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# Phospho*enol*pyruvate Carboxykinase Assayed at Physiological Concentrations of Metal Ions Has a High Affinity for CO<sub>2</sub><sup>1</sup>

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The effect of  $Mn^{2+}/Mg^{2+}$  concentration on the activity of intact, homogeneous phosphoenolpyruvate carboxykinase (PEPCK) from leaves of the C<sub>4</sub> grass, Guinea grass (*Panicum maximum*), have been investigated. Assay conditions were optimized so that PEPCK activity could be measured at concentrations of  $Mn^{2+}/Mg^{2+}$  similar to those found in the cytosol (low micromolar  $Mn^{2+}$  and millimolar  $Mg^{2+}$ ). PEPCK activity was totally dependent on  $Mn^{2+}$  and was activated at low micromolar concentrations of  $Mn^{2+}$  by millimolar concentrations of  $Mg^{2+}$ . Therefore, at physiological concentrations of  $Mn^{2+}$ , PEPCK has a requirement for  $Mg^{2+}$ . Assay at physiological concentrations of  $Mn^{2+}/Mg^{2+}$  led to a marked decrease in its affinity for ATP and a 13-fold increase in its affinity for CO<sub>2</sub>. The  $K_m$  (CO<sub>2</sub>) was further decreased by assay at physiological ATP to ADP ratios, reaching values as low as 20  $\mu$ M CO<sub>2</sub>, comparable with the  $K_m$  (CO<sub>2</sub>) of ribulose 1,5-bisphosphate carboxylase-oxygenase. This means that PEPCK will catalyze a reversible reaction and that it could operate as a carboxylase in vivo, a feature that could be particularly important in algal CO<sub>2</sub>-concentrating systems.

Phospho*enol*pyruvate carboxykinase (PEPCK-ATP; EC 4.1.1.49) is a Mn<sup>2+</sup>-dependent enzyme that catalyzes the reversible reaction:

oxaloacetate + ATP  $\leftrightarrow$  PEP + ADP + CO<sub>2</sub>

This reaction is important in plant metabolism because it lies at an interface between organic acid, amino acid, and sugar metabolism. In keeping with the importance of this reaction the presence of PEPCK in a wide range of plant tissues is now emerging, including structures involved in plant defense, such as trichomes and oil and resin ducts, in flowers, fruits, and developing seeds, and in the phloem of some plants. In at least some of these tissues, it plays a role in nitrogen metabolism and its abundance may change greatly and rapidly in response to changes in the nitrogen status of the tissue (Leegood and Walker, 1999; Walker et al., 1999, 2001). In addition, it is well established that PEPCK is involved gluconeogenesis, converting stored fats to sugars after germination in oil-storing seeds (Leegood and ap Rees, 1978) and in the photosynthetic  $CO_2$ -concentrating mechanisms present in both PEPCK-type and some NADP-malic enzyme-type C<sub>4</sub> plants and in plants with Crassulacean acid metabolism (Leegood et al., 1996; Walker and Leegood, 1996; Wingler et al., 1999). In higher plants it has always been thought that PEPCK acts as a decarboxylase in vivo because of its low affinity for  $CO_2$  (Ray and Black, 1976; Urbina and Avilan, 1989). In both gluconeogenesis and in  $C_4$  and Crassulacean acid metabolism photosynthesis, PEPCK certainly acts as a decarboxylase, but in some aquatic plants and algae it has also been proposed to act as a carboxylase (Reiskind and Bowes, 1991).

A difficulty with the suggestion that PEPCK acts as a carboxylase in some tissues has been the low affinity of the enzyme for CO<sub>2</sub> when measured in in vitro assay. It is possible that this low affinity is a result of assay of the enzyme at non-physiological concentrations of metal ions (Walker et al., 1997). For example, all previous studies have assayed PEPCK at completely unphysiological concentrations of Mn<sup>2+</sup> (>0.5 mm) and in the absence of  $Mg^{2+}$ .  $Mg^{2+}$  has also been shown to be inhibitory (Burnell, 1986). However, the concentration of Mn<sup>2+</sup> in the cytosol of plant and animal cells is submicromolar as, for example, in maize roots (Quiquampoix et al., 1993), and the concentration of  $Mg^{2+}$  in the cytosol of plant cells is millimolar as, for example, in mung bean (*Vigna* radiata) roots (Yazaki et al., 1988). In addition, previous studies of PEPCK have used the proteolytically cleaved form of the enzyme, the properties of which may differ from the intact enzyme (Walker et al., 1997, 2002).

