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**Plant responses to simulated Carbon  
Capture and Storage (CCS) CO<sub>2</sub> pipeline leakage: the effect of soil type.**

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**Running Title**

**soil type and extreme CO<sub>2</sub>**

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## Abstract

Carbon capture and storage (CCS) has been proposed as a bridging technology to enable the transition to an energy system based on renewable sources. Many high CO<sub>2</sub> emitting industries (e.g. power stations) are distant from potential carbon storage sites (such as offshore geological reservoirs) and therefore an infrastructure of CO<sub>2</sub> transportation must be developed to carry the CO<sub>2</sub> to safe storage. As such there is a need to understand the risks involved and the mitigation of potential leaks associated with CCS and dense-phase CO<sub>2</sub> transportation networks. Since 2012 a number of experimental studies have provided a mechanistic understanding of the risks posed to crops as a function of CO<sub>2</sub> leakage from CCS infrastructure. However, what remains largely unresolved is the role played by both soil type and soil structure in mitigating and/or enhancing plant stresses. In this study we provide an experimental framework to evaluate these effects. Wheat and beetroot were grown in four different experimental soils to test the effects of specific soil attributes (organic, low pH; organic, open structure; organic, limed; loam, neutral pH) on crop performance when exposed to high levels (~40%) of CO<sub>2</sub> in the soil environment. Comparison between treatment and controls and across the soil types reveals little difference in terms of biomass or plant stress chemistry. From a stakeholder perspective these findings suggest that soil type may play only a minor role in mitigating or amplifying plant stress in response to the unlikely event of a CO<sub>2</sub> leak from CCS infrastructure.



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**Keywords** Carbon capture and storage, CCS, Crops, Extreme CO<sub>2</sub>, Gas exchange, Roots,

48 Soils

# 1. Introduction

Anthropogenic climate change is driven by the acceleration of the long-term carbon cycle via  
51 the combustion of fossil fuel directly transferring carbon from the lithosphere to the  
atmosphere. Since the dawn of the industrial age this has resulted in atmospheric CO<sub>2</sub>  
increasing from ~280 ppm in the 1850s to 406 ppm in 2017  
54 (<https://www.esrl.noaa.gov/gmd/ccgg/trends/monthly.html>). This rise in CO<sub>2</sub> has seen a  
concomitant increase in global average temperature ([http://climate.nasa.gov/vital-](http://climate.nasa.gov/vital-signs/global-temperature/)  
signs/global-temperature/). The Paris Agreement in 2015 COP21 climate treatise (ratified  
57 November 2016) was designed to limit warming to "... 2°C above pre-industrial levels and  
pursuing efforts to limit the temperature increase to 1.5°C above pre-industrial levels,  
recognising that this would significantly reduce the risks and impacts of climate change".  
60 These ambitious goals require the development of multiple mitigation practices and  
eventually the removal of fossil fuel derived carbon from the energy system.

63 One potential mechanism identified as having a role in delivering these ambitious targets and  
which has been recognised as a bridging technology for transition from a fossil fuel carbon  
based energy system to a renewable energy infrastructure is the use of carbon capture and  
66 storage (CCS). This mitigation technique essentially allows for energy to be extracted as the  
carbon is moved from one geological reservoir to another. Many high CO<sub>2</sub> emitting industries  
(e.g. power stations) in the UK are distant from potential carbon storage sites (such as  
69 offshore geological reservoirs) and therefore an infrastructure of CO<sub>2</sub> transportation must be  
developed to carry the CO<sub>2</sub> to safe storage. As such there is a need to understand the risks  
involved and the mitigation of potential leaks associated with CCS and dense-phase CO<sub>2</sub>  
72 transportation networks into the environment. Whilst risks assessment studies have been  
undertaken, many have focused on marine benthic studies related to off-shore storage

reservoirs<sup>1-4</sup>. Of those undertaken in terrestrial environments, several have utilised natural  
75 CO<sub>2</sub> vent sites, which are not comparable to a sudden or recent release of CO<sub>2</sub>, as both the  
soil and biological components within have evolved over many years<sup>5, 6</sup>. Specific  
experimental systems include outdoor CO<sub>2</sub> gradient studies, which whilst giving a more  
78 realistic scenario with comparable CO<sub>2</sub> and O<sub>2</sub> levels of leakage in soils, do not fully replicate  
particular scenarios such as soil type and focus largely on leakage detection methods rather  
than direct effects on soil or bio-components<sup>7-10</sup>. Studies with the aim of measuring the  
81 effects of CO<sub>2</sub> leakage on soils have been undertaken, but have not specifically looked at  
different soil types under the same conditions<sup>11</sup>, an exception is that of<sup>12</sup> who did investigate  
two soil types and the effect of CO<sub>2</sub> on microbial communities in a long-term mesocosm  
84 study. The nearest equivalent study system is that of<sup>13, 14</sup> who did specifically measure  
vegetation responses, did not investigate different soil types.

87 Recent experimental work has highlighted that the effects of CO<sub>2</sub> leakage on agricultural land  
are highly localised<sup>15, 16</sup> as reviewed in<sup>17</sup> (e.g. these effects are also transient with recovery of  
vegetation close to complete after 12 months<sup>17</sup> and that this stress is induced by direct CO<sub>2</sub>  
90 exposure rather than as a function of O<sub>2</sub> depletion<sup>18</sup>. Further, using the system reported here  
we have recently demonstrated that the effects of impurities (specifically SO<sub>2</sub> and H<sub>2</sub>S)  
within the CO<sub>2</sub> gas stream are limited. Within our experimental system there are no additive  
93 toxicity effects when comparing plants gassed with a combination of CO<sub>2</sub> and SO<sub>2</sub> or CO<sub>2</sub>  
and H<sub>2</sub>S to control plants exposed just to CO<sub>2</sub><sup>19</sup>

96 However, what remains largely unresolved is the role played by both soil type and structure  
in mitigating and/or enhancing reported plant stresses. Closing this knowledge gap is an  
important step in the development and deployment of CCS transportation infrastructure as

99 any potential hazard requires full elucidation. This will aid the decision making process in  
where and how CCs technologies are deployed<sup>20</sup>. Typically sites suitable for the geological  
storage of CO<sub>2</sub> are distal to CO<sub>2</sub> emitters, consequently CO<sub>2</sub> pipelines will cross numerous  
102 soil types. To address this knowledge gap we build on our experimental protocols<sup>17-19</sup> to test  
for differences in plant stress/health as a function of soil type when exposed to high soil CO<sub>2</sub>  
concentrations that simulate CO<sub>2</sub> leakage analogous to the field based experiments conducted  
105 at the ASGARD (Artificial Soil Gassing And Response Detection) facility<sup>17, 21</sup>.

