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1 **Kinetic effects on the 660 km-phase transition in mantle upstreams and seismological**
2 **implications.**

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27

28

29 **Abstract**

30

31 The effects of reaction kinetics of bridgmanite and ferropericlae transforming to ringwoodite
32 on elastic properties in upwelling mantle are investigated using data of kinetic experiments and
33 internally self-consistent thermodynamic modelling of density and seismic velocities. The
34 kinetic experiments show inhibited grain growth of ringwoodite. At the initiation of ringwoodite
35 growth, bridgmanite completely transforms to a metastable pyrope-bearing garnet.
36 Ringwoodite then gradually grows from the metastable assemblage of ferropericlae and garnet.
37 The changes in mineralogy result in a low-velocity zone directly above the 660 km seismic
38 discontinuity due to the lower seismic velocities and densities of ferropericlae and garnet
39 compared to ringwoodite and bridgmanite. The modelling of the effects of reaction kinetics and
40 its effect on seismic structure at ~660 km depth shows more sensitivity to grain size than to
41 temperature and upwelling rate. Modelling 1-D synthetic seismograms of PP (SS) underside
42 reflections off the kinetically inhibited backward reaction to ringwoodite shows advanced travel
43 times of underside reflections off ~660 km depth of 0.2 - 0.8 s (1.2 - 1.6 s) for upwelling rates
44 of 50 cm/yr and initial grain sizes between 5 and 20 cm due to the low-velocity zone above the
45 660 km discontinuity. The finite width of the low-velocity layer results in frequency-dependent
46 behaviour of PP and SS underside reflection amplitudes, with higher amplitudes towards shorter
47 periods. The effect on the travel times of P-to-s conversions used for receiver function is small
48 (< 0.7 s) but the inhibited backward reaction leads to strong amplitude changes and significant
49 waveform variations. The effects of reaction kinetics in mantle upwellings might serve as an
50 additional means to map regions of large-scale upwellings and to constrain grain size in the
51 lower mantle.

52

53 **Key words:** Phase transitions, Composition and structure of the mantle, High-pressure
54 behaviour, Body waves

55

56 **1 Introduction**

57

58 The seismic structure of the Earth's mantle transition zone between upper and lower mantle is
59 dominated by two seismic discontinuities at which the seismic velocities and density increase
60 over short depth intervals. These discontinuities are found in most spherically averaged 1-D
61 Earth reference models, e.g., ak135 (Kennett et al., 1995), and are considered as the seismic
62 expressions of solid-solid phase transitions and mineral reactions (both hereafter called phase
63 transformations) of the olivine system (e.g., Ringwood, 1969; Helffrich, 2000; Shearer, 2000;
64 Weidner & Wang, 2000; Deuss, 2015). Thermodynamic calculations of mantle mineral phase
65 relations are frequently used to interpret observations of seismic velocities and seismic
66 discontinuity structure and help to identify and quantify their thermal and compositional origins
67 (e.g., Cobden et al., 2008; Ritsema et al., 2009; Saki et al., 2019; Vilella et al., 2021; Waszek et
68 al., 2021).

69

70 For thermodynamic modelling of phase equilibria and their seismic properties, the Earth's
71 mantle is usually assumed as vigorously convecting, homogeneous and well mixed. However,
72 geodynamic studies have demonstrated the long-term existence of thermal and compositional
73 heterogeneities throughout the mantle (e.g., Allègre & Turcotte, 1986; Kellogg et al., 2002;
74 Tackley et al., 2005; Ballmer et al., 2017). Small diffusivities of the solid mantle (e.g., Hofmann
75 & Hart, 1978; Farber et al., 1994) lead to the question: to what degree have the mineral
76 assemblages reached thermodynamic equilibrium and mineral transformations been
77 completed? The mechanisms of mineral transformations, i.e., their reaction kinetics, might
78 therefore affect the dynamics and seismic structure of the Earth's mantle (e.g., Kubo et al., 2002;
79 Kubo et al., 2008). Therefore, taking the reaction kinetics into account might be essential for
80 our interpretation of the seismic signals in terms of mantle composition and temperature.

81

82 The influence of reaction kinetics on mineral transformations was first investigated in detail
83 when seismologists looked for mechanisms of deep earthquakes in subducting slabs (e.g., Sung
84 & Burns, 1976; see Kirby, 1996, for a review). Motivated by observations of apparently
85 stagnant slabs at ~660 km depth in seismic tomography models (e.g., van der Hilst, 1991; Fukao
86 et al., 2001; Fukao et al., 2009), Kubo et al. (2002) and Kubo et al. (2008) investigated the
87 reaction kinetics of phase transformations of ringwoodite to ferropericlase and bridgmanite and
88 of garnet to bridgmanite at ~660 km depth and temperature conditions typical for subducting
89 slabs. The kinetic experiments by Kubo et al. (2002) and Kubo et al. (2008) show that low
90 temperatures within the slab decrease the mobility of atoms in the crystal lattice and hence

91 could reduce reaction rates of mineral transformations. The kinetic inhibition of the phase
92 transformations of ringwoodite to ferropericlase and bridgmanite and of garnet to bridgmanite
93 result in neutral to positive buoyancy which may lead to stagnant slabs. Recently it has also
94 been shown that kinetics of the bridgmanite to post-perovskite phase transition can affect
95 visibility of the D" reflector (Langrand et al., 2019).

96
97 Reaction mechanisms of mineral transformations can also change for higher temperatures.
98 Studies by Gasparik (1996a), Gasparik (1996b), Weidner & Wang (1998), Irifune et al. (1998)
99 and Hirose (2002) show that the reaction mechanism of ringwoodite to ferropericlase and
100 bridgmanite changes for mantle temperatures larger than 2100 K. This temperature is
101 considered as a reasonable estimate for the centre of hot mantle upwellings (e.g., Sleep, 1990;
102 Schilling, 1991; Weidner & Wang, 1998). For aluminium contents of approximately 5 wt% and
103 temperatures of at least 2100 K, ringwoodite transforms to bridgmanite and ferropericlase via
104 the dissociation of ringwoodite to ferropericlase and majorite garnet, and finally to
105 ferropericlase and bridgmanite (Gasparik, 1996a; Gasparik, 1996b; Weidner & Wang, 1998;
106 Irifune et al., 1998; Hirose, 2002).

107
108 Kinetic studies by Shimojuku et al. (2014) and Dobson & Mariani (2014) have investigated the
109 reaction kinetics for the reverse reaction of bridgmanite and ferropericlase to ringwoodite in
110 upwelling mantle. The kinetic experiments show that the growth of ringwoodite is kinetically
111 inhibited, and that bridgmanite transforms to pyrope-bearing garnet. The reverse reaction from
112 bridgmanite and ferropericlase to ringwoodite and garnet (in mantle upwellings) requires
113 diffusion on a much larger length-scale than the forward reaction because macroscopic grains
114 of bridgmanite and ferropericlase must recombine to form ringwoodite (Dobson & Mariani,
115 2014). In the case of aluminous garnets, the interface between ringwoodite and garnet develops
116 a fingering instability resulting in a complex intergrowth (Figure 1). Garnet incorporates the
117 aluminium oxide of the bridgmanite into its crystal lattice and has the same bulk chemistry as
118 the bridgmanite before the transformation, and hence this isochemical transformation is fast
119 compared to the diffusion-controlled reaction between bridgmanite and ferropericlase. Over
120 time, this chemically metastable garnet reacts with ferropericlase, and the stable phase,
121 ringwoodite, starts to nucleate and grow at the interface between ferropericlase and garnet.

122
123 The growth of ringwoodite on the ferropericlase grains is most likely controlled by the slow
124 diffusion of SiO₂ from garnet across the reaction layer into the grains of ferropericlase. The
125 ringwoodite growth rate is sensitive to the initial grain size of lower mantle minerals (Dobson

126 & Mariani, 2014). The initial ferropericlasite grain size is the typical length scale of diffusion.
127 The amount of ringwoodite formed depends on bulk composition and ambient temperature. The
128 bulk composition affects the individual molar amounts of the mineral solid solutions. The
129 ambient mantle temperature influences the equilibrium phase relations and mobility of atoms,
130 determining the diffusion rate of atoms across interfaces and grain boundaries (e.g., Putnis,
131 1992). The initial grain size controls the relative proportions of reactants and products for a
132 given amount of diffusion across the reacting interface.

133

134 Our study investigates the effects of reaction kinetics for the recombination of bridgmanite and
135 ferropericlasite to ringwoodite in mantle upwellings on density and seismic wave velocities to
136 assess the possibility to observe the effects of the kinematically inhibited back reaction
137 seismically. The kinetically inhibited growth of ringwoodite is determined for a range of mantle
138 temperatures at 660 km depth, varying grain sizes and mantle upwelling rates. The changed
139 volume fractions of the minerals as a function of pressure are used to calculate profiles of
140 density, P- and S-wave velocities for kinetically inhibited and equilibrated phase assemblages
141 using internally self-consistent thermodynamic modelling. These models are used to compute
142 1-D synthetic seismograms for PP and SS underside reflections off the resulting discontinuity
143 and for P-to-s conversions at the discontinuity (known as P receiver functions), two probes often
144 used to study upper mantle discontinuities, to assess the effects on travel times and waveforms.
145 Our aim is not to carry out a seismic study at this point to search for such effects in observed
146 seismic data. This work presents a previously unstudied effect of mantle phase transitions on
147 seismic data that should be kept in mind in future seismic studies.

148

149 **2 Methods**

150

151 The kinetically inhibited growth of ringwoodite and its effects on phase assemblages and
152 seismic properties are computed in five steps:

- 153 1. calculation of phase equilibria,
- 154 2. calculation of the thickness of the ringwoodite layer produced for a given temperature,
155 grain size and mantle upwelling rate,
- 156 3. mass balancing consumption of metastable ferropericlasite and garnet,
- 157 4. calculation of density and seismic wave velocities.
- 158 5. computation of 1-D synthetic seismograms

159

160 **2.1 Calculation of phase equilibria**

161

162 Phase equilibria for the CFMAS (calcium oxide, iron(II) oxide, magnesium oxide, aluminium
163 oxide and silica) system are calculated for a pyrolitic mantle composition (Table 1) with the
164 Perple_X program package (Connolly, 2005) that utilises the free-energy minimization
165 algorithm by Ita & Stixrude (1992) and Stixrude & Lithgow-Bertelloni (2005), and the elastic
166 and thermodynamic properties and the solid solution model of Stixrude & Lithgow-Bertelloni
167 (2011). From the calculated phase equilibria, profiles of P-wave velocity, S-wave velocity and
168 density are extracted along adiabatic mantle geotherms (e.g., Cobden et al., 2008). The values
169 of P-wave velocity, S-wave velocity, and density of the phase assemblage at a specific P-T-
170 point are calculated as the Hill average of Voigt and Reuss bounds of the constituent minerals
171 (Voigt, 1928; Reuss, 1929; Hill, 1965).

