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1    **Oral tribology: Update on the relevance to study astringency in wines**

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19 **Oral tribology: Update on the relevance to study astringency in wines**

20

21 **Abstract**

22 Oral tribology is emerging as a new paradigm to quantify friction and lubrication of food-saliva  
23 mixtures in the oral mucosa. Recently, oral tribology has captured research attention in  
24 quantifying wine astringency, a characteristic “dryness feeling”, which strongly impacts  
25 consumer preference. Hence, this paper aims to provide a concise review of oral tribology in the  
26 context of wine astringency. Firstly, the important roles of “biolubricant” saliva, salivary proteins  
27 and current tribo-pairs used in oral tribology measurements are reviewed. Then, we have  
28 discussed the key mechanisms of wine astringency involving polyphenol-salivary protein  
29 interactions (hydrogen bonding, hydrophobic interactions), rupture of the lubricating salivary film  
30 and oral sensation of discrete particles. Studies employing Stribeck curve analysis and  
31 microstructural characterization to understand polyphenol-salivary protein interactions are  
32 reviewed. Finally, we highlighted the need for bio-relevant tribo-pairs, simulated oral conditions  
33 and tribology-sensory correlation, before such quantification can be used to characterize wine  
34 astringency at a commercial level.

35

36           Keywords: Oral tribology; Wine; Astringency; Mucin; Saliva; Lubrication; Tannins;  
37           Proline-rich proteins

38

39 **Introduction**

40 Astringency is defined as “the complex of sensation due to shrinking, drawing or puckering of  
41 the oral epithelium as a result of exposure to substances, such as alums or tannins”<sup>1</sup>.  
42 Astringency is a very frequent sensory experience perceived upon consumption of various food  
43 and beverages, such as some unripe fruits (e.g. persimmon, chokecherry), soy-based foods,  
44 green and black tea, some herbs and spices (e.g. turmeric, marjoram, sage) and red wine. In

45 wine, astringency can be associated with different components, such as metals ions, alcohols,  
46 organic acids, but polyphenols are generally agreed to play the most important role. These  
47 polyphenols in wine come from the grapes (hydroxybenzoic and hydroxycinnamic acids,  
48 flavonol glycosides, flavan-3-ols/procyanidins, and stilbenes) and the oak barrels (hydrolysed  
49 tannins) in which the wine is stored for ageing.

50 Particularly, astringency is an important wine texture quality parameter. Till now,  
51 wine astringency research has mainly focused on identifying appropriate analytical methods,  
52 such as chromatography<sup>2</sup>. Although chromatographic tests have enabled successful  
53 identification of the relevant wine components that cause astringency, they do not allow  
54 quantifying the intensity or the evolution of the “astringent feeling”. That is why, the gold  
55 standard method of assessment of the astringency in wine is “tasting” by trained sensory panels  
56 using set of reference compounds and descriptors<sup>3</sup>. However, training a sensory panel is time-  
57 consuming and expensive. Furthermore, astringency is a complex sensory attribute as it builds  
58 in intensity over repeated exposure. Thus, it is difficult to clean the mouth between the samples  
59 with astringent components, latter can cause fatigue in sensory panel members and  
60 consequently assessments errors<sup>4, 5</sup>.

61 From mechanistic viewpoint, the term astringency comes from the Latin phrase “ad  
62 stringere” meaning ‘to bind, which is believed to be related to the ability of astringent  
63 substances, such as wine polyphenols to bind to and precipitate salivary proteins<sup>5</sup>. Although  
64 there have been several hypotheses on interactions between wine polyphenols and salivary  
65 proteins, the predominant mechanism by which solutions containing polyphenols are perceived  
66 as astringent is still not clear. Using psychophysical methods, Green<sup>6</sup> suggested that oral friction  
67 is the key underlying physiological mechanism behind the sensation of astringency. This oral  
68 friction has been postulated to be resulting from the loss of oral mucosal lubrication of the  
69 salivary film, on exposure to the polyphenol components<sup>7</sup>. Therefore, “tribology” i.e. the

70 science of friction, wear and lubrication appears as a promising approach that can be used  
71 quantify coefficient of friction in oral environment, former has gathered recent research  
72 attention in understanding astringency perception.