## 2. Materials and methods

### 108 2.1 Experimental setup

Soil chambers were constructed of acrylic plastic with pipe inlets to allow CO<sub>2</sub> gassing of the  
soil environment exclusively. The experimental system was housed in a controlled  
111 environment growth facility (UNIGRO, UK) to standardise all other environmental variables:  
irradiance was 300 μmol m<sup>-2</sup> s<sup>-1</sup> (at plant height), day/night as 12/12 hours; temperature  
21/18°C day/night; relative humidity 60%. Gas was supplied from either an integral supply  
114 (pure CO<sub>2</sub>) or a gas cylinder (N<sub>2</sub>) and separated prior to entering each individual soil chamber  
by 2 flow rate step-down manifolds. Gas was delivered to each individual chamber at a rate  
of 30 (±15) mL min<sup>-1</sup> to maintain CO<sub>2</sub> and N<sub>2</sub> levels at steady state. Gases were exhausted to  
117 atmosphere via a separate manifold to prevent build up within the growth room.

### 2.2 Soils types

120 To simulate a wide variety of soil types (yet maintain standardised growth conditions) a  
series of commercially available potting media were chosen and/or manipulated to deliver a  
number of experimental soils. Soil experimental treatments are as follows: (I) Levington's  
123 no.3 (L3) compost to represent an organic soil with a low pH; (II) L3 plus sand (25% by

volume), was designed to simulate an organic rich soil with an open structure; (III) L3 plus lime: organic soil with lime (lime was added to raise the pH by 1 unit) was chosen to see if  
126 the addition of lime acted as a potential buffer of CO<sub>2</sub> induced acidity and finally (IV) John Innes no.3 (JI 3) compost was chosen to simulate a standardised loamy soil with a neutral pH. We stress that these soils are used as an experimental system. They are not meant to represent  
129 actual soil types, but are used as standardised media to determine the specific effects of CO<sub>2</sub> exposure across a range of plausible soil types/ structures and to measure explore these responses in standardised a consistent experimental setting.

132

To fully elucidate the effect of soil type each experiment consisted of an experimental treatment and three levels of control: (I) CO<sub>2</sub>-gassed soil (the experiment); (II) N<sub>2</sub>-gassed soil  
135 (O<sub>2</sub>-depleted control); and (III) non-gassed soil (natural state control). In all experiments gas concentrations (CO<sub>2</sub> and O<sub>2</sub>) were measured daily using the GEOTECH GA5000 gas analyser (Geotech, Warwickshire, UK). Each experimental run had the following replication,  
138 six chambers were exposed to CO<sub>2</sub>, a further six chambers were exposed to N<sub>2</sub> and four chambers were used as non-gassed chambers (Fig. 1).

### 141 2.3 Soil pH

Soil samples were taken prior to and at the end of each experiment and dried at 40 ± 4°C. A solution of 0.01M calcium chloride dihydrate (CaCl<sub>2</sub>.2H<sub>2</sub>O analytical grade) was dissolved in  
144 de-ionised water and added to a soil sample to give a final solid to solution ratio of 1:2.5. The mixture was placed on a magnetic stirrer and stirred for at least 5 minutes. The suspension was allowed to settle for 15 minutes and measured with a pH electrode (Hanna combination  
147 electrode and Jenway PHM6 meter, Fisher Scientific, UK) until readings were stable.

## 2.4 Crop species

150 In all experiments the crop plants used were spring wheat (*Triticum aestivum* v Tybault - a  
monocotyledon, grass) and beetroot (*Beta vulgaris* v Pablo F1 - a dicotyledon, vegetable).  
The crops were sown and grown within an environmental controlled growth room (details  
153 above) for 1 to 2 weeks before being transplanted into the soil chambers. They were then left  
to allow sufficient root growth before gassing commenced (approximately 2 weeks later) with  
the gassing period lasting for up to 7 days. After that time, plants become pot-bound and  
156 performance becomes compromised via physiological changes, making direct comparison  
with field data (not pot-bound) problematic. Samples for biochemical analyses were  
immediately quenched in liquid nitrogen and stored at -80°C. Biomass (all above ground  
159 parts; leaves and stems) were measured as fresh weight (g).

## 2.5 Biochemical analyses

162 During harvest the plants were sub-sampled for analysis of the key biochemical compounds  
that are either necessary for functional integrity or associated with symptoms of stress.  
Chlorophyll content was measured following observational discolouration of leaves in field  
165 trials<sup>17, 18</sup>. Chlorophyll is a necessary compound for the ability of plants to photosynthesise  
efficiently and subsequently grow to produce a crop yield. A decrease in this compound  
would suggest that resources are diverted to produce compounds which enable a plant to  
168 mitigate stress. Build-up of anthocyanin is indicative of many stresses and is identified via a  
red discolouration of leaves and/or stems. In field studies it was observed<sup>17, 18</sup> that some  
leaves had turned red; consequently changes in this compound were investigated in this  
171 laboratory study. Phenylalanine lyase (PAL) is a compound which mediates the production of  
many stress compounds and is a generic indication that plants are suffering from  
environmental stress.



174 2.6 Chlorophyll analysis

Approximately 300 mg of fresh leaf material was ground in a pestle and mortar in 5 mL 80% acetone (volume to volume (v/v) with distilled water) solution and transferred to a 10 mL universal tube. The tube was covered with aluminum foil, stirred for 30 minutes, and then centrifuged for 15 minutes (at a speed of 3,000 rpm). The supernatant was transferred to a new tube, mixed thoroughly and pipetted into duplicate 1 cm path length cuvettes.

177

180 Absorbance of chlorophyll content was measured using spectrophotometry (Cecil 1100, manufactured by Camlab Ltd, Cambridge, UK) against 80% acetone as a blank.

