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1 Title: Progressive development of ocean anoxia in the end-Permian pelagic Panthalassa

2

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17

18 Keywords: carbon isotope, iron speciation, redox-sensitive metals, mass extinction, ocean  
19 anoxia, pelagic deep sea, Permian-Triassic boundary

20 Abstract

21 The end-Permian mass extinction (EPME) has been linked with the widespread  
22 development of oxygen-poor oceanic conditions. However, information on the spatial extent  
23 of anoxia in the Panthalassa super-ocean has been limited. This study reports oceanic redox  
24 records from a deep-sea chert succession (the Waiheke 1 section, WHK 1, New Zealand) that  
25 was located in southern mid-latitudes of Panthalassa. High-resolution carbon isotope ( $\delta^{13}\text{C}$ )  
26 correlation between Waiheke and the Permian-Triassic boundary (PTB) type section indicates  
27 that the EPME is recorded in a thin black claystone interbedded between siliceous mudstone  
28 beds at WHK 1. Pyrite-dominated enrichment in highly reactive iron, coupled with elevated  
29 U/Al and Mo/Al ratios, are prevalent through this black claystone bed and the overlying  
30 Permo-Triassic transition strata, suggesting the development of euxinic water column  
31 conditions. Similar redox variations across the EPME horizon have been reported from other  
32 Panthalassic deep-sea PTB sections. Comparison with these PTB sections indicates that  
33 euxinic conditions were widespread in low-latitude regions of the Panthalassan ocean, and

34 such conditions developed earlier than in mid-latitude settings, up to 100,000 years before the  
35 EPME. This suggests there was a gradual expansion of ocean anoxia from low to middle-high  
36 latitude regions during the Permo-Triassic transition. The extent of ocean anoxia resulted in a  
37 decrease in the seawater inventory of redox sensitive trace metals (e.g., Mo), which is evident  
38 in the earliest Triassic strata of the studied section and other PTB sections. Panthalassic  
39 anoxia during the EPME coincides with extreme climate warming and the associated effects  
40 (e.g., changes in ocean circulation, marine eutrophication intensified by terrestrial  
41 weathering) were likely critical triggers for ocean deoxygenation.

42

## 43 1. Introduction

44 The most significant biotic catastrophe of the Phanerozoic, the end-Permian mass extinction  
45 event (EPME), was associated with the widespread development of oceanic anoxia (e.g.,  
46 Wignall and Twitchett, 1996; Isozaki, 1997). Previous research has demonstrated that many  
47 shallow water and deep oceanic regions of Late Permian and Early Triassic age experienced  
48 anoxic conditions (e.g., Wignall and Twitchett, 2002; Kaiho et al., 2016). Organic molecular  
49 fossils (biomarkers) of anaerobic green sulphur bacteria have shown that euxinia (anoxic,  
50 sulphidic conditions) extended into the photic zone in the paleo-Tethys (e.g., Grice et al.,  
51 2005, locality 3 in Figure 1A) and Panthalassan oceans (Hays et al., 2007, locality 5 in Figure  
52 1A). Enrichments in pyrite and redox-sensitive trace elements have also been reported from  
53 PTB sections of paleo-Tethys (e.g., Lei et al., 2017, Xiang et al., 2020, Shen et al. 2016,  
54 Clarkson et al., 2016), pelagic Panthalassa (locality 2 in Figure 1A; Algeo et al., 2011;  
55 Takahashi et al., 2014; Onoue et al., 2021), and the North Pangean margin (e.g., locality 4 in  
56 Figure 1A; Grasby et al., 2009; Schobben et al., 2020), indicating the widespread  
57 development of anoxia/euxinia. The extent of oxygen-poor conditions may have caused a  
58 decrease in both seawater sulphate concentrations (Song et al., 2014; Schobben et al., 2017)  
59 and trace elements such as Mo (e.g., Takahashi et al., 2014; Xiang et al., 2020), due to  
60 enhanced pyrite burial and trace metal drawdown (Takahashi et al., 2014; Schobben et al.,  
61 2015). The extent of ocean anoxia also changed the U isotope ratio of seawater, as recorded  
62 in marine carbonate ( $\delta^{238}\text{U}_{\text{carbonate}}$ ) (Brennecka et al., 2011, Lau et al., 2016; Zhang et al.,  
63 2018ab, 2020).

64 While anoxia was clearly widespread, there remains a significant absence of  
65 environmental records from the Panthalassa (Fig. 1A). Since most of the sedimentary record  
66 of Permian-Triassic Panthalassic seafloor has been subducted, limited material is available to

67 allow reconstruction of redox conditions, particularly in terms of revealing temporal and  
68 spatial variability in oceanic redox during the EMPE. Previous research has, however,  
69 focussed on Panthalassic PTB sections from Japanese accretionary complexes. These formed  
70 in low latitude central Panthalassa (locality 2 in Figure 1A), based on measured paleo-  
71 latitudes of Middle Triassic chert from the Mino Belt of central Japan (cf. paleomagnetic  
72 studies by Ando et al., 2001 and Oda and Suzuki, 2000). It was at least one hundred million  
73 years before these sediments were accreted in the Middle to Late Jurassic (Mastuda and  
74 Isozaki, 1991; Takahashi et al., 2014). Since oceanic plates migrate at a rate of several cm/yr,  
75 this indicates that the Japanese PTB sections were originally thousands of kilometres from  
76 land.

77 According to reports from Japan (e.g., Isozaki, 2009, Muto et al., 2020), the EPME  
78 horizon occurs at the transition from Permian radiolarian chert to siliceous claystone and  
79 black claystone layers. Based on the concentrations of redox-sensitive trace elements (Mo, U  
80 and V) and framboidal pyrite populations in the PTB sections, a shift to oxygen-poor  
81 conditions during the EPME has been suggested, with the specific development of either  
82 anoxic through to euxinic bottom waters, or an oxygen minimum zone setting (Wignall et al.,  
83 2010; Algeo et al., 2011; Takahashi et al., 2014; Onoue et al., 2021).

84 Southern Panthalassa PTB sections have also been documented in the accretionary  
85 complex on the North Island of New Zealand (Spörli et al., 2007a; Hori et al., 2011; Grasby  
86 et al., 2021), where data from Lower Triassic chert suggests a paleolatitude of ca. 34°S  
87 (Kodama et al., 2007). The Permian-Triassic deposits accreted during the Late Triassic,  
88 suggesting the original depositional location was close to Gondwana (Spörli et al, 2007b;  
89 locality 1 in Figure 1A). Two sections from Arrow Rocks and Waiheke Island (Fig.1B)  
90 indicate that black, bedded chert and claystone occur around the EPME, similar to their  
91 occurrence in the Japanese sequences. Hori et al. (2007) measured trace element enrichments  
92 (Mo and U) in the Griesbachian black chert layers of the Arrow Rocks (Unit 2) and identified  
93 an anoxic water column event they termed “OAE  $\alpha$ ” (they described an “OAE $\beta$ ” in the  
94 Dienerian). However, this PTB section has minor faults around the basal part of the black  
95 chert layers, making the continuous redox history in relation to the EPME and PTB unclear.

96 The PTB section on Waiheke Island (Waiheke section) has a continuous succession across  
97 the PTB, which has been dated using conodonts (Hori et al., 2011). Organic carbon isotope,  
98 sulphide sulphur isotope, and trace element profiles through this section suggest the  
99 development of anoxic-sulphidic water column conditions during the EPME (Hori et al.,  
100 2007; Takahashi et al., 2013; Grasby et al., 2021). Here, we present new, high-resolution

101 paleoenvironmental records for the Waiheke section, incorporating multiple paleoredox  
102 proxies, including iron speciation and redox-sensitive trace element systematics. We aim to  
103 reveal how ocean anoxia developed prior to and after the EPME, as well as the associated  
104 changes in seawater chemistry. In addition, we present a compilation of pelagic Panthalassic  
105 PTB records to investigate the spatial and temporal development of redox conditions in the  
106 Permian-Triassic Panthalassa.

