mariani (2014), data are  
684 based on tools available in the public domain:

- 685 1. Perple\_X program package (Connolly, 2005), utilising the elastic and thermodynamic  
686 properties and the solid solution model of Stixrude & Lithgow-Bertelloni (2011)  
687 2. 1-D synthetic seismic modelling using the reflectivity method (Fuchs & Müller, 1971)  
688 3. and data processing with seismic waveform analysis tool SeismicHandler (Stammler,  
689 1993).

690 If you are interested in detailed data, please contact the corresponding author.

691

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693

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704 1.

705

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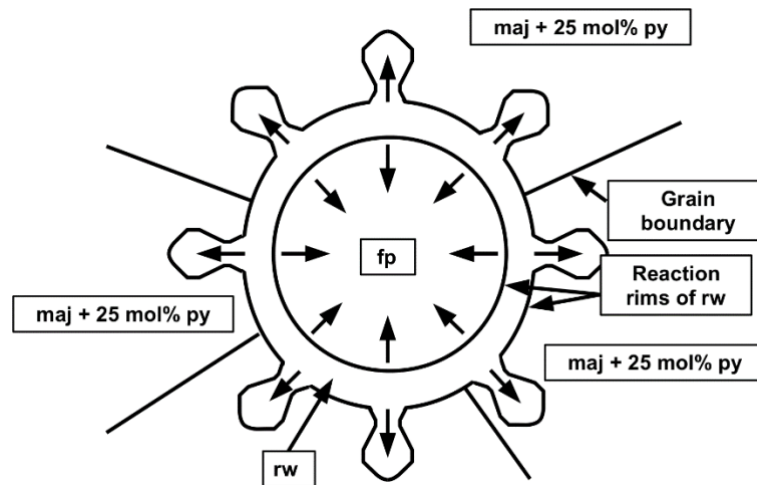
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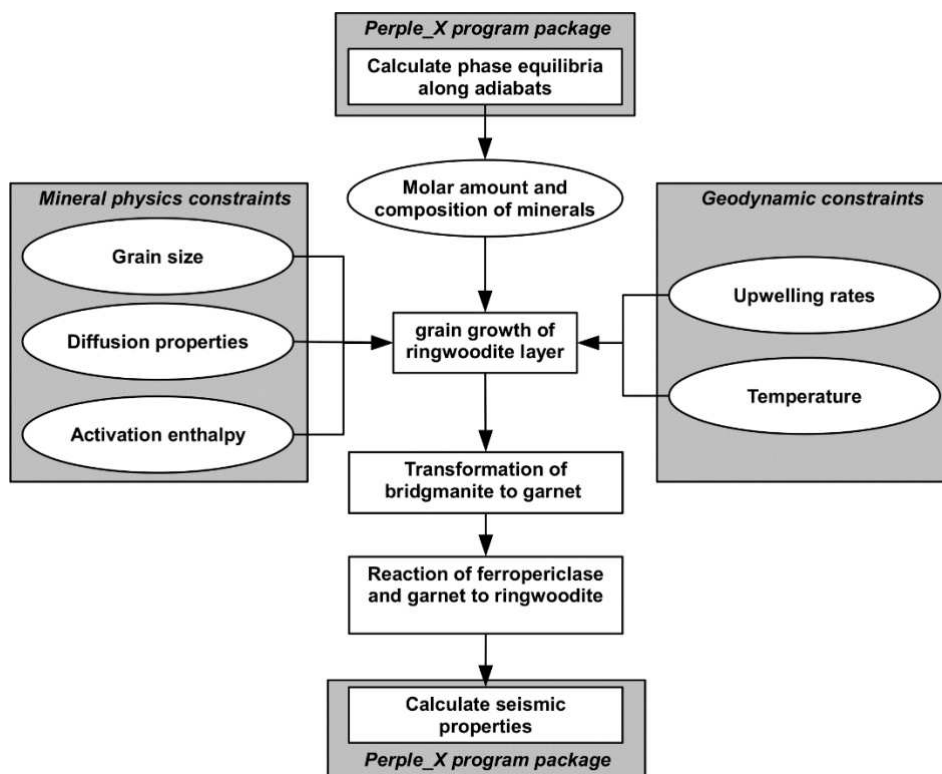
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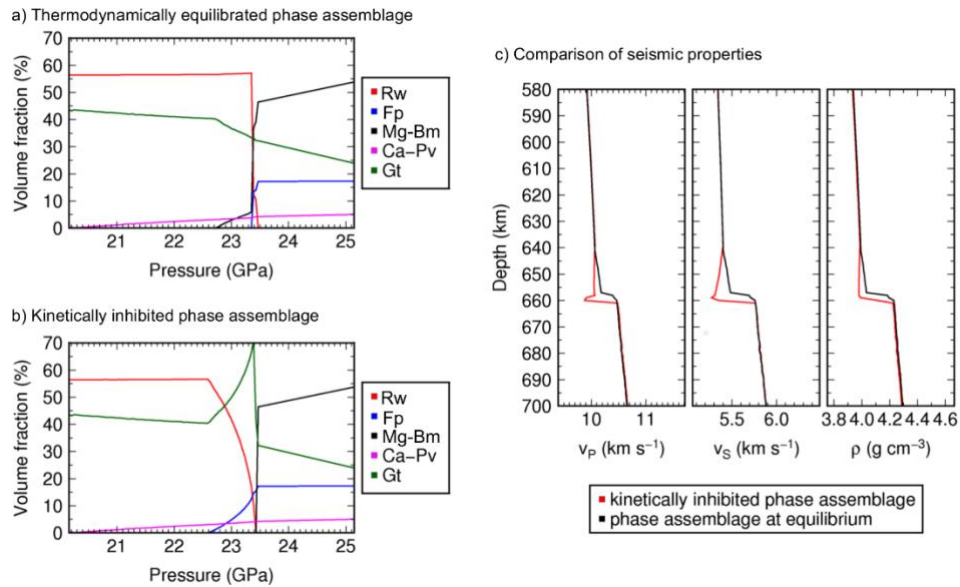
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983 **Figure 1:** Sketch of the reaction mechanism for the recombination of bridgmanite and  
 984 ferropericlasite (fp) to ringwoodite (rw). The ringwoodite layer nucleates on the fp grain surface  
 985 and grows into majorite garnet (maj) which contains 25 mol% Al-bearing pyrope (py) in a  
 986 fingering instability. After Dobson & Mariani (2014).



