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Substitution of PFAS chemistry in outdoor apparel and the impact on repellency performance

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5 Intensifying legislation and increased research on the toxicological and persistent nature of per-6 and polyfluoroalkyl substances (PFASs) have recently influenced the direction of liquid repellent 7 chemistry use; environmental, social, and sustainability responsibilities are at the crux. Without 8 PFAS chemistry, it is challenging to meet current textile industry liquid repellency requirements, 9 which is a highly desirable property, particularly in outdoor apparel where the technology helps to 10 provide the wearer with essential protection from adverse environmental conditions. Herein, 11 complexities between required functionality, legislation and sustainability within outdoor apparel 12 are discussed, and fundamental technical performance of commercially available long-chain (C8) 13 PFASs, shorter-chain (C6) PFASs, and non-fluorinated repellent chemistries finishes are evaluated 14 comparatively. Non-fluorinated finishes provided no oil repellency, and were clearly inferior in 15 this property to PFAS-finished fabrics that demonstrated good oil-resistance. However, water 16 repellency ratings were similar across the range of all finished fabrics tested, all demonstrating a 17 high level of resistance to wetting, and several non-fluorinated repellent fabrics provide similar 18 water repellency to long-chain (C8) PFAS or shorter-chain (C6) PFAS finished fabrics. The 19 primary repellency function required in outdoor apparel is water repellency, and we would propose

that the use of PFAS chemistry for such garments is over-engineering, providing oil repellency that is in excess of user requirements. Accordingly, significant environmental and toxicological benefits could be achieved by switching outdoor apparel to non-fluorinated finishes without a significant reduction in garment water-repellency performance. These conclusions are being supported by further research into the effect of laundering, abrasion and ageing of these fabrics.

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Keywords: Per- and polyfluoroalkyl substances; PFOA and PFOS; consumer products; outdoor
apparel; sustainability; SEM- EDX.

28

29 1. Introduction

30 A ubiquitous part of everyday life, liquid repellent finishes are used within a variety of sectors, both within consumer products and technical applications for oil and water resistance.¹⁻³ Use of 31 32 polymeric per- and polyfluoroalkyl substances (PFASs) has vastly grown since first use in the 1950s due to the high level of repellent functionality they provide;⁴ PFASs are used in many 33 applications, such as food-packaging, fire-fighting foams, cookware, electronics, medical products 34 and within consumer textiles.^{3,5-9} Repellent properties are essential for protection against harmful 35 36 liquids, for example within medical textiles and protective clothing in the oil and gas industry, and 37 are vital for health, safety and comfort of outdoor enthusiasts in inclement weather or extreme environmental conditions.¹⁰ The woven fabric used for outdoor repellent apparel, for both extreme 38 39 environments and casual leisure activities, is coated with an aqueous emulsion based on PFASs or side-chain fluorinated polymers that imparts a durable water and stain repellent finish.⁶ PFASs 40

41 comprise a polymeric backbone with branching fluorinated side-chains, where on one more carbon atoms and all hydrogens have been replaced by a perfluoroalkyl moiety $(C_n F_{2n+1})$.¹¹ There are two 42 43 main manufacturing processes to produce PFASs. In electrochemical fluorination (ECF) the 44 organic substance is reacted with anhydrous hydrogen fluoride (HF) by electrolysis, substituting 45 all hydrogen atoms with fluorine and creating a mix of linear and branched perfluorinated isomers and homologues.^{6,11-13} In telomerisation, perfluoroethylene (CF₂=CF₂) and perfluoroethyl iodide 46 47 (C_2F_5I) are reacted together to produce perfluorinated iodides with various chain lengths; these intermediates are used in subsequent reactions to synthesize fluorotelomer-based products, which 48 find use in food packaging, surfactants and textile treatments.^{6,11,12} In all compounds the degree of 49 50 fluorination, backbone chain length, and the packing of the side chains affects the characteristics of the compound.^{6,11,12,14} The high level of hydrophobicity and oleophobicity provided by PFASs 51 52 is due to the low surface energy provided by the orientation and packing of the terminal –CF₃ end groups within the side-chains.^{14,16} The wetting potential is dependent on the adhesive interaction 53 54 between the liquid surface, solid surface and the air interface. The structure of the fluorinated 55 polymer side-chain, with clustered hydrophobic groups, reduces the surface energy of the fabric; a –CF₃ surface construction has a surface tension of 6 dyn cm⁻¹ at 20 °C, which repels liquids with 56 a greater surface tension, including polar liquids (*e.g.* water with surface tension of 73 dvn cm⁻¹ at 57 20 °C) and non-polar liquids (e.g. octane with surface tension 22 dyn cm⁻¹ at 20 °C).^{6,17} 58

However, PFASs have been ubiquitously identified within wildlife, humans and found across the world in the environment;^{18,19} they are criticized as being toxic, carcinogenic and persistent within the environment.²⁰⁻²² Of high regulatory interest are long-chain PFASs: perfluoroalkyl carboxylic acids (PFCAs) with seven or more fluorinated carbons ($C_nF_{2n+1}COOH$; $n \ge 7$), for example perfluorooctanoic acid (PFOA; 1); and perfluoroalkane sulfonic acids (PFSAs) with six

64 or more fluorinated carbons ($C_nF_{2n+1}SO_3H$; $n \ge 6$), for example perfluorooctanesulfonic acid (PFOS; 2).^{11,15,19} A wealth of literature exists on the ubiquitous and bio-accumulative nature of 65 PFOA and PFOS and associated increased mortality rates, cancers, and toxic effects on liver and 66 immune systems.^{5,12,21,23-27} Bio-accumulation and bio-concentration of PFASs within humans and 67 the food chain are of primary concern,¹⁹ which increases with increasing fluorinated carbon chain 68 69 length; long-chain PFSAs and PFCAs have a higher bio-accumulation potential than their shorterchain analogues.^{11,19,28,29} PFASs have been acknowledged to have a greater bio-accumulative 70 71 nature than PFCAs of the same carbon chain length, which is thought to be due to the ability of PFASs to bind more strongly to serum proteins.^{29,30,31} 72



73

74 There is a continuing challenge to find an alternative chemistry and/or physical modifications to 75 provide equivalent liquid repellent functionality to that given by PFAS chemistry. Substitution to 76 'short-chain' PFAS chemistry has taken place with shorter fully fluorinated chain lengths as C6 or 77 C4 analogues. However, there is increasing concern on the persistent and bio-accumulative potential of these short-chain analogues, which have the capability to degrade to short-chain 78 perfluoroalkyl carboxylic acids (PFCAs) or perfluoroalkane sulfonic acids (PFSAs).^{32,33} An 79 80 increasing exposure trend to perfluorohexane sulfonate has been observed, and this compound 81 potentially has a similar or longer serum half-life, within mammals that have been tested, to PFOS.³³ Alternative non-fluorinated chemistries include hydrocarbons, silicones, and dendritic 82 83 structures, and product developers are increasingly cinched between fulfilment of technical 84 performance for the product, legislative requirements, and social and environmental responsibility.

