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**Published paper**
The linear stability of freely propagating, adiabatic, planar premixed flames is investigated in the context of a simple chain-branching chemistry model consisting of a chain-branching reaction step and a completion reaction step. The role of chain-branching is governed by a crossover temperature. Hydrodynamic effects, induced by thermal expansion, are taken into account and the results compared and contrasted with those from a previous purely thermal-diffusive constant density linear stability study. It is shown that when thermal expansion is properly accounted for, a region of stable flames predicted by the constant density model disappears, and instead the flame is unstable to a long-wavelength cellular instability. For a pulsating mode, however, thermal expansion is shown to have only a weak effect on the critical fuel Lewis number required for instability. These effects of thermal expansion on the two-step chain-branching model are shown to be qualitatively similar to those on the standard one-step reaction model. Indeed, as found by constant density studies, in the limit that the chain-branching crossover temperature tends to the adiabatic flame temperature, the two-step model can be described to leading order by the one-step model with a suitably defined effective activation energy.
Stability of premixed chain-branching flames

Although the one-step model has been very successful in describing and predicting many aspects of flame phenomena [1, 4], it fails to capture some aspects of hydrocarbon and hydrogen flame structures [4, 14]. In these flames, there are many intermediate steps between the conversion of fuel into products. These include chain-branching reactions, which produce a net increase in intermediates such as radicals. The chain-branching reactions tend to have high-activation energy, and are hence active in the high-temperature regions of the flame, where they convert the fuel into intermediate species [14]. The intermediate molecules produced may then diffuse forwards and backwards over the entire flame structure, so that the ‘pre-heat’ zone is in reality also chemically active [4, 14]. Completion reactions, which remove the intermediates and convert them into products, tend to be temperature insensitive but highly exothermic, so that heat release occurs throughout the flame [4], in contrast to in a narrow region at the downstream end of the flame structure as predicted by the one-step model. Indeed, the exothermic completion reactions continue even after the fuel has been completely converted into intermediates, so that the fuel is exhausted interior to the flame, i.e. before the adiabatic flame temperature is reached at the downstream end [14].

This discrepancy between the one-step model and real flame structures motivates the need for a chemistry model which better mimics the effects of chain-branching outlined above. For the purposes of the mathematical theory of flames, however, any improved model should still be sufficiently simple and generic such that transparent, fundamental insights can be obtained and that some analytical or asymptotic progress is still possible. In this spirit, Dold and co-workers [14–16] have suggested a two-step chemistry model, consisting of a single chain-branching step, \( F + Y \rightarrow 2Y \), and a single completion reaction step, \( Y + M \rightarrow P + M \), where \( Y \) represents a lumped or ‘pooled’ amalgam of intermediate species, and \( M \) is any species required to trigger the completion reactions, but is unchanged in the process. In the simplest version of this model, the branching reaction is assumed to have a high activation temperature but is thermally neutral, while the completion reaction is assumed to be temperature insensitive but releases all the heat. Dold [14] gives a very detailed discussion which puts the two-step model and its assumptions into the context of hydrogen and hydrocarbon oxidation, including how the model parameters can be fitted to hydrogen and hydrocarbon flame structures as predicted from detailed chemistry calculations.

Fundamental to the two-step model described above is the concept of a chain-branching crossover temperature, \( T_c \), which, in regards to flame structure, is the temperature at which the rate of chain-branching balances the rate of removal of intermediates by diffusion [14]. Thus above this temperature a chain-branching explosion occurs in which the fuel is converted rapidly into intermediates. For large activation energies of the branching step, the reaction is then active only in a narrow region occurring at temperatures close to \( T_c \). As for the one-step reaction, in the asymptotic limit of infinite activation energy, the branching reaction can be replaced to leading order by a reaction sheet across which appropriate jump conditions apply. For the two-step model, the reaction sheet occurs at the crossover temperature, where the fuel is consumed.

As well as being able to mimic the main features of real flames which the one-step model can not, the two-step model also has mathematical advantages over the one-step model. Firstly, the simple chain-branching model does not suffer from the well known ‘cold-boundary difficulty’ inherent in the one-step model [17]. Secondly, in the large activation energy asymptotic limit, the jump conditions across the reaction sheet are linear in the variables [14], as compared to the more complex jump conditions of the one-step model, which are non-linear in the temperature at which the reaction sheet occurs [1]. Further, in terms of the asymptotic flame stability, the two-step model does not require the NEF approximation and hence is valid for arbitrary values of the Lewis numbers [14].

Dold and co-workers [14–16] have applied the high-activation energy asymptotic limit of the two-step chemistry model to a number of flame problems, in the context of the purely thermal-diffusive constant density model. These include studies of the structure and stability of flame balls [16] and of premixed flames [14]. In this paper, we extend the asymptotic linear stability analysis of the two-step chain-branching premixed flames to include hydrodynamic effects induced by thermal expansion, by considering the Re-
active Navier-Stokes equations. The main purpose of the paper is hence to compare and contrast with both the two-step CDM [14] and the one-step chemistry Reactive Navier-Stokes predictions [9]. Indeed, the paper should be viewed as the two-step analog of the one-step linear stability study of Jackson and Kapila [9].

2 The Model

The governing equations of the model are the quasi-isobaric Navier-Stokes equations coupled to the two-step chemistry scheme \( F + Y \rightarrow 2Y, Y + M \rightarrow P + M \). The non-dimensional versions of these equations are, in two spatial dimensions, \( x \) and \( y \),

\[
\frac{\partial \rho}{\partial t} + \frac{\partial (\rho u)}{\partial x} + \frac{\partial (\rho v)}{\partial y} = 0, 
\]

\[
\rho \frac{\partial u}{\partial t} + \rho u \frac{\partial u}{\partial x} + \rho v \frac{\partial u}{\partial y} + \frac{\partial P}{\partial x} = Pr \left( \frac{4}{3} \frac{\partial^2 u}{\partial x^2} + \frac{\partial^2 u}{\partial y^2} + \frac{1}{3} \frac{\partial^2 v}{\partial x \partial y} \right),
\]

\[
\rho \frac{\partial v}{\partial t} + \rho u \frac{\partial v}{\partial x} + \rho v \frac{\partial v}{\partial y} + \frac{\partial P}{\partial y} = Pr \left( \frac{4}{3} \frac{\partial^2 v}{\partial y^2} + \frac{\partial^2 v}{\partial x^2} + \frac{1}{3} \frac{\partial^2 u}{\partial x \partial y} \right),
\]

\[
\rho \frac{\partial T}{\partial t} + \rho u \frac{\partial T}{\partial x} + \rho v \frac{\partial T}{\partial y} = \frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} + QW_C,
\]

\[
\rho \frac{\partial F}{\partial t} + \rho u \frac{\partial F}{\partial x} + \rho v \frac{\partial F}{\partial y} = \frac{1}{Le_F} \left( \frac{\partial^2 F}{\partial x^2} + \frac{\partial^2 F}{\partial y^2} \right) - W_B,
\]

\[
\rho \frac{\partial Y}{\partial t} + \rho u \frac{\partial Y}{\partial x} + \rho v \frac{\partial Y}{\partial y} = \frac{1}{Le_Y} \left( \frac{\partial^2 Y}{\partial x^2} + \frac{\partial^2 Y}{\partial y^2} \right) + W_B - W_C,
\]

\[
\rho T = 1,
\]