In this paper we show how PEPCK activity can be measured in in vitro assay at physiological concentrations of  $Mn^{2+}/Mg^{2+}$  and that under such condi-

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tions the affinity of PEPCK for  $CO_2$  is greatly increased.

### RESULTS

The effects of  $\ensuremath{\text{Mn}^{2+}}\xspace$  and  $\ensuremath{\text{Mg}^{2+}}\xspace$  concentration on the carboxylation and decarboxylation activities of pure, intact PEPCK from illuminated leaves of the C<sub>4</sub> plant, Guinea grass (Panicum maximum) were characterized (Fig. 1). For the carboxylation reaction, in the absence of 2-mercaptoethanol (ME), there was no activity at 10  $\mu$ M Mn<sup>2+</sup>, little activity at 100  $\mu$ M Mn<sup>2+</sup> and maximum activity at millimolar concentrations of  $Mn^{2+}$ . Under these conditions,  $Mg^{2+}$  was inhibitory, with more than 50% inhibition at 5 mM  $Mg^{2+}$ . In the presence of high concentrations of ME (500 mм), the characteristics of the carboxylation reaction were different. First, there was some activity at 10  $\mu$ M Mn<sup>2+</sup> and this was greatly increased by inclusion of  $Mg^{2+}$ . Second, there was substantial activity at 100 µM  $Mn^{2+}$ , but  $Mg^{2+}$  was inhibitory, as at 5 mM  $Mn^{2+}$ . For the decarboxylation reaction, in the absence of ME, there was essentially no activity at 10 or 100  $\mu$ M Mn<sup>2+</sup>, but activity was greatly stimulated by the presence of  $Mg^{2+}$ . Maximum activity was still observed at 5 mM  $Mn^{2+}$  and  $Mg^{2+}$  was inhibitory. In the presence of ME, the pattern of response to  $Mn^{2+}$ and Mg<sup>2+</sup> was much the same, except that there was an inhibition of PEPCK activity compared with the absence of ME. ME was therefore inhibitory to the decarboxylation reaction, but stimulated the carboxvlation reaction dramatically at low micromolar concentrations of Mn<sup>2+</sup> in the presence of millimolar concentrations of  $Mg^{2+}$  (Fig. 2).

Kinetic constants were determined for PEPCK from illuminated leaves (light) at both 5 mm  $Mn^{2+}$  and 10



 $\mu M Mn^{2+}/4 m Mg^{2+}$ . There were decreases in the affinities of PEPCK for phosphoenolpyruvate (PEP) and ADP in the carboxylation reaction and for oxaloacetate (OAA) and ATP in the decarboxylation reaction when assayed at micromolar concentrations of  $Mn^{2+}$  in the presence of 4 mM Mg<sup>2+</sup>. The increase in  $K_m$  was substantial in the case of ATP. However, there was a large increase in the affinity of PEPCK for  $CO_2$  when  $Mn^{2+}$  was lowered from 5 mm to 10  $\mu$ m. Figure 3 shows that at least part of the decrease in  $K_{\rm m}$  $(CO_2)$  was the result of the inclusion of Mg<sup>2+</sup> (compare plots with 5 mM  $Mn^{2+}$  and 5  $\mu M Mn^{2+}/4$  mM  $Mg^{2+1}$ ). There was no further decrease in the  $K_{\rm m}$  $(CO_2)$  when PEPCK was assayed in the presence of 1  $\mu$ M Mn<sup>2+</sup> (Fig. 3). However, inclusion of ATP as well as ADP in the carboxylation reaction resulted in a further 3-fold reduction in the  $K_{\rm m}$  (CO<sub>2</sub>) as the ATP to ADP ratio was increased (Fig. 4).

When the properties of PEPCK purified from darkened leaves (dark) were compared with that purified from illuminated leaves (light) at 10  $\mu$ M Mn<sup>2+</sup>/4 mM Mg<sup>2+</sup> (Table I), there was no difference in the affinity of the light and dark enzymes for CO<sub>2</sub> and there were no significant changes in affinities for PEP and ADP in the carboxylation reaction. There were substantial decreases in the affinities for OAA and ATP in the decarboxylation reaction with an approximate doubling of the  $K_m$ . These are the result of lightdependent changes in the phosphorylation state of PEPCK, as discussed by Walker et al. (2002).

## DISCUSSION

Previous investigators of the assay conditions and kinetics of PEPCK from plants have assayed it at unphysiological concentrations of  $Mn^{2+}$  (> 0.5 mM)

**Figure 1.** Effect of  $Mn^{2+}$  and  $Mg^{2+}$  concentration on the carboxylation and decarboxylation activities of Guinea grass PEPCK, measured in either the presence or absence of ME. Substrate concentrations for the carboxylation assay were 0.5 mm ADP/5 mm PEP and for the decarboxylation assay 0.5 mm ATP/200  $\mu$ m OAA.



