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# **Potential use of bile salts in lipid self-assembled systems for the delivery of phytochemicals**

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1           **ABSTRACT**

2           Phytochemicals show many desirable functions such as anti-oxidant, anti-microbial, anti-  
3 inflammatory, anti-carcinogenic, anti-aging, etc. activities. These biological functions can only  
4 be seen when the compounds are delivered at sufficient concentration. However, the  
5 bioavailability of phytonutrients suffers from their limited absorption, transformation, and  
6 rapid clearance from the circulation. In this review, we survey recent studies on the use of bile  
7 salts in lipid self-assembled systems to enhance the bioavailability of phytochemicals and  
8 consequently, their biological activity. The effect of bile salts on lipid self-assembled systems  
9 in terms of morphology, encapsulation efficiency, stability, bioavailability, and biological  
10 activities are discussed.

11           **Keywords:** Bile Salts, Lipid self-assembled system, Phytochemicals, Bioavailability,  
12 Biological activity

13

## 14           1. INTRODUCTION

15           In recent years, consumers have shown increased interest in including functional foods  
16 in their diet. Of particular interest are functional products containing health-promoting  
17 phytochemicals such as phenolic compounds, terpenes, phytosterols, etc [1, 2]. There is a  
18 growing body of literature that shows the potential of phytochemicals in the food, beverage,  
19 and pharmaceutical industries. They show many desirable biological functions including  
20 antioxidant, anti-microbial, anti-inflammatory, and anti-carcinogenic activity but these  
21 biological functions depend on their bioavailability. Phytochemicals can be included in the diet  
22 however when the amount of food and beverages that need to be consumed to reach 1  
23 g/person/day of resveratrol intake is calculated, a problem arises. It would require consuming  
24 795 kg of red Merlot grapes, 2584 kg of white Riesling grapes, and 505-2762 L of red wine  
25 (Figure 1). Thus, the recommended intake of these compounds may be unachievable by  
26 drinking beverages or consuming foods [3] and research is already being undertaken to improve  
27 the bioavailability of phytochemicals that are already reported as quite low after oral  
28 administration [4].

29           The bioavailability of these phytochemicals can be low because of their limited  
30 bioaccessibility or absorption, or because of fast metabolism in the body. The bioaccessibility  
31 of phytochemicals can be low because of limited liberation, poor solubility in the  
32 gastrointestinal tract (GIT), or through interaction with components in the GIT. Absorption can  
33 be significantly limited by the intestinal mucus layer, and also affected by innate properties of  
34 the compound including charge ( $\zeta$  potential), polarity, size, and hydrophobicity. Absorbed  
35 phytochemicals pass into systemic circulation via the liver or the lymphatic system [5].  
36 Phytochemicals and their metabolites that reach systemic circulation are absorbed by specific  
37 tissues and are finally excreted into the urine. Hence, their bioavailability may be limited by  
38 bioaccessibility, absorption, transformation, and rapid clearance from the circulation which  
39 causes low plasma concentration that is not enough to show significant pharmacological effects  
40 at target sites [6]. In order to overcome these limitations of phytochemical bioavailability , a  
41 promising solution is encapsulation technology and delivery systems [4].

42           Phytochemicals can be loaded into carriers and then released under controlled conditions  
43 at the target site. Encapsulation techniques are used to enhance the stability of active  
44 compounds during processing and storage, mask their undesirable taste, reduce their toxicity,  
45 preserve them from the harsh environment of the GIT, control their release and enhance their

46 cellular uptake and biological activities [2, 7, 8]. Different encapsulation technologies and  
47 materials can be used depending on the desired properties and function of active compounds  
48 such as increased surface area, solubility, homogeneity, stability, intracellular uptake, and  
49 bioavailability [2].

50 Among encapsulation techniques, there has been an expanding interest in encapsulation  
51 of bioactive compounds into the novel and promising lipid self-assembled systems (LSAS)  
52 such as phospholipid-based carriers (liposomes, bilosomes, transfersomes, cubosomes,  
53 hexosomes), surfactant-based carriers (niosomes), emulsions (single, double, microemulsions)  
54 and solid lipid nanocarriers (solid lipid nanoparticles, nanostructured lipid carriers (NLC),  
55 nano-oleogels) [9]. The properties of these LSAS can be enhanced with the use of different  
56 lipids, surfactants, or their combinations or by incorporation of several compounds such as  
57 cholesterol or bile salts (BSs) in the bilayer structure, and by modifying their surface with  
58 coatings [10]. The effect of phospholipid composition on liposome membrane properties has  
59 been investigated, showing that lipid type significantly affects the transition temperature ( $T_m$ ),  
60 elasticity, and fluidity of the membrane and consequently membrane stability and rigidity [11].  
61 The effect of sodium cholate (NaC) and sodium taurocholate (NaTC) on the membrane  
62 integrity of liposomes has been studied with a low  $T_m$  lipid (PC and 1,2-dimyristoyl-sn-glycero-  
63 3-phosphocholine (DMPC)) and high  $T_m$  (1,2-dipalmitoyl-sn-glycero-3-phosphocholine  
64 (DPPC), 1,2-distearoyl-sn-glycero-3-phosphocholine (DSPC), and sphingomyelin). The bile  
65 salts were shown to be less destructive to liposomes consisting of lipids with a high  $T_m$  [12].  
66 Coating with chitosan also shows promising results during digestion. The effect of low  
67 molecular weight chitosan coating on the bioavailability of curcumin-loaded liposomes [13]  
68 during *in vitro* digestion showed that after the oral and gastric phases of digestion, slightly  
69 higher curcumin concentration was detected compared to uncoated liposomes. Curcumin  
70 concentration was higher for chitosan-coated liposomes after the intestinal phase.

71 Bile salts are formed from conjugated bile acids that form complexes with sodium.  
72 Primary bile acids cholic acid (CA) and chenodeoxycholic acid are synthesized from  
73 cholesterol in the liver. These primary bile acids are conjugated to either taurine or glycine at  
74 the C-24 carboxyl group to increase their hydrophilicity. Intestinal bacteria convert primary  
75 bile acids to a secondary form (deoxycholic acid and lithocholic acid) [14]. BSs such as NaC,  
76 NaTC, and sodium deoxycholate (NaDC) are characteristic amphipathic molecules that contain  
77 a steroid nucleus. The presence of hydroxyl groups provides hydrophilic properties to the

78 concave side of the molecule, whereas the convex side is hydrophobic due to methyl groups.  
79 Because of their steroid skeleton, bile salts have a rigid structure that is very different compared  
80 to head-tail surfactants [15]. Due to their amphiphilic nature, bile salts form primary  
81 (aggregation number 2-10) or secondary (aggregation number 10-100) micelles in aqueous  
82 solutions above the critical micelle concentration (CMC) [16, 17]. Their CMC value is also not  
83 well defined so it is given as a range. They have low aggregation numbers of 4-6 molecules for  
84 NaC and 7-12 molecules for NaDC [18].

85 BSs show powerful solubilisation potential against phospholipid bilayers. When bile salt  
86 concentration reaches its CMC, bile salts solubilize the vesicles to mixed micelles [19]. BSs  
87 are incorporated into the outer layer of the phospholipid bilayer below their CMC and upon  
88 further increase of BS concentration, BSs transport to the inner leaflet of the bilayer. When the  
89 total BS concentration reaches the saturation phase boundary, the system transforms to  
90 coexisting micelles and vesicles. Upon further increase of BS concentration the solubilisation  
91 phase boundary is reached and BS-rich mixed micelles are formed. During solubilisation, all  
92 steps have aggregates with BS monomers [18]. Incorporating bile salts in the vesicle bilayer  
93 results in a negatively charged surface. Through electrostatic repulsion, BS/lipid systems are  
94 thought to stabilize the bilayer against intestinal BS adsorption [20, 21] ultimately leading to a  
95 slower vesicles-to-mixed micelles transition (Figure 2) and enhanced delivery of active  
96 compounds. BSs are also used as a permeability enhancer, which changes the membrane  
97 permeability and fluidity [22].

98 A considerable amount of literature has been published on BS-LSAS but this review  
99 primarily focuses on BS-LSAS for the delivery of phytochemicals. Therefore, the effect of the  
100 presence of BSs in LSAS in terms of the morphology, encapsulation efficiency, release profile  
101 and stability of the LSAS, and bioavailability and biological activities of loaded  
102 phytochemicals are discussed.