990 **Figure 2:** Forward modelling of seismic properties for kinetically inhibited growth of  
 991 ringwoodite and corresponding mineral reactions for ferropericlasite, bridgmanite and garnet.



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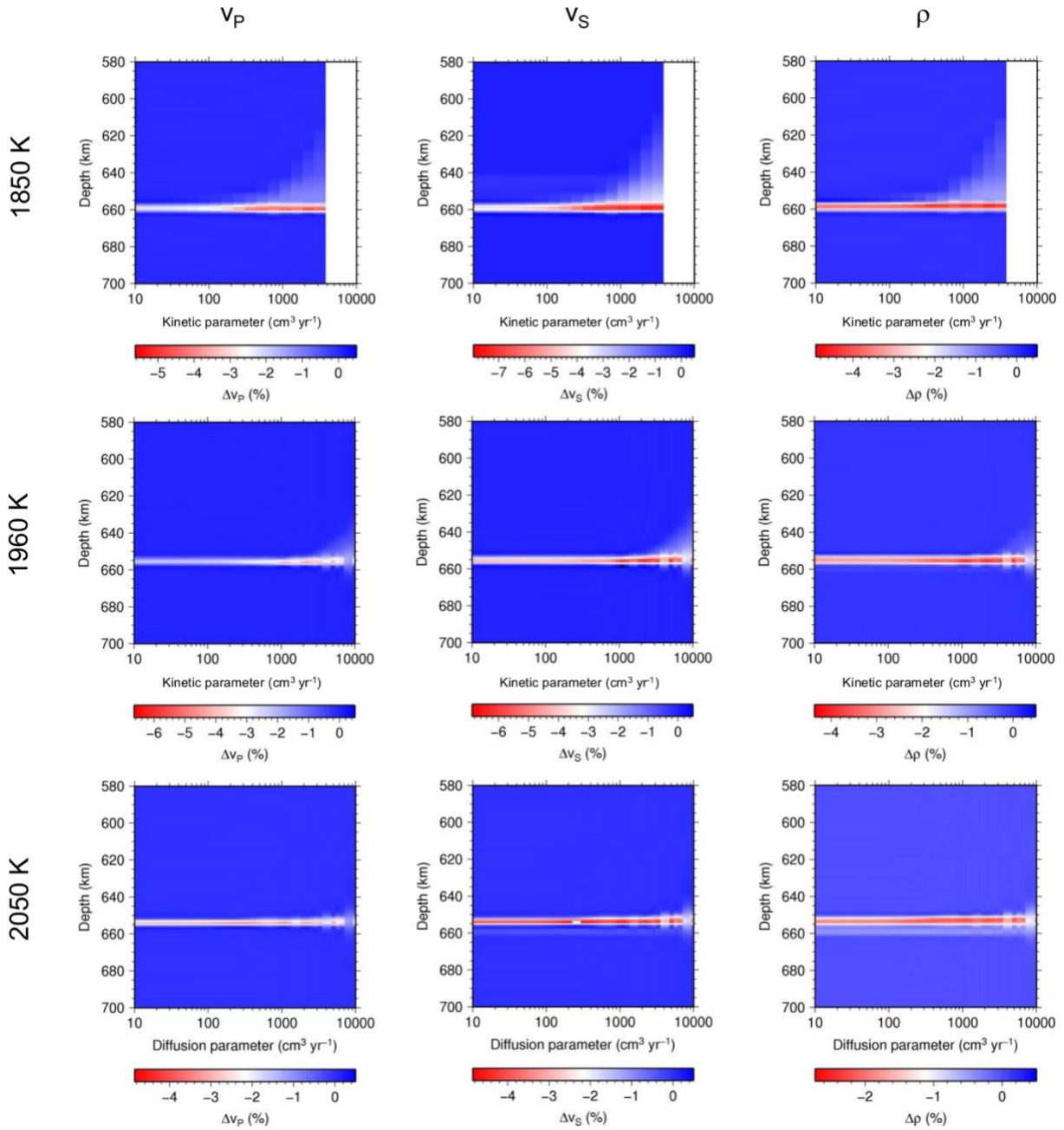
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995 **Figure 3:** Example for effects of kinetic inhibition of ringwoodite growth on volume fractions  
 996 of minerals and on seismic properties of phase assemblages for temperature of 1850 K  
 997 (following the 1300°C adiabat), 5 cm initial grain size and 50 cm/yr upwelling rate. a) Volume  
 998 fractions of equilibrated phase assemblages as a function of pressure for pyrolite composition  
 999 along the 1300°C adiabat. Mineral phases are abbreviated as follows: Rw: ringwoodite, Fp:  
 1000 ferropericlase, Mg-Bm: magnesium bridgmanite, Ca-Pv: calcium perovskite, Gt: garnet. b)  
 1001 Volume fractions of kinetically inhibited phase assemblages as a function of pressure for  
 1002 pyrolite composition along the 1300°C adiabat. c) Comparison of seismic velocities and density  
 1003 between equilibrated and kinetically inhibited phase assemblages.

