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2	Durability and degradation mechanisms of GFRP reinforcement subjected to severe environments and
3	sustained stress
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5	
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8	
9	Abstract
10	Despite the large amount of research that has been carried out to date on the use of glass fibre reinforced
11	polymer (GFRP) bars in concrete, one factor still hindering their widespread use in civil engineering
12	applications is the lack of comprehensive data on their long-term in-service performance. This paper presents
13	the test results of an experimental study investigating the physical and mechanical properties of GFRP bars
14	exposed to severe environments and subjected to different levels of sustained load. The test environments
15	included moist concrete, alkaline solution and tap water, with temperatures varying from 20°C to 60°C. The
16	mechanical properties of the bars were characterized through direct tension, flexural and inter-laminar shear
17	tests, while the physical and chemical properties were determined through the implementation of a series of
18	complementary techniques, including moisture absorption measurements, scanning electron microscope
19	(SEM), Fourier Transform Infrared Spectroscopy (FTIR) and Energy dispersive x-ray analyses (EDX). The
20	test results showed that the elevated temperatures play a key role in triggering and accelerating the
21	development of critical degradation mechanisms. The reduction in the tensile strength of all conditioned
22	samples subjected to a sustained stress equivalent to 3000 $\mu\epsilon$ was always within the limits recommended in
23	existing codes for high durability bars, while a lower average strength retention was observed for higher
24	levels of sustained stress (equivalent to 5000 $\mu\epsilon$). Finally, it can be conclude that the long-term mechanical
25	properties of the tested GFRP bars appeared to be mainly affected by moisture diffusion through the resin
26	rich layer and debonding at the fibre/matrix interfaces due to the dissolution of the silane coupling agents.
27	
28	Keywords
29	Environmental degradation; Fibre/matrix bond; Glass fibres; Mechanical properties; Microstructures

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32 1 Introduction

33 Corrosion of steel reinforcement is one of the main causes of degradation of reinforced concrete (RC) 34 structures. In order to reduce the high maintenance cost associated with this problem, alternative reinforcing 35 materials, such as Fibre Reinforced Polymer (FRP) reinforcements, have been investigated over the past 36 three decades [1-4]. Despite the large amount of research that has been carried out on the use of FRP in 37 concrete, the lack of comprehensive data on their long-term in-service performance [3, 5] still hinders the 38 widespread use of FRP bars in civil engineering applications. 39 The durability of Glass FRP (GFRP) bars in concrete has usually been evaluated by using accelerated test 40 methods that expose the bars to environments harsher than those they may encounter in service and by 41 studying the change with time of their mechanical properties (i.e. tensile strength, bond strength and modulus 42 of elasticity) [2, 6-10]. In an attempt to achieve a more fundamental understanding of their long-term 43 behaviour, several researchers also investigated changes in physical, chemical and microstructural properties 44 using scanning electron microscopy (SEM) [4,11], energy dispersive x-ray analyses (EDX) [11,12], dynamic 45 mechanical analysis (DMA) [13], thermogravimetric analysis (TGA) [13,14], differential scanning 46 calorimetry (DSC) [10,14] and Fourier transform infrared spectroscopy (FTIR) [10,15]. 47 Among the conditioning environments examined in the literature, moist and alkaline environments have been 48 reported to be the most detrimental for the GFRP matrix and the fibres [6, 16-19]. In particular, in moist 49 environments the free hydroxide ions (OH) attack the polymer matrix and the long molecular chains are 50 disassembled by hydrolysis, promoting the ingress of more OH⁻ ions and water molecules (H₂O) (Eq. 1). 51 These, in turn, can break the polymer-polymer chain secondary bond, creating localized voids (increase in 52 free volume) and ultimately affecting the physical and mechanical properties of the GFRP matrix (i.e. 53 plasticization) [6]. The presence of water can also lead to the degradation of glass fibres through dissolution 54 (i.e. leaching) (Eq. 2). This process, which consists in the extraction of alkalis from the glass structures, continues until alkalis are available. The hydroxide produced by the leaching process (Eq. 2) increases the 55 56 pH of the solution and, as the pH exceeds the threshold value of 9, the Si networks are affected (i.e., silica 57 dissolution in alkaline environment, Eq. 3). [19]

$$R = C + OH - H_{2O} + OH - H_{1O} + OH - R + OH - R$$

O = R O - R O - H Eq. 1

58

- $Si \longrightarrow O + H_2O \longrightarrow Si \longrightarrow OH + OH^-$
- $60 \qquad Si O Si + OH^{-} \longrightarrow Si OH + Si O^{-} Eq. 3$

61 While direct submersion of bare bars in an alkaline and water solution at elevated temperature is

recommended by current standards to accelerate aging [20-22], previous studies [11, 18, 19] have shown that

Eq. 2

63 these environments are excessively harsh and cause premature deterioration of GFRP bars, consequently lead

64 to conservative predictions of service life. As a result, researchers attempted to evaluate the performance of

65 GFRP bars by embedding them directly into concrete and exposing the specimens to high temperatures to

66 accelerate reaction rates, yet subjecting them only to the inherent alkaline and moist environment provided

67 by the concrete surrounding the bars [9,18,23].

68 In addition to the exposure environment, the presence of sustained stress might generate stress concentrations

around the micro-imperfections present in the matrix (e.g. pores) and accelerate the propagation of micro

racks [24]. This, in turn, would ease the ingress of the surrounding media (e.g. alkali solution and water)

and have a possible detrimental impact on the durability of the fibres [25]. Only a limited number of studies

examined the influence of sustained stress on the long-term properties of FRP and concluded that low values

of sustained stress (20% of the ultimate strength or less) did not significantly affect the residual mechanical

properties. Such studies, however, only examined the performance of bare composite specimens [26], or

75 were limited to relatively low values of sustained stress (equivalent to a strain level of about 2000με) and did

- not include any chemical/physical analysis of the microstructure [27].
- 77

78 2 Research significance

This study aims to provide a better understanding of the long-term performance of GFRP bars by employing accelerated conditioning methodologies that more closely reflect the in-service conditions of GFRP RC members including alkaline and moist environments as well as sustained stresses. A comprehensive experimental programme was designed to study the degradation mechanisms at both micro (chemical

83 properties) and macro levels (physical and mechanical properties). Ultimately, the goal of this research is to

84 develop an effective model to assess and predict the durability of GFRP bars, which in turn, would directly

assist in the development of more reliable and less conservative design tools for GFRP RC members.