## 103 **2. Effect of bile salts on the morphology of lipid self-assembled systems**

104 Liposomes are the most widely used carriers among LSAS (Figure 2). The properties of  
105 liposomes can be enhanced with the use of different compounds such as BS in the formulation.  
106 The terms used to describe BS-liposomes vary among researchers. For example, bilosomes [23,  
107 24], transfersomes [25, 26], ultradeformables or elastic liposomes [27, 28] are all used. BS-  
108 liposomes are designed based on the structure of liposomes, which also can encapsulate both

109 hydrophilic and lipophilic bioactive compounds [29]. BS interactions with phospholipid  
110 membranes can have a range of effects on the morphology of LSAS.

111 NaDC-nanotransfersomes were designed and optimized using response surface  
112 methodology to encapsulate sodium pravastatin and naringenin [25]. The amount of omega-3  
113 phospholipid, NaDC, and naringenin were selected as independent variables.  $D_h$ , encapsulation  
114 efficiency (EE(%)), cumulative % permeated, alanine amino transaminase (ALT) level (IU/L),  
115 and malondialdehyde (MDA) levels (mmol/mg protein) were determined as dependent  
116 variables. The impact of independent variables was reported as omega-3-  
117 phospholipid>NaDC>naringenin. Optimization results showed that increasing NaDC  
118 concentration significantly reduced the  $D_h$  of nanotransfersomes ( $D_h$ : 86-249 nm) ( $p<0.0001$ ).  
119 For optimum composition (Table 1), 76% of sodium pravastatin and naringenin were loaded  
120 in NaDC-nanotransfersomes ( $D_h$ : 191 nm). In the same way, the increased concentration of  
121 BSs resulted in the decreased  $D_h$  of resveratrol-loaded NaC-transfersomes [26] and  
122 notoginsenoside R1-loaded NaGC-liposomes [30]. Resveratrol loaded-NaC-transfersomes  
123 were optimized using face-centred central composite design (CCD). The amount of SPC and  
124 NaC were chosen as independent variables and their effect on  $D_h$  and EE(%) were chosen to  
125 optimise NaC-transfersomes. The  $D_h$  and polydispersity index (PDI) of the optimized  
126 formulation was  $178.9\pm 12.87$  nm and 0.132 respectively (Table 1). Their shape was found to  
127 be nearly spherical and uniform by transmission electron microscopy (TEM). According to the  
128 results, once the NaC concentration was increased, the  $D_h$  of transfersomes decreased [26].  $D_h$   
129 of notoginsenoside R1-loaded NaGC-liposomes (200 nm) was smaller than conventional  
130 liposomes and when the concentration of NaGC increased,  $D_h$  and EE(%) decreased [30].  
131 These findings indicate that the concentration of BSs in the formulations plays an important  
132 role in the  $D_h$  of vesicles. A decrease of the  $D_h$  can result from increased BSs concentration  
133 that can decrease the surface tension of the vesicles and stabilize the bilayer and enhance the  
134 elasticity of vesicles [20, 26].

135 Biflavonoid extract from *Selaginella doederleinii* was encapsulated using liposomes  
136 [31]. The SPC/biflavonoid extract ratios (15:1, 10:1 and 5:1 w/w) and SPC/NaDC ratios (15:1,  
137 10:1 and 5:1 w/w) were chosen for optimization to see the effect on EE(%). The  
138 SPC/biflavonoid extract ratios showed the lowest effect among the four variables and 10:1 w/w  
139 ratio was chosen for optimum formulation (Table 1). The SPC/NaDC ratio had the second-  
140 highest effect among the four variables. Optimized liposomes (SPC/NaDC, 10:1 w/w) had a

141  $D_h$  of  $249.77 \pm 15.68$  nm with PDI of  $0.184 \pm 0.002$ .  $91.39 \pm 0.88\%$  of the extract was encapsulated  
142 in the optimized liposomes. These results are similar to those reported by Yang *et al.* who  
143 studied tripterine-loaded hyaluronic acid (HA)-coated NaDC-bilosomes. Effects of  
144 concentration of NaDC (1, 2, and 3 mg/mL), on the  $D_h$  and EE(%) of bilosomes were evaluated  
145 [23]. With increased NaDC concentration from 1 mg/mL to 2 mg/mL,  $D_h$  decreased from ~140  
146 nm to ~90 nm [25, 26, 30]. However, when NaDC concentration was increased to 3 mg/mL,  
147 the  $D_h$  of bilosomes increased to ~110 nm. A possible explanation for these results, up to a  
148 specific concentration of BSs reduces the surface tension that induces membrane curvature and  
149 stabilizes the system and leads to decreased  $D_h$ . Further increase in BS concentration and an  
150 increase in the  $D_h$  may be related to aggregation in the system and increased medium viscosity  
151 [23, 24, 32]. The 2 mg/mL NaDC in the bilosome formulation had the lowest  $D_h$  (~90 nm) and  
152 the highest EE(%) (~100%). For the HA-coated particles, 2 mg/mL of NaDC was chosen and  
153 the  $D_h$  was 95.3 nm and 118.4 nm for the bilosomes and HA-functionalized bilosomes,  
154 respectively.

155 Barone *et al.* studied ammonium glycyrrhizate-loaded liposomes composed of SPC and  
156 NaC. Loading of ammonium glycyrrhizate did not affect the  $D_h$ , PDI,  $\zeta$  potential, and the  
157 deformability index (DI) of vesicles significantly.  $D_h$  increased from  $109 \pm 3.8$  to  $128 \pm 4.6$  nm  
158 and the  $\zeta$  potential decreased from  $-17 \pm 0.5$  to  $-22 \pm 0.6$  mV [27]. A decrease in the  $\zeta$  potential  
159 can result from the negative charge on the liposome surface arising from the interaction  
160 between hydrophilic groups of ammonium glycyrrhizate and water molecules and an increased  
161 number of hydrogen bonds. Moreover, the solvent layer can be increased with the interaction  
162 of hydrophilic groups and water that may result in an increased  $D_h$  [27, 33]. The lowering of  
163 the  $\zeta$  potential mainly results from the presence of NaC in the formulation. The LSAS with  $\zeta$   
164 potential values more than  $\pm 30$  mV shows moderate stability. As the  $\zeta$  potential comes close to  
165 zero, the system lost its stability, and agglomeration and precipitation were seen [34].

166 Surface-modified bilosomes were prepared by Waglewska *et al.* and triblock copolymer  
167 Pluronic P123 was used to stabilise the surface. Three different concentrations of NaC were  
168 added into formulations (0.25, 0.50, and 1.00 wt %) [24]. The effect of the presence of NaC in  
169 the bilosome formulation was similar to those shown by others [27]. The negative  $\zeta$  potential  
170 of the formulation results from the negative charge of NaC [35]. Sulphated polysaccharide-  
171 protein complexes of *Enteromorpha intestinalis*-loaded bilosomes were prepared using three  
172 different BSs (NaC, NaDC, NaTDC) and two different surfactants including Span 40



173 (hydrophilic-lipophilic balance (HLB): 6.7 [36] and Span 65 (HLB: 2.1) [37]. Span 40-  
174 bilosomes had more than double the  $D_h$  compared to Span 65-bilosomes for all BSs  
175 formulations. The lower  $D_h$  of Span 60-bilosomes resulted from the lower HLB value of Span  
176 65. Low HLB results in a decrease in the surface free energy and shows increased lipophilic  
177 affinity that leads to smaller vesicles [38]. Higher HLB values lead to water uptake into the  
178 vesicles and cause increased vesicle size due to hydrophilicity [21]. BSs had a significant effect  
179 on the  $D_h$  of bilosomes. The  $D_h$  of bilosomes were reported as NaTDC-bilosomes >NaDC-  
180 bilosomes>NaC-bilosomes. The EE(%) were ranged from 69.66% to 71.60% ( $p>0.05$ ). Span  
181 65/NaC-bilosomes had the highest EE(%) ( $71.60\pm 0.25\%$ ) with  $D_h$ :  $181\pm 16.80$  nm [39]. In  
182 phenylethyl resorcinol-loaded transfersomes, NaDC (HLB: 16), Tween 80 (HLB: 15), Tween  
183 20 (HLB: 16.7), Span 80 (HLB: 4.3), Span 20 (HLB: 8.6) were used as skin enhancers in the  
184 formulation [40]. In contrast to previous work [39], the  $D_h$  of NaDC-transfersomes was  
185  $353.70\pm 10.90$  nm, and the PDI,  $\zeta$  potential, and EE(%) of NaDC-transfersomes were  
186  $0.111\pm 0.038$ ,  $-41.86\pm 0.73$ , and  $92.49\pm 0.01$  respectively. TEM and SEM analysis also revealed  
187 that NaDC-transfersomes had a unilamellar to multilamellar structure with a smooth surface  
188 [40].

