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Shape and size transformations of biomass particles during combustion

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Abstract

Combustion of individual particles of different woody and agricultural residue biomass have been studied under a laboratory scale rapid-heating apparatus. Particles used were in the size range of 300- 1400 μm and weight 0.5-7 mg. A wire mesh element is used to radiatively heat the particle to 1200-1400K. The apparatus allows a high-speed camera to record the combustion of the individual particles directly. Examination of the resulting video images showed a sequential combustion of volatile matter followed by burn-out of the remaining char for all fuels. Analysis identified differences and patterns in burnout time, combustion behavior and the evolution of char size and shape transformations.

Heterogeneous behavior was observed between the different biomass samples and also among particles within some of the samples. Particles with initial prolate (fibrous) shapes have been observed to become more equant (quasi-spherical) during combustion. Measurement of the particle dimensions during its combustion extracted from the images of the high speed recording have allowed evaluation of the size and shape changes to be mapped. Particle size and shape appears to change only slightly during devolatilization and swelling was rarely observed. Following devolatilization, during the remaining char combustion, more pronounced

31 changes in the size and shape of the particle are apparent. In most cases the shrinking char
32 becomes more rounded as the char particle partially melts and contracts due to surface tension.
33 This transformation is more distinct in some of the biomass samples. Profiles and images of
34 the different fuels examined are presented.

35

36 Keywords: combustion, biomass, char shape

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40

41

42 HIGHLIGHTS

- 43 • Combustion of single particle of five different biomass fuels
- 44 • Thermal imaging camera was used during combustion
- 45 • Biomass char shape and aspect ratio was calculated during combustion.
- 46 • During char combustion, particle shape changes from elongated (prolate) to spheroid
47 (equant) are observed

48 **1. Introduction**

49 Biomass power plants can provide large scale reliable energy with the flexibility to meet
50 potentially unpredictable electricity demand. Biomass power is classed as a renewable source
51 and is a key option for achieving a low-carbon electricity supply while meeting environmental
52 targets on emissions and renewable energy. Many different biomass sources have been studied
53 as fuels with a wide range of physical and chemical properties. The standardization of biomass
54 fuel in the form of high energy-density pellets allows easier management and more sustainable
55 transport to all scales consumers [1]. This also facilitates reliable performance of combustion
56 plant with less variable ash content and calorific value of the fuel. Large power plants such as
57 Drax and Lynemouth in the UK have opted for high quality biomass pellets to minimize
58 problems in the combustion process. However this high standard on biomass fuel quality makes
59 it also more expensive. A more diverse range of source materials will benefit with lower prices.
60 However, lower quality fuels present technological challenges for biomass power plants in
61 maintaining combustion efficiency, avoiding ash fouling issues and complying with strict
62 pollutant emissions limits. Overcoming these challenges will enable better use of limited
63 biomass resources such as new biomass fuels, from energy crops and residues.

64 Single particle combustion (SPC) devices are relatively simple experimental apparatus that can
65 provide reliable information to study the main variables that affect combustion kinetics and
66 compare different fuels combustion behaviour [2, 3, 4, 5, 6]. Biomass has high volatile matter
67 content, less carbon, more oxygen and a lower heating value than coal. Moreover, due to higher
68 reactivity of biomass, a bigger particle size range can generally be used for biomass fuels
69 compared with pulverized coal. Previous studies on single particle device, provided an
70 approximation to milling requirements for biomass fuels in terms of maximum particle size for
71 efficient burn-out [7, 8]. Other single particle studies have identified the differences between
72 different fuels behaviour during combustion [7, 3, 8, 9], providing details on ignition [10, 11,

73 12, 13], volatile flame [14], char combustion [15], particle morphology [16], combustion
74 atmosphere [17,18] or temperature [8, 18]. Flower et al. [5] conducted biomass single particles
75 studies in a wire mesh single particle setup for particles between 5 and 30 mg and showed
76 relatively low dependency on the aspect ratio of the samples. Mason et al. [3] performed a
77 series of single particle experiments which identified useful empirical functions to describe the
78 relationship between particle size and burnout time, allowing the behaviour of different fuels
79 in a furnace to be compared.

80 Biomass shows significant differences in ignition and combustion temperatures and burnout
81 times [2]. The observations made on small scale lab combustion produce relevant information
82 on the combustion behaviour for comparison among different fuels. This information together
83 with other parameters can be used for computer fluid dynamics (CFD) modelling of the
84 combustion process at any scale. CFD has been increasingly improving the resolution of the
85 combustion process modelling in recent years, providing better prediction results for new
86 developments. This has enabled lower pollutant emissions such as NO_x reduction [19] to be
87 achieved through optimised process design.

88

89 The shape, size and density of the particle is important for fluid dynamic modeling as it can
90 affect the trajectory and residence time of the fuel particle inside the boiler [20]. Large particle
91 size implies ignition delay [21] and longer burnout time [3,7], and therefore higher amount of
92 carbon in ash in a furnace. The monitoring and control of fuel particle size distribution is
93 important to achieve a complete and efficient combustion. Panahi et al [21] reported on recent
94 studies the spheroidation of elongated biomass particles with an initial aspect ratio between 4
95 and 8 becoming almost spherical during pyrolysis. This spheroidation is explained because
96 measured temperatures during volatile mater combustion are higher than the melting point of
97 main biomass components as cellulose (732- 739 K). They also concluded higher aspect ratio

98 leads to faster ignition. Bonefacic et al [20] analyzed the influence of the geometry of particles
99 in the combustion process. Their model revealed that particle geometry has a significant impact
100 on the starting point and the rate of devolatilisation as well as the char burning. Higher aspect
101 ratio of the particles leads to increased scattering of fuel, accelerating the devolatilization
102 process and enhancing the mixing of volatiles with the air. Yang et al. [22] developed a
103 mathematical model that predicts the behavior of a range of particles of different sizes,
104 demonstrating the occurrence of the general features of combustion and the occurrence of a
105 combustion wave. It explained the differences observed between spherical and cylindrical
106 biomass particles. These observations are consistent with experimental observations made by
107 same authors using suspended biomass particles and other works in the literature. Gubba et al
108 [23] validated a CFD model including different particles shape distribution and found
109 significant influence of the internal thermal gradients on the surface temperature, residence
110 time and combustion performance. Momeni et al [24] also studied shapes and size effect on
111 combustion process of single biomass particles, cylindrical particles were found to lose mass
112 faster than spherical particles and the burnout time was shortened by increasing the particle
113 aspect ratio (surface area). Similarly, Lu et al [25] obtained differences during devolatilization
114 depending on the shape of the initial particle. The conversion times of cylindrical particles with
115 almost the same surface area/volume ratio are very close to each other.

116 It is clear from the literature, that assuming that fuel particles are approximately spherical is a
117 reasonable approach for modeling some solid fuels like coal, obtaining good agreement with
118 experimental results [26, 27]. However this assumption is generally not enough for
119 lignocellulosic biomass which tends to have elongated particle shapes. The model developed
120 by Yang et al. [22] predicts a change on particle shape. However more detailed experimental
121 data in the literature on the evolution of particle size, shape, or temperature changes during
122 combustion of different biomass fuels is needed.

123

124 Combustion experiments on a single particle combustion apparatus have been undertaken using
125 various biomass materials with the objective of evaluating the differences among particles of
126 different biomass fuels on their combustion; including volatile burning time, char combustion
127 time, particle temperature and adding new information regarding particle size and shape
128 changes. The dimensions of the particle during combustion have been measured providing
129 precise novel data regarding shape and size changes during combustion of different biomass
130 materials. The study examined a range of biomass and different particle sizes compare times
131 required for burnout for the different fuels. The information provided by the video observation
132 can also provide fundamental data for other researchers developing new models to more
133 accurately describe the changes in particle size and shape during combustion process at a
134 particle level.

135

136 2. Materials and methodology

137 2.1. Fuel samples used

138 A series of experiments have been made with five different biomass samples. The biomass
139 sample used were pine (PI), eucalyptus (EU), willow (WI), olive waste (OW) and steam
140 exploded wood ('black') pellet (BP). Each fuel was characterized in terms of chemical
141 composition by proximate analysis, ultimate analysis, and gross calorific value. Results are
142 presented in Table 1.