where \( \rho \) is the density, \( u \) and \( v \) are the \( x \)- and \( y \)-components of the fluid velocity, respectively, \( T \) the temperature, \( F \) and \( Y \) the mass fraction of fuel and intermediates. These equations have been non-dimensionalized using the standard scales employed in previous linear stability analysis of the one-step model [9,12], with which we seek to compare. Thus

\[
\rho = \bar{\rho}, \quad u = \bar{u} V_f, \quad v = \bar{v} V_f, \quad p = \bar{p} p_f, \quad T = \bar{T} T_f,
\]

\[
x = \frac{\bar{\rho} V_f c_p}{\bar{\kappa}} \bar{x}, \quad y = \frac{\bar{\rho} V_f c_p}{\bar{\kappa}} \bar{y}, \quad t = \frac{\bar{\rho} V_f^2 c_p}{\bar{\kappa}} \bar{t},
\]

where a bar (\( \bar{\cdot} \)) denotes dimensional quantities, an ‘\( f \)’ subscript denotes quantities in the fresh, unburnt gas upstream of the flame (and a ‘\( b \)’ subscript will be used to denote quantities in the completely burnt...
state downstream of the flame). Thus note that the non-dimensional temperature and density are unity in the fresh gas. Here \( p \) is the pressure, \( \bar{V}_f \) is the speed of the steady, planar flame, \( \bar{c}_p \) is the specific heat at constant pressure, \( \kappa \) is the co-efficient of thermal conductivity and \( \bar{\rho}_f \bar{V}_f \bar{c}_p / \kappa \) is a heat conduction lengthscale (characteristic of the ‘preheat zone’ of the flame in the one-step model). The quantity \( P \) appearing in equations (2) and (3) is the \( O(M_f^2) \) deviation of the pressure from the upstream value, i.e. \( p = 1 + \gamma M_f^2 P \), where \( M_f = \bar{V}_f (\bar{\rho}_f / (\gamma \bar{p}_f))^{\frac{1}{2}} \) is the Mach number of the flame, and \( \gamma \) is the ratio of specific heat. The non-dimensional parameters appearing in equations (2)-(6) are the Prandtl number, \( Pr \), the heat of reaction of the completion step, \( Q \), and the Lewis numbers of the fuel and of the intermediates, \( Le_F \) and \( Le_Y \), respectively.

Note that Dold [14] used alternative scalings for the non-dimensionalization. However, his non-dimensional quantities (denoted by a ‘D’ subscript) are simply related to those used here by

\[
T_D = \frac{T}{T_c}, \quad F_D = F, \quad Y_D = \frac{Le_F}{Le_Y} Y, \quad x_D = \sqrt{Le_Y \Lambda x}, \quad t_D = Le_Y t, \quad Q_D = \frac{Q}{T_c - 1},
\]

where \( \Lambda \) and \( T_c \) are defined below, and \( Q_D \) is an alternately defined ‘heat of reaction’ on which the steady, planar flame structure and CDM flame stability are found to depend [14].

The non-dimensional branching and completion reaction rates are assumed to be of the form

\[
W_B = \Lambda_B \rho^2 F Y \exp \left( \theta \left[ \frac{1}{T_c} - \frac{1}{T} \right] \right), \quad W_C = \Lambda \rho^2 T^n Y,
\]

where \( \Lambda_B \) and \( \Lambda \) are non-dimensional rate constants, \( \theta \) is the non-dimensional activation energy of the branching step, and a weak power law temperature dependence of the completion step is assumed for later analytical convenience. Here \( T_c \) is the ‘inhomogeneous’ crossover temperature defined by Dold [14], i.e. the temperature at which the chain-branching rate becomes equal to the rate of removal of intermediates by molecular diffusion. It can be seen that for large activation energy \( \theta \), the branching rate quickly becomes very small compared to diffusion as the temperature decreases below \( T_c \), while it becomes very large as \( T \) increases above \( T_c \). Thus while the reaction will be essentially frozen for temperatures below the crossover temperature, the fuel will rapidly be exhausted (and the branching-reaction complete) once \( T \) is above \( T_c \) due to the very high reaction rate. Hence, as its activation energy increases, the branching step will only be active in a increasingly narrow region around the crossover temperature.

In the asymptotic limit, \( \theta \to \infty \), it can be shown that the branching reaction can be replaced to leading order by an infinitesimally thin reaction sheet in which the fuel is consumed. For the chain-branching model, this surface occurs at \( T = T_c \), and suitable jump conditions need to be applied across it [14,15]. An analysis of the inner branching reaction zone structure [1,14,15] determines the outer jump conditions to be applied at the reaction sheet:

\[
[T] = [F] = F = [Y] = [u] = [v] = [P] = 0, \quad (8)
\]

\[
[T, n] = [F, n] + \frac{Le_F}{Le_Y} [Y, n] = [u, n] = [v, n] = 0, \quad (9)
\]

\[
T = T_c, \quad (10)
\]

where \([\cdot]\) denotes the jump in a quantity across the reaction sheet and \( \cdot, n \) denotes the derivative of a quantity in the direction normal to the reaction sheet (hence \([\cdot, n]\) denotes the jump in the normal gradient across the sheet). Note that the jump conditions for the two-step model are linear in the variables and their derivatives and are independent of \( \theta \) to leading order [14]. Note also that, as \( \theta \to \infty \), the inhomogeneous crossover temperature, \( T_c \), is related to the homogeneous crossover temperature, \( T_X \) (which
is the temperature where the branching and completion reaction rates balance) by

\[ T_X = T_c - 2T_c^2 \theta^{-1} \ln(\theta/T_c), \]

and hence, these inhomogeneous and homogeneous crossover temperatures are the same to leading order \([14]\).

3 Steady, planar flames

In the laboratory frame, the steady, planar flame is assumed to travel in the negative \(x\)-direction at unit non-dimensional speed, so that the fresh, unburnt fuel is approached as \(x \to -\infty\) and the completely burnt state approached as \(x \to \infty\). Here we work in the rest frame of the flame, such that the flow is steady (independent of \(t\)) and the upstream fuel is oncoming at unit speed. Denoting the steady flame solution by a zero subscript, after integrating once with respect to \(x\) and employing the boundary conditions \(T_0 = \rho_0 = u_0 = F_0 = 1\), \(P_0 = Y_0 = 0\) and \(dq_0/dx = 0\) as \(x \to -\infty\) (where \(q\) denotes any of the dependent variables), the governing equations (1)-(7) can be reduced to

\[
\begin{align*}
\frac{dT_0}{dx} &= T_0 - 1 + Q(X_0 + Z_0 - 1), & \frac{dF_0}{dx} &= Le_F(F_0 - X_0), & \frac{dY_0}{dx} = Le_Y(Y_0 - Z_0), \\
\frac{dX_0}{dx} &= 0, & \frac{dZ_0}{dx} &= -\Delta T^n - 2Y_0, \\
\end{align*}
\]

(11)

where \(X_0\) and \(Z_0\) are defined by the second and third of equations (11), cf. [18], (hence note \(X_0 = 1\) in the region \(x < 0\), and \(Z_0 = 0\) at the cold boundary), together with

\[
\begin{align*}
u_0 &= \frac{1}{\rho_0} = T_0, & P_0 &= \frac{4Pr}{3} \frac{dT_0}{dx} - (T_0 - 1) \\
\end{align*}
\]

(13)

In the fully burnt state \(x \to \infty\), the boundary conditions \(dq_0/dx = 0\) give \(T_0 = u_0 = 1/\rho_0 = 1 + Q = T_{ad}\), where \(T_{ad}\) denotes the adiabatic flame temperature, and \(F_0 = -Q\), \(F_0 = Y_0 = X_0 = Z_0 = 0\). Hence note that \(X_0 = 0\) in the region \(x > 0\) by equation (12).

