

This is a repository copy of *The role of Ca2+ ions on Ca/Fe carbonate products on X65 carbon steel in CO2 corrosion environments at 80 and 150 °C*.

White Rose Research Online URL for this paper: http://eprints.whiterose.ac.uk/146118/

Version: Accepted Version

Article:

Shamsa, A, Barker, R orcid.org/0000-0002-5106-6929, Hua, Y et al. (3 more authors) (2019) The role of Ca2+ ions on Ca/Fe carbonate products on X65 carbon steel in CO2 corrosion environments at 80 and 150 °C. Corrosion Science, 156. pp. 58-70. ISSN 0010-938X

https://doi.org/10.1016/j.corsci.2019.05.006

© 2019 Elsevier Ltd. All rights reserved. This manuscript version is made available under the CC-BY-NC-ND 4.0 license http://creativecommons.org/licenses/by-nc-nd/4.0/.

Reuse

This article is distributed under the terms of the Creative Commons Attribution-NonCommercial-NoDerivs (CC BY-NC-ND) licence. This licence only allows you to download this work and share it with others as long as you credit the authors, but you can't change the article in any way or use it commercially. More information and the full terms of the licence here: https://creativecommons.org/licenses/

Takedown

If you consider content in White Rose Research Online to be in breach of UK law, please notify us by emailing eprints@whiterose.ac.uk including the URL of the record and the reason for the withdrawal request.



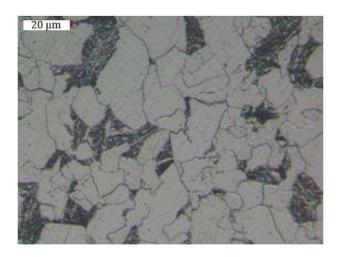


Figure 1: X65 carbon steel microstructure

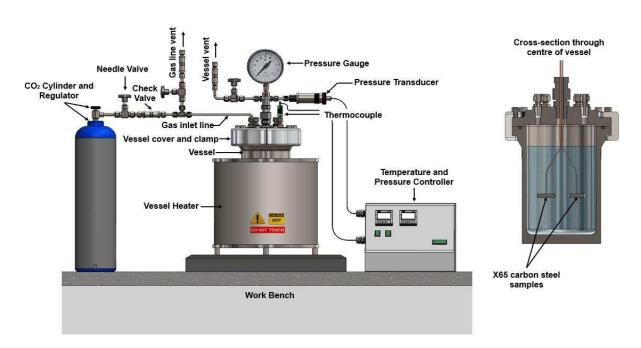


Figure 2: Schematic of autoclave set-up for evaluation of X65 carbon steel corrosion in CO_2 -containing environments at elevated temperature

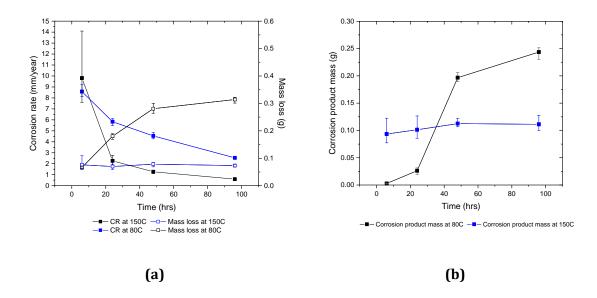
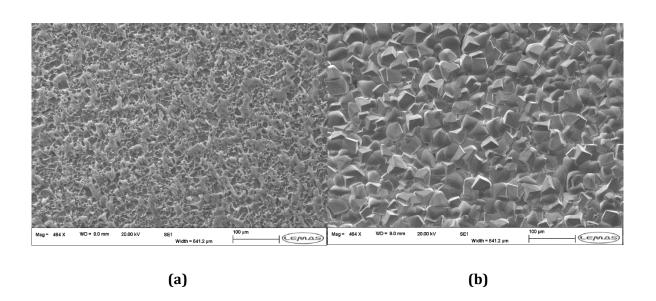
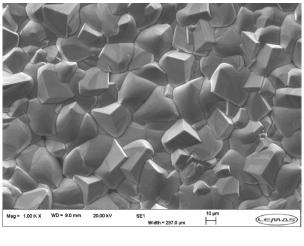


