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orcid.org/0000-0002-5249-9523 and Halcrow, MA orcid.org/0000-0001-7491-9034 (2017)
Gradual thermal spin-crossover mediated by a reentrant $Z' = 1 \rightarrow Z' = 6 \rightarrow Z' = 1$ phase transition. Inorganic Chemistry, 56 (6). pp. 3144-3148. ISSN 0020-1669

https://doi.org/10.1021/acs.inorgchem.7b00071

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Gradual Thermal Spin-Crossover Mediated by a Re-Entrant $Z' = 1 \rightarrow Z' = 6 \rightarrow Z' = 1$ Phase Transition

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ABSTRACT: The Fe[BF$_4$]$_2$ complex of the Schiff base podand $tris$-(4-{thiazol-4-yl}-3-aza-3-butenyl)amine exhibits gradual thermal spin-crossover with $T_{1/2} = 208$ K in the solid state. A weak discontinuity in the magnetic susceptibility curve at 190 K is associated with a re-entrant symmetry breaking transition involving a trebling of the unit cell volume (from $P2_1/c$, $Z = 4$ to $P2_1$, $Z = 12$). The intermediate phase contains six independent cations in puckered layers of low-spin, and high-spin or mixed-spin, molecules with an overall 30 % high-spin population at 175 K.

Spin-crossover compounds undergo a reversible transition between high-spin and low-spin electronic states under the influence of temperature or other physical stimuli. This can lead to changes in the color, magnetic moment, conductivity, dielectric constant and/or fluorescence of a material, as well as inducing a mechanical response. Spin-crossover transitions function in particles or thin films at the nanoscale, and have also been detected in single molecule junctions. The structural chemistry of molecular spin-crossover materials continues to be widely studied because of this functional flexibility. Thermal spin-transitions in bulk phases can exhibit varying degrees of cooperativity, occurring gradually or abruptly; with or without hysteresis; completely or incompletely; and, continuously (in one step) or discontinuously (in two or more steps). Crystal engineering of spin-crossover compounds is a fundamental challenge, for the design of new materials for application purposes.

Discontinuous spin-transitions can arise for a number of reasons. In particular, single or re-entrant crystallographic phase changes between the spin states can perturb the progress of a spin-transition, often affording a plateau region over a finite temperature range with a mixed high:low-spin population. Since the first discovery in $[Fe(pic)]_3Cl_2$·EtOH (pic = 2-{aminomethyl}-pyridine), a variety of re-entrant symmetry-breaking phase behaviors have been found in different spin-crossover compounds. The high- and low-spin molecules in the resultant intermediate phases are typically segregated into sub-lattices with 0D (checkerboard), 1D (chain) or 2D (layer) dimensionality.

Since crystallographic phase changes are cooperative phenomena, symmetry-breaking phase changes are usually associated with abrupt thermal spin-transitions. However, we report here a new compound 1 (Chart 1) exhibiting re-entrant symmetry breaking that, unusually, is associated with a gradual thermal spin-crossover.

Complex 1 was formed by reaction of a 3:1:1 ratio of thiazole-4-carboxaldehyde, $tris$-(2-aminoethyl)amine and Fe[BF$_4$]$_2$·6H$_2$O in refluxing ethanol. After the usual work-up, slow diffusion of diethyl ether vapor into nitromethane solutions of the compound yielded orange solvent-free crystals. Spin-crossover in 1 occurs gradually between the temperatures of 100-350 K, with $T_{1/2} = 208\pm2$ K from magnetic susceptibility data. The spin-crossover proceeds without thermal hysteresis but has a small discontinuity near 190 K, just below $T_{1/2}$ (Figure 1).

Figure 1 Variable temperature magnetic susceptibility data for polycrystalline 1, at scan rate 5 Kmin$^{-1}$, on cooling (black) and warming (gray) temperature ramps. The inset graph shows the first derivative of the curves.

At first glance, crystals of 1 retain the same space group at all temperatures ($P2_1/c$, $Z = 4$). The unit cell contains one formula unit of the compound, with no crystallographically imposed symmetry. Although podand complexes like 1 have a helical ligand conformation (Figure S3), 1 crystallizes as a racemate with equal numbers of A and D helical molecules in the centrosymmetric-
The unit cell volume decreases smoothly on cooling from 300 to 130 K, consistent with a gradual spin-crossover. However, this masks a discontinuous increase in $b$ and a decrease in $c$ between 150-200 K (Figures S1 and S2). That implies the spin-crossover discontinuity around 190 K in I is linked to a change in the thermal expansion properties of the crystal.

Five different crystals of I were examined, which were all isostructural at 300 and 120 K. The crystal reported here gave typical results, and was studied at seven temperatures within that range. The complex is essentially high-spin at 300 K and low-spin at 120 K according to its metric parameters. Structures at intermediate temperatures imply a mixed spin-state population, with a smooth high-low spin conversion upon cooling. This is accompanied by a conformational rearrangement of the podand ligand, such that the Fe...N distance to the non-coordinated bridgehead N atom is ca 0.7 Å longer in the low-spin state (Figure S3). That structure rearrangement is typical for spin-crossover in tren-based podand complexes.\(^{15,17}\)

More unusually, the atomic displacement parameters in the complex molecule grow steadily larger and more elongated on cooling between 250-175 K, before slowly contracting again at lower temperatures. At 150 K, the Fourier map resolved two equally occupied disorder sites for the complete complex molecule, which were refined with distance restraints applied to the ligand atoms; additional disorder at one thiazolyl ring was also included (Figures S4 and S5). The same disorder model refined successfully at 175, 200 and 225 K, each with improved refinement residuals compared to the ordered cation model. However, refinements based on a single wholly occupied cation site were superior at the other temperatures examined. The whole-molecule disorder sites at each intermediate temperature have comparable spin-state populations (Figure 2). Hence the disorder orientations do not simply correspond to the high-spin and low-spin forms of the molecule, but are a reflection of the lattice dynamics.