107

## 108 2. Geological setting

109 The Waiheke PTB section is an intertidal platform section in Island Bay, north-western  
110 Waiheke Island, North Island, New Zealand (Fig. 1B). It belongs to the Kiripaka Formation  
111 of the Waipapa composite terrain (Schofield, 1974, 1979), which is the accretionary complex  
112 forming part of the basement of New Zealand. The sedimentary sequence comprising the  
113 Waiheke section is approximately 10 m thick and divided into six litho-units (Unit A to F;  
114 Hori et al., 2011). This study focuses on the lowest part of these units (Unit A), in the  
115 Waiheke 1 section (abbreviated as “WHK 1” hereafter). It mainly consists of pale green  
116 tuffaceous claystone and mudstone towards the base, followed by black and yellowish-black  
117 chert with black claystone (Fig. 2).

118 The pale green siliceous and tuffaceous claystone and mudstone in the lower half of Unit  
119 A are composed of mm- to cm-scale alternations of siliceous microfossil enriched layers  
120 (radiolarians and sponges) and muddy layers (Fig. 3A). The black claystone consists of fine-  
121 grained clay materials and organic matter (Fig. 3B). Although siliceous microfossils are  
122 observed in the black claystone, fossils are rare and generally small (up to several  $\mu\text{m}$  in  
123 diameter). Black and black-yellow cherts in the upper part of WHK 1 also show mm- to cm-  
124 scale microfossil enriched layers and muddy layers (Figs 3C, 3D). The size of visible  
125 microfossils are similar to those from the pale green beds in the lower part of the study  
126 section. 88 single beds have been identified and labelled as Beds 1 to 88 (Hori et al., 2011;  
127 Takahashi et al., 2013). The conodont occurrences indicate that the section ranges from the  
128 uppermost Permian to the lowermost Triassic (Figs. 2 and 4; Hori et al., 2011). *Hindeodus* sp.  
129 and *Neogondolella carinata* were obtained from Bed 29 of the black-yellow chert. Above this  
130 horizon, *Ng. carinata* and *Sweetospathodus* cf. *kummeli* occur from Bed 40. These index  
131 fossils indicate that Bed 29 is Induan, and Bed 40 extends into the lower to upper part of  
132 Induan (uppermost Griesbachian or even the Dienerian). Ten previously published organic  
133 carbon isotope ( $\delta^{13}\text{C}_{\text{org}}$ ) analyses demonstrate a negative excursion from Bed 14 to Bed 22,

134 indicating the uppermost Permian carbon isotopic signal corresponding to the end-Permian  
135 mass extinction event (Hori et al., 2011). Based on this information, we place the Permian-  
136 Triassic transitional zone at this level. Here, we provide further  $\delta^{13}\text{C}_{\text{org}}$  data and discuss the  
137 timescale issue in more detail.

### 138 3. Methods

#### 139 *3.1. Organic carbon isotope analysis*

140 We examined organic carbon isotope ratios of bulk chert and claystone samples from 28  
141 horizons of WHK 1, using samples with no visible veinlets or staining. After washing with  
142 Ellix® water, all samples were ground into fine powder using an agate mill. The powdered  
143 samples were decarbonated with 10% HCl. Organic carbon isotopic compositions and  
144 organic carbon contents were measured using an elemental analyser (Flash EA 1112) coupled  
145 with a Thermo-Finnigan Delta plus Advantage isotope ratio mass spectrometer at the Center  
146 for Advanced Marine Core Research, Kochi University, Japan. Isotopic measurements were  
147 repeated up to four times for each horizon to check reproducibility. The carbon contents were  
148 calculated using standard sulfanilamide, and the precision of organic carbon isotope ( $\delta^{13}\text{C}_{\text{org}}$ )  
149 analyses was better than 0.1‰. Organic carbon isotope measurements were calibrated using  
150 histidine and alanine obtained from Sugito Ltd. Lab. All isotopic results are reported in  
151 conventional delta ( $\delta$ ) notation, defined as per mil (‰) deviation from the Pee Dee belemnite  
152 (PDB) standard value. Here we use the average values calculated from repeat (up to 4 times)  
153 measurements of the same sample.

154

#### 155 *3.2. Major and trace element analysis*

156 Major and trace elements were measured by inductively coupled plasma atomic emission  
157 spectrometry (ICP-AES; SPS3500, SII nanotechnology) and inductively coupled plasma  
158 mass spectrometry (ICP-MS; iCAPQc, Thermo scientific) at the Department of Earth and  
159 Planetary Science, University of Tokyo. For these analyses, 0.5 g of sample powder was  
160 treated with 4 mL of  $\text{HClO}_3 + \text{HNO}_3$  (1:1 mixture) in PTFE beakers for 45 min at 120°C,  
161 followed by two treatments with 4 mL of  $\text{HClO}_3 + \text{HF}$  (1:1 mixture) for 30 min at 120°C.  
162 The residues from these acid treatments were re-dissolved with 5 mL  $\text{HNO}_3$  and diluted to  
163 1:2000 using ultra pure water. Indium (In) and bismuth (Bi) were used as internal standards  
164 for the ICP-MS analyses. Working standards were prepared from a series of SPEX Multi-  
165 Element Plasma Standards supplied by SPEX Industries (SPEX). Repeated analysis on the  
166 reference material (GSJ-JR-1; rhyolite reference material supplied by Geological Survey of

167 Japan) confirms that values are reproducible to within 1.8% for Al, 0.8% for Fe, 12% for U,  
 168 and 9% for Mo, respectively (as relative standard deviation: RSD).

169

### 170 3.3. Normalization

171 To assess the enrichment or depletion of redox sensitive elements relative to typical  
 172 detrital material, enrichment factors ( $X_{EF}$ ) were calculated as follows (e.g., Tribovillard et al.,  
 173 2006):

$$174 \quad X/Al = X_{\text{sample}} \text{ (ppm)} / Al_{\text{sample}} \text{ (wt \%)} \quad (1)$$

$$175 \quad X_{EF} = (X_{\text{sample}} / Al_{\text{sample}}) / (X_{\text{PAAS}} / Al_{\text{PAAS}}) \quad (2)$$

176 Here,  $X_{\text{sample}}$ ,  $Al_{\text{sample}}$ ,  $X_{\text{PAAS}}$  and  $Al_{\text{PAAS}}$  are the weight concentrations of element X or Al  
 177 in the sample or in PAAS (the post-Archean average shale; Tayler and McLennan, 1985). To  
 178 compare our geochemical data with other PTB sections, Fe/Al, U/Al and Mo/Al ratios were  
 179 also calculated and compared with literature data (Algeo et al., 2011; Takahashi et al., 2014).

180

### 181 3.4. Iron speciation

182 A sequential extraction targeting different operationally defined Fe pools, including  
 183 carbonate ( $Fe_{\text{carbonate}}$ ), ferric (oxyhydr)oxides ( $Fe_{\text{OX}}$ ) and magnetite ( $Fe_{\text{magnetite}}$ ), was  
 184 performed on a split of 0.07–0.09 g of powdered sample (Poulton and Canfield, 2005;  
 185 Poulton, 2021).  $Fe_{\text{carbonate}}$  was first extracted using Na-acetate at pH 4.5 and 50 °C for 48 h,  
 186 followed by  $Fe_{\text{OX}}$  using Na-dithionite for 2 hours at room temperature, and finally  $Fe_{\text{magnetite}}$   
 187 was determined with an ammonium oxalate extraction for 6 h. Dissolved iron concentrations  
 188 (for  $Fe_{\text{carbonate}}$ ,  $Fe_{\text{OX}}$  and  $Fe_{\text{magnetite}}$  pools) were measured by atomic absorption spectrometry  
 189 (AAS; Thermo ice 3000 at the School of Earth and Environment, University of Leeds).  
 190 Replicate extractions of an international reference material (WHIT; Alcott et al., 2021) gave a  
 191 RSD of <5% for each Fe pool.

