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1 **DOC and Nitrate fluxes from farmland; impact on a dolostone aquifer KCZ**

2

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16

17 **Abstract**

18 DOC and nitrate in farmland represent key chemical species that determine the water  
19 quality in the Karst Critical Zone (KCZ). The work reported here focuses on quantifying  
20 fluxes of these species in an experimental farm site (University of Leeds Farm, UK)  
21 overlying a dolomitic karst aquifer of Permian age. In this research, the Transect Method  
22 was applied for the first time to farmland by combining hydrochemical data from soil and  
23 groundwater for computation of mass fluxes. The Transect Method, developed for  
24 management of industrially contaminated sites, was applied to a farm source due to the  
25 presence of localised contamination from application of pig slurry.

26 Required inputs for our approach include concentrations of nitrate and DOC in soil water  
27 and groundwater, net recharge flux (here derived from a MODFLOW-2005 model) and  
28 local hydraulic gradient and conductivity measurements. Key outputs are fluxes and  
29 downstream groundwater concentrations of DOC and nitrate. Downstream concentrations  
30 were validated against direct groundwater measurements, demonstrating the veracity of  
31 the approach. The approach shows that the localised contamination has a significant  
32 impact on both concentrations of nitrate and DOC in groundwater, although the DOC  
33 impact is greater, because the upstream land uses also produce nitrate as a result of  
34 agricultural practices that are widespread in the region.

35 The results of the study also constrain the zone vulnerable to contamination to the upper  
36 ~40 m below the ground surface. Future modelling efforts on solute contaminant transport  
37 should focus on this shallow vulnerability zone (0-40 mBGL) and the Transect Method  
38 applied in this work can be used to define boundary conditions.

39 Hence, following this research, we envisage to export a generic approach that combines  
40 physical flow parameters and hydrochemical analyses for computation of subsurface mass  
41 fluxes using the Transect Method, to identify the degree of impact of specific point sources  
42 and to support conceptualization and modelling of contaminant transport in the KCZ of  
43 farm areas.

44

45 **Keywords** Karst Critical Zone, Farmland, DOC, Nitrate, Mass Flux, Contaminant  
46 Transport

47

## 48 **1. Introduction**

49 The Critical Zone is the thin surface layer that extends from the top of the vegetation to the  
50 bottom of active groundwater circulation driven by meteorological recharge and supplies  
51 humans with most life-sustaining resources (Banwart et al. 2011, 2017; Anderson et al.

2014; Brantley et al. 2017; Keller 2019). Karst aquifers of carbonate origin represent a source of drinking water that supplies a quarter of the world's population (Ford and Williams 1989; Hartman et al. 2014). This category of fractured aquifers is particularly prone to dissolution that enlarges bedding planes, fractures and faults. Consequently, karst aquifers are characterized by a high degree of hydraulic connectivity with the land surface and transport of contaminants is therefore particularly rapid (Ford and Williams 1989; Worthington et al. 2012; Goldscheider and Drew 2014; Medici et al. 2016; Borović et al. 2019; Torresan et al. 2020). The Critical Zone in karst environment is considered as the most vulnerable to contamination due to the above mentioned dissolution processes in the vadose and shallow saturated zone of carbonate aquifers (Lian et al. 2011; Kogovsek and Petric 2014; Zhang et al. 2017; Jiang et al. 2019; Jourde et al. 2018; Green et al. 2019; Sullivan et al. 2019). Stress on groundwater resources of the KCZ has increased in recent decades in agricultural areas in terms of (i) quantity due to excessive abstraction of groundwater for irrigation, and (ii) water quality due to pollution from agricultural practices (Maheler and Massei 2007; Bicalho et al. 2011; Huebsch et al. 2013; Goldscheider and Drew 2014; Hartman et al. 2014). The research presented in this paper focuses on the second aspect of interest to the KCZ international community that is represented by protection of water resources from contamination.

Applications of N-fertilizer and manure have increased crop yields while also increasing nitrate concentrations in ground and surface water. For example, Defra (2002) estimated that between 70 and 80% of nitrate in English surface and groundwater derives from agricultural activities. Other sources of nitrate include atmospheric deposition, discharge from septic tanks and leaking sewers, the spreading of sewage sludge to land and seepage from landfills (Wakida and Lerner, 2005; Fezzi et al. 2010; Hutchins 2012). Elevated nitrate in groundwater has implications for human health. Therefore, quantification of nitrate fluxes in the unsaturated zone and groundwater across agricultural

78 areas is a key element of managing water resources. However, few studies have  
79 effectively integrated water quality data and hydrogeological information to estimate fluxes  
80 of nitrate, or other pollutants, from agricultural soils to the unsaturated zone and shallow  
81 groundwater (Liao et al. 2012; Green et al. 2018). To integrate these research areas  
82 dedicated to nitrate pollution, we present a baseline hydro-chemical analysis of soil and  
83 groundwater water which we use to compute both DOC and nitrate mass fluxes for the  
84 University of Leeds (UoL) Farm, UK (Figs. 1a-c).

85 The amount of nitrate available for leaching from the soil is related to the amount, timing  
86 and method of application of inorganic fertilizers, slurry and/or farmyard manure to  
87 agricultural land (Foster 1976; Sieling et al. 1997; Lord et al. 1999; Williams et al. 2000;  
88 Wang et al. 2016). Furthermore, the rate of nitrate leaching through the soil is controlled by  
89 texture, with sandy soils allowing more leaching through larger better connected pore  
90 spaces than clay rich soils (Goss et al. 1998). The fraction of applied nitrogen that is  
91 actually leached to groundwater (“leaching fraction”) from the soil zone is a key parameter  
92 linking nitrogen applications to groundwater nitrate concentrations and typically ranges  
93 between 5 and 50% (Liao et al. 2012; Green et al. 2018). Nitrate leaching will also depend  
94 on the extent of denitrification within the soil zone, which is function of the nitrate and DOC  
95 concentrations, fluid temperature, pH and alkalinity and is therefore strongly dependant on  
96 the characteristics of the study site (Panno et al. 2001; Rivett et al. 2008; Mellander et al.  
97 2012; Yang et al. 2020). Denitrification in groundwater in limestone aquifers can also occur  
98 (Panno et al. 2001) but is likely to be limited at our field site due to relatively low  
99 temperatures, dolostone lithology and rapid fracture flow (Rivett et al., 2008; Moon et al,  
100 2006). Furthermore, Siemens et al. (2003) found in their study that DOC leached from  
101 agricultural soils contributed negligibly to the denitrification of nitrate in groundwater  
102 because the DOC derived from the soil was not bio-available. Toxic hydrophobic organic  
103 contaminants are released by pesticides and form complexes with DOC that facilitates

104 their transport through the soil to the aquifer below (Huang et al. 1998; Weber et al. 1998;  
105 Wang et al. 2004, 2016, 2018). Given the role of DOC in influencing both sorption and  
106 biodegradation process its importance in influencing groundwater chemistry has been  
107 increasingly recognised (Allen-King et al. 2002). Hence, detection of the most vulnerable  
108 aquifer zone to infiltration and transport of both DOC and nitrate is included in this  
109 research.

110 To date, the Critical Zone scientific network has primary focused on weathering processes  
111 and the geochemical properties of soil and vadose zone using laboratory experimental and  
112 modelling approaches (e.g., Emblanch et al. 2003; Falcone et al. 2008; Peyraube et al.  
113 2012, 2013; Dong et al. 2018; Lerch et al. 2018; Zhou et al. 2019; Tremosa et al. 2020).  
114 Consequently, there is need for more field scale monitoring and hydraulic testing to  
115 characterize the unsaturated and saturated zone as recently highlighted by Critical Zone  
116 scientists Kuntz et al. (2011) and Jourde et al. (2018), and shown in this work.

117 Previous hydrogeological research at the UoL Farm consists of multi-level slug tests  
118 (Medici et al. 2019a), pumping tests (Allen et al. 1997) and development of a regional  
119 scale steady state MODFLOW-2005 groundwater flow model (Medici et al. 2019b) of the  
120 local dolostone aquifer of Permian age. The three formations (Cadeby, Edlington and  
121 Brotherton illustrated in Figure 1c) of the Mangnesian Limestone Group represents  
122 separate layers in the model. This groundwater flow model accounts for turbulent flow by  
123 inserting the Conduit Flow Process Type-1 (*sensu* Hill et al. 2010) developed by the USGS  
124 for karst systems in correspondence of streams and normal faults. Indeed, transmissivities  
125 from pumping tests in correspondence of streams and faults is one order of magnitude  
126 higher than the host rock due to the presence of karst conduits 0.10-0.20 m large of  
127 approximate pipe shape (Medici et al. 2019b). This modelling research has produced  
128 calibrated recharge and hydraulic conductivity values that represent inputs for the workflow  
129 presented here (Medici et al. 2019a, b).

130 Nitrate and DOC represent two chemical species, which are considered diagnostic of  
131 water quality pollution in karst environments in areas of the world characterized by intense  
132 agricultural activities (Ryan and Meiman 1996; Lastennet and Mudry 1997; Goldscheider  
133 and Drew 2014). The aim of this research is therefore to show how the combination of  
134 groundwater hydrological fluxes with soil water chemistry, baseline groundwater  
135 hydrochemistry and monitoring supports reliable computation of DOC and nitrate fluxes  
136 from specific agricultural activities which represent point source inputs, developing a  
137 method that can be applied in other farmlands overlying KCZs. The specific research  
138 objectives were: (i) determine baseline hydrochemical analysis of soil, spring and stream  
139 water and groundwater in the KCZ under the UoL Farm, (ii) detect depth of penetration of  
140 farm-derived nitrate and DOC in the sub-surface, and (iii) compute and validate nitrate and  
141 DOC mass fluxes and downstream concentrations in groundwater in an area of farming  
142 activity.

143

## 144 **2 Field site**

### 145 *2.1 Bedrock geology*

146 The experimental site of the UoL Farm (see Figs. 1a-d, 2) is located in Yorkshire (NE  
147 England, UK) between the cities of Leeds and York. Here, the Magnesian Limestone  
148 Group represents the major aquifer and the bedrock lithology. This geological group of  
149 Lower Permian age is comprised of dolomitic limestone, dolostone, halite and gypsum  
150 rocks derived from shallow water sedimentation at the margins of the Zechstein Basin (Fig.  
151 1a; Aldrick 1978; Smith et al. 1986). In NE England, the stratigraphic succession of the  
152 Magnesian Limestone Group is typically 120 m thick (Cooper and Lawley 2007). This  
153 geological group is formally sub-divided into three different formations: the Cadeby,  
154 Edlington and Brotherton formations (Fig. 2a; Smith et al. 1986). The Cadeby Formation  
155 that represents the focus of this research is characterized, with the exception of the basal

156 5 meters of marls, by thinly bedded dolostone showing ooids, peloids, corals and bivalves  
157 (Tucker 1991; Mawson and Tucker 2009). The Brotherton Formation above represents the  
158 uppermost part of the Magnesian Limestone Group in Yorkshire and is characterized by  
159 thinly bedded dolomitic limestones with ooids, algae and bivalves. However, the Edlington  
160 Formation is characterized by both halite and gypsum (Smith et al., 1986).