85 Recent statistics show that nearly 9 million people in England are active outdoors, with over 250,000 people either climbing or hill-walking at least once a month.³⁴ An increase in participation 86 87 and a diversity in the types of activities being undertaken, in terms of terrain, environment and 88 physical activity level, bring an increased and more varied demand on performance clothing 89 functionality; the wearer expects clothing to function and maintain comfort regardless of the climatic conditions encountered.³⁵ Durable water repellent (DWR) clothing is of high importance 90 91 for safety and wearer wellbeing in mountainous, often remote, environments during strenuous 92 activity, such as hiking, climbing or mountaineering, and in adverse weather conditions. Rainwear 93 should provide protection, keeping the wearer dry whilst allowing thermoregulation of the body.^{10,35,36} Wetting of the garment's outer fabric face, due to decreased repellency, saturates the 94 95 fabric rapidly, reducing evaporative cooling of perspiration and heat transfer away from the wearer's body;³⁷⁻³⁹ this results in a feeling of wearer discomfort, possible wetting of other clothing 96 layers, and accelerated cooling of the wearer, 39-41 consequently, the wearer's physiological 97 98 responses can be affected, potentially resulting in an issue of health and safety.

99 Multiple factors post-purchase affect the liquid repellent functionality of the garment such as 100 laundering durability, abrasion resistance (rocky terrain for outdoor consumers), and consumer care;⁴² only fluorinated repellent finishes have been used ubiquitously throughout consumer 101 102 repellent apparel achieving a high level of repellency and effective performance. For consumer 103 outerwear in less adverse conditions, a lower level of functionality may be appropriate, where a 104 high level of technical protection is not a key requirement. However, PFAS chemistry has been 105 widely used to fulfil this wide range of requirements and used in abundance due to its capability 106 to be applied to a range of fibre types and fabrics.

107 Concern on the use of long-chain PFAS chemistry started in the 1960s, notably with the 108 detection of organic fluorine within human serum by Taves in 1968.^{43,44} The substitution process 109 away from long-chain PFSAs and PFCAs began in 2000 when the first reports of the ubiquitous occurrence of PFOS within wildlife were published.^{11,33} These concerns led to the phase-out of 110 111 PFOS and related compounds by 3M, whose key components within the manufacture of their 112 Scotchgard stain products produced perfluorooctanesulfonamide derivatives by ECF with PFOS a 113 resulting products from the intermediate perfluorooctanesulfonyl fluoride (POSF) used in secondary synthesis. ^{11,33,45,46} Industry initiatives moved to shorter-chain analogues of side-chain 114 fluorinated polymers. with non-fluorinated alternatives progressively also being sought.^{15,17,33} 115 116 European legislation and NGO campaigning has driven the move away from long-chain PFSAs 117 and PFCAs; in 2006 the EU imposed a restriction on the use of PFOS to protect health and the environment;⁴⁷ in 2009 PFOS was classified as restricted on The Stockholm Convention's list of 118 119 Persistent Organic Pollutants (POPs); and in 2015 the European Chemicals Agency (ECHA) adopted a proposal to limit the marketing and use of PFOA European-wide.⁴⁸⁻⁵⁰ The 2013 120 121 Helsingør statement² raised concerns on the impact of PFASs on health, the environment, and degradation and exposure of fluorinated alternatives, while the Madrid statement^{9,51} raised similar 122 123 concerns on the production and release of PFASs, calling for a limit to its use, and requesting a 124 collaborative effort to develop non-fluorinated alternatives.

Since 2011, Greenpeace have concentrated their campaigning on the use of "toxic chemicals" on the apparel industry. In 2015, Greenpeace launched their 'Detox Outdoor' campaign with specific emphasis on use of PFAS chemistry within outdoor apparel; the 'Footprints in the Snow'⁵² study assessed snow and water samples from eight remote locations around the world; the 'Leaving Traces' report⁵³ utilized social media asking consumers to nominate certain products and brands to be analysed for long-chain PFAS content; and the latest report 'Hidden in Plain Sight'⁵⁴ tested air samples from outdoor apparel stores for evidence of PFAS degradation. This increased publicity specifically highlighting the outdoor apparel industry's chemical use has led to many manufacturers and brands seeking a move away from PFAS repellent chemistry.

134 Despite PFASs being used in a variety of aspects of daily life, the outdoor apparel industry have 135 explicitly been the primary target of this NGO activist attention, yet only a few research studies 136 on the use of PFASs in outdoor apparel have been published, and these have solely focused on exposure pathways and degradation routes;^{3,32,55-57} with many being non-peer-reviewed.^{53,54,58-60} 137 138 Whilst knowledge on degradation routes, exposure trends and analytical techniques remains 139 central to research on PFASs, there is sparse comparative literature on the repellent functionality 140 of PFAS chemistry and alternative, non-fluorinated chemistry, in outdoor apparel; one non-peer-141 reviewed study exists.⁶¹ This functionality is highly important to the end-use of the fabric and the 142 wearer.

143 The purpose of this work is to communicate the variation in functionality between long-chain 144 (C8) PFAS repellent chemistry, shorter-chain (C6) PFAS repellent chemistry, and non-fluorinated 145 repellent chemistry within outdoor apparel fabrics. The work aims to determine the necessary 146 chemistry of the finish in a DWR treatment by illustrating the user requirements of repellent 147 outdoor apparel and comparing repellent performance of finishes. Criticism has focused on the 148 outdoor apparel industry highlighting repellent performance clothing as a potential route for 149 exposure to PFASs. Considering the complex nature of balancing legislation, sustainability, and 150 functionality, this paper aims to report a novel comparison of currently commercially available 151 repellent fabrics for outdoor apparel and an assessment on their repellent functionality both for 152 water and oil resistance.