Since, apart from for the pressure, the steady, flame solution is independent of the Prandtl number, and \(Pr\) is known to have only a very weak effect on flame stability [6, 12], throughout this paper we set \(Pr = 0.75\). Here will consider the case \(n = 2\) in order to retain analytical simplicity of the steady flame structure (the equation for \(Y_0\) is then linear and decoupled from the \(T_0\) and \(F_0\) equations), and such that it has precisely the same structure as in the CDM considered previously [14], with which we seek to compare. Note that many of the exact thermic completion reaction rates given in the example of methane-oxidation in [14] are temperature insensitive, while others have positive or negative values of \(n\). As a lumped or pooled reaction step, it is unclear what value of \(n\) in the two-step model would give a best fit to a given real flame speed and structure. While general values of \(n \neq 2\) could also be considered, the steady flame structure would then need to be determined numerically. Furthermore, any weak temperature dependence in the completion reaction rate will not qualitatively effect the leading order high-activation energy results (which are themselves approximate) studied here. Hence it is hence worth retaining the mathematical simplicity of the flame structure equations (by considering \(n = 2\)) which was one of the original motivations for the two-step model [14, 15].

The spatial origin is chosen to occur at the point where the temperature equals the crossover temperature, \(T_0 = T_c\). The jump conditions (8)-(10) are thus to be applied at \(x = 0\), and for the steady flame give

\[
[T_{0,x}] = [T_0] = [Y_0] = [F_0] = F_0 = [X_0 + Z_0] = 0, \quad T_0 = T_c.
\]
In order to satisfy these and both sets of boundary conditions as $x \to -\infty$ and $x \to \infty$ simultaneously, $\Lambda$, which is related to the flame speed, must have a specific value. Hence $\Lambda$ is an eigenvalue of equations (11)-(12) which needs to be determined, for other parameters fixed.

The solution which satisfies the boundary conditions at $x \to \pm \infty$ and the jump conditions at $x = 0$ is

$$F_0 = \begin{cases} 1 - \exp(\text{Le}_F x) & x \leq 0 \\ 0 & x > 0 \end{cases},$$

$$Y_0 = \frac{\text{Le}_Y}{h_1 - h_2} \begin{cases} \exp(h_1 x) & x \leq 0 \\ \exp(h_2 x) & x > 0 \end{cases},$$

$$T_0 = \begin{cases} 1 - \frac{Q\Lambda \text{Le}_Y}{(h_1 - 1)(h_2 - 1)} \exp(x) & x \leq 0 \\ \frac{Q\Lambda \text{Le}_Y}{h_1(h_1 - 1)(h_1 - h_2)} \exp(h_1 x) & x > 0 \\ 1 + Q - \frac{Q\Lambda \text{Le}_Y}{h_2(h_2 - 1)(h_1 - h_2)} \exp(h_2 x) & x > 0 \end{cases},$$

cf. [14], where $h_1$ and $h_2$ are the positive and negative roots, respectively, of $h^2 - \text{Le}_Y h - \text{Le}_Y \Lambda = 0$. The eigenvalue $\Lambda$ (and hence the flame speed) is then determined by setting $T_0 = T_c$ at $x = 0$, giving

$$Q\Lambda \text{Le}_Y = (1 + Q - T_c) h_2(h_2 - 1)(h_1 - h_2),$$

which can be straightforwardly solved by Newton-Raphson iteration for $\Lambda$. Note that for $\text{Le}_Y = 1$ equation (17) has the analytical solution

$$\Lambda = \frac{Q^2}{4(1 + Q - T_c)^2} - \frac{1}{4},$$

which can be used as an initial guess for non-unity values of $\text{Le}_Y$. Note that in any case $\Lambda$ depends on $Q$ and $T_c$ only through the combination

$$Q/(1 + Q - T_c) = Q_D/(Q_D - 1) = (T_{\text{ad}} - 1)/(T_{\text{ad}} - T_c)$$

and does not depend on $\text{Le}_F$ [14].

Figure 1 shows some example spatial profiles of the steady, planar flame for $Q = 5$, $\text{Le}_F = \text{Le}_Y = 1$ and various values of the crossover temperature. Note, the peak in the intermediates occurs at the reaction.
sheet. The value of \( Y_0 \) at the peak is connected to the crossover temperature, \( T_c \), since \( T_{ad} - T_c \) determines the fraction of the heat still to be released after the fuel has been consumed in the chain-branching reaction sheet, while similarly the peak value of \( Y_0 \) represents the amount of intermediates still available for conversion into products (and hence heat). Thus the lower \( T_c \) is below the adiabatic temperature, the higher is the peak in the intermediates at \( x = 0 \), and hence the further the intermediates are able to diffuse from the reaction sheet. This results in the heat release and hence the temperature rise being spread over a larger distance for lower \( T_c \). On the other hand, as \( T_c \to T_{ad} \) (or equivalently as \( Q_D \to 1 \)), the peak in the radicals becomes very small and thus the fuel is largely converted directly to products. Indeed, it can be shown that in the limit \( Q_D \to 1 \), the planar flame structure tends to that of the one-step model (represented as a dotted line in figure 1a) in which intermediates have no role [14].

### 4 Linearized equations

We now suppose that the steady, planar flame is slightly perturbed such that the perturbed flame position with respect to that of the underlying steady, planar flame is

\[
x = X(y, t), \quad |X(y, t)| \ll 1.
\]

Since the jump conditions (8)-(10) are to be applied on a surface whose position is perturbed, we transform to a frame moving with this perturbed reaction sheet, cf. [9,19], via

\[
\begin{align*}
x_p &= x - X(y, t), \quad y_p = y, \quad t_p = t, \quad u_p = u - V(y, t),
\end{align*}
\]

where the ‘\( p \)’ subscript denotes the perturbed flame co-ordinate system, and

\[
V(y, t) = \frac{\partial X}{\partial t}
\]

is the speed of the perturbed reaction sheet in the \( x \)-direction. Hence in this new co-ordinate system, the perturbed reaction sheet remains stationary at \( x_p = 0 \). Note that, under this transformation, \( \partial \cdot / \partial x, \partial \cdot / \partial y, \partial \cdot / \partial t \) are replaced in the governing equations (1)-(6) by

\[
\begin{align*}
\frac{\partial X}{\partial x_p} \frac{\partial}{\partial y_p}, \quad \frac{\partial X}{\partial y_p} \frac{\partial}{\partial x_p}, \quad \frac{\partial X}{\partial t_p} \frac{\partial}{\partial x_p}, \quad \frac{\partial X}{\partial t_p} \frac{\partial}{\partial x_p},
\end{align*}
\]

respectively, so that \( \nabla^2 \) becomes

\[
\left( 1 + \left( \frac{\partial X}{\partial y_p} \right)^2 \right) \frac{\partial^2}{\partial x_p^2} + \frac{\partial^2}{\partial y_p^2} - 2 \frac{\partial X}{\partial y_p} \frac{\partial^2}{\partial x_p \partial y_p} - \frac{\partial^2 X}{\partial y_p^2} \frac{\partial}{\partial x_p}.
\]

Dold [14] chose to Taylor expand the jump conditions around the \( x = 0 \) instead of transforming to the frame of the perturbed flame, the two approaches being equivalent. Henceforth, for convenience we drop the ‘\( p \)’ subscript notation.