Figure 2 The evolution of the coordination sphere in the parent phase of I ($V_{Oh}$, white circles),\(^{19}\) compared with the progress of the spin-crossover transition (black spheres; Figure 1). Between 150-225 K, $V_{Oh}$ for the two molecular disorder sites is shown as green and yellow diamonds, and the white circles represent the average of the two.

The metric parameters from these crystallographic refinements reproduce the form of the gradual spin-crossover in I (Figures 2 and S7). While Figure 2 implies the cation disorder sites at intermediate temperatures may undergo spin-crossover at different rates, this should be interpreted with care because of the disordered nature of the crystal. None-the-less, the averaged $V_{Oh}$ values from the partial cations agree excellently with the spin-crossover measured from magnetic data, including the inflection near 190 K (Figure 2).

Additional weak reflections were observed in the diffraction images from I around 175 K, where the molecular disorder is most pronounced (Figure S6). These data could sometimes be merged to a larger monoclinic supercell, whose volume is three times larger than the original unit cell. The dimensions of the supercell relate to the parent unit cell as follows: $a' = \sqrt[3]{(\sqrt{2})}a$; $b' = b$; $c' = \sqrt{3}a$, and $\beta' = \beta.\(^{21}\)$ That would be consistent with a re-entrant symmetry-breaking transformation to an intermediate crystal phase during the progress of the spin-crossover. While the extra diffraction peaks were consistently observed in different experiments, in most cases data merged to the supercell could only be solved and refined after transformation into the original, reduced $P2_1/c$ unit cell.\(^{21}\) This was the case for experiments involving the crystal described above. However, a successful solution of the enlarged supercell at 175 K was obtained from one experiment using a different crystal. The refinement of the intermediate phase has relatively low precision, reflecting the size of the model and significant cation and anion disorder which was only partly resolvable. However, the basic features of the intermediate phase are clear (Figure 3).

Figure 3 Packing diagram of the intermediate phase of I, showing the segregation of the low-spin (purple) and high-spin/mixed-spin (white) molecules into puckered layers in the lattice. Only one orientation of the disordered residues is shown, and the $\text{BF}_4^-$ ions (green) are de-emphasized. The view shows the (010) lattice plane, with the $c$ axis vertical.
The asymmetric unit of the intermediate phase (P2₁, Z = 12) contains six unique complex molecules, with molecules A, B and C in the model having a δ helical ligand conformation molecules D, E and F a Λ conformation. The intermediate phase is also a perfect racemic twin, which is to be expected given the centro-symmetric nature of the parent phase. Based on their metric parameters three of the unique molecules are low-spin; one is predominantly high-spin; and the other two have a mixed high/low spin-state population (Table S4). That is consistent with the ca 30 % high-spin fraction predicted by the magnetic susceptibility curve at that temperature (\( \gamma M T = 0.98 \text{ cm} \cdot \text{mol}^{-1} \cdot \text{K} \) at 175 K). It also agrees well with the structure refinement of the parent phase at the same temperature (Tables S3 and S4). The low-spin cations (molecules B, E and F in the refinement) and the high-spin/mixed spin cations (molecules A, C and D) are grouped into puckered, interdigitated layers running parallel to the (102) lattice plane (Figure 3). There is no correlation between the spin-state and ligand helicity in the different cations.

Important intermolecular contacts in the lattice of I are listed in Tables S5-S7. There is a typical distribution of weak intermolecular C–H...X and S...F contacts between the cations and anions in the low-spin form of I. However, there is no long-range network of direct interactions between the complex cations, to propagate a spin-transition through a lattice. Rather, just centrosymmetric pairs of molecules in I are in direct contact, through one unique C–H...S interaction that is 0.2 Å shorter than the van der Waals sum of those atoms at 120 K (Figure 4). There is also no π–π overlap between thiazolyl rings in neighboring molecules. The same is true for the intermediate phase, except that there are fewer of the short intermolecular C–H...S contacts between cations. Relief of that steric clash may be one of the drivers for the local overlap between thiazolyl rings in neighboring molecules. The previous nature of spin-crossover in the asymmetric unit of the intermediate phase, which is among the largest known in a spin-crossover crystal. More fundamentally, the gradual nature of spin-crossover in I is also notable. As mentioned above, symmetry-breaking phase changes usually have a profound effect on a spin-transition, leading to abrupt, discontinuous and/or step-wise switching upon cooling. However, the symmetry breaking transition in I has a negligible effect on the progress of its spin-crossover, and is barely reflected in the magnetic susceptibility data (Figures 1 and 3). The weakly cooperative nature of I is consistent with its low-density crystal packing (Table S8), where the cations are well-separated from each other by the BF₄⁻ anions. That might explain the inconsistent resolution of the intermediate phase in different experiments. While significant rearrangement of the cations within their lattice sites between 150-225 K is evident from their crystallographic disorder (Figures S4 and S5), this rarely occurs with sufficient long-range order to be resolved in a single crystal dataset.

In conclusion, this study demonstrates that gradual spin-crossover transitions that are unexceptional at first glance can involve more complicated structural chemistry, that merits detailed characterization.

ASSOCIATED CONTENT
Supporting Information
Full experimental details, tables of metric parameters, additional crystallographic figures and variable temperature unit cell data. This material is available free of charge via the Internet at http://pubs.acs.org. CCDC 1519291–1519297 (parent phase) and 1526385 (intermediate phase) containing the supplementary crystallographic data can be obtained free of charge from the Cambridge Crystallographic Data Center via www.ccdc.cam.ac.uk/data_request/cif.

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