

This is a repository copy of *The 'allosteron' model for entropic allostery of self-assembly*.

White Rose Research Online URL for this paper:

<https://eprints.whiterose.ac.uk/id/eprint/143107/>

Version: Accepted Version

Article:

McLeish, Tom orcid.org/0000-0002-2025-0299, Schaefer, C. and von der Heydt, A. C. (2018) The 'allosteron' model for entropic allostery of self-assembly. *Philosophical Transactions of the Royal Society B: Biological Sciences*. 20170186. ISSN: 1471-2970

<https://doi.org/10.1098/rstb.2017.0186>

Reuse

Items deposited in White Rose Research Online are protected by copyright, with all rights reserved unless indicated otherwise. They may be downloaded and/or printed for private study, or other acts as permitted by national copyright laws. The publisher or other rights holders may allow further reproduction and re-use of the full text version. This is indicated by the licence information on the White Rose Research Online record for the item.

Takedown

If you consider content in White Rose Research Online to be in breach of UK law, please notify us by emailing eprints@whiterose.ac.uk including the URL of the record and the reason for the withdrawal request.

The ‘Allosteron’ Model for Entropic Allostery of Self-Assembly

T. C. B. McLeish,^{1,*} C. Schaefer,^{1,†} and A. C. von der Heydt^{1,‡}

¹*Department of Physics, Durham University,
South Road, Durham DH1 3LE, UK*

(Dated: October 6, 2017)

Abstract

Using the simple ‘allosteron’ model, we show that it is possible in principle to elicit pathways by which fluctuation allostery affects self-assembly of protein complexes. We treat the cases of (1) protein fibrils and nucleation, (2) n -mer protein complexes, and (3) show how the elastic and (weakly) attractive association interactions in a family of protein-like soft nanoscale objects can be tuned in a way that defines exclusive self-associating families.

* t.c.b.mcleish@durham.ac.uk

† charley.schaefer@durham.ac.uk

‡ alice.c.von-der-heydt@durham.ac.uk

I. GENERAL INTRODUCTION

An alternative, but less familiar, mechanism of allosteric signalling compared to that of structural change, exploits the modification that substrate binding in general applies to the amplitudes of thermal fluctuations around the mean structure of an allosteric protein. Since such a restriction of random motion constitutes in turn a change in entropy, the full allosteric free energy may contain components that arise purely by this route [1]. A feature of this mechanism is that to contribute to non-local allosteric interaction, the longer-wavelength low-frequency ‘global’ modes of motion are recruited, rather than the higher-frequency and more local motions (such as side-group oscillation). Successful models of protein dynamics that capture the effect are therefore coarse-grained rather than atomically-resolved. Specific models of particular protein systems at various degrees of coarse-graining have been constructed [2–5], which show that the orders of magnitude of real allosteric free energies can be generated by such restriction of dynamical correlations alone. An equivalent statement of the effect is a restriction of thermally-accessible states on binding [6]. Similar challenges are presented to experiments, which need to identify the large-scale dynamical changes at many points on a protein on substrate-binding, and to connect them to thermodynamics. NMR [7], X-ray analysis of B-factors and isothermal calorimetry [8] have been deployed in comparison with modelling.

At the simplest level of theoretical approach to this phenomenon of ‘fluctuation allostery’, use of the toy model of a protein at the coarsest possible level, with just one (harmonic) degree of freedom, has been extremely instructive. This simple unit, which can be represented without loss of generality as the scissor structure of Fig. 1 (a) (although other spatial and geometric representations are equally applicable, as any normal mode within the harmonic approximation is mathematically equivalent), possesses the minimal requirements of: (1) an internal structure that supports thermal structural fluctuations, (2) one or more potential effector binding sites with internal rules that modify the internal fluctuations, and (3) a route to bind to other units, see Fig. 1 (b).

It is straightforward to calculate the free energy $F(\kappa)$ of a single harmonic degree of freedom over its fluctuation spectrum, constrained by a spring constant κ . Since the relevance of this treatment of the normal modes of crude protein models is restricted to low-frequency, global modes, it is appropriate to use the classical (continuous-energy) approximation to the

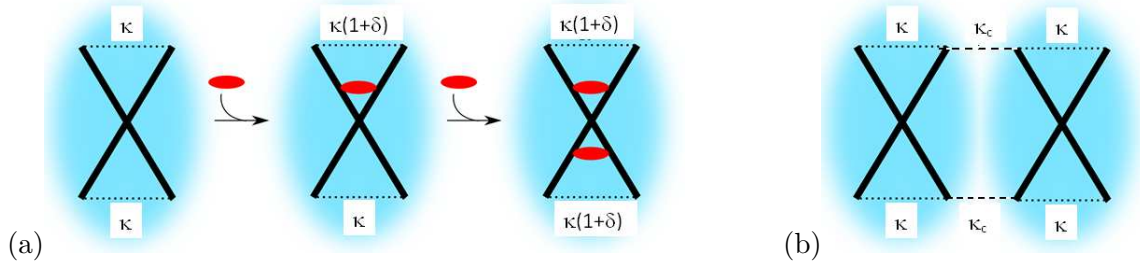


FIG. 1. (Colour online) Schematic of the allosteron (a), and its binding (b).

harmonic-oscillator partition function \mathcal{Z}_∞ :

$$\begin{aligned} \frac{F(\kappa)}{k_B T} &= -\ln \mathcal{Z}_\infty = -\ln \left\{ \int_{-\infty}^{\infty} \exp \left\{ -\frac{\kappa x^2}{2k_B T} \right\} d \left(\frac{x}{\lambda} \right) \right\} \\ &= -\frac{1}{2} \ln \left(\frac{2\pi k_B T}{\kappa \lambda^2} \right), \end{aligned} \quad (1)$$

with Boltzmann's constant k_B , temperature T , and a length scale λ for normalisation.