183 Chlorophyll concentrations were calculated as follows:

$$\text{Ca (mg/g)} = [12.7 \times A_{663} - 2.69 \times A_{645}] \times V / 1000 \times W \text{ (Chlorophyll a)} \quad (1)$$

186

$$\text{Cb (mg/g)} = [22.9 \times A_{645} - 4.86 \times A_{663}] \times V / 1000 \times W \text{ (Chlorophyll b)} \quad (2)$$

$$\text{Ca+b (mg/g)} = [8.02 \times A_{663} + 20.20 \times A_{645}] \times V / 1000 \times W \text{ (Chlorophyll a+b)} \quad (3)$$

189 Where A = absorbance wavelength, V = volume of the extract (mL), W = Weight of fresh leaves (g). Content is expressed as mg g<sup>-1</sup> fresh weight<sup>22</sup>.

192

## 2.7 Anthocyanin content

195 Pre flash-frozen plant material was ground in a pestle and mortar in 5mL of 1% HCl in methanol (%v/v) solution to yield 4 x 1 mL samples for duplicate samples at pH 1.0 and pH 4.5. Assays were performed using 0.5mL of sample added to 2.5mL of each of the following

198 buffers: Potassium chloride buffer: 0.025 M, pH 1.0 (1.86 g KCl added to 980 mL of distilled

water in a beaker, pH measured and adjusted to 1.0 with concentrated HCl and made up to 1 L with distilled water); and Sodium acetate buffer, 0.4 M, pH 4.5 (54.43 g CH<sub>3</sub>CO<sub>2</sub>Na·3 H<sub>2</sub>O added to 960 mL distilled water in a beaker, pH adjusted to 4.5 with concentrated HCl, and made up to 1 L with distilled water). The appropriate dilution factor for the sample was determined by diluting with potassium chloride buffer, pH 1.0, until the absorbance of the sample at the vis-max is within the linear range of the spectrophotometer (i.e. for most spectrophotometers the absorbance should be less than 1.2). The final volume was divided by the initial volume to obtain the dilution factor. In order to not exceed the buffer's capacity, the sample did not exceed 20% of the total volume. Two dilutions of the sample, one with a potassium chloride buffer, pH 1.0, and the other with sodium acetate buffer, pH 4.5, were prepared by diluting each by the previously determined dilution factor. Duplicates of each were pipetted into 1cm path length cuvettes. Dilutions were equilibrated for 15 minutes. Both are read at 510 and 700nm against a blank of distilled water on a spectrophotometer (Cecil 1100, manufactured by Camlab Ltd, Cambridge UK).

213

Anthocyanin content is expressed as mg g<sup>-1</sup> Gallic Acid equivalent and is calculated as follows:

216

$$A = (A_{533} - A_{700})_{\text{pH 1.0}} - (A_{533} - A_{700})_{\text{pH 4.5}} \quad (4)$$

## 219 2.8 Phenylalanine lyase (PAL)

50 mg of plant material was ground in a pestle and mortar in 2 mL 100mM Tris – HCl buffer with 12 mM mercaptoethanol (supplied by Fisher, UK), transferred to an eppendorf and centrifuged at a speed of 16,000 rpm for 5 minutes. The sample supernatant was used in the assay. A 500 µL sample, 450 µL 100mM Tris-HCl (pH 8.8) and 50 µL 100mM phenylalanine

was placed in a water bath for one hour at 37<sup>0</sup> C. The reaction was then stopped by the addition  
225 of 50  $\mu$ L 5 M HCl. Change in absorbance was measured on a spectrophotometer (Cary 50 UV-  
Visible Varian, manufactured by Northstar Scientific, UK) at 290 nm in 1 cm light path cells  
against blanks containing 50  $\mu$ L 5 M HCl before the addition of 50  $\mu$ L 100mM phenylalanine.  
228 The amount of PAL present is expressed as nmol *trans*-cinnamic acid gram<sup>-1</sup> plant tissue hour<sup>-1</sup>.

### 231 2.9 Biomass (shoot and root)

Plants were harvested between days 5 and 7 after gassing commenced. Shoots were taken  
from each plant, washed and dried at 80°C for 2 days. Biomass was measured as fresh and  
234 dry weight. Roots were carefully removed from the chambers, washed, patted dry, weighed  
and dried for 4 days at 50°C. They were then re-weighed. The beet (storage root) was  
separated from the lateral roots from the beetroot plants and analysed independently. Beets  
237 were dried until the constant dry weight was measured. The wheat roots were measured as  
dry weight only. All statistical analyses were carried out using Minitab v 12 (USA).

## 240 3. Results and discussion

This suite of experiments designed to simulate the unlikely event of a CO<sub>2</sub> leak from CCS  
infrastructure, set out to test whether established stress responses observed in earlier  
243 studies<sup>17,18, 23-25</sup> were alleviated or magnified when crop species were grown in different soils.  
Mean gas concentrations in both the CO<sub>2</sub> and N<sub>2</sub> gassed chambers show that reductions in O<sub>2</sub>  
level were comparable both across and between the soil treatments in both crop species  
246 (Table 1). N<sub>2</sub> gassed chambers were generally slightly lower in O<sub>2</sub> concentration than the  
CO<sub>2</sub> chambers. CO<sub>2</sub> and O<sub>2</sub> data are higher than those found in test sites such as the Otway  
Project in Australia at 10% CO<sub>2</sub> maximum<sup>26</sup> However, they are comparable to values

249 measured in both outdoor field facility ASGARD<sup>17, 18</sup> and the higher CO<sub>2</sub> levels in the  
gradient site reported by<sup>9</sup>. Laboratory based systems show similar levels to those reported<sup>12-</sup>  
14.

### 252 3.1 Biomass

Different soil types influenced the level of biomass decrease in both species (Fig. 2) with a  
mean decreases in biomass of ~40% - comparable to field grown counterparts for both crops  
255 indicating that biomass (and potentially yield) are affected within the first few days of  
exposure to CO<sub>2</sub> in the soil. The N<sub>2</sub>-induced O<sub>2</sub> depletion also impacted on biomass, leading  
to a ~10% mean decrease. This corroborates evidence that both elevated soil CO<sub>2</sub> and O<sub>2</sub>  
258 depletion have an effect on vegetation<sup>18</sup>, but that soil gassed with CO<sub>2</sub> exerts a greater impact  
and is responsible for the majority of the reduced biomass.

