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1 2 3	Deformation behavior of migmatites: Insights from microstructural analysis of a garnet- sillimanite-mullite-quartz-feldspar bearing anatectic migmatite at Rampura-Agucha, Aravalli Delhi Fold Belt, NW India
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31 ABSTRACT

32 In the present study we investigate the microstructural development in mullite, quartz and garnet in an anatectic migmatite hosted within a Grenvillian-age shear zone in the Aravalli-33 Delhi Fold Belt. The migmatite exhibits three main deformation structures and fabrics (S_1, S_2, S_3) 34 S_{3} . Elongated garnet porphyroblasts are aligned parallel to the metatexite S_2 layers and contain 35 crenulation hinges defined by biotite-sillimanite-mullite-quartz (with S₁ axial planar foliation). 36 Microstructural evidence and phase equilibrium relations establish the garnet as a peritectic 37 phase of incongruent melting by breakdown of biotite, sillimanite \pm mullite and quartz at peak 38 P-T of ~8 kbar, 730°C along a tight-loop, clockwise P-T path. Monazite dating establishes that 39 the partial melting occurred between ~1000-870 Ma. The absence of subgrains and systematic 40 crystal lattice distortions in these garnets despite their elongation suggest growth 41 pseudomorphing preexisting 3-D networks of S₁ biotite aggregates rather than high temperature 42 43 crystal plastic deformation which is noted in the S₁ quartz grains that exhibit strong crystallographic preferred orientation (CPO), undulatory extinction and subgrains. Mode-I 44 45 fractures in these garnet porphyroblasts induced by high melt pressure during late stage of partial melt crystallization are filled by retrograde biotite-sillimanite. Weak CPO and non-46 systematic crystal lattice distortions in the coarse quartz grains within the S₂ leucosome 47 domains indicate these crystallized during melt solidification without later crystal plastic 48 deformation overprint. In the later stages of deformation (D₃), strain was mostly accommodated 49 in the mullite-biotite-sillimanite rich restite domains that forming S₃ which warps around 50 garnet and leucosome domains; consequently fine-grained S₃ quartz does not exhibit strong 51 CPOs. 52

53 (252)

54 Keywords

Melt-present deformation; anatectic migmatites; crystal preferred orientation; quartz; garnet;
rheology.

57 INTRODUCTION

Anatectic migmatites record signatures of partial melting events in mid- to deep-crustal levels 58 at the roots of orogenic belts (Ashworth 1985; Brown 2001; Whitney et al. 2004; Beaumont et 59 al. 2006). The restitic parts preserved in many migmatites are archives of the geological history 60 that pre-dates the melting event (Guernina and Sawyer 2003; Sawyer 2008). Both experimental 61 62 and theoretical work demonstrate that the dynamic viscosity of partially molten rocks is 63 reduced by several orders of magnitude depending on the volume fraction of melt (Arzi 1978; Vigneresse et al. 1996; Berger and Kalt 1999; Takeda and Obata 2003). Therefore, in 64 situations, where the production, segregation and emplacement of melts are synchronous with 65 66 active deformation, strain is likely to be complexly partitioned into low-viscosity melt 67 dominated layers within the deforming matrix of the bulk rock. However, in such experiments strain rates and temperatures are unrealistic thus experimental results may not be directly 68 69 applied to study deformation patterns in migmatites (Paterson 1987; Rutter and Neumann 1995). Consequently, detailed studies of natural migmatites are crucial. 70

71 Deformation microstructures in minerals in anatectic migmatites in melt-dominated and melt-absent domains can be determined from quantitative orientation analysis including 72 73 internal deformation features (e.g. subgrains) and their CPOs using electron backscatter 74 diffraction (EBSD) studies (Venables and Harland 1973; Prior et al. 1999; Kleinschrodt and Duyester 2002; Ji et al. 2003; Mainprice et al. 2004; Storey and Prior 2005; Hasalová et al. 75 2008; Menegon et al. 2011; Cavalcante et al. 2013). The results of such EBSD studies may be 76 77 combined with quantitative phase equilibrium modeling of melting reactions in migmatites to understand how naturally-occurring minerals respond to far-field stress before, at and after 78 supra-solidus conditions. In this study, we integrate the results of phase equilibrium modelling 79

80 of anatectic migmatites using P-T pseudosection modelling and EBSD studies in a garnetbiotite-mullite migmatite of the Bhilwara Supergroup (Fig. 1a) from a Grenvillian-age, 81 Northern Indian shear zone to demonstrate the complex interplay between deformation strain, 82 83 mineralogical reactions and crystallization microstructures developed in quartz, mullite and garnet at subsolidus and supra-solidus condition. 84

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GEOLOGICAL BACKGROUND AND FIELD RELATIONSHIPS

The chosen field area lies in the Bhilwara Supergroup of Northern India (Fig. 1a). The Bhilwara 87 88 Supergroup rocks are exposed along the central and eastern parts of Aravalli-Delhi Fold Belt (Raja Rao 1976; Gupta et al. 1980; 1997, Fig. 1b). Two distinct NNE-SSW trending belts 89 90 constitute the Bhilwara Supergroup: (i) the western Meso- to Neoproterozoic amphibolite 91 facies schist belt that extends from Rajpura-Dariba in the south-west to Pur-Banera in the northeast and the (ii) eastern migmatite belt that extends up to the volcano-sedimentary successions 92 of the Hindoli Group and Jahazpur Belt in the east, the granitoids and metasedimentary units 93 94 of the North Delhi Fold Belt in the north (Fig. 1b; Gupta et al. 1997). The western schist belt consists of an interlayered sequence of garnet-staurolite-kyanite mica schists, calc-silicate 95 gneisses and quartzites (Hazarika et al. 2013; Ojha et al. 2016). The garnet-sillimanite-biotite 96 gneiss from which the sample studied stems forms the dominant lithology of the eastern 97 98 anatectic migmatite belt and is associated with porphyritic granitoid plutons (Hazarika et al. 99 2013). The Archaean (~2.5 Ga) Berach granite (also called Banded Gneissic Complex, BGC-I; Heron 1953) and the ~Palaeoproterozoic greenschist facies metasedimentary units of the 100 Aravalli Supergroup defines the southern margin of Bhilwara Supergroup. Along their western 101 102 margin, the migmatites of Bhilwara Supergroup share a NNE trending tectonic contact (marked by ductile shear zones) with the polycyclic Meso- to Neoproterozoic granulites of the 103 Mangalwar Complex and Sandmata Complex (together forming BGC-II, Heron 1953). Raja 104

Rao (1976) and Gupta et al. (1980, 1997) propose the BGC-I to represent the basement for theBhilwara Supergroup.

Geochronological data from the Bhilwara Supergroup are sparse. A single zircon age 107 108 of ~1.45 Ga is reported from the mylonitized porphyritic granitoid in Rampura-Agucha (Roy and Jhakar 2002). A Pb-Pb model age of ~1.8 Ga is obtained by Deb and Thorpe (2004) and 109 Deb et al. (1989) for the anatectic metapelites in Rampura-Agucha. Based on U-Th-Pb (total) 110 111 age determinations, Hazarika et al. (2013) obtained a single age population of ~1.0 Ga in monazites in the pelitic migmatites of the area. However, multiple monazite age populations 112 113 have been obtained from the garnet-staurolite-kyanite mica schists of the Rajpura-Dariba interpreted to date peak and retrograde metamorphic events (1.87 and 1.62 Ga; Hazarika et al. 114 2013) and Pur-Banera (1.37 and 1.05 Ga; Ojha et al. 2016). 115

116 The field area at Rampura-Agucha lies close to the north-western margin of the migmatite belt (Fig. 1b). It is characterized by an ensemble of calc-silicate gneisses, 117 amphibolites and anatectic garnet-biotite-sillimanite gneisses and mylonitized porphyritic 118 granitoids (Fig. 1c). Three fabric-forming events are recognized in the garnet-biotite-119 sillimanite migmatites (Fig. 2a). We use a numbering system in accordance to relative age 120 established in the field. The S₁ fabric is noted exclusively as inclusion trails within garnet 121 porphyroblasts (Table 1; Fig. 3a). The first set of mesoscopic fabric (S₂) is defined by garnet-122 123 bearing leucosomes in a mesocratic matrix dominated by biotite intergrown with sillimanite, 124 and coarse-grained quartz lenticles. The metatexite S₂ layers are co-axially refolded (hookshaped fold superposition structures are common) by isoclinal D₃ folds and locally developed 125 NNE-trending tight to open D₄ folds that plunge towards NNE (Fig. 2a). The axial planar fabric 126 127 (S_3) is mutually indistinguishable from S_2 , except at D_3 fold hinges (Figs 2b-c). In contrast, D_4 folding does not result in a recognizable fabric. 128

We selected a representative sample of the neosome part of the metatexite migmatite from a limb of a D_3 fold, with typical garnet bearing leucosome and restitic fine-grained layers warping around the leucosome layers (Figs 2b, c). No D_4 deformation features are seen at the sample location.

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134 ANALYTICAL METHODS

135 Mineral chemistry and bulk chemistry

Mineral–chemical analyses of an anatectic garnet-biotite-sillimanite gneiss (RAM-1; Fig 2b,
c) were carried out with a CAMECA SX (EPMA) at Indian School of Mines, Dhanbad.
Operating conditions for feldspar analyses were acceteration voltage 15 kV with 15 nA beam
current, and 15 kV, 20 nA were used for biotite, garnet, sillimanite and ilmenite. In built ZAF
corrections were applied.

For pseudosection modeling, the bulk composition for the sample RAM-1 was
determined by X-ray fluoresence at the National Geophysical Research Institute, Hyderabad.