73 Oral tribology is the study of friction and lubrication between two interacting surfaces, such  
74 as teeth–teeth, tongue–palate, tongue–teeth, tongue–food, lips, lips–food, bolus–palate, food  
75 particles–oral surfaces that are in relative motion in the oral cavity<sup>8, 9</sup>. Coefficient of friction  
76 and its relation to sensory smoothness and slipperiness in food research domain was first  
77 detailed by Kokini and co-workers<sup>10</sup> in 1977. The term “lubrication” as a determinant of food  
78 bolus formation and swallowing was used by Hutchings and Lillford<sup>11, 12</sup> after nine years.  
79 Lubrication in mouth was proposed to be dependent on saliva coating the oral surfaces before  
80 eating. Post food consumption, the changing properties of food and its interaction with the in-  
81 mouth environment was hypothesized to be the driver of oral lubrication. However, it is only  
82 recently that there has been an upsurge in research efforts in oral tribology, which can be  
83 evidenced by a power-law behaviour in the distribution of citations received by scientific papers  
84 over the last 10 years (Figure 1). Particularly, there has been some recent efforts to relate oral  
85 friction to sensory characteristics of “astringency”<sup>13-16</sup>, latter is an important quality  
86 characteristic in wine.

87 Hence, this review is aimed to provide a concise update on studies employing oral tribology  
88 as a quantitative tool to predict wine astringency. Firstly, we have provided a brief introduction  
89 on oral tribology with respect to definition and relevance of the tribo-pairs (i.e. pair of materials  
90 used to create the contact surfaces), load (i.e. tongue pressure against the hard upper palate) and  
91 chemistry of the “biolubricant” saliva. Then we have specifically focussed on wine and its  
92 components (polyphenols), which interact with saliva. Finally, we have provided an update of  
93 how tribology has been used as a tool to determine the loss of salivary lubricity on exposure to  
94 wine polyphenols and highlighted the research gaps in this area.

95     **Oral tribology**

96       The key parameter of tribology measurement is the friction coefficient, calculated as the  
97       ratio of the measured friction force against the normal load (Figure 2a)<sup>8, 9, 17-19</sup>. When two  
98       surfaces are in the relative motion at a steady speed of V, the frictional force ( $F_R$ ) can be  
99       expressed as  $F_R = \mu \times F_L$ , where  $\mu$  is the friction coefficient (dimensionless) and  $F_L$  is the  
100      normal force. Lubrication is a surface property, and the magnitude of  $\mu$  thus depends on the  
101      surface roughness and geometry of the interacting surfaces as well as nature of lubricant. A  
102      typical tribometer with ball on a rotating disc configuration during sliding is illustrated in Figure  
103      2b.

104       The friction coefficient is dependent on the lubricant film thickness ( $\delta$ ) between the two  
105      moving surfaces and is typically presented in a Stribeck curve (Figure 2c)<sup>17</sup>. The distinct friction  
106      scenarios that can occur between the tongue and palate is represented by three different regimes:  
107      the boundary regime, the mixed regime and the hydrodynamic regime. Details of these regimes  
108      can be found in previous reviews<sup>8, 17</sup>.

109     **Role of tribo-pairs and loads**

110       In order to understand the complex oral system (oral surfaces, saliva or saliva-wine mixtures as  
111      the lubricants), researchers have used different metallic, crystalline, polymeric and animal  
112      tissue-based tribo-pairs to mimic the topologies of real human tongue and oral palate. Pin-on-  
113      disc, ball-on-dics tribometers with tribo-pairs made up of steel<sup>20</sup>, tetrafluorethylene and  
114      zirconia<sup>21</sup>, glass<sup>22</sup> surfaces in a sliding or rotating configurations have been used. However, as  
115      one might imagine, contrasting to these surfaces, oral surfaces may vary significantly from  
116      highly keratinized bony palate to soft and rough tongue with papillae being in of order 20-100  
117       $\mu\text{m}$ <sup>12, 23</sup>.