A normal modes form of the perturbation is then assumed, such that

\[
X(y, t) = e^{\sigma t} e^{iky}, \quad \epsilon \ll 1,
\]

and the perturbed variables are hence of the form

\[
q(x, y, t) = q_0(x) + \epsilon q_1(x) e^{\sigma t} e^{iky},
\]
where \( q \) represents any of the dependent variables. Here \( \sigma \) is the (complex) growth rate and \( k \) is the wavenumber of the disturbance in the \( y \)-direction. For convenience, we define the following quantities in terms of the first derivatives of the perturbations with respect to \( x \):

\[
\tau_1 = \frac{dT_1}{dx}, \quad U_1 = \frac{du_1}{dx}, \quad V_1 = \frac{dv_1}{dx}, \quad X_1 = F_1 - \frac{1}{Le_F} \frac{dF_1}{dx}, \quad Z_1 = Y_1 - \frac{1}{Le_Y} \frac{dY_1}{dx}.
\]  

(23)

Substituting equations (21)-(23) into the governing equations and linearizing in \( \epsilon \), gives

\[
\frac{du}{dx} = Au + s,
\]

(24)

where \( u = (T_1, u_1, v_1, P_1, F_1, Y_1, U_1, V_1, X_1, Z_1)^T \) and

\[
A = \begin{pmatrix}
(T_0' - \sigma) & -T_0' & ik & 0 & 0 \\
T_0 & T_0 & 0 & 0 & 0 \\
0 & 0 & 0 & 0 & 0 \\
A_{41} & A_{42} & \frac{4ikPr(\sigma + T_0 - T_0')}{3T_0} & 0 & 0 \\
0 & 0 & 0 & 0 & Le_F \\
0 & 0 & 0 & 0 & 0 \\
A_{71} & -\sigma T_0' + T_0 T_0'' & ik(\sigma + T_0 - T_0') & 0 & 0 \\
0 & 0 & \frac{3\sigma + 4Prk^2T_0}{3PrT_0} & ik & 0 \\
0 & 0 & 0 & \frac{-Le_F\sigma + T_0k^2}{T_0Le_F} & 0 \\
F_0' & -F_0' & 0 & 0 & 0 \\
T_0' & T_0' & 0 & 0 & 0 \\
Y_0' & -Y_0' & 0 & 0 & 0 \\
T_0 & T_0 & 0 & 0 & 0
\end{pmatrix}
\]

\[
\begin{pmatrix}
0 & 1 & 0 & 0 & 0 \\
0 & 1 & 0 & 0 & 0 \\
0 & 0 & 1 & 0 & 0 \\
-4PrQA & 4Pr(\sigma + T_0) - 3T_0 & -ikPr & 0 & 0 \\
3 & 3T_0 & 0 & 0 & -Le_F \\
Le_Y & 0 & 0 & 0 & -Le_Y \\
-QA & \frac{\sigma + T_0}{T_0} & -ik & 0 & 0 \\
0 & \frac{-ik}{3} & \frac{1}{Pr} & 0 & 0 \\
-\frac{Le_Y\sigma - k^2T_0 - Le_Y\Delta T_0}{Le_Y T_0} & 0 & 0 & 0 & 0
\end{pmatrix}
\]

where

\[
A_{41} = \frac{4Pr(-\sigma^2 + T_0'\sigma + k^2T_0^2 - T_0 T_0'')}{3T_0^2} + \frac{T_0'}{T_0}.
\]
\[ A_{42} = \frac{-4(4 Pr T_0' + 3 T_0') \sigma}{3 T_0'^2} + \frac{Pr(4 T_0'' - 3 T_0 k^2)}{3 T_0} - \frac{T_0'}{T_0}, \]

\[ A_{71} = \frac{(-\sigma^2 + T_0' \sigma + k^2 T_0^2 - T_0 T_0'' )}{T_0^2}, \]

and

\[ s = \left( 0, 0, 0, -\frac{3\sigma^2 + 3 Pr T_0 k^2 \sigma + Pr T_0 k^2 T_0'}{3 T_0}, 0, 0, -k^2 T_0', \frac{i k}{Pr} (-P r T_0'' + T_0') \frac{k^2 F_0'}{Le_F}, \frac{k^2 Y_0'}{Le_Y} \right)^T, \]

where henceforth a prime denotes differentiation with respect to \( x \). Note that, since \( \rho = 1/T \), density has been eliminated from equations (1)-(6). Equation (1) then contains only first \( x \)-derivatives of the remaining variables, hence its linearized form has been used to eliminate \( \tau_1 \) in terms of the other perturbed quantities:

\[ \tau_1 = \frac{(T_0' - \sigma) T_1}{T_0} - \frac{T_0' u_1}{T_0} + ik v_1 + U_1, \quad (25) \]

Equation (24) is subject to boundedness conditions as \( x \to \pm \infty \). We hence now seek asymptotic solutions to equation (24) valid near the fresh or burnt states, as \( x \to \pm \infty \), which can then be used as initial conditions for numerical integration of the equation. In order to obtain higher order terms in these asymptotic solutions, which are necessary for implementation of straightforward shooting methods [12,19], it is beneficial to use one of the steady state solution variables as the independent variable [12,19]. Given the structure of the steady solution in equations (14)-(16), and given that \( Y_0 \to 0 \) as \( x \to \pm \infty \), here we choose \( Y_0 \) as the independent variable for both the fresh and burnt state analyses.

### 4.1 Asymptotic solution as \( x \to -\infty \)

Transforming to \( Y_0 \) as the independent variable by employing equation (15), then in the region \( x < 0 \) equation (24) becomes

\[ h_1 Y_0 \frac{du}{dY_0} = Au + s, \quad (26) \]

and the other steady variables can be written in terms of \( Y_0 \) as

\[ T_0 = 1 + \alpha_1 Y_0 + \beta_1 Y_0^{1/h_1}, \quad F_0 = 1 + \gamma_1 Y_0^{Le_F/h_1}, \]

where

\[ \alpha_1 = -\frac{QA}{h_1(h_1 - 1)}, \quad \beta_1 = -\frac{QA Le_Y}{(h_1 - 1)(h_2 - 1)} \left( \frac{h_1 - h_2}{Le_Y} \right)^{1/h_1}, \quad \gamma_1 = -\left( \frac{h_1 - h_2}{Le_Y} \right)^{Le_F/h_1}. \]

As \( Y_0 \to 0 \), corresponding to \( x \to -\infty \), we can thus expand equation (26) in the form

\[ h_1 Y_0 \frac{du}{dY_0} = (A_0 + A_1 Y_0 + A_2 Y_0^{1/h_1} + A_3 Y_0^{Le_F/h_1} + \cdots)u + s_0 + s_1 Y_0 + s_2 Y_0^{1/h_1} + s_3 Y_0^{Le_F/h_1} + \cdots, \quad (27) \]
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where the co-efficient matrices, $A_0$, etc, depend only on the parameters $Q$, $P_r$, $Le_Y$, $Le_F$ and $T_c$, and on

$$\sigma$$ and $k$. The homogeneous part of equation (27) has 10 independent asymptotic solutions of the form

$$u = \tilde{u}_i = Y_0^{\lambda_i} \left( a_0^i + a_1^i Y_0^{1/h_i} + a_2^i Y_0^{Le_F/h_i} + \cdots \right), \quad i = 1, \ldots, 10,$$