143 The analysis was done using a Phillips MagiX PRO Model 2440 XRF spectrometer.

144 Quantitative microstructural analysis and representation

145 Electron back-scatter diffraction (EBSD) analyses was done at Macquarie University on thin sections (sample RAM-1) cut perpendicular to S₂ foliation (Y-X plane) and parallel to the fold 146 axis lineation of F₃ (X-direction). Thin sections were carbon coated (~3 nm thick) following 147 mechano-chemical polishing using colloaidal silica. Full crystallographic orientation data 148 were collected using automatically indexed EBSD patterns, acquired with an HKL Nordlys 149 150 Nano high sensitivity EBSD detector and indexed using the Aztec analysis software (Oxford Instruments). The SEM was run at a high vacuum, with an accelerating voltage of 20kV, beam 151 current of 8.2 nA, and with an aperture of 70 µm at a working distance of 9.5 -10.5 mm. Data 152 153 was collected on rectangular grids with step size of 5-25 µm -(depending on the required spatial resolution of the crystallographic information) using a beam scan. Noise reduction on the raw data was performed following the procedure tested by Piazolo et al. (2006), Bestmann and Prior (2003) and Prior et al. (2002). In pole figures, data is represented as one point per grain, where a grain is defined as an area completely surrounded by high angle (>10°) boundaries. Grains which have a mean internal misorientation <1° are considered to be strain free (Piazolo et al. 2006).

160 Monazite analyses

Th-U-Pb (total) age determinations in monazites were computed from analytical data obtained 161 using the Cameca SX electron microprobe analyzer at the Indian School of Mines, Dhanbad 162 following the "age calibration" (Petrík and Konečný 2009). These age data are then corrected 163 against an age standard dated by SIMS method (Tomasck et al. 1998). Analytical settings were 164 set as follows: counting times for Pb 150 s, Th 45 s, U 75 s, Y 45 s, and for all other elements 165 166 25-35 s; sample current 100-130 nA; beam diameter ~3 µm. Standards used were the following: P - Apatite, Ca - Wollastonite, La- La Glass, Ce- Ce Glass, Eu-Eu Glass, Si- Albite, 167 Y-YAG, Al- Kyanite, Pb- Crocoite, Th-Thorium Glass, U- Uranium Glass, Pr-Pr Glass, Nd-168 Nd Glass, Sm- Sm Glass, Gd-Gd Glass, Dy -Dy Glass, Lu- Lu Glass. All errors are given as 169 2σ. Deconvolution histograms were calculated using Isoplot 4.15 (Ludwig 2001). 170

171 **RESULTS**

172 General microstructure phase relationships and mineral chemistry

173 S_2 foliation is defined by alternate melanocratic bands consisting of garnet porphyroblasts, 174 biotite (Bt₂), quartz (Qtz₂) and leucocratic bands consisting of quartz (Qtz₂), plagioclase (Pl₂), 175 potash feldspar (Kfs₂; Fig. 3a). Mineralogically similar leucocratic domains are also recognized 176 in the pressure shadows of the garnet porphyroblasts (Table 1; Fig. 3a). Garnet porphyroblasts 177 in the melanocratic bands of the penetrative S₂ metatexite layers are often elongated with aspect

ratios of 1.75 to 3.1 (Figs 3c, 4). Often bulbous and lobate coarse, strain free quartz (Qtz₂) 178 grains are intergrown with garnets. Fractures within garnet porphyroblasts are oriented 179 perpendicular to porphyroblast elongation (Table 1; Fig. 3c). These fractures in garnet are filled 180 with biotite (Fig. 3c) or biotite-mullite-sillimanite (Fig. 3a). In garnet porphyroblasts D_1 181 crenulation hinges are mainly defined by mullite (Mul₁) and sillimanite (Sil₁) (Figs 3a, b, 4). 182 Inclusion trails of quartz (Qtz_1) ribbons and resititic biotite (Bt_1) grains are axial planar to these 183 184 crenulations. Qtz₁ from the relict D₁ hinges in the matrix exhibit undulose extinction and distinct subgrains (Figs 3h-i). Rootless crenulation hinges (defined by Bt₁+Sil₁+Mul₁), that 185 186 formed during D₁ with orientation similar to those within garnet porphyroblasts also occur in the interfolial domains of S₃ (Fig. 3a). Coarse potash feldspar (Kfs₂) grains in the leucocratic 187 layers of S₂ contain sillimanite (Sil₁), biotite (Bt₁) and quartz (Qtz₁) grains (Table 1; Figs 3d, 188 189 e). Quartz (Qtz₁) inclusions in these potash feldspar (Kfs₂) grains are often rounded and form 190 rims around biotite (Bt₁) inclusions (Fig. 3e). In leucocratic domains, plagioclase (Pl₂) with low dihedral angles occurs at the margins of coarse potash feldspar grains (Fig. 3f). The 191 pervasive foliation in the sample is S_3 which is axial planar to the D_3 folds and transposes S_2 . 192 S₃ is defined by the shape-preferred aggregates of biotite (Bt₃), sillimanite (Sil₃) and mullite 193 (Mul₃) that warp around garnet porphyroblasts in the melanocratic bands of S₂ (Figs 3a, b, 4 194 Table 1). In these S₃ domains, quartz is fine-grained. The interfolial domains of S₃ consist of 195 196 the quartz-feldspar dominated S₂ leucosomes (Figs 3a, 4), with modally subordinate biotite, 197 Bt₂ (Fig. 3a). To distinguish between the Bt+Sil+Mul assemblages in the S₃ foliation from those seen in garnet fractures, we label the latter as Bt₄+Sil₄+Mul₄ (Figs 3a, 3c, 3g). 198

199 X_{Mg} in the cores of garnet porphyroblasts are tightly constrained between 0.19 and 0.20 200 (Table 2). In contrast, the corresponding values at the rims of the garnet grains (~0.15-0.17) 201 against S₃ biotite (Bt₃) and biotites in the fractures are variable, but against quartz grains, the 202 values are identical. Biotite inclusions in garnet (Bt₁) and matrix biotite (Bt₂) in the S₂ foliation show near-identical X_{Mg} (0.54 for one Bt₁ analyses and 0.55-0.56 for Bt₂) and TiO₂ contents (2.11 pfu for Bt₁ and 1.74-1.99 pfu for Bt₂; Table 2). X_{Mg} and TiO₂ values of coarse biotite grains (Bt₃) from the S₃ foliation range from 0.55-0.58 and 1.21-2.13 pfu, respectively. The X_{Mg} values of Bt₃ are similar to that of the Bt₁ and Bt₂, although their TiO₂ contents show significant variations. Biotites filling the fractures in garnets (Bt₄) are richer in Mg than Bt₁, Bt₂ and Bt₃. TiO₂ content of biotites in garnet fractures (Bt₄) is in the range 0.1-0.89 pfu (Table 2).

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211 Mineral paragenetic relationships: Their interpretation in terms of succession of 212 geological events

The succession of mineral assemblages that is stable with quartz and plagioclase is shown 213 214 schematically in Figure 4. The warping of Bt₃+Sil₃+Mul₃ aggregates around garnet porphyroblasts and S₃ continuity indicates pre-D₃ garnet (Fig. 3a) growth and fracturing. 215 Coarse K-feldspar (Kfs₂) grains in the leucosomes rimming the garnet porphyroblasts, and 216 those occurring in S_3 interfolial domains, host biotite (Bt₁), sillimanite (Sil₁), and quartz (Qtz₁) 217 inclusions, indicating K-feldspar and garnet crystallized from melt (Figs 3d-e). Bt₂ at the 218 contacts of garnets with globular quartz (Qtz_2) grains and with the leucocratic layers of S_2 are 219 also observed (Figs 3a, 3g). The occurrence of garnet within leucocratic domains indicate 220 221 garnet formation was syn-D₂. Relict sillimanite (Sil₁)-mullite (Mul₁)-biotite (Bt₁) bearing 222 crenulation hinges within garnet porphyroblasts, and in the S₃ interfolial domains indicates that 223 these are D_1 deformation features overgrown by D_2 garnet porphyroblasts. The presence of thin films of plagioclase (Pl₂) around coarse K-feldspar (Kfs₂) and plagioclase (Pl_{1/2}) in the 224 225 leucosome and the coarse quartz (Qtz_2) grains intergrown with garnet corroborate anatexis synchronous with D₂ (Figs 3e-f). Coarse quartz (Qtz₂) and plagioclase (Pl₂) within garnet grains 226 227 in Fig. 3j are interpreted to have crystallised from trapped melt.

Bt₄+Sil₄+Mul₄ assemblages present in garnet fractures or Bt₄ at the contact with garnet 228 margins and at the contacts of garnet and Qtz_2 are interpreted to represent the retrograde 229 assemblage generated by back reactions of melt in garnet fractures and boundaries. Similar 230 231 TiO₂ content (1.77 pfu) noted in one such biotite grains (Table 2) with some of the biotite grains (Bt₃; 1.74-1.88 pfu) from the S₃ layers indicate that they crystallized at comparable 232 temperatures. Hence we conclude that retrogression along garnet fractures is at the vaining 233 234 stages of D₂. The lower TiO₂ (0.1-0.59 pfu) and higher X_{Mg} (0.63-0.67) contents in most of the biotite grains seen in garnet fractures compared to the Bt₃ suggest that biotite crystallization 235 236 continued to lower temperatures after the melt had largely crystallised and solid state deformation occurred in S₃ fabric. 237

238 Constraints on P-T conditions of melting

239 A key factor in pseudosection analyses is to determine the 'effective bulk composition', which represents the equilibration volume of the rock (Spear 1993; Stüwe 1997). This 240 equilibration volume varies from grain scale e.g. zoned minerals and reaction coronae 241 (Nasipuri et al., 2009) to larger scales, i.e. in hand specimen scale (Stuwe 2007; Kelsey et al. 242 2005, Nasipuri et al. 2009). In the studied migmatite sample RAM-1, garnet, biotite, and 243 plagioclase in different layers show strong compositional homogeneity (Table 2). This suggests 244 that the length scale of mineral equilibration attained during metamorphism was larger than the 245 246 scale of the layers of mineralogical segregations (Figs 2b-c); the equilibration was possibly 247 aided by element transfer across neosomes of varying width (couple of mm to cm) which are seen to be commonly inter-layered and dispersed in biotire+sillimanite segregations. 248 Consequently, for pseudosection analyses we use the bulk rock composition derived from a 249 250 handspecimen to define the effective bulk composition (results given in Table 4).