118       Innovative approaches, such as everted dried dead tongues of pigs/ piglets have been also used  
119      in tribometers to represent human tongue surfaces<sup>12, 24</sup>. Besides ethical constraints, lack of

120 information about surface chemistry and biological heterogeneity of using animal tissues,  
121 papillae of the dried pig tongue ex vivo was not firm and erect during tribology measurements,  
122 which might be attributed to the biochemical changes (post-mortem) or dehydration process.  
123 Furthermore, the dead animal tissues were less hydrophobic and lubricating as compared to the  
124 living surfaces<sup>12, 25</sup>. It is also worth recognizing that the diameter of the hairs of the human  
125 filiform papillae (27 µm) is larger than that of the pig tongue (18 µm)<sup>26</sup>. Hence, the surface  
126 roughness of these dried animal tissue surface used in the tribology measurement was not  
127 representative of the real human tongue surface. Hence, the friction measurement interpretation  
128 for human tongue needs to be taken with precaution.

129 Instead of “hard” metallic surfaces and animal tissues, soft elastomeric substrates, such  
130 as polydimethylsiloxane (PDMS) that can be deformed by contact pressure are currently  
131 preferred as tribo-pairs<sup>19, 27, 28</sup>. Although tongue surface is significantly rougher than smooth  
132 PDMS surfaces, PDMS surfaces can be modified in deformability, roughness and  
133 hydrophobicity to represent tailored oral surfaces. For example, the hydrophobicity of PDMS  
134 surfaces can be tuned using plasma oxidation, surface coating with functional groups or layer-  
135 by layer<sup>29-31</sup>.

136 “Loads” in oral tribology context can be defined as the normal force that the tongue  
137 exerts on the hard upper palate. As compared to typical mechanical engineering context, a lower  
138 range of loads (1-10 N) has been used in oral tribology studies<sup>19, 27</sup>. Measurements of the loads  
139 of the tongue against the upper hard palate generally ranges from 0.01-90 N<sup>32</sup>. It is worth noting  
140 that the tongue pressure distribution is not uniform across different parts of the tongue-oral  
141 palate contacts and the load distribution might also vary with time<sup>33</sup>. Tongue pressure might  
142 also differ depending upon the population used for study, for instance, elderly population show  
143 significantly lower tongue pressures than younger adults group<sup>34-37</sup>. Hence, oral tribology study

144 for a particular wine consumer group needs to be carried out at a range of relevant loads rather  
145 than a single-point load to represent different oral conditions.

146 **Saliva: The potent “bio-lubricant”**

147 Saliva is composed of water (99.5%), proteins (highly glycosylated mucins, proline-rich  
148 proteins and enzymes, such as  $\alpha$ -amylase) (0.3%), and inorganic substances (0.2%) with pH  
149 around 6.8<sup>38-40</sup>.

150

151 Formation of salivary mucosal pellicle

152 Salivary mucosal pellicle is a viscoelastic gel that protects the oral mucosa from mechanical  
153 and chemical damages, such as exposure to microorganism, toxic materials, environmental  
154 insult, dehydration of oral mucosal epithelium and lubrication. The most prominent constituent  
155 of oral pellicles are mucins, a high molecular weight glycoprotein<sup>41 42</sup>. As Figure 3 shows,  
156 salivary mucosal pellicle comprises of two phases, an immobile pellicle retained on epithelial  
157 cells (membrane associated mucins: MUC1, MUC3, MUC4, MUC12) and a mobile salivary  
158 film (secreted soluble mucins: MUC2, MUC5A, MUC5B, MUC6, MUC7)<sup>43-46</sup>.

159 The MUC5B (high molecular weight) and MUC7 (low molecular weight) are the most  
160 important glycoproteins with regards to lubrication. Saliva is secreted to maintain saliva pellicle  
161 thickness of ~70-100  $\mu\text{m}$ <sup>47</sup>, but vary depending upon the oral location. The oral mucosa where  
162 saliva pellicle is created is generally hydrophobic until the salivary proteins bind. Upon  
163 adsorption to the tongue (hydrophobic), glycoproteins tend to bind with their hydrophobic sites  
164 towards the tongue, whilst hydrophilic sites point outwards for water retention. Salivary film  
165 reduces the “ $\mu$ ” in oral surfaces. Using AFM, human salivary pellicles have been shown to  
166 reduce the  $\mu$  by a factor of 20 between hard contact surfaces<sup>48</sup>, having  $\mu \approx 0.02$  i.e. two orders  
167 of magnitude lower than that of water<sup>49</sup>.