**Figure 2.** Effect of ME concentration on both the carboxylation and decarboxylation activities of Guinea grass PEPCK, measured at 10  $\mu$ M Mn<sup>2+</sup>/5 mM Mg<sup>2+</sup>. Substrate concentrations for the carboxylation assay were 0.5 mM ADP/5 mM PEP and for the decarboxylation assay 0.5 mM ATP/200  $\mu$ M OAA.

and have concluded that plant PEPCK is strongly inhibited by millimolar concentrations of Mg<sup>2+</sup> (Burnell, 1986; Walker et al., 1997). This behavior is different to that reported for the enzyme from non-plant tissues, in which a synergistic activation by a combination of  $Mn^{2+}$  and  $Mg^{2+}$  occurs, e.g. the enzyme from yeast (Cannata and Stoppani, 1963) and Trypanosoma cruzii (Jurado et al., 1996), which show substantial sequence similarity to the plant enzyme, and the enzyme from rat liver, which shows no significant sequence similarity (Foster et al., 1967). It has been suggested that, in the rat liver enzyme, Mg<sup>2+</sup> forms a MgITP<sup>2-</sup> complex that acts as a substrate for the reaction, whereas Mn<sup>2+</sup> acts as an activator at a separate site (Foster et al., 1967). The present results resolve this discrepancy between PEPCK from plant and non-plant tissues and show that PEPCK from



**Figure 3.** Hanes-Woolf plots, which show the effect of  $Mn^{2+}$  and  $Mg^{2+}$  concentration on the affinity of Guinea grass PEPCK for CO<sub>2</sub>. Substrate concentrations were 5 mM PEP/0.5 mM ADP.



**Figure 4.** Effect of ATP to ADP ratio (total adenylate concentration 1 mM) on the affinity of Guinea grass PEPCK for CO<sub>2</sub>. These assays used 10  $\mu$ M Mn<sup>2+</sup>/4 mM Mg<sup>2+</sup>/5 mM PEP. Michaelis constants were determined from Hanes-Woolf plots and are the means ± sE of three separate determinations.

Guinea grass is (a) totally dependent on  $Mn^{2+}$ , (b) that it can operate at physiological ( $\mu$ M) concentrations of  $Mn^{2+}$ , and (c) that physiological (mM) concentrations of  $Mg^{2+}$  activate the enzyme at physiological concentrations of  $Mn^{2+}$  (presumably forming an MgATP<sup>2-</sup> substrate complex). PEPCK thus has a dual requirement for both  $Mn^{2+}$  and  $Mg^{2+}$ . However, it is also clear that measurements of the maximum activity of PEPCK in plant extracts should still be made under conditions of saturating  $Mn^{2+}$  in the absence of  $Mg^{2+}$ .

The improvements made to the assay of PEPCK involve the inclusion of high concentrations of ME. A notable property of both ATP-dependent (plant) and GTP-dependent PEPCKs is their inactivation by thiol-modifying reagents (Chang and Lane, 1966), affecting nucleotide binding at the active site (Lewis et al., 1993). Although the sequence of the ATPdependent PEPCK from higher plants, yeast, and many bacteria shows little similarity to the GTP- or ITP-dependent enzyme found in animals and other bacteria, the active site has considerable homology in both ATP- and GTP- or ITP-dependent PEPCKs. Reactive thiol groups in ATP-dependent PEPCKs occur in yeast (Cardemil et al., 1990) and T. cruzii (Jurado et al., 1996). Cardemil et al. (1990) reported that the inactivation of yeast PEPCK by thiol reagents is caused by modification of both thiol and vicinal dithiol groups within the active site of each subunit and that loss of activity was effectively prevented by the combined presence of ATP plus Mn<sup>2+</sup> (see also Walker et al., 1997). Whether or not this loss of activity, that can be more severe for the carboxylation than for the decarboxylation reaction (e.g. Ray and Black, 1976), is related to the changes in metal ion dependence with ME for the carboxylation reaction, as shown in Figure 1, has yet to be resolved. Another possibility is that ME is acting to chelate inhibitory trace metals or to stabilize the concentration of  $Mn^{2+}$ .

 Table I. Kinetic constants for PEPCK from illuminated leaves of Panicum maximum

Carboxylase activity was measured using 10 mM CO<sub>2</sub>/0.5 mM ADP/5 mM PEP) and decarboxylase activity using 0.5 mM ATP/200  $\mu$ M OAA. Michaelis constants were determined from Hanes-Woolf plots and are the means  $\pm$  sE of three separate determinations. Measurements are for the enzyme purified from illuminated leaves except for the row labelled dark, which are for the enzyme purified from darkened leaves.