189 Catechin-loaded hexosomes for topical application have been prepared using NaC,  
190 NaDC, and NaTC in the formulations and the  $D_h$  of hexosomes was 148 nm without BSs but  
191 the  $D_h$  decreased to 107–110 nm after 0.6 wt% BS addition [41]. Moreover, small-angle X-ray  
192 scattering (SAXS) measurements, showed that after the addition of 0.02 wt% of BSs in the  
193 formulation of an inverse bicontinuous cubic phase dispersion (cubosomes) completely  
194 transitioned to unilamellar vesicles. The phase transition could have been caused by the  
195 amphiphilic structure of BSs that is in contact with the bulk water so decreases the negative  
196 interfacial curvature of the system [41].

197 Surfactant concentration (NaGC and Polysorbate 80) was chosen as one of the  
198 independent variables for the optimization of gypenosides loaded NLCs [42]. While the  
199 surfactant concentration did not show a significant effect on  $D_h$  it had a significant effect on  
200 EE(%). The EE(%) increased with increased surfactant concentration and the  $D_h$ , PDI, and  $\zeta$   
201 potential of the optimized NLCs formulation (Table 1) were reported as  $146.7\pm 6.8$  nm,  
202  $0.137\pm 0.011$ , and  $-56.0\pm 3.4$  mV respectively with 74.22% in EE(%).

203  $D_h$  of LSAS is a considerable parameter in the delivery of active compounds and,  
204 consequently, their passage through a biological membrane. According to the results, a

205 decrease of  $D_h$  is correlated to an increase in BSs up to a specific concentration (Figure 3). BSs  
206 stabilize the vesicles by reducing the interfacial tension of the vesicles. Further increased BSs  
207 concentration, BSs begin to show disruption effect on the LSAS.

### 208 **3. Effect of bile salts on encapsulation efficiency, release, and stability of lipid self-** 209 **assembled systems**

210 The encapsulation efficiency of LSAS varies depending on the hydrophobicity or  
211 hydrophilicity of encapsulated phytochemicals, stability of the lipid membrane, the surface  
212 area of LSAS and the methods and parameters chosen for vesicle loading. The general stability  
213 of LSAS comprises the stability of active material and the retention capacity of the membranes.  
214 Membrane stability that provides integrity of the system has a direct effect on the encapsulation  
215 efficiency [43].

216 NaDC-liposomes and HA-coated NaDC-liposomes were prepared to encapsulate  
217 tripterine [23]. Both formulations showed a narrow PDI and the  $\zeta$  potential decreased from 4.8  
218 mV to -34.2 mV due to the use of the negatively charged polysaccharide coating. EE(%) and  
219 DL(%) were 99.56% and 8.15% respectively. Doxorubicin-silybin-loaded liposomes were  
220 prepared using cholic acid (CA) to improve the oral treatment of actives [44]. The  $D_h$  of CA-  
221 liposomes was  $97.03 \pm 2.17$  nm with a narrow PDI and both active compounds had an EE(%)  
222 of approximately 95%. *In vitro* release was investigated in PBS buffers (pH: 2 and 7.4) to  
223 mimic the simulated gastric fluid (SGF) and simulated intestinal fluid (SIF) conditions. While  
224 ~25% of silybin was released from liposomes at pH 2, ~50% of silybin was released in pH 7.4  
225 in 12 h. In addition, the stability was determined in rat gastric fluid, rat intestinal fluid, and  
226 serum. After a 12 h incubation, the EE(%) of all liposomes in different steps was still more  
227 than 80% which shows that CA-liposomes protect their integrity during different steps of  
228 digestion. However, some active material leakage problems were also reported because of the  
229 presence of a high concentration of BSs. When the effect of NaC concentration on EE(%) of  
230 resveratrol-loaded liposomes was studied, high EE(%) was seen at intermediate NaC levels  
231 and a decrease in the EE(%) was seen as NaC was increased. Higher NaC amounts also can  
232 cause leakage of active material because of the formation of lipid pores in the membrane. The  
233  $\zeta$  potential of the optimized formulation (Table 1) was reported as  $-23.2 \pm 1.4$  mV [26].

234 The EE(%) of the system is affected by hydrophobicity or hydrophilicity of encapsulated  
235 phytochemicals. In vesicle preparations the EE(%) of hydrophilic phytochemicals is related to

236 the aqueous phase trapping ability of LSAS. When the desired rigidity of the lipid membrane  
237 could not be achieved, encapsulated hydrophilic phytochemicals may leak through the lipid  
238 membrane by diffusion [45]. The EE(%) of hydrophilic phytochemicals are generally reported  
239 as less than 30%. Hydrophobic phytochemicals can be incorporated into the lipid phase of  
240 LSAS during vesicle formation. Thus, the EE(%) of hydrophobic phytochemicals in LSAS  
241 mainly depends on the interaction between phytochemicals and lipids. High EE(%) can be  
242 reached with optimised lipid type and concentration in the formulation [46]. The EE(%)  
243 increases with increased lipid concentration and increases the Dh of particles so a higher  
244 amount of phytochemicals can be loaded [47].

245 The addition BSs below the CMC into the formulation can also enhance the EE(%) of  
246 LSAS (Figure 3). The flexibility of vesicles increases with the presence of BSs and the bilayer  
247 and the aqueous core can be flexed to encapsulate the drug more [48]. BSs above the CMC  
248 show solubilisation effect on vesicles so vesicles transit to micelles and increase the drug  
249 solubility consequently decrease the EE(%). In addition, drug leakage can be seen because of  
250 opened lipid membrane pores [49].

251 *In vitro* release studies are important to give a prediction about the *in vivo* release profile  
252 [33]. *In vitro* dissolution profiles of NLCs and NaGC-NLCs were compared with gypenosides  
253 powders. Cumulative dissolution of gypenosides powders (29.4%) was slightly higher than  
254 both NLCs in 6 h. Although NaGC-NLCs showed slightly higher release, the release of NLCs  
255 and NaGC-NLCs were not significantly different after 48 h. While ~55% of gypenosides was  
256 released from both NLCs at 48 h, the cumulative dissolution of gypenosides powders did not  
257 change significantly after 6 h. The most likely cause of slow-release from NLCs is due to lipid  
258 layers slowing the diffusion of the active compounds [42]. Sulphated polysaccharide-protein  
259 complexes-loaded bilosomes were prepared. Span 65-NaC-bilosomes and Span 40-NaC-  
260 bilosomes released ~70% and ~40% of active compounds respectively, while other  
261 formulations prepared with NaDC or NaTDC released less than ~20% in PBS (pH 6.8) at 37°C  
262 in 24 h [39]. It was reported that while ~50% of [6]-gingerol released, using NaC-liposomes  
263 significantly increase the cumulative release of [6]-gingerol (~90% in 24 h) (p<0.01) [50]. *In*  
264 *vitro* release from bilosomes and HA-coated bilosomes were investigated [23]. Bilosomes  
265 released only 40.51% of tripterin in phosphate-buffered saline (PBS) (pH 7.4) in 24 h, which  
266 may be due to the hydrophobic nature of tripterin. The HA-coating also slowed the release of  
267 actives from bilosomes; 11.35% and 23.24% of tripterin were released at 8 h and 24 h,

268 respectively. The release of active compounds can be controlled by loading the active into the  
269 system that acts as a barrier. Also, these barriers can be modified using coatings which provide  
270 additional layers to slow down the release. The presence of BSs in the self-assembled  
271 formulation enhances the physical stability of the system and thus can slow down the release  
272 [51].

273 The physical and chemical stability is important and to provide long-term shelf-life and  
274 to protect the structure of LSAS from the harsh conditions of the GIT and keep the structure in  
275 circulation longer[52]. Several factors such as lipid type, PDI, and  $\zeta$  potential of the system,  
276 etc. are known to affect the stability of LSAS. BSs are only one of many factors that can  
277 enhance stability.

