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Additively manufactured ultra-low sintering temperature, low loss $\text{Ag}_2\text{Mo}_2\text{O}_7$ ceramic substrates

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1. ABSTRACT

Ultra-low sintering temperature silver molybdenum oxide ($\text{Ag}_2\text{Mo}_2\text{O}_7$) ceramics have been printed using direct ink writing (a material extrusion additive manufacturing process) for the first time. An optimum densification conditions of $460^\circ\text{C} / 2 \text{ h}$ was determined, resulting in relative permittivity, $\epsilon_r = 13.45$, dielectric loss, $\tan\delta = 0.0005$, microwave quality factor, $Q \times f = 17,056 \text{ GHz}$ and the temperature coefficient of resonant frequency $\tau_f = -121 \text{ ppm}/^\circ\text{C}$. The results were comparable to the dielectric properties of conventionally fabricated ceramics. A series of metal/ceramic antenna designs were produced via dual-printing and co-firing, to demonstrate the potential of $\text{Ag}_2\text{Mo}_2\text{O}_7$, to be used as a co-firable dielectric material for functional integrated circuits and/or microwave RF devices through multi-material direct ink writing.

Keywords: additive manufacturing; 3D printing; LTCC; silver molybdenum oxide; microwave ceramics

1. INTRODUCTION

Microwave dielectric ceramics are a large part of the modern wireless telecommunications industry; being the materials of choice for many applications, such as substrates, antennas, filters, capacitors and couplers where a specific combination of properties are required to achieve the best performance [1,2]. Critical material properties are relative permittivity (ϵ_r) whose optimum value is specific to the application, low dielectric loss ($\tan\delta$), high-quality factor ($Q \times f$) and near zero value of the temperature coefficient of resonant frequency (τ_f) [3–5]. Sebastian et al. collated more than 4000 different electroceramic compounds with suitable properties for microwave applications [6]. However, most microwave ceramics have sintering temperatures above 1000 °C [7–9], limiting their potential to be co-fired with common electrode materials, such as aluminium, copper or silver. Low-temperature co-fired ceramics (LTCCs) can be densified below 1000°C, offering the potential for making passive integrated circuitry [10–13] with embedded electrodes and electronic components to achieve miniaturised, lighter and overall higher-performance RF devices [14].

Molybdenum oxide based compounds are important for the fabrication of ultra-low sintering temperature ceramics and devices [15–17]. Zhou et al. first investigated the $\text{Ag}_2\text{O}_3 - \text{MoO}_3$ binary system that contained the phases $\text{Ag}_2\text{Mo}_4\text{O}_{13}$, $\text{Ag}_2\text{Mo}_2\text{O}_7$ and Ag_2MoO_4 . They studied the wide band dielectric spectra and microwave dielectric properties of Ag_2MoO_4 , together with its chemical compatibility with common electrode materials. When sintered at 450 °C for 4 hours, Ag_2MoO_4 ceramics gave $\epsilon_r = 8.08$, $Q \times f = 17,000$ GHz and $\tau_f = -133$ ppm/°C [18]. Zhang et al. reported the synthesis, microwave dielectric and physical properties of both $\text{Ag}_2\text{Mo}_2\text{O}_7$ and $\text{Ag}_6\text{Mo}_{10}\text{O}_{33}$ ultra-low sintering temperature ceramic phases. $\text{Ag}_2\text{Mo}_2\text{O}_7$ ceramic was fired at 460 °C for 4 h resulting in $\epsilon_r = 13.3$, $Q \times f = 25300$ GHz and $\tau_f = -142$ ppm/°C. It was also shown that $\text{Ag}_2\text{Mo}_2\text{O}_7$ is chemically compatible with Ag and Al and therefore may be co-fired. [19]. The $\text{Ag}_2\text{Mo}_2\text{O}_7$ has attracted attention in the last few years due to its multifaceted potential to be used as a visible light active photocatalyst [20–22], an anode material for sodium-ion energy storage devices [23] and as an additive for improving the tribological properties of metal in high-temperature applications [24,25]. In this study however, we recognize that $\text{Ag}_2\text{Mo}_2\text{O}_7$ is an excellent candidate for metal/ceramic integrated microwave devices due to its low-temperature sintering and chemical compatibility with Ag, its high ϵ_r (~ 13 , the highest in the $\text{Ag}_2\text{O}_3 - \text{MoO}_3$ system) and its high $Q \times f$ ($\sim 20,000$ GHz).

Additive manufacturing (AM), commonly known as 3D printing, holds the potential to revolutionise electronics manufacturing by being able to combine printing of dissimilar materials in a single process and also realising complicated electronic devices that would normally require significant effort, manufacturing time and cost to integrate [26–28]. However, additively manufactured components do not always achieve bulk material properties due to process-specific limitations [29]. As a result, post-processing is often required to achieve desirable performance and repeatability.

This study investigates the microwave dielectric properties of additively manufactured $\text{Ag}_2\text{Mo}_2\text{O}_7$ samples through direct ink writing; a material extrusion additive manufacturing process. The effect of sintering temperature on the physical, microstructural and dielectric properties is discussed.