Modelling the fluctuation restriction of effector binding as a modulation of the effective spring constant for the internal mode by the dimensionless increment δ , allows a simple and direct calculation of the allosteric free energy $\Delta\Delta F$ of even a single degree of freedom. Since a single dynamical mode of a mechanical system is typically extended across the protein, it will in general offer more than one potential binding sites, which are also close to anti-nodes of the global mode, where substrate binding will, in general, affect the value of κ . If two successive effector binding events take $\kappa \longrightarrow \kappa + \delta \longrightarrow \kappa + 2\delta$:

$$\begin{aligned} \Delta\Delta F &= F((1+2\delta)\kappa) + F(\kappa) - 2F((1+\delta)\kappa) \\ &= k_B T \ln \left(\frac{1+2\delta}{(1+\delta)^2} \right) \end{aligned} \quad (2)$$

We have called this minimal fundamental building block for models of allosteric systems, the ‘allosteron’ [9]. In a natural extension of the single-mode allosteron unit, allosteric interaction between dimers can be modelled at the simplest possible level by associating two allosterons into a dimer [see Fig. 1 (b)], introducing a single new parameter of the harmonic coupling between them, κ_c . At any level of approximation within linear response, the elastic part of the Hamiltonian \mathcal{H} relevant to entropy-driven allostery (without contributions of momenta) is a quadratic form in the vector \mathbf{x} of allosterons’ degrees of freedom, with the (spring-constant) Hessian matrix H :

$$\mathcal{H} = \frac{\kappa}{2} \mathbf{x}^T \hat{H} \mathbf{x}, \quad (3)$$

written in terms of a dimensionless form, \hat{H} , of H in units of κ . If two allosterons with internal modes governed by spring constants κ_1 and κ_2 meet in association, the Hessians H_1 and H_2 in terms of the two single degrees of freedom (x_1, x_2) , one per monomer, before and after their weak association are:

$$H_1 = \begin{pmatrix} \kappa_1 & 0 \\ 0 & \kappa_2 \end{pmatrix} \quad H_2 = \begin{pmatrix} \kappa_1 + \kappa_c & -\kappa_c \\ -\kappa_c & \kappa_2 + \kappa_c \end{pmatrix} \quad (4)$$

The entropic free energy of binding, from the introduction of the coupling interaction between two free allosterons is

$$\begin{aligned} \Delta F &= k_B T \ln \left[\frac{\det(H_1)}{\det(H_2)} \right] \\ &= k_B T \ln \left[1 + \kappa_c \left(\frac{1}{\kappa_1} + \frac{1}{\kappa_2} \right) \right] \end{aligned} \quad (5)$$

This is, of course, positive definite as it represents a reduction in internal entropy of structural fluctuations on binding. The full free energy of binding will include an enthalpic term ΔU , so $\Delta F_{\text{bind}} = \Delta F + \Delta U$, for which we will develop a simple model below, in Section IV.

Such an approach has been used successfully as the coarsest-level model for allosteric protein dimers [2–4], in decorated form as a basis for theories of mode-coupling to local fast dynamics [3], and as a generator of fitness landscapes in the evolution of allosteric systems [8]. In spite of its extreme simplicity, this toy-model approach to protein dimers, for example, has delivered highly non-trivial insights, such as a region of its three-dimensional parameter space where a dimer may exhibit negative co-operativity [4].

Within a graded series of approximations to protein dynamics, the allosteron model sits at a similar level of coarse-graining to the Rotational Translational Block (RTB) approximation [10]. The strong elastic inhomogeneities within proteins that render the RTB picture appropriate also suggest that the dominance, for some purposes, of a few low frequency and long-range dynamic modes is a valid one. Allostery without structural change, driven by modulation of thermal fluctuations, which depends on long-range information transfer, is one such application. As the degree of approximation is refined, we arrive at models that resolve individual residues (such as Elastic Network Models — ENM’s [8, 11–13]), and finally at all-atomistic models [8].

In this work, we apply the allosteron model to elucidate the potential contributions of internal fluctuation entropy to protein self-assembly. The concatenation of fiber-forming

proteins, for example, is not usually thought of as an example of allostery. However, if we think of the two effector molecules for an allosteric protein as copies of the protein itself, then the connection is evident. There will, in general, be a modification of the internal fluctuations in a protein when it joins a self-assembling complex or fibre. In principle, this generates a non-equivalence of binding free-energies for each protein in the complex, even when these are structurally identical and when there is no mean change in structure on binding. In the following, we explore this insight, at the level of the allosteron model, in two cases: In Section II, for protein fibrils of arbitrary molecular weight; in Section III, for ligand-binding of finite protein n-mers. Section IV then explores another possibility identified by this approach — that of elastically-tuned families of proteins that possess weak mutual and exclusive association.

II. SELF-ASSEMBLY OF ALLOSTERON CHAINS

Protein fibrils that consist of reversibly bound proteins may be viewed as supramolecular ‘living’ polymers that grow and break up [14–16]. At steady state, the chain lengths of polymers obey equilibrium statistics, and may therefore be controlled by the thermodynamic properties of the building blocks. These properties include an enthalpic and entropic gain or penalty of monomer association to the polymer. Because of structural, hydrophobic and electronic effects, to name a few, the free energy of monomer association to a polymer often depends on the length of the oligomer. Consequently, polymerisation is often cooperative and the degree of polymerisation strongly depends on the *cooperativity factor* [14, 17–19].