278 Phenylethyl resorcinol-loaded NaDC-transfersomes were stored at 4°C, 30°C and 45°C  
279 for 4 months (RH: 75%). While NaDC-transfersomes was stable for 4 months at 4±1°C and  
280 30±1°C, they could retain their stability for 2 months at 45±1°C [40]. The accelerated stability  
281 of NLCs and NaGC-NLCs was monitored and both NLCs showed stable structures. The  $D_h$   
282 and EE(%) of NLCs did not change significantly during storage at 40±2°C, relative humidity:  
283 75±5% for 6 months [42]. Likewise, methylene blue and curcumin-loaded NaC-bilosomes  
284 showed minimum changes during storage at 4°C for 28 days. The  $D_h$  and PDI of the  
285 formulations were reported as ~100 nm and <0.25. EE(%) were 85% and 70% for methylene  
286 blue and curcumin, respectively [24]. The physical stability of ammonium glycyrrhizate-loaded  
287 NaC-liposomes (EE(%): 57.3±3.7%) was determined at 25°C and 37°C for 4 weeks. While the  
288  $D_h$  of vesicles did not change significantly at 25°C, a significant increase was observed at 37°C  
289 after 2 weeks, which could be related to the gel to fluid lamellar phase transition at 37°C [27].

290 PDI and  $\zeta$  potential of LSAS are important parameters that reflect the homogeneity  
291 (PDI<0.3) of the particles and stability (more than ±30 mV) of particles respectively.  
292 Aggregation, flocculation, or precipitation problems can be seen during storage with the  
293 decrease of  $\zeta$  potential below ±30 mV [53]. The net surface charge of particles comes from the  
294 combination of lipid, active compounds, and other compounds that are used to modify the  
295 structure. The addition of BSs in LSAS increases  $\zeta$  potential negatively and physically  
296 stabilizes the system due to the increased repulsion between particles. An increase of negative  
297 values of  $\zeta$  potential is correlated to an increased BS concentration due to the negative charge  
298 of BSs.

#### 299           **4. Effect of bile salts on the bioavailability of lipid self-assembled systems**

300           The biological fate of LSAS that aims to enhance the bioavailability of active compounds  
301 differs depending on their administration. While digestion is one of the main steps for the oral  
302 delivery of active materials, skin penetration is taken into consideration for the topical  
303 application of active materials. Much research is already being undertaken on LSAS for further  
304 exploiting the biological activities of phytochemicals.

305           After oral administration, LSAS pass through the oral cavity and reach the stomach. In  
306 the gastric compartment, LSAS interact with gastric fluids (pH 1-3) and a high concentration  
307 of calcium and sodium salts. Physical properties begin to change during this interaction and  
308 some lipophilic compounds in the structure begin to get digested by gastric lipases in the  
309 stomach (10–30%). After the gastric phase, LSAS that can survive the acidic environment of  
310 the stomach enter the small intestine (pH 6-7.5). Compounds mix with bile salts from the gall  
311 bladder and pancreatic secretions. Following hydrolysis of lipophilic compounds to free fatty  
312 acids and monoglycerides by pancreatic lipases in the small intestine (70–90%), bile salts, free  
313 fatty acids, monoglycerides, and phospholipids form mixed micelles which can be absorbed by  
314 the intestinal enterocytes [54, 55]. LSAS that are composed of digestible compounds lose their  
315 integrity after the gastric and intestinal phases. Hydrophilic phytochemicals pass to the  
316 enterocytes for absorption and are transported to the portal vein [54, 56]. Highly hydrophobic  
317 phytochemicals (partition coefficient ( $\log P$ )>5) are attached to chylomicrons. Chylomicrons  
318 enter the lymphatic system and carry these hydrophobic phytochemicals to the tissues.  
319 Hydrophobic phytochemicals are released from chylomicrons by hydrolysis of lipoprotein by  
320 lipases for the absorption of phytochemicals at the tissue [54, 56].

321           LSAS that are composed of indigestible coatings like chitosan, pectin, etc can remain  
322 intact through the small intestine and pass to colon where they may be hydrolysed by colonic  
323 microflora, encapsulated phytochemicals can then be absorbed in the colon [54].

324           Gypenosides are triterpenoid saponins from *Gynostemma pentaphyllum*. Gypenosides  
325 are used for medical purposes for many years however they are sensitive to the environment  
326 and suffer low solubility both in lipid and water [42]. In a study designed to assess the delivery  
327 of gypenosides for oral application, the time to peak plasma concentration ( $T_{max}$ ), maximum  
328 plasma concentration ( $C_{max}$ ), area under the plasma concentration ( $AUC_{0-\infty}$ ), and mean  
329 retention time ( $MRT_{0-\infty}$ ) were investigated in rats after oral administration of gypenosides

330 powders, NLCs and NaGC-NLCs. Loading of gypenosides into NLCs prolonged the  
331 absorption time 1.8-fold and retention time 1.9-fold ( $p < 0.05$ ). The  $C_{max}$  of NLCs ( $0.671 \pm 0.073$   
332 mg/mL.h) also increased ~1.5-fold compared to the powder so it shows that NLCs remained  
333 longer in the circulation. The addition of NaGC in the NLC formulation increases the  $T_{max}$  and  
334  $MRT_{0-\infty}$  significantly ( $p < 0.05$ ). The plasma concentration of gypenosides increased compared  
335 to gypenosides powders and NLCs ~1.9-fold and 1.3-fold, respectively. Intestinal absorption  
336 of NLCs and NaGC-NLCs was also compared with the gypenosides powder. The apparent  
337 absorption coefficient ( $P_{app}$ ) of NLCs showed increased permeability and bioavailability  
338 compared to gypenosides powder ( $p < 0.05$ ) in all sections of the intestine. The presence of  
339 NaGC in NLCs increased the  $P_{app}$  5.11-fold compared to gypenosides powder and ~2-fold  
340 compared to NLCs in the jejunum ( $p < 0.05$ ). In addition, Caco-2 cell viability was increased  
341 from ~70% to ~90% at 100  $\mu$ g/mL gypenosides due to loading the gypenosides into NaGC-  
342 NLCs that reduced the cytotoxicity of gypenosides [42].

343 The delivery of tripterin isolated from *Tripterygium wilfordii* was investigated [23].  
344 Tripterin, also known as celastrol, shows antioxidant, anti-angiogenic, and anti-rheumatic  
345 effects. However, it is poorly soluble in water and consequently shows low solubility in  
346 biological fluids [57]. Cellular uptake of free tripterine, bilosomes, and HA-coated bilosomes  
347 in RAW264.7 cells was compared. Loading tripterine into bilosomes doubled its cellular  
348 uptake from ~75 ng/mg of protein to ~150 ng/mg of protein ( $p < 0.01$ ). Compared to uncoated  
349 bilosomes, HA enhanced the internalization of bilosomes and provided higher cellular uptake  
350 (~275 ng/mg protein) ( $p < 0.01$ ). This enhanced internalization ability was further confirmed by  
351 confocal laser scanning microscopy (CLSM). Bioavailability and biodistribution results of free  
352 tripterine, bilosomes, and HA-coated bilosomes showed that  $C_{max}$ ,  $AUC_{0-\infty}$  and the elimination  
353 time of HA-coated bilosomes was significantly different from free tripterine and bilosomes.  
354 The  $C_{max}$  and  $AUC_{0-\infty}$  of bilosomes ( $23.57 \pm 0.49$   $\mu$ g/mL,  $76.19 \pm 1.13$   $\mu$ g/mL.h) and coated  
355 bilosomes ( $25.24 \pm 0.57$   $\mu$ g/mL,  $112.19 \pm 0.85$   $\mu$ g/mL.h) were higher than the  $C_{max}$  of free  
356 tripterine ( $22.62 \pm 0.71$   $\mu$ g/mL,  $35.86 \pm 0.53$   $\mu$ g/mL.h). In addition, the elimination time of the  
357 drug decreased 2.13-fold using bilosomes and 3.13 fold using coated bilosomes. Encapsulation  
358 improved the resistance of tripterine in circulation. The relative bioavailability of tripterine  
359 in bilosomes and coated bilosomes was 480.3% and 799.9% respectively, compared to free  
360 tripterine [23].

361 Notoginsenoside R1 derived from *Panax notoginseng* poorly dissolves in water and  
362 cannot exhibit its biological functions sufficiently [58]. For the oral application of  
363 notoginsenoside R1, *in vitro* cellular uptake in Caco-2 cells and intestinal absorption of  
364 notoginsenoside R1-loaded NaGC-liposomes and liposomes were compared. NaGC-liposomes  
365 showed 1.5-fold higher cellular uptake compared to liposomes ( $p < 0.01$ ) so the presence of  
366 NaGC enhanced the cellular uptake of notoginsenoside R1.