192 Pyrite iron ( $Fe_{\text{pyrite}}$ ) was determined stoichiometrically based on  $Ag_2S$  precipitates formed  
 193 after a boiling HCl and chromous chloride distillation (Canfield et al., 1986). All samples  
 194 were tested for the presence of acid volatile sulphide (AVS; predominantly FeS) via an initial  
 195 6 M HCl distillation, but in all cases AVS was below detection (<0.001 wt %). Replicate  
 196 pyrite extractions gave a RSD of <5%. The sum of these measured Fe pools defines the total  
 197 concentration of highly reactive Fe ( $Fe_{\text{HR}}$ ):

$$198 \quad Fe_{\text{HR}} = Fe_{\text{carbonate}} + Fe_{\text{OX}} + Fe_{\text{magnetite}} + Fe_{\text{pyrite}} \quad (1)$$

199

200 A 1 h 0.5 N HCl extraction was also conducted on ~0.5\_g splits of powdered samples to  
 201 assess for potential oxidation during weathering (see Matthews et al., 2017). This treatment  
 202 targets poorly crystalline ferric oxide minerals (Fe(III)<sub>HCl</sub>), such as ferrihydrite, in addition to  
 203 reduced Fe phases such as Fe<sub>AVS</sub> and potentially some ferrous carbonate/phosphate phases.  
 204 Since Fe<sub>AVS</sub> was not present in our samples (see above), this phase can be discounted. Thus,  
 205 this extraction targets the remaining non-sulphidized Fe(II) minerals (termed Fe(II)<sub>HCl</sub>), as  
 206 well as Fe(III)<sub>HCl</sub>. Poorly crystalline Fe(III)<sub>HCl</sub> minerals such as ferrihydrite are not stable on  
 207 geological timescales, and hence the Fe(III)<sub>HCl</sub> pool can be considered to dominantly result  
 208 from more recent weathering of reduced minerals (particularly pyrite). Thus, quantification of  
 209 this pool provides a maximum estimate of the extent of post-depositional pyrite oxidation  
 210 (e.g., Matthews et al., 2017).

211 The Fe(II)<sub>HCl</sub> pool was determined immediately by spectrophotometer using the ferrozine  
 212 assay (Stookey, 1970). The total amount of Fe dissolved by the 0.5 N HCl extraction was  
 213 determined following reduction of Fe(III) to Fe(II) using hydroxylamine hydrochloride,  
 214 followed by analysis by spectrophotometer using the ferrozine assay. Subtraction of Fe(II)<sub>HCl</sub>  
 215 from this Fe pool gives Fe(III)<sub>HCl</sub>. Corrected Fe pools were calculated as:

$$216 \quad \text{Corrected Fe}_{\text{carbonate}} (\text{Fe}^*_{\text{carbonate}}) = \text{Fe}_{\text{carbonate}} - \text{Fe(III)}_{\text{HCl}} \quad (2)$$

$$217 \quad \text{Corrected Fe}_{\text{pyrite}} (\text{Fe}^*_{\text{pyrite}}) = \text{Fe}_{\text{pyrite}} + \text{Fe(III)}_{\text{HCl}} \quad (3)$$

218

## 219 4. Results

220 All geochemical data are shown in Table S1 and Figure 2.

### 221 4.1. Organic carbon isotopes

222 Organic carbon isotope ratios in WHK1 vary between -32.8‰ and -25.4‰, but with  
 223 distinct trends up-section. Specifically,  $\delta^{13}\text{C}_{\text{org}}$  values in Beds 1-7 vary between -27.5‰ and  
 224 -25.4‰ (Fig. 2). Values then decrease to -28.6‰ in Bed 8, followed by an increase to-  
 225 26.1‰ in Bed 14. The most pronounced decrease in  $\delta^{13}\text{C}_{\text{org}}$  then occurs between beds 14 and  
 226 17, reaching the minimum value of -32.8 ‰. Low  $\delta^{13}\text{C}_{\text{org}}$  values persist up to Bed 28, before  
 227 a gradually increasing trend, interrupted by a low value in Bed 48 (-29.2‰), is established up  
 228 to bed 50. Above Bed 50,  $\delta^{13}\text{C}_{\text{org}}$  values are relatively stable around -26.9‰, with variability  
 229 within 0.6‰.

230

231 *4.2. Major and trace elements*

232 Concentrations of Al and Fe range between 0.74–5.2 wt % and 0.26–6.4 wt %, respectively  
233 (Table S1). Beds 35 and 38 have low values of 0.74 and 0.75 wt % for Al, and 0.42 and 0.40  
234 wt % for Fe, respectively. The Al content is relatively high in the tuffaceous claystone and  
235 mudstone (3.3–5.2 wt %) and the black claystone (1.7–4.2 wt %), but is low in chert beds (0.9–  
236 2.7 wt %). Iron is high in the black claystone (more than 4 wt %) and the black chert of Beds  
237 29 (6.4 wt %).

238 Concentrations of U and Mo vary between 1.0–39.1 ppm and 0.8–194.3 ppm, respectively.  
239 Normalized U/Al ratios (Fig. 2) and  $U_{EF}$  (Table S1) values range between 0.84–13.5 and 2.7–  
240 43.6, respectively. Notably, high U/Al ratios (>5) occur in Beds 17–35 and Bed 44. Mo/Al (Fig.  
241 2) and  $Mo_{EF}$  (Table S1) vary between 0.3–61.1 and 2.4–611.0, respectively, with particularly  
242 high Mo/Al ratios (>5) occurring in Beds 17–24 and Bed 29.

243

244 *4.3. Iron speciation*

245 Fe speciation profiles are shown in Figure 2. Relatively high concentrations of  $Fe(III)_{HCl}$   
246 (more than 0.7 wt %) were detected in the black claystone samples of Beds 22, 24 and 48, as  
247 well as the black chert of Beds 23 and 29, which for these samples has a significant effect on  
248 corrected  $Fe_{carbonate}$  and  $Fe_{pyrite}$  concentrations. However, this correction for oxidative  
249 weathering of pyrite brings the Fe speciation data more broadly in line with the trends evident  
250 in Mo/Al ratios (Fig. 2). Indeed, both  $Fe_{pyrite}/Fe_{HR}$  and Mo/Al reflect the availability of sulphide  
251 in the water column and during early diagenesis, and as such, the general similarity in the  
252 position of peaks in  $Fe^*_{pyrite}/Fe_{HR}$  and Mo/Al (Fig. 2) provides strong support that our approach  
253 is robust.

254 We distinguish the Fe speciation results for Beds 31, 35 and 50 (open circles in Figure 2)  
255 from the remainder of the data, as these samples contain less than 0.5 wt % total Fe, and as  
256 such should be viewed with caution when interpreting Fe speciation data (Clarkson et al., 2016).  
257 We thus do not discuss the Fe speciation characteristics of these samples further. For the  
258 remaining samples,  $Fe_{HR}/Fe_{total}$  ratios range from 0.15 to 1.0.  $Fe_{HR}/Fe_{total}$  ratios are higher than  
259 0.38, which represents the conventional boundary for distinguishing samples deposited from  
260 an oxic or anoxic water column (Raiswell and Canfield, 1998; Poulton and Canfield, 2011),  
261 across most horizons, with the exception of Beds 46 and 58.  $Fe^*_{pyrite}/Fe_{HR}$  ratios range from  
262 0.11 to 0.88, with low values (less than 0.22) in the siliceous siltstone and claystone beds of  
263 the lower part of the section (Beds 1, 6, 7, 8, and 13).  $Fe^*_{pyrite}/Fe_{HR}$  ratios exceed 0.6 in the

264 black claystone and chert beds in the middle and upper parts of the section, except for Beds 20  
265 and 28 which show moderate values (0.48 and 0.42, respectively).