161 Higher dolomitisation characterizes the Cadeby Formation (54%  $\text{CaCO}_3$ ; 46%  $\text{MgCO}_3$ ) that  
162 represents the focus of this study. In contrast, the dolomitic limestone of the Brotherton  
163 Formation is more abundant in calcite (65%  $\text{CaCO}_3$ ; 35%  $\text{MgCO}_3$ ), hence more prone to  
164 dissolution and permeability development (Allen et al. 1997; Lott and Cooper 2005). In NE  
165 England around the area of study (Figs. 1c, 2a), the tectonic structures which characterize  
166 the UK Magnesian Limestone Group are represented by normal faults and non-  
167 stratabound joints (*sensu* Odling et al. 1999). Outcrop studies and seismic lines carried out  
168 near the field site show how normal faults of Mesozoic age are mainly oriented ENE-WSW  
169 (Fig. 1c, 2a). Lack of significant effects of the Cenozoic Alpine orogenesis results in the  
170 gentle dip ( $< 5^\circ$  towards E) of the Permo-Mesozoic deposits in the study area (Fig. 1c, d,  
171 2a; Murphy 2000).

172 Boreholes drilled in non-faulted sections of the Cadeby Formation show high angle joints  
173 (dip  $50^\circ$ - $80^\circ$ ) which cross-cut the bedding parallel fractures (Medici et al. 2019a). Cavities  
174 of karstic origin up to 0.6 m large were detected in correspondence of fault zones in the  
175 study area (Cooper and Lawley 2007; Medici et al. 2019b). Vuggy porosity was detected at  
176 the field site in cores drilled in the vadose zone (Murphy 2000; Murphy and Cordingly  
177 2010). Discontinuity surveys indicate 0.7 and 0.3 mm average mechanical apertures for  
178 sub-vertical joints and bedding plane, respectively (Medici et al. 2019b). Note that,  
179 availability of a seismic survey, cores and quarry outcrops allows good spatial constraints  
180 on geological formations and presence of faults in the UoL Farm area (Fig. 2a; Cooper and  
181 Lawley 2007).

182

183 *2.2 Land use and hydrogeology*

184 The aquifer-unit of the Cadeby Formation is unconfined at the UoL Farm site due to the  
185 presence of only ~1 m thick Quaternary cover. The soil above the dolostone is a well-  
186 drained, loamy, calcareous brown earth type from the Aberford series of Calcaric Edoleptic  
187 Cambisols, and ranges in depth from 0.5 to 0.9 m (Holden et al. 2019). This soil type  
188 occurs extensively across the UK on gently sloping Permian and Jurassic Limestone and  
189 is mainly used for arable farming. The farm is comprised of 294 ha, the majority (264 ha)  
190 of which is arable with the remainder under grass. The arable fields have been in  
191 continuous cultivation and cropping since 1994 using a rotation of winter wheat (x2), spring  
192 or winter barley and oilseed rape, with the periodic inclusion of vining peas or potatoes.  
193 The grass fields are used for sheep grazing and some are cut for silage up to twice per  
194 year. Approximately 150 kg N ha<sup>-1</sup> is applied to the cereal crops as fertiliser in spring with  
195 an additional 40 kg N ha<sup>-1</sup> applied in the autumn as pig slurry after harvest. The grass  
196 fields receive 100-130 kg N ha<sup>-1</sup> as fertiliser in spring and an additional 50 kg N ha<sup>-1</sup> from  
197 pig slurry after silage harvest in June (Holden et al. 2019; Ward 2020). The UoL Farm is  
198 distinct from the immediate surrounding agricultural land in that it applies pig slurry to both  
199 the arable fields and pasture fields as it has indoor and outdoor herds of pigs.

200 Below the soil, groundwater flow occurs in the saturated zone of the Cadeby Formation.  
201 This dolostone aquifer of Permian age is characterized by interquartile interval ranges for  
202 intergranular hydraulic conductivity and porosity which are  $2.9 \times 10^{-4}$  to  $0.9 \times 10^{-3}$  m/day and  
203 8.5 to 18.7%, respectively (Allen et al. 1997). Flow occurs essentially in correspondence of  
204 fractures, i.e. evidenced by the large difference between permeability from pumping and  
205 core plug tests ( $K_{well-test}/K_{core-plug} \sim 10^4$ ; see Fig. 3). Groundwater flow is more vigorous in  
206 correspondence of normal faults and here the bedrock is heavily karstified and fault traces  
207 are characterized by alignment of springs and streams that are located 3-4 km away from

208 the study site (Fig. 1c). Fluid temperature and electrical conductivity are in the range of 9° -  
209 10° and 80 -110 mS/m respectively according to the probes installed in the boreholes of  
210 the UoL Farm (Medici et al. 2019a). Flow rate ranges of springs and streams are 0.1 - 1  
211 m<sup>3</sup>/s and 0.1 - 4 m<sup>3</sup>/s , respectively (Aldrick 1978).

212 Slug tests in the Cadeby Formation conducted at the UoL site show hydraulic  
213 conductivities ranging from 0.07 to 2.89 m/day (Medici et al. 2019a). Notably, higher  
214 values of hydraulic conductivities (K=0.83-2.89 m/day) from these tests characterize the  
215 first ~15 m below the water table (Medici et al. 2019a). Pumping tests indicate hydraulic  
216 conductivities ranging from 0.2 to 10 m/day with median values of 1.3 m/day across un-  
217 faulted areas for the Cadeby Formation of NE Yorkshire. The calibrated hydraulic  
218 conductivity for the Cadeby Formation from a MODFLOW-2005 steady state model for this  
219 area is 1.75 m/day (i.e. a similar value to those from pumping and shallow slug tests, see  
220 Fig. 3). Hence, the permeability of this dolostone aquifer primary comes from  
221 enhancement of fracture hydraulic aperture in first ~15 m below the water table (Medici et  
222 al. 2019a, b). At such depths, hydraulic apertures (0.33-0.43 mm) have been computed  
223 applying the cubic law combining slug and fluid and televiewer logging at the study site  
224 (Medici et al. 2019a). This information on enhancement of hydraulic conductivity due to  
225 karstification has been incorporated in the MODFLOW-2005 flow model of the study site  
226 (Medici et al. 2019a, b). Maximum fluctuations of the water table are typically 3.5 m during  
227 the hydrological year. As a consequence of the relatively small fluctuation of the water  
228 table, the model exclusively simulates steady state flow conditions and a constant rainfall  
229 recharge rate corresponding to the annual average of 0.134 m/year was applied.

### 230 **3. Material and Methods**

#### 231 *3.1 Hydrochemistry*

##### 232 *3.1.1 Water sampling*

233 Water samples were collected from a variety of different sources, in order to characterize  
234 the hydrochemistry of the study site at and around the UoL Farm (field areas shown in  
235 Figs. 1b-d, 2a, b). These sources include; soil (n=102), boreholes (n=123, maximum depth  
236 112m), springs (n=29) and streams (n=22) (Fig. 1a-d). However, it should be noted that no  
237 springs or streams occur within the UoL research farm.

238 Soil water was sampled from both arable (n=50) and pasture (n=52) fields every two  
239 weeks between January and October 2017 using a 5-cm MacroRhizo (Eijkelkamp,  
240 Holland) soil moisture sampler (0.02 m diameter, 0.09 m length) resulting in a total of 102  
241 samples. Six fields were studied; three fields were arable and three were improved  
242 permanent grassland (location of these fields A1-3 and P1-3 are shown in Figure 2b). Soil  
243 water samples were collected at the depth intervals of 0.05-0.10 m and 0.35-0.40 m.

244 The boreholes are divided into two groups that penetrate the Cadeby Formation to  
245 different depths. Group 1 are boreholes installed by the UoL at the Farm site (BH1, BH2  
246 and BH3, see Fig. 1d and 2b) where the water table is between 10 to 15 m below ground  
247 level (BGL). Group 1 boreholes are characterized by a diameter 0.20 m large at the UoL  
248 Farm. This group of boreholes were screened with 2.3 - 3 m long intervals between 8 and  
249 40 mBGL and were sampled 6 times from October 2017 to September 2018. Thus,  
250 groundwater samples were collected at multiple depths in each of the UoL boreholes.  
251 Group 2 are boreholes sampled by the Environment Agency of England (EA, see Fig. 1d)  
252 in the vicinity of the UoL Farm; these are typically deeper and are screened within the  
253 depth interval 15-112 mBGL and are characterized by diameters ranging from 0.15 and  
254 0.60 m. In addition, the EA also collected water samples from springs and streams near  
255 the farm (Fig 1d). The samples from springs, streams and Group 2 boreholes were  
256 collected two times per year by EA staff between 2006 to 2018.

257

### 258 3.1.2 Chemical analysis

259 Hydrochemical analysis of cations ( $\text{Na}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{K}^+$ ,  $\text{Ca}^{2+}$ ,  $\text{Mn}^{2+}$ ,  $\text{Fe}^{2+}$ ,  $\text{Al}^{3+}$ ), anions ( $\text{Cl}^-$ ,  $\text{SO}_4^-$ ,  
260  $\text{NO}_3^-$ ,  $\text{HCO}_3^-$ ) and DOC have been undertaken in streams, springs and groundwater within  
261 the dolostone of the Cadeby Formation to characterize the Critical Zone of the UoL Farm  
262 area (field area shown in Figures 1b, d; 2a, b). Soil water samples were analysed using a  
263 Mettler Toledo S20 pH meter, Horiba LAQUAtwin conductivity meter, and a Skalar San ++  
264 continuous flow analyser for  $\text{NO}_3^-$  concentrations. Dissolved organic and inorganic carbon  
265 (DOC and DIC, respectively) concentrations were determined using an Analytik Jena Multi  
266 N/C 2100C combustion analyser.

267 The temperature, electrical conductivity and pH of all groundwater, spring, and stream  
268 samples was measured using a 6PFCE Ultrameter 2 (Myron L Company). Groundwater  
269 alkalinity was measured by titration in the field. Calibration of probes was carried out at the  
270 beginning of each working day and maintenance of calibration was assessed prior to each  
271 measurement. All groundwater samples were filtered through a  $0.25\ \mu\text{m}$  pore membrane  
272 into a 50 ml bottle in the field using a syringe; a 10% nitric acid was added to those 50 mL  
273 bottles intended for cation analyses. All groundwater, spring and stream water samples  
274 were stored at  $4^\circ\text{C}$  in the fridge prior to laboratory analyses (following Piper 1953). Major  
275 anions ( $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{NO}_3^-$ ) and cations ( $\text{Na}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{K}^+$ ,  $\text{Ca}^{2+}$ ,  $\text{Mn}^{2+}$ ,  $\text{Fe}^{2+}$ ,  $\text{Al}^{3+}$ ) were  
276 determined using an ICS0 ion chromatographer and ICP, respectively. Charge balance  
277 errors were calculated and ranged from 0.1 up to 4.5%, which suggests good quality data.  
278 Dissolved organic carbon (DOC) concentrations in all groundwater samples was  
279 calculated from the difference between total dissolved carbon (DC) and dissolved  
280 inorganic carbon (DIC), which was measured by a Multi N/C Analyser. A comparison was  
281 made between the dissolved inorganic carbon concentration from the laboratory analysis  
282 and that derived from the alkalinity titration in the field; concentrations were very similar  
283 ( $\pm 2.5\%$  discrepancy) indicating good sample preservation.

