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1 **A chemical weathering control on the delivery of particulate iron to the continental shelf**

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16 **Abstract**

17 The delivery of potentially reactive iron minerals to the marine environment exerts a major
18 control on ocean and marine sediment biogeochemistry, including the preservation and burial
19 of organic carbon. One aspect of the global iron cycle that has received little attention concerns
20 the phase partitioning of iron in highly chemically-weathered sediments from (sub)tropical
21 mountainous regions, where the sediment is commonly deposited directly onto adjacent
22 continental margins. Here, we report the phase partitioning of particulate iron in surface
23 seafloor sediments deposited from oxic bottom waters along the east coast of China, including
24 the Bohai Sea, Yellow Sea and East China Sea, in addition to the continental shelf region
25 adjacent to the (sub)tropical mountainous island of Hainan in the northern South China Sea.

26 We document highly reactive (Fe_{HR}) to total iron (Fe_T) ratios for the Bohai Sea, Yellow Sea and
27 East China Sea that are typical of global marine sediments deposited under oxic water column
28 conditions. By contrast, Fe_{HR}/Fe_T ratios in Hainan offshore sediments are significantly elevated
29 above normal oxic marine sediments. These Hainan offshore sediments display a strong
30 positive correlation between Fe_{HR}/Fe_T ratios and the chemical index of alteration (CIA)
31 weathering proxy, suggesting that the Fe speciation systematics are controlled by the intensity
32 of chemical weathering. This observation is supported by additional geochemical data and Nd
33 isotope analyses, which suggest a common source lithology for all the studied regions. The
34 Hainan sediments thus document a specific scenario, whereby highly chemically weathered
35 sediments from the mountainous Hainan region bypass the inner shore settings that commonly
36 disproportionately trap Fe (oxyhydr)oxide minerals, depositing high concentrations of Fe_{HR} on
37 the adjacent continental margin. Such enrichments likely aid in the preservation of organic
38 carbon in marine sediments, suggesting that organic carbon preservation and burial may be
39 enhanced by elevated Fe_{HR} in such settings. We also establish a set of essential criteria for
40 potential recognition of chemical weathering-induced enrichments in Fe_{HR}/Fe_T ratios in ancient
41 marine sediments.

42 **Keywords**

43 continental margin sediments; east China coast; Fe speciation; chemical weathering; highly
44 reactive Fe

45 **1. Introduction**

46 The supply of potentially bioavailable iron to the marine environment exerts a major
47 control on ocean biogeochemistry. Iron is an essential limiting nutrient for phytoplankton, and

48 its supply exerts a significant impact on marine productivity in large areas of the world's oceans,
49 which in turn influences the carbon cycle and hence climate dynamics (Mackenzie and
50 Andersson, 2013). In addition, the supply of 'reactive' particulate iron exerts a major control
51 on the preservation and burial of organic carbon (e.g., Lalonde et al., 2012), as well as the
52 formation of pyrite in continental margin sediments (Berner, 1970; Canfield, 1989; Raiswell
53 and Canfield, 1998). Thus, given the significance for the long term regulation of the global
54 carbon, oxygen and sulfur cycles (Kump and Garrels, 1986; Berner, 2003; Canfield, 2005),
55 controls on the delivery of particulate iron phases to the marine environment are critical to
56 evaluate.

57 Particulate iron is delivered to the ocean by rivers, aeolian dust, icebergs and subglacial
58 runoff, while hydrothermal processes and recycling from shelf sediments provide dissolved
59 iron, which may subsequently precipitate in the water column (Poulton and Raiswell, 2002;
60 Raiswell et al., 2006; Raiswell and Canfield, 2012). The precise contribution from each of these
61 sources is difficult to quantify and will have evolved over Earth history, but transport by rivers
62 provides the dominant source of chemically-weathered sediment, and hence particulate iron, to
63 the global ocean (Milliman and Meade, 1983; Raiswell, 2006). Canfield (1997) quantified the
64 Fe (oxyhydr)oxide content of major US rivers in relation to the intensity of chemical weathering,
65 and found that the proportion of Fe (oxyhydr)oxide minerals (hereon termed highly reactive Fe,
66 Fe_{HR}) relative to total Fe (Fe_T) increased with enhanced chemical weathering. Poulton and
67 Raiswell (2002) subsequently produced a global budget for the delivery of Fe_{HR} (dominantly as
68 Fe (oxyhydr)oxide minerals) to the marine environment, and found that the average Fe_{HR}/Fe_T
69 ratio for global river particulates (0.43 ± 0.03 , 1σ) is substantially elevated relative to the

70 average for marine sediments (0.26 ± 0.08 , 1σ ; Raiswell and Canfield, 1998) deposited under
71 oxic water column conditions (note that for marine sediments, Fe_{HR} includes Fe sulfide phases
72 to account for sulfidation of Fe (oxyhydr)oxides during diagenesis). Indeed, the vast majority
73 of modern oxic marine sediments have Fe_{HR}/Fe_T ratios below 0.38 (Raiswell and Canfield,
74 1998), and this discrepancy between the chemical composition of river particulates and marine
75 sediments arises due to the preferential trapping of Fe_{HR} minerals in inner shore environments,
76 such as flood plains, subsiding deltas, salt marshes and lagoons (Poulton and Raiswell, 2002).

77 Currently, however, there is a missing link in our understanding of the global Fe cycle,
78 which relates to the delivery of Fe_{HR} to the ocean from (sub)tropical mountainous rivers, many
79 of which bypass inner shore settings to deposit sediment directly onto active margins (Milliman
80 and Syvitski, 1992). The relatively high intensity of chemical weathering expected in
81 (sub)tropical mountainous regions, combined with a general increase in sediment discharge
82 directly to the ocean, might be expected to result in an elevated supply of Fe_{HR} to adjacent
83 marine settings, and hence may contribute to increased primary productivity and organic carbon
84 burial (Lalonde et al., 2012; Tagliabue et al., 2017). However, this premise has not previously
85 been tested.