266

## 267 5. Discussion

### 268 5.1. Carbon isotope correlation

269 The Upper Permian strata of WHK1 exhibits a two-step decrease in  $\delta^{13}\text{C}_{\text{org}}$  values,  
270 resulting in an overall decrease of ca. 8‰. The first decrease occurs in Beds 7-8, followed by  
271 a larger decrease in Beds 14-17 (Figs. 2 and 4). The minimum  $\delta^{13}\text{C}_{\text{org}}$  value occurs in the thin  
272 black claystone bed (Bed 17) sandwiched between siliceous claystone beds. Then,  $\delta^{13}\text{C}_{\text{org}}$   
273 shows an increasing trend towards the overlying Lower Triassic black chert beds.

274 Inorganic carbonate carbon isotope ( $\delta^{13}\text{C}_{\text{carb}}$ ) values from low-latitude carbonate PTB  
275 sections show similar trends. One of the best documented examples is from the Meishan  
276 section, the type section for the Permian-Triassic boundary (Fig. 4; GSSP: Global strata-type  
277 section and point; Yin et al., 2001). The Meishan section also displays a two-step decrease in  
278  $\delta^{13}\text{C}_{\text{carb}}$  in the uppermost Permian strata below the base of the Triassic (Beds 23-25 in Fig 4;  
279 Jin et al., 2000; Cao et al., 2002; Kaiho et al., 2009). We label these two decreases in  $\delta^{13}\text{C}$  as  
280 “the first  $\delta^{13}\text{C}$  decrease” and “the second  $\delta^{13}\text{C}$  decrease”, respectively (marked by green and  
281 blue horizontal bars in Figure 4).

282 Based on U-Pb ages from zircon minerals in volcanic ash beds at the GSSP, these two  
283 isotopic steps are estimated to have occurred at  $251.999 \pm 0.039$  Ma and  $251.941 \pm 0.037$  Ma,  
284 respectively (Burgess et al., 2014), suggesting a 58 kyr separation. The first and second  
285 decreases are correlated with the two negative drops in  $\delta^{13}\text{C}_{\text{org}}$  observed at WHK 1 (Fig. 4).  
286 The second decrease in  $\delta^{13}\text{C}_{\text{carb}}$  at Meishan corresponds to the EPME horizon, and is  
287 associated with the loss of many marine taxa (Jin et al., 2000; Song et al., 2012). The  
288 Meishan section also shows a gradually increasing trend in  $\delta^{13}\text{C}_{\text{carb}}$  above the EPME horizon  
289 up to the Permian-Triassic boundary and into the overlying lowest Triassic strata, similar to  
290 the increasing trend seen in the strata of Beds 17-28 below the first Triassic conodont fossil  
291 occurrence from Bed 29 in WHK 1 (Figs. 2 and 4). This increasing  $\delta^{13}\text{C}$  trend above the  
292 EPME horizon provides a further correlative interval between PTB sections (marked by the  
293 pink horizontal bar in Figure 4). Therefore, based on this carbon isotope correlation within  
294 the biostratigraphic framework for the Permian-Triassic transition, defined at the GSSP  
295 Meishan section, we place the EPME horizon at the base of Bed 17 in WHK 1, which is a  
296 thin black claystone bed within the uppermost pale-green siliceous mudstone beds (Figs. 2

297 and 4). Another deep-sea PTB section from New Zealand (Arrow Rocks) also has a published  
298  $\delta^{13}\text{C}_{\text{org}}$  profile across the PTB (Fig. 4; Hori et al., 2007). This PTB section also shows  
299 decreasing trends in  $\delta^{13}\text{C}_{\text{org}}$  between a chert bed with the Changhsingian *Neogondolella*  
300 *prechangxingensis* in the lower part of the section (at -7 m, not shown in the Figure 4) and a  
301 Triassic black chert beds which yield conodonts of *Neogondolella carinata* and *Hindeodus*  
302 sp. of Triassic affinity (Yamakita et al., 2007). As the number of  $\delta^{13}\text{C}$  data from the lower half  
303 of the section are limited, the exact positions of the first and second  $\delta^{13}\text{C}$  decreases and their  
304 minimum troughs are uncertain. However, the  $\delta^{13}\text{C}_{\text{org}}$  profile in the Permian chert and  
305 siliceous mudstone beds shows a decrease at -0.6 to -0.3 m, and then a further decrease across  
306 the fault dividing a siliceous claystone unit (Unit 2a; Hori et al., 2007) and overlying black  
307 chert unit (Unit 2b; Hori et al., 2007). These  $\delta^{13}\text{C}$  trends are interpreted to correspond to the  
308 first  $\delta^{13}\text{C}$  decrease and the negative trough above the second  $\delta^{13}\text{C}$  decrease observed in the  
309 WHK 1 section (Fig. 4).

310 The two-step  $\delta^{13}\text{C}_{\text{org}}$  decrease during the Permian-Triassic transition, followed by a  
311 gradual increase toward the PTB, are also observed in the low latitude, deep-sea PTB  
312 sections of the Canadian Ursula Creek section (Wang et al., 1994), as well as the Japanese  
313 Akkamori-2 (Takahashi et al., 2010; Fig. 4), Ubara (Kaiho et al., 2012, Fig. 4) and NF1212C  
314 (Sano et al., 2012) sections. In the Japanese PTB sections, located in central Panthalassa,  
315 black claystone commonly occurs instead of siliceous claystone and bedded chert during the  
316 second  $\delta^{13}\text{C}$  decrease. This lithologic change reflects a significant decrease in siliceous  
317 microfossils (Takahashi et al., 2009). The black claystone facies in the Japanese PTB section  
318 continues during the Olenekian period, over a million years later (Takahashi et al., 2009). By  
319 contrast, WHK 1 has bedded chert beds just above the EPME horizon. The dominant black  
320 claystone facies of WHK 1 appears above these chert beds, with a return to chert beds below  
321 the PTB. This discrepancy in the timing of bedded chert deposition indicates a relatively  
322 short duration for the radiolarian chert gap in the southern mid-latitudes of Panthalassa.

323

## 324 5.2. Redox evaluation inferred from Fe speciation and trace elements in the Waiheke 1 325 section

### 326 5.2.1. Fe speciation systematics

327 Fe-speciation is a widely used proxy for reconstructing regional water column redox  
328 conditions (Poulton and Canfield, 2011; Poulton, 2021). Enrichments in  $\text{Fe}_{\text{HR}}$  ( $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}} >$   
329 0.38) commonly characterise deposition under anoxic water column conditions, and for

