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Pharmacokinetics of the SABRE agent 4,6-d₂-Nicotinamide and also Nicotinamide in rats following oral and intravenous administration

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Abbreviations: d₂-NA, 4,6-d₂-nicotinamide; NA, protio-nicotinamide; SABRE, Signal Amplification By Reversible Exchange; MRI, Magnetic Resonance Imaging; IV, Intravenous; IP, Intraperitoneal; PO, Per os (Oral); LLOQ, Lower Limit of Quantification; ULOQ, Upper Limit of Quantification.

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ABSTRACT

To prepare the way for using the isotopically labelled SABRE hyperpolarised 4,6-d₂-nicotinamide as an MRI agent in humans we have performed an *in-vivo* study to measure its pharmacokinetics in the plasma of healthy rats after intravenous and oral administration. Male Han Wistar rats were dosed with either 4,6-d₂-nicotinamide or the corresponding control, non-labelled nicotinamide, and plasma samples were obtained at eight time points for up to 24 hours after administration. Pharmacokinetic parameters were determined from agent concentration-versus-time data for both 4,6-d₂-nicotinamide and nicotinamide. 4,6-d₂-nicotinamide proved to be well tolerated regardless of route of administration at the concentrations used (20, 80 and 120 mg/kg). Pharmacokinetic parameters were similar after oral and intravenous administration and similar to those obtained for nicotinamide. Analysis of nicotinamide plasma concentrations after dosing 4,6-d₂-nicotinamide intravenously demonstrates a reversible exchange of endogenous nicotinamide by this labelled agent over the time-course of our assays. Supported by a large body of evidence for the safety of nicotinamide when dosed orally in humans, we conclude that 4,6-d₂-nicotinamide can also be safely administered intravenously, which will provide significant benefit when using this agent for planned imaging studies in humans.

1. Introduction

Over the last few decades Magnetic Resonance Imaging (MRI) has evolved to become an extremely important technique that allows researchers to obtain anatomical, functional and metabolic information. However, even after this tremendous success, its applicability is limited due to low sensitivity and high cost. The low sensitivity is derived from the fact that nuclei possess little intrinsic magnetisation and interact weakly with a magnetic field. Dynamic nuclear polarization (DNP) is now an established technique for dramatically increasing the sensitivity of magnetic resonance imaging and spectroscopy for the study of *in-vivo* metabolism (Kurhanewicz *et al.*, 2011; Gallagher *et al.*, 2011). Using pyruvate as a hyperpolarization agent, DNP has demonstrated that detailed clinical MRI images can be obtained in humans (Grist *et al.*, 2019; Daniels *et al.*, 2016). We have sought to

overcome the low sensitivity problem by developing similarly highly visible agents through a method of hyperpolarisation called SABRE (Adams *et al.*, 2009). SABRE's rapidity and potentially low cost mean it too has the potential to revolutionise clinical MRI, and related MR methods by improving the strength of the detected response by over four orders of magnitude from its normal level as found on a routine 1.5T clinical MRI system (Rayner *et al.*, 2017; Duckett *et al.*, 2013). SABRE reflects a novel approach that does not change the chemical identity of the agent it hyperpolarizes but instead equilibrates polarization between $p\text{-H}_2$ and the selected agent through binding to an iridium centre (Adams *et al.*, 2009) in a novel form of catalysis. With this sensitivity gain, SABRE hyperpolarization of ^1H , ^{13}C , ^{15}N , ^{31}P atoms (Iali *et al.*, 2018; Roy *et al.*, 2018; Olaru *et al.*, 2017) in labelled drugs or agents is expected to radically enhance their future detection by MRI and allow imaging of metabolic and physiologic processes of substrates *in vivo*. It may also enable *in vivo* nano-chemistry and metabolomics, and decreases the time needed to observe a metabolite concentration change (Viale, 2010; Kurhanewicz *et al.*, 2011).