86 The potential for local Fe_{HR} -enriched sediment delivery to the ocean from mountainous
87 (sub)tropical catchments also has implications for the use of Fe_{HR}/Fe_T ratios as a water column
88 paleoredox indicator (Poulton and Canfield, 2011; Raiswell et al., 2018; Poulton 2020).
89 Enrichment in Fe_{HR}/Fe_T above the upper range for oxic marine sediment deposition (>0.38)
90 commonly occurs in sediments deposited beneath an anoxic water column, due to mobilization
91 and precipitation of Fe in the water column (Canfield et al., 1996; Raiswell and Canfield, 1998).

92 This Fe_{HR}/Fe_T threshold generally provides a robust indication of water column anoxia (Poulton
93 and Canfield, 2011), despite the fact that the global riverine Fe_{HR}/Fe_T ratio is substantially
94 elevated relative to marine sediments (Poulton and Raiswell, 2002). Clearly, however, there is
95 potential for local sedimentary Fe_{HR} enrichments in oxic marine settings, in regions
96 experiencing relatively high rates of chemical weathering, and where the inner shore trap for
97 Fe_{HR} is circumvented (e.g., Schobben et al., 2020).

98 Here, we investigate the potential role of sub(tropical) mountainous regions in the supply
99 of Fe_{HR} to local marine sediments. We report Fe_{HR}/Fe_T ratios (using both sequential extraction
100 and single dithionite methods), major and trace elements, and neodymium isotope data for
101 surface sediments deposited along the east coast of China (Fig. 1). This includes the Bohai Sea,
102 Yellow Sea and East China Sea, which receive sediment from major low-lying rivers such as
103 the Huanghe (Yellow River) and Chang Jiang (Yangtze River), as well as the northern South
104 China Sea, in an area that receives sediment from the mountainous island of Hainan. We aim to
105 better unravel the potential effects of chemical weathering and detrital provenance on local
106 Fe_{HR}/Fe_T ratios in continental margin sediments, and in doing so, we provide new insight into
107 the modern global Fe cycle. We additionally discuss criteria for the potential recognition of
108 weathering-induced Fe_{HR} enrichments in ancient marine sediments deposited under oxic water-
109 column conditions.

110 **2. Geological setting and methods**

111 Forty-four surface sediments (0–4 cm) were sampled from the Bohai Sea, Yellow Sea, East
112 China Sea and the Hainan offshore area of the northern South China Sea (Fig. 1). The bulk
113 sediments are mainly silty and argillaceous, consisting of clay minerals, quartz, feldspar and

114 very small amounts of biogenic carbonate (a detailed description of locality, water-depth and
115 mean grain size for the samples is presented in the Supplementary Materials). Previous studies
116 have shown that these surface sediments were deposited beneath bottom waters with dissolved
117 O₂ concentrations of least 30 μM (e.g., Li and Qu, 2006; Chen et al., 2007; Li et al., 2015; Zhao
118 et al., 2017). Samples in this study were collected by a box corer, from which surface sediments
119 were taken and stored in a cold repository. Before analysis, the bulk sediments were freeze-
120 dried and sieved to avoid biotic carbonate components, and samples were then analyzed for Fe
121 speciation, major and trace elements, and Nd isotopes. Since our samples were not immediately
122 frozen and stored under oxygen-free conditions, oxidation of reduced phases is likely to have
123 occurred. However, while oxidation may have affected the speciation of the Fe_{HR} pool, it will
124 not have affected determination of Fe_{HR} concentrations or Fe_{HR}/Fe_T ratios, which are the
125 primary focus of this study.

126 For major (except Fe_T) and trace element analyses (including Al, K, Na, Ca, Ti, Sc, Zr, La,
127 Th), ~50 mg of sample was initially leached using 0.5 N acetic acid, in order to remove
128 carbonate phases and adsorbed seawater K and Na from bulk samples. Residues were then fully
129 digested using distilled HF–HNO₃–HCl, and diluted in 3% HNO₃. Total Fe was determined
130 directly using a full digestion method (i.e., without the initial 0.5 N acetic acid leach) with
131 distilled HF–HNO₃–HCl. Major elements were measured using a Thermo ICAP 6000 ICP-OES,
132 and trace elements were measured on an Agilent 7900 Quadrupole ICP-MS at the MOE Key
133 Laboratory of Surficial Geochemistry, Nanjing University. The RSD was <5% for all elements
134 based on repeat analyses of the rock standard BHVO-2, and the IAPSO standard seawater
135 solution was used to ensure accuracy.

136 Neodymium isotopes were analyzed on a Thermo Neptune Plus MC-ICP-MS at the State
137 Key Laboratory for Mineral Deposits Research, Nanjing University, following the procedures
138 outlined in Wei et al. (2019). The $^{143}\text{Nd}/^{144}\text{Nd}$ ratios were normalized to a $^{146}\text{Nd}/^{144}\text{Nd}$ ratio of
139 0.7219 in order to correct for mass drift. BHVO-2 and an in-house standard solution were
140 repeatedly measured to monitor the long-term reproducibility of Nd isotope analyses
141 ($^{143}\text{Nd}/^{144}\text{Nd}_{\text{BHVO-2}} = 0.512987 \pm 0.000002$ (SE); $^{143}\text{Nd}/^{144}\text{Nd}_{\text{Inhouse}} = 0.511699 \pm 0.000003$ (SE)).