(28)

where $h_1^i \lambda_i$ and $a_j^i$ are the eigenvalues and eigenvectors, respectively, of $A_0$, and the $a_1^i$, $a_2^i$, etc., are found by substituting (28) into equation (27) and equating powers of $Y_0$. The eigenvalues of $A_0$ are

$$\pm k, \quad \frac{1 \pm [1 + 4(\sigma + k^2)]^{\frac{1}{2}}}{2}, \quad \frac{1 \pm [1 + 4P_r(\sigma + Prk^2)]^{\frac{1}{2}}}{2P_r},$$

$$\frac{Le_F \pm [Le_F^2 + 4(\sigma Le_F + k^2)]^{\frac{1}{2}}}{2}, \quad \frac{Le_Y \pm [Le_Y^2 + 4(\sigma Le_Y + k^2 + Le_Y \Lambda)]^{\frac{1}{2}}}{2}.$$

For $Re(\sigma) > 0$, the solutions corresponding to the eigenvalues with a negative sign are unbounded as $Y_0 \to 0 (x \to -\infty)$, and hence must be discarded. We are left with 5 linearly independent, bounded, asymptotic solutions, denoted by $i = 1, \ldots, 5$, say.

Equation (26) has an exact particular integral, $u^p_L$, valid in the entire region to the left of $x = 0$, of the form

$$u^p_L = u_0^p + u_1^p Y_0 + u_2^p Y_0^{1/h_i} + u_3^p Y_0^{Le_F/h_i}.$$

We now therefore have the asymptotic general solution of equation (26), denoted by $\tilde{u}$, valid as $Y_0 \to 0$ or $x \to -\infty$, of the form

$$\tilde{u} = \sum_{i=1}^{5} a_i \tilde{u}_i + u^p_L,$$

(29)

where $a_i$ are (complex) constants of integration, to be determined.

4.2 Asymptotic solution as $x \to \infty$

In the region $x > 0$ equation (24) can be rewritten using $Y_0$ as the independent variable as

$$h_2 Y_0 \frac{du}{dY_0} = Au + s,$$

(30)

and in this region, in terms of $Y_0$, the other steady variables are, from equations (14)-(16),

$$T_0 = 1 + Q + \alpha_2 Y_0, \quad F_0 = 0,$$

where

$$\alpha_2 = -\frac{Q \Lambda}{h_2(h_2 - 1)},$$

As $x \to \infty$, corresponding to $Y_0 \to 0$, we can therefore expand equation (30) in the form

$$h_2 Y_0 \frac{du}{dY_0} = (A_0^* + A_1^* Y_0 + \cdots)u + s_0^* + s_1^* Y_0 + \cdots$$

(31)
where again the co-efficients matrices, $A_0^0$, etc, depend only on the parameters and on $\sigma$ and $k$. Noting that in the region $x > 0$, we must have $F_1 = X_1 \equiv 0$, the homogeneous part of equation (31) has 8 independent asymptotic solutions of the form

$$\mathbf{u} = \hat{\mathbf{u}}_i = Y_0^{\lambda_i} (b_0^i + b_1^i Y_0 + \cdots), \quad i = 1, \ldots, 8,$$  

(32)

where $h_2^\lambda_i$ and $b_i^0$ are the eigenvalues and eigenvectors, respectively, of $A_0^0$, and the co-efficients $b_1^i$, etc., are found by substituting (32) into equation (31) and equating powers of $Y_0$. The eigenvalues of $A_0^0$ are

$\pm k, \quad \frac{1 \pm [1 + 4(\sigma \phi + k^2)]^{\frac{1}{2}}}{2}, \quad \frac{1 \pm [1 + 4 \text{Pr}(\sigma \phi + \text{Pr} k^2)]^{\frac{1}{2}}}{2 \text{Pr}}, \quad \frac{\text{Le} Y \pm [\text{Le} k^2 + 4(\sigma \phi \text{Le} Y + k^2 + \text{Le} \lambda)]^{\frac{1}{2}}}{2}$

where $\phi = 1/(1 + Q) = 1/T_{ad}$. For $Re(\sigma) > 0$, the solutions corresponding to eigenvalues with a positive sign are unbounded as $Y_0 \to 0$ ($x \to \infty$), and again must be discarded. We are thus left with 4 linearly independent, bounded asymptotic solutions.

Equation (30) has an exact particular integral valid in the region the right of $x = 0$, $\mathbf{u}_R^p$, of the form

$$\mathbf{u}_R^p = b_0^p + b_1^p Y_0.$$

We thus now have the asymptotic general solution, denoted by $\hat{\mathbf{u}}$, valid as $Y_0 \to 0$ or $x \to \infty$, of the form

$$\hat{\mathbf{u}} = \sum_{i=1}^{4} b_i \hat{\mathbf{u}}_i + \mathbf{u}_R^p,$$  

(33)

where the $b_i$ are (complex) constants of integration, to be determined.

### 4.3 Jump conditions at $x = 0$

Upon linearization in $\epsilon$, the jump conditions (8)-(10) to be applied at the reaction sheet ($x = 0$), become

$$[T_1] = [u_1] = [v_1] = [P_1] = [F_1] = F_1 = [Y_1] = 0, \quad (34)$$

$$[U_1] = [V_1] = [X_1 + Z_1] = 0, \quad (35)$$

$$T_1 = 0. \quad (36)$$

### 4.4 Numerical solution of linearized equations

In summary, the linearized problem consists of equation (24) subject to $\mathbf{u} \to \mathbf{u}$ as $x \to -\infty$ and $\mathbf{u} \to \hat{\mathbf{u}}$ as $x \to \infty$, where $\mathbf{u}$ and $\hat{\mathbf{u}}$ are given by equations (29) and (33), respectively, and to the jump conditions (34)-(36) at $x = 0$.

The numerical solution of the problem is achieved as follows. Firstly, each of the five asymptotic linearly independent solutions, $\hat{\mathbf{u}}_i, \ i = 1, \ldots, 5$ which are valid as $x \to -\infty$, are used as initial conditions in the numerical integration of the homogeneous form of equation (24), starting from a sufficiently large negative value of $x$ (corresponding to a small value of $Y_0$), up to $x = 0$. A fourth-order Runge-Kutta routine with adaptive step size was used for the numerical integrations. These integrations thus give numerically the five bounded independent solutions of equation (24), denoted by $\mathbf{u}_L^i$, which are now valid throughout the
region left of $x = 0$. Hence the general solution in this region, $u_L$, is of the form

$$u_L = \sum_{i=1}^{5} a_i u_i^L + u_p^L.$$  

In particular, we can now evaluate numerically the general solution at $x = 0$ reached from the left, i.e. $u_L(x = 0)$.

Similarly, the four asymptotic linearly independent solutions, $\hat{u}_i$, $i = 1, \ldots, 4$, can be used as initial conditions for numerical integration of the homogeneous part of equation (24) back to $x = 0$ starting from a sufficiently large value of $x$. This gives numerically the four bounded linearly independent solutions in the entire region to the right of $x = 0$, denoted by $u_{iR}$, $i = 1, \ldots, 4$. The general solution for $x > 0$ is hence of the form

$$u = u_R = \sum_{i=1}^{4} b_i u_{iR} + u_p^R,$$

from which $u_R(x = 0)$ can be evaluated.

