

This is a repository copy of *Three-dimensional imaging of dislocation propagation during crystal growth and dissolution*.

White Rose Research Online URL for this paper: http://eprints.whiterose.ac.uk/89353/

Version: Accepted Version

Article:

Clark, JN, Ihli, J, Schenk, AS et al. (6 more authors) (2015) Three-dimensional imaging of dislocation propagation during crystal growth and dissolution. Nature Materials, 14 (8). pp. 780-784. ISSN 1476-1122

https://doi.org/10.1038/nmat4320

Reuse

Items deposited in White Rose Research Online are protected by copyright, with all rights reserved unless indicated otherwise. They may be downloaded and/or printed for private study, or other acts as permitted by national copyright laws. The publisher or other rights holders may allow further reproduction and re-use of the full text version. This is indicated by the licence information on the White Rose Research Online record for the item.

Takedown

If you consider content in White Rose Research Online to be in breach of UK law, please notify us by emailing eprints@whiterose.ac.uk including the URL of the record and the reason for the withdrawal request.



eprints@whiterose.ac.uk https://eprints.whiterose.ac.uk/

Three-dimensional imaging of dislocation propagation during crystal growth and dissolution

Jesse N. Clark1;#;y, Johannes Ihli2;y, Anna S. Schenk2, Yi-Yeoun Kim2, Alexander N. Kulak2, James M. Campbell3, Gareth Nisbet4, Fiona C. Meldrum2, & Ian K. Robinson1;5

1London Centre for Nanotechnology, University College, London WC1E 6BT, UK 2School of Chemistry, University of Leeds, Leeds LS2 9JT, UK. 3School of Physics and Astronomy, University of Leeds, Leeds, LS2 9JT, UK 4Diamond Light Source, Harwell Science and Innovation Campus, Didcot, Oxon OX11 0DE, UK

5Research Complex at Harwell, Didcot, Oxfordshire OX11 0DE, UK

Present address: Stanford PULSE Institute, SLAC National Accelerator Laboratory, 2575 Sand Hill Road, Menlo Park, California 94025, USA and Center for Free-Electron Laser Science (CFEL), Deutsches Elektronensynchrotron (DESY), Notkestrasse 85, 22607 Hamburg, Germany.

y These authors contributed equally to this work

Keywords: Calcite, Calcium Carbonate, Imaging, Coherent Diffraction, Screw Dislocation

Atomic-level defects such as dislocations play key roles in determining the macroscopic properties of crystalline materials1,2. Their effects range from increased chemical reactivity3,4 to enhanced mechanical properties5,6. Dislocations have been widely studied using traditional techniques such as X-ray diffraction and optical imaging. Recent advances have enabled atomic force microscopy to study single dislocations7 in two dimensions, while transmission electron microscopy (TEM) can now visualize strain fields in three dimensions with near-atomic resolution8–10. However, these techniques cannot offer three-dimensional imaging of the formation or movement of dislocations during dynamic processes. Here, we describe how Bragg coherent diffraction imaging (BCDI; refs

11,12) can be used to visualize in three dimensions, the entire network of dislocations present within an individual calcite crystal during repeated growth and dissolution cycles. These investigations demonstrate the potential of BCDI for studying the mechanisms underlying the response of crystalline materials to external stimuli.

Crystal growth and dissolution processes have been studied for over a century13, owing to their significance to fields such as geology, corrosion, catalysis and the synthesis of nanostructures, a desire to understand the link between microscopic and macroscopic processes, and to our innate fascination with such structures. These investigations confirmed that dissolution and growth proceed by analogous mechanisms14 and also identified that crystallographic defects, and in particular screw dislocations, are of fundamental importance to crystal growth and dissolution processes15. This can be attributed to the fact that screw dislocations cause deformation (and therefore strain) of the adjacent crystal lattice, which in turn alters the activation barrier for growth and dissolution in the vicinity of the dislocation16. A full picture of crystal growth and dissolution mechanisms can therefore be obtained only by studying the relationship between the evolution of the network of dislocations within a crystal and its morphology during these processes.