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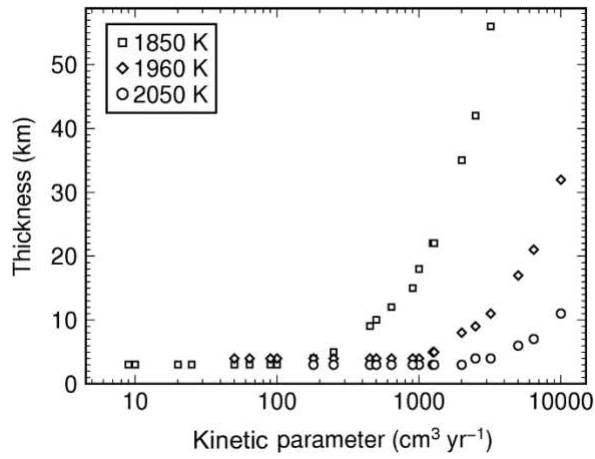
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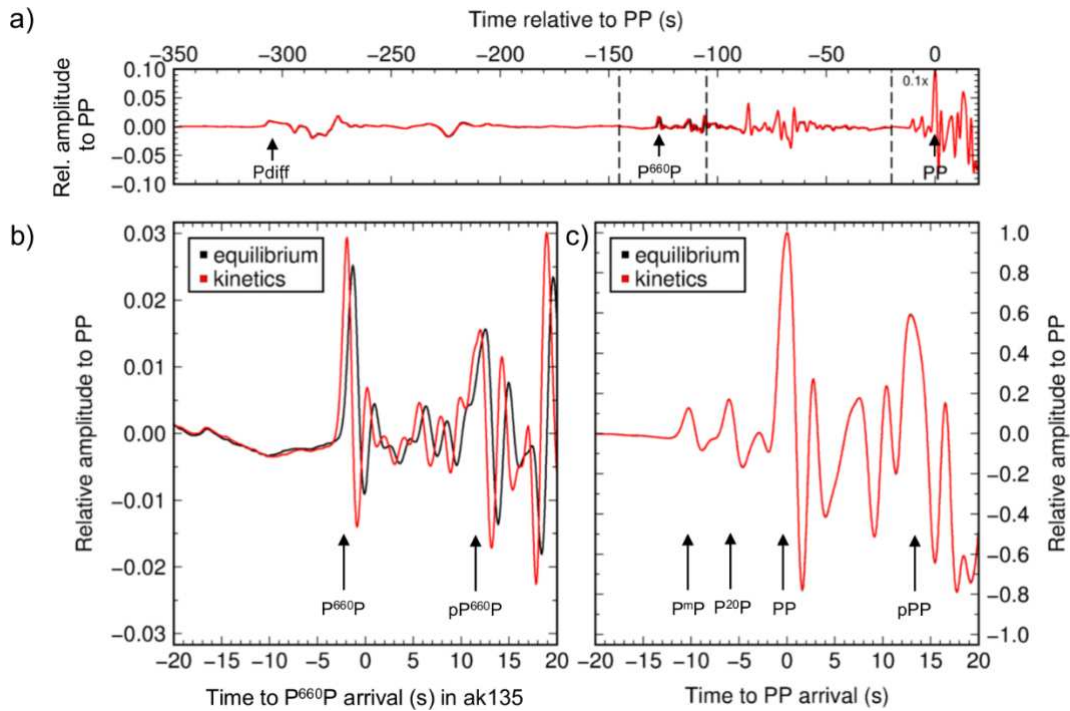
**Figure 4:** Relative differences of P wave velocity, S wave velocity and density due to the kinetically inhibited growth of ringwoodite and the presence of metastable ferropericlase and garnet for temperatures of 1850 K, 1960 K and 2050 K. No data are available for kinetic parameters  $> 4000 \text{ cm}^3 \text{ yr}^{-1}$  because the ringwoodite growth interferes with the ringwoodite-to-wadsleyite transition. To date, no studies with kinetic data for the ringwoodite-to-wadsleyite transition in mantle upstreams are known to the authors.