The entropic and energetic contributions that lead to (cooperative) supramolecular polymerisation can easily be included in the allosteron model. For instance, the association of a monomer can be entropically penalised by stiffening the monomer through an increase of their internal spring constant $\kappa \mapsto \alpha\kappa$ with $\alpha > 0$, even in absence of coupling, i.e., $K_c = 0$. The entropic penalty of dimerisation is then $\Delta S_{\text{nucl}} = 2k_B \ln \alpha$, whereas the entropic penalty of elongation is only $\Delta S_{\text{elong}} = k_B \ln \alpha$. Hence, the allosteron model captures the entropic origin for cooperativity by a statistical-mechanics description [20]. The novel feature of the allosteron model is that the internal modes of the monomers are coupled ($K_c > 0$), which enables entropic allosteric signalling along the backbone of the polymer. In this section, we show that coupling leads to an entropic interaction range of $\sqrt{K_c}$ monomers, as well as an

increase in the polymerisation concentration by a factor of $\sqrt{K_c}$.

We predict these phenomena from the allosteron model by investigating the equilibrium statistics of an allosteron solution in a volume V [21, 22]. The grand potential of the mixture is given by

$$\frac{\Omega}{Vk_B T} = \sum_{N=1}^{\infty} \rho(N) \left[\ln(\rho(N)v) - 1 - \ln(Z_N) - \frac{\mu N}{k_B T} \right]. \quad (6)$$

In this expression, $\rho(N)$ is the number density of chains of length N (note that we consider a monomer to be a chain of length $N = 1$), v is the interaction volume of a monomer, Z_N is the partition function of a chain of length N and μ is its chemical potential.

The partition function of the chain is obtained from the Hamiltonian where allosterons with spring constant κ bind through a binding energy ϵ at an entropic cost dictated by the coupling-spring constant κ_c . This bead-spring-like Hamiltonian is constructed by enhancing Eq (3) by an enthalpic term ϵ to give $\mathcal{H} = \frac{1}{2}\kappa \mathbf{x}^T \hat{H}_N \mathbf{x} - \epsilon(N-1)$. The dimensionless Hessian is a tridiagonal matrix of which the main-diagonal elements are given by $M_{11} = M_{NN} = 1 + K_c$ and by $M_{nn} = 1 + 2K_c$ for $1 < n < N$. Further, the first upper and lower diagonal elements equal $-K_c$, whereas all other matrix elements equal zero. The partition function is given by

$$\begin{aligned} \mathcal{Z}_N &= \int d^N x \exp \left(-\frac{\mathcal{H}}{k_B T} \right) \\ &= \mathcal{Z}_1^N \frac{1}{\sqrt{\det \hat{H}_N}} e^{-\frac{\epsilon}{k_B T}(N-1)}, \end{aligned} \quad (7)$$

with $\mathcal{Z}_1 \equiv \sqrt{2\pi k_B T / \kappa}$ the partition function of a free monomer, see Eq. (1).

We minimise the grand potential in Eq. (6) to find that we can write the solution for the distribution function as

$$\rho(N) = \frac{1}{v} K_N e^{-(g-\tilde{\mu})N/k_B T}, \quad (8)$$

where we have defined the free energy of aggregation $g \equiv -\epsilon - k_B T \ln(\sqrt{w})$, the chemical potential $\tilde{\mu} \equiv \epsilon - \mu - k_B T \ln \mathcal{Z}_1 / \sqrt{w}$ and the equilibrium constant $K_N \equiv \sqrt{w^{N-1} / \det \hat{H}_N}$. The subsidiary definition $w \equiv 1/2 + K_c + \sqrt{1 + 4K_c}/2$ guarantees that the equilibrium constant equals unity for the monomers, $N = 1$, but converges to a finite value for large N , see SI.

Indeed, from the Hamiltonian of the chain we deduce that the entropic interactions between monomers along the backbone extend over a distance of a persistence length $\sqrt{K_c}$. For chains shorter than this, the stiffness may be considered large and it can be shown that

$\det \hat{H}_N \approx NK_c^{N-1}$, see SI. For infinitely long chains, however, we find that $\det \hat{H}_N = \sigma w^{N-1}$, where σ is a constant *cooperativity factor*: If the cooperativity factor is unity then monomer association to a chain is independent of the chain length and polymerisation is ‘isodesmic’. For small values of σ , elongation is considerably easier than dimerisation and polymerisation is cooperative. In the present case, the value of the cooperativity factor follows from the fact that the two limits of $\det \hat{H}_N$ must crossover at $N = \sqrt{K_c}$. This gives the weak dependence $\sigma \propto K_c^{-1/4}$, which suggests that, in absence of ligand binding to the monomers, the allosteron model virtually predicts isodesmic polymerisation.

We confirm this by finalising our calculation of the chain-length distribution of the polymers via a calculation of the chemical potential, μ , which is implicitly given by the mass balance $\phi = \sum_{N=1}^{\infty} N\rho(N)v$, with ϕ the overall volume fraction of the monomers [21, 22]. By introducing the mass action (proportional to the experimentally controllable monomer concentration) $X = X(\phi, T; K_c, \epsilon) \equiv \phi \exp(g/k_B T)$, we cast the mass balance in the form

$$X(\phi, T; \epsilon, K_c) = \sum_{N=1}^{\infty} NK_N e^{-\tilde{\mu}N/k_B T}. \quad (9)$$

We numerically extract the chemical potential and insert it into Eq. (8) to obtain the length distribution of the chains.