In this article, we demonstrate how BCDI (refs 17,18) can be used to study the role of dislocations in dictating the mechanism of growth and dissolution of calcite crystals. Calcite was selected for study as it is one of the most-studied inorganic crystals, and the crystals readily grow to a few micrometres in size, with well-defined morphologies. BCDI is an imaging technique that uses coherent X-rays to image the density (and morphology) of a crystal and, importantly, the strain within it11,12. Illumination of a crystal that is smaller than the coherence volume of the beam generates a coherent X-ray diffraction (CXD) pattern due to scattering from all parts of the crystal. An image of the scattered wave is reconstructed using iterative phase retrieval algorithms11,12,17–19. The reconstructed density is complex valued, with the amplitude containing information about the electron density, $\rho(\mathbf{r})$. Phase shifts in the reconstructed complex density arise from strain (internal deformation) in the crystal lattice. The phase is proportional to the vector displacement

2

field, $\mathbf{u}(\mathbf{r})$ of the atoms from the ideal lattice points and the scattering vector \mathbf{Q} via $\varphi(\mathbf{r})=\mathbf{Q}\cdot\mathbf{u}(\mathbf{r})$ (see Supplementary Information). For a single Bragg peak, a single projection of \mathbf{u} is obtained and components of \mathbf{u} perpendicular to \mathbf{Q} will not be observed. It is this sensitivity to deformations that makes BCDI ideal for studying defects within crystals.

Calcite crystals were precipitated by placing 100 μ l droplets of a solution containing CaCl₂, urea and urease on hydroxyl-terminated, self-assembled monolayers (SAMs) supported on thin gold films (see Methods). CaCO3 precipitation then occurred on enzymatic hydrolysis of the urea to ammonium and carbonate20. This method was selected as it generated a high density of {104} oriented calcite rhombohedra with average diameters of \approx 1.25 µm (ref. 21). BCDI experiments were carried out at beamline I16 at the Diamond Light Source. Individual calcite crystals below 2 μm in size were illuminated with monochromatic, 8 keV X-rays, and diffraction was recorded at the {104} Bragg peak (see Methods). Three-dimensional (3D) diffraction data sets (Supplementary Fig. 1) were obtained by rocking an isolated calcite crystal through its Bragg peak, and the same crystal was monitored while undergoing cycles of growth and dissolution. Dissolution was achieved by depositing dilute acetic acid solution on the crystal, whereas growth was achieved by adding a drop of calcium bicarbonate solution (see Methods). Alignment of the crystal was maintained throughout, as the X-rays were nominally unfocussed and defined by slits with a square opening of 200 µm placed 0.3m before the sample. This large beam size relative to crystal size also ensured that the sample was coherently illuminated22.

Figure 1 shows 3D images of the crystals as iso-surface renderings of the reconstructed amplitudes (electron density, see Supplementary Information; Fig. 1a–d) and phase (projected lattice displacement; Fig. 1e–h) of the initial crystal (Fig. 1a,e, stage i), after growth (Fig. 1b,f, stage ii) and after two consecutive dissolution steps (Fig. 1c,g, stage iii, and d,h, stage iv). The initial crystal (Fig. 1a) is the expected rhombohedron, which is consistent with scanning electron micrographs (Supplementary Fig. 3). Growth of this crystal (Fig. 1b,c) leads to an increase in size and a smoothing of the faces exposed to the bulk solution, whereas the face in contact with the SAM remains unchanged. Interestingly, two of the faces directed into the solution (indicated with blue arrows) grow more rapidly than the other three. This is immediately indicative of a non-uniform distribution of defects, as all

faces of an entirely perfect rhombohedron would be expected to grow at the same rate. Images of the corresponding projected displacements are shown in Fig. 1e–h, where this is mapped onto an iso-surface with red and blue representing lattice contraction or expansion respectively by half a lattice spacing. Comparison of the projected displacements before (Fig. 1e) and after (Fig. 1f) crystal growth shows that these do not grow significantly with the crystal but remain maximal at the edges. This is indicative of the presence of active growth fronts23.

