



UNIVERSITY OF LEEDS

This is a repository copy of *Reduced anthropogenic aerosol radiative forcing caused by biogenic new particle formation*.

White Rose Research Online URL for this paper:
<http://eprints.whiterose.ac.uk/107267/>

Version: Accepted Version

Article:

Gordon, H orcid.org/0000-0002-1822-3224, Sengupta, K, Rap, A orcid.org/0000-0002-2319-6769 et al. (78 more authors) (2016) Reduced anthropogenic aerosol radiative forcing caused by biogenic new particle formation. *Proceedings of the National Academy of Sciences*, 113 (43). pp. 12053-12058. ISSN 0027-8424

<https://doi.org/10.1073/pnas.1602360113>

This is an author produced version of a paper published in *Proceedings of the National Academy of Sciences*. Uploaded in accordance with the publisher's self-archiving policy. In order to comply with the publisher requirements the University does not require the author to sign a non-exclusive licence for this paper.

Reuse

Items deposited in White Rose Research Online are protected by copyright, with all rights reserved unless indicated otherwise. They may be downloaded and/or printed for private study, or other acts as permitted by national copyright laws. The publisher or other rights holders may allow further reproduction and re-use of the full text version. This is indicated by the licence information on the White Rose Research Online record for the item.

Takedown

If you consider content in White Rose Research Online to be in breach of UK law, please notify us by emailing eprints@whiterose.ac.uk including the URL of the record and the reason for the withdrawal request.



eprints@whiterose.ac.uk
<https://eprints.whiterose.ac.uk/>

10 chamber show that particle formation in atmospheric conditions can
11 occur solely from biogenic vapours. Here we evaluate the potential
12 effect of this new source of particles on pre-industrial cloud conden-
13 sation nucleus (CCN) concentrations and on aerosol-cloud radiative
14 forcing over the industrial period. Model simulations show that the
15 pure biogenic particle formation mechanism has a much larger relative
16 effect on CCN concentrations in the pre-industrial atmosphere than
17 in the present atmosphere because of the lower aerosol concentrations.
18 Consequently, pre-industrial cloud albedo is increased more than under
19 present-day conditions, so the cooling forcing of anthropogenic
20 aerosols is reduced. The new mechanism increases CCN concentra-
21 tions by 20-100% over a large fraction of the pre-industrial lower at-
22 mosphere and the magnitude of annual global mean radiative forcing
23 caused by changes of cloud albedo since 1750 is reduced by 0.22 Wm^{-2}
24 (27%) to -0.60 Wm^{-2} . Model uncertainties, relatively slow formation
25 rates and limited available ambient measurements make it difficult to
26 establish the significance of a mechanism that has its dominant effect
27 under pre-industrial conditions. Our simulations predict more particle
28 formation in the Amazon than is observed. On the other hand, the
29 first observation of pure organic nucleation has now been reported for
30 the free troposphere. Given the potentially significant effect on an-
31 thropogenic forcing, effort should be made to better understand such
32 naturally-driven aerosol processes.

33 1 Significance Text

34 A new mechanism for the formation of atmospheric aerosols via the gas-to-
35 particle conversion of highly oxidised organic molecules is found to be the
36 dominant aerosol formation process in the pre-industrial boundary layer over
37 land. The inclusion of this process in a global aerosol model raises baseline
38 pre-industrial aerosol concentrations, and could lead to a reduction of 27%
39 in estimates of anthropogenic aerosol radiative forcing.

40 2 Article

41 Measurements in the CERN CLOUD chamber under atmospheric conditions
42 show that new particles can form purely from the oxidation products of α -
43 pinene, a compound emitted by the biosphere [1]. Nucleation of new aerosol

44 particles via gas-to-particle conversion has been studied for fifty years [2] and
45 is responsible for around half of global cloud condensation nuclei (CCN) [3],
46 which affect Earth’s radiation balance via aerosol-cloud interactions. The
47 involvement of oxidised organic molecules in the process, alongside sulphuric
48 acid, was proposed in early studies, and has been well-established for some
49 time [4, 5]. The new mechanism for organic particle formation without sul-
50 phuric acid presented in Ref. [1] could be important for Earth’s climate be-
51 cause it provides a way to form particles in the pristine pre-industrial atmo-
52 sphere, when the concentrations of sulphuric acid and ammonia were much
53 lower. The pre-industrial environment forms the baseline for calculations in
54 global models of the radiative forcing caused by anthropogenic emissions [6],
55 and uncertainties in this baseline are the largest component of the overall
56 uncertainty on aerosol radiative forcing [7]. This is because an incremental
57 increase in particle concentrations when they are low has a much stronger
58 radiative effect than when they are high. Previous model uncertainty anal-
59 yses suggested that the sensitivity of radiative forcing to particle formation
60 rates is low compared to many other factors [7]. However, these studies var-
61 ied the nucleation rate assuming that sulphuric acid is required for particle
62 production. Here we show that the inclusion of a new nucleation mechanism
63 that does not require sulphuric acid could have a more significant effect on
64 radiative forcing than previously thought [8, 7].

65 Our modelling study is inspired by and based on measurements in which
66 α -pinene, a volatile organic compound (VOC) emitted into the atmosphere
67 by vegetation, was oxidised by ozone and hydroxyl radicals in the CLOUD
68 chamber under ultra-clean conditions without sulphuric acid [1]. The mass
69 spectra of the highly oxidised multifunctional organic molecules (HOMs) pro-
70 duced from the VOCs closely resemble those observed in the atmosphere [9].
71 Therefore, while the concentrations of some reactive gases in the chamber
72 do not perfectly match those in the troposphere, we have confidence in our
73 assumption that the chamber results can be generalised to the atmosphere.
74 Particle counters show that typical atmospheric concentrations of the HOMs
75 produce particles at significant rates, even when sulphuric acid is absent from
76 nucleating clusters. We describe this process as pure biogenic nucleation.

77 In this paper, we examine the implications of pure biogenic nucleation
78 for atmospheric aerosol and Earth’s radiation balance using the GLOMAP
79 global model of aerosol microphysics [10]. A parametrisation of the pure
80 biogenic nucleation rate that depends on the HOM concentration and the
81 concentration of ions is provided in supplementary materials of Ref. [1]. We

82 assume for this study that this can be added linearly to parametrisations
 83 of the nucleation rate involving sulphuric acid only [11] and sulphuric acid
 84 with organics similar to HOMs [5]. Ref. [1] also provided the yields of HOMs
 85 from the oxidation of α -pinene by ozone (2.9%) and by the hydroxyl radical
 86 (1.2%). The yield of HOM from endocyclic monoterpenes such as α -pinene is
 87 higher than that from exocyclic monoterpenes, so we separate these classes in
 88 our model and use the yields from β -pinene in Ref. [12] to produce HOM from
 89 exocyclic monoterpenes. The rate of formation of 1.7 nm diameter aerosols by
 90 gas-to-particle conversion is therefore described by the sum of the following
 91 parametrisations:

92 1. Binary homogeneous nucleation of sulphuric acid and water [11].

93 2. Nucleation of organics with sulphuric acid [5], also used in Ref. [13]:

$$J_{\text{sa-org}} = k_{\text{sa-org}}[\text{H}_2\text{SO}_4]^2[\text{BioOxOrg}] \quad (1)$$

94 where BioOxOrg refers to the oxidation products of monoterpenes with
 95 OH and $k_{\text{sa-org}} = 3.27 \times 10^{-21} \text{ cm}^6\text{s}^{-1}$ (see Methods).

96 3. Pure biogenic nucleation, a sum of neutral (J_n) and ion-induced (J_{iin})
 97 components [1]:

$$J_{\text{org}} = J_n + J_{\text{iin}} \quad (2)$$

$$J_n = a_1[\text{HOM}]^{a_2+a_5/[\text{HOM}]} \quad (3)$$

$$J_{\text{iin}} = 2[n_{\pm}]a_3[\text{HOM}]^{a_4+a_5/[\text{HOM}]} \quad (4)$$

98 where HOMs are produced as described above but given here for con-
 99 venience in units of 10^7 molecules per cubic centimetre, n_{\pm} is the ion
 100 concentration and a are free parameters. Ions in the model are pro-
 101 duced from radon and galactic cosmic rays (see SI Appendix).

102 Ammonia and amines can also contribute to nucleation by stabilising sul-
 103 phuric acid clusters, but the binary homogeneous mechanism has been shown
 104 to be a reasonable representation of free tropospheric nucleation [14], and nu-
 105 cleation at low altitudes involving amines or ammonia is important only in
 106 polluted regions where the changes in radiative forcing calculated here are
 107 very insensitive to nucleation rates.

108 In our model, aerosols formed in this way, and those emitted directly from
 109 Earth's surface, grow by condensation and coagulation, are transported in

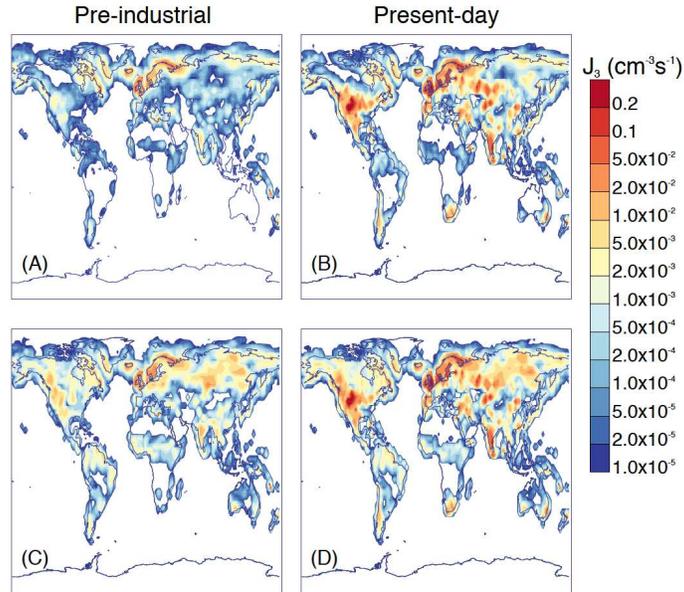


Figure 1: Nucleation rates at 3 nm diameter (J_3 , $\text{cm}^{-3}\text{s}^{-1}$) within approximately 500 m of the surface averaged over June without pure biogenic nucleation in (A) pre-industrial and (B) present-day conditions, and with pure biogenic nucleation in (C) pre-industrial and (D) present-day conditions.

110 the atmosphere, and are ultimately removed by dry or wet deposition. We
 111 consider the radiative forcing between 1750 and 2008 via the effect of these
 112 aerosols on cloud albedo, which is evaluated at the top of the atmosphere
 113 (0.03 Pa atmospheric pressure). To determine the effects of pure biogenic nu-
 114 cleation, particle formation rates, aerosol concentrations and radiative forcing
 115 from model runs with and without mechanism 3 are compared.

116 3 Biogenic nucleation rates and observational 117 evidence

118 Fig. 1 shows the effect of pure biogenic nucleation on the pre-industrial and
 119 present-day atmospheres. When sulphuric acid is required for nucleation
 120 to proceed, substantially less nucleation is expected for pre-industrial times
 121 (Fig. 1A) compared with the present (Fig. 1B). However, when pure bio-
 122 genic nucleation is included, the nucleation rates in pre-industrial (C) and
 123 present-day times (D) become more similar. While pure biogenic nucleation

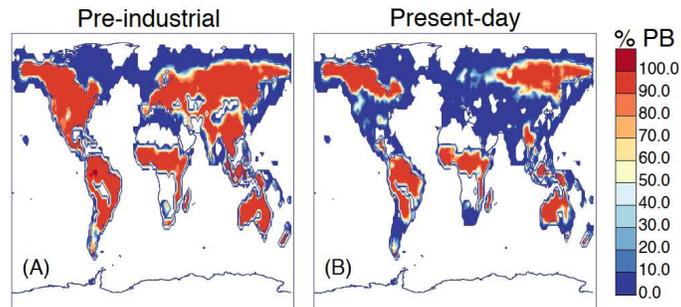


Figure 2: Percentage of particles produced via pure biogenic (PB) nucleation within approximately 500 m of the surface, averaged over June in (A) pre-industrial and (B) present-day conditions. We note that our model predicts large changes to particle formation at the surface and very little change above the boundary layer.