168

169 Use of saliva in oral tribology studies  
170 Use of saliva is becoming popular in oral tribological measurements in food research as saliva  
171 is a key “biolubricant” that can reduce “ $\mu$ ” significantly within the human oral surfaces.  
172 However, such lubricating properties of saliva (ex vivo) can vary significantly depending upon  
173 stimulation (unstimulated, mechanical, acid), collection (protein-binding properties and air  
174 exposure) and usage (immediate use, freeze-thaw-induced precipitation)<sup>50, 51</sup>. Also, within an  
175 individual, salivary protein amount varies and acidic and glycosylated proline-rich-proteins  
176 PRPs (gPRPs and aPRPs) may vary significantly throughout the day and is highly dependent  
177 on the type of food ingested<sup>52</sup>. Other factors influencing interactions with wine are pH,  
178 buffering capacity and concentrations of calcium and phosphate in saliva, latter shows huge  
179 variation over a day in unstimulated whole saliva<sup>53</sup> and even depends on how saliva has been  
180 handled after collection<sup>51</sup>.

181 The friction coefficient of stimulated and unstimulated saliva measured between two  
182 mucosal surfaces using loads (0.34-2.20 N) showed decrease of  $\mu$  with increase in load and  
183 speed for both types of saliva<sup>54, 55</sup>. The differences in  $\mu$  were due to the protein content and  
184 rheological properties of saliva, particularly, stimulated saliva produced by sublingual and  
185 submandibular gland had a higher protein content and lower viscosity as compared to  
186 unstimulated saliva<sup>54</sup>.

187 Saliva also changes its composition along the salivary film (Figure 3), and until now,  
188 the “mobile salivary phase” has only been studied. However, the most important lubricating  
189 proteins (MUC5B and MUC7) still remain attached to the mucosal epithelia even if the salivary  
190 film is ruptured. As these mucins may be important to understand “astringency”, it might be  
191 worth to consider collecting saliva from parotid glands or gently scraping the immobile salivary  
192 pellicle from the oral surfaces of the participants after ethics approval for tribological  
193 measurements.

194           Finally, the use of “artificial saliva” i.e. fluid mimicking the ionic composition, mucin  
195          and rheological properties of unstimulated human saliva has been quite common due to its ease  
196          of preparation and reproducibility<sup>38-40</sup>. However, the term “artificial saliva” has been argued by  
197          several authors as there has been no bio-mimetic that accurately simulates all of the properties  
198          of saliva<sup>50</sup>. In a recent study by Laguna and coworkers<sup>19</sup>,  $\mu$  of artificial saliva was measured in  
199          a PDMS-PDMS ball-dics set-up and the Stribeck pattern was found to be similar to real human  
200          saliva (unstimulated)<sup>54</sup>. Hence, use of at least mucin in a mimicked ionic composition can be a  
201          good starting point to understand wine-saliva interaction as compared to that without  
202          consideration of any aspects of salivary lubrication.

203

#### 204          **Wine and astringency**

205          Wines, derived from fermented grapes<sup>56</sup> (*Vitis Vinifera*) are essentially composed of 80-85%  
206          water, 9-20% ethanol and other minor compounds, such as phenolic compounds, esters, acids,  
207          nitrogenous compounds, volatiles, lipids, mineral salts etc. A well balanced-wine should  
208          contain optimum primary taste components (i.e. balance between sweetness and acidity), tactile  
209          elements (i.e. astringency) and flavour<sup>57</sup>. Among the different textural attributes, astringency  
210          has been considered to be one of the most important sensory characteristic in red wines.