261 Specific soil differences reveal that plants growing in L3 had a greater reduction in biomass  
than those grown in JI3 when gassed with CO<sub>2</sub>. A decrease in biomass, however, is still  
evident in JI3, with wheat showing a greater impact than beetroot. The addition of lime  
264 (CaCO<sub>3</sub>) to L3, produced the largest effect in terms of biomass reduction in both species (Fig.  
3). This large reduction in biomass with the addition of lime was an unexpected result, as it  
was reasoned that liming the soil would provide a buffer against CO<sub>2</sub>-induced acidity at the  
267 root interface. This finding suggests that acidification of soil pores through the interaction of  
CO<sub>2</sub> with water to produce carbonic acid is not a major factor responsible for the observed  
reduction in biomass, but rather that the amount of lime may have exceeded that suitable for  
270 the crops used in this soil type. The addition of sand to L3 produced an anomalous result  
(when compared to the other soil type experiments) as there was no statistically significant  
loss in biomass when comparing the CO<sub>2</sub> gassed wheat or the beetroot to their non-gassed  
273 control plants (Fig. 3 and Table 2). The L3 compost supplemented with sand was used to

simulate a soil with a more open structure. The more open structure of this soil may have provided a better growing medium for these plants thus they might have been buffered from the stress effects of high concentrations of CO<sub>2</sub> in the root zone. This may be the explanation for the beetroot as the non-gassed control has the highest biomass (Fig. 3). However this does not appear to provide an explanation for wheat as there was no increase in biomass in this soil treatment when compared to other non-gassed controls with the exception of L3 and the lime treatment that shows a reduction in biomass (Fig. 3). It is possible that the open soil structure could have minimised CO<sub>2</sub>/root contact time in this set up. Yet, the similarity in O<sub>2</sub> and CO<sub>2</sub> concentrations across soil types (Table 1) suggests this is unlikely and at the moment we are unable to explain these intriguing findings. In general the analysis of root biomass from both wheat and beetroot indicates a dramatic reduced root growth of >60% under CO<sub>2</sub> gassed soil when compared to controls. The reduction in root biomass provides a mechanism for the inability of plants to access sufficient nutrients and water. This response was investigated in more detail and found to be a whole plant response affecting the water status of the plant<sup>27</sup>.

The majority of the findings of this short-term study (a reduction in above and below ground biomass when CO<sub>2</sub> treatments are compared to controls) reflect those conducted on more long-term field trials such as work at the ZERT (Zero Emission Research and Technology) centre Montana, USA<sup>2</sup> and the ASGARD facility<sup>23-25</sup>.

### 3.2 Soil pH

For wheat plants in the L3 experiment with CO<sub>2</sub> and N<sub>2</sub> gassed treatments the soil pH was not significantly different, but both have a significantly different soil pH when compared to the L3 non-gassed control ( $p = <0.001$ ) (Fig.4). There is no difference in soil pH between treatments in L3 and sand. CO<sub>2</sub> gassed soils in the L3 and lime experiment are not

significantly different to N<sub>2</sub> gassed and non-gassed control, but N<sub>2</sub> gassed is significantly  
300 higher than control (p = 0.015). CO<sub>2</sub> and N<sub>2</sub> gassed JI3 are not significantly different, but  
both are higher than control (p = 0.03, p = 0.003 respectively).

303 In beetroot, L3 has the lowest pH (organic acidic soil). Added lime and JI3 have similar  
values between pH 6.0 and 7.0 under all treatments. There is no statistical difference between  
treatments on pH of L3 or L3 with lime. pH of L3 and sand under N<sub>2</sub> gassing is significantly  
306 lower than non-gassed control soil (p = 0.005) but not CO<sub>2</sub> gassed soil. pH of CO<sub>2</sub> and N<sub>2</sub>  
gassed soil in JI3 are both significantly lower than the non-gassed control soil (p = 0.003, p =  
0.012). Both CO<sub>2</sub> and N<sub>2</sub> gassing does have the potential to reduce pH compared to controls  
309 in JI3 and with the addition of sand. This suggests that different soil types do interact  
differentially with gasses in respect of acidity. Wheat exhibits a different result to beetroot.  
Plants are known to exude compounds that stabilise pH levels around the roots. Wheat  
312 appears to be more efficient in this process, as the pH levels in gassed plants are higher than  
the controls, except in L3, a soil that wheat prefers the least. There is no correlation between  
soil pH and biomass in either wheat or beetroot across all experimental soil types.

315

### 3.3 Plant biochemistry

Biochemical analysis was undertaken to test for plant stress as a function of treatment and to  
318 determine if soil type mitigated or amplified the plant stress response. Data shows both  
specific treatment effects (Table 3) and differences between soil types (Table 4). Results are  
presented by stress compound and subdivided by crop.

321

### 3.4 Chlorophyll

Chlorophyll analysis was undertaken to test for overall plant photosynthetic health, as a  
324 reduction in chlorophyll content would indicate plants reallocating resources from  
maintaining photosynthesis to stress mitigation. For wheat chlorophyll content in the L3 soil,  
the treatment (CO<sub>2</sub>) is not significantly different from either the N<sub>2</sub> (oxygen depletion  
327 control) or non-gassed control; in the L3 and sand combination the treatment (CO<sub>2</sub>) is  
significantly higher than the control, but not significantly different to N<sub>2</sub> control; which is  
suggestive of an O<sub>2</sub> depletion effect<sup>18</sup>. In the combined L3 and lime experiment the treatment  
330 (CO<sub>2</sub>) is not significantly different from either control and this finding is repeated in the JI3  
treatment (Table 3). Comparison between soils for a CO<sub>2</sub> effect indicates that the L3  
treatment produces statistically lower chlorophyll levels than all other soil types; L3 and sand  
333 (p = 0.005), L3 and lime (p = 0.044) and JI3 (p = 0.007) [Student's t-test] in wheat (Table 4).