172

173 **2.2 Kinetically inhibited growth of ringwoodite and mineral reactions**

174

175 The seismic structures of the kinetically inhibited phase assemblages are calculated from the
176 grain growth of ringwoodite for fixed temperature, grain size and upwelling rate (Figure 2).
177 Before we describe the calculations of kinetically inhibited growth of ringwoodite and its
178 impact on seismic properties, we compare the mineral reactions at thermodynamic equilibrium
179 with those observed by Dobson & Mariani (2014).

180

181 The recombination reaction at thermodynamic equilibrium is described as

182



ferropericlase + bridgmanite \leftrightarrow ringwoodite

183

184 Dobson & Mariani (2014) observe in the kinetic experiments that bridgmanite entirely
185 transforms to garnet, where the aluminium-bearing bridgmanite transform to pyrope garnet:

186



aluminous bridgmanite \leftrightarrow majorite garnet + pyrope garnet

187

188 The consumption of bridgmanite to ringwoodite comprises of several reactions. At the onset of
189 ringwoodite growth, the entire amount of bridgmanite transforms to garnet,

190



bridgmanite \rightarrow majorite garnet

191

192 creating an excess of garnet compared to conditions at thermodynamic equilibrium.

193 The difference between the decrease of ferropericlasite with kinetic inhibition effects and the

194 phase fraction of ferropericlasite is later used to infer the decrease of garnet which recombines

195 with ferropericlasite to give ringwoodite via the reaction

196



ferropericlasite + majorite garnet \leftrightarrow ringwoodite

197

198 The kinetically inhibited growth of ringwoodite and the dependent mineral transformations of

199 bridgmanite and garnet are assumed to leave calcium-bearing garnets and the exsolution of

200 calcium perovskite unaffected. Therefore, amounts of calcium-bearing garnets and calcium

201 perovskite are extracted from the thermodynamic calculations.

202

203 We now derive how we calculate the growth kinetics of ringwoodite and account for the mineral

204 reactions mentioned above. The growth kinetics of ringwoodite are controlled by two

205 parameters which are determined in experiments: its enthalpy ΔH_a and its reaction rate constant

206 k_0 which is closely linked to the reaction entropy ΔS_a .

207

208 For a mineral transformation from mineral assemblages A and B to a mineral assemblage C,

209 the reaction rate is proportional to the molar concentrations of mineral phases A and B, C_A and

210 C_B (Putnis, 1992):

211

$$\text{Reaction rate} \propto K^* C_A C_B \quad (5)$$

212

213 In our case, C_A and C_B are the concentration of ferropericlasite and majorite garnet.

214 The equilibrium constant of the reaction K^* is a function of the change of reaction Gibbs free

215 energy, ΔG_a ,

216

$$K^* = \exp(-\Delta G_a / (R T)) = \exp(\Delta S_a / R) \exp(-\Delta H_a / (R T)) \quad (6)$$

217

218 using the definition of the reaction Gibbs free energy $\Delta G_a = \Delta H_a - T\Delta S_a$ with the enthalpy of
 219 the reaction ΔH_a and the entropy of the reaction ΔS_a . R denotes the gas constant and T is
 220 temperature. The reaction rate can be rewritten as

$$221 \text{ Reaction rate} = c \exp(\Delta S_a/R) \exp(-\Delta H_a/(R T)) C_A C_B \quad (7)$$

222
 223 where c is a constant. As the concentration of the reactants changes during the reaction, the
 224 reaction rate will change accordingly. The reaction rate thus becomes a function of the phase
 225 fraction C

$$226 \text{ Reaction rate} = k f(C) \quad (8)$$

227
 228 where the reaction rate parameter k describes how the reaction rate depends on the concentration
 229 of the reactants. Therefore, the rate constant is

$$230 k = k(T) = c \exp(\Delta S_a/R) \exp(-\Delta H_a/(R T)) = k_0 \exp(-\Delta H_a/(R T)) \quad (9)$$

231
 232 with the reaction rate constant $k_0 = c \exp(\Delta S_a/R)$ where c is an experimentally determined factor.

233
 234 Dobson & Mariani (2014) show that the growth kinetics of ringwoodite are controlled by
 235 diffusion kinetics. In this case, the rate of growth of the interface is linear with the square-root
 236 of time and the thickness of reaction rims measured at each temperature can be converted into
 237 apparent rate constants (Watson & Price, 2002). Following Tammann (1920) and Watson &
 238 Price (2002), the temperature dependent rate constant $k(T)$ is given by

$$239 k(T) = \frac{1}{2} \frac{d}{dt} x^2 \quad (10)$$

240
 241 where x is the thickness of the growing interface and t is time. Integration of (10) gives the
 242 interface thickness evolution as a function of time and temperature:

$$243 x(t, T) = ((2 k(T) t)^{1/2}). \quad (11)$$

244
 245 To model the kinetic inhibition of ringwoodite growth, the interface grows until reaching the
 246 assumed initial grain size d . In this study, radially isotropic growth of the ringwoodite interface

247 is assumed. We therefore normalise the ringwoodite interface thickness evolution (11) with half
248 of the desired grain size (d) which results in a relative grain size: :

249

$$x_{\text{rel}}(t, T) = ((2 k(T) t)^{1/2}) / (0.5 d). \quad (12)$$

250

251 Since temperatures are approximately constant along the chosen adiabats, we assume a constant
252 temperature T_0 during the reaction:

253

$$x_{\text{rel}}(t) = ((2 k(T_0) t)^{1/2}) / (0.5 d). \quad (13)$$

254

255 Due to its inverse dependency on grain size d , eq. (13) indicates that growth of ringwoodite
256 takes more time in case of large grain size.

257

258 Assuming an upwelling rate v and utilizing the depth-to-pressure scale used by Cobden et al.
259 (2008), the time of grain growth is converted to a grain growth curve as a function of pressure,
260 $x_{\text{rel}}(P)$. In our calculations, upwelling rates in the mantle are estimated to range between 1 and
261 100 cm/yr considering current heat flux estimates in upwelling mantle (e.g., Sleep, 1990;
262 Courtillot et al., 2003) and geodynamic modelling experiments (Farnetani & Hofmann, 2010).
263 The average grain size in the lower mantle is estimated to range between 1 μm (Yamazaki et
264 al., 1996) and 10 cm (Solomatov, 1996; Korenaga, 2005) based on varying estimates of grain
265 growth kinetics of mineral transformations in downwelling, or primordial, mantle. In the
266 interior of a subducting slab, Yamazaki et al. (1996) and Yamazaki et al. (2005) find very slow
267 grain growth rates for ferropericlase and bridgmanite, with grain sizes ranging between 10 μm
268 to 100 μm for temperatures typical for subduction zones and for 10 million years after grain
269 nucleation, but Fei et al. (2021) find a much smaller grain size exponent compared to Yamazaki
270 et al. (1996). Their estimate for grain size in subducted slabs is on the order of 30-45 μm .
271 Solomatov et al. (2002) question estimates of grain size for the bulk mantle as they cannot be
272 reconciled with viscosity models or laboratory data for other materials. Solomatov et al. (2002)
273 argue that the experiments have not been run sufficiently long to evaluate grain growth kinetics
274 beyond transient effects. Solomatov (1996) and Korenaga (2005) have additionally suggested
275 that deep-mantle plumes might have a significantly larger grain size than the ambient lower
276 mantle due to (i) the temperature of the source region of the mantle upwelling, (ii) the kinetics
277 of grain growth from an early magma ocean, and (iii) seismic tomographic imaging of wide
278 slow structures which have been interpreted as wide plume roots (French & Romanowicz,
279 2015).

280

281 **2.3 Mass balancing consumption of metastable ferropericlaase and garnet**

282

283 We follow the workflow in Figure 2 to calculate volume fractions of the mineral phases affected
284 by the kinetically inhibited growth of ringwoodite for given grain size d , upwelling rate v and
285 constant temperature T_0 . Ringwoodite starts to grow at pressure P_{Start} where the molar amount
286 of ringwoodite is non-zero in the equilibrium phase assemblage. The kinetically inhibited
287 growth of ringwoodite is calculated as the convolution of the grain growth curve and the
288 derivative of the molar amounts of ringwoodite at thermodynamic equilibrium as a function of
289 depth. The growth curve x_{rel} is truncated where the relative growth of ringwoodite equals 1. At
290 this pressure, hereafter called P_{end} , the molar amount of ringwoodite with kinetically inhibited
291 growth conditions is considered to be equal to the amount of ringwoodite at thermodynamic
292 equilibrium. Using the molar amount of ringwoodite at P_{end} , $n_{\text{Rw,end}}$, we convert the grain growth
293 curve (12) into molar amount of ringwoodite:

294

$$n_{\text{Rw}}(P) = n_{\text{Rw,end}} x_{\text{rel}}(P) \quad (13)$$

295

296 We calculate the mass balances of ferropericlaase, majorite garnet and pyrope garnet, following
297 the mineral reactions in (2) – (4). Using molar volumes of the individual mineral phases from
298 our thermodynamic calculations in section 2.1, we convert molar amounts to volume fractions
299 within the pressure interval from P_{end} to P_{start} .

300

301 **2.4 Calculation of density and seismic wave velocities**

302

303 Utilizing their volume fractions calculated as described in section 2.3, we calculate the seismic
304 properties of the kinetically inhibited phase assemblages, for the Hill average of Voigt and
305 Reuss bounds of the constituent minerals (Voigt, 1928; Reuss, 1929; Hill, 1965). For each
306 mineral, the density and seismic velocity are determined at the given pressure and temperature.
307 The seismic properties of the kinetically inhibited phase assemblages are fit to the seismic
308 profiles of phase assemblages at thermodynamic equilibrium for the corresponding adiabatic
309 mantle geotherm.