P-T pseudosections were constructed (Table 4) in the NCKFMASH system using the
PERPLE_X (Connolly 2005) software built on the thermodynamic database of Holland and

253 Powell (1998,) modified in 2004. Manganese was excluded from the system as the spessartine contents in garnet is extremely low, and manganese is absent in other Fe-Mg phases (Table 2). 254 The TiO₂ content in the rock is low and, therefore, also excluded in the computation. Solution 255 256 models for phases used in P-T pseudosection modeling are: (i) garnet: hybrid model of Holland and Powell (1998), Engi and Wersin (1987); (ii) biotite: White et al. (2007), (iii) muscovite: 257 hybrid model of Coggon and Holland (2002) and Auzanneau et al. (2010); (iv) feldspar: 258 Benisek et al. (2004); (v) melt: hybrid model of Holland and Powell (2001) and White et al. 259 (2001). 260

261 Melt volume percentages depend on the availability of free H₂O during initiation and progress of partial melting reactions (Webb et al. 2015). H₂O influencing partial melting 262 reactions is primarily produced by dehydration reactions in surrounding rocks that may 263 264 eventually get trapped in the rocks or added to the rock along shear zones (Brown 2002; White and Powell 2002; 2010; Webb et al. 2015). In order to estimate the H₂O available in the rock 265 during partial melting under both H₂O-fluxed and H₂O-absent conditions, we constructed 266 267 isobaric T–M_{H2O} pseudosections (Appendix 1a and b; M_{H2O} representing mol% of water in the bulk). We chose reference pressures of 6 and 8 kbar based on the 6-8 kbar pressure reported 268 for the Grenvillian-age peak metamorphic condition obtained in mica schists of the Bhilwara 269 Supergroup rocks from the Pu-Banera area (Ojha et al. 2016). The T-M_{H2O} pseudosections 270 271 indicate stability of garnet-sillimanite-feldspar (ternary)-quartz-biotite-melt assemblage 272 during H₂O-deficient melting at temperature ranges of 690-840°C, and during H₂O-fluxed melting at 690-750°C. H₂O contents for H₂O-undersaturated and water-fluxed conditions 273 partial melting, as estimated from the T-M_{H2O} pseudosections have been considered for 274 275 construction of the P-T pseudosections in Figs. 5a-d.

The P-T pseudosections show stability of the garnet–sillimanite–feldspar–quartz– biotite–melt peak metamorphic assemblage in P–T ranges of 5–8 kbar, 720–740°C during

H₂O-fluxed melting and ~780-850°C, 5-10 kbar at H₂O-absent melting (Figs 5a, 5e). In both 278 cases, garnet modal isopleths increase with temperature and increasing melt modes and 279 decreasing biotite and sillimanite modes thus indicating formation of garnet as a peritectic 280 281 phase (Figs 5c-d). Within the field of the peak assemblage in the P-T pseudosections in Figs. 5b and 5f, (marked as yellow), the X_{Mg} isopleth of 0.2 for peritectic garnet (corresponding to 282 the measured core composition of garnet porphyroblasts, table 2) is bracketed between 730°C 283 284 and 780-800°C. The computed volume % of melt in equilibrium with garnet compositional isopleth of $X_{Mg} = 0.2$, in case of H₂O-fluxed melting is higher than the melt percolation 285 286 threshold limit (~10-12 vol%) and lower for H₂O-absent melting (Figs 5d-e). Since the sample studied here is a migmatite where melt-rich layers form distinct bands supporting melt 287 migration during formation of peritectic garnet porphyroblasts (Figs 2a-c), we propose that the 288 289 partial melting for the rock occurred under H2O-fluxed conditions when the mechanical 290 conditions for melt segregation and migration was imparted. Peak pressure for stability of the assemblage garnet-biotite-sillimanite-melt is estimated to be ~8 kbar where the X_{Mg} isopleth 291 (0.2) for garnet intersects the muscovite-in line (yellow field) in Figs. 5a-b (for water-fluxed 292 melting). Based on isopleth thermometry from garnet composition, the peak P-T estimated is 293 ~8 kbar, 730°C (Fig. 5b) for the following melting reaction (1): 294

Biotite
$$(Bt_1)$$
 + sillimanite (Sil_1) + quartz (Qtz_1) + plagioclase + H₂O

296

$$\rightarrow$$
 garnet + melt (eqn. 1).

The reaction is also supported by the textural observations like occurrences of inclusion trails of biotite (Bt₁) and sillimanite (Sil₁) in garnet porphyroblasts (Figs 3a-b). The X_{An} isopleth of ~0.3 is the same as that measured in the plagioclase grains (in the leucocratic domains) in textural equilibrium with garnet. This isopleth intersects the garnet compositional isopleth of X_{Mg} =0.2 at the peak P-T conditions of ~8 kbar, 730°C (Fig. 5b). For conventional thermometry, we have chosen the near peak pressure of 8 kbar obtained from isopleths thermobarometry as a reference pressure for calculating temperatures from garnet cores, rims and biotites (Bt₂) crystallized in melt rich domains. Average temperature obtained for these pairs is ~680°C (maximum and minimum values ranging from 670-690°C; Table 3). This temperature is within the temperature range obtained from the garnet core and biotite inclusion data (690°C). Any prograde path that may be proposed for reaction in equation 1 for the studied sample will cross the wet-solidus in the pseudosection, with increase in pressure and temperature (Fig. 5a).

Fractures in garnet porphyroblasts are filled with biotite and/or biotite + sillimanite + mullite (Figs 3a,c,g). High X_{Mg} values of biotites in these fractures indicate metamorphic origin rather than crystallization from any melt. So it is proposed that these biotites were formed by back reaction between melt trapped in fractures and garnets during melt-present reaction:

314 garnet + melt = biotite \pm sillimanite \pm mullite (eqn. 2). Because of variable equilibration volumes for such reactions which are smaller than 315 even a thin-section scale, their bulk composition cannot be determined and hence the reaction 316 in equation 2 could not be modeled by any pseudosection analyses. Nevertheless, it may be 317 predicted that the retrograde segment of any clockwise P-T path for such orogenic 318 metamorphism will cross the H₂O-saturated solidus for was H₂O-rich, leading to the formation 319 of the retrograde Bt₄±Sil₄±Mul₄ assemblage. For calculating the retrograde temperature range 320 321 we have chosen the reference pressure at 6 kbar, which is the minimum pressure obtained for 322 the retrograde section of P-T path in Fig. 10. At this reference pressure, rims of garnet against the adjoining biotites (Bt₃) in the S₃ fabric yield average temperature of ~570°C (maximum 323 and minimum temperatures being 550-580°C, Table 3); this estimate is somewhat higher than 324 325 the compositions of garnet and biotite (Bt₄) lodged in the garnet fracture (~540°C). biotite melting at ~6 kbar (Fig. 10). Hence, we infer that the retrograde section of the clockwise P-T 326 327 path corroborates the back reaction between garnet and residual melt, which

328 Crystallographic and microstructural characterization

In order to constrain deformation mechanisms of the phases in presence of melt detailed crystallographic and microstructural analyses were carried out on garnet, quartz and mullite occurring within the different microstructural domains of the migmatite (Fig. 3a).

332 General orientation characteristics of mullite, quartz and garnet

Mullite: Mullite (Mul₁) occurring along the two limbs of D₁ micro folds hosted within D₂ garnet porphyroblasts shows contrasting orientations (Fig. 6a; b). Elongate mullite mats (Mul₃) defining the S₃ foliation in the matrix show a strong CPO, with the c-axes parallel to the stretching lineation (X-direction) and (010) plane perpendicular to the foliation plane (Figs 6cd).

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Quartz: Quartz occurs in different textural domains (Table 1, Fig. 4). Quartz grains at the hinges 339 of D₁ rootless fold in the matrix (Qtz₁; Fig. 7a-b) exhibit a high number of subgrains and 340 dauphine twins (60° rotation around c-axis) and a distinct CPO with clustering of (0001) poles 341 to plane (Fig.7c). In contrast, quartz grains within leucocratic layers and intergrown with 342 garnet (Qtz₂; Fig. 7b, 7d) show very weak to random CPO (Fig. 7e). The orientation map of 343 the coarse quartz grains (Qtz_2) within S_2 leucosomes in the interfolial domains of S_3 is shown 344 345 in Fig. 7f. Here, quartz grains exhibit lattice distortions, few continuous subgrain boundaries and a near random CPO (Fig. 7g). In contrast, a weak, but distinct CPO (Fig. 7i) is documented 346 347 in fine-grained quartz (Qtz₃) in the biotite-mullite-sillimanite (Bt₃-Mul₃-Sil₃) dominated S₃ foliation (Figs 3a, 7h; 10). 348

349

Garnet: All analyzed garnet grains exhibit an elongate shape (e.g. Fig. 8a,b,d). All grains have very similar crystallographic orientations (Fig. 8e) even if these grains do not appear to be physically continuous in the plane of observation (Fig. 3a). The crystallographic orientation map of a representative elongate garnet grain from the S₂ melanocratic band shows an absence of distinct subgrain boundaries and/or systematic crystal lattice bending. However, distinct changes in orientation are seen associated with fractures filled or not filled by quartz, biotite and mullite (Mul₃-Bt₃) (Fig. 8b,c,d). Analysis of other garnet grains show the same absence of systematic crystal lattice distortion or subgrain boundaries.