143 **Table 1. Proximate and ultimate analysis of the samples used.**

		Pine	Eucalyptus	Willow	Olive residue	Black pellet	Basis	units
Proximate Analysis	moisture content	8.3	8.1	3.1	5.9	7.2	as received	%wt
	ash content	2	0.9	2	7.6	4.2	d.b.	%wt
	volatile content	80.85	82.55	81.93	74.01	76.07	d.b.	%wt
	fixed carbon	17.15	16,55	16.07	18.39	19.73	d.b.	%wt

		GCV (dry)	18.6	19.4	19.8	20.1	20.0	d.b.	MJ.kg ⁻¹
Particle density		0.48	0.67	0.52	1.3	1.26	d.b.	mg.mm ⁻³	
Elemental Analysis	C	46.45	48.76	49.78	48.79	50.49	d.b.	%wt	
	H	5.19	5.55	5.88	6.01	5.56	d.b.	%wt	
	O	44.98	44.69	41.85	36.13	39.47	d.b.	%wt	
	N	1.27	0.1	0.39	1.48	0.19	d.b.	%wt	

144

145 Each sample was milled, dried and sieved to obtain different ranges of sizes

146 The particle size ranges used for the experiments were 300--1400 μm . These sizes were
 147 selected to be high enough to have meaningful images for the shape analysis and targeting with
 148 the thermal imaging camera while being small enough to enable the assumption of uniform
 149 particle temperature (i.e. thermally thin). This upper range is also representative of the largest
 150 size fraction that would typically be used in a pulverized fuel boiler. Samples were dried in an
 151 oven at 115 °C for 2 hours to remove any moisture before combustion experiments in the wire
 152 mesh reactor.

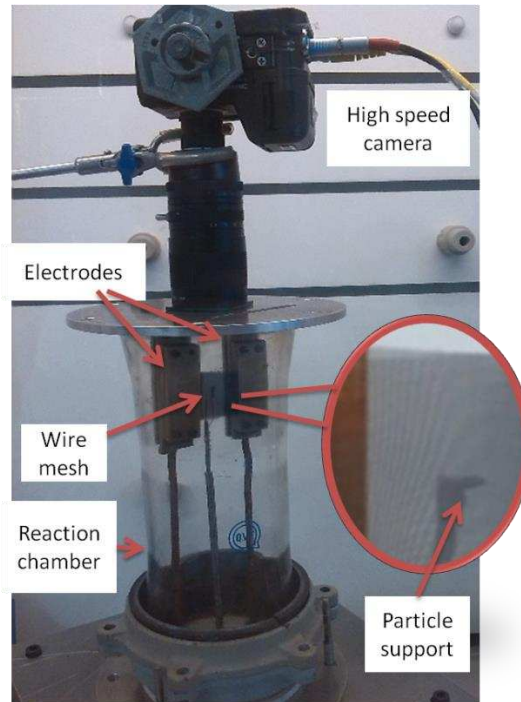
153

154 2.2. Experimental device

155 The experimental device used at the University of Edinburgh has been previously described at
 156 [7]. The wire mesh reactor (WMR) apparatus used in this work allows a stationary particle
 157 sample to be recorded with high speed video camera as it burns. The reactor was based on the
 158 WMR used by Flower et al. [5], and is shown schematically in Figure 1. The particle under test
 159 is supported between 2 vertical wire meshes that heat the particle by radiation, which is
 160 consistent and permits reproducible results. The electrical current through the elements can
 161 heat them to their operating temperature within 500 ms, which is small compared to particle
 162 burning times. The output current was limited to enable an operating mesh temperature of 900
 163 °C, and estimated heating rate of $\sim 10^3$ °C s⁻¹. The reactor temperature was measured with an
 164 k-Type thermocouple centred between the two meshes to ensure that the heat flux was
 165 consistent between runs, permitting valid particle-to-particle comparisons. The sample holder

166 was placed on the centre line between the meshes. The sample holder was made of the same
167 wire mesh material as the heating elements, forming a rectangle of 3 x 6 mm to place the sample
168 over it. The effect of mesh heating rate and final temperature affects the particle ignition delay,
169 but once the particle ignites the heat coming from the combustion becomes the main source of
170 energy for the particle temperature. Small errors on the mesh temperature will, therefore, have
171 little effect during the combustion performance and burnout time.

172 The high speed camera used in this study was a Phantom Miro eX4 with a zoom lens coupled
173 to a 20 mm expansion tube to give image magnification. It was placed on the top of the
174 apparatus. The camera-to-particle distance was fixed so that a consistent optical magnification
175 was achieved. The high speed video recording allowed a good temporal resolution to be
176 achieved. 500 frames per second were normally used for the recording exposure time of 3300
177 μ s and resolution 128X128 pixels. A PC was used to retrieve the images from the camera and
178 all the videos were analysed using Phantom Control Camera. The recording was played back
179 in real time and at reduced speed, allowing observation of much smaller particles and also
180 phenomena that would be missed with a normal camera. The times for the respective phases in
181 particle combustion were then accurately determined by processing the video image files and
182 representative rankings of burning times were obtained. All particles were weighed before the
183 experiment and the burnout time for each particle could be measured from observation of the
184 video.



185

186 **Figure 1. Single particle experimental setup.**

187

188 **2.3. Thermographic imaging**

189 Measurement of the surface temperature of the combusting particle was possible by use of
190 thermal imaging. In this study, a FLIR A600 series infra-red thermographic camera was used
191 for this purpose. This camera has a focal-plane-array thermal detector with a measurement
192 range up to 2273K and sensitivity specified as 0.05K with an accuracy tolerance of 2% of
193 reading (i.e. $\pm 30\text{K}$ at 1500K).

194 The images are derived from the intensity of the infra-red radiation detected by the thermal
195 detector in the camera. The temperature of the particle's surface can be derived from the data
196 although it is necessary to account for the emissivity of the surface. An emissivity of 0.85 was
197 adopted for the entire range of the combustion process being observed with thermal imaging
198 based on reported values for biomass char [28]

199

200 **3. Results and discussion**

201 **3.1. Ignition and combustion behaviour analysis**

202 All biomass fuels studied presented a sequential combustion with two well differentiated
203 stages, volatile flame and char combustion. This is in agreement with previous works
204 completed with this, and other single particle devices [2,3,4,7]. Upon ignition, the gases
205 released through the pores of the particle created a smooth volatile flame surrounding the
206 particle. This volatile flame became bigger and increasingly luminous and then shrinks to
207 extinction. Char ignition follows until completion of particle char burnout. Figure 2 shows
208 images from the videos of all biomass samples examined at different stages of combustion.
209 Video files can be found on supplementary material section. In order to measure the burnout
210 times and compare between the different tests, 0% burnout is the last frame before visible
211 ignition takes place and 100% the first frame where there is not any visible combustion at all.
212 The brightness and contrast of the images at 0% burnout were adjusted in order to have a clearer
213 image of the original particle. It is noted that OW was especially difficult to distinguish from
214 the background support so the particle perimeter is underlined in blue in the first image.

215 **3.1.1. Ignition delay**

216 The delay between the start of heating to particle ignition is very consistent between the
217 different runs and different samples. However the delay can increase significantly with higher
218 moisture content, as observed by previous studies [3].

219 **3.1.2. Devolatilisation**

220 After heating and drying, the pyrolysis reactions start to take place and they continue until
221 a stable char is formed. In fast pyrolysis, biomass decomposes very quickly generating mostly
222 gas, vapours, aerosols and char. The biomass pyrolysis starts releasing light hydrocarbons and
223 main products such as H₂, H₂O, CO, CO₂, and CH₄. These products start the ignition of the
224 volatile flame. The biomass particles appeared to start releasing volatiles a few milliseconds
225 before ignition, this was visible on the images of the olive waste which produced a blurry image

226 and darkening on the surroundings of the particle. Nevertheless, the ignition delay time is
227 negligible in respect to the overall burnout time.