154 **2. Materials and Methods**

155 2.1. Consumer survey

156 To illustrate consumer use of repellent apparel and their requirements, a consumer survey was 157 designed and launched through Bristol surveys, in affiliation with The University of Leeds. The 158 questions included demographic descriptors, indicators of the respondent's participation in 159 outdoor activities, inquiry on the preferences in purchasing decisions and user requirements of 160 personal apparel during activity. Respondents gained access to the survey through a URL address. 161 Completion was voluntary and respondents could withdraw at any time. It was believed that the 162 group of consumers targeted would have some knowledge of the criticism through brand 163 marketing, retailers or NGO literature. The survey was promoted within outdoor recreation 164 Internet forums and featured on an outdoor magazine's online website. The survey ran for 15 months from 15th May 2015 to 19th August 2016 and received a total of 575 responses. 165

166

167 *2.2. Materials*

168 Woven fabric samples were kindly supplied for the study by various manufacturers and brands: 169 according to manufacturer details, three of these fabrics were stated to be finished with long-chain 170 (C8) PFAS repellent chemistry, nine fabrics were stated to be finished with shorter-chain (C6) 171 PFAS repellent chemistry, nine fabrics were stated to be finished with non-fluorinated chemistry, 172 and one fabric was untreated. The non-fluorinated chemistries were, at the time of the study, 173 relatively new to the market, supplier information stated that samples P to U were hydrocarbon 174 hyper-branched polymers (dendrimers) with a polyurethane backbone, sample N was a fat-175 modified resin, and sample V was a hyper-branched polymer. All fabrics were commercially in

176 use at the time of the study, intended for use in repellent outerwear, with the majority either 100% 177 polyester (PET) or polyamide (PA) fibre content; some samples contained a laminate or membrane 178 (see Table S.1). The fabrics display a range of commercially used fibre and fabric types, within 179 outdoor apparel; all were synthetic monofilaments and the majority plain weave (only samples B180 and Q differ being twill weaves). The sample size stated within standard test methods to be used 181 throughout the experimental work were compared; the specimen size needed to be cross-functional 182 was calculated as 165 mm x 165 mm.

183

184 2.3. Energy-dispersive x-ray spectroscopy (EDS/EDX)

SEM-EDX (Jeol JSM 6610LV coupled to Oxford Instrument INCA X-Max 80 EDS system) was used to indicate elemental composition of the fabric sample surface, and therefore define the repellent finish type. Semi-quantitative elemental analysis (magnification x50, accelerating voltage 20 kV, spot size 50, working distance ~10 nm, and aperture 2) determined the elemental content of each fabric sample by weight percentage. Two specimens of each fabric sample, from different areas of the fabric, were analysed using ~1 cm² specimens.

191

192 2.4. Water repellency

AATCC 22-2014⁶² (similarly BS EN ISO 4920:2012⁶³) is a widely used test method to determine the resistance of a fabric to surface wetting by water. The procedure set out in the standard was followed using three different specimens, cut from separate places of the fabric sample, with 5 repeat tests. Each specimen was assessed according to the AATCC rating scale; intermediate ratings can be used for evaluation above water repellency grade of 50. Evaluation was carried out according to the water repellency grades as shown in Table 1 with inclusion of intermediate rating 95. According to AATCC 22-2014, a rating of '100' should be given where there is no sticking or
wetting of the specimen, however, in preliminary testing it was observed that there is always some
sticking to the fabric surface and therefore determined that a rating of 100 was unfeasible; a rating
of 97.5 was given, as a substitute, when few small sparse droplets were seen.

203

Table 1. Spray test water repellency grades, according to AATCC 22-2014 (BS EN ISO 4920).^{62,63}

Repellency grade	Description
97.5	Sparse small droplets visible on the specimen surface.
95	Few random sticking of water droplets clinging to the surface fibrils.
90	Slight random sticking or wetting of the specimen face
80	Wetting of specimen face at spray points
70	Partial wetting of the specimen face beyond the spray points
50	Compete wetting of the entire specimen face beyond the spray points
0	Complete wetting of the entire face of the specimen

205

206 The mode value of the repeat spray tests, for each fabric sample, was calculated. In addition, the 207 amount of water that adhered to each fabric sample, either by sticking to the surface or by 208 absorption by capillary action, was calculated as % change in comparing mass before and after 209 testing; this method has previously been used to discriminate between similar rated fabrics.⁶¹ The 210 mass of each dry and conditioned fabric sample was measured using a Precisa 310C-3010D 211 balance, and the mass of the sample following testing to two decimal figures. % Water adherence 212 (A) was calculated according to equation 1, where mi and mt are the mass of the sample before 213 and after testing, respectively. Average percentage mass increase was calculated for each fabric 214 sample.

$$215 \qquad A = \frac{m_i - m_t}{m_i} 100$$

217 2.5. Aqueous and oil repellency

BS ISO 23232:2009⁶⁴ determines aqueous liquid repellency using eight grades of water and 218 isopropyl alcohol solutions with surface tension values between 24.0-59.0 dyn cm⁻¹. BS EN ISO 219 14419:2010⁶⁵ determines oil repellency using eight test solutions of hydrocarbons with surface 220 tension values between 19.8-31.5 dyn cm⁻¹. These tests provide a wider range for greater 221 222 discrimination between similarly performing samples. Wetting was evaluated and assigned a grade 223 number 0-8, and assessed as a 'fail', 'pass' or 'borderline pass', where the grade was expressed to 224 the nearest 0.5 value. The grade number in agreement from two specimens was recorded, with a 225 third specimen tested where necessary.

226

227 **3. Results and Discussion**

228 *3.1. Consumer survey*

Of all the respondents, 526 were living within the UK (91.5%) with 35 other respondents from Europe and 14 from other countries worldwide. All age groups were represented. On a monthly basis, 83 respondents participated in outdoor recreation daily (14.4%), 76 participated 21-30 times per month (13.2%), 244 participated 6-20 times per month (42.4%) and 170 respondents participated 5 or fewer times per month (29.5%). Two respondents did not participate in outdoor recreation at all (0.3%). The main outdoor activity undertaken by respondents was hiking, trekking, mountaineering and hill-walking.

(1)

236 384 respondents (67%) said they participated in outdoor recreational activities in all weathers, 237 including rain and snow, with 268 respondents being outdoors in the rain more than 20 times per 238 year; advocating the need for a high, sustained level of water repellency on their apparel. 239 Respondents ranked purchasing factors by importance (Figure 1). None of the factors were ranked 240 as 'unimportant' but respondents, overall, stated water repellency, breathability, fit, durability and 241 wind resistance to be very important. Overall, the majority of respondents (82%) considered water 242 repellency to be the most important factor, compared to the majority of respondents being 243 indifferent to stain resistance (48%) and dirt and oil repellency (42%). Respondents were more 244 concerned with performance factors than appearance. This highlights the primary consumer 245 demand of a water repellent garment: protection from the rain and inclement conditions.