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1016 **Figure 5:** Thickness of the reaction interval where ringwoodite grows from the recombination  
 1017 of ferropericlasite and garnet as function of the kinetic parameter  $D$ . The kinetic parameter  $D$  is  
 1018 proportional to the upwelling rate  $v$  and the square of grain size  $d$ , for simplicity we set the  
 1019 proportion factor to unity so that  $D = v d^2$ . Scenarios which have different grain size and  
 1020 upwelling rates, but have the same kinetic parameter, show the same kinetic inhibition.

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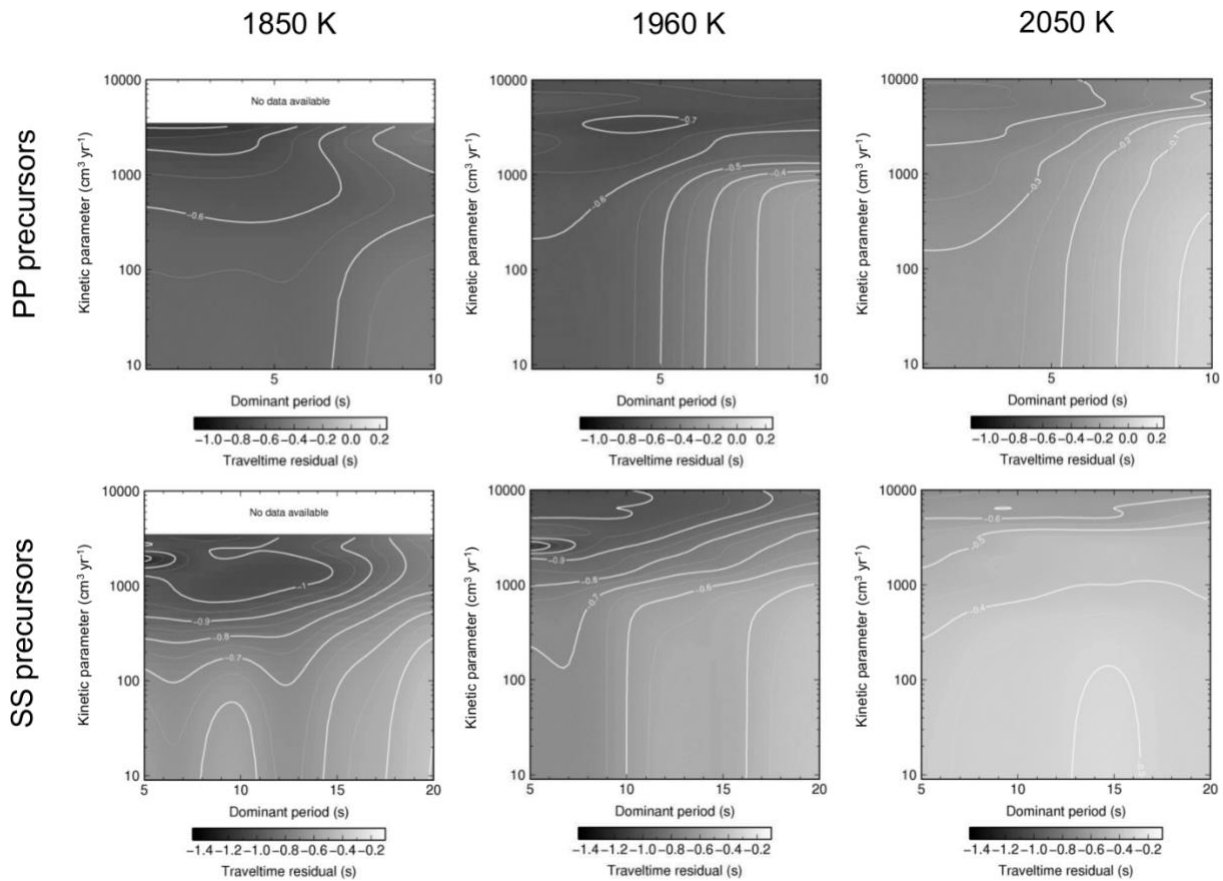