310

311 **2.5 Computation of 1-D synthetic seismograms**

312

313 The mineral physical results can be used to test if the effects of the kinetically inhibited

314 transformation of bridgmanite and ferropericlaase to ringwoodite can be observed in
315 seismological data. We focus on effects of PP and SS underside reflections and P-to-s
316 conversions as they are commonly used probes to study upper mantle discontinuities. For this,
317 the computed profiles of density, P wave and S wave velocity for equilibrated and kinetically
318 inhibited phase assemblages are used to calculate 1-D reflectivity synthetic seismograms (Fuchs
319 & Müller, 1971; Müller, 1985) for a range of dominant periods. Attenuation values are taken
320 from PREM (Dziewonski & Anderson, 1981). For comparison, 1-D reflectivity synthetic
321 seismograms using the 1-D velocity model ak135 (Kennett et al., 1995) and attenuation of
322 PREM (Dziewonski & Anderson, 1981) are computed. Dominant periods of PP underside
323 reflections are 1, 3, 5, 8 and 10 s. A source at 50 km depth is chosen and the data are analysed
324 on the vertical component. For SS underside reflections, dominant periods are 5, 8, 10, 15 and
325 20 s. The source is at 50 km depth and the data are analysed on the transverse component. 21
326 receivers are distributed as a linear array around 120° epicentral distance with an aperture of
327 5° . An epicentral distance of 120° is within typical distances used for PP and SS precursor
328 studies (e.g., Shearer, 1991; Thomas & Billen, 2009; Deuss, 2009). The seismograms are
329 aligned on the PP/SS arrival and are normalised to unit amplitude.

330

331 For the P-to-s conversions we use dominant periods of 0.5 s, 1 s and 2 s with a source depth of
332 2 km to avoid interference of the P660s (and P410s for reference) arrivals with depth phases.
333 We use a linear array with 1° receiver spacing between 40° and 80° . We calculate three-
334 component seismograms (Z, R, T) but only analyse the radial component. Due to the simple
335 source time function and the consistent waveforms, we abstain from deconvolving the vertical
336 component P-wave and interpret the relevant waveform variation in the Results section.

337

338 **3 Results**

339

340 **3.1 Effects of kinetic inhibition on seismic fine structure**

341

342 The effects of kinetic inhibition on phase assemblages are compared with phase assemblages at
343 thermodynamic equilibrium with regard to the mineral proportions and seismic properties. The
344 kinetically inhibited growth of ringwoodite and the metastable assemblage of ferropericlaase and
345 pyrope-bearing garnet change the phase proportions above the onset depth of the phase
346 transformation (Figure 3). For equilibrated phase assemblages at a temperature of ~ 1850 K
347 (following the adiabat with 1300°C potential surface temperature, abbreviated hereafter as
348 “ 1300°C adiabat”; please see Table 2 and Cobden et al., 2008 for details), ringwoodite

349 dissociates to ferropericlasite and bridgmanite within a pressure interval of ~0.1-0.2 GPa (i.e.,
350 ~2 km) at ~23.4 GPa (i.e., at ~659 km depth) (Figure 3a). The amount of garnet decreases due
351 to the exsolution of bridgmanite, starting at ~22.7 GPa (~642 km).

352

353 The thickness of inhibited growth of ringwoodite is sensitive to temperature, grain size and
354 mantle upwelling rate. For 5 cm grain size and 50 cm/yr upwelling rate, the kinetic inhibition
355 leads to a growth of ringwoodite over a pressure interval of 1 GPa (~25 km) (Figure 3b). The
356 gradual growth of ringwoodite results in a gradual consumption of ferropericlasite. At the
357 pressure where ringwoodite starts to grow, the entire amount of bridgmanite reacts to garnet,
358 thus its volume fraction increases to 70%. The recombination reaction of garnet and
359 ferropericlasite to ringwoodite gradually decreases the amount of garnet towards lower pressures.

360

361 The excess of garnet changes the seismic structure around 660 km depth (Figure 3c). Since
362 garnet has lower P- and S-wave velocities (garnet: $v_p \approx 9.8$ km/s, $v_s \approx 5.2$ km/s at 23.4 GPa and
363 1850 K) than ringwoodite and bridgmanite (ringwoodite: $v_p \approx 10.2$ km/s, $v_s \approx 5.6$ km/s;
364 bridgmanite: $v_p \approx 11.2$ km/s, $v_s \approx 6.3$ km/s at 23.4 GPa and 1850 K), the kinetically inhibited
365 growth of ringwoodite and the abundance of garnet results in a low-velocity layer above the
366 660 km discontinuity of $\Delta v_p \approx -5\%$, $\Delta v_s \approx -7\%$ (Figure 45). The low-velocity layer above the
367 lower boundary of the ringwoodite-forming reaction thus increases the P- and S-wave
368 impedance contrast. When the growth of ringwoodite has reached equilibrium conditions, the
369 profiles of density and seismic wave speeds intersect the profiles of the phase assemblages at
370 thermodynamic equilibrium.

371

372 The effects of initial grain size d and mantle upwelling rate v can be described by a kinetic
373 parameter D which is proportional to the upwelling rate and the squared grain size (Dobson &
374 Mariani, 2014):

375

$$D \sim v d^2.$$

377

378 Scenarios of kinetic inhibition with different initial grain sizes and upwelling rates give the
379 same results if they can be described by the same kinetic parameter. Therefore, the kinetic
380 parameter is an appropriate entity to compare scenarios of kinetically inhibited phase
381 assemblages. Since the kinetic parameter scales with the square of initial grain size d , the kinetic
382 inhibition of ringwoodite is more sensitive to grain size than to upwelling rate. In this study, we

383 set the proportional factor to unity and use

384

$$385 \quad D = v d^2.$$

386

387 in order to compare scenarios of initial grain size d and mantle upwelling rate v .

388

389 The thickness of the low-velocity layer, i.e., the depth interval between the onset and the
390 completion of ringwoodite growth, increases exponentially with increasing kinetic parameter
391 of the kinetic parameter (Figure 5). The exponential behaviour can be explained by the grain
392 growth equation which includes the Arrhenius relation for the reaction rate parameter $k(T)$ (cf.
393 Eq. (9)). For kinetic parameters D between 10 and 100 cm³/yr, kinetic inhibition occurs over a
394 pressure interval of ~0.1 GPa which is as narrow as the transformation pressure interval of the
395 recombination of ringwoodite from ferropericlase and bridgmanite. A transformation pressure
396 interval of ~0.1 GPa is equivalent to a low-velocity layer of thickness ~2 km which is likely
397 below the resolution threshold even of short period PP waves and at the resolution level for
398 receiver function (Ps). Seismic waves therefore sample the reflector and low-velocity layer
399 above in the same way as the reflector for phase assemblages at thermodynamic equilibrium.
400 The onset of seismically detectable kinetic inhibition in terms of the kinetic parameter D is D
401 > 100 cm³/yr for 1850 K, $D > 1000$ cm³/yr for 1960 K and $D > 2500$ cm³/yr for 2050 K. The
402 temperature dependence can be explained by the increased diffusion of ions at higher
403 temperatures.

404

405 **3.2 Synthetic seismograms**

406

407 We have calculated 1-D synthetic seismograms from the synthetic velocity and density profiles
408 using the reflectivity methods (Fuchs & Müller, 1971; Müller, 1985) as described in section
409 2.4. Here we discuss the effects of the kinematically inhibited velocity structures on PP/SS
410 underside reflections and P-to-s conversions, two probes regularly used for the study of upper
411 mantle discontinuities.

412

413 **3.2.1 PP and SS Precursors**

414

415 The seismograms calculated for the velocity models with kinetically inhibited recombination
416 reaction of ringwoodite from bridgmanite and ferropericlase are compared to seismograms for
417 velocity models with equilibrated phase assemblages. Using linear vespagrams (Davies et al.,

418 1971; Muirhead et al., 1976), PP/SS and P660P/S660S waveforms are extracted for travel times
419 predicted by ak135 (Kennett et al., 1995) (Figure 6).

420

421 For velocity models where the kinetic inhibition of the ringwoodite growth results in a thicker
422 low-velocity layer of the order of ~ 25 km, the PP/SS precursors off the reflector caused by
423 kinetically inhibited recombination of ringwoodite are noticeably advanced for reaction
424 intervals wider than 25 km compared to PP/SS precursors reflected off discontinuities of
425 equilibrated phase transformations (Figure 3b). The travel time advances also affect secondary
426 phases of the underside reflections such as depth phases, e.g., pP660P (Figure 6). For the kinetic
427 parameters tested here, the maximum travel time advances for PP underside reflections are ~ 0.8
428 s, whereas maximum travel time advances for SS underside reflections are ~ 1.3 s (Figure 7).
429 These travel time residuals correspond to apparent maximum shifts of the discontinuity depth
430 on the order of 5 km.

431

432 Amplitudes for PP precursors with a dominant period of 1 s reflected off the kinetically
433 inhibited formation of ringwoodite vary between 120% and 160% relative to the equilibrium
434 model at 1850 K. For the same temperature, amplitudes of SS precursors with a dominant period
435 of 5 s range between 120% and 160% relative to the equilibrium model. The precursor
436 amplitudes for the kinetically inhibited phase assemblages decrease with increasing periods
437 (Figure 8). The transition from enhanced to apparently unaltered precursor amplitudes is at a
438 dominant period of 5 s for PP and 10 s for SS. Moreover, the PP and SS precursor amplitudes
439 increase with increasing kinetic parameters and increasing thickness of the low-velocity layer.
440 The PP/SS precursors off the lower boundary of the phase transformation are less affected by
441 interference with PP/SS precursors off the gradient above the low-velocity layer.

442

443 While it appears that underside reflections of PP waves at the ringwoodite to bridgmanite plus
444 ferropericlase transition are more difficult to detect than the corresponding SS underside
445 reflection (Estabrook & Kind, 1996), there is evidence for P660P waves in some regional
446 studies (e.g., Deuss et al., 2006; Thomas & Billen, 2009; Schmerr & Thomas, 2011, Saki et al.,
447 2014) and usually the amplitudes of observed P660P waves are smaller than predicted by PREM
448 or a pyrolytic mantle. There is also a noted discrepancy between the detection of P660P and its
449 equivalent underside reflection of PKPPKP (P'660P') (Day & Deuss, 2013). The scarcity of
450 P660P detections in recorded data can likely be explained by compositional and thermal effects
451 that are not captured in our modelling and the assumed pyrolytic starting model (Lessing et al.,
452 2015; Guo & Zhou et al., 2020; Waszek et al., 2021).

453 Underside reflection of PP from the 660 km phase transition are rarely detected in recorded data
454 although such detections exist (Deuss et al., 2006; Day & Deuss, 2013).

455

456 The finite width of the low-velocity layer (Figure 3a) and the vertical resolution of the seismic
457 waves lead to a frequency-dependent behaviour of P660P/S660S precursor amplitudes (Figure
458 9). PP waves with 1 s dominant period appear to resolve the increased P wave impedance
459 contrast off the reflector caused by the kinetically inhibited recombination of ringwoodite. The
460 increased P wave impedance contrast results in increased PP precursor amplitudes compared to
461 the reflections for equilibrated phase assemblages. If the dominant period is increased, PP
462 precursor amplitudes off the reflector with kinetic inhibition decrease and have similar PP
463 precursor amplitudes off the reflector as those without kinetic inhibition.