367 Intestine absorption of notoginsenoside R1 was reported as  
368 duodenum>jejunum>ileum>colon. The  $P_{app}$  of notoginsenoside R1 was NaGC-liposomes>  
369 liposomes> notoginsenoside R1 solution for the same drug concentration (20  $\mu\text{g/mL}$  and 100  
370  $\mu\text{g/mL}$ ) and in the same section of the intestine ( $p < 0.05$ ). Addition of NaGC increased  $P_{app}$   
371 1.85-fold (20  $\mu\text{g/mL}$  drug) and 3.19 fold (100  $\mu\text{g/mL}$  drug) compared to free drug. NaGC-  
372 liposomes enhanced the  $C_{max}$  1.85-fold and  $T_{1/2}$  2.52-fold compared to liposomes. The presence  
373 of NaGC also improved  $AUC_{0-t}$  1.32-fold and 2.68-fold compared to liposomes and  
374 notoginsenoside R1 solution [30]. Loading into NaC-liposomes increased the relative  
375 bioavailability (16.4-fold) and elimination half-time ( $T_{1/2}$ ) of [6]-gingerol (~2.5-fold) which  
376 shows that [6]-gingerol can stay longer in the circulation [50]. Similarly, an *in vivo*  
377 pharmacokinetic study showed that while  $C_{max}$ ,  $AUC_{0-\infty}$  and  $MRT_{0-\infty}$  of all biflavonoids were  
378 increased,  $C_L$  decreased when loaded into NaDC-liposomes. Encapsulation did not show a  
379 significant difference in the  $T_{1/2}$  of biflavonoids. A pharmacokinetic study showed that the  
380 relative bioavailability of biflavonoids in the extract was enhanced between 191%-995% due  
381 to loading into NaDC-liposomes. Moreover, the cytotoxicity of the biflavonoids extract-loaded  
382 liposomes was evaluated on HT-29 cells. Encapsulation into NaDC-liposomes reduced the  
383 half-maximal inhibitory concentration ( $IC_{50}$ ) of the extract by 37% [31].

384 Ammonium glycyrrhizinate is isolated from *Glycyrrhiza glabra* and is used for  
385 supplements and medical purposes. Ammonium glycyrrhizinate shows an anti-inflammatory  
386 effect against inflammation on the skin [59]. The applicability of the ammonium glycyrrhizate-  
387 loaded NaC-liposomes for topical application was investigated and at 24 h, only 40% of  
388 ammonium glycyrrhizate was released from vesicles and showed pseudo-first-order kinetics  
389 [27]. Permeation and skin tolerability studies were done with the topical application of vesicles  
390 on humans. 134.9  $\mu\text{g/cm}^2$  ammonium glycyrrhizate penetrated through the *stratum corneum*  
391 and epidermis and permeation through the skin layers showed zero-order kinetics. For topical  
392 application, drugs need to pass through the *stratum corneum* and epidermis as well as protect

393 their structure [27]. Erythema index ( $\Delta EI$ ) values measured were below 4.5 and show that  
394 ammonium glycyrrhizate-loaded NaC-liposomes can be tolerated by human skin without any  
395 toxicity. In order to determine the effect of NaC on the deformability of  
396 dipalmitoylphosphatidylcholine (DPPC) liposomes, the deformability of liposomes and NaC-  
397 liposomes (DPPC/NaC 4:1) were compared using the vesicle-pore model where both  
398 liposomes are pushed to pass through narrow pores and deformability parameters are measured  
399 [28]. According to the results, NaC improves the flexibility of liposomes and decreases the  
400 deformability through pores.

401 Resveratrol (3,5,4'-trihydroxytrans-stilbene) is a well-known natural antioxidant [60].  
402 Spreadability, skin permeation, and deposition of a resveratrol-loaded NaC-transfersome  
403 cream was compared with a conventional resveratrol cream [26]. The spreadability of the  
404 vesicular cream was  $47.90 \pm 3.65\%$  spread by weight (conventional resveratrol cream:  
405  $39.14 \pm 1.59\%$ ). The permeation flux of the resveratrol-loaded NaC-transfersomes cream and  
406 conventional cream was determined as  $4.95 \pm 0.69$  and  $2.70 \pm 0.73 \mu\text{g}/\text{cm}^2$  respectively (at 24 h).  
407 When deposition in the skin layer was evaluated, drug deposition of the NaC-transfersome  
408 cream ( $335.2 \pm 4.12 \mu\text{g}/\text{cm}^2$  with skin retention:  $70.16 \pm 0.87\%$ ) was reported to be significantly  
409 higher than conventional cream ( $67.12 \pm 19.63 \mu\text{g}/\text{cm}^2$  with skin retention:  $14.05 \pm 4.11\%$ ). *In*  
410 *vitro* skin permeation and deposition of phenylethyl resorcinol-loaded NaDC-transfersomes  
411 were investigated. Compared to liposomes (skin permeation:  $20.65 \mu\text{g}/\text{cm}^2$ , accumulation:  
412  $28.18 \mu\text{g}/\text{cm}^2$ ), transfersomes (skin permeation:  $72.66 \mu\text{g}/\text{cm}^2$ , accumulation:  $71.21 \mu\text{g}/\text{cm}^2$ )  
413 provided higher permeation and accumulation of phenylethyl resorcinol in newborn pig skin  
414 ( $p < 0.05$ ) [40]. According to the results, the enhanced permeation ability of the resveratrol-  
415 loaded NaC-transfersome cream [26] and phenylethyl resorcinol-loaded NaDC-transfersomes  
416 [40] is possibly due to the presence of BSs that can enhance the penetration [16]. Sodium  
417 pravastatin and naringenin-loaded NaDC-transfersomes were optimized and NaDC showed a  
418 significant impact on the cumulative % permeated ( $< 0.0001$ ). When NaDC concentration (20-  
419 60 mg) increased, cumulative (%) permeated increased from 37% to 59%. Results also showed  
420 that NaDC did not show a considerable effect on ALT ( $p: 0.8767$ ) and MDA ( $p: 0.4490$ ) levels  
421 [25]. The deformability of phenylethyl resorcinol-loaded NaDC-transfersomes and liposomes  
422 was compared before and after filtering through a 200 nm polycarbonate filter [40]. While  
423 transfersomes ( $D_h: 398.37 \pm 9.82 \text{ nm}$ , PDI:  $0.06 \pm 0.08$ ) passed through the filter, liposomes ( $D_h:$   
424  $600.23 \pm 11.92 \text{ nm}$ , PDI:  $0.21 \pm 0.07$ ) couldn't pass because of their rigid structure. After  
425 filtration, the  $D_h$ , PDI, and DI of NaDC-transfersomes were  $371.97 \pm 8.72 \text{ nm}$ ,  $0.13 \pm 0.09$ , and



426 6.63% respectively. These results confirm that incorporating NaDC into the lipid bilayers  
427 enhances the flexibility of vesicles.

428 Catechin is a widely studied phenolic compound that shows mainly antioxidant activity.  
429 *In vitro* skin permeation performance of catechin-loaded NaTC-hexosomes was tested on the  
430 skin of newborn pigs and compared with catechin-loaded hexosomes and vesicles. Due to the  
431 presence of NaTC, NaTC-hexosomes showed better permeation through skin layers and higher  
432 drug accumulation in the skin layers including *stratum corneum*:~14%, epidermis:~7%,  
433 dermis:~8%, and receptor compartment:~9%. While liposomes showed max accumulation  
434 (~9%) in the dermis, no drug accumulation was detected in the receptor compartment.  
435 Hexosomes showed maximum of ~5% drug accumulation both in the *stratum corneum* and the  
436 receptor compartment [41].

437 Loading of active compounds into LSAS prolonged the absorption time and retention  
438 time compared to free active compounds. The addition of BSs to the LSAS further increases  
439 the absorption time and retention time of the active compounds compared to BS-free LSAS.  
440 Due to the presence of BSs, the  $C_{max}$  and  $AUC_{0-\infty}$  of circulating active compounds are  
441 increased, therefore improved relative bioavailability has been reported (Figure 3). In addition,  
442 BSs increase the elasticity of particles and enhance the penetration and accumulation of active  
443 compounds [48].

## 444 **5. Effect of bile salts on the biological activity of phytochemicals loaded into lipid** 445 **self-assembled systems**

446 Phytochemicals show numerous biological activities such as antioxidant, anti-  
447 inflammatory, anti-tumour, anti-arthritis, anti-tyrosinase activity, etc (Figure 4). LSAS can  
448 provide increased solubility, enhanced diffusion properties, and stability to phytochemicals and  
449 promote these biological functions [61].