358

359 Monazite dating

Monazites in the sample occur as inclusions in garnet, quartz (Qtz_2) and potash feldspar (Kfs₂) 360 in the S₂ layer, and overgrowing D₁ crenulation hinges (Figs 9a-d). Monazite grains are often 361 362 anhedral and vary in size from 40-100µm (Figs 9c-o). BSE images show that the monazite grains are mostly homogeneous in composition with some grains having yitrium rich cores. 363 Monazites included in garnet and potash feldspar (Kfs₂) yield 1028 Ma and 920-912 Ma 364 365 respectively (Figs 9a,d,g,k). Cores of monazites occurring within S₂ melt bearing layer yield ages ranging from 1037-930 Ma and ages recorded from their rims vary from 887-840 Ma (Figs 366 367 9 e-f, h, n; Table 5). Monazites from the S₃ foliations yield 965-889 Ma from the cores and 887-850 Ma from their rims (Figs 9i, 1-m, o; Table 5). The 33 monazite spot ages obtained 368 from 15 grains yield two unmixed populations: 914±11 Ma and 1285±64 Ma (Fig. 9r). The 369 older age population (~1285 Ma) is obtained from the core of one monazite included in a garnet 370 porphyroblast. 371

372 **DISCUSSION**

373 Conditions of migmatisation and synchronous deformation

Field and microscopic observations indicate that melting in the garnet-sillimanite migmatites of Rampura-Agucha initiated post- D_1 , and was pronounced during the D_2 deformation stage resulting in melt segregation and formation of S_2 layering including coarse-grained leucocratic, garnet-bearing bands and fine-grained, restititic layers dominated by biotite, mullite and some quartz. The absence of melts cross-cutting D₃ fabrics precludes anatexis during or after the D₃
deformation.

Microstructural observations (Figs. 3a-b) and pseudosection analyses indicate that the 380 381 D_2 peritectic garnet was formed during a H₂O-fluxed partial melting event at the expense of D_1 biotite-sillimanite aggregates at ~ 725°C and at pressure ~ 8 kbar under granulite facies 382 conditions indicated by isopleth thermometry involving garnet compositions (Figs 5b, 10). The 383 temperature recorded is thus close to the biotite melting reaction temperatures determined from 384 pseudosection modeling using average metapelite bulk composition (White et al., 2001; Saha 385 386 et al., 2008). A tight-loop clockwise P-T path constructed for the metamorphic event (Fig. 10), indicates rapid burial and exhumation. Since the garnet porphyroblasts are chemically 387 homogeneous, it is suggested that high degree of chemical equilibrium was attained at the 388 389 anatectic conditions as ubiqutous melt allowed rapid elemental exchange. Biotite and/or 390 sillimanite in the fractures of garnet porphyroblasts were formed by back reactions with the melts trapped in the fractures during late stages of partial melt crystallization which marks the 391 392 onset of retrogression (Figs. 3c; 10).

393

The Aravalli Delhi Fold Belt represents part of the Grenvillian–age orogeny in Northern India

Enclaves of polycyclic granulite facies migmatites from the Sandmata Complex of BGC-II (Fig. 1b) occur within charnockitic gneisses (Saha et al. 2008; Bhowmik et al. 2010). These migmatite enclaves record two partial melting events, namely M_1 at ~1.7-1.6 Ga in the sillimanite stability field, at 6-7 kbar and 850°C, followed by a second event (M_2) at ~1.0 Ga in kyanite stability field, at ~14 kbar, 850°C (Figs 5a, 5e; Saha et al. 2008). The high-pressure Grenvillian-age M_2 partial melting event recorded in the enclaves has also been recorded from the host felsic gneisses. Based on zircon and monazite ages from the gneisses of the Sandmata Complex and Mangalwar Complex (Fig. 1b), Buick et al. (2006; 2010) proposed recycling of
Paleo-Mesoproterozoic (~1.7-1.4 Ga) crustal components during the Grenvillian Delhi
orogeny. Bhowmik et al. (2010) reported partial melting of gneisses from the Mangalwar
Complex at ~12 kbar, 800°C during the Grenvillian orogeny (Figs 5a, 5e). In contrast, for the
Grenvillian metamorphism from the Pur-Banera supracrustal sequence (Fig. 1b), Ojha et al.
(2016) recorded peak P-T conditions of ~8 kbar, 600°C (Figs 5a, 5e).

In our sample, monazite aligned parallel to the S₂ and S₃ fabrics yield ages ranging between 1000-900 Ma (Fig. 9). Similar ages were also recorded by monazite grains sequestered within D₂ garnets suggesting an anataxis event at that time. Rims of monazites in the melt-rich layers yield spot ages of ~870-850 Ma, which may be interpreted as minimum age of melting and/or the age range of melt crystallization. Monazite yielding older ages of ~1286 Ma (Fig. 9) may be detrital.

In summary, the Rampura-Agucha metapelite migmatites underwent Grenvillian-age
anatexis at ~8 kbar and 730°C. Peak P-T condition were thus at substantially lower pressures
than those recorded westwards in the Sandmata Complex.

418

419 Evolution of deformation microstructures in the Rampura-Agucha migmatite: 420 Characteristics and rheological implications

The correlation of microstructural development with the P-T path vector is given in Fig. 10. Contrasting orientations of mullite (Mul₁) grains in the limbs of the micro-folds hosted within garnet porphyroblast are consistent with systematic re-orientation (by rigid body rotation) of initially highly aligned mullite grains that inherited strong crystallographic preferred orientation during growth syn-tectonic with D₁ prior to melting. Solid state, prograde high strain is manifested by coarse quartz grains (Qtz₁) with undulose extinction and subgrains (Figs 3h-i). EBSD analyses of these quartz grains (Qtz₁) in the relict D₁ crenulation hinge in the matrix show a distinct CPO (Figs 7a-c). The orientation of the c-axis relative to the reference frame along with the presence of undulatory extinction and distinct subgrains with continuous systematic subgrain boundaries suggests solid state deformation via dislocation creep. The patterns observed in the pole figure is consistent with a dominance of <c> slip suggesting high-T (650-700°C) deformation during D₁ (Blumenfild et al. 1986, Passchier and Trouw 2005; Law 2012; Fig. 7b).

434 Quartz grains in the leucocratic S_2 domains in the migmatite are coarser than 200 μ m and exhibit a very weak to random CPO. Interestingly, despite near absence of a CPO, 435 436 internally the grains show weak undulatory extinction, and few, discontinuous subgrain boundaries (Fig. 7f-g). We interpret these features to originate from the local stresses that occur 437 when a melt cools and quartz crystallizes and grows as one of the last phases. During this phase, 438 439 minor solid state crystal plastic deformation can be induced, resulting in the observed internal 440 deformation features, but since deformation is not due to regional differential stress, no distinct CPO develops. Similar features have been described by Hasalova et al. (2011) in frozen-in 441 partial melts. We suggest that the newly crystallized coarse quartz grains escaped deformation 442 post-anatexis because as the melt pool solidified, strain was partitioned first into adjacent melt-443 bearing rheologically weak domains; hence the solidified melt pool deformed only to a minor 444 extent. Once, all melt had solidified these coarse-grained domains were not subjected to the 445 446 significant solid state deformation. This lack of crystal plastic deformation imposed during 447 later retrograde deformation is similar to conclusions for similar migmatitic rocks (Hasalova et al. 2011, Menegon et al. 2011). 448

In contrast, in the S_3 realms the polygonised, finer-grained quartz aggregate (10-50 μ m; Fig. 10; Fig. 7h) shows a weak CPO (Fig. 7i). These aggregates are interleaved with relict prograde biotite, sillimanite and mullite layers that anastomose around the coarser-grained domains. The relict nature of the latter phases is inferred from the chemistry of the biotites and 453 is therefore also interpreted to represent the restitic parts of a migmatite. Based on to the geometric warping around the coarse-grained garnet bearing leucosome, these fine-grained, 454 phase mixed layers must have deformed post-migmatisation accommodating strain during the 455 456 retrograde path. We suggest that deformation occurred during solid state where most of the strain was taken up by the basal slip of biotite along with rigid body rotation of biotite and 457 mullite. Quartz is mostly "passive" as the rheologically weaker biotite and mullite concentrate 458 459 strain. Therefore, quartz exhibits only minor internal lattice distortions and a weak CPO (Fig. 7i). We prefer this interpretation to the possibility of dislocation glide accommodated grain 460 461 boundary sliding at high temperature (>700°C), aided by small grain sizes and possibily melt presence at grain boundaries (Schmid 1982; Behrmann and Mainprice 1987; Brodie and Rutter 462 2000; Song and Ree 2007; Killan et al. 2011; Svahnberg and Piazolo 2011). Our preference is 463 464 based on the fact, that biotite is known to be weak and in mixed, multiphase layers a weak CPO is possible to develop in the "harder" mineral phase. This is due to pronounced strain partitition 465 between easily deformed phases e.g. graphite or biotite and a phase requiring higher critical 466 467 resolved shear stress for activation of slip systems e.g. ice or quartz (Cyprych et al. 2016). Furthermore, microstructurally there are no indications for melt film presence during D_3 468 deformation. A consequence of our interpretation is that at the time of the last deformation 469 event, the rock was relatively weak, not due to presence of melt, but due to the presence of 470 471 fine-grained phase mixed layers that developed during the transient migmatistationn event. 472 These fine-grained, biotite, mullite rich, restite layers deform at relatively low stresses as suggested by experiments (e.g. of an analogues system of ice and graphite; Cyprych et al. 473 2016), hence are weak. 474

In summary, we interpret that high-strain deformation prevailed postdating anatexis.
Strain was accommodated by solid state deformation (D₃) within the fine-grained quartz
aggregates interleaved with biotite aggregates forming rheologically weak layers, and not by

the coarse-grained leucosome domains similar to strain partitioning seen in other layeredmetamorphic rocks (e.g. Smith et al. 2015).