We have chemically developed nicotinamide (NA) using NMR assessment as a SABRE agent for future MRI applications. NA belongs to the niacin family of compounds and is a common nutrient. It is also described as belonging to a class of compounds called histone deacetylase (HDAC) inhibitors, which have been shown to protect the central nervous system in rodent models of Parkinson's and Huntington's diseases (Didonna *et al.*, 2015). Clinical trials are currently underway to learn whether HDAC inhibitors help ALS and Huntington's patients (Wild *et al.*, 2014). Our SABRE agent, 4,6- d_2 -nicotinamide (d_2 -NA), has the same chemical structure as NA except that two protons at positions 4 and 6 on the aromatic ring have been replaced by deuterium (Rayner *et al.*, 2017). Though numerous publications have discussed the advantages and disadvantages of deuterated drugs (Liu *et al.*, 2017; Gant, 2014; Katsnelson, 2013; Sanderson *et al.*, 2009; Foster, 1985) and several companies focus on the use of deuterated drugs for various conditions, we use deuteration solely to enhance the SABRE hyperpolarization properties of NA. The deuterium positions are magnetically silent and serve to focus magnetization on the two remaining aromatic protons. The ability of SABRE to hyperpolarize protons, as in d_2 -NA, makes it applicable as a method to make agents for use in all hospital MRI scanners. One of the benefits of deuterium labelling is that the magnetic longitudinal relaxation time of d_2 -NA is considerably extended when compared to NA (Rayner *et al.*, 2017).

One factor that was important in selecting NA is that it is an intrinsically safe clinical agent that has already been used in pharmacological doses over many years with a low incidence of side effects and toxicity ([Final Report of the Safety Assessment of Niacinamide and Niacin. 2005](#)). Importantly, previous literature reports provide considerable evidence that NA reflects a safe therapy to use when given at adult doses of up to 10 g/day orally ([Knip, 2000](#)) and is commonly used in clinical trials at oral (PO) doses of ≥ 3 g/day in humans for various indications. Doses (up to 6 g/day) have been used in combination with radiotherapy ([Dragovic *et al.*, 1995](#)). The pharmacokinetics of NA are known to be dependent on dose, species, sex and route of administration. Safe doses in rats have previously been characterized: 750 mg/kg for subcutaneous injection and 1000 mg/kg for IP injection ([Ayouba *et al.*, 1999](#)). Furthermore, intravenous (IV) doses of 750 mg/kg in rat models of stroke ([Sakakibara *et al.*, 2000](#)) have been used. An *in vitro* cytotoxicity study demonstrated that it is possible to create a biocompatible SABRE bolus of d₂-NA and that deuteration of NA does not increase the toxicity compared to the NA ([Manoharan *et al.*, 2018](#)).

We are now seeking to progress d₂-NA to first-in-human exploratory MR imaging studies, where it is anticipated that this agent will be used intravenously as a bolus injection to observe perfusion in the heart or brain, for example. In order to pave the way for using d₂-NA we intend to conduct two *in-vivo* preclinical studies: a PK study and a single dose toxicity study in rats. At sub-therapeutic doses, guided by a body of clinical data for NA in humans and current regulatory guidelines, only rodent data may be required to support the use of d₂-NA in humans. We report here the results of the first of these whereby the pharmacokinetics of d₂-NA in the plasma of healthy rats after PO and IV administration was measured. Data describing IV administration of NA in human is limited and there is no corresponding information known for d₂-NA. The pharmacokinetics study of d₂-NA is required to predict its efficacy and the duration of its effects, especially in comparison to NA, and to provide supportive data on the mechanisms of drug action, to predict side-effects and for the extrapolation of results from laboratory animals to man. In this study, we used sub-therapeutic doses between 20 and 120 mg/kg in rats. The main objective is to measure plasma concentrations of d₂-NA following IV and PO administration and to compare these to the pharmacokinetics of NA.

2. Materials and Methods

This is a non-regulatory study for which a claim of GLP compliance is not been made. However, the laboratory procedures used were consistent with International Standards of GLP.

2.1 Animals. Fifteen male Han Wistar rats were obtained from Charles River Laboratories, Margate, UK. Animals were all in the target weight range of 163 to 215 g at dosing. Diet was removed at the end of the working day prior to each dose occasion, and returned 4 hours after dosing.

2.2 Agents used. Both d₂-NA and NA were formulated by dissolving appropriate amounts of each in 0.9% physiological saline and were assessed visually for complete dissolution. No cloudy suspension or particles were visible (Table 1).