464

465 **3.2.2 P-to-s conversions (P receiver functions)**

466

467 Examples of the synthetic waveforms for the kinetically inhibited models and pyrolitic models
468 at equilibrium conditions are shown in Figure 10. For varying temperatures at equilibrium, we
469 observe strong conversions from the 660 km discontinuity and other discontinuities (e.g., 410-
470 km as well as shallower) with minor waveform and amplitude variations. We calculated
471 synthetics for dominant periods of 0.5 s, 1 s, and 2 s and observe very minor variability of the
472 results. While absolute travel times for P- and P-coda arrivals vary, differential travel time
473 variations are minor, with 0.2 s and 0.6 s for the 1960 K and 2050 K simulations, respectively,
474 compared to the 1850 K simulations. Waveform variations are limited to larger moveout for a
475 deeper transition, arriving 5.1 to 8.9 s after P660s.

476

477 For the kinetically inhibited transformations we observe strong waveform variations with
478 temperature, where the waveforms of the high temperature models show a dominant precursor
479 about 7 s preceding the arrival from P660s (Figure 10b and c). The precursor due to the low
480 velocity zone will lead to a likely detectable negative polarity precursor in P-wave receiver
481 functions (see Figure S1 for an additional velocity profile at 2050 K). Those detectable negative
482 polarity precursors have been reported in some up-flowing regions of the mantle (e.g., Negi et
483 al., 2022). We observe amplitude variations dependent on grain size, upwelling rate, and
484 temperature (Figure 10d). Amplitudes vary between 52% and 122% relative to the equilibrium
485 model at 1960 K with the smallest amplitudes observed for the highest temperatures.
486 Interference effects lead to apparent inverted waveforms for the P-to-s conversion relative to P.
487 Both amplitude and waveform variations for these models would be large enough to be

488 observed in receiver function analyses.

489

490 Travel times relative to P vary slightly for the different models. We observe travel time
491 variations of ± 0.3 s relative to the equilibrium model at 1960 K. While detectable, these might
492 be difficult to detect for recorded data due to the 3-D velocity variations known to exist in upper
493 mantle.

494

495 **4 Discussion**

496

497 **4.1 Estimates of uncertainties**

498

499 The effects of initial grain size and upwelling rate on the seismic properties of kinetically
500 inhibited phase assemblages have been investigated in section 3.1. Among the kinetic input
501 parameters, i.e., kinetic rate parameter k_0 , reaction enthalpy ΔH_a and temperature T),
502 uncertainties of the reaction enthalpy predominantly affect the seismic properties of kinetically
503 inhibited phase assemblages (Figure 11a and b). Dobson & Mariani (2014) measure the
504 enthalpy of the reaction with an uncertainty of ± 40 kJ mol⁻¹ for the ringwoodite layer (Table 1).
505 Varying the enthalpy of the reaction with this value while keeping remaining parameters
506 constant in eq. (9) results in variations of the growth time of ringwoodite until equilibrium
507 conditions by one order of magnitude (Figure 11b).

508

509 Converting growth time into the thickness of the reaction interval using the upwelling rate gives
510 uncertainties of the reaction interval of one order of magnitude. Given the uncertainty of the
511 enthalpy of the reaction for the ringwoodite layer, the reaction interval could therefore range
512 from 1 km to ~ 250 km for kinetic parameters ranging from 1 to 10.000 cm³/yr. On the one hand,
513 reaction intervals with a width of 1-10 km are close to the vertical resolution of PP and SS
514 precursors (e.g., Thomas & Billen, 2009; Schmerr & Thomas, 2011) and it might be challenging
515 to discriminate kinetic effects from other influences on reaction intervals. On the other hand,
516 reaction intervals with widths of 100-200 km would give increased PP and SS precursor
517 amplitudes due to the increased P wave and S wave impedance contrast and due to the more
518 gradual change of seismic properties from the low-velocity region relative to seismic properties
519 at thermodynamic equilibrium. P-to-s conversions at the discontinuities beneath the receiver
520 typically show higher resolution and are likely able to detect even the thinner layers inferred
521 for the smaller kinetic parameters. Thicker reaction intervals would result in strongly reduced
522 amplitudes of Ps. Reaction intervals of 100-200 km would also affect phase transformations at

523 smaller pressures/shallower depths, e.g., the wadsleyite to ringwoodite transition, as well as the
524 phase transformations from pyroxene to garnet and from garnet to calcium silicate perovskite.
525 The kinetics of the olivine to wadsleyite phase transition (Poirier, 1982; Brearley et al., 1992;
526 Kirby et al., 1996) and of pyroxene to garnet (Hogrefe et al., 1994) have been studied for
527 harzburgitic mantle composition so far. However, kinetic data of other phase transformations in
528 the transition zone are necessary to allow for consistent treatment of kinetic effects on seismic
529 properties within and above the transition zone.

530

531 The reaction rate constant k_0 has a minor effect on the seismic properties of the kinetically
532 inhibited phase assemblage (Figure 11c). P- and S-wave velocity vary by ± 0.05 km/s, and
533 density is changed by ± 0.02 g cm⁻³ for varying k_0 to its lower and upper bounds. The thickness
534 of the low-velocity layer above the 660 km discontinuity shifts by ± 4 km.

535

536 **4.2 Seismological and geodynamic implications**

537

538 Patterns of mantle flow and dynamics have been investigated using seismic wave speed
539 anomalies in seismic tomography models (e.g., Li et al., 2008; Ritsema et al., 2011) and seismic
540 waves sampling the upper mantle and mantle transition zone discontinuities using receiver
541 functions (e.g., Vinnik, 1977; Farra & Vinnik, 2000; Rondenay, 2009), short period P'P'
542 reflections (e.g., Niazi & Anderson, 1965; Vidale & Benz, 1993; Day & Deuss, 2013), and PP
543 and SS precursors (e.g., Shearer, 1991; Deuss, 2009). Since the upper mantle and mantle
544 transition zone discontinuities are sensitive to thermal variations due to their Clapeyron slopes
545 (Bina & Helffrich, 1994; Helffrich, 2000; Weidner & Wang, 2000), variations of discontinuity
546 depth and of transition zone thickness (i.e., the depth between the 410 and 660 km discontinuity)
547 and constraints from mineral physics, are frequently used to infer thermal variations in the
548 mantle (e.g., Shearer, 2000, Ritsema et al. 2009).

549

550 The investigation of reaction kinetics of the reaction from ferropericlasite and bridgmanite to
551 ringwoodite can add further constraints on mantle temperature and flow patterns. The effect of
552 kinetic inhibition can be compared to seismic data using a two-step approach. In a first step,
553 through forward modelling of synthetic seismograms for a range of temperatures, upwelling
554 rates, and grain sizes. Relevant temperatures can be determined from adiabatic mantle
555 geotherms (Brown & Shankland, 1981; Spiliopoulos, 1984; Ita & Stixrude, 1992; Cobden et
556 al., 2008) to estimate excess temperatures (e.g., Sleep, 1990; Schilling, 1991; Herzberg et al.,
557 2007). Heat flux estimates (e.g., Sleep, 1990; Courtillot et al., 2003) can be converted into

558 estimates for upwelling rates using convection models for vertical upwellings. A steady-state
559 axisymmetric plume with temperature-dependent viscosity might serve as a starting point for
560 estimates of upwelling rates as a function of heat flux and excess temperatures (Loper & Stacey,
561 1983; Schubert et al., 2001). In a second step, the seismic recordings can be compared to the
562 set of synthetic seismograms and those models are taken which give the best match with regard
563 to frequency dependent travel time residuals and amplitude ratios. The set of matching models
564 gives the range of corresponding grain sizes and upwelling rates.

565

566 Considering the effects of reaction kinetics of ferropericlase and bridgmanite to ringwoodite on
567 the amplitudes and travel times may offer additional constraints on mapping mantle flow using
568 P-to-s receiver functions. P-to-s conversions typically have higher frequencies (Li et al., 2000;
569 Wölbern et al., 2006). Several studies (van der Meijde et al., 2003; Schmandt, 2012; Huckfeldt
570 et al., 2013) show frequency dependent behaviour of P-to-s conversions at mantle transition
571 zone discontinuities. We see little frequency dependence of P-to-s conversion in our synthetic
572 models within the range typically used for receiver functions, but observe strong waveform
573 variations that are likely to be observable in recorded data as well as strong amplitude variations
574 and small travel time variations that might add additional constraints for detecting regions with
575 mantle upstreams.

576

577 The analysis of PP and SS precursors might have some resolution limitations with respect to
578 the scale of the mantle upwellings which might hamper the detection of the predicted
579 waveforms. Geodynamic modelling indicates that deep thermal mantle plumes have typical
580 diameters of ~ 200 km (e.g., Sleep, 1990; Schubert et al., 2001; Steinberger & Antretter, 2006)
581 whereas thermochemical plumes might have larger diameters ranging from 600 to 800 km
582 (Farnetani, 2005; Farnetani & Hofmann, 2010; Stockmann et al., 2019). Lessing et al. (2015)
583 use the 2.5-D axisymmetric finite difference technique PSVaxi (Jahnke, 2008; Thorne et al.,
584 2013) to examine the effects of topography of the upper mantle discontinuities imaged by PP
585 underside reflections with dominant periods of 2 s. Their models show that the PP wavefield
586 reflected off a locally elevated 660 km with a diameter of 200 to 400 km does not differ
587 significantly from the PP wavefield reflected off a flat reflector at 660 km depth. The size of
588 the detailed topography of the perturbed reflector is smaller than the PP Fresnel zone at 660 km
589 depth and thus might not be resolved. The detection of effects of the kinetically inhibited
590 reaction from ferropericlase and bridgmanite to ringwoodite might therefore be challenging in
591 narrow mantle upwellings. Although conduction might lead to thermal halos around narrow
592 mantle plumes the effect might be small (Sleep, 2004) and might not influence the detectability

593 of the plume using underside reflection. Nonetheless, recent tomographic images show that
594 plumes might be broad upstreams (French and Romanowicz, 2015), very different to early
595 numerical and laboratory plume experiments which might make the upstream detectable
596 seismically. The pooling of plume material beneath the 660 km discontinuity due to the
597 endothermic phase transition might lead to larger scale temperature anomalies in the region of
598 plume-like mantle upstreams (Tackley et al., 1994, Schubert et al., 1995, Stockmann et al.,
599 2019). Possible candidates for upwellings with large, detectable lateral dimensions are large
600 transient domes near the mantle transition zone (Courtillot et al., 2003, Davaille et al., 2005,
601 Kumagai et al., 2007; Kumagai et al., 2008). The transient domes are characterised by
602 upwelling rates slower than deep mantle plumes due to their small buoyancy ratio (ratio of
603 chemically driven buoyancy to thermally driven buoyancy). Nonetheless, the transient domes
604 could have excess temperatures up to 300 K in their centre and form a thermal boundary layer
605 with the ambient mantle (Davaille et al., 2005). Upwelling mantle material might pond beneath
606 the 660 km discontinuity when encountering the phase transformation from bridgmanite and
607 ferropericlase to ringwoodite (e.g., Schubert & Turcotte, 1975; Schubert & Tackley, 1995;
608 Tackley, 1995; Steinbach & Yuen, 1997; Brunet & Yuen, 2000; Davaille et al., 2005, Farnetani
609 & Samuel, 2005). Due to the negative Clapeyron slope, the transformation pressure of the
610 warmer material is shifted towards lower pressure, resulting in a negative buoyancy force which
611 delays further upwelling. The material starts to pond beneath the phase transformation (Davaille
612 et al., 2005) and spreads laterally, growing with dimensions broader than plume conduits (e.g.,
613 Steinbach & Yuen, 1997; Davaille et al., 2005, Stockmann et al., 2019).