Large elongate garnet porphyroblasts described here seem at first sight to be similar to 480 481 those noted by several researchers in metamorphic rocks formed under granulite facies to eclogite facies conditions (Ji and Martignole 1994; Prior et al. 2000; Kleinschrodt and Mc Grew 482 2000; Ji et al. 2003; Mainprice et al. 2004; Storey and Prior 2005). These studies conclude that 483 484 under both subsolidus and suprasolidus conditions at different crustal depths coarse garnet crystals may achieve their elongated shape due to crystal-plastic deformation manifested by 485 486 clear subgrain boundaries and systematic lattice bending. In contrast, the elongate garnet in our sample is characterized by the absence of subgrain boundaries and/or crystal lattice bending 487 (Figs 8a-b). The occurrence of vermicular quartz intergrown with garnet (Fig. 3b), melt films 488 489 at feldspar-garnet interfacess (Figs 3e-f), and melt-bearing domains trapped within garnets (Fig. 8d, Fig. 10), unmistakably point to the presence of melt during garnet growth. 490 Furthermore, small plagioclase grains with low dihedral angles possibly represent melt 491 492 pseudomorphs (Fig. 3j; e.g. Sawyer, 2001). Hence, garnet is peritectic growing in response to incongruent melting of biotite-sillimanite aggregates. The extensional Mode-I fractures in the 493 494 elongate garnet porphyroblasts may have been induced by a combination of high fluid/melt pressure exerted by partial melt crystallization (Abe and Urai 2012; Komoroczi et al. 2013, 495 496 Rimsa et al. 2007, Tretiakova et al. 2016) as well as tectonic forces (Abe and Urai 2012; 497 Komoroczi et al. 2013). Subsequent to garnet fracturing, growth of biotite and mullite (Bt₄-Mul₄) along fractures and along garnet margins (Figs 3a, g) indicate that melt-rich domains 498 remained in contact with the garnet-bearing layers to promote their growth during cooling and 499 500 back-reaction. However, these domains remained shielded during subsequent solid-state deformation as garnet and surrounding coarse-grained quartz and orthoclase remained rigid at 501 502 the D₃ event.

We attribute the elongate shape of garnet to its growth via partial melting (reaction 1) pseudomorphing a pre-existing three–dimensional network of biotite-sillimanite/mullite clusters. Such a biotite-sillimanite network intergrowth explains the uniform crystallographic orientation of garnet as one large garnet grew within the three–dimensional network. We however concede that the statistics for making the inference is low (number of grains exposed at the surface of the thin section analyzed = 15).

509 Mullite stable in the S₃ foliation domains shows no substructure but strong CPOs (Figs 510 6c-d), indicating either rigid body rotation in presence of high shear stress under solid state 511 (see Piazolo and Jaconelli 2013) or possibly by epitaxial replacements of Kfs₂ in the presence 512 of differential stress inducing oriented growth during the same late-stage solid-state 513 deformation.

514

515 **CONCLUSIONS**

516 Our analyses of microstructural development in the garnet-sillimanite-mullite-quartz-feldspar-517 biotite migmatite in the high-grade Aravalli Delhi Fold Belt help to identify the suprasolidus 518 to subsolidus deformation mechanisms in a migmatite preserving signatures of its prograde, 519 peak and retrograde pressure-temperature-deformation path experienced by the migmatite. The 520 following conclusions can be drawn:

521 During the Grenvillian Delhi Orogeny (spanning from 1000 Ma-870 Ma), garnet-522 biotite-sillimanite-mullite-quartz-feldspar bearing migmatite from the shear zone in the 523 Rampura-Agucha area of Aravalli Delhi Fold Belt underwent crustal anatexis at ~8-9 kbar, 524 730°C which is at much lower depth (~30 km) compared to the high pressure crustal anatexis 525 of the migmatites from the SC and MC (~14 kbar, >40 km) in the west (Fig. 1). Three main 526 stages of fabric development (S₁-S₃) occurred in the rock during the orogeny, where the S₂ 527 metatexite layers were formed on its prograde path which underwent folding leading to

528 formation of biotite-sillimanite-mullite-quartz bearing axial planar foliations (S₃). During peak conditions, anatectic incongruent melting reaction involved breakdown of biotite, sillimanite 529 and formation of garnet bearing peritectic assemblage along a tight loop clockwise P-T path. 530 531 The preservation of sillimanite-mullite-biotite-quartz bearing inclusion trails in the Grenvillian age garnet porphyroblasts and strong CPOs, undulose extinction and subgrain boundaries in 532 quartz grains in the relict crenulation hinges in matrix and within garnets imply that the rock 533 534 was subjected to a pre-Grenvillian solid state, high-temperature deformation event (>650-700°C). 535

536 Peritectic garnet porphyroblasts are elongated with aspect ratios varying from 3.1-1.7 with Mode-I fractures perpendicular to the elongation. Presence of biotite-sillimanite-mullite 537 in these fractures indicate their crystallisation during back reaction of garnet and trapped 538 539 hydrous melt during cooling. Absence of subgrain boundaries and systematic crystal lattice 540 bending in the garnet porphyroblasts indicate that their elongated nature is not a result of ductile deformation. We suggest that the garnet grains attained elongated shape as a result of their 541 growth over a 3-D network of pre-existing (prograde) biotite-sillimanite-mullite foliations. 542 Post garnet growth, extensional Mode-I fractures formed due to local high melt pressure during 543 vaining stages of anataxis in combination with tectonic stresses. Microstructural analyses of 544 garnet from the study hence imply that elongated shapes of peritectic garnets especially those 545 546 crystallised in equilibrium with melts need not necessary be due to ductile, crystal-plastic 547 deformation and that under such conditions garnets may exhibit brittle behaviour.

Near-random two weak CPOs of coarse-grained quartz in the matrix S_2 indicate that crystallisation of melt outlasted high temperature (>700°C) D_2 deformation in these areas. Postmigmatisation solid state deformation was instead accommodated in the fine-grained quartz rich domains interleaved with biotite, sillimanite, mullite in the S_3 layers. These layers represent restitic layers formed during partial melting. During post-migmatisation deformation

these phase mixed, fine-grained layers act as rheological weak zones as they deform mainly by
basal slip in biotite and rigid body rotation of both biotite and mullite and minor dislocation
creep in quartz

556 Our study suggests that the inferred rheological weakness of migmatites may outlast 557 the actually melt-present time period, as biotite rich layers developed during partial melting 558 facilitates late stage, solid state deformation with low rheological strength.

559

560 **TABLE CAPTIONS**

561 Table 1.Characteristics of deformation microstructures and mineral assemblages.

Table 2. Electron probe microanalaytical data and structural formulae of silicate phases inRAM-1.

Table 3. Results of geothermobarometry for analysed sample RAM-1.

Table 4. Bulk rock major element oxides (in wt %) in analysed sample RAM-1.

Table 5. Electron probe microanalaytical data and spot age $(\pm 2\sigma)$ data in monazites in analysed sample RAM-1.

568

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577 FIGURE CAPTIONS

Figure 1. (a) Generalised tectonic map of India showing the Archean cratons and Proterozoic 578 mobile belts. Abbreviations used are ADFB: Aravalli Delhi Fold Belt, AC: Aravalli Craton, 579 580 BuC: Bundelkhand Craton, SC: Singhbhum Craton, BC: Bastar Craton, EDC: Eastern Dharwar Craton, WDC: Western Dharwar Craton, CITZ: Central Indian Tectonic Zone, CGC: 581 Chotanagpur Gneissic Complex, SPGC: Shillong Plateau Gneissic Complex, EGMB: Eastern 582 Ghats Mobile Belt, SIG: South Indian Granulite terrain, HOB: Himalayan Orogenic Belt, DV: 583 DeccanVolcanics; 1: Mahandi Rift, 2: Godavari Rift, 3: Closepet Granite. Grey regions show 584 585 the Grenvillian orogenic belt. The study area (ADFB) is shown in the box. (b) Map showing different lithological and tectonic components of Aravalli Craton and Aravalli Delhi Fold Belt 586 (modified after Heron, 1953; Roy and Jhakar, 2002). GF: Grenvillian Font after Ojha et al. 587 588 (2016). Box indicates location of the study area. (c) Geological map of Bhilwara Supergroup from Rampura-Agucha area (modified after Roy 2002). Boxes show locations from which 589 detailed structural analyses have been conducted and location (Rampura-Agucha mine pit) 590 from which sample RAM-1 has been collected. 591

592

Figure 2. (a) Field photograph of anatectic garnet-biotite-sillimanite gneiss showing different stages of deformation (for detailed description see text). (b, c) Photographs showing the neosome parts of anatectic garnet-biotite-sillimanite bearing migmatite (RAM-1). Mineral abbreviations in the figures and tables are after (Kretz 1983). See text for description of structural elements in the rock.

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Figure 3. Optical microphotographs (a-b, d-f, h-k), SEM image (c) and BSE image (g) displaying textures from RAM-1. (a) Photograph showing D_1 crenulation hinges (defined by sillimanite and mullite) within garnet and pervasive S_3 (defined by biotite, sillimanite, mullite), 602 warping around garnet porphyroblast. Quartz-feldspar bearing melt layers (S₂) occur within the interfolial domains of S_3 . Relict rootless D_1 crenulation hinges occur in the interfolial domains 603 of S₃; yellow arrow heads point to fracture filled with biotite and sillimanite/mullite (b) Garnet 604 605 porphyroblast preserving D₁ crenulation hinges and quartz inclusion trails axial planar to the crenulation. (c) An elongated garnet porphyroblast in contact with S₃. (d) A coarse potash 606 feldspar grain (at contact of S₃ biotites), containing rounded quartz inclusions and biotite 607 inclusions. (e) Quartzo-feldspathic layer at contact of garnet porphyroblast showing melt-608 related microstructures like randomly oriented sillimanite and mullite and biotite grains in 609 610 coarse potash feldspar, globular inclusions of quartz in potash feldspar and thin melt film (now represented by plagioclase) in between potash feldspar grains. (f) Thin films of plagioclase 611 along the margins of coarser plagioclase indicate sites of melt crystallization. (g) Intergrowth 612 613 of garnet with globular quartz grains indicate crystallization in presence of melt. (h and i) 614 Coarse quartz grains in the D₁ crenulation hinges are strained, show undulose extinction (yellow arrow) and presence of subgrains (red arrows). (j) Presence of euhedral plagioclase 615 grain at the contact of coarse quartz grains crystallised from melt trapped melt in D₂ garnet. 616 Note presence of relict sillimanite. 617

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Figure 4. Schematic sketch showing deformation microstructures, i.e. (a) D₁ deformation; (b)
D₂ deformation; (c) D₃ deformation.