330 samples that show such enrichment, the extent of pyritization of the  $Fe_{HR}$  pool provides an  
331 indication of whether the water column was ferruginous ( $Fe_{pyrite}/Fe_{HR} < 0.6-0.8$ ) or euxinic  
332 ( $Fe_{pyrite}/Fe_{HR} > 0.6-0.8$ ). The samples that record a euxinic water column obtain syngenetic  
333 pyrite from the water column, in addition to possible continued pyritization of  $Fe_{HR}$  during  
334 diagenesis. The threshold values for  $Fe_{HR}/Fe_{total}$  and  $Fe_{pyrite}/Fe_{HR}$  are not absolute, and may be  
335 influenced by high sedimentation rates (particularly during turbidite deposition; e.g., Canfield  
336 et al., 1996) or transfer of  $Fe_{HR}$  to clay minerals (such as glauconite) during diagenesis, the  
337 latter of which may be particularly prevalent in organic-rich, non-sulphidic sediments (e.g.,  
338 Poulton and Raiswell, 2002; Poulton et al., 2010). However, we found no evidence for  
339 glauconite in the samples from WHK 1, based on thin section observations, which would be  
340 consistent with the relatively high pyrite contents of many of the samples. Regarding  
341 sedimentation rates, deep sea cherts in general have slow sedimentation rates (order of  
342 mm/kyr; e.g., Mastuda and Isozaki, 1991). Indeed, the calculated sedimentation rate for  
343 WHK 1 (3.8 mm/kyr) roughly corresponds to  $0.001 \text{ g cm}^{-2}/\text{yr}$  ( $= 0.38 \text{ cm/kyr} \times 2.7 \text{ g/cm}^3$ ),  
344 based on dating for Beds 7 and 17 (see above), which would not be expected to impact Fe  
345 speciation systematics. However, the tuffaceous sedimentary layers in this interval may  
346 potentially have masked  $Fe_{HR}$  enrichments due to rapid sedimentation (see Poulton et al.,  
347 2004).

348

### 349 *5.2.2. Uranium and molybdenum systematics*

350 Uranium and Mo exist as soluble oxyanions in oxic seawater (e.g., Tribovillard et al.,  
351 2006), with a relatively homogenous distribution because of their longer residence times  
352 relative to the mixing time of the ocean. Under reducing conditions, U transforms to  
353 tetravalent U(IV), whereas in the presence of dissolved sulphide, Mo converts to  
354 thiomolybdate. This results in enrichments in these elements, with Mo removal primarily  
355 occurring under euxinic conditions (but a degree of drawdown can also occur in open system  
356 sulphidic porewaters), whereas U uptake occurs within anoxic sediments (McManus et al.,  
357 2006). Based on observations from modern environments, Mo enrichments may begin to  
358 occur when transformation to thiomolybdates occurs ( $>11 \mu\text{M H}_2\text{S}$ ; Helz et al., 2011). Scott  
359 and Lyons (2012) suggest that sediment Mo concentrations  $>25$  ppm commonly indicate at  
360 least intermittent water column euxinia, whereas concentrations  $>100$  ppm likely indicate a  
361 persistent, strongly euxinic water column. Bennett and Canfield (2020) recommend using  
362 U/Al and Mo/Al ratios as the most objective normalization approach for minimising

363 influence from detrital inputs. Furthermore, based on a compilation of modern settings,  
364 Bennett and Canfield (2020) suggest that euxinic conditions are indicated when U/Al and  
365 Mo/Al ratios exceed  $\sim 5$  ( $\mu\text{g/g}$ ), and oxygen-poor sea floor sediments ( $\text{O}_2 \geq 10 \mu\text{M}$ ) are  
366 indicated when U/Al ratios are  $>1$  ( $\mu\text{g/g}$ ).

367 Consistent with the emerging perspective that local baseline values for redox proxy  
368 thresholds should be assigned wherever possible (Algeo and Li, 2020; Poulton, 2021), we  
369 apply different threshold values to the very different lithologies that comprise the upper  
370 cherty (partly muddy) strata (Beds 19-88) and the lower tuffaceous strata (Beds 1-18) (see  
371 Figure 2). For the upper cherty sediments, two samples (Beds 46 and 58) that give a clear  
372 oxic signal based on all of our applied redox proxies (with  $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}} < 0.22$ , low U/Al (0.90  
373 and 0.84; average = 0.87) and low Mo/Al (0.93 and 0.77; average = 0.83), can be used to  
374 define the oxic baseline for these parameters. By contrast, the tuffaceous strata in the lower  
375 part of WHK 1 would have a different baseline value, due to dilution by tuffaceous materials  
376 with a very different chemical composition, as well as the potential for masking of  
377 enrichments due to rapid sedimentation (see above). Minimum U/Al and Mo/Al ratios are  
378 observed in Beds 1 and 6, respectively. We use these beds to define oxic baseline ratios of  
379 0.84 for U/Al, and 0.30 for Mo/Al. Although the elemental composition of volcanic ash may  
380 depend on multiple factors (including magma source and crystallization differentiation), U is  
381 commonly enriched in tuff, whereas Mo is relatively depleted (Zielinski, 1983; Pichavant et  
382 al., 1987). For example, the Macusani tuff in southeast Peru has a U/Al ratio of at least 0.79,  
383 and a Mo/Al ratio of 0.03 (Pichavant et al., 1987). Furthermore, Bennet and Canfield (2020)  
384 show that oxic marine sediments have U/Al ratios in the range of 0.2-1.0, while Mo/Al ratios  
385 are in the range of 0.08-0.4. Thus, our defined oxic baseline for U/Al is comparable to these  
386 values, whereas our Mo/Al oxic baseline value is only slightly higher than these values. In  
387 addition, our approach to defining local redox thresholds provides a high level of consistency  
388 when interpreting the different redox proxies we apply. Thus, taken together, these  
389 observations provide strong support that our approach is robust.

390

### 391 5.2.3. Redox evaluation of the Waiheke 1 section

392 The lower part of WHK 1 (Beds 1-15) shows slightly elevated  $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}}$  and U/Al  
393 ratios, both of which show an overall progressive increase through the Upper Permian strata  
394 (Fig. 2). This consistent behaviour between two independent proxies suggests a progressive  
395 decline of bottom-water oxygen levels through this interval. Low  $\text{Fe}^*_{\text{pyrite}}/\text{Fe}_{\text{HR}}$  and Mo/Al

396 ratios indicate that the water column did not become euxinic at this time. However, during  
397 deposition of the thin black claystone bed at the EPME horizon (Bed 17), enrichments in  
398  $Fe_{HR}/Fe_{total}$  and U/Al, combined with high  $Fe^*_{pyrite}/Fe_{HR}$  and Mo/Al (Fig. 2), strongly support  
399 the onset of euxinia. This redox state was dominant until deposition of the earliest Triassic  
400 bedded cherts, which contain the first occurrence of Triassic conodont fossils (Bed 29),  
401 although two intervals of low  $Fe^*_{pyrite}/Fe_{HR}$  and Mo/Al in Beds 19 and 28, combined with  
402 elevated  $Fe_{HR}/Fe_{total}$  and U/Al, suggest intermittent development of ferruginous conditions  
403 (Fig. 2).

404 Above these strata, up until around Bed 44,  $Fe_{HR}/Fe_{total}$  and U/Al ratios remain high  
405 (although U/Al ratios are somewhat lower in this interval, relative to the underlying euxinic  
406 sediments; Fig. 2), suggesting persistent anoxia (as discussed above, the low  $Fe_{total}$  data  
407 shown by open circles are not considered reliable). The  $Fe^*_{pyrite}/Fe_{HR}$  ratios are elevated,  
408 suggesting water column euxinia, while Mo/Al ratios are also slightly elevated, but at a lower  
409 level than in the underlying euxinic strata. On a local scale, a reduction in the extent of Mo  
410 drawdown could simply be a consequence of less intense euxinia in the water column (Nägler  
411 et al., 2011). However, since U also shows a reduction in the extent of drawdown, and since  
412 U responds to anoxia rather than specifically euxinia, less intense euxinia does not appear to  
413 provide a valid interpretation for the muted trace metal enrichments we observe.  
414 Alternatively, however, if euxinia was widespread, then the muted enrichments we observe  
415 under euxinic conditions would more likely reflect a significant decrease in the seawater trace  
416 metal inventory (e.g., Algeo, 2004; Takahashi et al., 2014; Goldberg et al., 2016), which we  
417 consider in more detail below.