From this distribution we obtain the fraction of polymerised material, f , and the number-averaged degree of polymerisation, \bar{N}_n . Figure 2 shows these quantities as a function of the mass action for a coupling constant K_c ranging from 0 to 10^4 . We have scaled the mass action using $X_p = 2.34$, at which half of the material is polymerised for the true isodesmic case, $K_c = 0$ [21]. As expected, supramolecular polymerisation remains close to isodesmic: The polymerisation curve maintains its symmetric S shape, and the degree of polymerisation for $X \gg X_p$, given by $\bar{N}_n \propto \sqrt{X/\sigma}$, only weakly depends on K_c through $\sigma \approx K_c^{-1/4}$.

Nevertheless, the crossover concentration and crossover temperature may be affected by the entropic coupling between the monomers through the coupling parameter K_c . Indeed, in the isodesmic limit ($K_N = 1$ for all N) the polymerisation temperature is

$$T_p = \frac{\Delta H_p}{\Delta S_p} \propto 1/\ln \sqrt{K_c}, \quad (10)$$

with $\Delta H_p \equiv \epsilon$ the enthalpy of polymerisation and $\Delta S_p \equiv k_{\{B\}} \ln(\sqrt{w}/4\phi)$ the entropy of polymerisation. We hence find that the polymerisation temperature is only weakly affected by the coupling constant. In contrast, the polymerisation concentration is much more

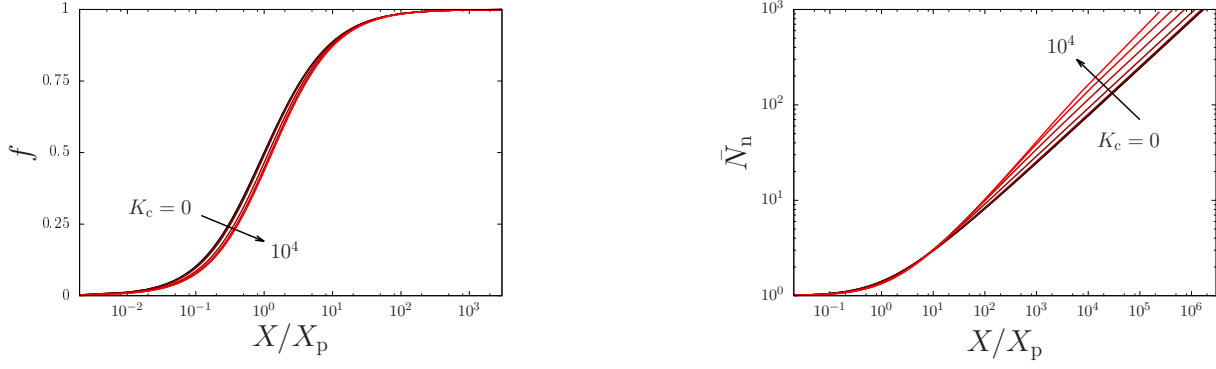


FIG. 2. (Colour online) Fraction of polymerised material f (top), and number-averaged degree of polymerisation \bar{N}_n (bottom), as a function of the mass action X for coupling constants K_c ranging from 0 to 10^4 . The mass action is scaled using $X_p = 2.34$, see main text.

strongly affected. This quantity is given by

$$\phi_p = \frac{1}{4} \exp \left(-\frac{\Delta H_p - T\Delta S_p}{k_B T} \right) \propto \sqrt{K_c}. \quad (11)$$

This shift in polymerisation concentration originates from the stiffening of not only the associating monomer, but also of the monomers within a distance $\propto \sqrt{K_c}$ from the chain end. The interaction range, as well as the shift in polymerisation concentration, may be reduced by increasing the internal spring constants of the monomers to $\kappa \mapsto \alpha\kappa$, which leads to $K_c \mapsto K_c/\alpha$. We speculate that this could be achieved by the binding of activators, such as ligands, to the monomers. The phenomena of ligand-induced dimerisation and polymerisation can potentially be addressed using the allosteron model. It should be noted that this generates a copolymer that consists of both activated and non-activated monomers. As we have seen, the range of entropic interaction exceeds the nearest-neighbour distance, and currently available Ising-like models cannot be applied directly [23]. We address this in forthcoming work. The importance of entropic signalling is not limited to supramolecular polymerisation alone, but is also of crucial importance to the allosteric properties of proteins, as we discuss in the following.

III. ALLOSTERIC INTERACTIONS ON RING COMPLEXES

In this section, we apply the model of coupled allosterons —as exploited for polymerisation above— to the ligand-binding cooperativity (allostery) of polymers of fixed size.

Specifically, we aim at the relative entropy changes induced by binding, conditional on prior binding events. Many proteins are known to occur as ring oligomers, such as the oxygen carrier, ferro-protein haemoglobin, or to form multi-protein complexes of ring topology. Due to their doubly connected and intrinsically stable structure, rings are promising candidates to display entropy-driven allostery without conformational change [1]. In fact, for the hetero-tetramer haemoglobin, the cooperative O₂ binding is well known [24], yet the underlying mechanism is still a matter of active research [25, 26]. From a modelling point of view, rings’ periodic boundary conditions suggest the possibility of generalising results obtained for small rings to larger ones.