142 The extraction procedures for determining Fe_{HR} in marine or riverine sediments have
143 evolved over recent decades, and are summarized in Table 1. The most recent scheme (Poulton
144 and Canfield, 2005) utilizes a sequential extraction approach and targets a variety of
145 operationally-defined phases (Table 1). In this approach, ~100 mg of sample was first treated
146 with a room temperature sodium acetate solution at pH 4.5 for 24 h (to extract Fe_{carb}). The
147 residue was then reacted with a room temperature sodium dithionite solution at pH 4.8 for 2 h
148 (to extract Fe_{ox}). Finally, the residue was reacted with a room temperature 0.2 M ammonium
149 oxalate and 0.17 M oxalic acid solution for 6 h (to extract Fe_{mag}). In addition, we also performed
150 a single sodium dithionite extraction (Table 1) to quantify non-sulfidized Fe_{HR} phases (termed
151 Fe_{D}), to allow direct comparison with existing data for modern marine sediments which were
152 analyzed in this way (Raiswell and Canfield, 1998). Extractions were performed at the State
153 Key Laboratory for Mineral Deposits Research, Nanjing University.

154 After drying and re-dissolution in 3% HNO_3 , all Fe solutions were analyzed on a Thermo
155 ICAP ICP-OES at the MOE Key Laboratory of Surficial Geochemistry, Nanjing University,
156 with a relative standard deviation (RSD) of <5% for all steps. Acid volatile sulfide (Fe_{AVS}) and
157 pyrite Fe (Fe_{Py}) concentrations were determined by the two-step Cr-reduction method of

158 Canfield et al. (1986), but concentrations were found to be below detection (<0.001 wt%) in all
159 cases, and are given a value of zero when determining Fe_{HR} concentrations.

160 It should be noted that all of the extractions are operationally-defined, and while specific
161 mineral groups are targeted in each step (see Table 1), the utility of the technique is not
162 predicated on the precise phases that are extracted. Rather, it is quantification of the total Fe_{HR}
163 pool, regardless of mineralogy (Alcott et al., 2020; Poulton, 2021), that has been calibrated in
164 terms of its reactivity towards dissolved sulfide in marine sediments (Canfield et al., 2002;
165 Raiswell and Canfield, 1998). We determined Fe_{HR} in two ways, whereby $Fe_{HR} = Fe_D + Fe_{py}$
166 and allows direct comparison with existing data for modern marine sediments (Raiswell and
167 Canfield, 1998) and riverine particulates (Canfield, 1997; Poulton and Raiswell, 2002), and
168 $Fe_{HR}^* = Fe_{carb} + Fe_{ox} + Fe_{mag} + Fe_{py}$, representing the latest approach for determining highly
169 reactive Fe (Poulton and Canfield, 2005).

170 **3. Results**

171 Results of major/trace element concentrations, Fe speciation and radiogenic Nd isotopes
172 (ϵNd) are presented in the Supplementary Materials. Sediments from the Bohai Sea, Yellow
173 Sea and East China Sea have low Fe_{HR}/Fe_T ratios (0.24 ± 0.04 , 1σ ; ranging from 0.15 to 0.32),
174 and higher Fe_{HR}^*/Fe_T ratios (0.34 ± 0.05 , 1σ ; ranging from 0.25 to 0.44) (Fig. 2). By contrast,
175 sediments from the Hainan continental margin have systematically higher Fe_{HR}/Fe_T ratios (0.46
176 ± 0.06 , 1σ ; ranging from 0.36 to 0.59), and Fe_{HR}^*/Fe_T ratios (0.59 ± 0.07 , 1σ ; ranging from 0.45
177 to 0.70).

178 The higher Fe_{HR}^*/Fe_T ratios relative to Fe_{HR}/Fe_T are as expected, due to the additional
179 extraction steps that are now performed to provide a more complete evaluation of phases that

180 are considered highly reactive towards dissolved sulfide (Poulton and Canfield, 2005).
181 However, a comparison of the two extraction procedures is instructive when considering the
182 nature of the phases extracted in each step. In some cases, Fe dissolved by the single dithionite
183 extraction (Fe_D) is similar to that dissolved by the sequential dithionite extraction (Fe_{ox} ; Fig. 2).
184 However, more often, Fe_D/Fe_T ratios (Bohai Sea, Yellow Sea and East China Sea = 0.24 ± 0.04 ;
185 Hainan Island = 0.46 ± 0.06) are closer to the sum of Fe_{carb} and Fe_{ox} (i.e., $(Fe_{carb}+Fe_{ox})/Fe_T$)
186 determined via the sequential extraction technique (Bohai Sea, Yellow Sea and East China Sea
187 = 0.28 ± 0.04 ; Hainan Island = 0.49 ± 0.06). Since these sediments have low carbonate contents,
188 these considerations suggest that the Fe_{carb} step is likely dominantly extracting dithionite-
189 soluble Fe, with only minor Fe dissolved from other phases. The Fe_D phases that are also
190 solubilized by the Fe_{carb} extraction may be in the form of surface-reduced Fe(II), which is
191 commonly prevalent during early diagenesis in locally reduced sediments (e.g., Poulton, 2003;
192 Xiong et al., 2019), or freshly-precipitated hydrous ferric oxide phases, formed either via in situ
193 diagenetic recycling or oxidation of reduced Fe phases during sampling and storage. As such,
194 these differences in the two extraction techniques are unlikely to be prevalent when applied to
195 ancient marine rocks, since surface-reduced Fe phases and freshly precipitated hydrous ferric
196 oxides ultimately age to more resistant phases during diagenesis, and hence would not be
197 dissolved as part of the Fe_{carb} fraction (Poulton and Canfield, 2005).