2. MATERIAL & METHODS

2.1. Materials

The $\text{Ag}_2\text{Mo}_2\text{O}_7$ (AMO) powder was synthesised using the solid-state reaction method. High purity raw chemicals, Ag_2O (99.5%, Acros Organics) and MoO_3 (>99%, Acros Organics) were weighed stoichiometrically ($\text{Ag}_2\text{O}:\text{MoO}_3 = 1:1$) and then ball-milled in isopropanol for a duration of 4 hours. Afterwards, the dried mixed powders were calcined 4 h at 430 °C. The calcined powder was then milled using a rotary mill and sieved through a 100 μm aperture laboratory sieve.

Silver paste (PMC3, Mitsubishi Materials Trading Corp., Japan) was used as the electrode material to 3D print the patch antenna. The paste comprised a 90 wt.% solid loading of spherical Ag particles of < 1 μm in diameter, mixed in a water-soluble binder system.

2.2. Ceramic paste synthesis

A binder mix consisting of ethylene glycol diacetate (3.3 wt.%, Sigma Aldrich, UK) as dispersant, methylcellulose (1.6 wt.%, Sigma Aldrich, UK) and propylene carbonate (0.9 wt.%, Sigma Aldrich) as viscosity modifier and binder, diisononyl phthalate (1.3 wt.%, Sigma Aldrich, UK) as plasticiser, and ammonium lauryl sulphate (< 0.5 wt.%, Sigma Aldrich, UK) as surfactant was prepared. The AMO powder was slowly incorporated into the binder followed by mixing and then topped up with de-ionised water. The paste was homogenised thrice using a planetary mixer (Thinky ARM 310, Thinky Inc., Laguna Hills, California USA) at 1500 rpm for 2 mins. The total solids content in the paste was 78.96 wt.%.

2.3. Additive manufacturing and post-processing.

All additively manufactured test samples were made using a multi-process additive manufacturing kit (High-Resolution Engine, Hyrel3D, Norcross, GA, USA), equipped with a syringe dispensing module (SDS-5, Hyrel3D, Norcross, GA, USA), using 5 ml luer-lock syringes (Becton Dickinson, Franklin Lakes, New Jersey, USA) and 500 μm metallic needles of 18.25 mm in length (Adhesive Dispensing, Milton Keynes, United Kingdom). A printing speed of 5 mm/s, layer thickness of 0.2 mm, 0.45 mm hatch spacing; providing a 10 % overlap of the extruded filaments, and a constant positive displacement value of 85 pulses per microlitre, were used to print cylindrical test samples of 15 mm in diameter and 4 mm thickness. The test samples were first modelled prior to printing using CAD and G-code and Hyrel3D's inbuilt slicing software.

3D printed samples were left to dry at room temperature for a minimum of 24 hours. Samples were thermally debinded in a furnace (RHF 1600, Carbolite Gero Ltd, Hope, UK) in static air using a ramp rate of $1^{\circ}\text{C}\cdot\text{min}^{-1}$ up to 400°C whilst holding at 100°C , 200°C , 300°C and 400°C for 2 hours each. The debinded samples were then sintered in static air in the temperature range of $460 - 490^{\circ}\text{C}$ for 2 hours with heating and cooling rates of $3^{\circ}\text{C}\cdot\text{min}^{-1}$.

2.4. Characterisation methods

The phase assemblage and crystal structure of the as-synthesised powder and sintered AMO ceramics were investigated using X-Ray diffraction (D2 Phaser, Brucker AXS, Karlsruhe, Germany) using $\text{CuK}\alpha$ radiation at $\lambda = 1.54054 \text{ \AA}$, operating at 30 kV and 10 mA. A 1 mm divergence and a 3 mm anti-scatter slits were used. Diffraction patterns were collected in the $10 - 60^{\circ} 2\theta$ interval, using a 0.02° step size and 15/min sample rotation. Data were analysed using Brucker's proprietary software (DIFFRAC.EVA 4.2.2, Brucker AXS, Karlsruhe, Germany). To investigate further the possibility of any phase transformation taking place within the $460 - 490^{\circ}\text{C}$, corroborative Differential Thermal Analysis (DTA) (SDT Q600) was performed on the powder in N_2 using $100 \text{ ml}\cdot\text{min}^{-1}$ purging rate at a heating rate of $10^{\circ}\text{C}\cdot\text{min}^{-1}$.

Rheological properties of the AMO paste were measured using a high shear rheometer (Physica MCR 101, Anton Paar) with a parallel-plate geometry of 25 mm in diameter and 0.5 mm gap. To determine the storage (G') and loss modulus (G''), an amplitude sweep was performed using a continuous strain ramp from 0.01 to 100s^{-1} .

The AMO paste was subjected to differential scanning calorimetry (DSC) (2920 Modulated DSC, TA Instruments Inc., USA), to verify the material's melting point and any thermophysical material interactions during heating. Thermogravimetric analysis (TGA) (Q5000-IR, TA Instruments Inc., USA) was used to identify the minimum temperature required for the complete burn-off of the paste's vehicle/organic matter. Both techniques were carried out under an Ar gas using a $200\text{ml}\cdot\text{min}^{-1}$ purging rate at a heating rate of $10^{\circ}\text{C}\cdot\text{min}^{-1}$.