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Figure 5. Pseudosections in NCKFMASH system from bulk composition of RAM-1. Molar proportions under H₂O-fluxed conditions are SiO₂:Al₂O₃:FeO:MgO:CaO:Na₂O:K₂O: H₂O = 48.33:9.63:5.22:8.48:1.15:1.15:2.90:23.14. (a) P-T pseudosection under H₂O-fluxed melting condition, showing stability field of garnet-sillimanite-feldspar-biotite-quartz-melt bearing assemblage noted in the sample, at P-T ranges of 5-8 kbar, 725-780°C. (b) Compositional

627 isopleths of garnet, biotite and feldspar in the P-T pseudosection of C and star sign indicates highest P-T for melting reaction. (c and d) Modal isopleths of different phases in the P-T 628 pseudosection of (a). (e) P-T pseudosection under H₂O-absent melting condition. Oxide molar 629 630 proportions of the bulk is SiO₂:Al₂O₃:FeO:MgO:CaO:Na₂O:K₂O:H₂O = 58.09:11.57:6.27:10.19:1.38:1.38:3.49:7.62. (f) Compositional isopleths of garnet, biotite and 631 feldspar in the P-T pseudosection of (e). The yellow, grey and pink shaded fields respectively 632 show stability of biotite±garnet, orthopyroxene and cordierite with melt. M_{MC}, M_{SC} and M_{PB} 633 show peak P-T conditions for the Grenvillian anatexis in the Mangalwar Complex (after 634 635 Bhowmik et al. 2010), Sandmata Complex (after Saha et al. 2008) and Pur-Banera supracrustals (after Ojha et al 2016). 636

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Figure 6. Orientation characteristics of different mullite types (a) Mul₁; orientation map of garnet grain (lilac) and mullite as inclusions in the grain; note hinge trace (dashed white lines); (b) 3D representation of mullite crystal orientation grains present as inclusion in garnet forming crenulation hinges shown in (a). (c) Orientation map of mullite Mul₃ intergrown with finegrained biotite (khaki) and fine-grained quartz (red), for mullite colouring signifies crystal orientations as shown in inset. (d) Pole figures showing the orientation of Mul₃ mullite grains shown in (c); one point per grain.

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Figure 7. Orientation characteristics of different quartz types. (a) Petrographic image of microfold made up by Qtz₁. (b) Orientation map of microfold limb shown in (a). (c) Pole figures of Qtz₁ corresponding to orientation map of microfold shown in (a,b). (d) Orientation map of Qtz₂ (quartz intergrwon with garnet porphyroblast); note presence of Dauphine twins (red lines) but absence of subgrain boundaries (yellow lines). (e) Pole figures and contour maps of quartz corresponding to orientation map (d). (f) Orientation map of Qtz₂ coarse-grained quartz from the leucocratic layers of S_2 . (g) Pole figures and contour maps of coarse-grained quartz corresponding to orientation map (f). (h) Orientation map of fine-grained quartz, Qtz₃ intergrown with mullite (Mul₃) in the S₃ foliation planes. (i) Pole figures and contour maps of fine-grained quartz corresponding to orientation map (h).

656 Figure 8. Orientation characteristics of garnet grains. (a) Map of elongate garnet grain with surrounding minerals; different phases are shown in different colours; (b) Map of orientation 657 variations within garnet grain shown in (a), colour variations indicate the progressive internal 658 659 misorientation 0-2°. p-p' outline misorientation profile shown in (c); note in (a) and (b) location of fractures is shown as stippled lines (c) Misorientation profile within grain along line p-p' 660 shown in (b). (d) photomicrograph of fracture within elongate garnet porphyroblast; (e) Pole 661 figures showing the orientation of all garnet grains by plotting data from 15 different garnet 662 grains; note the similarity in orientation. 663

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Figure 9. Mcirostructural relations (BSE images) of the monazites (a-d), BSE images of the
monazite grains with spot ages (e-o), X-Ray Th and Y images of the monazite grain in o (p-q)
and probability density plot of monazite grains from the analysed sample RAM-1.

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Figure 10. Summary of the P-T path inferred for the migmatite of Bilwara Supergroup from the study area. Also shown are the calculated wet solidus (WSS) from Figure 5a, granite wet solidus (GWS), granite dry solidus (GDS) curves (after Brown, 2002). Shematic drawings show typical microstructures developed at the different segments of P-T path. Labeling of deformation D_1 to D_3 according to discussion in text, in italic deformation mechanisms for noted minerals. Thin section photograph shows presence of polygonal fine-grained quartz from S_3 layers (marked with the box). See text for details. Abbreviations used for metamorphic facies 676 fields are: BS: blueschist, AmEc: amphibole eclogite, EpAm: epidote amphibole, GrtAm:677 garnet amphibolite, Gr: granulite.

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Appendix-1. (a) and (b) T-M_{H2O} pseudosections at 6 kbar and 8 kbar respectively, showing stability fields of different minerals under water-deficient and water-fluxed conditions. High modal percentages of melt are observed in garnet-sillimanite-bioite- feldspar-quartz-melt bearing fields. Bulk compositions for C₀ and C₁ (in molar proportions) are respectively SiO₂: Al₂O₃: FeO: MgO: CaO: Na₂O: K₂O: H₂O = 62.68: 12.56: 6.77: 10.99: 1.49: 1.49: 3.77: 0.26

684 and 41.18: 8.25: 4.45: 7.22: 0.98: 0.98: 2.47: 34.46.

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Figure 4



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1: Bt Ms Grt Ky Fsp Qtz

4:Bt Melt Fsp Grt Ky Qtz

5: Bt Ms(2) Grt Ky Fsp Qtz

7: Bt Fsp(3) Ms Grt Ky Qtz

9: Melt Fsp Grt Sill Qtz

11: Bt Opx Melt Fsp Grt

10: Bt Melt Fsp(2) Ky Qtz

12: Bt Fsp Ms(2) Ky Zo Qtz 13: Bt Fsp Ms(2) Sill Qtz

14: Bt Melt Fsp Grt Sill Qtz

15: Bt Melt Fsp(2) Grt Sill

16: Bt Melt Fsp(2) Grt Crd

19: Opx Melt Fsp Grt Spr

20: Opx Melt Grt Spr

21: Opx Melt Fsp Spr

17: Bt Melt Fsp(2) Grt Crd Qtz

18: Bt Opx Melt Fsp(2) Grt Crd

6: Bt Ms(2) Grt Ky Zo Fsp Qtz

8: Bt Fsp Ms(2) Grt Ky Zo Qtz

2: Bt Melt Fsp(2) Ms Grt Qtz

3: Bt Melt Fsp(2) Ms Grt Ky Qtz













600

800 Temperature(°C) Figure 10

Microstructure type (S = foliation, L – lineation, F- fold, D- deformation)	Mineral assemblage	Main features	Metamorphic reactions leading to assemblage and P-T conditions	Inferred deformation mechanism(s)	Additional Comments
D 1	Mul ₁ , Sil ₁ , Qtz ₁ , Bt ₁	Mul ₁ grains forming two limbs of the crenulations included in garnet porphyroblast show contrasting orientations (Figs 3a; 6a-b). Rootless fold hinge (Figs 3a, 7a). Qtz ₁ shows strong CPO (Figs 7b-c).	~ 650-700°C	Strong CPO indicates activity of prism <c> slip with solid state dislocation creep.</c>	High strain persisted before partial melting and formation of gneissic band S_{2} .
D ₂ ; S ₂	Grt, Qtz ₂ , Kfs ₂ , Pl ₂ , Bt ₂ , Mul ₂	Qtz ₂ and Kfs ₂ present in the pressure shadows of garnet and within the interfolial domains of S ₃ . Inclusions of rounded Qtz ₁ grain in Kfs ₂ , bulbous and lobate Qtz ₂ grains intergrown with garnet. Mode I fractures filled with Bt ₄ , Mul ₄ are observed in garnet porphyroblasts.(Fig. 3a, d) Absence of subgrain boundaries in garnet. (Fig. 8 a, c) Qtz ₂ grains show weak CPOs (Figs 7 d- g).	Bt ₁ + Sil ₁ + Qtz ₁ + Pl ₁ + H ₂ O \rightarrow garnet + melt (crystallized to form Qtz ₂ , Kfsp ₂) ~ 725°C and ~ 8 kbar	Weak CPOs of Qtz ₂ resulted due tocrystallization from the melt. No crystal plastic deformation is noted in garnet. Mode I fracturesin garnet interpreted to be formed due to high fluidpressure and differential pure shear stresses.	Elongated nature of garnet grains is not due to crystal plastic deformation but rather garnet porphyroblasts have replaced pre- existing foliation of Bt ₁ , Sil ₁ , Mul ₁ and Qtz ₁ . Weak CPOs of Qtz ₂ and fractures in Grt indicate melt present deformation.
S ₃	Bt ₃ , Sil ₃ , Mul ₃ , Qtz ₂ , Qtz ₃	 Foliation (Qtz₃, Bt₃, Mul₃) warps around garnet porphyroblast and leucosome domains ,Fig. 3a) Fine-grained Qtz₃ interleaved with Bt_{3b}, Mul_{3b} show weak CPO (Figs 7h-i). Mul_{3b} shows strong CPO (Figs 6c-d). 	≥500°C	Strong CPO of Mul _{3b} and weak CPO of Qtz ₃ (fine-grained) indicate solid state deformation in former restitic areas.	High-strain that prevailed post-anatexis was accommodated by the fine-grained polygonal (prograde) Qtz ₃ by diffusion creep.