198 Nevertheless, consistent with other studies that have applied the more recent Poulton and
199 Canfield (2005) extraction technique to modern marine sediments (Hardisty et al., 2018),
200 Fe_{HR}^*/Fe_T ratios are generally below the 0.38 threshold used to identify water column anoxia,
201 which was defined using the less complete extraction procedure of Raiswell et al. (1994). By

202 contrast, the Hainan continental margin sediments have Fe_{HR}/Fe_T and Fe_{HR}^*/Fe_T ratios that
203 generally exceed global modern marine sediments (Fig. 2), suggesting that they are
204 geochemically distinct, regardless of the precise Fe extraction technique utilized.

205 Chemical index of alteration (CIA) values were calculated as $CIA = \text{mol } Al_2O_3 / (Al_2O_3 +$
206 $CaO + Na_2O + K_2O) \times 100$ (Nesbitt and Young, 1982). Values range from 52.3 to 62.4 for the
207 Bohai Sea, Yellow Sea and East China Sea sediments, and from 62.8 to 79.1 for the Hainan
208 offshore sediments. Sediment TiO_2 , Zr, Sc, Th and La concentrations were relatively variable,
209 and range from 0.40–1.73 wt% for TiO_2 , 27.5–200.2 ppm for Zr, 2.9–22.9 ppm for Sc, 4.4–29.0
210 ppm for Th, and 9.8–68.4 ppm for La. All of the studied sediments show relatively uniform ϵNd
211 values, ranging from -9.6 to -12.9.

212 **4. Discussion**

213 **4.1 Geochemical characteristics**

214 The studied sediments exhibit a strong correlation between Fe_T and Al concentrations (R^2
215 = 0.80, Fig. 3A), with the trend in Fe_T/Al ratios (0.56 ± 0.09 , 1σ) being analogous to modern
216 oxic marine sediments ($Fe_T/Al = 0.55 \pm 0.11$, 1σ ; Clarkson et al., 2014) and slightly higher than
217 continental detrital materials ($Fe_T/Al = 0.472 \pm 0.051$, 1σ ; Cole et al., 2017), which is consistent
218 with a dominant terrestrial source for the iron.

219 The linear relationships between both Fe_{HR} and Fe_{HR}^* as a function of Fe_T (Fig. 3B, C)
220 support a continental chemical weathering control on the Fe speciation characteristics, since
221 both Fe_T and highly reactive Fe increase as the intensity of chemical weathering increases
222 (Canfield, 1997; Poulton and Raiswell, 2002; 2005). However, the much steeper trends for
223 Hainan offshore sediments (Fig. 3B, C) demonstrate enhanced accumulation of highly reactive

224 Fe, relative to the other sites in this study and average oxic marine sediments (Raiswell and
225 Canfield, 1998).

226 While the Fe speciation systematics imply a chemical weathering control on the
227 geochemistry of the marine sediments deposited in the two broad regions (i.e., the Bohai Sea,
228 Yellow Sea and East China Sea region, and the Hainan offshore region), it is also important to
229 evaluate the provenance of the sediments, as markedly different source material may have
230 different chemical weathering characteristics. Where the primary major element contents (e.g.,
231 Ca, Na, K) and mineralogy of silicate rocks are intensely altered during chemical weathering,
232 the source rocks for fine grained marine sediments may still be constrained by conservative
233 trace element systematics (e.g., TiO_2/Zr ratios, La–Th–Sc relationships) and Nd isotope
234 compositions (McLennan et al., 1993; Fralick and Kronberg, 1997; Cullers and Podkovyrov,
235 2000; Li et al., 2003; Wei et al., 2012). Our sediments exhibit uniformly low ϵNd values (-11.2
236 ± 0.9 , 1σ), and variable but low TiO_2/Zr ratios (from 56 to 114 ppm) (Fig. 4A), suggesting an
237 intermediate–felsic source rock (cf. Fralick and Kronberg, 1997; Li et al., 2003).

238 We additionally examine the relationship between Th/Sc and Zr/Sc ratios (Fig. 4B), to
239 consider the potential effects of sediment recycling on geochemical composition, which is
240 based on the assumption that recycled sediments are preferentially enriched in zircon (giving
241 higher Zr/Sc ratios relative to Th/Sc ratios) (McLennan et al., 1993). Our studied sediments
242 align with the compositional trend of the parent rock (Fig. 4B), suggesting a more felsic
243 provenance without appreciable sediment recycling. Furthermore, the La–Th–Sc distribution
244 indicates that the sediments are more likely derived from granite- and granodiorite-type source
245 rocks (Fig. 5A), which is consistent with previous provenance analyses for the sediments of

246 these areas (Wei et al., 2012). In conclusion, the marine sediments analyzed in this study are
247 dominantly derived from the chemical weathering of intermediate–felsic rocks, with no
248 significant local control on the provenance of the sediments deposited in different regions off
249 the east coast of China.

250 **4.2 A link between chemical weathering and Fe_{HR} accumulation in Hainan offshore** 251 **sediments**

252 Based on the above provenance analysis, the significant enrichments in highly reactive Fe
253 in Hainan offshore sediments are unlikely to be a consequence of different source rock types.
254 We can also discount a water column redox control on the relative enrichments, as all sediments
255 were deposited in bottom waters with $[O_2] > 30 \mu M$, despite seasonal seawater hypoxia in some
256 areas off the east China coast (Chen et al., 2007; Zhao et al., 2017).