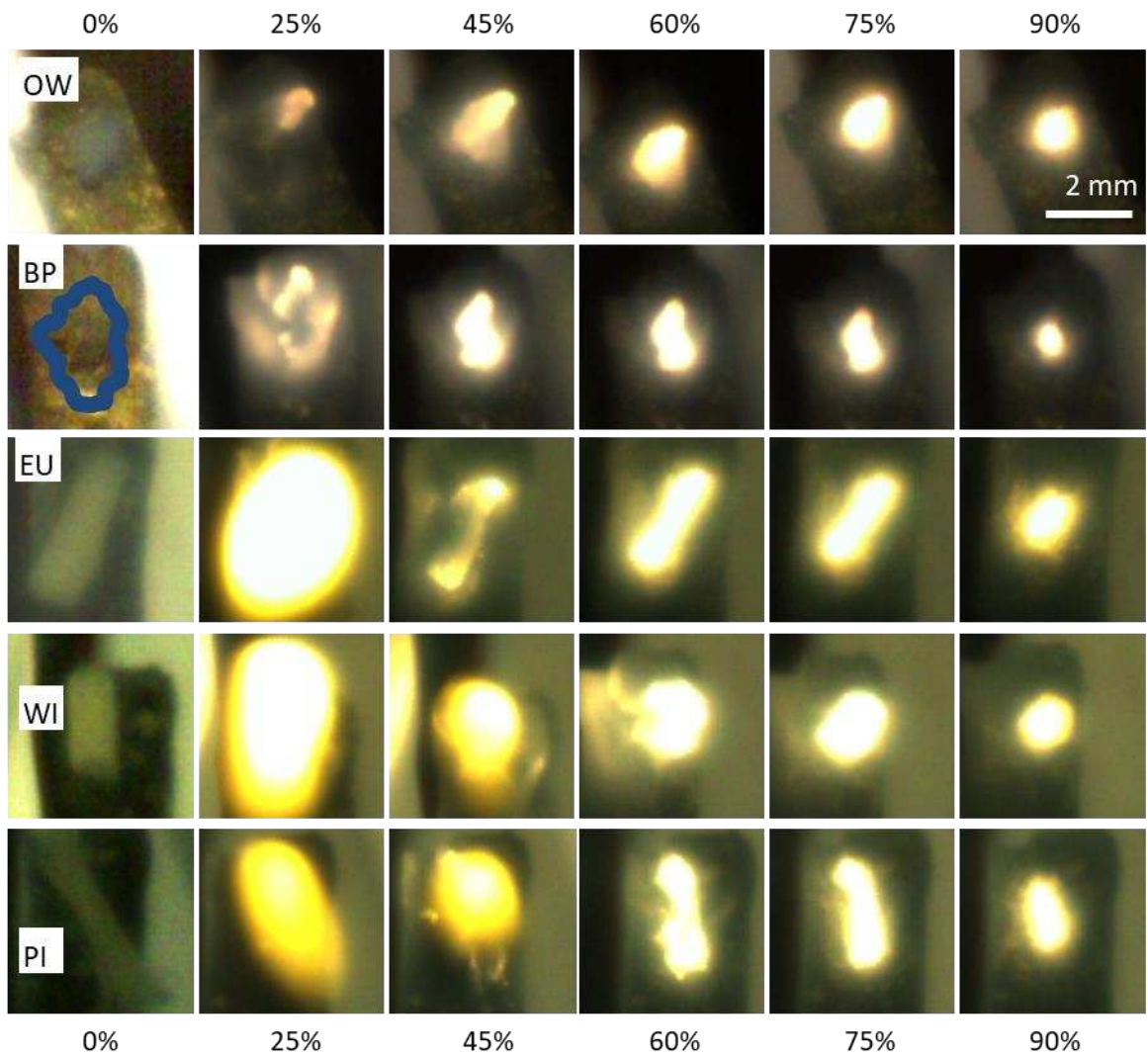
228 All samples showed homogeneous ignition and no significant differences were observed in
229 the volatile flame development. As the temperature increases, biomass fuels decompose into
230 primary volatiles. At temperatures above 773 K, the primary volatiles are subject to a secondary
231 pyrolysis, during which the tars are converted into a variety of gaseous species, especially CO,
232 light hydrocarbons, hydrogen and CO₂. The pores of the particle become open and connected
233 after a mass-loss front (moisture evaporation or devolatilisation) propagates from the particle
234 interior [22]. The gases from pyrolysis reactions flow through the porous medium inside the
235 particle reaching the surface and creating the volatile flame around the particle. As the pyrolysis
236 completes, the volatile flame shrinks as can be noted on the images for WI and PI comparing
237 25% and 45% burnout images on Figure 2. As the outward flow of pyrolysis products reduces,
238 the oxygen reaching the solid surface increases and char ignition is initiated.

239 There are notable differences between the samples regarding the times required for volatile
240 burning. The composition of biomass, the amount of volatiles and the pyrolysis kinetics of each
241 biomass, affects the duration of the volatile flame. Average volatile/char burning percentages
242 are presented in Figure 4. As shown on Figure 2, at 25% the video shows the volatile flame
243 surrounding the contour of the biomass particle very clearly in the case of EU, PI and WI. In
244 the case of BP the particle is in transition to char ignition, while the OW particle shown in the
245 figure has already completed the volatile combustion stage and the remaining char particle has
246 ignited at one end of the particle.

247 The EU particles had the most elongated shaped particles. Given that, it was observed that
248 the volatile flame in the centre of the particle and char ignition on both ends of the particle
249 overlapped for a few milliseconds. On the contrary, BP sometimes exhibited a few milliseconds
250 delay between the volatile flame extinction and the complete ignition of the char. This is not

251 the case on the BP particle on Figure 2 at 25% where there is still some blurriness due to the
252 end of the volatile flame while the char particle on the background has already ignited.

253 WI and PI continue showing volatile flame at 45% burnout, and even at 60% burnout the
254 WI particle shows a small volatile flame while the char has completely ignited.



255
256 **Figure 2. Images from the combustion recording of: olive waste (OW) steam torrefacted**
257 **wood pellet (BP), eucalyptus (EU), willow (WI) and pine (PI) particles at different**
258 **percentages of burnout time.**

259

260 3.1.3. Char combustion

261 For some cases the flame front could be observed during particle char ignition. Comparing
262 OW particle at 0% and 45% burnout in Figure 2, it can be observed that the particle moved or
263 changed during the devolatilization stage. In this particular case, the particle just rolled over
264 the support. Changes were occasionally observed with deformations and bending of some
265 particles. Swelling has also been observed in some cases; however there was not a significant
266 number of swelling cases at the volatile burning stage for any of the samples tested to quantify
267 this. The majority of them remain unchanged or shrink slightly on the major axis (length). An
268 increment in the size of the particle was observed on the minor axis (width) in some particles
269 during devolatilization. Rather than swelling, as some bituminous coals do, this phenomenon
270 can be attributed to the separation of fibres of the particle. Images from the char at the different
271 stages of combustion evidenced particle shape changes. The observations of particle shape
272 and projected area evolution observations are addressed in detail in section 3.4.

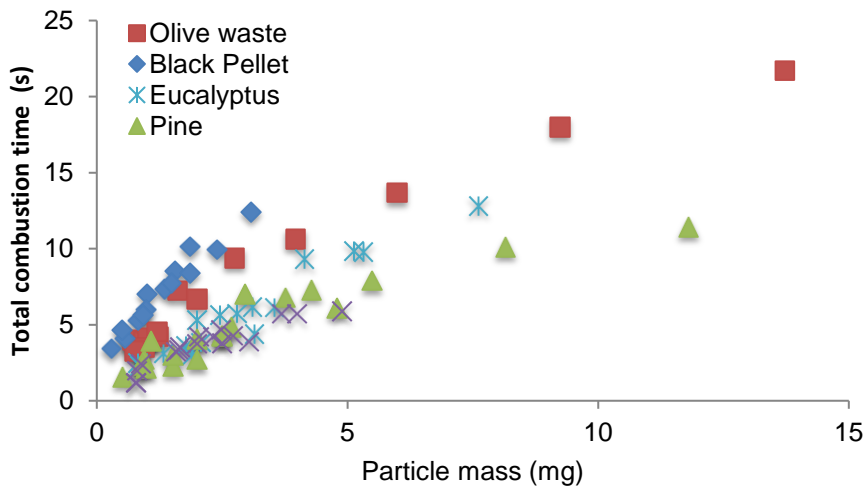
273 At the end of the particle combustion the ash remains incandescent over the support for a
274 few milliseconds before cooling down. It is observed that the ash distributed all over the particle
275 is transported during char combustion. While the particle is consumed the ash reaches the
276 particle surface. In some of the particles, the ash creates a coarse surrounding the particle. At
277 100% burnout, the coarse ash remains incandescent with a shape that is proportional to the char
278 particle on its last stage of combustion.