124 is much less important today (compare the change from B to D with that
 125 from A to C), it is still expected to be significant in some continental regions
 126 remote from pollution, for example boreal regions, Australia and, according
 127 to our simulations (discussed later), the Amazon. Within around 500 m of
 128 the surface pure biogenic nucleation increases total production of particles of
 129 at least 3 nm in diameter via nucleation by 2.1% globally in the present-day
 130 atmosphere, but by 90% in pre-industrial conditions.

131 Fig. 2 shows that pure biogenic nucleation is predicted to be the domi-
 132 nant mechanism for particle formation over large parts of the land surface
 133 above 50°N in summer even in the present-day. However, both pure biogenic
 134 and sulphuric acid particle formation rates are often insufficient to produce
 135 detectable nucleation events (see SI Appendix, Fig. S4). Pure biogenic nucle-
 136 ation has more effect in June than in January because terpene emissions are
 137 higher in June. The diurnal cycles of nucleation rates at Hyytiälä and Pal-
 138 las in Finland, shown in SI Appendix Fig. S4, indicate that nucleation rates
 139 in these areas are occasionally higher than around $0.1 \text{ cm}^{-3} \text{ s}^{-1}$. Experience
 140 from these boreal forest sites [15] suggests that nucleation rates above this
 141 value will result in detectable nucleation events. This is confirmed by the
 142 modelled size distributions shown in SI Appendix Fig. S6. As is observed,
 143 simulated nucleation rates are substantially higher during the day than at
 144 night.

145 To our knowledge, Hyytiälä and Jungfraujoch are the only locations with
 146 published measurements from the APi-TOF and CI-APi-TOF mass spec-

147 trometers needed to unambiguously detect pure biogenic nucleation [15]. There
148 is strong evidence in Ref. [16] that pure organic nucleation proceeds alongside
149 sulphuric acid-driven nucleation at Jungfraujoch. For example, their Fig. 2
150 shows that, on the Nucleation Day 3, most organic clusters of masses of up
151 to 400 amu contain no sulphuric acid, there is no inorganic nucleation, and
152 the nucleation rate exceeds $10 \text{ cm}^{-3} \text{ s}^{-1}$ when sulphuric acid concentrations
153 are less than $5 \times 10^5 \text{ cm}^{-3}$.

154 There are no measurements of pure biogenic nucleation so far from Hyytiälä
155 since almost all the nucleation rates measured in Ref. [15] are at $[\text{H}_2\text{SO}_4] >$
156 $1 \times 10^6 \text{ cm}^{-3}$. Observations at Hyytiälä were, however, used alongside those
157 from Melpitz and Hohenpeissenberg to derive parameterizations of particle
158 formation rates in Ref. [17]. The authors found that nucleation could be
159 described well by

$$J_2 = k_1[\text{H}_2\text{SO}_4]^2 + k_2[\text{H}_2\text{SO}_4][\text{org}] + k_3[\text{org}]^2, \quad (5)$$

160 for constant k_{1-3} , suggesting that pure biogenic nucleation is a statistically
161 detectable component of nucleation in these environments.

162 In addition to the Jungfraujoch observations, there is extensive circum-
163 stantial evidence for pure biogenic nucleation. The Amazon, where the lowest
164 SO_2 concentrations over land are found, is an obvious place to look. While
165 some nucleation mode particles are seen in pristine regions of the Amazon [18]
166 (on 19% of days sampled in the study referenced), no clear nucleation events
167 or conclusive evidence for biogenic nucleation have yet been published, and
168 growth of nucleation mode particles to CCN size is rarely observed there.
169 Our model does not produce Hyytiälä-like nucleation events (see SI Ap-
170 pendix Figs. S5-S7) but it does predict non-zero particle formation rates.
171 It slightly overestimates CCN concentrations compared to Ref. [19] in the
172 Amazon even without pure biogenic nucleation, and pure biogenic nucle-
173 ation further increases the discrepancy, by around a factor two. This may
174 point to a chemical suppression of HOM yields by isoprene [20] or NO_x [21],
175 but could also be due to other sources of model error, for example, under-
176 estimation of particle size and therefore condensation sink. Overprediction
177 of particle concentrations over the Amazon seems to be a common feature
178 among models [22]. Comparing models with observations in this region is
179 challenging due to large uncertainties in emissions of biogenic VOCs and a
180 complex wet scavenging environment.

181 Pure biogenic nucleation is also predicted to be the dominant source of
182 secondary particles in the cleanest high latitude boreal regions. Low SO_2

183 concentrations, often below 100 ppt, and nocturnal nucleation were reported
184 in a study at Värriö, Finland (67°N) at similar temperatures to the CLOUD
185 chamber [23]. Similar observations of nocturnal nucleation were made at
186 Abisko, Sweden [24] and Tumbarumba, Australia [25], although SO₂ concen-
187 trations were not reported. At Pallas, Finland, H₂SO₄ concentrations are
188 reported below $3 \times 10^5 \text{cm}^{-3}$ in a large number of new particle formation
189 events [26]. The air masses in Pallas are usually of marine origin, which
190 leads to low condensation sinks favourable to nucleation, but may also allow
191 halogens of marine origin to locally influence nucleation. Three instances of
192 new particle formation with $[\text{H}_2\text{SO}_4] < 3 \times 10^5 \text{cm}^{-3}$ shown in Ref. [26] Fig.
193 6 are unambiguously continental. This should also allow the contribution
194 of halogens to be excluded, making it highly likely the nucleation was pure
195 biogenic.

196 With only sparse or indirect observational evidence for pure biogenic nu-
197 cleation, an alternative strategy is to compare modelled particle concentra-
198 tions against observations. However, this is also inconclusive because there
199 are many compensating causes of model error [7], making attribution of bi-
200 ases ambiguous. Substantial changes in total particle number concentration
201 are caused by pure biogenic nucleation (SI Appendix Fig. S2). However,
202 when we compare the monthly mean model predictions to particle number
203 concentrations at 37 surface sites [27, 28], and the daily mean concentrations
204 to those measured during the ARCTAS aircraft campaign [29] in 2008 (SI Ap-
205 pendix Figs. S8 and S9), we find that the effect of pure biogenic mechanism,
206 increasing summertime particle concentrations by up to a factor 2, is also
207 comparable to, or smaller than, existing discrepancies between observations
208 and the model.

209 4 Impact on CCN and radiative forcing

210 Fig. 3 shows the effect of pure biogenic nucleation on present-day and pre-
211 industrial CCN concentrations, calculated at 0.2% supersaturation. When
212 pure biogenic nucleation is included, global annual average concentrations of
213 these particles at cloud base level (approximately 600 m altitude) increase by
214 4% in the present-day and 12% in the pre-industrial atmospheres. Although
215 nucleation rates are affected mostly close to sources of biogenic gases, CCN
216 are affected over much wider areas due to the slower removal rate of larger
217 aerosol particles. This spread is important because it carries the particles to

218 cloudy marine regions where most of the anthropogenic aerosol-cloud radia-
219 tive forcing occurs [30]. The change in CCN production across the pristine
220 pre-industrial atmosphere is particularly important for global climate be-
221 cause cloud droplet concentrations and albedo are both more sensitive to
222 CCN changes in pristine environments.

223 The change in aerosol radiative forcing from 1750 to 2008 attributable
224 to pure biogenic nucleation was calculated by comparing simulations with
225 and without pure biogenic nucleation. We only consider changes in the cloud
226 albedo effect. The aerosol direct forcing is unlikely to be substantially in-
227 fluenced by the new nucleation mechanism as it is not strongly affected by
228 the aerosol size distribution [31]. The change in radiative forcing when pure
229 biogenic nucleation is included is presented in Fig. 4. We estimate that the
230 global annual mean cloud albedo forcing since 1750, after including pure bio-
231 genic nucleation, is -0.60 Wm^{-2} . The *change* in calculated aerosol radiative
232 forcing due to pure biogenic nucleation is $+0.22 \text{ Wm}^{-2}$, corresponding to a
233 27% reduction in the negative forcing. This change is a result of the non-
234 linear dependence of the forcing on the baseline CCN concentration [7]. We
235 note that our simulations may underestimate the net effect since they do not
236 account for possible increases in cloud fraction and thickness, which, in pris-
237 tine regions (CCN below 100 cm^{-3}), may be highly sensitive to small changes
238 of CCN [32]. We also do not account for the possibility of pure biogenic nu-
239 cleation involving sesquiterpenes. However, we also emphasise that including
240 pure biogenic nucleation in our model leads to an over-prediction of CCN in
241 the Amazon region, which may indicate that it is chemically suppressed. In-
242 hibition of nucleation, if it happens, may be local to the tropical rainforest
243 environment or more widespread. If we artificially set pure biogenic nucle-
244 ation rates to zero within 10° latitude of the Equator, the effect on aerosol
245 forcing when pure biogenic nucleation is included changes only slightly, to
246 $+0.20 \text{ Wm}^{-2}$.

247 The largest changes in radiative forcing occur over the NH, especially over
248 oceans with high annual cloud cover (Fig. 4b) where CCN concentrations are
249 most strongly perturbed by anthropogenic emissions. The NH is also where
250 pure biogenic nucleation causes the largest reduction in contrast between
251 pre-industrial and present day CCN concentrations driven by the large con-
252 tinental source of biogenic gases. However, the relative change in forcing in
253 the SH is greater than the NH: pure biogenic nucleation reduces the annual
254 southern hemispheric mean from -0.25 Wm^{-2} to -0.14 Wm^{-2} (compared to
255 a change in the NH of -1.39 Wm^{-2} to -1.06 Wm^{-2}). In some tropical and

256 southern regions, there are higher CCN in pre-industrial times than today,
257 and a positive radiative forcing. In these regions and nearby, pre-industrial
258 OH· and HOMs were higher than today and particle condensation sinks were
259 lower, while SO₂ levels (largely marine) were comparable.

260 We consider the principal uncertainties in our analysis to be associated
261 with a) VOC, SO₂ and primary particle emissions as in Ref. [7], b) how
262 representative α -pinene and the pinanediol used in Ref. [5] are of VOCs in
263 the atmosphere, c) yields of HOM from α -pinene oxidation in the presence of
264 other vapours such as NO_x, and d) temperature dependence of the nucleation
265 rates.

266 To investigate the effect of a plausible temperature dependence we re-
267 ran the model multiplying all boundary-layer nucleation rates by $\exp(-(T -$
268 $278)/10)$. The charged nucleation rate remained limited by the ion produc-
269 tion rate and the overall rate by the kinetic limit. We find annually averaged
270 changes to cloud albedo radiative forcing over the industrial period from pure
271 biogenic nucleation are reduced to +0.14 Wm⁻² from +0.22 Wm⁻².