Figure 1. Respondents purchasing factors ranked by importance. No factors were ranked as
unimportant. Water repellency was the main requirement for survey participants, ranked as 'very
important' by 82% of respondents.

251	Respondents selected important factors they considered to be important in the 'environmentally
252	friendly production' of a repellent garment. The main priorities of consumer environmental
253	considerations in production were 'functionality to not be lessened' (310 respondents), product to
254	be 'ethically sourced' (255 respondents), a 'repairable product' (252 respondents) and 'non-toxic
255	chemicals' (242 respondents). Whilst this, again, highlights the importance of performance for the
256	consumer, it does suggest that social and environmental impact are of concern to the consumer.
257	

Table 2. Energy-dispersive x-ray spectroscopy analysis of all fabric samples; elemental content
shown in weight percentage (%) of elements carbon (C), oxygen (O), fluorine (F), titanium (Ti),
sulphur (S), silicone (Si), and chlorine (Cl).

Alleged repellent finish type	Sample label	C (%)	0 (%)	F (%)	Ti (%)	S (%)	Si (%)	Cl (%)
Long-chain (C8) PFAS	Α	74.2±0.2	22.6±0.5	2.05±0.30	0.78±0.10	0.19±0.00	n.d.	0.23±0.01
	В	73.4±0.1	23.7±0.4	2.11±0.36	0.58±0.04	0.18±0.06	n.d.	n.d.
	С	59.9±0.1	38.9±0.1	n.d.*	1.21±0.01	n.d.	n.d.	n.d.
Shorter-chain (C6) PFAS	D	73.8±0.9	22.4±0.8	2.77±0.29	0.84±0.21	0.22±0.05	n.d.	n.d.
	Ε	75.0±0.2	22.7±0.1	1.38±0.04	0.35±0.05	0.16±0.00	0.12±0.00	0.34±0.05
	F	74.7±1.8	22.6±2.6	2.01±0.64	n.d.	0.22±0.03	0.28±0.08	0.19±0.05
	G	76.2±0.6	21.0±0.3	1.62±0.25	0.85±0.12	0.20±0.01	0.10±0.00	0.13±0.02
	Н	77.8±0.4	21.1±0.1	0.74±0.00	0.20±0.01	0.17±0.01	0.14±0.00	0.23±0.02
	J	72.7±0.2	19.7±0.2	5.53±0.51	0.83±0.06	0.52±0.26	0.28±0.02	0.48±0.01
	K	75.0±0.1	24.1±0.1	n.d.	0.57±0.03	0.17±0.01	n.d.	0.19±0.02
	L	74.3±0.2	23.7±0.2	1.43±0.12	0.60±0.14	n.d.	n.d.	n.d.
	М	61.0±0.6	37.8±0.7	n.d.	1.16±0.01	n.d.	n.d.	n.d.
Non-F (fat-modified resin)	N	73.4±0.6	25.2±0.5	n.d.	1.48±0.05	n.d.	n.d.	n.d.

Non-F (specifics unknown)	0	72.6±0.4	26.8±0.4	n.d.	0.59±0.03	n.d.	n.d.	n.d.
Non-F (dendrimers with PU backbone)	Р	72.8±0.2	25.9±0.1	n.d.	1.30±0.12	n.d.	n.d.	n.d.
	Q	72.4±1.1	25.6±1.0	n.d.	2.01±0.08	n.d.	n.d.	n.d.
	R	65.8±0.4	33.4±0.3	n.d.	0.73±0.08	n.d.	n.d.	n.d.
	S	60.2±0.4	39.4±0.5	n.d.	0.44±0.11	n.d.	n.d.	n.d.
	Т	72.8±0.0	24.5±0.7	n.d.	2.73±0.80	n.d.	n.d.	n.d.
	U	72.1±0.3	26.4±0.4	n.d.	1.47±0.06	n.d.	n.d.	n.d.
Non-F (dendrimers)	V	58.5±0.5	39.5±0.6	n.d.	1.99±0.03	n.d.	n.d.	n.d.
Untreated	Z	75.24±0 .3	24.5±0.3	n.d.	0.12±0.00	0.11±0.01	n.d.	n.d.

261 Detection limit to parts per thousand (1×10^{-3}) .

262 Table 2 shows the semi-quantitative elemental composition of the repellent finish by EDX. 263 Twelve samples were allegedly finished with either long-chain (C8) PFAS or shorter-chain (C6) 264 PFAS repellent chemistry, however, no F content was detected on samples C, K, or M. On a few 265 samples F was seen as an emerging peak, but was below the levels of detection from the baseline 266 by the software. C, O, and Ti (originating from TiO₂ used for fabric whitening) were detected on 267 all 'non-fluorinated' repellent finished fabric samples suggesting a hydrocarbon-based surface 268 chemistry; no F nor Si was detected on any 'non-fluorinated' repellent finished fabrics. While 269 EDX is a surface analysis technique it is thought that several elements of the fabric bulk were 270 detected; Si detected in sample J is thought to be the laminate backing and in sample F it is thought 271 to be the polymeric coating.

All fabric samples showed a good level of resistance to surface wetting, assigned a spray rating of 90 or above (Figure 2). Untreated fabric (Sample *Z*) was completely wet by the water spray with movement of water by capillary action through the fibres (known as 'wicking') within the fabric structure and penetration of water through the fabric; Sample *Z* was assigned a spray rating of 0. Generally, long-chain (C8) and shorter-chain (C6) PFAS repellent fabric samples were rated either 95 or 97.5, with the exception of two shorter-chain (C6) examples (samples *G* and *H*), which were rated 90; for sample *H*, this may be due to the low fluorine content. In comparison, nonfluorinated repellent fabric samples were generally rated at 90, although two examples (samples *O* and *V*) were rated 95. Directly comparing samples *C*, *M* and *V*, which have the same fibre and fabric type, the long-chain (C8) repellent sample was rated at 97.5, while the shorter-chain (C6) repellent sample and the non-fluorinated repellent sample had an average spray rating of 95.



284

Figure 2. Spray rating of repellent outerwear fabric samples, measured according to AATCC 22-

286 2014 (BS EN ISO 4920),^{62,63} categorized by repellent chemistry type.