The dissolution steps (Fig. 1b,c and c,d) show that the crystal faces retreat along all directions, but is more pronounced at certain sites (indicated by red arrows). Initial signs of changes in the crystal shape and the onset of etch-pit formation are visible after the first stage of dissolution (Fig. 1c), leading to an increase in the specific surface area and roughness of the crystal. The etch pits are also associated with higher levels of deformation/strain (Fig. 1g). That relatively little change occurs in the crystal face adjacent to the substrate is consistent with the intimate contact of the SAM with this nucleation face. The second dissolution step (Fig. 1c,d) results in a significant change in the crystal morphology, and the production of a porous isometric form24 that is quite distinct from the original shape, and which can be attributed to the removal of defect outcrops at the crystal surface and the coinciding etch pits formed25,26. Looking at the lattice deformation, it is indicates that the least stable (more strained) regions dissolve first, leaving behind a more stable core (Supplementary Movies 1–4).

The projected displacement images also reveal a further feature, which is indicated by the grey arrows in Fig. 1e–g, and is present throughout the growth and dissolution of the crystal. This region possesses both a hollow core and a spiral phase, where this combination of features is characteristic of dislocations and agrees with 2D images of dislocations obtained with TEM (ref. 8) and Bragg ptychography27. A number of these dislocations are highlighted in Fig. 2. Dislocations are characterized by a Burgers vector that measures the topological shift of the crystal along the dislocation line, where this is usually a lattice vector of the crystal28. Whenever there is a component of the Burgers vector parallel to the dislocation line, it has a screw dislocation character causing the lattice to spiral around the

dislocation; in this way crystal growth (and dissolution) can be facilitated. Confirmation that this feature indeed corresponds to a screw dislocation was obtained by recording the polar angle dependence of the displacement associated with the core (indicated by the circles in Fig. 2) over the growth/ dissolution cycle of the crystal (Fig. 3a). Although calcite is elastically anisotropic, an approximately linear relationship was observed and is consistent with what is predicted by linear elasticity theory28 (see Supplementary Information). To further confirm the nature of the identified dislocation, a comparison is provided between a simulated screw dislocation and simulated screw after BCDI processing. The model screw dislocation used in the simulation is shown at atomic resolution in Fig. 4a, with the resulting displacement of atoms from their ideal lattice positions given in Fig. 4c. The BCDI experiment simulated screw dislocation is shown in Fig. 4b,d (see Supplementary Information). Figure 4e,f provide a comparison of BCDI-reconstructed displacement and simulated displacement, viewed along the dislocation line, and clearly show that the low-density core and spiral displacement are well-preserved after BCDI processing.

Further examination of regions that showed a spiral deformation and low-amplitude core enabled many more dislocations to be identified within the imaged crystal (see Supplementary Information). These have been rendered and are shown in Fig. 5 and Supplementary Movies 5–8. In many cases the exact type of dislocation cannot be identified from a single Bragg reflection, as edge, screw and mixed dislocations will all possess qualitatively similar deformation fields. The initial crystal (Fig. 5a) possesses several dislocations which are located relatively close to the crystal surface (down to 200 nm) and are found predominantly parallel to {104} planes. These dislocations are mainly associated with the faces exposed to the bulk solution, and are visible throughout a single crystal growth and dissolution cycle (Fig. 5a–c), during which they increase (Fig. 5b) and decrease (Fig. 5c) in length. The reproducibility of locating the same dislocation across independently reconstructed data sets provides further credence to the recovered images.

Dislocations are often found to occur in loops or pairs with opposite Burgers vector, as this reduces the lattice potential energy and resulting long-range strain associated with these features. They also are stabilized near crystal surfaces, as found in this work, such that dislocation motion can transport material into and out of crystals from the growth

5

solution29. As intuitively expected, the most rapidly growing crystal faces were observed to have the highest number of surface dislocations. The distribution of these defects within the crystal therefore plays a part in determining the morphology of the product crystal. Considering the dissolution process, the locations of the dislocations coincide with the position of the etch pits that appear at the crystal surface during dissolution. This suggests that the developing pits follow the cores of the dislocations, where this effect can be attributed to a reduced activation barrier to dissolution due to stored strain energy in the defects14. New dislocations could be identified during the dissolution (Fig. 5c,d), and by the final stage (Fig. 5d) many of the original dislocations had been annihilated. The loss of faceting and dislocations near the surface support the hypothesis that the least stable regions dissolve first, leaving behind a more stable core.