284

285 3.2 Calculation of nitrate and DOC mass fluxes

286 The Transect Method (*sensu* Goltz et al. 2007) was applied at the UoL Farm to determine  
287 the mass flux of DOC and nitrate that represent the principal groundwater quality  
288 indicators and contaminants in a KCZ (Goldscheider and Drew 2014; Figs. 2, 4). The  
289 transect was applied avoiding faults where the bedrock can be assumed homogeneous  
290 (Fig. 2a). The described approach represents the first application, as far as we are aware,  
291 of the Transect Method to a Farm that we refer to as TMF. Given the concentration of the  
292 chemical species in groundwater,  $C_i$ , the advective groundwater mass flux,  $M_{gw}$ , is  
293 calculated as:

294

295 (1) 
$$M_{gw} = C_{gw} \times Q_{gw}$$

296

297 where  $Q$  is the groundwater flux defined by the Darcy's law as the product of the hydraulic  
298 conductivity ( $K$ ) and the hydraulic gradient ( $i$ ) and transect area ( $A_2$ ). Note that in the case  
299 of the transect area defined in Figure 2, the hydraulic gradient (annual arithmetic average,  
300 0.0236) is known from the monitoring of the three boreholes at the UoL Farm and the  
301 aquifer hydraulic conductivity (arithmetic mean, 1.1 m/day; Fig. 3) from slug tests (Medici  
302 et al. 2019a). Slug test values representative of the horizontal hydraulic conductivity show  
303 arithmetic mean > geometric mean > median > harmonic mean. The arithmetic mean was  
304 selected for use in this study, representing the highest mean, because the most highly  
305 conductive layers dominate horizontal flow at the field site (Medici et al. 2019a).

306

307 Given the contaminant concentration ( $C_{sw}$ ) of the specific chemical specie in soil water  
 308 collected at the base of the soil profile at 0.35-0.45 m depth<sup>1</sup>, the contaminant mass flux  
 309 from the land surface through area  $A_1$  (Fig. 4) is then defined as:

310

$$311 \quad (2) \quad M_{inf} = C_{sw} \times Q_{inf}$$

312

313 where  $Q_{inf}$  is the product of the recharge rate (i.e. precipitation minus evapotranspiration)  
 314 and the area  $A_1$ . The annual average recharge rate used to calculate the hydrological flux  
 315 from the land surface ( $Q_{inf}$ ) is 0.134 m/year derived from the calibrated MODFLOW-2005  
 316 regional groundwater model previously used at the field site (Medici et al. 2019b). The use  
 317 of unique recharge value is supported by the selection of a transect area that is overlain by  
 318 uniform ~0.5 m thick well drained calcareous soil above a relatively homogenous bedrock,  
 319 avoiding fault zones (Fig. 2a).

320 A theoretical 3D block (see Figures 1d, 2 and 4) has been created with the longer side  
 321 oriented parallel to the annual groundwater Fisher mean vector (azimuth 68°; Medici et al.  
 322 2019a). The area used to compute the upstream groundwater flux ( $Q_{gw}$ ) is defined by the  
 323 parallelepiped side,  $L_1$  (Figure 2) and the aquifer saturated thickness of 30 m estimated by  
 324 the British Geological Survey core logs (Cooper and Lawley, 2007, Fig. 4).

325 Our analysis assumes that the total mass flux ( $M_{mix}$ ) of the two species leaving the  
 326 transect area within groundwater is given by the sum of the groundwater flux entering the  
 327 transect area ( $M_{gw}$ ) and that infiltrating through the soil zone ( $M_{inf}$ ) (Fig. 4). Median values  
 328 of DOC and nitrate concentration ( $C_i$ ) were calculated from the dataset available in the  
 329 area of the transect to apply equations (1) and (2). Concentrations in groundwater from the  
 330 Headley Hall Farm and BH2 borehole and the soil water concentrations collected at 0.40  
 331 mBGL (Figs. 2b, 4) were used to define  $C_{gw}$  and  $C_{sw}$  respectively. Note that, DOC and

---

<sup>1</sup> Samples were collected at the base of the soil zone to obtain concentrations of nitrate and DOC below the zone of potential denitrification and differential bacterial activity within the soil.

332 nitrate concentrations from Hadley Hall Farm and BH2 were selected due the upstream  
333 position of these boreholes with respect to the other boreholes of the UoL Farm, BH1 and  
334 BH3 (Fig. 2b). The Headley Farm is the shallowest borehole that the Environment Agency  
335 sample in the study area and is screened at the same depth interval of BH1, BH2 and  
336 BH3. The Headley Farm, BH1, BH2 and BH3 boreholes are all characterized by a 0.20 m  
337 diameter.

338 To test the validity of the TMF, the modelled DOC and nitrate mass fluxes were converted  
339 into concentrations dividing by the surface plus groundwater flux,  $Q_{\text{mix}} (Q_{\text{gw}}+Q_{\text{in}})$ . This  
340 conversion allows comparison with the measured concentration values acquired from the  
341 UoL Farm boreholes.

342

## 343 **4.0 Results**

### 344 *4.1 Aquifer physiochemical properties*

345 The mean, range and standard deviation for electrical conductivity, fluid temperature and  
346 pH for the different source waters are shown in Table 1. The mean pH of all the surface  
347 and groundwater samples was 7.4° and ranged from 6.5° to 8.0° (Fig. 1d). Mean fluid  
348 temperature collected from all the different sources of water was 10.0° and ranged from  
349 9.2° up to 19.9°C (Fig. 1d). Electrical conductivity of the water samples displayed a wide  
350 range; from 140 up to 1558  $\mu\text{S}/\text{cm}$  (n=276).

351 Comparison between the different sources of water indicates similarity in the pH and  
352 conductivity values between the shallow (0-40 mBGL) boreholes of the UoL and the soil  
353 waters from the arable and pasture fields as well as the springs. Overlap of electrical  
354 conductivity, fluid temperature and pH ranges indicates a good degree of hydraulic  
355 connectivity between the soil, shallow saturated aquifer zone and springs (Tab. 1). In  
356 contrast, pH, fluid temperature and electrical conductivity are higher in the groundwater  
357 sampled at depth from the EA boreholes, indicating a lower degree of hydraulic

358 connectivity with all sources of surface water. Fluid temperature and electrical conductivity  
359 from the stream samples were more scattered compared to the other water sources,  
360 probably due to contribution of external surface water sources and atmospheric control on  
361 surface water temperature (Tab. 1).

362 The groundwater of the Cadeby Formation, shows a  $\text{Ca}^{2+}$ - $\text{Mg}^{2+}$  bicarbonate-type  
363 composition (Fig. 5). The order of abundance is  $\text{Ca}^{2+}$ > $\text{Mg}^{2+}$ > $\text{Na}^{+}$ > $\text{Al}^{3+}$ > $\text{Mn}^{2+}$ > $\text{Fe}^{2+}$  and  
364  $\text{HCO}_3^{-}$ > $\text{SO}_4^{2-}$ > $\text{Cl}^{-}$ > $\text{NO}_3^{-}$  for cations and anions, respectively. Binary diagrams, which  
365 provide information on the chemical processes occurring in the Cadeby Formation, are  
366 illustrated in Figure 6a-c. The  $\text{Mg}^{2+}$ - $\text{Ca}^{2+}$  binary diagram shows positive correlation, which  
367 lies above the 1:1 ratio (Fig. 6a). This trend is due to the higher abundance of more highly  
368 soluble calcite with respect to dolomite (60%  $\text{CaCO}_3$ , 40%  $\text{Mg}_2\text{CO}_3$ ) in the Cadeby  
369 Formation of Yorkshire (Lott and Cooper, 2008).  $\text{Na}^{+}$ - $\text{Cl}^{-}$  regression lines (Fig. 6b) are  
370 parallel to the 1:1 ratio which indicates dissolution of halite, which is most likely related to  
371 dissolution of evaporites within the Edlington Formation. The latter evaporitic formation is  
372 juxtaposed against the Cadeby Formation by normal faults that behave as hydraulic  
373 connectors (see Figs. 1c, 2).

374 Similarly,  $\text{SO}_4^{2-}$ - $\text{Ca}^{2+}$  linear regression (Fig. 6c) indicates gypsum dissolution from the  
375 evaporitic strata of the Edlington Formation (Moussa et al. 2014; Re et al. 2017). Note that,  
376 the  $\text{SO}_4^{2-}$ - $\text{Ca}^{2+}$  regression lines do not superimpose exactly on the 1:1 line for gypsum  
377 dissolution due to additional  $\text{Ca}^{2+}$  input from dissolution of calcite in the Magnesian  
378 Limestone Group. Springs show a higher concentration of  $\text{Ca}^{2+}$  with respect to  
379 groundwater and streams (Fig. 6a, c).

380 The World Health Organization (WHO) drinking water quality limit for nitrate (50 mg/L) is  
381 breached in many of the EA boreholes, springs and streams, and in the pasture soil water  
382 samples (Fig. 7). The nitrate 75<sup>th</sup> percentile is also above the WHO limit in all the UoL  
383 boreholes (Fig. 8).

384 Overall, nitrate concentration shows medians which are respectively above and below the  
385 WHO limit in the groundwater sampled from the UoL shallow (15-40 mBGL) boreholes  
386 versus those sampled from the deeper (20-112 mBGL) EA boreholes. Note, no consistent  
387 variation in nitrate concentrations with depth was recognized from the multiple depth  
388 intervals (8-40 mBGL) in the UoL boreholes. Nitrate concentrations however substantially  
389 decrease at greater depths >40 mBGL seen in samples from the EA boreholes (Fig. 7).  
390 Nitrate concentrations in the springs are very similar to that of the shallow UoL boreholes,  
391 whereas, nitrate concentrations in streams are lower than those in both the UoL boreholes  
392 and springs (and the upper percentile lies below the WHO limit, Fig. 7).  
393 Nitrate concentrations in the soil water from the pasture fields are similar to those in the  
394 groundwater from the UoL boreholes (Fig. 7). This similarity suggests that leaching from  
395 these pasture soils which are upstream of the farm boreholes (see Fig 2) are a major  
396 nitrate source for groundwater of the studied Permian dolostone. Indeed, nitrate  
397 concentration is higher in pasture soil where livestock is present compared with arable  
398 land exclusively used to grow crops as shown in Figure 7. Nitrate concentrations are lower  
399 in streams compared with soil water and groundwater as illustrated in the box plot in  
400 Figure 7. This concentration difference may arise from dilution of the groundwater  
401 baseflow component of streamflow by freshwater runoff that is generated in  
402 correspondence of the low permeability units of the Pennine Coal Measure Group in the  
403 western sector of the study site (Fig. 1; Aldrick 1978).  
404 Ranges, interquartile ranges and median concentration (C) of DOC are shown in Figure 9.  
405 Similar and highest DOC concentrations were observed in the soil water from the pasture  
406 ( $C_{\text{median}} = 13.0$  mg/L) and arable ( $C_{\text{median}} = 11.0$  mg/L) fields. Concentrations of DOC in the  
407 saturated zone of the aquifer were considerably smaller, with median values of 5.5 mg/L in  
408 groundwater from the UoL boreholes (15 to 35 mBGL) and 0.5 mg/L from the deeper (>  
409 40 mBGL) EA boreholes. Here, the observed pattern of DOC in soil water, shallow and

410 deeper groundwater is consistent with infiltration from both the farm arable and pasture  
411 soils providing increased DOC to shallow (15 to 40 mBGL) groundwater. DOC  
412 concentrations observed at the shallower depth (8 - 40 mBGL) intervals sampled via the  
413 UoL Farm boreholes do not show evident depth variations. However, DOC varies  
414 seasonally from 1.0 up to 13.8 mg/L (Fig. 6) and shows a wider seasonal variation than the  
415 other solute chemical species, with higher concentrations observed in the spring-summer  
416 period from late April to June.