Figure 3. Water adherence measurements of the repellent outerwear fabric samples after spray
test. Samples are categorized by repellent chemistry type. Error bars show standard deviation of
5 repetitions.

291 In terms of water adherence, long-chain (C8) repellent samples all had less than 2% mass 292 increase post-testing with a low standard deviation (0.23-0.72%), suggesting a uniform, highly 293 water-repellent finish (Figure 3). shorter-chain (C6) repellent samples with a spray rating at 97.5 294 showed low % mass increase, whilst the two examples rated at 90 had a significantly higher % 295 mass increase (sample G 27.5% and sample H 30.9%) and greater standard deviation between 296 measurements, suggesting a non-homogenous fabric finish. There was also variation in water 297 adherence for non-fluorinated repellent samples; those assigned a spray rating of 95 had a low 298 percentage mass increase; of those assigned a spray rating of 90, sample N and sample P had a 299 relatively high % mass increase (26.1% and 25.4%, respectively) with significant standard 300 deviation between measurements.

Long-chain (C8) repellent fabric sample *A* showed the greatest level of repellency to aqueous staining, testing standard BS ISO 23232⁶⁴, (Figure 4), with a rating of 6.5 out of 8; shorter-chain (C6) repellent fabric samples varied from 2.5-5.0, and non-fluorinated repellent samples varied from 2.5-4.0, which was expected as the efficacy of repellency to liquids of surface tensions different to water decreased with reduction in fluorocarbon chain length (or presence of fluorine). The untreated fabric sample *Z* showed no resistance to aqueous staining.

307 All non-fluorinated repellent fabric samples demonstrated no resistance to oil-based 308 (hydrocarbon) liquids (Figure 5), which was expected, demonstrating the key differences between 309 repellent functionality provided by PFAS chemistry and repellency provided by non-fluorinated 310 alternative chemistries. The greatest level of repellency to hydrocarbon oil liquids was 311 demonstrated by long-chain (C8) repellent fabric sample A; variation between shorter-chain (C6) 312 repellent fabric samples was seen, with ratings ranging from 1.0-2.5, and shorter-chain (C6) 313 repellent fabric sample H demonstrated no repellency to hydrocarbon liquids, which may be 314 associated with the absence of fluorine in elemental detection. As expected, untreated fabric Z315 demonstrated no resistance to oil-based (hydrocarbon) liquids.



Figure 4. Repellency to aqueous liquids of the repellent outerwear fabric samples, according to
 BS ISO 23232:2009.⁶⁴ Higher grades signify a greater level of repellency. The samples are
 categorized by repellent chemistry type.





Figure 5. Repellency to hydrocarbons (oil repellency) of the repellent outerwear fabric samples, according to BS EN ISO 14419:2010.⁶⁵ Higher grades signify a greater level of repellency. The samples are categorized by repellent chemistry type.

327 4. Conclusions

This is the first study to report functionality specifically for repellent outerwear used by the outdoor apparel industry and a direct comparison of commercially available long-chain (C8) PFAS, shorter-chain (C6) PFAS and non-fluorinated repellent finishes. It was demonstrated that a DWR finishing treatment is required to provide a level of water repellency to woven apparel fabrics, exemplified by the untreated fabric showing no resistance to surface wetting by water, with associated high water adherence and absorption. EDX was employed as a semi-quantitative method to assess the type of repellent finish; analysis detected fluorine content in several of the

335 long-chain (C8) and shorter-chain (C6) PFAS repellent fabric samples, potentially showing 336 presence of PFASs. EDX analysis has vividly shown difference in elemental content between 337 fluorinated and non-fluorinated repellent fabric samples, and demonstrated that all non-fluorinated 338 repellent samples to be based on hydrocarbon chemistry. Information supplied with seven of the 339 non-fluorinated finishes stated a hyper-branched hydrocarbon polymer surface chemistry, which 340 is typical of dendrimer technology, wherein multiple hyper-branched (tree-like) alkyl end-groups 341 provide the function of aqueous repellency, but have a lower repellence to oil staining 342 (hydrocarbon test liquids) that have lower surface tension values than the critical surface tension 343 provided by the finish. Fluorine was not detected on any non-fluorinated samples, highlighting 344 sustainable substitution chemistries that may be adopted. Limits of detection, however, meant that 345 definite connections between elemental composition and functionality could not be made.

346 Water repellency ratings were similar across the range of fabrics tested (excluding the untreated 347 fabric); all demonstrating a high level of resistance to wetting, with only random sticking or minor 348 wetting of the fabric face observed. Measurements showed that several non-fluorinated repellent 349 fabric samples provide similar water repellency to long-chain (C8) or shorter-chain (C6) PFAS 350 finished fabrics. Using standard test method BS ISO 23232, some resistance to aqueous-based 351 staining by non-fluorinated repellent fabrics was observed, surface tension of each non-fluorinated fabric ranging between 46.0-33.0 dyn cm⁻¹; this can be associated with repellence of commonplace 352 353 polar liquids such as wine, coffee and fruit juice. Standard test method BS EN ISO 14419 was 354 used to evaluate the fabric's resistance to oil-based liquids corresponding to non-polar liquids used 355 within daily life such as cooking oil, butter, petrol, and sun cream. Non-fluorinated repellent 356 finished fabrics demonstrated no oil repellency, therefore no resistance to these commonplace 357 liquids; and were clearly inferior in this property to long-chain (C8) PFAS finished fabrics, two of 358 which demonstrated good oil-resistance (standard test method BS EN ISO 144419; sample *A* grade 359 4.5; sample *C* grade 5.5). Further investigation into the effect of laundering, abrasion and ageing 360 of these fabrics would provide further insight into the durability of the water repellency, and 361 whether oil repellency is necessary in practice for longevity of performance, and the authors are 362 currently conducting research into this.

363 For a majority sector of outdoor apparel consumers, non-fluorinated chemistry can currently 364 meet repellency requirements. As shown within the consumer survey study, the primary repellency 365 function required in outdoor apparel is water repellency, and we would propose that the use of 366 PFAS chemistry for such garments is therefore over-engineering, providing oil repellency that is 367 in excess of consumer requirements. Consumers ranked stain resistance and dirt and oil repellency 368 to be of lesser importance; evidencing that oil repellency is in excess of consumer requirements. 369 Performance functionality was of greater concern than appearance; however staining may 370 compromise repellent functionality and requires further investigation. From the consumer study, 371 it can also be reasoned that outdoor consumers have an interest in environmental and social impact. 372 Accordingly, significant environmental and toxicological benefits could be achieved by switching 373 outdoor apparel to non-fluorinated DWR chemistry, such as hydrocarbon chemistry, and our 374 further research into the effect of laundering, abrasion and ageing will help in confirming this.