279

280 **3.2. Burnout time for single particle combustion**

281 Particles of different weight and sizes were evaluated through the single particle
282 experimental device in order to study the effect of particle size on the combustion behavior of
283 the fuels as well as their burnout time. The variability on the aspect ratio of the woody biomass
284 particles makes the relationship between particle cut size and particle mass imprecise. As a
285 result, it is difficult to predict accurately a biomass particle mass for a given particle cut size.

286 Therefore this relationship needs to be taken with its limitations and results are represented
 287 based on particle weight not on their size. The corresponding range size will depend on
 288 roundness of each fuel particle. A significant number of experiments were done with each
 289 sample on a range of weight from 0.5mg to 5 mg. Some experiments were done with particles
 290 above 5 mg in order to ensure the trends obtained are in agreement with previous results [3, 5,
 291 7] The particle weight versus burnout time for each particle of the different biomass tried is
 292 plotted in Figure 3. The deviation of the trend is large for some of the particles also because
 293 the heterogeneity of biomass particles composition.



294

295 **Figure 3. Burnout time obtained for different biomass samples**

296 The burnout time increase with the increment of particle weight for all samples. The
 297 relationship between weight and burnout time results for biomass particles were consistent with
 298 that observed in previous SPC studies [3] where the data could be fitted to a regression function
 299 of the form $t=a \cdot \text{mass}^b$ where a and b are empirically derived coefficients. In this study, the best
 300 adjustments were obtained for a second order polynomial function. Equations and R^2 values
 301 are represented on Table 2. The burnout time trend lines versus particle mass cannot be directly
 302 transposed to large scale conditions burnout times, as the heat transfer and combustion
 303 conditions are different, but it provides a useful means to compare different biomass samples.

304 For a better comparison among the samples, values for extrapolated burnout time for volatiles
 305 and char for a particle of 2 mg are plotted on Figure 4.

306

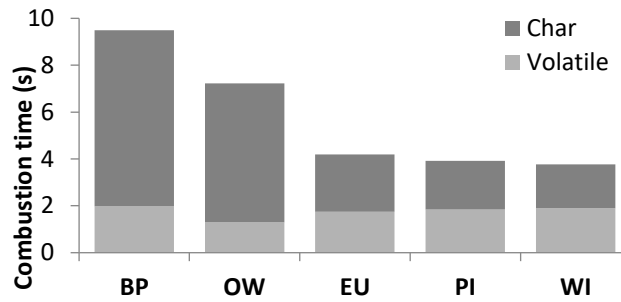
307 **Table 2. Burnout time, normalized projected area and char temperature observed for the**
 308 **different samples**

Sample	Burnout time	R2	Range	Char Temperature	Normalized Particle Projected Area	R2
BP	$t = -0.384m^2 + 4.454m + 2.11$	0.959	0.5 - 3 mg	-	$A = -79.93t^2 + 5.565t + 100$	0.965
OW	$t = -0.428m^2 + 4.34m + 0.25$	0.951	0.5 - 4 mg	-	$A = -169.49t^3 + 83.997t^2 + 19.314t + 100$	0.9864
EU	$t = -0.028m^2 + 1.857m + 0.59$	0.935	1 -7 mg	1405	$A = -88.83t^2 + 8.6t + 100$	0.959
PI	$t = -0.075m^2 + 1.787m + 0.64$	0.979	0.5 - 5.5 mg	1422	$A = -271t^3 + 319t^2 - 123t + 100$	0.9593
WI	$t = -0.246m^2 + 2.43m - 0.11$	0.943	1 -5 mg	1392	$A = -115.6t^2 + 55.5t + 100$	0.981

309

310 The differences observed between samples must be attributed to the differences on reactivity
 311 of the fuels. The regression functions for WI, EU and PI were almost overlapping. WI was
 312 slightly lower, which suggests a higher reactivity, being therefore the most reactive of all
 313 samples tried. OW was notably above those. Olive waste is been reported in the literature to
 314 have lower reactivity than other biomass due to high content in lignin [29, 30]. The steam-
 315 exploded wood pellet (BP) sample showed the lowest reactivity with the longest burnout times.
 316 The raw material before torrefaction was not available to compare in this study, so the low
 317 reactivity cannot be attributed to the process of torrefaction alone. However, Jones et al [31]
 318 reported that torrefaction produced less reactive biomass chars, similarly Vorobiev et al. [32]
 319 reported that the char burnout rate was found to decrease with torrefaction intensity of beech
 320 wood. Other works by Lu et al [33] did not find differences on intrinsic reactivity as determined
 321 by thermogravimetric analysis due to the torrefaction pretreatment, but the torrefaction
 322 increased char yield and char particle density.

323



324

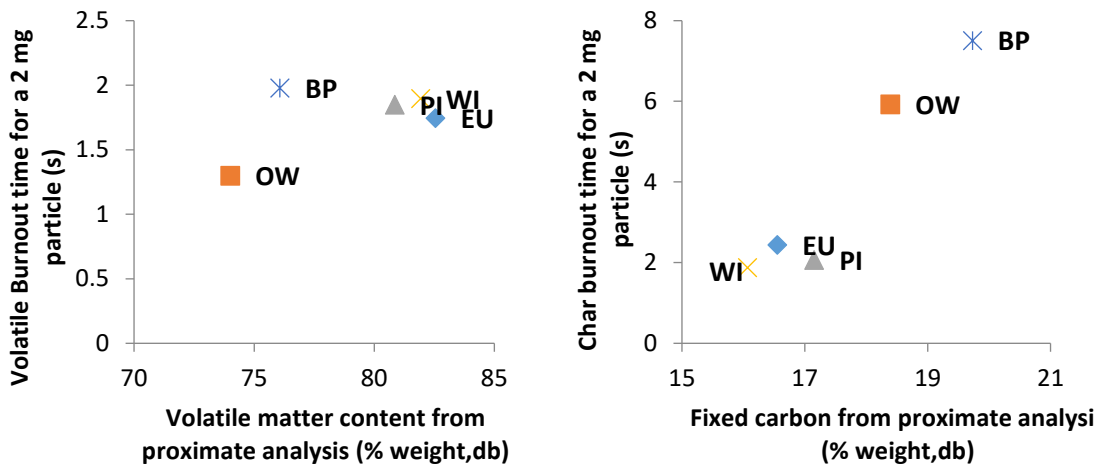
325 **Figure 4. Average burnout time obtained for a 2 mg particle of the different biomass**
 326 **samples.**

327 The average volatile/char time ratio remained constant along the range of particle sizes tried
 328 (1-5 mg) for all samples. For the biomass samples EU, PI and WI, the volatile pyrolysis and
 329 burnout time was practically equal to the char burnout time. These three samples showed
 330 similar times for volatile burning, aligned with very little differences on volatile matter content
 331 from the proximate analysis. The volatiles combustion times account for 40 to 50 % over the
 332 total combustion time, which is clearly distinctive to OW and BP that account 18 to 20 % on
 333 average. This is partially correlated with volatile matter content obtained on the proximate
 334 analysis that ranges 80.8-82.5 %, as it can be seen on Figure 5. Olive waste shows a shorter
 335 volatile burnout time, as expected due to lower volatile matter content from proximate analysis
 336 (74%). Torrefied pellet has a 78% volatile matter on proximate analysis, however the burnout
 337 time is slightly higher than the higher volatile matter biomass but the main difference is on the
 338 char burnout time, that may be attributed to low fuel reactivity.

339 Particle density is significative higher on both OW and BP, this has a worsening effect on the
 340 heat transfer from the mesh to the particle that could also delay the pyrolysis and char
 341 combustion reaction.

342

343 The correlation between parameters from proximate analysis, such as volatile content or fixed
 344 carbon is not enough to predict burnout times for volatile matter or char combustion times,
 345 reflecting the complexity of combustion process on pyrolysis and combustion kinetics.