Figure 3: (a) Mass loss measurements and the corresponding calculated corrosion rates in mm/year and (b) corrosion product mass determined from mass gain measurements as a function of time for X65 carbon steel exposed to a $\rm CO_2$ -saturated 3 wt.% NaCl solution at 80 and 150°C





(c)

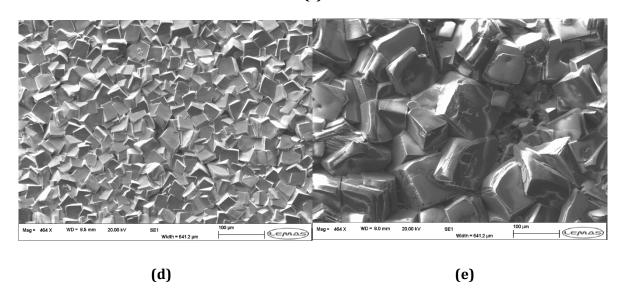
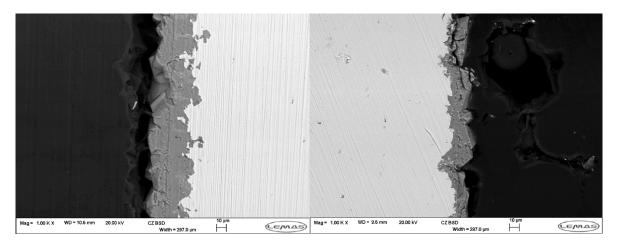


Figure 4: Top view SEM images of X65 carbon steel after exposure to a CO_2 -saturated 3 wt.% NaCl solution after (a) 6 h at 80° C; (b) 96 h at 80° C; (c) 96 h at 80° C at higher magnification; (d) 6 h at 150° C and (e) 96 h at 150° C



(a) (b)

Figure 5: Cross-section SEM images of X65 carbon steel after exposure to a CO_2 -saturated 3 wt.% NaCl solution after 96 h at (a) 80° C and (b) 150° C

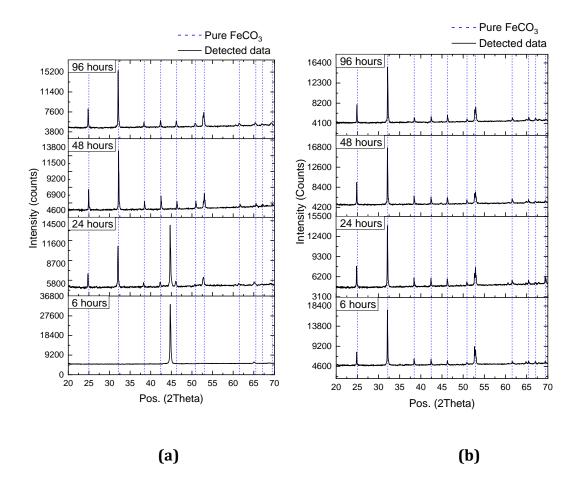
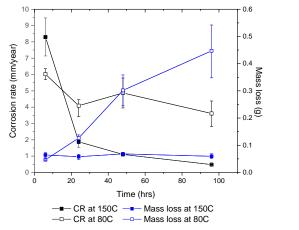
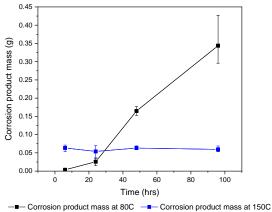


