Supporting Information

Microscopic and Macroscopic Properties of Carbohydrate Solutions in the Ionic Liquid 1-ethyl-3-methyl-imidazolium acetate

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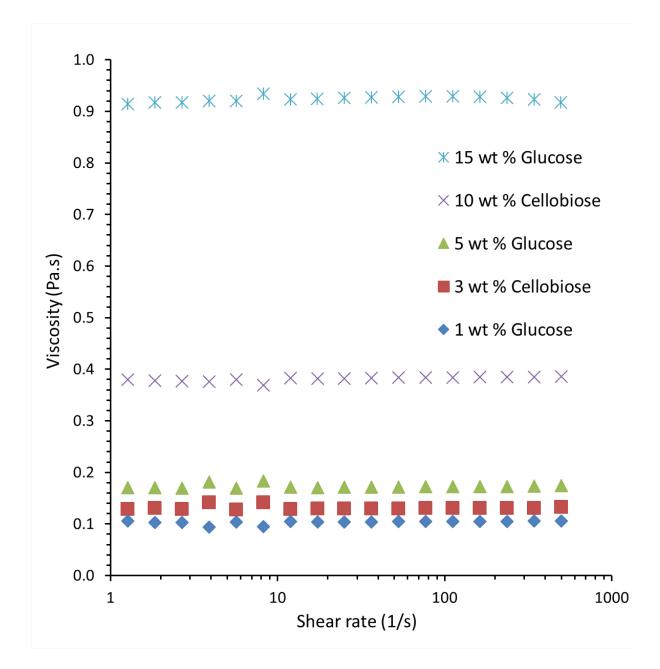


Figure S1. Viscosity against shear rate of carbohydrate-[C2mim][OAc] solutions at different concentrations of carbohydrates at 30 °C. Error bars are within the size of the symbols shown.

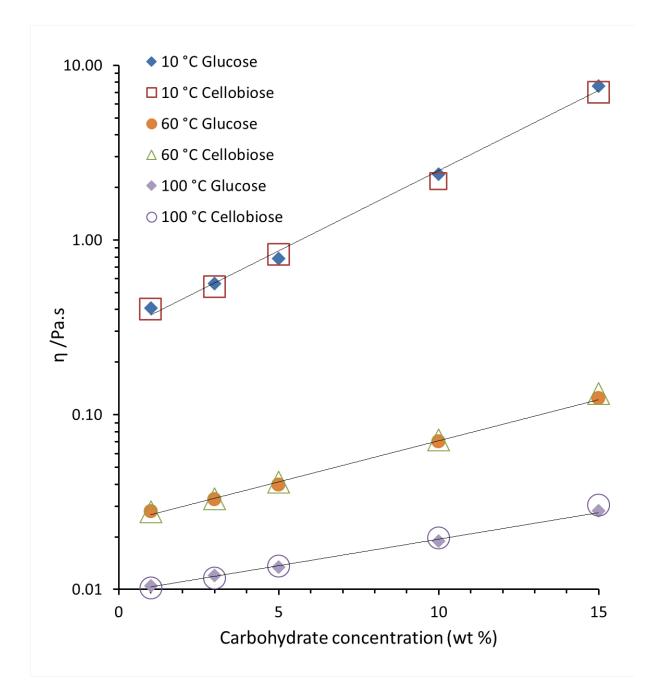


Figure S2. Viscosity at 10 °C, 60 °C and 100 °C for the glucose and cellobiose in [C2mim][OAc] solutions, against carbohydrate concentration. The straight lines are guides to the eye. Error bars are within the size of the symbols shown.