Analysis of the beetroot chlorophyll data shows that in the L3 experiment chlorophyll  
336 concentrations in the CO<sub>2</sub> treatment are not significantly different from either the N<sub>2</sub> control  
or non-gassed control and these findings are repeated in the L3 and sand experiment. In the  
L3 and lime experiment chlorophyll concentrations in the CO<sub>2</sub> treatment are significantly  
339 lower than both controls and in the JI3 experiment there is no statistical difference between  
the treatment and control (Table 3). Comparison between soils for a CO<sub>2</sub> effect indicates that  
there are no statistical differences in chlorophyll content between the soil types (Table 4).

342

### 3.5 Anthocyanin

Anthocyanin analysis was undertaken to test for generic plant health as anthocyanin up-  
345 regulation is a precursor to numerous plant stresses. For wheat the following was observed  
with anthocyanin content: In the L3 and L3 and sand experiment there was no statistical  
difference between the CO<sub>2</sub> treatment and the N<sub>2</sub> control or non-gassed control. In the L3 and

348 lime and the JI3 experiment the anthocyanin concentration was significantly lower than both  
controls (Table3). Comparison between soil types shows that the anthocyanin content in the  
L3 and sand experiment has statistically higher levels of anthocyanin than in the L3 and lime  
351 ( $p = 0.005$ ) and JI ( $p = <0.0001$ ) experiments [ Student's t-test] (Table 4).

For the beetroot anthocyanin content there was no significant difference observed between  
354 treatment and either level of control in the L3, the L3 and sand and the L3 and lime  
experiments. In the JI3 experiment anthocyanin levels were significantly higher in the CO<sub>2</sub>  
treatment than the non-gassed control but not different to the N<sub>2</sub> control (Table 3).  
357 Comparison between CO<sub>2</sub> treatment and different soil types shows that in the L3 and sand  
experiment the anthocyanin content is statistically lower than the JI3 ( $p = 0.001$ ) and that  
anthocyanin content in the L3 and lime experiment is statistically lower than JI3 ( $p = <0.000$ )  
360 [Student's t-test]. Data indicate that only CO<sub>2</sub>-gassed plants grown in JI3 have a higher  
anthocyanin content than control plants ( $p = <0.0001$ ) (Table 4), this could be an indicator of  
early onset of stress in this specific treatment when compared to other soil treatments.

363

### 3.6 Phenylalanine lyase (PAL)

PAL analysis was again performed to test for generic plant health as PAL up-regulation is a  
366 precursor to numerous plant stresses. In wheat there were no significant differences in PAL  
context between the CO<sub>2</sub> treatment and the controls in any of the soil type experiments (Table  
3). Comparisons of PAL data from CO<sub>2</sub> treatments across the soil experiments shows that in  
369 the L3 and sand experiment PAL expression is statistically lower than in the L3 and lime soil  
( $p = 0.04$ ) and the JI3 soil experiments ( $p = 0.019$ ) [Student's t-test] (Table 4).



372 Analysis of beetroot PAL levels indicates no significant differences between the CO<sub>2</sub>  
treatment and either set of controls in any of the soil type experiments (Table 3).  
Comparisons of PAL data from CO<sub>2</sub> treatments across the soil experiments shows that PAL  
375 concentration in the L3 and sand experiment is statistically greater than L3 and lime (p =  
0.029), JI3 (p = 0.009) experiments, while L3 and lime is greater than the JI3 experiment (p =  
0.001) and in the L3 experiment (p = 0.034) [Student's t-test] (Table 4).

378

Overall, there is little change in stress biochemistry with treatment (CO<sub>2</sub> compared to non-  
gassed control). Comparing our pot studies to those of our longer-term field studies<sup>17</sup>  
381 indicates leaves change colour (an indication of the up-regulation of stress compounds)  
approximately ten days after the initiation of CO<sub>2</sub> treatment<sup>17</sup>. The concentration of CO<sub>2</sub> in  
the soil of our pot experiments exceeds that found in our field experiments<sup>17</sup>. Consequently  
384 this lack of a stress response can't be explained purely as a function of CO<sub>2</sub> concentration. It  
is possible that the more stable environmental conditions in the plant growth room could have  
acted as a buffer to the specific CO<sub>2</sub> stress delaying the onset of stress. However, the short  
387 duration of these pot experiments may offer an alternative explanation for the lack of an  
observed stress. We did find clear differences in non-gassed control plants in different soil  
types (Table 5), showing that soil type will influence biochemical composition regardless of  
390 the presence of an experimental stress.

The similarity in CO<sub>2</sub> concentration between our field and laboratory data is important as  
393 field based experiments to manipulate soil type would be prohibitively expensive. Moving to  
a laboratory based system that broadly matches field manipulations allows for the analysis of  
more specific soil attributes with adequate experimental replication whilst minimising costs.  
396 We have previously demonstrated<sup>18</sup> a similar response to chamber experiments and field

399 trials undertaken at the ASGARD site in comparable soil types. Consequently we are confident that the results presented in this lab study are transferable to field situations when soil types are similar to those used in our experimental set up.

402 Via funding from the National Grid, UK and the European Union Energy Programme for Recovery (EEPR) under the COOLTRANS research programme we have developed an experimental programme designed to understand the impact on crops of CO<sub>2</sub> leakage from CCS infrastructure. This programme focussed either on catastrophic failure<sup>28</sup> or small scale leakage<sup>17-19</sup>. Synthesising these findings reveals that although there are noticeable effects on crops these affects across all experiments are minimal when placed into the context of farm scale agriculture. For example in field trials where biomass and yield decreased, the area of 408 vegetation that was affected was small, between 0.2 and 0.3 m<sup>2</sup> in spring barley and grass/clover and ~0.5m<sup>2</sup> for spring oilseed rape and autumn barley. In the context of an average arable field size in the UK of 12 ha, this represents an area of 0.00006% ha, with 411 yield losses corresponding to 0.0003% ha.

#### 4. Conclusions

414 The loss of biomass is broadly consistent across soil types for both species investigated, the exception being L3 and sand. From a stakeholder perspective these findings suggest that on the whole soil type does not amplify plant stress in response to the unlikely event of a CO<sub>2</sub> 417 leak from CCS pipelines. Intriguingly our data suggests that plants in a sand rich soil might be less susceptible to CO<sub>2</sub> induced stress. But the reasons behind this reduction in susceptibility are currently unknown so these findings should be interpreted with caution.