87 **3** Experimental programme

88 Fig. 1 presents a flow chart summarizing the complementary experimental program carried out in this 89 research. The flow chart is divided in three sections: Section 1 - Degradation; Section 2 - Investigation; and 90 Section 3 – Model. Section 1 shows the conditioning parameters (i.e. moist and alkali environments, high 91 temperature and sustained stress) considered in this research and the GFRP bar properties that are affected (at 92 both the macro level - tensile and bond properties - as well as at the micro level). Section 2 summarizes the 93 tests used to assess the investigated material properties. The use of this complementary set of tests was 94 implemented to enable a more comprehensive assessment of the degradation processes based not only on the 95 outcomes of the mechanical tests but also on the results of the physical and chemical tests. Section 3 is 96 presented in the chart as the logical convergence of this study, however a durability model still being 97 developed and therefore will not be discussed in this paper.

98

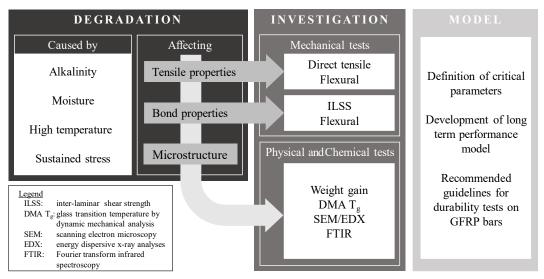




Fig. 1. Overview of experimental program.

101 3.1 Test Matrix

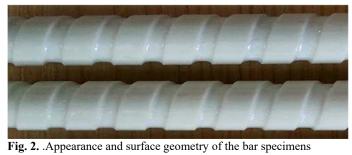
102 The test matrix (Table 1) consisted of 334 GFRP samples obtained from bars with a nominal diameter of

103 8mm. The GFRP bars used in this research work are commercially available and made of continuous E-

104 CR Glass fibres impregnated with a vinyl ester resin and a glass fibre content of approximately 88% by

105 weight. After the bars have gone through the pultrusion process, ribs are cut in the hardened bars and their

106 surface is subsequently coated (Fig. 2).



110	The program included tests on: unconditioned reference specimens (REF); unstressed and stressed bare bars
111	conditioned in alkaline solution (K and K-S3) or tap water (W and W-S3) at 20, 40 and 60°C; unstressed and
112	stressed bars embedded in concrete (i.e. moist and alkali environment) at ambient temperature (CON) and
113	immersed in tap water (M-CON, M-CON-S3 and M-CON-S5) at 20, 40 and 60°C. The exposure durations
114	taken into account varied from 1000 hr (~42 days) to 8760 hr (1 year).
115	Although it has been shown in previous research that direct exposure of GFRP to alkaline solution is
116	undoubtedly more aggressive than any realistic civil engineering application, such condition has been
117	included as it is recommended in current testing guidelines [21, 22]. The alkaline solution was prepared
118	according to the recommendations of the ACI 440 committee [22] using 118.5 g of Ca $(OH)_2$, 0.9 g of NaOH
119	and 4.2 g of KOH in 1 litre of deionised water. The pH level of this solution was periodically controlled and
120	kept at about 12.7, which is a representative value for a mature concrete pore solution [28].
121	A sustained stress inducing a tensile strain of 3000 $\mu\epsilon$ was applied to the FRP bars as recommended in ACI
122	440.3R and CAN/CSA-S806 [22,21], while a higher stress inducing a strain of 5000 $\mu\epsilon$ was examined to
123	assess the effect of less stringent serviceability limits (corresponding to larger allowable crack widths and
124	deflections) [29]. The sustained stress inducing the desired level of tensile strain was applied using two
125	different configurations: via a spring of adequate stiffness mounted in a stiff pre-tensioning rig (Fig. 2-a); or
126	by tying pairs of specimens at the two ends with a wedge interposed at mid-length to impose the required
127	curvature (Fig. 3-b). During the conditioning period, the strain level of the bars was measured periodically
128	using a caliper and demec gauge system. The length of the steel springs used in the loading system were also
129	checked regularly using a digital caliper with an accuracy of 0.02 mm to assure that the load was maintained
130	constant.
131	Table 1Test Matrix

	Test method								
Evenogunag	Temp	Time	MAb	TNS	ILSS	FLX	DMA	FTIR	SEM/EDX
Exposures	(°C)	(hr)	(N°)						
REF	20	0		5	5	5	3	5	5

Κ	20	1000, 2000, 6480,	3	9	3	3	1	3	2
	40	8760	3	9	3	3	1	3	2
	60		3	9	3	3	1	3	2
K-S3	20	1000, 2000, 6480,	4	9	3	3	N.A.	3	2
	40	8760	4	9	3	3		3	2
	60		4	9	3	3		3	2
W	20	1000, 2000, 6480,	3	N.A.	3	3	1	3	2
	40	8760	3		3	3	1	3	2
	60		3		3	3	1	3	2
W-S3	20	1000, 2000, 6480,	4	N.A.	3	3	N.A.	3	2
	40	8760	4		3	3		3	2
	60		4		3	3		3	2
CON	20	2000, 6480, 8760	N.A.	9	N.A.	N.A.	N.A.	N.A.	N.A.
M-CON	20	1000, 2000, 6480	N.A.	9	N.A.	N.A.	N.A.	N.A.	N.A.
	40			9					
	60			9					
M-CON-	20	1000, 2000, 6480	N.A.	9	N.A.	N.A.	N.A.	N.A.	N.A.
S3	40			9					
	60			9					
M-CON- S5	60	42, 90, 270	N.A.	9	N.A.	N.A.	N.A.	N.A.	N.A.