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1025 **Figure 6:** Synthetic seismograms of PP precursors with a dominant period of 3 s from seismic  
 1026 models of equilibrated and kinetically inhibited phase assemblages at an array with reference  
 1027 epicentral distance of 120 deg, a temperature of 1850 K, grain size of 5 cm and a vertical  
 1028 upwelling rate of 50 cm/yr. a) shows the linearly stacked traces of all calculated seismograms  
 1029 for equilibrated pyrolite phase assemblages (black) and kinetically inhibited phase assemblages  
 1030 (red) aligned with slowness for PP from ak135 (Kennett et al., 1995). Time windows around  
 1031 the predicted arrivals of P660P and PP from ak135 shown b) and c) and are marked by vertical  
 1032 lines and the phase names. Arrows indicate the arrivals of P660P and pP660P in zoomed  
 1033 diagram b) and the arrivals of PP and pPP in zoomed diagram c).

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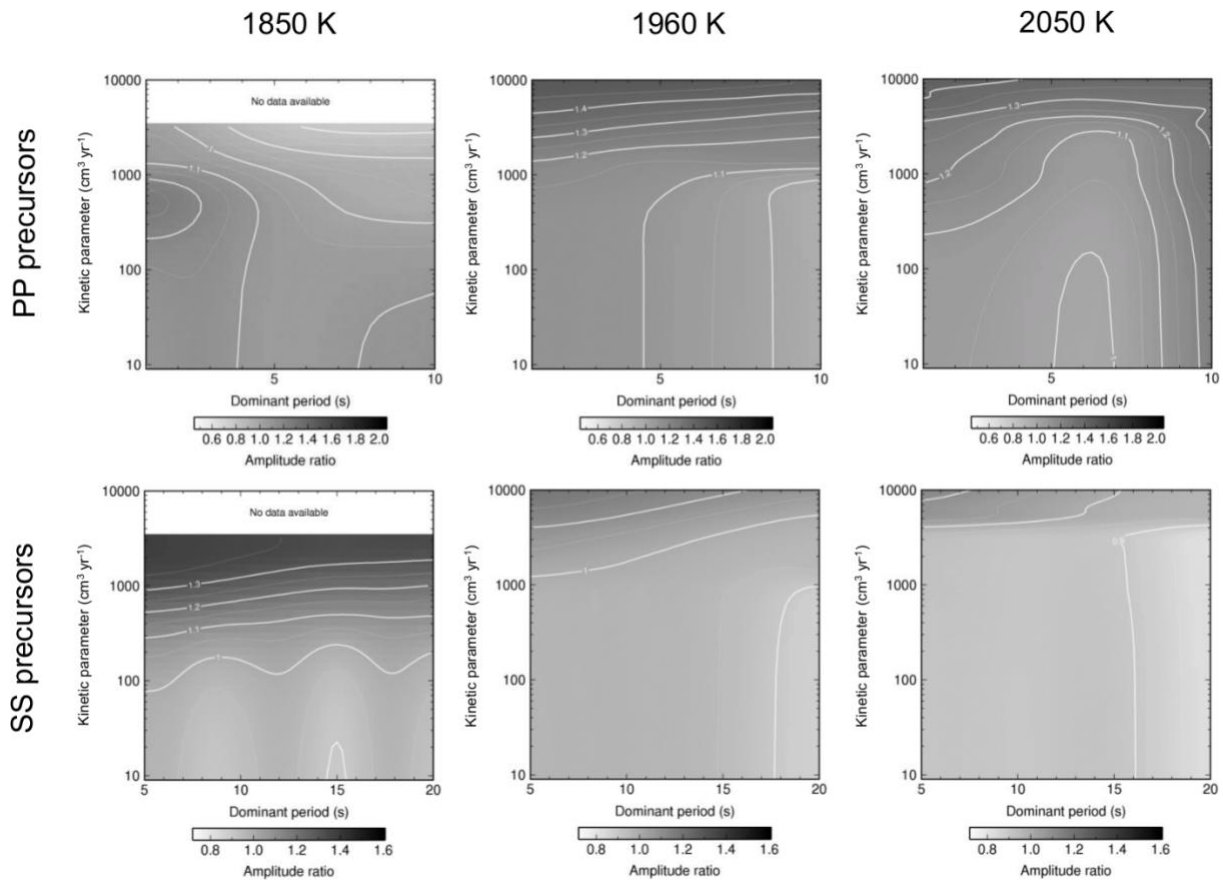