420 Looking more broadly across our work linked to CO<sub>2</sub> leakage from CCS infrastructure the impact of crop plants again appears to be localised.

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426

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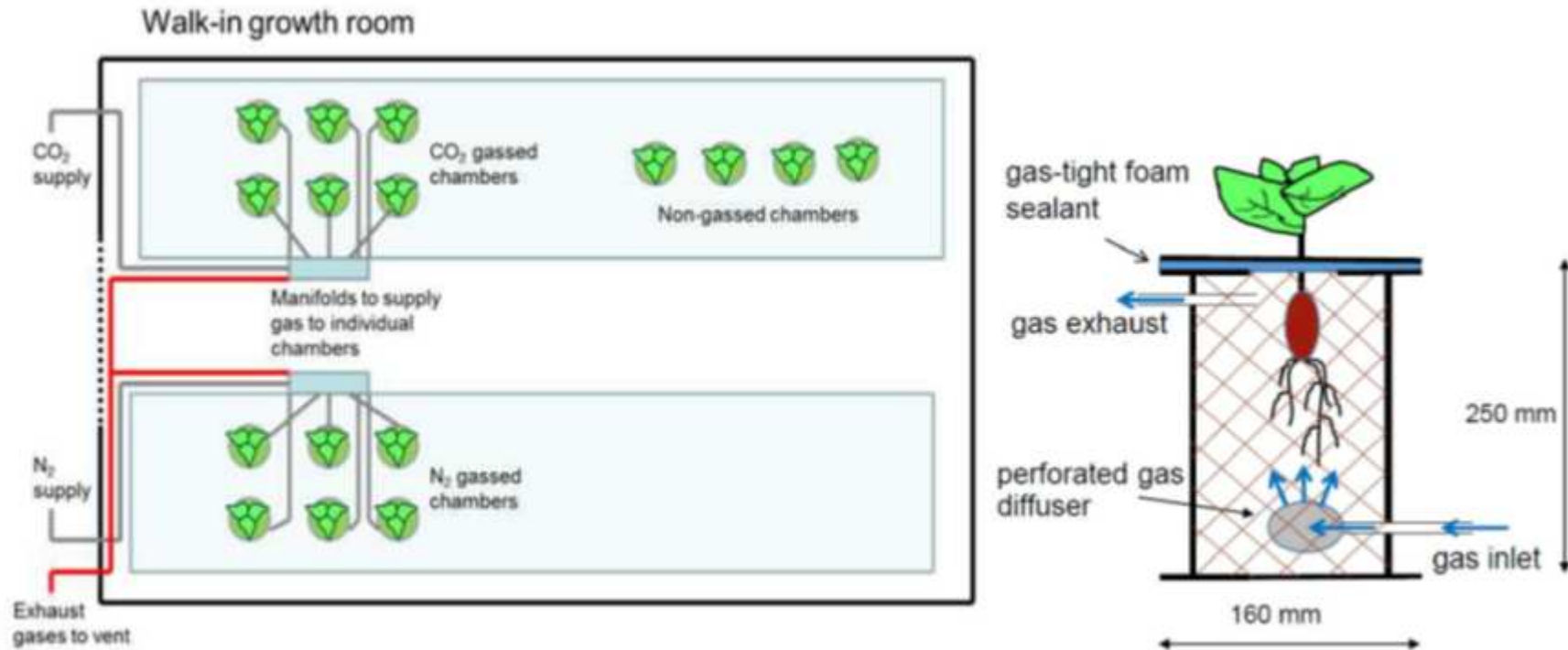
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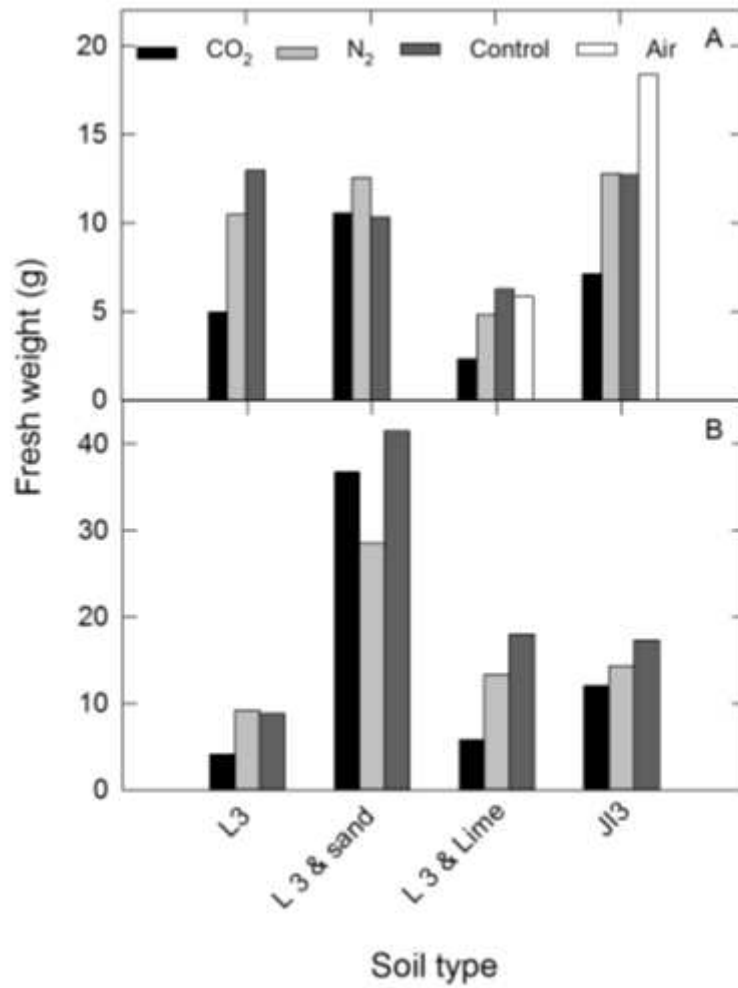
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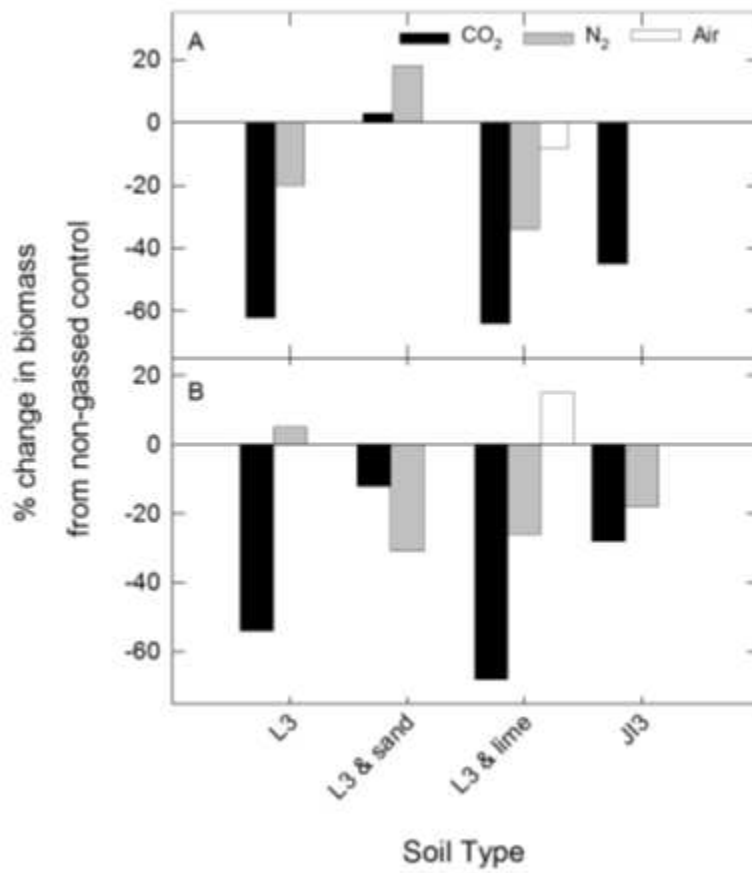
## Figures