450 *In vivo* anti-tumour activity of biflavonoids extract-loaded liposomes was determined on  
451 HT-29 colon cancer cells. NaDC-liposomes almost doubled the antitumor activity of  
452 biflavonoid extracts ( $p < 0.001$ ) without systemic toxicity [31]. In the same way, the anti-tumour  
453 activity of [6]-gingerol (100  $\mu\text{g/mL}$ , 15% inhibition) on HepG2 cells was improved  
454 significantly due to loading into NaC-liposomes (inhibition rate:~100%) ( $p < 0.01$ ) [50]. In  
455 addition, the *in vitro* anti-tumour effect of curcumin-loaded polyethylenimine-ethosomes (PEI)  
456 and doxorubicin-loaded NaC-ethosomes (7:3 v/v) on B16 cells was determined. PEI-loaded

457 NaC-ethosomes and doxorubicin-loaded NaC-ethosomes showed similar inhibition (~55%)  
458 against B16 cells. PEI-ethosomes and doxorubicin-loaded NaC-ethosomes (7:3 v/v) exerted  
459 stronger inhibition against B16 cells and cell survival was decreased to ~30% ( $p < 0.01$ ). Similar  
460 results were also seen for *in vivo* anti-tumor effects. Melanoma tumor inhibition rates were  
461 21.9% and 35.5% for curcumin-loaded PEI and doxorubicin-loaded NaC-ethosomes,  
462 respectively. PEI-ethosomes and doxorubicin-loaded NaC-ethosomes (7:3 v/v) inhibited  
463 46.38% of Melanoma tumor ( $p < 0.01$ ) [62]. Loading the phytochemicals in the BS-containing  
464 system enhanced their anti-tumour activity significantly.

465 The anti-inflammatory activity of ammonium glycyrrhizate-loaded NaC-liposomes was  
466 investigated in human volunteers [27]. Free compound and liposomes were applied topically  
467 on chemically (0.2% w/v methyl nicotinate) stimulate erythema and NaC-liposomes showed  
468 ~10-30-fold increase in anti-inflammatory activity compared to an equivalent of ammonium  
469 glycyrrhizate solution.

470 Anti-arthritic activity of tripterine-loaded HA-coated NaDC-bilosomes on arthritic mice  
471 was studied [23]. Mediators that show the inflammation level was measured and coated and  
472 uncoated bilosomes were compared with the model control. While bilosomes reduced the  
473 concentration of mediators in serum approximately double times compared to model control,  
474 HA-coated bilosomes reduced mediators approximately fourfold compared to model control  
475 ( $p < 0.01$ ).

476 Anti-tyrosinase activity of phenylethyl resorcinol-loaded NaDC-transfersomes and  
477 liposomes were evaluated after topical application on pig skin. Kojic acid was chosen as a  
478 positive control. Liposomes ( $95.54 \pm 0.13\%$ ) and NaDC-transfersomes ( $91.09 \pm 1.23\%$ ) showed  
479 better anti-tyrosinase activity than kojic acid ( $87.35 \pm 0.76\%$ ). Accumulation of phenylethyl  
480 resorcinol from NaDC-transfersomes ( $80.47 \pm 0.22\%$ ) was nearly the same as liposomes  
481 ( $79.53 \pm 0.45\%$ ). Tyrosinase activity and melanin content in B16 melanoma cells was also  
482 measured after NaDC-transfersome and liposome application. Compared to liposomes  
483 (tyrosinase activity: 71.27% and melanin content: 82.11%), NaDC-transfersomes decreased  
484 tyrosinase activity to 64.36% and melanin content to 64.85%. NaDC-transfersomes showed  
485 effective skin lightening properties [40].

486 The *in vitro* antioxidant activity of catechin-loaded NaTC-hexosomes was measured.  
487 Using the 1,1-diphenyl-2-picrylhydrazyl (DPPH) assay, NaTC-hexosomes, hexosomes,

488 vesicles, and catechin in methanol solution could all inhibit 88% of DPPH radicals [41].  
489 Similarly, the antioxidant activity of a resveratrol solution (95%) and resveratrol-loaded NaC-  
490 transfersomes (~92%), were not significantly different ( $p>0.05$ ). In addition, empty  
491 transfersomes also showed the ability to inhibit ~21% of 25  $\mu$ M DPPH radicals [26]. The  
492 presence of BSs in formulations and loading into BSs-systems did not show a significant effect  
493 on the antioxidant activity of catechin and resveratrol.

494 In general, the presence of BSs in LSAS increased the biological activity of the  
495 phytochemicals. Although increased permeation and accumulation of catechin and increased  
496 skin retention of resveratrol were reported due to loading into BS-LSAS, no significant changes  
497 were observed in the biological activities of resveratrol and catechin.

## 498 6. CONCLUSION

499 This review focused on the effect of BSs on phytochemicals-loaded-LSAS. The presence  
500 of BSs in LSAS showed effects on the morphology, encapsulation efficiency, stability,  
501 bioavailability, and biological activities of phytochemicals. Up to a specific concentration that  
502 differs depending on the BSs, the  $D_h$  of LSAS decreased as the concentration of BSs increased.  
503 This was due to BSs reducing the surface tension of the vesicles and stabilize the bilayers.  
504 While the EE(%) of hydrophilic phytochemicals is related to the trapping ability of the lipid  
505 membrane to hydrophilic phytochemicals dissolved in aqueous phases, the EE(%) of  
506 hydrophobic phytochemicals is mainly related to the interaction between the lipid membrane  
507 and phytochemicals. The addition of BSs to the formulations affected the EE(%) significantly.  
508 Increased BS concentration up to a specific concentration, resulted in increased EE(%). Upon  
509 further increase in BS concentration, BSs showed a solubilisation effect resulting in the  
510 transition from vesicles to micelles and disruption of the vesicles. It has been reported that  
511 EE(%) may decrease due to membrane pore formation caused by BSs which may cause leakage  
512 of phytochemicals. *In vitro* release studies showed that encapsulation of phytochemicals into  
513 BS-LSAS slows their release. The addition of negative charged BSs improved the stability of  
514 LSAS due to an increase negative  $\zeta$  potential that increased repulsion between particles.  
515 Pharmacokinetics studies reported that the addition of BSs increased the absorption time and  
516 plasma concentration and decreased the elimination time of phytochemicals. BSs also  
517 enhanced the flexibility of particles and favored the penetration and accumulation of  
518 phytochemicals. Increased relative bioavailability of phytochemicals was reported due to the  
519 loading of the compounds into BS-LSAS. With enhanced bioavailability, in general,

520 phytochemicals showed higher biological activity. The application of LSAS has potential in  
521 the food, beverage, and pharmaceutical industries. Hence, exploring the possible use of BSs in  
522 LSAS for the delivery of phytochemicals was outlined.

### 523 **Author Contributions**

524 AC: Conceptualization, Investigation, Identification, Methodology, Screening, Writing-  
525 Original draft preparation; AIT: Project Administration, Supervision, Validation, Writing-  
526 Review and editing; ARM: Project Administration, Supervision, Validation, Writing-Review  
527 and editing.