It remains to determine the (complex) constants of integration, $a_i$ and $b_i$. These are chosen so as to satisfy the jump conditions (34)-(35). Given the form of these jump conditions, we define a new reduced set of dependent variables by $q = (T_1, u_1, v_1, P_1, F_1, Y_1, U_1, V_1, X_1 + Z_1)^T$, which is hence straightforwardly constructed by adding the ninth and tenth entries of $u$. The jump conditions (34) and (35) can thus be written in the form

$$q_L(x = 0) = q_R(x = 0),$$

where $q_L$ and $q_R$ are just the reduced versions of $u_L$ and $u_R$, respectively. This is a system of 9 simultaneous linear equations for the 9 constants of integration, $a_1$ to $a_5$ and $b_1$ to $b_4$. These can hence be straightforwardly determined for given parameters such that the conditions (37) are satisfied.

However, determining the constants of integration in this manner also fixes the value of $T_1$ at $x = 0$. Only for certain discrete values of $\sigma$ (for fixed $k$) will the condition (36) be simultaneously satisfied. The eigenvalues are hence found by Newton-Raphson iteration on the condition $T_1 = 0$. Note that, for fixed parameters, equation (24) and the boundary and jump condition do not have any leading order dependence on the activation energy of the chain-branching step, unlike for the one-step model when $Le_F$ is fixed [9].

5 Linear stability results

As for the one-step Reactive Navier-Stokes model [9] and for the two-step CDM [14], we find there are only two possible distinct modes. The first of these is a cellular mode for which $\sigma$ is real, and the second a pulsating mode corresponding to a complex conjugate pair of eigenvalues. In this section, the linear neutral stability boundaries and linear dispersion relations of the two-step branching flame are examined for both these cellular and pulsating branches, when thermal expansion is taken into account. The effect of the various model parameters is explored, and the results compared and contrasted with both the two-step CDM predictions and the Reactive Navier-Stokes one-step chemistry results.

5.1 Effect of thermal expansion and comparison with CDM

We first examine the effect of varying thermal expansion (through the heat of reaction $Q$) on the stability of the flame. For the two-step CDM, in which hydrodynamical effects induced by thermal expansion are completely ignored, the linear stability is independent of $Q$ for fixed values of $Q_D = Q/(T_c - 1)$ [14]. Since the steady, planar flame speed also depends on $Q$ and $T_c$ only through the combination $Q_D$ (see §3), it is
Figure 2. Neutral stability boundary of the cellular branch in the $Le_F - k$ plane for $Le_Y = 1$ and $Q_D = 1.25$ ($T_c = 1 + Q/1.25$), when $Q = 5$ ($T_c = 5$) (solid line), $Q = 1$ ($T_c = 1.8$) (dashed line) and $Q = 0.1$ ($T_c = 1.08$) (dot-dashed line). Dotted line is the CDM prediction for $Q_D = 1.25$.

consistent to first examine the stability effect of varying $Q$ while keeping $Q_D$ fixed (and hence varying $T_c$ appropriately).

5.1.1 Cellular branch. Figure 2 shows the neutral stability boundaries in the $Le_F - k$ plane for various values of $Q$ when $Q_D = 5/4$ (hence $T_c = 1 + 4Q/5$) and $Le_Y = 1$. These neutral stability boundaries correspond to the loci on which $\sigma = 0$, and the flame is unstable to wavenumbers below the curve and stable to those above it. The dotted curve in figure 2 is the neutral stability boundary for $Q_D = 5/4$ as predicted by the two-step CDM. The CDM results predicts the flame is stable to all wavenumbers above a critical fuel Lewis number, while as $Le_F$ decreases below this critical value the flame becomes unstable to an increasingly wide band of wavenumbers [14]. Since the CDM ignores hydrodynamic effects, the instability predicted by the CDM is of a purely thermal-diffusive nature. However, the two-step Reactive Navier-Stokes model studied here simultaneously captures hydrodynamic effects, and figure 2 shows that the stability results are then qualitatively different to the predictions of the CDM, even when $Q$ is quite small. In particular the flame is unstable for all values of $Le_F$, at least for sufficiently long wavelength perturbations. While the CDM is valid in the asymptotic limit $Q \to 0$, figure 2 also shows that even for a physically very small heat release of $Q = 0.1$, this hydrodynamical effect is still important. Hence, while the $Q = 0.1$ stability boundary follows closely that of the CDM for fuel Lewis numbers sufficiently below the critical $Le_F$ predicted by the CDM, it then curves sharply to the right near this critical value. Hence the concept of an $Le_F$ stability boundary predicted by the CDM is not physically valid for gaseous flames.

A second difference is that, while the CDM results are independent of $Q$ for fixed $Q_D$, figure 2 shows that when hydrodynamic effects are taken into account, the stability does depend on $Q$ (with $Q_D$ fixed). Indeed, the higher the value of $Q$, the wider the range of unstable wavenumbers for any fixed $Le_F$, so that the neutral stability curve for a given $Q$ lies above those corresponding to lower values in the $Le_F - k$ plane.

For the cellular instability, a larger value of the neutrally stable wavenumber also corresponds to a more unstable flame, in that the maximum linear growth rate is also larger. This is demonstrated in figure 3, which shows the dispersion relations (growth rate, $\sigma$, as a function of wavenumber, $k$) for various values of $Q$ when $Le_F = 0.6$ and when $Le_F = 1.2$. Note, for the cellular mode, $\sigma$ is also zero at $k = 0$. As $Q$ increases, not only does the band of unstable wavenumbers increase for fixed $Le_F$, but so does the wavenumber with the maximum value of the growth rate, and the value of this maximum also rapidly increases. Thus weak perturbations to the planar flame will grow much more rapidly and become non-linear more quickly for larger $Q$. One would expect that cells would first appear on the flame with a wavenumber close to that with maximum linear growth rate. However, it should be noted that a linear analysis only gives information about stability boundaries, onset and initial growth stage of the instability. It is not relevant to the fully developed non-linear cells, which may be of a quite different characteristic wavelength to that predicted by the linear analysis [13].

The dependence of the cellular instability on $Q$, with fixed $Q_D$, described above is qualitatively the same as the effect of thermal expansion on the one-step model. The differences between the two-step CDM
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Figure 3. Dispersion relations of the cellular instability for $Le_Y = 1$, $Q_D = 1.25$ and (a) $Le_F = 0.6$ and (b) $Le_F = 1.2$, when $Q = 5$ ($T_c = 5$) (solid line), $Q = 1$ ($T_c = 1.8$) (dashed line) and $Q = 0.1$ ($T_c = 1.08$) (dot-dashed line). Dotted line in (a) is the CDM prediction for $Q_D = 1.25$.

Figure 4. Neutral stability boundary of the cellular branch in the $Le_F - k$ plane for $Le_Y = 1$, $T_c = 5$ and $Q = 5$ ($Q_D = 1.25$) (solid line) and $Q = 6$ ($Q_D = 1.5$) (dashed line).

and Reactive Navier-Stokes model predictions are also qualitatively the same in the one-step case, cf. Jackson and Kapila [9]. For the two-step model, however, the effect of varying $Q$ but keeping the crossover temperature, $T_c$, fixed is somewhat more complicated. Figure 4 shows the neutral stability boundaries for $T_c = 5$ and heats of reaction $Q = 5$ and 6 (corresponding to $Q_D = 1.25$ and 1.5, respectively). The stability response of the flame with the larger thermal expansion is less sensitive to the fuel Lewis number than for the smaller $Q$ case. The neutral stability curves for the two thermal expansions cross in the $Le_F - k$ plane, in this case at $Le_F = 0.97$, and the flame is more (less) unstable for the smaller degree of thermal expansion when $Le_F$ is above (below) this value. Thus, for $Le_F$ sufficiently below unity, the stabilizing effect of decreasing thermal expansion is outweighed by a destabilizing thermal-diffusive effect as $Q_D$ is lowered by the increase in $Q$ (cf. §5.2.1).