418 Above Bed 44, the limited available data show evidence for fluctuating oxic and anoxic  
419 redox conditions, with the possible development of euxinic conditions in some horizons (e.g.,  
420 Bed 71). When the redox state appears to have been euxinic (i.e., Bed 71), low trace metal  
421 drawdown (Fig. 2) may indicate prolonged seawater trace metal limitation through the  
422 earliest Triassic.

423

### 424 *5.3. Temporal relationship of end-Permian oceanic anoxia in the low latitude and south mid-* 425 *latitude Panthalassa*

426 To investigate the development of anoxia across the Panthalassic ocean, we have  
427 compiled redox proxy records for deep-sea PTB sections across the Permian to Triassic  
428 transition (Fig. 5). Available conodont-dated, Panthalassic deep-sea PTB sections carbon

429 isotope records and redox sensitive trace element data include Arrow Rocks in New Zealand  
430 (Hori et al., 2007), and the Akkamori 2 (Takahashi et al., 2014) and Ubara sections (Algeo et  
431 al., 2011) in Japan. In Figure 5, U/Al and Mo/Al ratios for WHK 1 and these additional PTB  
432 sections are compared based on  $\delta^{13}\text{C}$  chemostratigraphic correlation (Fig. 4).

433 WHK 1 and Arrow Rocks, which were deposited in the south mid-latitude of the  
434 Panthalassa Ocean, might be expected to have similar redox characteristics. During the first  
435  $\delta^{13}\text{C}$  decrease in the late Changhsingian, WHK 1 consists of pale green-coloured tuffaceous  
436 sedimentary rocks. Arrow Rocks also consist of similar tuffaceous strata in the late  
437 Changhsingian (Fig. 5). Given this lithologic similarity, we use the baseline values for U/Al  
438 and Mo/Al defined for WHK 1 to guide interpretation of the Arrow Rocks redox data.  
439 However, U concentrations were below detection (<1.5 ppm) in the lower part of the Arrow  
440 Rocks succession (Hori et al., 2007), preventing a direct redox interpretation. Furthermore,  
441 while Mo/Al ratios are relatively high, total sulphur concentrations are below detection (<1.8  
442 ppm), and thus it is difficult to provide a robust redox interpretation for this part of the  
443 succession. However, in the overlying Permian-Triassic transitional section at Arrow Rocks,  
444 enrichment of U/Al is evident, alongside very high peaks in Mo/Al (up to 670). This  
445 coincides with U and Mo enrichment in WHK 1 (Fig. 5), supporting the notion of widespread  
446 development of euxinic water column conditions in the south mid-latitude Panthalassa, at  
447 least from the EPME to the PTB.

448 Geochemical records from low latitude Panthalassa, represented by the Akkamori 2 and  
449 Ubara sections, provide further insight into temporal and spatial redox dynamics. To evaluate  
450 the redox records for these PTB sections, we again use the minimum measured U/Al  
451 (Akkamori = 0.55, Ubara = 0.11) and Mo/Al (Akkamori = 1.13, Ubara = 1.01) ratios to  
452 define oxic baseline values. Both sites show evidence for generally persistent anoxia  
453 (elevated U/Al) throughout the entire section, while elevated Mo/Al ratios suggest that the  
454 water column may have become euxinic prior to the first  $\delta^{13}\text{C}$  decrease (Fig. 5). Furthermore,  
455 while there was some fluctuation in the precise chemistry of the anoxic water column after  
456 the EPME, euxinia was a prevalent feature immediately after the EPME at the Akkamori site,  
457 and potentially also at the Ubara site.

458 Our interpretation of the redox dynamics at the Ubara site based on Al-normalized values  
459 of U and Mo contrast with those of Algeo et al. (2011), who considered suboxic conditions to  
460 have been prevalent. In their study, the elevated bulk concentrations of U and Mo at Ubara  
461 site are interpreted as being due to the slower sedimentation rate of Permian-Triassic black  
462 claystone relative to the Permian chert beds resulting in enrichment. There thus remains some

463 uncertainty in the precise redox state of the Ubara site. However, we note that our  
464 interpretation is more consistent with the redox conditions that are much more compellingly  
465 demonstrated by highly elevated enrichments in Mo/Al at the Akkamori site. Thus, taken  
466 together, it appears likely that the ocean oxygenation state of the Akkamori and Ubara sites in  
467 the low latitude Panthalassa deteriorated earlier, relative to the Waiheke site in the south mid-  
468 latitude Panthalassa.

469

#### 470 *5.4. Covariation between U and Mo, and Mo drawdown after the Permian-Triassic boundary*

471 The bulk concentrations of Mo and U in the PTB sections are recast as enrichment factors  
472 ( $M_{\text{OEF}}$  and  $U_{\text{EF}}$ ) and shown as a cross plot in Figure 6. Algeo and Tribovillard (2009) and  
473 Tribovillard et al. (2011) provide a framework for utilizing  $M_{\text{OEF}}$  and  $U_{\text{EF}}$  as a means to  
474 identify three enrichment patterns. Firstly, co-increases in  $M_{\text{OEF}}$  and  $U_{\text{EF}}$  may occur following  
475 a deterioration in redox conditions from suboxic to euxinic in open marine settings (Fig. 6).  
476 The second case occurs where the  $M_{\text{OEF}}$  is significantly elevated relative to the  $U_{\text{EF}}$ , which  
477 may be a consequence of active Mo transportation from the water column to the sediment, for  
478 example via Fe-Mn (oxyhydr)oxide shuttling. The third pattern occurs when smaller and  
479 relatively constant  $M_{\text{OEF}}$  are combined with high  $U_{\text{EF}}$ . This scenario is considered to represent  
480 the case where extensive Mo-depleted seawater arises because of a persistent sulphidic deep  
481 water mass, as may occur in a relatively restricted setting.

482 In the PTB sections of the low latitude Panthalassa during the Late Permian  
483 (Changhsingian), before the  $\delta^{13}\text{C}$  decrease,  $M_{\text{OEF}}$  values are high relative to  $U_{\text{EF}}$  values (Fig.  
484 6-1). This supports the redox observations based on U/Al and Mo/Al ratios (Fig. 5), and  
485 suggests anoxic, non-sulphidic water column conditions across this interval, with elevated  
486 Mo drawdown due to uptake and sequestration by Fe-Mn (oxyhydr)oxide minerals. At the  
487 WHK 1 site, the  $M_{\text{OEF}}$  and  $U_{\text{EF}}$  data (Fig. 6-1) document the slower development of anoxia in  
488 the south mid-latitude Panthalassa, with evidence for the very beginning of water column de-  
489 oxygenation, consistent with the Fe speciation, U/Al and Mo/Al (Fig. 2).

490 Between the two step decreases of  $\delta^{13}\text{C}$ , highly elevated  $M_{\text{OEF}}$ , and to a lesser extent  $U_{\text{EF}}$ ,  
491 are evident for the low latitude Panthalassa localities (Fig. 6-2). These values support the  
492 continued deterioration of water column redox conditions, with sporadic development of  
493 euxinic conditions (as documented in Figure 5) further enhancing trace metal drawdown  
494 (particularly Mo). In contrast, at the WHK 1 site in the south mid-latitude Panthalassa,  $U_{\text{EF}}$   
495 and  $M_{\text{OEF}}$  values document a later, progressive increase (Fig. 6-2). Across this interval, the

496 gradual co-increase in  $U_{EF}$  and  $Mo_{EF}$  values at the WHK 1 site suggests a classic redox  
497 transition from suboxic, through anoxic, to euxinic water column conditions (Fig. 6; Algeo et  
498 al., 2009; Tribouillard et al., 2012). This is generally consistent with the trends evident in the  
499 Fe speciation data (Fig. 2), supporting a progressive deterioration in the redox state of the  
500 water column and enhanced drawdown of U and Mo into the sediments.