257 Continental margin sediments do, however, commonly display relative enrichments in
258 Fe_{HR} in surface sediments, due to upwards diffusion of Fe^{2+} after mobilization in anoxic
259 sediment porewaters, followed by precipitation as Fe (oxyhydr)oxides close to the sediment-
260 water interface (Raiswell and Canfield, 1998; Poulton, 1998). Indeed, the database for modern
261 oxic continental margin sediments (Raiswell and Canfield, 1998) gives an average increase of
262 ~15% (with some increases at >50%) in Fe_{HR}/Fe_T ratios for surface sediments, relative to
263 sediments from deeper in the same core section. Depth profiles were not analyzed in the present
264 study, and thus it remains likely that the surface sediment Fe_{HR}/Fe_T ratios are inflated by
265 upwards diffusion of Fe^{2+} . Thus, the steady-state Fe_{HR}/Fe_T ratios preserved in deeper sediment
266 piles, and indeed in the average rock record, will likely be slightly lower than those of the
267 surface sediments. For example, a 15% decrease for the Hainan offshore region lowers the

268 range in Fe_{HR}/Fe_T ratios to 0.31–0.50. However, the global data-base of Raiswell and Canfield
269 (1998) also included surface sediment samples, with a maximum Fe_{HR}/Fe_T ratio of ~0.38, and
270 thus the measured Fe_{HR}/Fe_T ratios for Hainan region sediments (0.36 to 0.59) are clearly
271 generally elevated relative to normal marine sediments.

272 Soils on Hainan Island are characterized by widespread formation of intensely weathered
273 Fe-rich laterite and bauxitic laterite (e.g., Yu et al., 2019), consistent with a chemical weathering
274 control on Fe_{HR}/Fe_T ratios. To further investigate this potential control, we utilize the chemical
275 index of alteration (CIA) as a weathering proxy (Nesbitt and Young, 1982; McLennan, 1993;
276 Fedo et al., 1995). During chemical weathering, labile cations (e.g., Ca^{2+} , Na^+ , K^+) are
277 preferentially dissolved and transported from the weathering profile, leaving the regolith
278 enriched in less mobile elements, such as Al, Fe and Ti. Thus, utilizing an A–CN–K (i.e., Al–
279 Ca+Na–K) ternary diagram, where primary silicate minerals tend to have CIA values around
280 50 (Fig. 5B), increased CIA values correspond to elevated degrees of chemical weathering, due
281 to sequential loss of Ca, Na and K from primary minerals followed by the formation of
282 secondary clays (e.g., feldspar–illite/smectite–kaolinite series) (Nesbitt and Young, 1982; Fedo
283 et al., 1995).

284 Furthermore, since marine siliciclastic sediments comprise chemical weathering products
285 and fine-grained rock debris from physical erosion, the CIA values of marine sediments also
286 reflect the extent of chemical weathering in the source region (i.e., the relative proportions of
287 chemical weathering products deposited in bulk sediments). Given a granodiorite–granite
288 source rock type (Fig. 5A), changes in CIA values for the studied sediments reproduce a trend
289 consistent with an increasing degree of chemical weathering of the source rocks (solid arrow in

290 Fig. 5B; McLennan, 1993), without any discernable effect on the CIA values from K
291 metasomatism (dashed line in Fig. 5B; cf. Fedo et al., 1995).

292 Next, we investigate the relationship between highly reactive Fe (here, we focus on Fe_{HR}
293 concentrations to allow direct comparison with previously published data for river particulates,
294 but note that the same relationships are evident when considering Fe_{HR}^*) and the CIA proxy, in
295 order to document potential links between Fe_{HR} accumulation and chemical weathering (Fig.
296 6). Surface sediments from the east China coast exhibit a strong relationship between Fe_{HR}/Fe_T
297 and CIA values ($R^2 = 0.85$). However, in more detail, sediments from Bohai Sea, Yellow Sea
298 and East China Sea have relatively low CIA values (52–62) and Fe_{HR}/Fe_T ratios (0.15–0.32),
299 indicating relatively weak chemical weathering. By contrast, the Hainan region sediments have
300 significantly higher CIA values (63–79), coincident with the higher range observed in Fe_{HR}/Fe_T
301 ratios (0.36–0.59). This clearly supports the chemical weathering control on the speciation of
302 Fe in river particulates that has previously been documented for both individual US rivers
303 (Canfield, 1997), and continental-scale river catchments (Poulton and Raiswell, 2002).

304 Indeed, the global riverine data also show a clear positive correlation between Fe_{HR}/Fe_T
305 ratios and CIA values ($R^2 = 0.48$), and the marine sediments from the east coast of China overlap
306 with the range of global river particulate data (Fig. 6). This suggests that the source rocks in the
307 region weather in a manner that is typical of global scale weathering trends, specifically in
308 terms of the generation of increased Fe_{HR}/Fe_T ratios as chemical weathering intensity increases.
309 In this context, the observed enrichment in Fe_{HR} in the Hainan offshore sediments is consistent
310 with the widespread formation of lateritic or bauxitic lateritic soils in sub-tropical and tropical
311 regions, as found on both the southeastern mainland of China, and Hainan Island (Lan et al.,

312 2003; Qiu et al., 2014; Yu et al., 2019), with both soil types commonly being enriched in Fe
313 (oxyhydr)oxide minerals (Schwertmann, 1988; Widdowson, 2007).