614

615 Furthermore, the viscosity contrast across the 660 km discontinuity (e.g., Mitrovica & Forte,
616 2004) might support lateral mantle flow beneath the discontinuity. The geodynamic models of
617 Tosi & Yuen (2013) show additional evidence for horizontal flow whose scale is governed by
618 the viscosity contrast due to temperature variations. For small values of the viscosity contrast,
619 broad and highly viscous plumes are generated that tend to pass through the transition zone
620 relatively unperturbed. For higher values of the viscosity contrast ($10^2 \leq \Delta\eta_T \leq 10^3$), the
621 geodynamic simulations show horizontal channel flows beneath the 660 km discontinuity
622 which might extend up to 1500 km horizontally before tilting to more vertical flow. Therefore,
623 lateral mantle flow could change the fine-scale seismic structure at ~660 km depth in a way that
624 could allow for the detection of the effects of kinetically inhibited growth with seismic methods
625 sensitive to upper mantle discontinuity structure.

626

627 This study has predicted effects of kinetically inhibited growth of ringwoodite in mantle

628 upstreams and the expected effect on seismic probes for discontinuity structure. We find that
629 the effect on PP/SS underside reflection travel times is small; amplitudes, although harder to
630 assess in recorded data, might show a larger response to the altered discontinuity structure. For
631 P-to-s conversions travel time effects are minor but waveforms and amplitudes of conversions
632 are noticeably affected. Potentially P-to-s receiver function could be a suitable probe to detect
633 kinetic effects in the phase transitions and to map out mantle flow.

634

635 **5 Conclusions**

636

637 The effects of reaction kinetics of bridgmanite and ferropericlase to ringwoodite in upwelling
638 mantle on elastic properties were investigated using data of kinetic experiments and internally
639 self-consistent thermodynamic modelling. As the kinetic experiments show, growth of
640 ringwoodite is initiated at ferropericlase grain boundaries and is kinetically inhibited due to the
641 necessary diffusion of chemical components. At the initiation of ringwoodite growth,
642 bridgmanite transforms into metastable garnet. Ringwoodite grows from the metastable
643 assemblage of ferropericlase and pyrope-bearing garnet.

644

645 The results of the analysis of kinetically inhibited transformation at ~660 km depth predict
646 possible effects of kinetically inhibited growth of ringwoodite on seismic data which can be
647 summarized as follows:

648 1. The increased amount of ferropericlase and garnet results in a localized low velocity
649 zone of $\Delta V_P \sim -5\%$, $\Delta V_S \sim -7\%$ directly above the 660 km seismic discontinuity due to
650 the lower seismic velocities and density of ferropericlase and garnet compared to
651 ringwoodite and bridgmanite. The impedance contrasts for P and S waves are thus
652 increased compared with seismic velocity profiles of an equilibrated pyrolite mineral
653 assemblage.

654 2. The effects of reaction kinetics on the seismic fine structure at ~660 km depth depend
655 on the product of the squared grain size and upwelling rate. The change of seismic
656 properties correlates with the product of squared grain size and upwelling rate.
657 Modelling 1-D synthetic seismograms using the seismic profiles of kinetically inhibited
658 phase assemblages show travel time delays of 0.4-0.8 s for PP precursors and delays of
659 1.2-1.6 s for SS precursors for kinetic inhibition of the ringwoodite-forming reaction.
660 The finite width of the low-velocity layer above the seismic discontinuity results in
661 frequency dependent behaviour of PP and SS underside reflection amplitudes, with
662 higher amplitudes for 1-3 s PP waves and 5-8 s SS waves. However, high-frequency PP

663 and SS waves are rarely observed, and the predicted travel time differences are of
664 similar magnitude as typical picking uncertainties. Hence, it might be challenging to
665 utilize PP and SS underside reflections in order to observe kinetic effects.

- 666 3. Synthetic seismograms for P-to-s converted waves for velocity models for kinetically
667 inhibited transformations predict strong waveform variations for higher temperature
668 models and amplitude variations dependent on temperature, grain size and upwelling
669 rate. Travel time variations are maximum ± 0.3 s relative to models at equilibrium for
670 temperature of 1960 K. Receiver functions therefore might be a suitable seismic probe
671 to look for kinetic effects on the seismic structure of the 660 km discontinuity in mantle
672 upwellings. A future re-evaluation of P-to-s receiver functions at mantle upwellings
673 would be necessary to study the predicted effects described in this paper.
- 674 4. Possible candidates for mantle upwellings where reaction kinetics may become
675 important are transient domes with low buoyancy ratios and large lateral dimensions, or
676 broader scale plume upwellings. Material stalling at the phase transformation of
677 bridgmanite and ferropericlasite to ringwoodite might generate structure in the resolution
678 range of P-to-s converted waves.

679

680 **Data Availability**

681

682 The resulting data of this work and sources code developed during this work are not publicly
683 available. Except the results from kinetic experiments by Dobson & Mariani (2014), data are
684 based on tools available in the public domain:

- 685 1. Perple_X program package (Connolly, 2005), utilising the elastic and thermodynamic
686 properties and the solid solution model of Stixrude & Lithgow-Bertelloni (2011)
- 687 2. 1-D synthetic seismic modelling using the reflectivity method (Fuchs & Müller, 1971)
- 688 3. and data processing with seismic waveform analysis tool SeismicHandler (Stammler,
689 1993).

690 If you are interested in detailed data, please contact the corresponding author.

691

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693

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704 1.

705

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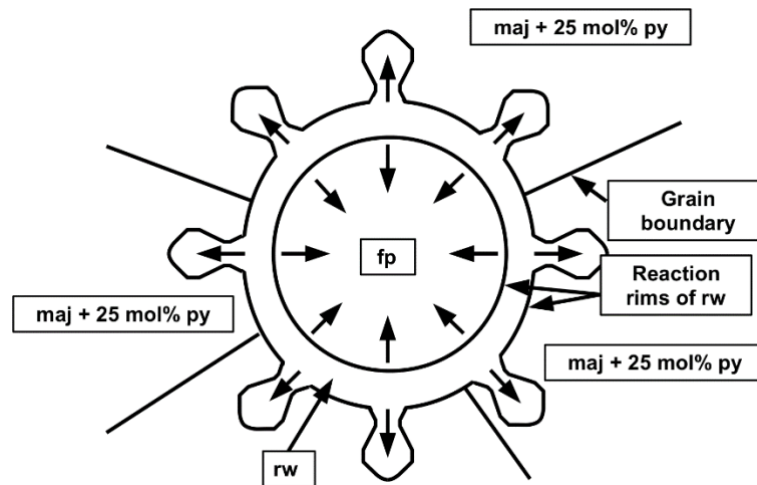
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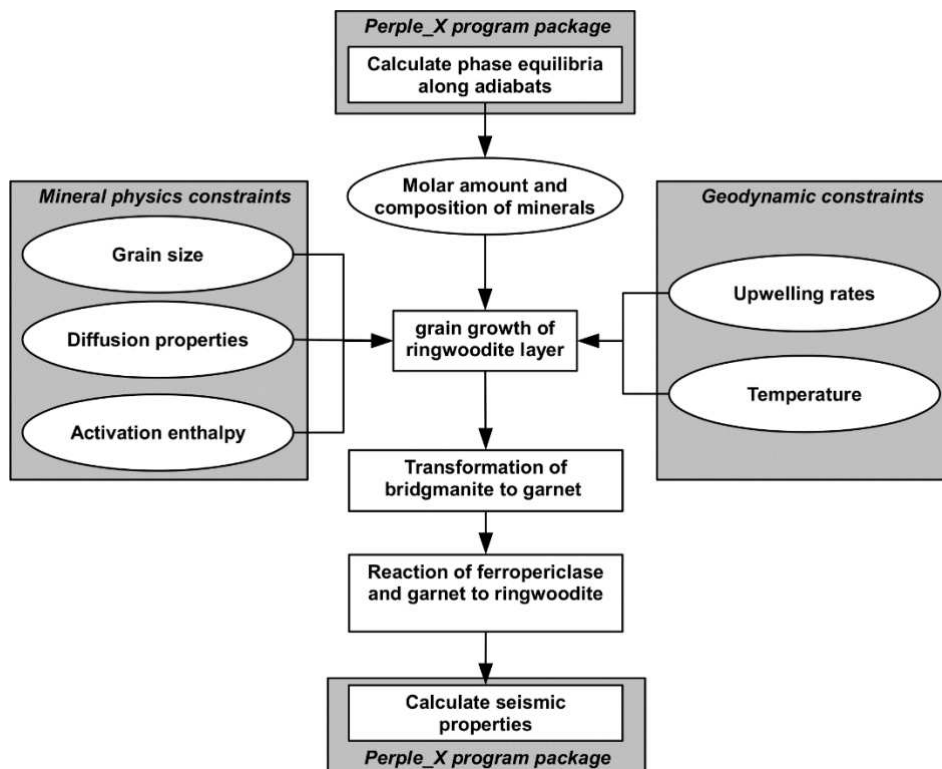
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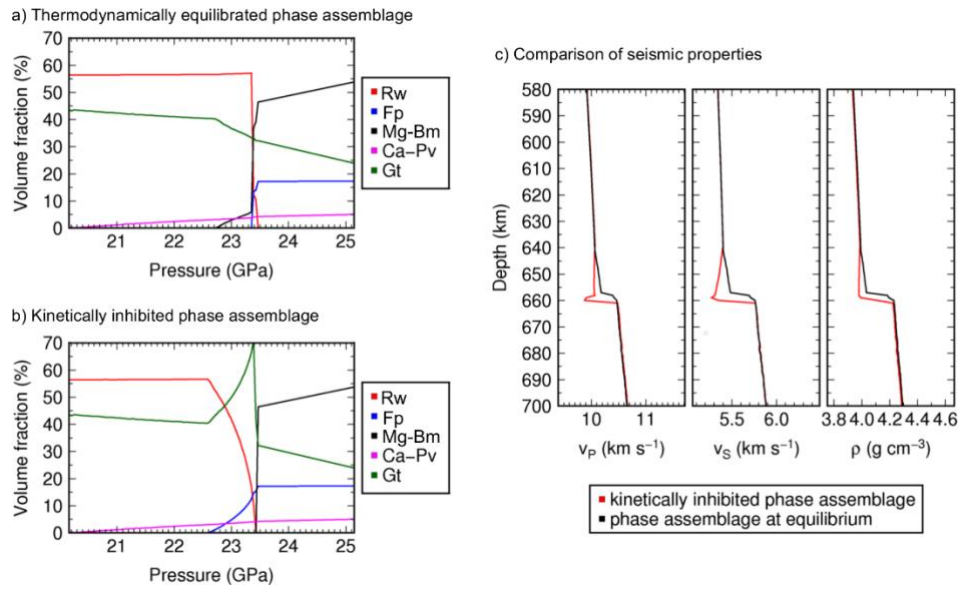
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983 **Figure 1:** Sketch of the reaction mechanism for the recombination of bridgmanite and
 984 ferropericlyase (fp) to ringwoodite (rw). The ringwoodite layer nucleates on the fp grain surface
 985 and grows into majorite garnet (maj) which contains 25 mol% Al-bearing pyrope (py) in a
 986 fingering instability. After Dobson & Mariani (2014).