	G	rain 1		Grain 2					Grain 3			
Min	Grt	Grt	Grt	Grt	Grt	Grt	Grt	Grt	Grt	Grt	Grt	
	С	R^Bt ₃	R^Qtz ₂	С	R^Bt ₄	С	С	R^Qtz ₂	С	С	С	
SiO ₂	38.17	37.87	38.2	38.15	37.68	38.19	38.32	38.12	38.31	38.2	38.2	
TiO ₂	0.02	0.03	0.11	0	0.02	0	0.01	0	0.03	0	0.05	
Al_2O_3	20.98	21.12	20.97	20.7	20.5	21.06	20.9	20.93	20.92	21.16	20.82	
Cr_2O_3	0	0.01	0	0.02	0.04	0.03	0.01	0	0	0	0.01	
Fe ₂ O ₃	0	0	0	0	0.45	0	0	1.26	0.45	0.92	0.6	
FeO	34.95	35.71	34.17	33.89	36.15	34.69	34.6	33.54	34.4	34.09	34.82	
MnO	0.51	0.5	0.38	0.38	0.45	0.51	0.46	0.33	0.36	0.41	0.46	
MgO	4.48	4.03	4.81	4.77	3.66	4.71	4.53	5.35	5.09	5.14	4.8	
CaO	1.37	1.39	1.41	1.4	1.4	1.36	1.34	1.41	1.43	1.39	1.47	
Na ₂ O	0.05	0.02	0.03	0.06	0.02	0.06	0.02	0.06	0.03	0.05	0	
K ₂ O	0	0	0	0.01	0.02	0	0	0.03	0.01	0	0	
Totals	100.53	100.68	100.08	99.38	100.38	100.61	100.19	101.03	101.02	101.36	101.23	
Oxygens	12	12	12	12	12	12	12	12	12	12	12	
Si	3.027	3.011	3.031	3.047	3.019	3.022	3.042	2.998	3.017	2.998	3.011	
Ti	0.001	0.002	0.007	0	0.001	0	0.001	0	0.002	0	0.003	
Al	1.961	1.979	1.961	1.949	1.936	1.965	1.956	1.941	1.942	1.958	1.935	
Cr	0	0.001	0	0.001	0.003	0.002	0.001	0	0	0	0.001	
Fe ³⁺	0	0	0	0	0.027	0	0	0.075	0.026	0.055	0.036	
Fe ²⁺	2.318	2.374	2.267	2.263	2.422	2.296	2.297	2.206	2.265	2.237	2.296	
Mn	0.034	0.034	0.026	0.026	0.031	0.034	0.031	0.022	0.024	0.027	0.031	
Mg	0.529	0.477	0.569	0.568	0.437	0.555	0.536	0.627	0.597	0.601	0.564	
Ca	0.116	0.118	0.12	0.12	0.12	0.115	0.114	0.119	0.121	0.117	0.124	
Na	0.008	0.003	0.005	0.009	0.003	0.009	0.003	0.009	0.005	0.008	0	
Κ	0	0	0	0.001	0.002	0	0	0.003	0.001	0	0	
Sum	7.995	7.999	7.984	7.984	8	7.999	7.981	8	8	8	8	
X _{Alm}	0.77	0.79	0.76	0.76	0.80	0.77	0.77	0.74	0.75	0.75	0.76	
X_{Prp}	0.18	0.16	0.19	0.19	0.15	0.19	0.18	0.21	0.20	0.20	0.19	
X _{Grs}	0.04	0.04	0.04	0.04	0.04	0.04	0.04	0.04	0.04	0.04	0.04	
X_{Sps}	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	
X_{Mg}	0.19	0.17	0.20	0.20	0.15	0.19	0.19	0.22	0.21	0.21	0.20	

Table 2: Representative mineral chemical analyses (EPMA) from RAM-1

Table 2 (c	continued)											
		Gr	ain 3		Grain 4							
Min	Grt	Grt	Grt	Grt	Grt	Grt	Grt	Grt	Grt	Bt_1	Bt ₂	
	R^Qtz_2	R^Qtz_2	С	R^Qtz ₂	R^Bt_2	С	С	С	С	I^Grt	M(C)	
SiO ₂	38.12	38.18	38.17	38.2	37.47	37.69	37.99	37.44	37.68	35.85	35.98	
TiO ₂	0	0	0	0	0	0	0	0.02	0	2.11	1.91	
Al_2O_3	20.43	20.79	20.48	21.06	20.86	20.7	21	20.83	20.74	17.67	17.77	
Cr_2O_3	0.01	0.04	0.05	0.03	0.06	0.08	0.08	0.1	0	0.01	0.06	
Fe ₂ O ₃	0.67	0.45	1.22	0.3	0.16	0	0.06	0.73	1.19	0	0	
FeO	35.5	36.06	35.46	35.7	32.6	32.67	32.87	32.3	32.06	16.87	17.2	
MnO	0.44	0.47	0.47	0.36	0.37	0.31	0.37	0.32	0.42	0.08	0	
MgO	4.41	4.29	4.49	4.43	5.57	5.62	5.77	5.78	5.83	11.28	11.7	
CaO	1.34	1.15	1.29	1.34	1.39	1.49	1.48	1.45	1.52	0.02	0.03	
Na ₂ O	0	0	0.01	0	0.02	0	0	0	0.05	0.09	0.18	
K ₂ O	0.02	0	0	0	0.01	0.02	0	0	0	9.58	9.38	
Totals	100.94	101.42	101.64	101.42	98.51	98.58	99.62	98.97	99.49	93.56	94.21	
Oxygens	12	12	12	12	12	12	12	12	12	11	11	
Si	3.025	3.017	3.011	3.012	3.008	3.022	3.014	2.992	2.996	2.751	2.742	
Ti	0	0	0	0	0	0	0	0.001	0	0.122	0.109	
Al	1.911	1.937	1.904	1.957	1.974	1.957	1.964	1.963	1.944	1.599	1.597	
Cr	0.001	0.002	0.003	0.002	0.004	0.005	0.005	0.006	0	0.001	0.004	
Fe ³⁺	0.04	0.027	0.072	0.018	0.009	0	0.003	0.044	0.071	0	0	
Fe ²⁺	2.356	2.383	2.339	2.354	2.189	2.191	2.181	2.159	2.132	1.083	1.096	
Mn	0.03	0.031	0.031	0.024	0.025	0.021	0.025	0.022	0.028	0.005	0	
Mg	0.522	0.505	0.528	0.52	0.666	0.672	0.682	0.688	0.691	1.29	1.329	
Ca	0.114	0.097	0.109	0.113	0.12	0.128	0.126	0.124	0.13	0.002	0.002	
Na	0	0	0.002	0	0.003	0	0	0	0.008	0.013	0.027	
Κ	0.002	0	0	0	0.001	0.002	0	0	0	0.938	0.912	
Sum	8	8	8	8	8	7.998	8	7.999	8	7.804	7.818	
X_{Alm}	0.78	0.79	0.78	0.78	0.73	0.73	0.72	0.72	0.72			
X_{Prp}	0.17	0.17	0.18	0.17	0.22	0.22	0.23	0.23	0.23			
X _{Grs}	0.04	0.03	0.04	0.04	0.04	0.04	0.04	0.04	0.04			
X _{Sps}	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.54	0.55	
X_{Mg}	0.18	0.17	0.18	0.18	0.23	0.23	0.24	0.24	0.24	0.54	0.55	

 Table 2 (continued)

Min	Bt_2	Bt ₂	Bt ₂	Bt ₃								
	M(C)	M(C)	M(C)	M(C)	M(C)	M(C)	M(C)	M(C)	M(C)	M^Grt Rim	M^Grt Rim	M^Grt Rim
SiO ₂	35.64	35.68	35.57	36.7	36.3	34.72	35.81	36.53	36.37	36.74	36.36	36.32
TiO ₂	1.87	1.99	1.74	1.88	1.21	1.24	1.18	1.94	2.13	2.21	2	2.13
Al_2O_3	17.93	17.71	17.18	17.55	18.38	18.37	18.51	17.84	18.08	18.62	18.86	18.88
Cr_2O_3	0.03	0.03	0.05	0.1	0	0.08	0	0.03	0	0.07	0.06	0.1
Fe_2O_3	0	0	0	0	0	0	0	0	0	0	0	0
FeO	16.38	16.58	16.53	17.07	16.3	16.35	16.74	17	17.28	16.62	16.87	16.1
MnO	0.04	0.03	0.07	0.03	0.02	0	0	0.04	0	0	0	0.03
MgO	11.73	11.71	11.67	11.93	12.4	11.61	12.15	12.02	11.76	12.1	11.78	11.97
CaO	0.12	0.19	0.02	0.11	0.04	0.22	0.03	0.05	0	0.04	0.02	0.03
Na ₂ O	0.13	0.11	0.11	0.14	0.17	0.45	0.16	0.19	0.17	0.25	0.16	0.18
K_2O	9.48	9.06	9.52	9.36	9.6	9.26	9.49	9.31	9.53	9.39	9.28	9.23
Totals	93.35	93.09	92.46	94.87	94.42	92.3	94.07	94.95	95.32	96.04	95.39	94.97
Oxygens	11	11	11	11	11	11	11	11	11	11	11	11
Si	2.735	2.742	2.762	2.771	2.747	2.7	2.726	2.754	2.738	2.73	2.722	2.722
Ti	0.108	0.115	0.102	0.107	0.069	0.073	0.068	0.11	0.121	0.124	0.113	0.12
Al	1.622	1.604	1.573	1.562	1.639	1.684	1.661	1.586	1.604	1.631	1.665	1.668
Cr	0.002	0.002	0.003	0.006	0	0.005	0	0.002	0	0.004	0.004	0.006
Fe ³⁺	0	0	0	0	0	0	0	0	0	0	0	0
Fe ²⁺	1.051	1.066	1.073	1.078	1.031	1.063	1.066	1.072	1.088	1.033	1.056	1.009
Mn	0.003	0.002	0.005	0.002	0.001	0	0	0.003	0	0	0	0.002
Mg	1.341	1.341	1.35	1.342	1.398	1.346	1.379	1.351	1.319	1.34	1.315	1.337
Ca	0.01	0.016	0.002	0.009	0.003	0.018	0.002	0.004	0	0.003	0.002	0.002
Na	0.019	0.016	0.017	0.02	0.025	0.068	0.024	0.028	0.025	0.036	0.023	0.026
Κ	0.928	0.888	0.943	0.902	0.927	0.919	0.922	0.895	0.915	0.89	0.886	0.883
Sum	7.82	7.793	7.829	7.8	7.841	7.877	7.848	7.804	7.81	7.792	7.786	7.776
X _{Mg}	0.56	0.56	0.56	0.55	0.58	0.56	0.56	0.56	0.55	0.56	0.55	0.57

C: Core, R: Rim, I: Inclusion; M: Matrix,^: against.