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376 **5. References**

Eschauzier, C.; de Voogt, P.; Brauon, H. J.; Lange, F. T. Polyfluorinated Chemical in
 European Surface Waters, Ground - And Drinking Waters. In *The Handbook of Environmental Chemistry: Polyfluorinated Chemicals and Transformation Products*.
 Knepper, T. P., Lange, F. T., Eds.; Springer-Verlag: Berlin Heidelberg 2012; pp.73-102.

381	2.	Scheringer, M.; Trier, X.; Cousins, I. T.; de Voogt, P,; Fletcher, T.; Wang, Z.; Webster, T. F.
382		Helsingør Statement on poly- and perfluorinated alkyl substances (PFASs). Chemosphere
383		2014 , <i>114</i> , 337-339.

- Kotthoff, M.; Muller, J.; Jurling, H.; Schlummer, M.; Fiedler, D. Perfluoroalkyl and
 Polyfluoroalkyl substances in consumer products. *Environmental Science and Pollution Research.* 2015, *22*, 14546-14559.
- Buck, R. C.; Murphy, P. M.; Pabon, M. Chemistry, Properties and Uses of Commercial
 Fluorinated Surfactants. In: *The Handbook of Environmental Chemistry: Polyfluorinated Chemicals and Transformation Products*. Knepper, T. P., Lange, F. T., Eds; Springer: London
 2012; pp. 1-19.
- 391 5. Giesy, J. P., Kannan, K. Gloabl Distribution of Perfluorooctane Sulfonate in Wildlife.
 392 *Environmental Science & Technology*. 2001, *35*, 1339-1342.
- 393 6. Kissa, E. *Fluorinated surfactants and repellents*. Marcel Dekker: New York, 2001.
- 394 7. Begley, T. H.; White, K.; Houingfort, P.; Twaroski, M. L.; Neches, R.; Walker, R.A.
 395 Perfluorochemicals: Potential sources of and migration from food packaging. *Food Additives*396 *and Contaminants*. 2005, *22* (10), 1023-1031.
- 397 8. OECD/UNEP Global PFC Group, Synthesis paper on per- and polyfluorinated chemicals
 398 (PFCs). 2013. www.oecd.org/env/ehs/risk-management/PFC FINAL-Web.pdf
- Bowman, J. Fluorotechnology Is Critical to Modern Life: The FluoroCouncil Counterpoint to
 the Madrid Statement. *Environmental Health Perspectives*. 2015, *123* (5), A112-113.
- 401 10. Watkins, S. *Clothing: The Portable Environment*. The Iowa State Uni Press: Iowa, 1984.

402	11. Buck, R. C.; Franklin, J.; Berger, U.; Conder, J. M.; Cousins, I.T.; de Voogt, P.; Jensen, A.
403	A.; Kannan, K.; Mabury, S. A.; van Leeuwen, S.P. Perfluoroalkyl and polyfluoroalkyl
404	substances in the environment: terminology, classification, and origins. Integrated
405	Environments Assessment and Management. 2011, 7 (4), 513-541.

- Lindstrom, A.; Strynar, M.; Libelo, E. Polyfluorinated Compounds: Past, Present, and Future. *Environmental Science & Technology*. 2011, 45, 7954-7961.
- 408 13. Conte, L., Gambaretto, G. Electrochemical fluorination: state of the art and future tendences.
 409 *Journal of Fluorine Chemistry*. 2014, *125*, 139-144.
- 410 14. Mahltig, B. Hydrophobic and oleophobic finishes for textiles. In: Functional Finishes for
- 411 *Textiles: Improving Comfort, Performance and Protection*; Paul, R. Ed.; Woodhead
 412 Publishing Limited: Cambridge 2015; pp. 387-428.
- 413 15. Holmquist, H.; Schellenberger, S.; van der Veen, I.; Peters, G. M.; Leonards, P. E. G.;
- Cousins, I.T. Properties, performance and associated hazards of state-of-the-art durable water
 repellent (DWR) chemistry for textile finishing. *Environment International*. 2016, *91*, 251264.
- 417 16. Sachin, B. Fluorochemicals in Textile Finishing. *International Textile Bulletin: Dyeing,*418 *Printing and Finishing.* 1996, 42 (3), 26-30.
- 419 17. Holme, I. Water-repellency and waterproofing. In: *Textile Finishing*. Heywood, D. Ed.;
 420 Society of Dyers and Colourists: Bradford 2003; pp. 135-213.

- 421 18. Lau, C. Perfluorinated Compounds: An Overview. In: *Toxicological Effects of Perfluoroalkyl*422 *and Polyfluoroalkyl substances*. De Witt, J. C. Ed.; Humana Press: Switzerland 2015; pp. 1423 22.
- 424 19. Krafft, M. P.; Riess, J. G. Per- and polyfluorinated substances (PFASs): Environmental
 425 challenges. *Current Opinion in Colloid & Interface Science*. 2015, 20 (3), 192-212.
- 426 20. Buck, R. C.; Schubert, K.- V. Textile Fluorochemicals What Users Need to Know. *AATCC*427 *Review.* 2009, 9 (5), 32-36.

428 21. Barry, V.; Winquist, A.; Steenland, K. Perfluorooctanoic acid (PFOA) exposurers and incident

- 429 cancers among adults living near a chemical plant. *Environmental Health Perspectives*. 2013,
 430 *121* (11-12), 1313-1318.
- 431 22. Birnbaum, L.; Grandjean, P. Alternative to PFASs: Perspectives on Science. *Environmental*432 *Health Perspectives*. 2015, *123* (5), A104-105.
- 433 23. Steenland, K.; Fletcher, T.; Savitz, D.A. Epidemiologic evidence on the health effects of
 434 perfluorooctanoic acid (PFOA). *Environmental Health Perspectives*. 2010, *118* (8), 1100435 1108.
- 436 24. Viberg, H.; Eriksson, P. Perfluorooctane sulfonate (PFOS) and perfluorooctanoic acid
 437 (PFOA). In: *Reproductive and Developmental Toxicology*. Gupta. R. C., Ed.; Elsevier:
 438 London 2011; pp. 627-630.
- 439 25. Domingo, J.L. Health risks of dietary exposure to perfluorinated compounds. *Environment*440 *International.* 2012, *40*, 187-195.