2.3 Experimental design. Each animal received a single administration (Table 2). PO doses were administered by gavage at a nominal dose volume of 5 mL/kg. IV doses were administered *via* a lateral tail vein, at a nominal dose volume of 2 mL/kg, as a bolus injection. Body weights were recorded the day after arrival and before dose administration and used for calculation of doses. Samples of blood (nominally 150 µL) were collected from the jugular vein of each animal. For PO dosing this was done at 0.25, 0.5, 1, 2, 4, 6, 8 and 24 hours post dose. For IV dosing this was at 0.016, 0.083, 0.5, 1, 2, 4, 8 and 24 hours post dose. The actual blood collection time was used in the subsequent PK data analysis. Blood was collected into tubes containing the K₂EDTA anticoagulant and stored at room temperature. Blood was centrifuged (1500 g, 10 minutes, 4°C) to produce plasma for analysis and residual blood cells were discarded.

2.4 Blood plasma assay for d₂-NA and NA. Initial NA concentration quantification revealed an endogenous NA background peak during calibration, the intensity of which is consistent with a level of approximately 50 ng/mL. Concentrations of d₂-NA and NA were subsequently determined by the now qualified LC-MS/MS method of Table 3 which involved the following steps:

1. Transfer of a 10 µL aliquot into a clean 2 mL 9-well collection plate
2. Addition of 25 µL I-B (500 ng/mL). Addition of 25 µL methanol to all blanks
3. Vortex mix plate (*ca.* 1 minute)

4. Addition of 80 μ L methanol to all wells
5. Vortex mix plate (*ca.* 2 minutes)
6. Centrifuge 2000 g plate for 10 minutes at room temperature
7. Transfer 40 μ L of supernatant to a clean 1.2 mL 96-well collection plate
8. Dilute with 160 μ L 10 mM of ammonium acetate:ammonium hydroxide (100:0.5 v/v) solution
9. Vortex mix plate (*ca.* 2 minutes)
10. Complete analysis by LC/MS-MS

2.5 Pharmacokinetic analysis. Pharmacokinetic analysis of concentrations of d₂-NA and NA in blood and plasma was performed using the validated non-compartmental pharmacokinetic analysis program Phoenix WinNonLin, version 6.4 (Certara, St Louis, Missouri, USA).

3. Results

No clinical signs were observed to be associated with the IV or PO administration of d₂-NA or NA at doses up to 120 mg/kg (Table 2) to fifteen animals. A complete list of the associated pharmacokinetic parameters determined for each of these dose groups are presented in Table 4. For all dose groups, the mean volume of distribution was high compared to the blood volume in the rat (*ca* 54 mL/kg (Davies *et al.*, 1993)), suggesting that test compound was distributed into body water or tissues. The mean plasma clearance of all groups was lower than liver blood flow in the rat (*ca* 3000 mL/h/kg (Davies *et al.*, 1993)) suggesting passive clearance from the systemic circulation and some distribution into body tissues and/or saturation of clearance mechanisms. The following results were calculated for each group.

3.1 Control Group A (120 mg/kg NA, IV):

Although used as a control group in this study, not much is known about the pharmacokinetics of NA after IV dosing. Almost all preclinical and clinical *in-vivo* data have resulted from PO dosing regimes. We report here that a mean plasma concentration of NA of 168000 \pm 9170 ng/mL is observed 0.016 hours post-dose,

which suggests a mean plasma C_0 of 178000 ± 16700 ng/mL. NA was still detected in the plasma of all three animals 8 hours post-dose (32500 ± 1100 ng/mL) but was below the LLOQ of 50 ng/mL by 24 hours post-dose. These data are presented in Table 4 and Figure 1. In comparison with group B, exposure to NA was 1.7 times higher than exposure to d_2 -NA after dosing of d_2 -NA at the same dose level of 120 mg/kg as judged by AUC data, even though C_{max} data are consistent (Table 4.). This difference is driven by a significantly longer half-life observed for NA (4.0 hours for Group A versus 2.3 hours for Group B). Our interpretation of this data is that d_2 -NA may be metabolised and cleared more quickly from plasma than NA.