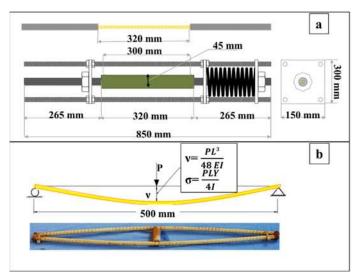
132 Note: TNS, tensile test; MAb, water absorption test; ILSS, inter-laminar shear strength; FLX, flexural tests; DMA, Tg by

dynamic mechanical analysis; FTIR, Fourier transform infrared spectroscopy; SEM, scanning electron microscopy; EDX,

134 energy dispersive x-ray analyses; REF, reference samples; K, alkaline solution; W, tap water; M, moisture; CON, concrete;

135 S3 and S5 sustained stress at 3000 με and 5000 με, respectively.

136



137

Fig. 3. Sketch illustrating the methods of application of sustained stress for (a) tensile and (b) moisture absorption test
 specimens.

140

141 **3.2 Test methodology**

142 Direct tensile tests (TNS) [22], inter-laminar shear strength tests (ILSS) [30] and flexural tests (FLX) [31]

- 143 were carried out to evaluate the mechanical properties. All tests were performed in displacement control
- 144 using hydraulic and mechanical actuators with a maximum capacity of 1000 kN (TNS) and 10 kN (ILSS and

- 145 FLX), respectively. Data were recorded at 1 Hz using an NI LabVIEW data acquisition system. Details of the
- setup of each test are shown in. **Fig. 4**.
- 147

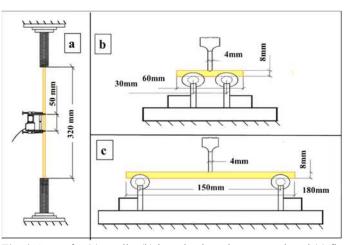


Fig. 4. Set up for (a) tensile, (b) inter-laminar shear strength and (c) flexural tests.

151 A series of tests on the physical and chemical properties of the samples was also carried out to investigate the 152 microstructure and the integrity of the GFRP material before and after exposure. Standard test methods were 153 implemented to measure the glass transition temperature (T_g) by dynamic mechanical analysis (DMA) [32] 154 (Fig. 5-a) as well as moisture absorption (MAb) [33]. DMA was performed using a Mettler Toledo 155 DMA/SDTA861e machine while moisture absorption was assessed monthly by measuring the weight gain due to fluid uptake into 300-mm-long (unstressed) and 450-mm-long (stressed) GFRP samples with the two 156 157 ends sealed using epoxy to prevent absorption from the cut edges. Prior to conditioning, the specimens were 158 dried in an oven at 55°C for 24 hr. The moisture uptake was monitored monthly using a digital scale with 159 accuracy of 0.001 g. For each set of measurements, the specimens were removed from their respective 160 conditioning environments at the same time, their surface was wiped dry with a cloth, and weighed 161 immediately. In addition, scanning electron microscopy (SEM), energy dispersive x-ray analyses (EDX) and Fourier transforms infrared spectroscopy analysis (FTIR) were carried out following procedures available in 162 163 the literature [12, 34] (Fig. 5-b).

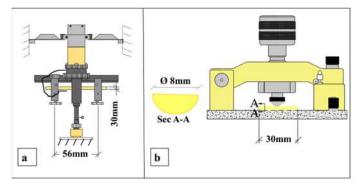


Fig. 5. Set up for (a) Tg and b) FTIR tests.

168	SEM and EDX were performed using a high magnification microscope (Phillips XL30 SEM) which uses a
169	focussed scanned electron beam to produce images of the sample. Prior to the analysis, 10-mm-long samples
170	were cut from the GFRP bars using a low speed diamond blade. The samples were subsequently placed
171	vertically in moulds, cold-mounted in epoxy resin and cured for 24 hr at room temperature. The cross
172	sections were then grinded using sand paper of increasing grit (400, 600, 800 and 1200) and polished with
173	$6 \ \mu m$ and $1 \ \mu m$ diamond pastes. The specimens were then carbon coated and analysed. Finally, FTIR
174	analyses were performed using an FTIR spectrometer to investigate the external surface and the internal
175	layers of the GFRP bars. In the former case, 30-mm-long samples were cut longitudinally from the bars up to
176	a maximum depth of 3 mm, while in the latter the samples were prepared by grinding the GFRP bars into a
177	fine powder (approximately 2 mg). The milled samples were pressed into a disc after being mixed with
178	approximately 200 mg of anhydrous Potassium Bromide (KBr). The disc analysis was carried out by treating
179	the KBr as the background reference. In both configurations, 32 scans were routinely obtained with an
180	optical retardation of 0.25 cm to yield a resolution of 4 cm ⁻¹ . Typical characteristic absorption bands
181	available in the literature are reported in Error! Reference source not found Past studies on FRP
182	examining civil engineering applications specifically focused on spectral zones in the band between 3600-
183	2900 cm ⁻¹ including the stretching mode of the hydroxyl group OH (3450 cm ⁻¹) and of the carbon-hydrogen
184	group CH (2928 cm ⁻¹) [12, 34]. In this research the acquisition of the spectra was performed ranging from
185	4000 to 600 cm ⁻¹ , so as to include the stretching modes of C-O and C=C (1295 cm ⁻¹ and 920 cm ⁻¹)
186	respectively).

Absorption bands	Assignment
(cm^{-1})	
3400	O-H bending vibration
3026	C-H stretch of the benzene
2960, 2928, 2871	CH, CH ₂ , CH ₃ stretch
1722	C=O- Stretching vibration in saturated aldehyde, ketone or act
1608, 1510	C=C- Stretching vibration of skeleton in benzene ring
1295	C-O stretch

1160	C-CO-C stretch	
920	C=C stretch	

189 4 Results and discussion

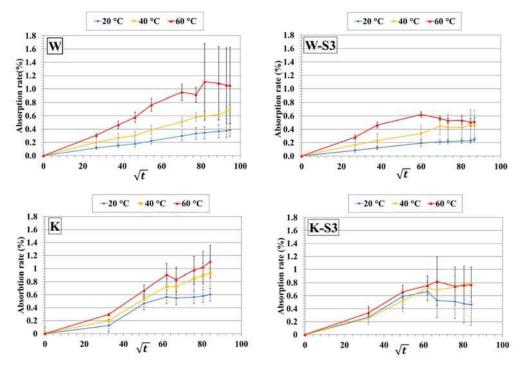
The results of the experimental programme are presented in this section. Firstly, the results from the moisture absorption tests, which can be used to investigate changes in the diffusion mechanism of water and alkaline solutions, are reported and used to support the results of physical and chemical tests, which assessed the deterioration of the micro structure (i.e., matrix, fibres and interface layer). Finally, the data acquired from all tests is utilized to discuss the changes in mechanical properties at the macro scale.