The density of the sintered samples was measured using the Archimedes principle. Results are reported as an average of three different 3D printed samples, together with standard error.

The effect of sintering temperature on the ceramic microstructure was assessed using a scanning electron microscope (SEM) (TM3030, Hitachi High Energy Technologies, etc.). 3D printed test samples were embedded in epoxy resin (EpoThin2, Buehler, Illinois, USA) and prepared using a standard metallographic regime; ground using silicon carbide paper of P1200 grit size, followed by a polishing stage using a polishing cloth and a 0.05 μm alumina suspension. To prevent charging during SEM, samples were sputter coated with a gold/palladium alloy in an 80:20 wt.% ratio, for 60 s at 25 mA (Quorum Q150T, Quorum, Edwards, Hastings, UK).

Relative permittivity (ϵ_r) and dielectric loss ($\tan\delta$) of additively manufactured sintered samples were determined by placing the cylindrical 3D printed test samples of 15 mm diameter and 4 mm thickness, into a 24 mm $\text{TE}_{01\delta}$ cavity resonator (QWED, Warsaw, Poland), measuring the transmission and reflection from the target sample, connected to a Vector Network Analyser (VNA) (MS465B22, Anritsu, Japan) using coaxial cables [30]. The properties were calculated using QWED's proprietary software. Results are reported as an average of 3 different printed samples, together with standard error.

A Peltier device was used to heat the cavity resonator for measuring the temperature coefficient of resonant frequency (τ_f) from 25 °C to 85 °C. The temperature coefficient of resonant frequency was calculated with the following formula (1) where the f_{85} and f_{25} were the $\text{TE}_{01\delta}$ resonant frequencies at 85 °C and 25 °C, respectively.

$$\tau_f = \frac{f_{85} - f_{25}}{f_{25}(85 - 25)} \times 10^6 \text{ (ppm/}^\circ\text{C)} \quad (1)$$

3. Results & Discussion

3.1. Raw material characterisation

Figure 4a, shows the room-temperature diffraction patterns for the AMO powder. The diffraction peaks were indexed to the triclinic (anorthic) AMO phase (ICDD PDF 75-1505) of the $P\bar{1}$ space group, with lattice parameters of $a = 6.095 \text{ \AA}$, $b = 7.501 \text{ \AA}$, $c = 7.681 \text{ \AA}$, $\alpha = 110.4^\circ$, $\beta = 93.3^\circ$ and $\gamma = 113.5^\circ$. The theoretical density using the lattice parameters was calculated at $\rho = 5.8 \text{ g}\cdot\text{cm}^{-3}$. No traces of impurities or secondary phases were detected. **Figure 1b** depicts the complementary thermal analysis of the AMO powder via DTA, from room temperature (RT) to 600°C . No thermophysical reactions associated with a phase transition were observed in DTA during heating, aside from an endothermic peak at 510°C due to melting.

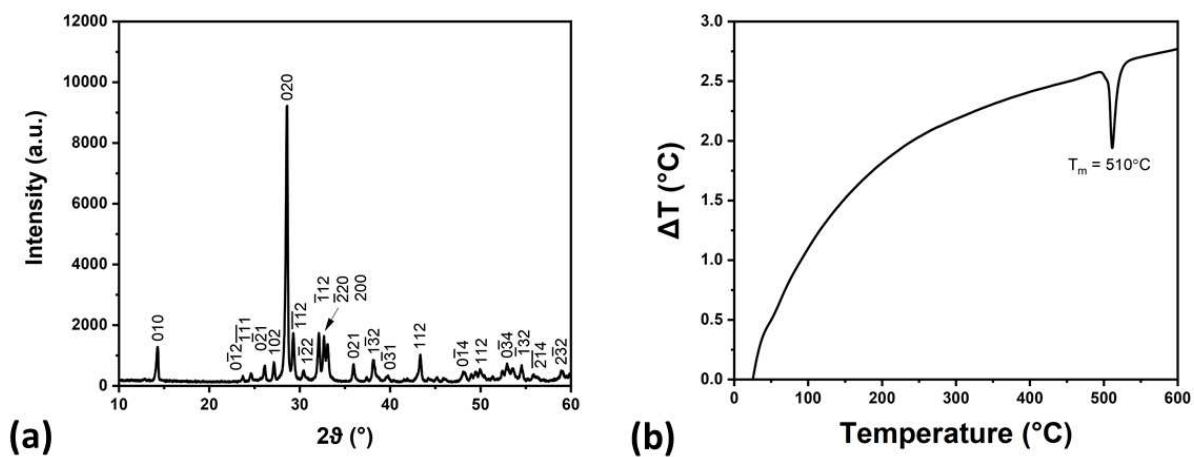


Figure 1 – (a) XRD pattern of the as-synthesised AMO powder, (b) DTA curve of the AMO powder.