A. Model

We briefly recall the main ingredients of the coupled-allosteron picture, located within the class of elastic network models (ENM) [11, 27]: Each allosteron, called ‘unit’ in what follows for the sake of brevity, possesses an internal ‘breathing’ mode [5, 28, 29], modelled as a single harmonic degree of freedom, of spring constant κ , see Fig. 1. One unit can represent a monomer, a protomer such as a helix dimer, or a protein, and interacts with its two neighbours on the ring via a harmonic mode of constant κ_c , of relative strength $K_c \equiv \kappa_c/\kappa$. The dimensionless Hessian matrix \hat{H} which encodes the network’s connectivity takes for a tetramer (four-unit) ring the form

$$\hat{H} = \begin{pmatrix} 1 + 2K_c & -K_c & 0 & -K_c \\ -K_c & 1 + 2K_c & -K_c & 0 \\ 0 & -K_c & 1 + 2K_c & -K_c \\ -K_c & 0 & -K_c & 1 + 2K_c \end{pmatrix} \quad (12)$$

A first approach to model ligand binding is to assume a bound unit’s internal mode to change in strength, according to $\kappa \mapsto \alpha\kappa$. By taking this to be the only binding-induced modification of interactions, we arrive at a single-parameter model of binding. A more differentiated picture of allostery emerges from invoking a two-parameter model, in which ligand-binding is assumed to affect also the strength of the harmonic coupling to the neighbouring units, according to $\kappa_c \mapsto \beta\kappa_c$. The mappings in \hat{H} which reflect this model of binding, exemplified for unit i , take $1 \mapsto \alpha$ and $K_c \mapsto \beta K_c$ in the i th diagonal entry, and multiply off-diagonal entries in both the i th row and the i th column by β .

B. Allosteric Free Energies

In order to quantify the entropy-induced cooperativity between successive binding events, we analyse differences of entropic free energies of binding, such as the allosteric free energy for two bindings,

$$\Delta\Delta F_{2,1} = F_2 - 2F_1 + F_0. \quad (13)$$

Herein, F are total free energies computed via the determinant of \hat{H} , and subscripts 0, 1, and 2 refer to the state with no, one ligand, and two ligands bound, respectively.

C. Trimer Rings

Despite its simplicity, a trimer ring may illustrate transparently the mechanism of cooperativity (of both signs) at work between states of different binding degrees. Indeed, allosteric interactions exist between the three units, pairwise neighbours in this case.

The values and spacing of free energies of binding, ΔF , for the i th binding defined as $\Delta F \equiv F_i - F_0$, for no, one site, two, and eventually all sites binding a ligand, computed within the one-parameter binding model, are plotted in Fig. 3 (a). Cumulative binding is seen to be cooperative, i.e., favoured by the differences of ΔF , for all values of α and K_c . The map of the particular allosteric free energy $\Delta\Delta F_{2,1}$ [cf., Eq. (13)] against α and K_c , in Fig. 3 (b), shows explicitly the purely cooperative allostery predicted by this model.

In part, this prediction may be traced back to the one-parameter model of binding, which rescales diagonal elements of the Hessian \hat{H} only, combined with the logarithmic dependence of the free energy on the (eigenfrequency) spectrum of \hat{H} [5]. A more trimer-specific explanation relates to the mentioned degeneracy as regards nearest neighbours: Local decrease [increase] of the extent of thermal fluctuations and the associated entropy loss [gain], induced by changing one internal-mode strength, automatically carries over to all remaining units, which in a trimer are all directly coupled. Therefore, the entropy change upon altering another internal-mode strength is smaller in modulus than that for a previous binding.

Within the two-parameter model of binding, we find that both cooperativity and anti-cooperativity of successive bindings (both signs of $\Delta\Delta F$) can occur, dependent on the set of modifiers, (α, β) , and the initial ratio K_c of coupling to internal-mode strength. Maps

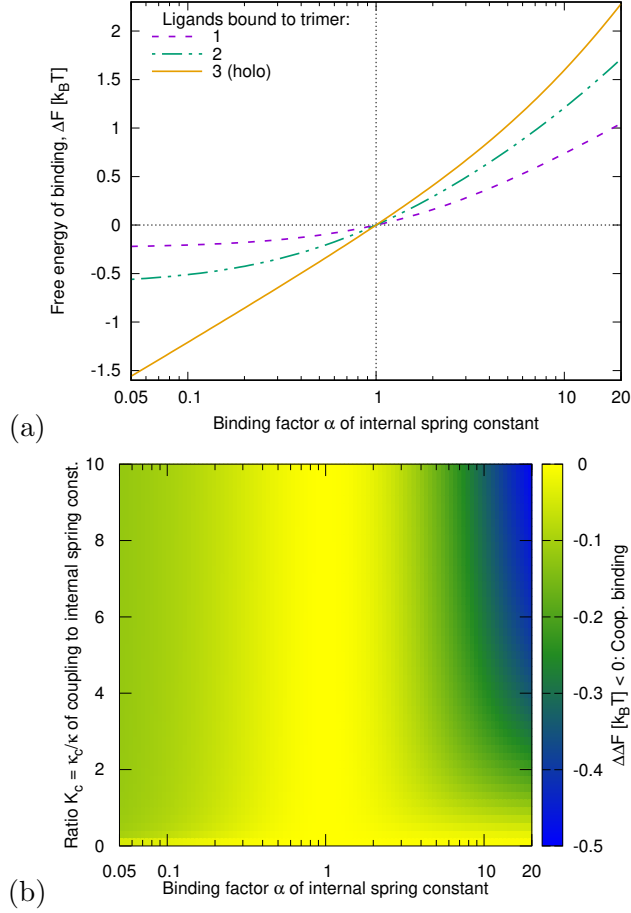


FIG. 3. (Colour online) Trimer ring, one-parameter binding model: (a) Binding free energies $\Delta F(\alpha)$ for $K_c = 5$. (b) Allosteric free energy $\Delta\Delta F_{2,1}(\alpha, r)$.

of $\Delta\Delta F_{2,1}(\alpha, \beta)$ such as in Fig. 4, parameterised by K_c , indicate anti-cooperative binding to arise particularly for strengthened internal mode and weakened coupling. Larger values of K_c or α cause more pronounced allosteric interactions, i.e., a larger range of values of $\Delta\Delta F_{2,1}$.