314 **4.3 Weathering-induced Fe_{HR} enrichments in modern and ancient marine sediments**

315 Our data demonstrate that it is possible for enhanced chemical weathering to result in a
316 locally high influx of Fe (oxyhydr)oxide minerals to the marine environment, which has
317 implications for the local biogeochemistry of marine waters and sediments. However, enhanced
318 chemical weathering is not sufficient, in isolation, to result in enrichments in highly reactive Fe
319 relative to normal oxic marine sediments. As discussed above, global river particulates are also
320 enriched in highly reactive Fe, but preferential trapping of Fe oxyhydr(oxide) minerals in inner-
321 shore environments results in marine sediments that have significantly lower Fe_{HR}/Fe_T ratios
322 on a more global scale (Poulton and Raiswell, 2002). Our data thus specifically document a
323 missing factor in understanding of the modern global iron cycle, whereby the supply of Fe_{HR}
324 from (sub)tropical mountainous regions (e.g., Hainan Island) may be transported directly onto
325 active margins, thus supplying sediment to the marine environment that is relatively enriched
326 in Fe (oxyhydr)oxide minerals.

327 We compare our data to other oxic and anoxic environments in Fig. 7, which shows a
328 global compilation of both Fe_{HR}/Fe_T (Fig. 7A) and Fe_{HR}*/Fe_T (Fig. 7B) ratios for different
329 sediment types and localities, plotted as a function of Fe_T/Al ratios. It is clear from Fig. 7 that
330 a very limited range of oxic environments record weathering-induced enrichments in Fe_{HR}. In
331 addition to the (sub)tropical mountainous region represented by Hainan Island, these include
332 restricted basins in (sub)tropical areas, such as the Gulf of Aqaba in the Red Sea. The defining
333 characteristic of these settings, is that not only does the sediment originate from highly

334 weathered (sub)tropical catchments, but the inner-shore trap for Fe_{HR} is bypassed, resulting in
335 a direct influx of Fe_{HR} -enriched sediment.

336 The sediments off Hainan Island thus appear to be influenced by a ‘perfect storm’ of
337 combining factors, whereby the source area was in a (sub)tropical locality, and the inner-shore
338 trap for Fe_{HR} was bypassed due to the mountainous terrain that extends to the coast. These
339 observations are consistent with the suggestion that significant enrichments in Fe_{HR} in ancient,
340 open marine oxic settings are likely to be relatively rare through the geological record (Poulton,
341 2021). However, such enrichments remain a distinct possibility on a local scale at times when
342 chemical weathering rates were particularly high (e.g., in the run-up to the Permian–Triassic
343 boundary; Schobben et al., 2020). Given the common utility of Fe_{HR}/Fe_T ratios as a proxy for
344 identifying local bottom water redox conditions (Poulton and Canfield, 2011), it is clearly
345 important to attempt to recognize weathering-induced enrichments in Fe_{HR}/Fe_T in the rock
346 record.

347 An important factor in identifying criteria for recognition of weathering-induced
348 enrichments in Fe_{HR}/Fe_T concerns the differing behavior of Fe_{HR}/Fe_T and Fe_T/Al ratios during
349 chemical weathering. While both proxies may increase under anoxic conditions, due to the
350 transport and subsequent precipitation of Fe_{HR} minerals in the water column (Canfield et al.,
351 1996; Raiswell and Canfield, 1998), only Fe_{HR}/Fe_T ratios are appreciably enhanced during
352 chemical weathering (Canfield, 1997). This is because both Fe_T and Al behave conservatively
353 during chemical weathering, to approximately the same degree (Martin and Meybeck, 1979;
354 Canfield, 1997). Thus, as demonstrated in Fig. 7, all samples from the east China coast, as well
355 as global river particulates and sediments from Gulf of Aqaba, plot in the average Fe_T/Al range

356 characteristic of oxic depositional conditions (0.44–0.66; Clarkson et al., 2014), despite some
357 of the samples having elevated Fe_{HR}/Fe_T ratios due to enhanced chemical weathering.

358 The above observations suggest that a combination of elevated Fe_{HR}/Fe_T (>0.38) and low
359 Fe_T/Al may provide an initial indication of a possible chemical weathering, rather than water
360 column anoxia, control on sedimentary Fe speciation characteristics. However, we stress here
361 that the local Fe_T/Al ratio of sediments may be highly variable (Cole et al., 2017; Raiswell et
362 al., 2018), and indeed, in settings where the Fe_T/Al ratio of terrestrial materials supplied to the
363 marine environment is low, the addition of Fe_{HR} from anoxic water column precipitation may
364 not be sufficient to increase Fe_T/Al ratios above the upper range that is typical of oxic marine
365 sediments. This may give a false indication of a chemical-weathering control on Fe speciation
366 characteristics. We demonstrate this likelihood in Fig. 7, where data from anoxic modern basins
367 are also plotted. While some anoxic modern basins show the expected elevated Fe_{HR}/Fe_T and
368 Fe_T/Al ratios diagnostic of anoxic depositional conditions, sediments from the anoxic Cariaco
369 Basin, as well as some sediments from the anoxic Orca Basin, have relatively low Fe_T/Al ratios.
370 Thus, wherever possible, a background oxic Fe_T/Al ratio should be defined for the particular
371 setting of interest, to allow a more refined evaluation of potential enrichments in Fe_T/Al
372 (Sperling et al., 2016; Cole et al., 2017; Raiswell et al., 2018; Poulton, 2021).