5.1.2 Pulsating branch. Recall that the pulsating instability branch corresponds to a complex conjugate pair of eigenvalues. For the CDM limit, this instability is found to occur only provided $Q_D$ is below a finite value, and then only if $Le_F$ is above a critical value which is greater than unity [14]. This critical value of the fuel Lewis number decreases and tends to unity as $Q_D \rightarrow 1$. However, $Q_D - 1$ still has to be sufficiently small for the pulsating instability to occur for values of $Le_F$ realistic to gaseous flames. For example, figure 5 shows the CDM prediction of the neutral stability boundary when $Q_D - 1 = 0.1$ (for $Le_Y = 1$), in which case $Le_F$ needs to be above 1.772 for the flame to be unstable to the pulsating mode. Note that the CDM predicts that the pulsating instability first appears at a non-zero wavenumber [14]. For example, when $Q_D = 1.1$, the wavenumber is 0.252 at the turning point of the CDM neutral stability curve in figure 5.

Figure 5 also shows the neutral stability boundaries, on which $Re(\sigma) = 0$, predicted by the Reactive Navier-Stokes model with $O(1)$ values of the thermal expansion, but with $Q_D = 1.1$ fixed. As for the CDM prediction, when hydrodynamic effects are taken into account, the pulsating instability still only occurs above a critical value of $Le_F$. Increasing thermal expansion lowers this critical value somewhat. For example, the instability occurs when $Le_F > 1.732$ for $Q = 1$ and when $Le_F > 1.683$ for $Q = 5$ in figure...
5.2 Effect of crossover temperature and comparison with one-step model

The two-step CDM results predict that as $T_c \to T_{ad}$, the flame becomes increasingly unstable in that the critical values of $Le_F$ at which the cellular and pulsating instability occurs both tend to unity [14]. In this section we examine the effect of $T_c$, for fixed $Q$, when realistic thermal expansion is taken into account.
1.2 non-dimensional

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stability boundary is only weakly dependent on number-wavenumber (\(l = k\)) when thermal expansion is taken into account. Figure 7(b) shows the neutral stability boundaries in the reduced Lewis number-wavenumber plane (\(l = k\)) for various \(T_e\) and \(Q_D\) as \(Q_D \rightarrow 1\) and \(T_e \rightarrow \text{ad} (\text{finite} \ \beta_{\text{eff}})\) when thermal expansion is taken into account. However, it remains to determine how predictive the one-step model is of the two-step results for non-zero values of \(Q_D\). Dold [14] showed that, for the constant density models, in the limit \(T_e \rightarrow \text{ad}\) there is a direct correspondence between the flame structure and stability of the one-step and two-step chemistry models. Indeed, under the asymptotic limit \(Q_D \rightarrow 1\), the two-step model can be described to leading order by a one-step model which has an effective Zel’ dovich number defined by \(\beta_{\text{eff}} = 2Q_D/(Q_D - 1)\). In other words, in the limit, the two-step CDM linear stability depends on leading order on \(Q_D\) only through a reduced Lewis number defined by \(l = \beta_{\text{eff}}(Q_D - 1)\), and this leading order dependence is then the same to that of the one-step model with Zel’dovich number \(\beta = \beta_{\text{eff}}\) [14].

Since the analysis in Dold [14] is based on comparing the two-step jump conditions in the temperature and fuel in the limit \(Q_D \rightarrow 1\) with those of the one-step model, and these jump conditions are unchanged when thermal expansion is taken into account, one would expect the analogy between the two models to still hold in the Reactive Navier-Stokes formulation. However, it remains to determine how predictive the one-step model is of the two-step results for non-zero values of \(Q_D - 1\) (finite \(\beta_{\text{eff}}\)) when thermal expansion is taken into account. Figure 7(b) shows the neutral stability boundaries in the reduced Lewis number-wavenumber plane (\(l = k\)) for various \(T_e\) and \(Q_D\) as \(Q_D \rightarrow 1\) and \(T_e \rightarrow \text{ad} (\text{finite} \ \beta_{\text{eff}})\) when thermal expansion is taken into account. However, it remains to determine how predictive the one-step model is of the two-step results for non-zero values of \(Q_D - 1\) (finite \(\beta_{\text{eff}}\)) when thermal expansion is taken into account. Figure 7(b) shows the neutral stability boundaries in the reduced Lewis number-wavenumber plane (\(l = k\)) for various \(T_e\) and \(Q_D\) as \(Q_D \rightarrow 1\) and \(T_e \rightarrow \text{ad} (\text{finite} \ \beta_{\text{eff}})\) when thermal expansion is taken into account. However, it remains to determine how predictive the one-step model is of the two-step results for non-zero values of \(Q_D - 1\) (finite \(\beta_{\text{eff}}\)).

5.2.1 **Cellular branch.** Figure 7(a) shows the neutral stability boundary when \(Q = 5\) for various \(T_e\) (and hence varying \(Q_D\) also), when \(L_e = 1\). Again, when thermal expansion is taken into account, the flame is unstable to the cellular instability for all values of \(L_e\), and there is no critical Lewis number as in the CDM prediction. For unit values of \(L_e\) and \(L_e = 1\) and realistic thermal expansions, the cellular instability is insensitive to \(T_e\). Thus the neutral stability curves for the different values of \(T_e\) cross at \(L_e = 1\) in figure 7(a). As \(T_e \rightarrow \text{ad}\) (or equivalently as \(Q_D \rightarrow 1\)) the flame becomes increasingly sensitive to thermal-diffusive effects associated with the Lewis number of the fuel. These effects are destabilizing when \(L_e < 1\), but stabilizing for \(L_e > 1\), and appear to have an enhanced role as \(Q_D \rightarrow 1\). Thus, for fixed values of \(L_e\) which are less than unity, the flame becomes increasingly unstable (the neutrally stable wavenumber becomes larger) as \(T_e\) increases. Conversely, for fixed values of \(L_e > 1\), the thermal-diffusive effect becomes more strongly stabilizing, and hence the neutrally stable wavenumber decreases.

These dependences of the cellular mode stability boundaries in the \(L_e - k\) plane on \(T_e\) are in fact qualitatively similar to the dependence of the one-step model boundaries on activation energy [12]. Indeed, in the high-activation energy asymptotic limit, the one-step flame stability depends on \(L_e\) and the activation energy only through a reduced Lewis number, \(l\), defined by \(l = \beta(\text{Le}_{L} - 1)\) [9], such that \(l = O(1)\) as \(\beta \rightarrow \infty\) (i.e. the NEF approximation), where \(\beta\) is the Zel’dovich number (the non-dimensional activation energy of the one-step reaction rate). Dold [14] showed that, for the constant density models, in the limit \(T_e \rightarrow \text{ad}\) there is a direct correspondence between the flame structure and stability of the one-step and two-step chemistry models. Indeed, under the asymptotic limit \(Q_D \rightarrow 1\), the two-step model can be described to leading order by a one-step model which has an effective Zel’dovich number defined by \(\beta_{\text{eff}} = 2Q_D/(Q_D - 1)\). In other words, in the limit, the two-step CDM linear stability depends on leading order on \(L_e\) only through a reduced Lewis number defined by \(l = \beta_{\text{eff}}(L_e - 1)\), and this leading order dependence is then the same to that of the one-step model with Zel’dovich number \(\beta = \beta_{\text{eff}}\) [14].