272 The yields of HOM have an experimental uncertainty around a factor
273 two (and were reported to be about a factor two higher in an earlier cham-
274 ber study [33]). These uncertainties are comparable to uncertainties in the
275 VOC emissions themselves [34]. The yields could be affected by nitrogen ox-
276 ides [21], and were found to differ substantially between monoterpenes [12].
277 To test the sensitivity to the uncertainty in yields, which is a proxy for the
278 overall intrinsic uncertainty on the experimental measurements, we repeated
279 our analysis with the yield of the HOMs that participate in pure biogenic
280 nucleation perturbed by a factor 3. This gives an uncertainty range for the
281 increase in CCN due to the pure biogenic mechanism of 4–19% in the pre-
282 industrial and 1–6% in the present-day, as shown in Table S2. The lower
283 limit still leads to a significant change to cloud albedo forcing of 0.10 Wm⁻²
284 when the corresponding parametrisation is added to the model.

285 We have also investigated the sensitivity of our radiative forcing estimate
286 to other sources of uncertainty. We perturb the pre-industrial volcanic SO₂
287 emissions and find this does not strongly affect our reported CCN changes.
288 When we perturb the biomass burning and sea spray emissions (see SI Ap-
289 pendix for details) we find larger changes both to CCN and forcing, especially
290 when emissions are reduced. The model becomes slightly more sensitive to
291 pure biogenic nucleation when different baseline nucleation mechanism from
292 Ref. [17] instead of Ref. [5] is used. The percentage changes to CCN from
293 including pure biogenic nucleation under these scenarios are given in SI Ap-

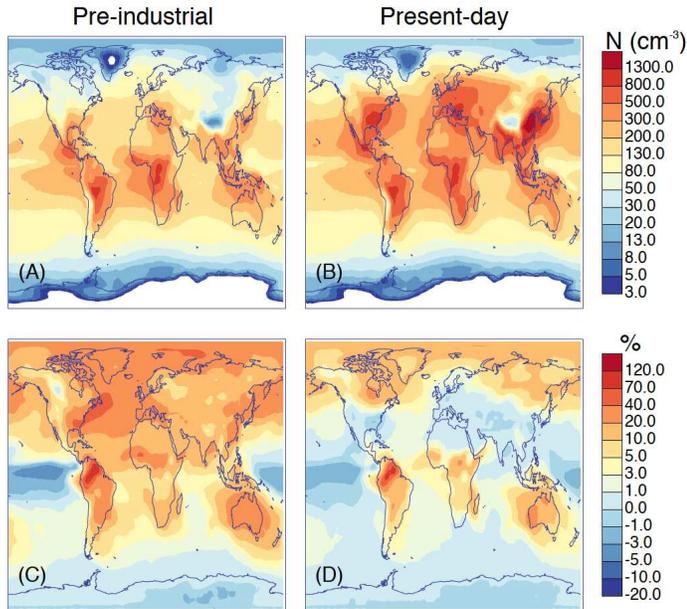


Figure 3: Concentrations of cloud condensation nuclei calculated at 0.2% supersaturation, in cm^{-3} , annually averaged at cloud base level in (A) pre-industrial and (B) present-day conditions, and (C, D) percentage changes to these concentrations when pure biogenic nucleation is introduced. In this Figure we assume HOM formation and pure biogenic nucleation proceed at the rates measured at the CLOUD chamber.

294 pendix Table S2, and the changes to forcing in Table S3.

295 5 Discussion and conclusions

296 Our global aerosol simulations indicate that pure biogenic nucleation [1] dom-
 297 inates particle formation in the pre-industrial boundary layer, producing 59%
 298 of new particles below approximately 500 m altitude and 36% below around
 299 1.5 km. For the organic system, laboratory measurements are currently the
 300 only route to a comprehensive understanding of the processes leading to
 301 particle formation. This is particularly the case for a mechanism that is
 302 difficult to decouple from sulphuric acid-driven nucleation pathways in the
 303 polluted present-day atmosphere. This mechanistic understanding is required
 304 to perform accurate extrapolations from present-day conditions back to the
 305 pre-industrial. Improving such extrapolations is of critical importance as un-

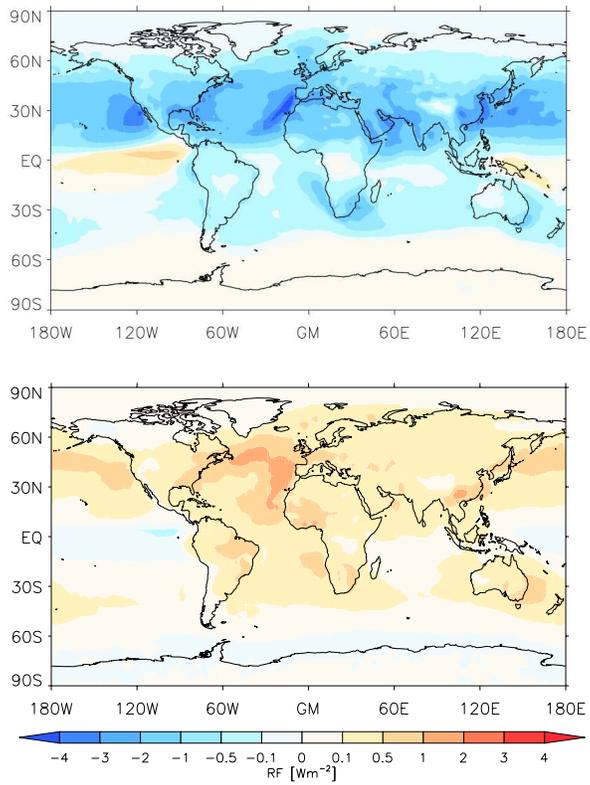


Figure 4: Distribution of (A) cloud albedo radiative forcing and (B) change to this distribution when pure biogenic nucleation is included in the model.

306 certainties in pre-industrial aerosol are a large component of the uncertainty
307 in IPCC estimates of radiative forcing. While nucleation in tropical environ-
308 ments is relatively unimportant for global mean cloud albedo radiative forc-
309 ing in our model, discrepancies between modelled and observed nucleation
310 in these regions suggest further investigation of Amazon aerosol chemistry
311 could significantly improve our understanding of pristine aerosol processes.

312 Based on the nucleation rates reported by CLOUD [1], we show here
313 that pure biogenic nucleation may reduce the magnitude of pre-industrial
314 to present-day aerosol cloud albedo forcing by as much as 0.22 Wm^{-2} , or
315 27%. This change in forcing is greater than the combined one standard
316 deviation uncertainty of twenty-eight parameters related to emissions and
317 aerosol processes in this model [7], which is 19%. Other forcing mechanisms
318 or uncertainties in the results quoted here could still lead to stronger effects.
319 Although the calculated change in forcing is comparable to the model para-
320 metric uncertainty, it shifts the entire probability distribution of forcing,
321 and therefore represents a significant downward revision in the likelihood
322 of high negative aerosol-cloud forcings in this model. Similar revisions are
323 likely to occur in other models [35] due to the same chain of processes: 1)
324 proportionally greater increases in aerosol concentrations in the cleaner pre-
325 industrial atmosphere than in the present day; 2) high sensitivity of cloud
326 albedo and adjustments on the pre-industrial aerosol concentrations; 3) re-
327 duction in the magnitude of anthropogenic aerosol radiative forcing by raising
328 the pre-industrial baseline aerosol concentration. To remain consistent with
329 the observed temperature rise over the industrial period, reduced aerosol
330 forcing implies reduced climate sensitivity [30, 36].

331 **6 Materials**

332 The modal version of the global aerosol model GLOMAP [10] is used to de-
333 termine the impact of the biogenic nucleation mechanism reported in Ref.
334 [1]. The model resolution is $2.8^\circ \times 2.8^\circ$ horizontally, and there are 31 vertical
335 levels from ground level to 10 hPa. GLOMAP is embedded within a chemical
336 transport model, TOMCAT [37], and simulates the formation or emission,
337 growth, coagulation, advection, cloud processing and deposition of aerosol in
338 seven log-normal size modes. Four modes (nucleation, Aitken, accumulation
339 and coarse) are hydrophilic, and there are also hydrophobic Aitken, accu-
340 mulation and coarse modes. The composition of each mode is determined

341 by the relative fractions of the sulfate, sea-salt, black carbon, and organic
 342 carbon compounds. Dust is not included, as it was not found to contribute
 343 significantly to CCN [38]. Meteorology is forced by fields from the Euro-
 344 pean Centre for Medium-range Weather Forecasting. Total monoterpene
 345 emissions are taken from Ref. [34] and the ratio of endocyclic to exocyclic
 346 monoterpenes was calculated from a run of the MEGAN model with the set-
 347 tings prescribed to follow Ref. [39]. Ref. [40] suggests that terpene emissions
 348 are (within uncertainties) unchanged through the industrial period.

349 While sulphuric acid, ammonia, amines, halogens and HOMs can all partici-
 350 pitate directly in nucleation, here we consider only sulphuric acid and
 351 HOMs. The HOMs are formed via the oxidation of monoterpenes (MT)
 352 by ozone (O_3) and hydroxyl radicals ($OH\cdot$). The concentrations of these ox-
 353 idants are read in every six hours from a dedicated TOMCAT simulation.
 354 Instead of modelling the full reaction mechanism, we represent the HOM
 355 concentrations by

$$[HOM] = (Y_{AP.O_3}k_{AP.O_3}[AP][O_3] + Y_{BP.O_3}k_{BP.O_3}[BP][O_3] + Y_{AP.OH}k_{AP.OH}[AP][OH\cdot] + Y_{BP.OH}k_{BP.OH}[BP][OH\cdot]) / CS$$

356 where $Y_{AP.O_3} = 2.9\%$ and $Y_{AP.OH} = 1.2\%$ are the yields of HOM from α -
 357 pinene (AP) oxidation with ozone and hydroxyl radicals in the CLOUD cham-
 358 ber, described below, $Y_{BP.O_3} = 0.12\%$ and $Y_{BP.OH} = 0.58\%$ are taken from
 359 Ref. [12] and CS is the condensation sink (s^{-1}), determined assuming the dif-
 360 fusion characteristics of a typical α -pinene oxidation product (see Appendix
 361 A1 of Ref. [10]). The temperature-dependent reaction rate constants k for
 362 oxidation of α and β -pinene by ozone and hydroxyl radicals are taken from
 363 IUPAC [41].

364 The ozonolysis yield is determined with chemical ionisation time-of-flight
 365 mass spectrometers in the presence of a hydroxyl scavenger (0.1% H_2), repli-
 366 cating the effect of atmospheric $OH\cdot$ sinks such as methane and carbon
 367 monoxide. The HOM yield from reaction with hydroxyl radicals is deter-
 368 mined from measurements in the absence of ozone, and where photolysed
 369 HONO provides the $OH\cdot$ source.

370 BioOxOrg in nucleation mechanism 2 and HOM in mechanism 3 play
 371 equivalent roles but the former refers to the parametrised oxidation products
 372 derived from pinanediol, a first-generation oxidation product of α -pinene. Its
 373 concentration, as described in Ref. [5], is

$$[\text{BioOxOrg}] = k_{\text{MT.OH}} \cdot [\text{MT}] \cdot [\text{OH}\cdot] / CS$$

374 where CS is the condensation sink. The BioOxOrg concentration was not
375 measured directly in a mass spectrometer, but calculated from the pinanediol
376 concentration assuming a yield of 100%. The nucleation rate in mechanism 2
377 is measured as a function of this BioOxOrg, so the yield is incorporated into
378 the rate constant for nucleation. In Ref. [5] monoterpenes are assumed to
379 be equivalent to α -pinene, and so we assume only endocyclic monoterpenes
380 participate in this nucleation mechanism.

381 Particles are formed according to the mechanisms described in the main
382 text at a critical diameter usually around 1.7 nm. Ion concentrations are
383 determined by balancing production from radon and galactic cosmic rays
384 with losses to pre-existing particles and to ion-ion recombination (see SI
385 Appendix). The formation rates are then adjusted to account for losses
386 during the initial growth with the Kerminen-Kulmala equation [42] using
387 growth rates taken from the parametrisation of Ref. [43].