1954.1Moisture absorption properties

196 Fig. 6 shows the absorption rate of stressed (S3) and unstressed GFRP bars in water (W) and alkaline (K)

197 environment against the square root of the exposure time. The absorption rate was calculated as the ratio

198 between the average moisture uptake and the initial weight.



199

Fig. 6. Moisture absorption rate of unstressed and stressed (S3) GFRP bars immersed in water (W) and alkaline solution
 (K) at different temperature.

In all samples, weight increased gradually with respect to the exposure time, following the Fick's law up to about 4500 hr (~67 \sqrt{hr}). At this stage, the diffusion mechanism started to become unstable (non-Fickian behaviour) resulting in absorption rate curves characterized by portions of weight loss, due to the degradation of the outer layer leading to material loss, and by portions of weight gain, possibly due to the filling of new

207 or pre-existing voids and micro cracks. While no saturation was reached at the end of the considered ageing 208 period (8760 hr - \sim 93 \sqrt{hr}), stressed samples consistently showed a lower moisture uptake than unstressed 209 ones. This phenomenon could be attributed to the method used to stress the samples (i.e. bending, Fig. 3b) 210 which reduces the number of open pores in the portion of the bar in compression. This is discussed further in 211 section 4.2.2. In addition, the exposure temperature played a critical role on the absorption process. The 212 measured data showed that the higher the temperature the higher the absorption rate. This can be attributed to 213 an increase in the chemical reactions rate as well as to a higher void pressure resulting from an increase in 214 the volume of gases as temperature rises. This pressure build-up favoured the propagation of micro cracks 215 and consequently increased the free volumes within the bars that could be filled by the surrounding solution. 216 It can be finally observed that, for all temperatures, specimens exposed to alkali solution showed slightly

217 higher moisture uptake than those exposed to tap water as the former contains more free hydroxide ions (OH-

218), which work as a solvent for the polymer, thus promoting the solution ingress.

219 4.2 Chemical and physical degradation of GFRP bars in wet environments

220 4.2.1 Effects on Matrix

The effect of the different conditioning environments, temperatures and applied stress levels, on the GFRP matrix was studied at different depths of the samples. In particular, the external surface was analysed using both SEM images and FTIR. FTIR was also employed to examine an inner portion of the bars, yet still

sufficiently close to the surface, while the core of the samples was analysed using DMA.

225 FTIR was employed to detect possible chemical changes in the polymeric chain after conditioning. Intensity

226 changes in the hydroxyl group (OH) (Eq. 1) and in the stretching bands of C=O and C-O were investigated to

227 verify that hydrolysis and oxidation reactions took place. Since the intensity of a FTIR spectrum can be

affected by several parameters, it is customary [19] to quantify the changes in a given band as a ratio with the

229 variations of the aromatic group CH, which is very stable and does not easily break apart and react with other

substances. Thus, changes in the OH/CH (I3400/I3026) and C=O/CH (I1722/I3026) ratios as well as in the

231 C-O/CH (I1295/I3026) ratio can be used to gage hydrolysis and oxidation reactions, respectively. Similarly,

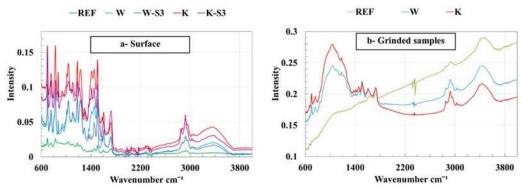
232 changes in the C=C peak, also referred to as curing index [35], were monitored to gain information on the

233 curing process.

234 The FTIR spectra were calculated on the surface of control (REF), unstressed (W and K) and stressed (W-S3

and K-S3) samples (Fig. 7-a) as well as on a deeper portion of control (REF) and unstressed (W and K)

samples (Fig. 7-b, grinded samples). All samples were conditioned for 8760 hr at 60°C.



237

Fig. 7. FTIR spectra for the resin matrix on (a) the surface of GFRP bars and on (b) grinded samples. Results for
 unstressed and stressed (S3) specimens conditioned in water (W) and alkali solution (K) at 60°C for 8760 hr are
 compared to benchmark samples (REF).

241

242 The results of the FTIR analysis on the surface of the samples (Table 3) showed that all the OH/CH ratios 243 increased with respect to REF and that such increase was more marked in unstressed samples than in stressed 244 ones. This directly confirms that moisture entered the polymer matrix promoting the hydrolysis reaction and 245 it also implies that stressed samples had lower chemical interaction (i.e., lower hydrolysis). As the intensity 246 of OH was similar for unstressed samples (only slightly lower for K), the C=O/CH (I1722/I3026) ratio was 247 used as an additional indicator of the hydrolysis reaction taking place and to assess the effect of the 248 conditioning environment. The reduction in C=O/CH (I1722/I3026) was due to the breakage of the end of 249 the vinyl ester chain and, as suggested by the lower value, the alkaline solution seemed to cause more 250 breakage of the ester chain than water (i.e. more hydrolysis). Conversely, while oxidation occurred in all 251 samples as all the C-O/CH (I1295/I3026) ratios increased with respect to REF, this reaction was more 252 evident in stressed samples than unstressed ones and in water rather than alkaline solution.