990 **Figure 2:** Forward modelling of seismic properties for kinetically inhibited growth of
 991 ringwoodite and corresponding mineral reactions for ferropericlyase, bridgmanite and garnet.



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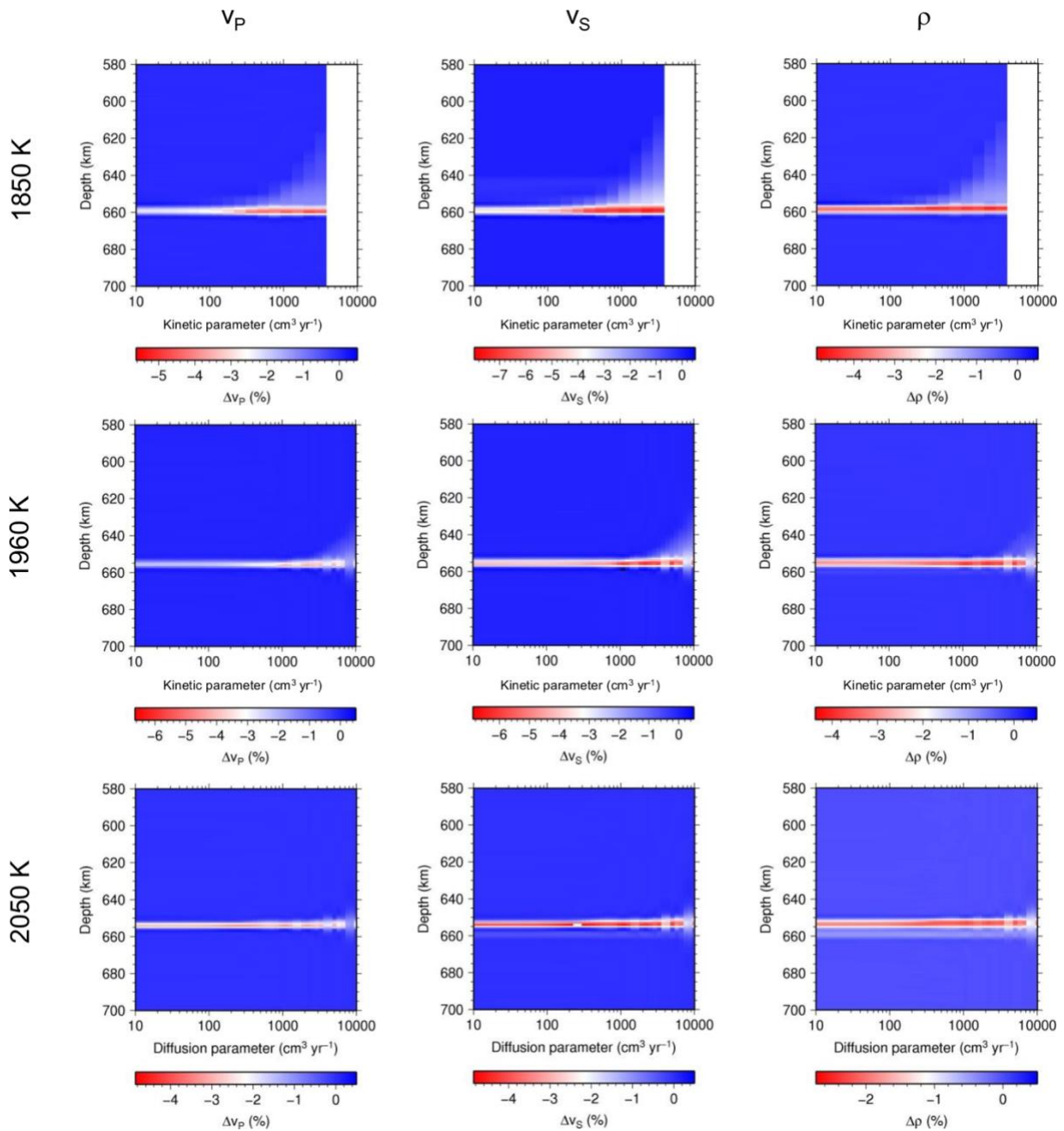
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995 **Figure 3:** Example for effects of kinetic inhibition of ringwoodite growth on volume fractions
 996 of minerals and on seismic properties of phase assemblages for temperature of 1850 K
 997 (following the 1300°C adiabat), 5 cm initial grain size and 50 cm/yr upwelling rate. a) Volume
 998 fractions of equilibrated phase assemblages as a function of pressure for pyrolite composition
 999 along the 1300°C adiabat. Mineral phases are abbreviated as follows: Rw: ringwoodite, Fp:
 1000 ferropericlase, Mg-Bm: magnesium bridgmanite, Ca-Pv: calcium perovskite, Gt: garnet. b)
 1001 Volume fractions of kinetically inhibited phase assemblages as a function of pressure for
 1002 pyrolite composition along the 1300°C adiabat. c) Comparison of seismic velocities and density
 1003 between equilibrated and kinetically inhibited phase assemblages.

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1008 **Figure 4:** Relative differences of P wave velocity, S wave velocity and density due to the

1009 kinetically inhibited growth of ringwoodite and the presence of metastable ferropericlase and

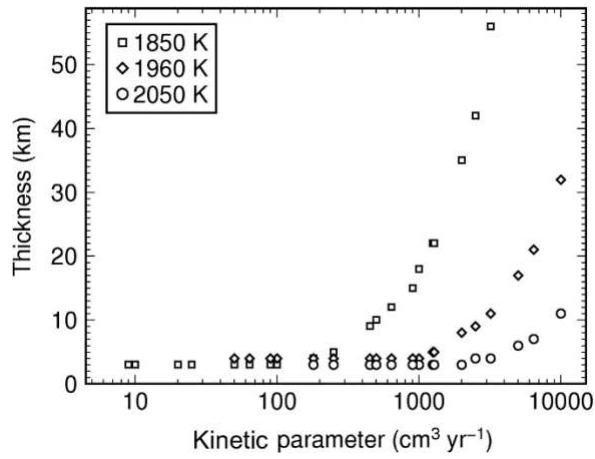
1010 garnet for temperatures of 1850 K, 1960 K and 2050 K. No data are available for kinetic

1011 parameters $> 4000 \text{ cm}^3 \text{ yr}^{-1}$ because the ringwoodite growth interferes with the ringwoodite-to-

1012 wadsleyite transition. To date, no studies with kinetic data for the ringwoodite-to-wadsleyite

1013 transition in mantle upstreams are known to the authors.

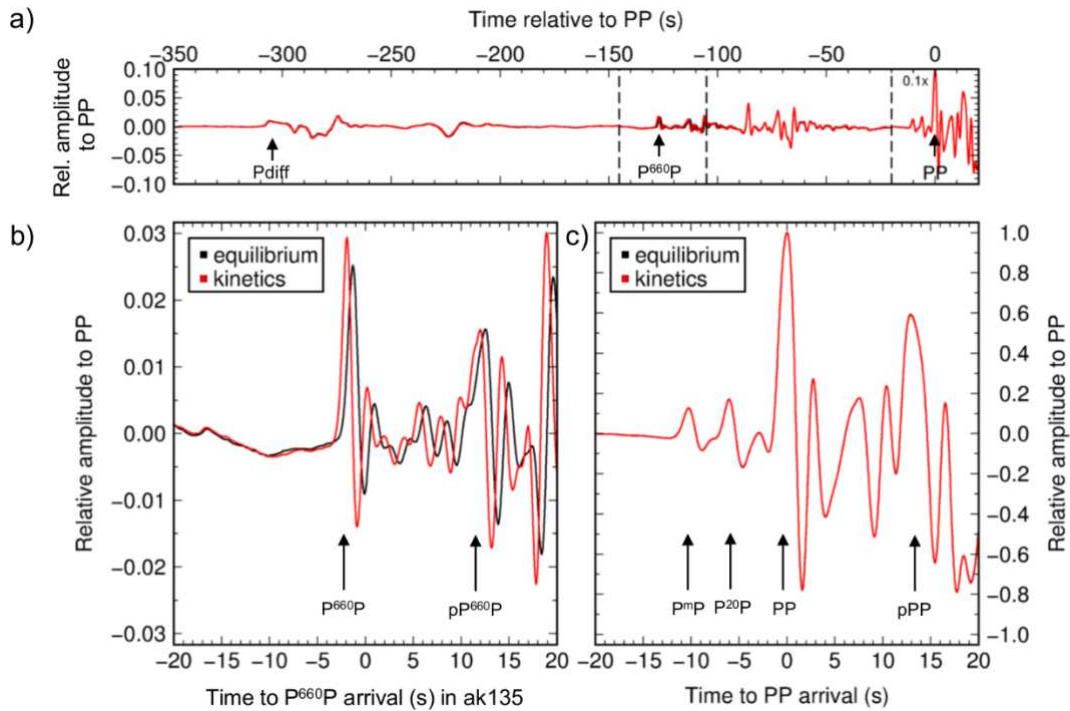
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1016 **Figure 5:** Thickness of the reaction interval where ringwoodite grows from the recombination
 1017 of ferropericlasite and garnet as function of the kinetic parameter D . The kinetic parameter D is
 1018 proportional to the upwelling rate v and the square of grain size d , for simplicity we set the
 1019 proportion factor to unity so that $D = v d^2$. Scenarios which have different grain size and
 1020 upwelling rates, but have the same kinetic parameter, show the same kinetic inhibition.