### 528 **Conflicts of interests**

529 The authors declare no competing financial interest.

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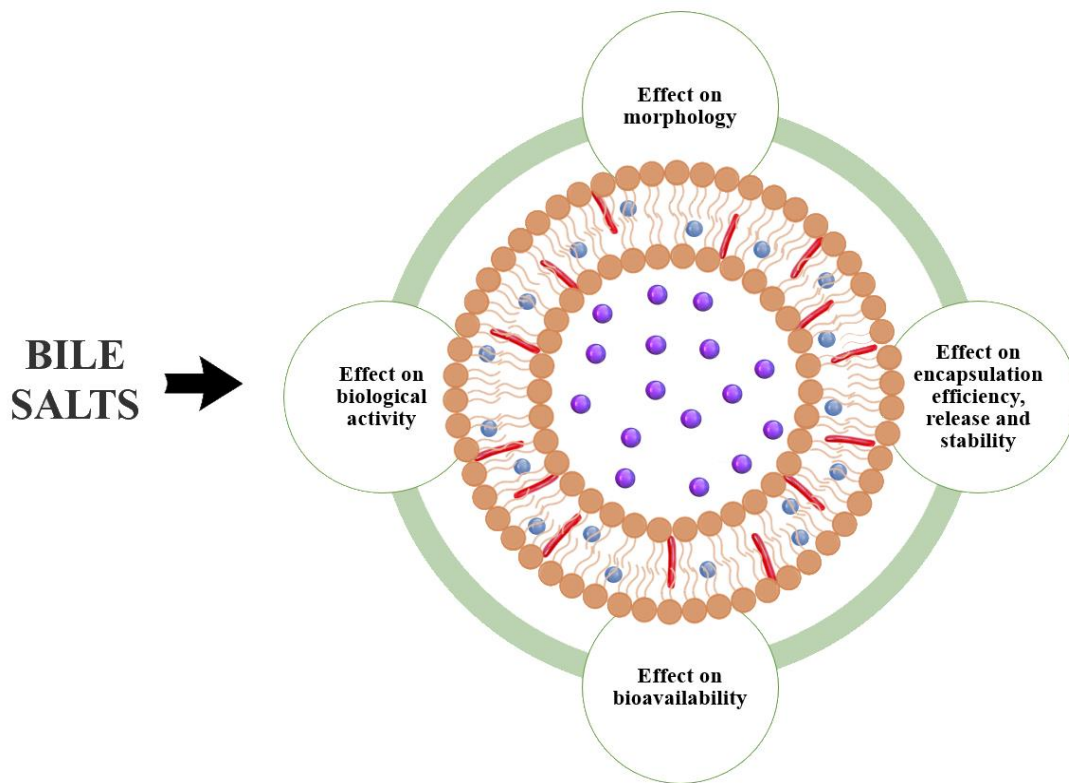
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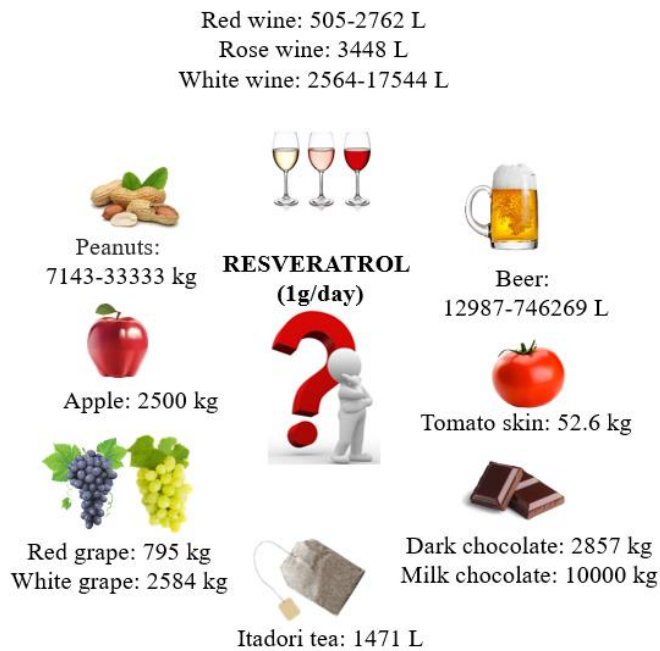
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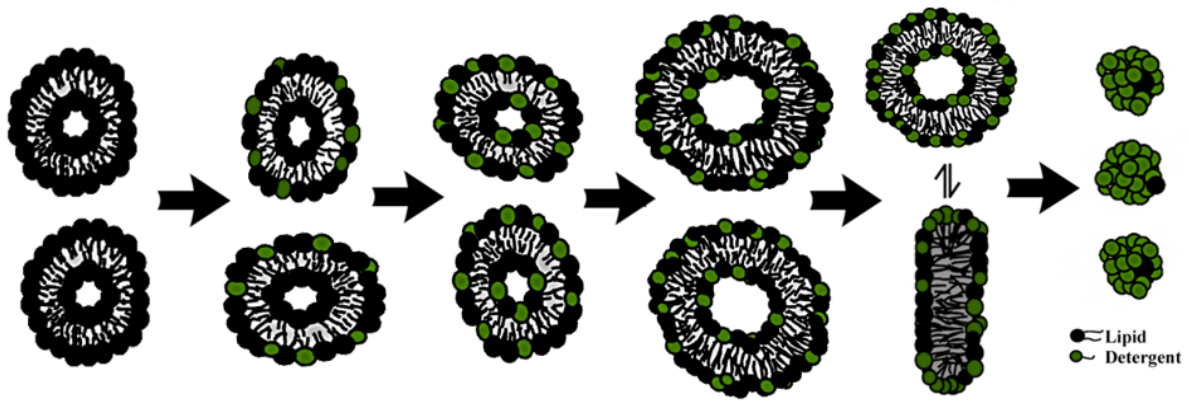
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766 **Graphical abstract**

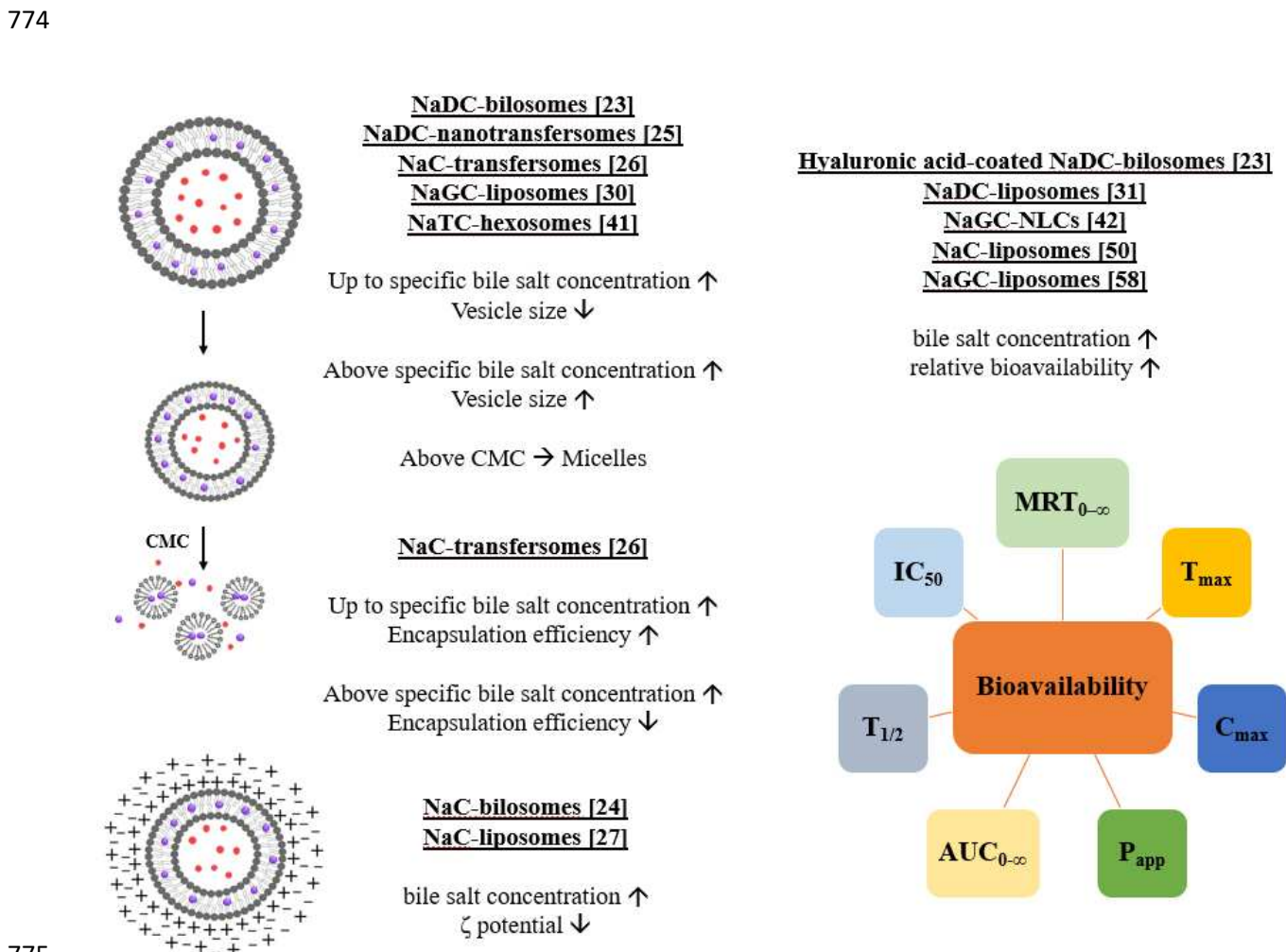


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768 **Figure 1.** Quantities of food and beverages **that** must be consumed to reach therapeutic  
 769 doses **of resveratrol**. If a person intends to ingest 1 g of resveratrol each day, this would require  
 770 consuming the depicted quantities of foods or beverages [3].