3.2. Rheological behaviour of pre-ceramic paste

Figure 2a shows the storage modulus (G'), loss modulus (G''), complex viscosity (η) of the ceramic paste used for the additive manufacturing of the cylindrical pellets that were then used for sintering and further characterisation. At low strains ($<1\%$), storage and loss moduli possess a linear viscoelastic (LVE) relationship with storage modulus being greater than the loss, suggesting the structural stability of the printing medium; i.e. suitable for material extrusion [31,32]. The intersection point between the moduli curves is the yield stress (γ_0) required to deform the ceramic paste and facilitate flow. Practically, this also translates to the minimum amount of pressure to be applied to the syringe plunger when a pneumatic-based extrusion system is used. The yield stress point of $\gamma_0 = 42.4 \text{ kPa}$ at 3.86% strain was calculated using

an intersection script (Origin 2020, OriginLab Corporation, Northampton, MA, USA) between the storage and loss modulus curves. The effect of shear strain on viscosity is also shown where the pre-ceramic paste exhibited a characteristic reduction of its viscosity with increasing rate of deformation. The latter is commonly known as shear thinning behaviour; a fundamental requirement for printing mediums to be used with material extrusion AM processing [33].

Figure 2b shows the thixotropy study of the AMO ceramic paste. Thixotropy is a rheological state in which viscosity decreases with shear but some time is required for the structure to reform as shear is slowly reduced. A paste test sample was subjected to a low (0.1 s^{-1}) – high (4 s^{-1}) – low (0.1 s^{-1}) shear deformation rate sequence where its ability to recover its initial internal structure and viscosity is assessed and the time required to do so is calculated accordingly. The AMO ceramic paste exhibited a thixotropic rebuild time (also known as hysteresis) of 30 s to regain its viscosity from $1.5 \text{ Pa}\cdot\text{s}$ to $10.0 \text{ kPa}\cdot\text{s}$. This is another critical parameter in material extrusion, suggesting the minimum time afforded to print a monolayer. When the printing time per layer is less than the thixotropic rebuild time, the printed layer will not have regained its strength and stability to provide support for the next layers. This will also define the maximum printing speed for a stable print. Based on the above, a printing speed of 5 mm/s , taking 52 s printing time per layer, was chosen as optimum printing speed to additively manufacture all the AMO test samples.

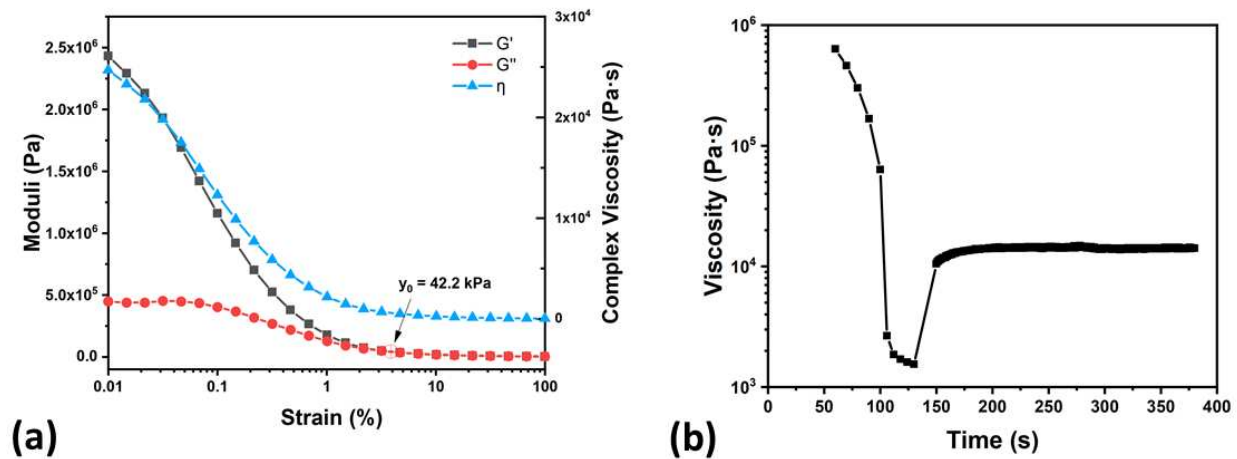


Figure 2 – Rheology data of the AMO ceramic paste formulation.

3.3. Thermal analysis of the AMO pre-ceramic paste

Figure 3 shows the DSC-TGA curves of the AMO ceramic paste from RT – 600 °C. In the DSC thermogram, the range from RT to 400°C contains several endo- and exothermic peaks ascribed to the thermophysical reactions of melting, boiling and evaporation of the ceramic paste’s organic components. A major endothermic peak at $T_m = 510$ °C is associated with melting (possibly incongruent), in agreement with DTA traces, **Figure 1b**. This is followed by an exothermic peak at the temperature of $T_c = 518$ °C, suggesting a $\text{Ag}_2\text{Mo}_2\text{O}_7 \rightarrow \text{Ag}_2\text{Mo}_4\text{O}_{13}$ phase transformation as indicated by the $\text{Ag}_2\text{O} - \text{MoO}_3$ phase diagram [34]. In the same figure, the inset TGA curve, revealed a trend of continuous weight loss of organic components from RT to 380 °C above which the weight remains constant at 78.96 %, representative of the wt % of AMO ceramic powder within the paste. 400 °C was therefore, chosen for the binder-removal stage prior to sintering of the additively manufactured test samples. The presence of any residual organics within the ceramic body are expected to outgas during sintering and contribute towards increased porosity. Additionally, organic materials are known to promote reduction of oxides, giving rise to oxygen vacancies and low Q_{xf} (high $\tan\delta$) [6].