388 Particles subsequently grow by kinetic condensation of organic molecules
389 produced from oxidation of terpenes or isoprene by nitrate or hydroxyl rad-
390 icals, or ozone, with a 13% assumed yield for terpenes [10] and a 3% yield
391 for isoprene [44]. They also coagulate, and hence the overall particle number
392 is determined by solving the coagulation-nucleation equation [10]. Finally,
393 particles may be lost by dry or wet deposition.

394 Present-day simulations are run for 2008 and pre-industrial simulations
395 are run with 2008 meteorology and 1750 emissions. For the 1750 simulation,
396 anthropogenic sources of SO_2 and H_2SO_4 were removed from the model,
397 OH , NO_3 and ozone concentrations were adjusted to pre-industrial levels
398 determined from a dedicated TOMCAT simulation, and black and organic
399 carbon primary emissions were adjusted to a representation of pre-industrial
400 levels.

401 Cloud condensation nuclei (CCN) and cloud droplet number concentra-
402 tions (CDNC) are calculated for each simulation from the particle size dis-
403 tributions using the parametrisation of Ref. [45], assuming for the CDNCs
404 constant updraft velocities of 0.15 ms^{-1} over sea and 0.30 ms^{-1} over land.
405 The hygroscopicity parameters assigned to each chemical component follow
406 Ref. [44]: sulphate (0.61, assuming ammonium sulphate), sea salt (1.28),
407 black carbon (0.0), and organics (0.1). The change in cloud droplet effective

408 radii corresponding to the CDNC change is calculated in accordance with
409 Ref. [31], while the cloud albedo is estimated using the radiative transfer
410 model of Ref. [46].

411 **7 Acknowledgments**

412 We would like to thank CERN for supporting CLOUD with important techni-
413 cal and financial resources, and for providing a particle beam from the CERN
414 Proton Synchrotron. We also thank P. Carrie, L.-P. De Menezes, J. Dumol-
415 lard, F. Josa, I. Krasin, R. Kristic, A. Laassiri, O.S. Maksumov, B. Marichy,
416 H. Martinati, S.V. Mizin, R. Sitals, A. Wasem and M. Wilhelmsson for their
417 important contributions to the experiment. We thank D. Veber from Envi-
418 ronment and Climate Change Canada for maintenance and calibrations of in-
419 struments at East Trout Lake and NOAA/ESRL for collaboration with data
420 collection and QA/QC software. We thank A. D. Clarke and C. L. S. Red-
421 dington for making available processed data from the ARCTAS campaign.
422 The global modelling simulations were performed on ARC1 and ARC2, part
423 of the High Performance Computing facilities at the University of Leeds,
424 U.K. This research has received funding from the EC Seventh Framework and
425 Horizon 2020 Programmes (Marie Curie Initial Training Network MC-ITN
426 CLOUD-TRAIN no. 316662, Marie Sklodowska-Curie grant agreements Nos.
427 656994 and 600377, ERC-Consolidator grant NANODYNAMITE no. 616075
428 and ERC-Advanced grant ATMNUCLE no.227463), the German Federal
429 Ministry of Education and Research (project no.01LK1222A), the Swiss
430 National Science Foundation (project nos.200020_135307, 200021_140663,
431 206021_144947/1 and 20FI20_149002/1), the Academy of Finland (Center
432 of Excellence project no. 1118615), the Academy of Finland (135054, 133872,
433 251427, 139656, 139995, 137749, 141217, 141451), the Finnish Funding Agency
434 for Technology and Innovation, the Väisälä Foundation, the Nessling Foun-
435 dation, the Austrian Science Fund (FWF; project no. L593), the Portuguese
436 Foundation for Science and Technology (project no. CERN/FP/116387/2010),
437 the Swedish Research Council, Vetenskapsrådet (grant 2011-5120), the Pre-
438 sidium of the Russian Academy of Sciences and Russian Foundation for
439 Basic Research (grant 12-02-91522-CERN), the U.K. Natural Environment
440 Research Council (grant NE/K015966/1), the Royal Society (Wolfson Merit
441 Award), the U.S. National Science Foundation (grants AGS1136479,AGS1439551,
442 AGS1447056 and CHE1012293), Caltech ESE Grant (Davidow Foundation),

443 Dreyfus Award EP-11-117, the French National Research Agency (ANR),
444 the Nord-Pas de Calais, the European Funds for Regional Economic Devel-
445 opment (FEDER, Labex-Cappa, ANR-11-LABX-0005-01), and the French
446 Civil Aviation Office (MERMOSÉ).

447 **References**

- 448 [1] Kirkby, J et al. (2016) Ion-induced nucleation of pure biogenic particles.
449 *Nature* **533**, 521–526.
- 450 [2] Went, F. W. (1960) Blue hazes in the atmosphere. **187**, 641–643.
- 451 [3] Merikanto, J, Spracklen, D. V, Mann, G. W, Pickering, S. J, & Carslaw,
452 K. S. (2009) Impact of nucleation on global CCN. *Atmospheric Chem-*
453 *istry and Physics* **9**, 8601–8616.
- 454 [4] Kavouras, I. G, Mihalopoulos, N, & Stephanou, E. G. (1998) Formation
455 of atmospheric particles from organic acids produced by forests. *Nature*
456 **395**, 683–686.
- 457 [5] Riccobono, F et al. (2014) Oxidation products of biogenic emissions
458 contribute to nucleation of atmospheric particles. *Science* **344**, 717–
459 721.
- 460 [6] Hansen, J et al. (1981) Climate impact of increasing atmospheric carbon
461 dioxide. *Science* **213**, 957–966.
- 462 [7] Carslaw, K. S et al. (2013) Large contribution of natural aerosols to
463 uncertainty in indirect forcing. *Nature* **503**, 67–71.
- 464 [8] Pierce, J. R & Adams, P. J. (2009) Uncertainty in global CCN concen-
465 trations from uncertain aerosol nucleation and primary emission rates.
466 *Atmospheric Chemistry and Physics* **9**, 1339–1356.
- 467 [9] Schobesberger, S et al. (2013) Molecular understanding of atmo-
468 spheric particle formation from sulfuric acid and large oxidized organic
469 molecules. *Proceedings of the National Academy of Sciences* **110**, 17223–
470 17228.

- 471 [10] Mann, G. W et al. (2010) Description and evaluation of GLOMAP-
472 mode: a modal global aerosol microphysics model for the UKCA
473 composition-climate model. *Geoscientific Model Development* **3**, 519–
474 551.
- 475 [11] Vehkamäki, H et al. (2002) An improved parameterization for sulfu-
476 ric acid-water nucleation rates for tropospheric and stratospheric condi-
477 tions. *Journal of Geophysical Research* **107**, 4622–4622.
- 478 [12] Jokinen, T et al. (2015) Production of extremely low volatile organic
479 compounds from biogenic emissions: Measured yields and atmospheric
480 implications. *Proceedings of the National Academy of Sciences* **112**,
481 7123–7128.
- 482 [13] Yu, F et al. (2015) Spring and summer contrast in new particle formation
483 over nine forest areas in north america. *Atmospheric Chemistry and*
484 *Physics* **15**, 13993–14003.
- 485 [14] Spracklen, D. V, Pringle, K. J, Carslaw, K. S, Chipperfield, M. P, &
486 Mann, G. W. (2005) A global off-line model of size-resolved aerosol
487 microphysics: Ii. identification of key uncertainties. *Atmospheric Chem-*
488 *istry and Physics* **5**, 3233–3250.
- 489 [15] Kulmala, M et al. (2013) Direct observations of atmospheric aerosol
490 nucleation. *Science* **339**, 943–946.
- 491 [16] Bianchi, F et al. (2016) New particle formation in the free troposphere:
492 A question of chemistry and timing. *Science*.
- 493 [17] Paasonen, P et al. (2010) On the roles of sulphuric acid and low-volatility
494 organic vapours in the initial steps of atmospheric new particle forma-
495 tion. *Atmospheric Chemistry and Physics* **10**, 11223–11242.
- 496 [18] Rizzo, L, Artaxo, P, Karl, T, Guenther, A, & Greenberg, J. (2010)
497 Aerosol properties, in-canopy gradients, turbulent fluxes and VOC con-
498 centrations at a pristine forest site in Amazonia. *Atmospheric Environ-*
499 *ment* **44**, 503 – 511.
- 500 [19] Martin, S et al. (2010) Sources and properties of amazonian aerosol
501 particles. *Reviews of Geophysics* **48**.

- 502 [20] Kiendler-Scharr, A et al. (2009) New particle formation in forests in-
503 hibited by isoprene emissions. *Nature* **461**, 381–384.
- 504 [21] Wildt, J et al. (2014) Suppression of new particle formation from
505 monoterpene oxidation by NO_x . *Atmospheric Chemistry and Physics*
506 **14**, 2789–2804.
- 507 [22] Mann, G. W et al. (2014) Intercomparison and evaluation of global
508 aerosol microphysical properties among aerocom models of a range of
509 complexity. *Atmospheric chemistry and physics* **14**, 4679–4713.
- 510 [23] Vehkamäki, H et al. (2004) Atmospheric particle formation events at
511 Värriö measurement station in Finnish Lapland 1998-2002. *Atmospheric*
512 *Chemistry and Physics* **4**, 2015–2023.
- 513 [24] Svenningsson, B et al. (2008) Aerosol particle formation events and
514 analysis of high growth rates observed above a subarctic wetland–forest
515 mosaic. *Tellus B* **60**, 353–364.
- 516 [25] Suni, T et al. (2008) Formation and characteristics of ions and charged
517 aerosol particles in a native Australian Eucalypt forest. *Atmospheric*
518 *Chemistry and Physics* **8**, 129–139.
- 519 [26] Asmi, E et al. (2011) Secondary new particle formation in Northern Fin-
520 land Pallas site between the years 2000 and 2010. *Atmospheric Chem-*
521 *istry and Physics* **11**, 12959–12972.
- 522 [27] Spracklen, D. V et al. (2010) Explaining global surface aerosol number
523 concentrations in terms of primary emissions and particle formation.
524 *Atmospheric Chemistry and Physics* **10**, 4775–4793.
- 525 [28] Torseth, K et al. (2012) Introduction to the European Monitoring and
526 Evaluation Programme (EMEP) and observed atmospheric composition
527 change during 1972-2009. *Atmospheric Chemistry and Physics* **12**, 5447–
528 5481.
- 529 [29] Jacob, D. J et al. (2010) The Arctic Research of the Composition of the
530 Troposphere from Aircraft and Satellites (ARCTAS) mission: design,
531 execution, and first results. *Atmospheric Chemistry and Physics* **10**,
532 5191–5212.