211          Since astringency can be perceived in the mouth where no taste receptors are present,  
212          it is considered to be tactile rather than a taste stimulation, contrary to the initial speculations<sup>6</sup>.  
213          Different phenolic compounds show different affinities towards human salivary proteins<sup>58</sup>.  
214          Polyphenols with extended structure have been reported to have higher affinity to PRPs<sup>58, 59</sup>. In  
215          other words, smaller polyphenols can bind with one phenolic ring, whilst larger polyphenols  
216          interact in a multi-dentate fashion, occupying two or three consecutive prolines increasing the  
217          degree of salivary protein precipitation. Despite the chemical differences in phenolic  
218          components, the astringency of polyphenols mixtures with different structures, such as,

phenolic acids and catechins were perceived to be of same astringency by a trained sensory panel<sup>60</sup>. The total phenolic content of wines depends on many factors and it can vary from 900-1400 mg/L in young red wines and 1600-2500 mg/L in aged red wines<sup>61</sup>. Astringency feeling evolves during aging, and it is generally higher in young wines and decreases over time, “softening” the wine. This is caused presumably by the soluble pectin fragments, associated with the grapes that might inhibit protein-tannin interactions and pectin might aggregate or encapsulate the tannins making the latter unavailable to the salivary proteins<sup>62, 63</sup>.

Three different mechanisms of wine astringency has been hypothesized that complement each other: protein precipitation, rupture of the lubricating salivary film and formation of mouth debris<sup>64</sup> (Figure 3). Firstly, wine polyphenolic compounds form complexes with salivary proteins, specially PRPs<sup>65</sup> due to hydrophobic interactions and hydrogen bonding, precipitating the salivary proteins and decreasing its viscosity<sup>66</sup>, latter affecting the integrity of the salivary film. Hydrogen bonding occurs between hydroxyl groups of phenolic compounds and carbonyl and amide group of the salivary protein, whereas hydrophobic interactions occur between the benzoic ring of phenolic compounds and the apolar side chains of amino acids such leucine, lysine or proline in the salivary proteins<sup>62</sup> (Figure 3). The rupture of the lubricating saliva film activates the mechanoreceptors, located within the mucosa connected with the trigeminal nerve that then transmits to brain the perception of astringency<sup>67</sup>. Furthermore, the increase in precipitated salivary proteins and other debris in saliva increases the sense of “discrete particles” in the mouth, which essentially relates to roughness and oral friction<sup>23</sup>. Due to the strong correlation between astringency perception and formation of insoluble salivary protein-wine polyphenol complexes, research has focused in finding analytical methods for quantification/qualification of these complexes. In the next section, we only focus on recent studies that used Stribeck curves to quantify astringency.

243

244 **Relevance of oral tribology to unravel wine astringency**

245 Salivary proteins are widely separated from each other due to mutually repulsive forces of  
246 negatively charged mucins<sup>38</sup> at neutral pH in saliva, latter is a highly diluted system<sup>23</sup>. However,  
247 when tannic acid was added, large flocs appeared in saliva (approx. 300 µm) (Figure 4a)<sup>23</sup>. In  
248 red wines-saliva mixtures, similar aggregates have been recently observed using light and  
249 transmission electron microscopy (Figure 4b)<sup>14</sup>. Furthermore, the microstructure of such  
250 aggregates varied depending upon the wine type and their polyphenol composition, specifically  
251 proanthocyanidin (grape skin) and tannin (seeds). Cabernet Sauvignon wines presented  
252 densely-packed aggregates whereas Carménère, Merlot wines showed smaller aggregates with  
253 much more open structure (Figure 4b)<sup>14</sup>. However, irrespective of the type of wines<sup>14</sup>, wine-  
254 saliva mixtures showed a significantly higher  $\mu$  as compared with human saliva in the boundary  
255 regime using a PDMS-steel contact surfaces (Figure 4c). Authors reported a high correlation  
256 ( $R^2=0.93$ ) between  $\mu$  and sensory “astringency” at a sliding speed of 0.075 mm/s linking  
257 astringency to salivary protein depletion by wine polyphenols.