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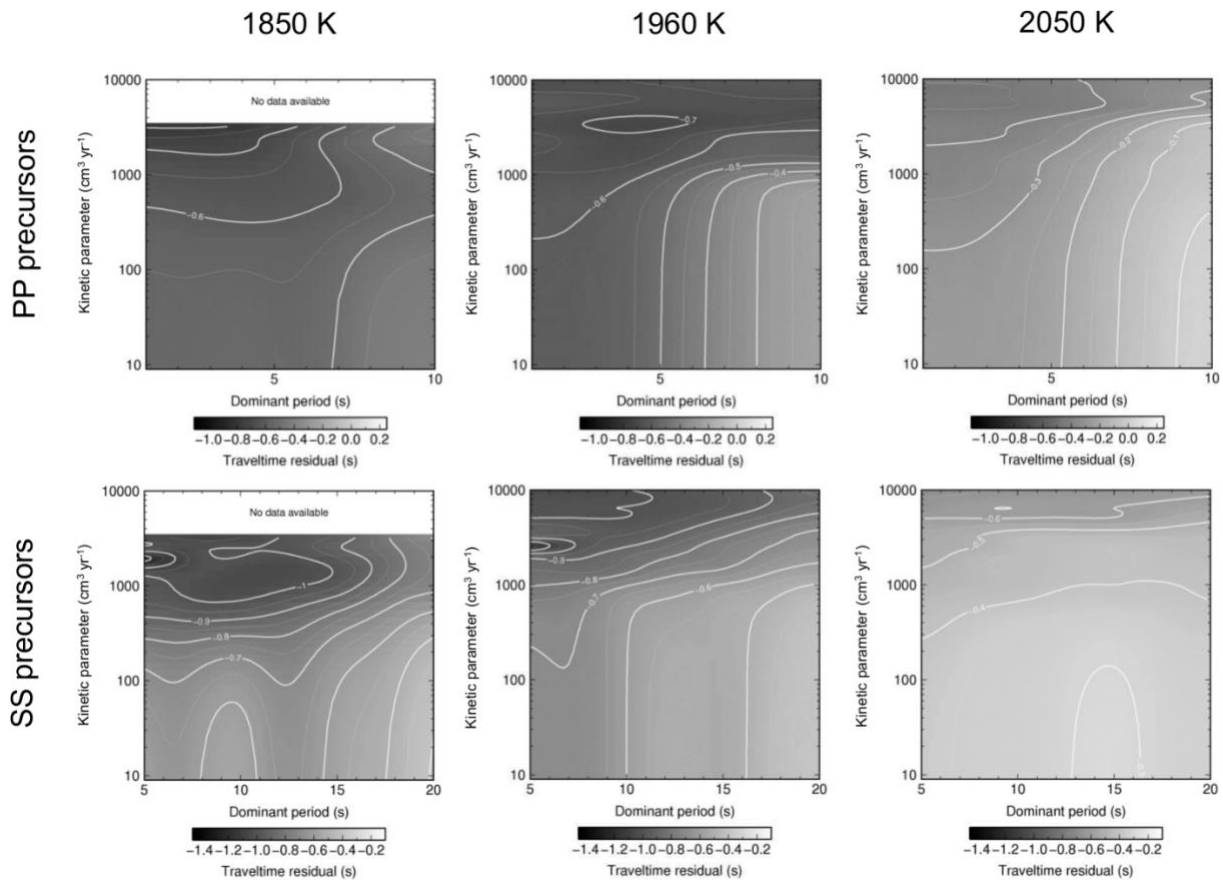


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1025 **Figure 6:** Synthetic seismograms of PP precursors with a dominant period of 3 s from seismic
 1026 models of equilibrated and kinetically inhibited phase assemblages at an array with reference
 1027 epicentral distance of 120 deg, a temperature of 1850 K, grain size of 5 cm and a vertical
 1028 upwelling rate of 50 cm/yr. a) shows the linearly stacked traces of all calculated seismograms
 1029 for equilibrated pyrolite phase assemblages (black) and kinetically inhibited phase assemblages
 1030 (red) aligned with slowness for PP from ak135 (Kennett et al., 1995). Time windows around
 1031 the predicted arrivals of P⁶⁶⁰P and PP from ak135 shown b) and c) and are marked by vertical
 1032 lines and the phase names. Arrows indicate the arrivals of P⁶⁶⁰P and pP⁶⁶⁰P in zoomed
 1033 diagram b) and the arrivals of PP and pPP in zoomed diagram c).

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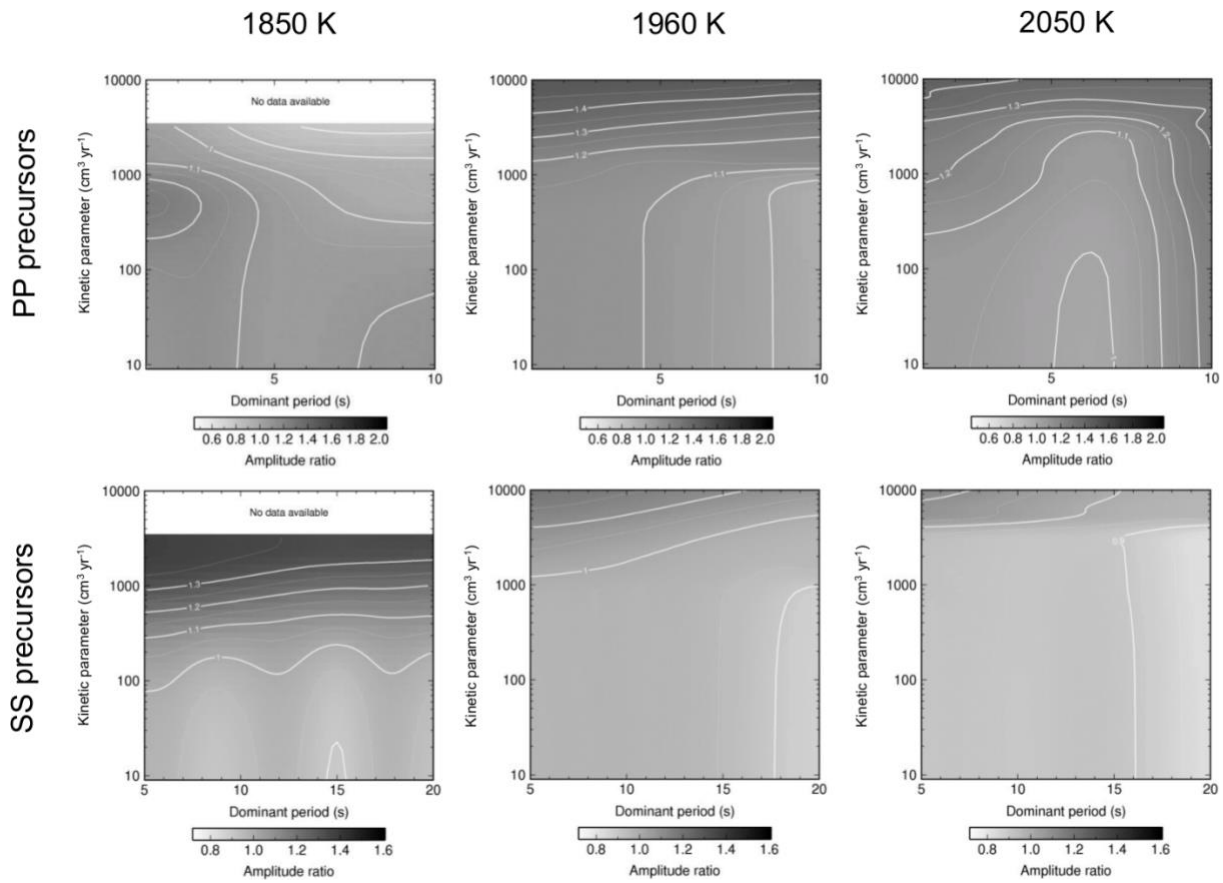


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1037 **Figure 7:** Travel time residuals of stacked PP and SS precursors between kinetically inhibited
 1038 and equilibrated backward reaction to ringwoodite as a function of temperature and kinetic
 1039 parameter for varying dominant periods at an epicentral distance of 120° . Bold lines denote
 1040 changes of travel time residuals by 0.1, thin lines denotes changes of travel time residuals by
 1041 0.05. Data are not available at temperature pf 1850 K and kinetic parameters $> 4000 \text{ cm}^3 \text{ yr}^{-1}$
 1042 (cf. Figure 4).

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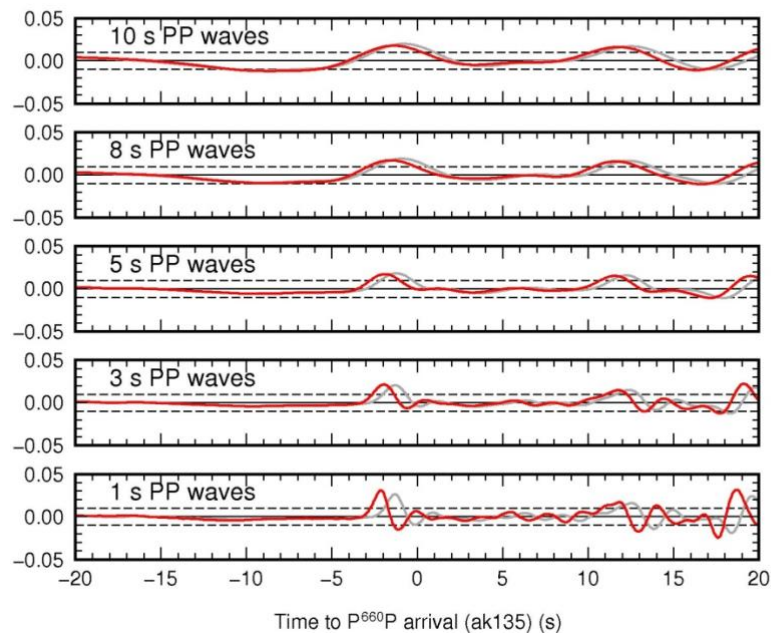
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Figure 8: PP and SS precursor amplitude ratios of reflections off the kinetically inhibited reaction vs. off the equilibrated reaction to ringwoodite as a function of temperature and kinetic parameter for varying dominant periods at an epicentral distance of 120° . Bold lines denote changes of amplitude ratios by 0.1, thin lines denote changes of amplitude ratios by 0.05. Data are not available at temperature pf 1850 K and kinetic parameters $> 4000 \text{ cm}^3 \text{ yr}^{-1}$ (cf. Figure 4).