346

347 **Figure 5. Burnout time for volatiles (a) and char (b) for a 2 mg particle of the different**
 348 **biomass samples versus volatile matter and fixed carbon from proximate analysis**
 349 **respectively.**

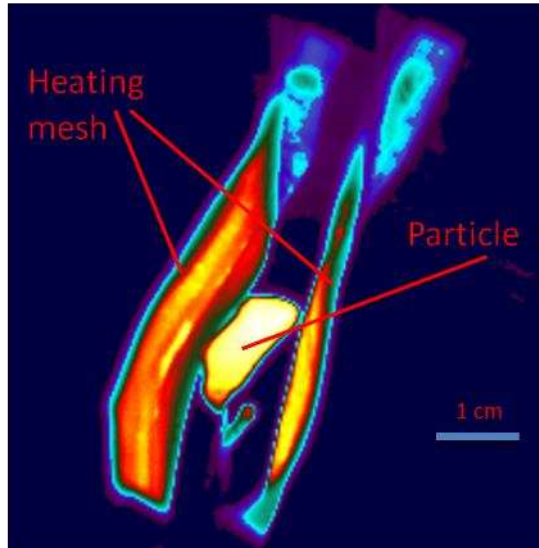
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351 Particle size distribution and milling process is important for an efficient combustion of fuels.
 352 This is especially important on fuels with low reactivity. However for biomass with high
 353 reactivity, the burnout time exhibits a lower variation with particle weight or size. This would
 354 allow the fuel processing system to be more flexible on its outcome and have less milling
 355 energy requirements. In those cases, other parameters such as moisture could be as relevant as
 356 the particle size.

357

358 3.3. Thermal imaging

359 The single particle combustion experiment was performed also using a thermal imaging
 360 camera instead of the high-speed camera. The experiments were done for three of the samples:
 361 PI, EU, WI. Images were obtained through the entire combustion process for a number of
 362 particles of each sample. An example image is presented in figure 6. The surface temperature
 363 of the particles was subsequently extracted from the image data and the results are presented
 364 in figure 7.

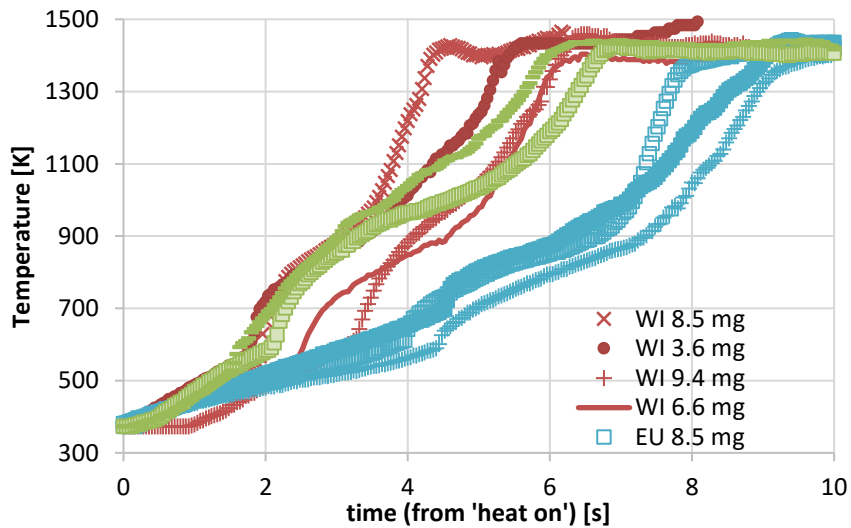


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366

367 **Figure 6 – Thermographic image of particle of EU biomass in wire-mesh reactor**

368



369

370 **Figure 7 . Surface temperature time-history profile of combusting particles of PI, WI**
 371 **and EU derived from thermal imaging.**

372

373 The surface temperature profile for a number of particles of PI, WI and EU samples is
 374 presented in Figure 4. After start heating, the particle surface temperature raises up relatively
 375 slow until volatile ignition. Two main increments of the heating rate are observed on the
 376 temperature-time profile, corresponding with two changes on slope trend. First notable

377 increment of the surface heating rate is observed when the volatiles ignite with surface
378 temperatures in the range of 500 to 600 K. Pine particles seem to develop a slower heating rate
379 on the surface of the particle. The surface temperature continues increasing on a slower heating
380 rate after volatile ignition. Second increase of the heating rate is related with char ignition, with
381 temperature raising fast until reaching a plateau at a temperature on the range of 1390- 1430
382 K. Average plateau temperatures for the three samples tested are listed in Table 2. No
383 significative differences were observed among the biomass samples on the plateau temperature
384 during char combustion.

385

386 **3.4. Shape change during particle combustion**

387 No regular behaviour was detected for all particles on their shape and size evolution during
388 combustion. Despite similar origin, shape, size and conditions, different behaviors were
389 observed on the particles of each fuel. In some aspects, even opposite trends were observed as,
390 for example, some particles swell and some shrink at the pyrolysis stage.

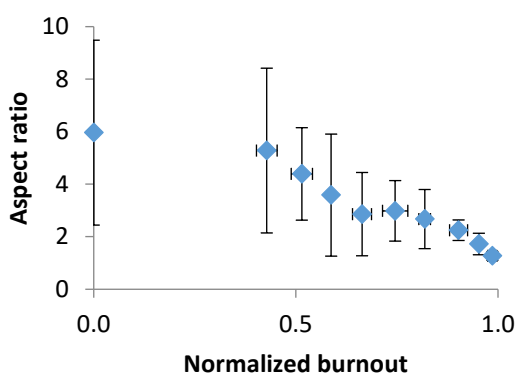
391 Length and width dimensions of the particle where recorded at different stages of combustion.

392 A proportional decrease of the particle size would be initially expected during the char
393 combustion. However, this is not what was observed for these samples. As the char combustion
394 progresses the particles instead become more rounded. The shape of the particles with higher
395 aspect ratios changes from a fibrous shape to an ellipsoid that appears to be consumed only in
396 the direction of the length. During volatile flame combustion the particle was not visible at all
397 due to flame screening, however measurements on particle size before ignition and during char
398 combustion where relatively straightforward to take from the video frames. In order to facilitate
399 the comparison among the particles and samples, combustion time, normalized to 100% of total
400 burnout time, for each particle was used. The aspect ratio defined as the length divided by width
401 was also calculated at the different stages of combustion. Some particles were observed to bend

402 over themselves and therefore had to be discarded for the aspect ratio analysis. Some particles
403 were cut manually so the initial dimensions do not necessarily represent the dimensions coming
404 from an industrial mill. The aspect ratio however is on the range of values that would be
405 expected on a large scale milling. Initial aspect ratio for EU, PI and WI were higher than BP
406 and much higher than OW.

407

408 The sample of pine biomass (PI) used had the largest aspect ratio of the samples tried with an
409 initial value of 5.96. No swelling was observed on the vast majority of PI particles. After
410 devolatilization most of the particles shrink slightly on their length, keeping normally the same
411 width as the initial particle. So the aspect ratio of the char obtained after devolatilization was
412 close to the size of the initial particle or slightly lower. During the char combustion stage, the
413 particles are consumed mainly on the tips, while the observed width of the particle remains on
414 similar values as the initial particle until well advanced burnout times. Therefore the aspect
415 ratio of the particle reduces due to length reduction of the burning char. Figure 8 represents the
416 average observed aspect ratio of PI particles as well as the standard deviation versus the
417 normalized burnout time.