Since the analysis in Dold [14] is based on comparing the two-step jump conditions in the temperature and fuel in the limit \(Q_D \rightarrow 1\) with those of the one-step model, and these jump conditions are unchanged when thermal expansion is taken into account, one would expect the analogy between the two models to still hold in the Reactive Navier-Stokes formulation. However, it remains to determine how predictive the one-step model is of the two-step results for non-zero values of \(Q_D - 1\) (finite \(\beta_{\text{eff}}\)) when thermal expansion is taken into account. Figure 7(b) shows the neutral stability boundaries in the reduced Lewis number-wavenumber plane (\(l = k\)) for various \(T_e\) and \(Q_D\) as \(Q_D \rightarrow 1\) and \(T_e = 5\) (\(Q_D = 1.064, \beta_{\text{eff}} = 33.333\)). Also shown in (b) is the high-activation energy one-step model result from Jackson and Kapila [9] (dot-dashed line).

Figure 7. Neutral stability boundary of the cellular branch (a) in the \(L_e - k\) plane and (b) in the \(l - k\) plane (\(l = \beta_{\text{eff}}(L_e - 1)\), \(\beta_{\text{eff}} = 2Q_D/(Q_D - 1)\), for \(Q = 5\), \(L_e = 1\) and \(T_e = 4\) (\(Q_D = 5/3, \beta_{\text{eff}} = 5\)) (solid lines), \(T_e = 5\) (\(Q_D = 1.25, \beta_{\text{eff}} = 10\)) (dashed lines) and \(T_e = 5.7\) (\(Q_D = 1.064, \beta_{\text{eff}} = 33.333\)) (dotted lines). Also shown in (b) is the high-activation energy one-step model result from Jackson and Kapila [9] (dot-dashed line).
chain-branching step), but is a function of the the crossover temperature [14]. In particular, $\beta_{\text{eff}}$ can be arbitrarily large compared to any physical Zel’dovich number.

### 5.2.2 Pulsating branch

Figure 8(a) shows the neutral stability curves of the pulsating branch for various values of $T_c$, when $Q = 5$ and $Le_Y = 1$. Unsurprisingly, since thermal expansion has a weak effect on the pulsating instability boundary, the dependence on $T_c$ (and hence on $Q_D$) follows the trends predicted by the CDM [14], i.e. the critical value of $Le_F$ decreases towards unity with $Q_D$. The dependence of the neutral stability curves in the $l - k$ plane is more interesting. Figure 8(b) shows the neutral stability boundaries in this plane, together with the one-step boundary for the pulsating branch when $Q = 5$ from Jackson and Kapila [9]. The values of $T_c$ used in figure 8 correspond to $\beta_{\text{eff}} = 22, 25$ and $28.57$. The stability of the pulsating branch in the $l - k$ plane can be seen to be much more sensitive to $\beta_{\text{eff}}$ than is the cellular instability (cf figure 7(b)). Furthermore, while the two-step model results do tend to those of the one-step model as $\beta_{\text{eff}} \to \infty (T_c \to T_{\text{ad}})$, it is apparent that a very high $\beta_{\text{eff}}$ would be required for the one-step and two-step model results to quantitatively agree.

Interestingly, these effects of finite $\beta_{\text{eff}}$ on the two-step model pulsating instability boundary in the $l - k$ plane are actually very similar to the role of finite activation energy in the one-step model. Lasseigne et al. [10] performed a numerical linear stability analysis of the one-step reaction CDM with finite activation energy (i.e. without invoking the high activation energy asymptotic limit). They also found that, for the pulsating instability, the results are sensitive to $\beta$, with the finite activation energy results lying further to the right in the $l - k$ plane the lower the Zel’dovich number, and that a very large $\beta$ would be required for the high-activation energy asymptotic one-step theory to be quantitatively predictive.

### 5.3 Effect of Lewis number of intermediates

Figure 9 shows the stability boundaries of the cellular mode for various values of the intermediates Lewis number for $Q = 5$ when $T_c = 5$ and when $T_c = 5.7$. Increasing $Le_Y$ can be seen to have a destabilizing effect on the cellular mode, in that the neutral stability curves for higher $Le_Y$ lie entirely above those of lower values. However, figure 9 also shows that the destabilizing effect becomes weaker as $T_c$ increases, and the results becomes independent of $Le_Y$ in the limit $Q_D \to 1$. This is to be expected, since as $T_c$ tends to $T_{\text{ad}}$, the amount of intermediates, and hence the role of their diffusion in the flame structure, decreases (see figure 1). For the pulsating branch, however, increasing $Le_Y$ is found to have a stabilizing effect, in that the critical fuel Lewis number for this instability becomes larger as $Le_Y$ increases, cf. the CDM prediction [14]. The dependence of the pulsating branch critical fuel Lewis number on $Le_Y$ again becomes weaker as $T_c$ tends to $T_{\text{ad}}$.
6 Conclusions

A leading order, high-activation energy asymptotic linear stability analysis of premixed flames has been performed in the context of a two-step chain-branching chemistry, Reactive Navier-Stokes model. The main purpose was to extend the results of a previous study, which employed a constant density model [14], to take into account hydrodynamic effects induced by thermal-expansion. The main difference is that, as found previously for the standard one-step chemistry model, when thermal expansion is taken into account, the flame always has a band of perturbation wavelengths which are unstable to a cellular mode. Hence the region of stable fuel Lewis numbers predicted by the constant density study does not exist for realistic thermal expansions. On the other hand, a sufficiently large, critical fuel Lewis number is still required for the flame to be unstable to a pulsating mode, as correctly predicted by the constant density model. For this mode, thermal expansion has a weak effect on the stability boundaries, but a large effect on the maximum linear growth rate.

Comparing the results with those of a previous one-step reaction Reactive Navier-Stokes study [9], the two-step behaviour is found to be broadly similar, and hence no qualitatively new behaviour is revealed by using the more realistic two-step model. Furthermore, in the limit that the branching cross-over temperature, $T_c$, tends to the adiabatic flame temperature, $T_{ad}$, the two-step model results can be identified to leading order with those of a one-step model with a suitably defined effective activation energy. This effective activation energy tends to infinity in the limit $T_c \to T_{ad}$, and hence unlike a physical activation energy, can be arbitrarily large. The one-step model approximation is found to be quantitatively good for the cellular instability even when the effective activation energy is moderately large. For the pulsating instability, however, it is found that a crossover temperature very close to the adiabatic temperature (corresponding to a very large effective activation energy) would be required for the effective one-step results to be quantitatively predictive of the two-step model results.

Here we have considered adiabatic, freely propagating premixed flames. Secondary effects, such as buoyancy, heat loss, endothermic branching reaction, finite activation energy effects, etc., could also be included in future linear stability studies. However, calculations of the fully non-linear stages of the evolution, and studies of how these compare and contrast to the fully non-linear one-step model results, would perhaps be a more important next step. Direct numerical simulations using the two-step model, along the lines of the one-step computations in Sharpe and Falle [13], will be presented in a future article.

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