258 In a separate study, tribological analysis in a soft PDMS ball/model mucin-adsorbed  
259 glass surface<sup>16</sup> indicated that astringency may arise from the temporary failure of the boundary  
260 lubrication of the adsorbed mucins by tannic acid. This loss of boundary lubrication showed  
261 concentration dependency on tannic acid. Authors suggested that interaction with tannic acid  
262 molecules might result in the change in conformation and hydration of adsorbed mucin, both  
263 leading to the marked rise in friction force. This is in agreement with a previous report, where  
264 “chemically pure” polyphenol (epigallocatechin gallate) appeared to partially deplete the thin  
265 lubricating human salivary film (mechanically stimulated whole saliva) from the smooth  
266 PDMS-PDMS contact surfaces in a tribological experiment performed at 37 °C<sup>15</sup>. This induced  
267 an increase in  $\mu$  and was correlated to a certain extent with the astringency perception.

268        Besides phenolic compounds, there are other components in wine, which can also  
269        contribute to astringency perception. For example, tartaric acid present in wines are known to  
270        lower the pH of wine significantly, which precipitates the salivary proteins<sup>2</sup> as well as increase  
271        the binding affinities of the salivary proteins with polyphenols. In contrast, the presence of  
272        ethanol in wine has been reported to modify the degree of hydrogen bonding between  
273        polyphenols and salivary proteins. This tends to modify the degree of protein folding and  
274        solubility of tannins<sup>68</sup>. Another key component in wines i.e. glycerol has been associated with  
275        oiliness, persistence and mellow ness<sup>69</sup>. Interestingly, tribological measurements of aqueous  
276        solutions of glycerol in steel tribo-pairs (ball/ disc) have suggested glycerol to be a potential  
277        “green lubricant” with its lubricating properties being better than those of rapeseed oil. Hence,  
278        contribution of wine components other than polyphenols in astringency should not be  
279        underestimated and the complex interplay of polyphenol, pH, ethanol, glycerol in wine  
280        astringency needs further investigation from tribological viewpoint<sup>70</sup>.

281

## 282        **Conclusions**

283        In summary, astringency studies in wine essentially rely on sensorial methods so far. Interaction  
284        between polyphenols in wine and salivary proteins is generally considered to be the main  
285        mechanism inducing astringency sensation. Oral tribology is a relatively recent approach that  
286        has been used to quantitatively study the loss of lubricity of saliva on exposure to polyphenols.  
287        Measurement of coefficient of friction of wine and specific polyphenols at certain sliding  
288        speeds have shown some correlation with sensory perception of astringency. This shows  
289        potential of oral tribology measurement as a promising quantitative tool for analysing  
290        astringency perception. However, lubrication is a surface property. Hence, the friction  
291        coefficient not only depends on the mechanical properties of the lubricant (e.g. saliva) but also  
292        on the surfaces used in tribology measurement to represent the tongue and the upper palate.

293 Currently, the contact surfaces used in oral tribology range from steel to glass to PDMS. The  
294 key requirement is the accurate development of bio-relevant tribo-contact surfaces that  
295 effectively represent the soft, micro-patterned tongue and bony upper palate surfaces. Use of  
296 accurate loads representative of real human tongue pressure values when consuming  
297 polyphenol-rich food need to be used in such measurements. Use of relevant tribo-pairs and  
298 loads need to be standardized across different laboratories to have comparable results. Most  
299 importantly, these quantitative friction measurements need appropriate correlation with sensory  
300 perception using trained sensory panel, before such quantification can be of use to characterize  
301 astringency in wine and other polyphenol rich foods at a commercial level.

302

### 303 **Notes on Contributors**

304 **Dr. Laura Laguna** works for the Spanish National Research Council (CSIC) at the Institute of  
305 Food Research (CIAL, Madrid). Over the last years, her research has been focused on the study  
306 of food oral processing, nowadays with special emphasis on studying the sensory perception of  
307 wine components.

308 **Dr. Anwesha Sarkar** is a Lecturer of Food Colloids at the University of Leeds. She has more  
309 than 36 scientific papers and 4 patents. Research interests: colloidal design (emulsion, emulsion  
310 gels, microgel, particles, protein complexes, coacervates), oral tribology, lubrication in soft  
311 contacts, multi-scale structural analysis, in vitro digestion.

312

### 313 **Disclosure statement**

314 No potential conflict of interest was reported by the authors.

315

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