- 441 26. Farre, M.; Llorca, M.; Perez, S.; Barcelo, D. Perfluorinated Compounds in Food. In: *The*442 *Handbook of Environmental Chemistry: Polyfluorinated Chemicals and Transformation*443 *Products.* Knepper, T. P., Lange, F. T., Eds. 2012; pp. 127-149.
- 444 27. Shiwanov, E. Chain Reactions: A deep look into the DWR dilemma and the current options
 445 for product designers. *Inside Outdoor Magazine*. 2015, 24-29.
- 446 28. Hekster, F. M.; Laane, R. W. P.; de Voogt, P. Environmental and Toxicity Effects of
 447 Perfluoroalkylated Substances. In: *Reviews of Environmental Contamination and Toxicology*.
- 448 Ware, G. W. Ed., 2003; 179; pp. 99-121.
- 29. Conder, J.M.;Hoke, R. A.; De Wolf, W.; Russell, M. H.; Buck, R. C. Are PFCAs
 Bioaccumulative? A Critical Review and Comparison with Regulatory Critieria and Persistent
- 451 Lipophilic Compounds. *Environmental Science Technology*. **2008**, *42* (4), 995-1003.
- 30. Jones, P. D.; Hu, W.; De Coen, W.; Newsfed, J. L.; Giesy, J. P. Binding of perfluorinated fatty
 acids to serum proteins. *Environmental Toxicology and Chemistry*. 2003, *22* (11), 2639-2649.
- 31. Ng, C. A.; Hungerbühler, K. Bioconcentration of Perfluorinated Alkyl Acids: How Important
 is Specific Binding? *Environmental Science Technology*. 2013, 47 (13), 7214-7223.
- 456 32. Liu, J.; Wang, N.; Szostek, B.; Buck, R. C.; Panciroli, P. K.; Folsom, P. W.; Sulecki, L. M.;
- Bellini, C.A. 6-2 Fluorotelomer alcohol aerobic biodegradation in soil and mixed bacterial
 culture. *Chemosphere*. 2010, 78 (4), 437-444.
- Wang, Z.; Cousins, I. T.; Scheringer, M.; Hungerbuhler, K. Fluorinated alternatives to longchain perfluoroalkyl carboxylic acids (PFCAs), perfluroalkane sulfonic acids (PFSAs) and
 their potential precursors. *Environment International*. 2013, 60, 242-248.

- 462 34. Gardner, T. Popularity and economic benefit of mountaineering: instant expert.
 463 www.thebmc.co.uk/participation-in-climbing-mountaineering.
- 464 35. Cloud, R. N.; Cao, W.; Song, G. Functional finishes to improve the comfort and protection of
- 465 apparel. In: Advances in the Dyeing and Finishing of Technical Textiles. Gulrajani, M. Ed.;
- 466 Woodhead Publishing Limited: Cambridge 2013; pp. 258-279.
- 467 36. Booth, J. E. 1983. *Principles of textile testing*. Butterworths: London, 1983.
- 468 37. Watkins, S.; Dunne, L. *Functional Clothing Design: From Sportswear to Spacesuits*.
 469 Bloomsbury: London, 2015.
- 470 38. Ea, J. *Water vapour transfer in Breathable Fabrics for Clothing*. PhD Thesis, University of
 471 Leeds: Leeds, 1988.
- 472 39. Golden, F., Tipton, M. *Essentials of Sea Survival*. Human Kinetics Publishers: London, 2002.
- 473 40. Rengasamy, R. S. Improving moisture management in apparel. In: *Improving Comfort in*474 *Clothing*. Song, G. Ed.; Woodhead Publishing Limited: Cambridge 2011; pp. 182-246.
- 475 41. Pavlidou, S.; Paul, S. Moisture management and soil release finishes for textiles. In:
 476 *Functional Finishes for Textiles*. Paul, R. Ed.; Woodhead Publishing Limited: Cambridge
 477 2015; pp. 99-121.
- 478 42. Knepper, T. P.; Fromel, T.; Gremmel, C.; Driezum, I.V.; Weil, H.; Vestergren, R.; Cousins,
- 479 I. Understanding the exposure pathways of per- and polyfluoralkyl substances (PFASs) via
- 480 use of PFASs-containing products risk estimation for man and environment. www.reach-
- 481 info.de/dokumente/pfc_expositionspfade.pdf.

- 482 43. Taves, D. R. Evidence that there are Two Forms of Fluoride in Human Serum. *Nature*. 1968,
 483 217, 1050-1051.
- 484 44. Frömel, T.; Knepper, T. P. Biodegradation of Fluorinated Alkyl substances. In: *Reviews of*485 *Environmental Contamination and Toxicology*. De Voogt, P. Ed., 2010; 208; pp. 161-177.
- 486 45. Ritter, S. K. Fluorochemicals Go Short. *Chemical and Engineering News*. 2010, *881* (5); 12487 17.
- 488 46. 3M: The Science of Organic Fluorochemistry. https://www.fluoridealert.org/wp489 content/pesticides/pfos.fr.final.docket.0006.pdf
- 490 47. The European Parliament and the Council of the European UnionDirective 2006/122/ECOF,
- 491 L 372/32. eur-lex.europa.eu/LexUriServ/LexUriServ.do?uri=OJ:L:2006:372:0032:0034:en
- 492 48. ECHA. Opinion on an Annex XV dossier proposing restrictions on Perfluorooctanoic acid
- 493 (PFOA), its salts and PFOA-related substances. echa.europa.eu/documents/10162/3d13de3a494 de0d-49ae-bfbd-749aea884966.
- 49. ECHA. Annex X V Restriction Report: Proposal for a Restriction. echa.europa.eu/documents/10162/e9cddee6-3164-473d-b590-8fcf9caa50e7
- 497 50. ECHA. Annex: RAC concludes on PFOA restriction. The Committee finalises two opinions
 498 for authorisation, and adopts six opinions on harmonised classification and labelling.
 499 echa.europa.eu/documents/10162/13579/annex_to_rac_news_alert_15_september_2015.pdf
- 500 51. Blum, A.; Balan, S. A.; Scheringer, M.; Trier, X.; Goldenman, G.; Cousins, I. T.; Diamond,
- 501 M.; Fletcher, T.; Higgins, C.; Lindeman, A. E.; Peaslee, G.; de Voogt, P.; Wang, Z.; Weber,