501 During the interval from the EPME to the PTB, high  $U_{EF}$  and  $Mo_{EF}$  values are recorded in  
502 both the south middle latitude and low latitude Panthalassa (Fig. 6-3). However, data from  
503 the WHK 1 site plot closer to the typical trend for euxinic settings, which supports the  
504 suggestion from trends in U/Al and Mo/Al (Figs. 2 and 5), that while euxinia was likely  
505 widespread, this redox state may have become more prevalent and persistent in south middle  
506 latitude regions, relative to the low latitude Panthalassa.

507 After the peaks across the EPME-PTB interval,  $U_{EF}$  and  $Mo_{EF}$  shift to generally lower  
508 values during the early Triassic (Fig. 6-4). Indeed, enrichment patterns do not return to the  
509 open-marine redox evolution trend. Instead, while the data show a wide range,  $Mo_{EF}$  values  
510 in particular are generally lower than during the EPME-PTB interval. Thus, we suggest that  
511 the development of widespread anoxia, with the common occurrence of euxinia, during the  
512 EPME-PTB interval subsequently resulted in limited Mo sequestration in the sediments due  
513 to a significant decrease in the seawater Mo inventory caused by intense Mo-bearing sulphide  
514 deposition (e.g., Algeo, 2004; Takahashi et al., 2014; Goldberg et al., 2016). Although euxinic  
515 water condition enhances uptake of both Mo and U into the sediment, relatively stronger  
516 accumulation of Mo results in significantly decreased residence time of Mo in the seawater  
517 and depletion of Mo relative to U. Alternatively, such a signal could be due to lower sulphide  
518 availability in the water column, which might result from seawater sulphate drawdown after  
519 EPME anoxia (Song et al., 2014; Schobben et al., 2017). However, the fact that several  
520 intervals have  $Fe_{HR}/Fe_{total}$  and  $Fe^*_{pyrite}/Fe_{HR}$  ratios suggestive of euxinia, implies that Mo  
521 depletion from the water column, rather than low sulphate availability, is a more likely  
522 explanation for the relatively subdued Mo enrichments evident in the earliest Triassic.

523

#### 524 5.5. Possible drivers of Panthalassic anoxia

525 This study indicates that the pelagic Panthalassa became anoxic, and was often euxinic, in  
526 south middle latitude and low latitude regions during the EPME. However, the development  
527 of anoxia occurred earlier at low latitudes. The progressive development of ocean anoxia  
528 recorded by our data is in agreement with a decreasing trend in  $\delta^{238}U_{carbonate}$  recorded in

529 multiple carbonate PTB sections (Fig. 7; Zhang et al., 2020). The  $\delta^{238}\text{U}_{\text{carbonate}}$  signal begins to  
530 decrease during the first  $\delta^{13}\text{C}$  decrease of the Changhsingian, which continues across the  
531 EPME, resulting in a negative trough in the earliest Triassic. This  $\delta^{238}\text{U}_{\text{carbonate}}$  variability is  
532 consistent with drawdown of U into Panthalassic anoxic deep-sea sediments through the Late  
533 Permian and Early Triassic.

534 In Figure 7, data that can promote marine anoxia are compiled for the EPME. The  
535 Siberian Traps were likely the main trigger for environmental change across the Permo-  
536 Triassic boundary (e.g., Wignall., 2001). A mercury (Hg) anomaly, taken as an indicator of  
537 enhanced volcanic activity, has been detected in late Changhsingian horizons in the Akkamori  
538 2 and Ubara sections in the low latitude Panthalassa (Fig. 7; Shen et al., 2019), and at WHK 1  
539 (Grasby et al., 2021), demonstrating a coincidence of major volcanism with the onset of  
540 euxinia. Identified Hg anomalies appear below the EPME horizon at the low latitude sites,  
541 but just above the EPME horizon at the south mid-latitude WHK 1 site (Fig. 7). One possible  
542 reason for this discrepancy is the south mid-latitude location was too far from the Siberian  
543 Traps to receive enough Hg before the EPME. The low sampling resolution across the WHK  
544 1 section could be another possibility for this discrepancy. Grasby et al. (2021) did not  
545 measure Hg in the thin black claystone bed at the EPME (Bed 17 of this study) and adjacent  
546 layers.

547 The development of ocean anoxia prior to the EPME and persistence into the earliest  
548 Triassic coincides with extreme climate warming (Fig. 7; Joachimski et al., 2012, 2019). This  
549 warming has been linked to massive release of  $\text{CO}_2$  from heated sediments and upper crusts  
550 by Siberian Traps sills (Svensen et al., 2009). Subsequently, seawater warming would  
551 contribute to marine deoxygenation in these oceanic regions through the decreasing solubility  
552 of dissolved  $\text{O}_2$ . Furthermore, thermohaline circulation driven by subduction of warm saline  
553 surface water formed in mid and low latitudes would potentially contribute to the  
554 development of a stratified ocean structure with anoxic deep waters (Kidder and Worsley,  
555 2004).

556 The development of anoxia during the EPME coincides the cessation of radiolarian chert  
557 deposition (Fig. 7). Our  $\delta^{13}\text{C}$  correlation of PTB sections reveals that this phenomenon  
558 occurred in the south mid-latitude Panthalassa around ten kiloyears later than in the low  
559 latitude Panthalassa (Fig. 7). This may be due to a decreased biogenic silica burial flux  
560 (declining radiolarian population and/or degraded preservation of siliceous tests) and dilution  
561 of biogenic silica by increased clastic material (Takahashi et al., 2009; Muto et al., 2018,  
562 2020). The environmental conditions during the EPME, including ocean anoxia and seawater

563 warming may have decreased the siliceous faunal population and increased sedimentation of  
564 clay-rich material relative to biotic silica. If warming is a major factor limiting deposition of  
565 biotic silica, then the different timing for the disappearance of radiolarian chert in the low  
566 latitude and south mid-latitude ocean is noteworthy. For example, the higher latitude ocean  
567 would warm more slowly, thus resulting in the observed delayed onset of both ocean anoxia  
568 and the decrease in biotic silica deposition. The subsequent return to lower temperatures  
569 favourable for the radiolarian chert deposition would also arrive earlier in mid-latitude  
570 regions.

571

## 572 6. Conclusions

573 This study provides new geochemical records across the end-Permian mass extinction  
574 event in the south middle latitude pelagic Panthalassa. Fe-speciation and redox-sensitive  
575 elements (U and Mo) show the progressive development of ocean anoxia, with euxinia  
576 occurring during the EPME. Comparison with previously reported paleo redox records from  
577 the low latitude Panthalassa indicates that anoxia developed earlier in low latitude regions.  
578 The pelagic Panthalassa PTB sections show Mo drawdown in the earliest Triassic, after the  
579 EPME likely attributable to high Mo burial flux into anoxic marine sediments. The Late  
580 Permian anoxia that occurred in the low latitude Panthalassa, and the progressive  
581 development of anoxia during the run-up to the EPME in the south middle latitude  
582 Panthalassa, coincides with Siberian Traps volcanic activity and de-vegetation trends on land.  
583 Furthermore, the persistent anoxia that developed in both latitudinal regions of Panthalassa  
584 coincide with severe seawater warming.

585 The results of this study shed light on spatiotemporal changes in open ocean redox  
586 conditions across the Permian-Triassic boundary and show that the redox history of this  
587 super-ocean varied according to latitude. Conditions for radiolarians were persistent and  
588 harsh in equatorial latitudes relative to the mid-latitude site discussed here, indicating that  
589 high temperatures were a major factor controlling their absence. Environmental records for  
590 mid-high latitudes and pelagic regions of the ocean remain highly limited, and this should be  
591 a prime target for further investigation.