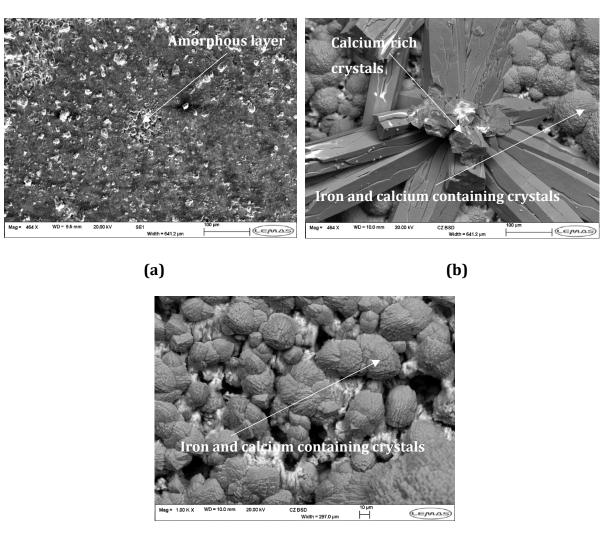
Figure 6: XRD patterns collected from X65 carbon steel surfaces exposed to a CO_2 -saturated 3 wt.% NaCl solution for different time periods between 6 h and 96 h at (a) 80° C and (b) 150° C





(a) (b)

Figure 7: (a) Mass loss measurements and the corresponding calculated corrosion rates in mm year-1 and (b) corrosion product mass determined from mass gain measurements as a function of time for X65 carbon steel exposed to a CO₂-saturated 1.54 wt.% NaCl solution with 1.83 wt.% CaCl₂-2H₂O at 80 and 150°C



(c)

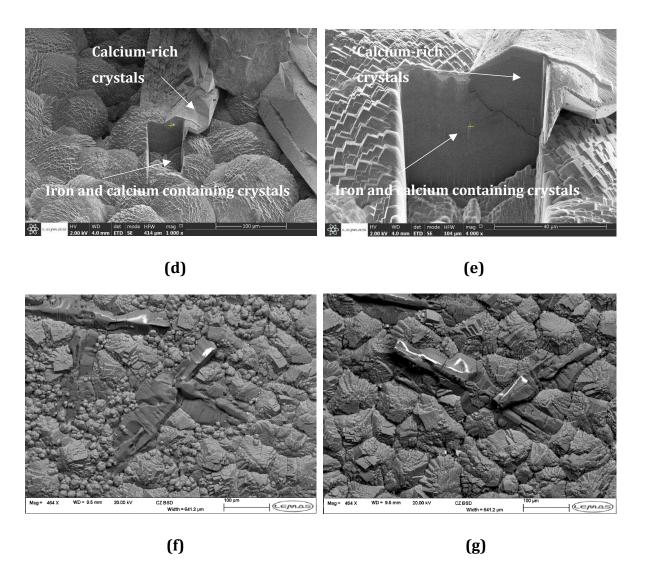
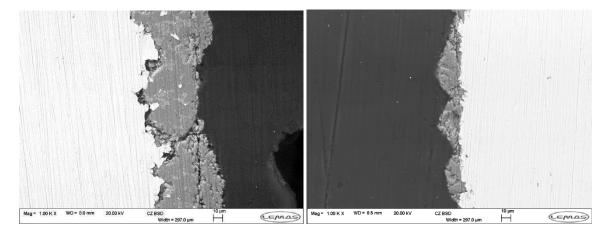


Figure 8: Top view SEM images of X65 carbon steel after exposure to a CO_2 -saturated 1.54 wt.% NaCl solution with 1.83 wt.% $CaCl_2 \cdot 2H_2O$ after (a) 6 h at $80^{\circ}C$; (b) 96 h at $80^{\circ}C$; (c) 96 h at $80^{\circ}C$ at higher magnification (d) FIB-SEM 96 h at $80^{\circ}C$; (e) FIB 96 h at $80^{\circ}C$; (f) 6 h at $150^{\circ}C$ and (g) 96 h at $150^{\circ}C$



(a) (b)

Figure 9: Cross-section SEM images of X65 carbon steel after exposure to a CO₂-saturated 1.54 wt.% NaCl solution with 1.83 wt.% CaCl₂-2H₂O after 96 h at (a) 80°C and (b) 150°C