3.2 Group B (120 mg/kg d_2 -NA, IV):

After an IV administration of d_2 -NA at a nominal dose level of 120 mg/kg, a mean plasma concentration of d_2 -NA of 166000 ± 12700 ng/mL was observed 0.016 hours post-dose, which is approximately 2-times higher than for group D where the dose is (60 mg/ml) and 6-times higher than for group E (20 mg/ml). This indicates dose response proportionality. A mean plasma C_0 level of 172000 ± 16700 ng/mL was detected which compares well with the C_0 of 178000 ± 16700 ng/mL of control group A (120 mg/kg NA, IV). Increased d_2 -NA levels were still detected in all three animals at 8 hours post-dose but the value falls to below 50 ng/mL in all three animals by 24 hours post-dose.

Interestingly, although only d_2 -NA was dosed in this group, low levels of NA were detected in two out of the three animals at the first sampling point (mean 73 ± 63.2 ng/mL), increasing to a mean C_{max} of 4400 ± 229 ng/mL (*ca* 2.7% of d_2 -NA) at 8 hours post-dose (Figure 3). An estimate of the $t_{1/2}$ and corresponding $AUC_{(0-\infty)}$ for NA could not be determined from a simple analysis of the NA pharmacokinetic profile.

3.3 Group C (120 mg/kg, d_2 -NA PO):

Following PO administration of d_2 -NA at a nominal dose level of 120 mg/kg, the plasma of all animals demonstrated exposure to the agent. A mean maximum (C_{max}) plasma concentration of 96000 ± 31500 ng/mL was reached within 0.5 to 4 hours (median of 0.5 hours). By 24 hours post dose, the mean plasma concentrations of d_2 -NA had declined to a baseline value of 21.7 ± 37.6 ng/mL, 0.02% of C_{max} . Hence the mean terminal elimination half-life is 2.00 hours suggesting rapid elimination, which is comparable to that found for group B, after an IV dose at the same initial level. The

mean $AUC_{(0-t)}$ was 20100 ± 1190 h.ng/mL, is also comparable to the systemic exposure observed after a 120 mg/kg IV dose (group B).

3.4 Group D (60 mg/kg d₂-NA, IV):

After an IV administration of d₂-NA at a nominal dose level of 60 mg/kg, a mean plasma concentration of d₂-NA of 79400 ± 4540 ng/mL was observed at (C_t), 0.016 hours post-dose. d₂-NA was still detected in all three animals at 8 hours post-dose but had been eliminated in two out of three by 24 hours post-dose. The terminal half-life ($t_{1/2}$) was 2.23 ± 0.78 hours. As for group B, NA was detected in all three animals at the first sampling point, albeit at very low concentrations: mean 225 ± 40.9 ng/mL. However, the concentration of NA increased to a mean of 3830 ± 97.1 ng/mL (ca 4.8% of d₂-NA) at 4 hours post-dose.

3.5 Group E (20 mg/kg d₂-NA, IV):

Following IV administration of d₂-NA at a nominal dose level of 20 mg/kg, a mean plasma concentration of d₂-NA of 27700 ng/mL, was observed at the first sampling point (C_t), 0.016 hours (1 minute) post-dose. When extrapolated back to time zero (C_0), this corresponds to a mean plasma concentration of 28700 ± 657 ng/mL. d₂-NA again proved to be eliminated in all animals by 24 hours post-dose. The mean areas under the plasma concentration time curves, $AUC_{(0-t)}$ and $AUC_{(0-\infty)}$ were 34000 ± 5370 h.ng/mL and 34800 ± 6460 h.ng/mL, respectively, with a corresponding terminal half-life ($t_{1/2}$) of 0.828 ± 0.175 hours. As now expected, NA was detected in all three animals at the first sampling point and a mean maximum concentration of 2310 ± 180 ng/mL (ca 8% of parent compound) was achieved at ca 2 hours post-dose. These data did not lend themselves to $t_{1/2}$ or $AUC_{(0-\infty)}$ determination.

4. Discussion

The dose normalised $AUC_{(0-\infty)}$ data suggests that at 60 and 120 mg/kg levels the exposure to d₂-NA is proportional to the dose administered (Figure 2). The sub-proportional exposure, in terms of $AUC_{(0-\infty)}$, at the lowest dose level, 20 mg/kg, is possibly due to concentrations being smaller relative to the animals NA pool, but we note they are also close to the limits of quantification, and hence there were fewer detectable data points; the corresponding pharmacokinetic parameters are therefore less well defined. The measured exposure level after a PO dose of d₂-NA was however directly comparable to that of IV exposure at 120 mg/kg and this is reflected in the

measured bioavailability of 98.7%. Hence, the IV and PO routes should result in similar agent response behaviour. In fact, C_{\max} concentrations of NA after IV administration were comparable to those of d₂-NA after d₂-NA administration.