373 While it may be possible to define a local oxic Fe_T/Al ratio in some paleoenvironmental
374 settings (e.g., Poulton et al., 2010; Sperling et al., 2016; Scholz et al., 2019a), in many cases
375 this is not straightforward. However, it is increasingly the case that a multi-proxy approach is
376 taken to reconstruct ancient water column redox conditions, whereby enrichments in redox
377 sensitive trace metals may provide independent evidence for anoxic depositional conditions

378 (for reviews see Robbins et al., 2016; Bennett and Canfield, 2020). In this context, however,
379 we note that anoxic enrichments in Fe_{HR}/Fe_T may arise in two distinct ways. Firstly, Fe_{HR}
380 enrichments may arise from precipitation in, and settling through, an anoxic water column (e.g.,
381 Canfield et al., 1996). Under such conditions, and notwithstanding possible diminished marine
382 trace metal inventories when anoxia is widespread (e.g., Algeo, 2004; Scott et al., 2008;
383 Goldberg et al., 2016), certain trace metals (dependent on the precise redox state) would be
384 expected to be enriched in the deposited sediments, and as such would provide supporting
385 evidence for an anoxic source for Fe_{HR} enrichments. However, Fe_{HR} enrichments may also arise
386 through upwelling of dissolved Fe^{2+} into shallow oxic waters (e.g., Wood et al., 2015). Under
387 such conditions, deposition occurs under oxic conditions, and Fe_{HR} enrichments actually reflect
388 deeper water anoxia (Poulton and Canfield, 2011; Poulton, 2021). In this case, trace metal
389 enrichments at the site of Fe_{HR} enrichment may therefore not necessarily occur, which adds an
390 extra dimension of complexity that should be considered when attempting to distinguish
391 chemical weathering-induced enrichments in Fe_{HR} from anoxic enrichments.

392 In addition, it is clearly important to have direct evidence for an enhanced degree of
393 chemical weathering, as in the present study, when invoking a chemical weathering rather than
394 water column anoxia control on Fe speciation systematics. Given that enrichments in Fe_{HR}/Fe_T
395 ratios due to enhanced chemical weathering are likely relatively rare in the geological record,
396 we propose that four principal criteria should be satisfied in order to invoke this control: 1.
397 Sediments should display persistent (not fluctuating) enrichments in Fe_{HR}/Fe_T (>0.38); 2.
398 Sediments should have low Fe_T/Al ratios, and where possible these ratios should be evaluated
399 relative to the local ratio for oxic deposition; 3. The sediments should display no independent

400 evidence for deposition under anoxic conditions, for example via additional consideration of
401 trace element (noting the caveat to this discussed above) or biomarker systematics; 4.
402 Independent evidence for a relatively high degree of chemical weathering should be provided,
403 for example via use of the CIA weathering proxy, mineralogical considerations, or isotope
404 systematics.

405 **4.4 Implications for the modern oceanic Fe cycle**

406 Our demonstration that (sub)tropical mountainous rivers may provide elevated fluxes of
407 highly reactive Fe to adjacent shelves has implications for the modern oceanic Fe cycle. Recent
408 work has demonstrated that the significance of glacial catchments has commonly been
409 underappreciated in terms of supply of highly reactive Fe to the ocean, with such sites
410 ultimately constituting an important source of potentially bioavailable Fe to fuel primary
411 productivity (e.g., Raiswell et al., 2016). Similarly, elevated oceanic influxes of highly reactive
412 Fe from (sub)tropical mountainous regions also constitute an additional source of potentially
413 bioavailable Fe to the marine environment, which through subsequent mobilization from
414 sediments may impact the global budget of bioavailable Fe and thus help to regulate global
415 ocean primary productivity (e.g., Tagliabue et al., 2017; Homoky et al., 2021). In addition, Fe
416 (oxyhydr)oxide minerals exert an important control on the preservation and burial of organic
417 carbon (e.g., Lalonde et al., 2012), and thus enrichments in Fe_{HR} in continental margin
418 sediments deposited adjacent to (sub)tropical mountainous regions may enhance regional rates
419 of organic carbon burial.

420 **5. Conclusions**

421 Our new iron speciation data for surface seafloor sediments deposited from oxic bottom
422 waters on the east coast of China, provide new insight into potential controls on highly reactive
423 Fe accumulation in continental margin settings. Surface sediments from the Bohai Sea, Yellow
424 Sea and East China Sea have low Fe_{HR} contents ($Fe_{HR}/Fe_T = 0.24 \pm 0.04$; $Fe_{HR}^*/Fe_T = 0.34 \pm$
425 0.05 , 1σ), which are similar to those of typical oxic marine settings ($Fe_{HR}/Fe_T = 0.26 \pm 0.08$, 1σ ;
426 Raiswell and Canfield, 1998). However, surface sediments from the Hainan offshore area in the
427 northern South China Sea have significantly elevated Fe_{HR} contents ($Fe_{HR}/Fe_T = 0.46 \pm 0.06$
428 and $Fe_{HR}^*/Fe_T = 0.59 \pm 0.07$, 1σ), despite deposition from oxic bottom waters.

429 Nd isotope compositions and TiO_2/Zr ratios demonstrate a similar dominant source rock
430 composition for sediments deposited in the two regions, and Th/Sc and Zr/Sc ratios additionally
431 suggest limited sediment recycling. Thus, based on a close relationship between Fe_{HR}/Fe_T ratios
432 and the CIA weathering proxy, we suggest that the elevated Fe_{HR}/Fe_T ratios of the Hainan region
433 are a direct result of an enhanced degree of chemical weathering, coupled with the mountainous
434 terrain, which results in Fe (oxyhydr)oxide minerals bypassing the inner shore trap and
435 depositing directly on the local continental margin. We propose that a carefully considered
436 combination of high Fe_{HR}/Fe_T , low Fe_T/Al , no independent evidence for water column anoxia,
437 and evidence for a high degree of chemical weathering, may be used to indicate a possible
438 chemical weathering control on Fe_{HR} enrichments in marine sediments preserved in the rock
439 record. The observed linkage between chemical weathering and enhanced terrestrial Fe_{HR}
440 delivery from (sub)tropical mountainous regions provides new insight into the modern oceanic
441 Fe cycle, with wider potential implications for a variety of biogeochemical elemental cycles in
442 the ocean.