To determine the overall effect that the growth/dissolution had on the strain and deformation of the crystal, the gradient of the displacement was calculated and its magnitude was plotted as a function of fractional size (Fig. 3b,c). The magnitude of the gradient was calculated over successively larger shells, averaged over all directions, and is plotted as a percentage of deformation (strain) relative to the lattice constant for the {104} reflection (see Supplementary Information) for the directions parallel and perpendicular to the scattering vector. What is evident (Fig. 3b,c) is that the initial crystallite experiences an increasing strain from the centre, outwards. This can be compared to a crystal formed after the first growth stage (Fig. 1f), which shows an overall flatter initial strain, and then increases more rapidly above a fractional size ≈0.6. The smoother transition and lower strain for the direction parallel suggest some relaxation of the crystal after the initial growth and can also be seen in the slices in Fig. 2. The non-zero strain perpendicular to the scattering vector is indicative of a shear strain, although we have only a single Bragg peak so the full strain tensor cannot be obtained and the exact nature of this cannot be determined. These data are consistent with the assertion that the effect of the surface penetrates only to a finite distance into a crystal. With the increased roughness and etch-pit formation that occurs in the first dissolution step (Fig. 1g), the overall strain is higher, demonstrating that these surface features affect the entire crystal, leading to an increase in overall deformation. After the final dissolution (Fig. 1h), when the majority of surface dislocations

have been removed, the strain becomes almost flat. This indicates that surface effects are less pronounced, despite the increased relative surface area.

With its ability to simultaneously generate 3D images of the strain within a crystal and the gross crystal morphology, BCDI provides an extremely powerful way of visualizing the network of dislocations present within an individual crystal. Importantly, BCDI can be performed in the absence of any sample preparation, which means that it can be used to elucidate the effects of dislocation networks on dynamic crystal behaviour. Using the example of crystal growth and dissolution, we have located dislocations within calcite single crystals, and then demonstrated that their distribution within the crystal dictates rapidly growing directions. Conversely, preferential dissolution and etch-pit formation was also observed within the vicinity of the dislocations, which provides evidence that the stored energy within the dislocations affects the crystal response. The ability to view these fundamental processes using BCDI represents an important step forward in elucidating the nanoscale mechanisms underlying crystallization processes. Importantly, BCDI also opens the door to 3D visualization of the role of dislocations in the response of a crystal to a huge range of experimental conditions, such as temperature changes or mechanical force.

Methods

Methods and any associated references are available in the online version of the paper.

Received 12 January 2015; accepted 4May 2015; published online 1 June 2015

Figure 1 | Growth and dissolution of calcite observed by BCDI. **a**–**h**, Iso-surface rendering of the electron density (reconstructed amplitude) (**a**–**d**) and projected displacement (phase) (**e**–**h**) from initially deposited calcite crystal (**a**,**e**, stage i), after secondary growth (**b**,**f**, stage ii), and after consecutive dissolution steps (**c**,**g**, stage iii, and **d**,**h**, stage iv). The scale bar is 1 µm. Three different viewing angles of the crystal are shown—top down, side and bottom up. The substrate would be on the bottom of the crystal shown in the side view. Perspectives highlight the shape transition that occurs during growth (**a**,**b** and **e**,**f**) and dissolution, red arrows) directions are shown. The grey arrow points towards the primary dislocation continuously identifiable during crystal growth and dissolution (**a**–**c** and **e**–**g**). The beam direction is along the *z* axis, with the *y* axis vertical. The scattering vector (**Q**) is shown in red.



Figure 2 | Slices showing the electron density and projected displacement during growth and dissolution. **a**–**d**, Crystal in the initial state (**a**), and after growth (**b**) and repetitive dissolution steps (**c**,**d**) for two viewing directions (top view) and (side view). The amplitude is shown in the top rows (black background) with the dislocations highlighted by white arrows featuring a low-amplitude core. The phase presented in the bottom rows also shows selected dislocations highlighted by the dark blue arrows. Particularly evident is the spiral phase (displacement) that is characteristic of a screw dislocation. The iso-surface to the right shows the location of the cut planes. The scale bar is 1 μ m.