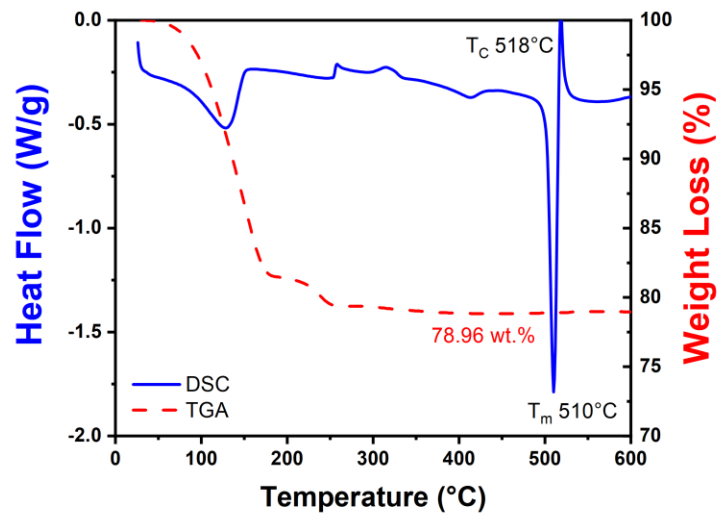


Figure 3 – DSC and TGA curves overlay of AMO paste sample.

3.4. Physical and microwave dielectric properties of AMO ceramics.

The RT XRD patterns of the AMO ceramic samples sintered in the 460 – 490°C temperature range are shown in **Figure 4a**. The XRD patterns do not reveal any new or secondary phases as a result of increasing sintering temperature. However, the intensity of the 020 diffraction peak at 28.3° increases at > 470 °C, indicating preferential grain growth normal to the 020 plane at the pellet's surface.

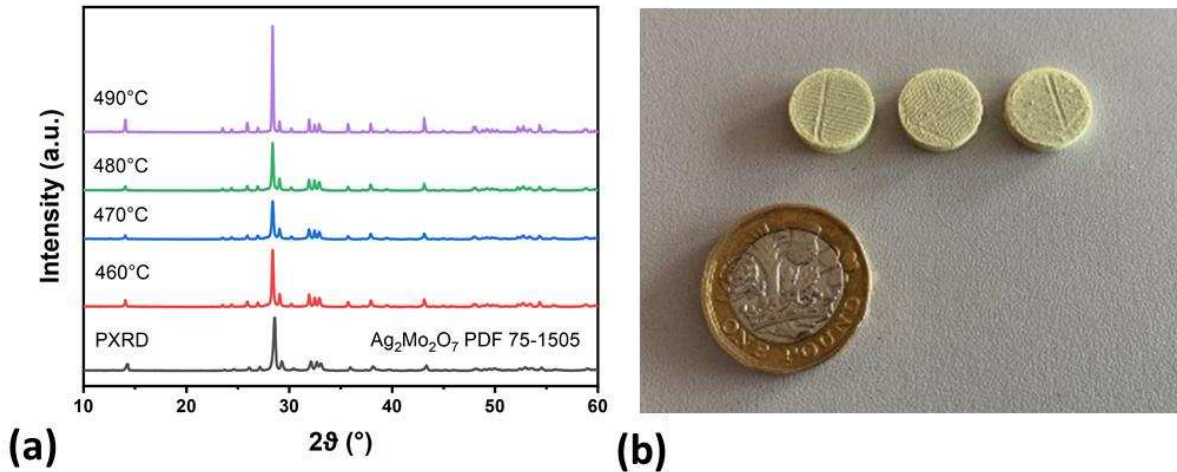


Figure 4 – (a) X-Ray diffraction patterns of the AMO and (b) photo of the sintered 3D printed samples

Figure 5 shows the SEM images of the surface and cross-section microstructure of the AMO ceramic samples, sintered for 2 hours between 460 – 490 °C which reveal a gradual increase in grain size as a function of sintering temperature. At 460 °C, there is notable intergranular porosity as seen in the Figures (5a) and (5e). At 470 °C and 480 °C, intergranular porosity at the surface reduces as the grains enlarge but the micrographs from the cross-sectioned samples reveal that the pore size is increasing. At 490 °C, larger pores are evident both at the surface and in the cross-section micrographs (Figure 5d and 5h). The increase in pore size as a function of temperature is ascribed to coalescence as the AMO grains grow which is exaggerated by their elongated morphology for which there is inefficient packing. The elongated surface grains are consistent with preferential growth normal to the 020 plane, observed by X-ray diffraction, Figure 4a.