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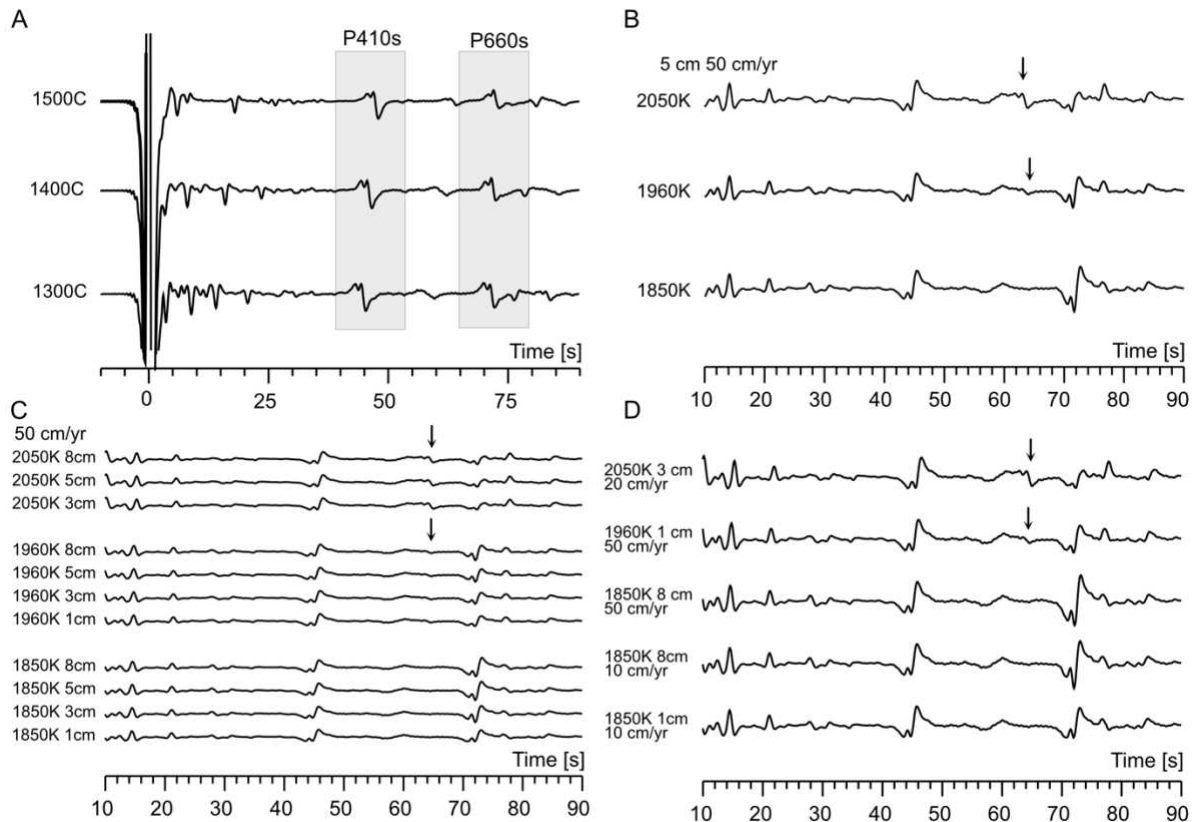
Frequency dependence of kinetic_model_1850k_5cm_50cm/yr at 120 deg for PP underside reflections



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1054 **Figure 9:** Frequency dependence of P⁶⁶⁰P underside reflections for equilibrated and kinetically
1055 inhibited phase assemblages at 1850 K, for 5 cm grain size and 50 cm/yr upwelling rate for
1056 dominant periods of 1 to 10 s. Waveforms of reflections off the kinetically inhibited backward
1057 reaction to ringwoodite are depicted in red, whereas waveforms of reflections off the
1058 equilibrated backward reaction to ringwoodite are depicted in black.

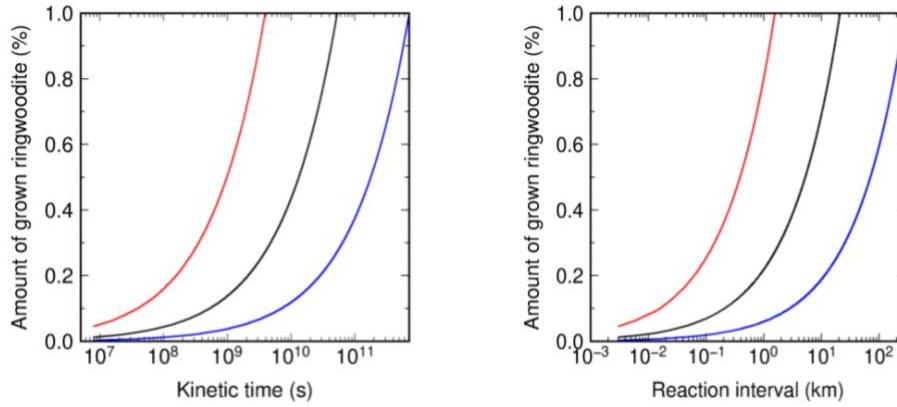
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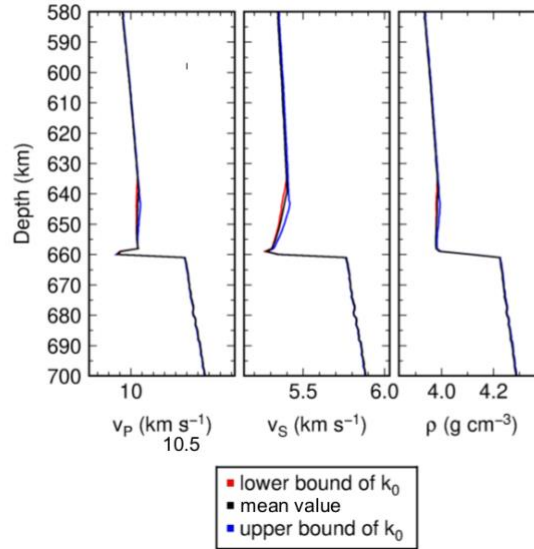
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Figure 10: Synthetic waveforms for P-to-S conversions arriving in the P-wave coda with dominant period of 1 s. All traces are recorded at 43 deg epicentral distance and aligned on the theoretical P-wave arrival. All times are given relative to P-wave arrival time. A) Waveforms for pyrolytic models at equilibrium conditions for mantle adiabats with surface potential temperatures ranging from 1300 to 1500 °C. P-to-s conversions from the 410-km discontinuity and the 660 km discontinuity are highlighted. B) Waveforms for kinetically inhibited transformation for models with 5 cm grain sizes and an upwelling rate of 50 cm/yr for geotherms at 1850 K, 1960 K and 2050 K. Amplitudes are normalized on the P-wave amplitude. Arrows indicate detectable precursors to P660s. C) Waveforms for kinetically inhibited transformation for 50 cm/yr upwelling rate and varying temperatures and grain sizes. Amplitudes are normalized on the P-wave amplitude. Arrows indicate detectable precursors to P600s. D) Waveforms for kinetically inhibited transformation for varying temperatures, upwelling rates and grain sizes. Amplitudes have not been normalized and the amplitude variations of the P-to-S conversion from the ferropericlase and bridgmanite to ringwoodite are observable. Arrows indicate detectable precursors to P660s.

a) Kinetic times as a function of uncertainties of ΔH_a b) Reaction interval as a function of uncertainties of ΔH_a



c) Influence of kinetic parameter on seismic properties



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1080 **Figure 11:** Effects of uncertainties of kinetic parameters on seismic properties of kinetically
 1081 inhibited phase assemblages and thickness of the low-velocity layer above the onset of
 1082 ringwoodite growth. The effects have been calculated for an ambient mantle temperature of
 1083 1850 K, an initial grain size of 5 cm and an upwelling rate of 50 cm/yr. a) Effects of uncertainties
 1084 of the enthalpy ΔH_a on the growth time of ringwoodite. The red curve denotes the calculation
 1085 for the lower bound of ΔH_a from Table 3 whereas the blue curve the upper bound of ΔH_a . b)
 1086 Effects of uncertainties of the enthalpy ΔH_a on the width of inhibited ringwoodite. c) Effects of
 1087 uncertainties of the diffusion coefficient k_0 on the seismic wave speeds and density. The lower
 1088 and upper bound of k_0 correspond to the uncertainties in Table 3.

1089

1090 **Tables:**

1091

1092 **Table 1:** Pyrolite composition by Sun et al. (1982) used in this study. The oxides of the CFMAS
1093 system are given as molar percentages.

1094

SiO ₂	MgO	Al ₂ O ₃	FeO	CaO
38.61	49.13	2.77	6.24	3.25

1095

1096 **Table 2:** Relationship between surface potential temperature of a mantle adiabat and their
1097 ambient temperatures at ~23.4 GPa, i.e., at ~660 km depth. Surface potential temperature are
1098 often given in °C (e.g., Cammarano et al., 2005; Cobden et al., 2008) whereas ambient
1099 temperatures are usually given in K.

1100

Potential temperature (°C)	1300	1400	1500
Ambient temperature (K)	1850	1960	2050

1101

1102 **Table 3:** Parameter range for calculating the seismic properties of kinetically inhibited phase
1103 assemblages.

1104

Parameter	Value(s)
From the kinetic experiments by Dobson & Mariani (2014):	
Enthalpy of the reaction ΔH_a (kJ mol ⁻¹)	456±40
Diffusion coefficient $\ln(k_0)$ (m ² /s)	-6.36±0.25
For modelling seismic properties:	
Temperatures (K)	1850, 1960, 2050
Grain size (cm)	1, 5, 10, 20
Upwelling rate (cm/yr)	1, 5, 10, 50, 100

1105