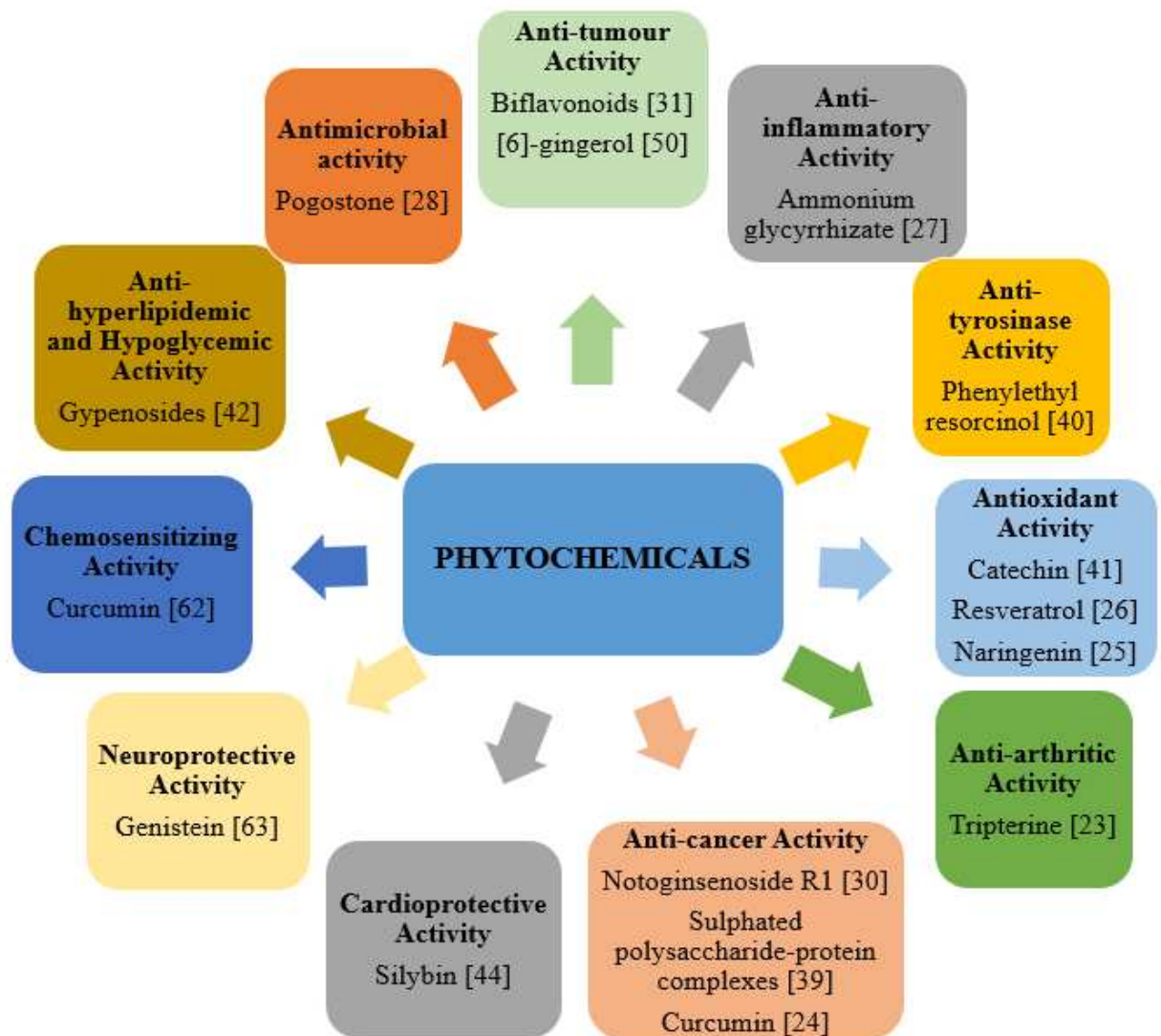


771  
 772 **Figure 2.** Schematic representation of the transition from vesicles to mixed micelles upon an  
 773 increase in bile salt concentration. Adapted from reference [18].



776 **Figure 3.** Effect of presence of bile salts on  $D_h$ , EE(%),  $\zeta$  potential of lipid self-assembled  
 777 systems and relative bioavailability of phytochemicals.

778  
 779



782 **Figure 4.** Biological activities of phytochemicals.

783 **Table 1.** A summary of the self-assembled system for delivery of phytochemicals.

Self-assembled System	Application	Active Compound	Biological Activity	Bile Salts	Composition	Reference
<b>Liposomes</b>	Topical Application	Ammonium glycyrrhizinate	Anti-inflammatory Activity	NaC	SPC: 88mg NaC: 12 mg in 6 mL water/EtOH Ammonium glycyrrhizate : 0.3% w/v of the drug in the lipid	[27]
<b>Liposomes</b>	Oral Application	Doxorubicin and Silybin	Cardioprotective Activity	Cholic acid	PC: 40 mg/mL DSPE-PEG- Cholic acid : 3 mg/mL Chol: 8 mg/mL Silybin: 5 mg/mL Doxorubicin: 1 mg/mL	[44]
<b>Liposomes</b>	Oral Application	Biflavonoids extract from <i>Selaginella doederleinii</i>	Anti-tumour Activity	NaDC	Optimum: SPC: 30 mg/mL Chol: 3 mg/mL NaDC: 3 mg/mL Extract: 3 mg/mL Isomalto-oligosaccharides/lipid, 2:1, w/w)	[31]
<b>Liposomes</b>	Oral Application	[6]-gingerol	Anti-tumour Activity	NaC	Optimum: PC: 75 mg Isopropyl myristate: 0.08 mg NaC: 0.04 mg [6]-gingerol: 5 mg	[50]
<b>Liposomes</b>	Oral Application	Notoginsenoside R1	-	NaGC	Optimum: SPC: 50 mg Chol: 10 mg DSPE-PEG2000: 10 mg NaGC: 6 mg Notoginsenoside R1: 8 mg	[30]
<b>Liposomes</b>	Topical Application	Pogostone	-	NaC	SPC, NaC, Chol, Vitamin E, Pogostone: 0.07%	[28]
<b>Transfersomes</b>	Oral Application	Genistein	Antioxidant and Neuroprotective Activity	NaDC	PC: 693 mg NaDC: 297 mg Genistein: 10 mg in 50 mL water	[63]



<b>Transfersomes</b>	-	Resveratrol	Antioxidant Activity	NaC	Optimum: SPC: 90.7 mg NaC: 12.5 mg Resveratrol: 100 mg in 5 mL PBS buffer (pH 7.4)	[26]
<b>Transfersomes</b>	Topical Application	Phenylethyl resorcinol	Anti-tyrosinase activity	NaDC	SPC: 3% (w/v) CHOL: 0.5% (w/v) phenylethyl resorcinol: 0.5% (w/v) NaDC: 15% (w/w) in water: up to 100% (v/v)	[40]
<b>Transfersomes</b>	Oral Application	Sodium pravastatin and Naringenin	Anti-hyperlipidemic and Antioxidant Activity	NaDC	Desirability values of the numerical optimization: Omega-3 PC: 0.7908 NaDC: 1 Naringenin: 1	[25]
<b>Bilosomes</b>	-	Tripterine	Anti-arthritis Activity	NaDC	SPC: 80 mg DOTAP: 20 mg NaDC: 2 mg/mL Tripterine: 10 mg HA: 10 mg in water: 10 mL	[23]
<b>Bilosomes</b>	Oral Application	Sulphated polysaccharide-protein complexes of <i>Enteromorpha intestinalis</i>	Anti-cancer activity	NaC, NaDC and NaTDC	Chol/ Span 65: 1:5 molar ratio Active compounds: 10 mg NaC: 0.5 M in 10 mL saline solution (0.9% w/v NaCl)	[39]
<b>Surface-modified Bilosomes</b>	Intravenous and Topical Application	Methylene blue and Curcumin	Anticancer Activity	NaC	Optimum: L- $\alpha$ -PC: 1.00 wt% Chol: 0.30 wt% Pluronic P123: 0.60 wt% NaC: 0.50 wt% in water: 97.60 wt%	[24]
<b>Hexosomes</b>	Topical Application	Catechin	Antioxidant Activity	NaC, NaGC and NaTC	Optimum: GMO: 3.0 wt% Oleic acid: 0.5 wt% NaTC: 0.3 wt% Pluronic F108: 0.3 wt% in water: 95.9 wt%	[41]
<b>Ethosomes</b>	Transdermal Application	Doxorubicin and Curcumin	Cytotoxic and Chemosensitizing Activity	NaC	Lecithin: 100 mg Chol: 10 mg PEI or SC: 5 mg in 10 ml of water/ethanol	[62]
<b>Nanostructured Lipid Carriers</b>	Oral Application	Gypenosides	-	NaGC	Oleoyl macrogolglycerides, glyceryl monolinoleate, glycerol monostearate, SPC, Gypenosides, NaGC, Polysorbate 80	[42]