D. Tetramer Rings

Haemoglobin is one of the most prominent proteins of tetramer-ring topology. In contrast to binding to a trimer ring, successive binding to the four units can proceed via more than one pathway, so that both number and *configuration* of bound ligands [29] have to be specified in ΔF . The energy-level plot for a tetramer within the one-parameter model of binding is provided in Fig. 5. It reveals that two ligands at diametrical sites of the tetramer ring, either

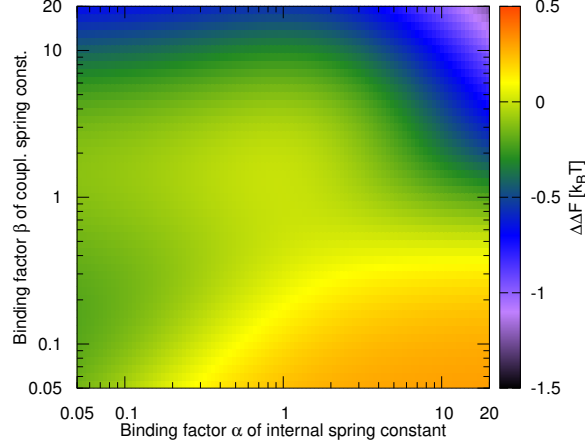


FIG. 4. (Colour online) Trimer ring, two-parameter model: Allosteric free energy $\Delta\Delta F_{2,1}(\alpha, \beta)$; initial coupling-to-internal strength ratio $K_c = 2$.

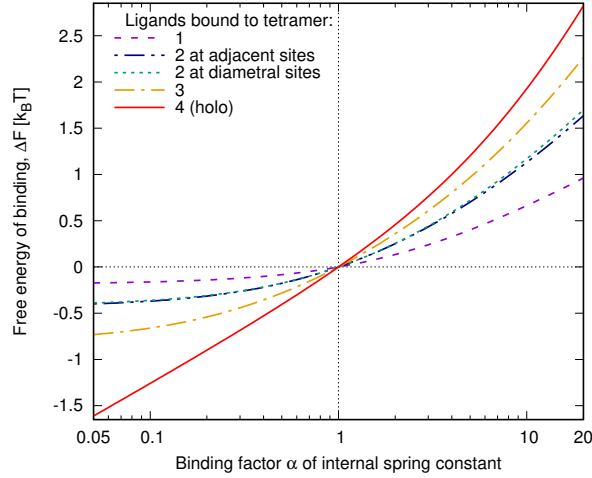


FIG. 5. (Colour online) Tetramer ring, one-parameter binding model: Binding free energies $\Delta F(\alpha)$ for $K_c = 5$.

stiffening or weakening the respective internal modes, are entropically slightly less favourable than two ligands at adjacent sites. If we recall the plausibility argument for cooperativity on a trimer, via entropy-cost transfer between nearest neighbours through local change of fluctuations, the ordering of the ΔF agrees with intuition. As in the trimer case, calculations based on the *one*-parameter model of binding predict only *positive* allostery.

Already these minimal examples of cyclic coupling demonstrate nontrivial allosteric interactions of subtle parameter dependence between units with binding sites.

IV. DYNAMIC ALLOSTERY GENERATES TUNED WEAK ATTRACTION

Weak association of proteins is determined by a balance of the attractive forces operative at the mutual surfaces of interaction, and entropic repulsion. The attractive forces may have for their origin van-der-Waals, hydrophobic or screened electrostatic effects. The entropic repulsion arises from the penalty that their association generates from the constraints it places on the amplitude of internal structural fluctuations. A general question of interest is whether this delicate balance can be ‘tuned’ in different ways. This might allow distinct families of weakly-associating nanoparticles (or proteins in the biological case) which bind reversibly among themselves, but which do not associate with other, differently-tuned, families. Here, we use the simple ‘allosteron’ model, with both coupling spring constants and enthalpic surface forces derived from Lennard-Jones potentials, to explore the possibility for such entropic tuning.

Starting from the entropic free energy of binding, ΔF , given in Eq. (5), and the total free energy $\Delta F_{\text{bind}} = \Delta F + \Delta U$, we construct a simple model for both the enthalpic term, ΔU , and the spring constant κ_c . As a general model for weakly attractive interactions between protein surfaces we choose the Lennard-Jones potential

$$U(r) = \frac{A}{r^{12}} - \frac{B}{r^6} \quad (14)$$

One of the advantages of a physics-based phenomenological model such as the L-J potential is that it is able to illustrate and work with the correlation between the enthalpic binding strength $\Delta U = \min[U(r)] = U(r_{\min})$ and $\kappa_c = U''(r_{\min})$. Using the parameterised form of the interaction gives

$$\begin{aligned} \Delta U &= -\frac{B^2}{4A} & \kappa_c &= U''(r_{\min}) = \gamma \frac{B^{7/3}}{A^{4/3}} \\ & & \text{with } \gamma &= \left[\frac{156}{2^{7/3}} - \frac{42}{2^{4/3}} \right] \simeq 14.3 \end{aligned} \quad (15)$$