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1037 **Figure 7:** Travel time residuals of stacked PP and SS precursors between kinetically inhibited  
 1038 and equilibrated backward reaction to ringwoodite as a function of temperature and kinetic  
 1039 parameter for varying dominant periods at an epicentral distance of 120°. Bold lines denote  
 1040 changes of travel time residuals by 0.1, thin lines denotes changes of travel time residuals by  
 1041 0.05. Data are not available at temperature pf 1850 K and kinetic parameters  $> 4000 \text{ cm}^3 \text{ yr}^{-1}$   
 1042 (cf. Figure 4).

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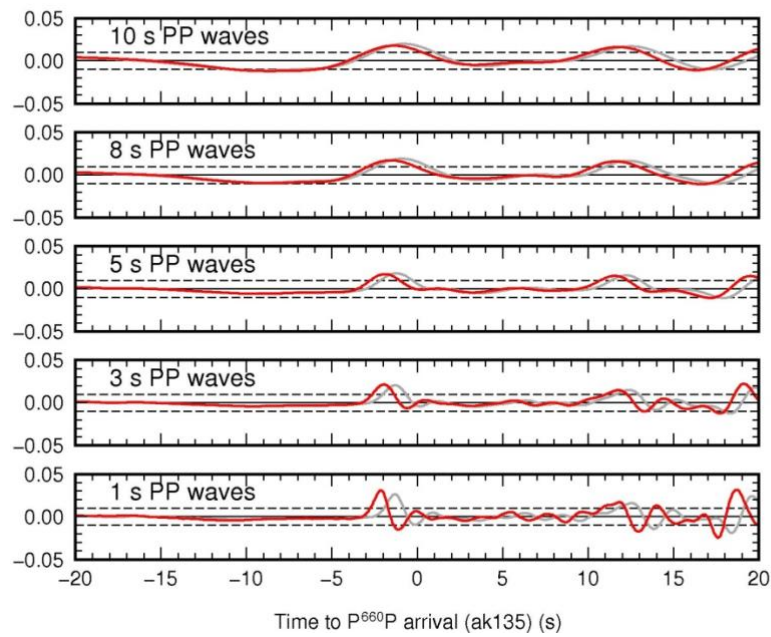
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**Figure 8:** PP and SS precursor amplitude ratios of reflections off the kinetically inhibited reaction vs. off the equilibrated reaction to ringwoodite as a function of temperature and kinetic parameter for varying dominant periods at an epicentral distance of  $120^\circ$ . Bold lines denote changes of amplitude ratios by 0.1, thin lines denote changes of amplitude ratios by 0.05. Data are not available at temperature pf 1850 K and kinetic parameters  $> 4000 \text{ cm}^3 \text{ yr}^{-1}$  (cf. Figure 4).