- 533 [30] Charlson, R. J et al. (1992) Climate forcing by anthropogenic aerosols.
534 *Science* **255**, 423–430.
- 535 [31] Rap, A et al. (2013) Natural aerosol direct and indirect radiative effects.
536 *Geophysical Research Letters* **40**, 3297–3301.
- 537 [32] Koren, I, Dagan, G, & Altaratz, O. (2014) From aerosol-limited to
538 invigoration of warm convective clouds. *Science* **344**, 1143–1146.
- 539 [33] Ehn, M et al. (2014) A large source of low-volatility secondary organic
540 aerosol. *Nature* **506**, 476–479.
- 541 [34] Sindelarova, K et al. (2014) Global data set of biogenic VOC emissions
542 calculated by the MEGAN model over the last 30 years. *Atmospheric
543 Chemistry and Physics* **14**, 9317–9341.
- 544 [35] Myhre, G et al. (2013) Radiative forcing of the direct aerosol effect from
545 AeroCom Phase II simulations. *Atmospheric Chemistry and Physics* **13**,
546 1853–1877.
- 547 [36] Knutti, R, Stocker, T. F, Joos, F, & Plattner, G.-K. (2002) Constraints
548 on radiative forcing and future climate change from observations and
549 climate model ensembles. *Nature* **416**, 719–723.
- 550 [37] Chipperfield, M. P. (2006) New version of the TOMCAT/SLIMCAT off-
551 line chemical transport model: Intercomparison of stratospheric tracer
552 experiments. *Quarterly Journal of the Royal Meteorological Society* **132**,
553 1179–1203.
- 554 [38] Manktelow, P. T, Carslaw, K. S, Mann, G. W, & Spracklen, D. V.
555 (2010) The impact of dust on sulfate aerosol, CN and CCN during an
556 East Asian dust storm. *Atmospheric Chemistry and Physics* **10**, 365–
557 382.
- 558 [39] Guenther, A et al. (2012) The Model of Emissions of Gases and Aerosols
559 from Nature version 2.1 (MEGAN2.1): an extended and updated frame-
560 work for modeling biogenic emissions. *Geoscientific Model Development*
561 **5**, 1471–1492.
- 562 [40] Acosta Navarro, J. C et al. (2014) Global emissions of terpenoid VOCs
563 from terrestrial vegetation in the last millennium. *Journal of Geophysical
564 Research: Atmospheres* **119**, 6867–6885.

- 565 [41] Atkinson, R et al. (2006) Evaluated kinetic and photochemical data
566 for atmospheric chemistry: Volume ii: gas phase reactions of organic
567 species. *Atmospheric Chemistry and Physics* **6**, 3625–4055.
- 568 [42] Kerminen, V.-M & Kulmala, M. (2002) Analytical formulae connecting
569 the real and the apparent nucleation rate and the nuclei number concen-
570 tration for atmospheric nucleation events. *Journal of Aerosol Science*
571 **33**, 609 – 622.
- 572 [43] Tröstl, J et al. (2016) The role of low-volatility organic compounds in
573 initial particle growth in the atmosphere. *Nature* **533**, 527–531.
- 574 [44] Scott, C. E et al. (2014) The direct and indirect radiative effects of
575 biogenic secondary organic aerosol. *Atmospheric Chemistry and Physics*
576 **14**, 447–470.
- 577 [45] Morales, R, Nenes, A, Jonsson, H, Flagan, R. C, & Seinfeld, J. H. (2011)
578 Evaluation of an entraining droplet activation parameterization using in
579 situ cloud data. *Journal of Geophysical Research: Atmospheres* **116**,
580 D15205.
- 581 [46] Edwards, J. M & Slingo, A. (1996) Studies with a flexible new radiation
582 code. i: Choosing a configuration for a large-scale model. *Quarterly*
583 *Journal of the Royal Meteorological Society* **122**, 689–719.

584 **8 Supplementary: Ion concentrations in the** 585 **GLOMAP aerosol model**

586 We consider two sources of ions in the atmosphere: radon and galactic cosmic
587 rays. Radon is dominant at the land surface, where most biogenic nucleation
588 is likely to happen. Ion production rates from radon are read in from look-
589 up tables [1]. Above the surface and over the ocean, cosmic ray ionisation is
590 more important. The ionization rates from cosmic rays are calculated from
591 lookup tables [2] which are provided for several solar cycles, so the effect
592 of the Sun’s magnetic field can be incorporated via the heliospheric modu-
593 lation potential. The technique of Fraser-Smith [3] is used to calculate the
594 geomagnetic cut-off rigidity from the International Geomagnetic Reference
595 Field coefficients. These are available with five-yearly time resolution so are
596 interpolated within the five-year periods, then the atmospheric depth (which
597 determines the interaction probability of a cosmic ray) and the heliospheric
598 modulation potential are spatially interpolated across the model grid-boxes.

The small-ion concentration of either sign, $[n_{\pm}] = [n_+] = [n_-]$, is calcu-
lated from the steady state solution of the ion balance equation [4]

$$d[n_{\pm}]/dt = q - \alpha[n_{\pm}]^2 - k_i[n_{\pm}] \quad (6)$$

599 where q is the ion pair production rate from GCRs and α is the ion-ion recom-
600 bination coefficient (cm^3s^{-1}). The factor 2 in Eq. 4 accounts for nucleation
601 from both positive and negative ions. The ion loss rate, k_i , is due to the con-
602 densation sink, CS , and ion-induced nucleation, so that $k_i = CS + J_{iin}/2 [n_{\pm}]$
603 where $J_{iin}/2 [n_{\pm}]$ is given by Eq. 4 and the steady state concentration of small
604 ions is $[n_{\pm}] = [(k_i^2 + 4\alpha q)^{0.5} - k_i]/2\alpha$. From Eq. 6, J_{iin} saturates at $2q$ at high
605 nucleation rates (see Ref. [5] Fig. 2).

606 **9 Supplementary: Simulating the pre-industrial** 607 **atmosphere**

608 The concentrations of key precursor gases for particle formation are compared
609 between present-day and pre-industrial in Fig. S1. The percentage changes
610 between pre-industrial and present-day are compared in Table S1 for sum-
611 mer and winter in the two hemispheres. The sulphuric acid concentration
612 is substantially higher in the present day atmosphere due to much higher

613 emissions, while the organic concentrations are higher in the pre-industrial
614 atmosphere due to lower sinks.

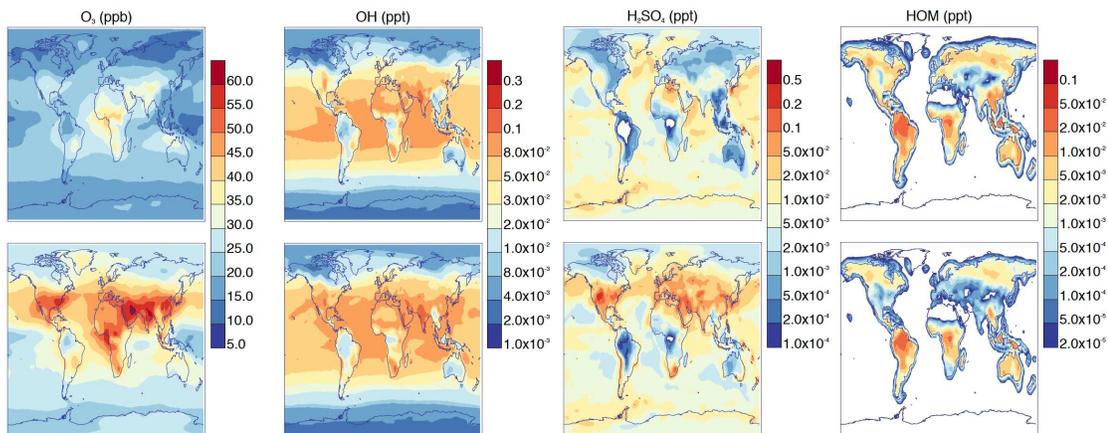


Figure S1: Concentrations of key gases: ozone, hydroxyl radicals, sulphuric acid and HOMs in pre-industrial (top row) and present-day atmospheres (bottom row) at cloud base level, annually averaged.

614

615 **10 Supplementary: modelled changes in par-** 616 **ticl e concentrations and further discussion** 617 **of particle numbers in the Amazon region**

618 In Fig. S2 we present the seasonal cycle in surface 3 nm particle concentra-
619 tions with and without pure biogenic nucleation, and the change when pure
620 biogenic nucleation is included. This figure shows that the strongest effects
621 are in the present-day in summertime in boreal regions, Australia, southern
622 Africa and the Amazon region. Fig. S3 shows the effect of pure biogenic
623 nucleation on cloud-level CCN concentrations in months chosen to reflect
624 the Amazon wet and dry seasons (February and August). In Sect. 12, we
625 further show that the present-day concentrations are in good agreement with
626 observations at a diverse range of surface sites.

627 As discussed in the main text, on average our model predicts greater
628 numbers of particles in the Amazon than observations suggest. According
629 to the review by Martin *et al* [6], the mean number concentration in the

Table S1: Relative changes between present-day and pre-industrial atmosphere at cloud level, averaged over the month for January and July, and averaged over the entire year in the “Annual” column. The value quoted is the percentage increase in the mean in the present-day compared to the pre-industrial atmosphere. Pure biogenic nucleation is included.

Quantity	Change w.r.t. pre-industrial (%)				
	Jan. NH	Jul. NH	Jan. SH	Jul. SH	Annual
O ₃	76.9	70.2	32.5	37.9	59.1
OH	41.7	21.3	-0.9	-0.1	14.6
H ₂ SO ₄	361.4	84.4	0.8	42.3	79.3
HOM	-69.4	-36.1	-21.6	-26.4	-39.8
total J	1255.0	445.3	23.0	341.5	491.6
J_R	1264.5	164.3	24.5	343.2	484.0
J_{org}	-90.3	-47.5	-33.6	-65.3	-57.8
N_3	136.4	36.8	3.0	10.6	42.9
N_{70}	117.0	36.9	6.0	7.5	41.8
CCN 1%	122.2	42.2	4.6	7.3	44.4
CCN 0.2%	99.3	72.0	7.5	9.5	54.2

630 Aitken mode is 239 cm^{-3} and that in the accumulation mode is 177 cm^{-3} ,
631 so the total concentration of particles of at least 70 nm in diameter (N_{70} , a
632 reasonable proxy for CCN, usually equivalent to a supersaturation between
633 0.4% and 1%) is $\sim 300 \text{ cm}^{-3}$ in the wet season. Our model predicts N_{70} of
634 $\sim 500 - 800 \text{ cm}^{-3}$ in the wet season (higher near Manaus, lower near the
635 coast, see Fig. S3) so it is still high, but within a factor 2 or 3, which may
636 be larger than the measurement uncertainty but is certainly within our best
637 estimate of the model parametric uncertainty [7]. In February in Manaus, we
638 predict CCN concentrations at 0.2% supersaturation to be 127 cm^{-3} with-
639 out pure biogenic nucleation, and pure biogenic nucleation increases this to
640 299 cm^{-3} . Observations in the wet season in the pristine forest near Man-
641 aus are considerably lower, at $30 - 80 \text{ cm}^{-3}$. However, our model averages
642 over both the pristine forest and the Manaus pollution plume in this area
643 so would be expected to yield higher concentrations than the pristine ob-
644 servations. There is the additional challenge that Manaus is so close to the
645 Equator that it is close to the boundary between the wet season and the
646 dry season. The observed transition season concentrations of $200 - 300 \text{ cm}^{-3}$
647 may be more appropriate, which would agree with our model. In the dry

648 season, our model is in reasonably good agreement with observations, with
649 on average $400 - 800 \text{ cm}^{-3}$ CCN 0.2% (Fig. S3, Ref. [6]).

650 Overprediction of nucleation rates in the Amazon region suggests that
651 pure biogenic nucleation may be suppressed there. The implications of the
652 CLOUD results for the global atmosphere thus depend on whether a mecha-
653 nism that suppresses pure biogenic nucleation exists, and if it does, whether
654 or not it has a seasonal dependence, or is localised to regions with similar
655 characteristics to the Amazon. For example, it could be associated with high
656 humidity which increases the effective condensation sink [8, 9], high isoprene
657 concentrations [10], high temperatures, or high levels of peroxy radicals, or
658 it could be present globally all year round.

659 We note that overprediction of CCN (even without pure biogenic nucle-
660 ation) is a feature of many global aerosol models. The AeroCom assessment
661 of 15 global aerosol models [12] shows annual mean N100 concentrations in
662 the Amazon of over 500 cm^{-3} and N30 concentrations of closer to 1000 cm^{-3} .
663 This suggests that there are general model weaknesses in the Amazon which
664 makes it difficult to say anything with confidence about the effects an addi-
665 tional aerosol source would have on this region.