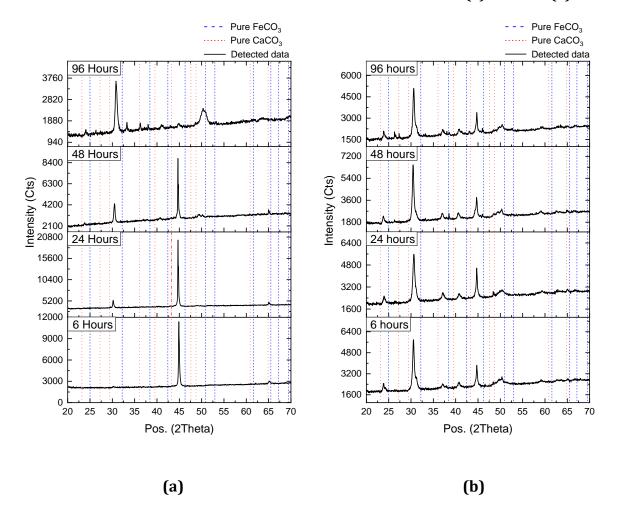


Figure 10: XRD patterns collected from X65 carbon steel surfaces exposed to a CO₂-saturated 1.54 wt.% NaCl solution with 1.83 wt.% CaCl₂·2H₂O for different time periods between 6 h and 96 h at (a) 80°C and (b) 150°C

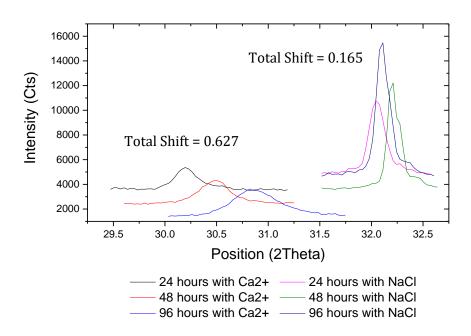


Figure 11: XRD patterns for X65 steel samples exposed to a CO_2 -saturated solution containing either 3 wt.% NaCl or 1.54 wt.% NaCl with 1.83 wt.% $CaCl_2$ -2 H_2O at 80°C after 24, 48 and 96 h of exposure

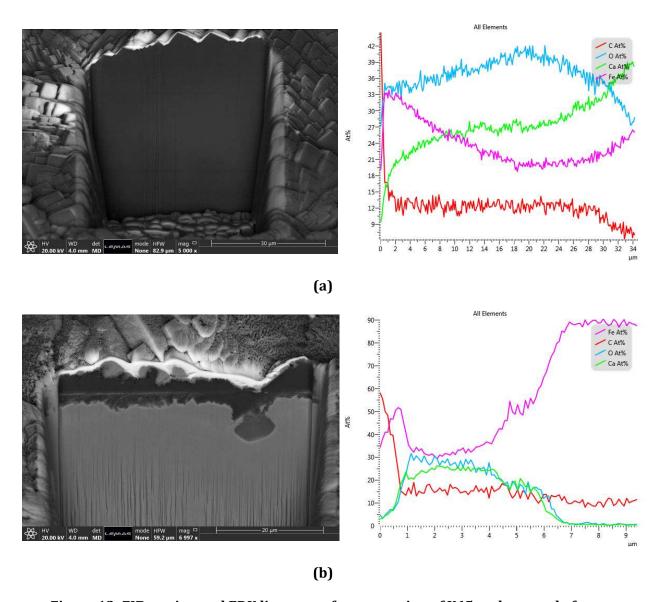


Figure 12: FIB section and EDX line scan of cross-section of X65 carbon steel after exposure to a CO_2 -saturated 1.54 wt. % NaCl solution with 1.83 wt.% $CaCl_2 \cdot 2H_2O$ after (a) 96 h at $80^{\circ}C$ and (b) 96 h at $150^{\circ}C$