417

#### 418 *4.2 DOC and Nitrate mass fluxes*

419 Mass fluxes of nitrate and DOC were modelled applying the TMF to a theoretical 3D block  
420 in the area of the UoL Farm (Figs. 2, 4). The inputs and outputs for this method are  
421 illustrated in Table 2 for 'background or inflowing' groundwater ( $Q_{gw}$ ,  $M_{gw}$ ), infiltrating  
422 recharge water ( $Q_{inf}$ ,  $M_{inf}$ ), and exiting groundwater ( $Q_{mix}$ ,  $M_{mix}$ ), in terms of DOC and  
423 nitrate fluxes. The initial step for this computation is the definition of the fluxes of recharge  
424 and groundwater flow (see Figure 4). The recharge flux ( $Q_{inf}$ ) is the product of the annual  
425 average recharge rate from the calibrated MODFLOW-2005 model (see Figure 1c) and  
426 land surface area ( $A_1$ ) applied at the top of the 3D block that represents a portion of the  
427 land surface (see Figure 4). The groundwater flux ( $Q_{gw}$ ) is calculated from the annual  
428 average hydraulic gradient (0.0236) and arithmetic mean hydraulic conductivity (1.1  
429 m/day) from slug tests applied to the saturated thickness of aquifer unit (Medici et al.  
430 2019a). The sum of these two contributions ( $Q_{inf}$ ,  $Q_{gw}$ ) provides the outflowing groundwater  
431 flux ( $Q_{mix}$ ) of 2292 m<sup>3</sup>/day for the selected transect (Fig. 2b; Tab. 2).

432 Mass fluxes of nitrate and DOC were modelled from  $Q_{in}$ ,  $Q_{out}$  by applying equations (1)  
433 and (2) (Tab. 2). Median concentrations of nitrate (Figs. 7, 8) and DOC (Fig. 9) from land  
434 surface and inflowing groundwater flow are assumed to be those from soil water (median

435 of all arable and pasture measurements) and from the EA Headley Hall Farm and BH2  
436 boreholes that are located upstream with respect to the other two UoL boreholes (Fig. 2b).  
437 The infiltrating mass fluxes ( $M_{inf}$ ) are 4526 and 10454 kg yr<sup>-1</sup> for DOC and nitrate,  
438 respectively according to the proposed model. The transect is 200 hectares and therefore  
439 infiltration mass fluxes can be expressed as 23 and 52 kg yr<sup>-1</sup> ha<sup>-1</sup> for DOC and nitrate,  
440 respectively to enable comparison at the UoL as well as other farm areas across the world.  
441 Applied nitrogen inputs from arable and pasture field are 190 and 180 kg yr<sup>-1</sup> ha<sup>-1</sup>,  
442 respectively. 52 kg yr<sup>-1</sup> ha<sup>-1</sup> of NO<sub>3</sub><sup>-</sup> corresponds to 12 kg yr<sup>-1</sup> ha<sup>-1</sup> of N flux. This N value of  
443 flux is derived from natural recharge and reaches the water table. Hence, ~6% of applied  
444 nitrogen infiltrates in the saturated part of the aquifer.

445 The outflowing nitrate flux derived from equations (1) and (2) is 45202 kg yr<sup>-1</sup> and contains  
446 the inflowing mass flux,  $M_{gw}$  (34748 kg yr<sup>-1</sup>; Tab. 2) as well as that infiltrating from the  
447 transect. DOC outflowing mass flux is 5146 kg yr<sup>-1</sup> as shown in Table 2. Notably, model  
448 outputs show that  $M_{inf} < M_{gw}$  for nitrate. This hydrochemical scenario contrasts the modelled  
449 DOC mass fluxes that indicate  $M_{inf} \gg M_{gw}$  (Tab. 2). The UoL Farm appears therefore to  
450 be an evident point source of DOC and a more mild nitrate-polluter.

451 To test the validity of the TMF, the modelled DOC and nitrate mass fluxes are converted  
452 into modelled downstream concentrations by dividing the groundwater flux leaving the site  
453 downstream boundary,  $Q_{mix}$  (2292 m<sup>3</sup>/day). The modelled concentrations for DOC and  
454 nitrate are shown as dashed lines in Figures 7 and 8 (for nitrate) and 9 (for DOC). While  
455 concentrations at the downstream boundary were not measured directly, these modelled  
456 concentrations can be compared to the measured concentration in groundwater at the  
457 BH1 and BH3 receptors (see Fig. 2b).

458 The modelled nitrate (54.0 mg/L) concentration from the transect calculation falls within the  
459 interquartile range (53.0-71.5 mg/L) of nitrate for BH1 (Fig. 8). However, BH3 nitrate  
460 concentrations lies above the model output of 54.0 mg/L, possibly due to the proximity of a

461 silage field that has received pig slurry (Figs. 2b, 8). The model also highlights that if the  
462 median soil water nitrate concentration was reduced by 40% from 40.0 mg/L to 24.0 mg/L  
463 the modelled out-flowing nitrate concentration in groundwater would fall below the 50 mg/L  
464 that represents the limit imposed by the World Health Organization. Thus agricultural  
465 practices in the arable and pasture fields need to be modified to reduce soil water nitrate  
466 concentrations by this amount.

467 The modelled DOC (6.2 mg/L) concentration from transect calculation fall within the  
468 interquartile ranges of DOC (3.0-10.3 and 2.2-11.7 mg/L for BH1 and BH3, respectively) in  
469 the two receptors of the UoL (Fig. 9). Notably, the model outputs closely match the  
470 arithmetic average of BH1 and BH3 boreholes for DOC. These results support validity of  
471 the presented TMF approach (Figs. 7-9; Tab. 2).

472

## 473 **5. Discussion**

474 The influence of soil-derived nitrate and DOC on groundwater quality was studied in a  
475 dolomitic KCZ at UoL Experimental Farm. From a hydrochemical point of view, analyses  
476 reveal a baseline  $\text{Ca}^{2+}$ - $\text{Mg}^{2+}$  bicarbonate-type water that represents the typical composition  
477 of dolostone aquifers (Seyhan et al. 1985; Barbieri et al. 2005; Xanke et al. 2015).  
478 Similarly, the range of nitrate (Fig. 7) and DOC (Fig. 9) concentrations from soil and  
479 groundwater indicate a high degree of hydraulic connectivity between these two elements,  
480 as commonly reported for a KCZ (Jourde et al. 2018); this is discussed further in section  
481 5.1.

482 Hydraulic conductivity values of the Cadeby Formation increase from the core-log to the  
483 scale of the field site. Groundwater flow models find calibration with values that overlap  
484 those of pumping tests (Fig. 3; Schulze-Makuch et al. 1999; Gleeson et al. 2011). This  
485 physical feature is typical for moderately karstified aquifers. As a consequence of the  
486 common physiochemical features of the studied aquifer, this research shows how physical

487 hydrogeological parameters from modelling groundwater flow can be combined with  
488 hydrochemical analyses to develop approaches for modelling contaminant fluxes via KCZs  
489 under farmland.

490

#### 491 *5.1 Aquifer vulnerability*

492 Nitrate and DOC are good indicators of aquifer vulnerability to contamination in areas  
493 dedicated to intense agriculture (Ducci 2010; Wachniew et al. 2016). DOC is an important  
494 parameter from the viewpoint of water quality at farm sites in karst areas (Moral et al.  
495 2008; Goldscheider and Drew 2014; Koit et al. 2020). One reason for its importance is that  
496 DOC forms complexes with hydrophobic organic contaminants released by pesticides  
497 facilitating their transport (Wang et al. 2004, 2018).

498 In this paper, we have used DOC and nitrate concentrations in surface, soil and  
499 groundwater to characterize the flux of  $\text{NO}_3^-$  and DOC through the Permian dolostone of  
500 the Cadeby Formation (see conceptual model illustrated in Figure 10) and in doing so  
501 identified the most vulnerable depth interval of the aquifer. Pasture soil waters have much  
502 higher concentrations of nitrate than arable soil waters (the latter show values below the  
503 WHO drinking water limit). Nitrate concentrations decrease with depth from the pasture  
504 soil water to the shallow (<~40 mBGL) groundwater, and at depths > 40 mBGL nitrate  
505 concentrations are even lower; mostly below the WHO drinking water limit (Fig. 7). DOC  
506 concentrations are similar in arable and pasture soil waters, with lower concentrations in  
507 shallow (0-40 mBGL) groundwater (Figs. 9, 10). Note that, DOC also decreases at depths  
508 > 40 mBGL in the deep aquifer. Other studies have also reported a sharp decrease in both  
509 nitrate and DOC concentrations with depth in the KCZ in agricultural areas (Foster et al.  
510 1982; Geyer et al. 1992).

511 Our research suggests that the zone of maximum aquifer vulnerability is in the first 40 m  
512 below the soil surface. This zone partially includes the most conductive ( $K=0.83-2.89$

513 m/day) part of the aquifer that is in the depth interval 0-25 mBGL as indicated by multi-  
514 level slug tests (Medici et al. 2019a). However, karstification is much more pervasive in  
515 correspondence of normal fault zones intercepted in the area of the relatively deep  
516 abstraction wells (Fig. 1c). These faults are more conductive in borehole tests ( $K_{\text{median}} = 35$   
517 m/day;  $n=7$ ; Allen et al. 1997) and characterized by much faster modelled groundwater  
518 velocities (3000-5500 m/day; Medici et al. 2019b) as typical in faulted dolostone aquifer  
519 portions (Bauer et al. 2016). Intense karstification persists up to 40 m depth in  
520 correspondence of these tectonized zones justifying a relatively deep vulnerable zone (see  
521 Fig. 10).