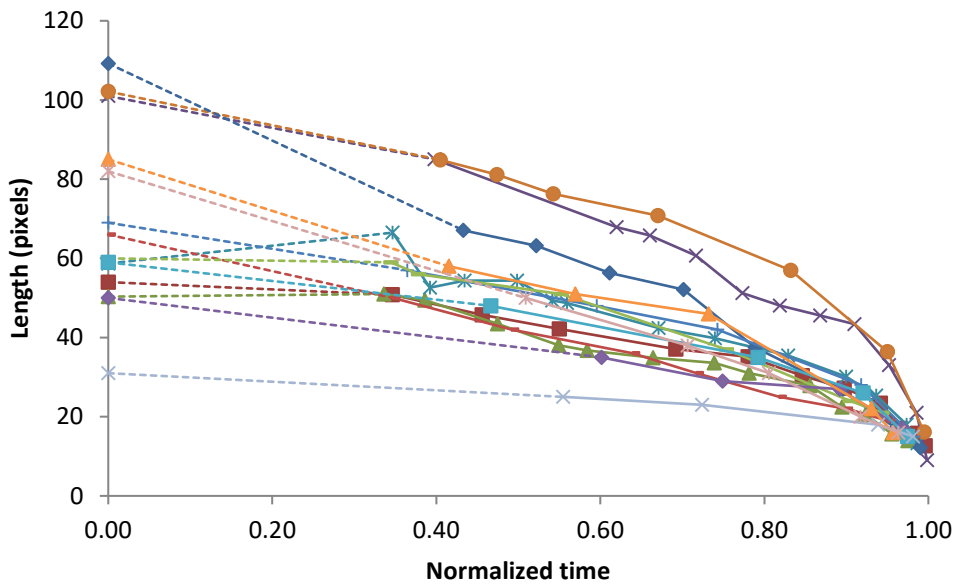


418

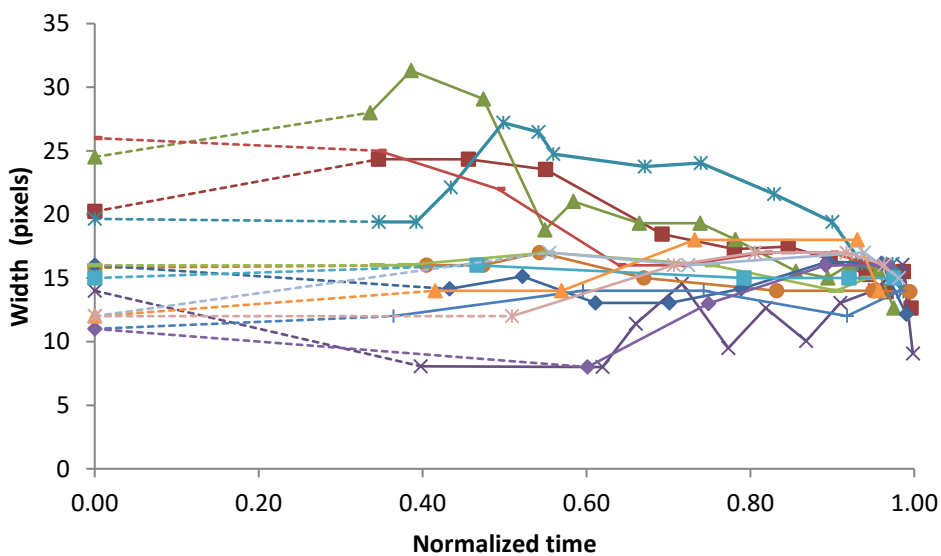
419 **Figure 8. Evolution of the average aspect ratio during combustion of PI particles.**

420

421 EU particles had an initial aspect ratio of 4.7, large differences are observed between length
422 and width. A few representative particles of the evolution of the dimensions of EU particles
423 are represented on Figure 9 (length) and Figure 10 (width). On the length the particles observed
424 gradually decrease on size, the dimension evolution on the largest size of the particle generally
425 trend to an asymptotic tendency towards a value around the width.



426
427 **Figure 9. Evolution of the length measurements during combustion for various EU**
428 **particles.**



430
21

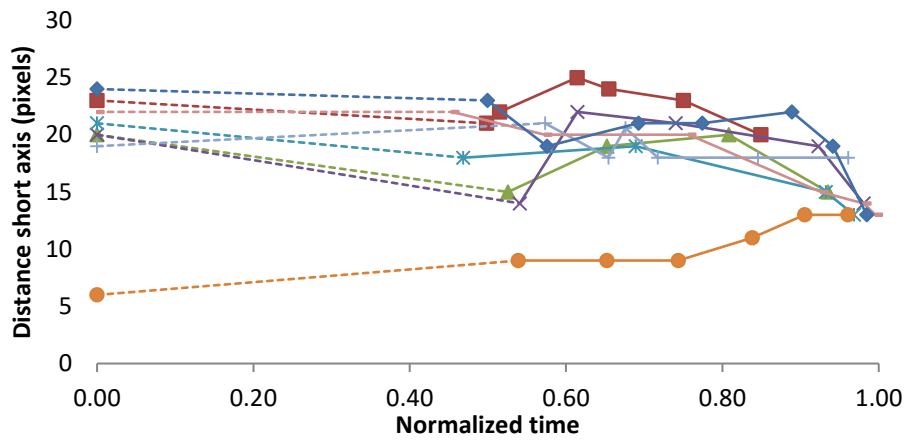
431 **Figure 10. Evolution of the width measurements during combustion for various EU**
432 **particles.**

433 Diverse behavior is observed on the width, some particles present apparent swelling while
434 others remain constant or decrease slightly. Those particles that swell during the first step of
435 volatile release, decrease shortly after the ignition of the char surface. The swelling of the
436 particle create a rapid enlargement of the particle dimension. The surface area and higher
437 particle volume created reduces the density considerably. Therefore, once ignited, the char
438 surface is rapidly consumed by the combustion reactions, which explains the fast change on
439 the width on the particles that swell during the volatile combustion. Some particles showed
440 swelling during different stages of combustion. Some particles showed swelling during the first
441 stages of char burning. This could be associated with a higher temperature of the particle during
442 char combustion compared to volatile combustion stage [8] and therefore a softening of the
443 particle while pyrolysis reactions continue occurring inside the particle during char
444 combustion. Particles presenting swelling are generally observed to have slightly shorter
445 burnout time than the average.

446

447 WI particles did not show significant swelling. The particles shrink gradually on their length
448 on the second half of the char combustion. As observed in the videos, the particle dimension
449 on the width remains constant during the combustion until very end of the combustion (over
450 95% burnout).

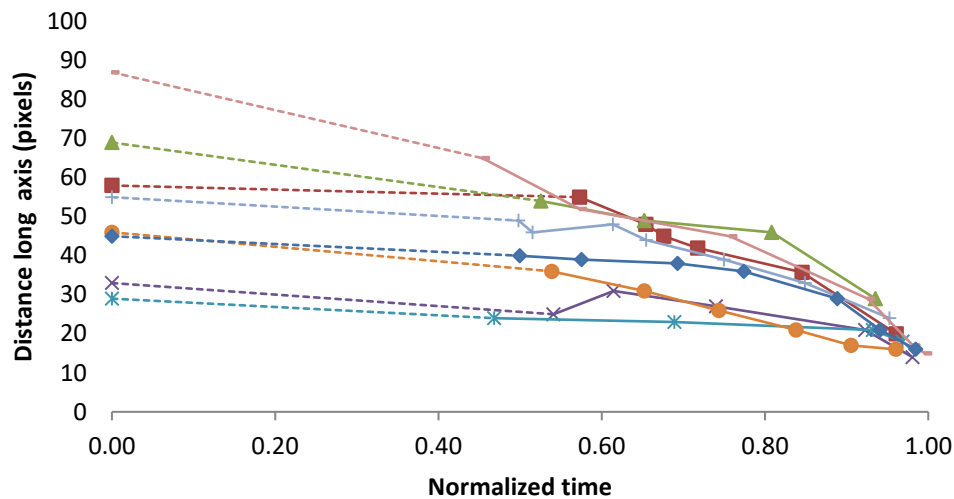
451



452

453 **Figure 11. Evolution of the width measurements during combustion of various WI**
 454 **particles.**

455



456

457 **Figure 12. Evolution of the length measurements during combustion of various WI**
 458 **particles.**

459

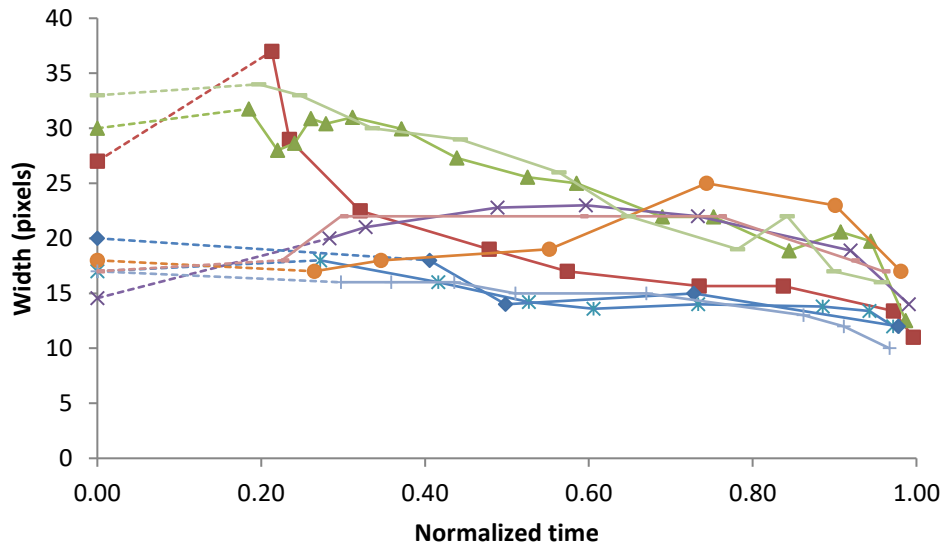
460 Neither WI nor PI samples showed any swelling. A slight decrease in the length was observed
 461 on the willow particles after devolatilization stage that on average takes half of the total burnout
 462 time. No evidence was seen for particle shape changes during the volatile combustion stage for
 463 willow sample.