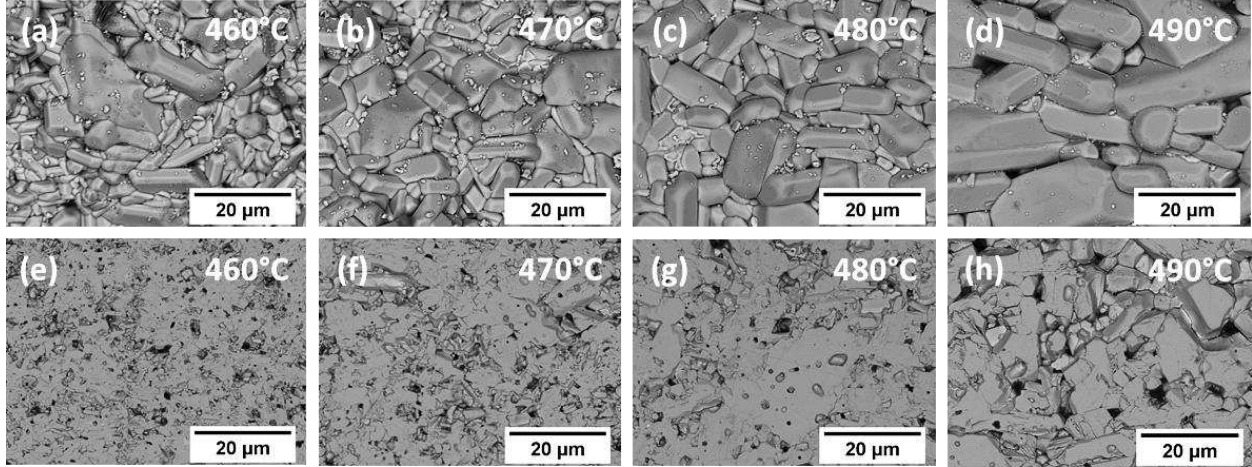


Figure 5 – SEM images of the (a – d) surface and (e – f) polished cross-sections of the sintered 3D printed ceramic samples.

Figure 6 shows the microwave dielectric properties and density measurements acquired from the additively manufactured sintered test samples at 460 – 490 °C. ϵ_r increases as a function of sintering temperature until 470 °C ($\epsilon_r = 13.45 \pm 0.1$) but then exhibits a progressive reduction, as shown in Figure 6a. ϵ_r values are in agreement with density measurements (Figure 6b) with the highest ρ ($5.40 \pm 0.02 \text{ g}\cdot\text{cm}^{-3}$, 93% relative density) also reported at 470 °C, suggesting that sintering at $> 470^\circ\text{C}$ has a negative effect on densification and permittivity. (Figure 5). Despite the low relative density (93%), $\epsilon_r = 13.45 \pm 0.1$ of the additively manufactured samples of this study is similar to that previously reported in the literature [19].

$Q \times f$ values of AMO ceramic samples increase from 5,963 GHz to 22,280 GHz, with increasing sintering temperature from 460 to 490 °C, as shown in Figure 6c. Qf is affected by a number of factors, such as anharmonic lattice vibrations porosity, grain size and grain boundaries, defects such as microcracks and impurities phases within the ceramic body [35-39]. In this study, the primary reason for the increase in $Q \times f$ is the decrease in porosity and an increase in grain size.

Finally, τ_f values decrease from -109 ppm/°C to -133 ppm/°C [14,36–38] as sintering temperature increases. Dependence of τ_f on sintering temperature is well known in the processing of microwave ceramics and presents an obstacle to the control of temperature stability in commercial resonators. τ_f is affected by numerous factors such as grain growth, density, preferential orientation and phase assemblage. Here, the most striking difference as a function of temperature is an increase in density and exaggerated in plane grain growth at the pellet's surface. τ_f is an anisotropic property with different orientations of the crystal having different values. Here, we postulate that the preferred 020 orientation

is the most likely factor influencing τ_f as a function of sintering temperature. We note however, that the changes in τ_f are comparatively modest to those of Qxf .

In summary, additively manufactured AMO ceramic samples sintered at 470°C for 2 hours exhibited optimal microwave dielectric properties of $\epsilon_r = 13.45$, $Qxf = 17,056$ GHz and $\tau_f = -121$ ppm/°C. **Table 1**, lists all measured properties of AMO ceramic samples.