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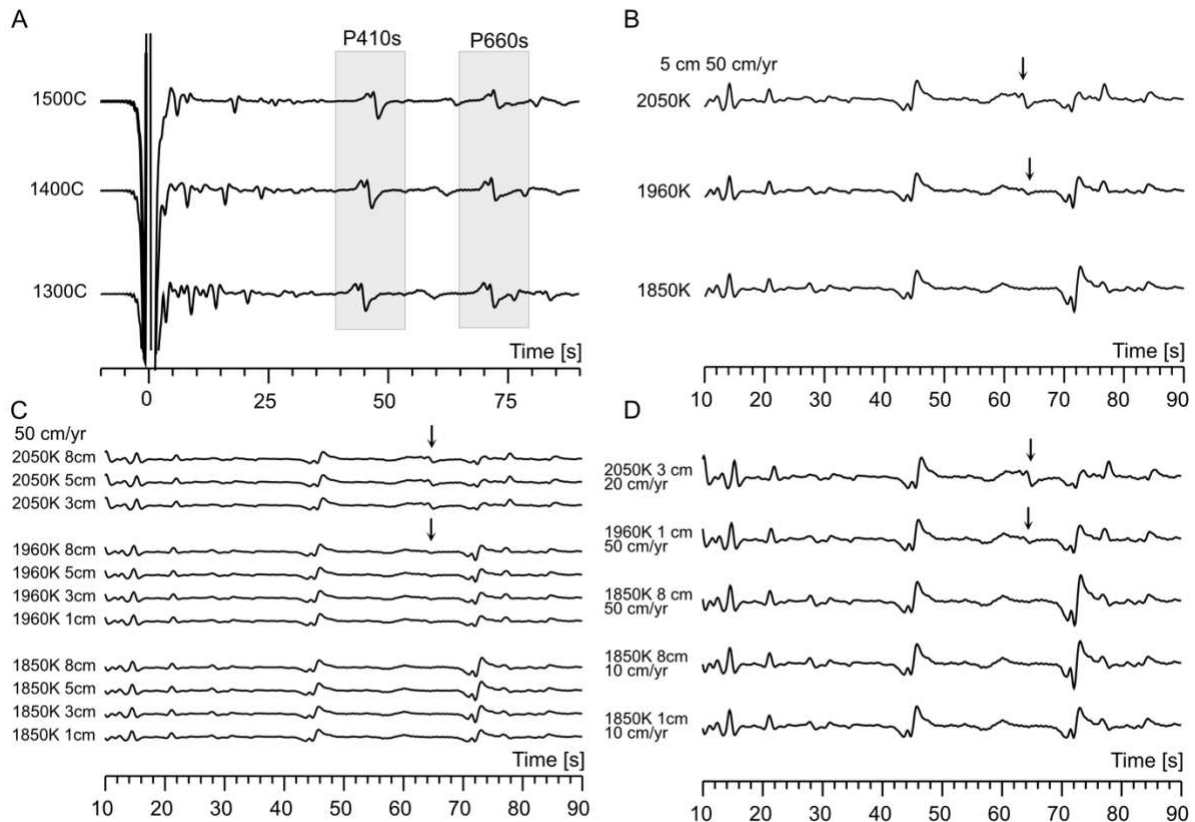
Frequency dependence of kinetic\_model\_1850k\_5cm\_50cm/yr at 120 deg for PP underside reflections



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1054 **Figure 9:** Frequency dependence of P<sup>660</sup>P underside reflections for equilibrated and kinetically  
1055 inhibited phase assemblages at 1850 K, for 5 cm grain size and 50 cm/yr upwelling rate for  
1056 dominant periods of 1 to 10 s. Waveforms of reflections off the kinetically inhibited backward  
1057 reaction to ringwoodite are depicted in red, whereas waveforms of reflections off the  
1058 equilibrated backward reaction to ringwoodite are depicted in black.