NA occurs as a component of a variety of biological systems and as a part of the coenzyme nicotinamide adenine dinucleotide (NADH / NAD⁺): it is crucial to life (Belenky *et al.*, 2007). Once ingested, nicotinamide undergoes a series of reactions that transform it into NAD, which can then undergo a transformation to form NADP⁺. However, rats and humans can produce NAD⁺ from the amino acid tryptophan and niacin without ingestion of nicotinamide (Williams *et al.*, 2015). As part of our analysis, we estimated the endogenous level of NA to be approximately 50 ng/mL in rat plasma and set the Lower Limit of Quantification (LLOQ) for our analysis at this level. The ULOQ was set at 250,000 ng/mL.

This study aims to determine the pharmacokinetics of d₂-NA after dosing in-vivo in comparison to a control group of NA. However, one of the most interesting findings is that dosing of d₂-NA elicits a release or production of NA *in-vivo*, isolated in blood plasma of rats from groups B, D and E (Figure 3). A significant level of NA (C_{\max} 2310 ± 180 ng/mL) was observed 2 hours after dosing d₂-NA at 20mg/kg (group E). This is approximately 8% of the level of d₂-NA dosed. This NA concentration reduced to zero after 8 hours and did not fit a normal pharmacokinetic profile. For the 60mg/kg d₂-NA group (D) the level of NA observed rose to 3830 ± 97.1 ng/mL (4.8% of d₂-NA) 4 hours post-dose while for a 120 mg/kg dose of d₂-NA (group B) a mean C_{\max} of 4400 ± 229 ng/mL (2.7% of d₂-NA) was found at 8 hours post-dose (Figure 3).

Our interpretation of these data is that at all dose levels used in this study d₂-NA exchanges and partially replaces the “pool” of endogenous NA in body tissues, resulting in a release of NA into the plasma during the time course of our assay. The data presented in Figure 3 shows how the concentrations of NA, which are significantly in excess of the estimated endogenous NA levels and LLOQ of 50 ng/mL, vary with time after dosing with d₂-NA. Endogenous NA, which is replaced by d₂-NA and released to the plasma, is linked to the response level in terms of C_{\max} to the levels of d₂-NA injected. In comparison to the concentration-versus-time plasma levels of d₂-

NA (Figure 2) there is a delay of 2-8 hours to reach C_{\max} for the released endogenous NA (Figure 3) which suggest that exchange with the endogenous pool is slow. The shapes of these NA concentration-versus-time profiles are consistent with this slow replacement of NA in the endogenous pool by d_2 -NA followed by simultaneous clearance of the excess released endogenous NA. Importantly, the plasma levels of NA after dosing d_2 -NA return to baseline by 24-hrs, indicating the clearance of all dosed d_2 -NA and the released endogenous NA.

The presence of NA when d_2 -NA is dosed in rats is not a result of contamination of the dose samples or loss of the deuterium label from d_2 -NA. In a bioanalytical study we have demonstrated that the deuterium label of d_2 -NA is stable in rat blood plasma for significantly longer than 24 hours (data not shown).

5. Conclusions

Importantly, this study has demonstrated that d_2 -NA is well tolerated regardless of route of administration at the concentrations used (20, 80 and 120 mg/kg bodyweight) and that pharmacokinetic parameters of d_2 -NA are similar after PO and IV administration and similar to those obtained for NA after IV dosing. Taken together the pharmacokinetic data suggest that there are no significant differences between dosing either NA or d_2 -NA. This is important as it helps us to bridge the large body of known clinical data for the use of NA via PO dosing with our intended IV administration of hyperpolarized d_2 -NA as an MRI imaging agent.

Concerning the significant presence of NA after dosing of d_2 -NA our conclusion is that an endogenous pool of NA is partially replaced by d_2 -NA in a dose dependent fashion (Figure 3). We have also shown that this process is reversible on the timescale of our study after a single dose at time zero and hypothesize that newly synthesized endogenous NA refills the endogenous cellular or body tissue pool. Both d_2 -NA and the released endogenous NA are completely cleared within 24 hours after IV dosing and a normal equilibrium is restored to the pool of endogenous NA.