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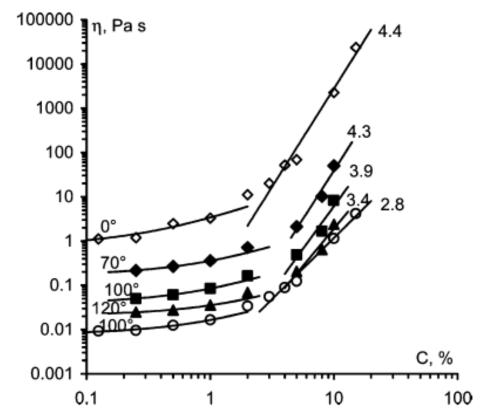


Figure 3. Viscosity-concentration dependence for cellulose-BMIMCl (dark symbols) and cellulose-EMIMAc (open symbols) solutions at various temperatures shown directly at the corresponding data. Lines are linear (in the dilute region) and power-law (in the semidilute region) approximations. Power-law exponents are shown for each set of data. The errors are of the size of symbols.

Figure S3. Viscosity of cellulose [C2mim][OAc] solutions against cellulose concentration. Reprinted with permission from {Journal of Physical Chemistry B, volume 114 page 7224-7228, 2010}. Copyright {2010} American Chemical Society.

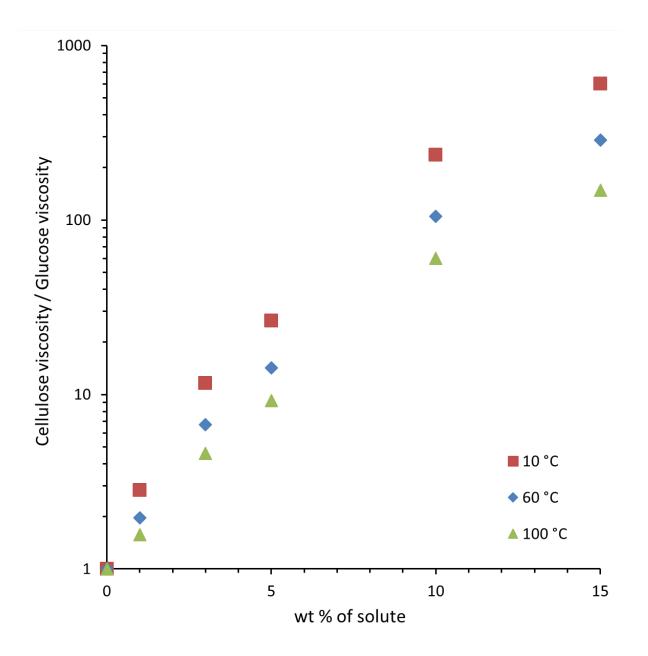


Figure S4. Viscosity ratio of cellulose to glucose in [C2mim][OAc] solutions (at the same solute concentrations), against solute carbohydrate concentration.

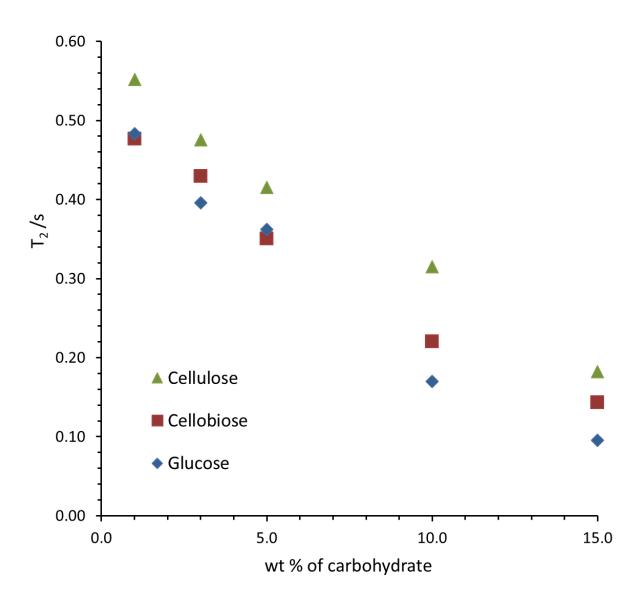


Figure S5. NMR spin-spin relaxation times T_2 for glucose, cellobiose and cellulose as a function of the wt % of carbohydrate in [C2mim][OAc] solutions at 70 °C. Uncertainties are within the size of the symbols used.