522

## 523 *5.2 TMF and solute contaminant transport*

524 Mass fluxes of nitrate and DOC arising from agricultural practices at the farm, including pig  
525 slurry spreading, were modelled using the TMF applying the rainfall recharge from the  
526 calibrated regional MODFLOW-2005 flow model (Fig. 3; Medici et al. 2019b). Modelled  
527 mean concentrations of DOC and nitrate in groundwaters down-gradient of the site are  
528 within the range of those measured (Figs. 7-9; Tab. 2). The modelled nitrogen mass flux  
529 ( $12 \text{ kg ha}^{-1} \text{ yr}^{-1}$ ) driven by rainfall recharge indicates that  $\sim 6\%$  of applied nitrogen (from pig  
530 slurry plus mineral fertilisers) reaches the saturated part of the aquifer; a value that is in  
531 the expected range of infiltration fractions (5%-50%) (Liao et al. 2012; Green et al. 2018).  
532 Literature also supports the modelled values of nitrate flux in farmland. Indeed, nitrogen  
533 fluxes applied to crops is  $\sim 10^1 \text{ kg ha}^{-1} \text{ yr}^{-1}$  in areas of the world dedicated to intense  
534 agriculture (Messer and Brenzonic 1983; Jordan et al. 1998; Green et al. 2018). The TMF  
535 therefore provides a reliable tool to compute solute mass fluxes leaving a farm site via the  
536 groundwater pathway. Note that at this field site denitrification below the soil zone is  
537 unlikely to be significant, which means that the assumption of the TMF that nitrate is  
538 conserved within the aquifer is valid. This scenario is related to low groundwater

539 temperature (9° - 10°) that typically inhibits denitrification (Rivett al. 2008). Furthermore,  
540 measured groundwater flow velocities which are very high (50 - 250 m/day, Medici et al.  
541 2019a) and hence residence times in the aquifer are too low to allow significant  
542 denitrification, as proposed by other authors for similar aquifers across the world (Moon et  
543 al. 2006; Goldscheider and Drew, 2014; Hartmann et al. 2014; Yang et al. 2020).

544 Nitrate and DOC concentrations in soil and groundwater are known to vary temporally and  
545 spatially due to variations in hydrology, soil physical and chemical properties, crop rotation  
546 and fertilizer inputs (Sieling et al. 1997, Lord et al. 1999; Williams et al. 2000). In recent  
547 years, there has been concerted efforts to improve the nitrogen use efficiency of both  
548 inorganic fertilisers and manure/slurry applications in order to help reduce nitrate leaching.  
549 In 2000, Chambers et al. (2000) reported that about 50% of pig and poultry manures are  
550 applied in the autumn (August-October) to cereal stubble in the UK. However, they also  
551 found that highest nitrate losses occurred following the application of slurry to winter  
552 cereals in the autumn. This practice is applied at the UoL Farm and here more nitrate is  
553 available for leaching. Pig slurry application is going to change at the studied farm in the  
554 near future. In fact, in recent years, advice to UK farmers has focussed on getting them to  
555 switch from autumn to spring applications for high readily-available-N manures and slurry  
556 (Chambers et al. 2000; Ball Coelho et al. 2006).

557 This research develops the application of the Transect Method, previously used for  
558 prediction of impacts of industrial sources on groundwater quality, to other point sources of  
559 contamination such as farms. Previous applications of the Transect Method focused on  
560 prediction of contaminant mass fluxes and concentration at industrial field sites related to  
561 release of chlorinated pollutants and heavy metals (Verreydt et al. 2012, 2013; Padgett et  
562 al. 2017). Here we show that the Transect Method can be used to define mass fluxes and  
563 hence concentrations of contaminants in groundwater. The approach provides an  
564 indication of whether a particular point source (in this case a farm where pig slurry

565 spreading occurs) has a significant impact on the contamination load in groundwater. In  
566 our case, the results show that the farm has a significant impact on groundwater DOC  
567 concentrations, and to a lesser extent nitrate concentrations (see Tab. 2). This scenario is  
568 related to surrounding farms use of inorganic fertilisers so regional groundwater nitrate  
569 concentrations are already relatively high, whereas pig slurry spreading is less common in  
570 the immediate vicinity of the UoL Farm. In fact, pig slurry is practiced at this farm because  
571 the indoor pig unit at the farm provides a ready supply.

572 The TMF could also be used to define the boundary conditions for solute transport models  
573 (e.g., Goltz et al. 2007) for prediction of downstream impacts for example at well  
574 abstractions at the study site. Thus, the future modelling of reactive contaminant transport  
575 at the UoL Farm must primarily focus on the first and highly vulnerable ~40 meters below  
576 the ground using different concentrations of nitrate and DOC in soil and groundwater (Fig.  
577 10).

578

## 579 **6.0 Conclusions**

580 Nitrate and DOC are considered two key chemical species that determine water quality in  
581 the CZ in karstic environments. Here, we propose an approach to demonstrate how the  
582 combination of hydrochemical analyses of different water sources from multiple depths,  
583 combined with physical groundwater flow model characterisation allows prediction of mass  
584 fluxes of DOC and nitrate from farmland via the groundwater pathway. A robust hydraulic  
585 characterisation reveals the mechanisms of solute transport and facilitates future  
586 modelling scenarios. In this paper, the KCZ of the Permian dolostone of NE Yorkshire (NE  
587 England, UK) has been used and the Transect Method applied to compute nitrate and  
588 DOC mass fluxes at the study site (the University of Leeds Farm, Yorkshire, UK). This  
589 methodology is used to predict the impacts of farm activity on concentrations of pollutant  
590 species in groundwater. The transect approach is here applied for the first time to a farm-

591 source of contaminations where pig slurry is applied in the autumn when leaching losses  
592 are highest to both cereal crops, and pasture fields (post cutting for silage). The use of this  
593 methodology was validated by comparing the modelled concentrations of DOC and nitrate  
594 with measured concentrations in groundwater. The results show that the farm activity  
595 influences the 'background' concentrations of both DOC and nitrate in groundwater. The  
596 influence on DOC is most marked, because this farm uses pig slurry rather than only  
597 inorganic fertilisers, which is more common for the surrounding farms.

598 The hydrochemical analysis of groundwater highlights that the zone of highest vulnerability  
599 to contamination to the first ~40 meters below the ground surface is due to higher  
600 concentrations of nitrate and DOC being observed in this zone. This zone also had high  
601 hydraulic conductivity of karst fissures and conduits in the first ~15 meters below the water  
602 table that most likely vertically extends to greater depths in faulted areas. Hence, future  
603 modelling of KCZ contaminant transport should primarily focus in the first ~40 meters  
604 below the ground.

605 Following this research, we envisage to export the TMF that combines a groundwater  
606 model-derived recharge, baseline soil water and groundwater analyses and computation of  
607 mass fluxes to support conceptualization and modelling of contaminant transport in other  
608 karst areas.

609

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628

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## 885 **Tables**

886 **Table 1** Fluid temperature, electrical conductivity and pH arithmetic mean, range and  
887 standard deviation ( $\sigma$ ) for the UoL and the EA boreholes, springs, streams and arable and  
888 pasture fields

889 **Table 2** Input and output parameters for computation of mass fluxes of DOC and nitrate.

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## 891 **Figures**

892 **Fig. 1** The study site of the UoL Farm in the area of Leeds and York and the dolostone  
893 aquifer-unit of the Cadeby Formation. (a) The Permian Zechstein Basin and deposition of  
894 the shallow water deposits of the Cadeby Formation (from Mawson and Tucker 2009); (b)  
895 Great Britain with location of the study area (basemap from GeoMapApp); (c) Map of the  
896 bedrock lithology with the location of the UoL Farm site, transect for computation of nitrate-  
897 DOC mass fluxes and MODFLOW-2005 model of the study area (Medici et al. 2019b); (d)  
898 Location of sampled streams, springs and boreholes.

899 **Fig. 2** Study site and detail of the transect. (a) Bedrock geology and studied boreholes, (b)  
900 Sampling points used for computation of nitrate-DOC mass fluxes.

901 **Fig. 3** Hydraulic conductivity vs. scale for the Cadeby Formation at the study site.

902 **Fig. 4** Theoretical 3D block for calculation of nitrate and DOC mass fluxes at the study  
903 site.

904 **Fig. 5** Piper diagram for water samples from boreholes, springs and streams.

905 **Fig. 6** Binary diagrams for springs, boreholes and streams; **a** Cl<sup>-</sup>-Na<sup>+</sup>, **b** Mg<sup>2+</sup>-Ca<sup>2+</sup>, **c** SO<sup>4-</sup>  
906 - Ca<sup>2+</sup>

907 **Fig. 7** Model output vs. concentrations of nitrates in samples from soil moisture, springs,  
908 streams and boreholes

909 **Fig. 8** Model output vs. concentrations of nitrate in the EA Headley Farm wells, and BH1,  
910 BH2 and BH3 boreholes of the UoL Farm.

911 **Fig. 9** Model output vs. concentrations of DOC in samples from soil moisture, springs and  
912 boreholes

913 **Fig. 10** Hydrogeological conceptual scheme of the KCZ of at the UoL Farm site

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Data source	pH	Fluid Temperature (°C)	Electrical Conductivity ( $\mu\text{S}/\text{cm}$ )
	Mean; Range; $\sigma$	Mean; Range; $\sigma$	Mean; Range; $\sigma$
University of Leeds boreholes	7.3; 7.0-7.6; 0.1	10.2; 9.2-11.4; 0.11	880; 775-1086; 95
EA Borehole	7.9; 7.4- 8.0; 0.2	18.1; 16.2-19.9; 3.4	890; 660-1170; 167
Spring	7.2; 7.0-7.7; 0.2	12.1; 9.8-19.3; 2.3	972; 640-1558; 281
Stream	7.4; 7.4-7.9; 0.2	13.9; 10.5-17.9; 1.5	817; 443-1154; 817
Arable Soil	7.6; 6.9-7.9; 0.3	N/A	670; 280-1070; 208
Pasture Soil	7.1; 6.5-8.0; 0.9	N/A	540; 140-790; 93

921 Table 1

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Water Fluxes (Input)		$Q_{\text{inf}}$ ( $\text{m}^3/\text{day}$ )	$Q_{\text{gw}}$ ( $\text{m}^3/\text{day}$ )	$Q_{\text{mix}}$ ( $\text{m}^3/\text{day}$ )
				734
Mass Fluxes (Output)	Chemical Specie	$M_{\text{in}}$	$M_{\text{gw}}$	$M_{\text{mix}}$
	Nitrate ( $\text{kg yr}^{-1}$ )	16900	34748	51648
	DOC ( $\text{kg yr}^{-1}$ )	4526	620	5146