This pharmacokinetic study is the first of two *in-vivo* rat studies we will be conducting. It will be followed by a single-dose IV bolus study of d_2 -NA in rats followed by a 14-day recovery period, necropsy and histopathology, to determine the safety

toxicity profile of this d₂-NA after hyperpolarization and extraction from the SABRE catalyst reaction mix. We will use a biphasic separation procedure to do this (modified from [Manoharan *et al.*, 2019](#)). This toxicity study will be a pivotal preclinical study to determine if this d₂-NA agent can be progressed clinically in humans as an MRI imaging agent of well perfused organs or tissues such as the heart, brain or kidney.

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Table 1: Agents used for this PK study

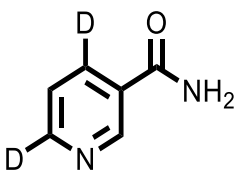
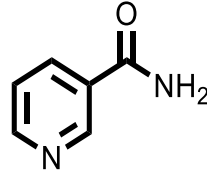
Test Substance Name:	4,6-d ₂ -nicotinamide	nicotinamide
Acronym	d ₂ -NA	NA
Chemical Form:	Free compound	Free compound
Physical Form:	Solid	Solid
Chemical Structure:		
Molecular Formula:	C ₆ H ₄ D ₂ N ₂ O	C ₆ H ₆ N ₂ O
Molecular Weight:	124.1 g/mol	122.1 g/mol
Chemical Purity:	99%	99%

Table 2: Dose groups for d₂-NA and NA

Dose Group	Agent used	Dose route	Dose level	Number of animals
			mg/kg	Males
A	NA	IV	120	3
B	d ₂ -NA	IV	120	3
C	d ₂ -NA	PO	120	3
D	d ₂ -NA	IV	60	3
E	d ₂ -NA	IV	20	3

Table 3: Mass Spectrometer setup

Mass Spectrometer	Sciex 6500
Ionisation Interface and Temperature	TISP 650°C
LC System	Shimadzu Nexera
Mobile Phase A	10 mM ammonium acetate: ammonium hydroxide (100:0.5)
Mobile Phase B	methanol
Weak Wash	methanol: water (50:50)
Strong Wash	2-propanol: methanol: THF: water (25:25:25:25 v/v/v/v)
Auxiliary Wash	methanol: water (50:50)
Injection Volume	20 µL
Analytical Column	Phenomenex Gemini 3 µm NX-C18, 50 x 2 mm
Column Temperature	40°C
Flow Rate	0.6 mL/min
Sample Temperature	5°C

Gradient
Profile

Time (min)	% Aqueous Phase
0	95
0.2	95
2	2
2.5	2
2.6	95
3.2	95

Table 4: Pharmacokinetic parameters

Dose Group	C_{max}	t_{1/2}	AUC_(0-t)	AUC_(0-∞)	C₀	V_z	Cl
	ng/mL	hours	h.ng/mL	h.ng/mL	ng/mL	mL/kg	mL/h/kg
A	168000	4.06	573000	774000	178000	911	163
B	166000	2.33	416000	459000	172000	877	262
C	96000	2.00	418000	386000	-	-	-
D	79400	2.23	196000	197000	82700	963	307
E	27700	0.83	34000	34800	28700	691	586

C_{max} maximum plasma concentration observed

t_{1/2} the terminal elimination phase half-life was determined by linear regression of at least three data points (not including C_{max}) the terminal portion of the log-linear concentration vs time curve.

AUC_(0-t) area under the concentration-time curve calculated from 0-t, where t is the time of the last measurable concentration, was calculated by non-compartmental analysis using the log/linear trapezoidal rule.

AUC_(0-∞) area under the concentration- time curve extrapolated to infinite time was calculated by non-compartmental analysis using the log/linear trapezoidal rule.

C₀ plasma concentration at time zero following the IV dose was obtained by back extrapolation of the first two plasma concentrations.