666 In Ref. [6], it is observed that particles in the nucleation and Aitken modes
667 in the wet season rarely grow to larger sizes (although it is certainly the case
668 that some condensational growth does occur). This is *a priori* surprising
669 since the terpene and condensable organic concentrations are very high in this
670 area. The most likely explanation must be very high losses: high precipitation
671 frequency in the wet season and high condensation sink in the dry season,
672 or very strong vertical mixing that prevents particle growth being observed
673 adequately from a single surface location. Any losses are also likely to be
674 inhomogenously distributed, and therefore their non-linear effects could be
675 easily underestimated in a low-resolution model like ours.

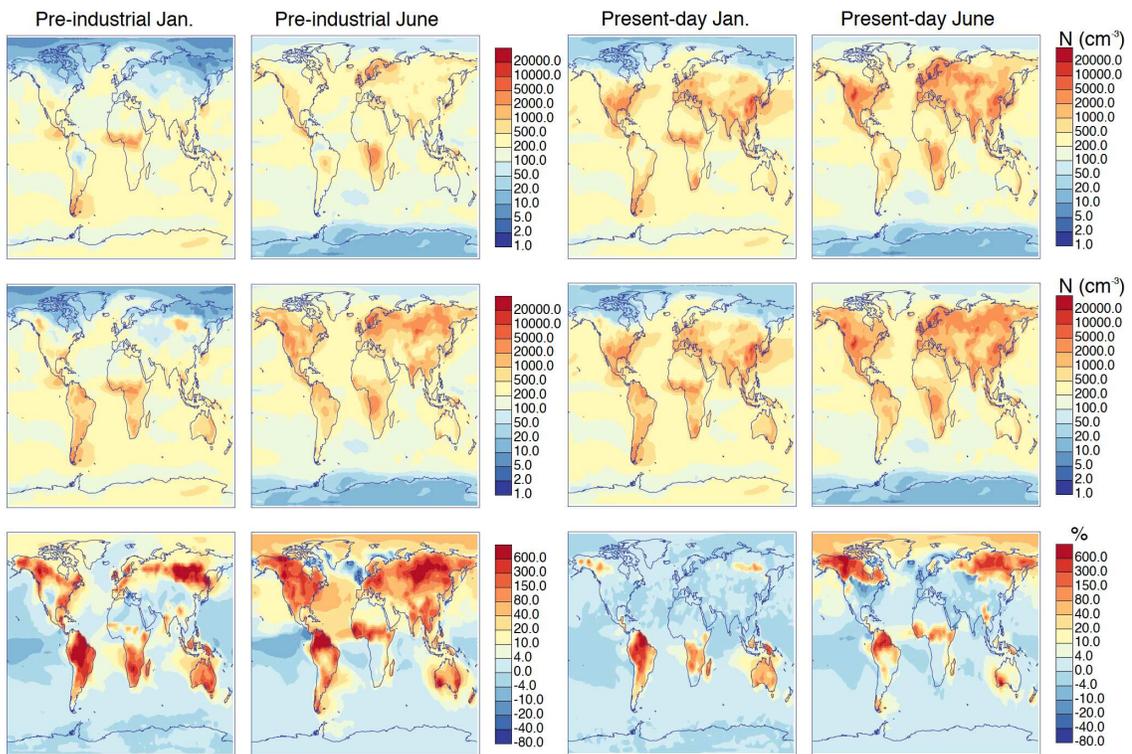


Figure S2: Monthly average surface level concentrations of particles larger than 3 nm in diameter, along each row pre-industrial January, pre-industrial June, present-day January and present-day June. Top: without pure biogenic nucleation. Middle: with pure biogenic nucleation. Bottom: percentage changes when pure biogenic nucleation is added.

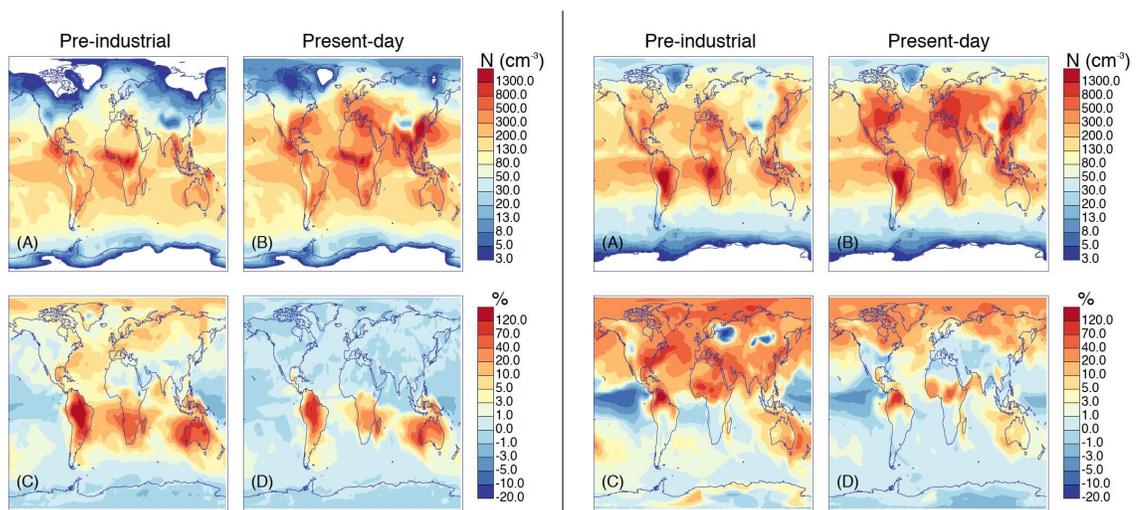


Figure S3: Concentrations of cloud condensation nuclei calculated at 0.2% supersaturation, in cm^{-3} , in February (left four panels) and August (right four panels). On either side of the line, average CCN concentrations at cloud base level over the month in (A) pre-industrial and (B) present-day conditions are shown, and, below these in subfigures (C, D), the percentage changes to these concentrations when pure biogenic nucleation is introduced.

676 **11 Supplementary: diurnal cycles of particle** 677 **formation rates**

678 In Fig. S4, we show model predictions of the diurnal cycle of particle for-
679 mation in July at Pallas and at the most studied field site, Hyytiälä, also in
680 Finland. We predict that pure biogenic nucleation contributes significantly
681 to the nucleation rate at both sites in July. One would expect APi-TOF data
682 at Hyytiälä in July to show clusters of HOMs both with and without sul-
683 phuric acid. However, identifying the absence of sulphuric acid from clusters
684 large enough to be equivalent to nucleated particles in mass spectra from
685 field measurements at Hyytiälä has not been possible [13]. Furthermore, nu-
686 cleation measurements at Hyytiälä are usually made in spring, when pure
687 biogenic nucleation is predicted to make a much smaller contribution (see
688 also Fig. S8, below). At Pallas, on the other hand, nucleation events will be
689 rarer and signals are likely to be smaller, but the background particle con-
690 centrations responsible for the condensation sink are lower. Therefore, the
691 peaks in nucleation at 200 and 520 hours into the month in Fig. S4 might
692 well lead to observable ‘banana’-type events, and our model suggests that
693 all nucleation at this site should be dominated by pure biogenic processes.
694 We note that while the pure biogenic nucleation mechanism can in principle
695 operate both day and night, the model does not predict any nucleation at
696 night, principally because terpene emissions are higher during daytime.

697 We also show the diurnal cycle in February (wet season) and August (dry
698 season) at the most studied observation site in the Amazon, Manacapuru,
699 in Fig. S5. Manacapuru is in the same model gridbox as the Amazon Tall
700 Tower Observatory and as Manaus. At the surface level, approximately 0 to
701 30 m above ground level, we predict quite a substantial amount of nucleation,
702 but rarely, perhaps never, enough to produce observable banana-type events,
703 especially in the dry season.

704 Further, we show the evolution of the size distributions at the surface
705 level at Hyytiälä and Pallas in August in Fig. S6, and in Manacapuru for
706 February and August in Fig. S7. Clear nucleation events are seen at both
707 Finnish sites which are similar to the observations detailed in, for example,
708 Refs. [14] or [15] for Hyytiälä and [16] or [17] for Pallas. The size distributions
709 also show that in the dry season in the Amazon essentially no nucleation is
710 predicted, while in the wet season very weak additions to the Aitken mode are
711 predicted. We speculate that these signals would rarely, if ever, be observable

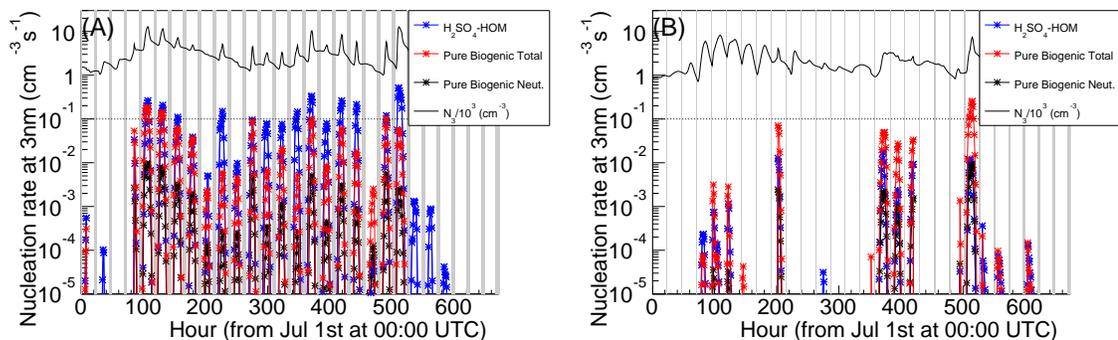


Figure S4: Modelled diurnal cycles of nucleation rates and particle concentrations at present-day (A) Hyytiälä (61.85°N, 24.28°E) and (B) Pallas (68.00°N, 24.23°E) in the first four weeks of July 2008. July is the month where the pure biogenic nucleation rate at Hyytiälä is strongest. Intervals between sunset and sunrise are marked in grey. The wind changes direction around 90 hours into the month. Observable nucleation events are likely when the nucleation rate is above around $0.1 \text{ cm}^{-3} \text{ s}^{-1}$, indicated by the dotted line. At Pallas, Aitken mode particles are transported from nucleation happening elsewhere, which explains the daytime peaks in particle number concentration even when nucleation rates at Pallas are very low.

712 as nucleation events due to the complicated and inhomogenous meteorology.

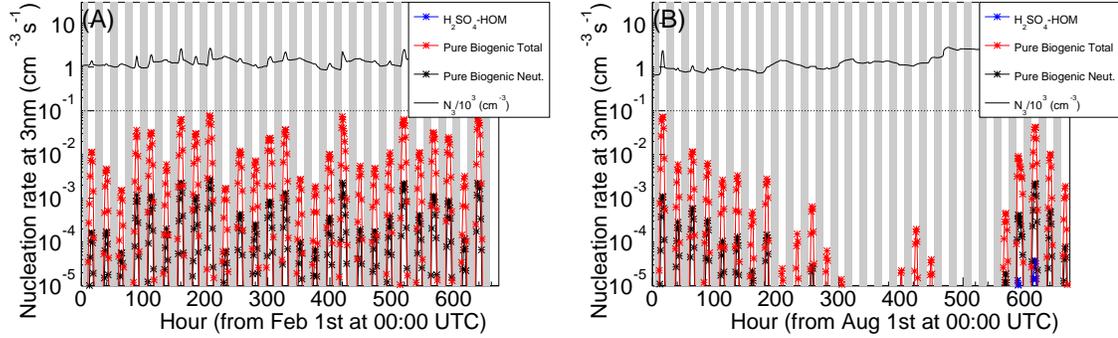


Figure S5: Modelled diurnal cycles of nucleation rates at present-day Manacapuru (3.30°S, 60.62°W) in (A) the wet season (February) and (B) the dry season (August), in 2008. The particle number concentration shows small spikes during nucleation events, often increasing from around 1000 cm^{-3} to around 2000 cm^{-3} due to nucleation. However, in these figures, our spatial model resolution smears out much larger, more local fluctuations in the particle concentrations (which can easily be a factor 10), and so in reality such small spikes would be difficult, and probably impossible, to discern in atmospheric observations.