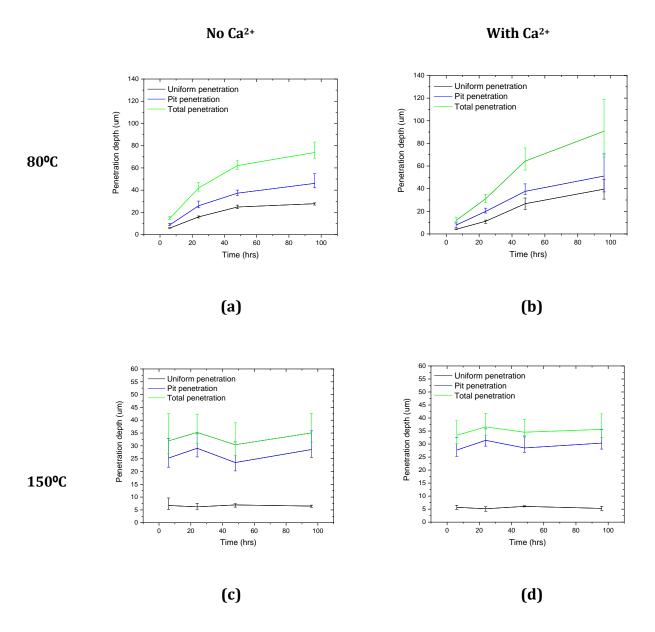


Figure 13: Uniform thickness loss (calculated from mass loss measurements), average of the top 10 pit depths relative to the corroded steel surface (from profilometry analysis) and total penetration depth (summation of the two aforementioned measurements) on X65 carbon steel as a function of time for a CO₂-saturated solution containing (a) 3 wt.% NaCl at 80°C; (b) 1.54 wt.% NaCl and 1.83 wt.% CaCl₂-2H₂O at 80°C; (c) 3 wt.% NaCl at 150°C; (d) 1.54 wt.% NaCl and 1.83 wt.% CaCl₂-2H₂O at 150°C

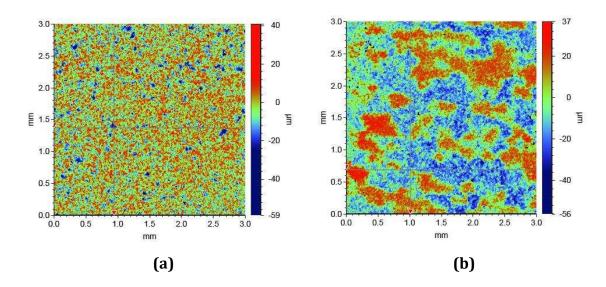


Figure 14: Surface profilometry images of carbon steel surfaces after exposure to a CO₂-saturated solution for 96 h at 80°C containing (a) 3 wt.% NaCl and (b) 1.54 wt.% NaCl and 1.83 wt.% CaCl₂·2H₂O. Images are acquired after removal of the corrosion product layer

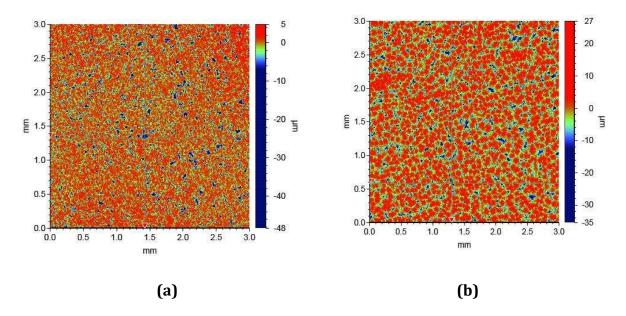


Figure 15: Surface profilometry images of carbon steel surfaces after exposure to a CO₂-saturated solution for 96 h at 150°C containing (a) 3 wt.% NaCl and (b) 1.54 wt.% NaCl and 1.83 wt.% CaCl₂·2H₂O. Images are acquired after removal of the corrosion product layer