Figure 3 | Displacement and strain magnitude line plots. **a**, Recovered projected displacement measured at development stages (i–iv) as a function of theta for the dark blue circles with arrows from Fig. 2a. This is compared to the displacement expected by linear elasticity theory (LET). The error bars represent the standard deviation $(\pm \sigma)$ of the displacement over adjacent layers spanning a region of 200nm along Q. **b**,**c**, The magnitude of the strain component parallel (**b**) and perpendicular (**c**) to the scattering vector over the growth/dissolution cycle (i–iv) plotted as a function of fractional crystal size (centre of the crystal 0 to crystal surfaces 1). These graphs highlight the diminishing relevance of surface effects with growth (ii) of the crystal, and its increasing relevance with dissolution (iii and iv). The strain is calculated as the magnitude of the gradient of the displacement and averaged over successively larger shells.



Figure 4 | Simulation of a screw dislocation. **a**, Iso-surface rendering of a screw dislocation with atomic resolution. **b**, Iso-surface rendering by filtering the Fourier transform of **a** with a Gaussian centred on the (0,1,0) Bragg peak, replicating the BCDI experiment. **c**,**d**, The displacement is rendered onto the iso-surface for the atomic resolution (**c**) and the phase is rendered onto the surface for the BCDI simulation (**d**) showing the spiral phase centred around the dislocation core. **e**,**f**, Another view of the dislocation for the atomic resolution (**e**) and BCDI experiment (**f**), revealing the low-density region at the core in the BCDI experiment simulation. It should be noted that for the atomic resolution the displacement is mapped onto the iso-surface, whereas for the BCDI example the resultant phase is mapped onto the iso-surface.



Figure 5 | Iso-surface rendering of defect network within a calcite crystal. **a**–**d**, Iso-surface renderings of dislocations present within the crystal imaged before (**a**) and after (**b**) growth, and after dissolution (**c**,**d**). The scale bar is 1 μ m. The evolution of the dislocations is evident through crystal growth (**a**,**b**) and dissolution (**b**,**c** and **c**,**d**), where these are shown from left to right. Three different viewing angles of the crystals are provided (vertically, top to bottom)—from the base, above and the side. Dislocations are predominantly located near the surface and on the faces that grow most rapidly. The phase has been mapped to the iso-surfaces of present dislocations, showing the characteristic spiral.



References

1. Stoneham, A. Theory of Defects in Solids (Clarendon, Oxford, 1985).

2. Burton, C. A theory concerning the constitution of mater. Phil. Mag. 33, 191–203 (1892).

3. Lasaga, A. C. & Luttge, A. Variation of crystal dissolution rate based on a dissolution stepwave model. Science 291, 2400–2404 (2001).

4. De Yoreo, J. J. & Vekilov, P. G. Principles of crystal nucleation and growth. Reviews in Mineralogy and Geochemistry 54, 57–93 (2003).

5. Ma, E., Shen, T. D. & Wu, X. L. Nanostructured metals: Less is more. Nat Mater 5, 515–516 (2006).

6. Kunitake, M. E., Mangano, L. M., Peloquin, J. M., Baker, S. P. & Estroff, L. A. Evaluation of strengthening mechanisms in calcite single crystals from mollusk shells. Acta Biomaterialia 9, 5353–5359 (2013).

7. Davis, K. J., Dove, P. M. & De Yoreo, J. J. The role of Mg2+ as an impurity in calcite growth. Science 290, 1134–1137 (2000).

8. Hytch, M. J., Putaux, J.-L. & Penisson, J.-M. Measurement of the displacement field of dislocations to 0.03[thinsp]a by electron microscopy. Nature 423, 270–273 (2003).

9. Barnard, J. S., Sharp, J., Tong, J. R. & Midgley, P. A. High-resolution three-dimensional imaging of dislocations. Science 313, 319 (2006).

10. Chen, C.-C. et al. Three-dimensional imaging of dislocations in a nanoparticle at atomic resolution. Nature 496, 74–77 (2013).

11. Pfeifer, M. A., Williams, G. J., Vartanyants, I. A., Harder, R. & Robinson, I. K. Three dimensional mapping of a deformation field inside a nanocrystal. Nature 442, 63–66 (2006).

12. Robinson, I. & Harder, R. Coherent x-ray diffraction imaging of strain at the nanoscale. Nat Mater 8, 291–298 (2009).