 Table 2 (continued)

Min	Bt_4	Bt_4	Kfs ₂	Kfs ₂	Kfs ₂	Pl_2	Pl_2	Pl_2	Pl_2	Pl_2
	^Grt fracture	^Grt fracture	M(C)	M(C)	M(C)	M(C)	M(C)	M(C)	M(C)	M(C)
SiO ₂	36.35	35.87	63.53	63.86	63.96	60.83	61.46	60.76	61.03	60.84
TiO ₂	0.1	0.59	0.04	0.06	0.07	0.02	0	0.02	0.01	0.06
Al_2O_3	18.93	19.57	17.27	17.59	17.79	23.29	23.34	23.12	23.43	23.21
Cr_2O_3	0.03	0.01	0.04	0	0.04	0.04	0	0	0	0.03
Fe_2O_3	0.91	0.86	0.18	0	0	0.03	0.04	0.06	0.11	0.01
FeO	12.63	14.1	0	0	0	0	0	0	0	0
MnO	0	0	0	0	0	0	0	0	0	0.07
MgO	14.35	13.42	0	0.01	0	0	0	0	0	0.02
CaO	0.04	0.08	0.06	0	0.02	5.82	5.75	5.82	5.93	5.94
Na ₂ O	0.16	0.61	1.98	1.79	1.58	8.25	8.23	8.44	8.32	8.21
K ₂ O	9.67	8.96	13.69	13.37	13.69	0.19	0.3	0.17	0.18	0.2
Totals	93.17	94.08	96.79	96.68	97.15	98.47	99.12	98.39	99.01	98.59
Oxygens	11	11	8	8	8	8	8	8	8	8
Si	2.744	2.695	3.014	3.019	3.013	2.744	2.753	2.746	2.74	2.744
Ti	0.006	0.033	0.001	0.002	0.002	0.001	0	0.001	0	0.002
Al	1.685	1.733	0.966	0.981	0.988	1.239	1.233	1.232	1.24	1.234
Cr	0.002	0.001	0.002	0	0.001	0.001	0	0	0	0.001
Fe ³⁺	0.053	0.049	0.006	0	0	0.001	0.001	0.002	0.004	0
Fe ²⁺	0.797	0.886	0	0	0	0	0	0	0	0
Mn	0	0	0	0	0	0	0	0	0	0.003
Mg	1.614	1.503	0	0.001	0	0	0	0	0	0.001
Ca	0.003	0.006	0.003	0	0.001	0.281	0.276	0.282	0.285	0.287
Na	0.023	0.089	0.182	0.164	0.144	0.722	0.715	0.74	0.724	0.718
Κ	0.931	0.859	0.829	0.807	0.823	0.011	0.017	0.01	0.01	0.012
Sum	7.859	7.855	5.004	4.974	4.974	5.001	4.996	5.011	5.005	5.001
X _{Mg}	0.67	0.63								
X _{An}						0.28	0.27	0.27	0.28	0.28

C: Core, R: Rim, M: Matrix,^: against; mineral abbreviations are after Kretz (1983).

Table 3: Results of geothermometry

S. No.	Grt	Bt	Bh(°C)	Dg (°C)	FS(°C)	Н (°С)	HP(°C)	Average Grt-Bt (°C)
At 8 kbar re	ference pressure							
1	С	Bt ₁ ^Grt	630	597	731	708	794	692
2	С	$Bt_2(C)$	631	640	706	699	763	688
3	С	$Bt_2(C)$	628	632	697	693	760	682
4	С	$Bt_2(C)$	626	632	688	692	746	677
5	R ^ Qtz (M)	$Bt_2(C)$	617	612	671	680	754	667
At 6 kbar re	ference pressure							
6	R^Bt_4	Bt_4	531	518	515	596	517	535
	$\mathbf{D} \wedge \mathbf{D}_{t}$	$Bt_{2}(C)$	561	518	564	619	615	576
7	K Dl ₃	Dig(C)			201		010	
7 8	$R^{A}Bt_{3}$	Bt ₃ (C) Bt ₃ (R^Grt)	556	508	557	609	588	564
7 8 9	$R^{A}Bt_{3}$ $R^{A}Bt_{3}$ $R^{A}Bt_{3}$	$Bt_{3}(C)$ Bt_{3}(C)	556 547	508 491	557 536	609 597	588 567	564 548
7 8 9 10	$\begin{array}{c} R & Bt_3 \\ R^{A}Bt_3 \\ R^{A}Bt_3 \\ R^{A}Bt_3 \end{array}$	$Bt_{3}(R^{Grt})$ $Bt_{3}(C)$ $Bt_{3}(C)$	556 547 570	508 491 530	557 536 577	609 597 620	588 567 615	564 548 583

Table 4	
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Table 4	Table 4: XRF data of RAM-1								
SiO_2	55.550								
Al_2O_3	18.780								
Fe ₂ O ₃	8.860								
MnO	0.030								
MgO	6.520								
CaO	1.250								
Na ₂ O	1.360								
K ₂ O	5.240								
TiO ₂	0.880								
P_2O_5	0.070								
LOI	1.387								
Sum	99.927								

	Texture	Pb	Th	U	Age (Ma)	2 sigma	%error
1	Contact of Otz in S	0.54	10.01	0.83	030	40	5 27
2	Contact of Qt_2 in S_2	0.34	10.01	1 31	866	50	6.81
2	Contact of garnet and Otz_i in S	0.30	4.0	1.51	840	56	6.67
5 1	Contact of garnet and Qt_2 in S_2	0.57	4.77	0.82	1027	50	4.82
-	Contact of Qtz_1 in S_2	0.02	10.20	0.82	870	50 62	4.02
5	Contact of Qt_3 in S_3	0.33	4.70	0.02	074	59	7.24
7	Contact of Qtz_3 in S_3	0.44	5.12	0.93	9/4 802	50	7 20
0	Contact of Qt_3 in S_3	0.32	5.12	0.82	093	60	6.09
0	Contact of $Q(Z_3 \text{ in } S_3)$	0.35	5.28	1	888	02 71	0.98
9	Inclusion in KIs_2 in S_2	0.31	5.45	0.58	920	/1	7.11
10	Inclusion in KIs_2 in S_2	0.34	5.12	0.98	900	04 (2	/.11
11	Inclusion in KIs_2 in S_2	0.36	0 5 02	0.75	916	03 59	0.88
12	Inclusion in Kis_2 in S_2	0.4	5.02	1.30	912	58 (4	0.30
13	Overgrowing Bl_3 in S_3	0.33	5.41	0.85	889	04	7.20
14	Contact of Qtz_3 in S_3	0.31	4.83	0.62	991	/6	/.0/
15	Contact of Qtz_3 in S_3	0.33	4.67	0.84	965	/1	7.30
16	Contact of Qtz_3 in S_3	0.31	5.11	0.81	8/6	66	1.53
17	Contact of Qtz_3 in S_3	0.39	4.95	1.5	854	55	6.44
18	Contact of garnet and Qtz_2 in S_2	0.36	4.62	1.31	867	60	6.92
19	Inclusion in Grt in S_2	0.37	5.84	0.98	896	59	6.58
20	Overgrowing Bt_3 in S_3	0.33	5.26	0.96	859	62	7.22
21	Overgrowing Bt_3 in S_3	0.36	4.31	1.25	921	63	6.84
22	Overgrowing Bt_3 in S_3	0.49	8.07	0.92	956	53	5.54
23	Overgrowing Bt_3 in S_3	0.35	4.57	1.21	890	61	6.85
24	Contact of Qtz_3 in S_3	0.37	4.85	1.34	883	59	6.68
25	Contact of Qtz_2 in S_2	0.31	4.43	0.75	963	74	7.68
26	Overgrowing Bt ₃ in S ₃	0.33	5.25	0.88	887	64	7.22
27	Overgrowing Bt ₃ in S ₃	0.55	9.51	0.65	1028	53	5.16
28	Overgrowing Bt ₃ in S ₃	0.4	4.72	1.66	853	54	6.33
29	Contact of garnet and Qtz_2 in S_2	0.36	3.15	0.84	1273	92	7.23
30	Inclusion in Grt in S ₂	0.51	9.34	0.71	954	52	5.45
31	Inclusion in Grt in S ₂	0.39	5.05	0.41	1296	89	6.87
32	Contact of Qtz_2 in S_2	0.32	5.2	0.82	880	65	7.39
33	Contact of Qtz ₂ in S ₂	0.35	5.26	0.92	912	64	7.02

Table 5: Analytical data (EPMA) for the dated monazites of sample RAM-1