253 254

Table 3 Band ratios results for FTIR analysis on the outer surface of the samples aged at 60°C for 8760 hours. OH/C-H C=O/C-H C-O/C-H C=C/C-H C=C/C-H (13400/3026) (I1722/I3026) (I1295/I3026) (1920/13026) (I1608/I3026) (2)(4)(5)Samples (1)(3) (%) (%) (%) (%) Value Value Value Value Value (%) REF 1.05 2.07 1.79 1.49 3.45 _ -_ _ W 1.40 1.82 2.80 1.56 2.47 2.92 0.85 1.33 0.88 1.66 **W-S3** 2.73 1.16 1.10 2.10 1.01 2.941.64 1.83 3.32 0.96 Κ 1.98 1.32 1.26 1.52 0.73 2.16 1.21 1.33 2.89 0.84 K-S3 1.16 1.10 1.19 2.46 1.37 1.87 1.26 3.36 0.97 2.46

256 Error! Reference source not found. shows that, for all grinded deeper samples, there was a small reduction

257 of the OH/CH ratio and a negligible variation of the C=O/CH (I1722/I3026) ratio with respect to the REF

samples, suggesting that no hydrolysis reaction occurred in this portion of the bar. The test results also

²⁵⁵

showed an increase in the ratio of C-O/CH (I1295/I3026) ratio in both K and W samples due to the oxidation

260 of the methylene group (CH_2) at the end of the vinyl ester chain.

hours.										
	OH/O	C-H	C=O/C-H		C-O/0	C-H	C=C/	C-H	C=C/C-H	
	(I3400/	[3026]	(I1722/I3026)		(I1295/I3026)		(1920/13026)		(I1608/I3026)	
Samples	(1)	(2)	(3))	(4)	(5)
	Value	(%)	Value	(%)	Value	(%)	Value	(%)	Value	(%)
REF	1.17	-	0.57	-	0.70	-	0.79	-	0.62	-
W	1.13	0.97	0.57	1.00	1.00	1.43	0.98	1.24	1.08	1.74
К	1.13	0.97	0.57	1.00	1.19	1.70	1.11	1.41	1.41	2.27

Table 4 Band ratios results for FTIR analysis on a grinded portion of the inner layer of the samples aged at 60°C for 8760 hours.

263

For both surface and grinded specimens, an increase in the C=C/CH (I920/I3026) ratio with respect to REF

was observed (column 4 in Tables 3 and 4). This may be attributed to additional repeating units connecting to

the backbone chain and increasing the number of benzene rings, and consequently the stiffness of the resin.

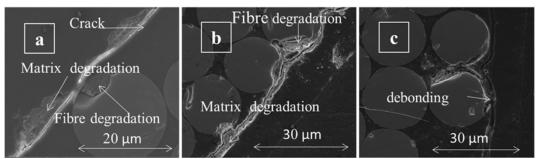
267 The ratio of C=C/CH at different locations of the polymer chain (i.e. 1608/I3026) (column 5 in Table 3,4)

showed a reduction in the outer layer and an increase in the inner part of the bars, thus providing evidence of

269 degradation on the surface and post curing of the core portion of the bars.

270 Additional evidence of matrix deterioration (i.e. cracks, voids and de-bonding from the fibres) was provided

- by the SEM images of the outer layers of GFRP bars conditioned in water (W) and alkaline solution (K) at
- 272 60°C for 8760 hr (**Fig. 8**).
- 273



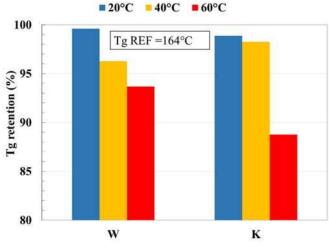
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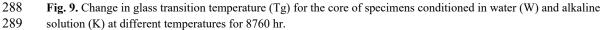
Fig. 8. SEM images of the cross-sections of specimen conditioned in (a) water and (b and c) alkaline solution at 60°C for

276 8760 hr.

- 278 The analysis of the innermost part of the GFRP bars was carried out by determining the glass transition
- 279 temperature (Tg) of samples obtained from the core of specimens aged for 8760 hr in water (W) and alkaline
- solution (K) at 20°C, 40°C and 60°C. After the specimens were removed from the conditioning environment,
- 281 they were stored in sealed plastic bags up to the time of testing. The values of T_g were determined according

- to the storage modulus (E') method. Fig 9 shows that the change in T_g was negligible for low temperatures,
- while a reduction of more than 5% was observed at 60° C for both environments. Since a reduction in T_g
- 284 indicates a reduction of transfer properties of resin matrix, it can be concluded that, at least at elevated
- temperatures, the matrix properties in the core of the specimens were also affected.
- 286





287

All the results confirm that in presence of moisture the higher the conditioning temperatures the higher the deterioration of the resin matrix and indicate that the polymer chain undergoes disruption (hydrolysis) and reconstruction (oxidation) of the bond links with possible repercussion on the physical and mechanical properties of the polymer matrix.

295 4.2.2 Effects on matrix/fibre interface

296 The interface between the matrix and the fibres is a very thin layer (approximately1 µm) of coupling agent,

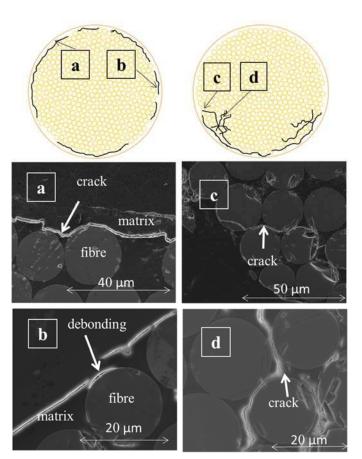
297 consisting of a saline-based material susceptible to moisture and alkaline environments. The assessment of

the deterioration of such a thin layer was only possible by using highly magnified SEM images.

Fig. 10 shows, at the top, the typical crack pattern observed in the cross-section of unstressed (left) and

- 300 stressed (right) samples, as well as the approximate locations where the SEM images (a, b, c and d) were
- 301 taken. As no significant deterioration was seen in samples conditioned at low temperatures, only images of
- 302 samples conditioned in water and alkaline solution at 60°C for 8760 hr are presented. These results confirm
- that moisture dissolved the saline substances at the interface causing local loss of bond between fibres and
- 304 matrix (i.e., cracks and debonding). In addition, it can be noticed that for the unstressed sample the cracks

- tend to propagate circumferentially around the outermost layer of fibres, while the cracks are concentrated on
- 306 the tension side of the stressed samples and propagate both radially and circumferentially as a result of the
- 307 three-point bending setup used to stress the specimens (Fig. 3-b).
- 308



309

Fig. 10. SEM images of the cross sections of unstressed specimen (left) and stressed specimens (right) conditioned in water at 60°C for 8760 hr.