Combining these allows a single-degree-of-freedom parameterisation of the set of possible L-J couplings, using ΔU as the free variable, and finding:

$$\kappa_c = 4^{4/3} \gamma B^{-1/3} (\Delta U)^{4/3} \quad (16)$$

The dimensionless free energy at the potential minimum $f = \Delta F_{\text{bind}}/k_B T$ can be written in terms of a dimensionless potential energy minimum $x = U(r_{\min})/k_B T$ as

$$f(x) = \ln[1 + ax^{4/3}] - x \quad (17)$$

Fig. 6 plots this function for various values of the renormalised coupling constant, a . Note

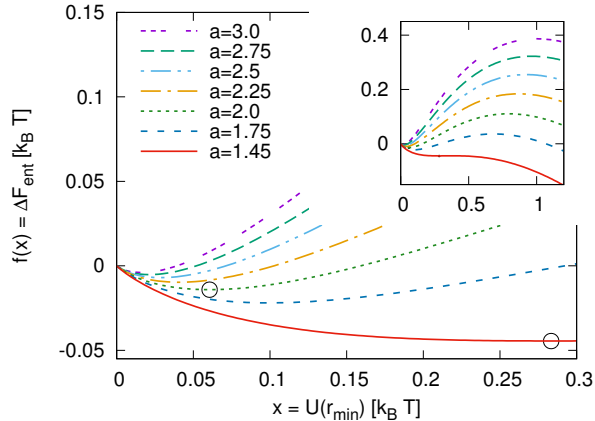


FIG. 6. (Colour online) Dimensionless free-energy function $f(x)$ from Eq. (17), for different values of a .

that the asymptotic form decreases linearly with x as it must from the dominance of the enthalpic interaction, but there exist other values of x , corresponding to a ‘tuning’ of the L-J coupling to the inner elastic strength of the proteins (the κ_i), which lead to a family of shallow local minima. The depth and value of this minimum increases with decreasing a (see Fig. 6). There is a lower critical value of $a \simeq 1.4$ below which the minimum disappears.

A word is necessary on the values of the interaction strengths, which are of course very small (of order $k_B T/50$). However, this corresponds to the response of a single dynamical mode of internal motion only. In practice, there will be many internal modes that are coupled to the surface interaction, sufficient to create an associative energy of the order of $k_B T$. Should two proteins from different families attract, the mutual free energy of attraction will have the same form as the entropy/enthalpy function within a single family, but will not be at the tuned minimum potential. This permits the emergence of interprotein potentials that generate weakly associated clusters for proteins within ‘tuned’ families exclusively.

V. DISCUSSION

The allosteron model constitutes the maximal level of coarse-graining within models for protein dynamics that still captures the physics of thermally excited modes of deformation within proteins and their complexes. In spite of this level of simplicity, the model is never-

theless able to make non-trivial predictions about the contribution of structural fluctuations to the statistical mechanics of complexation, in terms of the ratio of coupling to internal interactions K_c .

When applied as a building block of supramolecular polymers, the allosteron model is not only capable of capturing cooperativity in the usual way, but also predicts entropic allosteric signalling along the backbone of the chain. When a monomer associates to the chain, a number of $\sqrt{K_c}$ monomers near the chain end are entropically penalised, which causes an increase of the polymerisation concentration by a factor $\sqrt{K_c}$. We find that a reduction of the polymerisation concentration may be triggered through ‘activation’ of the monomers, e.g., by stimuli such as ligand binding that modify the coupling constant to $K_c \mapsto K_c/\alpha$.

We have also investigated ligand binding to allosteron complexes that represent ring-type proteins. In this case, binding is assumed to modify the strengths of both internal and coupling interactions and predicts a palette of cooperative binding behaviour even for small rings. In particular, maps of the allosteric free energy for two binding events in the parameter space of the two spring modifiers show regions of negative allostery and an intriguing non-monotonic dependence. Further studies along these lines might extend to inter-ring coupling, possibly adding to the understanding of allostery without conformational change reported for chaperonins [30].

When an allosteron coupling model is derived from a specific model for intermolecular potentials, we find the possibility to ‘tune’ the surface attraction and the internal elasticity of soft nanoparticles in general, and proteins in particular. Simultaneous tuning of the dimensionless interparticle coupling K_c and the enthalpy of binding creates ‘families’ of particles that possess weak mutual associativity (and thereby potential allosteric activity of other, enzymatic, kinds).

The application of the allosteron model to the three cases of small, specifically-bound clusters, large specifically-bound fibrils, and non-specific binding have all revealed subtle phenomena that are open to experimental investigation.

ACKNOWLEDGMENTS

The authors acknowledge the Engineering and Physical Sciences Research Council [grant number EP/N031431/1] for funding of this project.