923 Table 2

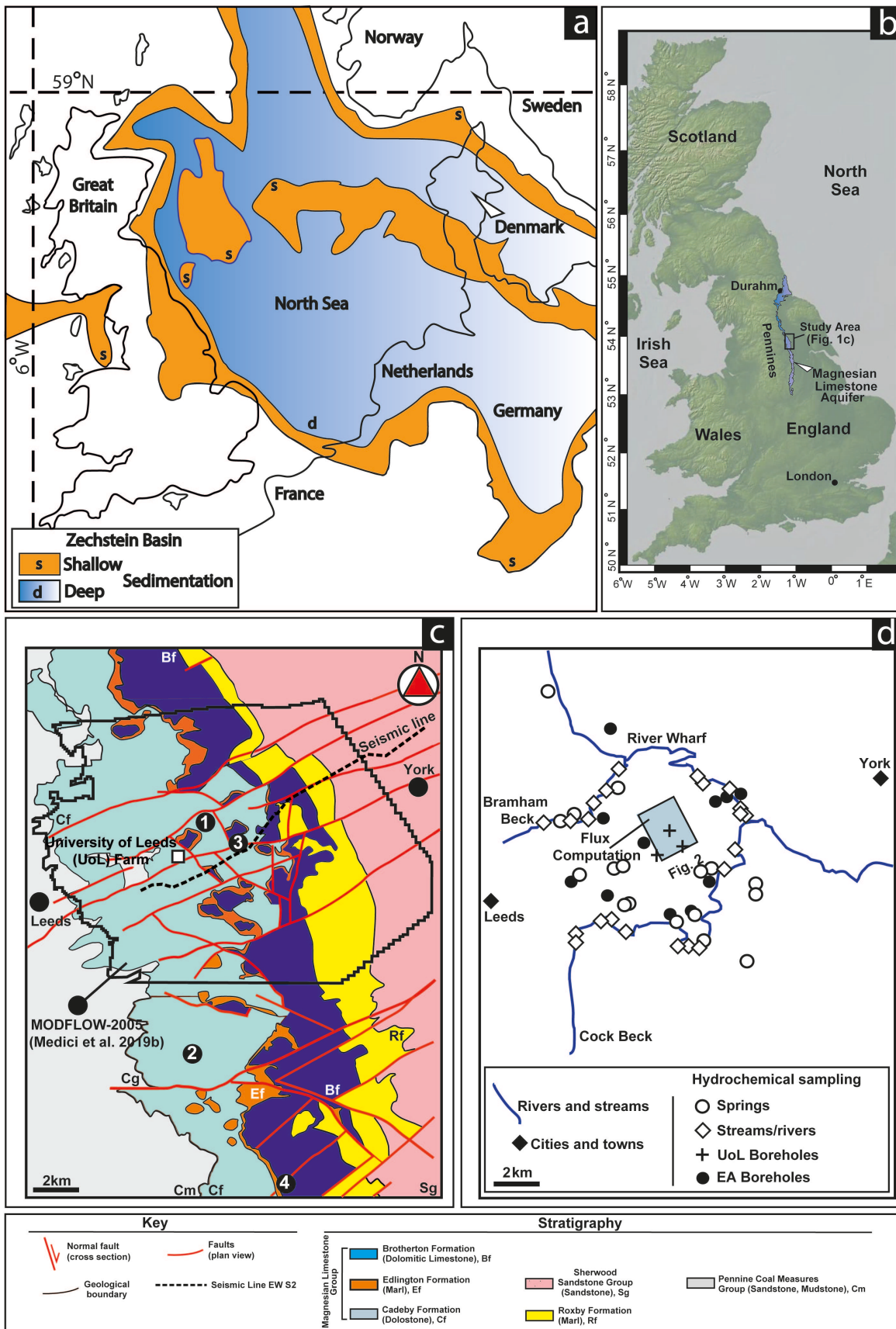
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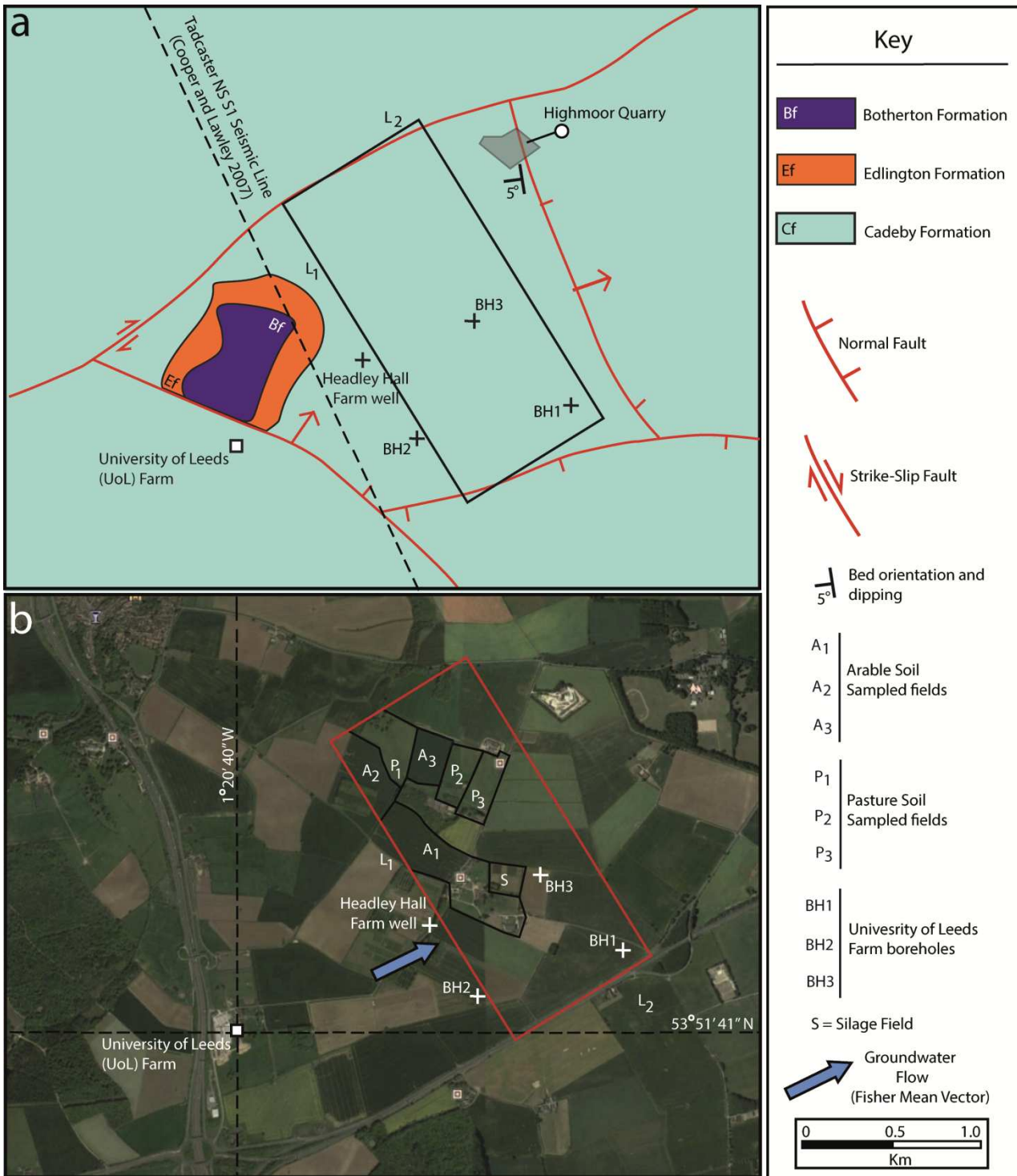
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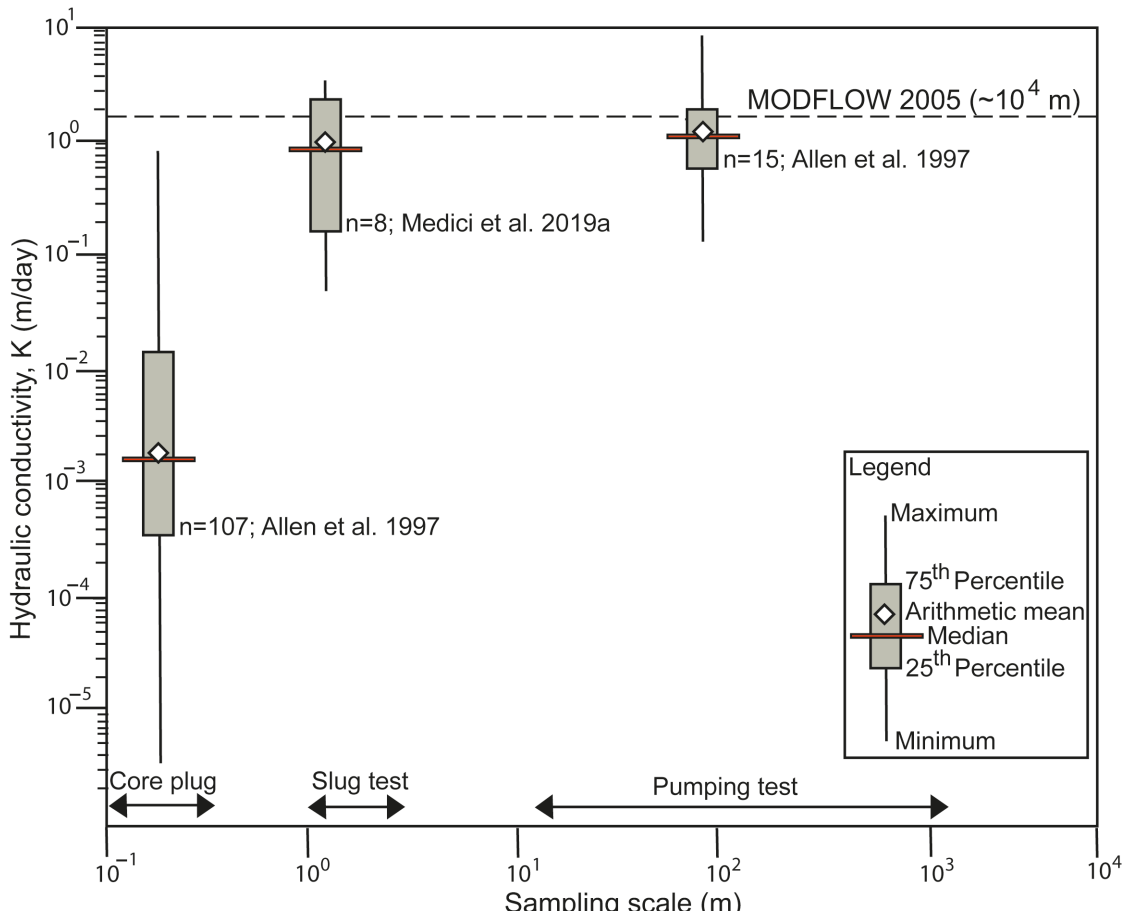
930 Fig. 1