313 4.2.3 Effects on glass fibres

- 314 Highly magnified SEM images were used to study the degradation of the glass fibres in water and alkaline
- 315 solution (Eq. 2 and Eq. 3). In addition, as silica (Si), aluminium (Al) and calcium (Ca) are not chemical
- 316 elements present in a vinylester matrix, a precise elemental analysis using EDX was carried out to detect the
- 317 presence of such elements as a result of the glass fibres dissolution.
- 318 The results of the EDX analysis carried out on the matrix and the fibres of a reference sample (Fig 11-a and
- 319 Fig 11-b, respectively) and on the matrix of samples conditioned in water and alkaline solution at 60°C for
- 320 8760 hr (Fig 11-c and Fig 11-d, respectively) are presented in the form of the spectra displaying the
- 321 concentration (peaks) of each chemical element identified in the analysis.
- 322

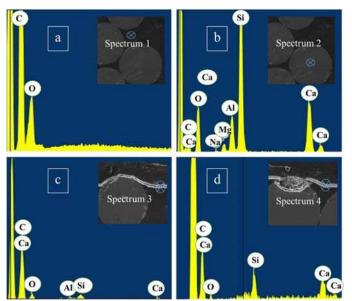


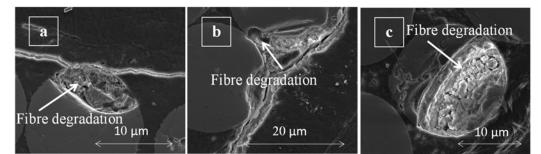
Fig. 11. Results of EDX scans for specimens conditioned (c) in water and (d) alkaline solution at 60°C for 8760 hr are compared to benchmark samples for (a) matrix and (b) fibres.

327 In all figures, the location where the analysis was performed is indicated by a target. The EDX analysis

328 detected the aforementioned chemical elements in the resin matrix near to the fibre/matrix interface both for

329 specimens conditioned in water (W, Fig 11-c) and alkaline solution (K, Fig 11-d). This confirmed that water

- 330 molecules and hydroxide ions, through micro-cracks in the matrix, reached the glass fibres deteriorating
- them. Additional evidence of the fibre deterioration within the outer layer was provided by SEM images as
- 332 shown in **Fig. 12**.
- 333



334

Fig. 12. SEM images showing the degradation in glass fibres after conditioning in alkaline solution at 60°C for 8760 hr

336

4.3 Mechanical degradation of GFRP bars in wet environments under sustained stress

338 4.3.1 Tensile properties

339 The results of the tensile tests are summarized in Table 5, including average (Avg) tensile strength and

- 340 modulus of elasticity, their coefficients of variation (COV) and their retention calculated as percentage of
- 341 results for REF. The tensile strength was calculated using a nominal diameter of 8 mm while the modulus of

342 elasticity was calculated as the slope of the stress-strain curve between 20% and 60% of the ultimate tensile

343 strength.

Cor	nditioning		Те	Tensile Strength				Modulus of Elasticity			
Evene	Temp	Time	Avg	COV	Retention	Avg	COV	Retention			
Exposures	(°C)	(hr)	(MPa)	(%)	(%)	(GPa)	(%)	(%)			
REF	20	0	1542	1.8	100	56.1	2.7	100			
K	20	1000	1472	0.3	95	56.7	0.7	101			
	20	2000	1454	0.2	94	53.6	0.6	96			
	20	6480	1498	n.a.*	97	58.5	n.a.*	104			
	40	1000	1407	1.9	91	57.5	4.0	102			
	40	2000	1390	1.3	90	57.0	4.5	102			
	40	6480	1310	n.a.*	84	60.4	n.a.*	108			
	60	1000	1310	3.1	84	59.2	6.2	106			
	60	2000	1139	11.6	73	58.0	1.7	103			
	60	6480	997	n.a.*	64	66.4	n.a.*	118			
K-S3	60	1000	1327	5.4	86	58.9	4.4	105			
	60	2000	1164	4.5	75	56.1	4.2	100			
CON	20	6480	1540	0.7	100	56.7	6.4	101			
M-CON	20	2000	1421	5.3	92	55.3	3.4	99			
	20	6480	1392	6.9	90	55.5	3.5	99			
	40	1000	1458	1.4	95	56.5	0.1	101			
	40	2000	1370	1.0	89	57.5	1.9	102			
	40	6480	1148	18.0	74	58.4	5.4	104			
	60	1000	1244	5.9	80	59.8	6.0	107			
	60	2000	1227	1.1	79	58.8	4.2	105			
	60	6480	917	1.4	59	55.4	1.9	99			
M-CON-S3	20	1000	1473	0.2	96	57.6	1.0	103			
	20	2000	1416	1.8	92	61.6	6.5	110			
	20	6480	1365	11.5	89	55.5	3.0	99			
	40	1000	1228	17.0	87	58.5	0.3	104			
	40	2000	1335	3.6	87	57.0	4.2	102			
	40	6480	1177	3.6	76	57.2	2.7	102			
	60	1000	1233	2.4	82	58.3	2.8	104			
	60	2000	1073	11.0	74	58.8	2.7	105			
	60	6480	919	4.2	59	61.1	3.4	109			
M-CON-S5	60	1000	1190	14.0	77	59.4	2.2	106			
	60	2000	1023	0.6	66	59.9	0.9	107			
	60	6480	971	4.8	63	58.3	5.0	104			

344 Table 5 Tensile test results

* COV not available since only one specimen was tested.

345

Fig. 13 shows the tensile strength and moisture absorption rate over the conditioning time for unstressed and stressed specimens exposed to moist concrete (M-CON, M-CON-S3, and M-CON-S5) and alkaline solution

- 348 (K and K-S3) at 20, 40 and 60°C. The results indicate that tensile strength decreased as moisture uptake
 349 increased and these effects were accelerated by increasing the temperature. It should be mentioned that
 350 moisture absorption and strength retention were measured on different specimens configurations (see Fig. 3)
- and the different stress distribution within the direct tension specimens might have affected moisture
- absorption, which was possibly underestimated as a result of the different induced cracking pattern.
- 353

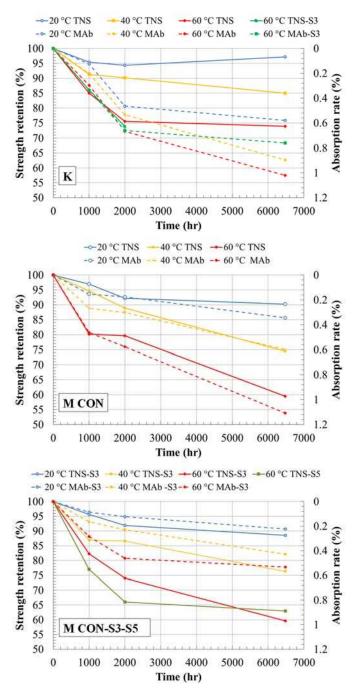




Fig. 13. Tensile strength (TNS) retention and moisture absorption (MAb) over time for unstressed and stressed (S3 and
 S5) bars conditioned in alkaline solution (K) and concrete (M-CON).