- [1] A. Cooper and D. T. F. Dryden, “Allostery without conformational change: A plausible model,” *Eur. Biophys. J.* **11**, 103–109 (1984).
- [2] R. J. Hawkins and T. C. B. McLeish, “Coarse-grained model of entropic allostery,” *Phys. Rev. Lett.* **93**, 098104 (4pp) (2004).
- [3] R. J. Hawkins and T. C. B. McLeish, “Coupling of global and local vibrational modes in dynamic allostery of proteins,” *Biophys. J.* **91**, 2055–2062 (2006).
- [4] Hedvika Toncrova and Tom C. B. McLeish, “Substrate-modulated thermal fluctuations affect long-range allosteric signaling in protein homodimers: Exemplified in CAP,” *Biophys. J.* **98**, 2317–2326 (2010).
- [5] T. C. B. McLeish, T. L. Rodgers, and M. R. Wilson, “Allostery without conformation change: modelling protein dynamics at multiple scales,” *Phys. Biol.* **10**, 056004 (9pp) (2013).
- [6] V. J. Hilser, J. O. Wrabl, and H. N. Motlagh, “Structural and energetic basis of allostery,” *Annu. Rev. Biophys.* **41**, 585–609 (2012).
- [7] S. R. Tzeng and C. G. Kalodimos, “Protein activity regulation by conformational entropy,” *Nature* **488**, 236–240 (2012).
- [8] T. L. Rodgers, Philip D. Townsend, David Burnell, Matthew L. Jones, Shane A. Richards, Tom C. B. McLeish, Ehmke Pohl, Mark R. Wilson, and Martin J. Cann, “Modulation of global low-frequency motions underlies allosteric regulation: Demonstration in CRP/FNR family transcription factors,” *PLoS Biol* **11**, e1001651 (2013).
- [9] T. C. B. McLeish, “Slow Brownian fluctuations for allosteric signalling without structural change,” in *Royal Society Discussion Meeting ‘Allostery and Molecular Machines’, London* (2017).
- [10] F. Tama and Y.-H. Sanejouand, “Conformational change of proteins arising from normal mode calculations,” *Protein Eng. Des. Sel.* **14**, 1–6 (2001).
- [11] M. M. Tirion, “Large Amplitude Elastic Motions in Proteins from a Single-Parameter, Atomic

- Analysis,” Phys. Rev. Lett. **77**, 1905 (4pp) (1996).
- [12] I. Bahar, A. R. Atilgan, M. C. Demirel, and B. Erman, “Vibrational dynamics of folded proteins: Significance of slow and fast motions in relation to function and stability,” Phys. Rev. Lett. **80**, 2733–2736 (1998).
 - [13] T. L. Rodgers, D. Burnell, P. D. Townsend, E. Pohl, M. J. Cann, M. R. Wilson, and T. C. B. McLeish, “ $\Delta\Delta$ PT: a comprehensive toolbox for the analysis of protein motion,” BMC Bioinformatics **14**, 183 (9pp) (2013).
 - [14] Tom F. A. de Greef, Maarten M. J. Smulders, Martin Wolffs, Albert P. H. J. Schenning, Rint P. Sijbesma, and E. W. Meijer, “Supramolecular polymerization,” Chem. Rev. **109**, 5687–5754 (2009).
 - [15] Christina Rest, Ramesh Kandanelli, and Gustavo Fernández, “Strategies to create hierarchical self-assembled structures via cooperative non-covalent interactions,” Chem. Soc. Rev. **44**, 2543–2572 (2015).
 - [16] Pol Besenius, “Controlling supramolecular polymerization through multicomponent self-assembly,” J. Polymer Sci. A: Polymer Chem. **55**, 34–78 (2017).
 - [17] Robert F. Goldstein and Lubert Stryer, “Cooperative polymerization reactions: Analytical approximations, numerical examples, and experimental strategy,” Biophys. J. **50**, 583–599 (1986).
 - [18] C. Schaefer, I.K. Voets, A.R.A. Palmans, E.W. Meijer, P. van der Schoot, and P. Besenius, “Controlling the cooperativity in the supramolecular polymerization of ionic discotic amphiphiles via electrostatic screening,” ACS Macro Lett. **1**, 830–833 (2012).
 - [19] Chidambar Kulkarni, E. W. Meijer, and Anja R. A. Palmans, “Cooperativity scale: A structure-mechanism correlation in the self-assembly of benzene-1,3,5-tricarboxamides,” Acc. Chem. Res. **50**, 1928–1936 (2017).
 - [20] L. Saroleá-Mathot, “Thermodynamic and spectroscopic properties of associated solutions,” Trans. Faraday Soc. **49**, 8 (1953).
 - [21] Paul van der Schoot, *Supramolecular polymers*, 2nd ed., edited by A. Ciferri (CRC Press, Boca Raton, 2005).
 - [22] Paul van der Schoot, *Advances in chemical engineering*, edited by Rudy J. Koopmans, Vol. 35 (Elsevier, Amsterdam, 2009).
 - [23] Sara Jabbari-Farouji and Paul van der Schoot, “Theory of supramolecular co-polymerization

- in a two-component system,” J. Chem. Phys. **137**, 064906 (2012).
- [24] C. Bohr, K. Hasselbach, and A. Krogh, “Ueber einen in biologischer Beziehung wichtigen Einfluss, den die Kohlensäurespannung des Blutes auf dessen Sauerstoffbindung übt,” Skand. Arch. Physiol. **15**, 401–412 (1904).
 - [25] C. Ciaccio, A. Coletta, G. De Sanctis, S. Marini, and M. Coletta, “Cooperativity and Allostery in Haemoglobin Function,” IUBMB Life **60**, 112–123 (2008).
 - [26] W. A. Eaton, E. R. Henry, J. Hofrichter, and A. Mozarelli, “Is cooperative oxygen binding by hemoglobin really understood?” Nat. Struct. Biol. **6**, 351–358 (1999).
 - [27] I. Bahar, T. R. Lezon, A. Bakan, and I. H. Shrivastava, “Normal mode analysis of biomolecular structures: functional mechanisms of membrane proteins,” Chem. Rev. **110**, 1463 (2010).
 - [28] R. J. Hawkins, *Coarse-grained models of dynamic allostery in proteins*, Ph.D. thesis, University of Leeds (2005).
 - [29] H. Toncova, *Coarse-grained models of biomolecule dynamics and allostery*, Ph.D. thesis, University of Leeds (2010).
 - [30] W. Zheng, B. R. Brooks, and D. Thirumalai, “Allosteric Transitions in the Chaperonin GroEL are Captured by a Dominant Normal Mode that is Most Robust to Sequence Variations,” Biophys. J. **93**, 2289–2299 (2007).