592

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602

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854 Figure captions

855

856 Figure 1. Location map of the studied Permian-Triassic section (Waiheke 1 section, New  
857 Zealand). (A) Paleo-geographical map of the Permian-Triassic showing the depositional  
858 position of pelagic deep-sea Permian-Triassic boundary sections. Labeled numbers (1-7)  
859 indicate the following localities: (1) pelagic sections from New Zealand accretionary  
860 complexes, such as the Waiheke section (Spörli et al., 2007); (2) pelagic sections from  
861 Japanese accretionary complexes; (3) shallow marine sections such as Meishan, South China;  
862 (4) sections in Sverdrup Basin, arctic Canada (Grasby and Beauchamp, 2011); (5) sections in  
863 Peace River Basin, western Canada (Hays et al., 2007). (B) Locality of the Waiheke-1 section  
864 (WHK1) in the North Island of New Zealand. Distribution of the Triassic-Jurassic  
865 accretionary complex is based on Black (1994) and Begg and Johnson (2000). Locality of  
866 Arrow Rocks is also shown.

867 Figure 2. Lithologic column of the Waiheke 1 section, including organic carbon isotopes  
868 ( $\delta^{13}\text{C}_{\text{org}}$ ), iron speciation pools ( $\text{Fe}_{\text{pools}}$ ), highly reactive iron normalised to total iron  
869 ( $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}}$ ), corrected pyrite iron as a function of highly reactive iron ( $\text{Fe}^*_{\text{pyrite}}/\text{Fe}_{\text{HR}}$ ), U,  
870 U/Al and Mo/Al. The error ranges on the organic carbon isotope values ( $\delta^{13}\text{C}_{\text{org}}$ ) are based on  
871 repetitions of carbon isotope measurements (Table S1). Open circles in the plots of  $\text{Fe}_{\text{pools}}$ ,  
872  $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}}$  and  $\text{Fe}^*_{\text{pyrite}}/\text{Fe}_{\text{HR}}$  mean the samples with low Fe content (less than 0.5 wt %  
873  $\text{Fe}_{\text{total}}$ ). We did not use these samples for paleoenvironmental reconstruction. Local  
874 background values of U/Al and Mo/Al for tuffaceous strata (Beds 1-18) and cherty strata  
875 (Beds 19-88) are shown as vertical dashed lines with “baseline” notation. The values are  
876 based on the minimum values in the tuffaceous strata and the values in the cherty strata with  
877 oxic  $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}}$  signals.

878 Figure 3. Thin section photographs from the Waiheke 1 section. (A) tuffaceous siliceous  
879 claystone of Bed 8; (B) black claystone of Bed 22; (C) bedded chert of Bed 50; (D) bedded  
880 chert from around Bed 60.

881

882 Figure 4. Stratigraphic correlation of the Permian-Triassic boundary sections of the Meishan  
883 section, Waiheke 1 section, Arrow Rocks section, Akkamori 2 section and Ubara section.  
884 Compiled data of the Meishan section are based on Jin et al. (2000), Cao et al. (2002), Kaiho  
885 et al. (2009) and Burgess et al. (2014). Data for Arrow Rocks, Akkamori and Ubara sections  
886 are modified from Hori et al. (2007), Takahashi et al. (2010) and Kaiho et al. (2012),  
887 respectively. The horizontal bars indicate correlative stratigraphic zones around the Permian-  
888 Triassic boundary. Green: the first decrease in carbon isotope ratio ( $\delta^{13}\text{C}$ ) in the latest  
889 Changhsingian, Blue: the second decrease in  $\delta^{13}\text{C}$  in the latest Changhsingian coinciding with  
890 the EPME, Pink: the interval from the  $\delta^{13}\text{C}$  minimum during the EPME to the  $\delta^{13}\text{C}$  increase  
891 toward the PTB.

892

893 Figure 5. Vertical plots of U/Al and (Mo/Al) for the Waiheke 1 and other pelagic deep-sea  
894 sections. The data for Arrow Rocks, Akkamori 2 and Ubara are modified from Hori et al.  
895 (2007), Takahashi et al. (2014) and Algeo et al. (2011), respectively. The local background  
896 value (baseline) for U/Al and Mo/Al are based on the WHK 1 baseline values for Arrow  
897 Rocks, and are based on the minimum values for the Akkamori and Ubara sections.

898

899 Figure 6. Cross plots of molybdenum (Mo) and uranium (U) enrichment factors ( $\text{Mo}_{\text{EF}}$  and  
900  $\text{U}_{\text{EF}}$ , respectively) for Panthalassic PTB sections from Waiheke 1, Arrow Rocks, Akkamori  
901 and Ubara. The data for Arrow Rocks, Akkamori 2 and Ubara are modified from Hori et al.  
902 (2007), Takahashi et al. (2014) and Algeo et al. (2011), respectively. The data are plotted in  
903 four different panels representing the time intervals based on the carbon isotope correlation  
904 demonstrated in Figures 4 and 5. 1: Late Permian (Changhsingian) before the  $\delta^{13}\text{C}$  decrease;  
905 2: between the first and second steps decreases of  $\delta^{13}\text{C}$  in the latest Permian; 3: the interval  
906 from the  $\delta^{13}\text{C}$  minimum at the EPME to the  $\delta^{13}\text{C}$  increase toward the PTB; 4: Earliest  
907 Triassic (Induan) after the PTB. Superimposed grey-coloured areas representing modern  
908 oceanic sediments are from Algeo and Tribovillard (2009) and Tribovillard et al. (2012). The  
909  $\text{U}/\text{Mo}_{\text{sw}}$  is the weight ratio in modern seawater (SW), which ranges from 3.0 to 3.1 (modern  
910 Pacific and Atlantic, respectively; Anderson, 1987; Bruland, 1983; Chen et al., 1986;  
911 Emerson and Huested, 1991).

912

913 Figure 7. Compilation of oceanic environmental records from Panthalassa, seawater  
914 temperature for the paleo-Tethys reconstructed by oxygen isotope ratios in conodont apatite  
915 ( $\delta^{18}\text{O}_{\text{apatite}}$ ; Joachimski et al., 2009, 2019), global anoxic water extent referred from uranium  
916 isotopes in carbonate ( $\delta^{238}\text{U}_{\text{carbonate}}$ ; Zhang et al., 2020). Time scale through Permian and  
917 Triassic is from Burgess et al. (2014). Information on oceanic environmental records for  
918 Panthalassa is based on litho- and chemo-stratigraphy by this study, Hori et al. (2007, 2011),  
919 Algeo et al., (2011), Takahashi et al. (2014), Kaiho et al. (2016), Shen et al., (2019) and  
920 Grasby et al., (2021). Thin lines indicate the range of each Permian-Triassic boundary  
921 section. The bars coloured black, orange, purple and grey refer to Hg anomaly ( $\text{Hg}/\text{TOC} >$   
922 100), radiolarian chert occurrence, geochemical signals of euxinia (high  $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}}$ ,  
923  $\text{Fe}_{\text{pyrite}}/\text{Fe}_{\text{HR}}$  and/or  $\text{Mo}/\text{Al}$  and  $\text{U}/\text{Al}$ ), and seawater Mo drawdown (lower  $\text{Mo}/\text{Al}$  and  $\text{Mo}_{\text{EF}}$   
924 despite high  $\text{Fe}_{\text{HR}}/\text{Fe}_{\text{total}}$  and  $\text{Fe}_{\text{pyrite}}/\text{Fe}_{\text{HR}}$  and/or  $\text{U}/\text{Al}$  and  $\text{U}_{\text{EF}}$ ), respectively.  
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