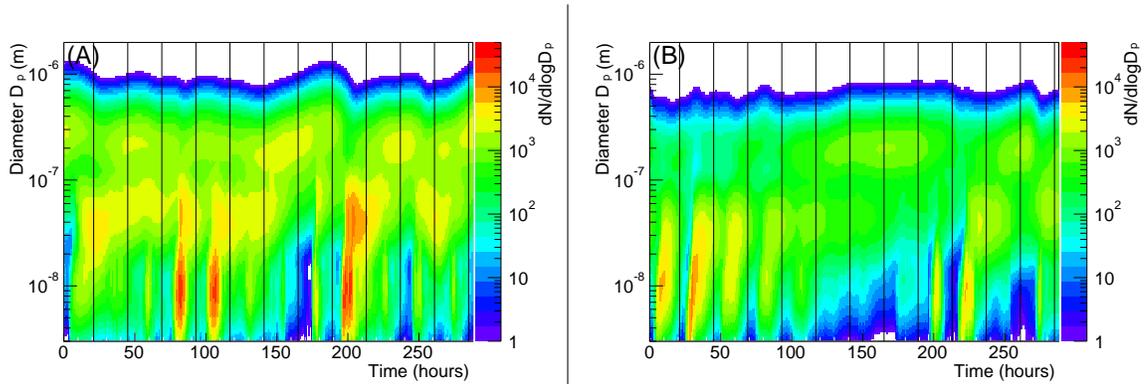


Figure S6: Modelled diurnal cycles of particle size distribution $dN/d\log D_p$ at present-day (A) Hyytiälä and (B) Pallas, in the first twelve days of August (UTC time). The vertical lines mark midnight Finnish local time (UTC+3). In this figure $d\log D_p = 0.02$.

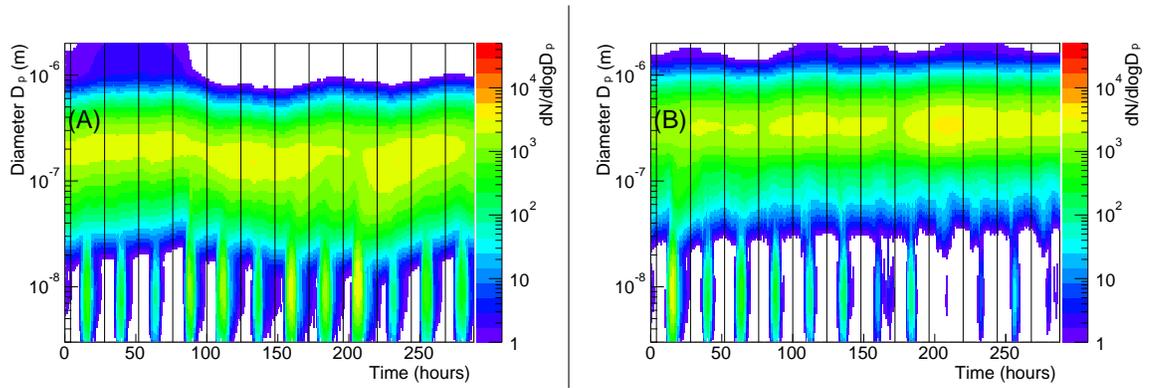


Figure S7: Modelled diurnal cycles of particle size distribution at present-day Manacapuru in (A) the wet season (the first twelve days of February, UTC time) and (B) the dry season (the first twelve days of August). Like the fluctuations in particle number concentration shown in Fig. S5, the small spikes in the nucleation mode would be difficult to see in observation data. The vertical lines mark midnight local time (UTC-4). In this figure $d \log D_p = 0.02$.

713 **12 Supplementary: Model evaluation against** 714 **measurements**

715 The model was evaluated by comparing the particle number concentrations
716 it predicts to those measured at 37 surface sites (Fig. S8). The first thirty-
717 six are those used in Ref. [18]. We also added previously unpublished data
718 recorded in 2010, 2011 and 2013 from a condensation particle counter with
719 a 4 nm cut-off diameter at the East Trout Lake Global Atmosphere Watch
720 station (54.35° N 104.98° W) because our model predicts a significant contri-
721 bution from pure biogenic nucleation in central Canada. When measurements
722 from multiple years are available, the data from the months in each year were
723 averaged. Averaging over all sites and over the whole year, we find including
724 pure biogenic nucleation leads to a modest improvement in the model bias
725 from -42% to -41%. In summer the bias changes from -36% to -34% when
726 pure biogenic nucleation is included and in winter it is unchanged at -53%.
727 The overall low bias, particularly in winter, is likely to be because we do not
728 include the effects of ammonia or anthropogenic organic molecules on nu-
729 cleation in our model. Fig. S8 shows that pure biogenic nucleation strongly
730 affects particle concentrations only at East Trout Lake, Listvyanka and Point
731 Barrow.

732 We also compared the daily mean particle number concentrations from the
733 model to observations made during the ARCTAS campaign [19] in spring and
734 summer 2008 (Fig. S9). We interpolate these modelled particle concentra-
735 tions within model grid boxes to match the locations of one-minute-averaged
736 condensation particle counter measurements from a NASA P3-B aircraft.
737 We note that the campaign during summer, when pure biogenic nucleation
738 has most effect, was designed to investigate the influence of boreal forest
739 fires. Our low model resolution and averaged fire emissions inventory mean
740 that close agreement between the model and the measurements is not ex-
741 pected. However, we attempt the comparison anyway as the measurements
742 are, unusually, in a particularly relevant region where pure biogenic nu-
743 cleation is predicted to have a strong effect. Averaged over all altitudes, the
744 model without pure biogenic nucleation is biased low (-57.9%); including
745 pure biogenic nucleation leads to a smaller bias of -37.4%. Most of the pure
746 biogenic contribution is in a band of latitudes from 51 to 62°N, correspond-
747 ing to flights starting from Cold Lake between 26 June and 14 July 2008. While
748 the relatively large underprediction of particle number without pure biogenic

749 nucleation suggests a particle formation pathway for pristine regions like ours
750 may be needed, large local deviations of the model and measurements evi-
751 dent in Fig. S9 preclude any firm conclusion being drawn. These deviations
752 could be attributed to the fire emissions or low model resolution discussed
753 earlier, temperature or chemistry effects on new particle formation that we
754 did not account for, or uncertainties in the observation data.

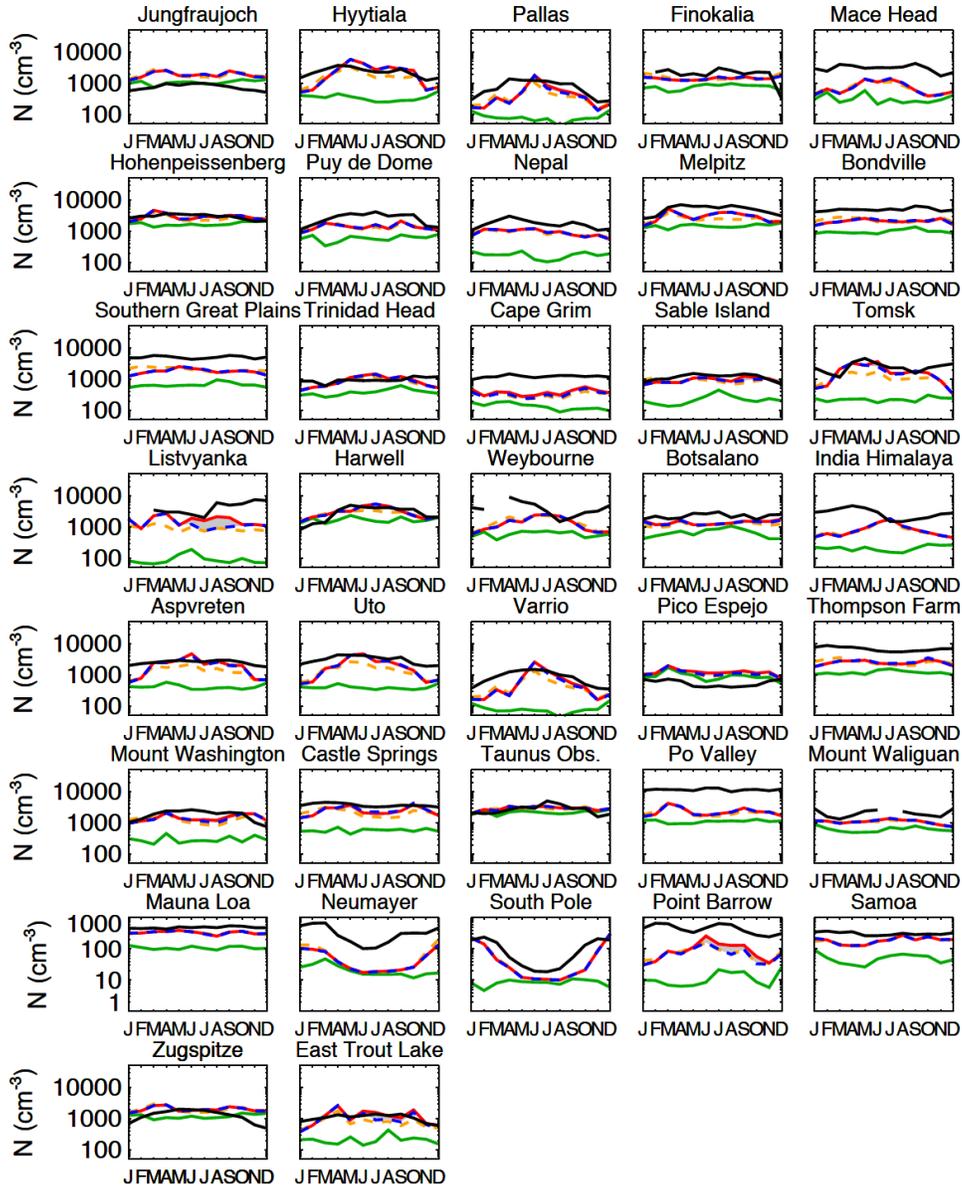


Figure S8: Particle concentrations at selected measurement sites [18, 20], in black, measured by counters with cut off sizes varying from 3 nm to 14 nm, compared to model predictions. The red curve shows the particle concentrations predicted by the baseline nucleation mechanisms, numbered 1-3 in the main text and including pure biogenic nucleation. The blue dotted curve shows the particle concentrations predicted without pure biogenic nucleation. A grey band is drawn between these two curves. The orange dotted curve shows particle concentrations predicted by the parametrisation of Paasonen et al [21], including a component of pure biogenic nucleation proportional to the square of the organic concentration. The dark green curve shows particles from primary emissions only.