464 It is notable that particle width remains practically constant during combustion, even increasing
465 in some particles. This increase does not fit on the trend of a swelling particle, and is more
466 likely due to partial melting and spherification process. This was observed by Panahi et al. [21],
467 where much longer and thinner particles were observed to fuse and bend over themselves
468 leading to a almost round char. The different initial aspect ratio on the samples tried in this
469 study and [21] indicates that thin and long particles seem to have higher propensity to
470 spherification during the pyrolysis stage, although this process is not observed for all samples.
471

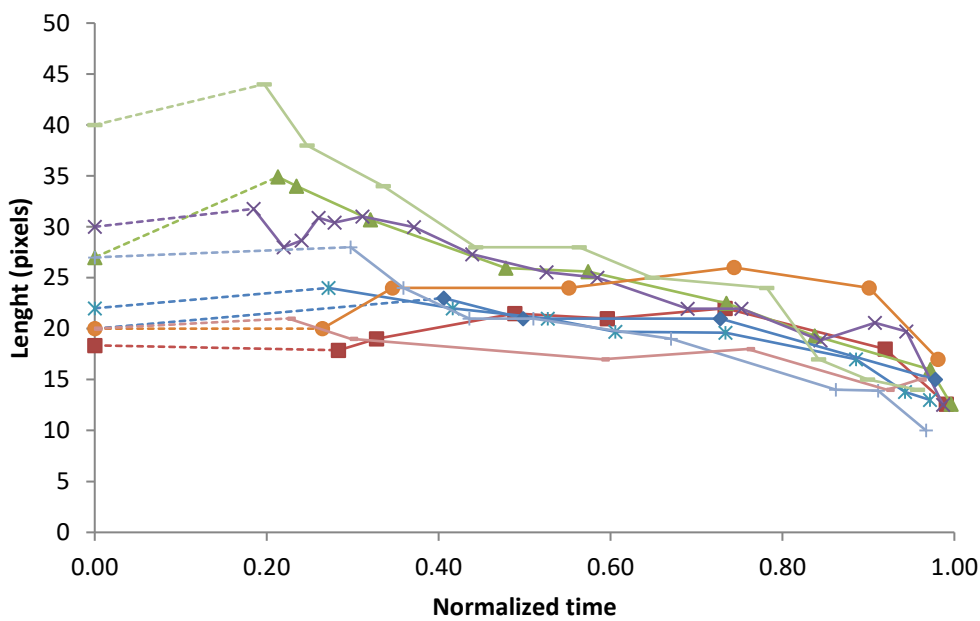
472 The olive waste has been extensively pressed in order to extract the oil so the raw particles are
473 quite round with initial aspect ratio of 1.3 ± 0.36 . Differences on the dimensions of width and
474 length were small. The size of the particle is reduced during combustion, however it doesn't
475 decrease proportionally during time. Particles remain on proportionally large until almost the
476 end of the combustion. At 90% of the combustion time, some particles remain with similar
477 volumes, while some have been reduced to half and others are even larger due to particle
478 swelling. The volatile flame stage is much shorter on the olive waste sample. It was noted that
479 some particles showed swelling during char combustion stage. As previously described for the
480 EU sample, a rapid swelling on the pyrolysis stage is normally followed by a quicker
481 consumption of the particle and a large decrease in particle size shortly after the swelling. This
482 was also the case in a few OW particles, as represented in Figure 13. Both OW and BP present
483 a high particle density and more compact structure and much lower aspect ratio than the rest
484 of the samples. As described by Biagini et al [34], woody biomass samples present more
485 particles with a fibrous structure that is likely to shrink and fragment. While dense particles
486 with a compact structure are expected to experience minor variations in size

487

488 Particles presenting thin aspect do not decrease in size on the short edge size during
489 combustion. This observation was noted on all the samples studied. As observed in Figures 11
490 and 13, particles with initial small width, remain constant or decrease during combustion and
491 end with very similar width dimensions as the initial particle.



492
493 **Figure 13. Evolution of the width measurements during combustion of various OW**
494 **particles.**

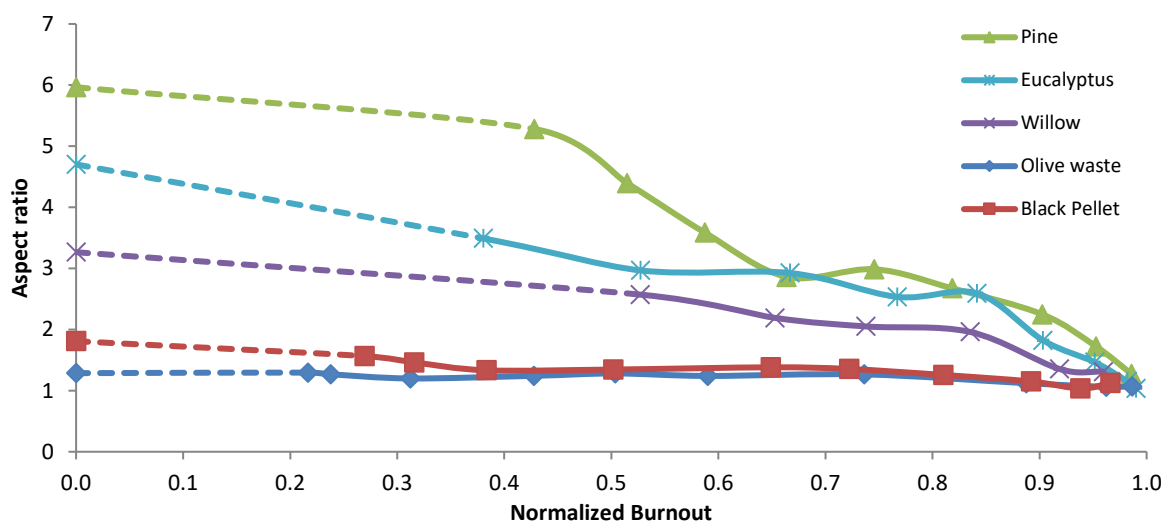


496

497 **Figure 14. Evolution of the length measurements during combustion of various olive**
 498 **waste particles.**

499

500 Torrefaction tends to make biomass particles brittle [35]. So, during the processes of milling,
 501 pelletization and pellet milling, the particles break, resulting in a lower aspect ratio than a non-
 502 torrefied biomass fuel. The fuel sample from the steam-exploded wood pellets reflects this
 503 brittleness with an initial average aspect ratio of 1.8, which is low for a woody biomass. The
 504 aspect ratio evolution reflects a tendency for the particles to become more rounded at early
 505 stages of char combustion. Figure 14 represents the average aspect ratio of the different samples
 506 examined. During the volatiles combustion stage the particle was not completely visible and
 507 the measurements could not be taken, so the linear evolution assumed during that time is
 508 represented with dotted line.