- 502 R. The Madrid Statement on Poly- and perfluoroalkyl substances (PFASs). Environmental 503 Health Perspectives. 2015, 123 (5), A107-111.
- 504 52. Greenpeace. Footprints in the Snow: Hazardous PFCs in remote locations around the globe.
- 505 www.greenpeace.org/international/Global/international/publications/toxics/2015/Footprints-
- 506 in-the-Snow-Executive-Summary-EN.pdf
- 507 53. Greenpeace. Leaving Traces. www.greenpeace.org/sweden/Global/sweden/miljogifter/-508 dokument/2016/Leaving Traces report.pdf.
- 509 54. Greenpeace. Hidden in Plain sight. www.greenpeace.org/international/Global/international/-510 publications/detox/2016/Hidden in Plain Sight.pdf.
- 511 55. Van der Veen, I.; Weiss, J. M.; Hanning, A.; de Boer, J.; Leonards, P. E. G. Development and 512 validation of a method for the quantification of extractable perfluoroalkyl acids (PFAAs) and 513 perfluorooctane sulfonamide (FOSA) in textiles. Talanta. 2016, 147, 8-15.
- 514 56. Gremmel, C.; Fromel, T.; Knepper, T.P. Systematic determination of perfluoroalkyl and 515 polyfluoroalkyl substances (PFASs) in outdoor jackets. Chemosphere. 2010, 160, 173-180.
- 516 57. Liu, X.; Krebs, K.; Guo, Z.; Roache, N. Method development for liquid 517 chromatographic/triple quadrupole mass spectrometric analysis of trace level 518 perfluorocarboxylic acids in articles of commerce. Journal of Chromatography A. 2009, 1216, 519 3910-3918.
- 520 58. Hanssen, L.; Herzke, D. Investigation of outdoor textiles with respect to determine the content 521
- of ionic perfluorinated substances (PFASs): Evaluation of Results. www.miljodirektoratet.-
- 522 no/Documents/publikasjoner/M315/M315.pdf

524		Articles of Commerce. U.S. Environmental Protection Agency, Washington, DC, EPA/600/R-
525		09/033, 2009.
526	60.	Knepper, T. P.; Fromel, T.; Weil, H.; Gremmel, C.; van Driezum, I.; Vestergreen, R.; Cousins,
527		I. Outdoor jackets as a source of PFASs in the environment. www.umweltbundesamt.de-
528		/sites/default/files/medien/376/dokumente/reach_in_der_praxis_ws_iii_7_praesentation_tho
529		mas_knepper.pdf.
530	61.	Davies, A. Durable Water Repellency - Study Phase 1: An evaluation of the test methods used
531		for assessing durable water repellent fabrics within the outdoor industry.
532		www.europeanoutdoorgroup.com/files/DWR-Study_Alice_Daviesdigitalpdf.
533	62.	AATCC 2014. Water Repellency: Spray Test (AATCC Test method 22-2014), 2014.
534	63.	BSI (British Standards Institution). BS EN ISO 4920:2012 Textile Fabrics - Determination of
535		resistance to surface wetting (spray test), 2012.
536	64.	BSI (British Standards Institution). BS ISO 23232:2009 Textiles - Aqueous liquid repellency
537		- Water/alcohol solution resistance test, 2009.
538	65.	BSI (British Standards Institution). BS EN ISO 14419:2010 Textiles - Oil repellency -
539		Hydrocarbon resistance test, 2010.
540		
541		

59. Guo, X.; Liu, X.; Krebs, K. A.; Roache, N. F. Perfluorocarboxylic Acid Content in 116

Supporting Information.

- **Table S.1.** Characterization of repellent outerwear fabrics supplied to the study; PA, polyamide;
- 544 PET, polyester.

Sample	Alleged repellent finish type	Weight (g m ⁻²)	Thickness (mm)*	Fabric face structure	Fibre type	No of layers	Membrane or laminate
A	Long-chain PFAS (C8)	239.5	0.70	Plain weave	100% PA	2	-
В	Long-chain PFAS (C8)	146.8	0.31	Twill weave	100% PA	3	Membrane
С	Long-chain PFAS (C8)	165.4	0.22	Plain weave	100% PET	2	Laminate
D	Shorter-chain PFAS (C6)	135.9	0.22	Plain weave	100% PA	2	PU coating
E	Shorter-chain PFAS (C6)	133.6	0.35	Plain weave	PA/elastan e	1	-
F	Shorter-chain PFAS (C6)	47.1	0.09	Plain weave	100% PA	1	Wax coating
G	Shorter-chain PFAS (C6)	53.4	0.05	Plain weave	100% PA	1	Acrylic coating
Н	Shorter-chain PFAS (C6)	50.2	0.04	Plain weave	100% PA	1	Coating
J	Shorter-chain PFAS (C6)	77.2	0.12	Plain weave	100% PA	2	Laminate
K	Shorter-chain PFAS (C6)	121.8	0.26	Plain weave	100% PA	3	Membrane
L	Shorter-chain PFAS (C6)	193.3	0.38	Plain weave	96% PA/ 4% elastane	3	Laminate
М	Shorter-chain PFAS (C6)	167.4	0.24	Plain weave	100% PET	2	Laminate
N	Non-fluorinated (fat- modified resin)	136.9	0.21	Plain weave	100% PA	2	PU coating
0	Non-fluorinated (specifics unknown)	139.5	0.37	Plain weave	PA/elastan e	1	-

Р	Non-F (dendrimers with PU backbone)	54.6	0.09	Plain weave	100% PA	1	-
Q	Non-F (dendrimers with PU backbone)	121.5	0.25	Twill weave	100% PET	2	PU laminate
R	Non-F (dendrimers with PU backbone)	138.5	0.19	Plain weave	65% PA/ 35% PET	2	PU coating
S	Non-F (dendrimers with PU backbone)	198.0	0.63	Plain weave	100% PET	3	Laminate
Τ	Non-F (dendrimers with PU backbone)	222.5	0.54	Plain weave	100 % PA	3	Laminate
U	Non-F (dendrimers with PU backbone)	138.5	0.33	Plain weave	85% PA/ 15% elastane	1	
V	Non-F (dendrimers)	165.1	0.20	Plain weave	100% PET	2	Laminate
Ζ	Untreated	114.6	0.18	Plain weave	100% PA	1	-

*Thickness measured using a FAST-1 Compression Tester at two fixed loads: 2 g cm⁻² and 100 g cm⁻² over a 10 cm² area of the fabric.

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559 Table of Contents Graphic and Synopsis

560 PFAS chemistry for outdoor apparel is over-engineering; significant environmental and 561 toxicological benefits can be achieved by switching to non-fluorinated finishes without significant 562 reduction in water-repellency performance.