V_z volume of distribution based on the terminal phase following IV dosing, calculated using the equation:

$$V_z = \frac{Dose}{\lambda_z \times AUC_{(0-\infty)}}$$

Cl total body clearance following IV dosing was calculated using the equation:

$$Cl = \frac{Dose}{AUC_{(0-\infty)}}$$

Figure 1: Mean Concentrations of NA in the plasma of male rats (n=3) following a single IV dose at a nominal dose level of 120 mg/kg (group A)

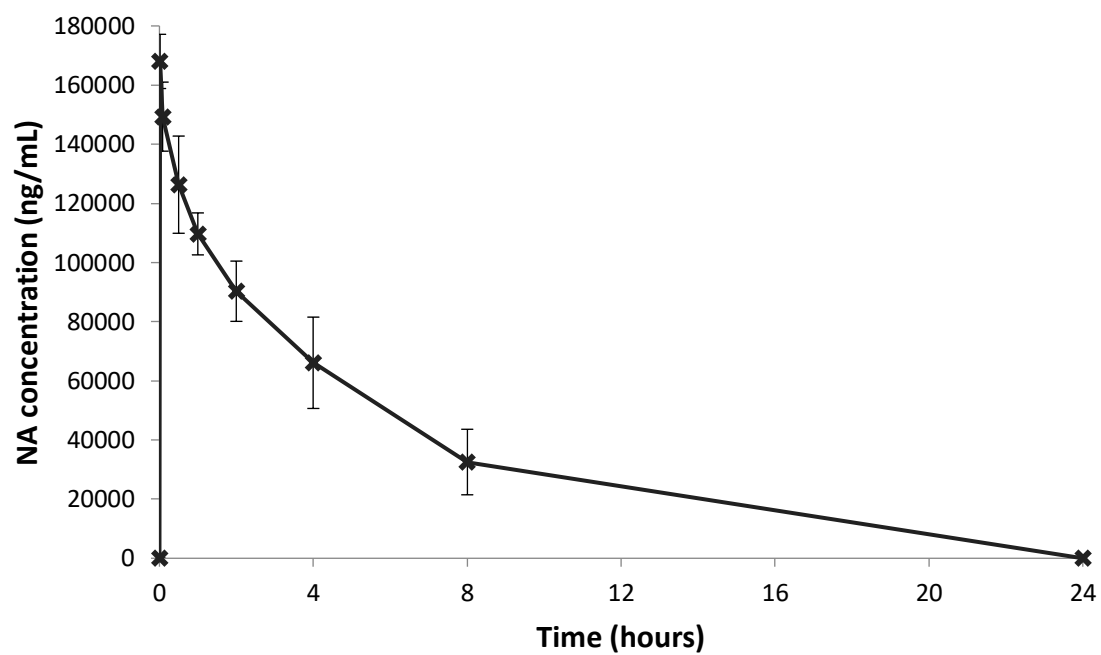


Figure 2: Mean concentrations of d₂-NA in the plasma of male rats (n=3) following a single IV administration of d₂-NA at nominal dose levels of 20 mg/kg (group E ◆), 60 mg/kg (group D ■) and 120 mg/kg (group B ▲)