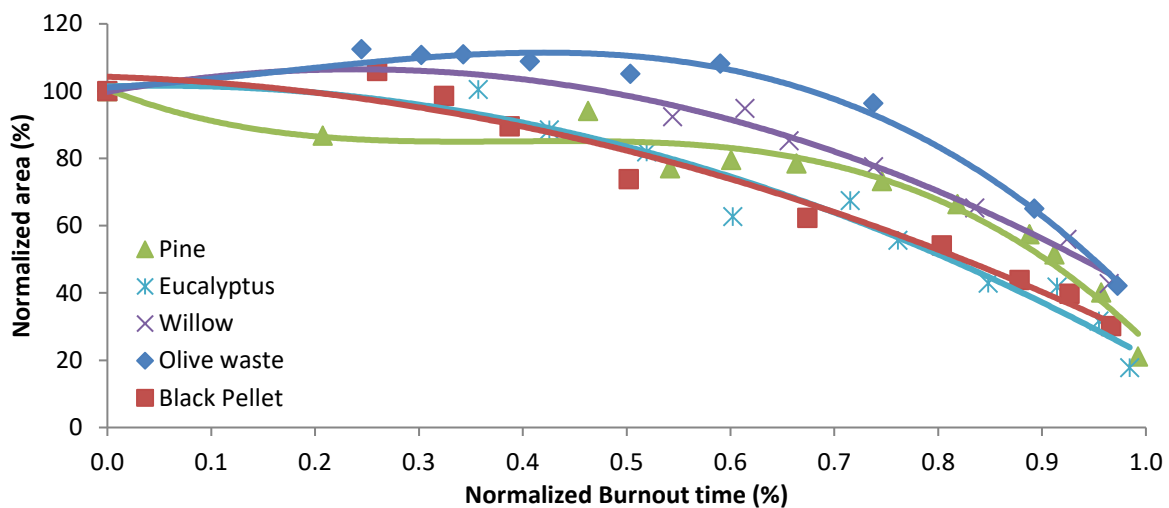
509

510 **Figure 15. Evolution of the average aspect ratio of all samples during combustion.**

511

512 Particle density is a key factor during combustion process. The samples with higher particle
 513 density tried were OW and BP. These samples presented very small aspect ratios compared to
 514 the rest of the samples, with close valued for width and length. The projected areas even
 515 increase slightly during pyrolysis and sometimes during char combustion as well, as observed

516 average measurements on Figure 16. The char density has been previously reported to change
 517 dramatically during combustion [30], not only due to pyrolysis but also due to the combustion
 518 process on the char surface. The heterogeneous combustion reactions take place all over the
 519 porous surface of the char and as progresses the oxygen reaches areas closer to the particle
 520 centre. Thus the particle volume does not decrease proportionally to the combustion rate. As
 521 observed from the projected area on the images, the particles with well advanced combustion
 522 rates still have relatively large volumes. Projected area has been calculated assuming the
 523 particles as ellipsoids, The equations for the projected area for normalized area of the different
 524 samples are listed on Table 2, Regression lines of 2nd and 3th order polynomial have been
 525 selected for best fit on each case according to R² values,



526

527 **Figure 16. Evolution of the normalized projected area during combustion.**

528

529 Li and Zang [36] developed a model for combustion of spherical particles that predicts the
 530 particle shrinking during the char combustion. The model demonstrated that char consumption
 531 is not uniform on the particle surface, being higher at the semi-major axis than at the semi-
 532 minor axis. This leads to a decrease in the ellipsoidal particle size. The shape of the particle
 533 has been also modelled in previous works, such as Yang et al [22], this works describe with
 534 detail a model for a single particle combustion comparing shapes such as sphere and cylinder.

535 The combustion rate on the tip of the cylinder shape is predicted to be higher due to the thinner
536 boundary layer of the reactions front wave. Gradually the particle is assumed to become more
537 round due to higher combustion rate on the thermally thinner parts of the particle. The
538 observations on the particle shape from the videos of the combustion experiments presented in
539 this study do not match with a regular shrinkage. The changes on the shape cannot be only
540 attributed to a higher combustion rate on the tips of elongated particles. This is due to two main
541 observations: first, some particles preserve or even enlarge their width during combustion;
542 secondly the length of the particle shrinks on an irregular rate, accelerating by the end of
543 combustion. This second observation can be attributed to the progressive reduction of the
544 density of the particle. As the biomass char mass is consumed in the combustion process it
545 becomes lighter. The projected area does not reflect a linear reduction during combustion time.
546 As the width of the particles remains practically constant until the end of combustion, the
547 projected area is directly linked with the length of the particle. This relation is stronger for more
548 elongated particles. At the end of the combustion, with 90% burnout the particles are generally
549 equant in width and length. This observation is in agreement with other works such as [37],
550 where the wood particles were observed to change shape looking like droplets at the end of
551 pyrolysis at 1573 K on a drop tube reactor.

552 The projected area then remains circular and a final, very quick, reduction of the area is
553 observable after 90-95% burnout until the end. This has been previously observed by Levendis
554 et al. [38] and Khatami et al. [6] and is described as shrinking core combustion behavior by
555 Shan et al [39].

556 Particle temperature recorded with the thermal camera on the range of 1200-1400 K is enough
557 for melting of the main components of biomass (cellulose and hemicellulose). The melting is
558 likely to happen close to the surface, where combustion takes place and temperatures are high,
559 however the heat transfer between the surface and the centre of the particle is slower than the

560 combustion reaction and prevents the particle to become completely melted under the
561 conditions of this study. Mineral matter among the biomass can play a role in char softening
562 and shape changes, inorganic elements could be spread through the biomass material, and as
563 would be in the particles. Some elements such as Mg, P or K are present in higher concentration
564 in agricultural residues and can decrease the melting point of the ash due to eutectic melting
565 points. Proximate analysis on the different biomass shows high ash content on olive (7.2%),
566 and steam exploded pellet (3.9%) being low in willow and pine (1.8% both) and very low
567 (0.8%) in Eucalyptus. All of them showed shape transformation at different steps of the
568 combustion

569 The temperatures recorded during combustion can be enough to melt certain elements of the
570 ash by mid to end of combustion, as normally ash deformation temperatures for some biomass,
571 can start at 1200 K for biomass rich in K, being higher for those rich in Si or Ca. However the
572 main constituent will be organic material and therefore the main effect this inorganic elements
573 is the catalytic effect that can take part during pyrolysis. Some inorganic elements will catalyze
574 the pyrolysis reactions, accelerating the cracking reactions and allowing the char to get
575 plasticity at low temperatures.

576 Relatively low temperatures have been proven enough to part-melt biomass char compounds
577 as bituminous coals. Trubetskaya et al. [40] observed softening of the solid char matrix already
578 at 350 °C, being greater with increasing temperature and heating rate. This temperature is far
579 below the normal range of ash deformation temperatures. This can explain the slight increase
580 on the width of OR at early stages of the combustion, when temperatures are not expected to
581 be enough to melt main biomass initial constituents.

582 Some particles with prolate shape showed an irregular shape change in various stages. As it
583 can be observed in figure 15 for all samples and particularly for pine in figure 8, the aspect
584 ratio does not decrease following a linear regression. This can be explained due to the different

585 temperature on the out layer of the particle, allowing the particle to deform in the more exposed
586 areas

587 On a first stage the tips get rounded, decreasing the aspect ratio. However the inner structure
588 of the char at the center of the particle remains at lower temperature keeping the particle on an
589 ellipsoid shape until the combustion is well advanced. At the end of combustion, the
590 temperature gradient is negligible and the whole particle is at enough temperature to allow
591 plastic deformation to an almost spherical shape.

592 Surface tension of melted components of the char particle can play an important role on the
593 conformation of particle shape leading to a more rounded particle. This effect is stronger at the
594 end of combustion when particle has been heated for longer times and the combustion front
595 wave has reached the particle centre. Temperature gradient inside the particle will be then
596 negligible and high temperatures, close to the temperature measured on surface are expected at
597 the entire particle.

598

599 **4. Conclusions**

600 All samples presented homogeneous ignition and sequential burning of volatiles and char. BP
601 and OW had shorter volatile burning time, and longer char and total burnout times, .PI, EU,
602 WI were very close to each other with the half of total burnout time of BP. Temperatures
603 measured with the thermal imaging camera for PI, EU, WI show a plateau during char
604 combustion on the range of 1390-1430 K, with no significative differences among the samples.

605 The observed evolution of the dimensions of the particles differs from the gradually reduction
606 on size that would be initially expected. Little changes on the shape and size were observed
607 after the pyrolysis stage. Nor WI nor PI samples presented any swelling, while some particles
608 of OW presented slight swelling in some cases during pyrolysis or char combustion stages.
609 During char combustion the particles become more round. The shape of the particles with initial

610 fibrous shape, changes to an ellipsoid that apparently seems to be consumed only in the
611 direction of the major axis. During char combustion process, long fibrous particles reduce their
612 length, while the observed width of the particle remains on similar values as the initial particle
613 until well advanced burnout times. The projected area of the char particles with well advanced
614 combustion rates still have relatively large values. At the end of the combustion, with 90%
615 burnout the particles are generally equant in width and length. Partial melting of the particle
616 during char combustion allows the surface tension to modify the particle shape, leading to a
617 more round particle. The projected area then remains as a circle and finally a very quick
618 reduction of the area is observable after 90-95% burnout until the end.

619

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626

627 **Supplementary material**

628 Video files are provided as supplementary material associated with this article and can be
629 found in the online version research data repository Mendeley Data.

630

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