Metal lons in Assay	K <sub>m</sub>				
	CO <sub>2</sub>	PEP	ADP	OAA	ATP
	$\mu$ M	тм	тм	μм	μм
5 mм Mn <sup>2+</sup>	732 ± 90	$1.38 \pm 0.21$	$18 \pm 2.0$	85 ± 15	$1.5 \pm 0.2$
5 mм Mn <sup>2+</sup> + 4 mм Mg <sup>2+</sup>	$238 \pm 55$	$1.67 \pm 0.30$	$17 \pm 2.5$	$100 \pm 16$	$1.1 \pm 0.3$
10 µм Mn <sup>2+</sup> + 4 mм Mg <sup>2+</sup>					
Light	$55 \pm 18$	$2.10 \pm 0.25$	$36 \pm 3.3$	$156 \pm 20$	$25.7 \pm 3.5$
Dark	$55 \pm 19$	$2.65 \pm 0.31$	$44 \pm 3.1$	$304 \pm 25$	$43.5 \pm 3.3$

However, the fact that ME stimulated the carboxylation reaction but was inhibitory to the decarboxylation reaction suggests that this explanation is less likely.

An important consequence of assay of PEPCK at physiological concentrations of metal ions is the effect on its substrate affinities, in particular a marked decrease in its affinity for ATP and a 13-fold increase in its affinity for CO<sub>2</sub>. PEPCK from higher plant and algal sources has always been notable for its low affinity for  $CO_2$ . For example, the  $K_m$  ( $CO_2$ ) in Guinea grass has been estimated at 1.36 and 1.61 mм (calculated from Urbina and Avilan [1989] and Ray and Black [1976], respectively), 0.497 mм in pineapple (calculated from Daley et al. [1977]) and 0.175 to 1.21 mм in a range of brown and green algae (Table II in Johnston and Raven, 1983). This low affinity for CO<sub>2</sub> has been particularly controversial in the algae in which it has been proposed that PEPCK may sometimes act as a carboxylase (Reiskind and Bowes, 1991; Johnston and Raven, 1983). The present results suggest that the  $K_{\rm m}(\rm CO_2)$  of PEPCK can be many-fold lower than previous estimates, reaching as low as 20  $\mu$ M CO<sub>2</sub> at physiological ATP to ADP ratios (an ATP to ADP ratio above 4:1 is typical in the cytosol of wheat leaf mesophyll protoplasts in the light; Stitt et al., 1982). It is not clear how the ATP to ADP ratio affects the  $K_{\rm m}(\rm CO_2)$ . Although chelation of metal ions could be involved, ATP is known to be an effector of PEPCK (Walker et al., 2002). The  $K_m(CO_2)$  of PEPCK at high ATP to ADP ratios is comparable with the  $K_{\rm m}({\rm CO}_2)$  of Rubisco from C<sub>3</sub> and Crassulacean acid metabolism plants (12–26  $\mu$ M) or from C<sub>4</sub> plants (28–63 μm; Yeoh et al., 1980, 1981). This means that PEPCK will catalyze a reversible reaction and that it could operate as a carboxylase in vivo. Clearly the balance between carboxylation and decarboxylation will be strongly influenced by the ATP to ADP ratio, that will affect both the equilibrium of the reaction and the  $K_{\rm m}$  (CO<sub>2</sub>), and by the availability of reductant or amino donors that will determine the concentration of OAA. Further studies of algal PEPCK are now required.

### MATERIALS AND METHODS

#### **Plant Material**

Seeds of Guinea grass (*Panicum maximum*) were obtained from the Kew Seed Bank (Royal Botanical Gardens, Kew, UK). Plants were grown in soil in a greenhouse during the summer with no supplementary light.

#### **Purification of PEPCK**

PEPCK was purified from both darkened and illuminated leaves of Guinea grass as described by Walker et al. (2002).

#### Assay of PEPCK

Carboxylase activity was measured as described by Walker et al. (1995), and decarboxylase activity was measured as described by Lee et al. (1981). Modifications to these procedures in individual experiments are described in the text. For determination of  $CO_2$  affinity, solutions were made up in boiled acidified water that had been purged with N<sub>2</sub>. The  $CO_2$  concentration was calculated using a pK of 6.365 at 25°C (Umbreit et al., 1972). One unit of PEPCK activity corresponds to the production of 1  $\mu$ mol product/min at 25°C.

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