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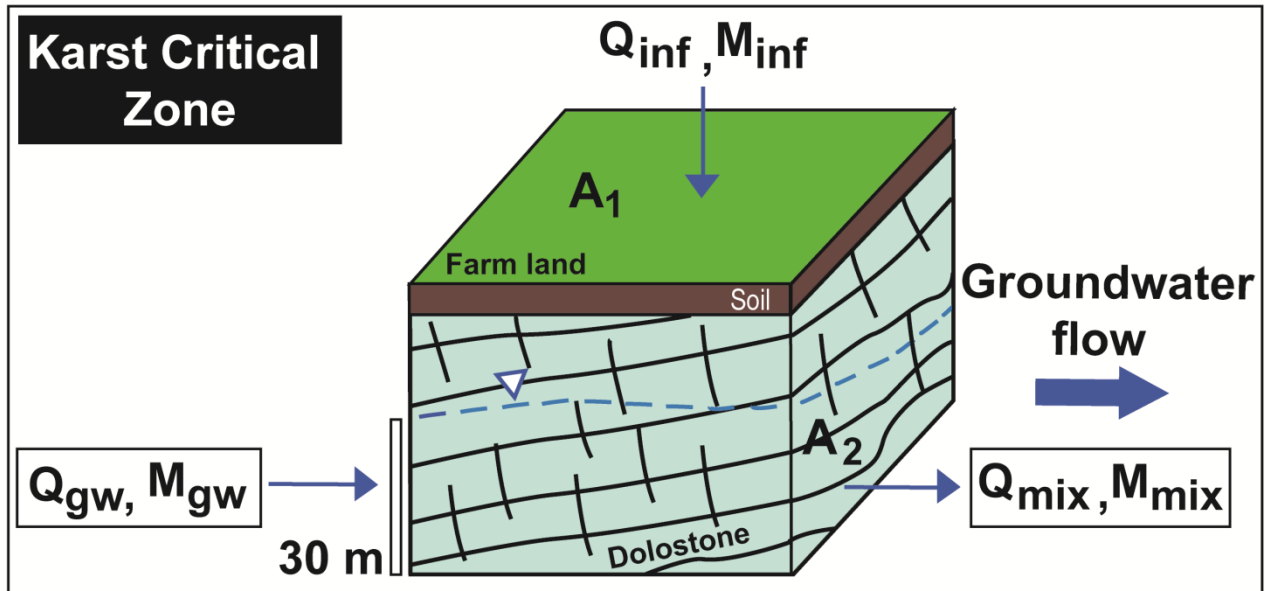
932 Fig. 2

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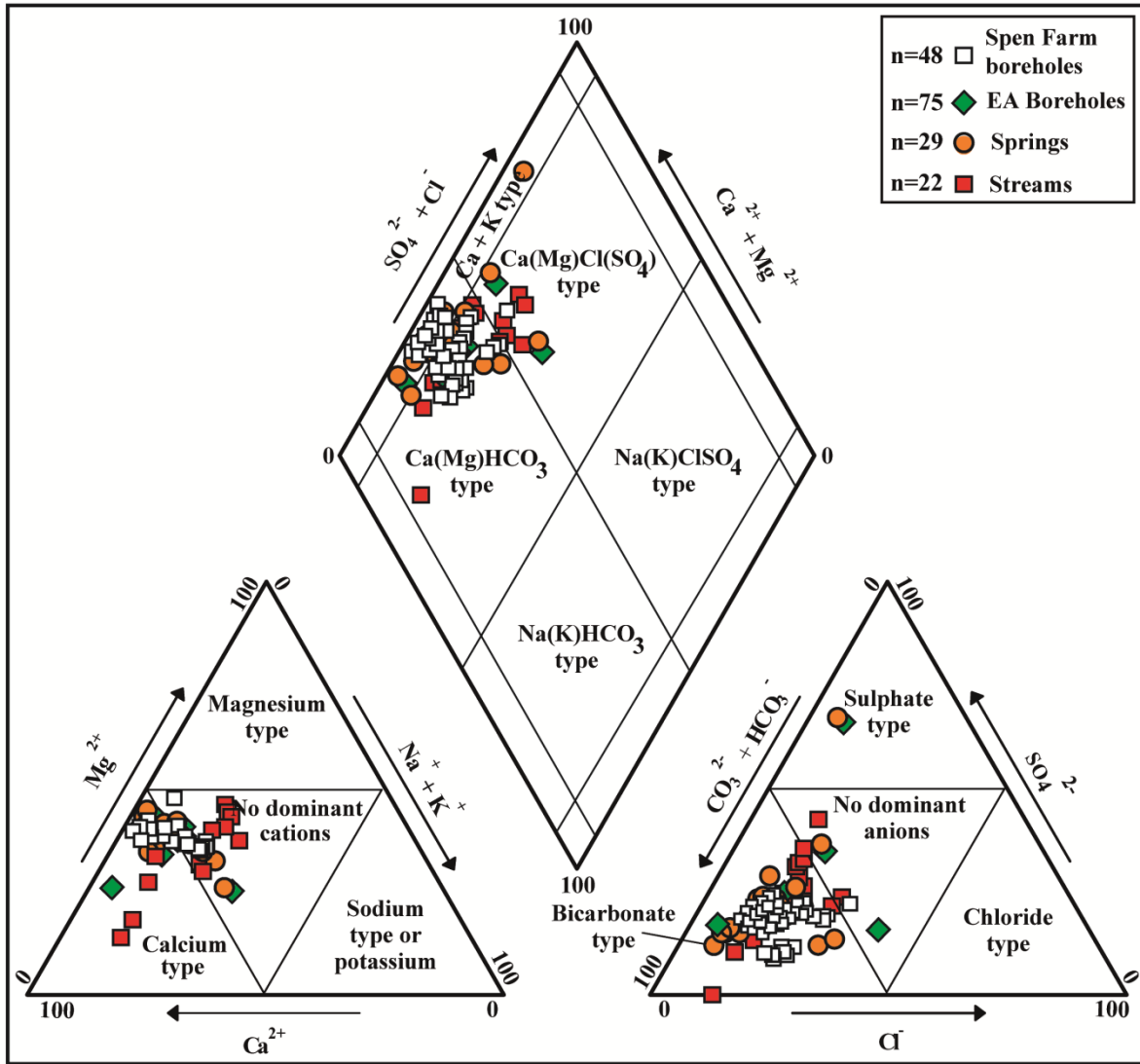
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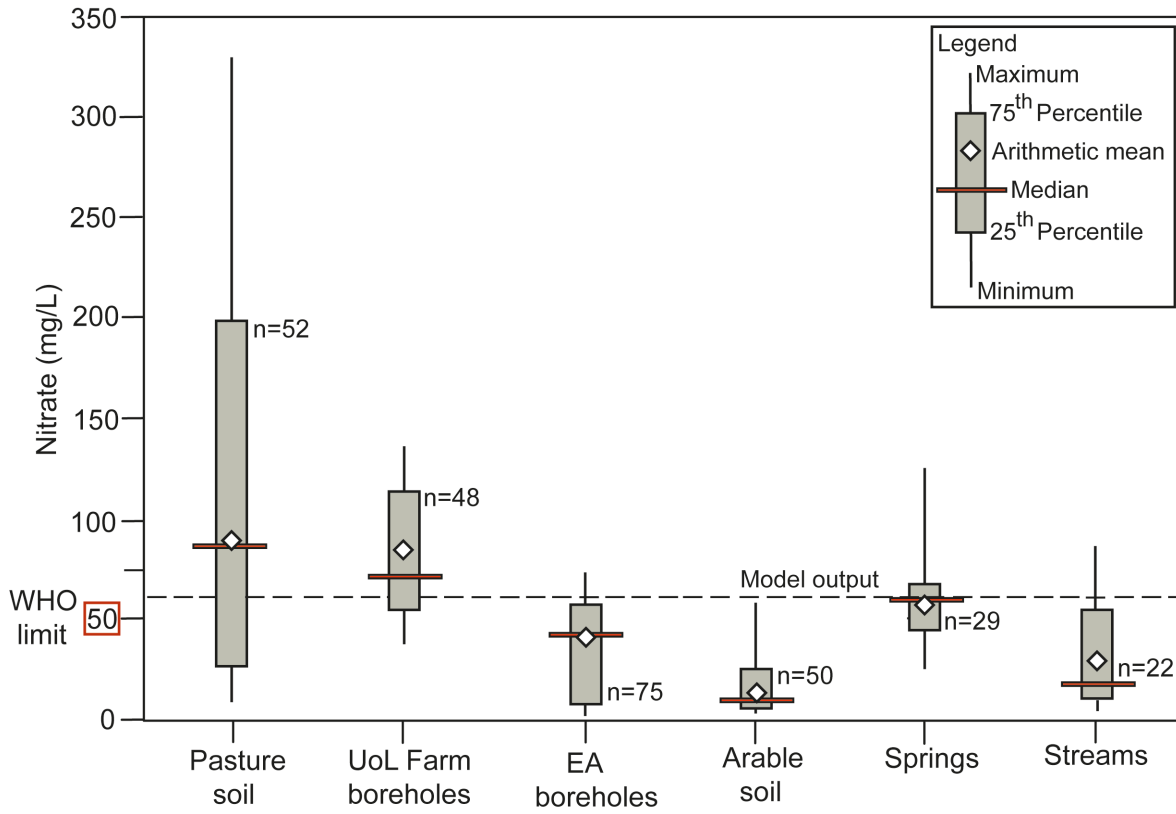
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948 Fig. 6