The reduction in tensile properties of GFRP bars can be interpreted in light of the results obtained by FTIR

359 and SEM/EDX. The deterioration of the matrix allowed the ingress of hydroxide ions with a two-fold effect:

360 1) causing de-bonding between fibres and matrix, consequently reducing the GFRP ability to transfer

361 effectively the load from the matrix to the fibres; 2) dissolving the glass fibres, thus resulting in a local

362 reduction of the number of fibres effectively contributing to carry the load.

363 In addition, even if the test results showed that the changes in the average tensile modulus of elasticity of

364 most conditioned specimens is negligible, as it varied within the variability of the experimental data, a

365 general increasing trend can be noticed. This beneficial effect on the stiffness was particularly evident in

366 specimens directly exposed to alkaline solution (K) for an extended period of time (6480 hr), showing that

the higher the temperature the higher the modulus of elasticity. This could be interpreted as post-curing of

the matrix, phenomenon that was also previously captured by FTIR analysis in the inner layers of the bar. In

addition, and as supported by the analysis of the SEM micrographs, only a small portion of the bar was

370 subjected to a significant level of deterioration after exposure to the conditioning environments. The majority

371 of the tensile strength reduction occurred within the first 2000 hr for all environments, with a slightly higher

372 degradation rate being observed for the specimens conditioned in alkaline solution (K). The unstressed

373 specimens embedded in concrete (M-CON), exhibited a slightly higher retention rate up to 2000 hr but their

374 strength at 6480 hr was overall lower than that of specimens immersed in alkaline solution. This can be

375 attributed to the formation of salt barriers on the surface of the K specimens that inhibited further diffusion

and damage to the microstructure. The application of a sustained stress during conditioning, equivalent to

377 $3000 \ \mu\epsilon$ and $5000 \ \mu\epsilon$, affected slightly only the tensile strength, possibly because the level of imposed strain

378 was small compared to the ultimate strain of the REF samples (i.e. $27000 \ \mu\epsilon$).

379 4.3.2 Flexural properties and inter-laminar shear strength

380 Error! Reference source not found. summarizes the results of flexural (flexural strength and modulus of

elasticity) and inter-laminar shear tests (ILSS) in terms of average values, coefficients of variation and

382 retention for unstressed and stressed bars directly exposed to water (W and W-S3) and alkaline solution (K

383 and K-S3) at 20, 40 and 60°C for 6480 hr.

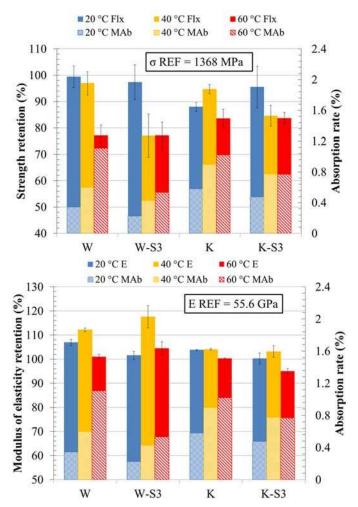
LAP.											
	(°C)	(MPa)	(%)	(%)	(GPa)	(%)	(%)	(MPa)	(%)	(%)	
REF	20	1368	7.1	100	55.6	1.5	100	73	2.5	100	_

W	20	1360	4.0	99	59.5	1.3	107	66	4.8	90
	40	1326	4.3	97	62.4	0.7	112	63	3.3	86
	60	1124	4.0	82	56.2	1.0	101	55	5.4	75
W-S3	20	1331	6.6	97	56.1	1.7	101	65	3.8	89
	40	1054	8.2	77	65.4	4.6	115	64	3.7	87
	60	1054	5.0	77	58.2	2.7	104	63	4.0	86
Κ	20	1203	1.6	88	57.8	0.2	103	70	1.6	95
	40	1296	3.5	94	57.9	0.3	103	68	3.5	93
	60	1143	4.0	83	55.8	0.2	100	54	4.0	74
K-S3	20	1306	7.9	95	55.8	2.2	100	66	6.2	90
	40	1157	3.9	84	57.4	2.3	103	64	1.7	87
	60	1145	2.2	83	52.92	2.2	95	61	1.3	85

The retention of mechanical properties is also shown in Fig. 14 and Fig. 15 along with the moisture

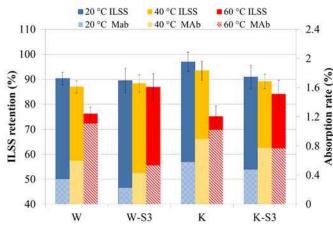
absorption (MAb) rate results to illustrate the effects of moisture ingress on flexural and inter-laminar

properties.



391

Fig. 14. Flexural strength (Flx) and modulus of elasticity (E) are compared to moisture absorption results (MAb) for different conditioning environments after 6480 hr.