- 3 **Fig. 1.** Schematic representation of the experimental arrangement within a walk-in controlled environment facility and a soil chamber with a beetroot plant. Gases were exhausted to atmosphere via a separate manifold to prevent build up within the growth room.

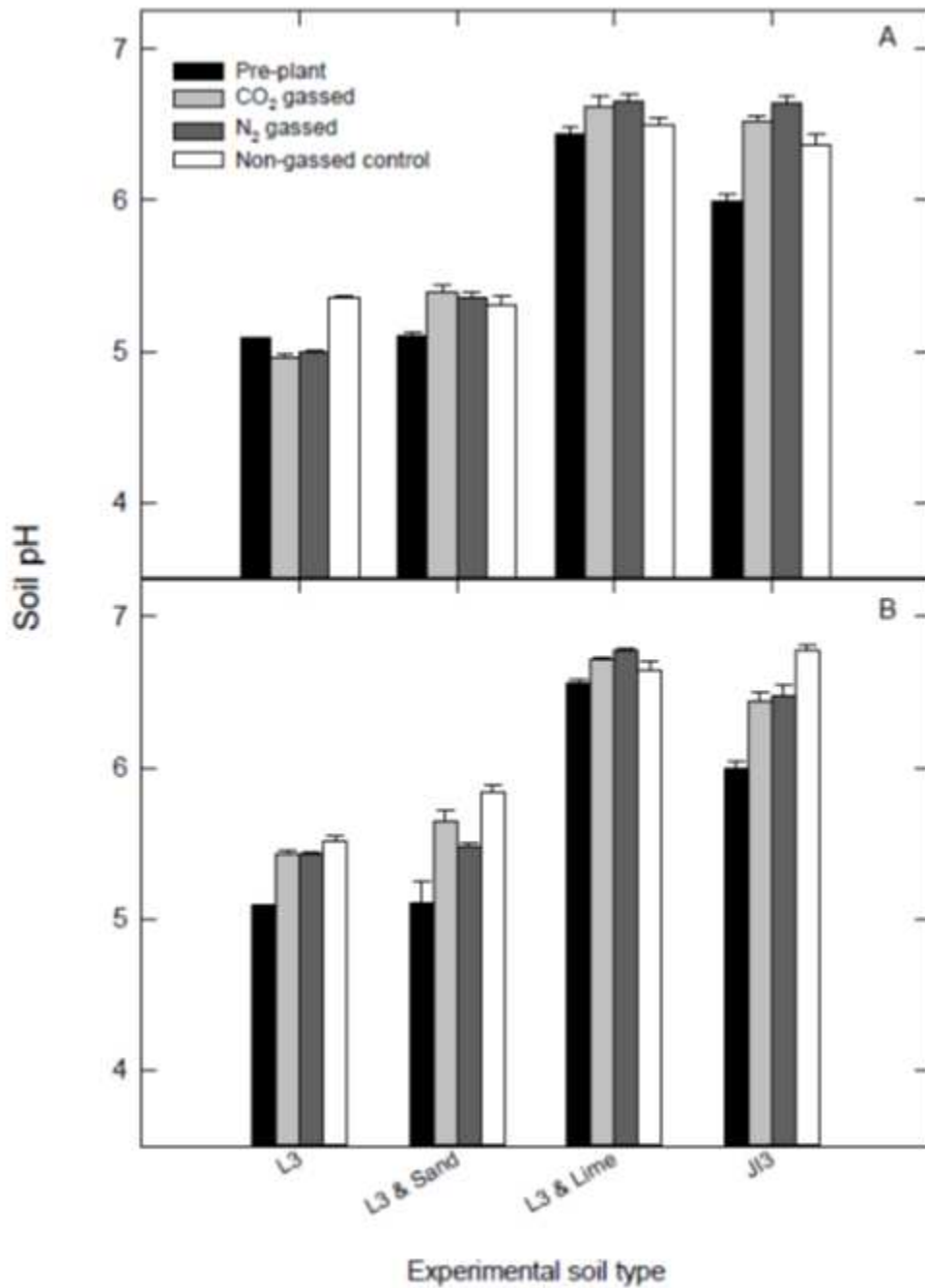


**Fig. 2.** Fresh weight biomass at harvest across experimental treatment and soils type. (A) Wheat and (B) beetroot. A full statistical break down of results is given in the text. See materials and methods for experimental set up.