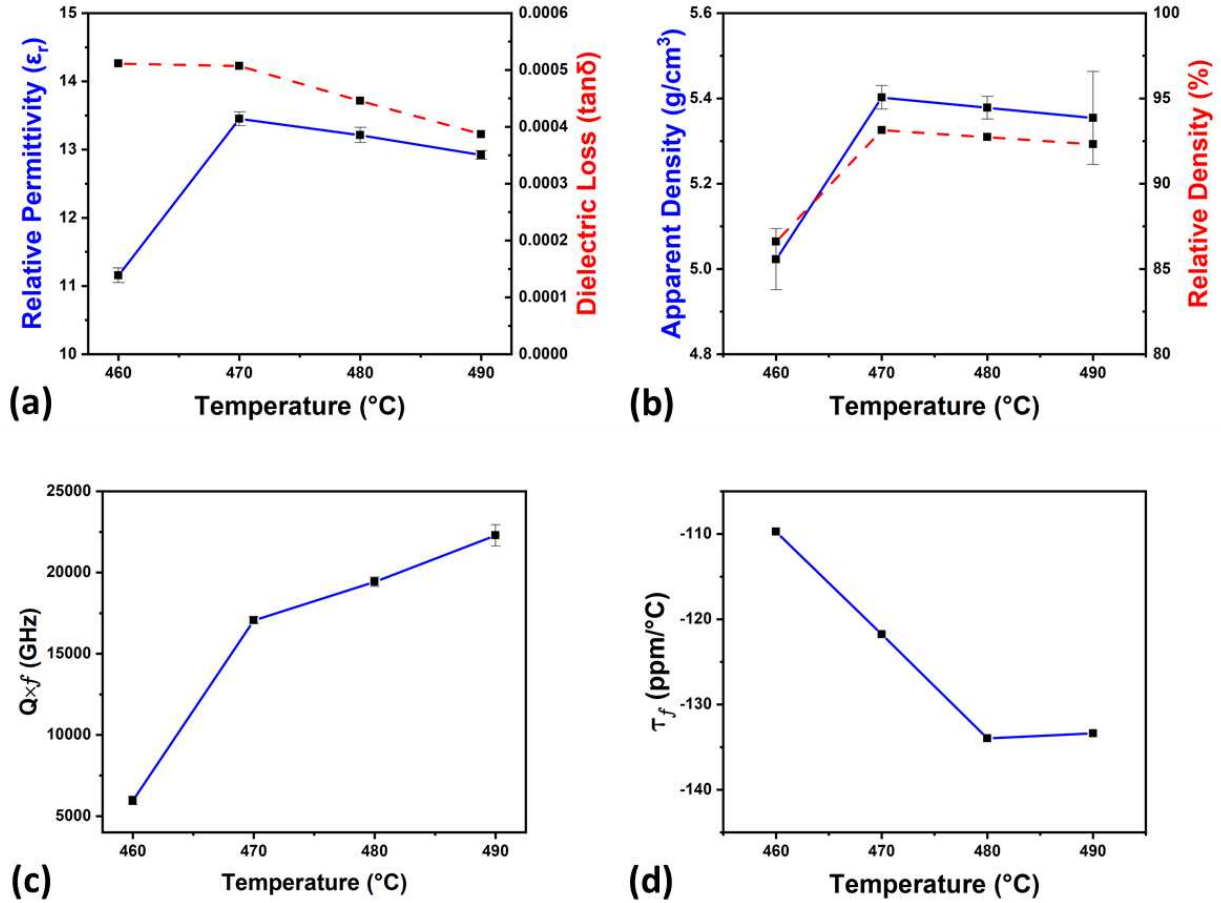


Figure 6 – The effect of sintering temperature on the microwave dielectric properties of additive manufactured AMO ceramic samples: (a) relative permittivity (ϵ_r) and dielectric loss ($\tan \delta$), (b) apparent and relative density, (c) quality factor (Qxf) and (d) temperature coefficient of resonant frequency (τ_f).

Table 1 – The sintering temperatures, microwave dielectric properties of relative permittivity (ϵ_r), dielectric loss ($\tan\delta$), quality factor ($Q \times f$), resonance frequency coefficient (τ_f) and densities of the additively manufactured AMO ceramic samples.

Temperature (°C)	ϵ_r	$\tan\delta$	Apparent Density ($\text{g} \cdot \text{cm}^{-3}$)	Relative Density (%)	$Q \times f$ (GHz)	τ_f (ppm/°C)
460	11.16 ± 0.11	0.0005 ± 0.00002	5.02 ± 0.07	86.59	5963 ± 265	-109
470	13.45 ± 0.10	0.0005 ± 0.000005	5.40 ± 0.03	93.14	17057 ± 207	-121
480	13.21 ± 0.11	0.0004 ± 0.00001	5.38 ± 0.03	92.72	19419 ± 289	-133
490	12.92 ± 0.06	0.0004 ± 0.00001	5.35 ± 0.11	92.31	22280 ± 649	-133

3.5. Additive manufacture of a metal/ceramic antennas

Figure 7, depicts the first ever metal/ceramic antennas made using AMO and Ag, that were designed and additively manufactured using the direct ink writing process. Given the materials’ known chemical compatibility, a silver electrode paste was chosen to print the radiating element. Both an exposed (**Figure 7a – c**) and a packaged version (**Figure 7d – f**) of microstrip antennas were manufactured using sequential printing of the conductive and dielectric materials, in a single process. Both metal/ceramic antennas were successfully co-fired at 470 °C for 2 hours, with no macroscopic failures such as cracking after sintering.

In the first metal/ceramic antenna where the electrode patch was nested on top of the ceramic substrate, the fired structure appeared to have slightly deformed. This is ascribed to either the mismatch between the shrinkage rates and/or coefficient of thermal expansion (CTE) of the two materials. We propose that this may be addressed in future studies either by tailoring the CTE of the electrode and ceramic formulations or by controlling the heat treatment profiles of the cofired body. However, no macroscopic failures or structural deformations were evident in the case of the packaged antenna, where the electrode patch was nested inbetween the AMO substrate and superstrate. The substrate also contained a groove of 0.2 mm in depth, to host the metallic patch of 0.2 mm thickness. It is presumed that the superstrate, the additional ceramic layer above the conductive element of the antenna, helped to reduce stresses involved during the firing. No further characterisations were performed, in regards to the radiation profile and efficiency of the AMO/Ag co-fired antennas, as this was not the scope of this study.