393

Fig. 15. Inter laminar shear strength retention (ILSS) is compared to moisture absorption results (MAb) for different 396 conditioning environments after 6480 hr

398 It can be noticed that the average flexural strength of all conditioned specimens decreased compared to the 399 reference specimens. In particular, for exposures at 20 and 60°C, the higher the temperature, the lower the 400 flexural strength, which is in line with the findings previously discussed for TNS and MAb tests. Conversely, 401 at 40°C, the application of sustained stress appears to affect strength retention significantly, as the strength of 402 stressed samples was much lower than that of unstressed ones. However, this behaviour was only recorded 403 for flexural tests and the same trend was not observed in TNS and ILSS tests results. This can be attributed to 404 the different state of stress and failure mode generated within specimens subjected to different tests 405 configurations. Although flexural tests are relatively simple to perform, the different behaviour of composite 406 materials in tension and compression (with unidirectional FRP being typically characterised by a lower 407 compressive strength) can result in underestimating tensile strength and in an increased variability of results. 408 The variability in the results obtained from flexural tests can also increase for conditioned specimens, 409 especially when deterioration occurs at the fibre/matrix interface and can affect compression behaviour. As 410 stiffness is always determined within the perfectly elastic load-deformation range, the values determine for 411 flexural stiffness generally agree well with those obtained from direct tension tests. 412 The average values of the flexural modulus of elasticity increased marginally for most of the specimens, with 413 the exception of samples conditioned in water at 40°C for which the flexural stiffness increased more than 414 10%. The overall higher stiffness could have been caused by post-curing of the matrix during conditioning. 415 This is in line with the results obtained from the direct tension tests as well as the FTIR findings, albeit at 416 lower temperature.

417 The results from the ILSS tests show that the higher the conditioning temperature, the higher the strength

418 degradation, but also that stressed specimens experienced lower reduction in ILSS than unstressed ones.

419 Such findings are in good agreement with previous tests carried out in this study, suggesting that the

420 increased temperature favoured the ingress of moisture leading to a premature inter-laminar de-bonding and

that the state of stress induced using the test setup described in Fig 3b limited moisture ingress mitigating the

- 422 reduction in ILSS.
- 423

424 **5** Conclusions

The present study investigated the mechanical performance and physical and chemical characteristics of GFRP bars conditioned in different environments with and without sustained stress. The mechanical and micro structural properties were evaluated by direct tension test, flexural tests, inter-laminar shear tests, moisture absorption, SEM-EDX, FTIR, and DMA. On the basis of the discussion presented above, the

429 following conclusions can be drawn:

430 1) Moisture uptake of the tested GFRP bars occurred at a faster rate up to 4600 hr. The different rate of

431 diffusion can be attributed to the moisture absorption via capillaries and micro-cracks in the resin matrix.

432 This mechanism continues until moisture reaches the fibres, which provide a barrier and restrict moisture

flow. The tests results also confirmed that diffusion rate during the initial stages is highly dependent on the

434 temperature, and higher temperatures lead to higher absorption rates. However, the specimens exposed to

435 alkaline solution showed slightly higher moisture uptake than those exposed to tap water as the former

436 contains more free hydroxide ions (OH-), which work as a solvent for the polymer, thus promoting ingress of

the solution.

438 2) For temperature up to 40°C the degradation observed in specimens exposed to K and M-CON

439 environments were similar, whereas for higher temperatures and after 2000hr of exposure less degradation

440 was observed in specimens exposed to K environment. This can be attributed to deposition of salts on the

441 bars surface, which provided a barrier and restricted moisture diffusion into the bars, thus potentially leading

to underestimate strength retention.

443 3) Unexpectedly, the stressed specimens exhibited lower absorption rate than the unstressed counterparts.

444 This may be attributed to the method that was used to induce the desired strain (i.e. imposing the required

445 curvature via elastic bending) and the closing of micro cracks within the compression zone of the specimens.

446 Further tests on bars subjected to different stress distributions (e.g. uniform tension) should be carried out to447 examine this aspect in more details.

448 4) The tensile test results clearly show that the tensile strength of the tested GFRP bars was affected by the 449 conditioning environments. In general, higher temperatures and higher levels of sustained stress lead to 450 higher strength degradation. Similar trends were observed for all types of exposure with most of the tensile 451 strength reduction occurring within the first 2000 hr. This can be attributed to the high diffusion of water 452 molecules through the resin matrix outer layer inducing damage to the fibre/resin interface. After 2000 hr 453 exposure, bars embedded in concrete under sustained stress equivalent to 5000 µc conditioned in water at 454 60°C exhibited the higher reduction in strength due to the acceleration of moisture uptake caused by the high 455 temperature and as a result of the higher level of damage in the matrix caused by the sustained stress. 456 5) The reduction in the tensile strength of all conditioned samples subjected to a sustained stress equivalent 457 to 3000 µE was within the limits recommended in the Canadian code for high durability bars [21]. A lower 458 average strength retention was observed only for the specimens subjected to the higher sustained stress 459 equivalent to 5000 µE.

6) No significant change was observed in the elastic modulus of the tested GFRP bars regardless of the

461 conditioning environment or exposure period. Although a small degree of degradation was observed through

462 SEM and EDX in the outer resin matrix layer and some of the glass fibres closer to the surface of the bar, the

d63 observed level of damage was too small to result into an appreciable stiffness reduction. Some of the samples

exhibited a slight enhancement in stiffness, possibly due to the post curing of the resin matrix, as confirmedby FTIR results.

7) The change in flexural properties (e.g. strength, stiffness) was in line with what observed as a result ofdirect tension tests.

468 8) No significant fibre deterioration was observed through SEM/EDX within the cross-section of the bars,

469 with the majority of the damage being limited to a small percentage of glass fibres within the outer surface

470 layer. Microstructural changes in the matrix and debonding between fibre/matrix seem to be the main issues

471 affecting the long term mechanical properties of the tested GFRP bars.

472 9) The EDX scan detected additional elements such as silicon (Si), aluminium (Al), and calcium (Ca) in the

473 interface zone and in the resin matrix. This is attributable to the dissolution of the fibres and leaching of their

474 components into the resin matrix.

10) The FTIR analysis showed a significant increase in the amount of hydroxyl groups at the surface of the

476 material. This is an indication of chemical degradation in the resin matrix.

- 477 In summary, the long term mechanical properties of GFRP bars appear to be mainly affected by diffusion of
- 478 moisture through the resin rich layer and the debonding between the fibre/matrix interfaces due to the
- 479 dissolution of the silane coupling agents. Therefore, more developmental work could focus on these aspects
- to improve further the durability of GFRP bars for civil engineering application. It should be noted that the
- 481 above conclusions are based on the characterization of a specific bar with a given glass fibre/resin
- 482 combination and may not directly extend to other types of FRP rebars. Additional tests should be performed
- 483 on GFRP bars with different types of resin matrixes to provide statistically significant results and
- 484 conclusions.

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