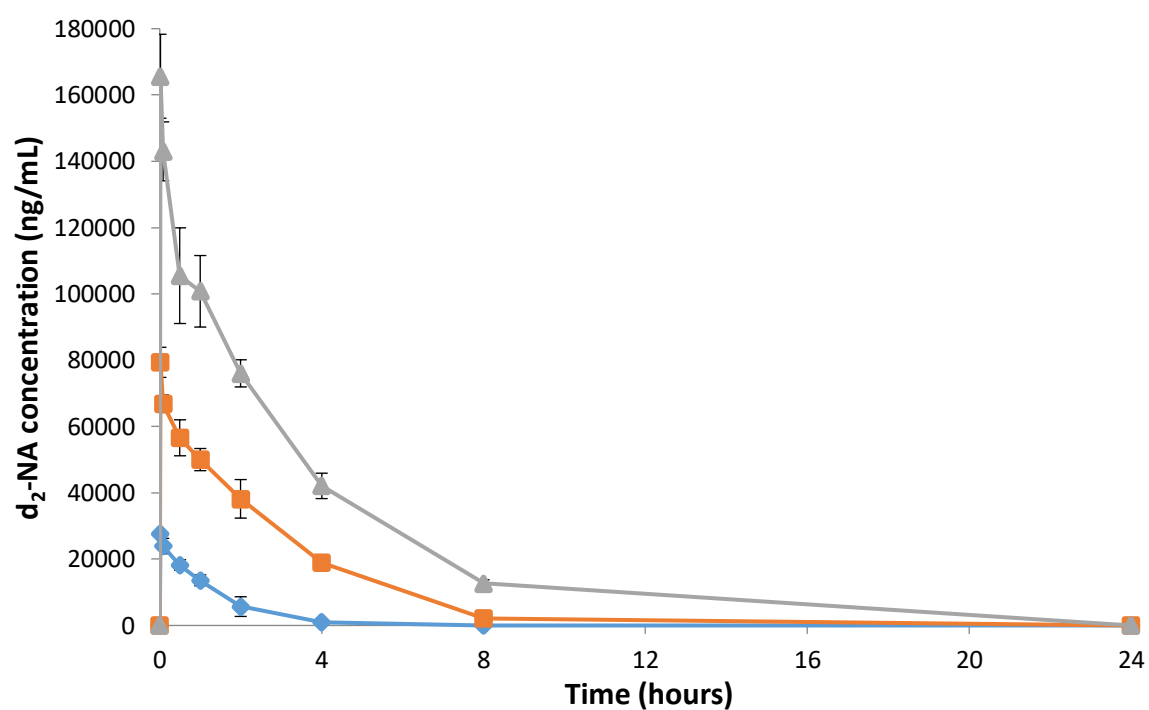


Figure 3: Mean concentrations of NA in the plasma of male rats (n=3) following a single IV administration of d₂-NA at nominal dose levels of 20 mg/kg (group E ◆), 60 mg/kg (group D ■) and 120 mg/kg (group B ▲). Endogenous NA is released after intravenous administration of d₂-NA.

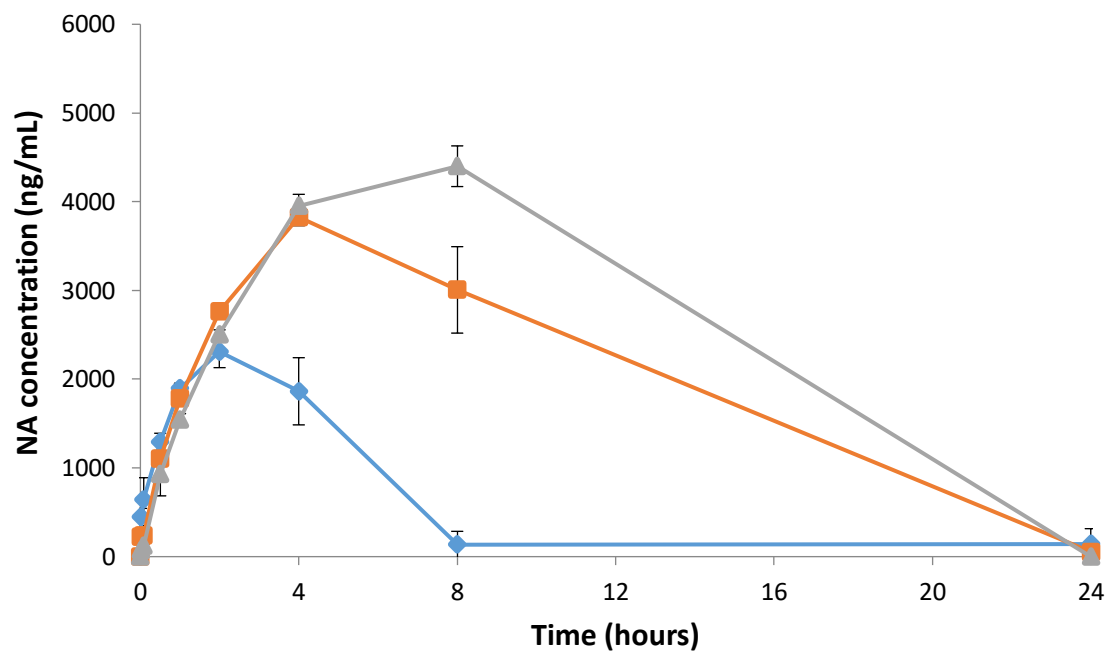


Figure 4: Mean concentrations of d₂-NA in the plasma of male rats (n=3) following a single PO administration at a nominal dose level of 120 mg/kg (group C)