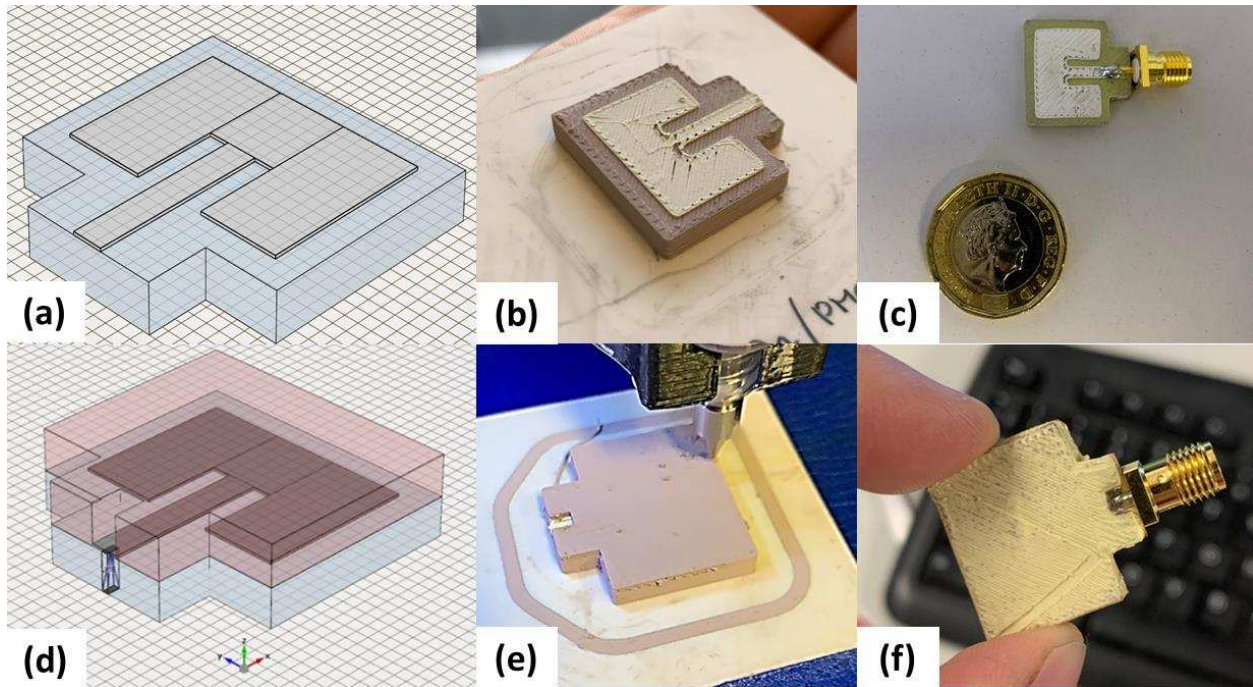


Figure 7 – Additive manufactured (a-c) exposed and (d-f) packaged metal/ceramic antennas.

Figure 8, shows the SEM image of cross-section for the AMO/Ag antenna. The inset elemental maps were acquired via EDS, revealing a stable co-fired structure with no chemical reaction nor Ag diffusion between the top and bottom AMO layers. However, several micropores are evident in between the layers of the superstrate that are close to the electrode layer. Any apparent porosity is expected to hinder the resultant ϵ_r of the packaging constituents of the antenna assembly, thereby affecting its performance and overall efficiency which will be optimized in future work.

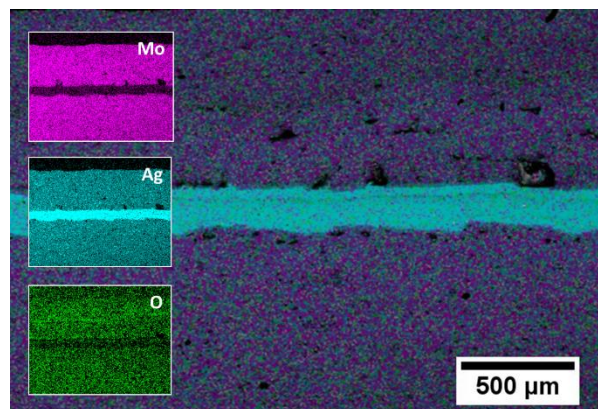


Figure 8 – Combined element EDS with inset individual element maps of the packaged metal/ceramic antenna.

4. Conclusions

Material extrusion, an additive manufacturing process has been used to fabricate three-dimensional test samples, using $\text{Ag}_2\text{Mo}_2\text{O}_7$, an ultra-low sintering temperature microwave ceramic. $\text{Ag}_2\text{Mo}_2\text{O}_7$ ceramic green bodies were printed using a 0.2 mm layer thickness, 5 mm/s printing speed, 0.45 hatch spacing, using a 0.5 mm nozzle. The green bodies were fired at 470 °C for 2 hours which resulted in relative density $\approx 93\%$, $\epsilon_r = 13.45$, $\tan\delta = 0.0005$, $Q \times f = 17,065$ and $\tau_f = -121$ ppm/°C. A series of metal/ceramic patch antenna designs were printed and co-fired using the $\text{Ag}_2\text{Mo}_2\text{O}_7$ ceramic and Ag electrode. No chemical incompatibility of interdiffusion was evident allowing for the first metal/ceramic antennas to be fabricated via single process multi-material extrusion.

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