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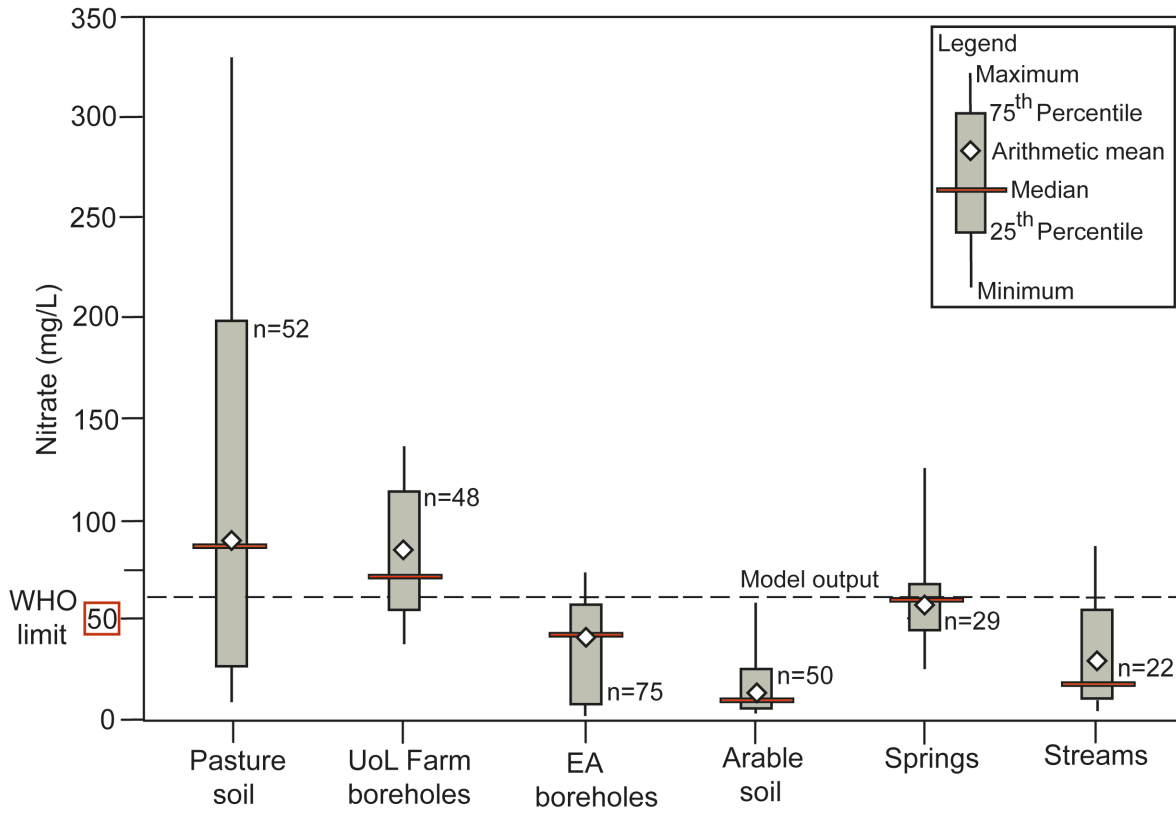
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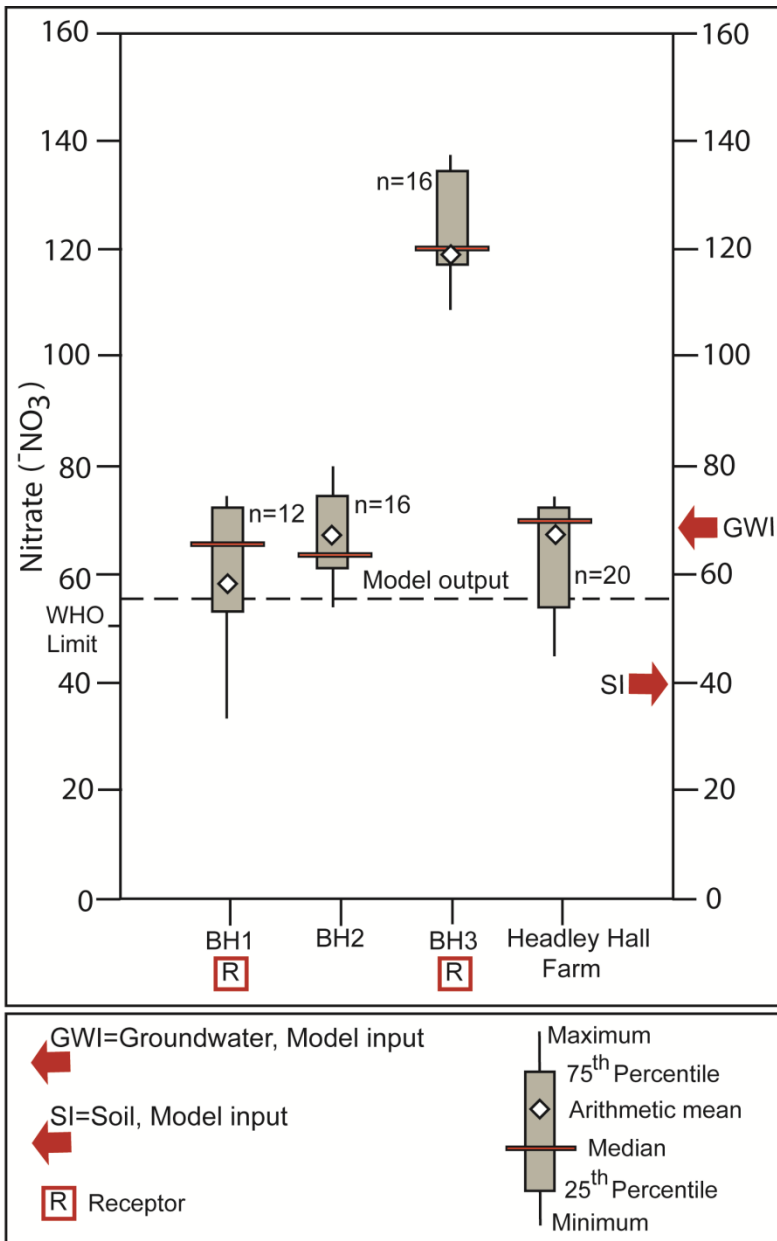
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960 Fig. 7



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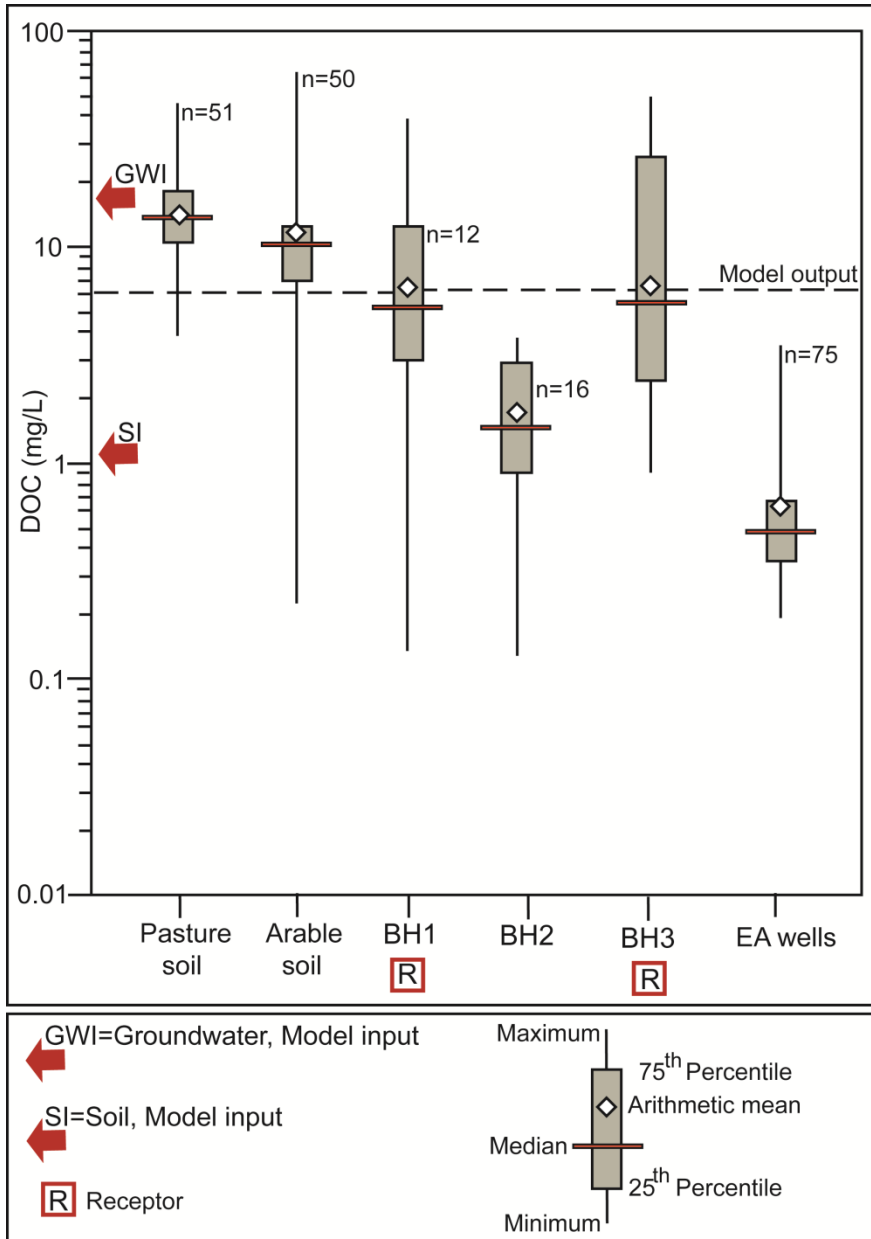
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969 Fig. 9

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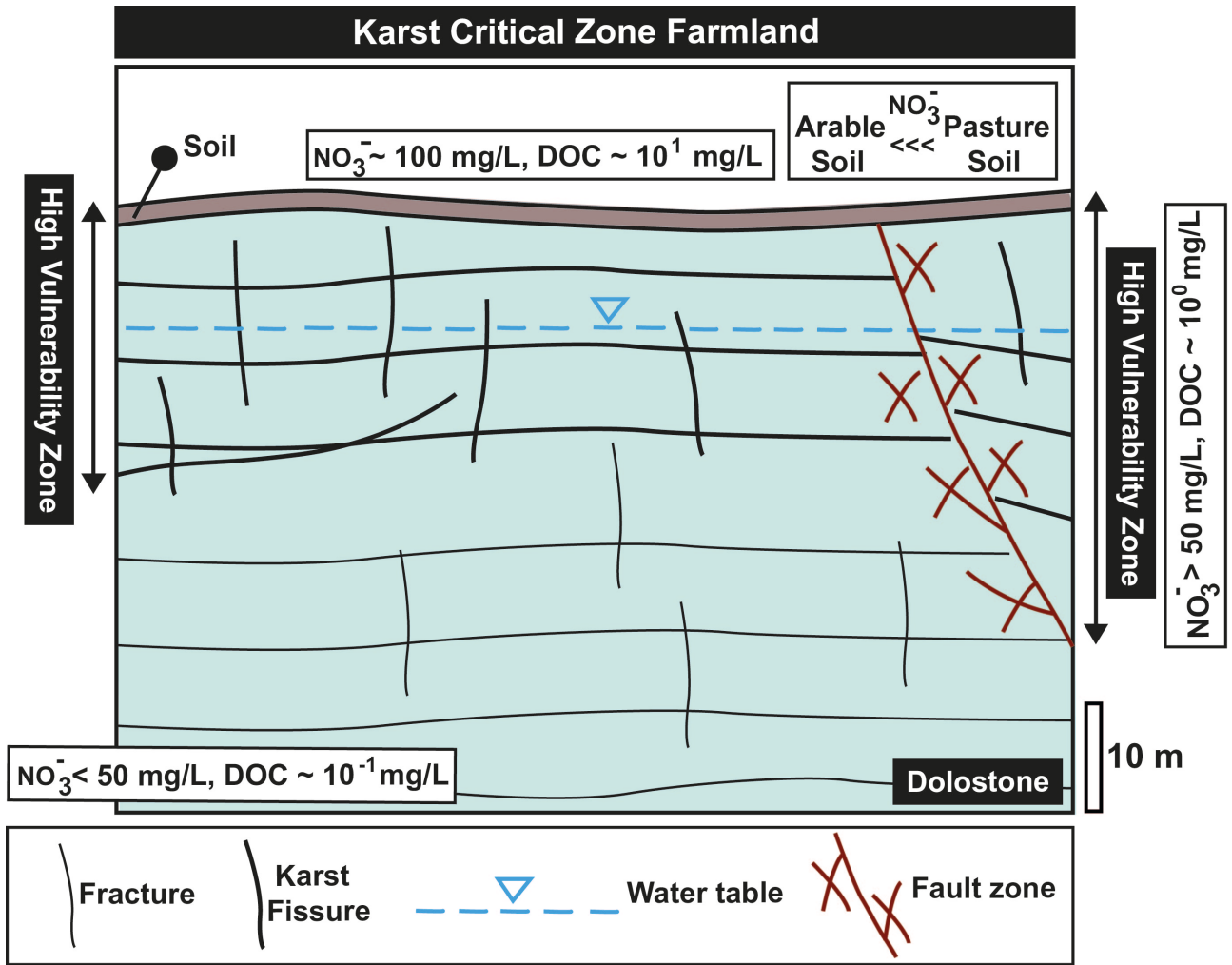
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977 Fig. 10

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