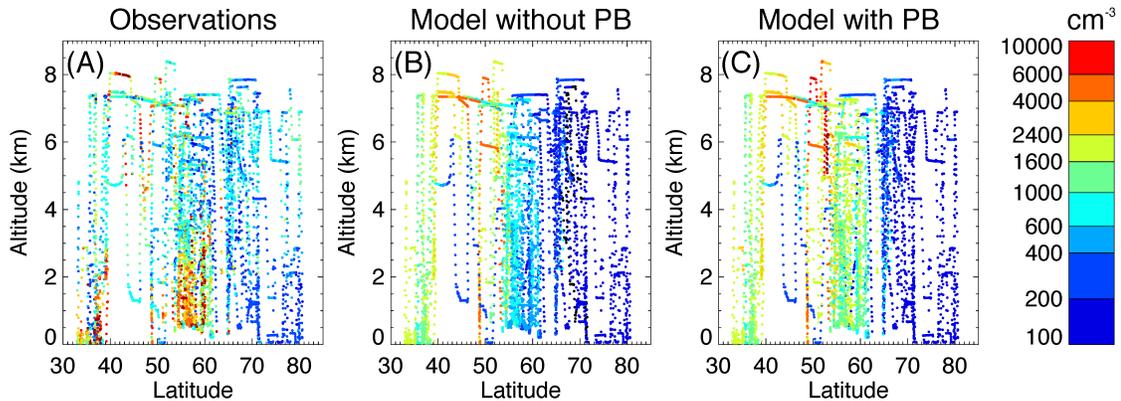


Figure S9: Vertical profiles of particle number concentrations calculated at S.T.P. (all particles of at least 3 nm in diameter), (A) measured during the ARCTAS campaign [19] in 2008, (B) modelled without pure biogenic nucleation (PB) and (C) modelled including pure biogenic nucleation.

755 **13 Supplementary: further discussion of un-**
 756 **certainties and summary tables of sensi-**
 757 **tivity studies**

758 Table S2 gives the changes to CCN concentrations due to pure biogenic
 759 nucleation in different scenarios (e.g. if organic nucleation is temperature-
 760 dependent, or if primary emissions are higher in our simulation than in re-
 761 ality). Following Ref. [7], the high primary emissions scenario corresponds
 762 to a doubling of biomass burning primary particulate emissions, a reduction
 763 in the mode diameter of biomass burning primary emissions from 150 nm to
 764 100 nm, and a factor 2.5 increase in the sea spray flux. The low emissions
 765 scenario corresponds to a halving of biomass burning emissions, an increase
 766 in the mode diameter to 175 nm and a reduction in the sea spray flux by
 767 60%. Table S3 gives the corresponding changes to radiative forcing.

768 In the main text, we discussed various sources of uncertainty, such as
 769 uncertainties in the CLOUD experimental measurements or HOM yields,
 770 in the possible temperature dependence of pure biogenic nucleation, and in
 771 primary emissions. The experiment in which we determine the sensitivity to

Table S2: Annual average global mean changes to the concentrations of particles larger than 70 nm in diameter, a proxy for CCN, when pure biogenic nucleation is introduced. The differences between the contents of the first row and the subsequent rows give an indication of the sensitivity of the analysis to different perturbations. The temperature dependence is added for both the baseline organic and the pure biogenic nucleation mechanisms, so changes both the control and perturbed simulations.

Variation	Change to CCN 0.2% (PI) (%)	Change to CCN 0.2% (PD) (%)
Add pure biogenic nucleation (PBN)	12	4
Add PBN with trebled yield	19	6
Add PBN with one-third yield	4	1
Add T dependence	7	2
Double pre-industrial volcanic SO ₂	11	-
Halve pre-industrial volcanic SO ₂	13	-
High primary emissions	7	2
Low primary emissions	14	5
Use baseline J from Ref. [21]	14	5

Table S3: Annual average global mean radiative forcings (including pure biogenic nucleation) and absolute changes to forcing when pure biogenic nucleation is introduced.

Variation	Forcing (PD-PI) Wm ⁻²	Change to forcing with pure biogenic nucleation (Wm ⁻²)
Add pure biogenic nucleation (PBN)	-0.60	+0.22
Add PBN with trebled yield	-0.52	+0.30
Add PBN with one-third yield	-0.72	+0.10
Add T dependence	-0.64	+0.14
High primary emissions	-0.63	+0.17
Low primary emissions	-0.54	+0.33
Use baseline J from Ref. [21]	-0.63	+0.24
Exclude area close to Equator	-0.62	+0.20

772 primary emissions also tests our sensitivity to uncertainty in the condensation
773 sink.

774 In the particularly interesting Amazon region, this uncertainty will be
775 strongly influenced by the quantity and mode diameter of biomass burning
776 emissions [7]. The 150 nm diameter used, following Ref. [22], is already larger
777 than that of fresh smoke particles (averaged over all vegetation types) of
778 117 nm [23], and in theory condensation of secondary organic vapours in our
779 model should increase this diameter to the aged diameter of 235 nm in a few
780 days. However, the aged diameter in reality depends on cloud processing of
781 organics as well as condensation, and this is not included in our model. To
782 compensate for this, the larger fresh diameter of 150 nm is used by default in
783 GLOMAP, but it has a large uncertainty [7]. We increase it to 175 nm in our
784 sensitivity study with high primary emissions, which is still well within the
785 uncertainty and the ranges for different phases of typical Amazon burning
786 found in Ref. [24].

787 Another possible source of uncertainty in the Amazon region, which we
788 are not able to quantify directly, is the condensation sink during periods of
789 high humidity [8, 9] or aerosol-cloud interaction [25]. Clouds in the Amazon
790 region are likely to be strongly affected by both of these. For the latter,
791 impaction scavenging in GLOMAP is due to raindrops but the only loss
792 mechanism to cloud droplets is nucleation scavenging. In a cloud, the coag-
793 ulation sink should surely dramatically increase. This is not modelled. The
794 effective condensation sink in a monodisperse cloud with 100 droplets cm^{-3}
795 of diameter $15.6 \mu\text{m}$ (corresponds to LWC 0.2 gm^{-3}) is 0.16 s^{-1} . We investi-
796 gated this further by including a crude treatment of this effect in our model.
797 The perturbation to the model results in this test was small, mostly because
798 clouds rarely cover more than half a model gridbox in areas where pure bio-
799 genic nucleation is important, and therefore the average effect is smeared out.
800 With higher model resolution, however, the effect would likely be stronger
801 due to the nonlinear nature of nucleation and survival probability.

802 **References**

- 803 [1] Zhang, K et al. (2011) Radon activity in the lower troposphere and
804 its impact on ionization rate: a global estimate using different radon
805 emissions. *Atmospheric Chemistry and Physics* **11**, 7817–7838.

- 806 [2] Usoskin, I. G, Kovaltsov, G. A, & Mironova, I. A. (2010) Cosmic ray
807 induced ionization model CRAC:CRII: An extension to the upper at-
808 mosphere. *Journal of Geophysical Research* **115**, 6.
- 809 [3] Fraser-Smith, A. C. (1987) Centered and eccentric geomagnetic dipoles
810 and their poles, 16001985. *Reviews of Geophysics* **25**, 1–16.
- 811 [4] Franchin, A et al. (2015) Experimental investigation of ion-ion recom-
812 bination under atmospheric conditions. *Atmospheric Chemistry and*
813 *Physics* **15**, 7203–7216.
- 814 [5] Kirkby, J et al. (2016) Ion-induced nucleation of pure biogenic particles.
815 *Nature* **533**, 521–526.
- 816 [6] Martin, S et al. (2010) Sources and properties of Amazonian aerosol
817 particles. *Reviews of Geophysics* **48**.
- 818 [7] Lee, L. A et al. (2013) The magnitude and causes of uncertainty in global
819 model simulations of cloud condensation nuclei. *Atmospheric Chemistry*
820 *and Physics* **13**, 8879–8914.
- 821 [8] Rose, C et al. (2015) Frequent nucleation events at the high altitude
822 station of Chacaltaya (5240 m a.s.l.), Bolivia. *Atmospheric Environment*
823 **102**, 18 – 29.
- 824 [9] Falvey, M & Garreaud, R. D. (2005) Moisture variability over the South
825 American Altiplano during the South American low level jet experiment
826 (SALLJEX) observing season. *Journal of Geophysical Research: Atmo-*
827 *spheres* **110**.
- 828 [10] Kiendler-Scharr, A et al. (2009) New particle formation in forests in-
829 hibited by isoprene emissions. *Nature* **461**, 381–384.
- 830 [11] Crounse, J. D, Nielsen, L. B, Jørgensen, S, Kjaergaard, H. G, &
831 Wennberg, P. O. (2013) Autoxidation of organic compounds in the
832 atmosphere. *The Journal of Physical Chemistry Letters* **4**, 3513–3520.
- 833 [12] Mann, G. W et al. (2014) Intercomparison and evaluation of global
834 aerosol microphysical properties among AeroCom models of a range of
835 complexity. *Atmospheric chemistry and physics* **14**, 4679–4713.

- 836 [13] Schobesberger, S et al. (2013) Molecular understanding of atmo-
837 spheric particle formation from sulfuric acid and large oxidized organic
838 molecules. *Proceedings of the National Academy of Sciences* **110**, 17223–
839 17228.
- 840 [14] Dal Maso, M et al. (2005) Formation and growth of fresh atmospheric
841 aerosols: eight years of aerosol size distribution data from SMEAR II,
842 Hyytiälä, Finland. *Boreal Environment Research* **10**, 323.
- 843 [15] Kulmala, M et al. (2012) Measurement of the nucleation of atmospheric
844 aerosol particles. *Nature Protocols* **7**, 1651–1667.
- 845 [16] Väänänen, R et al. (2013) Analysis of particle size distribution changes
846 between three measurement sites in northern Scandinavia. *Atmos.*
847 *Chem. Phys* **13**, 11887–11903.
- 848 [17] Komppula, M, Sihto, S.-L, Korhonen, H, Lihavainen, H, Kerminen, V.-
849 M, Kulmala, M, & Viisanen, Y. (2006) New particle formation in air
850 mass transported between two measurement sites in northern Finland.
851 *Atmospheric Chemistry and Physics* **6**, 2811–2824.
- 852 [18] Spracklen, D. V et al. (2010) Explaining global surface aerosol number
853 concentrations in terms of primary emissions and particle formation.
854 *Atmospheric Chemistry and Physics* **10**, 4775–4793.
- 855 [19] Jacob, D. J et al. (2010) The Arctic Research of the Composition of the
856 Troposphere from Aircraft and Satellites (ARCTAS) mission: design,
857 execution, and first results. *Atmospheric Chemistry and Physics* **10**,
858 5191–5212.
- 859 [20] Torseth, K et al. (2012) Introduction to the European Monitoring and
860 Evaluation Programme (EMEP) and observed atmospheric composition
861 change during 1972–2009. *Atmospheric Chemistry and Physics* **12**, 5447–
862 5481.
- 863 [21] Paasonen, P et al. (2010) On the roles of sulphuric acid and low-volatility
864 organic vapours in the initial steps of atmospheric new particle forma-
865 tion. *Atmospheric Chemistry and Physics* **10**, 11223–11242.
- 866 [22] Stier, P et al. (2005) The aerosol-climate model ECHAM5-HAM. *At-*
867 *mospheric Chemistry and Physics* **5**, 1125–1156.

- 868 [23] Janhäll, S, Andreae, M. O, & Pöschl, U. (2010) Biomass burning aerosol
869 emissions from vegetation fires: particle number and mass emission fac-
870 tors and size distributions. *Atmospheric Chemistry and Physics* **10**,
871 1427–1439.
- 872 [24] Costa, M. A. M et al. (2012) Real-time sampling of particulate matter
873 smaller than $2.5\ \mu\text{m}$ from Amazon forest biomass combustion. *Atmo-*
874 *spheric environment* **54**, 480–489.
- 875 [25] Pierce, J. R, Croft, B, Kodros, J. K, D’Andrea, S. D, & Martin,
876 R. V. (2015) The importance of interstitial particle scavenging by
877 cloud droplets in shaping the remote aerosol size distribution and global
878 aerosol-climate effects. *Atmospheric Chemistry and Physics* **15**, 6147–
879 6158.