13. Weeks, J. D. & Gilmer, G. H. Dynamics of crystal growth. In Advances in Chemical Physics, 157–228 (John Wiley and Sons, Inc., 2007).

14. Dove, P. M. & Han, N. Kinetics of mineral dissolution and growth as reciprocal microscopic surface processes across chemical driving force. AIP Conference Proceedings 916, 215–234 (2007).

15. Frank, F. C. The influence of dislocations on crystal growth. Discussions of the Faraday Society 5, 48–54 (1949).

16. Brantley, S. L. Kinetics of mineral dissolution. In Brantley, S. L., Kubicki, J. D. & White, A. F. (eds.) Kinetics of Water-Rock Interaction, 151–210 (Springer New York, 2008).

17. Clark, J. N. et al. Ultrafast three-dimensional imaging of lattice dynamics in individual gold nanocrystals. Science 341, 56–59 (2013).

18. Cha, W. et al. Core–shell strain structure of zeolite microcrystals. Nat Mater 12, 729–734 (2013).

19. Fienup, J. R. Phase retrieval algorithms: a comparison. Applied Optics 21, 2758–2769 (1982). 20. Antipov, A. et al. Urease-catalyzed carbonate precipitation inside the restricted volume of polyelectrolyte capsules. Macromolecular Rapid Communications 24, 274–277 (2003).

21. Lee, J. R. I. et al. Structural development of mercaptophenol self-assembled monolayers and the overlying mineral phase during templated $CaCO_3$ crystallization from a transient amorphous film. J. Am. Chem. Soc. 129, 10370–10381 (2007).

22. Clark, J. N., Huang, X., Harder, R. & Robinson, I. K. High-resolution three-dimensional partially coherent diffraction imaging. Nat Commun 3, 993 (2012).

23. Paquette, J. & Reeder, R. J. Relationship between surface structure, growth mechanism, and trace element incorporation in calcite. Geochimica et Cosmochimica Acta 59, 735–749 (1995).

24. Snyder, R. C. & Doherty, M. F. Faceted crystal shape evolution during dissolution or growth. AIChE Journal 53, 1337–1348 (2007).

25. MacInnis, I. N. & Brantley, S. L. The role of dislocations and surface morphology in calcite dissolution. Geochimica et Cosmochimica Acta 56, 1113–1126 (1992).

26. MacInnis, I. N. & Brantley, S. L. Development of etch pit size distributions on dissolving minerals. Chemical Geology 105, 31–49 (1993).

27. Takahashi, Y. et al. Bragg x-ray ptychography of a silicon crystal: Visualization of the dislocation strain field and the production of a vortex beam. Phys. Rev. B 87, 121201 (2013).

28. Hirth, J. P. & Lothe, J. Theory of Dislocations (McGraw-Hill, New York, 1968).

29. Pun, G. P. & Mishin, Y. A molecular dynamics study of self-diffusion in the cores of screw and edge dislocations in aluminum. Acta Materialia 57, 5531 – 5542 (2009).

Acknowledgements This work was supported by FP7 advanced grant from the European Research Council (J.N.C. and I.K.R.) and an Engineering and Physical Sciences Research Council Leadership Fellowship (F.C.M. and J.I.). It was also funded through an EPSRC Programme Grant (A.S.S. and F.C.M., EP/I001514/1) which funds the Materials in Biology (MIB) consortium, and EPSRC grants EP/J018589/1 (YYK) and EP/K006304/1 (ANK).We thank Diamond Light Source for access to Beamline I-16 (MT 8187, MT 7654 and MT 7277) that contributed to the results presented here.

Author contributions

J.N.C. and J.I. designed the project; J.I. prepared samples; J.N.C., J.I., J.M.C., A.S.S., Y-Y.K., J.M.C., G.N. and I.K.R. performed the experiments; J.N.C. performed image reconstructions; J.N.C. and I.K.R. analysed the data, J.N.C., J.I., F.C.M. and I.K.R. wrote the paper. All the authors read and commented on the manuscript.

Additional information

Supplementary information is available in the online version of the paper. Reprints and permissions information is available online at www.nature.com/reprints. Correspondence and requests for materials should be addressed to J.N.C. or F.C.M.

Competing financial interests

The authors declare no competing financial interests.