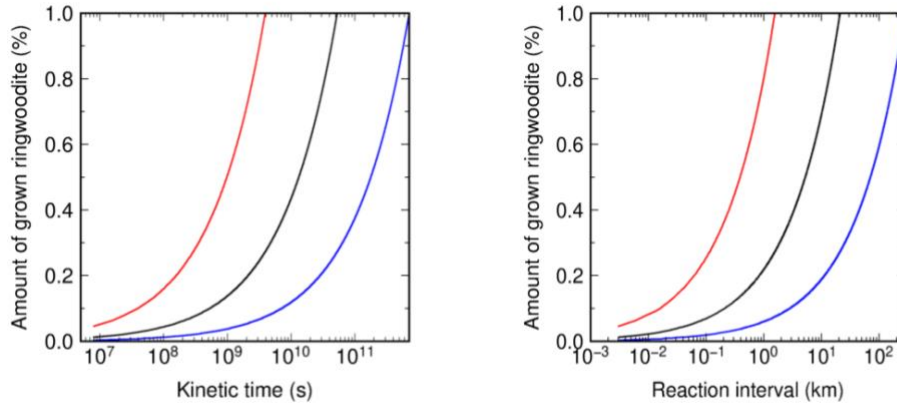
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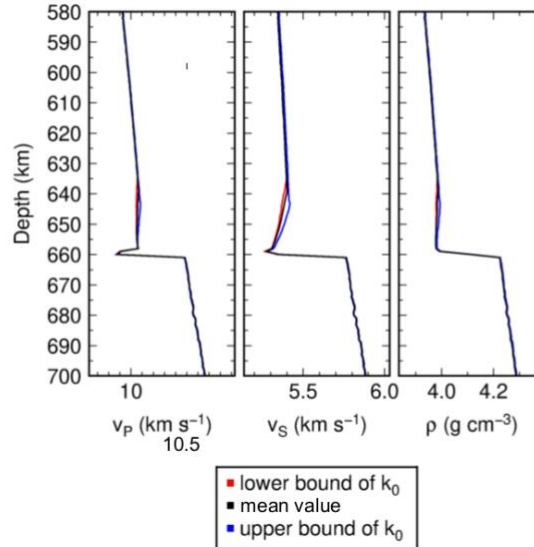
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**Figure 10:** Synthetic waveforms for P-to-S conversions arriving in the P-wave coda with dominant period of 1 s. All traces are recorded at 43 deg epicentral distance and aligned on the theoretical P-wave arrival. All times are given relative to P-wave arrival time. A) Waveforms for pyrolytic models at equilibrium conditions for mantle adiabats with surface potential temperatures ranging from 1300 to 1500 °C. P-to-s conversions from the 410-km discontinuity and the 660 km discontinuity are highlighted. B) Waveforms for kinetically inhibited transformation for models with 5 cm grain sizes and an upwelling rate of 50 cm/yr for geotherms at 1850 K, 1960 K and 2050 K. Amplitudes are normalized on the P-wave amplitude. Arrows indicate detectable precursors to P660s. C) Waveforms for kinetically inhibited transformation for 50 cm/yr upwelling rate and varying temperatures and grain sizes. Amplitudes are normalized on the P-wave amplitude. Arrows indicate detectable precursors to P600s. D) Waveforms for kinetically inhibited transformation for varying temperatures, upwelling rates and grain sizes. Amplitudes have not been normalized and the amplitude variations of the P-to-S conversion from the ferropericlase and bridgmanite to ringwoodite are observable. Arrows indicate detectable precursors to P660s.

a) Kinetic times as a function of uncertainties of  $\Delta H_a$  b) Reaction interval as a function of uncertainties of  $\Delta H_a$



c) Influence of kinetic parameter on seismic properties



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1080 **Figure 11:** Effects of uncertainties of kinetic parameters on seismic properties of kinetically  
 1081 inhibited phase assemblages and thickness of the low-velocity layer above the onset of  
 1082 ringwoodite growth. The effects have been calculated for an ambient mantle temperature of  
 1083 1850 K, an initial grain size of 5 cm and an upwelling rate of 50 cm/yr. a) Effects of uncertainties  
 1084 of the enthalpy  $\Delta H_a$  on the growth time of ringwoodite. The red curve denotes the calculation  
 1085 for the lower bound of  $\Delta H_a$  from Table 3 whereas the blue curve the upper bound of  $\Delta H_a$ . b)  
 1086 Effects of uncertainties of the enthalpy  $\Delta H_a$  on the width of inhibited ringwoodite. c) Effects of  
 1087 uncertainties of the diffusion coefficient  $k_0$  on the seismic wave speeds and density. The lower  
 1088 and upper bound of  $k_0$  correspond to the uncertainties in Table 3.

1089

1090 **Tables:**

1091

1092 **Table 1:** Pyrolite composition by Sun et al. (1982) used in this study. The oxides of the CFMAS  
1093 system are given as molar percentages.

1094

SiO <sub>2</sub>	MgO	Al <sub>2</sub> O <sub>3</sub>	FeO	CaO
38.61	49.13	2.77	6.24	3.25

1095

1096 **Table 2:** Relationship between surface potential temperature of a mantle adiabat and their  
1097 ambient temperatures at ~23.4 GPa, i.e., at ~660 km depth. Surface potential temperature are  
1098 often given in °C (e.g., Cammarano et al., 2005; Cobden et al., 2008) whereas ambient  
1099 temperatures are usually given in K.

1100

Potential temperature (°C)	1300	1400	1500
Ambient temperature (K)	1850	1960	2050

1101

1102 **Table 3:** Parameter range for calculating the seismic properties of kinetically inhibited phase  
1103 assemblages.

1104

Parameter	Value(s)
From the kinetic experiments by Dobson & Mariani (2014):	
Enthalpy of the reaction $\Delta H_a$ (kJ mol <sup>-1</sup> )	456±40
Diffusion coefficient $\ln(k_0)$ (m <sup>2</sup> /s)	-6.36±0.25
For modelling seismic properties:	
Temperatures (K)	1850, 1960, 2050
Grain size (cm)	1, 5, 10, 20
Upwelling rate (cm/yr)	1, 5, 10, 50, 100

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