3 **Fig. 3.** Percentage change in biomass across experimental treatment and soils type relative to controls. (A) Wheat and (B) beetroot. See materials and methods for experimental set up.

6



3 **Fig. 4.** Comparison in soil pH across the different soil treatments. (A) Wheat and (B) beetroot. A full statistical break down of results is given in the text.

## Tables

**Table 1.** Mean gas concentrations measured as % CO<sub>2</sub> and % O<sub>2</sub> (v/v) within the soil

3 chambers.

6	Crop & soil type	CO <sub>2</sub> concentration (%)		O <sub>2</sub> concentration (%)	
		CO <sub>2</sub> gassed	N <sub>2</sub> gassed	CO <sub>2</sub> gassed	N <sub>2</sub> gassed
	<b>Wheat</b>				
	L3	48.4 (12.9)	0.15 (0.2)	9.57 (2.7)	8.67 (3.9)
9	L3 plus sand	40.5 (0.6)	0.14 (0.05)	11.83 (0.09)	9.18 (0.4)
	L3 plus lime	42.03 (3.2)	0.24 (0.3)	12.15 (0.4)	10.2 (3.2)
	J13	44.14 (5.9)	0.95 (0.03)	11.48 (1.5)	10.1 (1.7)
12	<b>Beetroot</b>				
	L3	48.0 (4.9)	0.60 (0.5)	8.97 (1.5)	8.76 (1.7)
	L3 plus sand	51.9 (4.9)	0.17 (0.02)	8.81 (0.7)	6.74 (0.5)
15	L3 plus lime	32.0 (17.2)	0.60 (0.4)	13.68 (0.4)	9.04 (1.9)
	J13	32.81 (4.2)	0.59 (0.2)	12.08 (1.4)	12.59 (1.1)

18 [n = 5; (SEmean)]

21

3 **Table 2.** Percent change in biomass for wheat and beetroot grown in different substrates and  
 6 gassed with either CO<sub>2</sub>, N<sub>2</sub> or air compared to non-gassed (control).

6	Soil type	Wheat			Beetroot		
		CO <sub>2</sub>	N <sub>2</sub>	Air	CO <sub>2</sub>	N <sub>2</sub>	Air
9	L3	-62	-20	n/a	-54	5	n/a
	L3 & sand	+3	+18	n/a	-12	-31	n/a
	L3 & lime	-64	-34	-8	-68	-26	15
12	JI 3	-45	0	n/a	-28	-18	n/a

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**Table 3.** Treatment effects on biochemical compounds associated with stress. Values given are the content of each compound found in CO<sub>2</sub> gassed leaves. Statistical comparison is between treatments (CO<sub>2</sub> gassed) are compared to non-gassed control plants within each soil experiment.

6

**Soil pH and significance level (difference from control)**

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Treatment	pre-plant	treatment			non-gassed control		
		CO <sub>2</sub> gassed	N <sub>2</sub> gassed	<i>p</i> value	pH	<i>p</i> value	pH
<b>Crop</b>	<b>Soil type</b>	<b>pH</b>	<b>pH</b>	<b><i>p</i> value</b>	<b>pH</b>	<b><i>p</i> value</b>	<b>pH</b>
12 <b>Wheat</b>	L3	5.09	4.96	<0.001	4.99	<0.001	5.35
	L3 & sand	5.10	5.39	NS	5.35	NS	5.31
	L3 & lime	6.44	6.62	NS	6.66	0.015	6.5
	JI3	5.99	6.52	0.03	6.64	0.003	6.36
18 <b>Beetroot</b>	L3	5.09	5.43	NS	5.43	NS	5.51
	L3 & sand	5.10	5.65	NS	5.48	0.005	5.84
	L3 & lime	6.56	6.72	NS	6.77	NS	6.64
	JI3	5.99	6.44	0.003	6.48	0.012	6.78

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**Table 4.** Soil type effects on biochemical compounds associated with stress. Statistical comparisons are between treatments (CO<sub>2</sub> gassed) across each soil type.

<b>Biochemical Compound and significance level</b>								
6	<b>Crop</b>	<b>Soil type</b>	<b>Chlorophyll</b> (mg g <sup>-1</sup> )	<b>p value</b>	<b>Anthocyanin</b> (mg g <sup>-1</sup> GA equivalent)	<b>p value</b>	<b>PAL</b> (nmol trans-CA g <sup>-1</sup> hr <sup>-1</sup> )	<b>p value</b>
9	Wheat	L3	15.84	<0.05	5.56	NS	133483.1	NS
		L3 & sand	22.49	NS	9.84	<0.05	88453.7	<0.05
		L3 & lime	22.15	NS	4.94	NS	72428.7	NS
12		Jl3	22.16	NS	2.99	NS	118975.4	NS
	Beetroot							
		L3	16.92	NS	27.8	NS	75463.3	NS
15		L3 & sand	18.06	NS	4.50	0.001	198100.7	NS
		L3 & lime	16.27	NS	4.06	<0.0001	81539.4	NS
		Jl3	19.51	NS	10.91	NS	54079.8	<0.05

[mean values of 4 to 6 replicate plants]



**Table 5.** Comparison of biochemical analysis of non-gassed controls only in all soil types.  
Influence of soil type alone.

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<b>Biochemical Compound (non-gassed control only)</b>				
<b>Crop</b>	<b>Soil type</b>	<b>Chlorophyll</b> (mg g <sup>-1</sup> )	<b>Anthocyanin</b> (mg g <sup>-1</sup> GA equivalent)	<b>PAL</b> (nmol trans-CA g <sup>-1</sup> hr <sup>-1</sup> )
<b>Wheat</b>				
	L3	19.92	10.75	116857.2
	L3 & sand	22.30	10.63	94030.9
	L3 & lime	22.33	10.01	104607.0
	Jl3	22.33	8.47	95222.4
<b>Beetroot</b>				
	L3	16.91	17.55	81463.3
	L3 & sand	20.1	3.72	287984.2
	L3 & lime	21.96	7.01	25080.9
	Jl3	16.98	3.09	72205.8