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654 **Figure Captions**

655 **Figure 1.** Location of the surface sediment samples from the Bohai Sea, Yellow Sea and East
656 China Sea (A), and the Hainan offshore area (B). The dashed lines represent major rivers in the
657 studied regions.

658 **Figure 2.** Iron speciation data shown as a proportion of total Fe for the Bohai, Yellow and
659 East China Seas (A and B), and for Hainan offshore sediments (C and D). Data are reported
660 for sequential extractions (A and C) using the method of Poulton and Canfield (2005), and for
661 a single sodium dithionite extraction (B and D) using the method of Raiswell et al. (1994).
662 See Table 1 for operationally-defined target phases. Blue lines in (A) and (C) represent Fe_{HR}^* ,
663 calculated as the sum of $Fe_{carb} + Fe_{ox} + Fe_{mag} + Fe_{py}$ (Fe_{py} were below detection). Blue lines in
664 (B) and (D) represent Fe_{HR} determined via a single sodium dithionite extraction. Dashed lines
665 represent the upper limit for normal marine sediments deposited under oxic conditions
666 ($Fe_{HR}/Fe_T = 0.38$; Raiswell and Canfield, 1998), and the average value for oxic marine
667 sediments ($Fe_{HR}/Fe_T = 0.26$ Raiswell and Canfield, 1998).

668 **Figure 3.** Cross-plots of (A) Fe_T vs. Al, (B) Fe_{HR} vs. Fe_T , (C) Fe_{HR}^* vs. Fe_T , and (D) Fe_{HR}^* vs.
669 Fe_{HR} . The R^2 values represent the coefficient of determination for the correlations (2σ
670 uncertainties, dashed lines). The red lines in (A) represent the overall trend for modern oxic
671 marine sediments ($Fe_T/Al = 0.55$; Raiswell and Canfield, 1998; Raiswell et al., 2018) and
672 average terrestrial materials ($Fe_T/Al = 0.472$; Cole et al., 2017). The red lines in (B) represent
673 the overall trend for modern riverine sediments ($Fe_{HR}/Fe_T = 0.43$; Poulton and Raiswell, 2002)
674 and oxic marine sediments ($Fe_{HR}/Fe_T = 0.26$; Raiswell and Canfield, 1998). The red line in
675 (C) represents the upper threshold value for oxic depositional conditions ($Fe_{HR}^*/Fe_T = 0.38$)
676 (Poulton and Canfield, 2011).

677 **Figure 4.** Cross-plots of trace elements and Nd isotope values for the studied sediments. (A)
678 ϵNd vs. TiO_2/Zr ; (B) Th/Sc vs. Zr/Sc . The dashed lines in (B) represent varying trends for
679 compositional changes in parent rocks and for sediment recycling (after McLennan et al., 1993).

680 **Figure 5.** Ternary plots of the studied sediments. (A) La–Th–Sc diagram; (B) A–CN–K (Al–
681 Ca + Na–K) diagram with calculated CIA values. Red stars represent the distributions of
682 molar Al₂O₃, CaO+N₂O, and K₂O for different primary silicate minerals and igneous rocks.
683 The solid arrow represents the trend for increasing degree of chemical alteration of primary
684 silicate minerals in this study. The dashed arrow denotes the chemical alteration process with
685 effects of K addition via K metasomatism in sedimentary rocks or paleosols (after Fedo et al.,
686 1995).

687 **Figure 6.** Relationship between the Fe_{HR}/Fe_T ratios and CIA values for the China east coast
688 sediments analyzed in this study and global riverine sediments (modified from Canfield,
689 1997; Poulton and Raiswell, 2002). The division of different chemical weathering degrees is
690 modified from Rieu et al. (2007), Schoenborn and Fedo (2011), and Wang et al. (2020). The
691 R² values represent the coefficient of determination for the correlations (2σ uncertainty;
692 dashed lines).

693 **Figure 7.** Compilation of Fe_{HR}/Fe_T (A) and Fe_{HR}*/Fe_T (B) ratios as a function of Fe_T/Al for
694 the studied sediments, modern riverine sediments, and published marine sediment data
695 (presented as average values with 1σ uncertainty). The upper threshold Fe_{HR}/Fe_T value for
696 recognition of oxic sedimentation (0.38) is shown in both A and B, since this value is
697 currently used to identify anoxic deposition, regardless of the precise Fe extraction scheme
698 used. The range in Fe_T/Al for average oxic modern marine sediments is 0.55 ± 0.11 (1σ)
699 (Clarkson et al., 2014). The range in Fe_T/Al for terrestrial materials is 0.472 ± 0.051 (1σ)
700 (Cole et al., 2017), and is shown as a gray area in (A) and (B). Data sources: China east coast
701 sediments (this study), global riverine sediments (Canfield, 1997; Poulton and Raiswell,

702 2002), oxic marine sediments (average continental margin and deep sea; Raiswell and
703 Canfield, 1998), dysoxic marine sediments (Raiswell and Canfield, 1998), tropical semi-
704 enclosed bays (Praslin and Savannes Bay), Saint Lucia, Lesser Antilles (Ku et al., 2008), Gulf
705 of Aqaba, Red Sea (Steiner et al., 2017; Boyko et al., 2019), FOAM, Long Island Sound
706 (Hardisty et al., 2018), Black Sea, Orca Basin, Effingham Inlet (Andersen and Raiswell,
707 2004; Lyons and Severmann, 2006), Cariaco Basin (Lyons et al., 2003), Mediterranean
708 sapropels (Benkovitz et al., 2020), Peru margin (Scholz et al., 2014), Lake Cadagno (Xiong et
709 al., 2019). Note that basins influenced by strong hydrothermal activity (e.g., Guaymas Basin,
710 Gulf of California; Scholz et al., 2019b) are excluded from the compilation.